

Supporting Information

Photocatalytic Oxygenation of Substrates by Dioxygen with Protonated Manganese(III) Corrolazine

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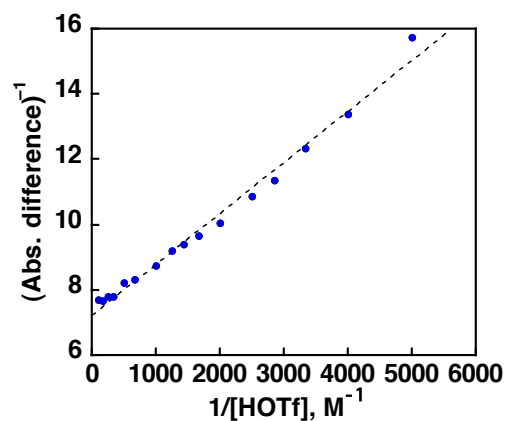


Figure S1. Plot of $(\Delta\text{Abs.})^{-1}$ vs $[\text{HOTf}]^{-1}$ to determine the second binding constant ($K_2 = \frac{[\text{Mn}^{\text{III}}(\text{OTf})(\text{H}_2\text{O})(\text{TBP}_8\text{Cz}(\text{H})_2)][\text{OTf}]}{[\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))][\text{H}^+]}$) upon addition of HOTf ($0.0 - 1.0 \times 10^{-2}$ M) into the solution of $\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))$ (1.0×10^{-5} M) in PhCN.

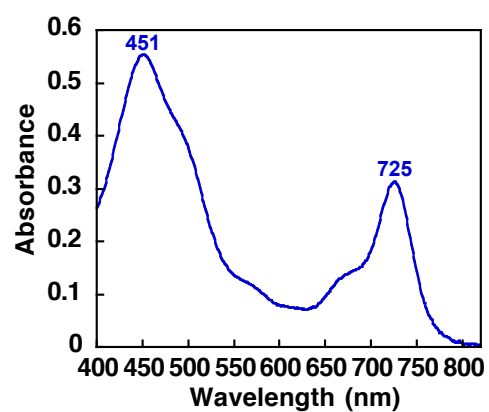


Figure S2. (a) UV-vis spectrum of crystalline $[\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))]$ dissolved in PhCN (2.0 mL) at 23 °C.

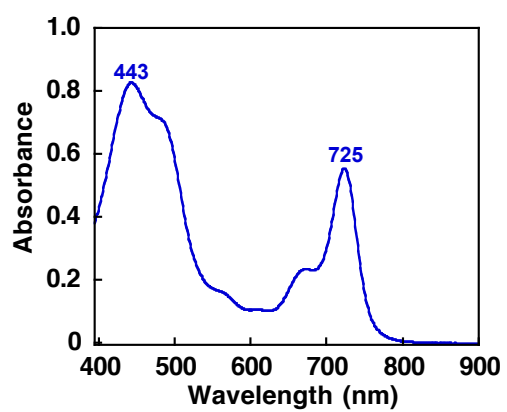


Figure S3. (a) UV-vis spectrum of crystalline [Mn^{III}(OTf)(TBP₈Cz(H))] dissolved in CH₂Cl₂ (2.0 mL) at 23 °C.

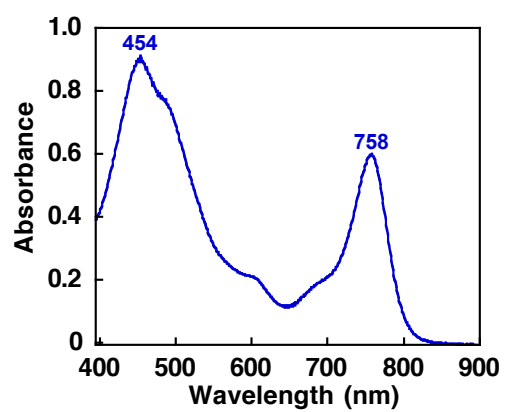


Figure S4. (a) UV-vis spectrum of crystalline $[\text{Mn}^{\text{III}}(\text{OTf})(\text{H}_2\text{O})(\text{TBP}_8\text{Cz}(\text{H})_2)][\text{OTf}]$ dissolved in CH_2Cl_2 (2.0 mL) at 23 °C.

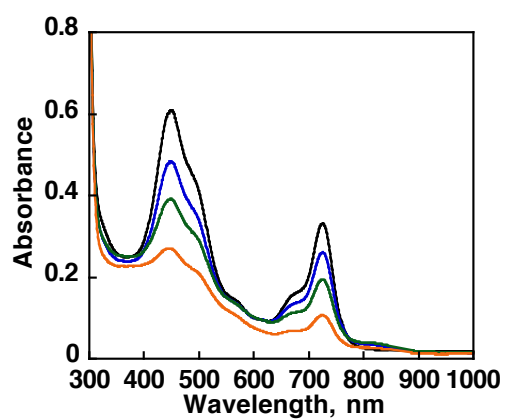


Figure S5. UV-vis spectral change of Mn^{III}(OTf)(TBP₈Cz(H)) (1.0×10^{-5} M) during photocatalytic oxidation reaction [black: 0 h, blue: 2 h, green: 4 h, and orange: 6 h] of HMB (0.3 M) in the presence of 4.0×10^{-5} M of HOTf under photoirradiation (white light) in an O₂-saturated PhCN solution at room temperature.

