Auranofin and Related Heterometallic Gold(I)-Thiolates as Potent Inhibitors of Methicillin-Resistant Staphylococcus aureus Bacterial Strains

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Supplementary Material

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1. Crystallographic Data for Compounds 3a and 3d

Single crystals of **3a** and **3d** (see details below) were mounted on a glass fiber in a random orientation. Data collection was performed at room temperature on a Kappa CCD diffractometer using graphite monochromated Mo-Ka radiation (I=0.71073 Å). Space group assignments were based on systematic absences, E statistics and successful refinement of the structures. The structures were solved by direct methods with the aid of successive difference Fourier maps and were refined using the SHELXTL 6.1 software package. All non-hydrogen atoms were refined anisotropically. Hydrogen atoms were assigned to ideal positions and refined using a riding model. Details of the crystallographic data are given in Table S1 (below). These data can be obtained free of charge from The Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/data_request/cif. (CCDC 977988 for compound **3a**, and 977989 for compound **3d**). **3a**: Crystals of **3a** (orange prisms with approximate dimensions 0.25 x 0.23 x 0.23 mm) were obtained from a solution of **3a** in CH₂Cl₂ by slow diffusion of Et₂O at RT. **3d**: Crystals of **3d** (orange prisms with approximate dimensions 0.25 x 0.24 x 0.22 mm) were obtained from a solution of Et₂O at RT.



Figure S1. Crystal structure of 3a.

Compound	3a	3d		
	[{AuSPh}(μ-DPPF)]	[{AuSthiazoline}₂(µ-DPPF)]		
formula	$C_{46}H_{38}Au_2FeP_2S_2$	$C_{40}H_{36}Au_2FeN_2P_2S_4$		
fw	1161.61	1184.67		
Т [К]	293 (2)	293(2)		
λ (Μο_{Κα})[Å]	0.71073	0.71073		
crystal system	Triclinic	Monoclinic		
space group	P-1	P2(1)/c		
<i>a</i> [Å]	8.6320(17)	10.8500(7)		
b [Å]	10.140(2)	9.5950(3)		
<i>c</i> [Å]	13.014(3)	19.8390(13)		
α [°]	101.87(3)	90		
β [°]	106.29(3)	105.962(2)		
γ [^o]	105.48(3)	90		
V [Å] ³	1004.1(3)	1985.72(19)		
Z	1	2		
D _{calcd} (g cm ⁻³)	1.929	1.981		
μ (mm⁻¹)	7.860	8.053		
GOF	1.134	1.049		
$R_1[l > 2\sigma]$	0.0328	0.0582		
wR ₂ (all data)	0.085	0.1861		

Table S1. Crystal Data and Structure Refinement for Complexes 3a and 3d.

3a		3d		
Au(1)-P(1)	2.2546(14)	Au(1)-P(1)	2.259(3)	
Au(1)- S(1)	2.3000(16)	Au(1)- S(1)	2.307(3)	
Fe-C	2.032(4)-2.063(5)	Fe-C	2.025(15) -2.050(15)	
S-C(41)	1.783(6)	S-C(1)	1.761(14)	
P(1)-C(14)	1.781(5)	P(1)-C(12)	1.788(12)	
P(1)-C(31)	1.826(5)	P(1)-C(31)	1.815(13)	
P(1)-C(21)	1.827(5)	P(1)-C(21)	1.810(13)	
P(1)-Au(1)-S(1)	171.33(5)	C(1)-N(1)	1.273(18)	
C(14)-Fe(1)-C(14A)	180.0(3)	C(1)-S(2)	1.759(16)	
C(41)-S(1)-Au(1)	100.93(19)	N(1)-C(3)	1.44(2)	
		S(2)-C(2)	1.82(2)	
		C(2)-C(3)	1.41(3)	
		P(1)-Au(1)-S(1)	177.75(11)	
		C(14)-Fe(1)-(C14A)	179.999(2)	
		C(1)-S(1)-Au(1)	104.0(5)	

Table S2. Selected bond lengths [Å] and angles [°] for complexes 3a and 3d.

2. DFT Studies for compounds 4a and 4d

The calculations have been performed using the hybrid density functional method B3LYP, as implemented in Gaussian09. Geometries were optimized with the 6-31G(d,p) basis set for the C, N, P, S, and H elements and the SDD pseudopotential for the iron and gold centers. Frequency calculations have been done at the same level of theory as the geometry optimizations to confirm the nature of the stationary points.





