

**Fluorinated Indeno-quinoxaline bearing thiazole moieties as hypoglycemic agents targeting  $\alpha$ -amylase, and  $\alpha$ -glucosidase: synthesis, molecular docking, and ADMET studies.**

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## S1 Material and methods

### 3.1. Chemistry

Inaccurate melting points (MPs) of the novel designed compounds are determined and reported on the digital Gallen Kamp MFB-595 instrument. A Shimadzu 440 spectrophotometer is used to measure the IR spectra between 400 – 4000  $\text{cm}^{-1}$ . A Bruker (400 and 101 MHz) spectrometer is used to evaluate the  $^1\text{H}$  and  $^{13}\text{C}$  signals in the NMR spectra and is recorded relative to deuterated solvent signals only in dimethyl sulfoxide ( $\text{DMSO-}d_6$ ). Values of the chemical shift were listed as  $\delta$  ppm units. Also, the starting material 7-fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-one (**3**) was constructed according to the previous work in a good yield = 65 %, as pale-orange color and melting point in arrange of M.p. = 253–254  $^{\circ}\text{C}$ .

#### 3.1.1. Synthesis of Schiff's bases (4-6)

To a solution of starting material (**3**) (0.01 mol) and amino thiazole derivatives (0.01 mol) namely, 2-aminothiazol-4(5*H*)-one, 1-(2-amino-4-methylthiazol-5-yl)ethan-1-one or ethyl 2-amino-4-methylthiazole-5-carboxylate in ethanol / acetic acid mixture (5 / 15 mL) was heated under reflux 3-4 h (monitored by TLC). Upon the reaction's completion, the solid formed was collected, filtered out, and recrystallized from the appropriate solvent.

#### 3.1.2. 2-((7-Fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)amino)thiazol-4(5*H*)-one (**4**)

Yield: 76 % as light-orange crystals from EtOH; mp. 190–192  $^{\circ}\text{C}$ ; IR:  $\nu/\text{cm}^{-1}$  = 3051 (CH-Arom), 2981 (CH-Aliph), 1736 (C=O), 1604 (C=N);  $^1\text{H}$  NMR  $\delta/\text{ppm}$  = 4.08 (2H, s,  $\text{CH}_2$ ), 7.69 (1H, t,  $J=8.0$  Hz, H-Arom), 7.76 (1H, t,  $J=8.0$  Hz, H-Arom), 7.87 (1H, s, H-Arom), 7.89 (1H, d,  $J=8.0$  Hz, H-Arom), 7.94 (1H, d,  $J=8.0$  Hz, H-Arom), 8.07 (1H, d,  $J=8.0$  Hz, H-Arom), 8.24 (1H, dd,  $J=8.6, 4.8$  Hz, H-Arom);  $^{13}\text{C}$  NMR  $\delta/\text{ppm}$  37.46 ( $\text{CH}_2$ ), 113.94, 114.16, 115.78, 117.28, 118.68, 118.85, 120.59, 121.35, 122.96, 124.76, 133.60, 133.81, 137.46, 145.60, 163.57 (C=N), 173.55 (C-F), 178.41 (C=O), 179.02 (S-C=N). Anal. Calcd. for  $\text{C}_{18}\text{H}_9\text{FN}_4\text{OS}$  (348.36): C, 62.06; H, 2.60; N, 16.08; Found: C, 62.48; H, 2.34; N, 16.54 %.

### 3.1.3. 1-(2-((7-Fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)amino)-4-methylthiazol-5-yl)ethan-1-one (5)

Yield: 72 % as orange crystals from EtOH; mp. 198–200 °C; IR:  $\nu/\text{cm}^{-1}$  = 3053 (CH-Arom), 2950 (CH-Aliph), 1735 (C=O), 1604 (C=N);  $^1\text{H}$  NMR  $\delta/\text{ppm}$  = 2.33 (3H, s, CH<sub>3</sub>), 2.41 (3H, s, CH<sub>3</sub>), 7.72 (1H, t,  $J=8$  Hz, H-Arom), 7.79 (1H, d,  $J=6.1$  Hz, H-Arom), 7.87 (1H, s, H-Arom), 7.90 (1H, d,  $J=5.3$  Hz, H-Arom), 7.96 (1H, d,  $J=6.8$  Hz, H-Arom), 8.08 (1H, t,  $J=4$  Hz, H-Arom), 8.25 (1H, dd,  $J=9.2, 6.0$  Hz, H-Arom);  $^{13}\text{C}$  NMR  $\delta/\text{ppm}$  18.15 (CH<sub>3</sub>), 29.44 (CH<sub>3</sub>), 113.48, 113.70, 119.85, 120.10, 121.22, 122.25, 122.48, 124.29, 133.11, 133.34, 133.45, 136.83, 136.96, 139.13, 140.66, 157.33 (C=N), 170.41 (C-F), 188.26 (S-C=N), 189.04 (COCH<sub>3</sub>). Anal. Calcd. for C<sub>21</sub>H<sub>13</sub>FN<sub>4</sub>OS (388.42): C, 64.94; H, 3.37; N, 14.42; Found: C, 64.68; H, 3.44; N, 14.24 %.

### 3.1.4. Ethyl 2-((7-fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)amino)-4-methylthiazole-5-carboxylate (6)

Yield: 70 % as red crystals from EtOH; mp. 180–182 °C; IR:  $\nu/\text{cm}^{-1}$  = 3076 (CH-Arom), 2925 (CH-Aliph), 1735 (C=O), 1605 (C=N);  $^1\text{H}$  NMR  $\delta/\text{ppm}$  = 1.21 (3H, t, CH<sub>3</sub>CH<sub>2</sub>O-), 2.38 (3H, s, CH<sub>3</sub>), 4.13 (2H, q, CH<sub>3</sub>CH<sub>2</sub>O-), 7.72 (1H, t,  $J=8$  Hz, H-Arom), 7.79 (1H, d,  $J=8.6$  Hz, H-Arom), 7.88 (1H, s, H-Arom), 7.89 (1H, d,  $J=7.5$  Hz, H-Arom), 7.95 (1H, t,  $J=8.0$  Hz, H-Arom), 8.08 (1H, d,  $J=7.2$  Hz, H-Arom), 8.25 (1H, dd,  $J=9.1, 6.3$  Hz, H-Arom);  $^{13}\text{C}$  NMR  $\delta/\text{ppm}$  14.07 (CH<sub>3</sub>CH<sub>2</sub>O-), 27.42 (CH<sub>3</sub>), 61.12 (CH<sub>3</sub>CH<sub>2</sub>O-), 113.38, 113.63, 119.57, 122.52, 130.03, 132.67, 135.22, 139.49, 146.69, 155.28, 158.27 (Cs.Arom), 165.00 (C=N), 166.36 (C-F), 169.24 (CO.ester), 173.68 (S-C=N); Anal. Calcd. for C<sub>22</sub>H<sub>15</sub>FN<sub>4</sub>O<sub>2</sub>S (418.45): C, 63.15; H, 3.61; N, 13.39; Found: C, 63.27; H, 3.41; N, 13.12 %.

### 3.1.5. Synthesis of thiazolyl hydrazone derivatives (7-9)

7-fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-one (3) (0.01 mol) was dissolved in absolute ethanol (15 mL) and the appropriate hydrazine derivatives (0.01 mol) such as 2-hydrazineylthiazol-4(5*H*)-one, 1-(2-hydrazineyl-4-methylthiazol-5-yl)ethan-1-one, or ethyl 2-hydrazineyl-4-methylthiazole-5-carboxylate were added. The reaction mixture was heated under reflux for 1 h (monitored by TLC). Upon the reaction's completion, the solid formed was collected, filtered out, and recrystallized from the appropriate solvent.

### 3.1.6. 2-(2-(7-Fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)thiazol-4(5*H*)-one (7)

Yield: 78 % as light-yellow crystals from EtOH; mp. 171–173 °C; IR:  $\nu/\text{cm}^{-1}$  = 3130 (NH), 3049 (CH-Arom), 2967 (CH-Aliph), 1709 (C=O), 16125 (C=N);  $^1\text{H}$  NMR  $\delta/\text{ppm}$  = 4.06 (2H, s, CH<sub>2</sub>), 7.05 (1H, m, H-Arom), 7.68 (2H, m, H-Arom), 7.87 (1H, t,  $J=4.0$  Hz, H-Arom), 7.90 (1H, s, H-Arom), 8.04 (1H, d,  $J=7.7$  Hz, H-Arom), 8.22 (1H, dd,  $J=16.8, 8.2$  Hz, H-Arom), 11.77 (1H, s, NH, disappeared by D<sub>2</sub>O);  $^{13}\text{C}$  NMR  $\delta/\text{ppm}$  33.03 (CH<sub>2</sub>), 113.64, 120.03, 122.07, 122.44, 130.00,

132.27, 134.61, 136.58, 136.75, 138.93, 142.23, 145.59 (Cs.Arom), 158.70, 160.05 (C=N), 162.35 (C-F), 168.75 (S-C=N), 173.28 (C=O); Anal. Calcd. for **C<sub>18</sub>H<sub>10</sub>FN<sub>5</sub>OS** (363.37): C, 59.50; H, 2.77; N, 19.27; Found: C, 59.65; H, 2.78; N, 19.03 %.

### 3.1.7. 1-(2-(2-(7-Fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)-4-methyl-thiazol-5-yl)ethan-1-one (8)

Yield: 67 % as light-orange crystals from EtOH; mp. 174–176 °C; IR:  $\nu/\text{cm}^{-1}$  = 3427 (NH), 3076 (CH-Arom), 2980 (CH-Aliph), 1735 (C=O), 1605 (C=N); <sup>1</sup>H NMR  $\delta/\text{ppm}$  = 1.94 (3H, s, CH<sub>3</sub>), 2.29 (3H, s, CH<sub>3</sub>), 7.72 (1H, t,  $J=6.0$  Hz, H-Arom), 7.77 (1H, t,  $J=5.8$  Hz, H-Arom), 7.87 (1H, s, H-Arom), 7.90 (1H, d,  $J=5.6$  Hz, H-Arom), 7.94 (1H, d,  $J=8.0$  Hz, H-Arom), 7.96 (1H, s, NH, disappeared by D<sub>2</sub>O), 8.07 (1H, d,  $J=8$  Hz, Ar-H), 8.23 (1H, dd,  $J=9.2, 6.0$  Hz, Ar-H); <sup>13</sup>C NMR  $\delta/\text{ppm}$  16.18 (CH<sub>3</sub>), 27.76 (CH<sub>3</sub>), 114.21, 118.72, 120.10, 121.48, 122.37, 123.72, 124.21, 131.49, 132.08, 133.52, 135.18, 136.04, 137.21, 138.38, 141.00 (Cs.Arom), 155.36 (C=N), 171.49 (C-F), 188.30 (S-C=N), 190.27 (CH<sub>3</sub>CO); Anal. Calcd. for **C<sub>21</sub>H<sub>14</sub>FN<sub>5</sub>OS** (403.44): C, 62.52; H, 3.50; N, 17.36; Found: C, 62.43; H, 3.41; N, 17.75 %.

### 3.1.8. Ethyl 2-(2-(7-fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)-4-methyl-thiazole-5-carboxylate (9)

Yield: 69 % as deep-red crystals from EtOH; mp. 151–153 °C; IR:  $\nu/\text{cm}^{-1}$  = 3425 (NH), 3073 (CH-arom), 2985 (CH-Aliph), 1714 (C=O), 1605 (C=N); <sup>1</sup>H NMR  $\delta/\text{ppm}$  = 1.27 (3H, t, CH<sub>3</sub>CH<sub>2</sub>O-), 2.57 (3H, s, CH<sub>3</sub>), 4.23 (2H, q, CH<sub>3</sub>CH<sub>2</sub>O-), 7.71 (1H, t,  $J=7.5$  Hz, H-Arom), 7.76 (1H, t,  $J=5.6$  Hz, H-Arom), 7.85 (1H, d,  $J=6.8$  Hz, H-Arom), 7.88 (1H, s, H-Arom), 7.90 (1H, s, NH, disappeared by D<sub>2</sub>O), 7.94 (1H, d,  $J=6.8$  Hz, H-Arom), 8.07 (1H, d,  $J=7.4$  Hz, H-Arom), 8.24 (1H, dd,  $J=9.2, 6.0$  Hz, H-Arom); <sup>13</sup> NMR  $\delta/\text{ppm}$  19.23 (CH<sub>3</sub>CH<sub>2</sub>O-), 21.02 (CH<sub>3</sub>), 61.94 (CH<sub>3</sub>CH<sub>2</sub>O-), 104.53, 113.45, 113.68, 114.69, 117.55, 119.83, 120.08, 122.46, 124.27, 133.09, 133.33, 133.43, 136.81, 136.95, 137.03, 139.13, 140.63, 148.26 (Cs.Arom), 149.52 (C=N), 157.31 (C-F), 161.95 (S-C=N), 164.62 (C=O); Anal. Calcd. for **C<sub>22</sub>H<sub>16</sub>FN<sub>5</sub>O<sub>2</sub>S** (433.46): C, 60.96; H, 3.72; N, 16.16; Found: C, 60.57; H, 3.55; N, 16.63 %.

### 3.1.9. Synthesis of 5-(7-Fluoro-11*H*-indeno[1,2-*b*]quinoxalin-11-ylidene)-2-hydrazineyl-thiazol-4(5*H*)-one (10)

To a solution of compound **3** (0.01 mol) and 2-hydrazineylthiazol-4(5*H*)-one (0.01 mol) in acetic acid (15 mL) as solvent containing sodium acetate (0.3 g) as a catalyst. The reaction mixture was heated under reflux for 4 h (monitored by TLC). Upon the reaction's completion, left to cool and poured onto ice-cold water, then neutralized with a few drops of 1 M HCl. The solid that formed was collected by filtration, washed with water, and recrystallized from the appropriate solvent.

Yield: 82 % as a pale-white powder from EtOH; mp. 100–102 °C; IR:  $\nu/\text{cm}^{-1}$  = 3415 (NH<sub>2</sub>), 3190 (NH), 3076 (CH-Arom), 1711 (C=O), 1624 (C=N), 1570 (C=C); <sup>1</sup>H NMR  $\delta/\text{ppm}$  = 5.43 (2H, s, NH<sub>2</sub>, disappeared by D<sub>2</sub>O), 7.73 (1H, t,  $J=7.8$  Hz, H-Arom), 7.78 (1H, t,  $J=8.0$  Hz, H-Arom), 7.84 (1H, d,  $J=7.3$  Hz, H-Arom), 7.87(1H, s, H-Arom), 7.89 (1H, s, NH, disappeared by D<sub>2</sub>O), 7.92 (1H, d,  $J=7.9$  Hz, H-Arom), 8.05 (1H, d,  $J=7.4$  Hz, H-Arom), 8.13 (1H, dd,  $J=9.2, 6.0$  Hz, H-Arom); C <sup>13</sup> NMR  $\delta/\text{ppm}$  112.70, 112.93, 113.38, 121.76, 122.40, 124.19, 130.10, 131.02, 133.03, 134.36, 135.51, 136.64, 153.36, 154.83 (Cs.Arom), 157.13 (C=N), 158.91 (S-C=N), 161.58 (C-F), 172.16 (C=O); Anal. Calcd. for C<sub>18</sub>H<sub>10</sub>FN<sub>5</sub>OS (363.37): C, 59.50; H, 2.77; N, 19.27; Found: C, 59.46; H, 2.56; N, 19.46 %.

### 3.1.10. Synthesis of 5-(7-fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)-2-(2-((substituted)-hydrazineyl)thiazol-4(5H)-one (12 &13)

To a solution of compound **3** (0.01 mol) and the corresponding reagent **11a** or **11b** (0.01 mol) in acetic acid (15 mL) as solvent containing sodium acetate (0.3 g) as a catalyst. The reaction mixture was heated under reflux for 4 h (monitored by TLC). Upon the reaction's completion, left to cool and poured onto ice-cold water, then neutralized with a few drops of 1 M HCl. The solid that formed was collected by filtration, washed with water, and recrystallized from the appropriate solvent.

### 3.1.11. 5-(7-Fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)-2-(2-((1-phenyl-3-(*p*-tolyl)-1H-pyrazol-4-yl)methylene)hydrazineyl)thiazol-4(5H)-one (12)

Yield: 64 % as a pale-brown powder from AcOH; mp. 270–272 °C; IR:  $\nu/\text{cm}^{-1}$  = 3422 (NH), 3052 (CH-Arom), 2924 (CH-Aliph), 1729 (C=O), 1630 (C=N), 1570 (C=C); <sup>1</sup>H NMR  $\delta/\text{ppm}$  = 2.37 (3H, s, CH<sub>3</sub>.tolyl), 7.29 (2H, d,  $J=6.5$  Hz, H-Arom), 7.35 (1H, t,  $J=8.0$  Hz, H-Arom), 7.41 (1H, d,  $J=8.0$  Hz, H-Arom), 7.51 (2H, d,  $J=8.2$  Hz, H-Arom), 7.56 (2H, d,  $J=8.2$  Hz, H-Arom), 7.63 (1H, t,  $J=7.4$  Hz, H-Arom), 7.68 (1H, s, methine-CH), 7.70 (1H, s, H-Arom), 7.80 (1H, d,  $J=8.0$  Hz, H-Arom), 7.85 (1H, d,  $J=9.0$  Hz, H-Arom), 7.89 (1H, s, H-Arom), 7.95 (2H, t,  $J=7.4$  Hz, H-Arom), 8.01 (1H, t,  $J=7.1$  Hz, H-Arom), 8.19 (1H, dd,  $J=8.7, 6.5$  Hz, H-Arom), 8.90 (1H, s, NH, disappeared by D<sub>2</sub>O); <sup>13</sup>C NMR  $\delta/\text{ppm}$  21.05 (CH<sub>3</sub>.tolyl), 113.51, 116.71, 118.95, 119.36, 122.58, 124.38, 127.21, 127.86, 128.62, 129.22, 129.79, 132.68, 133.23, 134.95, 136.79, 137.10, 138.20, 138.71, 139.03, 140.64, 149.45, 151.73 (Cs.Arom), 152.83 (C=N), 157.32(C=C), 164.99 (S-C=N), 172.31 (C-F), 174.44 (C=O); Anal. Calcd. for C<sub>35</sub>H<sub>22</sub>FN<sub>7</sub>OS (607.67): C, 69.18; H, 3.65; N, 16.14; Found: C, 69.21; H, 3.45; N, 16.12 %.

### 3.1.12. 5-(7-Fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)-2-(2-(2-oxoindolin-3-ylidene)-hydrazineyl)thiazol-4(5H)-one (13)

Yield: 65 % as a light-red powder from AcOH; mp. 283–285 °C; IR:  $\nu/\text{cm}^{-1}$  = 3178 (br. 2NH), 3075 (CH-Arom), 1695 (br. 2C=O), 1615 (C=N), 1569 (C=C); <sup>1</sup>H NMR  $\delta/\text{ppm}$  = 6.84 (1H, t,  $J=6.0$  Hz, H-Arom), 6.99 (1H, d,  $J=7.4$  Hz, H-Arom), 7.03 (1H, d,  $J=7.6$  Hz, H-Arom), 7.29 (1H, t,  $J=8.0$  Hz, H-Arom), 7.36 (1H, d,  $J=7.7$  Hz, H-Arom), 7.69 (1H, t,  $J=7.4$  Hz, H-Arom), 7.75 (1H, t,  $J=6.0$  Hz, H-Arom), 7.88 (1H, s, H-Arom), 7.91 (1H, d,  $J=9.5$  Hz, H-Arom), 8.04 (1H, d,

$J=6.5$  Hz, H-Arom), 8.21 (1H, dd,  $J=11.9, 5.7$  Hz, H-Arom), 10.67 (1H, s, NH, disappeared by D<sub>2</sub>O), 11.05 (1H, s, NH, disappeared by D<sub>2</sub>O); <sup>13</sup>C NMR  $\delta$ /ppm 110.29, 113.42, 117.34, 119.78, 120.56, 121.61, 122.42, 124.21, 127.94, 128.19, 131.25, 132.35, 132.74, 133.03, 136.89, 138.35, 139.10, 140.58, 142.92, 143.94 (Cs.Arom), 157.24 (C=N), 164.97 (S-C=N), 168.86 (C-F), 171.92, 171.97 (2C=O); Anal. Calcd. for C<sub>26</sub>H<sub>13</sub>FN<sub>6</sub>O<sub>2</sub>S (492.49): C, 63.41; H, 2.66; N, 17.06; Found: C, 63.55; H, 2.41; N, 17.01%.

### 3.1.13. Synthesis of 2-(7-fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazine-1-carbo-thioamide (14)

To a solution of the starting material (**3**) (0.01 mol) in ethanolic solution (15 mL) containing a few drops of acetic acid, thiosemicarbazide (0.01 mol) was added. The reaction mixture was heated under reflux for 1.5 h (monitored by TLC). Upon the reaction's completion, the solid that formed was left to cool and collected by filtration, washed with cold ethanol, and recrystallized from the appropriate solvent.

Yield: 77 % as a light-green crystal from AcOH; mp. 210–212 °C; IR:  $\nu/\text{cm}^{-1}$  = 3448, 3396 (NH<sub>2</sub>), 3275 (NH), 3158 (CH-Arom), 2950 (CH-Aliph), 1610 (C=N); <sup>1</sup>H NMR  $\delta$ /ppm = 7.64 (d,  $J=7.3$  Hz, 1H, H-Arom), 7.67 (t,  $J=7.4$  Hz, 1H, H-Arom), 7.77 (t,  $J=8.6$  Hz, 1H, H-Arom), 7.96 (d,  $J=9.2$  Hz, 1H, H-Arom), 8.06 (d,  $J=6.4$  Hz, 1H, H-Arom), 8.15 (s, 2H, NH<sub>2</sub> exchangeable by D<sub>2</sub>O), 8.79 (d,  $J=7.5$  Hz, 1H, H-Arom), 9.05 (s, 1H, H-Arom), 12.48 (s, 1H, NH exchangeable by D<sub>2</sub>O); <sup>13</sup>C NMR  $\delta$ /ppm 113.94, 114.17, 115.77, 120.19, 122.94, 124.75, 129.88, 132.97, 133.58, 137.43, 141.11, 142.88, 144.48 (Cs-Arom), 154.76 (C=N-NH), 161.93 (C-F), 182.02 (C=S); Anal. Calcd. for C<sub>16</sub>H<sub>10</sub>FN<sub>5</sub>S (323.35): C, 59.43; H, 3.12; N, 21.66; Found: C, 59.52; H, 3.22; N, 21.21%.

### 3.1.14. The reaction of thiosemicarbazone derivatives (14) with a different alkylating agent (15-17)

A solution of compound **14** (0.01 mol), an appropriate alkylating agent (0.01 mol) namely, ethyl 2-chloropropanoate, ethyl 2-chlorobutanoate, and diethyl but-2-ynedioate in ethanol / acetic acid mixture (15 / 5 mL) containing sodium acetate (0.2 gm) were added. The reaction mixture was heated under reflux for 3-5 h (monitored by TLC). Upon the reaction's completion, left to cool and poured onto ice-cold water, then neutralized with a few drops of 1 M HCl. The solid that formed was collected by filtration, washed with water, and recrystallized from the appropriate solvent.

### 3.1.15. 2-(2-(7-Fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)-5-methyl-thi-azol-4(5H)-one (15)

Yield: 66 % as a light-yellow crystal from AcOH; mp. 265–267 °C; IR:  $\nu/\text{cm}^{-1}$  = 3060 (CH-Arom), 2928 (CH-Aliph), 1722 (C=O), 1631 (C=N); <sup>1</sup>H NMR  $\delta$ /ppm = 1.45 (3H, d, CH<sub>3</sub>.thiazole), 3.81 (1H, q, CH. thiazole), 7.54 (1H, t,  $J=7.4$  Hz, H-Arom), 7.64 (1H, d,  $J=7.3$  Hz, H-Arom), 7.93 (1H, s, H-Arom), 7.99 (1H, d,  $J=9.6$  Hz, H-Arom), 8.08 (1H, d,  $J=7.4$  Hz, H-Arom), 8.17 (1H, t,  $J=8.2$  Hz, H-Arom), 8.26 (1H, dd,  $J=9.1, 5.9$  Hz, H-Arom), 8.84 (1H, s, 1NH disappeared by

D<sub>2</sub>O); <sup>13</sup>C NMR δ/ppm 19.85 (CH<sub>3</sub>.thiazole), 57.87 (CH.thiazole), 113.39, 113.61, 119.71, 119.96, 120.51, 120.75, 121.87, 122.04, 122.40, 131.14, 131.89, 132.49, 135.02, 135.99, 136.91, 138.29 (Cs.Arom), 154.29 (C-F), 178.79 (C=O); Anal. Calcd. for **C<sub>19</sub>H<sub>12</sub>FN<sub>5</sub>OS** (377.40): C, 60.47; H, 3.21; N, 18.56; Found: C, 60.46; H, 3.56; N, 18.34%.

### 3.1.16. 5-Ethyl-2-(2-(7-fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)thiazol-4(5H)-one (16)

Yield: 69 % as a yellow crystal from AcOH; mp. 255–257 °C; IR: ν/cm<sup>-1</sup> = 3239 (NH), 3054 (CH-arom), 2965 (CH-aliph), 1720 (C=O), 1631 (C=N); <sup>1</sup>H NMR δ/ppm = 0.99 (3H, t, CH<sub>3</sub>CH<sub>2</sub>), 2.02 (2H, p, CH<sub>3</sub>CH<sub>2</sub>), 4.33 (1H, t, CH.thiazole), 7.66 (1H, t, *J*=5.6 Hz, H-Arom), 7.69 (1H, d, *J*=5.0 Hz, H-Arom), 7.73 (1H, t, *J*=8.0 Hz, H-Arom), 7.89 (1H, s, H-Arom), 7.94 (1H, d, *J*=4.7 Hz, H-Arom), 8.05 (1H, d, *J*=3.9 Hz, H-Arom), 8.12 (1H, dd, *J*=9.2, 5.9 Hz, H-Arom), 12.64 (1H, s, NH disappeared by D<sub>2</sub>O); <sup>13</sup>C NMR δ/ppm 10.42 (CH<sub>3</sub>), 25.44 (CH<sub>2</sub>), 49.88 (CH.thiazole), 112.97, 113.18, 119.30, 119.55, 121.98, 122.19, 129.90, 131.85, 132.45, 135.30, 138.66, 139.25, 141.83, 146.34 (Cs-Arom), 154.80 (C=N), 161.63 (C-F), 164.12 (C=O); Anal. Calcd. for **C<sub>20</sub>H<sub>14</sub>FN<sub>5</sub>OS** (391.42): C, 61.37; H, 3.61; N, 17.89; Found: C, 61.65; H, 3.21; N, 17.15%.

### 3.1.17. Ethyl -2-(2-(2-(7-fluoro-11H-indeno[1,2-*b*]quinoxalin-11-ylidene)hydrazineyl)-4-oxo-thiazol-5(4H)-ylidene)acetate (17)

Yield: 79 % as deep-orange crystal from AcOH; mp. 199–197 °C; IR: ν/cm<sup>-1</sup> = 3263 (NH), 3054 (CH-Arom), 2987 (CH-Aliph), 1727, 1700 (2C=O), 1638 (C=N); <sup>1</sup>H NMR δ/ppm = 1.27 (3H, t, CH<sub>3</sub>CH<sub>2</sub>O-), 4.24 (2H, q, CH<sub>3</sub>CH<sub>2</sub>O-), 6.64 (1H, s, vinylic-CH), 7.65 (1H, d, *J*=7.5 Hz, H-Arom), 7.68 (1H, t, *J*=9.2 Hz, H-Arom), 7.73 (1H, t, *J*=8.0 Hz, H-Arom), 7.90 (1H, d, *J*=6.8 Hz, H-Arom), 7.96 (1H, s, H-arom), 7.98 (1H, s, NH disappeared by D<sub>2</sub>O), 8.15 (1H, dd, *J*=9.2, 5.9 Hz, H-arom); <sup>13</sup>C NMR δ/ppm 14.57 (CH<sub>3</sub>CH<sub>2</sub>O-), 61.55 (CH<sub>3</sub>CH<sub>2</sub>O-), 96.97 (vinylic-CH), 113.65, 119.69, 120.76, 122.53, 130.04, 132.77, 134.65, 135.23, 136.79, 138.30, 139.50, 143.07, 146.70, 155.28 (Cs-Arom), 158.28 (C=N), 161.62 (S-C=N), 165.01 (C-F), 166.24, 166.38 (2C=O); Anal. Calcd. for **C<sub>22</sub>H<sub>14</sub>FN<sub>5</sub>O<sub>3</sub>S** (447.44): C, 59.06; H, 3.15; N, 15.65; Found: C, 59.24; H, 3.03; N, 15.55%.

## 3.2. Biology

### 3.2.1. Animals and experimental design:

96 Male Swiss albino mice weighing 25 – 30 g were purchased from the National Scientific Research Center (Giza, Egypt). Mice were maintained at 20–25 °C in a 12 h light/dark cycle, with a commercial normal rodent diet and water freely available. They were divided into 16 groups, 6 mice each. First, the normal control group (received citrate buffer, ip for 2 weeks). Second; the diabetic control group (STZ-induced diabetic mice & received normal saline, ip, for 2 weeks).

