

# Supporting Information:

## Nitrogen Oxide Atom-Transfer Redox Chemistry; Mechanism of NO<sub>(g)</sub> to Nitrite Conversion Utilizing $\mu$ -oxo Heme-Fe<sup>III</sup>-O-Cu<sup>II</sup>(L) Constructs

Shabnam Hematian,<sup>†,§</sup> Isabell Kenkel,<sup>†,§</sup> Tatyana E. Shubina,<sup>†</sup> Maximilian Dürr,<sup>†</sup> Jeffrey J. Liu,<sup>†</sup> Maxime A. Siegler,<sup>†</sup> Ivana Ivanovic-

Burmazovic,<sup>\*,†</sup> Kenneth D. Karlin<sup>\*,†</sup>

<sup>†</sup> Department of Chemistry, Johns Hopkins University, Baltimore, Maryland 21218, United States

<sup>§</sup> Department of Chemistry and Pharmacy, University of Erlangen-Nürnberg, Erlangen 91058, Germany

### Contents:

#### 1. Reaction of [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(L)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] (L = TMPA, AN or MePY2) with NO<sub>(g)</sub>.

- **Figure S1.** Comparison of molecular structures of [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup> and [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup>.
- **Figure S2.** UV-vis spectra of the reaction of [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] or [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(AN)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in acetone at RT.
- **Figure S3.** EPR spectrum of the reaction of [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(AN)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in MeCN/toluene (1:1) at 20 K.
- **Figure S4.** EPR spectrum of the reaction of [(F<sub>8</sub>)Fe<sup>III</sup>-O-Cu<sup>II</sup>(MePY2)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in acetone at 12 K.

#### 2. Reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub>.

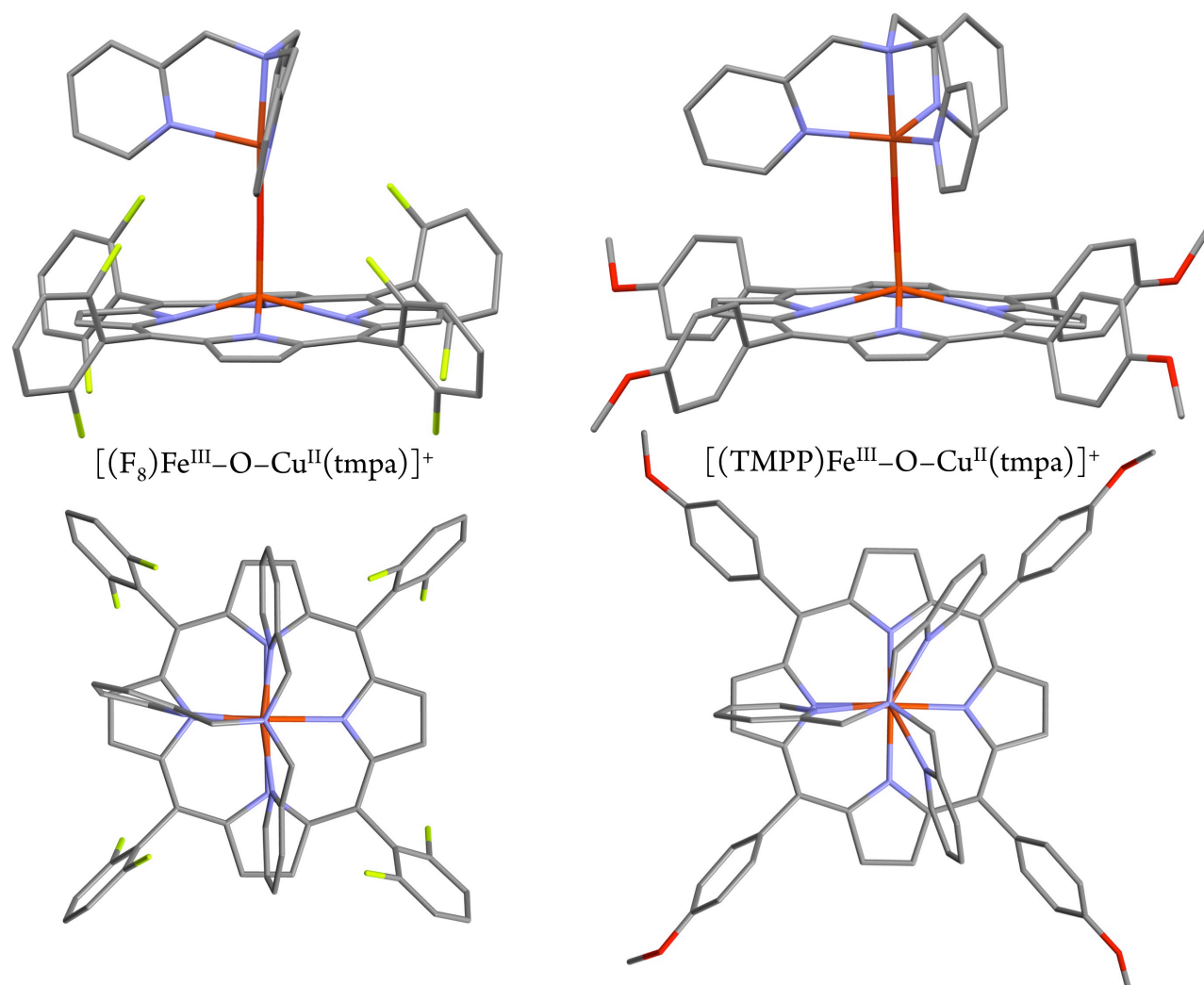
- **Figure S5.** UV-vis spectra of the reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in acetone at RT.
- **Figure S6.** EPR spectrum of the reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in acetone and MeTHF at 12 K.
- **Figure S7.** UV-vis spectra of [(TMPP)Fe<sup>III</sup>(NO)](SbF<sub>6</sub>) and (TMPP)Fe<sup>III</sup>(NO)(NO<sub>2</sub>) in acetone at -20 °C.
- **Figure S8.** Eyring plot for the final step, slow dissociation of the “intermediate”, [(NO)(TMPP)Fe<sup>II</sup>(NO<sub>2</sub>)-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] to final products in acetone.
- **Figure S9.** UV-vis spectrum of [(NO)(TMPP)Fe<sup>II</sup>(NO<sub>2</sub>)-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] in MeTHF generated from cooling a 1:1 mixture of (TMPP)Fe<sup>II</sup>(NO) and [(tmpa)Cu<sup>II</sup>(NO<sub>2</sub>)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] to -125 °C.
- **Figure S10.** EPR spectrum of the “intermediate” of the reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub> in MeTHF prepared at -80 °C recorded at 12 K.
- **Figure S11.** EPR spectra and simulated plots for product mixture and the “intermediate” of the reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with NO<sub>(g)</sub>.
- **Table S1.** Observed rate constants for the first and second reaction steps as a function of temperature and concentration.
- **Figure S12.** Kinetic traces with a two-exponential fit for the reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)][B(C<sub>6</sub>F<sub>5</sub>)<sub>4</sub>] with 2.3 mM NO at -60 °C in acetone monitored at different wavelengths.
- **Figure S13.** Eyring plots for the forward and reverse reactions for the first step, reaction of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup> with the first NO<sub>(g)</sub>.
- **Figure S14.** Eyring plot for the forward reaction of the second step, binding of the second NO<sub>(g)</sub>.
- **Figure S15.** Absorbance at 545 nm (at the end of the first reaction step) as a function of NO<sub>(g)</sub> concentration at -74 °C.
- **Figure S16.** Mass spectrum of the  $\mu$ -oxo compound, [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup> in acetone at -60 °C.
- **Figure S17.** Mass spectrum of [(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup> in acetone 20 min after bubbling with NO<sub>(g)</sub> at -60 °C.
- **Figure S18.** SOMO of the *mono*-NO adduct.
- **Figure S19.** Optimized geometry of *bis*-NO complex.
- **Figure S20.** Optimized geometry of [(tmpa)Cu<sup>II</sup>(NO<sub>2</sub>)]<sup>+</sup> complex.
- **Figure S21.** Optimized geometry of (TMPP)Fe<sup>II</sup>(NO) complex.

### 3. $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2](\text{SbF}_6)$ .

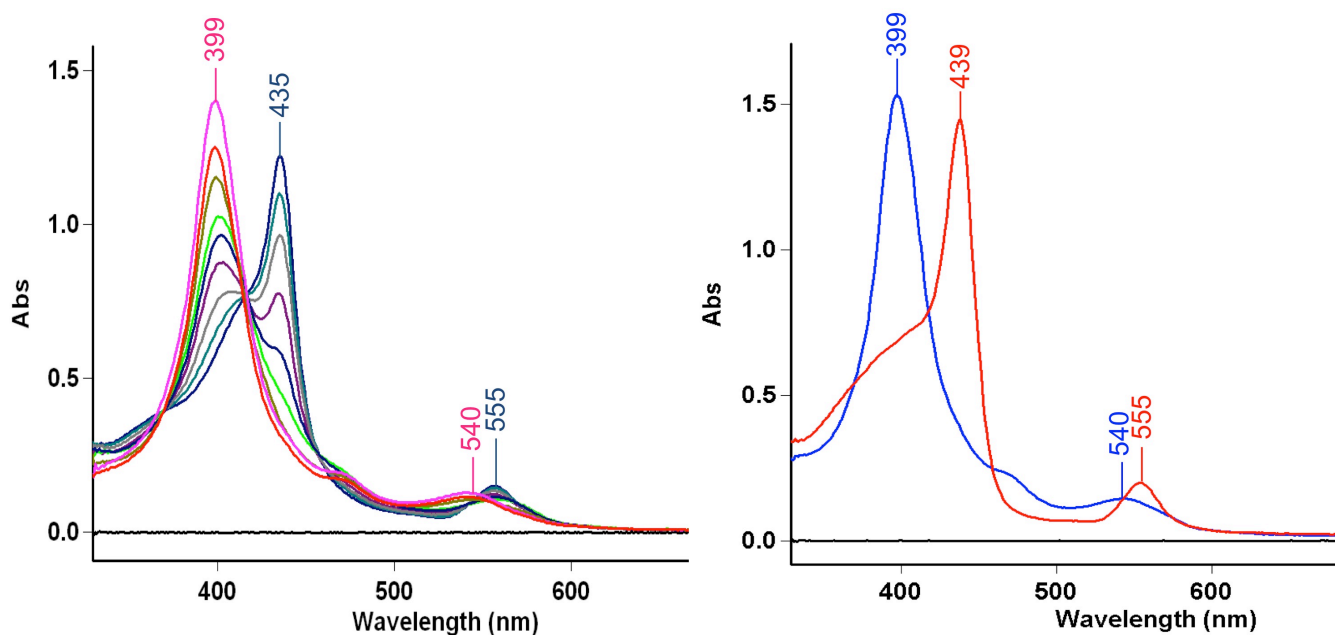
- **Figure S22.** Crystal Structure of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2]^+$  and selected bond lengths and angles.
- **Figure S23.** UV-vis spectra of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2](\text{SbF}_6)$  in acetone and MeCN at RT.
- **Figure S24.** EPR spectrum of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2](\text{SbF}_6)$  in THF/MeTHF (1:4) at 12 K.

### 4. Computational details

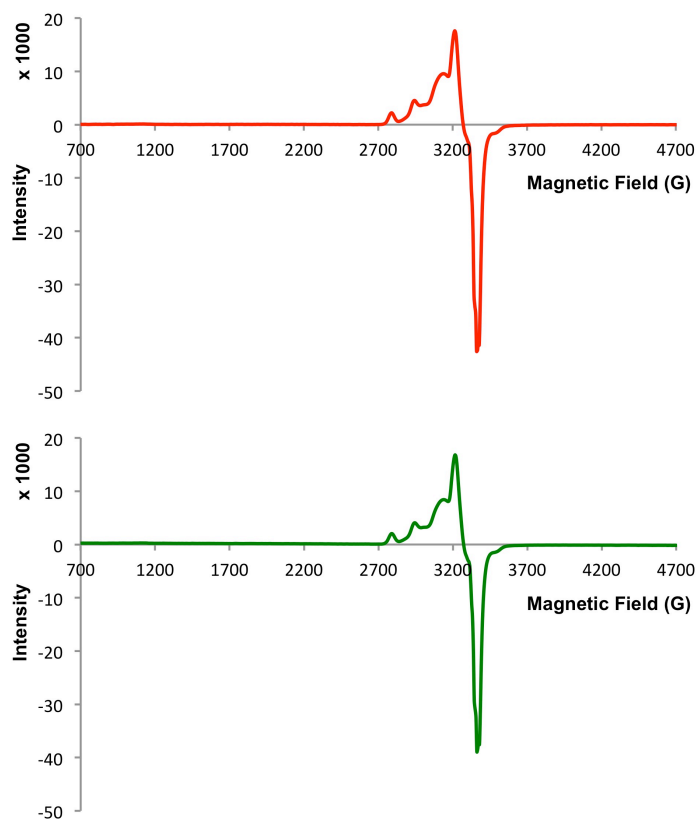
- **Table S2.** Absolute energies of studied compounds at the OLYP/6-311+G(d,p) level
- **Table S3.** Absolute energies and zero-point correction energy of studied compounds at the BP86/6-31G(d) level
- **XYZ coordinates**



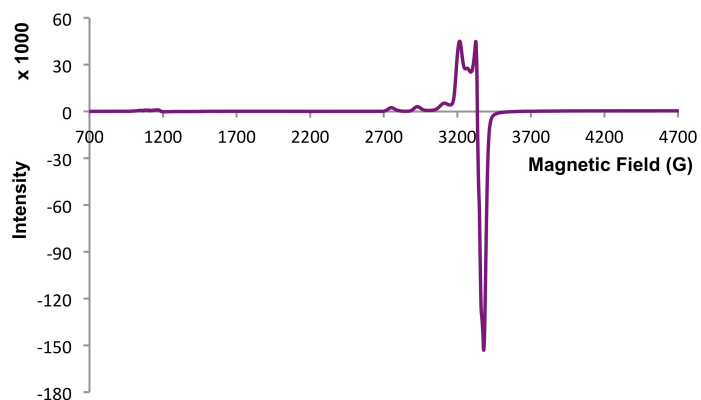
**Figure S1.** Molecular structures (side and top views) of the oxo-bridged heteronuclear cations  $[(\text{F}_8)\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$  (adapted from text reference #14) and  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$  showing that the cupric center adapted different geometries; in the former compound Cu(II) adjusts to a distorted square pyramidal geometry ( $\tau = 0.3$ ) while in the latter it is present in a trigonal bipyramidal arrangement ( $\tau = 0.9$ ). All the hydrogen atoms are omitted for clarity.



