

Supporting Information

Bacteriorhodopsin folds through a poorly organized transition state

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Appendices

Derivation of Eq. 2.

Proof that $\lambda_1 > k_h$ and $\lambda_2 < k_h$.

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Appendices

Derivation of Eq. 2. When a system is comprised of three species interconverting through unimolecular reactions, the relaxation of the system proceeds with two macroscopic rate constants (λ_1 and λ_2), which are solutions of the following equation:

$$\begin{vmatrix} \lambda - (k_{12} + k_{13}) & k_{21} & k_{31} \\ k_{12} & \lambda - (k_{21} + k_{23}) & k_{32} \\ k_{13} & k_{23} & \lambda - (k_{31} + k_{32}) \end{vmatrix} = 0. \quad (\text{S1})$$

In Eq. S1, k_{ij} is the microscopic rate constant for the conversion from the i -th species to the j -th species. The kinetic mechanism of bR unfolding and refolding involves three species (bR_F, bR_U, and bO) with three kinetic constants (k_f , k_u , and k_h). By replacing k_{12} , k_{21} , and k_{23} with k_u , k_f , and k_h , respectively, and set other kinetic constants to 0, Eq. S1 is applied to the kinetic mechanism of bR unfolding and folding.

$$\begin{vmatrix} \lambda - k_u & k_f & 0 \\ k_u & \lambda - (k_f + k_h) & 0 \\ 0 & k_h & \lambda \end{vmatrix} = 0, \quad (\text{S2})$$

Eq. S2 can be arranged into a quadratic equation:

$$\lambda^2 - (k_f + k_u + k_h)\lambda + k_u k_h = 0 \quad (2)$$

Proof that $\lambda_1 > k_h$ and $\lambda_2 < k_h$. A quadratic function $f(\lambda)$ can be defined using Eq. 2:

$$f(\lambda) = \lambda^2 - (k_f + k_u + k_h)\lambda + k_u k_h$$

When $\lambda = k_h$,

$$f(k_h) = k_h^2 - (k_f + k_u + k_h)k_h + k_u k_h = -k_f k_h \quad (\text{S3})$$

Because k_f and k_h have non-zero positive values, the left side of Eq. S3 should have a negative value, which indicates that $f(k_h) < 0$. Because the graph of the function has an upward curvature, k_h is between the two roots of the equation $f(\lambda) = 0$. As λ_1 and λ_2 are defined to be the macroscopic rate constant for the fast phase and the slow phase, respectively, λ_1 is greater than λ_2 . Therefore, λ_1 is greater than k_h and λ_2 is smaller than k_h .