Figure S2. Optimized structure of 4a.





Figure S3. Optimized structure of 4d.

Table S3. Selected structural parameters of complexes **4a** and **4d** obtained from DFT calculations and the comparison with experimental values of **3a** and **3d**. Bond lengths in Å and angles in °.

3a		4a	3d		4d
Au(1)-P(1)	2.2546(14)	2.335	Au(1)-P(1)	2.259(3)	2.333
Au(1)- S(1)	2.3000(16)	2.345	Au(1)- S(1)	2.307(3)	2.353
Fe-C	2.032(4)-	2.070-2.083	Fe-C	2.025(15)-	2.070-2.084
	2.063(5)	1.796		2.050(15)	1.765
S-C(41)	1.783(6)	178.201	S-C(1)	1.761(14)	177.842
P(1)-Au(1)-	171.33(5)	156.270	P(1)-Au(1)-	177.75(11)	166.650
S(1)	180.0(3)		S(1)	179.999(2)	
C(14)-Fe(1)-			C(14)-Fe(1)-		
C14(1)#			C14(1)#		
C(41)-S(1)-	100.93(19)	103.791	C(1)-S(1)-	104.0(5)	103.953
Au(1)			Au(1)		

Optimized geometries and cartesian coordinates of stationary points.

4a				Н	0.408445	-0.208660	2.601860
Charge: 0			С	1.398505	1.136051	3.964722	
Multiplicity: 1		Н	1.228247	0.501263	4.829310		
manapa	onyr i			С	2.057965	2.357866	4.108959
Fe	3 174301	-1 294432	-0 625860	Н	2.407172	2.675770	5.087051
Ĉ	2 439271	-3 219095	-0.335296	С	2.255076	3.178325	2.995756
н	1 633419	-3 658183	-0.907181	Н	2.755623	4.135950	3.106071
C	2 293533	-2 457363	0.861465	С	1.801676	2.776005	1.739081
н	1 354704	-2 224275	1 344373	Н	1.943681	3.429505	0.883775
C	3 590489	-2 026350	1 275263	С	0.025224	2.483243	-0.938494
н	3 810330	-1 407094	2 134249	С	0.887078	3.216346	-1.767461
C	3 828446	-3 263876	-0 661127	Н	1.907507	2.880813	-1.923990
Ċ	4 540117	-2 527824	0 334242	С	0.432434	4.372546	-2.404976
ц	5 607556	-2.327024	0.352775	Н	1.105611	4.931366	-3.048724
	-1 108365	-2.00000	0.021766	С	-0.881581	4.805286	-2.220765
P	0 567429	0.004042	-0.063069	Н	-1.233743	5.702586	-2.721267
s S	-2 7/5009	-2 3/2651	0.003003	С	-1.746489	4.076522	-1.401062
0	2.743003	-2.342031	-0.645406	Н	-2.773693	4.401277	-1.264685
С Ц	3 77/516	1 3/3353	-0.043400	С	-1.298856	2.918222	-0.766760
C	J 254501	0.165516	-1 644031	Н	-1.981772	2.342809	-0.147571
Ц	4.234331 5.224970	0.100010	1 692206	С	-4.292545	-1.430942	0.110507
	2 /10020	0.200210	-1.003200	С	-4.558042	-0.427374	1.057982
Ц	2 729///	1 1 2 9 4 5 0	2 401005	Н	-3.791808	-0.165700	1.781571
	2 064407	-1.120409	-3.401095	С	-5.793959	0.219503	1.080660
C	2.004407	0.420409	-0.934536	Н	-5.978199	0.995288	1.819781
	2.002070	-0.391269	-2.119420	С	-6.794342	-0.131735	0.171078
	1.100421	-0.027441	-2.379110	Н	-7.757731	0.369634	0.193634
	1.149288	1.542040	1.303490	С	-6.543311	-1.137326	-0.764145
U	0.941660	0.731372	2.709153				

