



Supporting Information

Hydrogen-Bonding Interactions Trigger a Spin-Flip in Iron(III) Porphyrin Complexes**

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Supporting Information

Instrumentation.

UV-vis spectra were recorded on a PerkinElmer UV/vis spectrometer. Electron paramagnetic resonance (EPR) spectra were obtained on a Bruker EMX EPR spectrometer. Elemental (C, H, and N) analyses were performed on a Perkin-Elmer 2400II elemental analyzer. ¹H NMR spectra were recorded on a JEOL 500 MHz instrument. The spectra for paramagnetic molecules were recorded over a 100- kHz bandwidth with 64 K data points and a 5-ms 90° pulse. For a typical spectrum between 2000 and 3000 transients were accumulated with a 50-μs delay time. The residual ¹H resonances of the solvents were used as a secondary reference. ⁵⁷Fe Mössbauer spectra were recorded using a Wissel 1200 spectrometer and a proportional counter. ⁵⁷Co(Rh) in a constant acceleration mode was used as the radioactive source. Isomer shifts (δ) are given related to α -iron foil at room temperature. Magnetic susceptibility data were collected using a Quantum Design MPMS SQUID magnetometer over the temperature range 5 to 300K.

X-ray Structure Solution and Refinement.

Single-crystal X-ray data were collected at 100 K on a Bruker SMART APEX CCD diffractometer equipped with CRYO Industries low temperature apparatus and intensity data were collected using graphite-monochromated Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$). The data integration and reduction were processed with SAINT software.^[1] An absorption correction was applied.^[2] The structure was solved by the direct method using SHELXS-97 and was refined on F2 by full-matrix least-squares technique using the SHELXL-2014 program package.^[3] Non-hydrogen atoms were refined anisotropically. In the refinement, hydrogens were treated as riding atoms using SHELXL default parameters. Crystal data and data collection parameters are given in Table S1. X-ray structure of **1** contains severely disordered solvent molecules (twelve cyclohexane and eight water molecules in the unit cell) which could not be modeled properly due to weakly diffracting nature of the crystal. SQUEEZE^[4] routine of PLATON was therefore used to remove such unbound and highly disordered solvent molecules but the reported formula includes the solvents.

Computational Details.

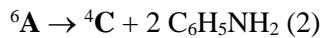
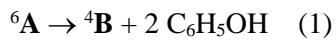
Calculations were initiated from the crystal structure coordinates reported in this paper. Complex **1** represents [Fe^{III}(TDMOETPP)Cl], TDMOETPP=3,5-Dimethyl-2,3,7,8,12,13,17,18-octaethyl-5,10,15,20-tetraphenylporphyrin. Hydrogen atoms were added to this complex to give the molecular formula C₆₈H₇₆N₄FeCl and a total atom count of 150. The second structure **2** uses [Fe^{III}(TDMOETPP)Cl] in complex with two molecules of phenol, acting as the H-bond donors in the system. After adding the hydrogen atoms to crystal

structure **2**, we were therefore left with a system that has the molecular formula C₈₀H₈₈N₄O₂FeCl and an atom count of 176. The crystal structure **4** has two aniline molecules in place of the phenol groups of structure **2** and therefore has the molecular formula C₈₆H₁₀₂N₆FeCl with a total atom count of 178. All structures were optimized using the Jaguar 7.9 software package.^[5] No constraints were placed on complex **1** with the H donor bonding distances being maintained for complexes **2** and **4**. Analytical frequencies were also carried out on all optimised structures using the Gaussian-09 software package.^[6] Free energies were taken from the Gaussian frequencies and contain, zero-point, thermal and entropic corrections to the energy at 298.15 K and 1 atm.

Initial geometry optimisations were carried out on the doublet, quartet and sextet spin states for each structure using both the unrestricted hybrid density functional method B3LYP,^[7] as well as the pure density functional method BP86.^[8] A basis set containing the 6-31G+ double- ζ basis set on all atoms except for the iron were a double- ζ LACVP basis set, that includes an effective core potential, was used (basis set BS1).^[9] Using the optimised structures for each spin state in complex **2** a geometry scan was run to increase the Cl---HOPh bond lengths from close to 2 Å to close to 3 Å, by moving the phenol molecules away in steps of 0.1 Å. All scan energies and geometries for the structure **2** complex were obtained using BS1 and energies were corrected using data from the vibrational frequencies. Additional gas phase, solvent corrected and dispersion corrected calculations were done with the triple- ζ LACV3P+ basis set on iron (with core potential) and 6-311+G* on the rest of the atoms (basis set BS2).

All three systems were also optimised using a slightly improved initial basis set (BS3) containing the polarised double- ζ basis 6-31G+* on all atoms except for the iron were the double- ζ basis with an effective core potential LACVP* was used. After crystal structure coordinate were optimized in BS3, using the same protocol as for BS1, a four basis set (BS4) were the triple- ζ LACV3P++** basis set on iron (with core potential) and 6-311++G** on the rest of the atoms was used. Single point gas phase, solvent corrected and dispersion corrected calculation were run applying the BS4 energies to BS3 optimisations.

The hydrogen bond strengths of the phenol and aniline bound complexes were determined from the reaction energies for equations 1 and 2 at UB3LYP/BS3 and include zero-point energies.



Experimental Section:

Materials:

Porphyrin free base 3,5-dimethyl-2,3,7,8,12,13,17,18-octaethyl-5,10,15,20-tetraphenylporphyrin was prepared from 3,5-dimethyl benzaldehyde and the diethyl pyrrole using a literature procedure.^[10] Reagents and solvents were purchased from commercial sources and purified by standard procedures before use.

Synthesis of **1**:

3,5-Dimethyl-2,3,7,8,12,13,17,18-octaethyl-5,10,15,20-tetraphenylporphyrin (100 mg, 0.105 mmol) was dissolved in 100 mL N, N-dimethylformamide. Excess FeCl₂ (66 mg, 0.51 mmol) was added and reflux for 2.5 hours under nitrogen. The resulting solution was then cooled to room temperature and transferred into a separatory funnel. 100 mL chloroform was added to the solution which was then washed well with 0.2(N) HCl (2×100 mL). Organic layer was separated and dried over anhydrous Na₂SO₄. The resulting solution was evaporated to dryness and purified by column chromatography using silica gel. The major fraction eluted with chloroform was collected and vacuum dried to obtain brown solid. Yield: 70 mg (64%). Anal. Calcd (found) for C₇₄H₉₀N₄OFeCl: C, 77.77 (78.00); H, 7.94 (8.22); N, 4.90 (5.11). UV-vis (benzene) [λ_{max} , nm (ϵ , M⁻¹ cm⁻¹)]: 443(2.4×10⁴), 401(2.2 ×10⁴), 578(2.0×10¹); ¹H NMR (CDCl₃, 295K): o-H, 10.72, 9.48; p-H, 8.21; H-CH₃(m), 3.81, 3.41; -CH₂-, 45.43, 37.65, 34.47 and 23.52 ppm. EPR data: in solid (77 K), g_⊥= 5.78 and g_{II}= 1.99; in toluene (77 K), g_⊥=5.75 and g_{II}=1.99.

Synthesis of 2:

Compound **1** (100 mg, 0.096 mmol) was dissolved in chloroform (10 mL); phenol (45 mg, 0.48 mmol) was added and the mixture was stirred for 15 min under nitrogen atmosphere. The resulting solution was then carefully layered with n-hexane and kept for slow diffusion. On standing for 6–8 days in air at room temperature, dark-purple crystals were formed, which was collected by filtration, washed thoroughly with the mother liquor, and dried under vacuum. Yield: 71 mg (60%). Anal. Calcd (found) for C₉₄H₁₀₂N₄O₄FeCl₇: C, 68.18 (68.40); H, 6.21 (6.43); N, 3.38 (3.57). UV-vis (benzene) [λ_{max} , nm (ϵ , M⁻¹ cm⁻¹)]: 440 (3.2×10⁴), 581 (2.2 ×10¹). ¹H NMR (CDCl₃, 295 K): o-H, 10.83, 10.29; p-H, 7.51; -CH₂-, 49.29, 24.93, 24.13 and 13.13 ppm. EPR data: in solid (77 K), g_⊥=4.16 and g_{II}=2.01; in toluene (77 K), g_⊥ = 4.18 and g_{II} = 2.00.

Synthesis of 3:

Compound **1** (100 mg, 0.096 mmol) was dissolved in chloroform (10 mL); 2-isopropyl aniline (64 mg, 0.48 mmol) was added and the mixture was stirred for 15 min under nitrogen atmosphere. The resulting solution was then carefully layered with n-hexane and kept for slow diffusion in air at room temperature. On standing for 6–8 days, dark-purple crystals were formed, which was collected by filtration, washed thoroughly with the mother liquor, and dried under vacuum. Yield: 70 mg (55%). Anal. Calcd (found) for C₉₂H₁₀₉N₇FeCl: C, 78.69 (78.44); H, 7.82 (8.05); N, 6.98 (6.74). UV-vis (benzene) [λ_{max} , nm (ϵ , M⁻¹ cm⁻¹)]: 437(2.1×10⁴). ¹H NMR (CDCl₃, 295 K): o-H, 11.01, 10.23; p-

H, 8.99; -*CH*₂-, 50.07, 25.07, 24.09 and 13.41 ppm. EPR data: in toluene (77 K), g_⊥ = 4.19 and g_{II} = 1.96.

References:

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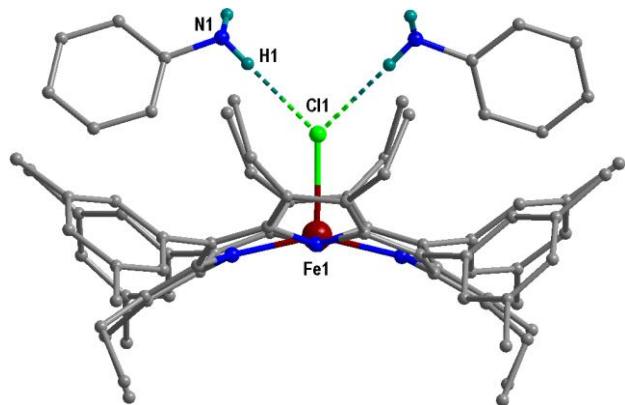


Figure S1. Molecular structures (at 100 K) of **4** (all the H atoms except for the aniline have been omitted for clarity). Selected bond distances (\AA) and angle (deg) for **4**: Fe1–N1, 1.997(5); Fe1–N2, 1.976(4); Fe1–Cl1, 2.312(3); N1–Fe1–N2, 88.74(19); N1–Fe1–Cl1, 103.18(14); N2–Fe1–Cl1, 95.77(15). H-bonding interactions: N1L…Cl1, 3.318(7); N2L…N1L, 3.147(9).