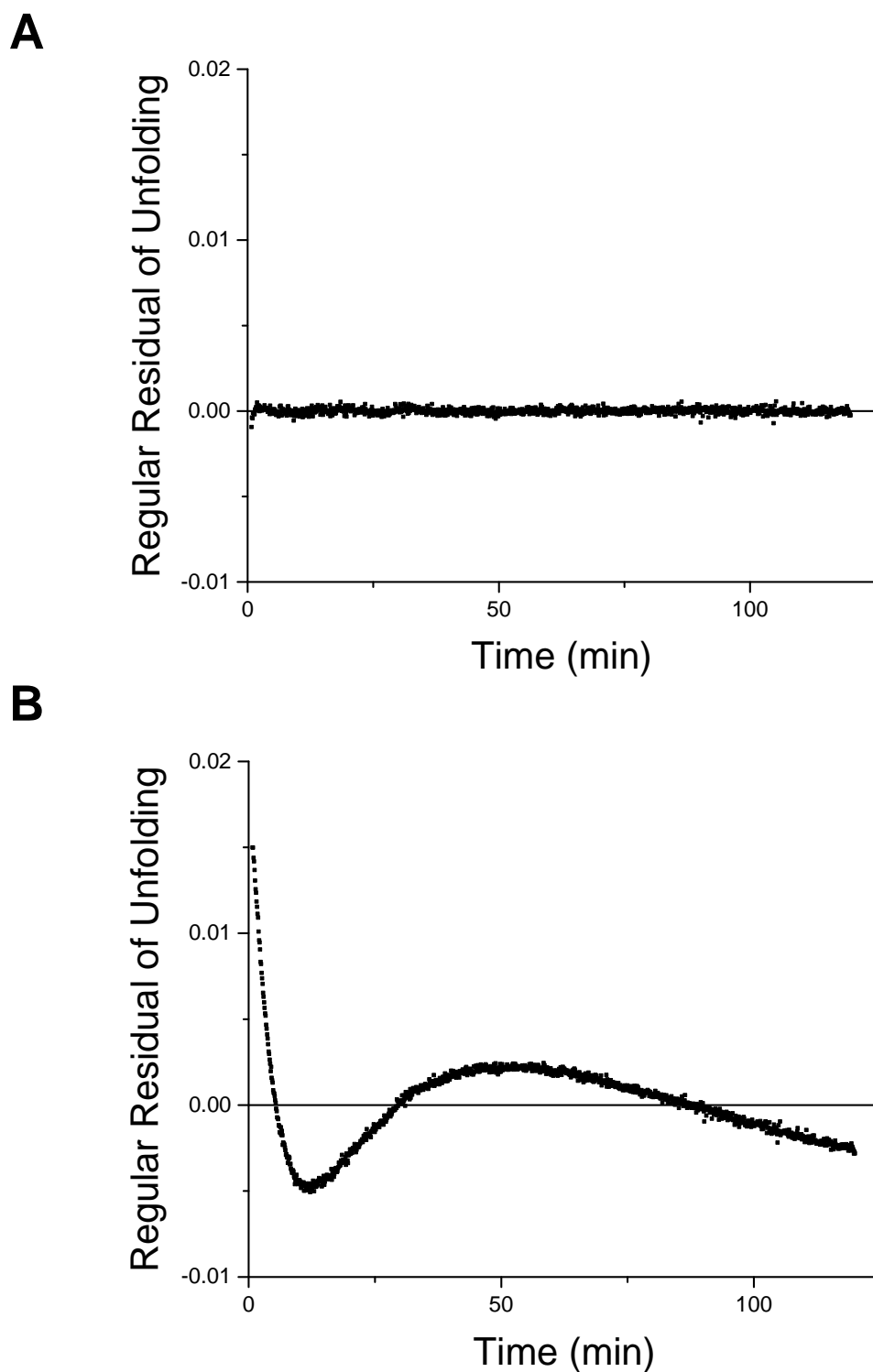


Figure S1. Residuals from curve-fitting of bR unfolding (A) The residuals from fitting the unfolding of bR_F at $X_{SDS} = 0.73$ with a double exponential equation. (B) The residuals from fitting the unfolding of bR_F at $X_{SDS} = 0.73$ with a single exponential equation.

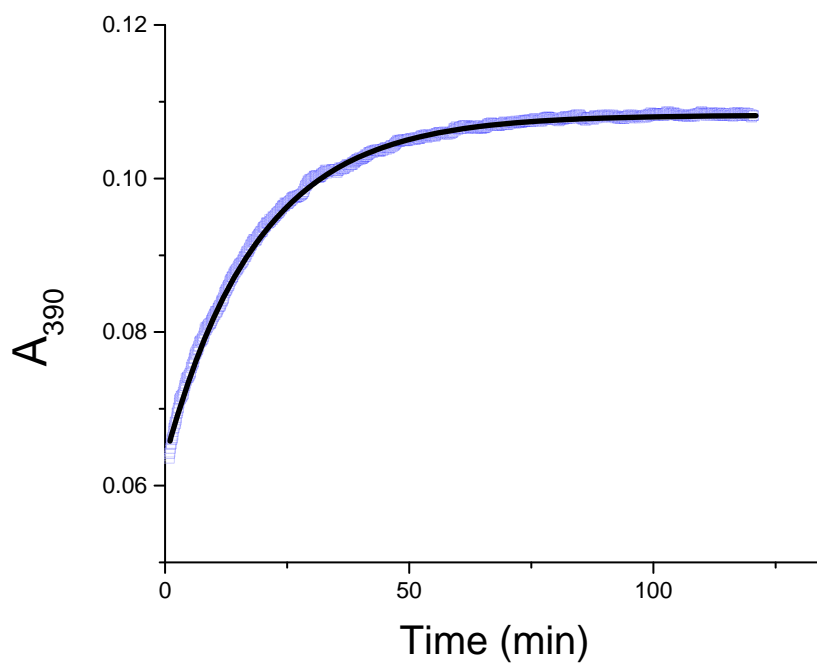


Figure S2. Kinetics of retinal hydrolysis. A single exponential fit (black) of the hydrolysis of retinal from bR_U at $X_{\text{SDS}} = 0.83$ as monitored by the increase at A_{390} (blue squares).

