

Supporting Information for

Quinary structure modulates protein stability in cells

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Author contributions: W.B.M., A.E.S., R.D.C and G.J.P. designed research; W.B.M.,
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wrote the paper.

Table S1. Backbone amide proton exchange rates (k_{obs} , s^{-1}) and corresponding $\Delta G_{\text{op}}^{\circ}$ (kcal/mol) values for wild type (T2Q) GB1 in buffer (20 mM citrate, 150 mM NaCl, pH 5.8, 37 °C). The in-cell data have been published (1). ^aSDM: standard deviation of the mean from three trials. pH is corrected for the isotope effect. For glass electrodes in D₂O, pH = pH_{read} + 0.4 (2).

residue	$k_{\text{Obs,buff}}$	SDM^a	$\Delta G_{\text{op,buff}}^{\circ}$	SDM^a
Y3	3.88E-05	4.92E-06	6.31	0.10
K4	4.87E-05	9.70E-06	6.46	0.02
V26	2.18E-05	2.22E-06	7.04	0.12
K28	4.97E-05	6.37E-06	5.97	0.04
V29	1.84E-05	3.78E-06	6.00	0.04
K31	1.32E-05	3.10E-06	7.17	0.09
A34	2.23E-05	2.55E-06	6.77	0.05
T44	1.34E-05	1.02E-06	6.89	0.06
D46	1.37E-05	2.74E-06	6.74	0.02
T51	2.79E-05	1.99E-06	6.64	0.10
F52	1.80E-05	5.34E-06	6.72	0.13
T53	2.54E-05	4.99E-06	6.73	0.14

Table S2. Backbone amide proton exchange rates (k_{obs} , s^{-1}) and corresponding $\Delta G_{\text{op}}^{\circ}$ (kcal/mol) values for D40K GB1 in cells and in buffer (20 mM citrate, 150 mM NaCl pH 5.8, 37 °C). pH is corrected for the isotope effect. For glass electrodes in D_2O , $\text{pH} = \text{pH}_{\text{read}} + 0.4$ (2). ^aSDM: standard deviation of the mean from three trials. ^b $\Delta\Delta G_{\text{op,cell,D40K}}^{\circ} = \Delta G_{\text{op,cell,D40K}}^{\circ} - \Delta G_{\text{op,buff,D40K}}^{\circ}$. ^cFrom propagation of error, $\sigma =$

$$\sqrt{(\delta_{\Delta G_{\text{op,cell}}^{\circ}})^2 + (\delta_{\Delta G_{\text{op,buff}}^{\circ}})^2}.$$

residue	$k_{\text{obs,cells}}$	SDM ^a	$k_{\text{obs,buff}}$	SDM ^a
Y3	9.73E-04	±4.46E-05	1.80E-05	±1.52E-06
A26	9.45E-04	±5.48E-05	9.43E-06	±2.27E-06
K28	9.73E-04	±4.74E-05	3.41E-05	±1.48E-06
V29	6.30E-04	±4.87E-05	1.61E-05	±1.62E-06
K31	9.28E-04	±1.92E-05	1.16E-05	±1.95E-06
A34	7.17E-04	±8.10E-05	1.06E-05	±1.88E-06
T44	6.77E-04	±7.96E-06	9.54E-06	±3.14E-06
D46	6.84E-04	±5.68E-05	1.07E-05	±2.45E-06
T51	9.59E-04	±1.79E-05	1.35E-05	±2.79E-06
F52	8.60E-04	±1.09E-04	1.02E-05	±2.59E-06
T53	9.42E-04	±3.27E-05	1.08E-05	±2.24E-06

residue	$\Delta G_{\text{op,cell,D40K}}^{\circ}$	SDM ^a	$\Delta G_{\text{op,buff,D40K}}^{\circ}$	SDM ^a	$\Delta\Delta G_{\text{op,cell,D40K}}^{\circ}$ ^b	uncertainty ^c
Y3	4.20	±0.03	6.59	±0.02	-2.39	±0.03
A26	4.60	±0.04	7.43	±0.11	-2.82	±0.12
K28	4.07	±0.03	6.07	±0.02	-2.00	±0.03
V29	3.75	±0.05	5.94	±0.02	-2.19	±0.05
K31	4.36	±0.01	7.02	±0.07	-2.65	±0.07
A34	4.57	±0.07	7.11	±0.07	-2.55	±0.10
T44	4.27	±0.01	6.94	±0.18	-2.67	±0.18
D46	4.23	±0.05	6.78	±0.11	-2.55	±0.12
T51	4.38	±0.01	6.97	±0.08	-2.59	±0.08
F52	4.33	±0.08	7.04	±0.12	-2.71	±0.14
T53	4.31	±0.02	7.03	±0.09	-2.72	±0.09