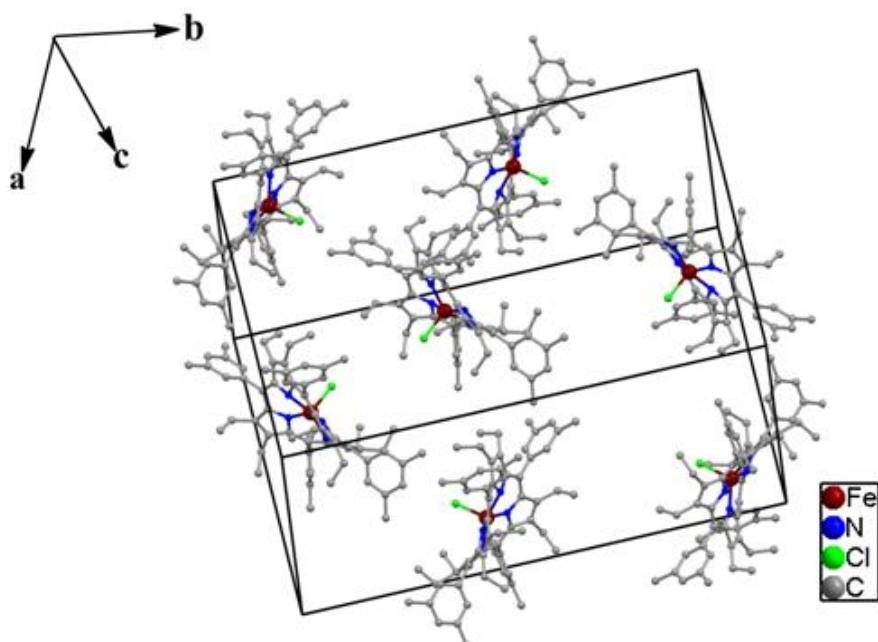


Figure S2. Diagram illustrating the packing of complex **1** in the unit cell (H atoms have been omitted for clarity).

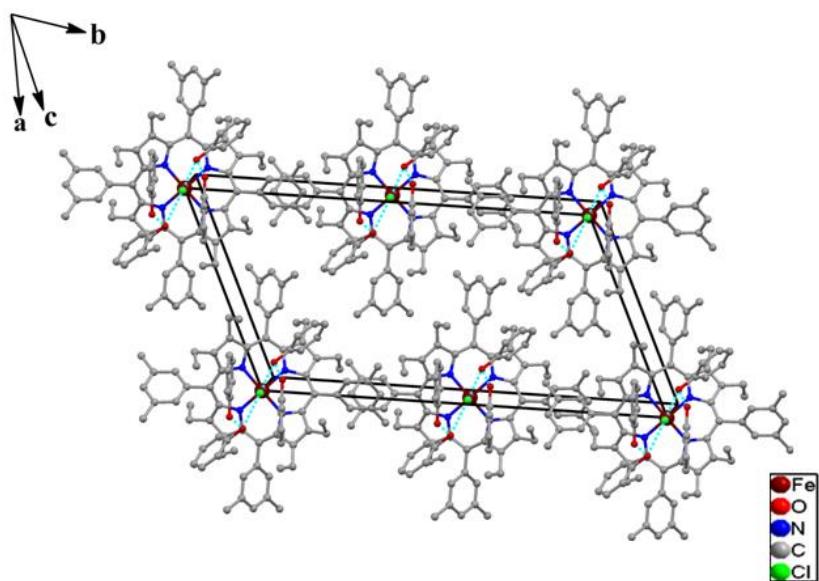


Figure S3. Diagram illustrating the packing of complex **2** in the unit cell (H atoms and CHCl₃ have been omitted for clarity).

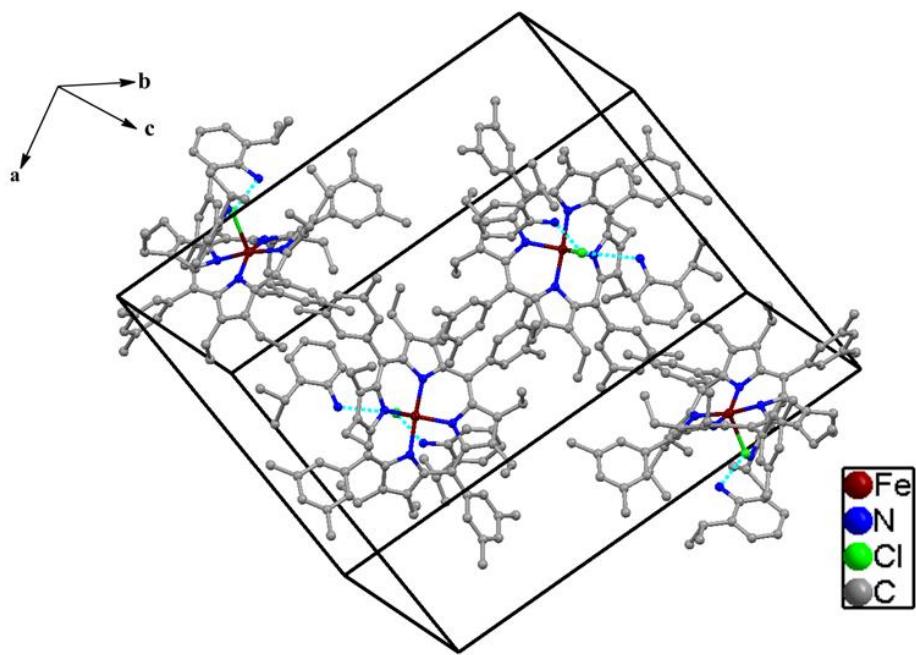


Figure S4. Diagram illustrating the packing of complex **3** in the unit cell (H atoms have been omitted for clarity)

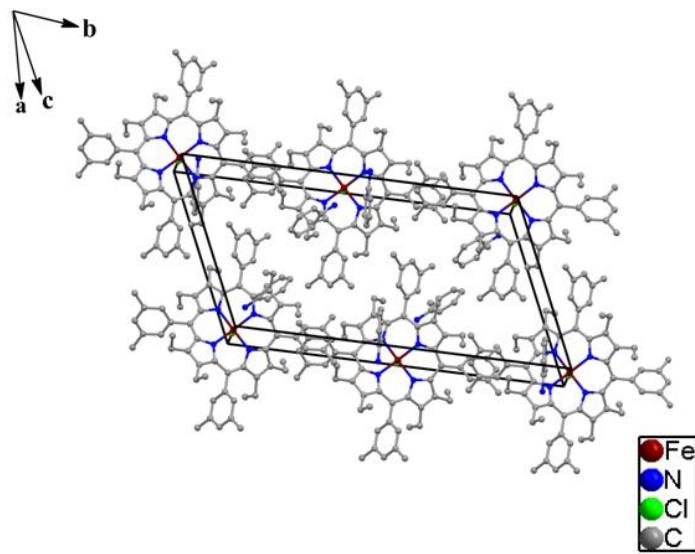


Figure S5. Diagram illustrating the packing of complex **4** in the unit cell (CH_2Cl_2 and H atoms have been omitted for clarity)

Table S1. Crystal data and data collection parameters.

	1	2	3	4
Formula	C ₇₇ H ₉₆ Cl N ₄ Fe O	C ₉₄ H ₁₀₂ Cl ₇ N ₄ O ₄ Fe	C ₉₂ H ₁₀₉ Cl Fe N ₇	C ₉₄ H ₁₀₈ Cl ₅ Fe N ₈
T, (K)	100(2)	100(2)	100(2)	100(2)
Formula weight	1184.87	1655.79	1404.16	1582.98
Crystal system	Orthorombic	Monoclinic	Monoclinic	Monoclinic
Space group	Pbca	C 2	P 21/c	C 2
a, Å	19.510(5)	26.129(5)	16.617(7)	26.066(5)
b, Å	24.761(5)	12.607(5)	19.957(9)	12.554(5)
c, Å	31.415(5)	13.960(5)	25.419(11)	13.936(5)
α, deg	90.000(5)	90.000(5)	90	90.000(5)
β, deg	90.000(5)	114.586(5)	108.342(9)	114.244(5)
γ, deg	90.000(5)	90.000(5)	90	90.000(5)
V, Å ³	15176(6)	4182(2)	8001(6)	4158(2)
Radiation (λ, Å)	Mo Kα (0.71073)	Mo Kα (0.71073)	Mo Kα (0.71073)	Mo Kα (0.71073)
Z	8	2	4	2
dcalcd, g.cm ⁻³	1.037	1.315	1.166	1.264
F(000)	5096	1742	3012	1678
μ, mm ⁻¹	0.275	0.459	0.272	0.394
No. of unique data	13322	6310	15658	7496
No. of parameters, refined	684	511	895	507
GOF on F ²	0.899	1.048	0.982	1.036
R1 ^a [I > 2σ(I)]	0.0734	0.0802	0.1114	0.0610
R1 ^a (all data)	0.1259	0.0914	0.3061	0.0702
wR2 ^b (all data)	0.2077	0.2332	0.3746	0.1732

$$^a R1 = \frac{\sum |F_O - F_C|}{\sum |F_O|}; \quad ^b wR2 = \sqrt{\frac{\sum [w(F_O^2 - F_C^2)]}{\sum [w(F_O^2)]}}$$

Table S2. Selected Bond Distance (\AA) and Angles ($^\circ$) for **1**, **2**, **3** and **4**.

	1	2	3	4
Fe(1)-N(1)	2.040(3)	1.997(6)	1.965(7)	1.997(5)
Fe(1)-N(2)	2.044(3)	1.960(5)	1.970(7)	1.976(4)
Fe(1)-N(3)	2.031(3)		1.937(7)	
Fe(1)-N(4)	2.046(3)		1.968(7)	
Fe(1)-Cl(1)	2.2287(12)	2.384(4)	2.357(3)	2.312(3)
N(1)-Fe(1)-N(2)	87.05(12)	89.5 (2)	89.2(3)	88.74(19)
N(1)-Fe(1)-N(4)	87.17(12)		88.5(3)	
N(2)-Fe(1)-N(4)	145.46(12)		155.7(3)	
N(3)-Fe(1)-N(1)	159.09(12)		169.9(3)	
N(3)-Fe(1)-N(2)	86.61(12)		87.9(3)	
N(3)-Fe(1)-N(4)	86.82(11)		90.2(3)	
N(1)-Fe(1)-Cl(1)	100.26(9)	102.3(2)	95.6(2)	103.18(14)
N(2)-Fe(1)-Cl(1)	107.27(9)	94.6(2)	100.8(2)	95.77(15)
N(3)-Fe(1)-Cl(1)	100.66(9)		94.4(2)	
N(4)-Fe(1)-Cl(1)	107.27(9)		103.5(2)	

Table S3. Selected structural parameters for $[\text{Fe}^{\text{III}}(\text{TDMOETPP})\text{Cl}]$, **1** and related species