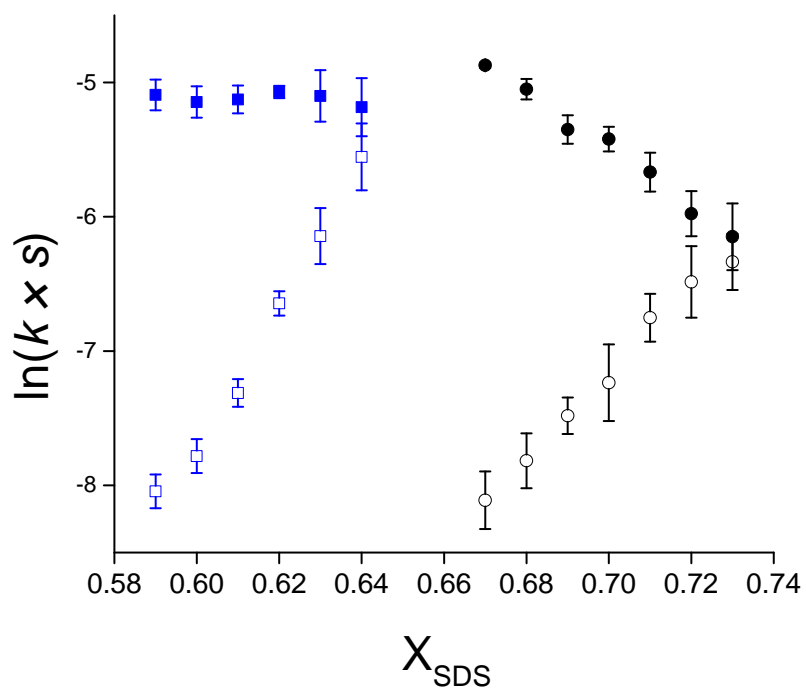


Figure S3. Folding and unfolding kinetics of I119A bR. The large shift in the folding slope indicates the change in the folding pathway of I119A bR. The natural log of k_f (filled blue squares) and k_u (open blue squares) of the bR mutant I119A is plotted with the natural log of k_f (filled black circles) and k_u (open black circles) of wild-type bR for reference.

