

1 **Supporting Information**

2 **Evolutionary expansion of the amidohydrolase superfamily in bacteria in response to**  
3 **synthetic compounds: the molinate and diuron hydrolases.**

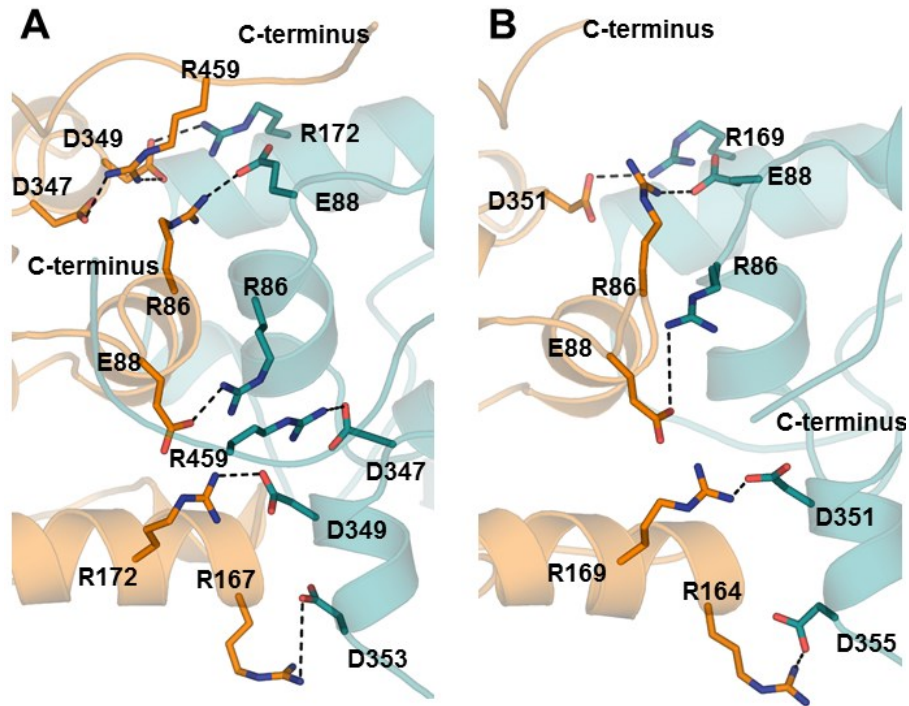
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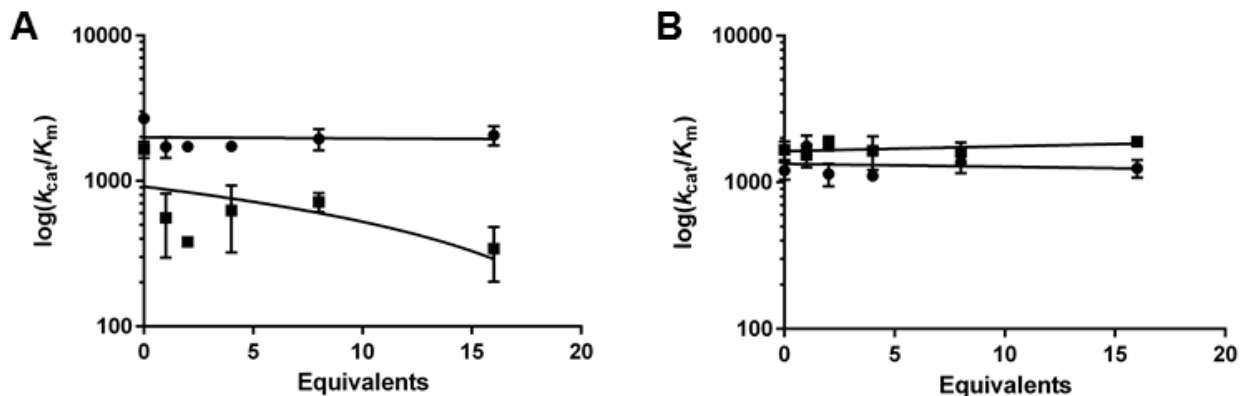
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11 **FIG. S1** Both MolA (**A**) and PuhB (**B**) have extensive salt bridge formation across the dimer-  
 12 dimer interface (orange, blue). The extended length of the MolA C-terminus allows for  
 13 additional salt-bridge formation *via* R459 (**A**).



