

## Article

## Phosphatase Specificity and Pathway Insulation in Signaling Networks

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**ABSTRACT** Phosphatases play an important role in cellular signaling networks by regulating the phosphorylation state of proteins. Phosphatases are classically considered to be promiscuous, acting on tens to hundreds of different substrates. We recently demonstrated that a shared phosphatase can couple the responses of two proteins to incoming signals, even if those two substrates are from otherwise isolated areas of the network. This finding raises a potential paradox: if phosphatases are indeed highly promiscuous, how do cells insulate themselves against unwanted crosstalk? Here, we use mathematical models to explore three possible insulation mechanisms. One approach involves evolving phosphatase  $K_M$  values that are large enough to prevent saturation by the phosphatase's substrates. Although this is an effective method for generating isolation, the phosphatase becomes a highly inefficient enzyme, which prevents the system from achieving switch-like responses and can result in slow response kinetics. We also explore the idea that substrate degradation can serve as an effective phosphatase. Assuming that degradation is unsaturable, this mechanism could insulate substrates from crosstalk, but it would also preclude ultrasensitive responses and would require very high substrate turnover to achieve rapid dephosphorylation kinetics. Finally, we show that adaptor subunits, such as those found on phosphatases like PP2A, can provide effective insulation against phosphatase crosstalk, but only if their binding to substrates is uncoupled from their binding to the catalytic core. Analysis of the interaction network of PP2A's adaptor domains reveals that although its adaptors may isolate subsets of targets from one another, there is still a strong potential for phosphatase crosstalk within those subsets. Understanding how phosphatase crosstalk and the insulation mechanisms described here impact the function and evolution of signaling networks represents a major challenge for experimental and computational systems biology.

## INTRODUCTION

Signaling networks allow cells to sense changes in their environment and respond adaptively. One of the most common motifs in eukaryotic signaling networks consists of a kinase that phosphorylates another protein in the network. Phosphorylation often alters the function of the target protein; for instance, the target itself might be a kinase that only becomes active when it is phosphorylated. A second enzyme, called a phosphatase, generally catalyzes the removal of the phosphoryl group. Although they are generally less well studied than kinases, phosphatases play a crucial role in controlling the phosphorylation levels of target proteins and thus the response of signaling networks to external stimuli (1–3).

Metazoan signaling networks are often very complex, exhibiting a high degree of crosstalk where many enzymes are shared among multiple pathways (3–9). Crosstalk studies have generally concentrated on the interactions made by kinases, and the potential for phosphatases to contribute to signaling complexity has been largely overlooked (10). For instance, when developing mathematical models of signaling networks, and a phosphatase has not yet been identified for a particular phosphoprotein, investi-

gators will often add an anonymous, independent, and often unsaturable phosphatase to the model to fill in the gap (11–17). This approach obviously ignores any contribution that phosphatase-mediated crosstalk might make to the behavior of the network.

We recently used a set of mathematical models to explore whether phosphatases acting on multiple substrates could impact signaling dynamics. We found that the responses of substrates to incoming signals can be strongly coupled if they share a phosphatase (2). In particular, we considered a case in which two different substrate proteins in the network,  $S_1$  and  $S_2$ , have two completely independent kinases but share a single phosphatase. In this case, signals that activate only one of the kinases can cause both substrates to respond in a switch-like manner. This occurs because the phosphorylated substrate (say  $S_2^*$ ) will act as a competitive inhibitor of the phosphatase, causing the other substrate to become active. This significantly increases  $S_1$  phosphorylation even when the kinase specific to that substrate has very low activity.

Although more than 500 distinct kinases have been identified in the human genome, there are only ~150 phosphatases (18,19). Thus, there are not even enough phosphatases to assign a unique one to each kinase, let alone to each unique substrate in the network, as has commonly been assumed in modeling studies (11–17). In fact, it is well

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established that phosphatases are often inherently promiscuous: well-characterized phosphatases have been shown to act on tens, if not hundreds, of substrates (20–22). This fact suggests a potential paradox: since by virtue of their relatively small numbers, each phosphatase must act on a large number of targets, it is unclear how the cell avoids rampant phosphatase-mediated crosstalk between distinct parts of the network (2).

In this work, we used mathematical models to investigate a variety of mechanisms that cells could deploy to prevent shared phosphatases from resulting in unwanted crosstalk. These models focus on a simplified scenario in which substrates share a single phosphatase but are otherwise unrelated (e.g., Fig. 1 A). Although this ignores crosstalk at the kinase level (2–9) and the fact that phosphatase activity is often itself regulated by the signaling network (23,24), it allows us to isolate a particular source of crosstalk and characterize various mechanisms the cell might use to prevent it.

Since phosphatases can only couple substrate responses if they are saturated (2), one natural approach to limiting the impact of the phosphatase crosstalk would be to evolve phosphatases with Michaelis constants ( $K_M$ ) so large that they essentially cannot be saturated by their substrates. We showed that in this scenario, the substrates can no longer respond ultrasensitively to incoming signals, and the phosphatases become highly inefficient enzymes that must be expressed at high levels to ensure rapid substrate responses. A second mechanism that cells might employ involves disposing of specific phosphatases altogether and instead employing degradation of the substrate as a means of removing phosphorylated molecules from the system. This alternative approach to effective dephosphorylation would have the benefit of reducing the required number of targets per phosphatase, decreasing the potential for phosphatase

coupling in the rest of the network. We found that this dephosphorylation mechanism also cannot generate ultrasensitive responses. Additionally, in order for degradation to yield rapid response kinetics, the phosphorylated substrate would have to be highly unstable, with half-lives on the order of tens of minutes, which would involve high energetic costs to the cell.

Phosphatase promiscuity is likely a larger problem for serine/threonine phosphatases than it is for tyrosine phosphatases (20,21,25–27). Interestingly, serine/threonine phosphatases such as PP2A often act as holoenzymes comprised of a catalytically active subunit, a scaffolding subunit, and an adaptor subunit that recruits specific substrates to the complex (20,27–33). Using our models, we demonstrated that these adaptor subunits can insulate signaling pathways from phosphatase crosstalk while still allowing each independent substrate to respond ultrasensitively. We found that the ability of adaptor subunits to insulate signals between different substrates depends upon the manner of adaptor binding. In particular, the adaptor must be able to bind the substrate independently of whether or not it is already bound in an active holoenzyme complex with the catalytic subunit. Focusing on the example of PP2A, it is likely that the substrate specificities of its adaptor subunits have evolved to functionally couple subsets of targets within the signaling network (20,27–33).

Overall, our work demonstrates that although certain mechanisms can allow cells to avoid widespread phosphatase crosstalk, each of these mechanisms involves a set of functional trade-offs that likely dictate which mechanism will evolve in any given situation. Although these mechanisms almost certainly reduce the overall level of crosstalk in the cell, our analysis of the PP2A example indicates that at least some phosphatase coupling likely remains. Characterizing the functional role of phosphatase-mediated

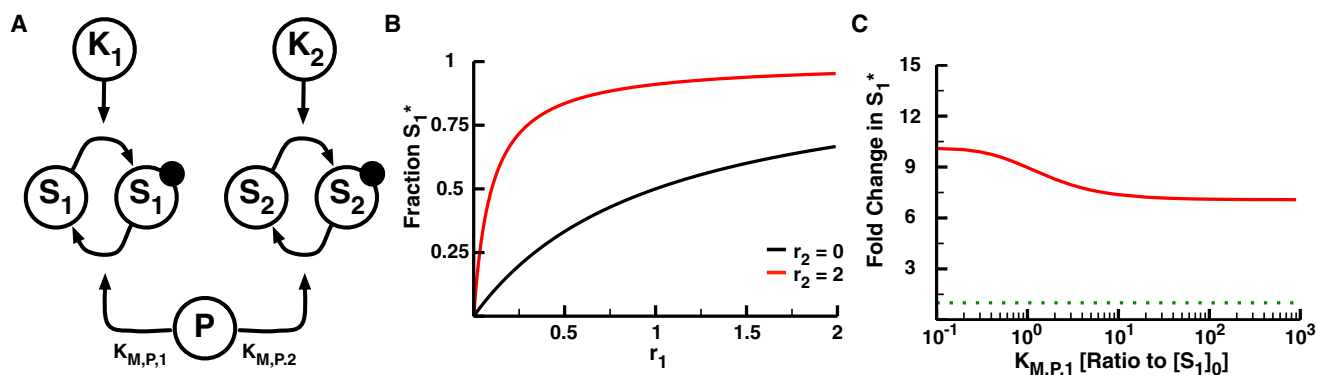


FIGURE 1 The 2-Kinase/1-Phosphatase loop. (A) Two independent kinases ( $K_1$  and  $K_2$ ) phosphorylate their respective substrates ( $S_1$  and  $S_2$ ). The solid circles indicate the phosphorylation of the substrates. A single shared phosphatase,  $P$ , dephosphorylates both substrates. (B) The fraction of phosphorylated  $S_1$  (denoted as  $S_1^*$ ) as a function of response parameter  $r_1$  when  $r_2 = 0$  (black) and  $r_2 = 2$  (red).  $r_1$  and  $r_2$  represent the ratio of the maximum velocity of the respective kinase to the maximum velocity of the phosphatase and are the dominant response parameters for the system. The initial concentrations of both  $S_1$  and  $S_2$  are set at  $10 \mu\text{M}$ ,  $K_{M,P,1}$  at  $1 \text{ mM}$ , and  $K_{M,P,2}$  at  $1 \mu\text{M}$ . As such,  $S_1$  does not saturate either the kinase  $K_1$  or phosphatase, whereas  $S_2$  saturates  $K_2$  and  $P$ . Note the increase in phosphorylated  $S_1$  in response to activation of the second loop, as described previously for this motif (2). (C) The fold change in  $S_1^*$  as a function of  $K_{M,P,1}$ . The fold change in  $S_1^*$  is calculated as the fraction  $S_1^*$  at  $r_2 = 2$  divided by the fraction  $S_1^*$  at  $r_2 = 0$ . The dotted green line represents no change in  $S_1^*$ . To see this figure in color, go online.

crosstalk in shaping network dynamics represents a major experimental and theoretical challenge in systems biology.

## MATERIALS AND METHODS

Our models of 2-Kinase/1-Phosphatase dynamics, the corresponding systems of ordinary differential equations (ODEs), and details regarding the simulations are described in the [Supporting Material](#). We used the CVODE library from SUNDIALS (34) to numerically integrate the systems of ODEs. The analytical solutions and subsequent derivations are also provided in the [Supporting Material](#).

The half-lives of signaling proteins were taken from a published data set of the half-lives of proteins in mouse C2C12 cells (35). We analyzed the UniProt entry for each protein in this data set and checked for Gene Ontology (GO) annotations describing the protein as being a phosphoprotein. We also obtained the density curve for signaling protein half-lives using the default density estimator in R (36).

## RESULTS

### The promiscuity of phosphatases

As mentioned above, kinases vastly outnumber phosphatases in the human genome (18,19). To characterize the generality of this kinase/phosphatase mismatch across different species, we searched the UniProt database for ratios of kinases to phosphatases, and phosphoproteins to phosphatases (see [Supporting Material](#) for details) (37). We found that for most eukaryotes, there is no way to achieve a single, independent phosphatase per kinase, let alone substrate (as is often assumed, if implicitly, in modeling studies) (27). These findings are consistent with a variety of experimental studies in which phosphatases were shown to target tens to hundreds of phosphoproteins (20,21,25–27).