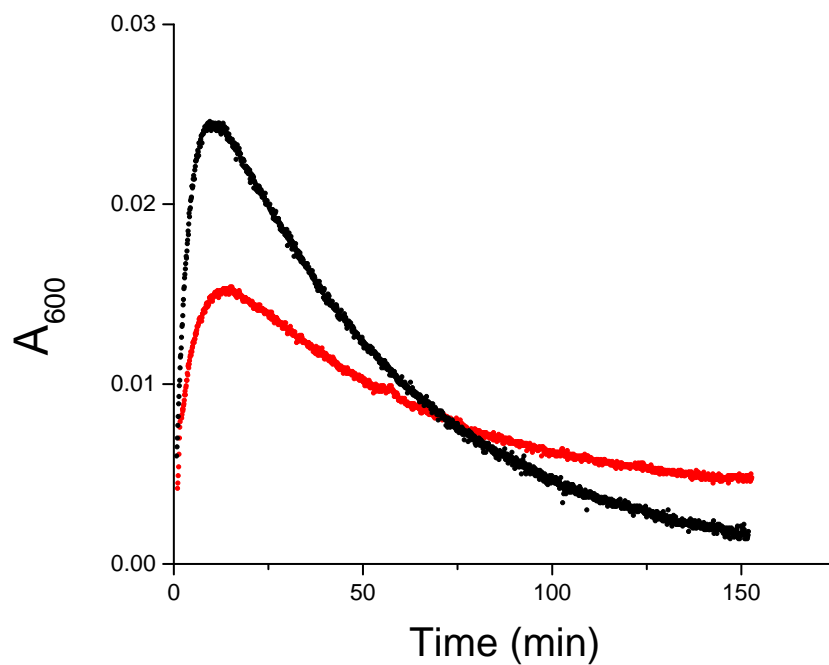
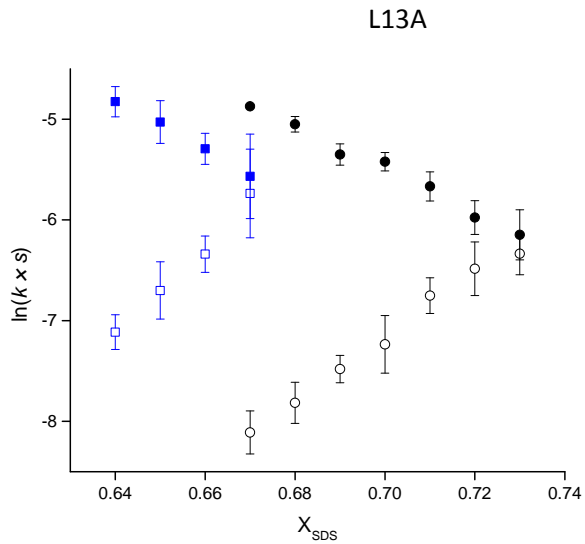
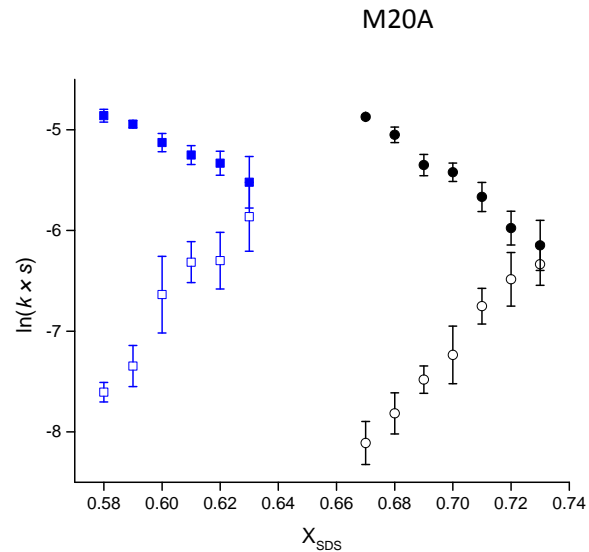
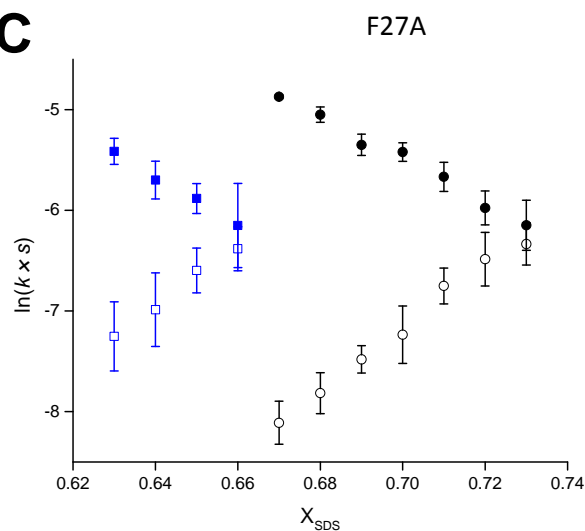