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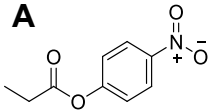
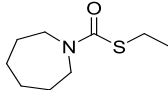
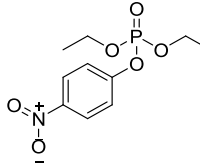
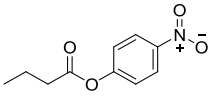
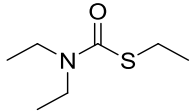
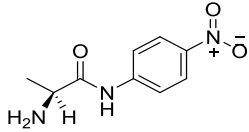
15 **FIG. S2 (A)** An exogenous supply of cobalt (●) and zinc (■) does not increase the activity of  
 16 MolA. **(B)** Exogenous bicarbonate addition has no effect on Co-MolA (●) or Zn-MolA (■), in  
 17 the presence of two equivalents of the respective metals. Error bars represent the standard  
 18 deviation of two independent replicates. Equivalents refer to the relative concentration of  
 19 exogenous metal or bicarbonate to protein concentration.

20 **TABLE S1** Effect of the metal cation on molinate hydrolase (MolA) thermostability

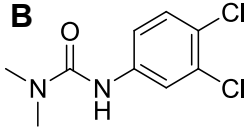
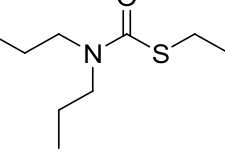
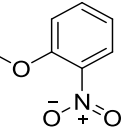
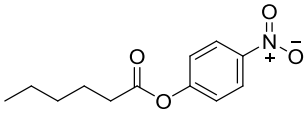
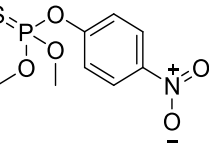
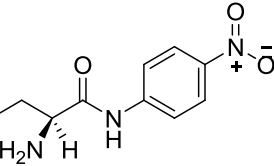
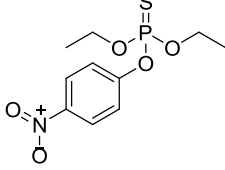
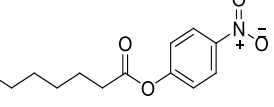
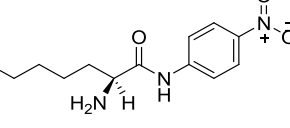
Enzyme	Melting point (°C) <sup>21</sup>
Zinc-MolA	46.0 ± 0.34 <sup>22</sup>
Apo-MolA	50.8 ± 0.31 <sup>23</sup>