We previously demonstrated that phosphatases acting on multiple substrates could contribute to network crosstalk. Using a mathematical model in which a phosphatase is shared between two substrates with independent kinases (diagrammed in [Fig. 1 A](#)), one can show that the shared phosphatase couples the responses of the substrates so that the activation of one kinase increases the phosphorylation of both substrates through phosphatase inhibition (2). It is straightforward to derive the fraction of phosphorylated substrate  $S_1$  at steady state for this system:

$$S_1^* = \frac{(r_1 - 1) - (K_{K,1} + r_1 \alpha_{P,1} K_{P,1}) + \sqrt{((r_1 - 1) - (K_{K,1} + r_1 \alpha_{P,1} K_{P,1}))^2 + 4(r_1 - 1)r_1 \alpha_{P,1} K_{P,1}}}{2(r_1 - 1)}, \quad (1)$$

where  $S_1^* \equiv [S_1^*]/[S_1]_0$  is the mole fraction of phosphorylated substrate  $S_1$ ;  $K_{K,1} \equiv K_{M,K,1}/[S_1]_0$  and  $K_{P,1} \equiv K_{M,P,1}/[S_1]_0$  are the Michaelis constants of the substrate  $S_1$  for the kinase  $K_1$  and the shared phosphatase  $P$  divided by the total

concentration of  $S_1$ ; and  $r_1 \equiv k_{cat,K,1}[K_1]_0/k_{cat,P,1}[P]_0$ . Since protein concentrations remain constant over relevant time-scales,  $r_1$  serves as the response parameter that drives  $S_1$  phosphorylation (2,38). The  $\alpha_{P,1} \equiv 1 + [S_2^*]/K_{M,P,2}$  term represents the influence of  $S_2$  on the phosphorylation of  $S_1$ . The solution for the fraction of phosphorylated  $S_2$  at steady state is the same as Eq. 1 with different indices (e.g.,  $r_2 \equiv k_{cat,K,2}[K_2]_0/k_{cat,P,2}[P]_0$ ). Upon activation of the second loop ( $K_2$ - $S_2$ ), the phosphorylated  $S_2$  acts as a competitive inhibitor of the phosphatase, increasing  $\alpha_{P,1}$  if  $[S_2^*]$  is large relative to  $K_{M,P,2}$ . This phosphatase inhibition results in an increase in  $S_1$  phosphorylation ([Fig. 1 B](#)). The difference in  $S_1$  phosphorylation due to  $K_2$ - $S_2$  activity can be illustrated as the fold change in  $S_1$  phosphorylation upon  $K_2$  stimulation, defined here as the concentration of phosphorylated  $S_1$  at  $r_2 = 2$  divided by the concentration of phosphorylated  $S_1$  at  $r_2 = 0$ . Given this definition, we observe up to a 10-fold increase in phosphorylation of the first substrate at low values of  $r_1$ . Additionally, making  $P$  a poor phosphatase for  $S_1$  by increasing  $K_{M,P,1}$  does not remove the crosstalk, since  $\alpha_{P,1}$  depends only on the saturation of the phosphatase by  $S_2^*$  ([Fig. 1 C](#)). Although the results in [Fig. 1](#) focus on a case in which there is a single competing substrate  $S_2$ , multiple substrates can collectively saturate the phosphatase, leading to indirect activation of  $S_1$  even when none of the competing substrates is at a high enough concentration to individually saturate the enzymes (see [Supporting Material](#) for details) (2).

The findings above indicate that on average, phosphatases are quite promiscuous, with tens to hundreds of substrates, and that this promiscuity can cause indirect activation of substrates due to phosphatase crosstalk ([Fig. 1](#)). In the following sections, we characterize a number of possible mechanisms that cells could use to minimize the impact of this crosstalk on the response of the network.

### Removing coupling with unsaturatable phosphatases

Since phosphatase coupling is dependent upon the collective saturation of the phosphatase by its substrates, it follows that an unsaturatable phosphatase could insulate substrate responses. To investigate the effects of phosphatase saturation

on crosstalk, we simultaneously increased both  $K_{M,P,1}$  and  $K_{M,P,2}$  in our 2-Kinase/1-Phosphatase model ([Fig. 1 A](#)). When the  $K_{M,P}$  values are smaller than the total concentrations of the substrates, we see that the phosphorylation of  $S_1$

at a low value of  $r_1$  is increased  $\sim 10$ -fold upon activation of the second kinase. As the phosphatase  $K_M$  values are increased, however, the fold increase in  $S_1$  phosphorylation drops until it reaches one, indicating that  $S_1$  becomes insensitive to  $K_2$  activity at  $K_{M,P}$  values above  $\sim 10$  times the total substrate concentration (Fig. 2 A).

The insulation provided by an unsaturated phosphatase comes at the cost of the loss of an ultrasensitive response of the substrates to incoming signal (2,39). The unsaturated phosphatase can no longer operate at its maximum velocity, and as such it takes very little active kinase to phosphorylate a significant fraction of  $S_1$  at steady state (Fig. 2 B). One can understand these results by treating the system analytically. With an unsaturated phosphatase, the solution for the fraction of phosphorylated substrate becomes

$$S_1^* = \frac{(1 + K_{K,1} + r_1 K_{P,1}) - \sqrt{(1 + K_{K,1} + r_1 K_{P,1})^2 - 4r_1 K_{P,1}}}{2} \quad (2)$$

Note the lack of the  $\alpha$  inhibition term from the shared phosphatase, since  $K_{M,P,2} \gg [S_2]_0$ ,  $\alpha_{P,1} \approx 1$ , resulting in a lack of phosphatase-mediated crosstalk. One can show that the analytical solution in Eq. 2 is strictly hyperbolic in  $r_1$  regardless of the values of the kinetic parameters, confirming the lack of any possible ultrasensitive response (see Supporting Material for derivation) (39).

Another complication with an unsaturated phosphatase is the timescale on which it can dephosphorylate a pool of substrate molecules. To explore this issue, we initialized our system with a fully phosphorylated pool of  $S_1$  mole-

cules and no activity for the second kinase (i.e.,  $r_2 = 0$ ). This represents a pathway that has been fully activated by an incoming signal. We then ran this system with absolutely no kinase activity ( $r_1 = 0$ ) to simulate the system after the removal of input. When  $K_{M,P}$  is small, it takes  $< 1$  h to fully dephosphorylate all substrate. However, it takes longer as the phosphatase becomes unsaturated; for example, it takes  $> 100$  h to completely dephosphorylate the substrate when the  $K_{M,P}$  is 100 times the total substrate concentration (Fig. 2 C). When  $K_{M,P}$  is large, the time it takes to dephosphorylate half the substrates ( $t_{1/2}$ ) follows

$$t_{1/2} = \frac{\log 2 \cdot K_{M,P,1}}{k_{cat,P}[P]_0} \quad (3)$$

The dependence of  $t_{1/2}$  on  $K_{M,P,1}$  is thus linear, which can result in very long timescales when the enzyme becomes highly unsaturated. In cases where fast dephosphorylation of the substrate is important (say, in tightly controlling the duration of a cellular response, or when fast oscillations are necessary), the system can compensate for this increase in timescale by expressing more phosphatase (Fig. 2 D).

Interestingly, overexpression of the phosphatase alone can insulate substrate responses even when the  $K_M$  values are small, as long as the concentration of phosphatase becomes so high that the traditional Michaelis-Menten assumption that the enzymes are at much lower concentration than their substrates is broken (1). In this regime, however, the enzymes tend to sequester their substrates, reducing the concentration of unbound, phosphorylated

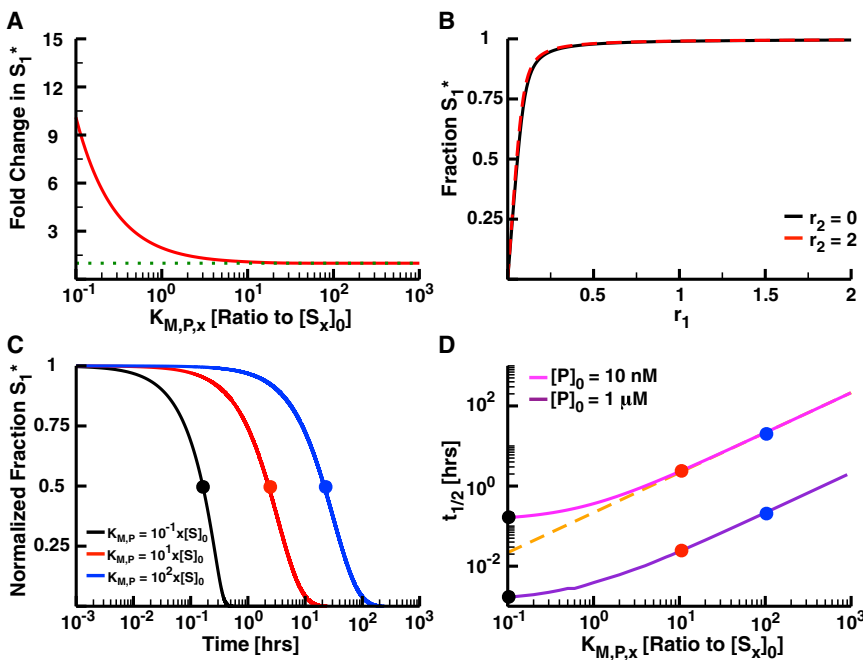


FIGURE 2 Removing coupling with unsaturable phosphatases. (A) The fold increase in  $S_1$  phosphorylation as a function of the  $K_{M,P}$  of the shared phosphatase for both substrates in a 2-Kinase/1-Phosphatase loop. (B) The fraction  $S_1^*$  as a function of  $r_1$  at  $r_2 = 0$  (black) and  $r_2 = 2$  (red) when  $K_{M,P} = 10 \times [S]_0$ . Note that there is very little difference between these curves. (C) The normalized fraction of phosphorylated substrate  $S_1^*$  as a function of time after the removal of input signal. In these simulations, the concentration of  $S_2$  and  $K_2$  are 0. The systems were allowed to run to steady state at high  $K_I$  activity ( $r_1 = 2$ ); at  $t = 0$ , the activity of the kinase was set to 0 (i.e.,  $r_1 = 0$ ). The y axes were normalized by  $y_1 = (y - \min_y) / (\max_y - \min_y)$ , where  $\min_y$  is the fraction  $S_1^*$  at  $r_1 = 0$  and  $\max_y$  is the fraction  $S_1^*$  at  $r_1 = 2$  at steady state. (D) The half-life of  $S_1$  phosphorylation as a function of  $K_{M,P}$  with two total concentrations of  $P$  (10 nM, green, and 1  $\mu$ M, purple). Note that the black, red, and blue dots are shown to illustrate the relationship between (D) and (C). The dashed orange line shows the linear approximation of  $t_{1/2}$  for highly unsaturated phosphatases ( $t_{1/2} = K_{M,P}/k_{cat,P}[P]_0$ ). To see this figure in color, go online.

substrate available to participate in downstream reactions within the network (see Fig. S1). Thus, while phosphatase overexpression can result in insulation and reduce dephosphorylation timescales (Fig. 2 D), it can also reduce the capacity of the system to respond to incoming signals. As a result, many phosphatases (such as Msg5, which dephosphorylates the MAPK Fus3 in yeast) are at least an order of magnitude lower in concentration than their substrates (40,41).

An increased phosphatase  $K_M$  is the only mechanism cells could use to unsaturate the phosphatase. Previous experimental studies have demonstrated that binding of phosphorylated sites by SH-2 domain-containing or 14-3-3 proteins can shield the phosphoprotein from dephosphorylation, creating a reservoir phosphorylated substrate (42,43). The presence of such proteins would effectively reduce the concentration of substrate available to the phosphatase (and thus its saturation level) without influencing the saturation of the kinase. To consider the effects of this shielding phenomenon, we added substrate-specific reservoir proteins to our 2-Kinase/1-Phosphatase model. As expected, these reservoir proteins do insulate the substrates from crosstalk even when the  $K_M$  values are relatively small (see Fig. S2, A and B). However, since the phosphatase is effectively unsaturated, the substrates always respond hyperbolically to inputs. The dephosphorylation kinetics are also very slow in the presence of high concentrations of reservoir proteins, since the phosphatase must essentially wait for a substrate to be unbound long enough for it to bind and catalyze the dephosphorylation reaction (Fig. S2 C).

Thus, although unsaturating the phosphatase can insulate a substrate from the response of another substrate, this mechanism clearly involves a set of trade-offs. For one, none of the substrates can respond in a switch-like manner to incoming signals, so this mechanism cannot be deployed in cases where ultrasensitive responses are crucial (11,16,44–47). In addition, achieving fast dephosphorylation timescales may require high levels of phosphatase expression, which may become impractical (or limit the capacity of the system to respond at all) in cases where the  $K_M$  needed to achieve insulation is very large (Figs. 2 D and S1 A). These trade-offs likely limit the number of cases in which insulation via an unsaturated phosphatase represents an evolutionarily favored mechanism.