Complex	Fe-N _p ^a	Fe-Cl	$\Delta_{\text{Fe}-\text{4N}}$ ^b	$\Delta_{\text{Fe}-24}$ ^c	Δ_{24} ^d	Ref
1	2.040(3)	2.2287(12)	0.49	0.52	0.52	tw
2	1.978(6)	2.384(4)	0.29	0.27	0.56	tw
3	1.960(7)	2.357(3)	0.29	0.30	0.56	tw
4	1.986(5)	2.312(3)	0.32	0.31	0.56	tw
$[\text{Fe}^{\text{III}}(\text{OMTPP})\text{Cl}]$	2.034(6)	2.24 (3)	0.46	0.51	0.51	9
$[\text{Fe}^{\text{III}}(\text{OETPP})\text{Cl}]$	2.031(5)	2.2418(23)	0.47	0.43	0.56	9
$[\text{Fe}^{\text{III}}(\text{detpp})\text{Cl}]$	2.075(4)	2.220(2)	0.491	0.54	0.14	10
$[\text{Fe}^{\text{III}}(\text{trans-tetpp})\text{Cl}]$	2.064(3)	2.219(1)	0.494	0.54	0.35	10
$[\text{Fe}^{\text{III}}(\text{cis-tetpp})\text{Cl}]$	2.065(1)	2.226(1)	0.477	0.52	0.34	10
$[\text{Fe}^{\text{III}}(\text{hetpp})\text{Cl}]$	2.053(2)	2.239(1)	0.490	0.52	0.45	10
$[\text{Fe}^{\text{III}}(\text{DPP})\text{Cl}]$	2.056(5)	2.238(2)	0.514	0.57	0.47	11
$[\text{Fe}^{\text{III}}(\text{tn-OEP})\text{Cl}]$	2.067(3)	2.1918(10)	0.46	0.47	0.41	12
$[\text{Fe}^{\text{III}}(\text{OEP})\text{Cl}]$	2.065(3)	2.243(13)	0.468	0.518	0.047	13
$[\text{Fe}^{\text{III}}(\text{TEtP})(\text{Cl})]$	2.053(3)	2.2644(13)	0.47	0.45	0.027	14

^aAverage value in Å. ^bDisplacement of iron from mean plane containing four porphyrinic nitrogen. ^cDisplacement of iron from mean plane containing C₂₀N₄ porphyrinic core. ^dAverage displacement of the 24 atoms from the least-squares plane of the porphyrin.

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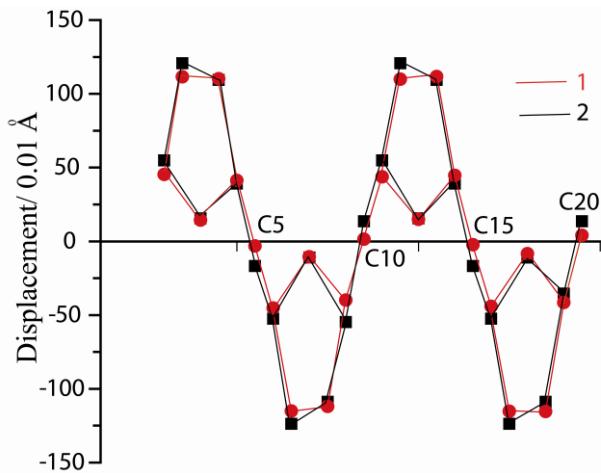


Figure S6. Atom deviations (in units of 0.01 Å) from the least-squares plane of the $C_{20}N_4$ porphyrinato core of **1** and **2**. The horizontal axis represents the atom number in the macrocycle showing the bond connectivity between atoms.

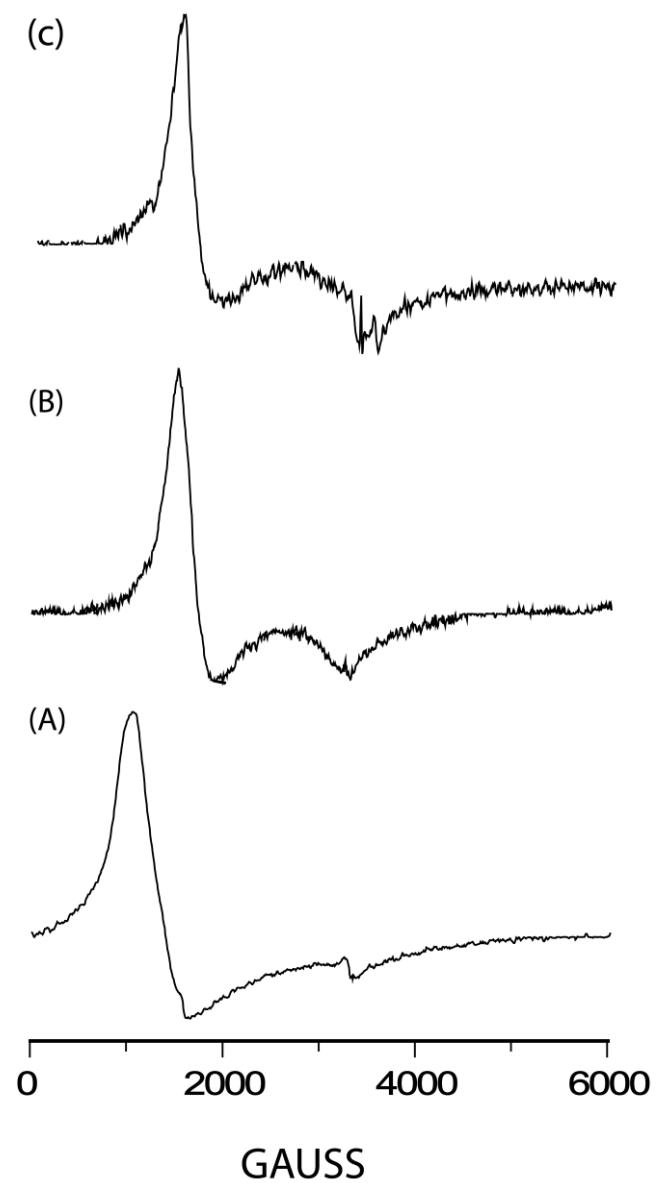


Figure S7. EPR spectra of (A) **1**, (B) **2** and (C) **3** in solid at 77 K.

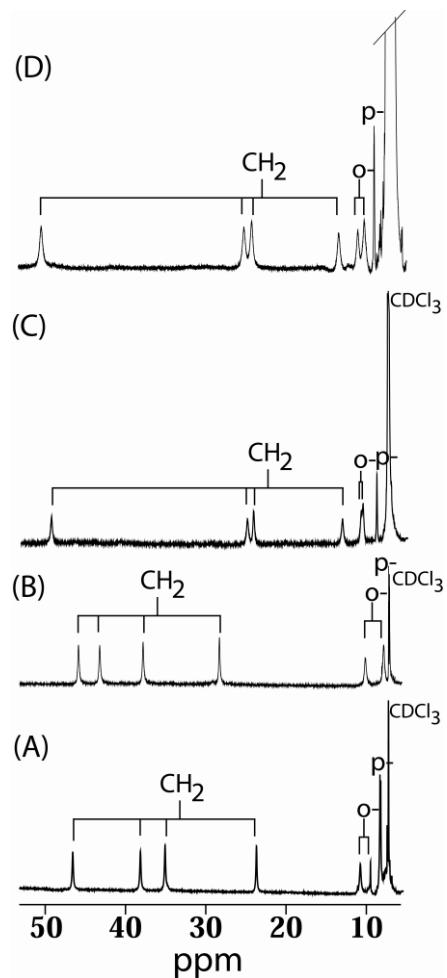


Figure S8. ^1H NMR (at 295 K in CDCl_3) spectra of (A) **1** and (B) five-coordinate phenolato iron(III)octaethyltetraarylporphyrin, (C) **2** and (D) **3**.

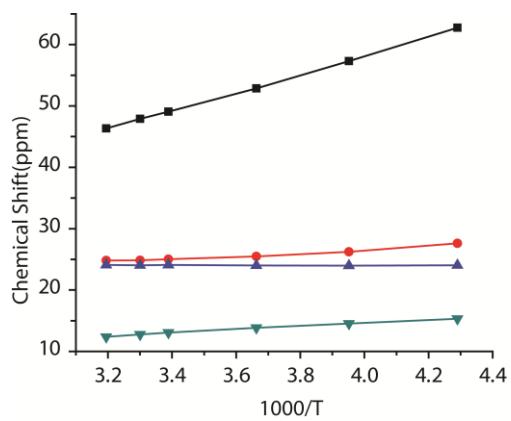


Figure S9. Curie plots (chemical shift versus $1000/\text{T}$) of the methylene proton signals for complex **2**.

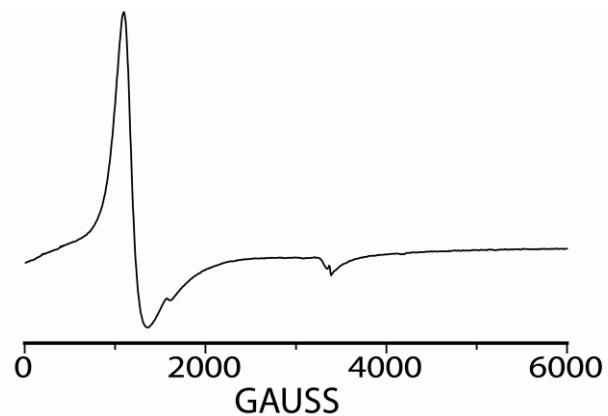
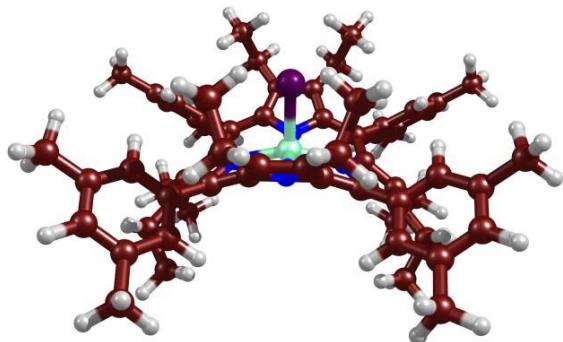


Figure S10. EPR spectrum of five-coordinate phenolato iron(III)octaethyltetraarylporphyrin in solid at 77 K. Simulation of the spectra gives the following values: $g_{\perp} = 5.91$ and $g_{\parallel} = 1.97$.



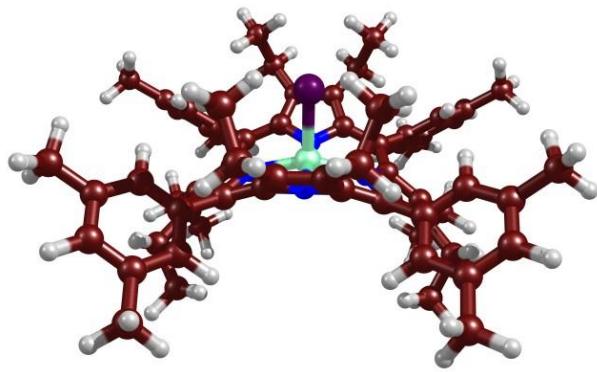
$$\text{Fe,Cl} = 2.344 \{2.405\} [2.347]$$

$$\text{Fe,N}_{\text{ave}} = 1.965 \{1.975\} [2.066]$$

$$\Delta^{\text{Fe}} = 0.266 \{0.306\} [0.273]$$

$$^2\text{A} \{^4\text{A}\} [{}^6\text{A}]$$

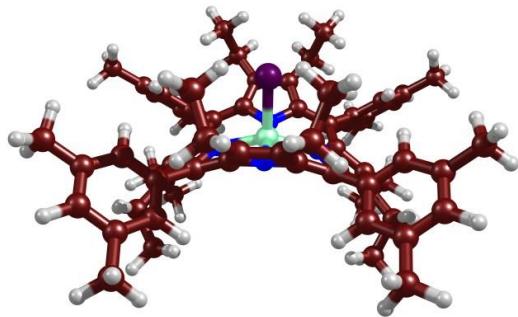
Figure S11. Bonding data of the complex **1** with no H bond donor calculated at B3LYP/BS1.