3<sup>rd</sup>: diabetic mice treated with gliclazide as a reference standard for 2 weeks. Other diabetic mice (groups 4-16) received new synthesized compounds; dissolved in tween 80 (25 mg/kg, oral, for 2 weeks).

### **3.2.2. Diabetes induction:**

Mice (groups 2-16) were fasted overnight for 12 hours, then intraperitoneally injected with 50 mg/kg streptozotocin (STZ) which was freshly dissolved in citrate buffer solution {pH 4.5}, 15 minutes after IP injection of nicotinamide (NA) 110 mg/kg. NA is used to partially protect against the damaging effects of STZ on  $\beta$  cells. This model is used to develop a model of type II DM. After 3 days blood glucose was checked by one-touch Glucometer (using blood from the tail). Mice with blood glucose level  $\geq 250$  mg/dL were selected for diabetic groups <sup>34</sup>. At the end mice were fasted overnight, then blood samples were collected on EDTA tubes. Plasma was isolated by centrifugation at 10,000 rpm/min, for 10 min.

### **3.2.3. Chemicals**

STZ, NA, and ELISA kits were purchased from Sigma Aldrich Co (St Louis, MO, USA). Spectrophotometric kits were purchased from Biodiagnostic Co, Egypt.

### **3.2.4. Biochemical analysis**

#### **3.2.4.1. Measurements of blood glucose and plasma insulin levels**

Fasting blood glucose was assessed spectrophotometrically using blood from the lateral tail vein on the last day of the experiment; by a commercial kit <sup>35</sup>. Glucose oxidase catalyzes the oxidation of glucose to hydrogen peroxide and gluconic acid. Then hydrogen peroxide liberated reacts with 4-amino antipyrine (4-AAP) and a phenol derivative producing red colored compound. The color intensity is directly proportional to glucose concentration in the sample.

**The provided reagents:** 1-R<sub>1</sub> solution: standard glucose solution. 2-R<sub>2</sub> solution: phosphate buffer, glucose oxidase, peroxidase, 4-AAP and phenol.

**Procedure:** R<sub>2</sub> solution (1 ml) was added to sample or standard (R<sub>1</sub>) (10  $\mu$ l) and incubated 10 minutes at 37°C for. Then the absorbance of samples and standard were measured colorimetry at 546 nm.

Plasma insulin for the most efficient compounds in lowering blood glucose level (**4,6,8,16**) was assessed using the ELISA technique. Standards and samples (5  $\mu$ l) were put in the wells, and incubated with monoclonal antibody conjugated to horseradish peroxidase enzyme (100  $\mu$ l). Then plate was sealed, incubated for 90 minutes with shaking at 600 rpm at room temperature. After wash, TMB substrate (3,3',5,5'-Tetramethylbenzidine) (100  $\mu$ l) was added and incubated at room



temperature for 15 minutes. Then color developed in proportion to the amount of insulin bound initially. The reaction was stopped by adding Stop solution (100  $\mu$ l). Absorbance of was measured at 450 nm.

#### 3.2.4.2. Measurements of kidney functions

Plasma urea was determined according to Orsonneau et al. (1992)<sup>36</sup> using a commercial reagent kit. The assay depends on the fact that, urea in the plasma is hydrolyzed by urease into ammonium and carbon dioxide. The ammonia liberated reacts with salicylate and sodium hypochlorite forming indophenol; a green colored compound. **The Provided reagents:** 1-R<sub>1</sub>: standard solution. 2-R<sub>2</sub>: urease. 3-R<sub>3</sub>: phosphate buffer, EDTA, sodium salicylate and sodium nitroprusside (catalyst). 4-R<sub>4</sub>: sodium hypochlorite and sodium hydroxide. **Procedure:** One drop of R<sub>2</sub> solution was added to 1 ml of R<sub>3</sub> solution followed by the addition of 10  $\mu$ l of sample or standard solution (R<sub>1</sub>). The contents of tubes were mixed and incubated for 3 minutes at 37°C, and then 200  $\mu$ l of R<sub>4</sub> solution were added, mixed and incubated for 5 minutes at 37°C. The absorbance was measured colorimetry at 578 nm.

Plasma creatinine was measured using a colorimetric kit according to Saibaba et al. (1997)<sup>37</sup>. Creatinine in the sample is reacted with sodium picrate in alkaline medium forming a red complex. **Provided reagents in the kit:** 1-R<sub>1</sub>: standard solution. 2-R<sub>2</sub>: picric acid. R<sub>3</sub> (alkaline reagent): sodium hydroxide.

Equal volumes of R<sub>2</sub> and R<sub>3</sub> are mixed to prepare working reagent. **Procedure:** 1 ml of R<sub>2</sub> was added to 100  $\mu$ l of sample or standard. The absorbance of the reagent blank, samples or standard was measured at 492 nm after 30 and 120 seconds and the change in absorbance was calculated.

#### 3.2.4.3. Measurements of oxidative stress biomarkers

Plasma GSH content was assessed colorimetry according to Beutler et al.<sup>38</sup>. Where 0.5 ml of sample and 1 ml of buffer were added followed by 100  $\mu$ l of Ellman's reagent [5,5'-dithio-bis (2-nitrobenzoic acid)]. The absorbance of the formed yellow color was measured within 5 minutes at 405 nm.

Plasma MDA was assessed colorimetry. 1 ml of chromogen solution provided from a commercial kit was added to 200  $\mu$ l of sample or standard MDA solution. Tubes were mixed; covered; heated in a boiling water bath for 30 minutes; then cooled. The absorbance of samples was measured at 534 nm<sup>39</sup>.

#### **3.2.4.4. Liver functions markers & cholesterol**

Plasma AST & ALT were measured colorimetry at 510 nm according to Piyachaturawat *et al.*, 1988<sup>40</sup> using a commercial kits. Cholesterol was measured colorimetry using a commercial kit according to Cox and García-Palmieri, 1990<sup>41</sup>. Where procaine hydrochloride reacted with sodium nitrite in acid medium to form diazonium salt; which is coupled with cholesterol in alkaline medium to form soluble colored product. Absorption was measured at 428nm.

#### **3.2.5. Statistical analysis**

The data were expressed as means  $\pm$  S.E. Results were calculated by one-way ANOVA, followed by Tuckey's test as a post hoc test.  $P < 0.05$  was considered significant. All the analyses were carried out by using SPSS software (version 22.0) for Windows 8.1 (SPSS, Inc., Chicago, IL, USA).

#### **3.2.6. *In vitro* assay of $\alpha$ -glucosidase inhibitory activity**

The assay was carried out following the  $\alpha$ -Glucosidase Inhibitor Screening Colorimetric Kit protocol supplied by BioVision Company, USA

##### **3.2.6.1. Assay Protocol**

###### **1. Screening Compounds, Inhibitor Control & Background Control preparations: Samples [S] and Inhibitor Control [IC]:**

Dissolve test samples to 100X in a proper solvent. Further, dilute to 10X using  $\alpha$ -Glucosidase Assay Buffer. Add 10  $\mu$ L of Diluted test compound, 10  $\mu$ L of acarbose into wells of 96-well clear plate designated as test samples [S] or Inhibitor Control [IC], respectively.

###### **Enzyme Control [EC] and Background Control [BC]:**

Add 10 and 20  $\mu$ L of  $\alpha$ -Glucosidase Assay Buffer into the designated well(s) of 96-well clear plate, respectively.

**IC<sub>50</sub> estimation (Optional):** Prepare several dilutions of candidate(s) in  $\alpha$ -Glucosidase Assay Buffer. Add 10  $\mu$ L of each dilution into designated wells.

###### **2. $\alpha$ -Glucosidase Enzyme Solution Preparation:**

Prepare a 20-fold dilution of  $\alpha$ -glucosidase (i.e. Dilute 2  $\mu$ L of  $\alpha$ -glucosidase with 38  $\mu$ L of  $\alpha$ -Glucosidase Assay Buffer), mix thoroughly, and keep on ice. Add 10  $\mu$ L of Diluted  $\alpha$ -Glucosidase Enzyme Solution to each well-containing Test Sample(s) [S], Inhibitor Control [IC], Enzyme Control [EC], and Solvent Control [SC]. Adjust the volume of each well to 80  $\mu$ L/well with  $\alpha$ -Glucosidase Assay Buffer. Mix well and incubate at room temperature for 15-20 min. Protect from light.

**3. Reaction Mix Preparation:** Mix enough reagents for the number of assays to be performed. For each well, prepare 20  $\mu$ L Reaction Mix containing:

#### **Reaction Mix**

**$\alpha$ -Glucosidase Assay Buffer** 17  $\mu$ L

**$\alpha$ -Glucosidase Substrate Mix** 3  $\mu$ L

Mix & add 20  $\mu$ L Reaction Mix to test sample(s) [S], Inhibitor Control [IC], Enzyme Control [EC], Solvent Control [SC] and Background.

Control [BC] wells and mix well.

**4. Measurement:** Measure absorbance immediately at OD: 410 nm in kinetic mode for 60 min at room temperature. Choose two-time points (t1 & t2) in the linear range of the plot and obtain the corresponding values for the absorbance (OD1 and OD2).

**5. Calculation:** Calculate the slope for all test samples [S], Enzyme Control [EC], Solvent Control [SC], and Background Control [BC] by dividing the net  $\Delta$ OD (A2-A1) values with the time  $\Delta$ t (t2-t1). Subtract the Slope of Background Control from [S], [EC] and [SC]. If [SC] slope is significantly different when compared to [EC], use [SC] values to determine the effect of the tested compound.

% Relative Inhibition =  $\{(\text{Slope of [ EC ]} - \text{Slope of [S]}) / \text{Slope of [EC]}\} \times 100$

% Relative Activity =  $\{\text{Slope of [S]} / \text{Slope of [EC]}\} \times 100$

#### **3.2.7. *In vitro* assay of $\alpha$ -amylase inhibitory activity**

The assay was carried out following the  $\alpha$ -Amylase Inhibitor Colorimetric Kit protocol supplied by BioVision Company, USA

The literature description <sup>42</sup> was used to calculate the proportion of enzyme inhibition caused by the synthesized compounds. Typically, a 0.01 M buffer with a pH of 6.9 was used to create a starch solution (Sigma Aldrich, 1% w/v). Then, in separate test tubes, mixes of enzyme and sample were pre-incubated at 37 °C by adding aliquots (300 L) of the solution. Another 30 minutes were spent incubating these combinations. After adding 300 L of a 2 M solution of dinitro salicylic acid substrate (DNSA), the reaction was stopped by immediately submerging each test tube in a pot of boiling water. After reaching room temperature, the test tubes were diluted with 1 mL of distilled water. To adjust the absorbance of colored test samples, a blank was created for each concentration by incubating samples without the enzyme solution. As a control, a test tube with 300 L of dimethyl sulfoxide (DMSO) was utilized. Each final solution's absorbency was measured at 540 nm using a UV-vis spectrophotometer. The IC<sub>50</sub> values were calculated from the gathered data using Microsoft Excel Professional Plus 2019 software. The following formula was used to compute the percentage inhibition of  $\alpha$ -amylase:  $((A_{\text{control}} - A_{\text{sample}}) / A_{\text{control}}) \times 100$ .

### **3.2.8. *In Silico* Studies**

#### **3.2.8.1. *In silico* Pharmacokinetic Studies:**

In the current study, the Swiss ADMET website (<http://www.sib.swiss>) interface was used to screen all the synthesized compounds as well as the standard reference medicines <sup>43</sup>. Drug-likeness and *in silico* ADME features were predicted. ChemDraw 19.0 was used to transform the molecular structures into the SMILES database. Then, to calculate the physicochemical descriptors, lipophilicity, pharmacokinetics qualities, ADME parameters, and medicinal chemistry friendliness, these SMILES were entered as input into the SwissADME website <sup>44-48</sup>.

#### **3.2.8.2. Docking studies**

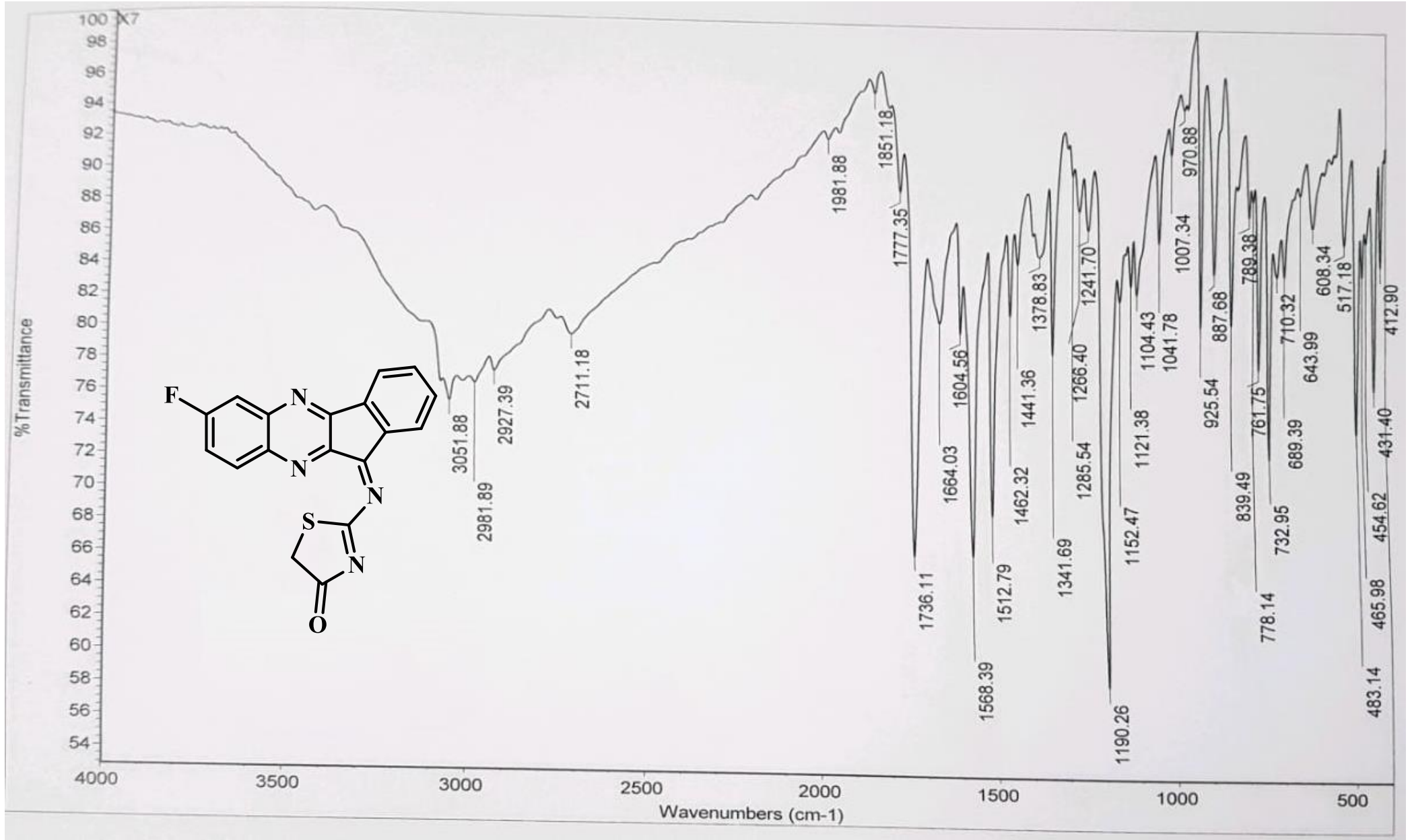
For this docking study, the 3D crystal structures of  $\alpha$ -amylase and  $\alpha$ -glucosidase in association with acarbose (PDB ID: 1B2Y, resolution: 3.2 and PDB ID: 5NN8, resolution: 2.45) <sup>49</sup> were retrieved in.pdb format from the RCSB PDB (<https://www.rcsb.org>). The protein crystal structures were all imported into the Molecular Operating Environment (MOE 2014.09) (Montreal, QC, Canada) <sup>50</sup>. At MMFF94x, the forcefield was configured. We eliminated far-off water molecules, fixed structural issues, added missing hydrogen atoms, and calculated partial charges. Using the default RMS gradient (0.1 kcal/mol/Å), energy reduction was carried out. One defined the binding pocket <sup>49</sup>. ChemDraw Ultra 19.0 was used to sketch newly synthesized compounds, which were subsequently stored in .mol file format and entered a database. By fixing structural issues and inserting missing hydrogens, the database was prepared. Charges in part were computed. Following the selection of the proper tautomeric states for each molecule, energy reduction was performed <sup>51-55</sup>. Triangle Matcher was used as a placement method and London dG as a scoring mechanism for docking. The Rigid Receptor technique was used for refinement.

## List of Abbreviations

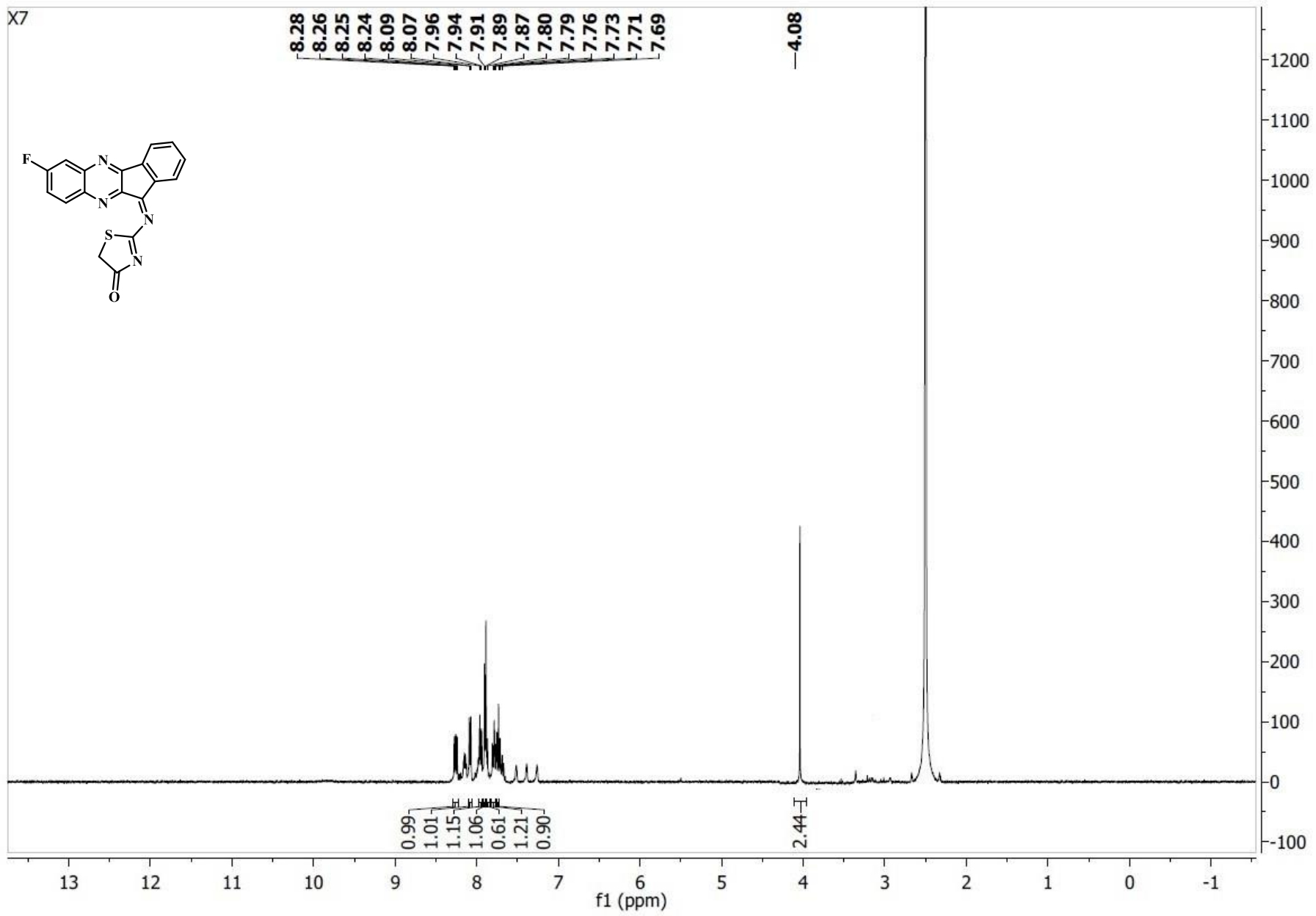
<b>ABS</b>	Absorption
<b>AGIs</b>	$\alpha$ -glucosidase inhibitors
<b>AST</b>	Aspartate Aminotransferase
<b>ALT</b>	Alanine Transaminase
<b>CNS</b>	Central nervous system
<b>DM</b>	Diabetes mellitus
<b>GC/ MS</b>	Gas Chromatograph-Mass Spectrometry
<b>GSH</b>	Glutathione
<b>HBA</b>	Hydrogen bond acceptor
<b>HBD</b>	Hydrogen bond donor
<b>IC<sub>50</sub></b>	The half-maximal inhibitory concentration
<b>IR</b>	Infrared

<b>MR</b>	Molar refractivity
<b>MDA</b>	Malondialdehyde
<b>NC</b>	Normal control
<b>PC</b>	Positive control
<b>P-gp</b>	P-glycoprotein
<b>PDB</b>	Protein data bank
<b>PPM</b>	Part per million
<b>RMSD</b>	Root Mean Square Deviation
<b>SAR</b>	Structure-activity relationship
<b>STZ</b>	Streptozotocin
<b>TLC</b>	Thin layer chromatography
<b>TMS</b>	Tetramethyl silane
<b>TPSA</b>	Topological polar surface area