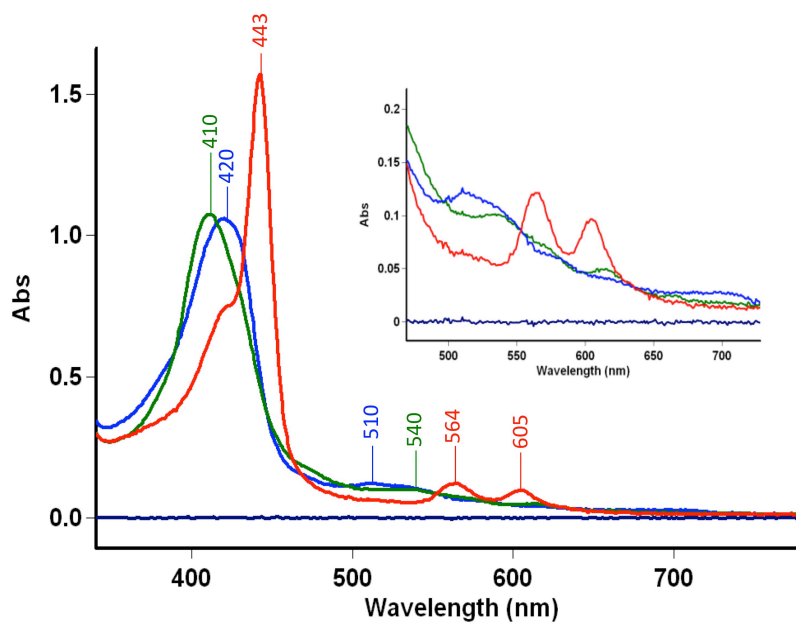
**Figure S2.** UV-vis spectra of: *Left*) (adapted from text reference # 11)  $[(F_8)Fe^{III}-O-Cu^{II}(tmpa)][B(C_6F_5)_4]$  (blue),  $(F_8)Fe^{II}(NO)$  (red) generated from addition of 1 mL of  $NO(g)$ . Addition of second mL of  $NO(g)$  to the solution resulted in completion of the reaction (purple) 10  $\mu M$  in acetone at RT. *Right*)  $[(F_8)Fe^{III}-O-Cu^{II}(AN)][B(C_6F_5)_4]$  (red) and  $(F_8)Fe^{II}(NO)$  (blue) immediately generated after addition of 1 mL  $NO(g)$  into the  $\mu$ -oxo complex solution 10  $\mu M$  in acetone at RT.



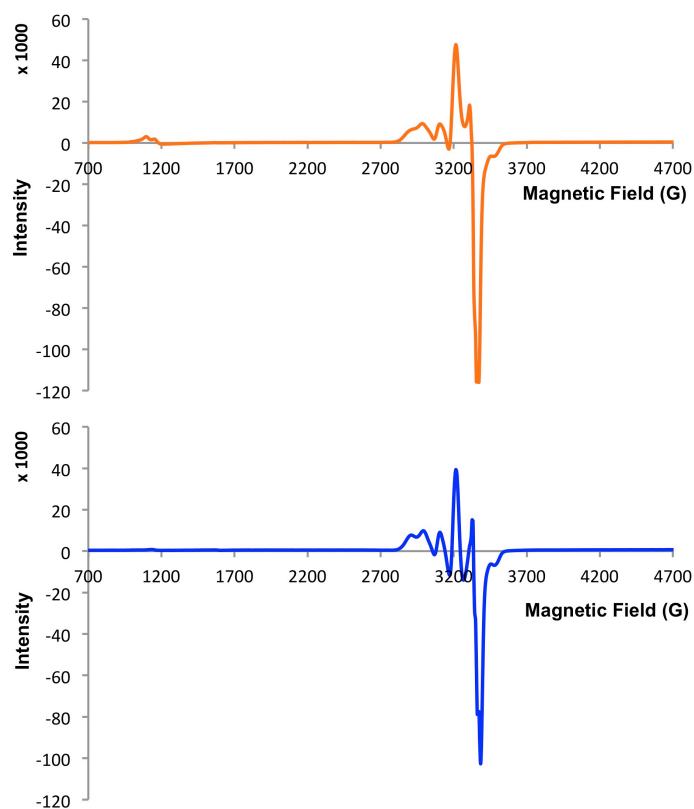
**Figure S3.** EPR spectrum comparison between the reaction mixture and an authentic sample: the products of the reaction of  $NO(g)$  and  $[(F_8)Fe^{III}-O-Cu^{II}(AN)][B(C_6F_5)_4]$ , giving signals of  $(F_8)Fe^{II}(NO)$  and Cu(II) (red); an authentic sample of a 1:1 mixture of  $(F_8)Fe^{II}(NO)$  and  $[(AN)Cu^{II}(NO_2)](CF_3SO_3)$  (green) 1 mM in MeCN/toluene (1:1) at 20 K.



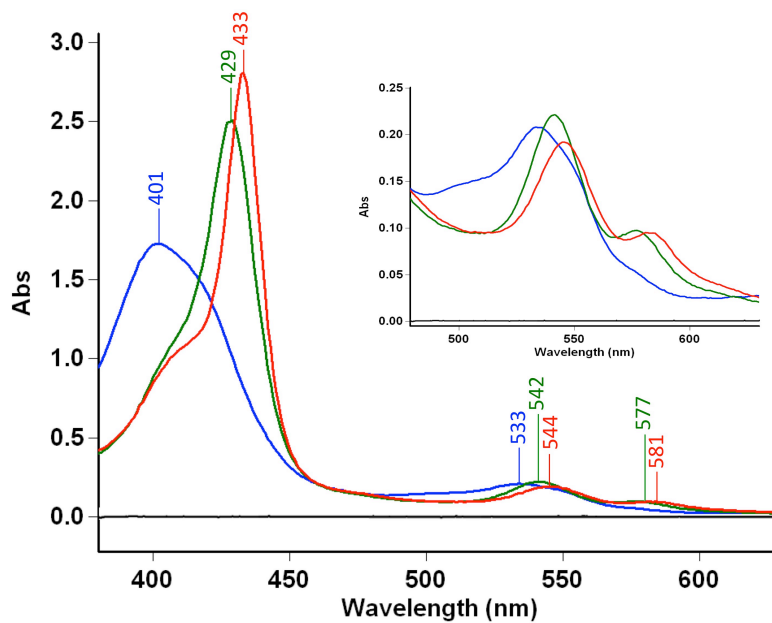
**Figure S4.** EPR spectrum of the products of the reaction of  $\text{NO}_{(g)}$  and  $[(\text{F}_8)\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{MePY}2)][\text{B}(\text{C}_6\text{F}_5)_4]$  (purple), giving signals of  $(\text{F}_8)\text{Fe}^{\text{II}}(\text{NO})$  and  $\text{Cu}(\text{II})$  in acetone 2 mM at 12 K.



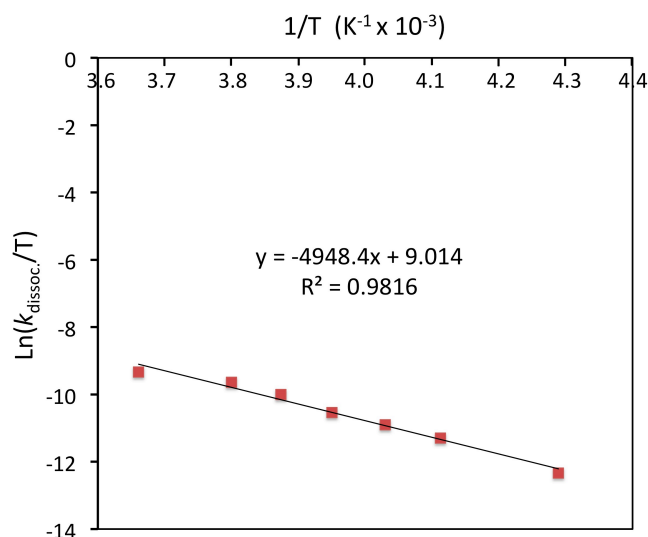
**Figure S5.** UV-vis spectra of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  (red) 35  $\mu\text{M}$  in acetone in a 2-mm cuvette at RT, right after addition of 1 mL of  $\text{NO}_{(g)}$  into the solution (blue), after stirring for 4 min forming  $(\text{TMPP})\text{Fe}^{\text{II}}(\text{NO})$  (green).



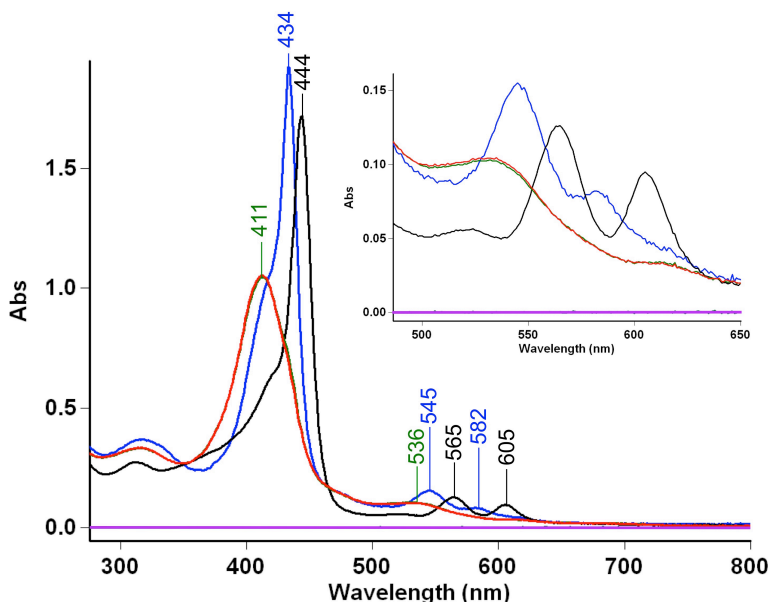
**Figure S6.** EPR spectra of the products of the reaction of  $\text{NO}_{(g)}$  and  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$ , giving signals of  $(\text{TMPP})\text{Fe}^{\text{II}}(\text{NO})$  and  $\text{Cu}(\text{II})$  in acetone (orange) and in MeTHF (blue) 2 mM at 12 K.



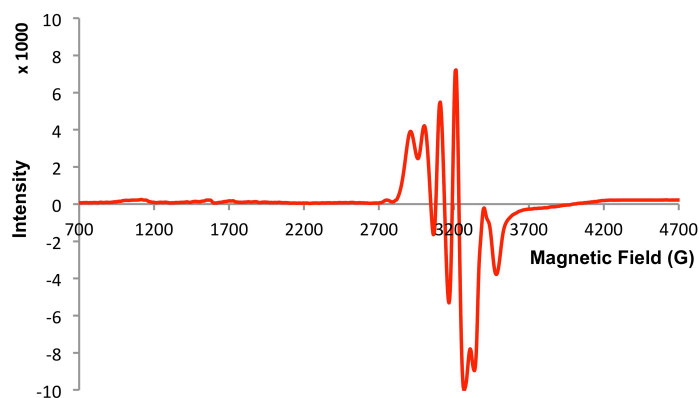
**Figure S7.** UV-vis spectra of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2]\text{SbF}_6$  (blue) 50  $\mu\text{M}$  in acetone in a 2-mm cuvette at  $-20\text{ }^\circ\text{C}$ , right after addition of 1 mL of  $\text{NO}_{(g)}$  into the solution generating  $(\text{TMPP})\text{Fe}^{\text{III}}(\text{NO})$  (green), after addition of 10 equiv  $(n\text{Bu})_4\text{N}(\text{NO}_2)$  forming  $(\text{TMPP})\text{Fe}^{\text{III}}(\text{NO})(\text{NO}_2)$  (red).



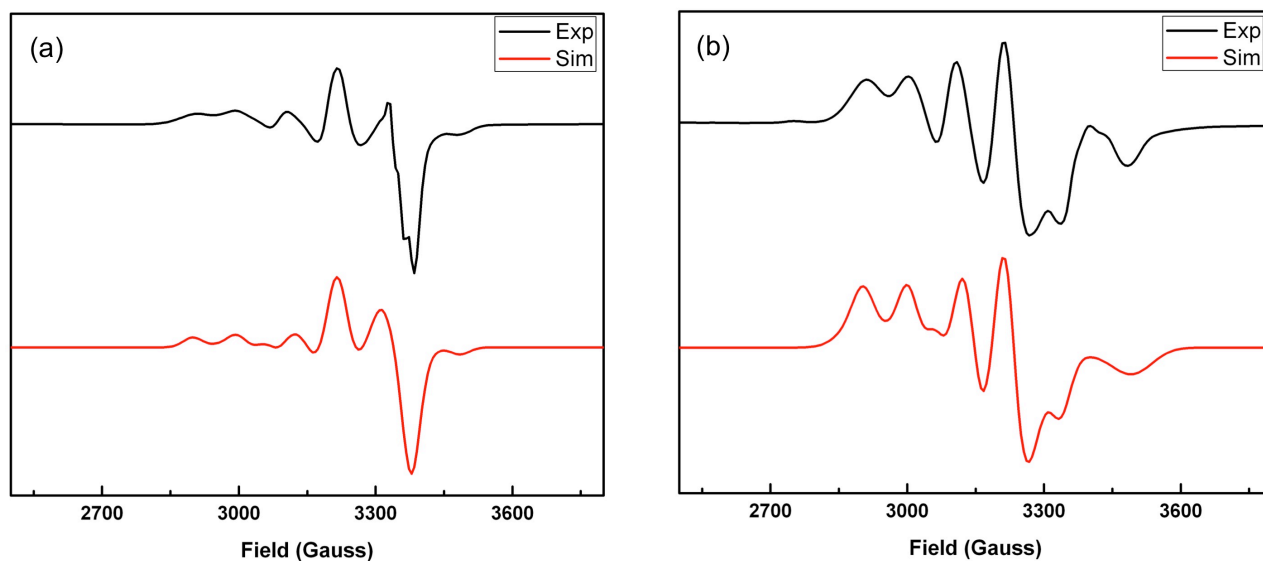
**Figure S8.** Eyring plot,  $\ln(k_{\text{obs}}/T)$  vs  $1/T$ , for the final step, slow dissociation of the “intermediate”,  $[(\text{NO})(\text{TMPP})\text{Fe}^{\text{II}}-(\text{NO}_2)-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$ , generated from reaction of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  with  $\text{NO}_{(\text{g})}$  to the two final Cu-nitrite and heme- $\text{Fe}^{\text{II}}(\text{NO})$  products. The rate constants ( $k_{\text{dissoc.}}$ ) were obtained from monitoring of the decay of the 433 nm intermediate in acetone at 0, -10, -15, -20, -25, -30, -40 °C. From the plot, activation parameters for the final step were obtained,  $\Delta S^{\ddagger}_{\text{dissoc.}} = R(\text{intercept} - \ln k_{\text{B}}/h) = -123 \pm 2 \text{ J mole}^{-1}\text{K}^{-1}$  and  $\Delta H^{\ddagger}_{\text{dissoc.}} = -R(\text{slope}) = +41,141 \pm 100 \text{ J mol}^{-1}$ .



**Figure S9.** UV-vis spectra of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  (black) 35  $\mu\text{M}$  in MeTHF in a 2-mm cuvette at RT, after addition of 1 mL of  $\text{NO}_{(\text{g})}$  into the solution forming final products,  $(\text{TMPP})\text{Fe}^{\text{II}}(\text{NO})$  and  $[(\text{tmpa})\text{Cu}^{\text{II}}(\text{NO}_2)][\text{B}(\text{C}_6\text{F}_5)_4]$  (green). The excess  $\text{NO}_{(\text{g})}$  was then removed and the sample was cooled to -125 °C generating a new species formulated as  $[(\text{NO})(\text{TMPP})\text{Fe}^{\text{II}}-(\text{NO}_2)-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  (blue). Warming up to RT resulted the reformation of initial spectrum of the final products (red).