Fe,Cl = 2.290 {2.351} [2.294]
Fe,N_{ave} = 1.975 {1.994} [2.081]
 $\Delta^{\text{Fe}} = 0.273 \{0.326\} [0.513]$

²A {⁴A} [⁶A]

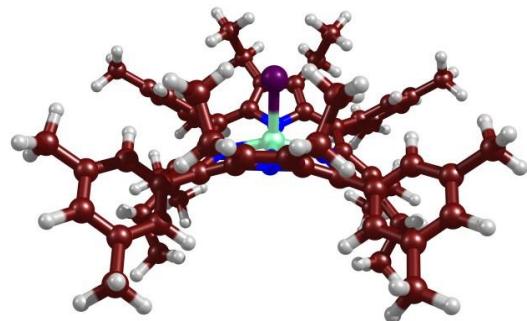
Figure S12. Bonding data of the complex **1** with no H bond donor calculated at UB3LYP/BS3.



Fe,Cl = 2.320 {2.399} [2.329]
Fe,N_{ave} = 1.951 {1.967} [2.081]
 $\Delta^{\text{Fe}} = 0.240 \{0.284\} [0.500]$

²A {⁴A} [⁶A]

Figure S13. Bonding data of the complex **1** with no H bond donor calculated at BP86/BS1.



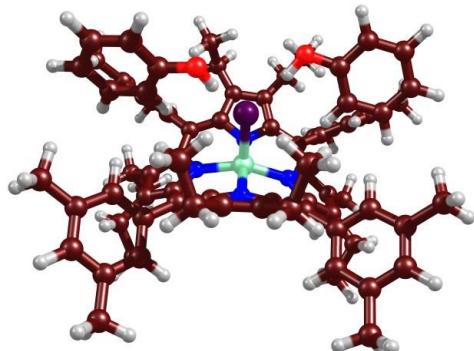
$$\text{Fe,Cl} = 2.262 \{2.371\} [2.292]$$

$$\text{Fe,N}_{\text{ave}} = 1.966 \{1.973\} [2.086]$$

$$\Delta^{\text{Fe}} = 0.243 \{0.277\} [0.496]$$

$${}^2\text{A} \{{}^4\text{A}\} [{}^6\text{A}]$$

Figure S14. Bonding data of the complex **1** with no H bond donor calculated at BP86/BS1.



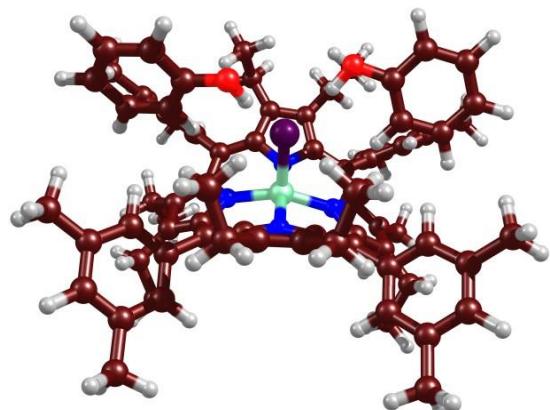
$$\text{Fe,Cl} = 2.450 \{2.511\} [2.438]$$

$$\text{Fe,N}_{\text{ave}} = 1.963 \{1.963\} [2.054]$$

$$\Delta^{\text{Fe}} = 0.268 \{0.259\} [0.426]$$

$${}^2\text{B} \{{}^4\text{B}\} [{}^6\text{B}]$$

Figure S15. Bonding data of the complex **2** with two phenol H Donor molecules calculated at UB3LYP/BS1.



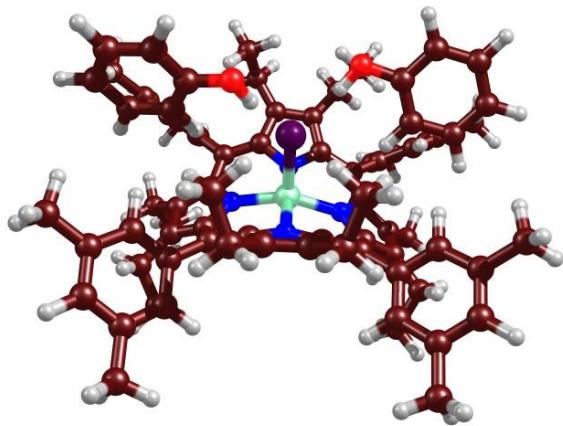
$$\text{Fe,Cl} = 2.554 \{2.485\} [2.392]$$

$$\text{Fe,N}_{\text{ave}} = 1.986 \{1.975\} [2.070]$$

$$\Delta^{\text{Fe}} = 0.193 \{0.253\} [0.452]$$

$${}^2\text{B} \{{}^4\text{B}\} [{}^6\text{B}]$$

Figure S16. Bonding data of the complex **2** with two phenol H Donors added as calculated at UB3LYP/BS3.



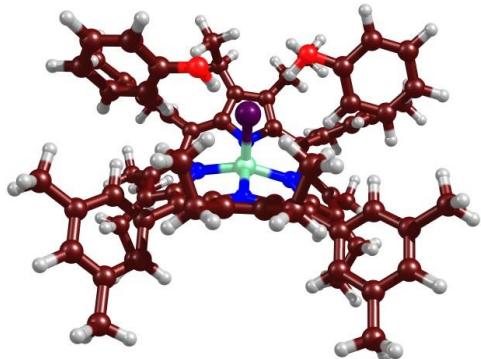
$$\text{Fe,Cl} = 2.420 \{2.523\} [2.440]$$

$$\text{Fe,N}_{\text{ave}} = 1.945 \{1.946\} [2.057]$$

$$\Delta^{\text{Fe}} = 0.247 \{0.239\} [0.411]$$

$${}^2\text{B} \{{}^4\text{B}\} [{}^6\text{B}]$$

Figure S17. Bonding data of the complex **2** with two hydrogen bonded phenol molecules as calculated at BP86/BS1.



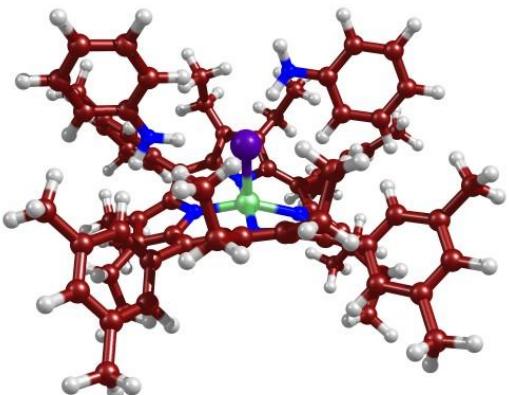
$$\text{Fe,Cl} = 2.375 \{2.471\} [2.398]$$

$$\text{Fe,N}_{\text{ave}} = 1.962 \{1.969\} [2.069]$$

$$\Delta^{\text{Fe}} = 0.249 \{0.233\} [0.448]$$

$${}^2\text{B} \{{}^4\text{B}\} [{}^6\text{B}]$$

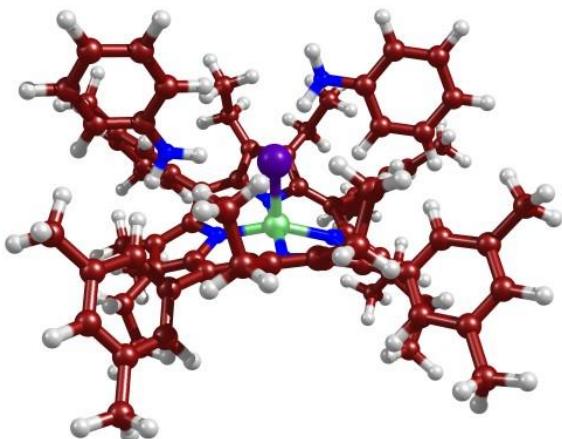
Figure S18. Bonding data of the complex **2** with two hydrogen bonded phenol molecules added, as calculated at UBP86/BS3.



Fe,Cl = 2.400 {2.472} [2.482]
Fe,N_{ave} = 1.966 {1.967} [1.975]
 $\Delta^{\text{Fe}} = 0.251 \{0.271\} [0.250]$

${}^2\text{A} \{{}^4\text{A}\} [{}^6\text{A}]$

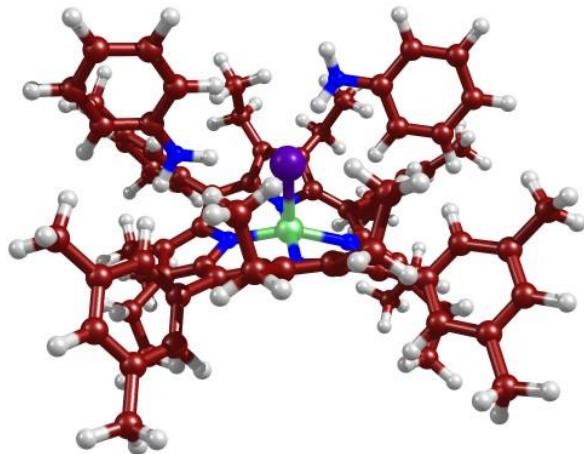
Figure S19. Bonding data of the complex **4** with two hydrogen bonded aniline molecules added, as calculated at UB3LYP/BS1.



Fe,Cl = 2.360 {2.440} [2.359]
Fe,N_{ave} = 1.978 {1.983} [2.071]
 $\Delta^{\text{Fe}} = 0.257 \{0.273\} [0.441]$

${}^2\text{A} \{{}^4\text{A}\} [{}^6\text{A}]$

Figure S20. Bonding data of the complex **4** with two hydrogen bonded aniline molecules added, as calculated at UB3LYP/BS3.



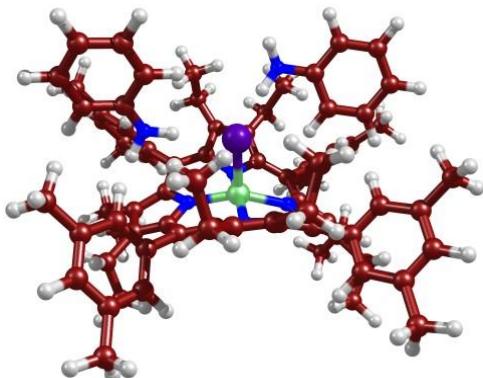
$$\text{Fe,Cl} = 2.372 \{2.474\} [2.508]$$

$$\text{Fe,N}_{\text{ave}} = 1.953 \{1.956\} [1.962]$$

$$\Delta^{\text{Fe}} = 0.240 \{0.242\} [0.229]$$

$${}^2\text{A} \{{}^4\text{A}\} [{}^6\text{A}]$$

Figure S21. Bonding data of the complex **4** with two hydrogen bonded aniline molecules added, as calculated at UBP86/BS1.



$$\text{Fe,Cl} = 2.327 \{2.448\} [2.469]$$

$$\text{Fe,N}_{\text{ave}} = 1.967 \{1.970\} [1.969]$$

$$\Delta^{\text{Fe}} = 0.237 \{0.237\} [0.219]$$

$${}^2\text{A} \{{}^4\text{A}\} [{}^6\text{A}]$$

Figure S22. Bonding data of the complex **4** with two hydrogen bonded aniline molecules added, as calculated at UBP86/BS3.