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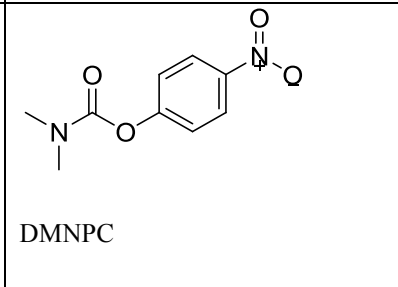
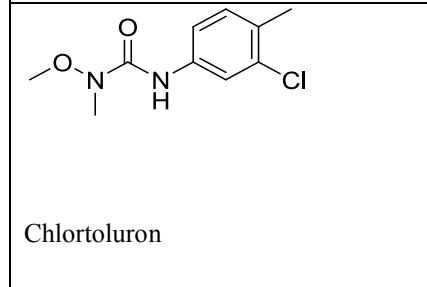
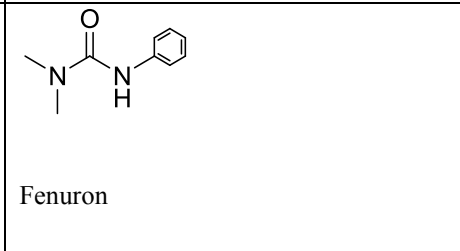
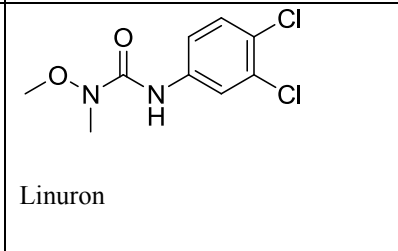
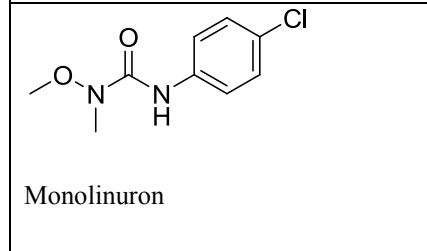
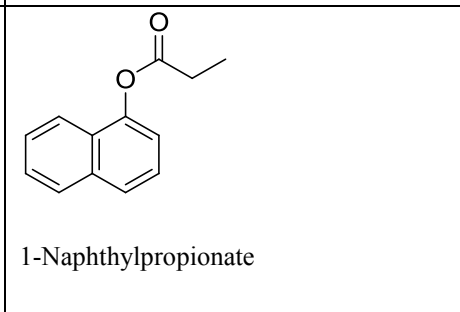
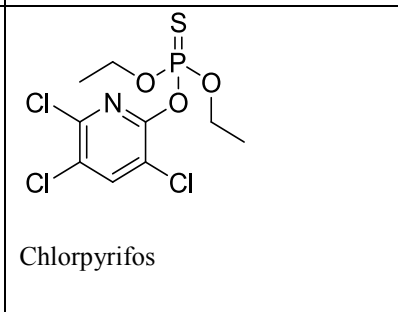
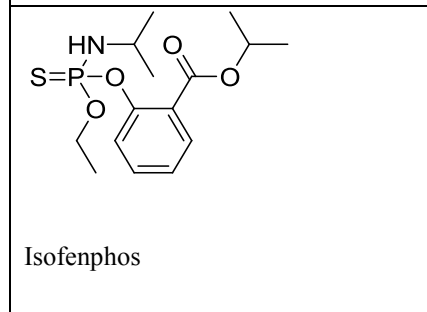
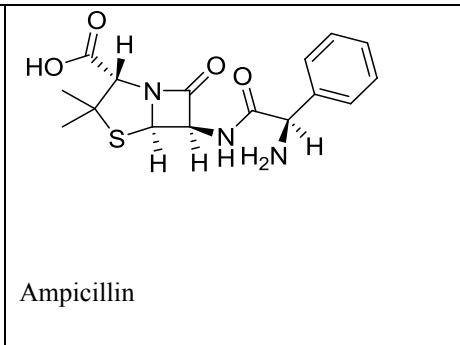
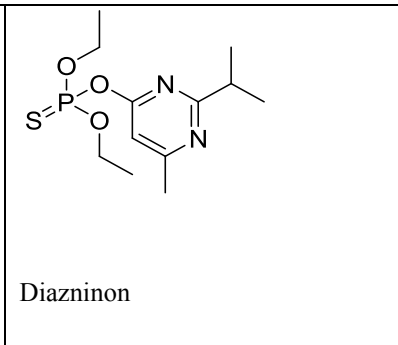
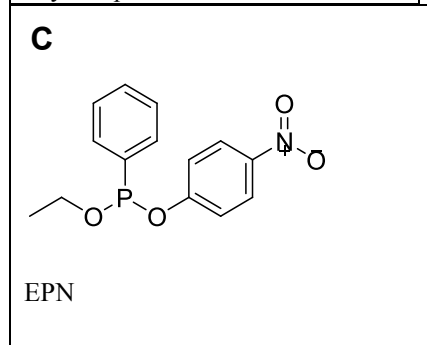
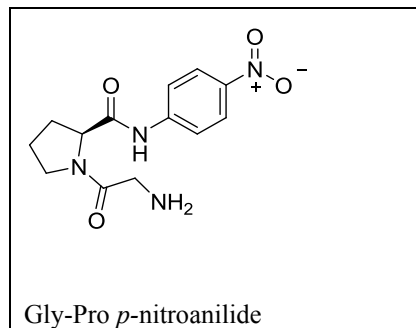
25 **FIGURE S3 (A)** Substrates from which kinetic parameters could be calculated with molinate  
 26 hydrolase. **(B)** Substrates with which there was detectable hydrolysis. **(C)** Substrates with which  
 27 there was no detectable hydrolysis.