**Figure S10.** EPR spectrum of the “intermediate” for the reaction of  $\text{NO}_{(g)}$  and  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  at 12 K giving signals of  $\text{Cu}(\text{II})$  and ferrous heme nitrosyl species, but with a different hyperfine coupling observed in the spectrum of products mixture (Figure S9). Sample was prepared by addition of 1 mL  $\text{NO}_{(g)}$  into the 1 mM solution of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$  in MeTHF at  $-80^\circ\text{C}$  (dry ice-acetone bath). Then excess  $\text{NO}_{(g)}$  was removed via vacuum/Ar-purge cycles and the EPR sample was frozen in liquid nitrogen prior to measurement.



**Figure S11.** Experimental (black) and simulated (red) spectra (for the reaction of  $\text{NO}_{(g)}$  with  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})][\text{B}(\text{C}_6\text{F}_5)_4]$ ) of the product species (a)  $g_{1,\text{Cu}} = 2.153$ ,  $g_{2,\text{Cu}} = 2.195$ ,  $g_{3,\text{Cu}} = 1.987$ ,  $A_{1,\text{Cu}} = 82.1$  G,  $A_{2,\text{Cu}} = 115$  G,  $A_{3,\text{Cu}} = 69.2$  G,  $g_{1,\text{FeNO}} = 2.094$ ,  $g_{2,\text{FeNO}} = 2.009$ ,  $g_{3,\text{FeNO}} = 1.993$ ,  $A_{1,\text{FeNO}} = 10.3$  G,  $A_{2,\text{FeNO}} = 22.8$  G,  $A_{3,\text{FeNO}} = 7.1$  G and the “intermediate” species (b)  $g_{1,\text{Cu}} = 2.161$ ,  $g_{2,\text{Cu}} = 2.187$ ,  $g_{3,\text{Cu}} = 2.056$ ,  $A_{1,\text{Cu}} = 73.1$  G,  $A_{2,\text{Cu}} = 110$  G,  $A_{3,\text{Cu}} = 51.2$  G,  $g_{1,\text{FeNO}} = 2.340$ ,  $g_{2,\text{FeNO}} = 1.920$ ,  $g_{3,\text{FeNO}} = 2.066$ ,  $A_{2,\text{NO}} = 18.9$  G,  $A_{2,\text{Nitrite}} = 7.74$  G. EPR conditions: microwave frequency, 9.41 GHz; microwave power, 0.2 mW; modulation frequency, 100 kHz; modulation amplitude, 10 G; temperature, 12 K. EPR simulations were performed using EasySpin v. 4.5.5 (Stoll, S.; Schweiger, A. *J. Magn. Reson.* **2006**, 178, 42; <http://www.easyspin.org>).

**Table S1. Observed Rate Constants for the First Reaction Step ( $k_{1(\text{obs})}$ , Binding of 1<sup>st</sup> NO) and Second Reaction Step ( $k_{2(\text{obs})}$ , Binding of 2<sup>nd</sup> NO) as a Function of Temperature and Concentration.**

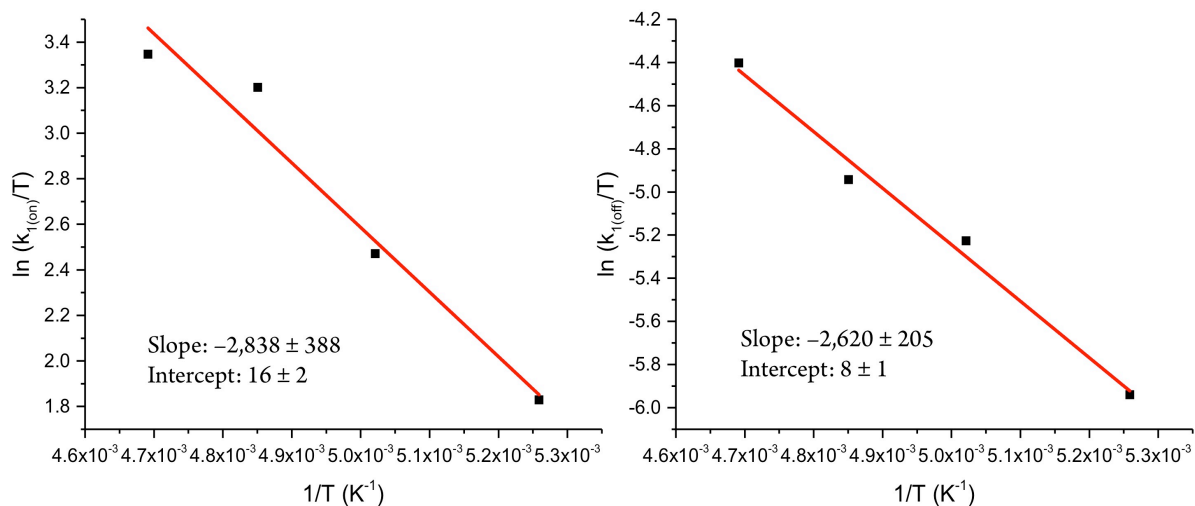
[NO] mM	Temperature						
	-60 °C		-67 °C		-74 °C		-83 °C*
	$k_{1(\text{obs})}$	$k_{2(\text{obs})}$	$k_{1(\text{obs})}$	$k_{2(\text{obs})}$	$k_{1(\text{obs})}$	$k_{2(\text{obs})}$	$k_{1(\text{obs})}$
<b>0.281</b>	4.35 ± 0.15	0.25 ± 0.02	3.11 ± 0.26	0.20 ± 0.04	1.81 ± 0.26	0.16 ± 0.03	0.92 ± 0.04
<b>0.563</b>	6.11 ± 0.66	0.45 ± 0.06	4.36 ± 0.23	0.37 ± 0.03	2.34 ± 0.03	0.23 ± 0.01	1.16 ± 0.03
<b>0.750</b>	6.72 ± 0.06	0.53 ± 0.04	4.64 ± 0.26	0.50 ± 0.04	2.90 ± 0.03	0.38 ± 0.02	1.38 ± 0.05
<b>1.13</b>	9.84 ± 0.86	0.82 ± 0.11	7.43 ± 0.18	0.72 ± 0.05	3.66 ± 0.06	0.46 ± 0.02	1.88 ± 0.10
<b>1.69</b>	12.76 ± 0.19	1.19 ± 0.03	10.33 ± 0.36	0.98 ± 0.08	4.91 ± 0.13	0.77 ± 0.06	2.52 ± 0.08
<b>2.25</b>	16.19 ± 0.37	1.65 ± 0.11	12.67 ± 1.47	1.51 ± 0.10	6.49 ± 0.09	1.16 ± 0.15	3.22 ± 0.11

\* The second step at -83 °C was very slow and  $k_{2(\text{obs})}$  could not be accurately determined.

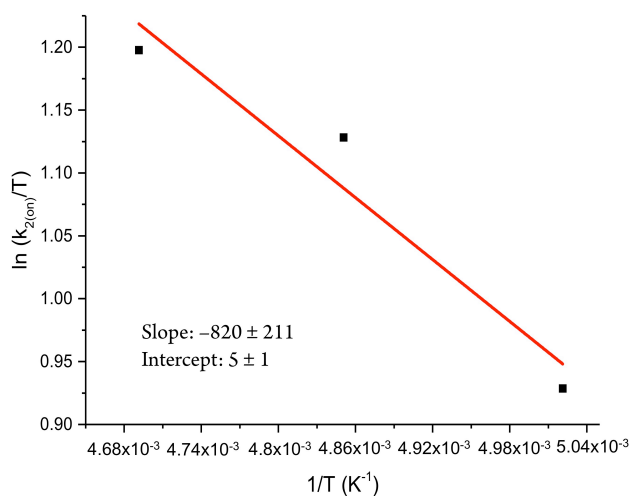


**Figure S12.** Kinetic traces with a two-exponential fit for the reaction of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tpma})][\text{B}(\text{C}_6\text{F}_5)_4]$  with 2.3 mM NO at -60 °C in acetone monitored (a) at 405 and 443 nm, as well as (b) at 475, 563 and 605 nm.

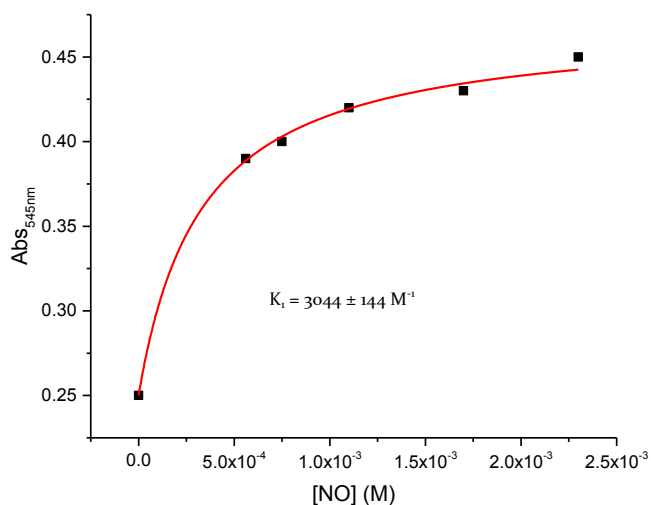




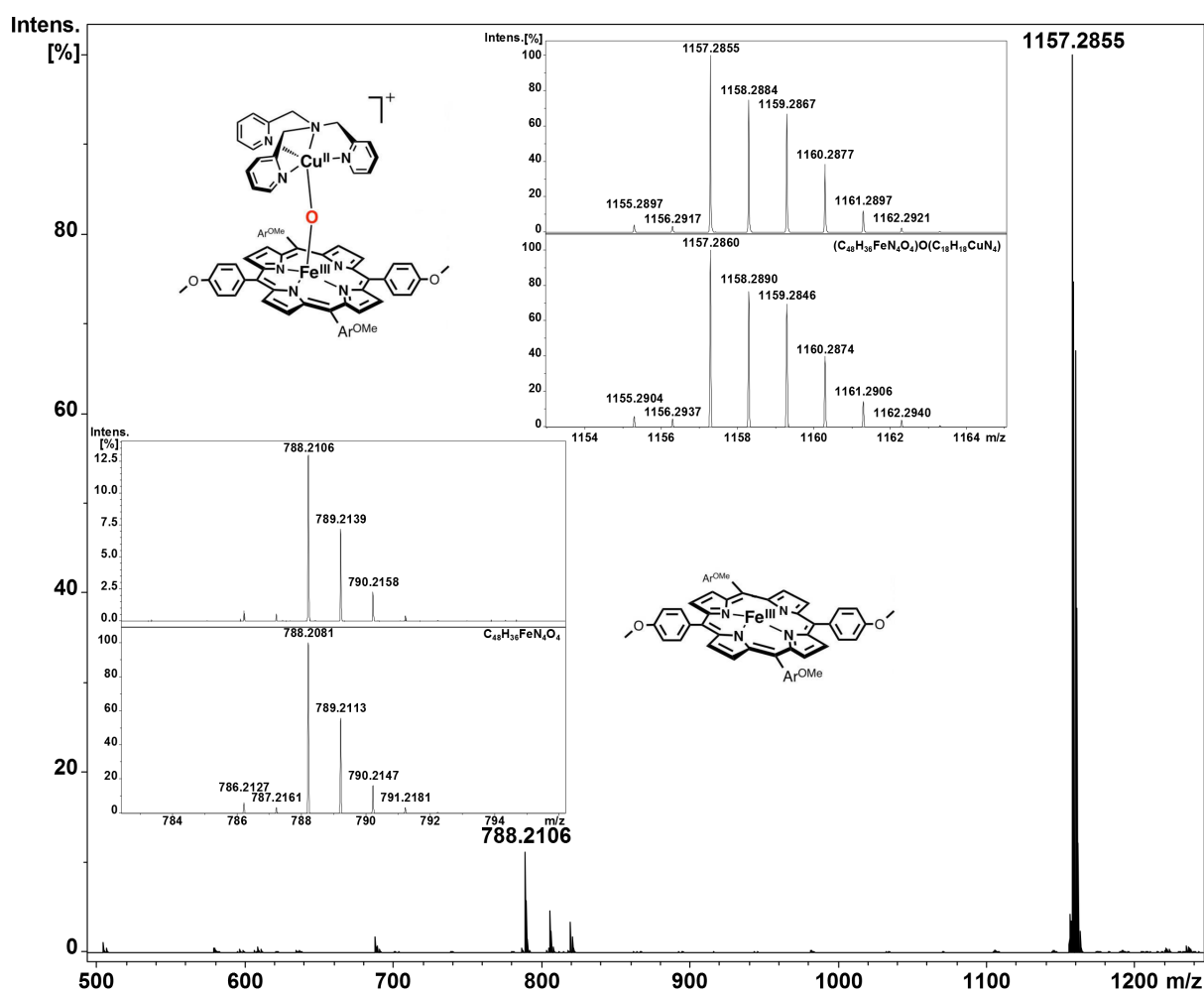
**Figure S13.** Eyring plots ( $\ln(k/T) = -(\Delta H^\ddagger/R)(1/T) + \Delta S^\ddagger/R + \ln(k_b/h)$ ;  $k = k_{1(\text{on})}$  or  $k_{1(\text{off})}$ ) for the forward and reverse reactions for the first reaction step, reaction of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$  with the first  $\text{NO}_{(\text{g})}$ .



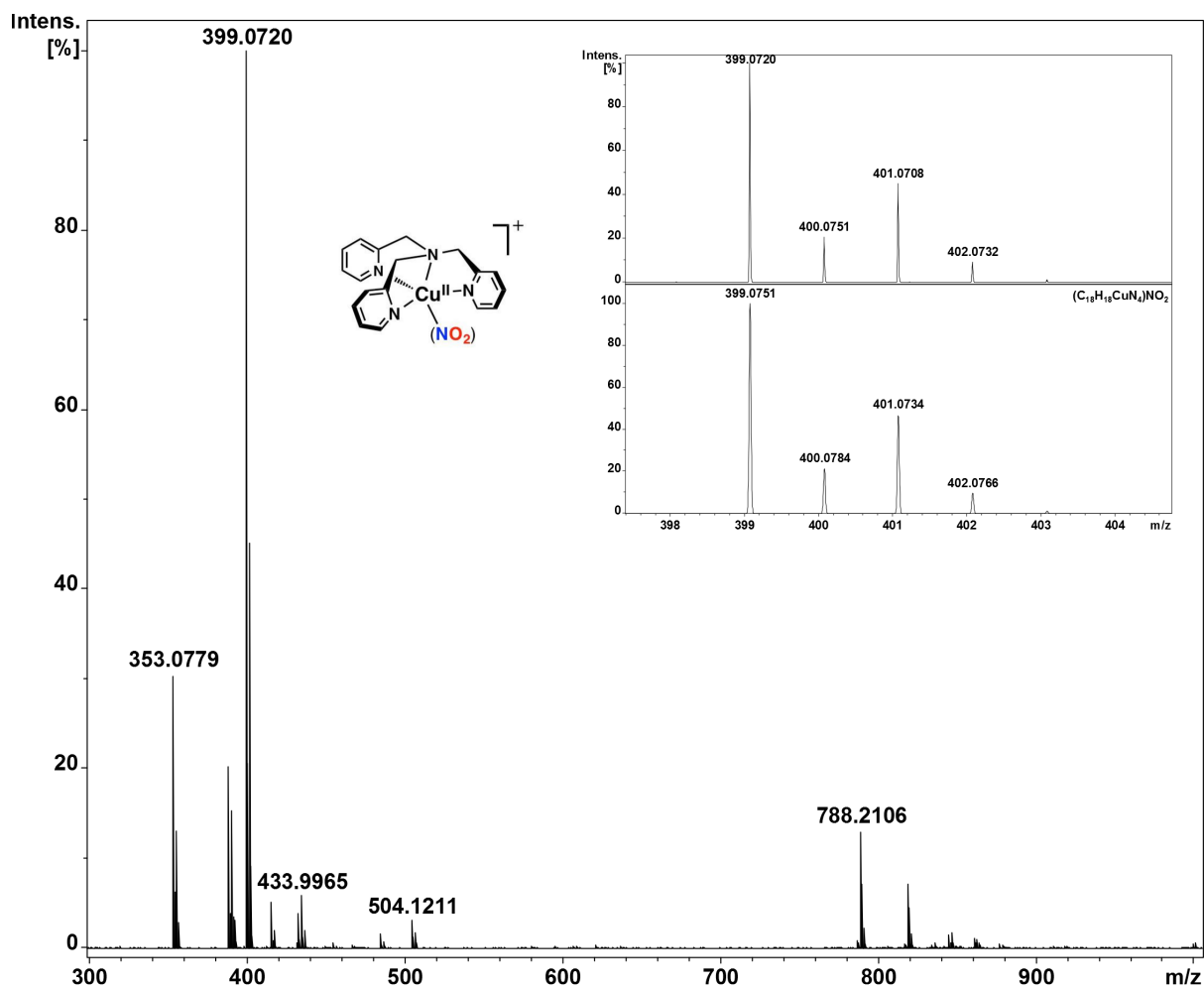
**Figure S14.** Eyring plot ( $\ln(k/T) = -(\Delta H^\ddagger/R)(1/T) + \Delta S^\ddagger/R + \ln(k_b/h)$ ;  $k = k_{2(\text{on})}$ ) for the forward reaction for the second reaction step, binding of the second  $\text{NO}_{(\text{g})}$ .



