Supplementary Figures



Supplementary Figure 1. Initial screening of chiral Brønsted acids as catalyst candidates for silylative kinetic resolution of substrate 2a. The reactions were conducted with 0.2 mmol of *rac*-2a. Selectivity factors (*s*) were calculated according to Kagan's equation ($s = \ln[(1-\text{conv.})(1+\text{ee}_{(S)-3a}] / \ln[(1-\text{conv.})(1-\text{ee}_{(S)-3a}])^1$. The conversions were monitored by ¹H NMR analysis and calculated by the following equation: $c = \text{ee}_{rsm}/(\text{ee}_{prod}+\text{ee}_{rsm})$. We presumed that a Brønsted acid moiety could catalyze the silylation reaction by protonating the amine of HMDS, however, we observed only low conversion of the starting material with disappointing enantioselectivities (See above). The lack of a proper binding functional group in a secondary alcohol promoted us to explore more sterically demanding hydrogen bonding donor catalyst, such as catalyst 1.



Supplementary Figure 2. Desilylation reaction of TMS-protected secondary alcohol with KF and Amberlite[®] CG 50 as additives. Under the reaction condition (0.01 mol% catalyst loading) using KF and Amberlite[®] CG 50 as additives, the desilylation process of a TMS-protected secondary alcohol such as (R)-**3a** did not occur. As we reported², the rate of the desilylation of TMS-protected secondary alcohols is much slower than those of TMS-protected phenols.



Supplementary Figure 3. Silylative kinetic resolution of substrate 2a with ppb-level catalyst loading (100 ppb). The reactions were performed in the presence of 1 equiv of KF and 0.8 equiv of Amberlite[®] CG 50 (CG 50). *Calculated conversion, $c = ee_{(R)-2a}/(ee_{(R)-2a} + ee_{(S)-3a})$. e.r. = enantiomeric ratio. We applied our catalytic system to gram-scale enantioselective kinetic resolution with ppb-level catalyst loading (100 ppb). Even at ppb level catalyst loading, the catalyst was still highly active (TOF/h, 4464), and thus the racemic alcohol was resolved with a reasonable *s*-factor (*s* = 14).



Supplementary Figure 4. Silylative kinetic resolution of 1-cyclohexylethanol. *Calculated conversion, $c = ee_{(R)-2a}/(ee_{(R)-2a} + ee_{(S)-3a})$. e.r. = enantiomeric ratio. Comments: The absolute configuration of silylether product was established to be (*S*)-form by its conversion to the corresponding (*S*)-1-cyclohexylethyl-2-naphthoate. The chiral HPLC condition: Chiralcel OD-H, Hexane/IPA = 98/2, flow rate: 0.7 mL/min, 220 nm, $t_R = 7.5 min$, $t_S = 8.2 min$.



Supplementary Figure 5. Study of the effect of catalyst optical purity on product optical

purity. To verify reaction mechanism, we confirmed that there is no non-linear effects during the catalysis. This indicates that there is only single chiral catalyst involved in the enantio-determining step of the silylation reaction. Although many polyether-based catalysts can form complex dimer or oligomers in the presence of metal salts, our system showed essentially linear relationship between enantiopurity of catalyst **1e** and enantioselectivity of product **5a**.



Supplementary Figure 6. ¹H, ¹³C and ¹⁹F NMR spectra of catalyst 1f





Supplementary Figure 7. ¹H, ¹³C and ¹⁹F NMR spectra of catalyst 1g





Supplementary Figure 8. ¹H and ¹³C NMR spectra of catalyst 1h



Supplementary Figure 9. ¹H, ¹³C and ²⁹Si NMR spectra of catalyst 1k





Supplementary Figure 10. ¹H, ¹³C and ²⁹Si NMR spectra of catalyst 11





Supplementary Figure 11. ¹H and ¹³C NMR spectra of catalyst 1m



Supplementary Figure 12. ¹H, ¹³C and ¹⁹F NMR spectra of the remaining alcohols 2a–2r

S16













S19































































Supplementary Figure 13. ¹H, ¹³C and ¹⁹F NMR spectra of the TMS-ether products 3a–3r


S37















80 70 Π (ppm) ů.

-10











































80: 70 (1 (ppm) -10







Supplementary Figure 14. ¹H and ¹³C NMR spectra of the *meso*-diols 4a–4c







Supplementary Figure 15. ¹H and ¹³C NMR spectra of the TMS-ether products 5a–5c







Supplementary Figure 16. ¹H NMR and HPLC Spectra for Figure 3

We monitored the conversion by ¹H-NMR spectrum and calculated by the following equation $(c = ee_{rsm}/(ee_{prod}+ee_{rsm}))^1$ in Figure **3** – Figure **5a**.



¹H NMR spectrum of the silylation reaction mixture of 2a after 7% conversion: cat 1c



¹H NMR spectrum of the silylation reaction mixture of 1a after 11% conversion: cat 1d

HPLC spectrum of the *rac*-2a (Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm)





[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2a after 49% conversion: cat 1e

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

¹H NMR spectrum of the silylation reaction mixture of 2a after 44% conversion: cat 1f





[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2a

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2a after 8% conversion: cat 1g

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

¹H NMR spectrum of the silylation reaction mixture of 2a after 3% conversion: cat 1h





¹H NMR spectrum of the silylation reaction mixture of 2a after 6% conversion: cat 1k

HPLC spectrum of the TMS-ether product (S)-3a

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

Supplementary Figure 17. ¹H NMR and HPLC Spectra for Table 1

¹H NMR spectrum of the silylation reaction mixture of 2a after 4% conversion: entry1



¹H NMR spectrum of the silylation reaction mixture of 2a after 17% conversion: entry2


HPLC spectrum of the TMS-ether product (*S*)-**3a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2a

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2a after 15% conversion: entry3

HPLC spectrum of the TMS-ether product (*S*)-**3a**



HPLC spectrum of the remaining alcohol (*R*)-**2a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



¹H NMR spectrum of the silylation reaction mixture of 2a after 18% conversion: entry4



HPLC spectrum of the TMS-ether product (*S*)-**3a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (R)-2a





¹H NMR spectrum of the silvlation reaction mixture of 2a after 52% conversion: entry5

HPLC spectrum of the TMS-ether product (*S*)-**3a**



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

¹H NMR spectrum of the silylation reaction mixture of 2a after 51% conversion: entry6



HPLC spectrum of the TMS-ether product (*S*)-**3a**



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (R)-2a

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2a after 50% conversion: entry8

HPLC spectrum of the TMS-ether product (*S*)-**3a**



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

¹H NMR spectrum of the silylation reaction mixture of 2a after 52% conversion entry9



HPLC spectrum of the TMS-ether product (*S*)-**3a**



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (R)-2a





¹H NMR spectrum of the silylation reaction mixture of 2a after 33% conversion entry10

HPLC spectrum of the TMS-ether product (*S*)-**3a**



HPLC spectrum of the remaining alcohol (*R*)-2a



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





HPLC spectrum of *rac*-2a (Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2a





¹H NMR spectrum of the silylation reaction mixture of 2b after 54.5% conversion

HPLC spectrum of *rac*-2b (Chiralcel OD-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3b** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2b

[Chiralcel OD-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2c after 55.8% conversion

HPLC spectrum of *rac*-2c (Chiralcel OJ-H, Hexane/IPA = 95/5, 1.0 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-3c



[Chiralcel OJ-H, Hexane/IPA = 95/5, 1.0 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2c

[Chiralcel OJ-H, Hexane/IPA = 95/5, 1.0 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2d after 52.6% conversion

HPLC spectrum of *rac*-2d (Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3d**



[Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2d

[Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2e after 53.5% conversion

HPLC spectrum of *rac*-2e (Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3**e



[Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2e





¹H NMR spectrum of the silylation reaction mixture of 2f after 51.4% conversion

HPLC spectrum of *rac*-2f (Chiralcel OD-H, Hexane/IPA = 95/5, 0.8 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-3f



[Chiralcel OD-H, Hexane/IPA = 95/5, 0.8 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2f** [Chiralcel OD-H, Hexane/IPA = 95/5, 0.8 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2g after 53.7% conversion

HPLC spectrum of *rac*-2g (Chiralcel OJ-H, Hexane/IPA = 99.8/0.2, 1.0 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3g** [Chiralcel OJ-H, Hexane/IPA = 99.8/0.2, 1.0 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-2g

[Chiralcel OJ-H, Hexane/IPA = 99.8/0.2, 1.0 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2h after 54.8% conversion

HPLC spectrum of *rac*-2h (Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-3h



[Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2h





¹H NMR spectrum of the silylation reaction mixture of 2i after 55.7% conversion

HPLC spectrum of *rac*-2i (Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-3i



[Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2i** [Chiralcel OB-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2j after 57.5% conversion

HPLC spectrum of *rac*-2j (Chiralcel OD-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3**j



[Chiralcel OD-H, Hexane/IPA = 98/2, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2j





¹H NMR spectrum of the silylation reaction mixture of 2k after 50.9% conversion

HPLC spectrum of *rac*-**2k** (Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3**k



[Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2k** [Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2l after 52.9% conversion

HPLC spectrum of *rac*-2l (Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-31



[Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (R)-21




¹H NMR spectrum of the silylation reaction mixture of 2m after 49.7% conversion

HPLC spectrum of *rac*-2m (Chiralcel OJ-H, Hexane/IPA = 99/1, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3m**



[Chiralcel OJ-H, Hexane/IPA = 99/1, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (R)-2m

[Chiralcel OJ-H, Hexane/IPA = 99/1, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2n after 53.0% conversion

HPLC spectrum of *rac*-2n (Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3n**



[Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2n

[Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 20 after 52.6% conversion

HPLC spectrum of *rac*-20 (Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**30** [Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-20

[Chiralcel OB-H, Hexane/IPA = 90/10, 0.5 ml/min, 220 nm]



¹H NMR spectrum of the silvlation reaction mixture of 2p after 58.5% conversion (cat 1e)



HPLC spectrum of *rac*-2p (Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm)



S115

HPLC spectrum of the TMS-ether product (*S*)-**3**p [Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)-**2p** [Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm]



¹H NMR spectrum of the silylation reaction mixture of 2p after 48.5% conversion (cat 1d)



HPLC spectrum of *rac*-2p (Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm)



S117

HPLC spectrum of the TMS-ether product (*S*)-**3p**



[Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2p** [Chiralcel OB-H, Hexane/IPA = 99.8/0.2, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silylation reaction mixture of 2q after 52.8% conversion

HPLC spectrum of *rac*-2q (Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (S)-**3** \mathbf{q}



[Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-2q

[Chiralcel OB-H, Hexane/IPA = 95/5, 0.5 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 2r after 54.9% conversion

HPLC spectrum of *rac*-2r (Chiralcel OD-H, Hexane/IPA = 90/10, 1.0 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3r** [Chiralcel OD-H, Hexane/IPA = 90/10, 1.0 ml/min, 220 nm]



HPLC spectrum of the remaining alcohol (*R*)- $2\mathbf{r}$ [Chiralcel OD-H, Hexane/IPA = 90/10, 1.0 ml/min, 220 nm]







HPLC spectrum of *rac*-2a (Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product (*S*)-**3a**



[Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]

HPLC spectrum of the remaining alcohol (*R*)-**2a** [Chiralcel OD-H, Hexane/IPA = 90/10, 0.7 ml/min, 220 nm]





Supplementary Figure 20. ¹H NMR and HPLC Spectra for Figure 5b ¹H NMR spectrum of the silylation reaction mixture of 4a after 95% conversion

HPLC spectrum of *rac*-**5a** (Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm)



HPLC spectrum of the TMS-ether product **5a**



[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]

¹H NMR spectrum of the silylation reaction mixture of 4b after 92% conversion



HPLC spectrum of *rac*-**5**b



[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]

HPLC spectrum of the TMS-ether product **5b**

[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]





¹H NMR spectrum of the silvlation reaction mixture of 4c after 91% conversion at 20°C

HPLC spectrum of *rac*-5c

[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]



HPLC spectrum of the TMS-ether product 5c



[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]

1H NMR spectrum of the silylation reaction mixture of 4c after 93% conversion at 0°C



HPLC spectrum of the TMS-ether product **5**c



[Chiralcel OD-H, Hexane/IPA = 98/2, 1.0 ml/min, 220 nm]

Supplementary Table

ĺ	ОН	cat 1e (1 mol%) silylating reagent (0.7 equiv	<i>i</i> .)		
		CH ₂ Cl ₂ (0.2 M), 20 ^o C, 1h			
	<i>rac-2a (1.0 mmol)</i>		(<i>R</i>)-2a	(S)- 3a	
	Me Me Me Si Me Me N Me	Me Me Ph I Ph Ma Me Si N Me Pl H	Ph Ph e I Me Si Si h N Ph H	Me O Me Me Si Me Me Si Me Me Me	
	S <i>i</i> -1	Si-2	Si-3	Si-4	
F ₃ C	O Me Si Me N ²	Me / Me / Me	Me N-Si-Me Me	O Me Me │	Ме
 	Si-5	Si-6 5	Si-7	Si-8 Si-9	 !
Entry	Si-reagent	Conv. (%) ^[a]	e.r. of (<i>R</i>) -2a	e.r. of (S)-3a	s ^[b]
1	<i>Si</i> -1	48.5	90:10	85:15	41.1
2*	S <i>i</i> -1	28.2 (after 24 h)	68.5:31.5	97:3	46.4
3	Si-2	26.3	65:35	92:8	15.4
4	Si-3	n.r.	-	-	-
5	Si-4	1.5	-	racemic	n.d.
6	Si-5	28.6	racemic	racemic	n.d.
7	Si-6	19.8	racemic	racemic	n.d.
8	Si-7	15.4	52:48	61:39	1.6
9	Si-8	4.0	racemic	racemic	n.d.
10	Si-9	n.r.	-	-	-

Supplementary Table 1. Silylating reagent screening

^[a] Determined by ¹H-NMR analysis of the unpurified reaction mixture. ^[b] $s = \ln[1-\text{conv.}(1+ee_{(S)-3a}] / \ln[1-\text{conv.}(1-ee_{(S)-3a}]) + 0.25$ equiv. of **Si-1** was used. n.d., not determined. e.r., enantiomeric ratio.

Comments: We observed reasonable conversion and enantioselectivity only with HMDS (*Si*-**1**, Supplementary Table 1). Most of silylating reagents are not capable of producing silylprotected alcohols in the presence of catalyst **1e**.

Supplementary Table 2. Solvent screening

	OH	cat 1e (1 mol%) HMDS (0.7 equiv.)		OH	отмs
Ĺ		solvent (0.2 M), -30 ^o C,	1 h		
(1	<i>rac-2a .</i> 0 mmol)		(<i>R</i>)- 2 a	(S)- 3a	
		[-]			[b]
Entry	Solvent	Conv. (%) ^[a]	e.r. of (<i>R</i>)- 2a	e.r. of (S)- 3a	S ^[D]
1	CH_2CI_2	51.6	98:2	95:5	89.1
2	Toluene	35.3	74:26	95:5	30.9
3	THF	5.1	52:48	97:3	29.0
4	Et ₂ O	12.8	56:44	91:9	11.4
5	DCE	53.0	99:1	93:7	66.1

^[a] Determined by ¹H-NMR analysis of the unpurified reaction mixture. ^[b] $s = \ln[1-\text{conv.}(1+e_{(S)-3a}] / \ln[1-\text{conv.}(1-e_{(S)-3a}]]$. e.r., enantiomeric ratio.

Comments: We observed lower enantioselectivity in the case of ether-based solvents, which can be as ascribed to the competitive interactions of solvents and the polyether-based catalyst. In non-polar solvents, superior enantioselectivities were obtained under identical reaction conditions.

Supplementary Table 3. Concentration effect

(OH	cat 1e (* HMDS (0 CH ₂ Cl ₂ ,	l mol%) .7 equiv.) -30 °C	OH T	OTMS	
	<i>rac-2a (1.</i> 0 mmol)			(<i>R</i>)- 2 a	(S)- 3a	
Entry	Conc. (M)	Time (min)	Conv. (%) ^[a]	e.r. of (<i>R</i>)- 2a	e.r. of (S)- 3a	S ^[b]
1	0.1	90	52.4	98.5:1.5	94:6	66.1
2	0.2	60	51.6	98:2	95:5	89.1
3	0.4	30	51.1	97.5:2.5	95.5:4.5	79.2

^[a] Determined by ¹H-NMR analysis of the unpurified reaction mixture. ^[b] $s = \ln[1-\text{conv.}(1+e_{(S)-3a}] / \ln[1-\text{conv.}(1-e_{(S)-3a}]]$. e.r., enantiomeric ratio.