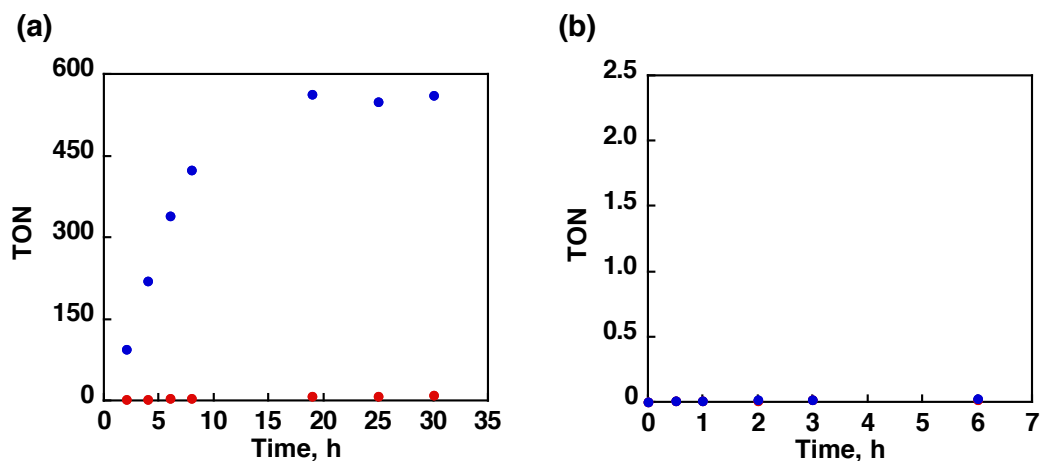


Figure S6. Time profiles of the amount of produced pentamethylbenzyl alcohol for photocatalytic oxidation reaction of HMB (0.3 M) by $(\text{TBP}_8\text{Cz})\text{Mn}^{\text{III}}$ (1.0×10^{-5} M) (a) in the presence of 4.0×10^{-5} M of HOTf or (b) in the presence of 5.0×10^{-3} M of HOTf under photoirradiation (white light) in an O_2 -saturated PhCN solution at room temperature.

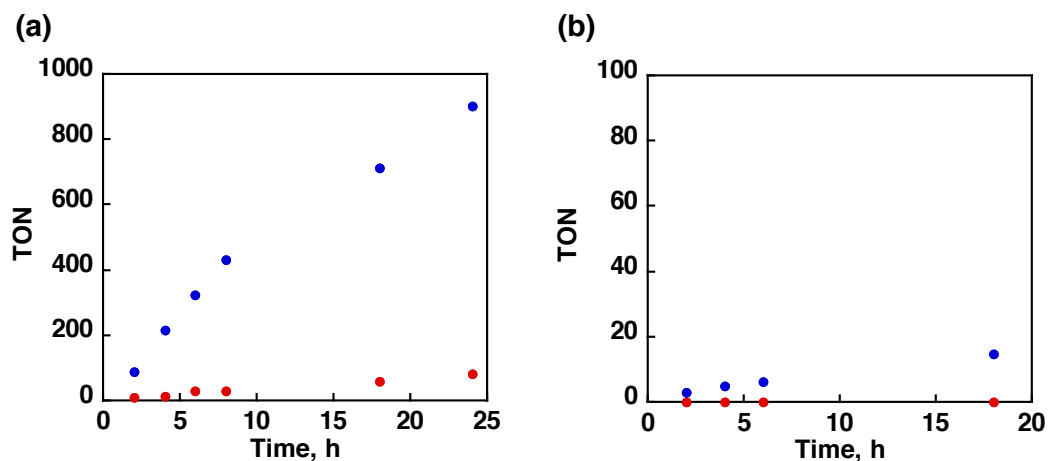


Figure S7. Time profiles of the amount of products [methylphenylsulfoxide (blue) and methylphenylsulfone (red)] for photocatalytic oxidation reaction of thioanisole (2.0×10^{-2} M) by $\text{Mn}^{\text{III}}(\text{TBP}_8\text{Cz})$ (1.0×10^{-5} M) (a) in the presence of HOTf (2.0×10^{-5} M) and (b) in the absence of HOTf under photoirradiation (white light) in an O_2 -saturated PhCN at room temperature.