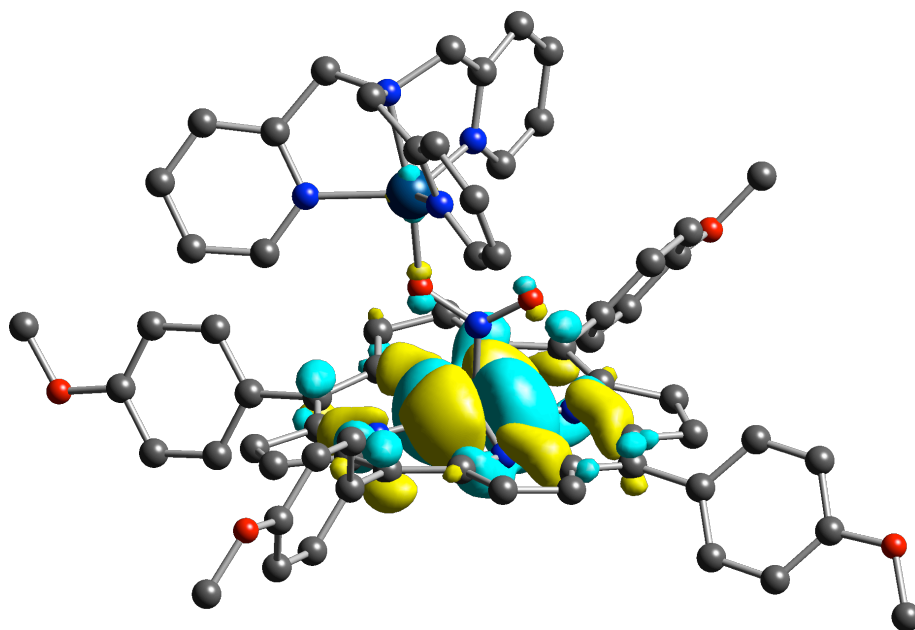
**Figure S15.** Absorbance at 545 nm (at the end of the first reaction step at  $-74^\circ C$ ) as a function of  $NO_{(g)}$  concentration (data fitted by Eq. 1 given in the main text).



**Figure S16.** Mass spectrum of  $[(TMPP)Fe^{III}-O-Cu^{II}(tmpa)]^+$  in acetone before bubbling with  $NO_{(g)}$ , spray gas temperature  $-60^\circ C$ , dry gas temperature  $-55^\circ C$ ; main species  $m/z = 1157.2855$ , which is assigned to the  $\mu$ -oxo compound.



**Figure S17.** Mass spectrum of  $[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$  in acetone 20 min after bubbling with  $\text{NO}_{(\text{g})}$ . Experimental conditions: spray gas temperature  $-60\text{ }^{\circ}\text{C}$ ; dry gas temperature  $-55\text{ }^{\circ}\text{C}$ . (Main species  $m/z = 399.0720$ , which can be assigned to the  $[(\text{tmpa})\text{Cu}^{\text{II}}(\text{NO}_2)]^+$  complex; the ferrous heme nitrosyl  $(\text{TMPP})\text{Fe}^{\text{II}}(\text{NO})$  compound cannot be observed because it is not charged.)



**Figure S18.** SOMO of the *mono*-NO adduct (BP86/6-31G(d)). Isodensity value  $\alpha = 0.02$ .

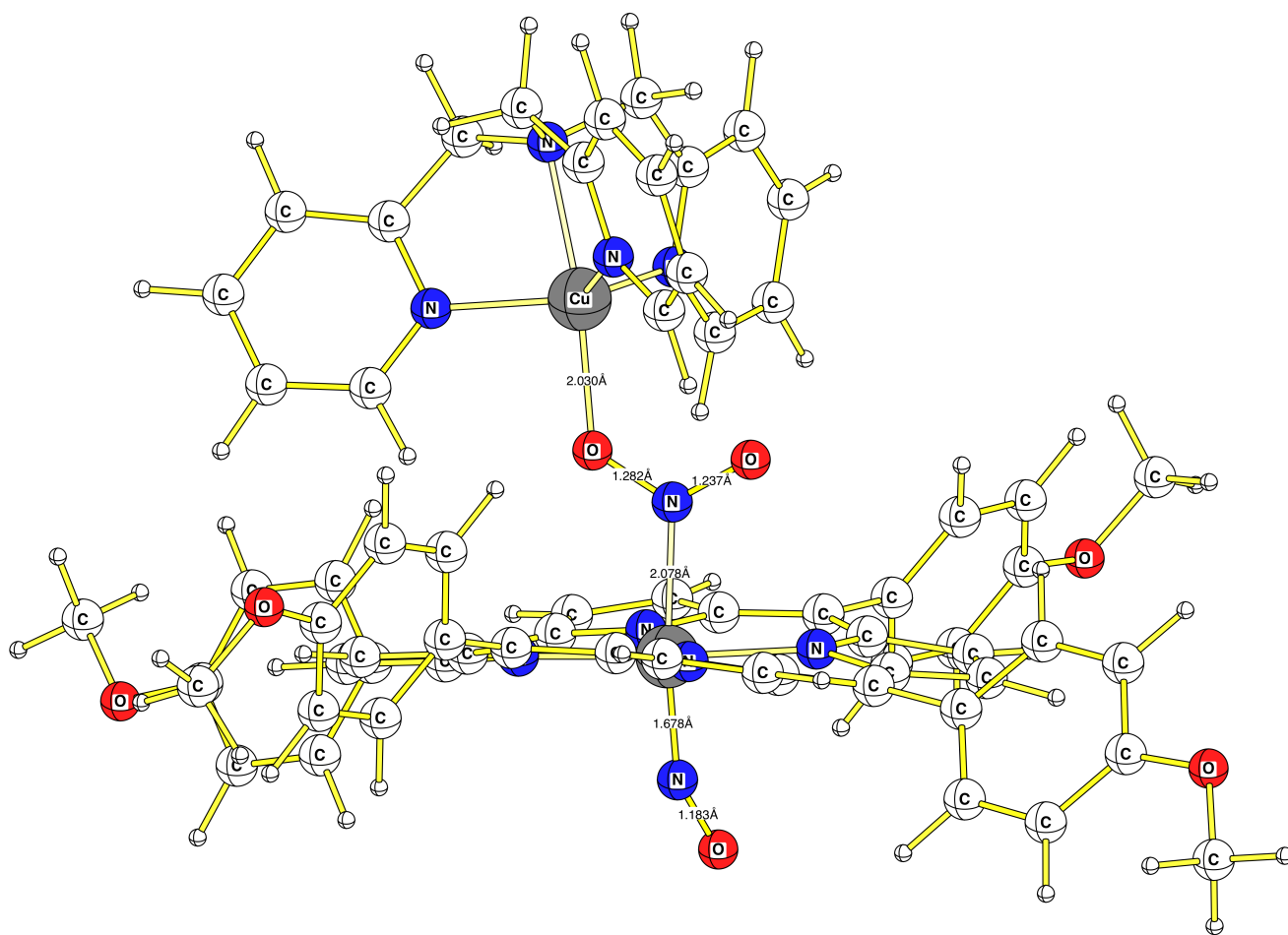


Figure S19. Optimized geometry of the *bis*-NO complex, BP86/6-31G(d).

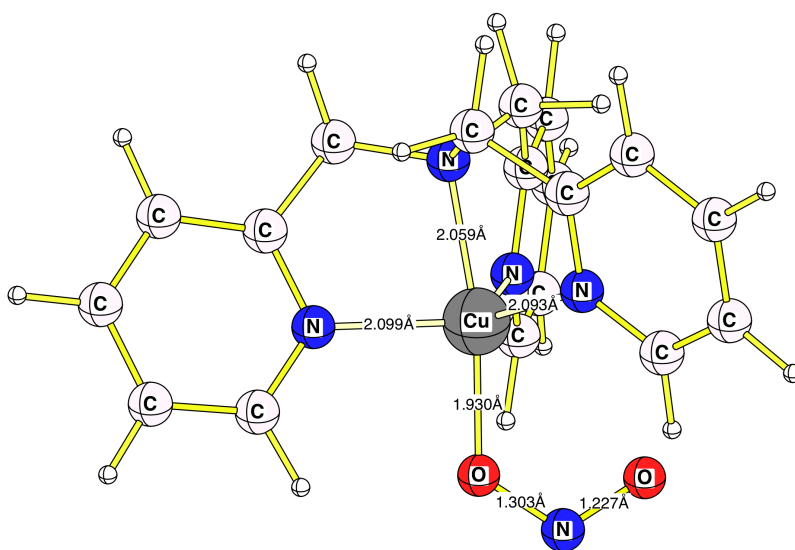
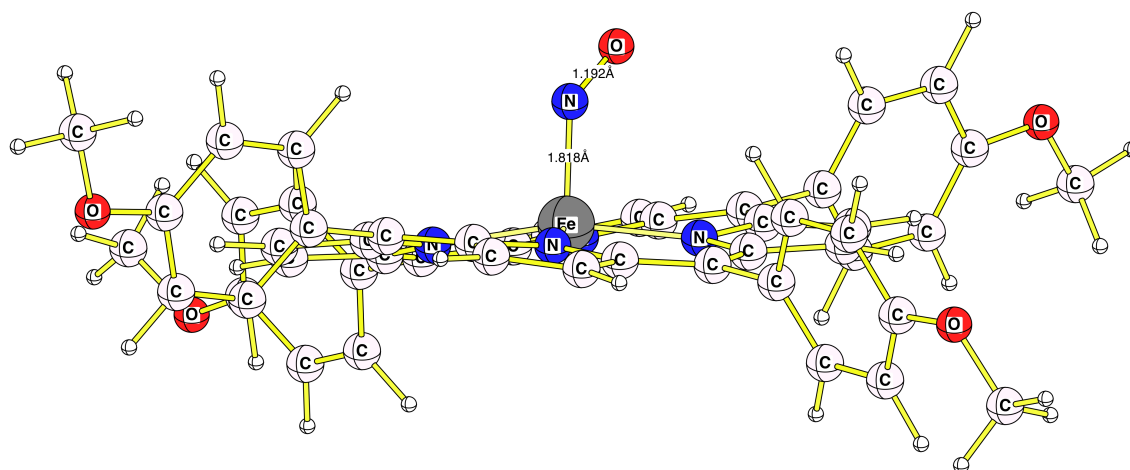
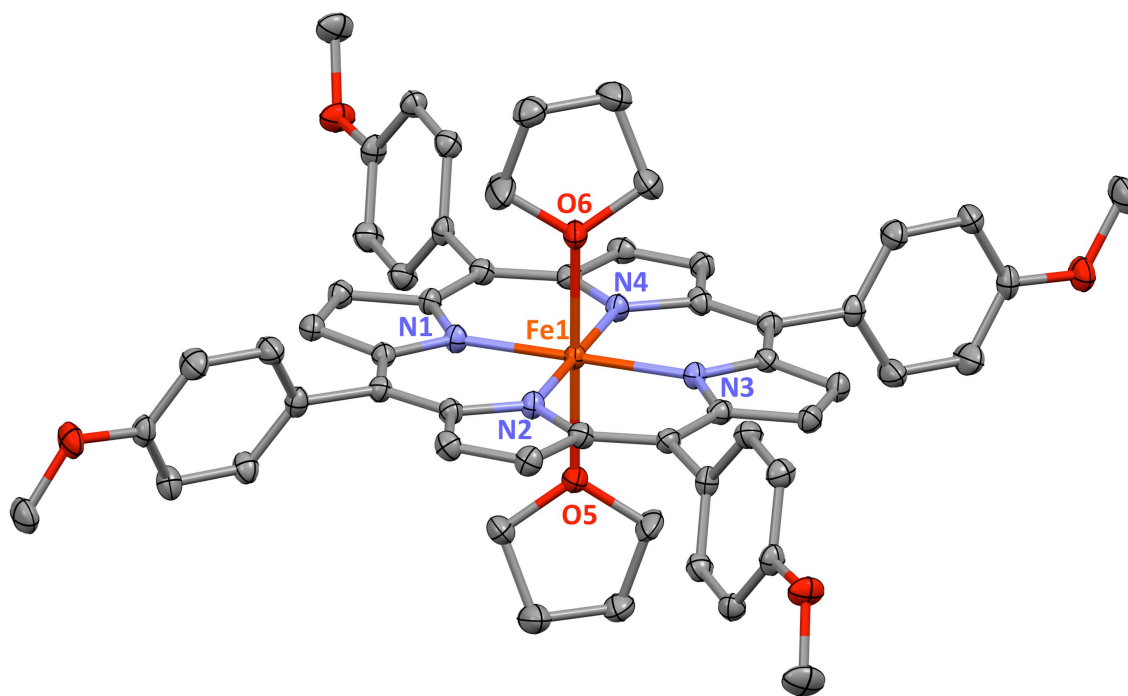


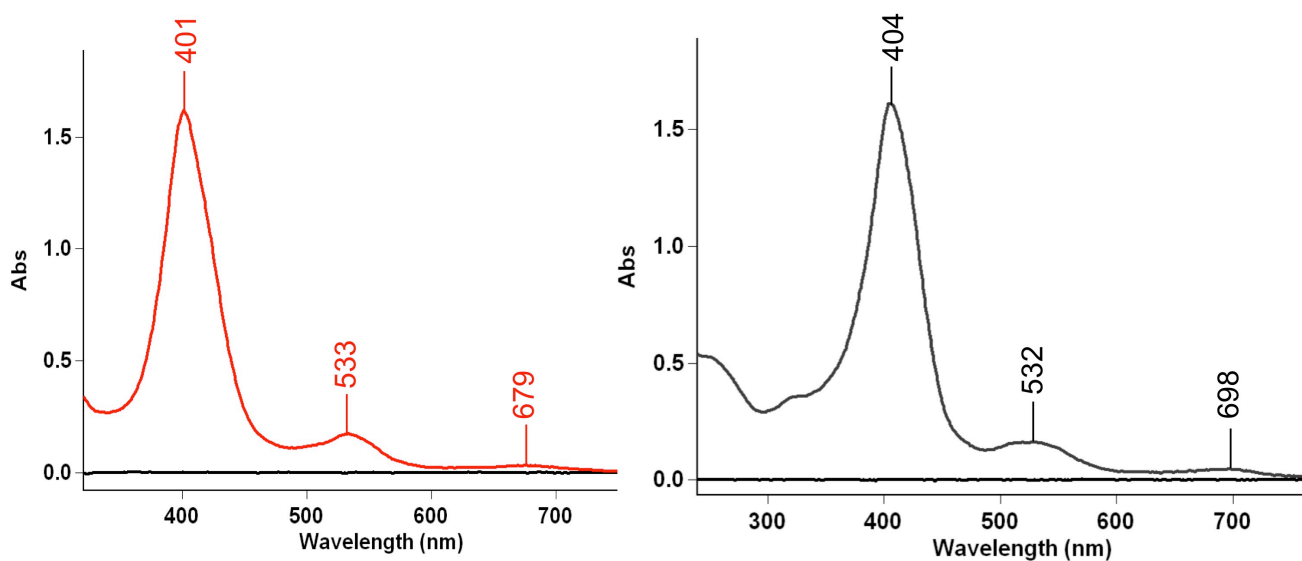
Figure S20. Optimized geometry of  $[(\text{tmpa})\text{Cu}^{\text{II}}(\text{NO}_2)]^+$  complex, BP86/6-31G(d).