Table S3. $\Delta\Delta G_{\text{op,mut}}^{\circ}$ ($\Delta G_{\text{op,D40K}}^{\circ} - \Delta G_{\text{op,wt}}^{\circ}$, kcal/mol) caused by the D40K mutation in cells and in buffer (20 mM citrate, pH 5.8, 37 °C). pH is corrected for the isotope effect. For glass electrodes in D₂O, pH = pH_{read} + 0.4 (2). ^aFrom propagation of error.

residue	$\Delta\Delta G_{\text{op,mut}}^{\circ}$	uncertainty^a	$\Delta\Delta G_{\text{op,mut}}^{\circ}$	uncertainty^a
Y3	-1.18	±0.04	0.34	±0.10
A26	-1.41	±0.06	0.44	±0.16
K28	-1.08	±0.05	0.09	±0.04
V29	-1.35	±0.05	-0.09	±0.04
K31	-1.37	±0.07	-0.06	±0.11
A34	-1.29	±0.08	0.34	±0.09
T44	-1.63	±0.07	-0.03	±0.19
D46	-1.49	±0.12	0.04	±0.11
T51	-1.23	±0.06	0.35	±0.13
F52	-1.50	±0.11	0.20	±0.17
T53	-1.22	±0.07	0.42	±0.16

Table S4. Backbone amide proton exchange rates (k_{obs} , s^{-1}) and corresponding ΔG_{op}^{\ddagger} ($kcal/mol$) values for I6L GB1 in buffer (20 mM citrate, 150 mM NaCl pH 5.8, 37 °C). The in-cell data have been published (1). pH is corrected for the isotope effect. For glass electrodes in D_2O , $pH = pH_{read} + 0.4$ (2). ^aThe k_{obs} values are the average of two trials. ^bThe uncertainty is the range. ^cFrom propagation of error.

residue	k_{obs}	uncertainty^b	ΔG_{op}^{\ddagger} ^c	uncertainty^c
Y3	9.13E-05	±1.80E-06	5.59	±0.06
K4	6.39E-05	±3.15E-06	5.93	±0.10
A26	3.85E-05	±4.50E-06	6.51	±0.14
K28	7.42E-05	±4.30E-06	5.59	±0.10
V29	4.85E-05	±6.50E-06	5.26	±0.15
K31	4.50E-05	±9.00E-06	6.17	±0.20
A34	6.97E-05	±7.00E-07	5.93	±0.06
T44	6.75E-05	±1.15E-05	5.63	±0.18
D46	6.50E-05	±1.50E-05	5.64	±0.21
T51	6.65E-05	±1.65E-05	5.97	±0.23
F52	6.00E-05	±1.60E-05	5.91	±0.24
T53	6.90E-05	±1.21E-05	5.86	±0.18

Table S5. $\delta\Delta\Delta G_{\text{op,int}}^{\text{oi}}$ ($=\Delta\Delta G_{\text{op,mut,cell}}^{\text{oi}} - \Delta\Delta G_{\text{op,mut,buff}}^{\text{oi}}$, kcal/mol) values for I6L and D40K

GB1. ^aFrom propagation of error.

residue	$\delta\Delta\Delta G_{\text{op,int,I6L}}^{\text{oi}}$	uncertainty^a	$\delta\Delta\Delta G_{\text{op,int,D40K}}^{\text{oi}}$	uncertainty^a
Y3	-1.00	±0.13	-1.46	±0.11
K4	-0.67	±0.15	-	-
A26	-0.84	±0.19	-1.79	±0.17
K28	-0.62	±0.12	-1.17	±0.06
V29	-0.64	±0.16	-1.29	±0.07
K31	-0.53	±0.23	-1.22	±0.13
A34	-0.83	±0.10	-1.63	±0.12
T44	-2.71	±0.21	-3.98	±0.20
D46	-0.63	±0.24	-1.54	±0.16
T51	-0.75	±0.26	-1.56	±0.14
F52	-1.05	±0.29	-1.82	±0.21
T53	-0.56	±0.25	-1.52	±0.18