Table S4: Optimized geometries of $^{2,4,6}\mathbf{1}$, $^{2,4,6}\mathbf{2}$ and $^{2,4,6}\mathbf{4}$ using different computational methods and basis sets.

BS1_{OPT}				BP86			
B3LYP	Fe,Cl	Fe,N	Δ_{Fe}	BP86	Fe,Cl	Fe,Nave	Δ_{Fe}
$^2\mathbf{1}$	2.344	1.965	0.266	$^2\mathbf{1}$	2.320	1.951	0.240
$^4\mathbf{1}$	2.405	1.975	0.306	$^4\mathbf{1}$	2.400	1.967	0.284
$^6\mathbf{1}$	2.347	2.066	0.273	$^6\mathbf{1}$	2.329	2.081	0.500
$^2\mathbf{2}$	2.450	1.963	0.268	$^2\mathbf{2}$	2.420	1.950	0.247
$^4\mathbf{2}$	2.511	1.963	0.259	$^4\mathbf{2}$	2.523	1.953	0.239
$^6\mathbf{2}$	2.438	2.054	0.426	$^6\mathbf{2}$	2.440	2.057	0.411
$^2\mathbf{4}$	2.400	1.966	0.251	$^2\mathbf{4}$	2.372	1.953	0.240
$^4\mathbf{4}$	2.472	1.967	0.271	$^4\mathbf{4}$	2.474	1.956	0.242
$^6\mathbf{4}$	2.482	1.975	0.250	$^6\mathbf{4}$	2.508	1.962	0.229
BS3_{OPT}				BP86			
B3LYP	Fe,Cl	Fe,Nave	Δ_{Fe}	BP86	Fe,Cl	Fe,Nave	Δ_{Fe}
$^2\mathbf{1}$	2.289	1.975	0.273	$^2\mathbf{1}$	2.262	1.966	0.243
$^4\mathbf{1}$	2.351	1.994	0.326	$^4\mathbf{1}$	2.371	1.973	0.277
$^6\mathbf{1}$	2.294	2.081	0.513	$^6\mathbf{1}$	2.292	2.086	0.496
$^2\mathbf{2}$	2.554	1.986	0.193	$^2\mathbf{2}$	2.375	1.962	0.249
$^4\mathbf{2}$	2.485	1.975	0.253	$^4\mathbf{2}$	2.471	1.969	0.233
$^6\mathbf{2}$	2.392	2.070	0.452	$^6\mathbf{2}$	2.398	2.069	0.448
$^2\mathbf{4}$	2.360	1.978	0.257	$^2\mathbf{4}$	2.327	1.967	0.237
$^4\mathbf{4}$	2.440	1.983	0.273	$^4\mathbf{4}$	2.448	1.970	0.237
$^6\mathbf{4}$	2.359	2.071	0.441	$^6\mathbf{4}$	2.469	1.969	0.219

Table S5. Absolute energies B3LYP/BS1 optimised

	E BS1	Zero- point BS1	Gibbs BS1	E BS2	E_{solv} (Kcal/mol) BS2	$E_{\text{D}3}$ BS2
² 1	-3439.123	1.282	-3437.965	-3440.458	-16.110	-0.221
⁴ 1	-3439.141	1.282	-3437.978	-3440.477	-18.015	-0.221
⁶ 1	-3439.130	1.281	-3437.974	-3440.475	-16.727	-0.220
² 2	-4053.930	1.495	-4052.569	-4055.560	-20.126	-0.273
⁴ 2	-4053.952	1.495	-4052.594	-4055.584	-19.979	-0.271
⁶ 2	-4053.935	1.494	-4052.576	-4055.575	-20.364	-0.270
² 4	-4014.215	1.521	-4012.829	-4015.808	-18.666	-0.264
⁴ 4	-4014.235	1.521	-4012.856	-4015.829	-19.771	-0.264
⁶ 4	-4014.176	1.517	-4012.797	-4015.771	-18.264	-0.262

Table S6. Absolute energies B3LYP/BS3 optimised

	E BS3	zero BS3	Gibbs BS3	E BS4	E_{solv} BS4	$E_{\text{D}3}$ BS4
² 1	-3439.770	1.275	-3438.613	-3440.567	-15.887	-0.223
⁴ 1	-3439.787	1.275	-3438.629	-3440.585	-17.602	-0.223
⁶ 1	-3439.778	1.273	-3438.626	-3440.583	-15.034	-0.222
² 2	-4054.716	1.485	-4053.369	-4055.695	-14.182	-0.275
⁴ 2	-4054.741	1.487	-4053.389	-4055.726	-14.080	-0.273
⁶ 2	-4054.720	1.486	-4053.371	-4055.715	-18.244	-0.270
² 4	-4014.979	1.513	-4013.604	-4015.948	-15.120	-0.281
⁴ 4	-4014.999	1.513	-4013.623	-4015.970	-15.486	-0.280
⁶ 4	-4014.985	1.511	-4013.618	-4015.964	-16.036	-0.275

Table S7. Relative energies B3LYP/BS1 optimised

	ΔE [BS1]	ΔE_{ZPE} [BS1]	ΔE [BS2]	ΔE_{ZPE} [BS2]	ΔG [BS1]	ΔG [BS2]	ΔE_{S+ZPE} [BS2]	ΔG_S [BS2]	ΔE_{D3+ZPE} [BS2]	ΔG_{D3} [BS2]
² 1	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 1	-11.29	-11.36	-12.01	-12.08	-8.26	-8.98	-13.99	-10.88	-11.78	-8.68
⁶ 1	-4.43	-5.37	-10.35	-11.29	-5.72	-11.63	-11.90	-12.25	-10.60	-10.94
² 2	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 2	-14.04	-14.04	-14.62	-14.62	-15.29	-15.87	-14.48	-15.72	-13.72	-14.97
⁶ 2	-3.24	-4.20	-9.11	-10.08	-4.14	-10.01	-10.32	-10.25	-8.42	-8.36
² 4	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 4	-12.89	-12.57	-13.21	-12.89	-16.85	-17.17	-14.00	-18.27	-12.51	-16.78
⁶ 4	24.15	22.08	22.95	20.87	19.79	18.58	21.27	18.99	22.12	19.84

Table S8. Relative energies B3LYP/BS3 optimised

	ΔE [BS3]	ΔE_{ZPE} [BS3]	ΔE [BS4]	ΔE_{ZPE} [BS4]	ΔG [BS4]	ΔG [BS4]	ΔE_{S+ZPE} [BS4]	ΔG_S [BS4]	ΔE_{D3+ZPE} [BS4]	ΔG_{D3} [BS4]
² 1	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 1	-10.68	-10.69	-11.74	-11.76	-9.83	-10.89	-13.47	-12.61	-11.54	-10.67
⁶ 1	-5.26	-6.18	-10.49	-11.40	-8.05	-13.27	-10.55	-12.42	-11.08	-12.95
² 2	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 2	-15.19	-14.40	-19.53	-18.74	-12.31	-16.64	-18.64	-16.54	-17.37	-15.27
⁶ 2	-2.25	-2.00	-12.51	-12.26	-1.40	-11.66	-16.32	-15.72	-8.94	-8.34
² 4	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
⁴ 4	-12.71	-12.81	-14.13	-14.23	-11.51	-12.93	-14.59	-13.29	-14.12	-12.82
⁶ 4	-3.89	-5.07	-10.35	-11.52	-8.51	-14.96	-12.44	-15.88	-8.04	-11.48

Table S9. Group spin densities and charges obtained from the UB3LYP/BS2/Gas-phase calculations.

Spin	charge									
	Fe	Cl	lig	HD1	HD1	Fe	Cl	lig	HD1	HD1
² 1	1.22	-0.07	-0.14			0.18	-0.31	0.13		
⁴ 1	2.75	0.25	-0.01			0.20	-0.36	0.16		
⁶ 1	4.12	0.27	0.61			0.22	-0.27	0.05		
² 2	1.31	-0.07	-0.23	-0.01	0.00	0.43	-0.46	0.19	-0.08	-0.08
⁴ 2	2.82	0.15	0.02	0.01	0.01	0.44	-0.49	0.22	-0.09	-0.09
⁶ 2	4.13	0.16	0.69	0.01	0.01	0.51	-0.37	0.04	-0.08	-0.09
² 4	1.29	-0.08	-0.21	0.00	-0.01	0.35	-0.40	0.52	-0.43	-0.04
⁴ 4	2.80	0.17	0.01	0.01	0.01	0.37	-0.44	0.58	-0.45	-0.05

Table S10. Group spin densities and charges obtained from the UBP86/BS2/Gas-phase calculations.

Spin	charge										
	BP86	Fe	Cl	lig	HD1	HD1	Fe	Cl	lig	HD1	HD1
² 1		1.04	0.00	-0.04			-0.02	-0.26	0.28		
⁴ 1		2.58	0.31	0.10			0.01	-0.31	0.29		
⁶ 1		4.00	0.32	0.68			0.02	-0.23	0.21		
² 2		1.04	-0.03	-0.01	-0.01	0.00	0.23	-0.42	0.35	-0.08	-0.08
⁴ 2		2.62	0.16	0.20	0.01	0.01	0.25	-0.49	0.40	-0.08	-0.08
⁶ 2		4.02	0.17	0.79	0.01	0.01	0.35	-0.37	0.18	-0.07	-0.08
² 4		1.04	-0.02	-0.02	0.01	-0.01	0.13	-0.35	0.72	-0.44	-0.06
⁴ 4		2.61	0.20	0.18	0.01	0.01	0.19	-0.39	0.78	-0.51	-0.07

Table S11. Group spin densities and charges obtained from the UB3LYP/BS2/Solvent calculations.

Spin						Charge					
	Fe	Cl	lig	HD1	HD1	Fe	Cl	lig	HD1	HD1	
² 1	1.19	-0.06	-0.13						0.27	-0.48	0.21
⁴ 1	2.81	0.19	0.00						0.32	-0.54	0.22
⁶ 1	4.13	0.23	0.64						0.33	-0.42	0.10
² 2	1.25	-0.06	-0.18	-0.01	0.00	0.46	-0.48	0.21	-0.10	-0.09	
⁴ 2	2.85	0.13	0.01	0.01	0.01	0.50	-0.53	0.23	-0.10	-0.10	
⁶ 2	4.14	0.15	0.69	0.01	0.01	0.55	-0.39	0.05	-0.10	-0.11	
² 4	1.23	-0.07	-0.16	0.00	-0.01	0.40	-0.41	0.52	-0.46	-0.04	
⁴ 4	2.82	0.16	0.00	0.01	0.01	0.41	-0.46	0.59	-0.48	-0.06	

Table S12. Group spin densities and charges obtained from the UB3LYP/BS2/Solvent calculations.