4: IR

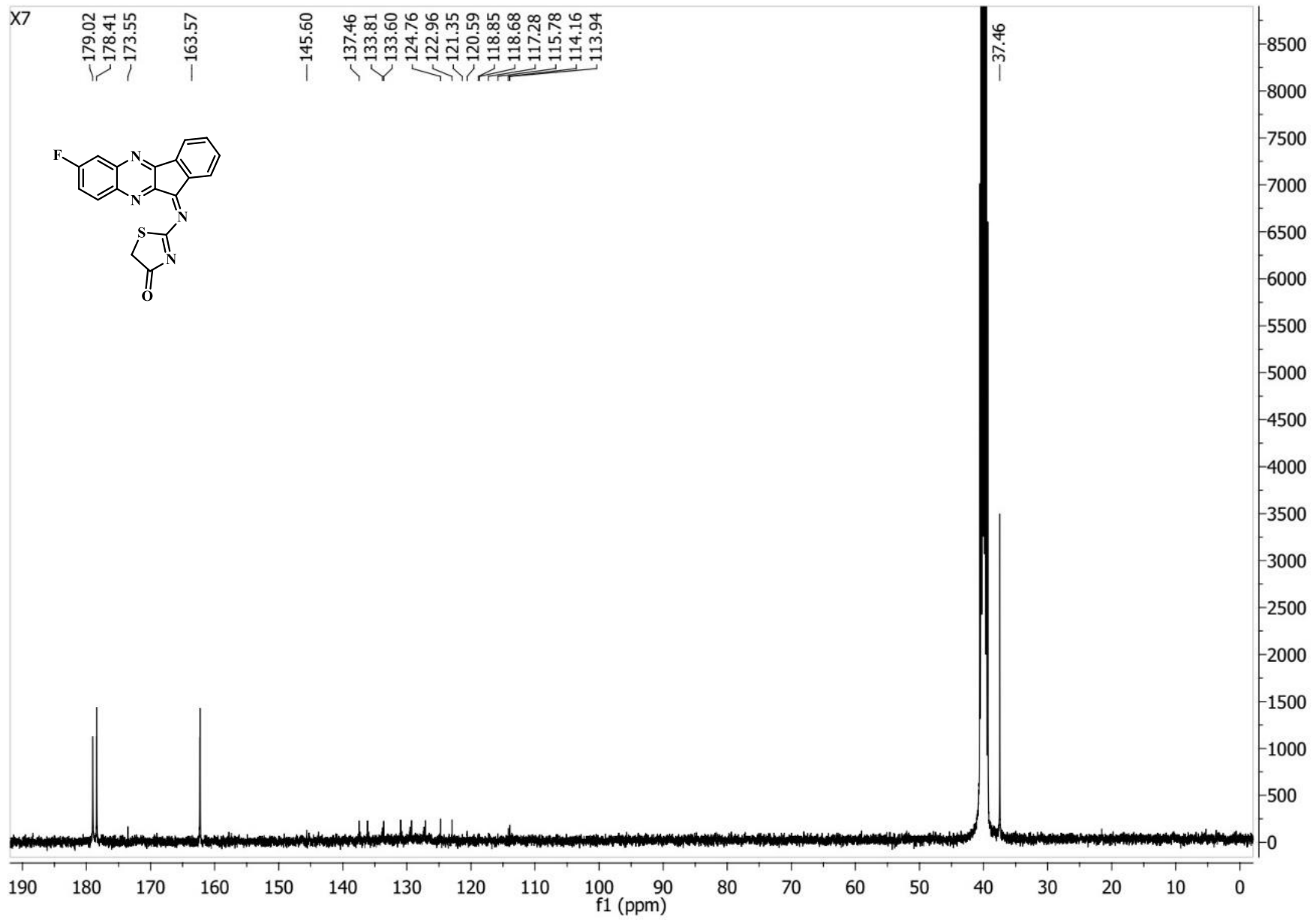


4: <sup>1</sup>H NMR

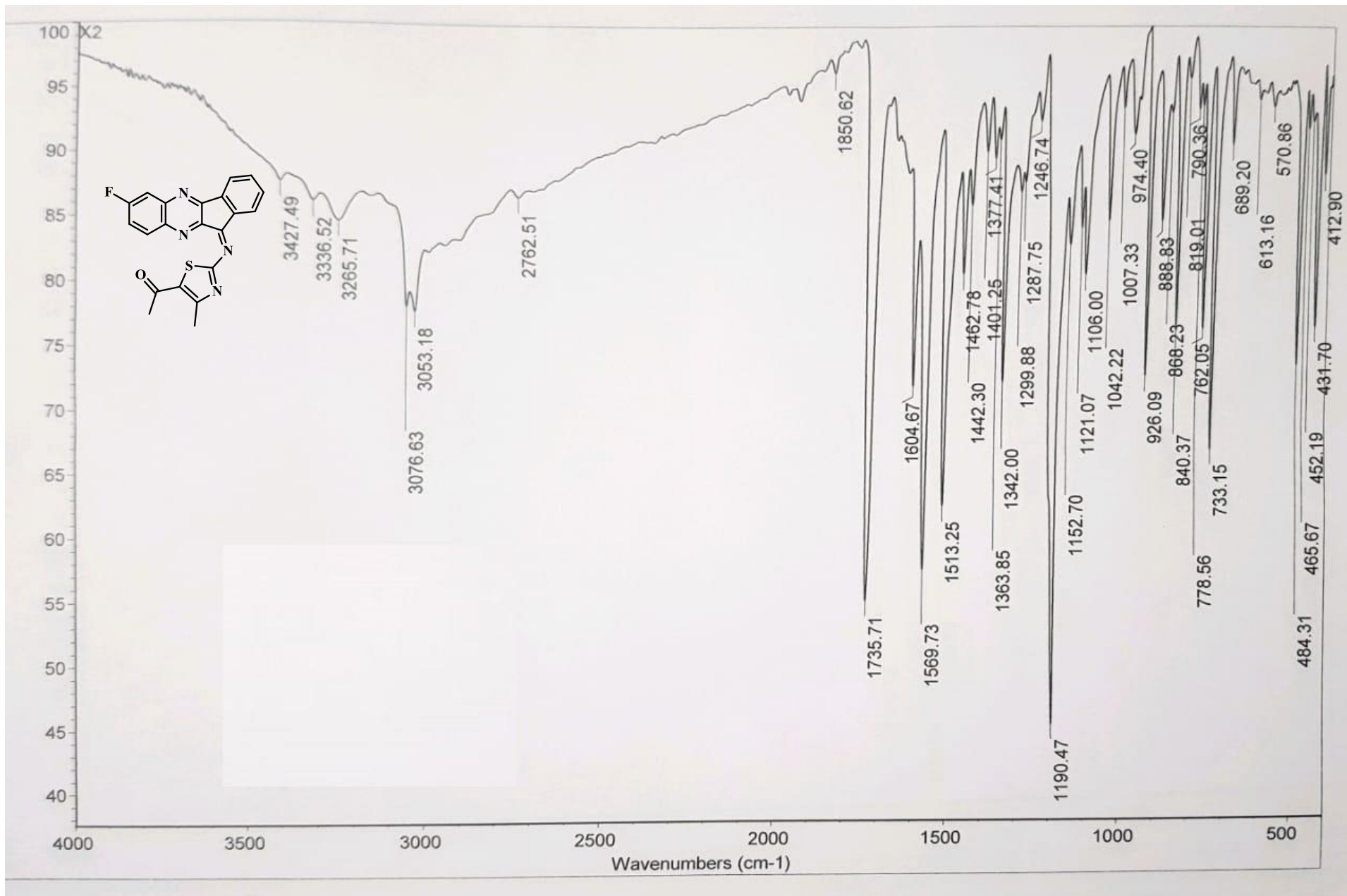


4: <sup>13</sup>C NMR

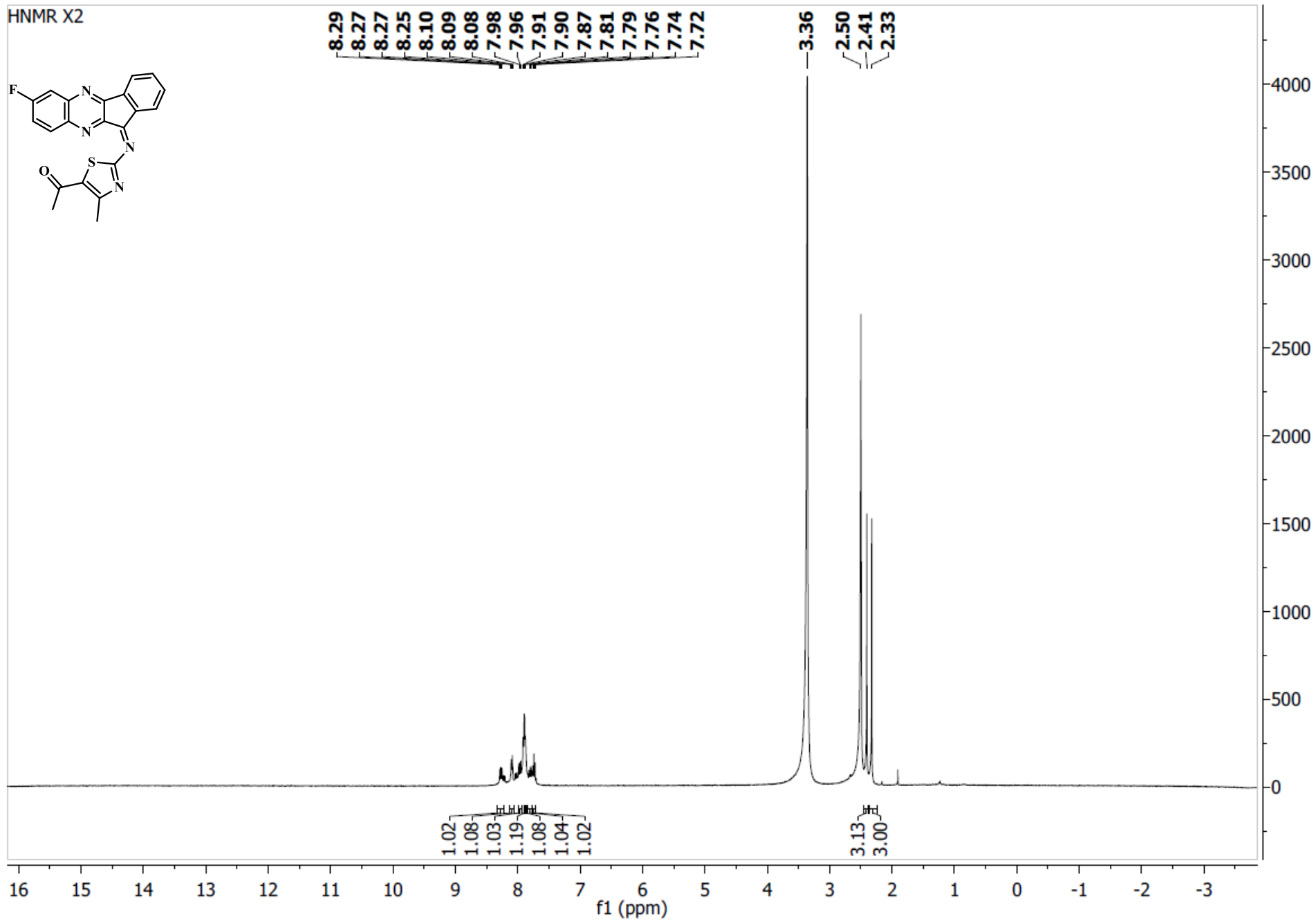




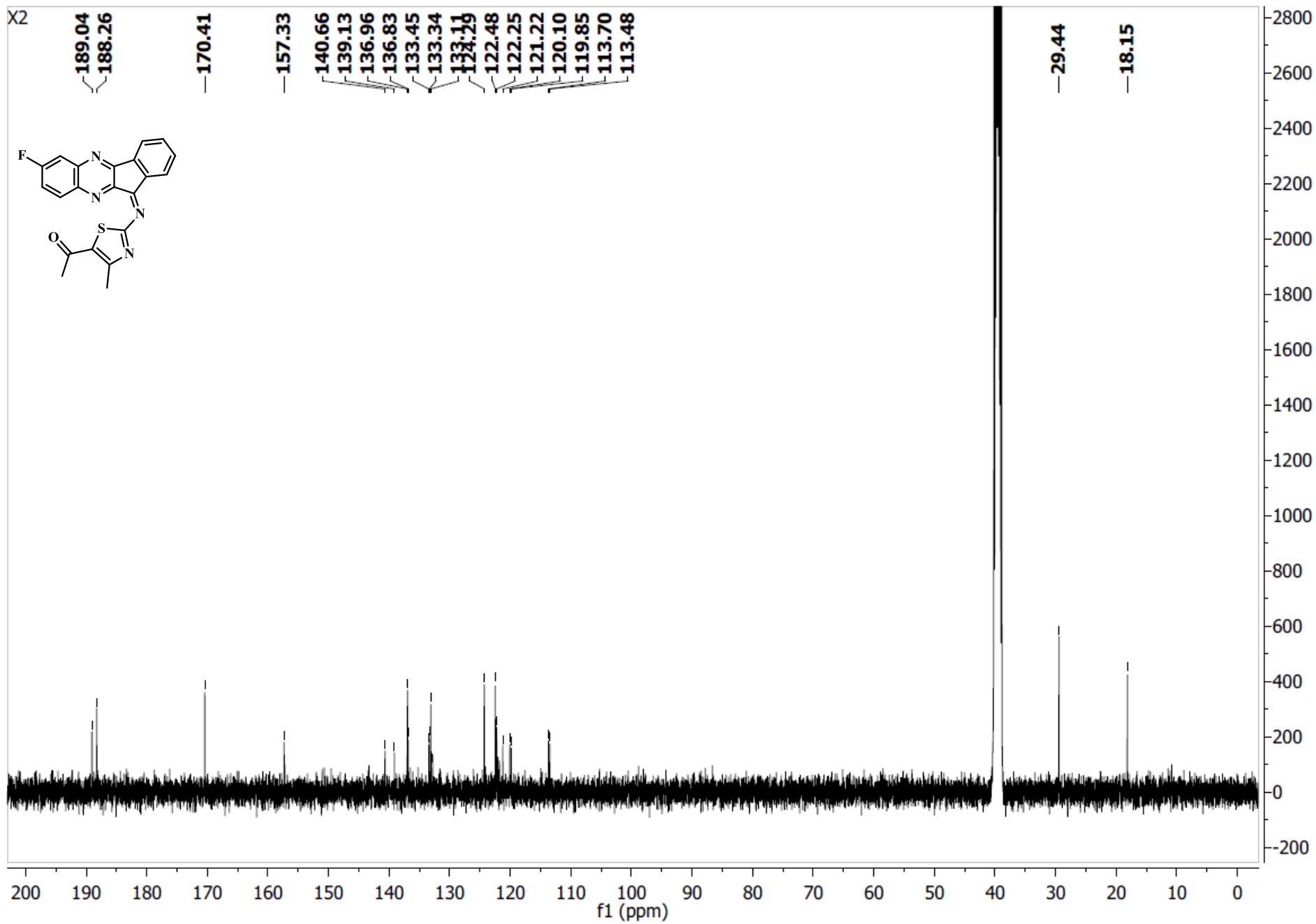
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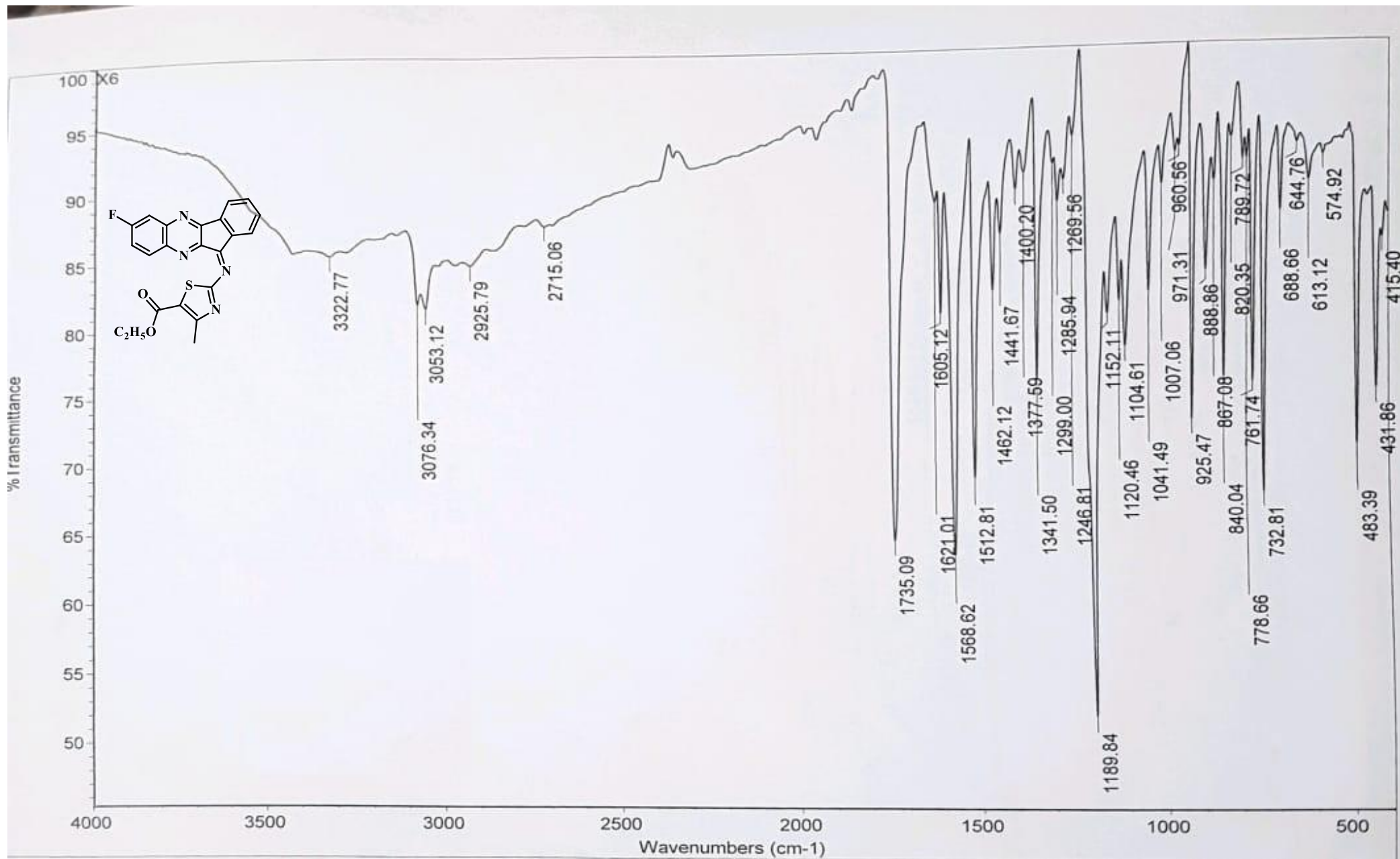
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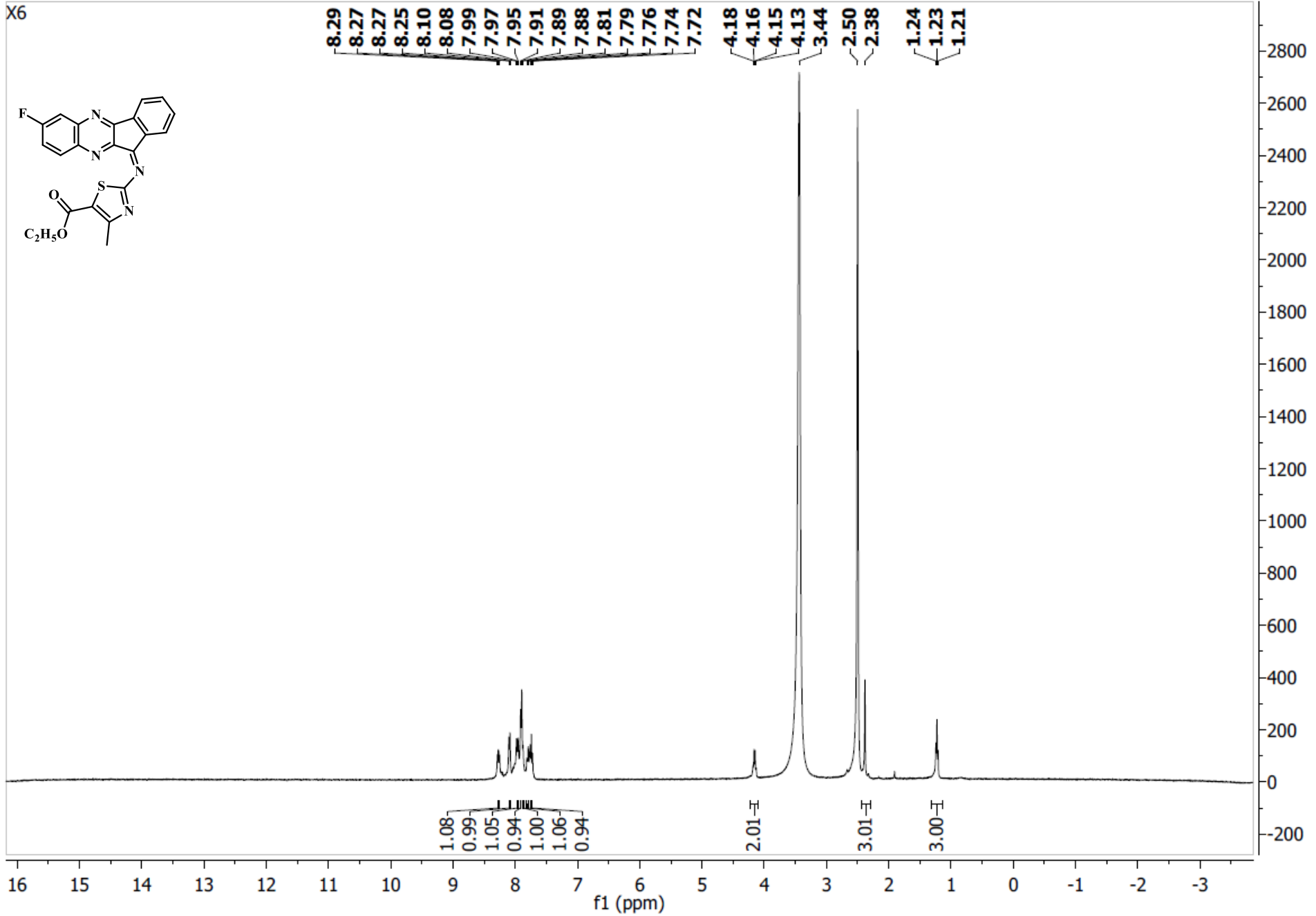
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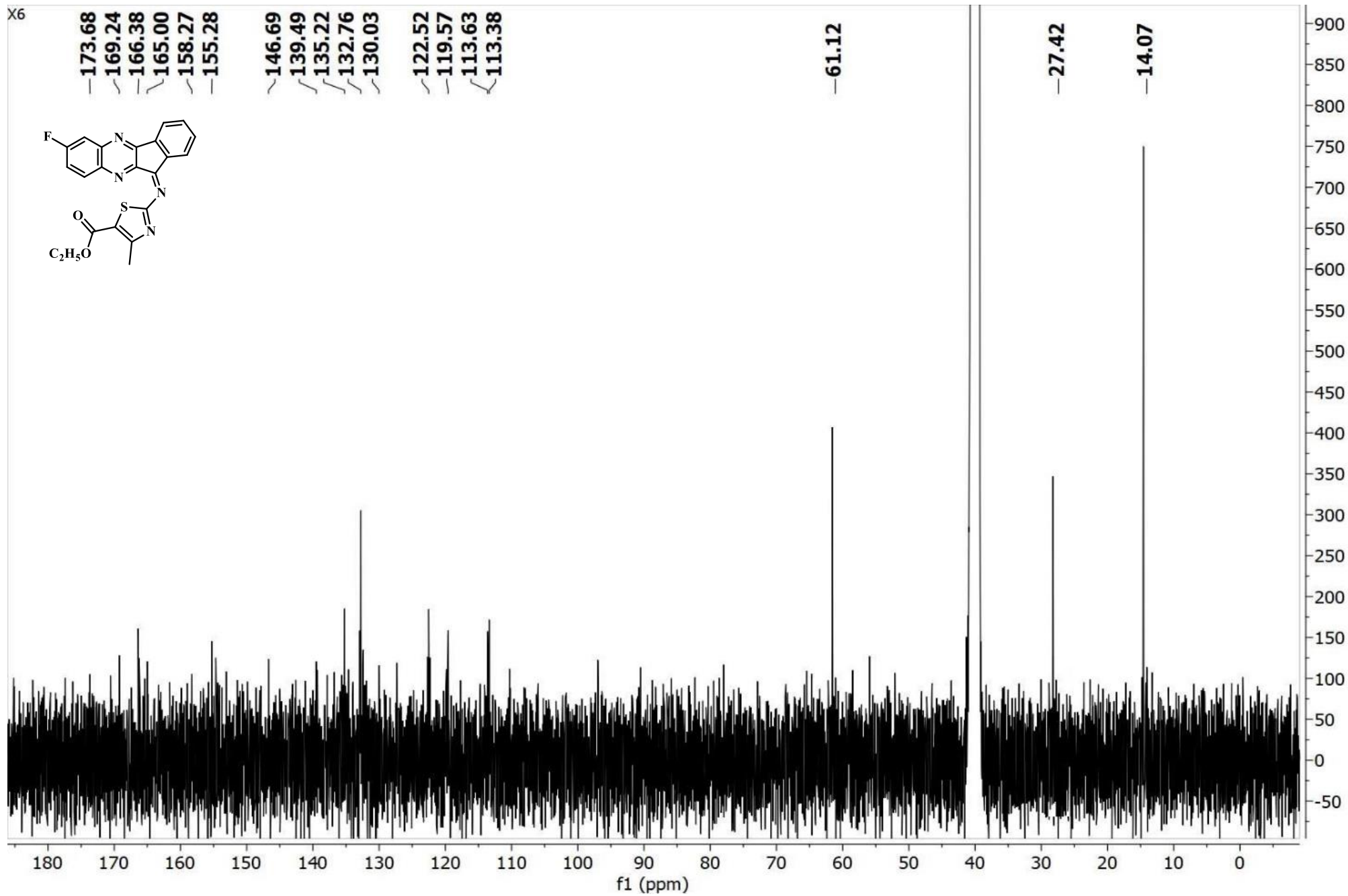
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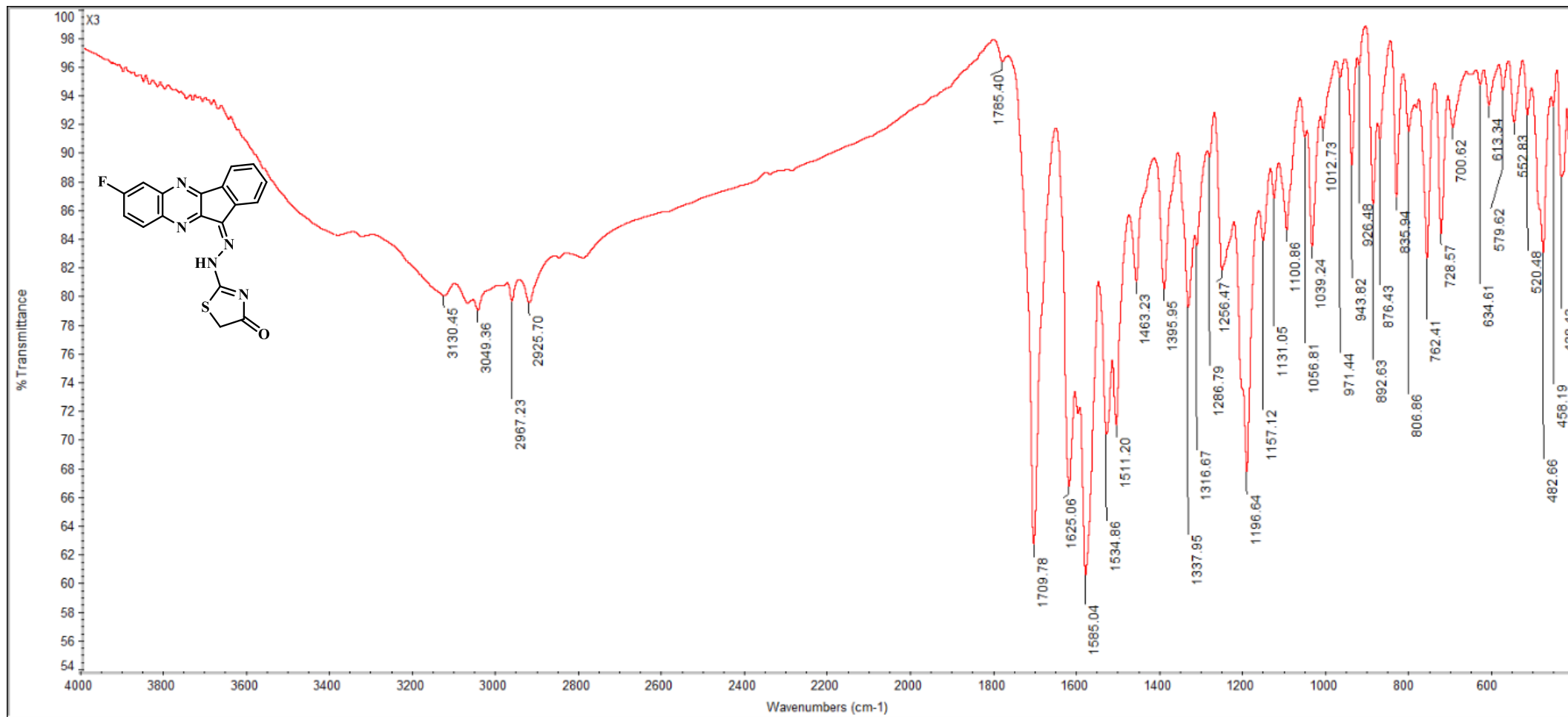
6: <sup>1</sup>H NMR



6:  $^{13}\text{C}$  NMR

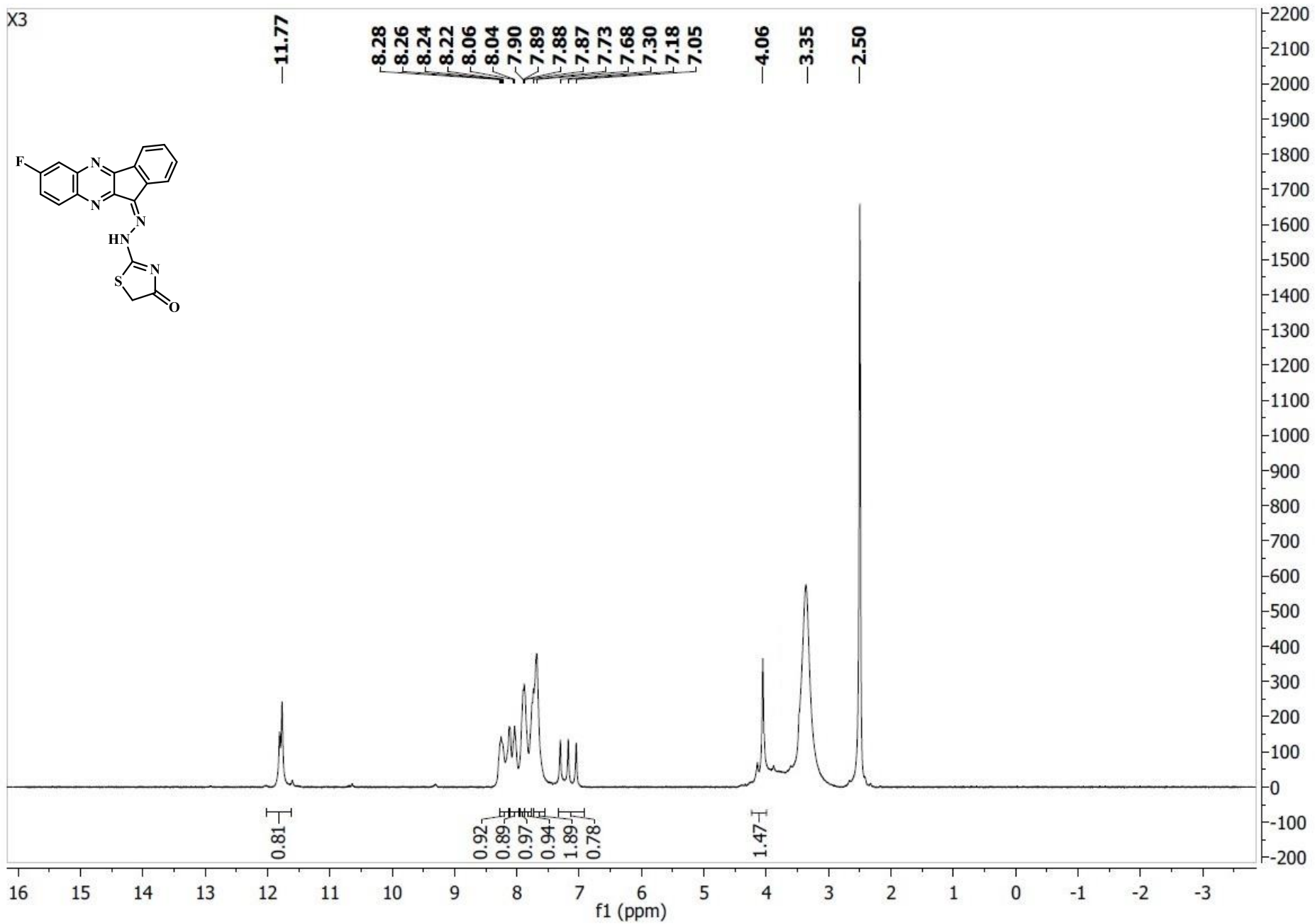


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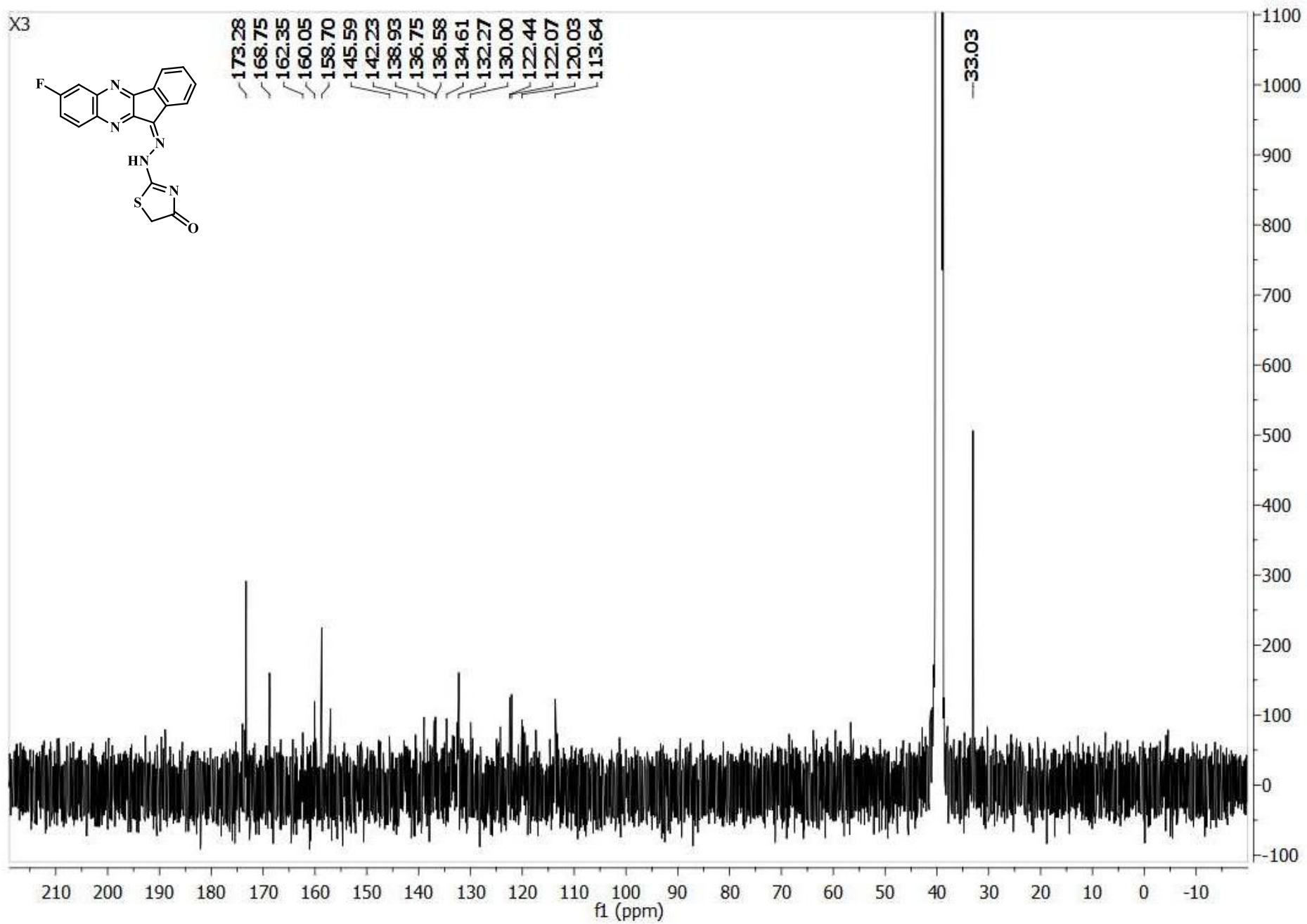