uncomplexed				com	plexed	
OH HO Catalyst 1e) +	OH * 2a / (<i>R</i>)-2a				
(0.03 mmol, 37 mg)	(0.03 mmol, 5	mg, 1 equiv)				
(0.03 mmol, 37 mg)	(0.03 mmol, 5	mg, 1 equiv) T ₁ [S]		T ₁ (unc	ompl.)/ <i>T</i> 1(compl.)
(0.03 mmol, 37 mg) Sample (ratio of cat 1e : 2a = 1 : 1)	(0.03 mmol, 5	mg, 1 equiv) T ₁ [S] C ₂	C ₃	T₁(unc C₁	ompl.)/ <i>T</i> ₁ (i C ₂	compl.) C ₃
(0.03 mmol, 37 mg) Sample (ratio of cat 1e : 2a = 1 : 1) cat 1e	(0.03 mmol, 5 C ₁ 0.364	mg, 1 equiv) T_1 [s] C_2 0.397	C ₃ 0.604	<i>T</i> ₁ (unc C ₁	ompl.)/ <i>T</i> ₁ ((C ₂ -	compl.) C ₃
(0.03 mmol, 37 mg) Sample (ratio of cat 1e : 2a = 1 : 1) cat 1e cat 1e & (<i>R</i>)- 2a	(0.03 mmol, 5 C ₁ 0.364 0.396	mg, 1 equiv) T_1 [s] C_2 0.397 0.492	C ₃ 0.604 0.505	<i>T</i> ₁ (unc C ₁ - 0.91	ompl.)/ <i>T</i> ₁ (0 <u>C</u> 2 - 0.80	compl.) C ₃ - 1.19
(0.03 mmol, 37 mg) Sample (ratio of cat 1e : 2a = 1 : 1) cat 1e cat 1e & (<i>R</i>)-2a cat 1e & <i>rac</i> -2a	(0.03 mmo), 5 C ₁ 0.364 0.396 0.403	mg, 1 equiv) T_1 [s] C_2 0.397 0.492 0.424	C ₃ 0.604 0.505 0.488	<i>T</i> ₁ (unc C ₁ - 0.91 0.90	ompl.)/ <i>T</i> ₁ (6 C ₂ - 0.80 0.94	compl.) C ₃ - 1.19 1.24

Supplementary Table 4. ¹³C Spin-lattice relaxation measurements (catalyst 1e)

All NMR experiments were carried out in CDCI₃ at 22 °C, [1e] = 0.04 M.

Comments: To verify the catalyst-secondary alcohol interactions, we conducted ¹³C Spinlattice relaxation measurements. A significant decrease in the T_1 value of C₃ was observed upon complexation of catalyst **1e** with alcohols **2a**. This significant decrease in the T_1 value of C₃ indicates that the complexation of the alcohol with catalyst **1e** significantly reduces the mobility of the ether units. Furthermore, the fast reacting (*S*)-configured substrate ((*S*)-**2a**) gave more decreased T_1 value of C₃ than that of the slow reacting (*R*)-**2a**, directly indicating a selective association of the catalyst with the fast reacting (*S*)-**2a**



Supplementary Table 5. ¹³C Spin-lattice relaxation measurements (catalyst 1m)

All NMR experiments were carried out in CDCl₃ at 22 $^{\circ}$ C, [1e] = 0.075 M.

Comments: No change of T_1 value was observed when the alcohol **2a** was mixed with **1m** in which ether chain was replaced with the alkyl chain, verifying the crucial role of the polyether backbone.

Supplementary Table 6. ²⁹ Si NMR data of HMDS in the pro-	esence of diverse phenols
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	²⁹ Si chemical shift, ppm* (Δυ in Hz)			
Sample	HMDS : phenol = 1 : 1 (molar ratio)	HMDS : phenol = 1 : 5		
HMDS	2.457	2.457		
HMDS & 2,6-dimethylphenol	2.475 (1.8 Hz)	2.511 (5.4 Hz)		
HMDS & 2,6-dibromophenol	2.479 (2.2 Hz)	2.539 (8.2 Hz)		
HMDS & 2,6-difluorophenol	2.545 (8.8 Hz)	n.d.*		

*Using tetramethysilane as an internal standard. [HMDS] = 0.4 M. n.d. = not determined; Under this condition, 2,6-difluorophenol was silylated by HMDS very fast.

Comments: ²⁹Si NMR experiments were used to establish the interaction between phenolic proton of the catalyst and HMDS. As shown in Supplementary Table 6, ²⁹Si NMR data of HMDS in the presence of diverse phenols exhibit ²⁹Si chemical shifts that are shifted downfield relative to the uncoordinated HMDS, depending on the acidity of phenols. This result clearly indicates that the acidic phenolic proton interacts with HMDS, enhancing the electrophilicity of silicon atom.

\cap HMDS (0.05 mmol) С OR OH HO R'C CH₂Cl₂ (1.0 ml) -30 °C, 10 min catalyst 1 (0.002 mmol) non-TMS-substituted catalyst (R,R'=H) catalyst 1c (X=CI), catalyst 1d (X=Br); mono-TMS-substituted catlyst (R=TMS, R'=H) di-TMS-substituted catalyst (R=TMS, R'=TMS) catalyst 1e (X=I), catalyst 1f (X=CF₃) Ratio (%)* di-TMS Entry Catalyst mono-TMS non-TMS 1 1c (X=CI) 9 64 27 1d (X=Br) 2 31 32 37 3 1e (X=I) 27 20 53 4 1f (X=CF₃) 18 39 43

Supplementary Table 7. Steric demand of 3,3'-substituents of the catalysts

*Determined by ¹H-NMR analysis of the unpurified reaction mixture in CDCl₃ at 22 °C.

Comments: After observing the TMS-protected catalyst is inactive towards the silvlation reaction, we conducted silvlation reaction of various catalyst **1c–1f** under the same reaction conditions. As summarized in Supplementary Table 7, steric hindrance of substituents on 3,3'-positions can inhibit the silvlation of phenolic alcohol.

Supplementary Table 8. The relative catalytic activity of **1a–1f** under the optimized reaction conditions

$\begin{array}{c} & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$							
<i>rac-2a</i> (1 mmol s	a cale)		(<i>R</i>)- 2a	(\$	S) -3a		
Entry	Catalyst	Conv. (%) ^[a]	e.r. of (<i>R</i>)- 2a	e.r. of (S)- 3a	s ^[b]		
1	1a (X=H)	n.r.	-	-	-		
2	1b (X=Ph)	n.r.	-	-	-		
3	1c (X=CI)	16.8	59:41	94.5:5.5	20.8		
4	1d (X=Br)	36.7	77:23	96.5:3.5	48.4		
5	1e (X=I)	50.3	98:2	97:3	132		
6	1f (X=CF ₃)	50.8	95.5:4.5	94:6	50.1		
7	1g (X=C ₂ F ₅)	34.1	73:27	95:5	30.8		

^[a] Determined by ¹H-NMR analysis of the unpurified reaction mixture. ^[b] $s = \ln[1-\text{conv.}(1+ee_{(S)-3a}] / \ln[1-\text{conv.}(1-ee_{(S)-3a}]. e.r., enantiomeric ratio. n.r., no reaction.$

Comments: As shown from the results in Supplementary Table 8, the relative catalytic activity of **1a–1f** under the optimized reaction conditions is almost similar with that shown in Figure 3.

Supplementary Methods

General information

All starting materials and solvents were obtained from commercial suppliers and used without purification. The organocatalysts 1a-1d, 1i and 1j were obtained starting from BINOL according to the literature procedure^{3,4}. The catalyst **1e** was purchased from C-Tri Co. Ltd. (www.c-tri.co.kr). The *meso*-diol **4b** and **4c** were prepared from the corresponding aldehyde according to the literature procedure⁵. Thin-layer chromatography (TLC) was performed using silica gel plates (Merck, Kieselgel 60 F254 0.25 mm). Chromatographic purification of the products was performed by using silica gel 60 (230–400 mesh, Merck). ¹H NMR (300 and 500 MHz), ¹³C NMR (75.4 and 125.7 MHz) and ²⁹Si NMR (99.2 MHz) spectra were recorded using a Varian 300 or Bruker 500 spectrometer. Chemical shifts are reported in ppm downfield from tetramethylsilane (TMS) as the internal standard or with the solvent reference as the internal standard (CDCl₃: δ 7.26 for ¹H NMR and δ 77.0 for ¹³C NMR). ¹⁹F NMR (470.4 MHz) spectra were recorded using a Bruker 500 spectrometer with benzotrifluoride (C₆H₅CF₃) as the external standard. HPLC analyses for the determination of the enantiomeric excess (ee) of the products were performed using a Varian Pro Star Series instrument equipped with an isostatic pump using a chiral column (CHIRALPACK AD-H, Chiralcel OD-H, OB-H or OJ-H; 250×4.6 mm). IR spectra were recorded using a Bruker Vertex 70 spectrometer with the MIRacle Micro ATR accessory. High-resolution mass spectra (HRMS) spectra were recorded using the FAB method using a Jeol JMS-700 MStation. ESI-MS spectra were recorded using a Finnigan Ultra Mass TSQ 7000. Melting points (Mps) were determined using a Buchi B-540 melting point apparatus and are uncorrected. Optical rotations were measured using a PerkinElmer Polarimeter 343 plus. Abbreviations: e.r. (enantiomeric ratio), s (selectivity factor), n.d. (not determined), n.r. (no reaction), DMF (N,N-dimethylformamide), THF (tetrahydrofuran), DCE (1,2-dichloroethane), HMDS (1,1,1,3,3,3-hexamethyldisilazane), TMS (trimethylsilyl), TBAF (tetra-*n*-butylammonium fluoride).