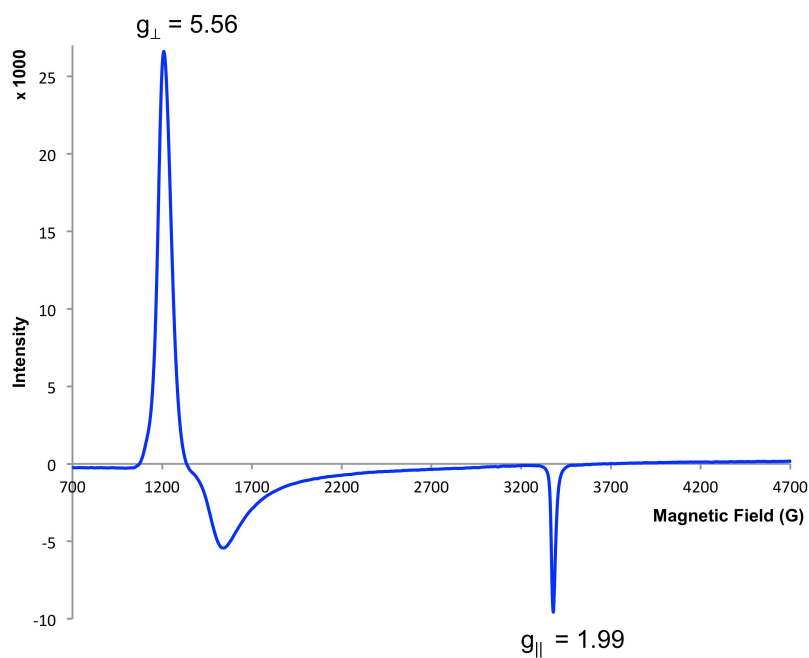
**Figure S21.** Optimized geometry of (TMPP)Fe<sup>II</sup>(NO) complex, BP86/6-31G(d).



**Figure S22.** Displacement ellipsoid plot (50% probability level) of [(TMPP)Fe<sup>III</sup>(THF)<sub>2</sub>]<sup>+</sup>, showing the atom-labeling scheme. Lattice solvent molecules and hydrogen atoms have been omitted for the sake of clarity. Selected bond lengths (Å) and angles (deg): Fe1–O6, 2.156(2); Fe1–O5, 2.154(2); Fe1–N1, 2.015(3); Fe1–N2, 2.025(3); Fe1–N3, 2.032(3); Fe1–N4, 2.014(3); O5–Fe1–O6, 179.68(10); N1–Fe1–N2, 89.40(10); N1–Fe1–N3, 179.70(11); N1–Fe1–N4, 90.98(10); N1–Fe1–O5, 90.58(9); N1–Fe1–O6, 89.67(9); N2–Fe1–N3, 90.30(10); N2–Fe1–N4, 179.26(10); N2–Fe1–O5, 90.68(9); N2–Fe1–O6, 89.13(9); N3–Fe1–N4, 89.32(10); N3–Fe1–O5, 89.38(9); N3–Fe1–O6, 90.37(9); N4–Fe1–O5, 89.96(9); N4–Fe1–O6, 90.24(9).



**Figure S23.** UV-vis spectra of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2](\text{SbF}_6)$  in acetone (left, red) and MeCN (right, black) 50  $\mu\text{M}$  in a 2-mm cuvette at RT.



**Figure S24.** EPR spectrum of  $[(\text{TMPP})\text{Fe}^{\text{III}}(\text{THF})_2](\text{SbF}_6)$  (1 mM) in THF/MeTHF (1:4) at 12 K.

## Computational details

All structures were fully optimized in the presence of solvent (acetone, PCM model) within Gaussian 09 program.<sup>1</sup> The BP86<sup>2,3</sup> functional with the 6-31G(d)<sup>4-18</sup> basis sets for all the atoms were used. In all case the nature of stationary points were checked (full optimization, NIMag= 0 for minima and NIMag=1 – for transition state).

Various spin states of the complexes were considered in our calculations (high spin (HS), intermediate spin (IS) and low spin (LS)). Single-point energy evaluation was performed at the OLYP<sup>19-22</sup>/6-311+G(d,p) level of theory, including solvent correction via PCM model.

- Gaussian 09 Revision A.2 Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, Millam, N. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R. E.; Stratmann, O.; Yazyev, A. J.; Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. Martin, R. L., Morokuma, K., Zakrzewski, V. G., Voth, G. A., Salvador, P., Dannenberg, J. J., Dapprich, S., Daniels, A. D., Farkas, O., Foresman, J. B., Ortiz, J. V., Cioslowski, J., and Fox, D. J. Gaussian, Inc., Wallingford CT **2009**.
- Becke, A. D. *Phys. Rev. A At. Mol. Opt. Phys.* **1988**, 38, 3098.
- Perdew, J. P. *Phys. Rev. B Condens. Matter Mater. Phys.* **1986**, 33, 8822.
- Ditchfield, R.; Hehre, W. J.; Pople, J. A. *J. Chem. Phys.* **1971**, 54, 724.
- Hariharan, P. C.; Pople, J. A. *Theor. Chim. Acta* **1973**, 28, 213.
- Hariharan, P. C.; Pople, J. A. *Mol. Phys.* **1974**, 27, 209.
- Francl, M. M.; Pietro, W. J.; Hehre, W. J.; Binkley, J. S.; Gordon, M. S.; DeFrees, D. J.; Pople, J. A. *J. Chem. Phys.* **1982**, 77, 3654.
- Rassolov, V. A.; Pople, J. A.; Ratner, M. A.; Windus, T. L. *J. Chem. Phys.* **1998**, 109, 1223.
- Rassolov, V. A.; Ratner, M. A.; Pople, J. A.; Redfern, P. C.; Curtiss, L. A. *J. Comput. Chem.* **2001**, 22, 976.
- Hehre, W. J.; Ditchfield, R.; Pople, J. A. *J. Chem. Phys.* **1972**, 56, 2257.
- Hehre, W. J.; Ditchfield, R.; Pople, J. A. *J. Chem. Phys.* **1972**, 56, 2257.
- Hariharan, P. C.; Pople, J. A. *Theor. Chim. Acta* **1973**, 28, 213.
- Hariharan, P. C.; Pople, J. A. *Mol. Phys.* **1974**, 27, 209.
- Gordon, M. S. *Chem. Phys. Lett.* **1980**, 76, 163.
- Francl, M. M.; Pietro, W. J.; Hehre, W. J.; Binkley, J. S.; DeFrees, D. J.; Pople, J. A.; Gordon, M. S. *J. Chem. Phys.* **1982**, 77, 3654.
- Binning, R. C. J.; L. A. Curtiss *J. Comput. Chem.* **1990**, 11, 1206.
- Blaudeau, J.-P.; McGrath, M. P.; Curtiss, L. A.; Radom, L. *J. Chem. Phys.* **1997**, 107, 5016.
- Rassolov, V. A.; Ratner, M. A.; Pople, J. A.; Redfern, P. C.; Curtiss, L. A. *J. Comput. Chem.* **2001**, 22, 976.
- Handy, N.C.; Cohen, A.J. *Mol. Phys.*, **2001**, 99, 403.
- Hoe, W.-M.; Cohen, A.J.; Handy, N.C. *Chem. Phys. Lett.*, **2001**, 341, 319.
- Lee, C.; Yang, W.; Parr, R.G. *Phys. Rev. B*, **1988**, 37, 785.
- Miehlich, B.; Savin, A.; Stoll, H.; Preuss, H. *Chem. Phys. Lett.*, **1989**, 157, 200.

**Table S2. Absolute energies (E, Hartree) of studied compounds at the OLYP/6-311+G(d,p) level, for different (HS – high spin, IS – intermediate spin, LS – low spin) states in the presence of solvent (acetone).**

Compounds	HS	IS	LS
	E	E	E
$[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$	-6266.15854	-6266.15260	–
<b>INIT1</b>	-6396.06924	-6396.07014	–
<b>TS1</b>	-6396.06936	-6396.06993	–
<b>mono-NO</b>	-6396.08632	-6396.08955	-6396.08947
<b>bis-NO</b>	-6525.96359	-6526.02858	-6526.03250
<b>NO</b>	–	–	-129.895151

**Table S3. Absolute energies (E, Hartree) and zero-point correction energy (ZPE, kcal/mol) of studied compounds at the BP86/6-31G(d) level, for different (HS – high spin, IS – intermediate spin, LS – low spin) states in the presence of solvent (acetone).**

Compounds	HS		IS		LS	
	E	ZPE	E	ZPE	E	ZPE
$[(\text{TMPP})\text{Fe}^{\text{III}}-\text{O}-\text{Cu}^{\text{II}}(\text{tmpa})]^+$	-6266.00970	648.85	-6266.021454	649.37	–	–
<b>INIT1</b>	-6395.90702	652.47	-6395.925409	653.50	–	–
<b>TS1</b>	-6395.90517	652.52	-6395.923805	653	–	–
<b>mono-NO</b>	-6395.92571	655.67	-6395.94428	654.00	-6395.96122	655.55
<b>bis-NO</b>	-6525.85606	658.31	-6525.90377	659.56	-6525.90605	659.87
<b>NO</b>	–	–	–	–	-129.895151	2.70

## XYZ Coordinates

NO

2

N -0.62483 -0.00000 0.00000  
O 0.54711 0.00000 0.00000

[(TMPP)Fe<sup>III</sup>-O-Cu<sup>II</sup>(tmpa)]<sup>+</sup>

135

Fe -0.05373 -1.86193 0.08131  
C 2.61924 -2.22350 -1.55535  
C 4.04803 -2.39887 -1.38933  
H 4.77127 -2.46632 -2.20086  
C 4.28455 -2.51253 -0.04429  
H 5.23772 -2.69142 0.45137  
C 3.00175 -2.40947 0.62202  
C 2.82882 -2.55675 2.02081  
C 1.56911 -2.64379 2.66265  
C 1.38831 -2.82795 4.08915  
H 2.18986 -2.89372 4.82354  
C 0.03886 -2.92076 4.31330  
H -0.46858 -3.09267 5.26165  
C -0.61637 -2.76179 3.03117  
C -2.01830 -2.81947 2.84496  
C -2.66140 -2.72701 1.58709  
C -4.09966 -2.73301 1.40812  
H -4.83292 -2.82589 2.20826  
C -4.33768 -2.58422 0.06521  
H -5.30241 -2.53262 -0.43774  
C -3.04741 -2.49163 -0.58701  
C -2.87259 -2.31388 -1.98098  
C -1.61301 -2.19795 -2.61638  
C -1.43287 -2.10925 -4.05086  
H -2.23537 -2.10702 -4.78725  
C -0.08275 -2.05000 -4.28383  
H 0.42414 -1.97490 -5.24488  
C 0.57355 -2.12880 -2.99450  
C 1.97837 -2.13560 -2.81458  
N 1.99153 -2.23690 -0.31583  
N 0.33267 -2.59287 2.03064  
N -2.03011 -2.57881 0.35683  
N -0.37605 -2.20875 -1.98219  
C 1.62965 1.00128 2.68957  
H 1.62106 0.03262 2.18152  
C 2.43973 1.25559 3.80188  
H 3.09366 0.46864 4.18804  
N -0.43510 3.82910 0.68812  
C 0.66336 4.38774 -0.14022  
C 0.81960 3.57900 -1.41423  
C 1.28110 4.12826 -2.61678  
H 1.49448 5.19964 -2.68137  
C 1.46509 3.28350 -3.72224  
H 1.82509 3.68896 -4.67302  
C 1.17504 1.91760 -3.58900  
H 1.30448 1.22152 -4.42225  
C 0.70354 1.44568 -2.35870  
H 0.45308 0.39396 -2.19248  
N 0.52584 2.25627 -1.29098  
O -0.20015 -0.14262 0.28478

Cu -0.31088 1.71869 0.47822  
N 0.80028 1.93255 2.16715  
C 2.39907 2.52877 4.38921  
H 3.02718 2.76630 5.25353  
C 1.54345 3.50000 3.84746  
H 1.48601 4.50539 4.27597  
C 0.75119 3.16441 2.74228  
C -0.26241 4.11597 2.13440  
C -1.77903 4.21294 0.18707  
C -2.80515 3.17404 0.59865  
C -4.15202 3.48038 0.82859  
H -4.49584 4.51734 0.76334  
C -4.55160 1.12769 1.21246  
H -5.20504 0.28370 1.44911  
C -3.18962 0.90271 0.98231  
H -2.74535 -0.09527 1.03358  
N -2.32958 1.90200 0.68167  
C -5.04156 2.43964 1.13735  
H -6.09916 2.65355 1.32089  
H 0.51098 5.46249 -0.35695  
H 1.59497 4.30344 0.44995  
H -2.06911 5.22724 0.52238  
H -1.72207 4.23881 -0.91726  
H 0.01385 5.17276 2.31368  
H -1.24114 3.94997 2.62196  
C -4.09755 -2.26097 -2.84040  
C 4.06345 -2.66088 2.86087  
C 4.34691 -3.81502 3.62137  
H 3.65253 -4.66192 3.59171  
C 5.51019 -3.91852 4.40115  
H 5.69481 -4.83573 4.96662  
C 6.42580 -2.84642 4.43403  
C 6.16136 -1.68396 3.67561  
H 6.87836 -0.85718 3.70938  
C 5.00199 -1.59966 2.90144  
H 4.81138 -0.69205 2.31825  
C 7.89699 -3.99533 5.94114  
H 7.12141 -4.18138 6.70708  
H 8.85446 -3.77373 6.43411  
H 8.00862 -4.89238 5.30425  
C -4.42162 -1.10380 -3.57930  
H -3.76962 -0.22545 -3.51455  
C -5.56952 -1.03993 -4.38588  
H -5.78761 -0.11977 -4.93446  
C -6.42515 -2.15763 -4.46923  
C -6.11376 -3.32714 -3.73998  
H -6.78010 -4.19221 -3.82057  
C -4.96983 -3.37265 -2.94003  
H -4.73642 -4.29046 -2.38927  
C -7.91981 -1.04510 -5.97978  
H -8.08954 -0.17202 -5.32243  
H -8.85446 -1.30020 -6.49967  
H -7.13880 -0.79959 -6.72321  
O 7.58867 -2.83206 5.15677  
O -7.56979 -2.21221 -5.21975  
C -2.87794 -2.99311 4.05924  
C -3.66324 -4.15095 4.23870  
H -3.63432 -4.94204 3.48110  
C -4.47113 -4.32632 5.37409  
H -5.05654 -5.24352 5.47835  
C -4.50893 -3.32261 6.36363