7: <sup>1</sup>H NMR

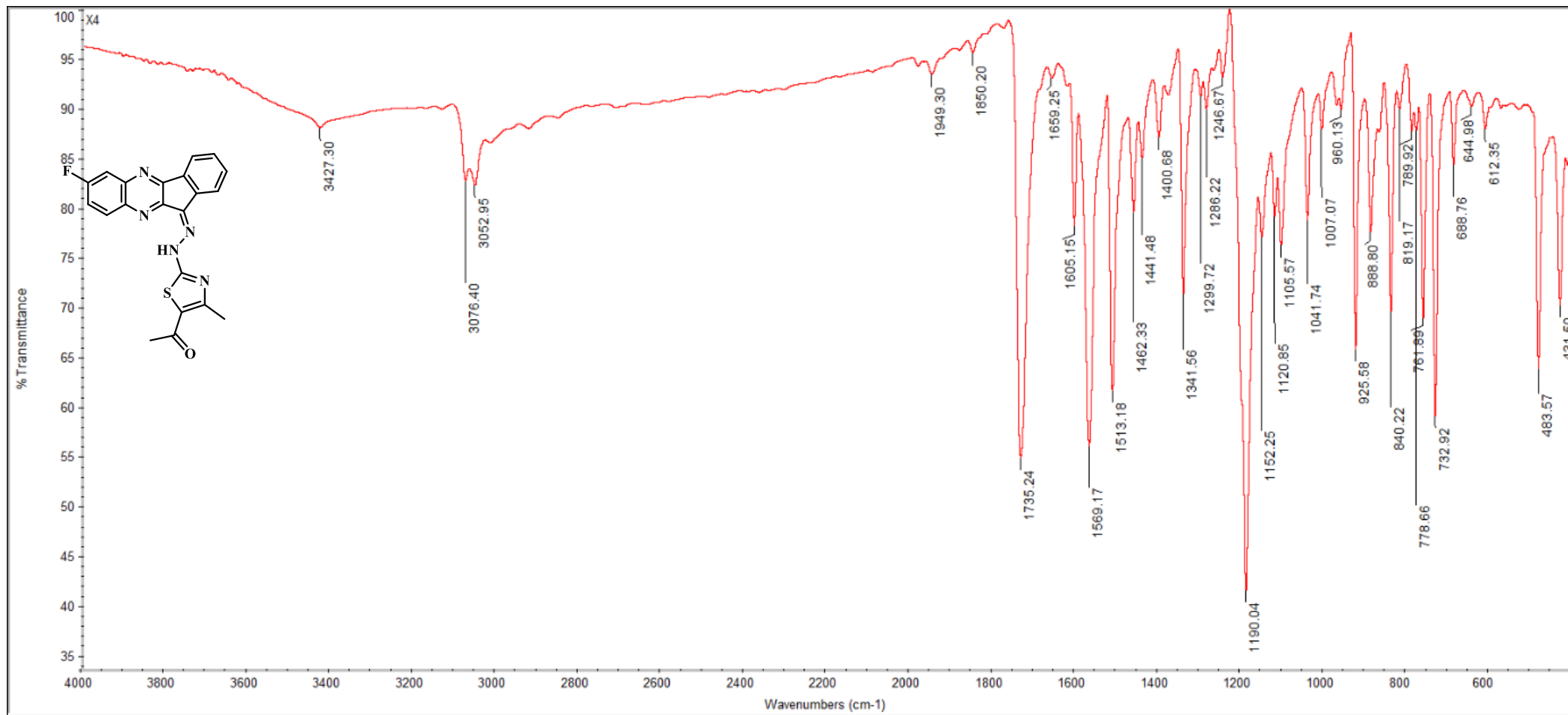




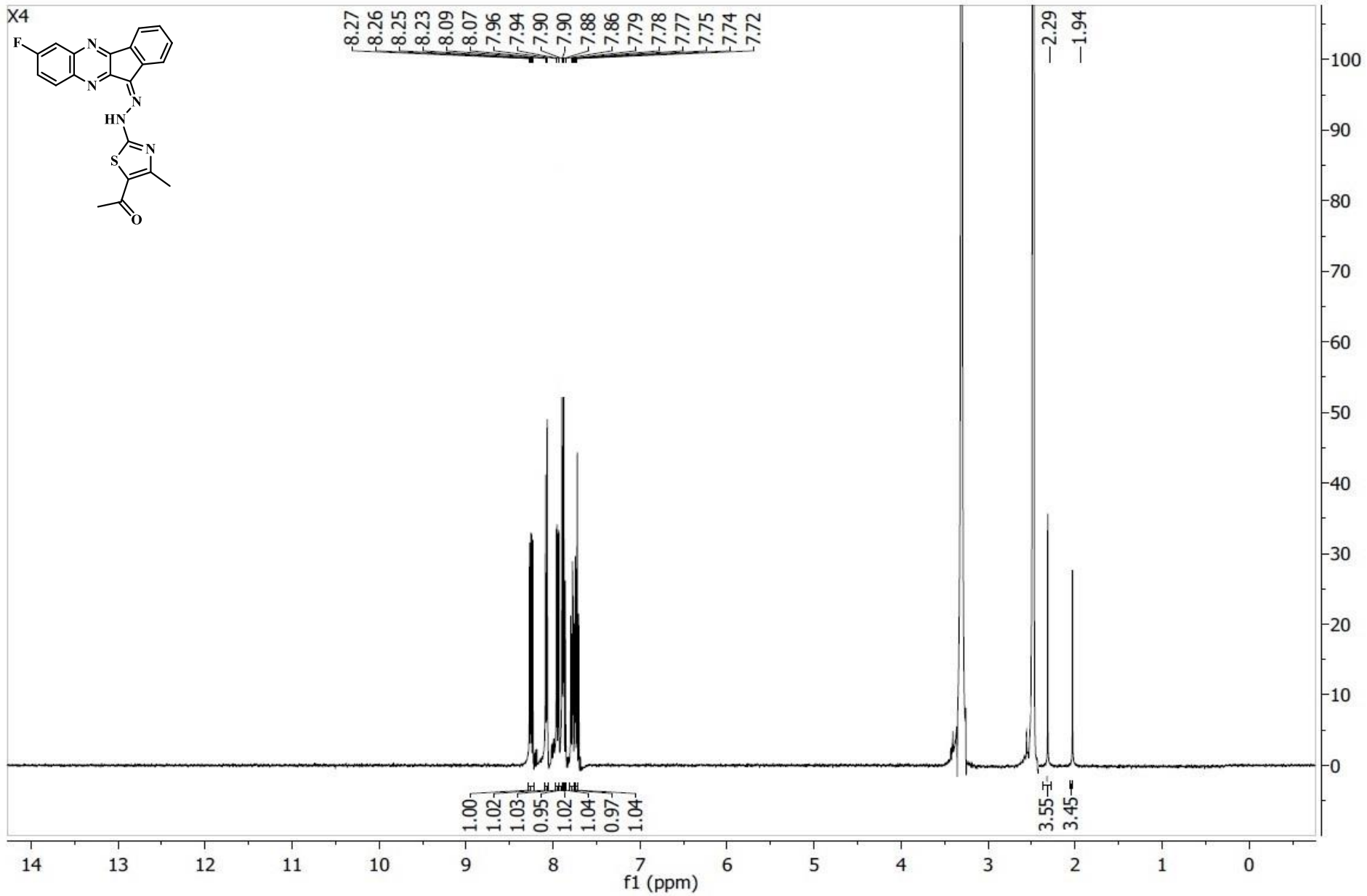
7: <sup>13</sup>C NMR



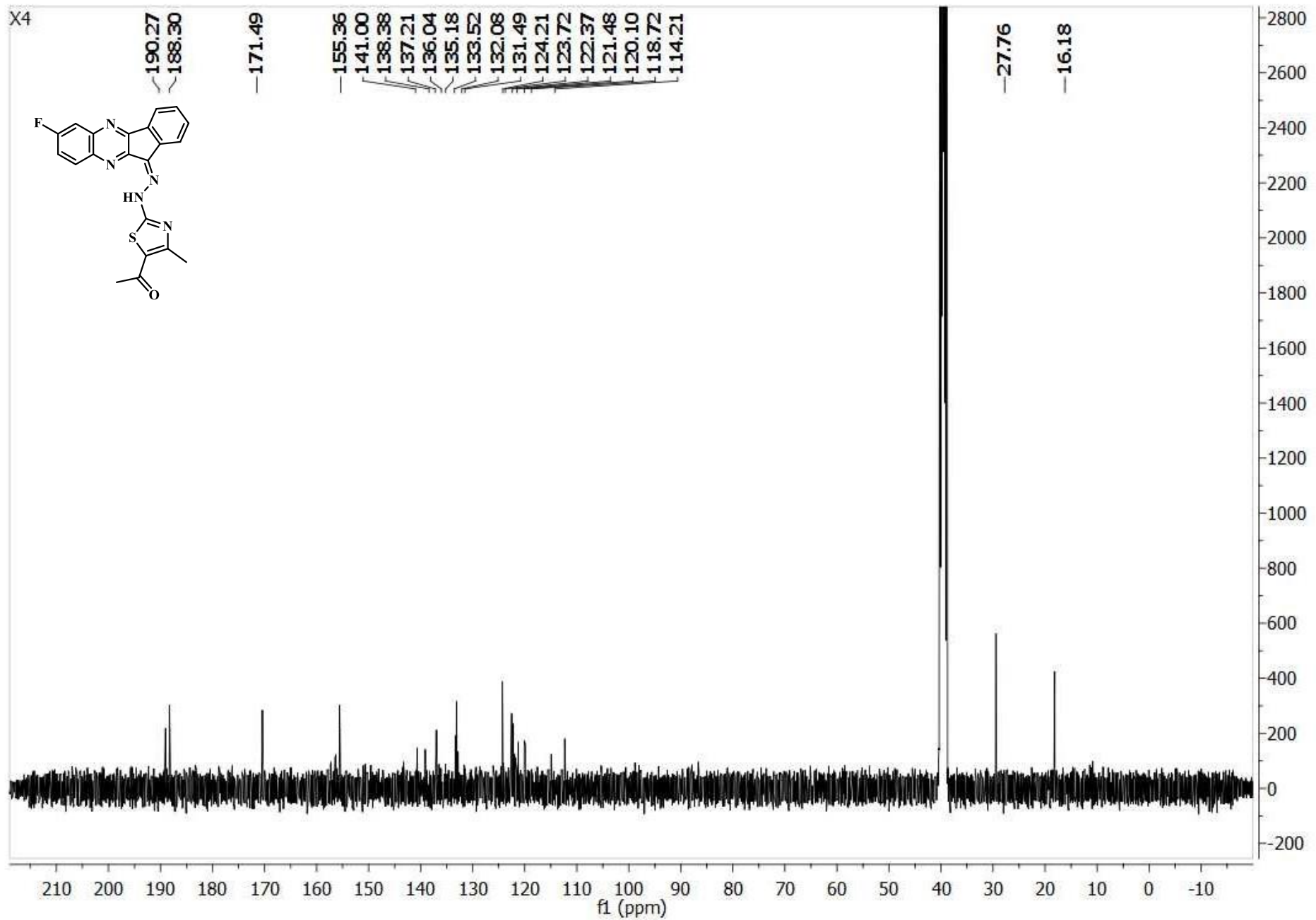
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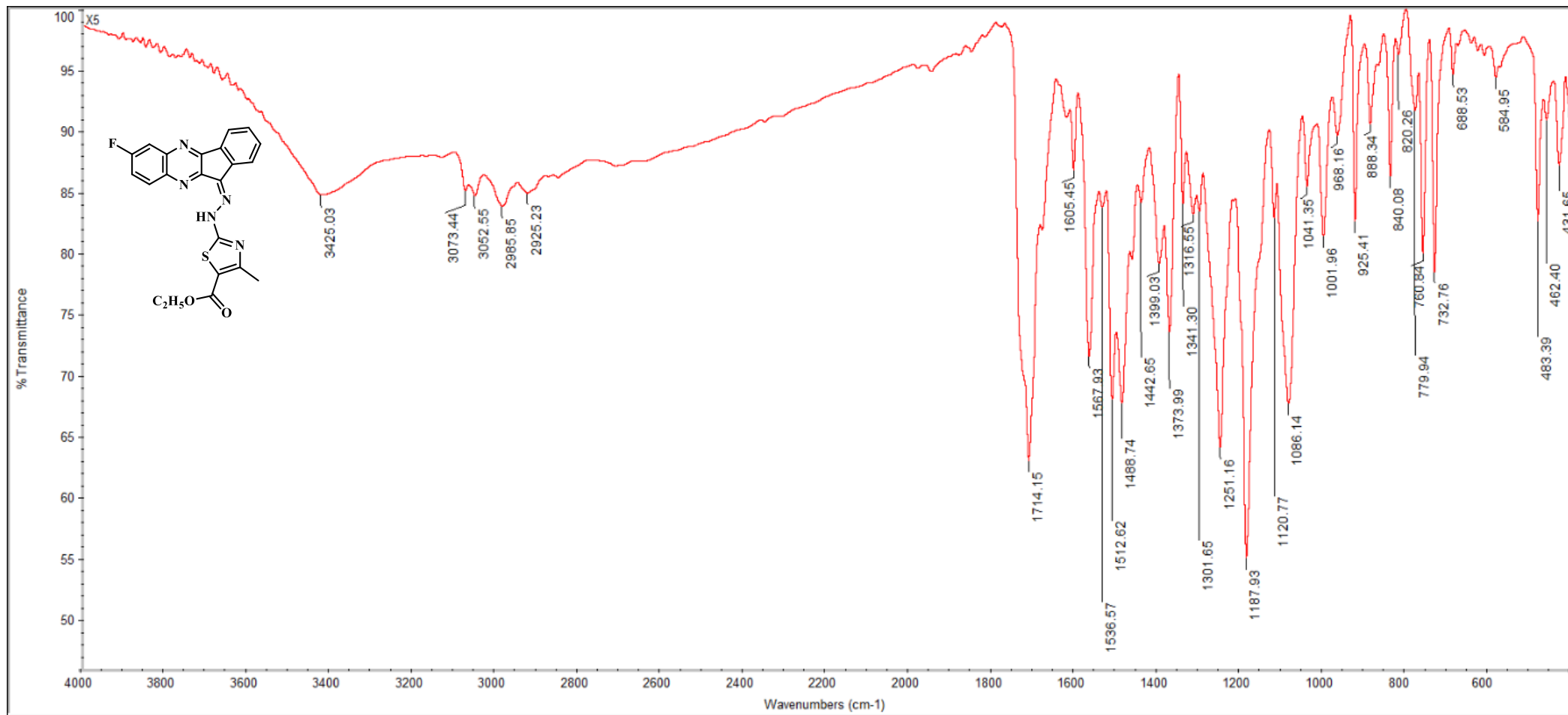
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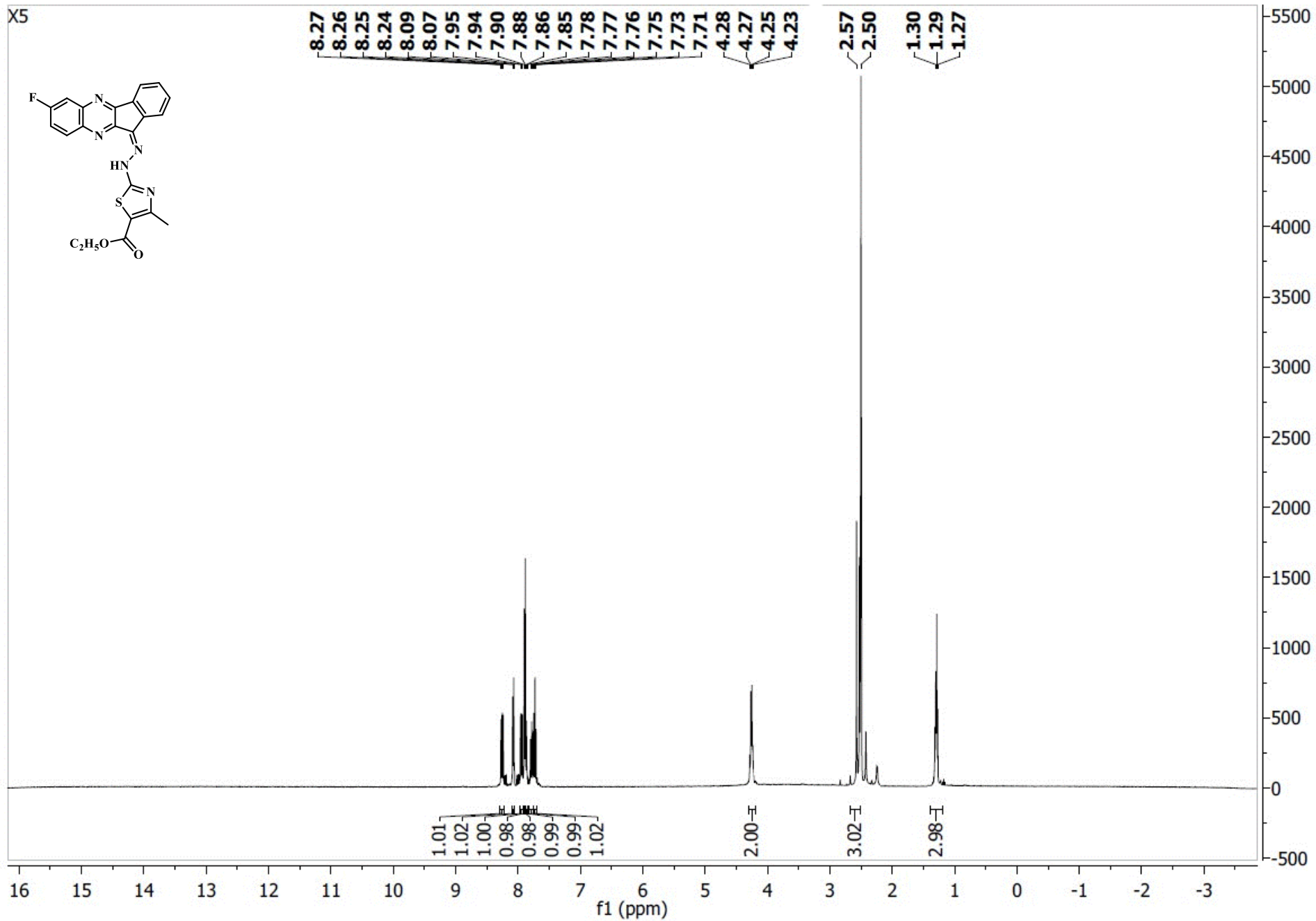
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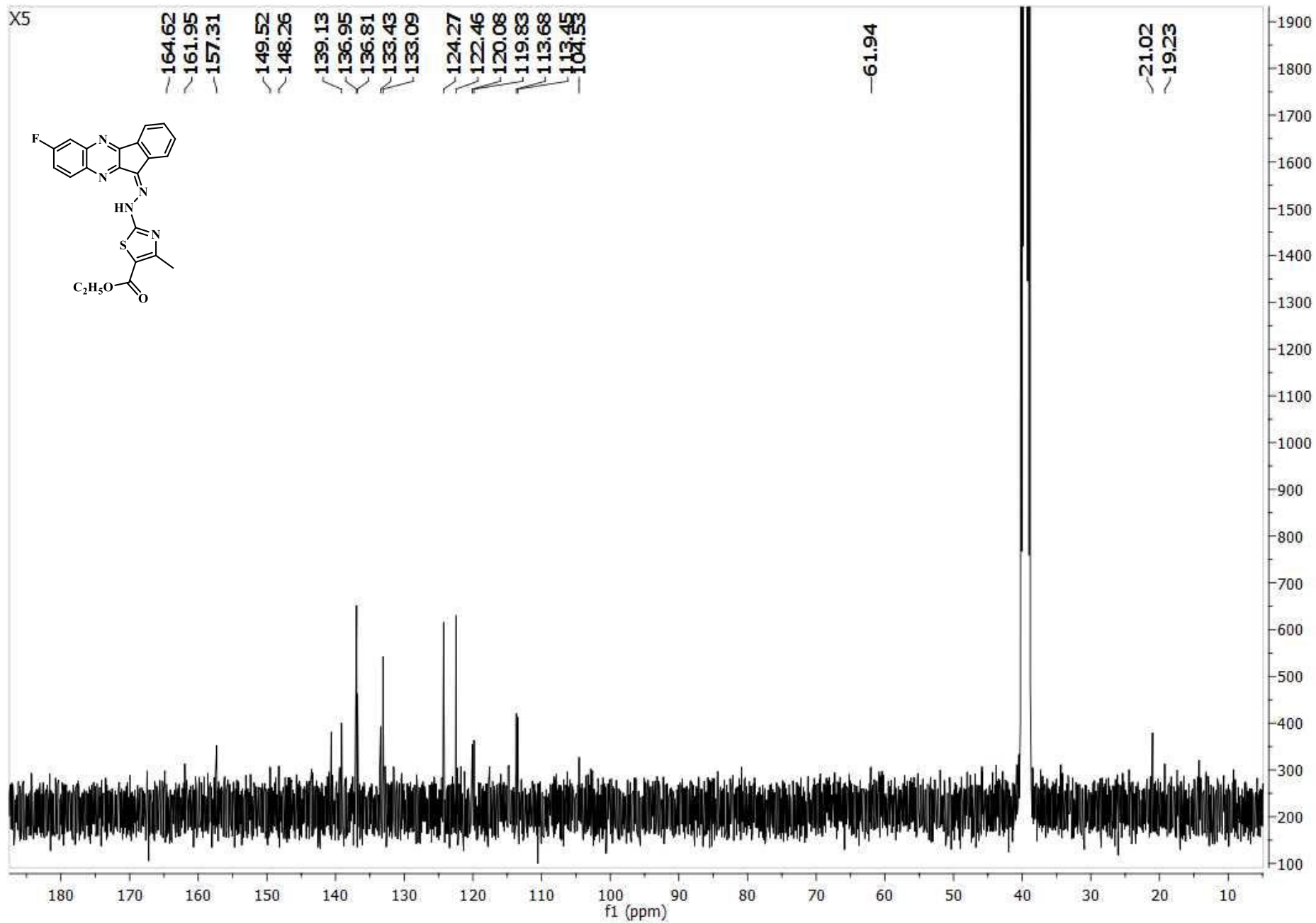
9: IR