Preparation and characterization of chiral catalysts (1f-1h, 1k-1m)



Chiral catalyst 1f

To a solution of the mono-MOM-protected $3,3'-CF_3-(R)$ -BINOL (A)⁶ (466.4 mg, 1 mmol) in DMF (10 mL), finely powdered K₂CO₃ (344.7 mg, 2.5 mmol) was added, and the reaction mixture stirred for 2 h at room temperature. To this mixture, a solution of bis-tosylated triethylene glycol (252.1 mg, 0.55 mmol) in DMF (5 mL) was then added via a syringe pump for 24 h at 60 °C. The resulting reaction mixture was allowed to cool down to room temperature, followed by the addition of concentrated HCl (35%) (5 mL). The reaction mixture was stirred for an additional 2 h and then diluted with H₂O and EtOAc. The organic layer was separated, washed with brine, dried over anhydrous NaSO₄, filtered, and concentrated *in vacuo* to afford a dark brown solid. The crude solid was purified by column chromatography (acetone/hexanes = 1/10) to afford the chiral catalyst **1f** as a white solid.

White solid; mp: 166–168 °C; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.21$; $[\alpha]_D^{20} = -16.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 8.33 (s, 2H), 8.28 (s, 2H), 7.97 (d, J = 8.0 Hz, 2H), 7.92 (d, J = 8.0 Hz, 2H), 7.52–7.48 (m, 2H), 7.41–7.33 (m, 6H), 7.13 (d, J = 8.5 Hz, 2H), 7.04 (d, J = 8.5 Hz, 2H), 5.79 (bs, 2H), 3.59–3.51 (m, 4H), 3.11–3.03 (m, 4H), 2.99–2.90 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 153.12, 148.33, 135.02, 134.80, 129.83 (q, ³J_{C-F} = 5.5 Hz)), 129.40, 129.35, 129.34, 129.28, 129.22, 129.12 (q, ³J_{C-F} = 5.5 Hz), 127.39, 126.47, 124.94, 124.89, 124.65, 123.66 (q, ²J_{C-F} = 30.2 Hz)), 123.49 (q, ¹J_{C-F} = 271.0 Hz), 123.42 (q, ¹J_{C-F} = 271.0 Hz), 122.36, 119.16 (q, ²J_{C-F} = 30.2 Hz), 116.92, 72.99, 69.68, 69.51; ¹⁹F NMR

 $(470 \text{ MHz}, \text{CDCl}_3): \delta -60.94 \text{ (s, 6F)}, -62.41 \text{ (s, 6F)}; \text{IR}: 2960, 2895, 1627, 1603, 1503, 1454, 1325, 1294, 1206, 1130, 1045, 1019, 907, 752, 725 \text{ cm}^{-1}; \text{HRMS} (m/z, \text{FAB}): [M + H]^+ \text{ calcd.}$ for C₅₀H₃₅F₁₂O₆, 959.2242; found, 959.2240.



To a solution of the di-MOM-protected compound of $1e(B)^3$ (1.2785g, 1 mmol) in DMF (6 mL), KF (464.8 mg, 8 mmol) and CuI (1.1388 g, 6 mmol) were added and the reaction mixture stirred for 10 min at room temperature. To this reaction mixture, TMSC₂F₅ (1.5376 g, 8 mmol) was added and stirred for 48 h at 100 °C. The reaction mixture was allowed to cool down to room temperature and concentrated HCl (35%) was added (5 mL). The resulting reaction mixture was stirred for an additional 2 h, and then diluted with H₂O and EtOAc. The organic layer was separated, washed with brine, dried over anhydrous NaSO₄, filtered, and concentrated *in vacuo* to afford a dark brown solid. The solid thus obtained was purified by column chromatography (acetone/hexanes = 1/10) to afford the chiral catalyst **1g** as the yellow solid.

Yellow solid; mp: 90–92 °C; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.26$; $[\alpha]_D^{20} = -6.4$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 8.30 (s, 2H), 8.25 (s, 2H), 7.92 (d, J = 8.0 Hz, 2H), 7.52–7.48 (m, 2H), 7.41–7.33 (m, 6H), 7.25 (d, J = 8.5 Hz, 2H), 7.10 (d, J = 8.0 Hz, 2H), 5.78 (bs, 2H), 3.55–3.50 (m, 4H), 3.02–2.97 (m, 4H), 2.90–2.82 (m, 4H); ¹³C NMR (125 MHz, CDCl₃): δ 153.57, 148.81, 135.10, 134.90, 131.87 (t, ⁴J_{C-F} = 8.7 Hz), 131.24 (t, ⁴J_{C-F} =

8.7 Hz), 129.65, 129.51, 129.38, 129.35, 129.26, 127.64, 126.51, 124.94, 124.81, 124.59, 123.02, 121.80 (t, ${}^{3}J_{C-F} = 21.2 \text{ Hz}$), 119.50 (qt, ${}^{1}J_{C-F} = 306.2 \text{ Hz}$ and ${}^{1}J_{C-F} = 33.7 \text{ Hz}$), 119.28 (qt, ${}^{1}J_{C-F} = 306.2 \text{ Hz}$ and ${}^{2}J_{C-F} = 33.7 \text{ Hz}$), 117.44, 117.11 (t, ${}^{3}J_{C-F} = 21.2 \text{ Hz}$), 113.78 (tq, ${}^{1}J_{C-F} = 254.3 \text{ Hz}$ and ${}^{2}J_{C-F} = 40.0 \text{ Hz}$), 113.72 (tq, ${}^{1}J_{C-F} = 254.3 \text{ Hz}$ and ${}^{2}J_{C-F} = 40.0 \text{ Hz}$), 73.23, 69.66, 69.26; ${}^{19}\text{F}$ NMR (470 MHz, CDCl₃): δ -83.02 (s, 6F), -83.43 (s, 6F), -110.52 (s, 2F), -110.593 (s, 2F), -111.375 (s, 2F), -111.43 (s, 2F); IR: 3072, 2889, 1626, 1599, 1501, 1454, 1357, 1278, 1194, 1144, 1128, 1092, 1071, 1029, 969, 905, 829, 751, 718, 640 \text{ cm}^{-1}; HRMS (m/z, FAB): [M + Na]⁺ calcd. for C₅₄H₃₄F₂₀O₆Na₁, 1181.1934; found, 1181.1934.





To a solution of the mono-MOM-protected 3,3'-I-(*R*)-H₈-BINOL (**C**) (590.2 mg, 1 mmol) in DMF (10 mL), finely powdered K₂CO₃ (344.1 mg, 2.5 mmol) was added and the reaction mixture stirred for 2 h at room temperature. To this reaction mixture, a solution of bistosylated triethylene glycol (251.1 mg, 0.55 mmol) in DMF (5 mL) was then added via a syringe pump for 24 h at 60 °C. The reaction mixture was allowed to cool down to room temperature and concentrated HCl (35%) was added (5 mL). The resulting reaction mixture was stirred for an additional 2 h, and then diluted with H₂O and EtOAc. The organic layer was separated, washed with brine, dried over anhydrous NaSO₄, filtered, and concentrated *in vacuo* to afford a dark brown solid. The solid thus obtained was purified by column chromatography (acetone/hexanes = 1/10) to afford the chiral catalyst **1h** as a white solid. White solid; mp: 220–222 °C; TLC (acetone : n-hexane, 1:5 v/v): R_f = 0.22; $[\alpha]_D^{20} = -30.0$ (*c*

= 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.57 (s, 2H), 7.45 (s, 2H), 5.16 (s, 2H), 3.88–3.84 (m, 2H), 3.61–3.57 (m, 2H), 3.44 (t, J = 5.0 Hz, 4H), 3.28 (s, 4H), 2.75–2.69 (m, 8H), 2.37–2.28 (m, 4H), 2.06–1.95 (m, 4H), 1.72–1.56 (m, 16H); ¹³C NMR (125 MHz, CDCl₃): δ 154.19, 149.17, 139.69, 138.29, 138.00, 137.59, 136.03, 132.17, 129.62, 123.30, 89.09, 81.15, 72.10, 70.17, 70.08, 29.18, 29.02, 27.18, 26.75, 22.85, 22.78, 22.75, 22.59; IR: 3464, 2934, 2856, 1736, 1654, 1573, 1441, 1354, 1241, 1137, 1070, 1012, 955, 877, 823, 723, 673 cm⁻¹; HRMS (m/z, FAB): [M + H]⁺ calcd. for C₄₆H₅₁I₄O₆, 1206.9865; found, 1206.9866.