C	-3.73015	-2.15507	6.19947
H	-3.77291	-1.37976	6.97148
C	-2.92803	-1.99899	5.06678
H	-2.33539	-1.08488	4.94994
C	-6.06709	-4.54870	7.71505
H	-6.81833	-4.66564	6.91187
H	-6.57815	-4.38874	8.67537
H	-5.44705	-5.46249	7.77458
C	2.84484	-2.08464	-4.03427
C	2.79552	-3.10805	-5.01287
H	2.11767	-3.95668	-4.86999
C	3.60844	-3.06602	-6.14746
H	3.57335	-3.86450	-6.89581
C	4.50254	-1.98919	-6.34232
C	4.57031	-0.96122	-5.37956
H	5.24971	-0.11397	-5.50497
C	3.74862	-1.02115	-4.24209
H	3.80397	-0.21276	-3.50462
C	6.17917	-0.96714	-7.71968
H	5.65841	0.00513	-7.80115
H	6.66937	-1.20219	-8.67537
H	6.93919	-0.91167	-6.91836
O	-5.25810	-3.37998	7.50903
O	5.25216	-2.03916	-7.48739

### INIT1\_IS

137

Initial complex in acetone

Fe	0.22176	-1.83067	-0.27991
C	2.67357	-1.84956	-2.06462
C	4.05870	-2.25149	-2.07368
H	4.70494	-2.23912	-2.95031
C	4.36444	-2.66594	-0.80356
H	5.31202	-3.05397	-0.43283
C	3.17451	-2.48862	-0.00788
C	3.13883	-2.65632	1.38732
C	1.96248	-2.45224	2.12964
C	1.90603	-2.48718	3.57092
H	2.76494	-2.62844	4.22527
C	0.59044	-2.32964	3.92257
H	0.15650	-2.33268	4.92171
C	-0.16162	-2.19868	2.69760
C	-1.56546	-2.21970	2.64349
C	-2.26302	-2.25610	1.42284
C	-3.68714	-2.45676	1.31837
H	-4.35952	-2.59497	2.16395
C	-4.00160	-2.43476	-0.01636
H	-4.98102	-2.54574	-0.47949
C	-2.77626	-2.17596	-0.73161
C	-2.73412	-1.87679	-2.10353
C	-1.54313	-1.49902	-2.74723
C	-1.47186	-1.06769	-4.12052
H	-2.32775	-0.94958	-4.78318
C	-0.14767	-0.85098	-4.40242
H	0.29027	-0.51387	-5.34084
C	0.59977	-1.21251	-3.22273
C	1.99941	-1.35909	-3.19745
N	2.11446	-2.04277	-0.80089
N	0.69369	-2.22941	1.59298
N	-1.69609	-2.09045	0.15521
N	-0.26660	-1.54937	-2.17920
C	0.25199	1.56896	3.22825
H	0.45143	0.59479	2.77320
C	0.54026	1.81449	4.57418

H	0.97396	1.02021	5.18794
N	-1.03590	4.39381	0.58581
C	0.16418	5.09025	0.07001
C	0.65182	4.39552	-1.18562
C	1.29839	5.07160	-2.22723
H	1.44267	6.15472	-2.16601
C	1.75046	4.34037	-3.33565
H	2.25978	4.84589	-4.16201
C	1.52858	2.95603	-3.36808
H	1.85832	2.34337	-4.21173
C	0.86910	2.35293	-2.29160
H	0.67736	1.27680	-2.26793
N	0.44045	3.05076	-1.21390
O	0.45322	0.48554	0.24469
Cu	-0.50620	2.31943	0.40437
N	-0.28702	2.50309	2.40819
C	0.27463	3.08782	5.09949
H	0.50439	3.32378	6.14323
C	-0.29341	4.05635	4.26085
H	-0.52361	5.05994	4.63227
C	-0.57773	3.72583	2.92845
C	-1.31315	4.68535	2.01010
C	-2.22645	4.57470	-0.27816
C	-3.17935	3.40888	-0.09230
C	-4.57037	3.56371	-0.14764
H	-5.00135	4.55717	-0.30652
C	-4.78330	1.18911	0.22113
H	-5.37399	0.27951	0.35784
C	-3.38500	1.11907	0.26856
H	-2.87194	0.16944	0.44185
N	-2.58600	2.20046	0.11426
C	-5.38853	2.43454	0.00645
H	-6.47827	2.53046	-0.03021
H	-0.02484	6.16634	-0.11424
H	0.94403	5.02026	0.85190
H	-2.73306	5.54081	-0.08899
H	-1.87686	4.59087	-1.32661
H	-1.08478	5.73805	2.26511
H	-2.39808	4.54861	2.17452
N	1.77261	0.60500	0.51265
O	2.13738	1.74441	0.79433
C	-4.00195	-1.95298	-2.89830
C	4.38951	-3.04248	2.10992
C	4.46364	-4.25300	2.83076
H	3.59502	-4.92068	2.84965
C	5.62954	-4.63631	3.51261
H	5.64621	-5.58797	4.05019
C	6.75976	-3.79348	3.48769
C	6.70150	-2.57368	2.77661
H	7.58169	-1.92237	2.77345
C	5.53552	-2.20993	2.09921
H	5.50237	-1.25637	1.56098
C	8.04773	-5.28880	4.85188
H	7.31270	-5.32379	5.67741
H	9.06608	-5.30263	5.26631
H	7.90500	-6.16634	4.19425
C	-4.62231	-0.79822	-3.41582
H	-4.17779	0.18421	-3.22201
C	-5.80651	-0.87219	-4.16854
H	-6.25754	0.04833	-4.54831
C	-6.39727	-2.12823	-4.41584
C	-5.79180	-3.29500	-3.89807
H	-6.25823	-4.26505	-4.09896
C	-4.61384	-3.20397	-3.15262
H	-4.15012	-4.11842	-2.76621

C	-8.18489	-1.16669	-5.69487
H	-8.50913	-0.46458	-4.90421
H	-9.06608	-1.54552	-6.23247
H	-7.51818	-0.64044	-6.40326
O	7.94810	-4.06040	4.11422
O	-7.54544	-2.32510	-5.13732
C	-2.33459	-2.29485	3.92775
C	-3.00611	-3.47425	4.30979
H	-2.96200	-4.35328	3.65703
C	-3.72165	-3.55801	5.51578
H	-4.22348	-4.49357	5.77638
C	-3.77345	-2.44017	6.37356
C	-3.10325	-1.25177	6.00717
H	-3.15293	-0.38945	6.68018
C	-2.39617	-1.18544	4.80385
H	-1.88498	-0.25507	4.53288
C	-5.12647	-3.59277	7.98515
H	-5.92065	-3.86837	7.26648
H	-5.57949	-3.34947	8.95705
H	-4.42848	-4.44212	8.10544
C	2.79442	-1.06268	-4.42853
C	2.57756	-1.76846	-5.63811
H	1.81558	-2.55450	-5.67364
C	3.33301	-1.49732	-6.78047
H	3.16908	-2.05325	-7.70939
C	4.33544	-0.50158	-6.75135
C	4.57167	0.21005	-5.55705
H	5.34018	0.98570	-5.50523
C	3.80756	-0.08002	-4.41478
H	3.99764	0.47970	-3.49236
C	6.05009	0.68623	-7.93697
H	5.64462	1.69092	-7.71509
H	6.46010	0.67286	-8.95705
H	6.85081	0.44419	-7.21369
O	-4.43658	-2.40295	7.57204
O	5.01510	-0.30974	-7.92416

**TS1\_IS**

137

Fe	-0.645382	0.077981	-1.339409
C	-3.525808	-0.912272	-1.482112
C	-4.878476	-0.413685	-1.510218
H	-5.761994	-1.019227	-1.705228
C	-4.811111	0.926968	-1.232023
H	-5.627915	1.643666	-1.158440
C	-3.415492	1.257223	-1.1086668
C	-2.953549	2.551626	-0.781396
C	-1.600979	2.902717	-0.925737
C	-1.104081	4.253361	-0.829248
H	-1.689236	5.107905	-0.492259
C	0.180868	4.242118	-1.309906
H	0.856763	5.085240	-1.447024
C	0.499602	2.873207	-1.630180
C	1.723568	2.456912	-2.179837
C	2.037349	1.095527	-2.346226
C	3.329625	0.627061	-2.776588
H	4.133152	1.269699	-3.134568
C	3.341880	-0.735299	-2.604409
H	4.158842	-1.430252	-2.794031
C	2.045856	-1.110701	-2.103094
C	1.684123	-2.426101	-1.762245
C	0.357776	-2.767544	-1.441250
C	-0.096436	-4.126019	-1.278357

H	0.554352	-4.998740	-1.247452
C	-1.467993	-4.088885	-1.246679
H	-2.163222	-4.923735	-1.169163
C	-1.855962	-2.708007	-1.383272
C	-3.191201	-2.277090	-1.493901
N	-2.622456	0.135855	-1.298356
N	-0.588142	2.042669	-1.357083
N	1.232381	0.018847	-1.960741
N	-0.727490	-1.887795	-1.444225
C	0.351262	2.798922	2.741260
H	-0.063161	2.571103	1.755343
C	0.164438	4.051543	3.334128
H	-0.402079	4.821280	2.802394
N	2.135594	-0.386491	4.600700
C	1.068796	-1.055127	5.364262
C	0.395058	-2.117859	4.512876
C	-0.130347	-3.297713	5.052887
H	-0.029348	-3.499554	6.124801
C	-0.786552	-4.203265	4.204713
H	-1.207356	-5.131956	4.603647
C	-0.885426	-3.894884	2.840787
H	-1.380355	-4.566500	2.133414
C	-0.329274	-2.697948	2.375893
H	-0.389424	-2.409701	1.322445
N	0.302076	-1.817525	3.188794
O	-0.106554	0.130104	1.002243
Cu	1.164187	-0.094768	2.606744
N	1.044312	1.796817	3.334981
C	0.698366	4.282273	4.609530
H	0.553783	5.242331	5.115622
C	1.425651	3.254728	5.226826
H	1.866962	3.394522	6.219889
C	1.596132	2.037673	4.553916
C	2.496239	0.943812	5.114721
C	3.296711	-1.240568	4.294700
C	3.967223	-0.774860	3.010864
C	5.355160	-0.813376	2.834264
H	5.998601	-1.161951	3.649453
C	5.031990	0.037447	0.601112
H	5.400279	0.367731	-0.373727
C	3.654711	0.052581	0.856268
H	2.941497	0.389824	0.098490
N	3.118464	-0.338835	2.036634
C	5.900891	-0.402441	1.607517
H	6.983768	-0.421820	1.447046
H	1.431622	-1.487938	6.321585
H	0.321583	-0.278028	5.617235
H	4.029821	-1.284350	5.127729
H	2.917248	-2.269026	4.146084
H	2.508070	0.977732	6.224450
H	3.531256	1.156180	4.785493
N	-1.385153	0.359075	1.408083
O	-1.535664	0.408816	2.623738
C	2.708560	-3.514474	-1.837697
C	-3.950081	3.574188	-0.343221
C	-4.225615	4.731302	-1.100311
H	-3.707484	4.886135	-2.053303
C	-5.173530	5.676719	-0.677549
H	-5.369516	6.553376	-1.300899
C	-5.868281	5.475201	0.532984
C	-5.604828	4.319541	1.302151
H	-6.151820	4.179741	2.239930
C	-4.666185	3.384752	0.865106
H	-4.463056	2.494981	1.471220
C	-7.124529	7.500993	0.294378

H	-6.240616	8.157278	0.179031	C	-0.02	0.01	-0.01
H	-7.895892	8.025666	0.876644	C	-0.02	0.00	-0.02
H	-7.526753	7.247153	-0.704935	H	-0.03	-0.01	-0.04
C	3.122574	-4.202635	-0.679976	C	-0.03	0.00	0.00
H	2.697763	-3.920296	0.290340	H	-0.04	0.00	0.00
C	4.070381	-5.237841	-0.736659	C	-0.03	0.01	0.02
H	4.368445	-5.744656	0.185341	C	-0.04	0.02	0.02
C	4.625923	-5.606427	-1.978746	C	-0.05	0.02	0.00
C	4.218821	-4.929156	-3.149597	C	-0.05	0.02	0.01
H	4.648678	-5.236776	-4.108039	H	-0.05	0.03	0.04
C	3.274728	-3.904102	-3.076186	C	-0.06	0.01	-0.02
H	2.951964	-3.402807	-3.995353	H	-0.05	0.00	-0.03
C	5.978611	-7.328937	-1.008976	C	-0.06	0.00	-0.05
H	6.473414	-6.670419	-0.268923	C	-0.06	0.01	-0.07
H	6.702348	-8.069149	-1.379945	N	-0.05	0.00	-0.11
H	5.131483	-7.852786	-0.525779	N	-0.03	0.00	-0.07
O	-6.809309	6.323826	1.046277	N	-0.03	0.01	0.01
O	5.552524	-6.595810	-2.161299	N	-0.06	0.01	-0.05
C	2.739899	3.483894	-2.572404	C	-0.03	0.00	0.09
C	3.052827	3.703679	-3.928790	H	-0.02	0.00	0.09
H	2.531355	3.124702	-4.699342	C	-0.04	0.00	0.10
C	4.007042	4.657260	-4.318633	H	-0.05	-0.01	0.10
H	4.214459	4.803303	-5.382080	N	0.04	0.02	0.04
C	4.674569	5.417592	-3.336910	C	0.05	0.01	0.05
C	4.373887	5.207469	-1.972621	C	0.05	0.01	0.05
H	4.906100	5.801870	-1.223097	C	0.05	0.00	0.04
C	3.422368	4.255495	-1.601705	H	0.05	-0.01	0.03
H	3.205701	4.096138	-0.539128	C	0.04	0.02	0.03
C	5.954108	6.626880	-4.961505	H	0.04	0.01	0.01
H	6.371232	5.726465	-5.451857	C	0.03	0.04	0.03
H	6.718727	7.416854	-4.936757	H	0.03	0.05	0.02
H	5.075488	6.982932	-5.532544	C	0.03	0.04	0.04
C	-4.271381	-3.303081	-1.596477	H	0.03	0.06	0.05
C	-4.296963	-4.212762	-2.683280	N	0.03	0.03	0.06
H	-3.524380	-4.145367	-3.457045	O	0.36	0.02	0.04
C	-5.300416	-5.174448	-2.798585	Cu	0.09	0.02	0.12
H	-5.328844	-5.866636	-3.645906	N	0.00	0.01	0.09
C	-6.315343	-5.263292	-1.819498	C	-0.03	0.00	0.09
C	-6.304855	-4.372151	-0.726530	H	-0.03	0.00	0.09
H	-7.072102	-4.424751	0.050750	C	0.00	0.01	0.09
C	-5.291654	-3.405785	-0.627901	H	0.00	0.02	0.08
H	-5.284710	-2.726023	0.231459	C	0.01	0.02	0.08
C	-8.310489	-6.357416	-1.071147	C	0.03	0.02	0.06
H	-7.924123	-6.602606	-0.063161	C	0.07	0.03	0.07
H	-8.941132	-7.183043	-1.431486	C	0.08	0.01	0.06
H	-8.912138	-5.429942	-1.018889	C	0.07	-0.01	0.05
O	5.620455	6.371536	-3.592291	H	0.08	-0.01	0.05
O	-7.249091	-6.239771	-2.024829	C	0.06	-0.05	0.03
				H	0.05	-0.08	0.02
				C	0.06	-0.02	0.05
				H	0.05	-0.03	0.05
				N	0.08	0.01	0.06
				C	0.06	-0.04	0.03
				H	0.06	-0.06	0.02
				H	0.07	0.01	0.04
				H	0.05	0.00	0.07
				H	0.05	0.06	0.08
				H	0.08	0.02	0.09
				H	0.05	0.00	0.05
				H	0.02	0.04	0.04
				N	0.28	-0.15	-0.30
				O	-0.08	-0.04	-0.35
				C	-0.04	0.01	0.02
				C	-0.02	-0.02	-0.03
				C	-0.01	-0.01	-0.01
NORMALMODES 405							
Fe	-0.04	0.00	-0.09				
C	-0.05	0.00	-0.09				
C	-0.07	0.00	-0.06				
H	-0.07	0.00	-0.05				
C	-0.06	0.00	-0.06				
H	-0.05	0.00	-0.05				
C	-0.04	-0.01	-0.07				
C	-0.03	-0.02	-0.05				
C	-0.03	-0.02	-0.02				
C	-0.02	-0.03	0.02				
H	-0.02	-0.04	0.06				
C	-0.01	-0.01	0.03				
H	-0.01	-0.01	0.07				
C	-0.02	0.00	-0.03				
C	-0.02	0.00	-0.02				