Spin						Charge					
	Fe	Cl	lig	HD1	HD1	Fe	Cl	lig	HD1	HD1	
² 1	1.04	-0.01	-0.03						0.06	-0.42	0.36
⁴ 1	2.63	0.24	0.14						0.11	-0.51	0.39
⁶ 1	4.00	0.26	0.73						0.11	-0.38	0.27
² 2	1.04	-0.03	-0.01	-0.01	0.00	0.28	-0.46	0.38	-0.10	-0.10	
⁴ 2	2.64	0.14	0.20	0.01	0.01	0.30	-0.52	0.43	-0.10	-0.10	
⁶ 2	4.02	0.16	0.80	0.01	0.01	0.39	-0.40	0.20	-0.09	-0.10	
² 4	1.04	-0.02	-0.02	0.01	-0.01	0.15	-0.36	0.75	-0.48	-0.06	
⁴ 4	2.62	0.18	0.18	0.01	0.01	0.22	-0.41	0.81	-0.53	-0.09	

Table S13. Group spin densities and charges obtained from the UB3LYP/BS4/Gas-phase calculations.

Spin	charge									
	Fe	Cl	Lig	HD1	HD2	Fe	Cl	lig	HD1	HD2
² 1	1.21	-0.05	-0.16			0.24	-0.34	0.11		
⁴ 1	2.78	0.26	-0.05			0.23	-0.39	0.15		
⁶ 1	4.10	0.29	0.61			0.25	-0.29	0.04		
² 2	1.95	-0.02	-0.93	0.00	0.00	0.35	-0.46	0.23	-0.06	-0.05
⁴ 2	2.83	0.14	0.02	0.00	0.00	0.35	-0.42	0.19	-0.06	-0.07
⁶ 2	4.12	0.17	0.70	0.01	0.01	0.51	-0.31	-0.06	-0.07	-0.07
² 4	1.32	-0.07	-0.25	0.00	0.00	0.33	-0.34	0.34	-0.27	-0.05
⁴ 4	2.83	0.18	-0.02	0.01	0.00	0.35	-0.41	0.38	-0.28	-0.04
⁶ 4	4.12	0.20	0.67	0.01	0.00	0.46	-0.29	0.15	-0.27	-0.04

Table S14. Group spin densities and charges obtained from UB3LYP/BS4/Gas-phase calculations.

Spin Multiplicity	Charge									
	Fe	Cl	lig	HD1	HD2	Fe	Cl	lig	HD1	HD2
² 1	1.04	0.02	-0.06			0.06	-0.31	0.25		
⁴ 1	2.59	0.33	0.08			0.07	-0.33	0.25		
⁶ 1	3.99	0.34	0.67			0.09	-0.26	0.17		
² 2	1.07	-0.03	-0.04	0.00	0.00	0.16	-0.36	0.23	-0.02	0.00
⁴ 2	2.64	0.18	0.17	0.01	0.01	0.26	-0.44	0.27	-0.04	-0.05
⁶ 2	4.00	0.18	0.80	0.01	0.01	0.30	-0.34	0.14	-0.06	-0.04
² 4	1.06	-0.02	-0.04	0.00	0.00	0.16	-0.28	0.49	-0.30	-0.07
⁴ 4	2.63	0.21	0.14	0.01	0.01	0.17	-0.34	0.51	-0.30	-0.04

Table S15. Group spin densities and charges obtained from UB3LYP/BS4/Solvent calculations.

Spin						Charge					
	Fe	Cl	lig	HD1	HD2	Fe	Cl	lig	HD1	HD2	
² 1	1.19	-0.04	-0.15						0.33	-0.50	0.17
⁴ 1	2.83	0.21	-0.04						0.36	-0.56	0.21
⁶ 1	4.11	0.24	0.65						0.35	-0.42	0.07
² 2	1.94	-0.02	-0.92	0.00	0.00	0.41	-0.48	0.20	-0.07	-0.06	
⁴ 2	2.85	0.14	0.00	0.00	0.00	0.41	-0.43	0.14	-0.06	-0.07	
⁶ 2	4.13	0.16	0.70	0.01	0.01	0.55	-0.33	-0.05	-0.08	-0.09	
² 4	1.27	-0.06	-0.20	0.00	0.00	0.37	-0.36	0.37	-0.32	-0.06	
⁴ 4	2.84	0.18	-0.03	0.01	0.00	0.40	-0.43	0.40	-0.33	-0.04	
⁶ 4	4.14	0.20	0.65	0.01	0.00	0.50	-0.30	0.17	-0.32	-0.05	

Table S16. Group spin densities and charges obtained from UBP86/BS4/Solvent calculations.

Spin						Charge					
	Fe	Cl	lig	HD1	HD2	Fe	Cl	lig	HD1	HD2	
² 1	1.05	0.01	-0.06						0.15	-0.46	0.31
⁴ 1	2.62	0.26	0.12						0.17	-0.51	0.34
⁶ 1	3.99	0.29	0.73						0.18	-0.40	0.22
² 2	1.24	-0.03	-0.21	0.00	0.00	0.20	-0.37	0.22	-0.04	-0.02	
⁴ 2	2.66	0.15	0.17	0.01	0.01	0.32	-0.47	0.29	-0.06	-0.07	
⁶ 2	4.01	0.17	0.81	0.01	0.01	0.34	-0.36	0.14	-0.07	-0.05	
² 4	1.05	-0.02	-0.03	0.00	0.00	0.18	-0.29	0.52	-0.35	-0.07	
⁴ 4	2.64	0.20	0.14	0.01	0.01	0.23	-0.36	0.53	-0.35	-0.05	

Table S17. Bonding data for complex 2 throughout scan, whereby the Cl–HO distance is fixed.

²B3LYP	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}	²BP86	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}
scan01	2.436	2.284	1.964	0.261	scan01	2.41	2.28	1.95	0.25
scan02	2.435	2.384	1.964	0.261	scan02	2.41	2.38	1.95	0.25
scan03	2.428	2.484	1.966	0.259	scan03	2.40	2.48	1.95	0.25
scan04	2.419	2.584	1.965	0.258	scan04	2.39	2.58	1.95	0.25
scan05	2.417	2.684	1.964	0.260	scan05	2.38	2.68	1.95	0.25
scan06	2.410	2.784	1.964	0.259	scan06	2.38	2.78	1.95	0.25
scan07	2.405	2.884	1.966	0.258	scan07	2.38	2.88	1.95	0.25
scan08	2.399	2.984	1.966	0.257	scan08	2.38	2.98	1.95	0.25
⁴B3LYP	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}	⁴BP86	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}
scan01	2.50	2.28	1.96	0.27	scan01	2.51	2.28	1.95	0.24
scan02	2.49	2.38	1.96	0.27	scan02	2.50	2.38	1.95	0.25
scan03	2.49	2.48	1.97	0.27	scan03	2.50	2.48	1.96	0.25
scan04	2.48	2.58	1.97	0.27	scan04	2.49	2.58	1.96	0.25
scan05	2.48	2.68	1.97	0.28	scan05	2.48	2.68	1.96	0.25
scan06	2.47	2.78	1.97	0.28	scan06	2.48	2.78	1.96	0.25
scan07	2.47	2.88	1.97	0.28	scan07	2.47	2.88	1.96	0.26
scan08	2.46	2.98	1.97	0.28	scan08	2.47	2.98	1.96	0.26
⁶B3LYP	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}	⁶BP86	Fe,Cl	Fe,H _{Sub}	Fe,N _{ave}	Δ _{Fe}
scan01	2.430	2.284	2.055	0.429	scan01	2.430	2.281	2.058	0.417
scan02	2.424	2.384	2.056	0.432	scan02	2.424	2.381	2.061	0.428
scan03	2.419	2.484	2.057	0.442	scan03	2.418	2.481	2.063	0.428
scan04	2.413	2.584	2.057	0.437	scan04	2.413	2.581	2.065	0.427
scan05	2.409	2.684	2.058	0.444	scan05	2.407	2.681	2.064	0.432
scan06	2.404	2.784	2.058	0.443	scan06	2.405	2.781	2.067	0.432
scan07	2.401	2.884	2.060	0.444	scan07	2.400	2.881	2.068	0.433
scan08	2.397	2.984	2.061	0.443	scan08	2.395	2.981	2.068	0.429

Table S18. Orbital energies for $^4\text{6}\text{1}$, $^4\text{6}\text{2}$ and $^4\text{6}\text{4}$ (in au) as obtained from UB3LYP/BS3.

	$^6\text{1}$		$^6\text{2}$		$^6\text{4}$		$^4\text{1}$		$^4\text{2}$		$^4\text{4}$	
	alpha	beta										
δ_{xy}	-0.3103	-0.0938	-0.3865	-0.1062	-0.3866	-0.1035	-0.3027	-0.2421	-0.3120	-0.2561	-0.3127	-0.2568
σ^*_{z2}	-0.2680	-0.0895	-0.2888	-0.1152	-0.2819	-0.1066	-0.2106	-0.0642	-0.2668	-0.0871	-0.2637	-0.0848
π^*_{yz}	-0.2479	-0.0827	-0.2751	-0.0957	-0.2680	-0.0934	-0.2242	-0.0525	-0.2544	-0.0933	-0.2475	-0.0941
π^*_{xz}	-0.2476	-0.0964	-0.2756	-0.1101	-0.2700	-0.1084	-0.2221	-0.0466	-0.2597	-0.0985	-0.2502	-0.0980
σ^*_{x2-y2}	-0.2328	-0.0469	-0.2430	-0.0541	-0.2393	-0.0518	-0.0782	-0.0071	-0.0858	-0.0103	-0.0886	-0.0123
a_{1u}	-0.1780		-0.1885		-0.1860		-0.1786		-0.1841		-0.1883	
a_{2u}	-0.1701		-0.1823		-0.1779		-0.1753		-0.1820		-0.1840	