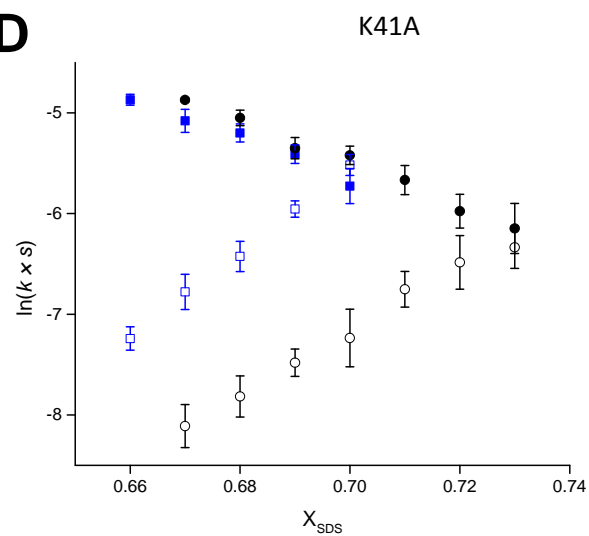
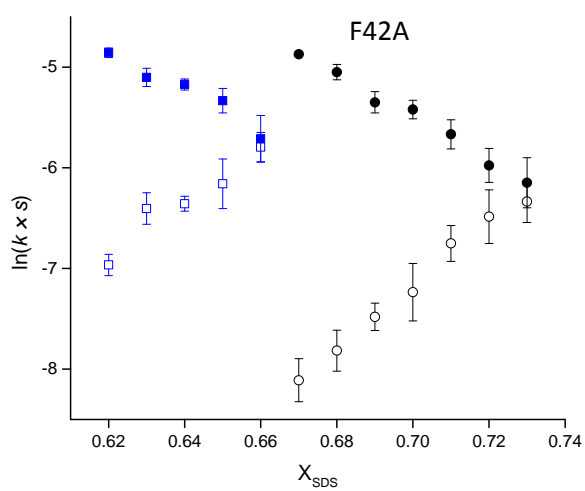
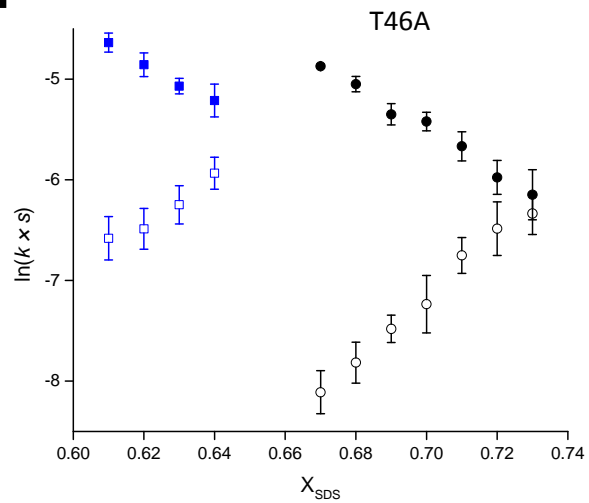
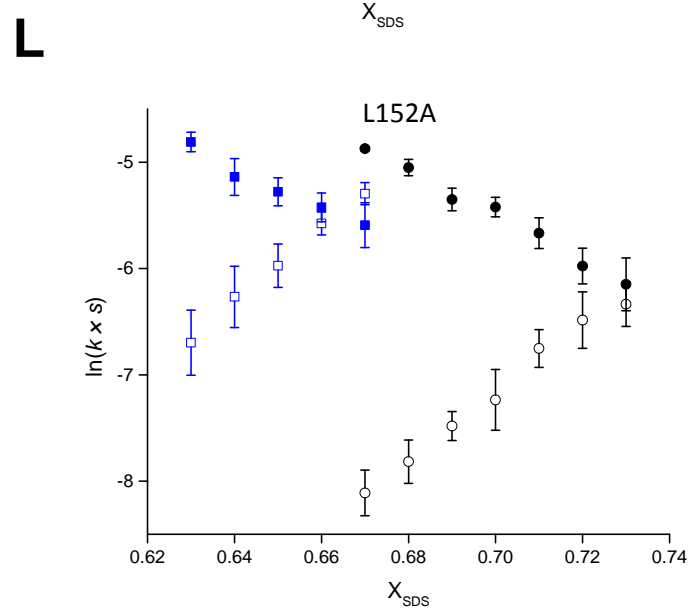
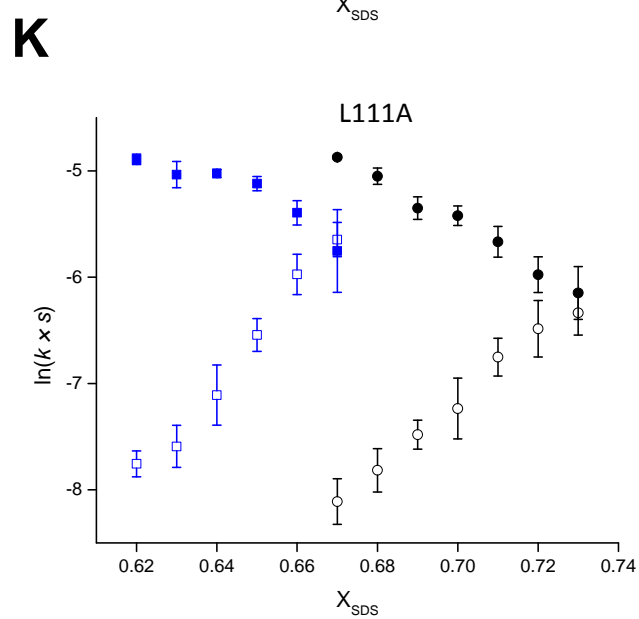