<b>A</b>  <i>p</i> -nitrophenyl propionate	 Molinate	 Paraoxon ethyl
 <i>p</i> -nitrophenyl butyrate	 Ethiolate	 L-alanine <i>p</i> -nitroanilide

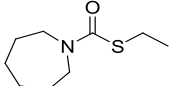
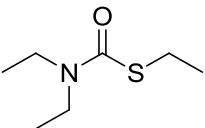
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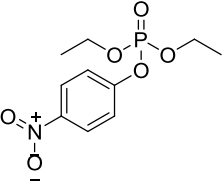
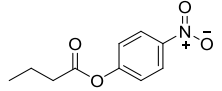
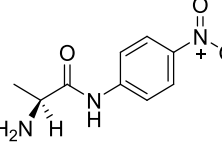
<b>B</b>  Diuron	 EPTC	 <i>o</i> -nitrophenyl acetate
 <i>p</i> -nitrophenyl hexanoate	 Parathion methyl	 L-leucine <i>p</i> -nitroanilide
 Parathion ethyl	 <i>p</i> -nitrophenyl octanoate	 L-lysine <i>p</i> -nitroanilide

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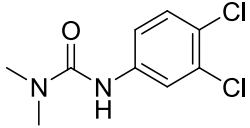


31 **TABLE S2** The kinetic parameters for wild-type MolA and variants with molinate, ethiolate,  
 32 paraoxon ethyl, *p*-nitrophenyl butyrate, L-alanine *p*-nitroanilide and diuron. The  $k_{\text{cat}}/K_m$  of diuron  
 33 was estimated using a single fixed substrate concentration well below the  $K_m$ . Values shown are  
 34 the mean and 95% confidence interval of the calculated kinetics.

Substrate	Variant	$K_m$ ( $\mu\text{M}$ )	$k_{\text{cat}}$ ( $\text{s}^{-1}$ )	$k_{\text{cat}}/K_m$ ( $\text{M}^{-1}\text{s}^{-1}$ )
Molinate 	Wild-type	6.88±1.85	8.24x10 <sup>-3</sup> ±6.54x10 <sup>-4</sup>	120 x10 <sup>1</sup>
	D327E	27.09±1.27	1.22x10 <sup>-2</sup> ±4.62x10 <sup>-4</sup>	450
	S319G	14.79±4.67	5.57x10 <sup>-2</sup> ±3.21x10 <sup>-3</sup>	376 x 10 <sup>1</sup>
	N69H	0	0	0
	H215N	0	0	0
	F318V	161.27±15.4	3.29x10 <sup>-3</sup> ±1.75x10 <sup>-5</sup>	20.4
	I73M	17.12±4.54	1.17x10 <sup>-2</sup> ±8.42x10 <sup>-4</sup>	153 x10 <sup>1</sup>
	V217C	59.42±15.56	1.17x10 <sup>-2</sup> ±2.54x10 <sup>-4</sup>	197
	F318M	62.17±5.40	1.16x10 <sup>-2</sup> ±3.16x10 <sup>-4</sup>	189
Ethiolate 	Wild-type	1976.37±322.46	6.66x10 <sup>-2</sup> ±1.79x10 <sup>-3</sup>	33.7
	D327E	1965.10±722.71	2.91x10 <sup>-2</sup> ±3.31x10 <sup>-3</sup>	14.8
	S319G	6266.73±1079.17	6.06x10 <sup>-2</sup> ±6.20x10 <sup>-3</sup>	9.67
	F318V	2914.30±2205.85	1.03x10 <sup>-2</sup> ±2.40x10 <sup>-3</sup>	3.54
	I73M	6099.57±2524.58	1.72x10 <sup>-2</sup> ±2.93x10 <sup>-3</sup>	2.81
	V217C	6234.20±286.81	3.15x10 <sup>-2</sup> ±9.23x10 <sup>-4</sup>	5.05
	F318M	7621.10±3023.30	2.43x10 <sup>-2</sup> ±3.95x10 <sup>-3</sup>	3.18
	Wild-type	1298.90±132.24	2.09x10 <sup>-2</sup> ±2.32x10 <sup>-3</sup>	16.1