Table S6. Backbone amide proton exchange rates (k_{obs} , s^{-1}) and corresponding $\Delta G_{\text{op}}^{\circ}$ (kcal/mol) values for D40N GB1 in cells and in buffer (20 mM citrate, 150 mM NaCl pH 5.8, 37 °C). pH is corrected for the isotope effect. For glass electrodes in D_2O , $\text{pH} = \text{pH}_{\text{read}} + 0.4$ (2). ^aSDM: standard deviation of the mean from three trials. ^b $\Delta\Delta G_{\text{op,cell,D40N}}^{\circ} = \Delta G_{\text{op,cell,D40N}}^{\circ} - \Delta G_{\text{op,buff,D40N}}^{\circ}$. ^cFrom propagation of error.

residue	$k_{\text{obs,cells}}$	SDM ^a	$k_{\text{obs,buff}}$	SDM ^a		
Y3	5.23E-04	±4.12E-05	1.64E-05	±1.60E-07		
K4	1.17E-03	±3.00E-04	2.58E-05	±2.10E-07		
A26	4.76E-04	±5.64E-05	6.88E-06	±4.25E-07		
K28	5.25E-04	±4.34E-05	3.53E-05	±2.05E-07		
V29	4.31E-04	±7.97E-05	1.52E-05	±8.00E-08		
K31	4.87E-04	±5.33E-05	7.95E-06	±4.50E-07		
A34	4.91E-04	±5.11E-05	1.11E-05	±2.40E-07		
T44	4.05E-04	±9.15E-05	7.65E-06	±1.50E-07		
D46	4.27E-04	±7.99E-05	7.77E-06	±1.30E-07		
T51	5.30E-04	±4.36E-05	1.14E-05	±1.50E-07		
F52	5.35E-04	±2.87E-05	7.14E-06	±5.05E-07		
T53	4.84E-04	±5.67E-05	8.94E-06	±3.15E-07		

residue	$\Delta G_{\text{op,cell,D40N}}^{\circ}$	SDM ^a	$\Delta G_{\text{op,buff,D40N}}^{\circ}$	SDM ^a	$\Delta\Delta G_{\text{op,cell,D40N}}^{\circ}$ ^b	uncertainty ^c
Y3	4.59	±0.05	6.72	±0.006	-2.13	±0.05
K4	4.25	±0.19	6.55	±0.005	-2.30	±0.19
A26	5.03	±0.07	7.64	±0.04	-2.61	±0.08
K28	4.45	±0.05	6.11	±0.004	-1.66	±0.05
V29	4.00	±0.11	6.05	±0.003	-2.04	±0.11
K31	4.77	±0.07	7.30	±0.03	-2.53	±0.08
A34	4.80	±0.06	7.13	±0.011	-2.33	±0.06
T44	4.62	±0.13	7.04	±0.01	-2.42	±0.13
D46	4.54	±0.11	6.99	±0.01	-2.45	±0.11
T51	4.75	±0.05	7.12	±0.01	-2.37	±0.05
F52	4.62	±0.03	7.28	±0.04	-2.66	±0.05
T53	4.72	±0.07	7.18	±0.02	-2.45	±0.08

Table S7. $\Delta\Delta G_{\text{op,mut}}^{\circ}$ ($\Delta G_{\text{op,D40N}}^{\circ} - \Delta G_{\text{op,wt}}^{\circ}$, kcal/mol) caused by the D40N mutation in cells and in buffer (20 mM citrate, pH 5.8, 37 °C). pH is corrected for the isotope effect.

For glass electrodes in D₂O, pH = pH_{read} + 0.4 (2). ^aFrom propagation of error.

residue	$\Delta\Delta G_{\text{op,mut,cell}}^{\circ}$	uncertainty^a	$\Delta\Delta G_{\text{op,mut,buff}}^{\circ}$	uncertainty^a
Y3	-0.79	±0.06	0.41	±0.10
K4	-1.01	±0.19	0.10	±0.02
A26	-0.99	±0.09	0.60	±0.12
K28	-0.74	±0.06	0.14	±0.04
V29	-1.14	±0.11	0.05	±0.04
K31	-1.02	±0.10	0.13	±0.09
A34	-1.11	±0.08	0.36	±0.06
T44	-1.32	±0.15	2.44	±0.06
D46	-1.23	±0.15	0.26	±0.02
T51	-0.91	±0.08	0.47	±0.10
F52	-1.05	±0.08	0.55	±0.13
T53	-0.85	±0.10	0.45	±0.14

Table S8. $\delta\Delta\Delta G_{\text{op,int}}^{\text{oi}}$ ($\Delta\Delta G_{\text{op,mut,cell}}^{\text{oi}} - \Delta\Delta G_{\text{op,mut,buffer}}^{\text{oi}}$, kcal/mol) values for D40N GB1.