9: <sup>1</sup>H NMR

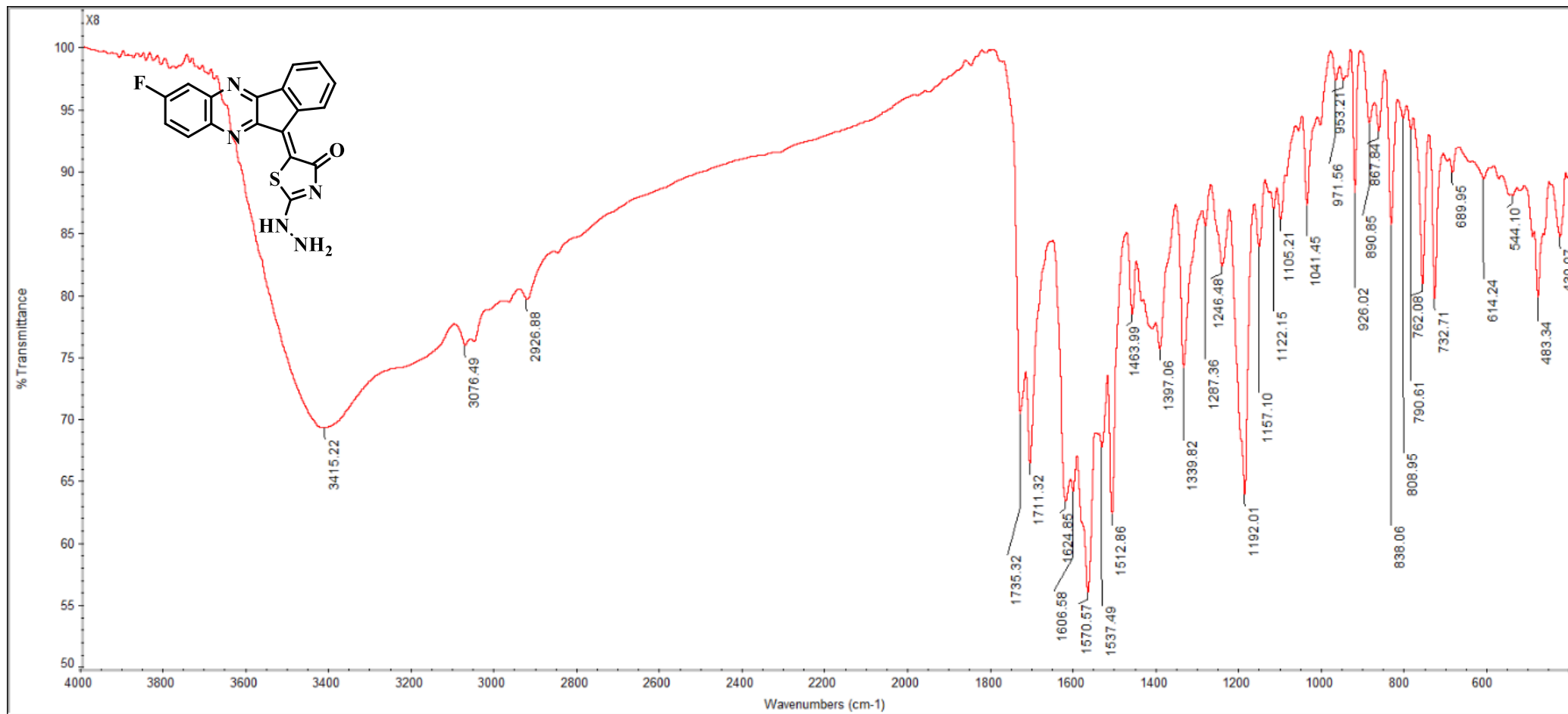


9:  $^{13}\text{C}$  NMR

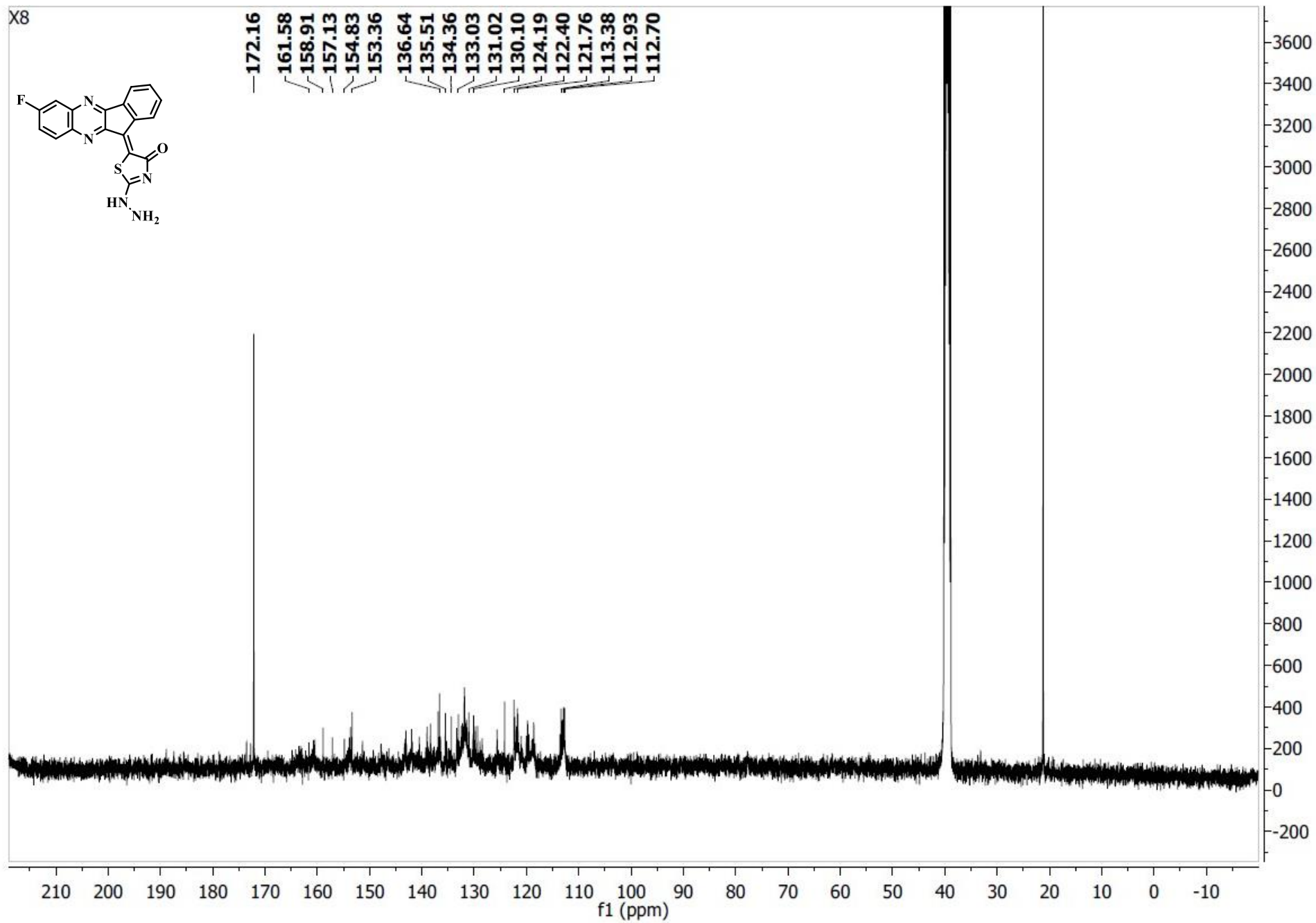


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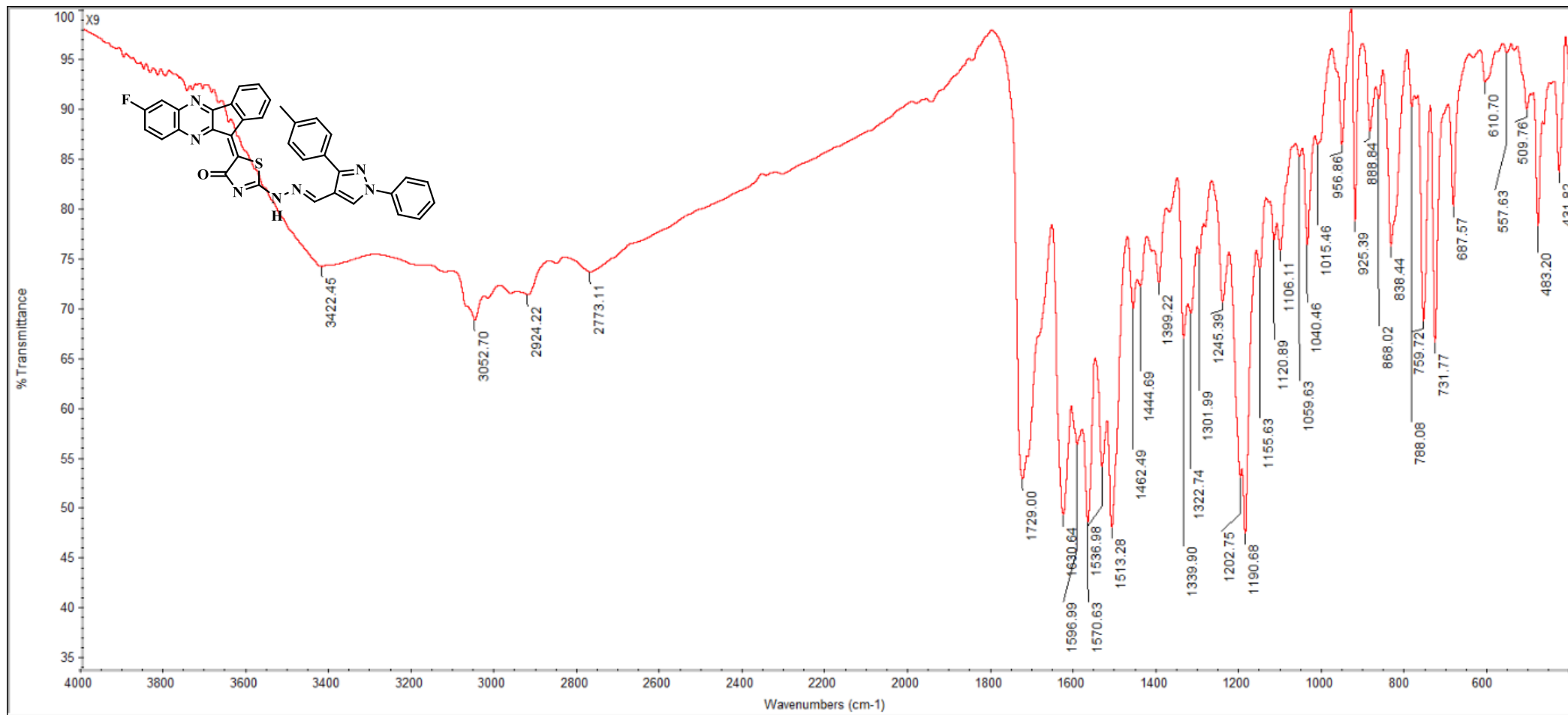




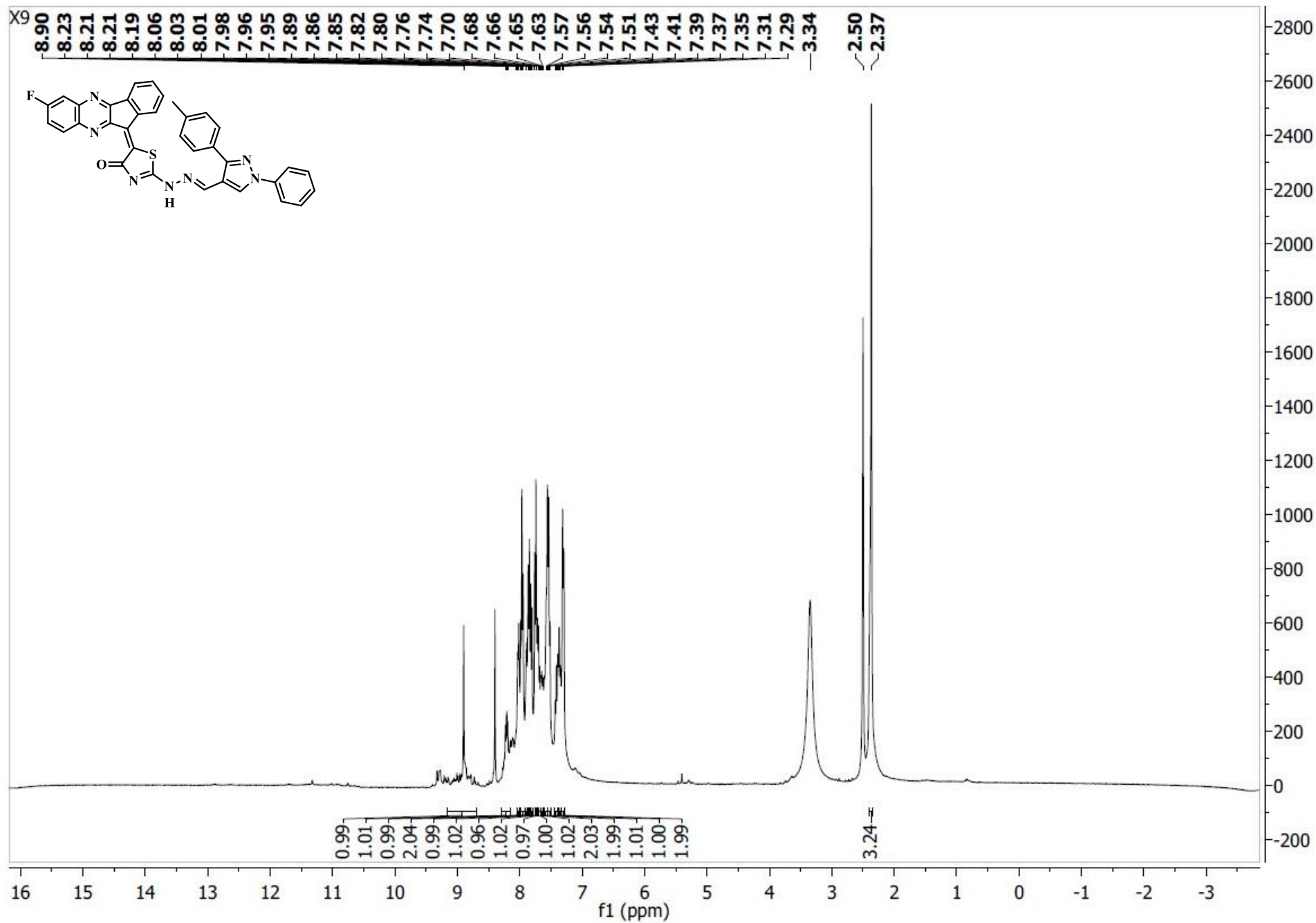
10: <sup>13</sup>C NMR



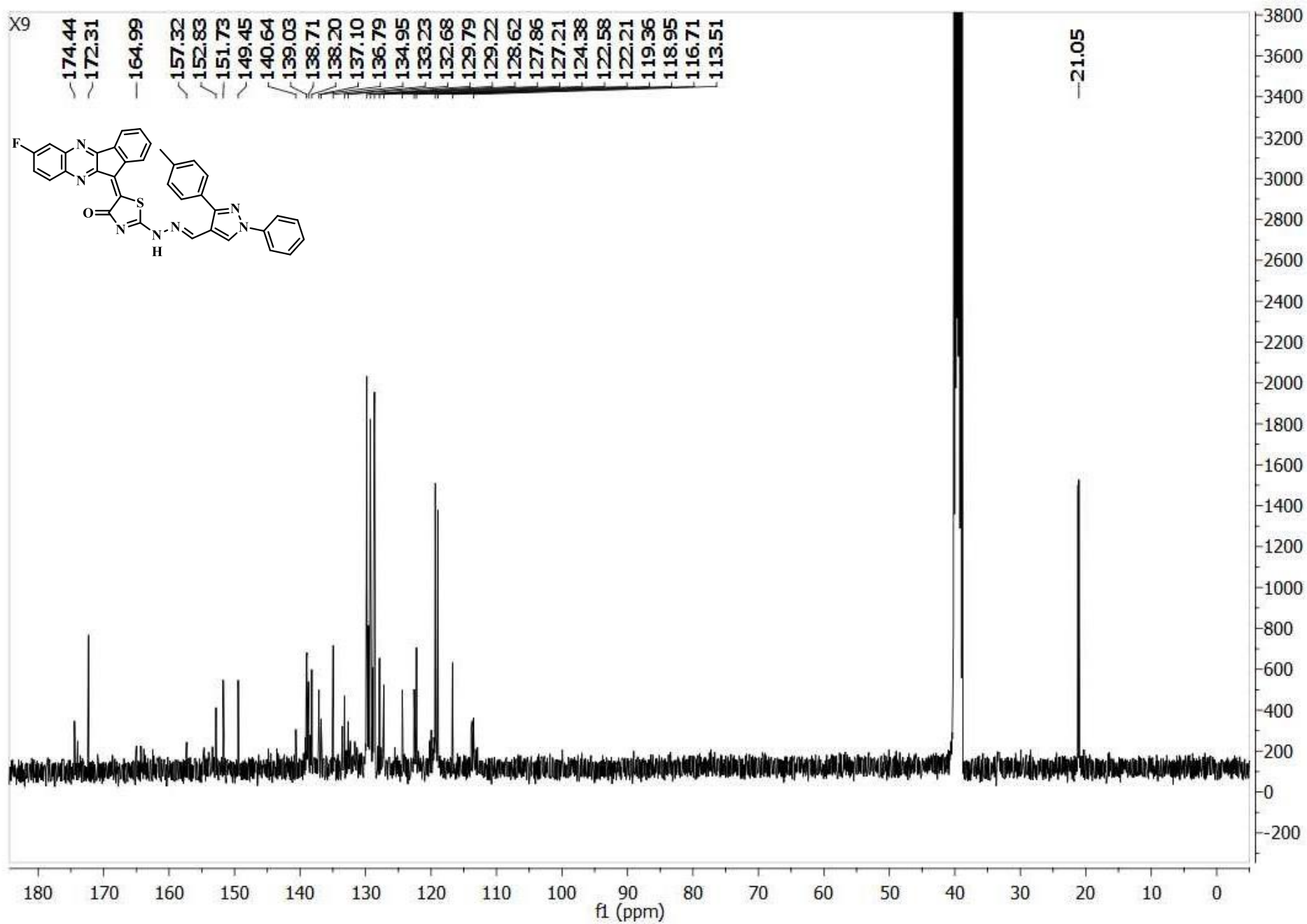
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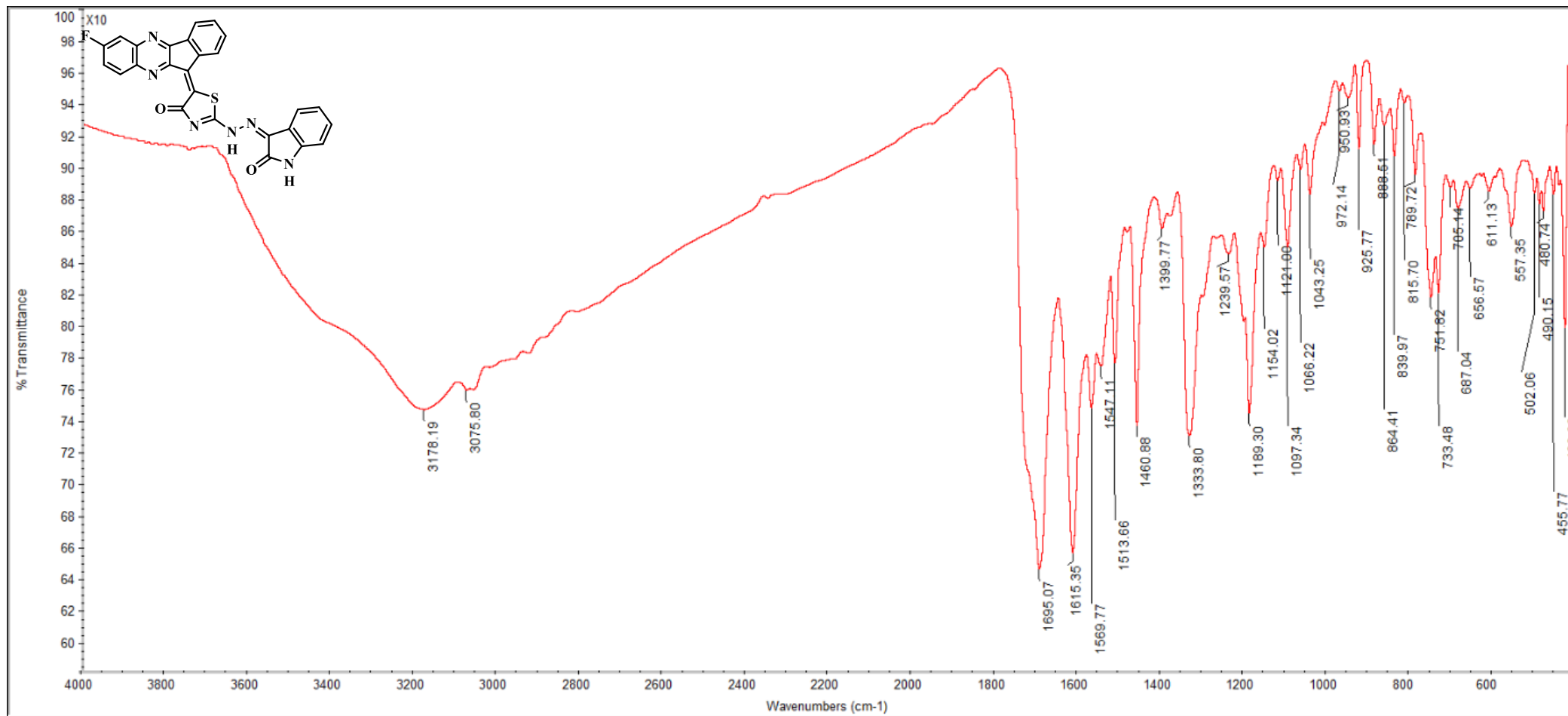