### Degradation as a phosphatase substitute

The vast majority of work on modeling signal transduction has assumed that dephosphorylation occurs through the catalytic activity of a phosphatase (1,2,11–17). Although there are many clear cases in which phosphatases play this role, it is also possible for substrate degradation to serve as a dephosphorylation mechanism. The idea in this case is straightforward: phosphoproteins are synthesized in their

unphosphorylated state, but both the unphosphorylated and phosphorylated states of the protein may be lost from the system due to degradation. When the total protein concentration (regardless of state) remains constant in time, the effect of synthesis and degradation effectively amounts to a first-order dephosphorylation term (see the Supporting Material for details). Substrates that rely on degradation as their phosphatase would not need a separate phosphatase enzyme, which would reduce the number of substrates upon which each phosphatase would have to act.

To characterize degradation as a phosphatase substitute, we built a mathematical model with a substrate responding to a single kinase without a phosphatase. In this model, we assume that the degradation process is completely unsaturable. As a result, the degradation terms are all taken to be first order, and there is no degradation- or phosphatase-mediated crosstalk between pathways. In some cases, phosphorylation of a protein changes its half-life; for instance, the phosphorylated state of the protein may be less stable than the unphosphorylated state (48,49). To capture this possibility in our model, the unphosphorylated and phosphorylated substrates are degraded at different rates ( $k_{deg,U}$  and  $k_{deg,P}$  for the unphosphorylated and phosphorylated states, respectively). When the substrate is bound to a kinase, we assume that the kinase falls off the complex as the substrate is degraded. Unphosphorylated substrate is synthesized at a rate necessary to maintain a constant total substrate concentration at steady state (see Supporting Material for details). To parameterize the model, we obtained a range of half-lives by using UniProt to identify phosphoproteins from a published data set of protein half-lives in mouse C2C12 cells (35,37). Phosphoproteins in this data set have half-lives ranging from ~10 to 187 h, with a median of 31 h (Fig. 3 A). We used these values to set the range of biologically relevant degradation rates for the substrates.

We then ran this model to steady state using each of the phosphoprotein half-lives for a range of kinase concentrations. For these initial simulations, we assumed that the phosphorylation state does not influence the degradation rate (i.e.,  $k_{deg,U} = k_{deg,P}$ ). The response of the substrate to incoming signal is dependent on the half-life of the substrate (e.g., more stable phosphoproteins are more highly phosphorylated than less stable phosphoproteins; Fig. 3 B). However, substrates with any half-life in the mouse data set become completely phosphorylated when the kinase concentration is  $\geq 1$  nM, making these substrates highly sensitive to incoming signal. Note that the catalytic rate of the kinase in these simulations is  $0.9 \text{ s}^{-1}$ , which is close to the kinase catalytic rates that have been determined experimentally (50,51). Faster kinase catalytic rates would further reduce the kinase concentrations necessary to completely phosphorylate substrates with the observed half-lives.

To understand how the different degradation rates may affect substrate phosphorylation and sensitivity, we derived

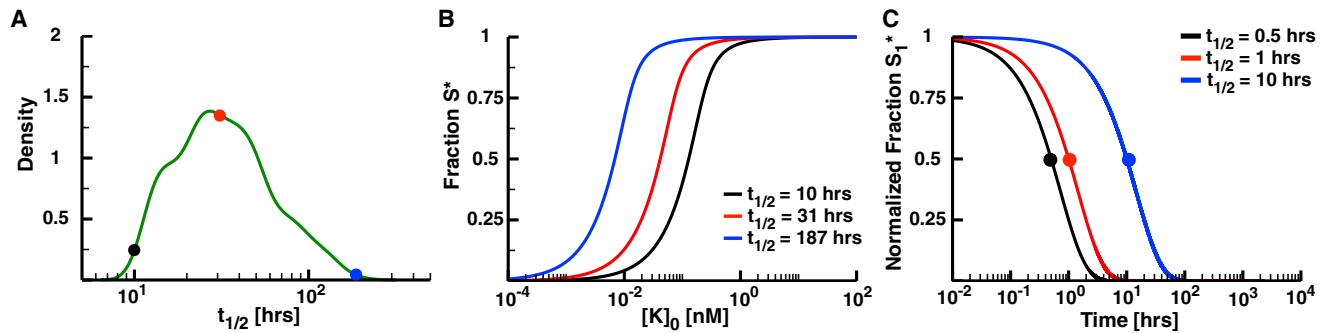


FIGURE 3 Degradation as a phosphatase. (A) The range of signaling protein half-lives in mouse C2C12 cells (35). The minimum reported half-life is 10 h (black dot), maximum is 187 h (blue), and median is 31 h (red). (B) The fraction  $S^*$  as a function of the concentration of active kinase,  $[K]_0$ . Note that the model for these panels includes only the kinase and degradable substrate.  $[K]_0$  directly affects the parameter  $r_{deg}$ , which represents the ratio of the maximum velocity of the kinase to the maximum velocity of degradation ( $k_{deg} \times [S]_0$ ). The value of  $r_{deg}$  will thus change for a constant concentration of kinase as the half-life changes. As such, any  $r_{deg}$  for a protein with a longer half-life will involve a lower concentration of kinase than that for a protein with a shorter half-life. (C) The normalized fraction  $S^*$  as a function of the time after setting kinase activity to 0 for substrates with half-lives of 0.5, 1, and 10 h. To see this figure in color, go online.

a solution for the fraction of phosphorylated  $S^*$  in this model:

$$S_1^* = \frac{(1 + r_{deg} + K_{deg}) - \sqrt{(1 + r_{deg} + K_{deg})^2 - 4r_{deg}}}{2}, \quad (4)$$

where  $r_{deg} \equiv k_{cat,K}[K]_0/k_{deg,P}[S]_0$  is the ratio of the maximum velocity of the kinase to the maximum velocity of substrate degradation. This solution includes a modified Michaelis constant  $K_{M,deg} \equiv (k_{-,K} + k_{cat,K} + k_{deg,U})/k_{+,K}$ , taking into account the degradation of the kinase-bound substrate, which is divided by the total substrate concentration to obtain  $K_{deg}$ . As expected based on our results for an unsaturable phosphatase, one can show that the phosphorylation response of the substrate in this model can only be hyperbolic in  $r_{deg}$  (see Supporting Material for the derivation). As such, a signaling pathway that relies on degradation to remove phosphorylated substrate could never respond ultrasensitively to incoming signals.

Finally, the timescales required to degrade a pool of phosphorylated substrate based on the half-lives of phosphoproteins from the C2C12 data set would be very long. To demonstrate this, we ran the model at high levels of kinase activity ( $r_{deg} = 2$ ) to steady state. We then set  $r_{deg} = 0$  to simulate the system after removal of the input. Even with the shortest substrate half-life, the system requires  $\sim 100$  h to completely dephosphorylate the substrate (Fig. 3 C). Complete dephosphorylation of the substrate in  $<1$  h would require the phosphorylated state to have a half-life on the order of minutes. If the system needs to recover quickly from incoming signals, utilizing degradation as a phosphatase subunit would likely be quite inefficient, requiring a very unstable phosphorylated substrate. Maintaining a reasonable concentration of total substrate would in turn require a high rate of protein synthesis, resulting in a high energetic cost for the cell. Thus, although degradation could reduce

the total substrate burden of phosphatases in the cell, it is likely to be employed only in cases where an ultrasensitive response is not necessary and when either slow dephosphorylation kinetics or the energetic costs of high protein turnover are functionally acceptable.

### Role of phosphatase regulatory subunits in pathway isolation

Serine phosphatases such as PP1 and PP2A exist as holoenzymes: the catalytic subunit of PP2A binds to a scaffold subunit, which makes up the catalytic core. This core can then bind to one of many possible regulatory adaptor subunits, with each adaptor recruiting PP2A to a specific set of substrates (Fig. 4 A) (20,27–33). A previous study examined the mechanism of targeting of substrates by a regulatory subunit, as well as the effects of concentration and binding constants (52). However, that study did not consider whether regulatory subunits might provide insulation from phosphatase-mediated crosstalk. To investigate the role of these adaptors in isolating different pathways that share the same phosphatase catalytic core, we built two models, one ordered and one unordered, for binding of the phosphatase adaptor subunits. In the ordered model, the phosphatase catalytic core can bind two one of two adaptor subunits, creating two distinct holoenzymes. This holoenzyme can then specifically bind to and dephosphorylate a cognate substrate (diagrammed in the inset of Fig. 4 B). This model represents the standard case in which the holoenzyme first assembles and then acts on its substrates (30,33).

We first tested the ordered model to determine whether the interaction specificity of adaptor subunits is sufficient to isolate the responses of different substrates. To examine how the concentrations of the adaptor subunits affect substrate phosphorylation, we ran this model to steady state with a range of total adaptor concentrations (keeping the

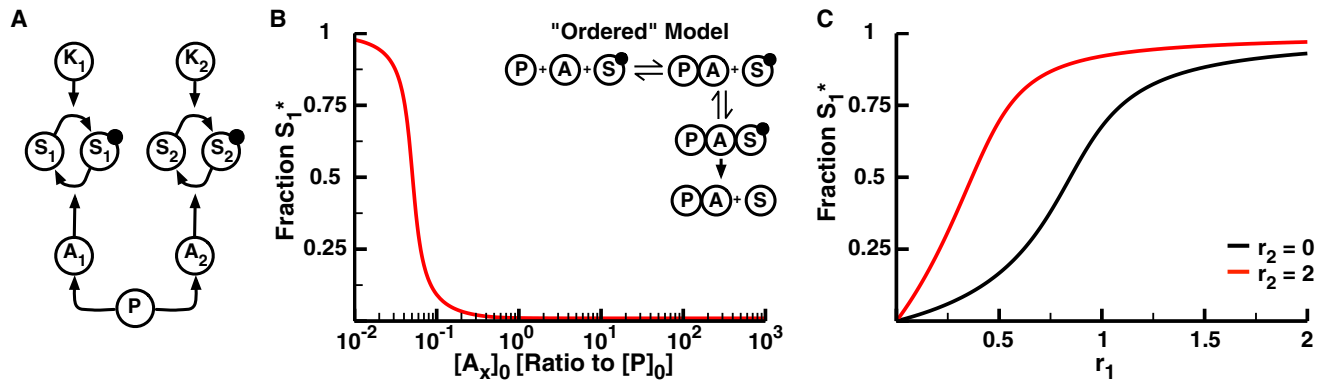


FIGURE 4 Ordered model of phosphatase regulatory adaptor subunits. (A) Diagram of the regulatory adaptor model. The previously described 2-Kinase/1-Phosphatase model (Fig. 1 A) was expanded to include two regulatory adaptor subunits,  $A_1$  and  $A_2$ , that target the phosphatase to substrates  $S_1$  and  $S_2$ , respectively. The phosphatase must bind an adaptor domain to bind and dephosphorylate either substrate. (B) The fraction  $S_1^*$  as a function of the initial concentration of each adaptor subunit with low levels of  $K_1$  activity ( $r_1 = 0.05$ ) and high levels of  $K_2$  activity ( $r_2 = 2$ ). The concentrations of both adaptor subunits are equal and are varied simultaneously. Inset: Diagram of the ordered model. The phosphatase must first bind to an adaptor subunit. This complex can then bind to and dephosphorylate a phosphorylated substrate. (C) The fraction  $S_1^*$  as a function of  $r_1$  at  $r_2 = 0$  (black) and  $r_2 = 2$  (red) with  $[A_x]_0 = 100$  nM. In this model, the adaptor subunits do not eliminate the potential for phosphatase-mediated crosstalk even though they are perfectly specific for their respective substrates. To see this figure in color, go online.

concentrations of the two adaptors equivalent, i.e.,  $[A_1] = [A_2]$ ) and measured the fraction of phosphorylated  $S_1$  at low levels of kinase 1 activation ( $r_1 = 0.05$ ) and high levels of kinase 2 activity ( $r_2 = 2$ ) (Fig. 4 B). In this model, a phosphatase can only act on its substrate when bound to an adaptor subunit. When the concentration of adaptors is less than that of the catalytic core, the total number of active phosphatases is thus limited by the adaptor concentration. As that concentration decreases, so does the concentration of active phosphatases, making the apparent value of  $r$  very high for both substrates and leading to an increase in  $S_1$  phosphorylation (Fig. 4 B). Once the concentration of the adaptors exceeds that of the catalytic core, however, the concentration of the adaptors has little influence on  $S_1$  phosphorylation (Fig. 4 B). At higher levels of kinase 1 activity, adaptor subunits provide no insulation between different substrates in this model:  $S_1$  phosphorylation increases considerably as  $r_2$  goes from 0 to 2 (Fig. 4 C). This indicates that although regulatory adaptor domains may help to target different phosphoproteins, their specificity cannot insulate substrates from phosphatase-mediated crosstalk in the ordered model. This is because the large pool of phosphorylated  $S_2$  makes it far more likely that the holoenzyme will remain intact in order to bind and further dephosphorylate  $S_2$ . As such, the catalytic core is prevented from disassociating from its  $S_2$ -specific adaptor subunit and forming the holoenzyme specific to  $S_1$ , decreasing the concentration of phosphatase available to act on  $S_1$ .

