## **Supplementary Information**

## Perovskite LaFeO<sub>3</sub>/montmorillonite nanocomposites: synthesis, interface characteristics and enhanced photocatalytic activity

Kang Peng, Liangjie Fu, Huaming Yang<sup>\*</sup>, Jing Ouyang<sup>\*</sup>

Samples	La(NO <sub>3</sub> ) <sub>3</sub> ·6H <sub>2</sub> O	Fe(NO <sub>3</sub> ) <sub>3</sub> ·9H <sub>2</sub> O	Co(NO <sub>3</sub> ) <sub>3</sub> ·6H <sub>2</sub> O	Ni(NO <sub>3</sub> ) <sub>3</sub> ·6H <sub>2</sub> O	Citric acid	MMT
	(mol)	(mol)	(mol)	(mol)	(mol)	(g)
LaFeO <sub>3</sub> /MMT	0.005	0.005	0	0	0.010	2.000
LaCoO <sub>3</sub> /MMT	0.005	0	0.005	0	0.010	2.000
LaNiO <sub>3</sub> /MMT	0.005	0	0	0.005	0.010	2.000
La(FeCo)O <sub>3</sub> /MMT	0.005	0.0025	0.0025	0	0.010	2.000
La(FeNi)O <sub>3</sub> /MMT	0.005	0.0025	0	0.0025	0.010	2.000
La(CoNi)O <sub>3</sub> /MMT	0.005	0	0.0025	0.0025	0.010	2.000
LaFeO <sub>3</sub> /MMT-0.5	0.0025	0.0025	0	0	0.005	2.000
LaFeO <sub>3</sub> /MMT-2	0.010	0.010	0	0	0.020	2.000
LaFeO <sub>3</sub>	0.005	0.005	0	0	0.010	0

**Table S1**The material formulation for the samples.



Figure S1 SEM image of LaFeO<sub>3</sub>.



Figure S2 (a) Si 2p and (b) Al 2p XPS spectra of MMT and LaFeO<sub>3</sub>/MMT.



Figure S3 XRD patterns of LaMO<sub>3</sub>/MMT (M=Fe,Co,Ni).



Figure S4 (a) UV–vis diffuse reflectance spectra and (b) band gap energies of LaMO<sub>3</sub>/MMT

(M=Fe,Co,Ni).



**Figure S5** (a) Photocatalytic degradation of RhB with LaMO<sub>3</sub>/MMT (M=Fe,Co,Ni), (b)

Photocatalytic activities compare of LaMO<sub>3</sub>/MMT.



**Figure S6** (a) Photocatalytic degradation of RhB with LaFeO<sub>3</sub>/MMT-0.5, LaFeO<sub>3</sub>/MMT and LaFeO<sub>3</sub>/MMT-2, (b) Photocatalytic activities compare of MMT, LaFeO<sub>3</sub>/MMT-0.5, LaFeO<sub>3</sub>/MMT,

LaFeO<sub>3</sub>/MMT-2 and LaFeO<sub>3</sub>