12:  $^1\text{H}$  NMR

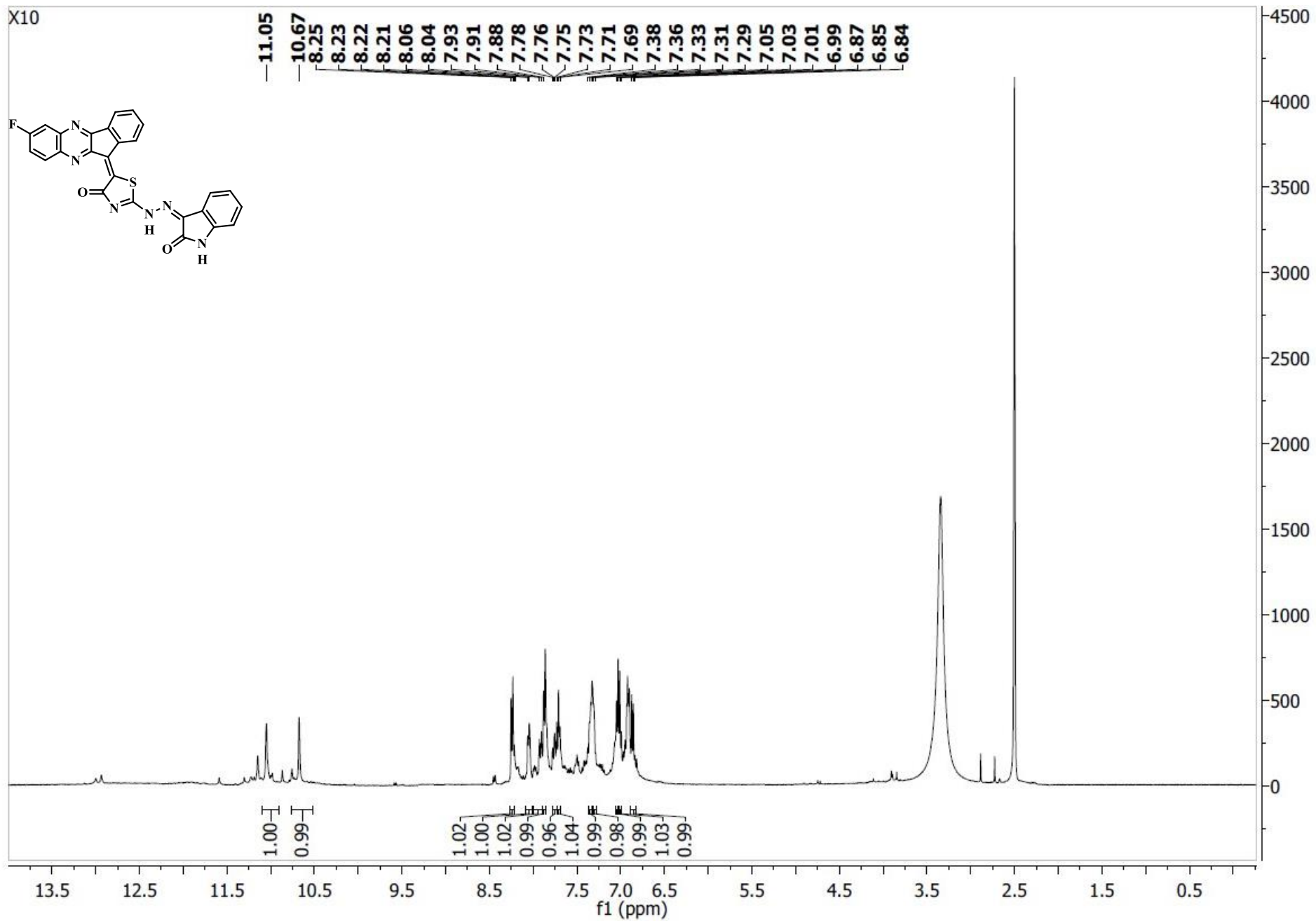


12: <sup>13</sup>C NMR

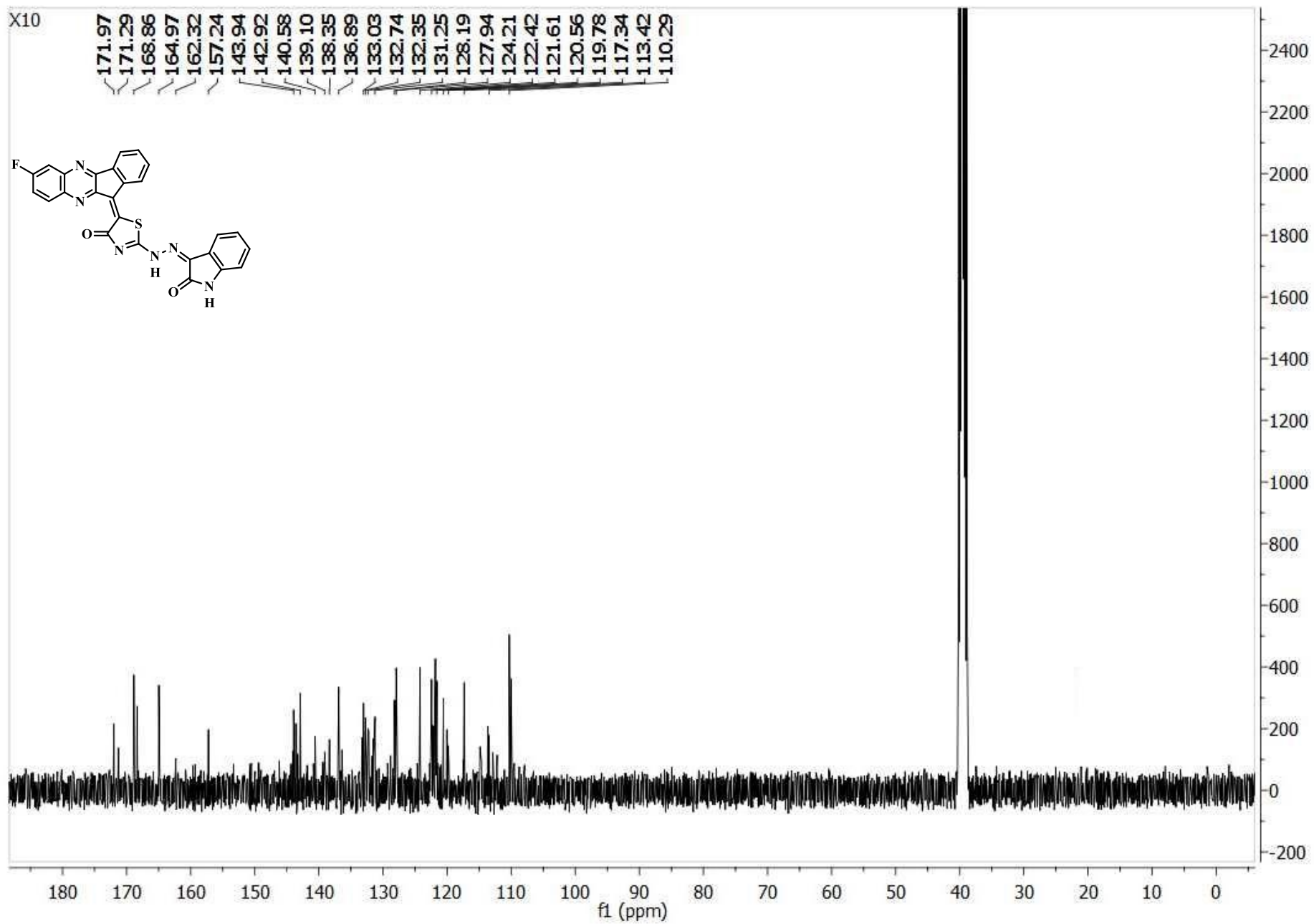




13: <sup>1</sup>H NMR

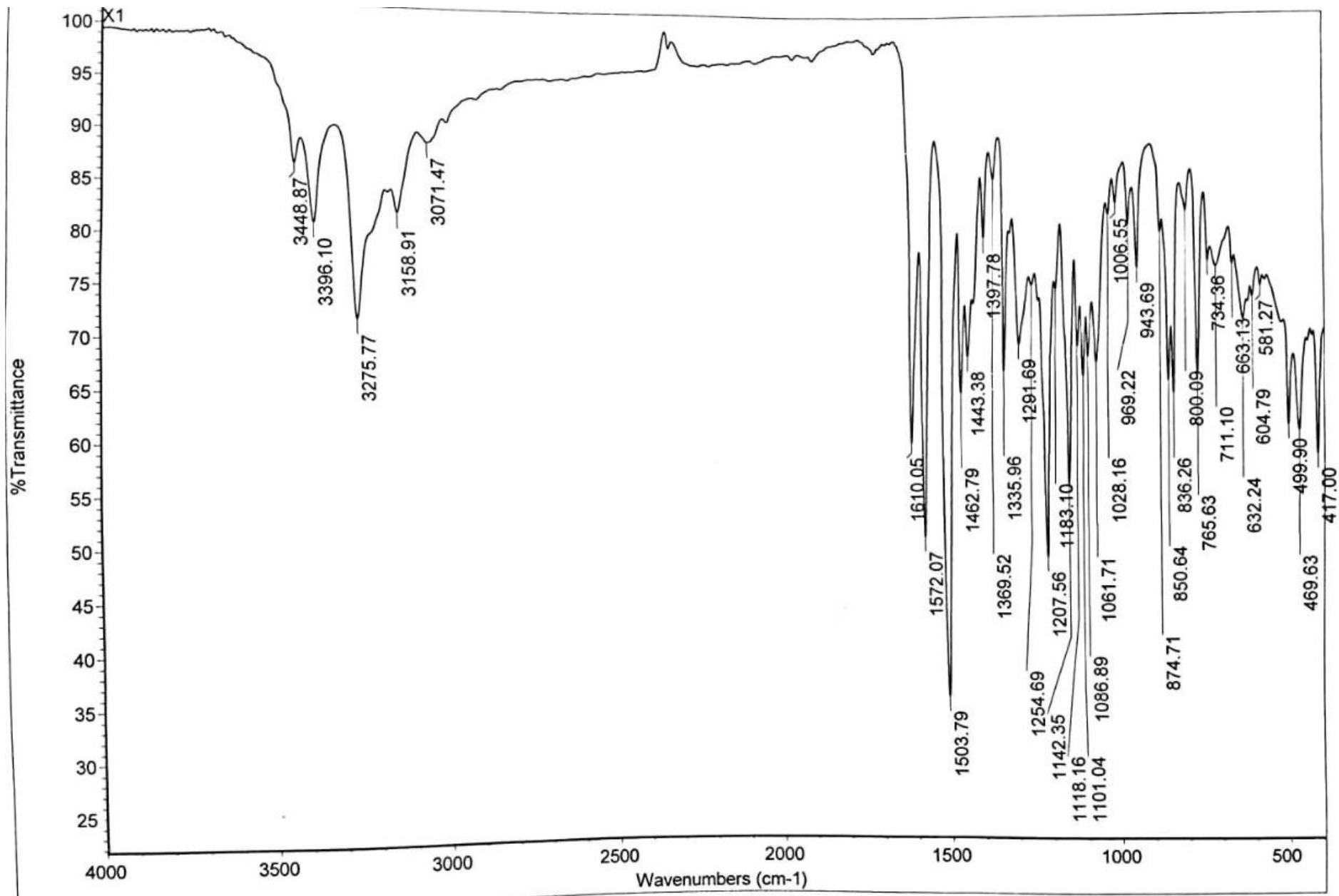


13: <sup>13</sup>C NMR

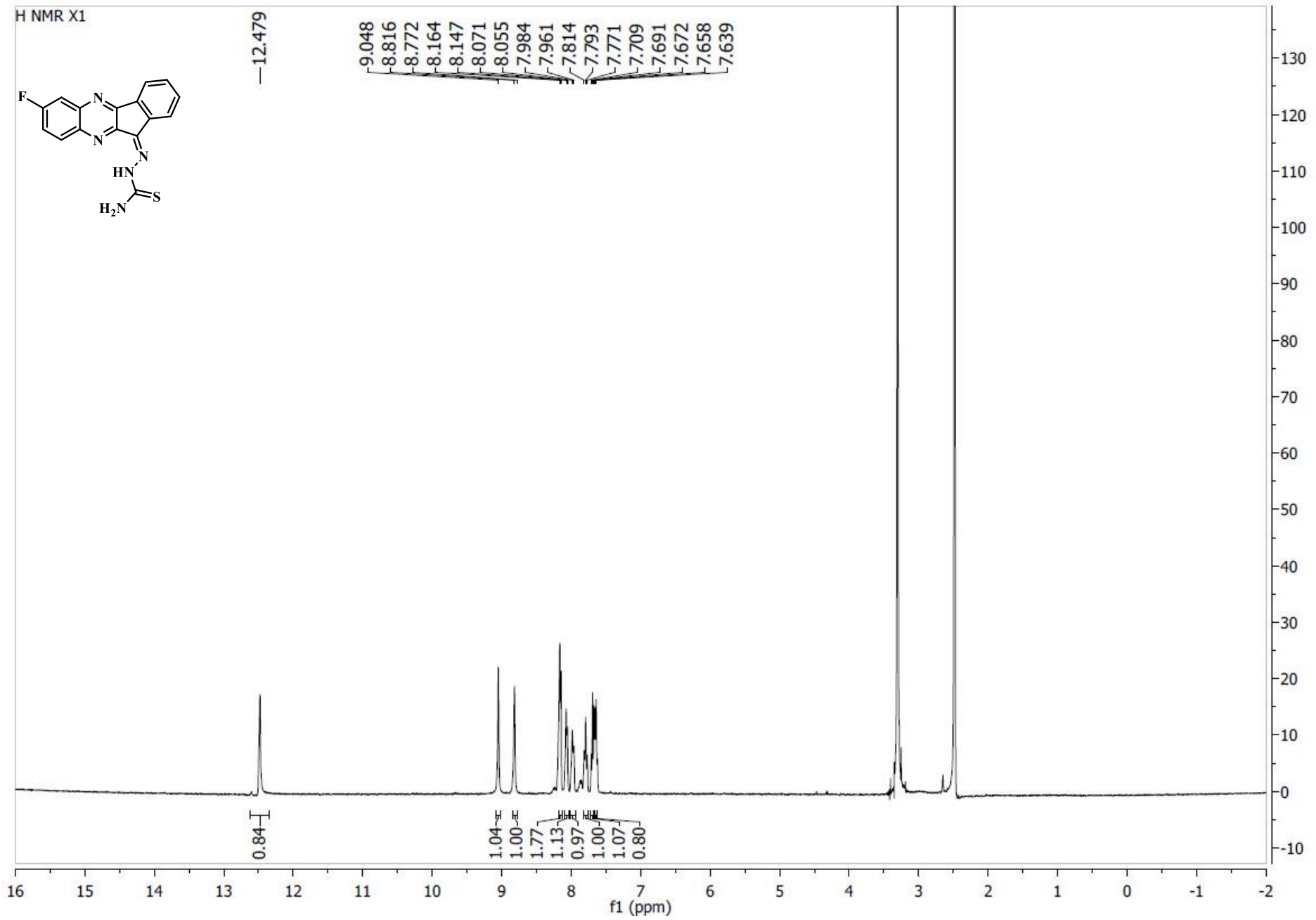


14: IR

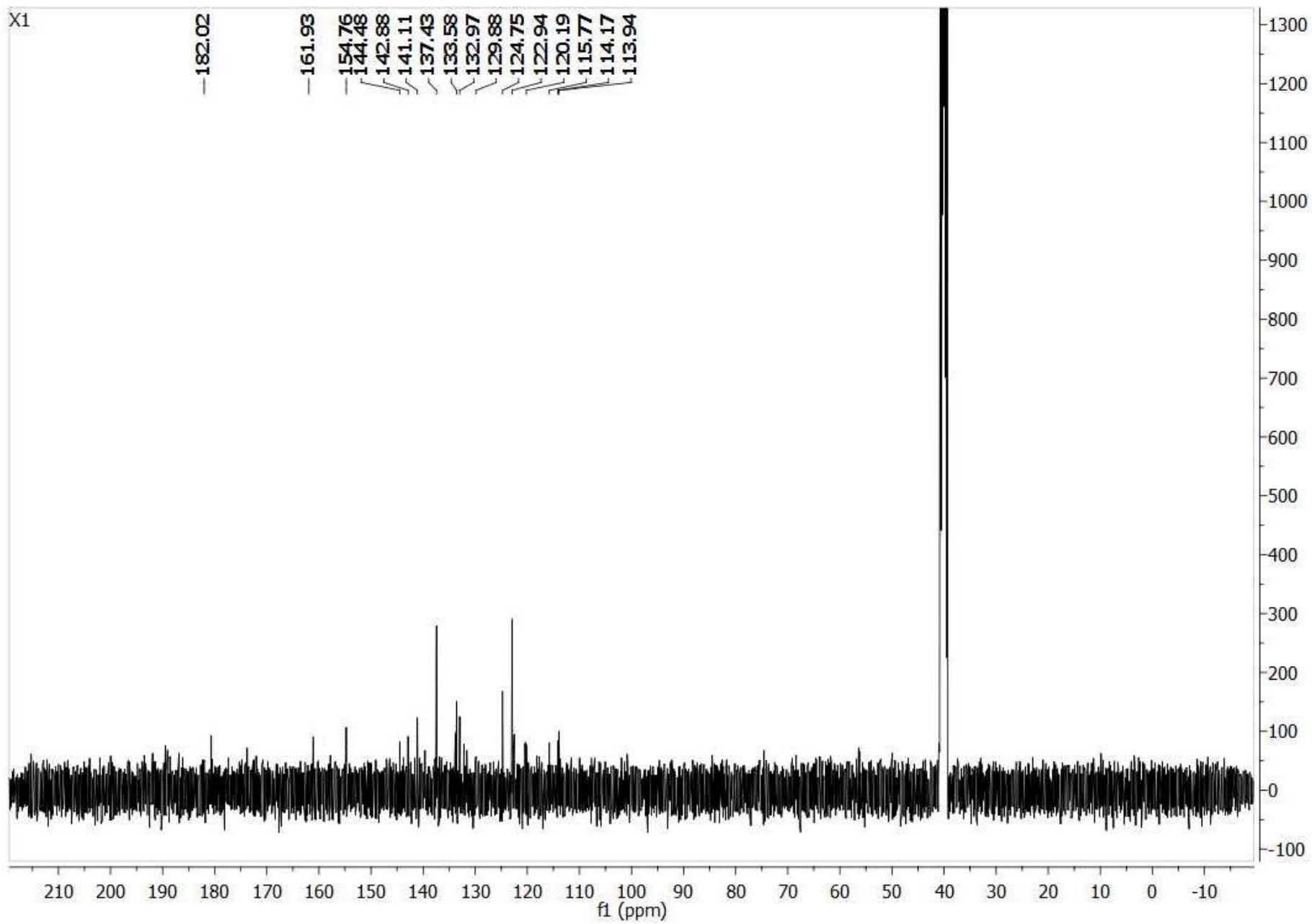




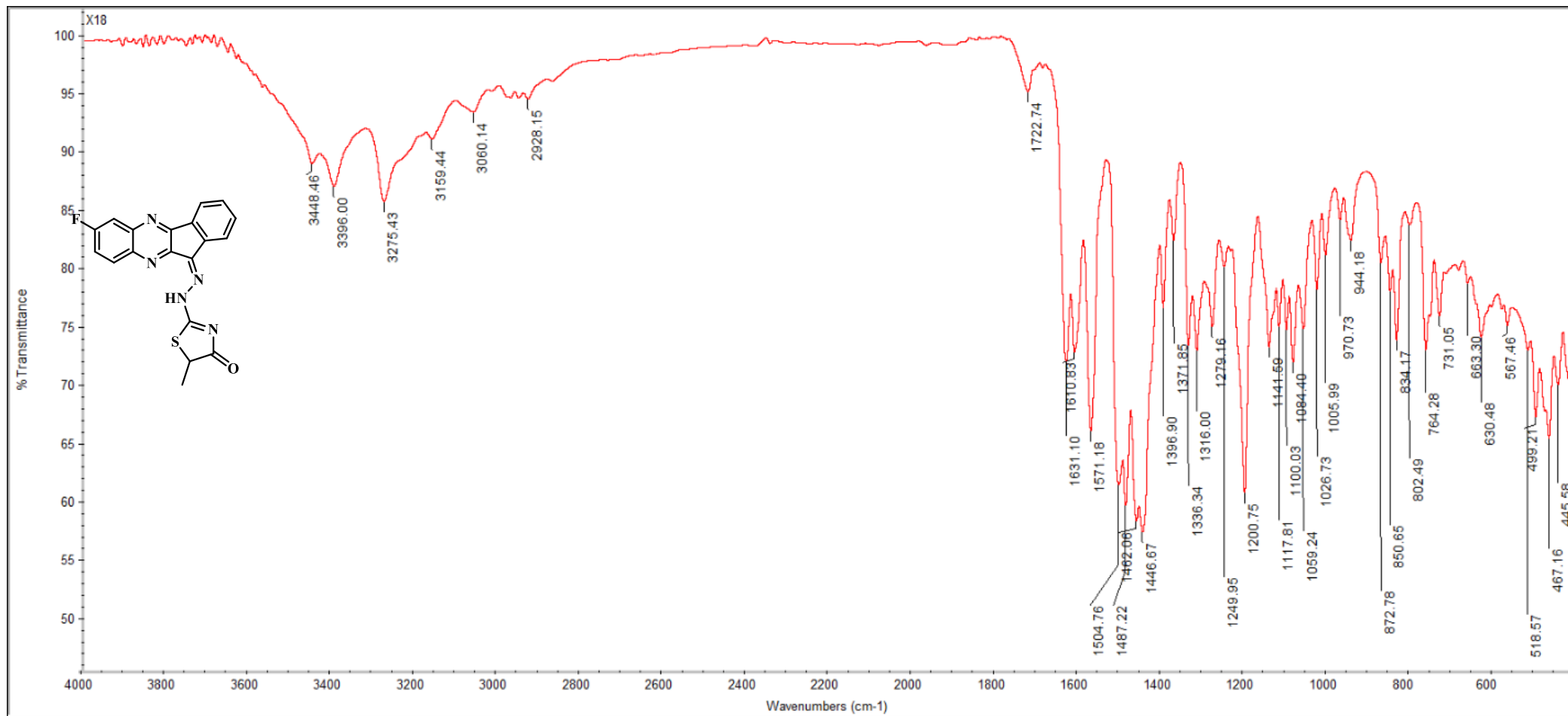
14: <sup>1</sup>H NMR



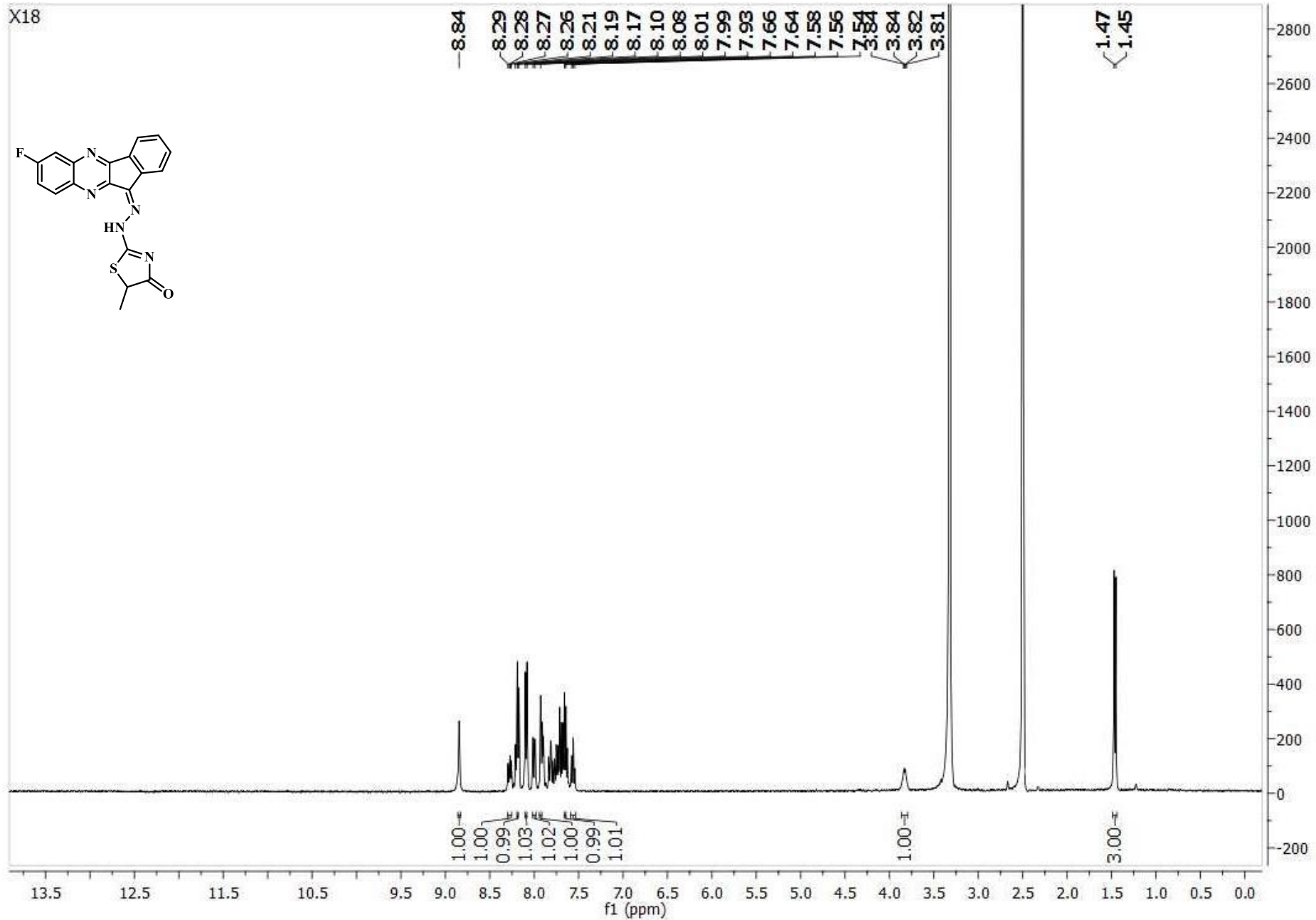
14:  $^{13}\text{C}$  NMR



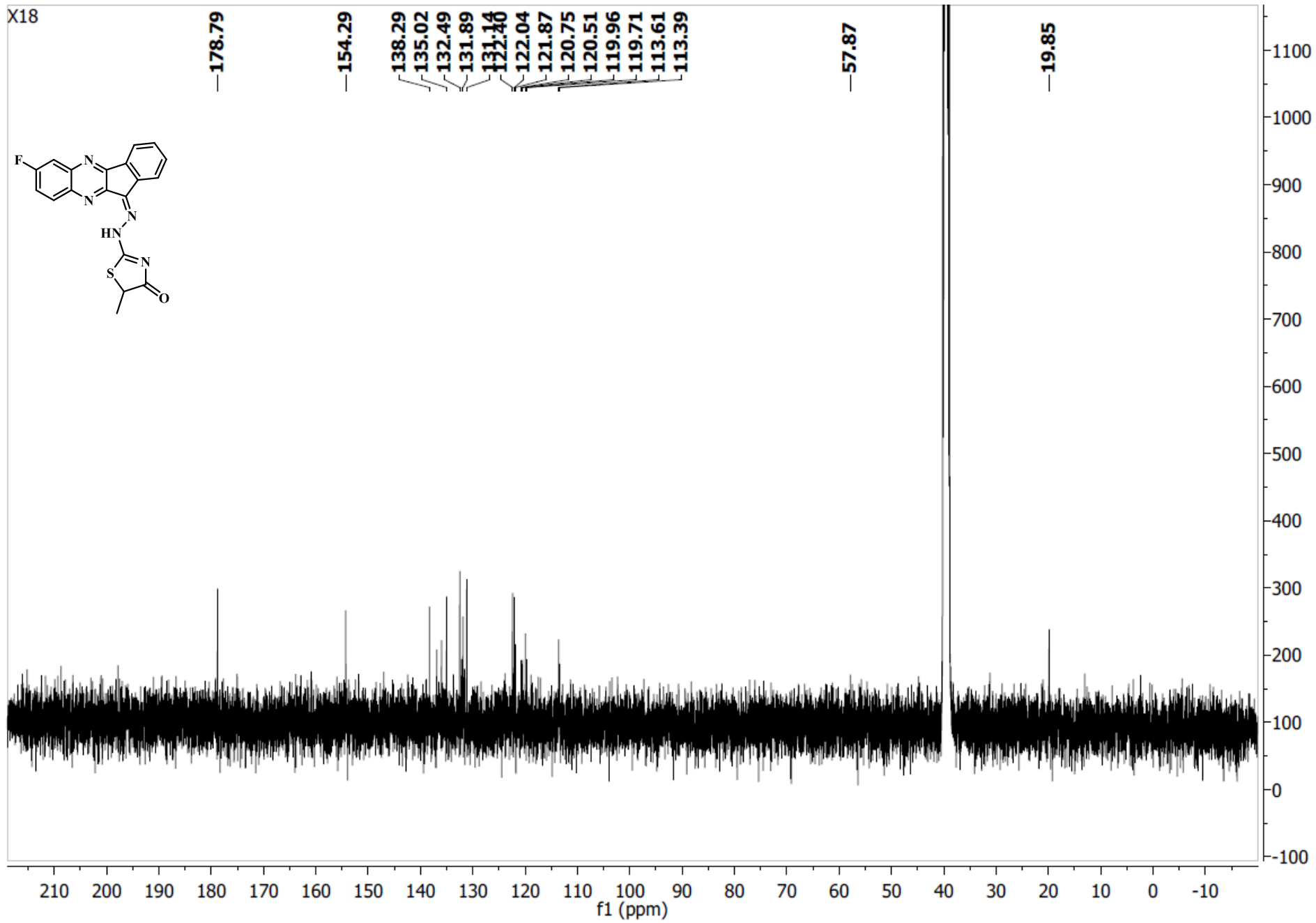
15: IR



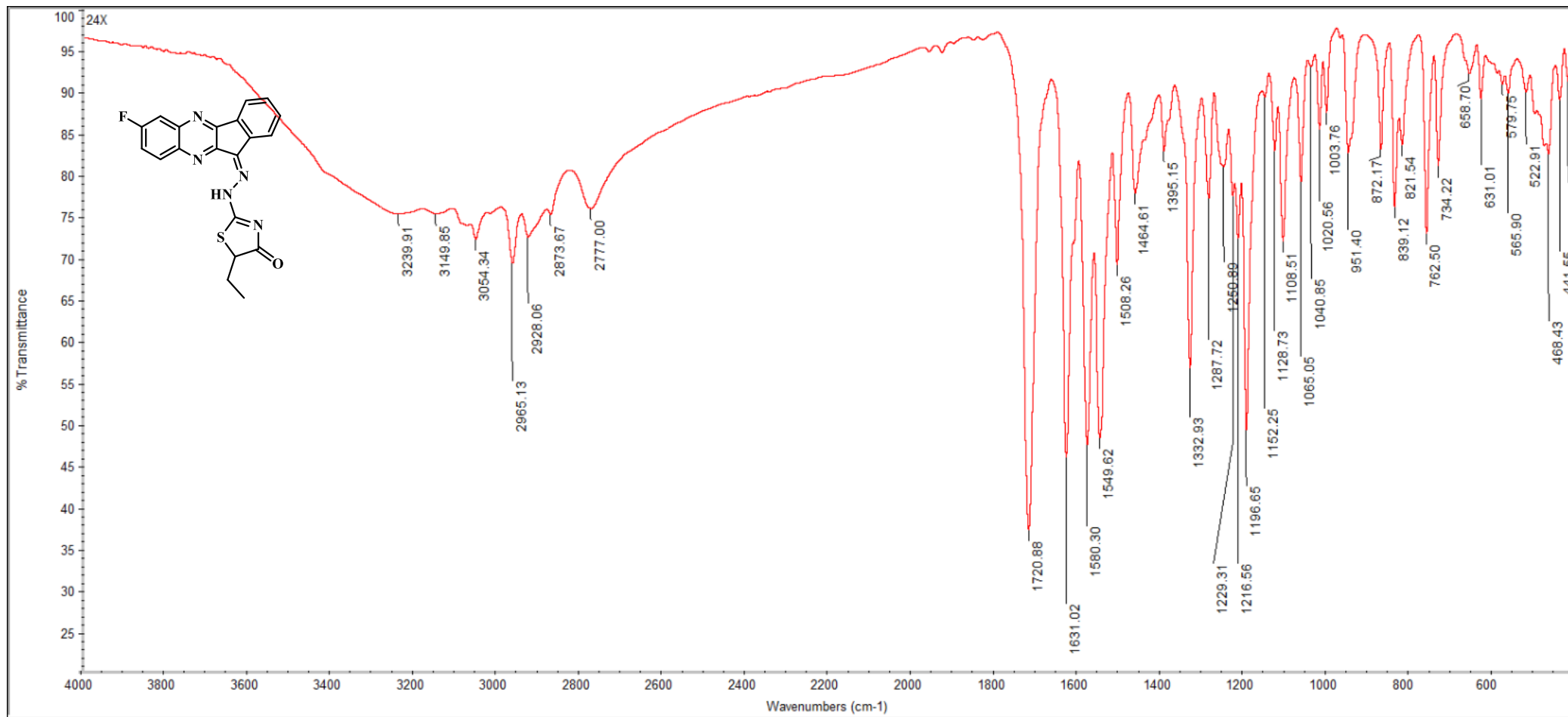
15: <sup>1</sup>H NMR



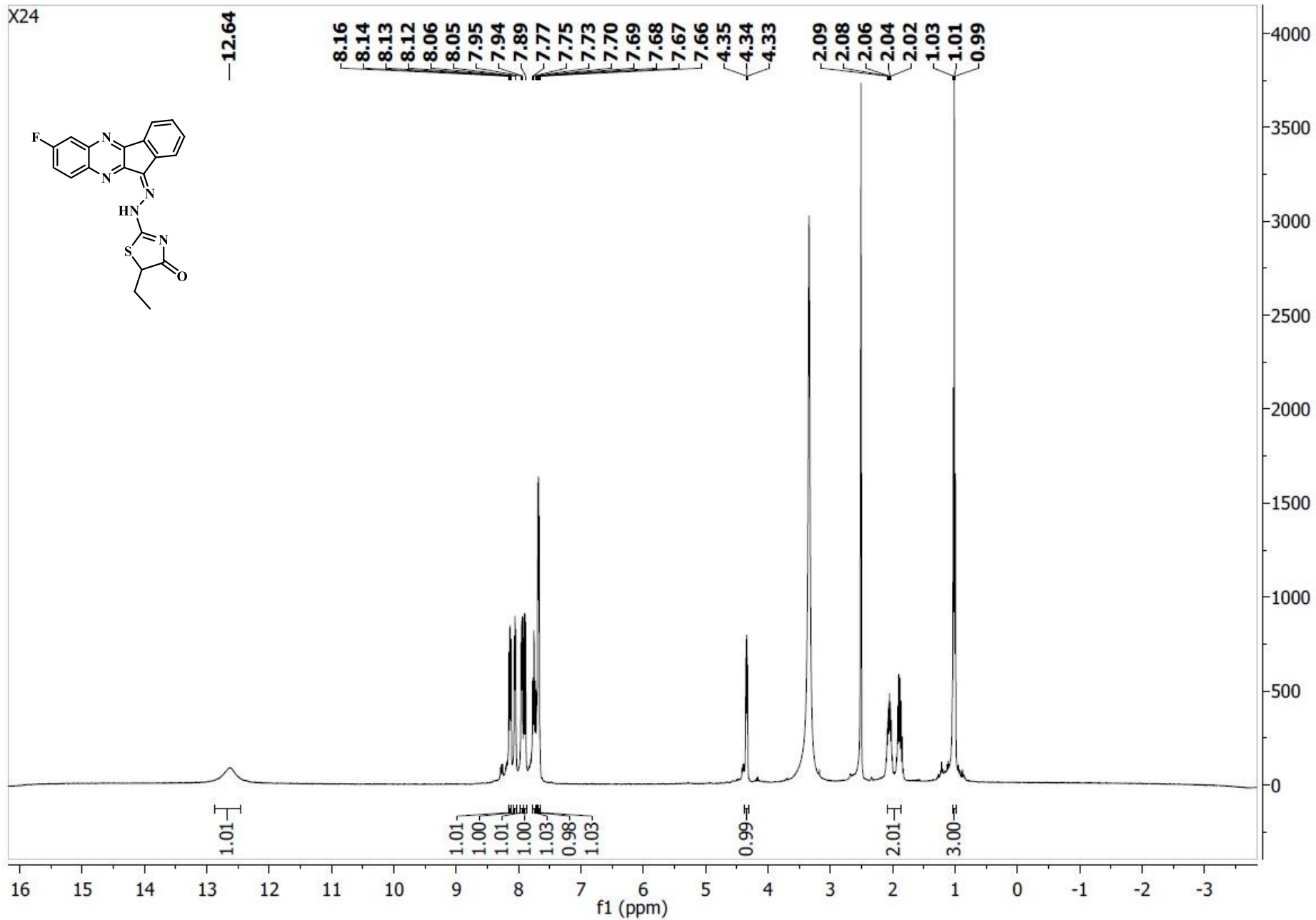
15: <sup>13</sup>C NMR



16: IR

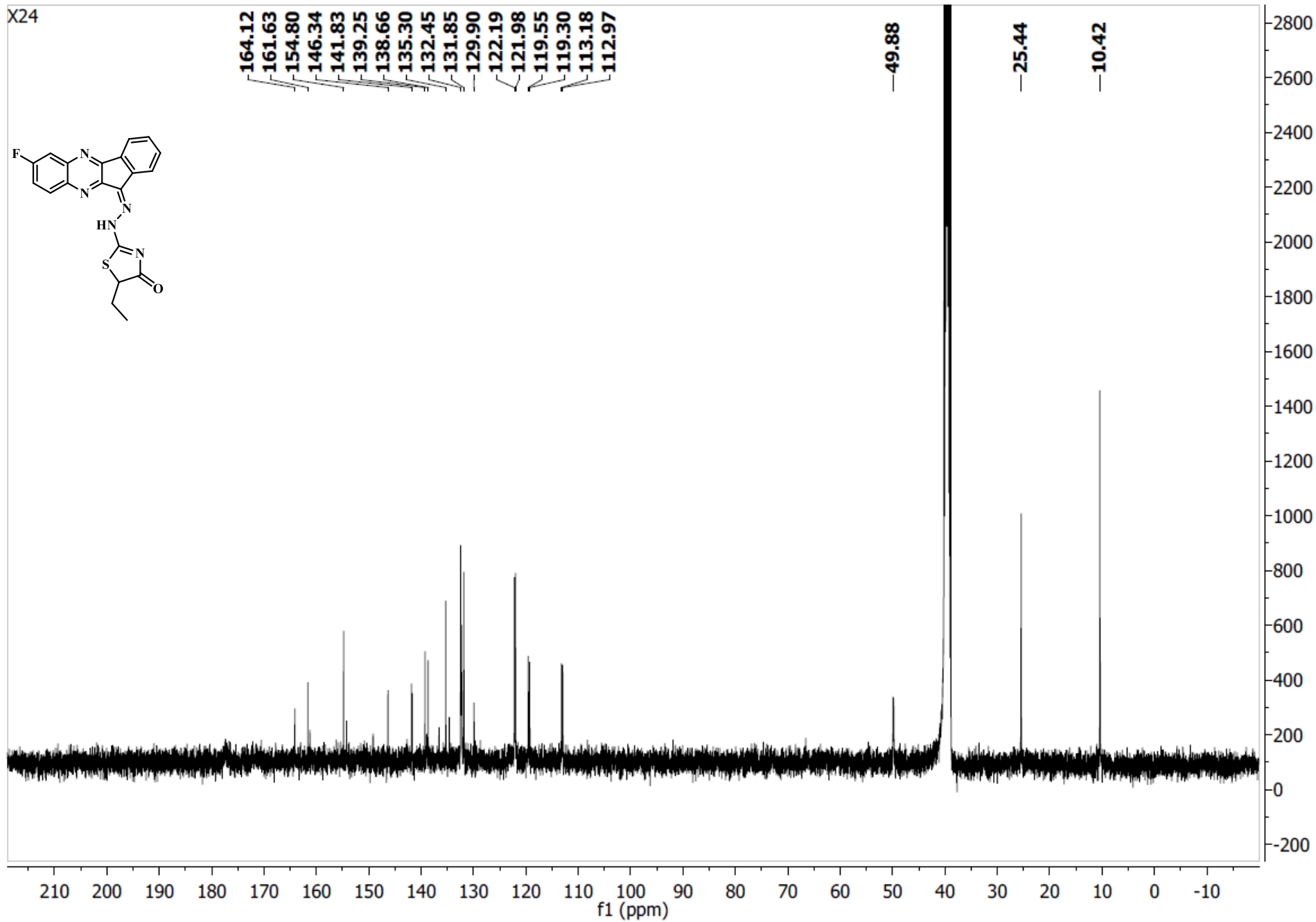


16: <sup>1</sup>H NMR

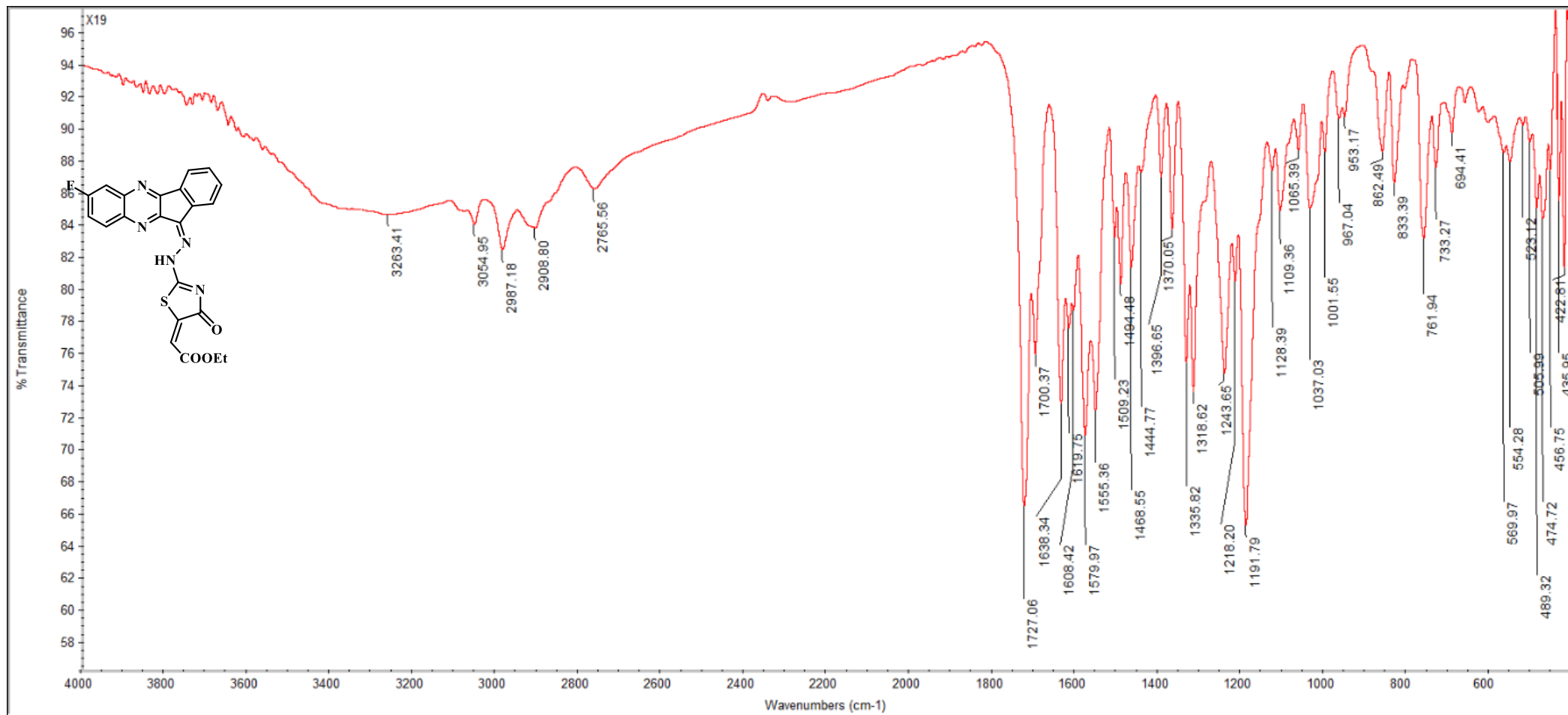


16:  $^{13}\text{C}$  NMR



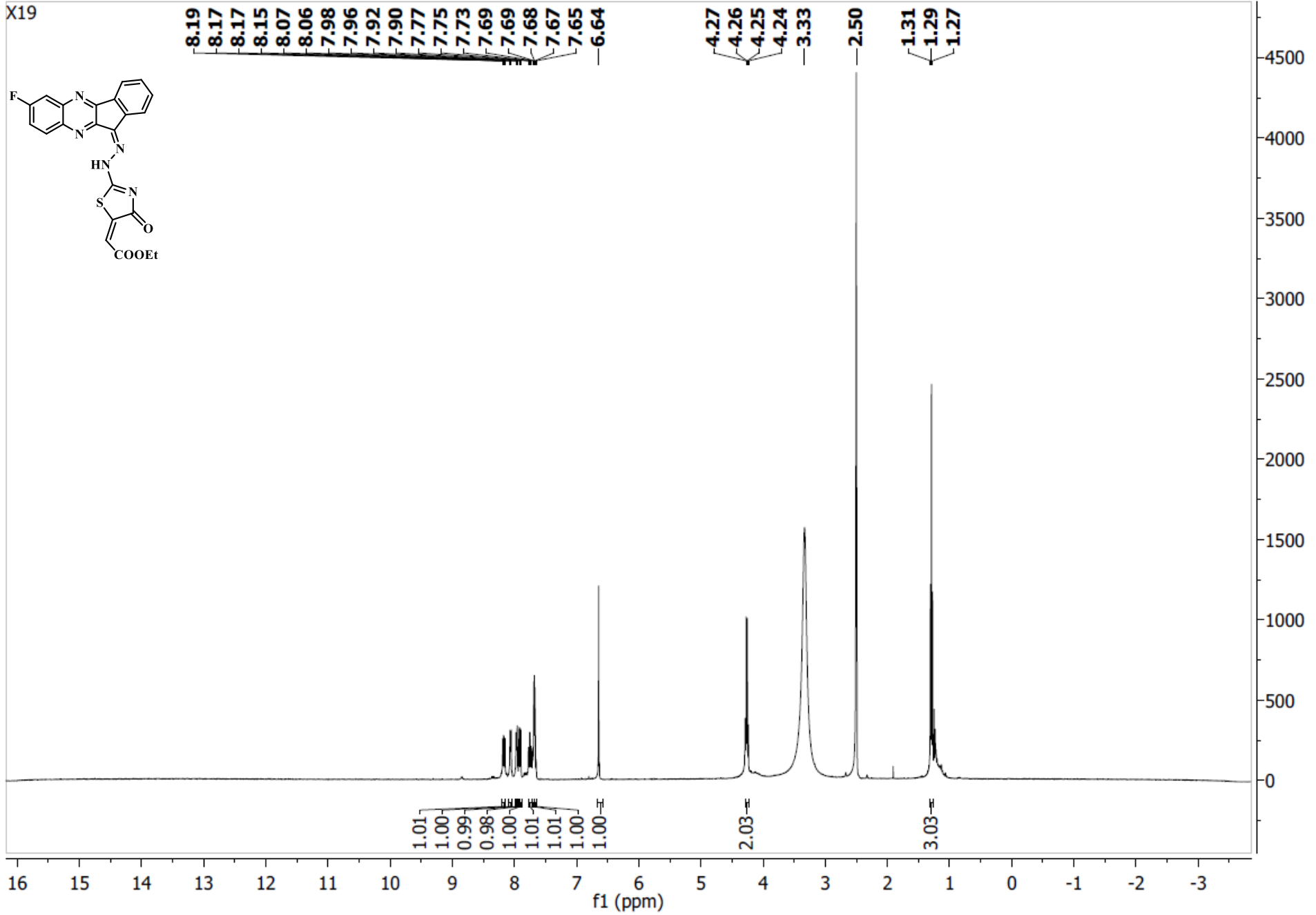