The unordered model differs in that the adaptor subunit can bind its specific phosphoprotein without being bound to the catalytic core first. The adaptor-substrate dimer can then bind the catalytic core, resulting in substrate dephosphorylation (diagrammed in the inset of Fig. 5 A). This removes the ability of one substrate to sequester the catalytic

core in the holoenzyme, since the core can dissociate from the regulatory adaptor subunit even in the presence of the phosphorylated substrate. At low concentrations of the adaptor subunits, the unordered model acts similarly to the ordered model (Fig. 5 A). However, as the adaptor concentration becomes very large, there is an increase in substrate phosphorylation. This is due to the prozone effect, i.e., as the concentration of adaptor increases, the system starts to produce many phosphatase-adaptor dimers and phosphorylated substrate-adaptor dimers. These dimers cannot bind to one another, which prevents the catalytic core from interacting with the substrate (41,52–57). Even though the prozone effect can influence the response at high adaptor concentrations, there is still a wide range of adaptor concentrations ( $>2$  orders of magnitude) that provide robust phosphatase activity (Fig. 5 A). The unordered model also provides effective pathway insulation since there is essentially no change in  $S_1$  phosphorylation as the second kinase switches between inactive and active ( $r_2 = 0$  and  $r_2 = 2$ , respectively) (Fig. 5 B). Additionally,  $S_1$  can respond ultrasensitively in  $r_1$ , although the apparent  $r_1$  is about half of what is expected since the adaptor subunits split the available phosphatase between the two substrates.

PP2A is known to interact with a large number of substrates, yet only  $\sim 18$  different regulatory subunits have been identified. To understand how this handful of subunits could allow the PP2A catalytic core to dephosphorylate so many substrates while maintaining a degree of pathway insulation, we built a network based on previously identified interactions between the regulatory subunits and the substrates of PP2A (Fig. 5 C) (20). This network depicts the interactions between 42 PP2A substrates and 18 regulatory subunits. Strikingly, only six of the substrates have interactions with more than one of the adaptors,





inherently creates phosphatases that are highly inefficient enzymes. As a result, the response of a substrate to upstream signals can no longer exhibit a switch-like, ultrasensitive character. Thus, it is unlikely that phosphatase inefficiency would be employed in cases where ultrasensitivity is a key component of the functional response (1,2,11,16,44–47). A second consequence of this inefficiency is slow dephosphorylation kinetics, which in turn means that high phosphatase expression levels are required before a particular substrate can quickly return to baseline activity levels after a signal is removed (Fig. 3, C and D). Other mechanisms that effectively desaturate the phosphatase, such as the expression of high levels of reservoir proteins that bind the phosphorylated state and prevent phosphatase binding, could also provide insulation but would have similar effects on the ultrasensitivity and dephosphorylation kinetics.

Protein degradation can also assume the role of an effective phosphatase, reducing the number of substrates that phosphatase enzymes might have to act upon in the cell. If we assume that degradation is both efficient and unsaturable, then degradation can indeed prevent phosphatase crosstalk between substrates. As with the inefficient phosphatase mechanism, however, the lack of an ultrasensitive response and slow dephosphorylation kinetics might represent an issue for this particular mechanism in some cases. Even the least stable phosphoprotein in mouse cells would still require >100 h to return to baseline after removal of the signal (Fig. 3 C). Using degradation to fill the role of a phosphatase would thus require a very high (and very costly) rate of protein turnover in cases where fast response kinetics are necessary.

The final mechanism we considered here involved separate regulatory subunits recruiting a catalytic core to particular substrates. Although these subunits can clearly provide substrate specificity, it is not clear that they can actually insulate those substrates from one another. Indeed, if the phosphatase preassembles into a holoenzyme before interacting with the substrate (which is essentially the classical picture for this sort of enzyme (20,27–32)), even perfectly specific regulatory subunits cannot prevent phosphatase-mediated crosstalk (Fig. 4). If the regulatory subunits can bind their substrates independently of binding the catalytic core, however, this mechanism could provide insulation while still maintaining the possibility of an ultrasensitive response and fast dephosphorylation kinetics (Fig. 5). Deploying this mechanism, however, would require the evolution and expression of a distinct regulatory subunit for every set of phosphoproteins that the cell needs to isolate, which could represent a costly and evolutionarily complex solution to the problem of phosphatase coupling.

Our findings thus demonstrate that cells have a considerable degree of flexibility in the mechanisms they might

use to insulate substrates from one another, despite the (relatively) small number of phosphatases in eukaryotic genomes. The question then arises: which of these mechanisms are deployed in any given situation, and to what extent are substrates truly isolated from one another? For instance, the well-characterized serine/threonine phosphatase PP2A acts as a holoenzyme with a regulatory subunit (20,27–32), and it is currently unclear whether the assembly of this holoenzyme follows the ordered or unordered model. Our results suggest that an experiment in which the adaptor concentration is increased (either directly in an *in vitro* setting or through overexpression *in vivo*) could establish which of these mechanisms is utilized by PP2A (Figs. 4 B and 5 A). Even if the assembly mechanism is unordered, PP2A-mediated crosstalk is still a strong possibility. Although the data presented in Fig. 5 C are certainly not complete (for instance, KSR and Akt share an additional regulatory subunit (63,64)), PP2A clearly does not have a distinct regulator for every substrate with which it interacts. Instead, these regulators have clearly evolved to interact with a specific subset of proteins, possibly coupling their responses in functionally meaningful ways (Fig. 5 C).

A major component of systems biology is the construction of formal mathematical or computational models of cellular regulatory systems, with the goal of understanding how cells process information from their environment and respond appropriately (65–68). In the case of complex eukaryotic signaling networks, a major barrier to this goal is the fact that dephosphorylation, whether by phosphatases or through some other mechanism, has been comparatively poorly characterized for most phosphoproteins in the network. The addition of anonymous and perfectly specific phosphatases to cover this gap may produce effective models of individual cascades or pathways (11–17), but it is unlikely that this practice will remain effective as larger, more genome-wide models of signaling networks are formulated. Experimentally determining the phosphatase structure of signaling systems and gaining a theoretical understanding of how that structure has evolved to generate and regulate crosstalk among pathways (Fig. 5 C) thus represents a major challenge for systems biology.

## SUPPORTING MATERIAL

Supporting Materials and Methods and two figures are available at [http://www.biophysj.org/biophysj/supplemental/S0006-3495\(14\)04755-9](http://www.biophysj.org/biophysj/supplemental/S0006-3495(14)04755-9).

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# Phosphatase specificity and pathway insulation in signaling networks

## Supporting Information

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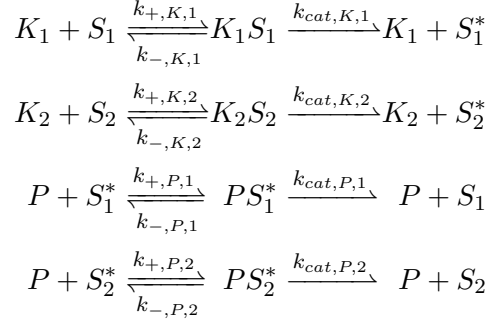
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# 1 Systems of Ordinary Differential Equations

## 1.1 2-Kinase/1-Phosphatase Loop with 2 Substrates

In order to characterize the effects of phosphatase saturation and competition on phosphatase-mediated crosstalk we used the 2-Kinase/1-Phosphatase Loop with 2 Substrates model that we have previously characterized [1]. The equations described below were derived and analyzed in our previous work; we include them here for completeness. The set of enzymatic reactions for the 2-Kinase/1-Phosphatase Loop with 2 Substrates model are:



Each contain three rates: the complex formation ( $k_+$ ), the rate of complex dissociation ( $k_-$ ), and catalytic rate ( $k_{cat}$ ). These reactions are diagrammed in Fig. 2A of the main text. The set of ODEs describing the free enzymes are:

$$\begin{aligned}
 \frac{d[K_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [K_1 S_1] k_{cat,K,1} - [K_1][S_1] k_{+,K,1} \\
 \frac{d[K_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [K_2 S_2] k_{cat,K,2} - [K_2][S_2] k_{+,K,2} \\
 \frac{d[P]}{dt} &= [P S_1^*] k_{-,P,1} + [P S_2^*] k_{-,P,2} + [P S_1^*] k_{cat,P,1} + [P S_2^*] k_{cat,P,2} - [P][S_1^*] k_{+,P,1} - [P][S_2^*] k_{+,P,2}
 \end{aligned}$$

The set of ODEs describing the free unphosphorylated substrates are:

$$\begin{aligned}
 \frac{d[S_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [P S_1^*] k_{cat,P,1} - [K_1][S_1] k_{+,K,1} \\
 \frac{d[S_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [P S_2^*] k_{cat,P,2} - [K_2][S_2] k_{+,K,2}
 \end{aligned}$$

The set of ODEs describing the free phosphorylated substrates are:

$$\begin{aligned}
 \frac{d[S_1^*]}{dt} &= [P S_1^*] k_{-,P,1} + [K_1 S_1] k_{cat,K,1} - [P][S_1^*] k_{+,P,1} \\
 \frac{d[S_2^*]}{dt} &= [P S_2^*] k_{-,P,2} + [K_2 S_2] k_{cat,K,2} - [P][S_2^*] k_{+,P,2}
 \end{aligned}$$

The set of ODEs describing the enzyme-substrate complexes are:

$$\begin{aligned}\frac{d[K_1S_1]}{dt} &= [K_1][S_1]k_{+,K,1} - [K_1S_1]k_{-,K,1} - [K_1S_1]k_{cat,K,1} \\ \frac{d[K_2S_2]}{dt} &= [K_2][S_2]k_{+,K,2} - [K_2S_2]k_{-,K,2} - [K_2S_2]k_{cat,K,2} \\ \frac{d[PS_1^*]}{dt} &= [P][S_1^*]k_{+,P,1} - [PS_1^*]k_{-,P,1} - [PS_1^*]k_{cat,P,1} \\ \frac{d[PS_2^*]}{dt} &= [P][S_2^*]k_{+,P,2} - [PS_2^*]k_{-,P,2} - [PS_2^*]k_{cat,P,2}\end{aligned}$$

For purposes of display we used the following rate parameters:

Parameter	Value
$k_{+,K,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,K,i}$	0.1-999.1 s <sup>-1</sup>
$k_{cat,K,i}$	0.9 s <sup>-1</sup>
$k_{+,P,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,P,i}$	0.1-999.1 s <sup>-1</sup>
$k_{cat,P,i}$	0.9 s <sup>-1</sup>

where  $i = 1$  or  $2$ . The ranges listed for the dissociation rates (i.e.  $k_{-,K,i}$ ) are used to set the  $K_M$ 's of the enzymes in different simulations. Note that, while the values of these parameters are not meant to describe any specific enzyme, they are within the range of values obtained for kinases and phosphatases experimentally [2-4].

We used the following initial conditions for all of our simulations:

Molecular Species	Initial Concentration
$K_1$	0-20 nM
$K_2$	0-20 nM
$P$	10 nM-1mM
$S_1$	10 $\mu$ M
$S_2$	0,10 $\mu$ M

with the remaining molecular species having initial concentrations of 0. The ranges of concentrations of  $K_1$  and  $K_2$  are used to vary the values of  $r_1$  and  $r_2$ .

This model was used to generate Fig. 1B and C of the main text. The concentration of  $K_2$  was set to 0 for  $r_2 = 0$  and to 20 nM for  $r_2 = 2$ . The concentration of  $K_1$  was set between 0-20 nM to vary  $r_1$  between 0 and 2. Both substrates are present at a concentration of 10 $\mu$ M. The values of  $k_{-,K,1}$  and  $k_{-,P,1}$  were both set to 999.1 s<sup>-1</sup> so that  $K_{M,K,1} = K_{M,P,1} = 100 \times [S_1]_0$ , while  $k_{-,K,2}$  and  $k_{-,P,2}$  were set to 0.1 s<sup>-1</sup> so that  $K_{M,P,2} = K_{M,K,2} = 100 \times [S_2]_0$ .