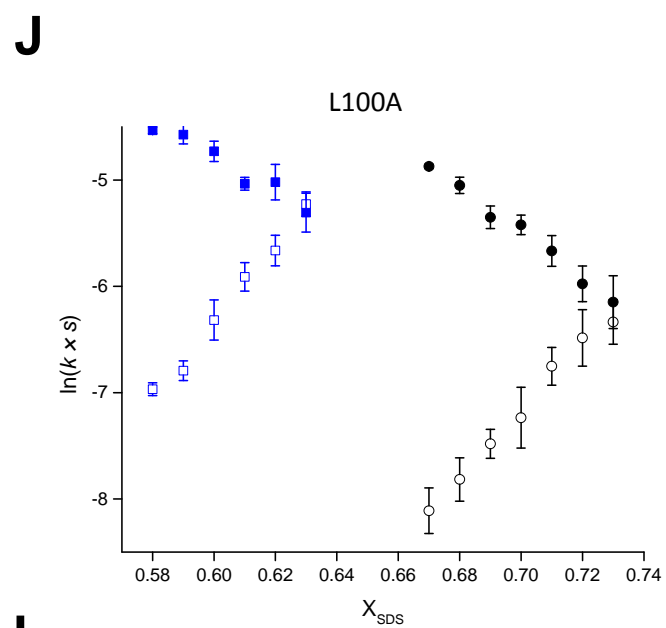
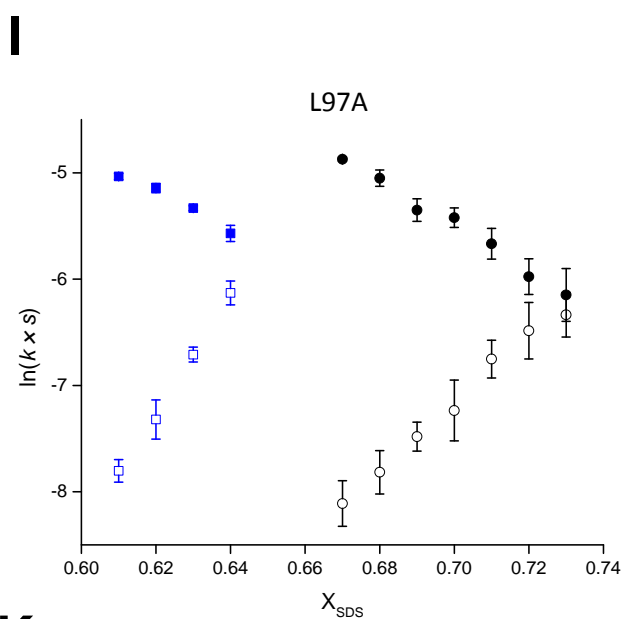
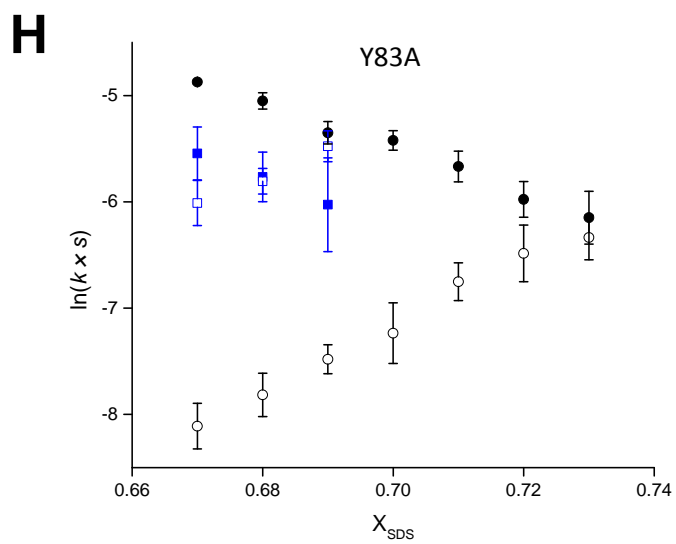
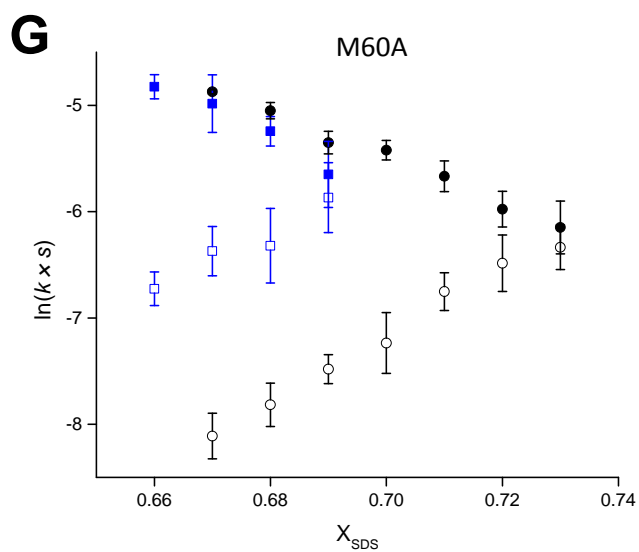