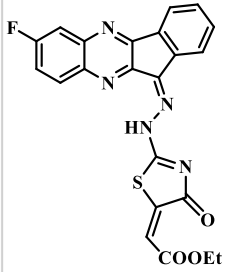


17: IR

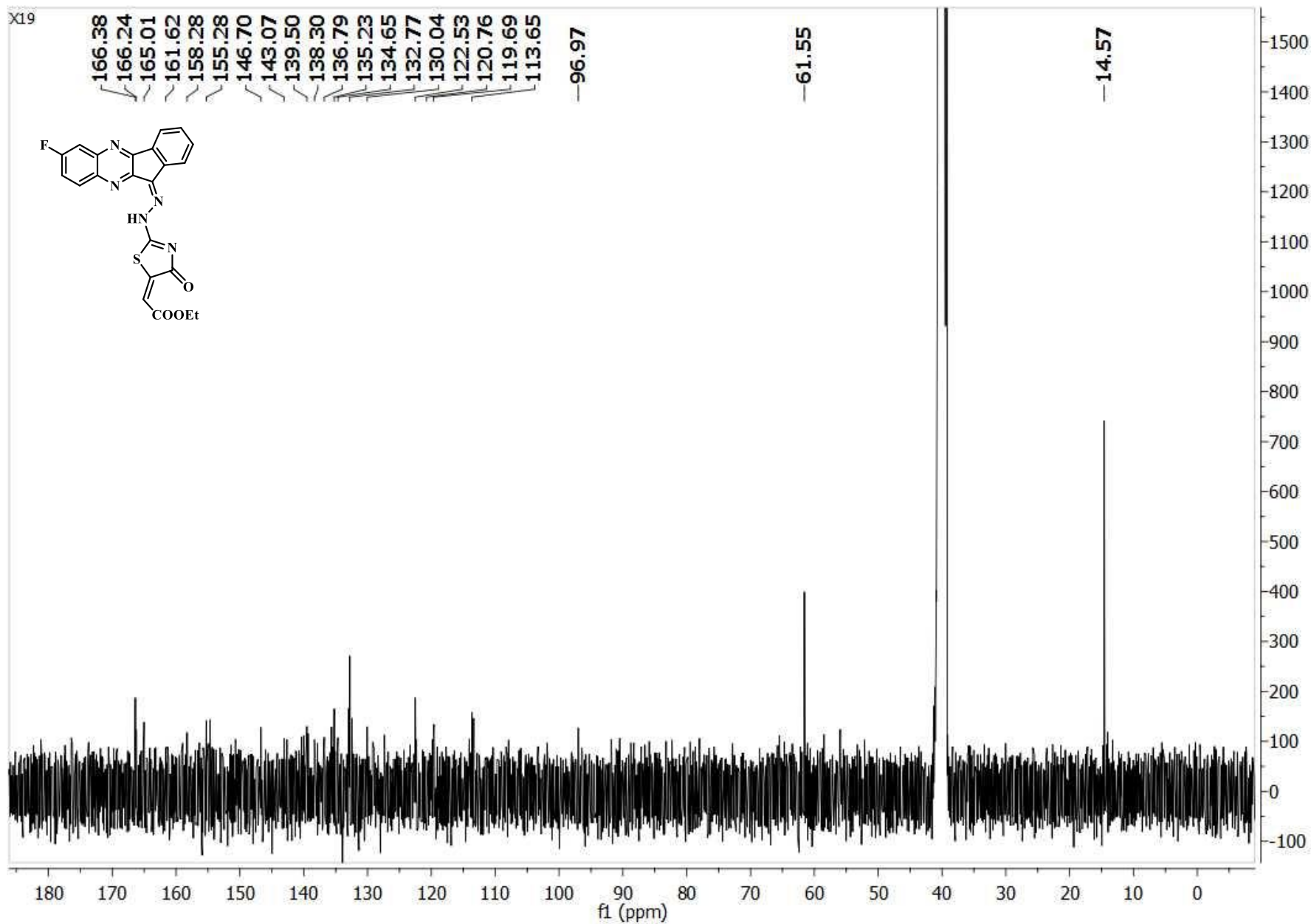


17: <sup>1</sup>H NMR

X19

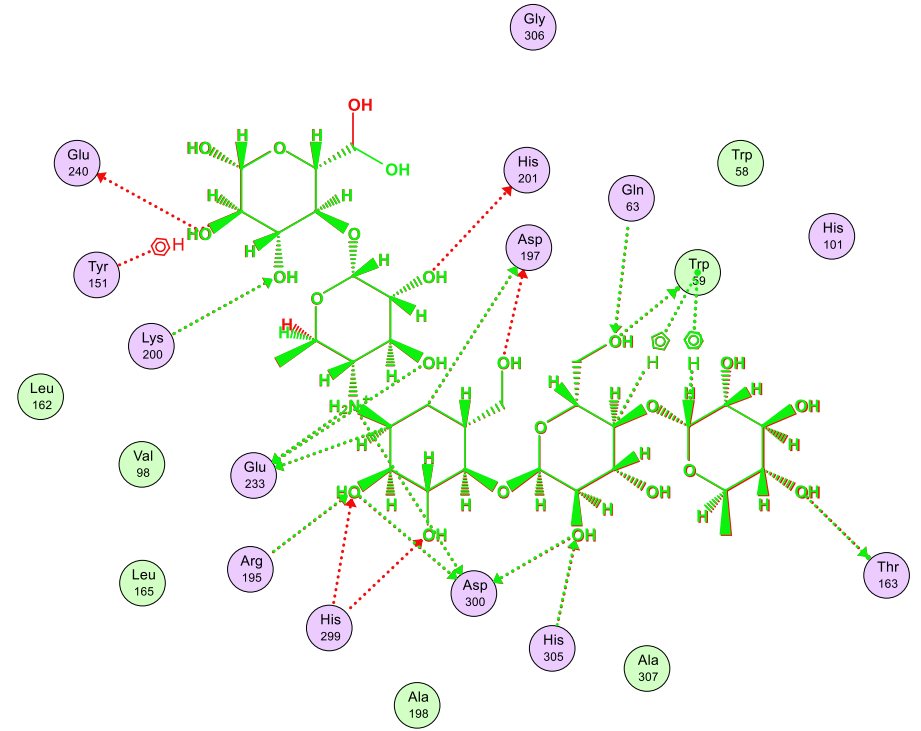
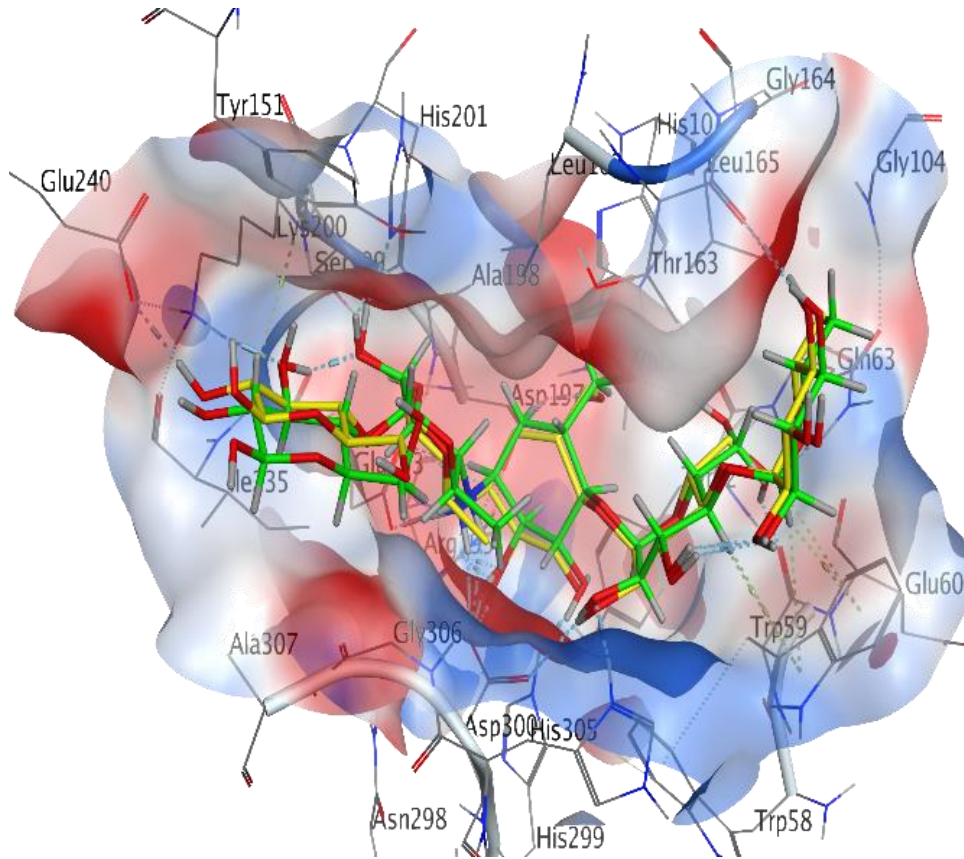


17: <sup>13</sup>C NMR

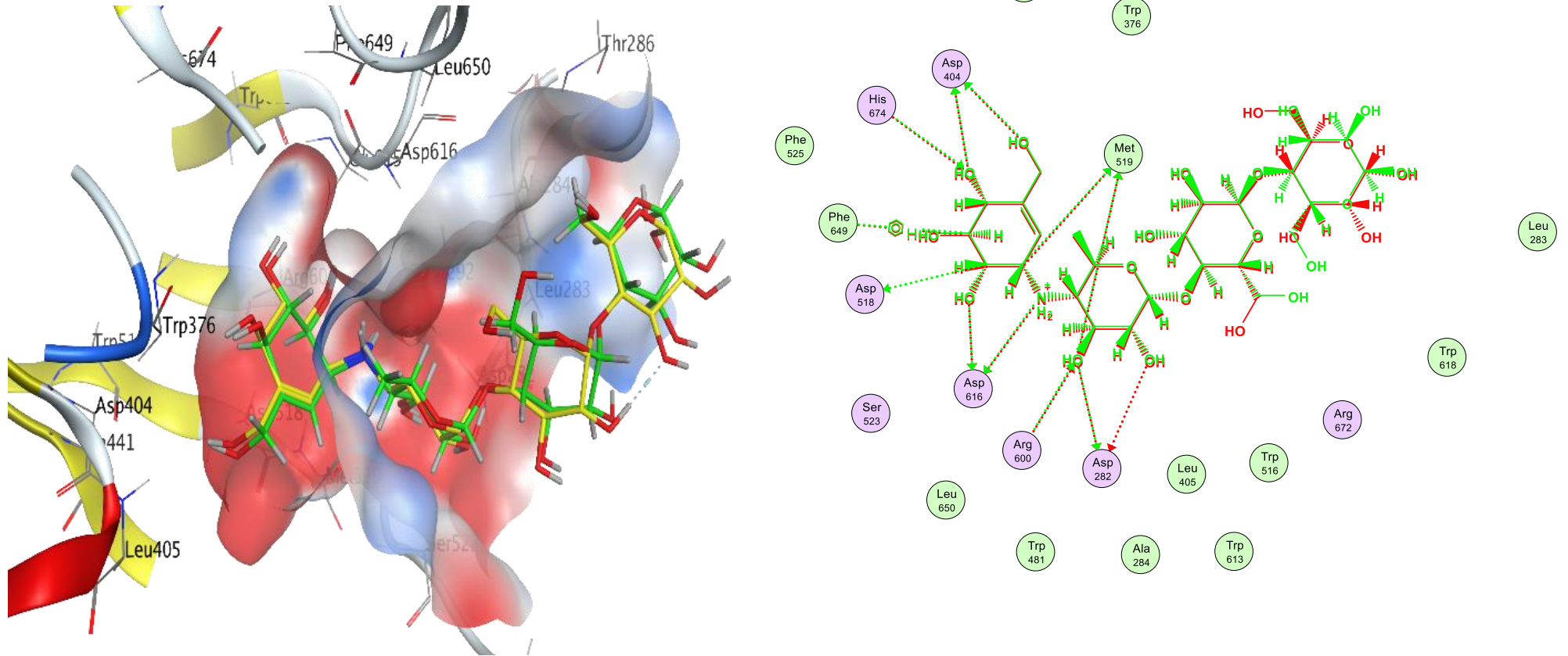


# Docking Studies

A)

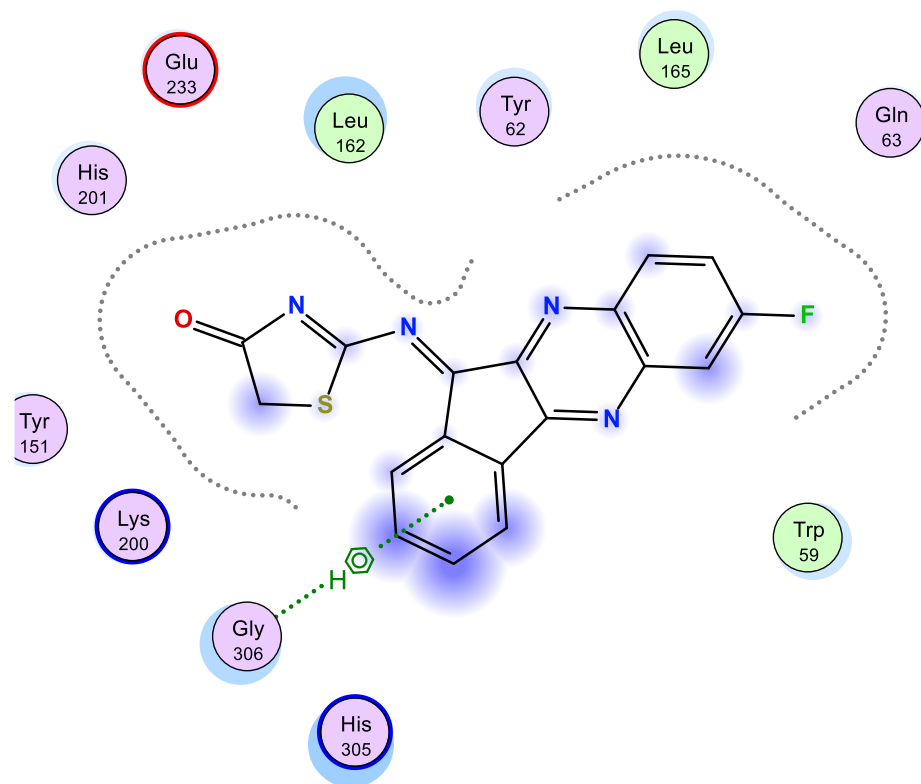
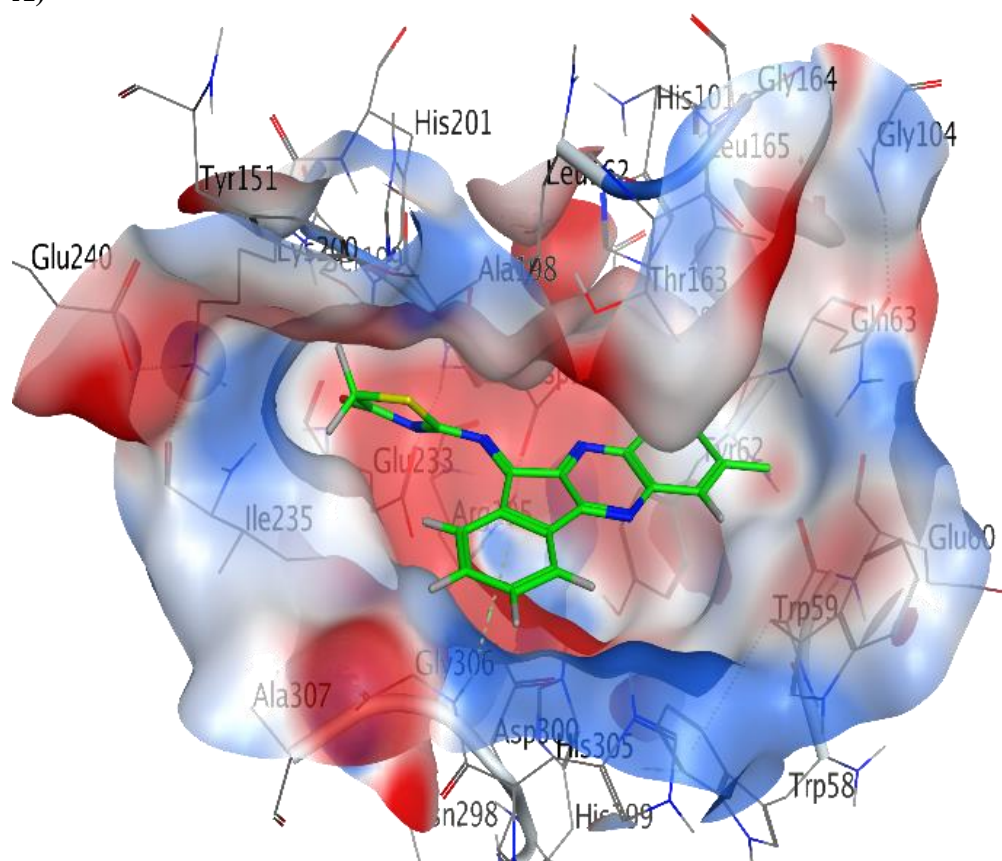


B)

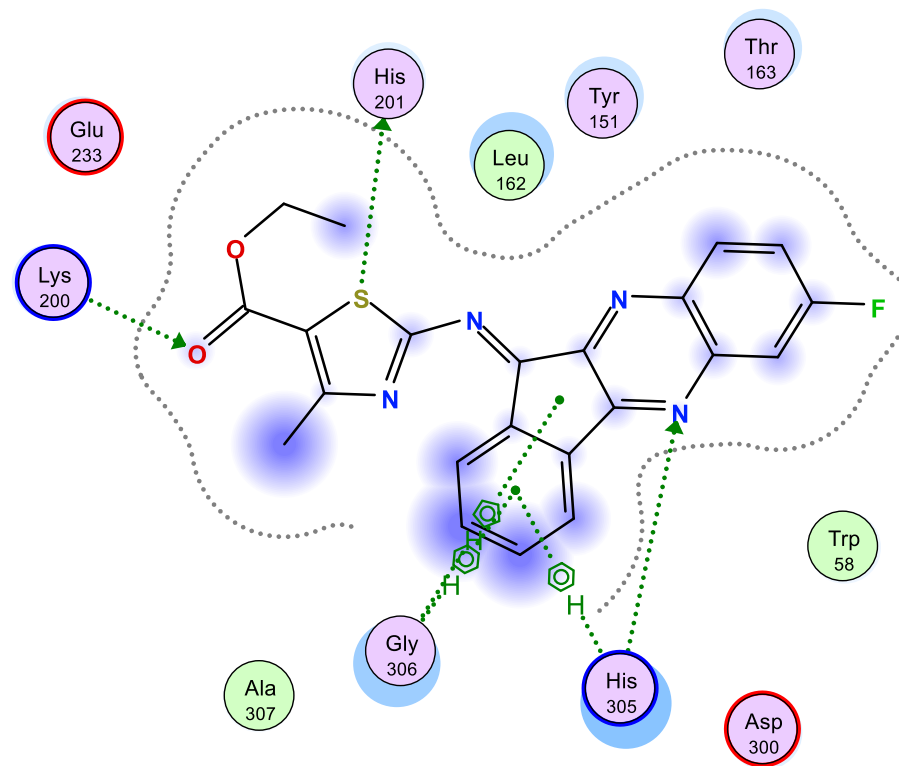
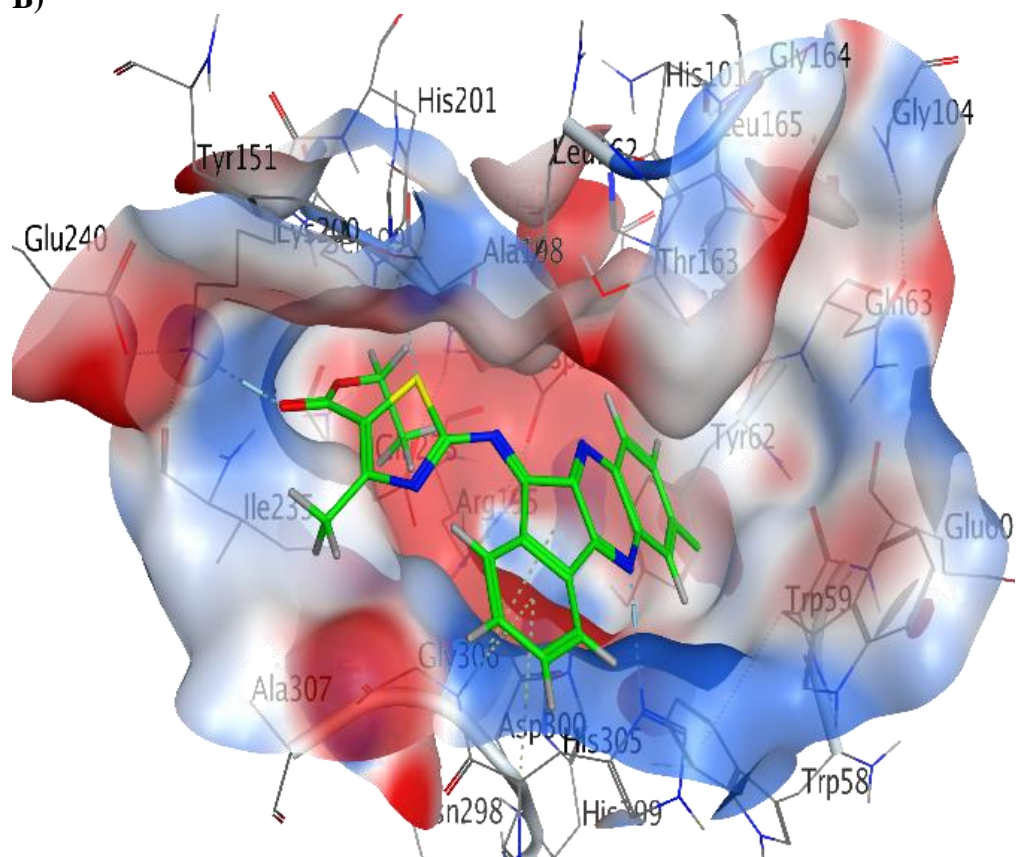


**Figure 3:** A) Overlay between co-crystallized ligand (yellow) and re-docked pose (green) of Acarbose (RMSD= 0.66 Å) in  $\alpha$ -amylase (PDB ID: 1B2Y), complex overlay and ligand interactions. B) Overlay between co-crystallized ligand (yellow) and re-docked pose (green) of Acarbose (RMSD= 0.56 Å) in  $\alpha$ -glucosidase (PDB ID: 5NN8), complex overlay and ligand interactions.