Paraoxon ethyl 	D327E	1756.13±138.53	$4.73 \times 10^{-2} \pm 5.52 \times 10^{-3}$	26.9
	S319G	1591.67±73.84	$6.04 \times 10^{-2} \pm 3.48 \times 10^{-3}$	38.0
	F318V	1965.10±605.32	$2.03 \times 10^{-2} \pm 5.16 \times 10^{-3}$	10.3
	I73M	1930.97±833.69	$1.86 \times 10^{-2} \pm 5.14 \times 10^{-3}$	9.61
	V217C	539.47±88.89	$1.04 \times 10^{-2} \pm 5.43 \times 10^{-4}$	19.3
	F318M	580.65±433.36	$3.72 \times 10^{-3} \pm 2.38 \times 10^{-3}$	6.40
<p><i>p</i>-Nitrophenyl butyrate</p> 	Wild-type	407.20±18.21	$1.37 \pm 5.47 \times 10^{-2}$	336 x10 <sup>1</sup>
	D327E	400.31±49.65	$1.67 \pm 1.02 \times 10^{-1}$	418 x10 <sup>1</sup>
	S319G	464.57±28.55	$8.97 \pm 2.42 \times 10^{-1}$	193 x10 <sup>2</sup>
	F318V	484.63±15.77	$2.28 \times 10^{-1} \pm 5.37 \times 10^{-3}$	470
	I73M	474.67±28.09	$6.49 \times 10^{-1} \pm 1.76 \times 10^{-2}$	137 x10 <sup>1</sup>
	V217C	345.06±48.44	$8.28 \times 10^{-1} \pm 5.60 \times 10^{-2}$	240 x10 <sup>1</sup>
	F318M	433.24±11.97	$1.15 \pm 2.98 \times 10^{-2}$	266 x10 <sup>1</sup>
L-Alanine <i>p</i> -nitroanilide 	Wild-type	1586.40±605.27	$1.65 \times 10^{-3} \pm 3.21 \times 10^{-4}$	1.04
	D327E	1583.23±331.47	$1.83 \times 10^{-3} \pm 2.22 \times 10^{-4}$	1.15
	S319G	1548.77±258.10	$2.58 \times 10^{-3} \pm 2.33 \times 10^{-4}$	1.67
	F318V	0	0	0
	I73M	816.55±46.60	$1.78 \times 10^{-3} \pm 3.44 \times 10^{-5}$	2.19
	V217C	696.18±18.09	$1.72 \times 10^{-3} \pm 5.69 \times 10^{-5}$	2.47
	F318M	0	0	0
	Wild-type			0.43



<p style="text-align: center;">Diuron</p> 	D327E	Could not be determined due to low solubility.	1.67
	S319G		2.43
	F318V		1.20
	I73M		0.28
	V217C		0
	F318M		0.09

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