Chiral catalyst 1k



To a solution of catalyst **1e** (119.0 mg, 0.1 mmol) in CH_2Cl_2 (2.5 mL), triethylamine (12.2 mg, 0.12 mmol) and chlorotrimethylsilane (13.0 mg, 0.12 mmol) were added, and the reaction mixture was stirred for 4 h at room temperature for completion, and then the reaction mixture was purified by short column chromatography (acetone/hexane = 1/5) without work-up, affording chiral catalyst **1k** as a white solid.

White solid; mp: 140–142 °C; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.35$; $[\alpha]_D^{20} = -44.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 8.53 (s, 1H), 8.49 (s, 2H), 8.43 (s, 1H), 7.78 (d, J = 8.5 Hz, 2H), 7.72–7.70 (m, 2H), 7.41–7.39 (m, 2H), 7.31–7.17 (m, 6H), 7.11–7.08 (m, 2H), 7.01–6.96 (m, 2H), 5.76 (s, 1H), 3.83–3.55 (m, 3H), 3.41–3.23 (m, 3H), 3.17–2.93 (m, 6H), -0.27 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 154.21, 154.11, 151.35, 149.47, 140.37, 139.71, 139.59, 139.53, 133.95, 133.90, 133.52, 133.40, 132.48, 132.31, 130.63, 130.32, 127.48, 127.43, 127.05, 127.03, 126.92, 126.85, 126.84, 126.10, 126.05, 125.97, 125.62, 125.60, 125.37, 124.97, 124.63, 124.34, 123.14, 121.80, 115.44, 93.74, 93.06, 92.81, 87.20, 87.13, 72.59, 72.34, 69.97, 69.93, 69.86, 69.47, 0.91; ²⁹Si NMR (100 MHz, CDCl₃): δ 20.69; IR: 3059, 2951, 2892, 2358, 1615, 1571, 1558, 1491, 1446, 1412, 1391, 1348, 1249, 1224, 1136, 1046, 1008, 946, 882, 848, 775, 747, 696 cm⁻¹; HRMS (m/z, FAB): [M]⁺ calcd. for

C₄₉H₄₁O₆I₄Si₁, 1260.8845; found, 1260.8846.

Chiral catalyst 11



To a solution of catalyst **1e** (119.0 mg, 0.1 mmol) in CH_2Cl_2 (2.5 mL), triethylamine (24.4 mg, 0.24 mmol) and chlorotrimethylsilane (26.0 mg, 0.24 mmol) were added, and the reaction mixture stirred for 4 h at room temperature for completion. The reaction mixture then was purified by short column chromatography (acetone/hexane = 1/5) without work-up, affording chiral catalyst **1l** as a white solid.

White solid; mp: 128–130 °C; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.54$; $[\alpha]_D^{20} = -128$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 8.49 (s, 2H), 8.48 (s, 2H), 7.78 (d, J = 8.0 Hz, 2H), 7.70 (d, J = 8.0 Hz, 2H), 7.43–7.38 (m, 2H), 7.31–7.24 (m, 4H), 7.21–7.16 (m, 6H), 7.10 (d, J = 8.5 Hz, 2H), 6.97 (d, J = 8.5 Hz, 2H), 3.85–3.81 (m, 2H), 3.41–3.37 (m, 2H), 3.24–3.15 (m, 4H), 3.07–3.00 (m, 4H), -0.27 (s, 18H); ¹³C NMR (125 MHz, CDCl₃): δ 154.16, 151.38, 139.72, 139.57, 133.96, 133.90, 132.30, 130.64, 127.03, 126.92, 126.85, 126.84, 126.12, 126.05, 125.62, 125.58, 124.63, 121.78, 93.74, 93.08, 72.31, 69.92, 69.54, 0.91; ²⁹Si NMR (100 MHz, CDCl₃): δ 20.70; IR: 3054, 2951, 2898, 1559, 1489, 1445, 1411, 1394, 1346, 1249, 1150, 1135, 1046, 1008, 946, 846, 800, 747, 693, 656 cm⁻¹; HRMS (m/z, FAB): [M + H]⁺ calcd. for C₅₂H₅₁O₆I₄Si₂, 1334.9398; found, 1334.9398.
Chiral catalyst 1m



To a solution of the mono-MOM-protected 3,3'-I-(*R*)-BINOL (**D**) (582.1 mg, 1 mmol) in DMF (10 mL), finely powdered K₂CO₃ (344.7 mg, 2.5 mmol) was added, and the reaction mixture stirred for 2 h at room temperature. To this mixture, a solution of bis-tosylated octanediol (250.0 mg, 0.55 mmol) in DMF (5 mL) was then added via a syringe pump for 24 h at 60 °C. The reaction mixture was allowed to cool down to room temperature, followed by the addition of concentrated HCl (35%, 5 mL). The reaction mixture was stirred for an additional 2 h and then diluted with H₂O and EtOAc. The organic layer was separated, washed with brine, dried over anhydrous NaSO₄ filtered, and concentrated in vacuo afforded a dark brown solid. The solid thus obtained was purified by column chromatography (acetone/hexane = 1/10) to afford chiral catalyst **1m** as a white solid.

White solid; mp: 204–206 °C; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.26$; $[\alpha]_D^{20} = -36.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 8.55 (s, 2H), 8.42 (s, 2H), 7.81 (d, J = 8.0 Hz, 2H), 7.70 (d, J = 8.0 Hz, 2H), 7.44–7.41 (m, 2H), 7.31–7.27 (m, 4H), 7.25–7.21 (m, 2H), 7.14 (d, J = 8.5 Hz, 2H), 7.02 (d, J = 8.5 Hz, 2H), 5.50 (s, 2H), 3.68–3.64 (m, 2H), 3.43–3.38 (m, 2H), 1.31–1.18 (m, 4H), 0.73–0.60 (m, 6H), 0.57–0.51 (m, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 154.38, 149.39, 140.45, 139.41, 133.76, 133.50, 132.57, 130.32, 127.45, 127.42, 127.13, 127.06, 126.02, 125.51, 125.09, 124.31, 123.61, 115.57, 92.98, 87.01, 74.09, 29.66, 28.80, 25.23; IR: 3507, 3056, 2929, 2853, 1614, 1572, 1494, 1446, 1352, 1226, 1197, 1144, 1080, 1009, 952, 886, 796, 747, 679, 636 cm⁻¹; HRMS (m/z, FAB): [M + Na]⁺ calcd. for C₄₈H₃₈I₄O₄Na₁, 1208.8847; found, 1208.8846.

Analytical data of the remaining alcohols ((*R*)-2a–2r)



(R)-**2b**

Colorless liquid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.39$; ¹H NMR (300 MHz, CDCl₃): δ 7.38–7.22 (m, 5H), 4.86 (qd, J = 6.3 Hz and J = 3.3 Hz, 1H), 2.07 (d, J = 3.3 Hz, 1H), 7.71 (d, J = 6.3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 145.78, 128.43, 127.39, 125.33, 70.32, 25.09.



MeO

White solid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.25$; ¹H NMR (500 MHz, CDCl₃): δ 7.85–7.79 (m, 4H), 7.51–7.44 (m, 3H), 5.06 (qd, J = 6.5 Hz and J = 4.0 Hz, 1H), 1.94 (d, J = 3.5 Hz, 1H), 1.58 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.19, 133.34, 132.94, 128.35, 127.95,

127.70, 126.18, 125.83, 123.82, 70.57, 25.17.



 OH
 Colorless liquid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.22$; ¹H NMR

 (500 MHz, CDCl₃): δ 7.32–7.28 (m, 2H), 6.90–6.85 (m, 2H), 4.85 (q, J = 6.0 Hz, 1H), 3.80 (s, 3H), 1.82 (s, 1H), 1.47 (d, J = 6.5 Hz, 3H); ¹³C NMR

 (R)-2e
 (125 MHz, CDCl₃): δ 158.94, 137.96, 126.63, 113.80, 69.95, 55.26, 24.99.

White solid⁷; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.13$; ¹H NMR (500 MHz, CDCl₃): δ 7.61–7.55 (m, 4H), 7.47–7.41 (m, 4H), 7.37–7.32 (m, 1H), 4.95 (qd, J = 6.5 Hz and J = 3.5 Hz, 1H), 1.85 (d, J = 4.0 Hz, 1H), 1.54 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 144.81, 140.85, 140.47, 128.77, 127.27, 127.09, 125.85, 70.19, 25.16.

