

# ***Inelastic X-Ray Scattering (IXS) of a Transition Metal Complex (FeCl<sub>4</sub>)<sup>-</sup> – A Vibrational Spectroscopy for All Normal Modes***

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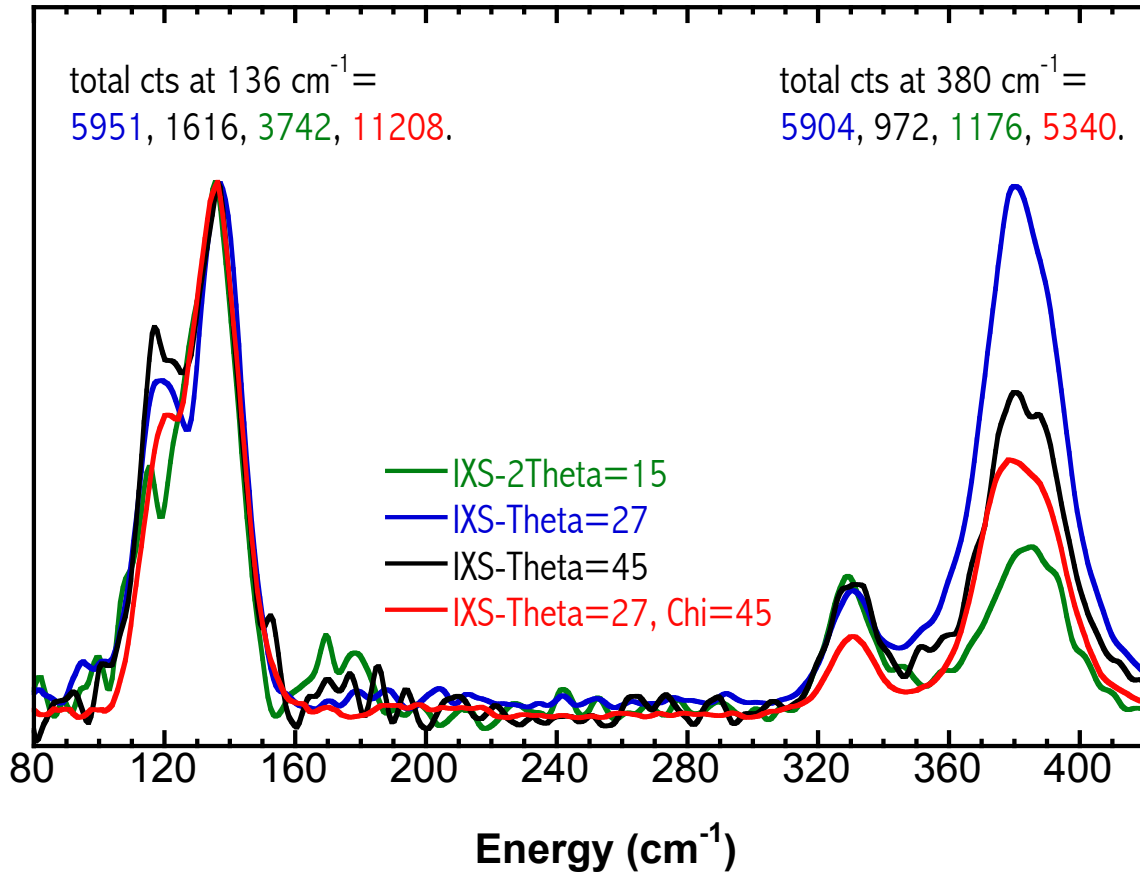
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## Supporting Information

**Preparation of (NEt<sub>4</sub>)(FeCl<sub>4</sub>).** A stoichiometric quantity of NEt<sub>4</sub>Cl in 5mL of ethanol was added to a solution containing 5 mmol (0.81g) of FeCl<sub>3</sub> in 5mL of absolute ethanol. Upon mixing and stirring for 1 hour, an amorphous precipitate of (NEt<sub>4</sub>)(FeCl<sub>4</sub>) appeared. In this experiment, 1.25 g of (NEt<sub>4</sub>)(FeCl<sub>4</sub>) in the form of yellow powder was isolated after filtration of the amorphous precipitate. The powders were re-crystallized from a tetrahydrofuran (THF) solution at ambient temperature. After 5 days of re-crystallization, greenish-yellow colored crystals of various sizes were obtained. For this particular experiment, a hexagonal pyramid crystal of ~ 4 mm x 4 mm (base dimensions) x 5 mm height was selected for IXS measurement.