^aFrom propagation of error.

residue	$\delta\Delta\Delta G_{\text{op,int,D40N}}^{\text{oi}}$	uncertainty^a
Y3	-1.19	± 0.11
K4	-1.11	± 0.19
A26	-1.59	± 0.15
K28	-0.88	± 0.08
V29	-1.19	± 0.12
K31	-1.15	± 0.14
A34	-1.47	± 0.10
T44	-3.76	± 0.16
D46	-1.49	± 0.15
T51	-1.38	± 0.13
F52	-1.60	± 0.16
T53	-1.30	± 0.17

Table S9. Backbone amide proton exchange rates (k_{obs} , s^{-1}) and corresponding $\Delta G_{\text{op}}^{\circ}$ (kcal/mol) values for D40A GB1 in cells and in buffer (20 mM citrate, 150 mM NaCl pH 5.8, 37 °C). pH is corrected for the isotope effect. For glass electrodes in D_2O , $\text{pH} = \text{pH}_{\text{read}} + 0.4$ (2). ^aSDM: standard deviation of the mean from three trials. ^b $\Delta\Delta G_{\text{op,cell,D40A}}^{\circ} = \Delta G_{\text{op,cell,D40A}}^{\circ} - \Delta G_{\text{op,buff,D40A}}^{\circ}$. ^cFrom propagation of error.

residue	$k_{\text{obs,cells}}$	SDM^a	$k_{\text{obs,buff}}$	SDM^a		
Y3	6.04E-04	±1.15E-04	3.71E-05	±6.04E-06		
K4	9.65E-04	±2.63E-04	4.10E-05	±4.75E-06		
A26	4.67E-04	±6.70E-05	1.48E-05	±2.37E-06		
K28	6.09E-04	±1.03E-04	5.84E-05	±9.67E-06		
V29	2.81E-04	±4.91E-05	2.64E-05	±4.80E-06		
K31	5.22E-04	±1.96E-04	1.74E-05	±3.24E-06		
A34	4.89E-04	±1.21E-04	1.95E-05	±1.38E-06		
T44	3.23E-04	±4.81E-05	1.15E-05	±1.20E-06		
D46	3.30E-04	±4.03E-05	1.36E-05	±1.82E-06		
T51	5.72E-04	±1.08E-04	1.97E-05	±2.62E-06		
F52	5.14E-04	±9.10E-05	1.61E-05	±2.44E-06		
T53	5.37E-04	±1.03E-04	1.91E-05	±2.84E-06		

residue	$\Delta G_{\text{op,cell,D40A}}^{\circ}$	SDM^a	$\Delta G_{\text{op,buff,D40A}}^{\circ}$	SDM^a	$\Delta\Delta G_{\text{op,cell,D40A}}^{\circ}$^b	uncertainty^c
Y3	4.52	±0.11	6.45	±0.02	-1.94	±0.12
K4	4.36	±0.15	6.49	±0.05	-2.13	±0.16
A26	5.05	±0.09	7.40	±0.03	-2.35	±0.09
K28	4.37	±0.10	6.03	±0.02	-1.66	±0.10
V29	4.26	±0.10	5.95	±0.02	-1.68	±0.10
K31	4.80	±0.22	7.06	±0.01	-2.26	±0.22
A34	4.83	±0.14	7.00	±0.08	-2.17	±0.16
T44	4.74	±0.09	7.00	±0.07	-2.27	±0.11
D46	4.69	±0.07	6.86	±0.03	-2.17	±0.08
T51	4.72	±0.11	7.01	±0.04	-2.29	±0.12
F52	4.66	±0.11	7.01	±0.03	-2.36	±0.11
T53	4.67	±0.11	6.94	±0.03	-2.27	±0.12

Table S10. $\Delta\Delta G_{\text{op,mut}}^{\circ}$ ($\Delta G_{\text{op,D40A}}^{\circ} - \Delta G_{\text{op,Wt}}^{\circ}$, kcal/mol) caused by the D40A mutation in cells and in buffer (20 mM citrate, 150 mM NaCl pH 5.8, 37 °C). pH is corrected for the isotope effect. For glass electrodes in D₂O, pH = pH_{read} + 0.4 (2). ^aFrom propagation of error.