A)

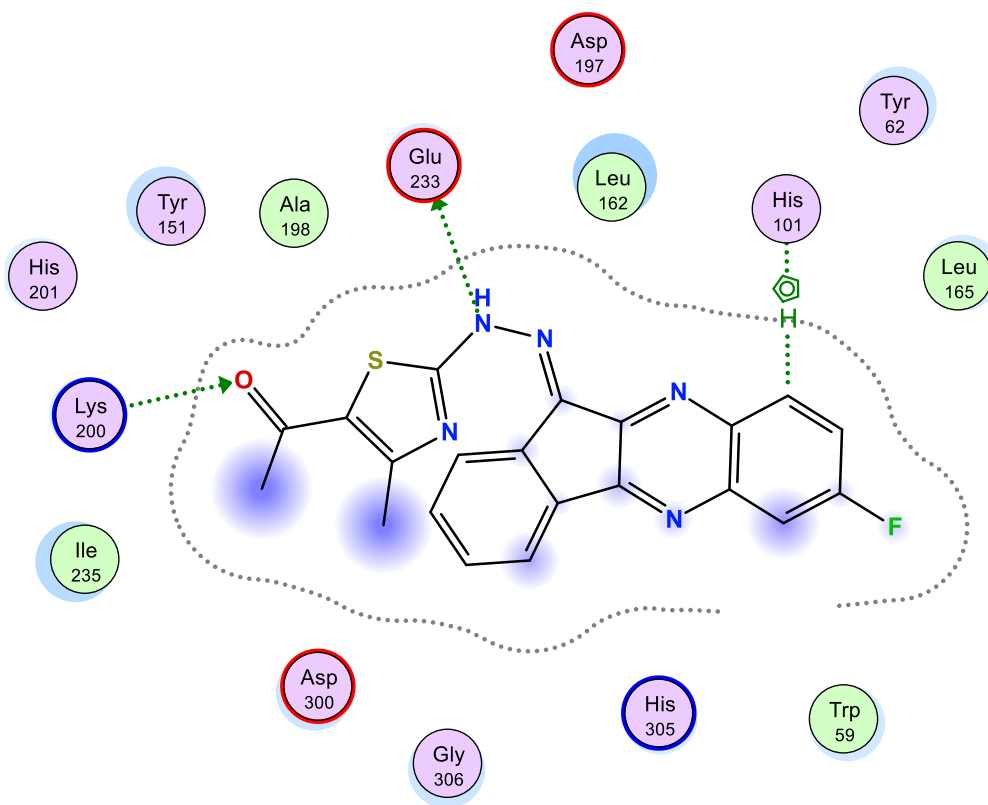
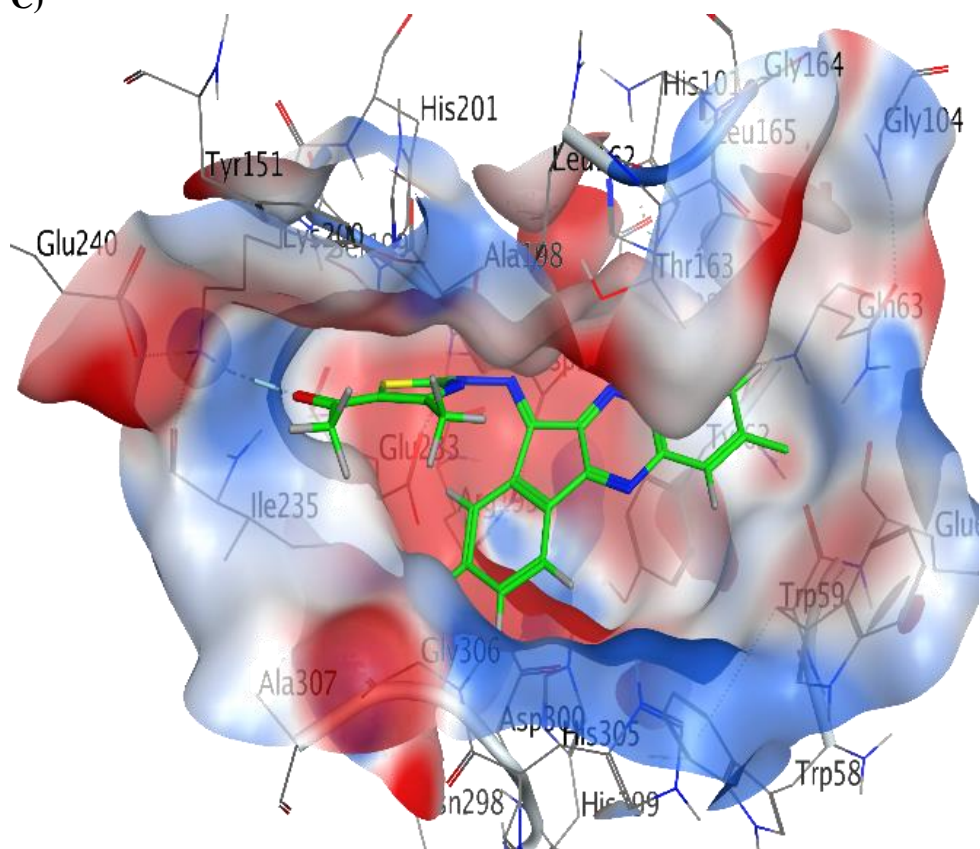


B)





c)



D)

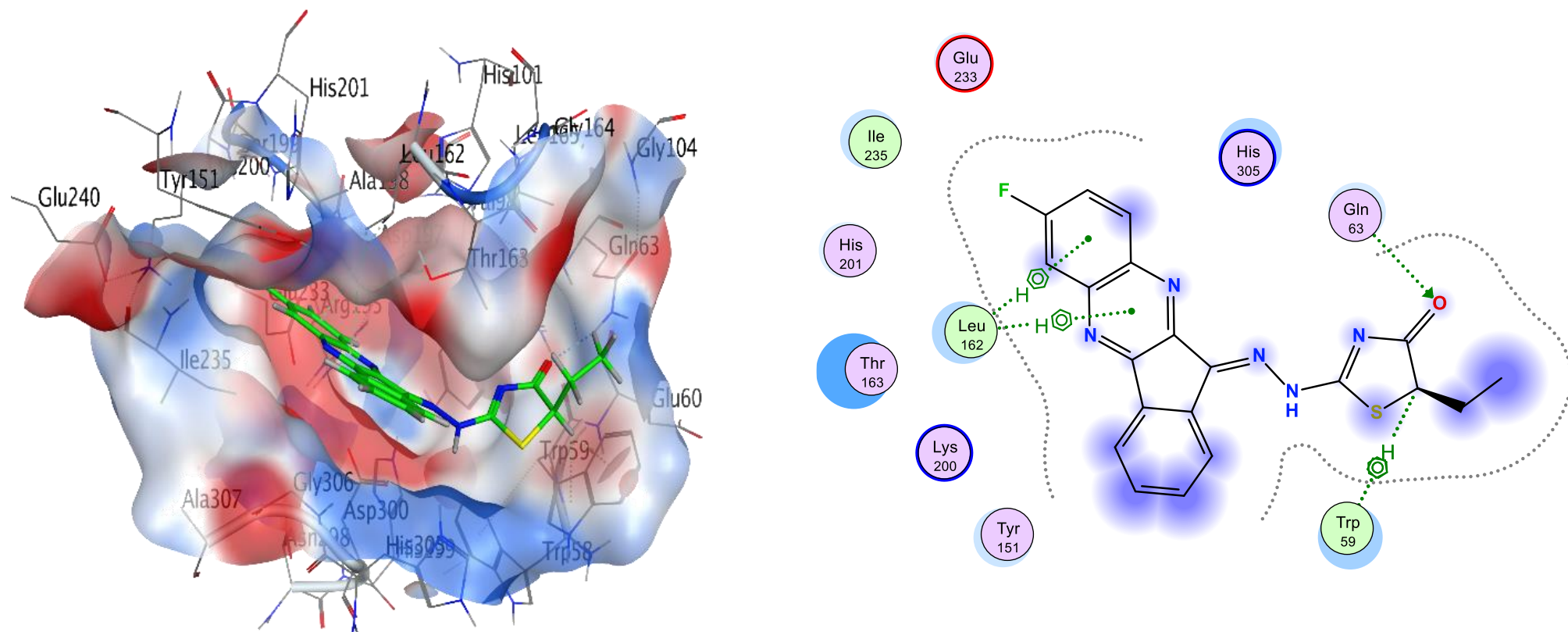
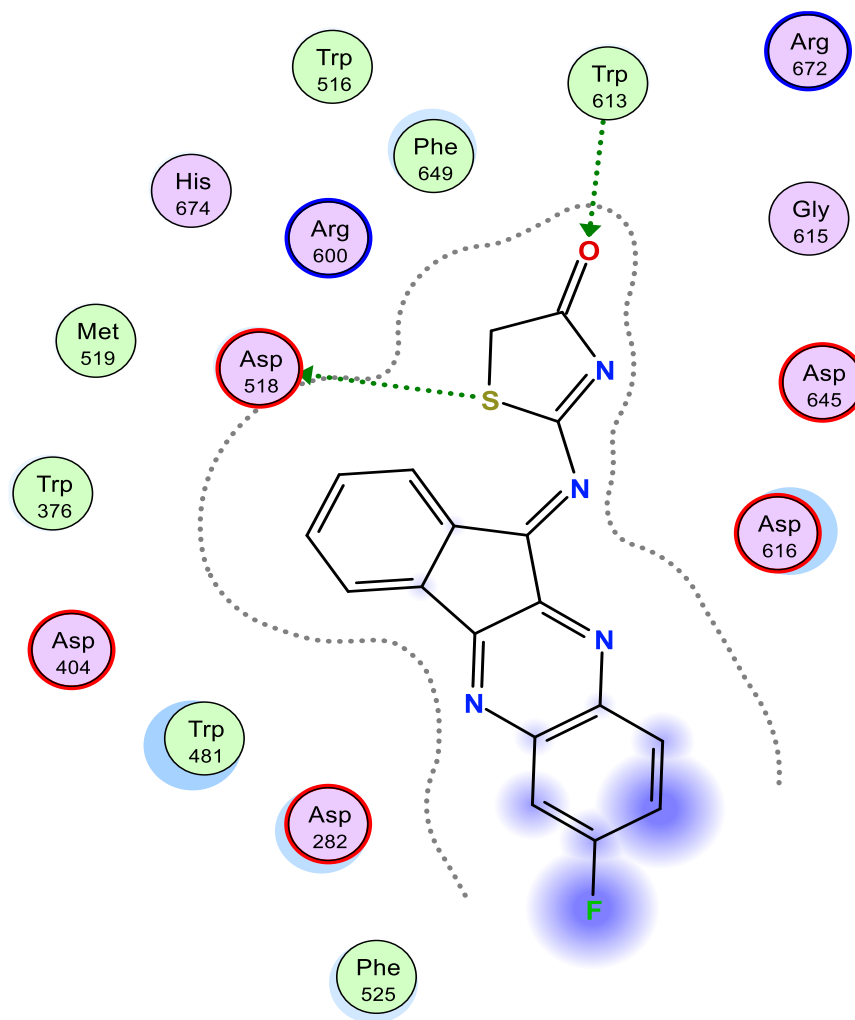
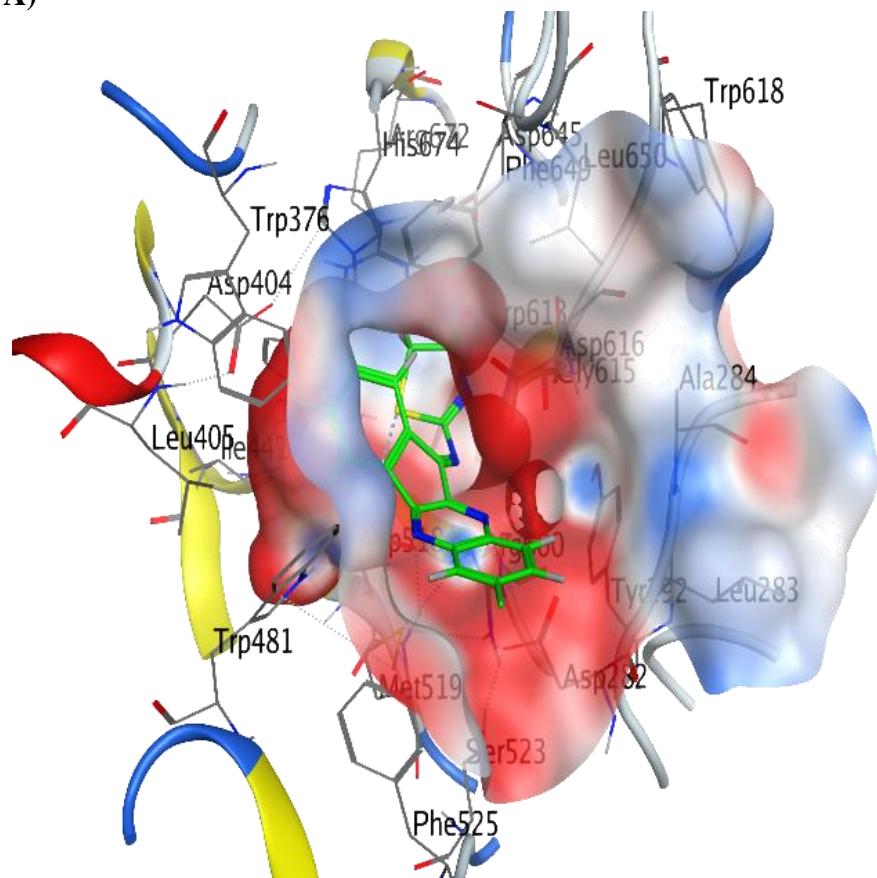
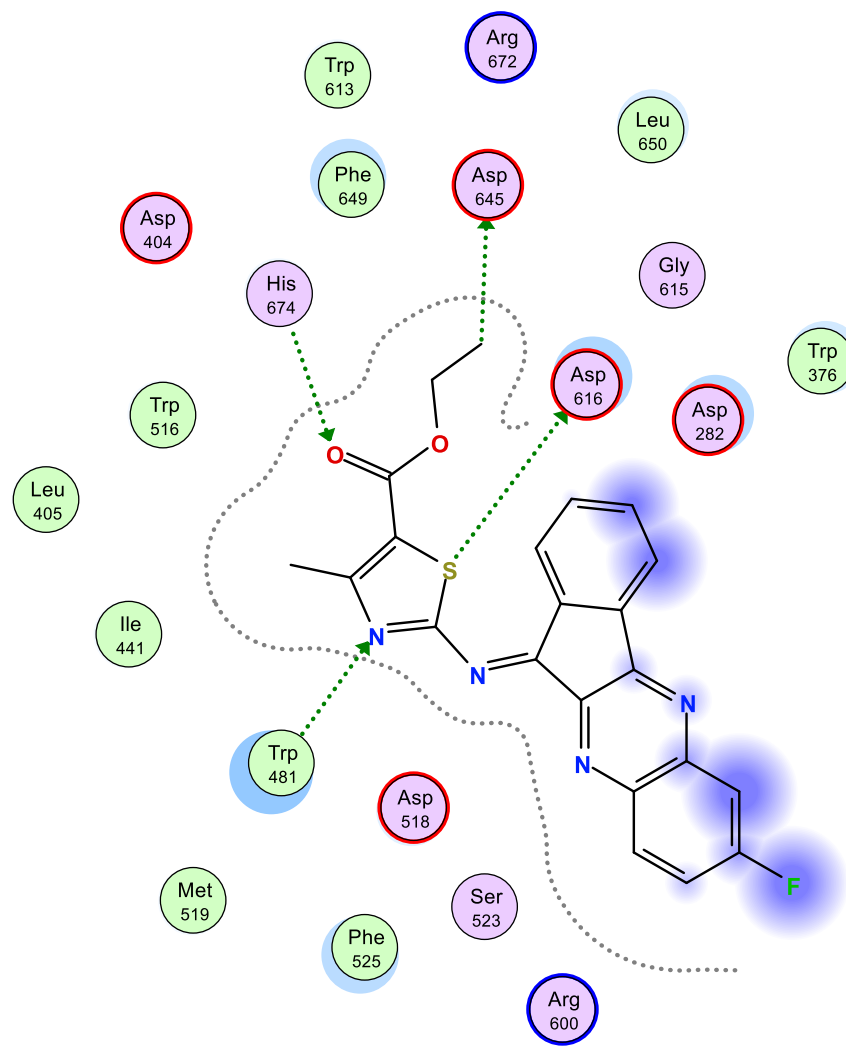
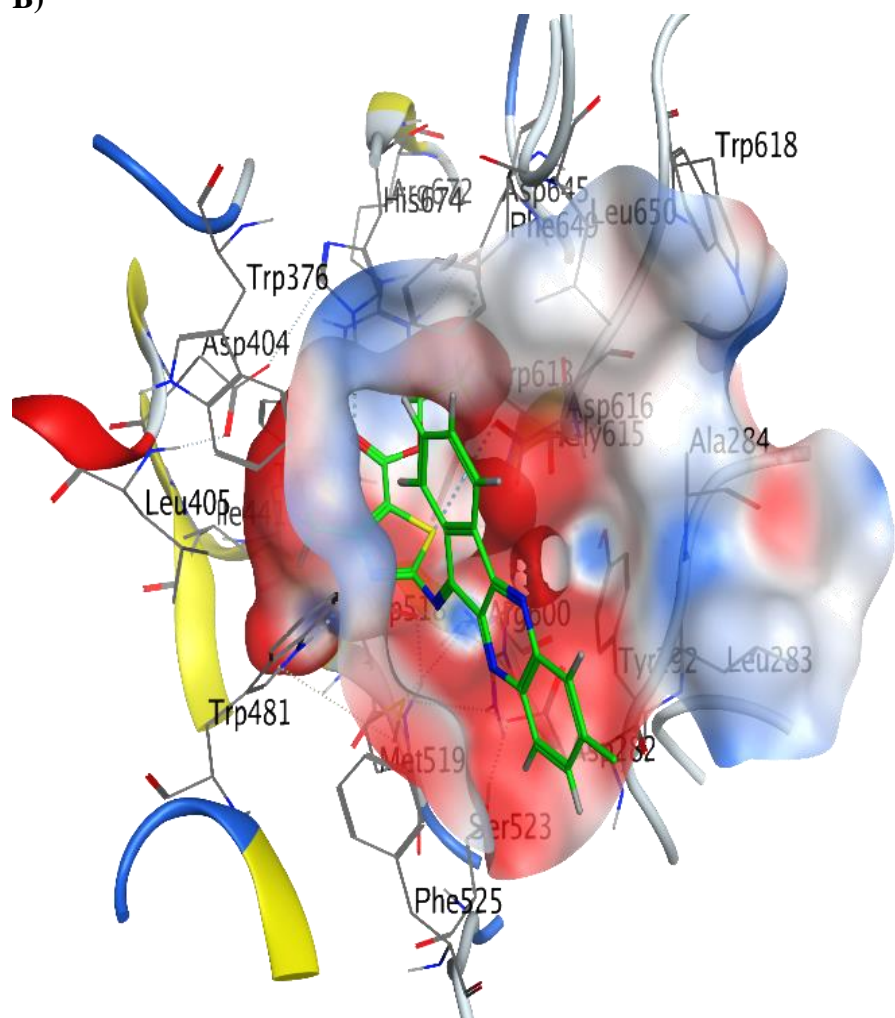


Figure 4: 3D binding modes & 2D ligand interactions with  $\alpha$ -amylase (PDB ID: 1B2Y) of the most active compounds, (A) Compound 4, (B) Compound 6, (C) Compound 8, (D) Compound 16

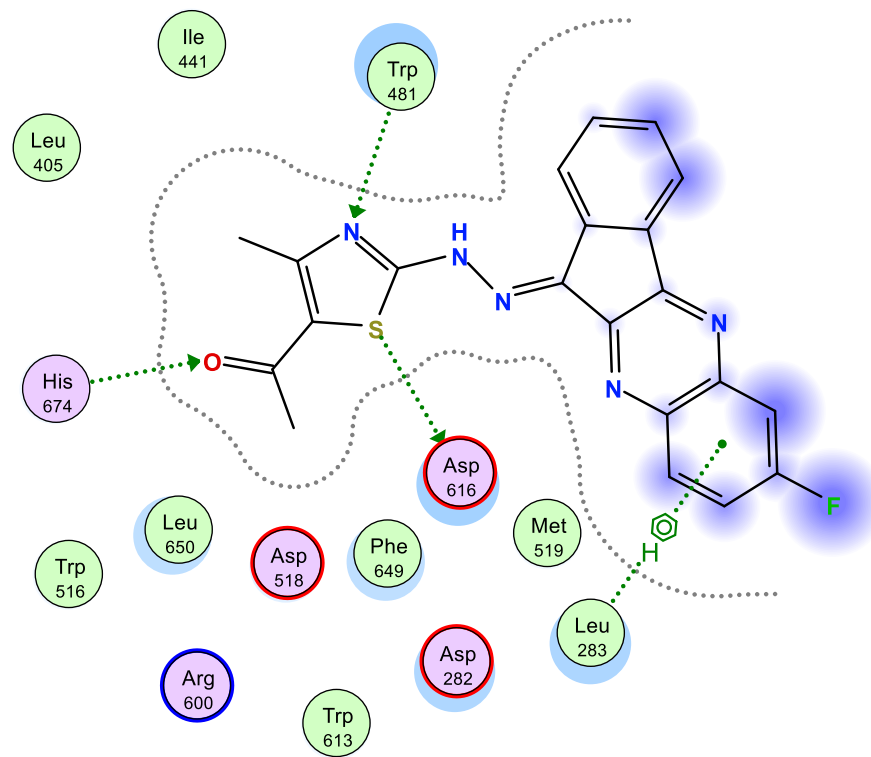
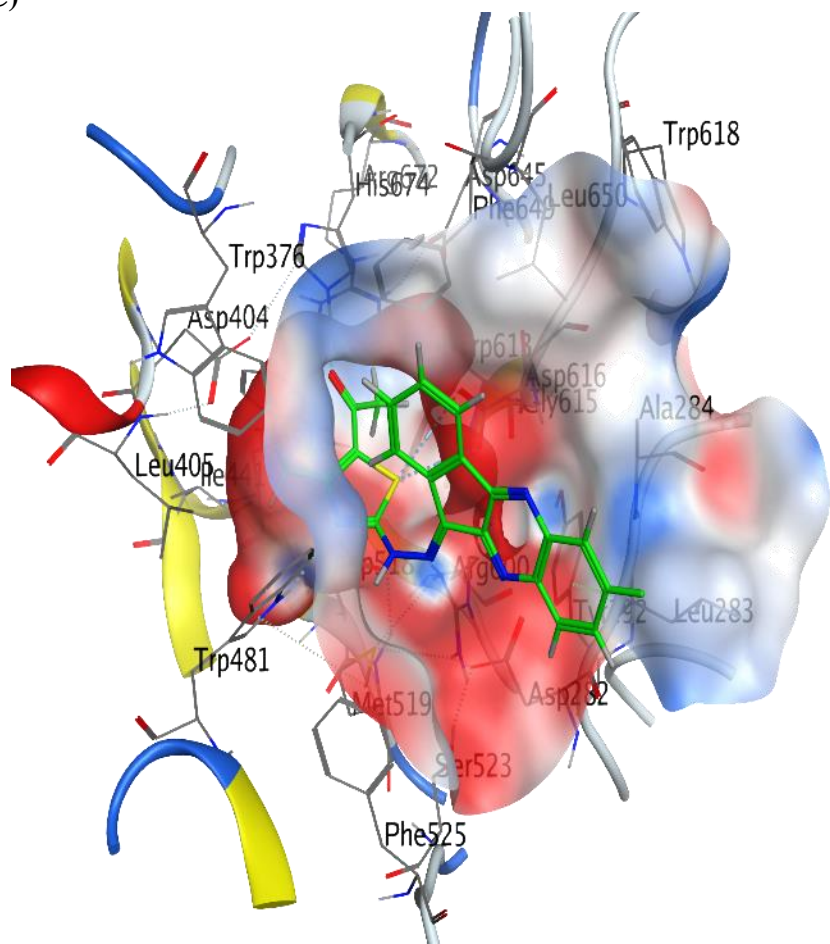
A)



B)



C)



D)

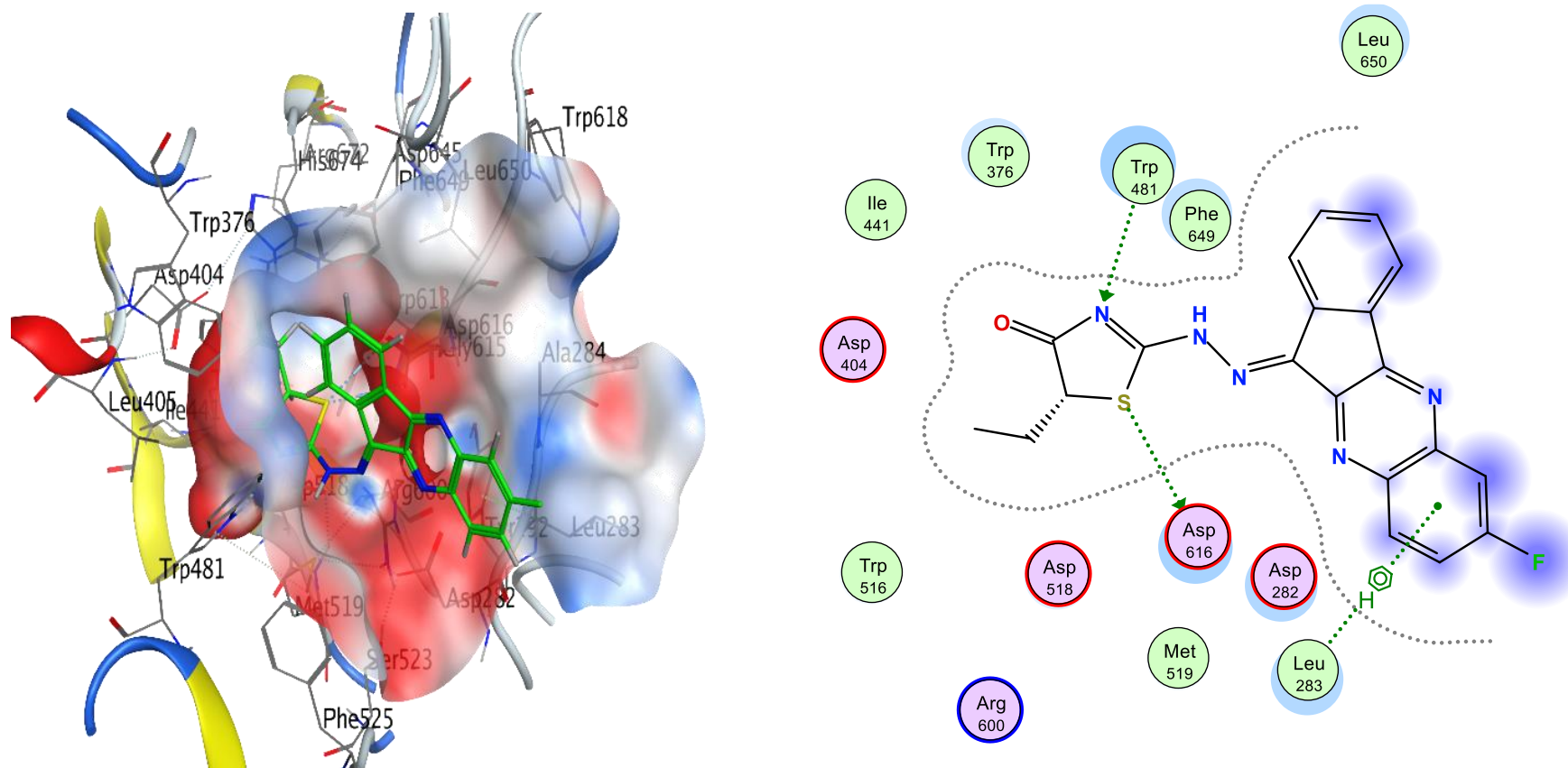


Figure 5: 3D binding modes & 2D ligand interactions with  $\alpha$ -glucosidase (PDB ID: 5NN8) of the most active compounds, (A) Compound 4, (B) Compound 6, (C) Compound 8, (D) Compound 16