6	10.850729740	30.684954113	39.692860970	7	10.881045022	33.652250287	35.063894154	1	9.945091542	41.127132987	32.187051822
1	11.166515718	30.578718092	40.474306953	17	8.035700447	32.178639028	36.231281522	1	9.604865220	39.657288391	31.229220566
1	10.852387394	31.758298031	39.435765425	6	8.524037036	34.974006813	33.961136806	1	9.563240353	41.057668243	34.733618314
9	9.811962144	30.323796188	39.599278738	6	7.166091938	35.482185344	33.816830333	1	61{BP86}optBS1		
6	14.482106452	34.604048610	35.683847250	6	6.437698094	34.496680930	33.156002481	6	8.192455633	30.065308499	31.116458167
1	14.809902575	35.659501690	35.632036694	6	7.360296842	33.399666245	32.896580257	6	7.776428691	30.063766504	29.661526707
1	14.749375252	34.249343147	36.692202681	6	7.182457276	32.280282272	32.055438873	1	7.838992135	29.031304409	29.268769763
6	15.281041287	33.781316051	34.636939896	6	8.085397315	31.186946240	32.072427290	1	6.723855923	30.369170568	29.550001151
1	16.368186052	33.874789896	34.814983142	6	8.969550450	29.10947128	31.714607570	6	8.661135683	30.986142587	28.780867734
1	15.014221123	32.711087113	34.692416085	6	9.332580239	29.613400986	33.033827124	1	8.339323759	30.939695682	27.724103667
1	15.066246878	34.130580138	33.610435522	6	9.867252977	28.892322159	34.128765360	1	8.589439282	32.035685672	29.116458544
6	12.754794292	36.830184746	34.276370096	6	10.368018542	29.549930517	35.274889464	6	9.727136138	30.686470328	28.833603531
1	12.089586228	37.673605601	34.522591310	6	10.727272141	28.979350210	36.566508591	6	9.510748588	27.842512485	31.043467324
1	13.727102887	37.055384534	37.547533388	6	11.427412404	29.964509634	37.257508462	1	8.888664861	27.622443027	30.155181833
6	12.953574423	36.782881037	32.7367145585	6	11.496433609	31.122423506	37.3676325154	6	10.986010747	28.007505367	30.589928709
1	13.371064765	37.738478595	32.367806014	6	12.308947951	32.271881434	36.496234306	1	11.137441457	27.096694067	30.078290556
1	13.644304851	35.969461537	32.449631400	6	12.102701982	33.415198711	35.685895018	1	11.096283646	28.863834143	29.900687968
1	11.991615151	36.606450570	32.222182110	6	12.973085897	34.573283661	35.520296728	6	6.001221826	32.215478105	31.071252224
1	5.170443610	30.436645200	32.087142163	6	12.213625572	35.557893538	34.896319361	6	5.884006018	33.134882790	30.004007217
6	9.886155513	27.374198633	34.092550717	6	10.885502802	34.991748749	34.689893051	1	6.650343493	33.908225986	29.880045957
6	6.8862079744	26.635542780	34.042016828	6	9.687582818	35.671252406	34.357014185	6	4.801515683	33.063855848	29.092609448
1	7.727768819	27.173998041	34.032935243	6	6.576074578	36.738579639	34.414741402	6	3.828181928	32.057340119	28.22297196
6	8.695876797	25.222225200	34.004288468	1	5.619870830	36.963634248	33.906860247	1	2.985598232	31.993514448	28.582013850
6	11.164711428	25.271080379	34.054706455	1	7.227837078	37.608543134	34.240047369	6	3.912582387	31.128752453	30.348121650
6	11.120743786	26.680873262	34.099905571	6	6.312683003	36.603869490	35.933008739	1	1.096283646	28.863834143	29.900687968
1	12.052748930	27.254408059	34.143379322	1	5.856476486	37.529843469	36.329376547	6	5.008490689	31.216481328	31.230886489
1	13.293851012	25.124960732	33.610573350	1	5.633034513	35.760703020	36.146869099	6	4.679275623	34.067047577	27.961835367
6	13.425309019	32.276785894	37.517829834	1	7.252233860	36.418419815	36.481367353	1	5.665388373	34.302317614	27.522630705
6	13.421365311	33.194203986	38.592697822	6	4.944004630	34.52812647	32.9229707630	1	4.027803305	33.687239387	27.155389421
1	12.582795101	33.891911759	36.943968628	1	4.590640731	35.572893493	33.006162622	6	1.424428654	35.021945595	28.316037371
6	14.477250297	33.219212634	39.532972478	1	4.686924270	34.193453591	35.911864715	6	2.846742494	30.060745003	30.526779820
6	15.543941506	32.307399613	39.375098832	6	4.170077600	33.663610867	33.952029069	1	1.843611097	30.509509112	30.654633630
1	6.3686167182	32.315951974	40.09970869	1	3.081606880	33.742759267	33.777561429	1	2.792933645	29.393101798	29.645951628
6	15.736490906	31.374450495	38.310492496	1	4.457157979	32.601576340	33.868814111	1	3.050118199	29.43427186	31.412120635
6	14.503255442	31.3655474081	37.392602041	1	4.382418340	33.98390967	34.986835567	6	9.956511188	24.529114361	34.000910812
1	4.497427874	30.646347749	36.566981670	1	9.369278332	26.976956153	31.699278615	1	9.979720497	23.433362119	33.954748522
6	16.733826761	30.404200865	38.167577980	1	11.606579461	28.134970311	31.427444862	6	12.505588400	24.510735251	34.018007006
1	16.877238180	29.809492684	39.089220418	6	10.290315697	27.656303192	37.151134758	1	12.862830232	24.316681822	35.048250618
1	16.556840934	29.700359412	37.334268604	1	10.430051329	26.835509777	34.039019117	6	1.124672621	23.353394358	33.509299574
1	17.684266553	30.937009650	37.974230853	1	10.933605155	27.417017426	38.018204160	6	7.407477850	24.402749514	34.003257232
6	14.470442808	34.223706808	40.6726178740	6	8.814993402	27.672610405	37.615679369	1	7.584434492	23.325203584	34.161545826
1	13.448919137	34.3875763968	40.059353035	1	8.539928810	26.699124916	38.062178247	1	7.90165492	32.159846648	36.402690533
6	15.103389665	33.886724785	41.511982775	1	8.641889267	28.462507185	38.367143900	6	7.9827220087	32.25720078	34.645295644
1	14.856066486	35.208145659	40.423219370	1	8.136557791	27.869113711	36.767883242	6	6.838051295	33.714589144	33.445289385
6	9.639154629	37.170050607	44.565069039	6	11.855230070	29.872123425	38.704050671	7	8.806055126	30.869434132	33.221659928
6	9.398835394	37.948909374	33.299001925	1	11.882076059	28.807941084	33.00919117	7	10.810310730	30.831516693	35.254235386
1	9.223396003	37.442134532	32.343243630	1	12.877894417	30.255439118	38.845131459	7	10.849472640	33.690737325	35.155024865
6	9.378523609	39.359291813	33.360709673	1	8.079831984	30.630685883	33.501269896	1	7.90165492	32.159846648	36.402690533
6	9.602364107	39.985003735	36.093937375	1	8.030934093	30.222495509	30.512698690	6	8.482441959	35.026481435	33.933148987
6	9.841396904	39.234500102	35.783001670	1	8.0782386636	31.708061040	39.415649080	6	7.108272182	35.524555034	33.780136717
6	9.857970955	37.824760746	35.691251201	1	9.864233584	30.253660811	39.562617965	6	6.389605955	34.532854487	33.102975456
1	10.04376620	37.221173579	36.586919095	1	14.455803047	34.661297445	35.800391311	6	7.319779504	33.428115378	32.839795921
6	6.668774519	24.788196062	31.212165138	1	14.763152639	35.723213191	37.458915745	6	7.156206061	32.275639195	32.021591565
6	10.103446868	39.930736219	37.107946899	1	14.694964700	34.287875906	36.806861926	6	8.037465836	31.156454032	32.086614436
1	11.176952539	40.178254543	37.225387066	1	15.301294936	33.884463944	34.761544153	6	8.968576278	29.076019701	31.732959977
1	9.539450330	40.877209670	37.185278092	1	16.379642784	33.994837767	34.981953310	6	9.289344269	29.5603636143	33.081526971
1	9.818274622	39.292159822	37.961983623	1	15.056339222	32.808140183	34.785221892	6	9.864053160	28.841680465	34.168314729
6	9.091771330	40.189171631	32.121265138	1	15.115115632	34.253541548	34.740563134	6	6.10416545454	29.487844778	35.307475543
1	8.034428077	40.517847048	32.096008367	1	12.744953489	36.878976000	34.390116497	6	10.791307059	28.927403973	36.612804226
1	9.716603421	41.100092690	32.090370210	1	12.058945066	37.705163401	34.6347223440	6	11.507031481	29.920734440	37.292823462
1	9.280539630	39.613902993	31.198408937	1	13.696284983	37.105986513	34.905814701	6	11.571410694	31.087005993	36.402283590
1	9.588160161	41.080669495	34.669274730	1	12.995785844	36.871847763	32.886636031	6	12.352193539	32.269294011	36.513579432
1	13.707508491	36.075283111	32.581072086	1	13.707508491	36.075283111	32.581072086	6	12.974888767	34.589457966	35.506923259
6	12.056061336	36.695547574	32.312460278	1</td							

1	6.856309000	3.450098000	17.271099000	6	7.045701690	6.529169777	7.843995549	6	12.147955082	7.037013594	12.496915848
1	5.496495000	2.542495000	16.569060600	6	6.186391406	5.571506151	7.267983343	6	12.982027654	6.242692797	13.309641515
6	9.584789000	8.657862000	17.151649000	1	5.600781720	4.917299772	7.925686218	1	12.523527169	5.545656973	14.022010214
1	10.641732000	8.900570000	17.358903000	6	6.080697314	5.438300131	5.870863517	6	14.383159886	6.324445087	13.211781416
1	9.224977000	8.098072000	18.029283000	6	6.849366550	6.293878106	5.056335786	6	14.937600374	7.243781315	12.299016614
6	8.774035000	9.975834000	17.029364000	1	6.780309582	6.194361789	3.964177720	1	16.030762788	7.320972269	12.217570149
1	8.873132000	10.578914000	17.950440000	6	7.708295925	7.267929834	5.601470549	6	0.350277910	7.954080424	13.805826911
1	7.701776000	9.764553000	16.871089000	6	7.804223453	7.367109365	7.002837587	6	1.755341647	7.864331284	13.757644906
1	9.128793000	10.583767000	17.166787000	1	8.486224251	8.099348490	7.452841974	1	2.366036510	8.520728625	14.390557592
6	11.844161000	8.450709000	14.978546000	6	5.149700119	4.411854839	5.255354947	6	15.276129549	5.425989121	14.044635050
1	12.664602000	7.797412000	14.642958000	1	5.563245681	4.000592777	4.316945197	1	14.739541220	5.012050460	14.913740695
1	12.070798000	8.712543000	16.029111000	1	4.165859885	4.858119512	5.010213492	1	16.164179751	5.967579615	14.418897630
6	11.853849000	9.749494000	14.128640000	1	4.965114486	3.569049372	5.944294983	1	15.64527772	4.568833581	13.448891243
1	12.832439000	10.257757000	14.207361000	6	8.502861118	8.194665793	4.702480531	6	-0.331080191	8.982287570	14.687759242
1	11.07980000	10.451912000	14.466995000	1	9.483508140	8.445423466	5.14121488	1	1.340340992	8.651213331	14.990174910
1	11.668432000	9.524008000	13.063643000	1	7.967267446	9.149490196	4.536405033	1	0.447576139	9.949582511	14.160945028
6	7.0306682000	6.583458000	16.723389000	1	8.680187302	7.742391857	3.710948916	1	0.252696766	9.179805990	15.604395461
6	8.222324000	5.748201000	18.307126000	6	2.396527310	6.941477159	12.907480051	8	5.272735923	1.787156638	13.879080066
1	8.875199000	5.083765000	17.731642000	6	1.612477615	6.089936592	12.1019988720	6	5.907928310	0.688064118	14.400552731
6	8.282764000	5.739517000	19.717836000	1	2.113851387	5.379132694	11.433442571	6	7.310351798	0.584702362	14.490863589
7	7.420075000	6.599131300	20.465525000	6	0.207016833	6.137601306	12.148181340	1	7.944650430	1.391896487	14.105513850
1	7.474555000	6.5999696000	21.523981000	6	0.404316269	7.07192810	13.000872530	6	7.884051343	-0.572838633	15.037279957
6	6.503051000	7.450929000	17.779949000	1	-1.501389278	7.131018093	13.039289314	1	8.976560014	-0.650714998	15.093024388
6	6.448769000	7.425186000	18.368139000	6	14.131758200	8.063615360	11.484618421	6	7.077994317	-1.625454392	15.499141396
1	5.732528000	8.063254000	17.838929000	6	12.732591816	7.940247360	11.584788998	1	7.533558020	-2.527474625	15.923423202
6	9.258906000	4.831438000	20.449416000	1	12.082264984	8.549160846	10.944454722	6	5.680120247	-1.513931059	15.401187830
1	8.772615000	4.314826000	21.292330000	6	-0.6376670124	5.184181611	11.321968585	1	5.037710416	-2.330712800	15.75112501
1	10.109762000	5.406659000	20.859759000	1	-0.0586668377	4.763479720	10.478951607	6	5.091578204	-0.365488994	14.856698992
1	9.671449000	4.064374000	19.768015000	1	-1.529715869	5.683898599	10.109686861	1	4.003716377	-0.264414061	14.772499308
6	5.609427000	8.385149000	20.576630000	1	-0.988940670	4.332566805	11.934339672	1	5.933658864	2.492700101	13.658572590
4²{BP86}_{optBS1}											
7	5.915141221				6.652517469			11.337060120			
7	8.639893439				6.265985013			11.18347249			
6	6.172157061				7.042824810			10.006742130			
6	5.03016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
7	7.077994317				1.625454392			15.499141396			
7	8.639893439				6.265985013			11.18347249			
6	6.172157061				7.042824810			10.006742130			
6	5.03016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
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6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
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6	5.031016192				7.800397232			9.500513672			
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6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
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6	5.031016192				7.800397232			9.500513672			
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6	5.031016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
6	4.559503592				6.942323031			11.584192471			
6	5.031016192				7.800397232			9.500513672			
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6	5.031016192				7.800397232			9.500513672			
6	4.024759782				7.727345646			10.475662290			
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6	5.031016192				7.800397232			9.500513672			
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6	4.559503592				6.94232303						