residue	$\Delta\Delta G_{\text{op,mut,cell}}^{\circ}$	uncertainty^a	$\Delta\Delta G_{\text{op,mut,buff}}^{\circ}$	uncertainty^a
Y3	-0.86	±0.12	0.14	±0.10
K4	-0.90	±0.16	0.04	±0.06
A26	-0.98	±0.17	0.36	±0.12
K28	-0.82	±0.10	0.06	±0.04
V29	-0.88	±0.11	-0.05	±0.04
K31	-0.98	±0.11	-0.11	±0.09
A34	-1.08	±0.23	0.23	±0.10
T44	-1.20	±0.15	2.41	±0.09
D46	-1.08	±0.11	0.12	±0.04
T51	-0.94	±0.13	0.36	±0.10
F52	-1.01	±0.13	0.29	±0.13
T53	-0.90	±0.13	0.22	±0.14

Table S11. $\delta\Delta\Delta G_{\text{op,int}}^{\text{oi}}$ ($\Delta\Delta G_{\text{op,mut,cell}}^{\text{oi}} - \Delta\Delta G_{\text{op,mut,buff}}^{\text{oi}}$, kcal/mol) values for D40A GB1.

^aFrom propagation of error.

residue	$\delta\Delta\Delta G_{\text{op,int,D40}}^{\text{oi}}$	uncertainty^a
Y3	-1.00	± 0.16
K4	-0.94	± 0.17
A26	-1.34	± 0.16
K28	-0.88	± 0.11
V29	-0.83	± 0.11
K31	-0.88	± 0.24
A34	-1.31	± 0.18
T44	-3.61	± 0.15
D46	-1.20	± 0.14
T51	-1.30	± 0.16
F52	-1.30	± 0.18
T53	-1.12	± 0.19

Fig. S1. Expanded thermodynamic cycle from Fig. 1, depicting equilibria between native (N) and denatured (D) states of wt GB1 and its variants. Intracellular interactions (red) cause deviations in values of $\Delta\Delta G_{mut,cell}^{\circ}$ and $\Delta\Delta G_{mut,buff}^{\circ}$.