We also used this model to generate Fig. 2 of the main text. In Fig. 2A,  $K_{M,P,i}$  was set by changing the values of  $k_{-,P,1}$  and  $k_{-,P,2}$  between 0.1-999.1 s<sup>-1</sup>. The value of  $r_1$  was set in Fig. 2B by setting the concentration of  $K_1$  between 0-20 nM, and the concentration of  $K_2$  was set to 0 for

$r_2 = 0$  and to 20 nM for  $r_2 = 2$ . For Fig. 2C, we first ran the model with  $K_2 = S_2 = 0$  to steady state with the initial concentration of  $K_1$  at 20 nM ( $r_1 = 2$ ). We then removed all  $K_1$  molecules from the system. The bound  $S_1$  was added back to the concentration of unphosphorylated  $S_1$ . The simulations were then resumed to obtain the time courses visualized in Fig. 2C.  $K_{M,P}$  was set by using values of  $k_{-,P,2,1} = 0.1 \text{ s}^{-1}$ ,  $9.1 \text{ s}^{-1}$ , and  $99.1 \text{ s}^{-1}$ . The fraction of phosphorylated  $S_1^*$  was normalized so that  $\hat{S}_1^*(t) = (\max(S_1^*) - S_1^*(t)) / (\max(S_1^*) - \min(S_1^*))$ . Fig. 2D was obtained using the same procedures as in Fig. 3C, setting  $K_{M,P}$  by using values of  $k_{-,P,2,1} = 0.1-999.1 \text{ s}^{-1}$ . The half-time of  $S_1^*$  phosphorylation was obtained by finding the time  $t_{1/2} = t$  where  $\hat{S}_1^*(t) = 0.5$ . The total concentration of the phosphatase was set to either 10 nM (magenta curve), or 1  $\mu\text{M}$  (purple curve).

To test the effectiveness of an increased phosphatase concentration in insulating substrates against phosphatase crosstalk while maintaining strong  $K_{M,P,i}$  values, we set  $K_{M,P,i} = 1 \mu\text{M}$ ,  $r_1 = 0.05$ ,  $r_2 = 2$  and varied the concentration of the phosphatase from 10 nM to 1 mM (Fig. S1).

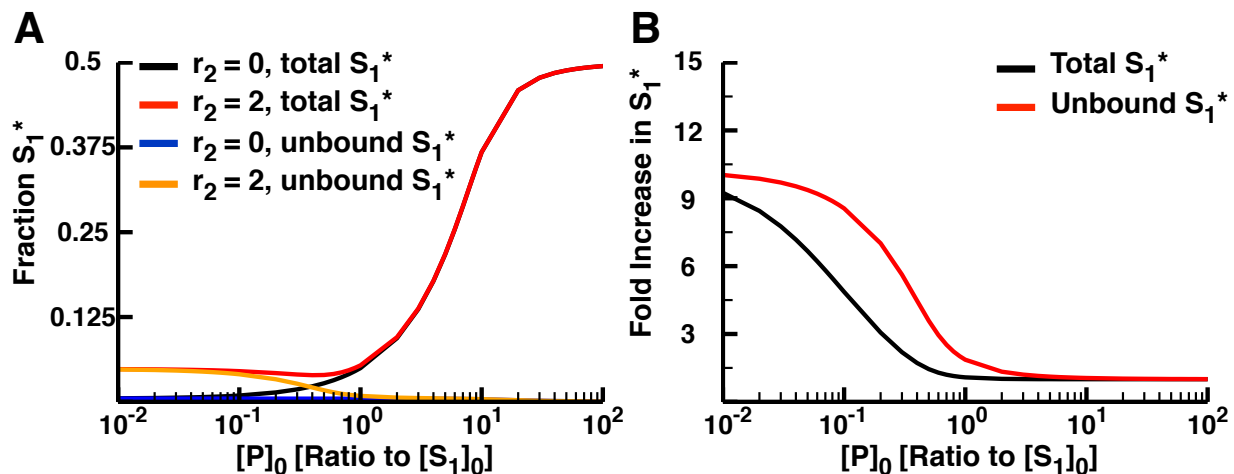
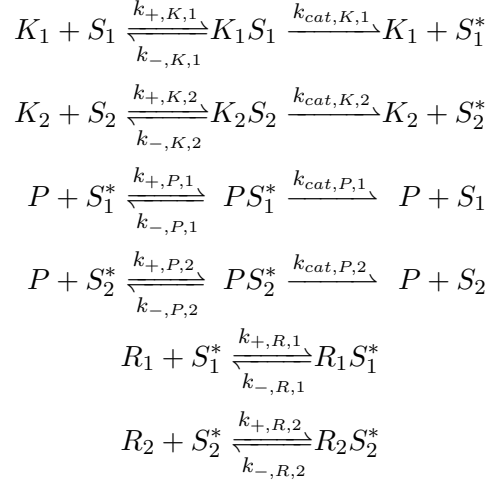


Figure S1: The effects of increased phosphatase concentration on substrate crosstalk with strong  $K_{M,P,i}$  (A) The fraction  $S_1^*$  as a function of the concentration of the phosphatase. The concentration of the kinase is increased in order to maintain the values of  $r_1$  and  $r_2$  with constant catalytic rates for different concentrations of the phosphatase. At low concentrations of  $P$ ,  $S_1$  phosphorylation is increased by activation of the  $S_2$  pathway, moving from  $r_2 = 0$  (black) to  $r_2 = 2$  (red). As  $P$  is expressed in concentrations greater than the substrates, the difference between the curves is removed. However, the fraction  $S_1^*$  is greatly increased. Additionally, the fraction of unbound  $S_1^*$  decreases with  $[P]_0$ , indicating that the increase in total fraction  $S_1^*$  is likely due to it being bound to the phosphatase. (B) The fold increase in  $S_1^*$  as a function of the concentration of the phosphatase. As stated above, the crosstalk between  $S_1$  and  $S_2$  is removed when the phosphatase is present in concentrations larger than those of the substrates.

## 1.2 2-Kinase/1-Phosphatase Loop with 2 Substrates and 2 Reservoir Proteins

In order to characterize the effects of reservoir proteins that bind to and shield phosphorylated substrates from dephosphorylated on phosphatase-mediated crosstalk we expanded the 2-Kinase/1-Phosphatase Loop with 2 Substrates model to include two substrate-specific reservoir

proteins,  $R_1$  and  $R_2$ . The set of enzymatic reactions for the model are:



The set of ODEs describing the free enzymes are:

$$\begin{aligned}
\frac{d[K_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [K_1 S_1] k_{cat,K,1} - [K_1][S_1] k_{+,K,1} \\
\frac{d[K_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [K_2 S_2] k_{cat,K,2} - [K_2][S_2] k_{+,K,2} \\
\frac{d[P]}{dt} &= [P S_1^*] k_{-,P,1} + [P S_2^*] k_{-,P,2} + [P S_1^*] k_{cat,P,1} + [P S_2^*] k_{cat,P,2} - [P][S_1^*] k_{+,P,1} - [P][S_2^*] k_{+,P,2}
\end{aligned}$$

The set of ODEs describing the free unphosphorylated substrates are:

$$\begin{aligned}
\frac{d[S_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [P S_1^*] k_{cat,P,1} - [K_1][S_1] k_{+,K,1} \\
\frac{d[S_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [P S_2^*] k_{cat,P,2} - [K_2][S_2] k_{+,K,2}
\end{aligned}$$

The set of ODEs describing the free phosphorylated substrates are:

$$\begin{aligned}
\frac{d[S_1^*]}{dt} &= [P S_1^*] k_{-,P,1} + [K_1 S_1] k_{cat,K,1} + [R_1 S_1^*] k_{-,R,1} - [P][S_1^*] k_{+,P,1} - [R_1][S_1^*] k_{+,R,1} \\
\frac{d[S_2^*]}{dt} &= [P S_2^*] k_{-,P,2} + [K_2 S_2] k_{cat,K,2} + [R_2 S_2^*] k_{-,R,2} - [P][S_2^*] k_{+,P,2} - [R_2][S_2^*] k_{+,R,2}
\end{aligned}$$



The set of ODEs describing the enzyme-substrate complexes are:

$$\begin{aligned}\frac{d[K_1S_1]}{dt} &= [K_1][S_1]k_{+,K,1} - [K_1S_1]k_{-,K,1} - [K_1S_1]k_{cat,K,1} \\ \frac{d[K_2S_2]}{dt} &= [K_2][S_2]k_{+,K,2} - [K_2S_2]k_{-,K,2} - [K_2S_2]k_{cat,K,2} \\ \frac{d[PS_1^*]}{dt} &= [P][S_1^*]k_{+,P,1} - [PS_1^*]k_{-,P,1} - [PS_1^*]k_{cat,P,1} \\ \frac{d[PS_2^*]}{dt} &= [P][S_2^*]k_{+,P,2} - [PS_2^*]k_{-,P,2} - [PS_2^*]k_{cat,P,2}\end{aligned}$$

The set of ODEs describing the free reservoir proteins are:

$$\begin{aligned}\frac{d[R_1]}{dt} &= [R_1S_1^*]k_{-,R,1} - [R_1][S_1^*]k_{+,R,1} \\ \frac{d[R_2]}{dt} &= [R_2S_2^*]k_{-,R,2} - [R_2][S_2^*]k_{+,R,2}\end{aligned}$$

The set of ODEs describing the reservoir-substrate complexes are:

$$\begin{aligned}\frac{d[R_1S_1^*]}{dt} &= [R_1][S_1^*]k_{+,R,1} - [R_1S_1^*]k_{-,R,1} \\ \frac{d[R_2S_2^*]}{dt} &= [R_2][S_2^*]k_{+,R,2} - [R_2S_2^*]k_{-,R,2}\end{aligned}$$

For purposes of display we used the following rate parameters:

Parameter	Value
$k_{+,K,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,K,i}$	0.1-999.1 s <sup>-1</sup>
$k_{cat,K,i}$	0.9 s <sup>-1</sup>
$k_{+,P,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,P,i}$	0.1-999.1 s <sup>-1</sup>
$k_{cat,P,i}$	0.9 s <sup>-1</sup>
$k_{+,R,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,R,i}$	0.01 s <sup>-1</sup>

where  $i = 1$  or  $2$ . The ranges listed for the dissociation rates (i.e.  $k_{-,K,i}$ ) are used to set the  $K_M$ 's of the enzymes in different simulations. Note that, while the values of these parameters are not meant to describe any specific enzyme, they are within the range of values obtained for kinases and phosphatases experimentally [2-4].

We used the following initial conditions for all of our simulations:

Molecular Species	Initial Concentration
$K_1$	0-20 nM
$K_2$	0-20 nM
$P$	10 nM, 1 $\mu$ M
$S_1$	10 $\mu$ M
$S_2$	0,10 $\mu$ M
$R_1$	0-100 $\mu$ M
$R_2$	0-100 $\mu$ M

with the remaining molecular species having initial concentrations of 0. The ranges of concentrations of  $K_1$  and  $K_2$  are used to vary the values of  $r_1$  and  $r_2$ .

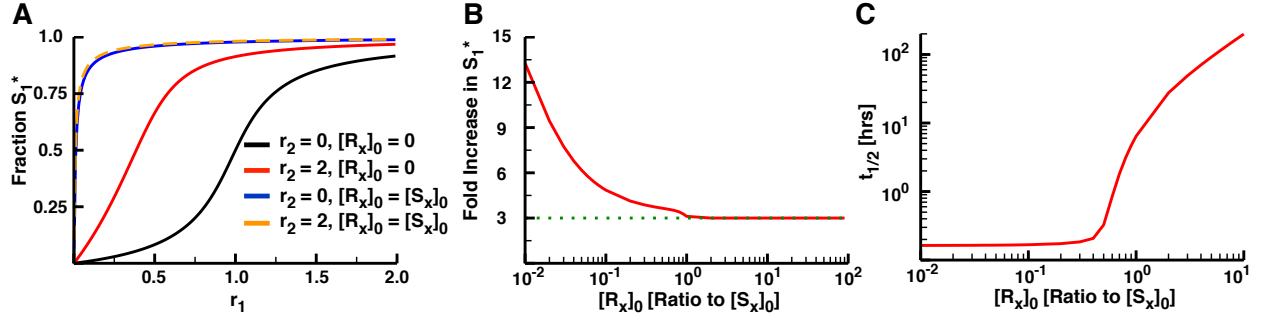


Figure S2: The effects of reservoir proteins on substrate crosstalk (A) The fraction  $S_1^*$  as a function of  $r_1$ . Without the reservoir proteins,  $S_1$  responds to signals from  $S_2$  (red versus black curves). The crosstalk is removed with the addition of the reservoir proteins (orange versus blue curves). Note, however, that the response becomes hyperbolic in  $r_1$ . (B) The fold increase in  $S_1^*$  as a function of reservoir protein concentration. As the concentrations of the reservoir proteins increases, the crosstalk between the substrates is gradually removed. (C) The half-life of  $S_1$  phosphorylation as a function of the concentration of reservoir proteins. Note that when the reservoir proteins are at stoichiometric or greater concentrations, the time required to completely dephosphorylate the substrates greatly increases.