**IXS Measurements.** IXS spectra were recorded on a (NEt<sub>4</sub>)(FeCl<sub>4</sub>) single crystal using standard procedures at the SPring-8 high-resolution IXS beamline, BL35-XU<sup>1,2</sup>. The photon flux on the sample was ~10<sup>10</sup> photons/sec in a 0.8 meV bandwidth at 21.75 keV with a spot size of ~0.1 mm diameter (FWHM) from a backscattering (89.8°) Si(11 11 11) monochromator. The scattered radiation was collected by a 2-dimensional array of 12 analyzer crystals. Each analyzer has an independent detector, allowing 12 spectra to be collected simultaneously, and the over-all (analyzer plus monochromator) energy resolution was 1.5 to 1.8 meV (different analyzers behaved differently). The energy scale was determined as discussed in [13] (?) and is expected to be accurate to better than 0.5%. The  $Q$  resolution was 0.48 nm<sup>-1</sup>, where  $Q = 4\pi / \lambda \sin\theta$ ,  $2\theta$  is the scattering angle and  $\lambda$  is the wavelength of x-rays. During the IXS measurements, the sample crystal was maintained at ~27K using a closed cycle He cryostat. The inelastically scattered radiation from each of the 12 analyzers was recorded with a separate CdZnTe detector. For (NEt<sub>4</sub>)(FeCl<sub>4</sub>), the typical total signal count rate on the 380 cm<sup>-1</sup> peak was ~0.2-0.9 cps.

The actual counts in each spectrum are shown below.



**IXS Data Analysis and Normal Mode Calculations.** The IXS signal was extracted from the raw averaged spectra by subtracting a combination of Lorentzian and Gaussian background lineshape functions. No smoothing was used for any of the spectra. Normal mode calculations were done using a modification of the program Vibratz<sup>3</sup> and a Urey-Bradley force field<sup>4</sup>. The parameters for this force field determine the potential energy  $V$  as previously described<sup>5</sup>. The Vibratz program was used to calculate the relevant phonon polarization eigenvectors (complex phonon polarization) including the Fe and Cl motion in a given normal mode from the normal mode eigenvectors. The calculations used the Evans *et al.*, *Acta Cryst.* 46, 1818 (1990) CSD structure file TABPIV, with a slightly larger force constant for the shorter Fe-Cl bond. To simulate the IXS, Gaussian components were generated at the calculated frequencies, with an amplitude determined by the relative dynamic structure for a particular normal mode and scattering geometry

and a FWHM determined by the  $14 \text{ cm}^{-1}$  instrumental resolution. We note that there are no adjustable scale factors in these simulations.

**X-Ray Crystallography.** The tetrahedral  $[\text{FeCl}_4]^-$  ion was chosen for its extremely simple set of well-studied normal modes, as well as the straightforward arrangement of the  $(\text{FeCl}_4)^-$  ions in the crystal with  $(\text{NEt}_4)^+$  counterions. An additional complication discovered *ex post facto* is that, with flash cooling,  $\text{NEt}_4\text{FeCl}_4$  undergoes a first order phase transition at  $\sim 229 \text{ K}$  that changes the space group from  $P6_3mc$  (Phase I) to  $P31c$  (Phase III). In Phase I there is one Fe in the asymmetric unit and the Fe resides on a site with crystallographic  $3m$  symmetry. In Phase III there are five Fe's in the asymmetric unit, two of which have no crystallographic symmetry and three of which have 3-fold symmetry. As a consequence of the low temperature phase, the structure is not as simple, although the structure bears the same relationship with respect to the  $c$  axis as in the room temperature phase. The most noticeable change is that two of the  $(\text{FeCl}_4)^-$  units are tipped by  $6.7^\circ$  and  $8.0^\circ$ , respectively, with respect to  $c$ . With slow cooling, an intermediate phase is observed (Phase II) both Phase II and Phase III show less disorder in the  $(\text{NEt}_4)^+$  group than in Phase I.

Table S1. Crystal data for three phases of  $(\text{NEt}_4)(\text{FeCl}_4)$

	Phase I <sup>ref [1]</sup>	Phase II <sup>slow cooling</sup>	Phase III <sup>flash cooling</sup>
Crystal system	hexagonal	orthorhombic	trigonal
Space group	$P6_3mc$ (No. 186)	$Pca2_1$ (No. 29)	$P31c$ (No. 159)
$a, \text{Å}$	8.198(1)	13.9690(3)	24.3337(9)
$b, \text{Å}$		8.1116(3)	
$c, \text{Å}$	13.183(3)	12.7889(5)	12.8909(5)
$V, \text{Å}^3$	767.3(2)	1449.12(10)	6610.4(6)
$Z$	2	4	18
$T, \text{K}$	295(2)	90(2)	90(2)
		3 component twin	inversion twin
$R_1 [I > 2\sigma(I)]$	0.053	0.015	0.031

Fe cryst symm, tilt to <i>c</i> axis, °			
Fe1	3m· ( $C_{3v}$ ), 0	1 ( <i>CI</i> ), 4.0	1 ( <i>CI</i> ), 6.7
Fe2			1 ( <i>CI</i> ), 8.0
Fe3			3 ( <i>C3</i> ), 0
Fe4			3 ( <i>C3</i> ), 0
Fe5			3 ( <i>C3</i> ), 0

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