Colorless liquid⁸; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.23$; ¹H NMR (500 MHz, CDCl₃): δ 7.63–7.58 (m, 2H), 7.52–7.46 (m, 2H), 4.96 (qd, J = 6.5 Hz and J = 4.0 Hz, 1H), 1.98 (d, J = 3.5 Hz, 1H), 1.50 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 149.68, 129.63 (q, ²J(C-C-F) = 32.3 Hz)), 125.65, 125.46 (q, ³J(C-C-C-F) = 3.7 Hz)), 124.16 (q, ¹J(C-F) = 271.2 Hz)), 69.84, 25.40. ¹⁹F NMR (470 MHz, CDCl₃): δ –62.45 (s, 3F).

Colorless liquid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.31$; ¹H NMR (300 MHz, CDCl₃): δ 7.37–7.30 (m, 2H), 7.06–6.99 (m, 2H), 4.87 (qd, J = 6.6 Hz and J = 3.6 Hz, 1H), 1.87 (d, J = 3.3 Hz, 1H), 1.47 (d, J = 6.6 Hz, 3H); (R)-2h ¹³C NMR (75 MHz, CDCl₃): 162.11 (d, ¹J(C-F) = 243.8 Hz), 141.51 (d, ⁴J(C-C-C-F) = 3.2 Hz), 127.01 (d, ³J(C-C-C-F) = 7.9 Hz), 115.23 (d, ²J(C-C-F) = 21.2 Hz), 69.77, 25.28.



(R)-2i

Colorless liquid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.31$; ¹H NMR (300 MHz, CDCl₃): δ 7.33–7.26 (m, 4H), 4.86 (q, J = 6.6 Hz, 1H), 1.99 (bs, 1H), 1.46 (d, J = 6.3 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 144.22, 133.03, 128.56, 126.75, 69.70, 25.22.

Br



White solid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.31$; ¹H NMR (300 MHz, CDCl₃): δ 7.49–7.43 (m, 2H), 7.26–7.21 (m, 2H), 4.84 (qd, J = 6.3 Hz and J = 3.6 Hz, 1H), 1.96 (d, J = 3.6 Hz, 1H), 1.46 (d, J = 6.6 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃): δ 144.74, 131.51, 127.11, 121.12, 69.74, 25.21.



Colorless liquid¹⁰; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.20$; ¹H NMR (500 MHz, CDCl₃): δ 7.29–7.24 (m, 1H), 6.97–6.92 (m, 2H), 6.84–6.79 (m, 1H), 4.87 (qd, J = 6.5 Hz and J = 3.5 Hz, 1H), 3.82 (s, 3H), 1.87 (d, J = 3.5 Hz, 1H), 1.49 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 159.80, 147.60, 129.56, 177.69, 112.91, 110.90, 70.37, 55.24, 25.16.

Colorless liquid¹⁰; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.24$; ¹H NMR (500 MHz, CDCl₃): δ 7.66–7.63 (m, 1H), 7.57–7.51 (m, 2H), 7.49–7.44 (m, 1H), 4.97 (q, J = 6.5 Hz, 1H), 1.93 (s, 1H), 1.52 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 146.71, 130.82 (q, ²J(C-C-F) = 31.8 Hz)), 128.94, 128.78, 124.24 (q, ⁴J(C-C-C-C-F) = 3.8 Hz)), 124.15 (q, ¹J(C-F) = 270.6 Hz)), 122.21 (q, ³J(C-C-C-F) = 3.9 Hz)), 69.85, 25.39. ¹⁹F NMR (470 MHz, CDCl₃): δ –62.59 (s).



OH Colorless liquid⁸; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.36$; ¹H NMR (500 MHz, CDCl₃): δ 7.52–7.48 (m, 1H), 7.25–7.21 (m, 1H), 7.19–7.10 (m, 2H), 5.12 (qd, J = 6.0 Hz and J = 3.0 Hz, 1H), 2.34 (s, 3H), 1.80 (d, J = 3.5 Hz, 1H), 1.46 (R)-20 (d, J = 6.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.85, 134.25, 130.39, 127.20, 126.40, 124.47, 66.84, 23.95, 18.93. OH Colorless liquid¹¹; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.32$; ¹H NMR (500 MHz, CDCl₃): δ 7.84–7.81 (m, 1H), 7.62–7.56 (m, 2H), 7.39–7.34 (m, 1H), 5.36–5.30 (m, 1H), 2.01–1.97 (m, 1H), 1.49 (d, J = 6.0 Hz, 3H); ¹³C NMR (125 (R)-2p MHz, CDCl₃): δ 145.02, 132.41, 127.37, 127.09, 126.48 (q, ²J(C-C-F) = 29.9 Hz)), 125.31 (q, ³J(C-C-C-F) = 6.2 Hz)), 124.38 (q, ¹J(C-F) = 272.5 Hz)), 65.70 (q, ⁴J(C-C-C-C-C-F) = 2.5 Hz)), 25.44. ¹⁹F NMR (470 MHz, CDCl₃): δ –58.32 (s, 3F).

Colorless liquid⁸; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.30$; ¹H NMR (500 MHz, CDCl₃): δ 7.60–7.57 (m, 1H), 7.33–7.27 (m, 2H), 7.22–7.17 (m, 1H), 5.27 (qd, J = 6.5 Hz and J = 3.5 Hz, 1H), 2.16 (d, J = 3.5 Hz, 1H), 1.48 (d, J = 6.0 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.06, 131.64, 129.41, 128.41, 127.23,

(*R*)-2q 3H); ¹³C NMR (125 MHz, CDCl₃): δ 143.06, 131.64, 129.41, 128.41, 127.23 126.42, 66.97, 23.52.



(R)-2r

White solid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.28$; ¹H NMR (500 MHz, CDCl₃): δ 7.45–7.42 (m, 2H), 7.40–7.35 (m, 4H), 7.32–7.28 (m, 3H), 7.25–7.21 (m, 1H), 6.72–6.66 (m, 1H), 6.41–6.36 (m, 1H), 5.39 (q, J = 3.0 Hz, 1H), 2.04 (d, J = 3.5 Hz, 1H); ¹³C NMR (125)

MHz, CDCl₃): δ 142.74, 136.49, 131.48, 130.55, 128.62, 128.56, 127.81, 127.78, 126.60, 126.33, 75.14.

Analytical data of the TMS-ether products ((S)-3a–3r)

Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.76$; ¹H NMR (500 MHz, CDCl₃): δ 8.11–8.07 (m, 1H), 7.88–7.85 (m, 1H), 7.76–7.72 (m, 1H), 7.69–7.66 (m, 1H), 7.52–7.44 (m, 2H), 5.59 (q, J = 6.5 Hz, 1H), 1.59 (d, J = 6.5 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 142.20,

133.71, 129.86, 128.85, 127.29, 125.60, 125.56, 125.19, 123.25, 122.72, 68.06, 26.47, 0.08.

OTMS

(S)-**3b**

Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.78$; ¹H NMR (500 MHz, CDCl₃): δ 7.35–7.29 (m, 4H), 7.25–7.20 (m, 1H), 4.85 (q, J = 6.5 Hz, 1H), 1.43 (d, J = 6.5 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 128.13, 126.82, 125.35, 70.57, 26.86, 0.09.



Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.77$; ¹H NMR (500 MHz, CDCl₃): δ 7.84–7.78 (m, 3H), 7.77–7.74 (m, 1H), 7.50–7.41 (m, 3H), 5.02 (q, J = 6.0 Hz, 1H), 1.51 (d, J = 6.5 Hz, 3H), 0.10 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 143.94, 133.29, 132.70, 127.89, 127.63,

125.89, 125.45, 124.08, 123.60, 70.75, 26.88, 0.13.



Colorless liquid³; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.75$; ¹H NMR (500 MHz, CDCl₃): δ 7.23–7.20 (m, 2H), 7.13–7.10 (m, 2H), 4.83 (q, J = 6.0 Hz, 1H), 2.33 (s, 3H), 1.42 (d, J = 6.5 Hz, 3H), 0.07 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 143.50, 136.35, 128.80, 125.29, 70.43, 26.88, 21.07,

0.10.



Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.71$; ¹H NMR (500 MHz, CDCl₃): δ 7.27–7.23 (m, 2H), 6.87–6.83 (m, 2H), 4.81 (q, J = 6.5 Hz, 1H), 3.79 (s, 3H), 1.41 (d, J = 6.5 Hz, 3H), 0.06 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 158.49, 138.66, 126.52, 113.47, 70.19,

55.21, 26.82, 0.11.