1	14.746231081	8.021817699	10.102734935	7	5.904736326	6.725195475	11.318631635	1	3.730340804	2.487363468	14.842543376
8	6.537625687	1.409858685	10.579907095	7	8.660095493	6.399073613	11.308666082	1	3.141382167	3.175172142	13.309470097
6	5.486134685	1.754115197	9.744907874	6	6.148005054	7.080042268	9.990783542	6	4.903748846	4.659259181	17.308862813
4	4.879017064	3.034290051	9.752697197	6	4.982695725	7.762136226	9.444237766	1	3.839126757	4.465481139	17.535613500
1	5.231365912	3.797867084	10.453690750	6	3.981989437	7.684018788	10.413275122	1	5.275418279	5.300236049	18.121889012
6	3.829151338	3.308593212	8.852744651	6	4.553071133	6.975119135	11.549523774	6	5.670533533	3.316074856	17.334441487
1	3.371012132	4.304618415	8.850540694	6	3.871477046	6.431018190	12.666934289	1	5.538662139	2.818322709	18.312462090
6	3.372466288	2.324238847	9.795196784	6	9.977174944	6.008360280	11.562910328	1	6.751117581	3.468253851	17.173200062
1	2.554297384	2.543140407	7.258026146	6	10.511871769	5.292153556	10.412418468	1	5.313122713	2.629538672	16.547711990
6	3.980015810	1.049537161	7.960043866	6	9.545697492	5.368652098	9.409135759	6	9.633090518	8.579410170	17.211757052
1	3.6332281831	0.273728081	7.269406582	6	8.421467671	6.110035756	9.9647353616	1	10.698496872	8.767446189	17.437960630
6	5.030090892	0.761648285	8.846992892	6	7.286872973	6.615374450	9.290871559	1	9.234433533	8.016034674	18.068385232
1	5.513837737	-0.218402023	8.866309467	6	11.778997008	4.471916864	10.322645042	6	8.898668927	9.936784756	17.114347863
1	6.782190816	2.161502310	11.198688738	1	12.634045918	5.004123233	10.764767840	1	9.016508756	10.510280468	18.052061762
26	7.261743384	6.215577779	12.691816114	1	12.030508253	4.320800968	9.256742761	1	7.818782554	9.790151109	16.937060498
17	7.252694833	3.692242148	12.683615864	6	11.640932401	3.088441281	11.002313820	6	9.299558434	10.548744664	16.284530037
7	8.610437366	6.636061616	14.050350929	1	12.580380273	2.515054658	10.89851865	6	11.895406974	8.363725045	15.068865466
5	5.882499985	6.261638481	14.064408928	1	10.824936436	2.497899799	10.550796151	1	12.693810210	7.692957562	14.717709441
6	8.352597233	7.022115285	15.382388887	1	11.419629639	3.192431176	12.078372745	1	12.115995925	8.569623499	16.132176671
6	9.495014894	7.776184391	15.892123942	6	9.642385768	4.666580499	10.873803624	6	11.965661597	6.961453173	14.279859589
10	5.10298421	7.704067051	9.417725410	1	10.709196875	4.487755083	7.84482639	1	12.96105898	10.159121428	14.392848148
6	9.969620005	6.925416072	13.804643621	1	9.260473000	5.302320063	7.261624216	1	11.205934277	10.409904987	14.636750337
6	10.663548394	3.6556601048	12.708228760	6	8.894774262	3.317295648	8.049622016	1	11.787863908	9.521861120	13.203216535
6	4.550136534	8.879206248	13.806895958	1	9.025590834	2.820231903	7.068849066	6	7.276577733	6.619488169	17.586905813
4	4.03460779	5.129422115	14.949725762	1	7.81302690	3.448996155	8.22071141	6	8.121919969	5.724556701	18.273247731
6	5.010811323	5.193372284	15.955770527	1	9.269150610	2.628970164	8.830864504	1	8.744240484	5.027010011	17.699007192
6	6.126443344	5.969437482	14.197529292	6	4.884244580	8.554362481	8.160942610	6	8.163027776	5.701170423	19.680445605
6	7.237686383	5.620244603	16.098193878	1	3.818074859	8.733614223	7.931488317	6	7.347292899	6.603468098	20.389875908
6	7.257292149	4.294287904	15.015516969	1	5.288505931	7.987075056	10.79393203	1	7.368110396	6.592972626	21.488715677
1	1.92224641	4.823864197	14.564168043	6	5.610739022	9.916559637	8.249354662	6	6.501189876	7.518445553	19.731106071
1	2.508204756	4.126945474	16.076032653	1	5.490772935	10.482785248	7.307507837	6	6.468215989	7.507210774	18.324165259
6	2.939004623	2.916362377	14.316952187	1	6.691386498	9.776943030	8.428700001	1	5.801848910	8.194774842	17.788459616
1	2.005804449	2.329912687	14.401406497	1	5.209523527	10.532319111	9.074490084	6	9.079447817	4.740178678	20.411980079
1	3.758410603	2.333289340	14.771819669	6	6.261748120	8.356709119	10.319981481	1	8.593140160	4.322481563	21.311879979
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6	4.948425444	4.452133653	17.271727724	1	2.048559475	8.565331232	9.257750561	1	9.384061063	3.901267086	19.763163599
1	3.899920590	4.158243099	14.765662065	6	5.659805383	9.683251718	11.111960096	6	5.6565796711	8.496233743	20.524001655
1	5.243504455	5.106612200	18.107850291	1	1.576014659	10.155316730	11.003634817	1	4.849446104	8.926558357	19.906992456
6	5.836490107	3.178897563	12.793121063	1	3.331321170	10.399201604	10.753735650	1	6.267710592	9.337871578	20.903665062
5	5.702094449	2.6366072368	18.264622944	1	2.750975959	9.510897721	12.187566823	1	5.193641582	8.011890383	21.402709858
6	6.904429718	3.437579698	17.197623233	6	7.257611414	6.615124993	7.794817166	6	12.147950726	6.274394531	12.807290545
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6	9.557072049	8.617176939	17.147404852	1	5.801857115	5.011590900	7.608191801	1	12.002848297	4.674281183	14.265511101
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1	9.188031412	8.055914718	18.020270463	1	6.568898971	5.221896017	4.589889074	1	14.92046473	5.913997563	12.992317415
6	8.749988590	9.937915961	17.023178503	1	1.7819594657	6.577197159	3.892298620	1	16.009314112	5.768297350	13.057754860
1	8.844705559	10.537561943	17.946782901	6	8.038415015	7.513383310	5.651578203	6	0.120658917	6.895583637	13.275332376
1	1.678623232	9.728496280	16.8588091913	6	8.064352904	7.506593058	7.058963797	6	1.517645646	7.059062514	13.395937030
1	9.112898633	10.547648418	15.176593836	1	1.7825032790	8.198914026	7.596206662	6	1.517645646	7.059062514	13.395937030
6	11.824079400	8.436223832	14.977039202	6	5.482407518	4.715097345	4.961648231	6	1.539892160	4.017204638	14.665370922
1	12.646961231	7.782827255	14.636708948	1	5.985332151	4.272939913	4.084138118	1	13.867411575	3.571590632	15.306189323
12	12.048357007	8.691059806	16.029785036	1	4.568898971	5.221896017	4.589889074	1	15.447334953	4.410516899	15.310581465
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12	12.804054559	10.253979824	18.264622944	1	8.882493303	8.493265323	4.860727217	1	6.812217533	7.769428923	14.097089092
11	11.041362903	10.436573008	14.476062281	6	1.685386339	5.909506400	12.048206413	1	1.756442162	7.246311701	14.323886365
11	11.649847515	7.9122128578	13.069005709	1	8.270069484	9.331915104	4.746783919	1	1.077564193	8.694101682	13.5429178740
6	7.252786775	6.553253957	15.798423784	1	1.9351715010	8.009143250	3.985176201	1	0.348078376	8.078051207	15.044151444
6	8.172130784	5.756126250	18.18910499	6	2.386406236	6.268413822	12.583054853	8	7.884606982	1.397677422	14.588532089
1	8.859966301	5.102864210	17.771329887	6	1.847251949	5.295962857	11.714067080	6	8.5946497907	1.630864179	15.379773468
6	1.678744849	5.773747810	19.732072679	1	2.526462741	4.671841330	11.209260633	6	9.569382409	2.911823598	15.491813554
6	7.287737470	6.612592402	20.413695703	6	0.457660276	5.097069732	11.624738876	1	9.172989275	3.749655050	14.906259723
1	1.296322608	6.630222911	21.510823412	6	-0.387863399	5.909506400	12.048206413	6	10.668093558	3.086974334	16.348849713
6	6.360319106	7.424531055	17.918935637	1	-1.475490869	5.764261648	12.346717867	1	1.119429672	4.083759312	16.439205630
6	6.348484091	7.379282034	18.307643588	6	14.417058029	6.902273693	12.126187248	6	11.183081334	2.010969797	17.092056970
1	5.628933213	7.987872485	17.748721797	6	13.020508644	7.065534423	12.086514493	1	12.041328661	2.155026176	17.759363494