Figure S4. Refolding of F219A bR. The change in the kinetic trace shows the change in folding pathway in F219A bR. The refolding trace of the F219 A bR at $X_{\text{SDS}} = 0.67$ (red dots) is plotted with that of wild-type bR at $X_{\text{SDS}} = 0.73$ (black dots) for reference.

A**B****C****D****E****F**



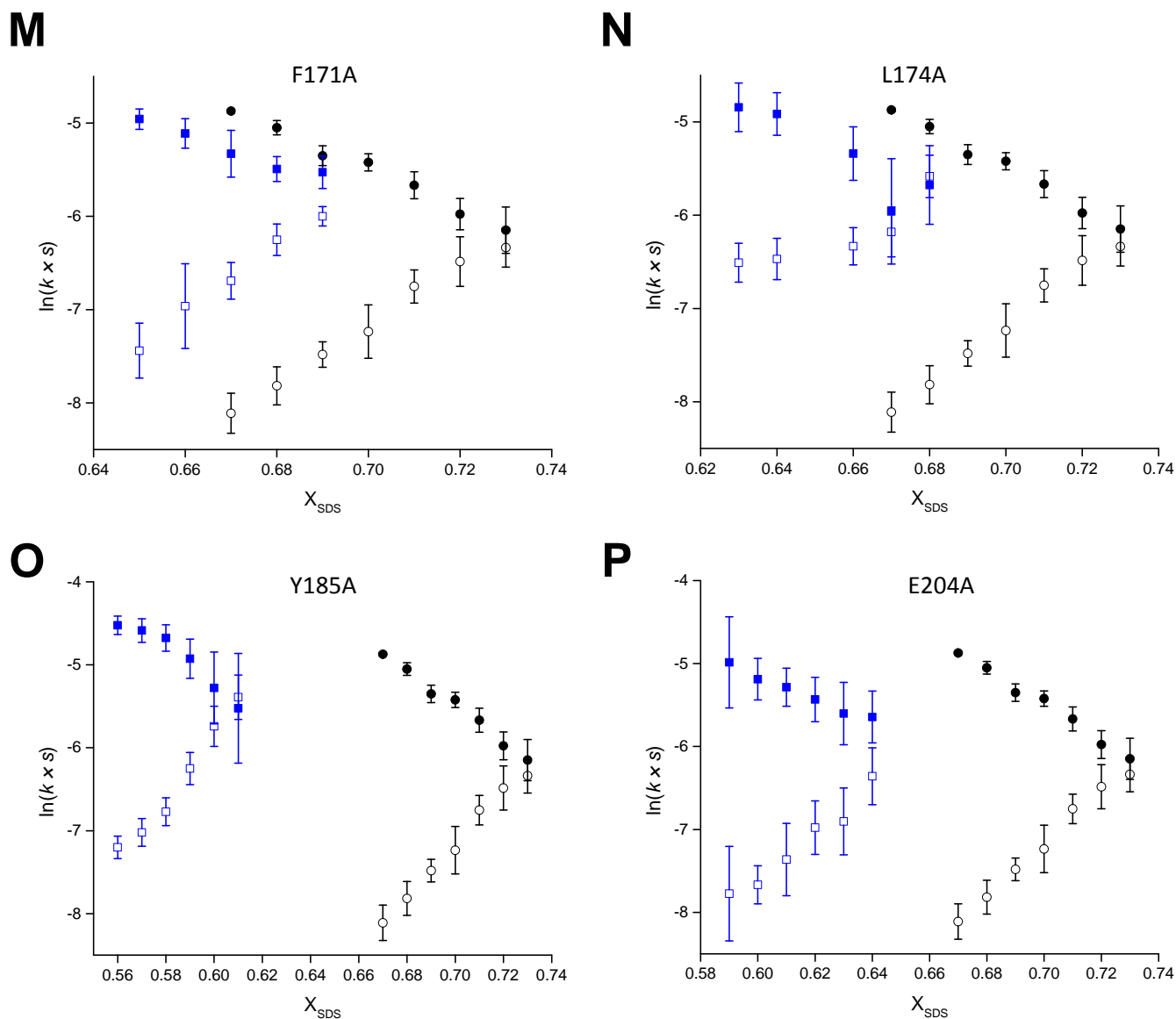


Figure S5. The extracted rates constants of the bR mutants used for ϕ -value analysis. The natural log of k_f (filled blue squares) and k_u (open blue squares) of the bR mutants are plotted with the natural log of k_f (filled black circles) and k_u (open black circles) of wild-type bR for reference. (A) L13A (B) M20A (C) F27A (D) K41A (E) F42A (F) T46A (G) M60A (H) Y83A (I) L97A (J) L100A (K) L111A (L) L152A (M) F171A (N) L174A (O) Y185A (P) E204A