Colorless liquid¹²; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.75$; $[\alpha]_D^{20} = -40.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.60–7.54 (m, 4H), 7.44–7.39 (m, 4H), 7.35–7.31 (m, 1H), 4.91 (q, J = 6.5 Hz, 1H), 1.47 (d, J = 6.5 Hz, 3H), 0.11 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 145.54, 141.03, 139.75, 128.69, 127.08, 127.05, 126.91, 125.78, 70.32, 26.83, 0.13.

New compound: colorless liquid; TLC (acetone : n-hexane, 1:20 v/v): R_f = 0.72; $[\alpha]_D^{20} = -46.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.59–7.55 (m, 2H), 7.46–7.42 (m, 2H), 4.90 (q, J = 6.5 Hz, 1H), 1.42 (d, J (s)-3g = 6.5 Hz, 3H), 0.09 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 150.55, 129.07 (q, ²J(C-C-F) = 32.0 Hz)), 125.55, 125.15 (q, ³J(C-C-C-F) = 3.8 Hz)), 124.27 (q, ¹J(C-F) = 270.2 Hz)), 70.02, 26.85, 0.02. ¹⁹F NMR (470 MHz, CDCl₃): δ –62.33 (s, 3F); IR: 2961, 2868, 1620, 1416, 1325, 1253, 1165, 1125, 1095, 1032, 958, 840, 750 cm⁻¹; HRMS (m/z, EI): [M]⁺ calcd. for C₁₂H₁₇O₁F₃Si₁, 262.0995; found, 262.0996.

Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.73$; ¹H NMR (300 MHz, CDCl₃): δ 7.31–7.27 (m, 2H), 7.01–6.97 (m, 2H), 4.83 (q, J = 6.5 Hz, 1H), 1.41 (d, J = 6.5 Hz, 3H), 0.07 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 161.77 (d, ¹J(C-F) = 242.6 Hz)), 142.24 (d, ⁴J(C-C-C-F) = 3.0 Hz)), 126.86 (d, ³J(C-C-C-F) = 7.7 Hz)), 114.88 (d, ²J(C-C-F) = 21.1 Hz)), 69.98, 26.93, 0.05. ¹⁹F NMR (470 MHz, CDCl₃): δ –116.39 (s, 1F).



Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.74$; ¹H NMR (500 MHz, CDCl₃): δ 7.29–7.25 (m, 4H), 4.82 (q, J = 6.5 Hz, 1H), 1.40 (d, J = 6.0 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 145.05, 132.39, 128.27, 126.71, 69.93, 26.86, 0.04.



Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.75$; ¹H NMR (500 MHz, CDCl₃): δ 7.44–7.42 (m, 2H), 7.22–7.19 (m, 2H), 4.80 (q, J = 6.0 Hz, 1H), 1.40 (d, J = 6.0 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 145.58, 131.21, 127.09, 120.49, 69.96, 26.83, 0.04.

New compound: colorless liquid; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.77$; mp: 148-150 °C; $[\alpha]_D^{20} = -45.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.21–7.18 (m, 1H), 7.14–7.11 (m, 2H), 7.05–7.03 (m, 1H), 4.82 (q, J = 6.5 Hz, 1H), 2.35 (s, 3H), 1.42 (d, J = 6.5 Hz, 3H), 0.08 (s, 3H)

9H); ¹³C NMR (125 MHz, CDCl₃): δ 146.38, 137.66, 128.02, 127.56, 126.04, 122.43, 70.58, 26.86, 21.48, 0.11; IR: 2973, 2926, 2867, 1609, 1488, 1369, 1251, 1163, 1095, 1036, 964, 840, 750, 702 cm⁻¹; HRMS (m/z, EI): [M]⁺ calcd. for C₁₂H₂₀O₁Si₁, 208.1278; found, 208.1280.

New compound: colorless liquid; TLC (acetone : n-hexane, 1:20 v/v): R_f = 0.73; $[\alpha]_D^{20} = -45.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.24–7.20 (m, 1H), 6.92–6.89 (m, 2H), 6.78–6.76 (m, 1H), 4.83 (q, J = 6.5 Hz, 1H), 3.81 (s, 3H), 1.42 (d, J = 6.5 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 159.53, 148.26, 129.11, 117.75, 112.15, 110.93, 70.466, 55.15, 26.85, 0.08; IR: 2957, 2868, 1602, 1486, 1456, 1252, 1159, 1095, 1035, 969, 840, 750, 698 cm⁻¹; HRMS (m/z, EI): [M]⁺ calcd. for C₁₂H₂₀O₂Si₁, 224.1227; found, 224.1225.

Colorless liquid¹²; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.71$; $[\alpha]_D^{20} = -36.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.62–7.59 (m, 1H), 7.52–7.48 (m, 2H), 7.44–7.41 (m, 1H), 4.91 (q, J = 6.5 Hz, 1H), 1.44 (d, J = 6.0 Hz, 3H), 0.09 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 147.55, 130.50 (q, ²J(C-C-F) = 31.8 Hz)), 128.67 (d, ⁴J(C-C-C-F) = 1.0 Hz)), 128.60, 124.27 (q, ¹J(C-F) = 270.6 Hz)), 123.67 (q, ³J(C-C-C-F) = 3.8 Hz)), 122.13 (q, ³J(C-C-C-F) = 3.8 Hz)), 70.02, 26.86, 0.01. ¹⁹F NMR (470 MHz, CDCl₃): δ –62.57 (s, 3F).

Colorless liquid¹³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.72$; $[\alpha]_D^{20} = -47.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.33–7.32 (m, 1H), 7.25–7.18 (m, 3H), 4.81 (q, J = 6.5 Hz, 1H), 1.40 (d, J = 6.5 Hz, 3H), 0.09 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 148.67, 134.03, 129.44, 126.93, 125.53, 123.45, 69.95, 26.83, 0.04.

Colorless liquid¹⁴; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.71$; ¹H NMR (500 MHz, CDCl₃): δ 7.52–7.48 (m, 1H), 7.21–7.18 (m, 1H), 7.14–7.07 (m, 2H), 5.04 (q, J = 6.0 Hz, 1H), 2.31 (s, 3H), 1.38 (d, J = 6.5 Hz, 3H), 0.06 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 144.58, 133.00, 129.91, 126.48, 126.00, 125.30, 67.38, 25.58, 18.87, -0.01.

New compound: colorless liquid; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.70; [\alpha]_D{}^{20} = -16.0 (c = 0.05 in CHCl_3); {}^{1}H NMR (500 MHz, CDCl_3): \delta 7.83-7.81 (m, 1H), 7.57-7.53 (m, 2H), 7.33-7.30 (m, 1H), 5.26-5.22 (m, 1H), 1.40 (s)-3p (d, J = 6.0 Hz, 3H), 0.03 (s, 9H); {}^{13}C NMR (125 MHz, CDCl_3): \delta 146.23, 132.08, 127.71, 126.76, 125.57 (q, {}^{2}J(C-C-F) = 29.9 Hz)), 124.93 (q, {}^{3}J(C-C-C-F) = 5.9 Hz)), 124.47 (q, {}^{1}J(C-F) = 272.1 Hz)), 66.15 (q, {}^{4}J(C-C-C-F) = 2.0 Hz)), 27.62, -0.16. {}^{19}F NMR (470 MHz, CDCl_3): \delta -58.39 (s, 3F); IR: 2979, 2960, 2902, 1609, 1454, 1373, 1313, 1253, 1164, 1121, 1092, 1060, 1027, 957, 841, 767 cm⁻¹; HRMS (m/z, EI): [M]⁺ calcd. for C₁₂H₁₇O₁F₃Si₁, 262.0995; found, 262.0993.$

New compound: colorless liquid; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.73$; $[\alpha]_D^{20} = -49.0$ (c = 0.1 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ 7.61– 7.59 (m, 1H), 7.29–7.25 (m, 2H), 7.18–7.14 (m, 1H), 5.22 (q, J = 6.5 Hz, 1H), (S)-3q 1.40 (d, J = 6.0 Hz, 3H), 0.08 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 144.03, 130.75, 128.96, 127.88, 127.14, 126.95, 67.16, 25.35, -0.06; IR: 3069, 2959, 2899, 1574, 1473, 1439, 1370, 1252, 1203, 1099, 1029, 957, 840, 752, 691 cm⁻¹; HRMS (m/z, EI): [M]⁺ calcd. for C₁₁H₁₇O₁Cl₁Si₁, 228.0732; found, 228.0730.