Н	-7.312689 -1.422106 -1.477278	С	0.974123	1.390648	1.659186
С	-5.304038 -1.778693 -0.798896	С	0.633081	0.498460	2.686668
Н	-5.110214 -2.552299 -1.535550	Н	0.103063	-0.418917	2.448015
Н	4.263233 -3.744992 -1.526861	С	0.953987	0.794625	4.012679
		Н	0.679472	0.099081	4.800208
4d		С	1.610700	1.986238	4.324500
		Н	1.854439	2.219146	5.356963
Charge	e: 0	С	1.940320	2.886486	3.308351
Multipl	icity: 1	Н	2.438662	3.821196	3.548491
		С	1.621368	2.593808	1.982023
Fe	3.263397 -1.233927 -0.476218	Н	1.863203	3.308622	1.201303
С	2.658664 -3.223362 -0.383392	С	0.112607	2.546472	-0.874612
Н	2.016464 -3.700454 -1.111017	С	1.043404	3.335045	-1.566473
С	2.224813 -2.528529 0.783562	Н	2.071676	3.001868	-1.666609
Н	1.195081 -2.391410 1.083779	С	0.648168	4.545670	-2.139504
С	3.379024 -2.011992 1.447800	Н	1.374343	5.148279	-2.677406
Н	3.377961 -1.412816 2.347832	С	-0.674084	4.977199	-2.025244
С	4.082786 -3.141529 -0.440656	Н	-0.979692	5.917882	-2.474064
С	4.527995 -2.394595 0.691831	С	-1.606502	4.192629	-1.342281
Н	5.553521 -2.134780 0.917535	Н	-2.639385	4.518159	-1.261510
Au	-1.098268 -0.659184 -0.297698	С	-1.219955	2.979417	-0.773790
Р	0.574126 0.953184 -0.082508	Н	-1.956174	2.362766	-0.264421
S	-2.725897 -2.340927 -0.550638	С	-4.239110	-1.535833	-0.127936
С	3.481102 0.826773 -0.418990	Н	4.712262	-3.547657	-1.221004
Н	3.734402 1.388811 0.468706	С	-6.029721	0.176414	0.525662
С	4.398998 0.274849 -1.357615	Н	-6.196109	0.866350	-0.306830
Н	5.477340 0.338034 -1.298388	Н	-6.530623	0.564110	1.415076
С	3.652919 -0.421800 -2.356419	С	-6.481278	-1.264774	0.173512
Н	4.067127 -0.978637 -3.185990	Н	-7.332582	-1.244535	-0.515667
С	2.150076 0.473797 -0.837275	Н	-6.811851	-1.790560	1.081264
С	2.269905 -0.308537 -2.039159	S	-4.225879	0.022175	0.829842
Н	1.446040 -0.748048 -2.584610	Ν	-5.381086	-2.016004	-0.415151

3. ¹H, ³¹P{¹H} and ¹³C{¹H} NMR (CDCI₃) spectra for the new compounds 2, 3b-3e and 3a-3e



Figure S4. ¹H NMR spectrum for compound 2 in CDCl₃.



Figure S5. ³¹P{¹H} NMR spectrum for compound **2** in CDCI₃.



Figure S6. ¹³C {¹H } NMR spectrum for compound 2 in CDCI₃.



Figure S7. ¹H NMR spectrum for compound 3b in CDCl₃.



Figure S8. ³¹P{¹H} NMR spectrum for compound **3b** in CDCI₃.



Figure S9. ^{13}C { ^{1}H } NMR spectrum for compound 3b in CDCl3.



Figure S10. ¹H NMR spectrum for compound 3c in CDCl₃.



Figure S11. ³¹P{¹H} NMR spectrum for compound **3c** in CDCl₃.



Figure S12. ^{13}C { ^{1}H } NMR spectrum for compound 3c in CDCl₃.



Figure S13. ¹H NMR spectrum for compound 3d in CDCI₃.



Figure S14. ³¹P{¹H} NMR spectrum for compound **3d** in CDCl₃.



Figure S15. ^{13}C { ^{1}H } NMR spectrum for compound 3d in CDCl3.



Figure S16. ¹H NMR spectrum for compound 3e in CDCl₃.



Figure S18. ^{13}C { ^{1}H } NMR spectrum for compound 3e in CDCl₃.



Figure S19. ¹H NMR spectrum for compound 4a in CDCl₃.



Figure S20. ³¹P{¹H} NMR spectrum for compound 4a in CDCl₃.