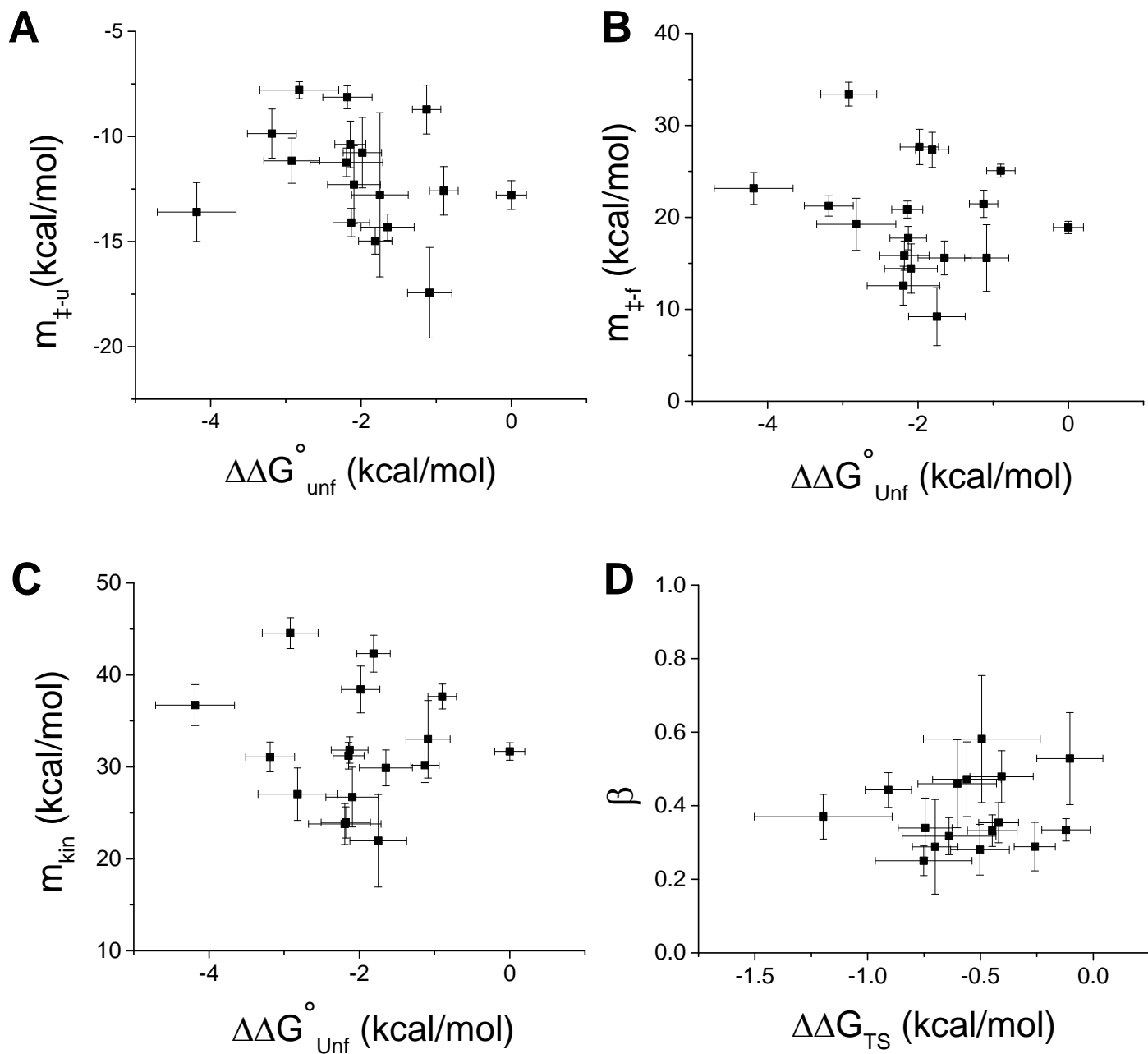


Figure S6. Lack of dependence of the folding and unfolding behavior of bR on the degree of destabilization upon mutation. (A) No effect of destabilization on the folding slope (B) No effect of destabilization on unfolding slope. (C) No effect of destabilization on the kinetic m -value (D) No correlation between the change in folding rate and β -value.