Colorless liquid³; TLC (acetone : n-hexane, 1:20 v/v): $R_f = 0.66$; ¹H NMR (500 MHz, CDCl₃): δ 7.40–7.29 (m, 8H), 7.26–7.19 (m, 2H), 6.63–6.58 (m, 1H), 6.33–6.27 (m, 1H), 5.33 (d, J = 6.5 Hz, 1H), 0.14 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 143.50, 136.81, 132.79, 129.24, 128.49, 128.29, 127.49, 127.21, 126.54, 126.19, 75.49, 0.28.

Analytical data of the meso-diols (4a-4c)



White solid¹⁵; TLC (acetone : n-hexane, 1:2 v/v): $R_f = 0.31$; ¹H NMR (500 MHz, CDCl₃): δ 7.29–7.24 (m, 10H), 4.81 (s, 2H), 2.26 (s, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 139.71, 128.20, 128.08, 127.05, 78.05.



White solid¹⁶; TLC (acetone : n-hexane, 1:2 v/v): $R_f = 0.28$; ¹H NMR (500 MHz, CDCl₃): δ 7.23–7.21 (m, 2H), 7.17–7.15 (m, 2H), 6.90–6.89 (m, 2H), 6.82–6.80 (m, 2H), 5.26–5.23 (m, 2H), 3.69 (s, 6H), 3.11–3.08 (m, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 156.88, 128.55,

128.49, 128.26, 120.48, 110.22, 73.78, 55.24.



White solid¹⁶; TLC (acetone : n-hexane, 1:2 v/v): $R_f = 0.24$; ¹H NMR (500 MHz, CDCl₃): δ 7.36–7.33 (m, 2H), 7.20–7.17 (m, 4H), 7.10–7.08 (m, 2H), 5.22–5.19 (m, 2H), 2.14–2.11 (m, 2H); ¹³C NMR (125 MHz, CDCl₃): δ 138.16, 136.19, 130.10, 127.80, 126.64, 126.12, 73.42, 19.21.

Analytical data of the TMS-ether products (5a-5c)



Colorless liquid¹⁷; TLC (acetone : n-hexane, 1:5 v/v): R_f = 0.59; ¹H NMR (500 MHz, CDCl₃): δ 7.29–7.19 (m, 10H), 4.72–4.70 (m, 1H), 4.67–4.65 (m, 1H), 2.27 (d, J = 3.5 Hz, 1H), -0.12 (s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 140.64, 127.85, 127.73, 127.67, 127.48, 127.26, 127.11, 79.05, 0 28

78.46, 77.21, -0.28.



New compound: colorless liquid; TLC (acetone : n-hexane, 1:5 v/v): $R_f = 0.54$; $[\alpha]_D^{20} = +42.0$ (c = 0.1 in benzene); ¹H NMR (500 MHz, CDCl₃): δ 7.35–7.33 (m, 1H), 7.17–7.13 (m, 2H), 6.95–6.94 (m, 1H), 6.90–6.86 (m, 1H), 6.80–6.76 (m, 2H), 6.69–6.65 (m, 1H), 5.53–5.50

(m, 1H), 5.11–5.07 (m, 1H), 3.77 (s, 3H), 3.49 (s, 3H), 3.20 (d, J = 6.5 Hz, 1H), -0.02(s, 9H); ¹³C NMR (125 MHz, CDCl₃): δ 156.99, 156.27, 129.24, 128.90, 128.13, 128.10, 127.88, 127.83, 120.02, 119.94, 109.69, 109.47, 74.06, 69.87, 55.26, 55.05, -0.25; IR: 2956, 2835, 1602, 1588, 1491, 1462, 1438, 1286, 1242, 1188, 1112, 1073, 1049, 1030, 901, 840, 751, 661 cm⁻¹; HRMS (m/z, FAB): [M]⁺ calcd. for C₁₉H₂₅O₄Si₁, 345.1516; found, 345.1517.



General procedure for silylative kinetic resolution of racemic alcohols

On a 1 mmol scale (Figure 4)

Catalyst stock solution: 11.9 mg of 1e in 50 mL of CH₂Cl₂

To a solution of the catalyst stock solution (0.5 mL = 0.119 mg, 0.01 mol%) and racemic secondary alcohol **2a** (172.2 mg, 1.0 mmol) in CH₂Cl₂ (4.5 mL), spray dried KF (58.1 mg, 100 mol%) and Amberite CG 50 (80 mg, 10 mmol H⁺ per gram) were added in one portion. The reaction mixture was stirred at -30 °C for 20 min, followed by adding hexamethyldisilazane (0.15 mL, 0.7 equiv.) to the reaction mixture. After stirring for 24 h at -30 °C, the reaction mixture was filtered, and the filtrate was concentrated in vacuo to afford a colorless oil. The residue thus obtained was purified by short silica gel chromatography (EA/hexanes = 1/5) and analyzed by chiral HPLC (Chiralcel OD-H, hexanes/IPA = 90/10, 0.7 mL/min, 220 nm). The enatiomeric excess of TMS-ether product **3a** was determined after the deprotection of its trimethylsilyl group with TBAF.

The absolute configurations of remaining alcohols 2a-2r were determined by comparison of the retention time of HPLC with the literature data^{3,7-11}. Silyl ethers 3a-3f, 3i-3j, 3m-3o and 3r are known compounds in literature, and their spectroscopic data were consistent with previously reported values^{3,12-14}.

Selectivity factors (k_{rel}) were calculated according to Kagan's equation¹: $k_{rel} = \ln((1-c)(1-ee_{rsm}))/\ln((1-c)(1+ee_{rsm})) = \ln(1-c(1+ee_{prod}))/\ln(1-c(1-ee_{prod}))$, wherein *c* is conversion of the reaction, ee_{prod} is the enantiomeric excess of the silvl ether product and ee_{rsm} is the enantiomeric excess of the recovered alcohol. Conversions (*c*) were calculated by the following equation: $c = ee_{rsm}/(ee_{prod}+ee_{rsm})$.

On a 10 mmol scale (Figure 5a)

Catalyst stock solution: 1.2 mg of 1e in 250 mL of CH₂Cl₂

To a solution of the catalyst stock solution (2.5 mL = 0.012 mg, 0.0001 mol%) and racemic secondary alcohol **2a** (1722 mg, 10 mmol) in CH₂Cl₂ (22.5 mL), spray dried KF (290 mg, 50 mol%) and Amberite CG 50 (200 mg, 10 mmol H⁺ per gram) were added in one portion. The reaction mixture was stirred at -15 °C for 20 min followed by adding HMDS (1.5 mL, 0.7 equiv) to the reaction mixture. After stirring for 14 d at -15 °C, the reaction mixture was filtered, and the filtrate was concentrated in vacuo to afford a colorless oil. The crude product was purified by short silica gel chromatography (EA/hexanes = 1/5) and analyzed by chiral HPLC (Chiralcel OD-H, hexanes/IPA = 90/10, 0.7 mL/min, 220 nm) to afford 900 mg of silyl ether (*S*)-**3a** (41% yield, 94 % *ee*) and 774 mg of recovered alcohol (*R*)-**2a** (44% yield, 77% *ee*) as colorless oils. The selectivity factor was 77 at 45.0% conversion.

General procedure for silvlative desymmetrization of meso-diols

To a solution of chiral catalyst **1e** (1.2 mg, 1.0 mol%) and *mseo*-diol **4a** (21.4 mg, 0.1 mmol) in distilled dichloroethane (2.0 mL), spray dried KF (5.8 mg, 100 mol%) and Amberite CG 50 (6.0 mg) were added in one portion. The reaction mixture was stirred at 20 °C for 10 min, followed by slow addition of HMDS (0.04 mL, 2.0 equiv) to the reaction mixture. After stirring for 48 h at 20 °C, the reaction mixture was filtered, and the filtrate was concentrated in vacuo to afford a colorless oil. The conversion of the reaction was determined from ¹H-NMR spectrum of the residue, (95.5% conv.); mono-silyl ether **5a**/di-silyl ether = 94/6.; The product was purified by silica gel chromatography (dichloromethane/hexanes = 1/20) and analyzed by chiral HPLC (Chiralcel OD-H, hexanes/IPA = 98/2, 1.0 mL/min, 220 nm) to afford 25.8 mg of silyl ether **5a** (90% yield, 92:8 e.r.) as a colorless oil.

Determination of the absolute configuration of 5a

The absolute configuration of **5a** was established to be (*S*)-1-silyloxy-(*R*)-2-hydroxy-1,2diphenylethane by its conversion to the corresponding (*R*)-1-acetoxy-(*S*)-2-hydroxy-1,2diphenylethane^{15,16,18}. The absolute configuration of TMS-ether **5b** and **5c** were assigned by analogy.



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