### 1.3 2-Kinase/1-Phosphatase Loop with Many Substrates

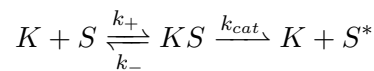
The expression for the fraction of phosphorylation  $S_1^*$  in a 2K1P loop in which kinase  $K_2$  and  $P$  act upon  $N$  substrates is similar to the expression for a 2K1P loop with 2 substrates we have previously derived (equation 1 in the main text) [1]. The only difference in this case is the value of  $\alpha_{P,1}$ , the inhibitory term that captures the effects of the competing substrates on the phosphatase. In a model with  $N$  substrates, this term becomes:

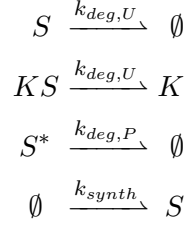
$$\alpha_{P,1} = 1 + \sum_{i \neq 1} \frac{[S_i^*]}{K_{M,P,i}} \quad (1.1)$$

Note that the value of  $\alpha_{P,1}$ , and thus the inhibition of the phosphatase, depends on the total saturation of the phosphatase across all its substrates. A set of substrates that all respond to the same signal can thus cause phosphatase crosstalk with other proteins in the network, even if none of those substrates is at high enough concentration to saturate the phosphatase individually.

### 1.4 1-Kinase/1-Substrate Model with Synthesis and Degradation

In order to characterize the effectiveness of synthesis and degradation of a substrate as a replacement for the phosphatase, we created the 1-Kinase/1-Substrate model with Synthesis and Degradation. The set of enzymatic reactions for this system are:





where  $k_{deg,U}$  and  $k_{deg,P}$  are the degradation rates of unphosphorylated and phosphorylated  $S$ . This model includes separate degradation rates for the unphosphorylated and phosphorylated substrate ( $k_{deg,U}$  and  $k_{deg,P}$ ), since phosphorylation of the substrate might either increase or decrease the stability of the protein. The ODE describing the free kinase is:

$$\frac{d[K]}{dt} = [KS]k_- + [KS]k_{cat} + [KS]k_{deg,U} - [K][S]k_+$$

The ODE describing the free unphosphorylated substrate is:

$$\frac{d[S]}{dt} = [KS]k_- + k_{synth} - [K][S]k_+ - [S]k_{deg,U} \quad (1.2)$$

The ODE describing the free phosphorylated substrate is:

$$\frac{d[S^*]}{dt} = [KS]k_{cat} - [S^*]k_{deg,P}$$

The ODE describing the concentration of kinase-substrate complex is:

$$\frac{d[KS]}{dt} = [K][S]k_+ - [KS]k_- - [KS]k_{cat} - [KS]k_{deg,U}$$

In order to understand the effects of degradation at steady state, note that the total substrate concentration is defined as:

$$[S]_T = [S] + [S^*] + [KS]$$

and the change in total substrate concentration can thus be written:

$$\frac{d[S]_T}{dt} = \frac{d[S]}{dt} + \frac{d[S^*]}{dt} + \frac{d[KS]}{dt}$$

At steady state,  $d[S]_T/dt = 0$ . By substituting the above ODEs and simplifying, we get:

$$\frac{d[S]_T}{dt} = k_{synth} - ([S] + [KS])k_{deg,U} - [S^*]k_{deg,P} = 0$$

We can then solve for  $k_{synth}$ :

$$k_{synth} = ([S] + [KS])k_{deg,U} + [S^*]k_{deg,P}$$

We can substitute this equation into the original differential equation for  $[S]$  (equation 1.2):

$$\frac{d[S]}{dt} = [KS]k_- - [K][S]k_+ + [KS]k_{deg,U} + [S^*]k_{deg,P} \quad (1.3)$$

Equation 1.3 is useful for two reasons. For one, there is a positive term in the equation corresponding to the degradation of phosphorylated substrate ( $[S^*]k_{deg,P}$ ). This term reflects the fact that, in order for  $[S]_T$  to remain constant at steady state, new, unphosphorylated substrate molecules must be synthesized to replace  $S^*$  molecules that are degraded. There is thus an “effective” dephosphorylation rate in this system where  $S^*$  molecules are converted to  $S$ , which corresponds mathematically to an unsaturateable first-order phosphatase. Secondly, we used equation 1.3 instead of 1.2 in our numerical integration, so  $[S]_T = [S]_0$  for all time; in other words, while the concentration of unphosphorylated and phosphorylated substrate might change in our simulations, the total concentration of substrate remains constant. This allows us to control total substrate levels by setting the initial substrate concentration, as we do in our other models. One could of course simulate equation 1.2 with a constant  $k_{synth}$  that allows total substrate concentration to vary with time; while such transients might have interesting effects on the system, we leave consideration of those effects to future work.

For purposes of display we used the following rate parameters:

Parameter	Value
$k_+$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_-$	0.1 s <sup>-1</sup>
$k_{cat}$	0.9 s <sup>-1</sup>
$k_{deg,U}$	1x10 <sup>-7</sup> - 1x10 <sup>-4</sup> s <sup>-1</sup>
$k_{deg,P}$	1x10 <sup>-7</sup> - 1x10 <sup>-4</sup> s <sup>-1</sup>

The ranges of  $k_{deg,U}$  and  $k_{deg,P}$  were used to vary the degradation rate of the substrate across simulations. The simulations started with the following initial concentrations:

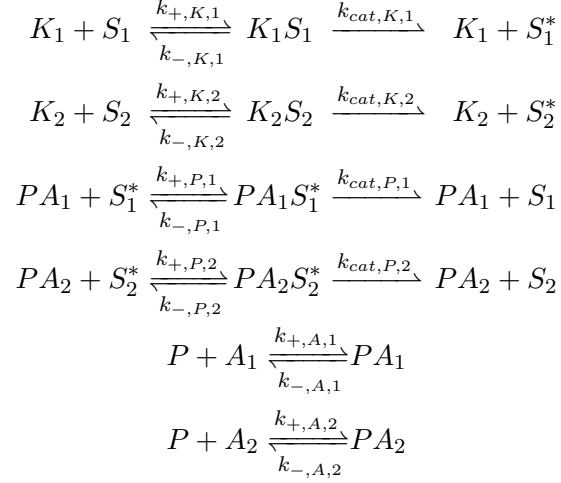
Molecular Species	Initial Concentration
$K$	0 - 1 nM
$S$	10 μM
$KS$	0

The range of  $K$  was used to set the value of  $r_{deg}$  across simulations.

This model was used to generate Fig. 3B and C of the main text. In Fig. 3B we used  $k_{deg,U} = k_{deg,P} = \log 2/t_{1/2}$  for  $t_{1/2} = 10, 31, \text{ and } 187$  hrs. The value of  $r_{deg}$  was set between 0 and 2 by changing the initial concentration of  $K$  so that  $[K] = (r_{deg}[S]_0 k_{deg,P})/k_{cat}$  (where  $r_{deg}$  was set to the desired value, see Section 3 for the derivation of  $r_{deg}$ ), which ends up giving a value between 0-0.186 nM. In Fig. 3C we first ran the model to steady state with an initial concentration of  $K$  at 0.186 nM ( $r_{deg} = 2$ ). All kinase molecules were then removed from the system, and the simulation continued in order to obtain the time courses shown in Fig. 3C.

## 1.5 2-Kinase/1-Phosphatase Loop with “Ordered” Phosphatase Adaptors

In order to characterize the effectiveness of phosphatase adaptors in insulating pathways from phosphatase crosstalk, we first developed the 2-Kinase/1-Phosphatase Loop with Ordered Phosphatase Adaptors. The set of enzymatic reactions for this model are:



where  $k_{-,A,i}$  represents the dissociation rate of the phosphatase-adaptor complex and  $k_{+,A,i}$  represents the association rate. The reactions that include a phosphatase molecule are diagrammed in the inset of Fig. 4A of the main text. The set of ODEs describing the concentration of free enzymes are:

$$\begin{aligned}
 \frac{d[K_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [K_1 S_1] k_{cat,K,1} - [K_1][S_1] k_{+,K,1} \\
 \frac{d[K_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [K_2 S_2] k_{cat,K,2} - [K_2][S_2] k_{+,K,2} \\
 \frac{d[P]}{dt} &= [P A_1] k_{-,A,1} + [P A_2] k_{-,A,2} - [P][A_1] k_{+,A,1} - [P][A_2] k_{+,A,2}
 \end{aligned}$$

The set of ODEs describing the concentration of free unphosphorylated substrates are:

$$\begin{aligned}
 \frac{d[S_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [P A_1 S_1^*] k_{cat,P,1} - [K_1][S_1] k_{+,K,1} \\
 \frac{d[S_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [P A_2 S_2^*] k_{cat,P,2} - [K_2][S_2] k_{+,K,2}
 \end{aligned}$$

The set of ODEs describing the concentration of free phosphorylated substrates are:

$$\begin{aligned}
 \frac{d[S_1^*]}{dt} &= [P A_1 S_1^*] k_{-,P,1} + [K_1 S_1] k_{cat,K,1} - [P A_1][S_1^*] k_{+,P,1} \\
 \frac{d[S_2^*]}{dt} &= [P A_2 S_2^*] k_{-,P,2} + [K_2 S_2] k_{cat,K,2} - [P A_2][S_2^*] k_{+,P,2}
 \end{aligned}$$

The set of ODEs describing the concentration of adaptor-bound phosphatase are:

$$\begin{aligned}\frac{d[PA_1]}{dt} &= [P][A_1]k_{+,A,1} + [PA_1S_1^*]k_{-,P,1} + [PA_1S_1^*]k_{cat,P,1} - [PA_1]k_{-,A,1} - [PA_1][S_1^*]k_{+,P,1} \\ \frac{d[PA_2]}{dt} &= [P][A_2]k_{+,A,2} + [PA_2S_2^*]k_{-,P,2} + [PA_2S_2^*]k_{cat,P,2} - [PA_2]k_{-,A,2} - [PA_2][S_2^*]k_{+,P,2}\end{aligned}$$

The set of ODEs describing the concentration of enzyme-substrate complexes are:

$$\begin{aligned}\frac{d[K_1S_1]}{dt} &= [K_1][S_1]k_{+,K,1} - [K_1S_1]k_{-,K,1} - [K_1S_1]k_{cat,K,1} \\ \frac{d[K_2S_2]}{dt} &= [K_2][S_2]k_{+,K,2} - [K_2S_2]k_{-,K,2} - [K_2S_2]k_{cat,K,2} \\ \frac{d[PA_1S_1^*]}{dt} &= [PA_1][S_1^*]k_{+,P,1} - [PA_1S_1^*]k_{-,P,1} - [PA_1S_1^*]k_{cat,P,1} \\ \frac{d[PA_2S_2^*]}{dt} &= [PA_2][S_2^*]k_{+,P,2} - [PA_2S_2^*]k_{-,P,2} - [PA_2S_2^*]k_{cat,P,2}\end{aligned}$$

For purposes of display we used the following rate parameters:

Parameter	Value
$k_{+,K,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,K,i}$	0.1 s <sup>-1</sup>
$k_{cat,K,i}$	0.9 s <sup>-1</sup>
$k_{+,P,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,P,i}$	0.1 s <sup>-1</sup>
$k_{cat,P,i}$	0.9 s <sup>-1</sup>
$k_{+,A,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,A,i}$	0.1 s <sup>-1</sup>