H -0.01 0.00 -0.01  
C 0.00 0.00 0.00  
H 0.01 0.01 0.01  
C 0.00 -0.01 0.00  
C -0.01 -0.02 -0.01  
H -0.01 -0.03 -0.01  
C -0.02 -0.03 -0.03  
H -0.03 -0.04 -0.04  
C 0.01 0.00 0.02  
H 0.01 0.00 0.02  
H 0.01 0.00 0.02  
H 0.01 0.01 0.02  
C -0.03 0.02 0.01  
H -0.02 0.02 0.02  
C -0.03 0.02 0.01  
H -0.02 0.02 0.01  
C -0.04 0.02 0.01  
C -0.05 0.01 0.01  
H -0.05 0.01 0.01  
C -0.05 0.01 0.01  
H -0.06 0.00 0.01  
C -0.02 0.02 0.00  
H -0.02 0.02 0.00  
H -0.02 0.02 0.00  
H -0.02 0.02 0.01  
O 0.01 -0.01 0.01  
O -0.03 0.02 0.00  
C -0.01 0.00 -0.02  
C -0.02 0.02 -0.02  
H -0.03 0.04 -0.02  
C -0.01 0.02 -0.01  
H -0.02 0.03 -0.01  
C 0.00 0.00 0.00  
C 0.00 -0.02 0.00  
H 0.01 -0.03 0.00  
C 0.00 -0.01 -0.01  
H 0.00 -0.03 -0.01  
C 0.00 0.01 0.00  
H -0.01 0.01 -0.01  
H 0.01 0.00 0.01  
H 0.00 0.02 0.01  
C -0.05 0.00 -0.05  
C -0.06 -0.01 -0.04  
H -0.06 -0.01 -0.05  
C -0.05 -0.02 -0.02  
H -0.05 -0.03 -0.02  
C -0.04 -0.02 -0.01  
C -0.03 -0.01 -0.02  
H -0.03 -0.01 -0.01  
C -0.04 0.00 -0.04  
H -0.04 0.01 -0.05  
C -0.02 -0.02 0.01  
H -0.02 -0.01 0.01  
H -0.02 -0.03 0.03  
H -0.03 -0.03 0.01  
O 0.00 0.00 0.00  
O -0.03 -0.03 0.01  
-49.5795

**TS1\_HS**

137

Fe -0.638549 0.168130 -1.105222  
C -3.018175 -1.870703 -1.436591

C -4.463211 -1.941471 -1.354220  
H -5.060000 -2.838697 -1.513321  
C -4.909277 -0.677823 -1.059778  
H -5.939785 -0.345385 -0.940141  
C -3.741475 0.176475 -0.983571  
C -3.800069 1.572643 -0.750197  
C -2.705425 2.456584 -0.907148  
C -2.793210 3.897209 -0.787399  
H -3.682154 4.452621 -0.491308  
C -1.569789 4.405457 -1.146927  
H -1.276596 5.452928 -1.206764  
C -0.710057 3.276918 -1.436975  
C 0.646688 3.381360 -1.827301  
C 1.492248 2.269223 -2.055297  
C 2.899458 2.370698 -2.382272  
H 3.446746 3.301643 -2.524957  
C 3.390076 1.090354 -2.449150  
H 4.413076 0.780350 -2.658666  
C 2.284992 0.196040 -2.172637  
C 2.394539 -1.215343 -2.141070  
C 1.304552 -2.098010 -1.943430  
C 1.392234 -3.535747 -2.074566  
H 2.308188 -4.096518 -2.255960  
C 0.115916 -4.029579 -1.959890  
H -0.200332 -5.070615 -2.013274  
C -0.762798 -2.895835 -1.769198  
C -2.172452 -2.984212 -1.659497  
N -2.595076 -0.563656 -1.237149  
N -1.416101 2.091051 -1.280220  
N 1.128936 0.932115 -1.928720  
N -0.017269 -1.720094 -1.726541  
C 0.064480 2.425563 3.565224  
H -0.388749 2.401720 2.571856  
C -0.198961 3.476885 4.448330  
H -0.866207 4.287703 4.142905  
N 2.261338 -0.890594 4.507090  
C 1.323911 -1.840437 5.145027  
C 0.682940 -2.722596 4.091600  
C 0.286338 -4.040174 4.350347  
H 0.466961 -4.477153 5.337317  
C -0.337305 -4.777279 3.332825  
H -0.657974 -5.808306 3.511409  
C -0.526917 -4.171171 2.082444  
H -0.996043 -4.705996 1.251796  
C -0.094816 -2.853832 1.898509  
H -0.209902 -2.346658 0.936390  
N 0.492422 -2.128265 2.880759  
O 0.071523 0.153899 1.053222  
Cu 1.239332 -0.260897 2.727948  
N 0.881775 1.387093 3.870289  
C 0.396283 3.455756 5.717931  
H 0.199549 4.250546 6.444162  
C 1.255306 2.395964 6.038453  
H 1.750766 2.345870 7.013221  
C 1.491230 1.393100 5.088238  
C 2.524682 0.306239 5.332636  
C 3.497638 -1.527540 3.996778  
C 4.087892 -0.677126 2.885622  
C 5.467561 -0.522938 2.705107  
H 6.164704 -1.015654 3.390053  
C 4.997271 0.890389 0.805846  
H 5.307172 1.524740 -0.028736  
C 3.633489 0.688831 1.052102  
H 2.870515 1.154888 0.421921  
N 3.174586 -0.078339 2.068701

C	5.932829	0.271428	1.645722	H	-4.677340	-8.904798	-0.319811
H	7.006856	0.410592	1.487754	H	-5.640014	-9.709229	-1.611889
H	1.813713	-2.448390	5.931501	H	-6.177344	-8.122209	-0.950938
H	0.537784	-1.237990	5.638634	O	2.855997	8.613578	-2.387569
H	4.239025	-1.705636	4.799966	O	-4.540620	-8.138683	-2.278892
H	3.216439	-2.515891	3.588421				
H	2.594902	0.057268	6.409116				
H	3.512480	0.706451	5.037768	NORMALMODES 405			
N	-1.256870	0.279553	1.270520	Fe	0.05	0.00	0.04
O	-1.636941	0.125159	2.421967	C	0.03	0.00	0.02
C	3.740154	-1.825504	-2.391135	C	0.03	0.00	0.00
C	-5.122320	2.140686	-0.337028	H	0.03	0.01	-0.02
C	-5.829941	3.057481	-1.142000	C	0.03	0.00	0.00
H	-5.410865	3.360975	-2.107822	H	0.03	0.01	-0.01
C	-7.073185	3.578250	-0.746792	C	0.03	0.00	0.02
H	-7.594115	4.278102	-1.405401	C	0.02	0.00	0.02
C	-7.634460	3.183914	0.485244	C	0.02	0.00	0.02
C	-6.939779	2.265297	1.303885	H	0.01	0.00	-0.02
H	-7.383549	1.970163	2.260388	C	0.02	0.00	0.00
C	-5.707895	1.751962	0.893013	H	0.01	0.00	-0.02
H	-5.174842	1.046359	1.539711	C	0.02	0.00	0.03
C	-9.576103	4.558673	0.173355	C	0.02	-0.01	0.02
H	-9.006884	5.493104	0.012552	C	0.02	-0.01	0.01
H	-10.489475	4.778775	0.744573	C	0.01	0.00	0.00
H	-9.848857	4.120837	-0.804859	H	0.01	0.00	0.01
C	4.399086	-2.562397	-1.385706	C	0.02	0.00	-0.01
H	3.924648	-2.673233	-0.404235	H	0.02	0.00	-0.02
C	5.656401	-3.149401	-1.604173	C	0.03	-0.01	-0.01
H	6.135401	-3.706177	-0.794563	C	0.03	-0.01	-0.02
C	6.282578	-3.006046	-2.859265	C	0.02	0.00	-0.03
C	5.634559	-2.274771	-3.879895	C	0.02	-0.01	-0.02
H	6.125999	-2.179339	-4.853695	H	0.02	-0.01	-0.01
C	4.385043	-1.695979	-3.645302	C	0.02	0.00	-0.01
H	3.889339	-1.143349	-4.451016	H	0.02	0.00	0.01
C	8.191701	-4.292432	-2.183925	C	0.03	0.00	-0.02
H	8.413944	-3.676010	-1.292917	C	0.03	0.00	0.00
H	9.133443	-4.615421	-2.650607	N	0.04	0.00	0.04
H	7.606757	-5.180561	-1.880352	N	0.03	-0.01	0.05
O	-8.834877	3.625771	0.975234	N	0.03	-0.01	0.01
O	7.504085	-3.531494	-3.188943	N	0.03	0.00	-0.04
C	1.229346	4.750952	-1.997859	C	0.06	0.05	-0.08
C	1.652936	5.205278	-3.264131	H	0.11	0.08	-0.11
H	1.542126	4.548927	-4.134575	C	0.03	0.03	-0.07
C	2.200869	6.486007	-3.443789	H	0.05	0.04	-0.07
H	2.508594	6.801051	-4.444288	N	-0.02	0.00	-0.04
C	2.337602	7.346451	-2.335307	C	-0.02	0.00	-0.04
C	1.921788	6.906571	-1.058601	C	-0.01	0.00	-0.04
H	2.041007	7.580438	-0.203834	C	-0.01	0.00	-0.03
C	1.376656	5.630453	-0.897919	H	-0.02	0.01	-0.02
H	1.068949	5.300323	0.100423	C	-0.01	0.00	-0.02
C	3.293082	9.098243	-3.666931	H	-0.01	0.00	-0.02
H	4.107400	8.472770	-4.077610	C	-0.01	-0.01	-0.03
H	3.667769	10.115872	-3.485235	H	-0.02	-0.01	-0.02
H	2.457459	9.136114	-4.390326	C	-0.01	-0.01	-0.03
C	-2.804216	-4.334606	-1.791687	H	-0.03	0.00	-0.03
C	-2.684794	-5.081053	-2.990040	N	0.00	0.00	-0.04
H	-2.132075	-4.655983	-3.834918	O	-0.40	-0.03	-0.06
C	-3.274249	-6.339980	-3.122103	Cu	-0.05	0.00	-0.06
H	-3.188162	-6.907631	-4.054399	N	0.04	0.03	-0.08
C	-4.004291	-6.899803	-2.049382	C	-0.02	0.00	-0.05
C	-4.135268	-6.175076	-0.847142	H	-0.04	-0.02	-0.03
H	-4.689025	-6.584925	0.001512	C	-0.03	-0.01	-0.04
C	-3.540937	-4.907509	-0.732931	H	-0.06	-0.03	-0.03
H	-3.645303	-4.356762	0.208572	C	0.00	0.01	-0.06
C	-5.300314	-8.740270	-1.218658	C	-0.02	0.00	-0.04

C -0.03 0.00 -0.05  
 C -0.04 0.00 -0.04  
 C -0.03 0.01 -0.03  
 H -0.04 0.01 -0.02  
 C -0.02 0.02 -0.02  
 H -0.02 0.03 -0.01  
 C -0.03 0.01 -0.04  
 H -0.02 0.01 -0.04  
 N -0.04 0.00 -0.05  
 C -0.03 0.02 -0.02  
 H -0.03 0.03 -0.01  
 H -0.02 0.01 -0.03  
 H -0.02 0.00 -0.04  
 H -0.03 -0.01 -0.05  
 H -0.04 0.00 -0.05  
 H -0.03 0.00 -0.04  
 H -0.01 -0.01 -0.03  
 N -0.36 0.05 0.42  
 O 0.11 -0.06 0.55  
 C 0.03 0.00 -0.01  
 C 0.02 0.00 0.00  
 C 0.02 0.00 0.00  
 H 0.02 0.00 0.00  
 C 0.02 0.00 -0.01  
 H 0.02 -0.01 -0.01  
 C 0.01 0.00 -0.01  
 C 0.01 0.00 -0.01  
 H 0.00 0.00 -0.01  
 C 0.01 0.00 0.00  
 H 0.00 0.00 0.00  
 C 0.01 0.00 -0.01  
 H 0.01 0.00 -0.01  
 H 0.01 0.00 -0.02  
 H 0.01 0.00 -0.02  
 C 0.02 0.00 -0.01  
 H 0.02 0.00 -0.01  
 C 0.02 0.00 0.00  
 H 0.02 0.00 0.00  
 C 0.02 -0.01 0.00  
 C 0.03 -0.01 -0.01  
 H 0.03 -0.01 -0.01  
 C 0.03 -0.01 -0.01  
 H 0.03 -0.01 -0.01  
 C 0.02 -0.01 0.00  
 H 0.02 -0.01 0.00  
 H 0.02 -0.01 0.01  
 H 0.02 -0.01 0.00  
 O 0.01 0.00 -0.01  
 O 0.02 -0.01 0.00  
 C 0.02 -0.01 0.01  
 C 0.01 -0.01 0.01  
 H 0.01 -0.01 0.01  
 C 0.00 -0.01 0.00  
 H 0.00 -0.01 0.00  
 C 0.00 0.00 0.00  
 C 0.00 0.00 0.00  
 H 0.00 0.01 0.00  
 C 0.01 0.00 0.00  
 H 0.01 0.00 0.01  
 C 0.00 0.00 -0.01  
 H 0.00 0.00 0.00  
 H 0.00 0.00 -0.01  
 H 0.00 -0.01 -0.01  
 C 0.02 0.00 0.00  
 C 0.02 0.00 0.00