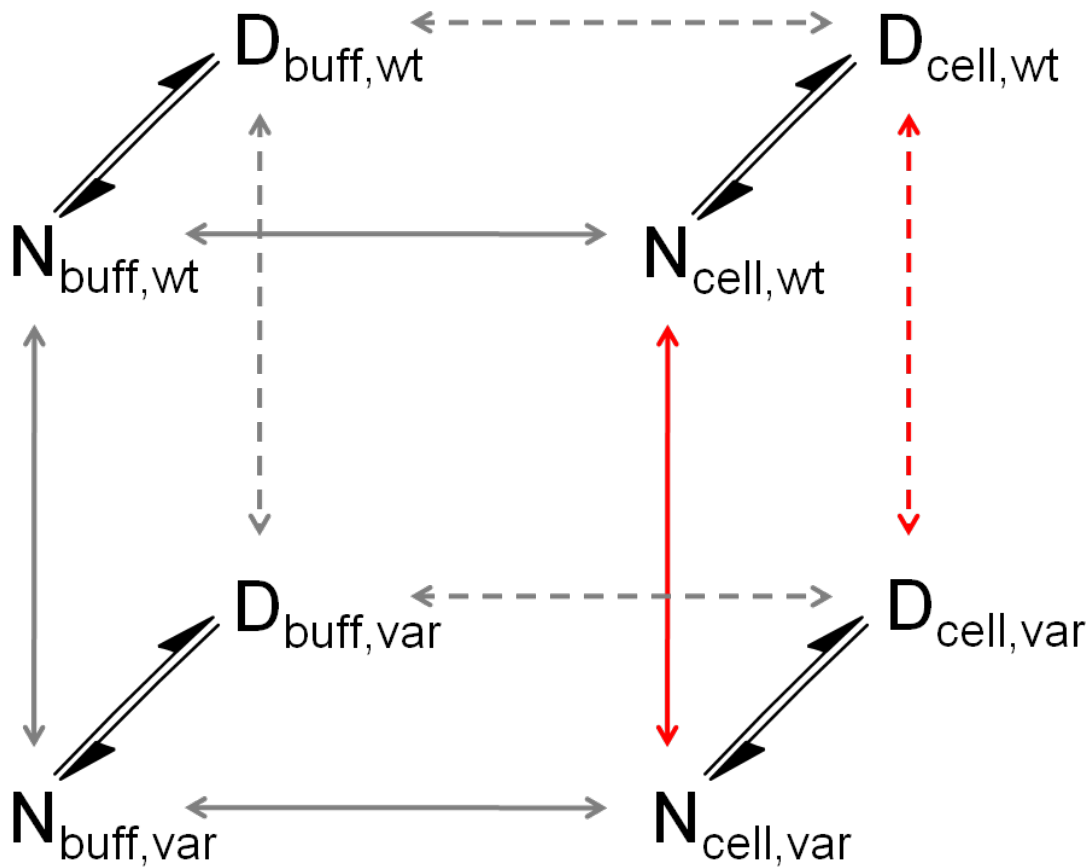
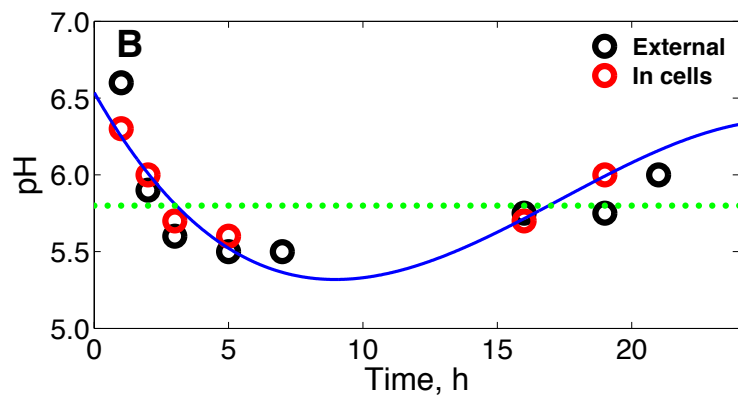
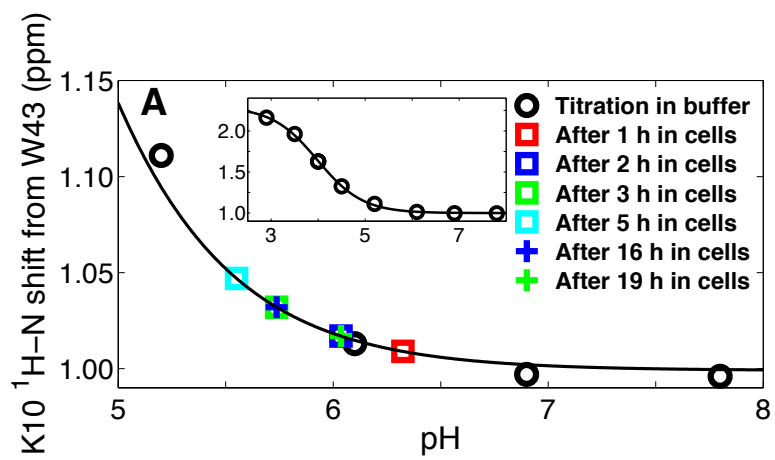


Fig. S2 Cytoplasmic pH. A) pH dependence of the K10 amide proton shift relative to that of the pH-insensitive amide proton of W43 (10.507 ppm). pH is corrected for the isotope effect. For glass electrodes in D₂O, pH = pH_{read} + 0.4 (2). The inset shows the entire titration curve constructed from GB1-expressing lysates in 50 mM citrate, 50 mM bis-tris propane, 50 mM HEPES, 50 mM borate, 5% D₂O, 0.1% DSS, 298 K. The curve is from a fit to the Henderson-Hasselbach equation, $\delta = \delta_{low} - \frac{\delta_{low} - \delta_{high}}{1 + 10^{n(pK_a - pH)}}$, where δ_{low} and δ_{high} are the low and high pH shift plateaus and n is the number of protons. The derived pK_a, 4.0, which probably reflects that of E56 to which the backbone amide forms a hydrogen bond, is consistent with literature values (3). In-cell spectra were acquired as a function of time and shift changes overlaid onto the curve. **B)** Time dependence of the external (glass electrode) and internal (from data in A) pH. The dashed horizontal line represents the pH used to derive the intrinsic rates. The curve is of no theoretical significance.



S1

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