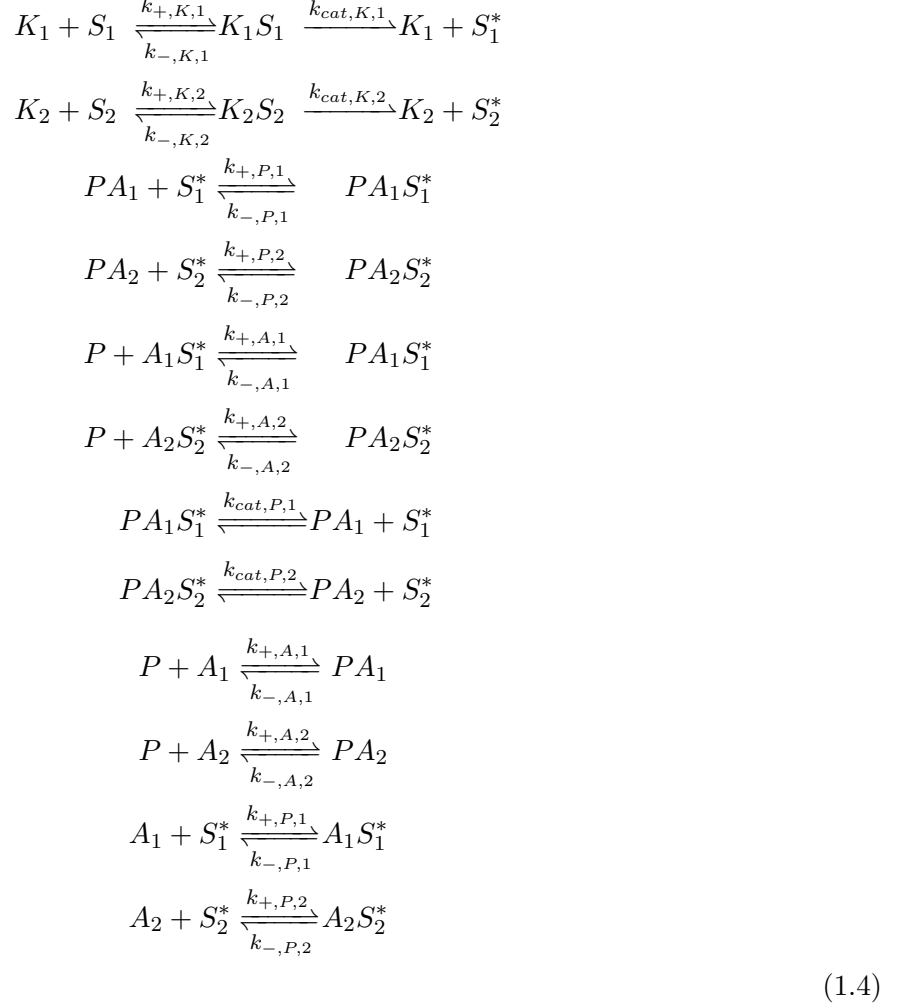
where  $i = 1$  or  $2$ . Our simulations started with the following initial concentrations:

Molecular Species	Initial Concentration
$K_1$	0-20 nM
$K_2$	0-20 nM
$P$	10 nM
$S_1$	10 μM
$S_2$	10 μM
$A_1$	10 <sup>-1</sup> -10 <sup>4</sup> nM
$A_2$	10 <sup>-1</sup> -10 <sup>4</sup> nM

This model is used to generate Fig. 4 of the main text. In Fig. 4A,  $r_1$  is set to 0.1 by having an initial concentration of  $K_1$  of 1 nM and  $r_2$  is set to 2 by having an initial concentration of  $K_2$  of 20 nM. The initial concentration of the adaptors  $A_1$  and  $A_2$  were then concurrently varied between 10<sup>-1</sup> to 10<sup>4</sup> nM. In Fig. 4B,  $A_1$  and  $A_2$  were initialized with total concentration of 10 nM each. The values of  $r_1$  and  $r_2$  were set between 0 and 2 by setting the initial concentrations of  $K_1$  and  $K_2$  between 0 and 20 nM.

## 1.6 2-Kinase/1-Phosphatase Loop with “Unordered” Phosphatase Adaptors

We also developed a model of a 2-Kinase/1-Phosphatase Loop with Unordered Phosphatase Adaptors. The set of enzymatic reactions for this model are:



This model differs from the “ordered” model in that the adaptors can first bind the phosphorylated substrate without needing to be bound to a phosphatase catalytic core. These reactions are diagrammed in the inset of Fig. 4A of the main text. The set of ODEs describing the concentration of free enzymes are:

$$\begin{aligned}
 \frac{d[K_1]}{dt} &= [K_1 S_1] k_{-,K,1} + [K_1 S_1] k_{cat,K,1} - [K_1][S_1] k_{+,K,1} \\
 \frac{d[K_2]}{dt} &= [K_2 S_2] k_{-,K,2} + [K_2 S_2] k_{cat,K,2} - [K_2][S_2] k_{+,K,2}
 \end{aligned}$$



$$\begin{aligned}\frac{d[P]}{dt} &= [PA_1]k_{-,A,1} + [PA_2]k_{-,A,2} + [PA_1S_1^*]k_{-,A,1} + [PA_2S_2^*]k_{-,A,2} \\ &\quad - [P][A_1]k_{+,A,1} - [P][A_2]k_{+,A,2} - [P][A_1S_1^*]k_{+,A,1} - [P][A_2S_2^*]k_{+,A,2}\end{aligned}$$

The set of ODEs describing the concentration of free unphosphorylated substrates are:

$$\begin{aligned}\frac{d[S_1]}{dt} &= [K_1S_1]k_{-,K,1} + [PA_1S_1^*]k_{cat,P,1} - [K_1][S_1]k_{+,K,1} \\ \frac{d[S_2]}{dt} &= [K_2S_2]k_{-,K,2} + [PA_2S_2^*]k_{cat,P,2} - [K_2][S_2]k_{+,K,2}\end{aligned}$$

The set of ODEs describing the concentration of free phosphorylated substrates are:

$$\begin{aligned}\frac{d[S_1^*]}{dt} &= [PA_1S_1^*]k_{-,P,1} + [A_1S_1^*]k_{-,P,1} + [K_1S_1]k_{cat,K,1} - [PA_1][S_1^*]k_{+,P,1} - [A_1][S_1^*]k_{+,P,1} \\ \frac{d[S_2^*]}{dt} &= [PA_2S_2^*]k_{-,P,2} + [A_2S_2^*]k_{-,P,2} + [K_2S_2]k_{cat,K,2} - [PA_2][S_2^*]k_{+,P,2} - [A_2][S_2^*]k_{+,P,2}\end{aligned}$$

The set of ODEs describing the concentration of adaptor-bound phosphatase are:

$$\begin{aligned}\frac{d[PA_1]}{dt} &= [P][A_1]k_{+,A,1} + [PA_1S_1^*]k_{-,P,1} + [PA_1S_1^*]k_{cat,P,1} - [PA_1]k_{-,A,1} - [PA_1][S_1^*]k_{+,P,1} \\ \frac{d[PA_2]}{dt} &= [P][A_2]k_{+,A,2} + [PA_2S_2^*]k_{-,P,2} + [PA_2S_2^*]k_{cat,P,2} - [PA_2]k_{-,A,2} - [PA_2][S_2^*]k_{+,P,2}\end{aligned}$$

The set of ODEs describing the concentration of adaptor-bound phosphorylated substrate are:

$$\begin{aligned}\frac{d[A_1S_1^*]}{dt} &= [A_1][S_1^*]k_{+,P,1} + [PA_1S_1^*]k_{-,A,1} - [A_1S_1^*]k_{-,P,1} - [P][A_1S_1^*]k_{+,A,1} \\ \frac{d[A_2S_2^*]}{dt} &= [A_2][S_2^*]k_{+,P,2} + [PA_2S_2^*]k_{-,A,2} - [A_2S_2^*]k_{-,P,2} - [P][A_2S_2^*]k_{+,A,2}\end{aligned}$$

The set of ODEs describing the concentration of enzyme-substrate complexes are:

$$\begin{aligned}\frac{d[K_1S_1]}{dt} &= [K_1][S_1]k_{+,K,1} - [K_1S_1]k_{-,K,1} - [K_1S_1]k_{cat,K,1} \\ \frac{d[K_2S_2]}{dt} &= [K_2][S_2]k_{+,K,2} - [K_2S_2]k_{-,K,2} - [K_2S_2]k_{cat,K,2} \\ \frac{d[PA_1S_1^*]}{dt} &= [PA_1][S_1^*]k_{+,P,1} + [P][A_1S_1^*]k_{+,P,1} - [PA_1S_1^*]k_{-,P,1} - [PA_1S_1^*]k_{-,A,1} - [PA_1S_1^*]k_{cat,P,1} \\ \frac{d[PA_2S_2^*]}{dt} &= [PA_2][S_2^*]k_{+,P,2} + [P][A_2S_2^*]k_{+,P,2} - [PA_2S_2^*]k_{-,P,2} - [PA_2S_2^*]k_{-,A,2} - [PA_2S_2^*]k_{cat,P,2}\end{aligned}$$

For purposes of display we used the following rate parameters:

Parameter	Value
$k_{+,K,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,K,i}$	0.1 s <sup>-1</sup>
$k_{cat,K,i}$	0.9 s <sup>-1</sup>
$k_{+,P,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,P,i}$	0.1 s <sup>-1</sup>
$k_{cat,P,i}$	0.9 s <sup>-1</sup>
$k_{+,A,i}$	0.001 nM <sup>-1</sup> s <sup>-1</sup>
$k_{-,A,i}$	0.1 s <sup>-1</sup>

where  $i = 1$  or  $2$ . We set the  $K_{D,A,i} = k_{-,A,i}/k_{+,A,i} = 100$  nM to represent a reasonably high affinity of the phosphatase catalytic core  $P$  with the adaptor domains  $A_i$ . We used the same affinity for the binding of the adaptor domain to the substrate. Note that, in this model, the affinity of the phosphatase for the adaptor domain, and the affinity of the adaptor domain for the substrate, does not depend on whether the adaptor is bound to its other partner. Our simulations started with the following initial concentrations:

Molecular Species	Initial Concentration
$K_1$	0-20 nM
$K_2$	0-20 nM
$P$	10 nM
$S_1$	10 $\mu$ M
$S_2$	0-10 $\mu$ M
$A_1$	10 <sup>-1</sup> -10 <sup>4</sup> nM
$A_2$	0-10 <sup>4</sup> nM

This model is used to generate Fig. 5 of the main text. In Fig. 5A,  $r_1$  is set to 0.1 by having an initial concentration of  $K_1$  of 1 nM and  $r_2$  is set to 2 by having an initial concentration of  $K_2$  of 20 nM. The initial concentration of the adaptors  $A_1$  and  $A_2$  were then concurrently varied between 10<sup>-1</sup> to 10<sup>4</sup> nM. In Fig. 5B,  $A_1$  and  $A_2$  were initialized with total concentration of 10 nM each. The values of  $r_1$  and  $r_2$  were set between 0 and 2 by setting the initial concentration of  $K_1$  and  $K_2$  to between 0 and 20 nM.