H 0.02 0.00 0.00  
 C 0.01 0.00 0.00  
 H 0.01 0.00 0.00  
 C 0.01 0.01 0.00  
 C 0.01 0.01 0.00  
 H 0.01 0.01 0.00  
 C 0.01 0.01 0.00  
 H 0.01 0.01 0.00  
 C 0.00 0.01 0.00  
 H 0.00 0.01 0.00  
 H 0.00 0.01 0.00  
 H 0.00 0.01 0.00  
 O 0.00 0.00 -0.01  
 O 0.00 0.01 0.00  
 -82.9500

**mono-NO LS**

137

Final complex in acetone

Fe 0.17437 -2.30689 -0.21933  
 C 2.88849 -2.56758 -1.56501  
 C 4.25724 -2.93830 -1.29179  
 H 5.03843 -3.03346 -2.04460  
 C 4.34479 -3.17310 0.05316  
 H 5.21394 -3.49418 0.62548  
 C 3.03590 -2.91964 0.60960  
 C 2.76652 -2.92672 1.98733  
 C 1.48054 -2.68614 2.49927  
 C 1.18997 -2.56587 3.90991  
 H 1.93044 -2.63087 4.70585  
 C -0.16067 -2.37693 4.02346  
 H -0.75290 -2.26940 4.93130  
 C -0.69902 -2.38249 2.68315  
 C -2.07371 -2.35242 2.40126  
 C -2.56802 -2.51084 1.09582  
 C -3.97200 -2.63400 0.78075  
 H -4.77887 -2.61551 1.51194  
 C -4.06422 -2.77446 -0.57758  
 H -4.96167 -2.88840 -1.18415  
 C -2.71916 -2.69141 -1.10051  
 C -2.44171 -2.59321 -2.47346  
 C -1.14917 -2.34297 -2.96325  
 C -0.85016 -2.11607 -4.35822  
 H -1.58762 -2.09388 -5.15908  
 C 0.50417 -1.94373 -4.45309  
 H 1.09402 -1.74732 -5.34718  
 C 1.04365 -2.13000 -3.12577  
 C 2.41670 -2.24768 -2.84950  
 N 2.12022 -2.60856 -0.39968  
 N 0.31823 -2.55767 1.74100  
 N -1.79113 -2.55967 -0.06433  
 N 0.01722 -2.33001 -2.19805  
 C -0.19153 1.87645 2.55991  
 H 0.17758 0.88728 2.27834  
 C -0.09602 2.35012 3.87352  
 H 0.33174 1.70914 4.64930  
 N -1.22906 4.17273 -0.61523  
 C 0.08890 4.79135 -0.91087  
 C 0.86880 3.91134 -1.86834  
 C 1.80015 4.41857 -2.78232  
 H 1.96198 5.49840 -2.85416  
 C 2.51190 3.52230 -3.59356  
 H 3.24630 3.89366 -4.31502  
 C 2.25727 2.14846 -3.47260



C	0.08313	2.23358	2.40363	C	-5.63928	-2.93048	-5.34110
H	0.40464	1.21882	2.15654	C	-5.25432	-4.00737	-4.51151
C	0.30682	2.77904	3.67250	H	-5.78894	-4.95883	-4.59976
H	0.79258	2.17512	4.44410	C	-4.20453	-3.85386	-3.60291
N	-1.21229	4.38481	-0.79483	H	-3.91217	-4.70008	-2.97128
C	0.08554	4.92928	-1.24725	C	-7.09711	-2.11599	-7.06370
C	0.75916	3.95618	-2.19830	H	-7.45664	-1.24472	-6.48512
C	1.59578	4.37830	-3.23813	H	-7.92451	-2.52701	-7.65989
H	1.75191	5.44815	-3.40814	H	-6.27900	-1.79863	-7.73685
C	2.22068	3.41787	-4.04723	O	7.01157	-3.90278	5.73093
H	2.87962	3.72591	-4.86492	O	-6.67814	-3.18234	-6.19794
C	1.97357	2.06095	-3.79509	C	-3.00535	-2.05673	3.50470
H	2.43137	1.27295	-4.39954	C	-3.91240	-3.07367	3.86641
C	1.11711	1.71540	-2.74468	H	-3.90977	-4.01888	3.31216
H	0.88495	0.66976	-2.52518	C	-4.81130	-2.91425	4.93410
N	0.51851	2.63602	-1.95010	H	-5.49271	-3.73100	5.18611
O	-0.71696	0.27277	-0.34538	C	-4.81397	-1.71081	5.66875
Cu	-0.83739	2.29081	-0.52436	C	-3.91065	-0.68180	5.32036
N	-0.51325	2.92147	1.39736	H	-3.92487	0.24984	5.89552
C	-0.08744	4.10200	3.91863	C	-3.02183	-0.85731	4.25693
H	0.08270	4.56687	4.89475	H	-2.32967	-0.04901	3.99573
C	-0.69777	4.82423	2.88373	C	-6.57195	-2.46442	7.11677
H	-1.01593	5.86130	3.03035	H	-7.27732	-2.70059	6.29847
C	-0.90696	4.20235	1.64601	H	-7.12619	-2.04948	7.97098
C	-1.64470	4.90327	0.51780	H	-6.04853	-3.38709	7.42915
C	-2.27293	4.43903	-1.82405	C	3.50069	-1.92493	-3.93778
C	-3.29319	3.34014	-1.57953	C	3.56181	-2.84437	-5.01395
C	-4.64777	3.48381	-1.90275	H	2.86861	-3.69222	-5.03822
H	-5.01217	4.43184	-2.31065	C	4.50475	-2.70386	-6.03433
C	-5.00036	1.21296	-1.16294	H	4.55469	-3.42362	-6.85791
H	-5.63748	0.34387	-0.97795	C	5.42374	-1.63056	-6.01173
C	-3.63651	1.14842	-0.85523	C	5.38300	-0.70553	-4.94836
H	-3.18678	0.24488	-0.43366	H	6.08008	0.13538	-4.90479
N	-2.78992	2.18657	-1.05626	C	4.43125	-0.86420	-3.92740
C	-5.51751	2.40213	-1.69611	H	4.40944	-0.14016	-3.10532
H	-6.58056	2.49175	-1.94075	C	7.26983	-0.51692	-7.06375
H	-0.01585	5.93170	-1.70817	H	6.77698	0.47198	-7.10935
H	0.72063	5.04733	-0.34903	H	7.87205	-0.66898	-7.97098
H	-2.75992	5.43262	-1.87393	H	7.92451	-0.56160	-6.17369
H	-1.78981	4.26599	-2.80409	O	-5.64169	-1.44203	6.72691
H	-1.53062	6.00279	0.59160	O	6.30606	-1.58195	-7.05754
H	-2.72484	4.68612	0.61809				
N	0.34845	-4.15589	-0.46592				
O	0.90478	-5.11455	-0.87900				
N	0.31777	-0.42834	-0.06165				
O	1.34846	0.13213	0.33192				
C	-3.50175	-2.62941	-3.49262				
C	3.91352	-3.15018	2.98321				
C	3.88535	-4.32561	3.76309				
H	3.05881	-5.03373	3.63645				
C	4.89761	-4.62157	4.68984				
H	4.84091	-5.54861	5.26628				
C	5.97180	-3.72363	4.85775				
C	6.01392	-2.53877	4.08881				
H	6.84748	-1.84414	4.23563				
C	5.00107	-2.26169	3.16782				
H	5.04115	-1.33170	2.59018				
C	7.00290	-5.09117	6.53772				
H	6.10994	-5.12915	7.18896				
H	7.90830	-5.03300	7.15884				
H	7.04196	-6.00279	5.91291				
C	-3.90331	-1.56325	-4.32337				
H	-3.39073	-0.59818	-4.24574				
C	-4.95798	-1.70008	-5.24101				
H	-5.24159	-0.84624	-5.86190				

(TMPP)Fe<sup>II</sup>(NO)

95

Fe	-0.022283	0.025569	0.170534
C	2.995722	-0.038785	0.094107
C	4.142632	0.829757	0.064419
H	5.163409	0.515156	0.221960
C	3.687326	2.082146	-0.205030
H	4.262170	2.990937	-0.303561
C	2.253922	1.999327	-0.299248
C	1.427030	3.116177	-0.455927
C	0.033865	3.044867	-0.348610
C	-0.830984	4.195243	-0.380686
H	-0.506726	5.211519	-0.547044
C	-2.092786	3.749602	-0.142160
H	-2.999521	4.331962	-0.075407
C	-2.018383	2.316232	-0.026264
C	-3.138420	1.488286	0.109393
C	-3.059510	0.092813	0.029877
C	-4.214098	-0.757617	-0.102481



H -5.238331 -0.415439 -0.096427  
 C -3.755862 -2.024533 -0.282929  
 H -4.332372 -2.922812 -0.445822  
 C -2.318941 -1.964133 -0.221489  
 C -1.487914 -3.086705 -0.303176  
 C -0.098347 -3.005424 -0.170247  
 C 0.764612 -4.154326 -0.097851  
 H 0.438167 -5.179735 -0.186672  
 C 2.026276 -3.692991 0.112717  
 H 2.935255 -4.267088 0.214602  
 C 1.949474 -2.256846 0.151978  
 C 3.068973 -1.424712 0.268635  
 N 1.840072 0.690782 -0.118801  
 N -0.706702 1.895852 -0.143664  
 N -1.902712 -0.657316 -0.047188  
 N 0.642335 -1.844744 -0.024577  
 O 0.120438 0.029165 1.983374  
 O 0.932479 -0.343465 2.772123  
 C -2.107803 -4.433722 -0.502276  
 C 2.070607 4.444801 -0.701925  
 C 2.068414 5.452728 0.270259  
 H 1.589259 5.270134 1.228136  
 C 2.678708 6.689655 0.045655  
 H 2.659275 7.440146 0.827244  
 C 3.310696 6.937276 -1.179528  
 C 3.322940 5.936945 -2.164433  
 H 3.813347 6.141827 -3.111273  
 C 2.713425 4.711836 -1.924468  
 H 2.729362 3.948593 -2.697586  
 C 3.958389 9.153240 -0.545012  
 H 2.944124 9.482817 -0.289544  
 H 4.499653 9.976123 -1.013953  
 H 4.481946 8.846415 0.368444  
 C -1.916301 -5.140883 -1.695834  
 H -1.317714 -4.699107 -2.487757  
 C -2.483331 -6.401432 -1.901009  
 H -2.315069 -6.911304 -2.842355  
 C -3.261553 -6.982489 -0.891958  
 C -3.461288 -6.288002 0.311818  
 H -4.061138 -6.751485 1.089189  
 C -2.892458 -5.034427 0.499139  
 H -3.050302 -4.513780 1.439515  
 C -3.685598 -8.957802 -2.177275  
 H -4.093982 -8.428280 -3.046436  
 H -4.237119 -9.886930 -2.027144  
 H -2.627826 -9.186342 -2.354661  
 O 3.936262 8.102809 -1.508829  
 O -3.860511 -8.203869 -0.979765  
 C -4.490104 2.109596 0.271650  
 C -5.193031 1.974755 1.475343  
 H -4.742344 1.425541 2.297432  
 C -6.460347 2.536194 1.652353  
 H -6.966410 2.413579 2.602752  
 C -7.053672 3.249040 0.603076  
 C -6.364955 3.388740 -0.612441  
 H -6.838760 3.936358 -1.421504  
 C -5.104369 2.826684 -0.771406  
 H -4.589752 2.935685 -1.721962  
 C -9.031946 3.718583 1.867752  
 H -9.247738 2.669466 2.103057  
 H -9.967723 4.250352 1.690161  
 H -8.505430 4.179777 2.711967  
 C 4.398088 -2.058828 0.531317  
 C 4.666174 -2.661369 1.774220  
 H 3.903581 -2.650370 2.547945

C 5.891331 -3.262689 2.032934  
 H 6.097065 -3.721305 2.995387  
 C 6.890827 -3.283069 1.047237  
 C 6.642051 -2.692152 -0.198036  
 H 7.391685 -2.698986 -0.980712  
 C 5.404935 -2.089412 -0.441568  
 H 5.221341 -1.643337 -1.415111  
 C 9.107000 -3.949720 0.433597  
 H 8.800431 -4.501999 -0.462872  
 H 9.930065 -4.475573 0.919447  
 H 9.436181 -2.943904 0.146190  
 O -8.282432 3.836491 0.660511  
 O 8.056548 -3.897510 1.396113

[(tmpa)Cu<sup>II</sup>(NO<sub>2</sub>)<sup>+</sup>

44

C 1.777276 -2.540808 0.888299  
 H 1.333709 -2.579644 1.875153  
 C 2.683553 -3.505088 0.459913  
 H 2.974788 -4.306497 1.129695  
 N 0.101618 0.239174 -1.516887  
 C -0.990768 -0.576704 -2.106883  
 C -2.240151 -0.466233 -1.260553  
 C -3.521996 -0.596580 -1.788363  
 H -3.658709 -0.746635 -2.853979  
 C -4.614058 -0.529037 -0.923620  
 H -5.623588 -0.628450 -1.309663  
 C -4.391421 -0.321865 0.437471  
 H -5.213140 -0.254650 1.141621  
 C -3.081101 -0.192714 0.885714  
 H -2.853789 -0.023382 1.931905  
 N -2.027778 -0.268710 0.056418  
 O -0.079340 0.029224 2.449548  
 Cu -0.018795 -0.024587 0.521278  
 N 1.381551 -1.523116 0.103030  
 C 3.187801 -3.421385 -0.837232  
 H 3.887881 -4.163190 -1.208443  
 C 2.772579 -2.373682 -1.657741  
 H 3.137550 -2.279493 -2.675074  
 C 1.876263 -1.436836 -1.149299  
 C 1.445124 -0.225284 -1.944066  
 C -0.081321 1.690722 -1.779428  
 C 0.640898 2.510106 -0.732658  
 C 1.151440 3.780455 -0.985954  
 H 1.085956 4.205757 -1.981856  
 C 1.808452 3.899208 1.326200  
 H 2.260975 4.414755 2.165918  
 C 1.285313 2.620686 1.495254  
 H 1.308522 2.110021 2.451821  
 N 0.715327 1.941776 0.486908  
 C 1.741797 4.486127 0.062749  
 H 2.147452 5.478495 -0.107176  
 H -1.183741 -0.287299 -3.146440  
 H -0.654436 -1.619518 -2.115078  
 H 0.247926 1.949760 -2.792225  
 H -1.153556 1.906524 -1.720725  
 H 1.469164 -0.423841 -3.021680  
 H 2.150725 0.589924 -1.750019  
 N -0.506911 -0.954104 3.190115  
 O -0.853581 -1.964382 2.587114