Figure S22. ¹H NMR spectrum for compound 4b in CDCI₃.



Figure S23. ³¹P{¹H} NMR spectrum for compound 4b in CDCl₃.



Figure S24. ¹³C {¹H } NMR spectrum for compound 4b in CDCI₃.



Figure S25. ¹H NMR spectrum for compound 4c in CDCl₃.



Figure S26. ³¹P{¹H} NMR spectrum for compound **4c** in CDCl₃.



Figure S27. ^{13}C { ^{1}H } NMR spectrum for compound 4c in CDCl₃.



Figure S28. ¹H NMR spectrum for compound 4d in CDCI₃.



Figure S30. ^{13}C { 1H } NMR spectrum for compound 4d in CDCl₃.



Figure S31. ¹H NMR spectrum for compound 4e in CDCl₃.



Figure S32. ³¹P{¹H} NMR spectrum for compound **4e** in CDCl₃.



Figure S33. ¹³C {¹H } NMR spectrum for compound 4e in CDCI₃.

4. Selected ¹H and ³¹P{¹H} NMR spectra (DMSO-d⁶) overtime for complexes 4a, 3b, 3c 4d and 4e.





Figure S34. ¹H NMR spectrum in DMSO-d⁶ of 4a over time.

Figure S35. ³¹P{¹H} NMR spectrum in DMSO-d⁶ of 4a over time.



Figure S36. ¹H NMR spectrum in DMSO-d⁶ of **3b** over time.



Figure S37. ³¹P{¹H} NMR spectrum in DMSO-d⁶ of **3b** over time.



Figure S38. ¹H NMR spectrum in DMSO-d⁶ of 3c over time.



Figure S39. ³¹P{¹H} NMR spectrum in DMSO-d⁶ of **3c** over time.



Figure S40. ¹H NMR spectrum in DMSO-d⁶ of 4d over time.



Figure S41. ³¹P{¹H} NMR spectrum in DMSO-d⁶ of 4d over time.



Figure S42. ¹H NMR spectrum in DMSO-d⁶ of 4e over time.



8 days

Figure S43. ³¹P{¹H} NMR spectrum in DMSO-d⁶ of **4e** over time.

5. ESI Mass spectra of compounds 3a-e and 4a-e in CH_2CI_2 .



Figure S44. Mass spectrum of compound 3a.



Figure S45. Isotopic distribution of compound 3a for $[M - SR]^+$, m/z = 1057.05 (1%).



Figure S46. Mass spectrum of compound 3b.



Figure S47. Isotopic distribution of compound **3b** for $[M - SR]^+$, m/z = 1024.05 (100%).



Figure S48. Isotopic distribution of compound **3b** for $[M + Au]^+$, m/z = 1295.08 (10%)



Figure S49. Mass spectrum of compound 3c.



Figure S50. Isotopic distribution of compound **3c** for $[DPPF + Au]^+$, m/z =751.10 (10%).



Figure S51. Isotopic distribution of compound **3c** for $[DPPF + 2Au]^{2+}$, m/z = 981.10 (35%).



Figure S52. Isotopic distribution of compound 3c for $[M - SR]^+$, m/z = 1311.10 (100%)



Figure S53. Mass spectrum of compound 3d.



Figure S54. Isotopic distribution of compound **3d** for $[M - SR]^+$, m/z = 1066.10 (57%).



Figure S55. Mass spectrum of compound 3e.



Figure S56. Isotopic distribution of compound **3e** for $[M - SR]^+$, m/z = 1099.10 (100%).



Figure S57. Mass spectrum of compound 4a.



Figure S58. Isotopic distribution of compound **4a** for $[M + H]^+$, m/z = 677.04 (0.1%).



Figure S59. Mass spectrum of compound 4b.



Figure S60. Isotopic distribution of compound **4b** for $[M + H]^+ m/z = 643.05 (0.5\%)$.







Figure S61. Isotopic distribution of compound **4c** for $[M + H]^+ m/z = 931.10 (2\%)$.



Figure S63. Isotopic distribution of compound **4d** for $[M + H]^+ m/z = 686.17 (50\%)$.



Figure S64. Mass spectrum of compound 4e.



Figure S65. Isotopic distribution of compound **4e** for $[M + H]^+$ m/z = 719.03 (2%).