## 2 The responses of the substrates of an unsaturated phosphatase are strictly hyperbolic in $r$

We have previously shown [1], following the derivation of Goldbeter and Koshland [5], that the change in product concentration  $S_1^*$  for an enzyme  $E$  with multiple substrates can be defined as:

$$\frac{d[S_1^*]}{dt} = \frac{V_{max,E,1}[S_1]}{\alpha_{E,1}K_{M,E,1} + [S_1]} \quad (2.1)$$

where  $\alpha_{E,1} \equiv 1 + \sum_{i \neq 1} \frac{[S_i]}{K_{M,E,i}}$  is an inhibitory constant for substrate competition with  $S_1$  for  $E$ . For a futile cycle at steady state we will have  $d[S_1^*]/dt = d[S_1]/dt$ . For an unsaturated 2K1P loop,  $\alpha_{P,i} = 1$  since  $K_{M,P,i} \gg [S_i]_0$ . Given 2.1, at steady state we have:

$$\frac{V_{max,K,1}[S_1]}{K_{m,K,1} + [S_1]} = \frac{V_{max,P,1}[S_1^*]}{K_{M,P,1} + [S_1^*]} \quad (2.2)$$

Following the standard Michaelis-Menten approach [1, 5], we assume that  $[S_i]_0 \gg [K]_0, [P]_0$ , giving us  $[S_i]_0 = [S_i] + [S_i^*]$ , which can be substituted into 2.2:

$$\frac{V_{max,K,1}(1 - S_1^*)}{K_{K,1} + 1 - S_1^*} = \frac{V_{max,P,1}S_1^*}{K_{P,1} + S_1^*} \quad (2.3)$$

where  $S_1 \equiv [S_1]/[S_1]_0$ ,  $S_1^* \equiv [S_1^*]/[S_1]_0$ ,  $K_{K,1} \equiv K_{M,K,1}/[S_1]_0$ , and  $K_{P,1} \equiv K_{M,P,1}/[S_1]_0$ . Dividing both sides by  $V_{max,P,1}$  we obtain:

$$\frac{r_1(1 - S_1^*)}{K_{K,1} + 1 - S_1^*} = \frac{S_1^*}{K_{P,1} + S_1^*} \quad (2.4)$$

where  $r_1 \equiv V_{max,K,1}/V_{max,P,1}$ . Since we are assuming the phosphatase to be unsaturated,  $K_{P,1} \gg S_1^*$ , and as such 2.4 can be simplified to:

$$\frac{r_1(1 - S_1^*)}{K_{K,1} + 1 - S_1^*} = \frac{S_1^*}{K_{P,1}} \quad (2.5)$$

Expanding 2.5, we get:

$$\begin{aligned} r_1 K_{P,1} - r_1 K_{P,1} S_1^* &= K_{K,1} S_1^* + S_1^* - (S_1^*)^2 \\ (S_1^*)^2 - (1 + K_{K,1} + r_1 K_{P,1}) S_1^* + r_1 K_{P,1} &= 0 \end{aligned} \quad (2.6)$$

Solving for  $S_1^*$ , we obtain the expression:

$$S_1^* = \frac{(1 + K_{K,1} + r_1 K_{P,1}) - \sqrt{(1 + K_{K,1} + r_1 K_{P,1})^2 - 4r_1 K_{P,1}}}{2} \quad (2.7)$$

where the negative branch of the quadratic solution is chosen to obtain physically realistic values of  $S_1^*$  (i.e.  $0 \leq S_1^* \leq 1$ ). We can show that  $S_1^*$  for a 2K1P with an unsaturated phosphatase is strictly hyperbolic in  $r_1$  by taking the second derivative of 2.7 with respect to  $r_1$ :

$$\frac{d^2 S_1^*}{dr_1^2} = -\frac{2K_{K,1}K_{P,1}^2}{(-4r_1 K_{P,1} + (1 + K_{K,1} + r_1 K_{P,1})^2)^{3/2}} \quad (2.8)$$

Note that, for positive real values of both the rate constants and the concentrations, both the numerator and denominator in the above equation are positive. The second derivative is thus always negative (i.e. the curvature is concave), and the variation of  $S_1^*$  with  $r_1$  lacks an inflection point. As a result, a system with unsaturated phosphatases cannot exhibit the sigmoidal behavior characteristic of an ultrasensitive, switch-like response.

### 3 Analytical solution for 1-Kinase/1-Substrate Model with Synthesis and Degradation

From the ODEs derived in 1.4, we have at steady state:

$$\frac{d[K]}{dt} = [KS]k_- + [KS]k_{cat} + [KS]k_{deg,U} - [K][S]k_+ = 0 \quad (3.1)$$

$$\frac{d[S]}{dt} = [KS]k_- + k_{synth} - [K][S]k_+ - [S]k_{deg,U} = 0 \quad (3.2)$$

$$\frac{d[S^*]}{dt} = [KS]k_{cat} - [S^*]k_{deg,P} = 0 \quad (3.3)$$

$$\frac{d[KS]}{dt} = [K][S]k_+ - [KS]k_- - [KS]k_{cat} - [KS]k_{deg,U} = 0 \quad (3.4)$$

Additionally, we can define the conservation of mass of the kinase  $K$ :

$$[K]_0 = [K] + [KS] \quad (3.5)$$

From 3.4 we can define the concentration of  $KS$  as:

$$[KS] = \frac{[K][S]k_+}{k_- + k_{cat} + k_{deg,U}} \quad (3.6)$$

Equation 3.6 can then be substituted into 3.5 to define the concentration of free kinase  $K$ :

$$\begin{aligned} [K]_0 &= [K] + \frac{[K][S]k_+}{k_- + k_{cat} + k_{deg,U}} \\ [K]_0 &= [K] \left( 1 + \frac{[S]k_+}{k_- + k_{cat} + k_{deg,U}} \right) \\ [K] &= \frac{[K]_0}{1 + \frac{[S]k_+}{k_- + k_{cat} + k_{deg,U}}} \end{aligned} \quad (3.7)$$

We can then substitute 3.7 into 3.6 to get:

$$[KS] = \frac{[K]_0[S]}{\frac{k_- + k_{cat} + k_{deg,U}}{k_+} + [S]} \quad (3.8)$$

We can simplify 3.8 by defining:

$$K_{M,deg} \equiv \frac{k_- + k_{cat} + k_{deg,U}}{k_+}$$

where  $K_{M,deg}$  is the analogue of the Michaelis-Menten constant for the kinase  $K$ , taking into account the effects of substrate degradation. Equation 3.8 then can be written:

$$[KS] = \frac{[K]_0[S]}{K_{M,deg} + [S]} \quad (3.9)$$

From 3.3 we obtain the expression:

$$[KS]k_{cat} = [S^*]k_{deg,P}$$

into which we can substitute 3.9 to get:

$$\frac{[K]_0[S]k_{cat}}{K_{M,deg} + [S]} = [S^*]k_{deg,P} \quad (3.10)$$

We can then multiply both sides of 3.10 by  $[S]_0/[S]_0$ :

$$\frac{[K]_0S \cdot k_{cat}}{K_{deg} + S} = S^*k_{deg,P}[S_0]$$

where  $S \equiv [S]/[S]_0$ ,  $S^* \equiv [S^*]/[S]_0$ , and  $K_{deg} \equiv K_{M,deg}/[S]_0$ . We can then define  $r_{deg} \equiv [K]_0k_{cat}/[S]_0k_{deg,P}$ , the ratio of the maximum velocity of the kinase to the maximum velocity of phosphorylated substrate degradation, to get:

$$\frac{r_{deg}S}{K_{deg} + S} = S^* \quad (3.11)$$

Following the standard Michaelis-Menten assumptions [1, 5], we have  $[S]_0 \gg [K]_0$ . This gives us  $[S]_0 = [S] + [S^*]$ , or  $1 = S + S^*$ , which can be substituted into 3.11:

$$\begin{aligned} \frac{r_{deg}(1 - S^*)}{K_{deg} + 1 - S^*} &= S^* \\ r_{deg} - r_{deg}S^* &= K_{deg}S^* + S^* - (S^*)^2 \\ (S^*)^2 - (1 + r_{deg} + K_{deg})S^* + r_{deg} &= 0 \end{aligned} \quad (3.12)$$

We can then solve 3.12 for  $S^*$ :

$$S^* = \frac{1 + r_{deg} + K_{deg} - \sqrt{(1 + r_{deg} + K_{deg})^2 - 4r_{deg}}}{2} \quad (3.13)$$

where we have again chosen the negative branch of the solution to ensure  $0 \leq S^* \leq 1$ . From this derivation, we can see that degradation has two effects on the fraction of phosphorylated substrate. The degradation rate of unphosphorylated substrate can change the saturation of the kinase  $K$  by the substrate through altering the Michaelis-Menten-like constant  $K_{M,deg}$ , in the same way as altering the dissociation or catalytic rates. Additionally, the degradation rate of phosphorylated substrate can modify the magnitude of the fraction of phosphorylated substrate by changing  $r_{deg}$ .

We can show that  $S^*$  is strictly hyperbolic in  $r_{deg}$  by taking the second derivative of 3.13 with regard to  $r_{deg}$ :

$$\frac{d^2S^*}{dr_{deg}^2} = -\frac{2K_{deg}}{(-4r_{deg} + (1 + r_{deg} + K_{deg})^2)^{3/2}} \quad (3.14)$$

As for the response with an unsaturated phosphatase (equation 2.8), both the numerator and denominator of the above equation are strictly positive, so the second derivative is always negative. A system that relies on degradation to achieve effective dephosphorylation thus has no inflection point in  $r_{deg}$  and cannot exhibit a sigmoidal response to signals.

## 4 UniProt Data

We searched the UniProt database for the number of serine/threonine and tyrosine kinases and phosphatases found in all complete eukaryotic genomes [6]. For each genome, we searched for UniProt for reviewed entries that included the enzyme classification numbers for kinases and phosphatases (see Table 1). We then counted the number of entries for each genome in the search results. In order to prevent genomes with small numbers of reviewed kinases or phosphatases from unduly influencing our results, we ignored genomes with less than 5 phosphatases or 5 kinases for any given residue class (i.e. serine/threonine or tyrosine). This resulted in 16 genomes for serine/threonine enzymes (See Table 2) and 9 genomes for tyrosine enzymes (See Table 3).

Table 1: Enzyme classification numbers used to search UniProt

Enzyme	E.C. Number
Serine/Threonine Phosphatases	3.1.3.3 / 3.1.3.16
Serine/Threonine Kinases	2.7.11.x
Tyrosine Phosphatases	3.1.3.48
Tyrosine Kinases	2.7.10.x

Table 2: The numbers and ratios of serine/threonine kinases and phosphatases from UniProt used in Figure 1 of the main text.

Species	Serine/Threonine Phosphatases	Serine/Threonine Kinases	Ratio
Arabidopsis thaliana	115	559	0.206
Bos taurus	26	81	0.321
Caenorhabditis elegans	15	89	0.169
Danio rerio	14	40	0.350
Dictyostelium discoideum	21	222	0.095
Drosophila melanogaster	22	66	0.333
Gallus gallus	9	36	0.250
Homo sapiens	79	372	0.212
Mus musculus	72	374	0.193
Oryctolagus cuniculus	7	23	0.304
Oryza sativa	94	120	0.783
Pongo abelii	10	37	0.270
Rattus norvegicus	40	188	0.213
Saccharomyces cerevisiae	24	124	0.194
Schizosaccharomyces pombe	20	100	0.200
Xenopus laevis	17	72	0.236

Table 3: The numbers and ratios of tyrosine kinases and phosphatases from UniProt used in Figure 1 of the main text.

Species	Tyrosine Phosphatases	Tyrosine Kinases	Ratio
<i>Arabidopsis thaliana</i>	9	16	0.563
<i>Bos taurus</i>	13	9	1.444
<i>Caenorhabditis elegans</i>	12	11	1.091
<i>Drosophila melanogaster</i>	14	21	0.667
<i>Gallus gallus</i>	11	32	0.344
<i>Homo sapiens</i>	87	95	0.916
<i>Mus musculus</i>	81	95	0.853
<i>Rattus norvegicus</i>	39	49	0.796
<i>Xenopus laevis</i>	11	25	0.440

Additionally, we used UniProt to determine the number of phosphoproteins in complete eukaryotic genomes. We searched UniProt for reviewed entries with keyword ‘Phosphoprotein’. We then analyzed the search results and computed the number of entries for each of the 16 species from the serine/threonine enzyme results and the 9 species from the tyrosine enzyme results. The phosphatase numbers represent the total number of phosphatases from tables 2 and 3.

Table 4: The numbers and ratios of the total number of phosphatases and substrates from UniProt used in Figure 1 of the main text.

Species	Total Phosphatases	Total Substrates	Ratio
<i>Arabidopsis thaliana</i>	124	1295	10.444
<i>Bos taurus</i>	39	1759	45.103
<i>Caenorhabditis elegans</i>	27	89	3.296
<i>Danio rerio</i>	14	195	13.929
<i>Dictyostelium discoideum</i>	21	160	7.619
<i>Drosophila melanogaster</i>	36	833	23.139
<i>Gallus gallus</i>	20	282	14.100
<i>Homo sapiens</i>	166	5924	35.687
<i>Mus musculus</i>	153	5313	34.725
<i>Oryctolagus cuniculus</i>	7	282	40.286
<i>Oryza sativa</i>	94	150	1.596
<i>Pongo abelii</i>	10	819	81.900
<i>Rattus norvegicus</i>	79	2691	34.063
<i>Saccharomyces cerevisiae</i>	24	2425	101.042
<i>Schizosaccharomyces pombe</i>	20	1067	53.350
<i>Xenopus laevis</i>	28	277	9.893

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