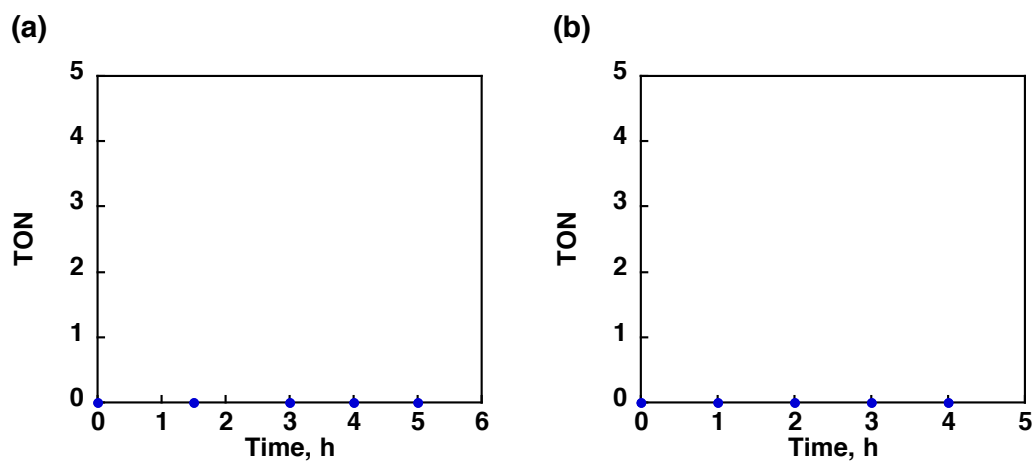


Figure S8. Time profiles of the amount of methylphenylsulfoxide for (a) photocatalytic reaction of thioanisole (2.0×10^{-2} M) in the presence of HOTf (2.0×10^{-5} M) under photoirradiation (white light) in O_2 -saturated PhCN at room temperature and (b) the reaction of thioanisole (2.0×10^{-2} M) by $Mn^{III}(TBP_8Cz)$ (1.0×10^{-5} M) in the presence of HOTf (2.0×10^{-5} M) without photoirradiation in O_2 -saturated PhCN at room temperature.

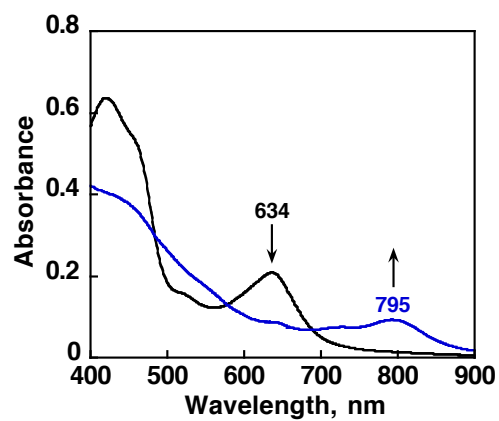


Figure S9. (a) UV-vis spectral changes from $\text{Mn}^{\text{V}}(\text{O})(\text{TBP}_8\text{Cz})$ (black line, $1.0 \times 10^{-5} \text{ M}$) to $[\text{Mn}^{\text{IV}}(\text{OH})(\text{TBP}_8\text{Cz}^+)](\text{OTf})$ in the presence of HOTf (blue line, $2.0 \times 10^{-4} \text{ M}$) in PhCN at room temperature.

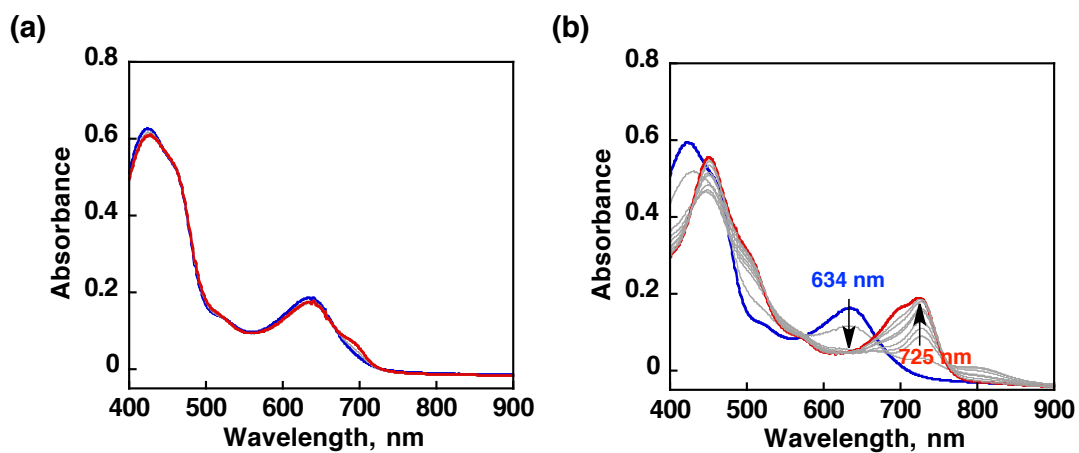


Figure S10. UV-vis spectral changes from $\text{Mn}^{\text{V}}(\text{O})(\text{TBP}_8\text{Cz})$ (blue line, $1.0 \times 10^{-5} \text{ M}$) with HMB ($2.0 \times 10^{-2} \text{ M}$) to $\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))$ (red line) (a) in the absence of HOTf and (b) upon addition of HOTf ($2.0 \times 10^{-5} \text{ M}$) in a PhCN solution for 1 h at room temperature.

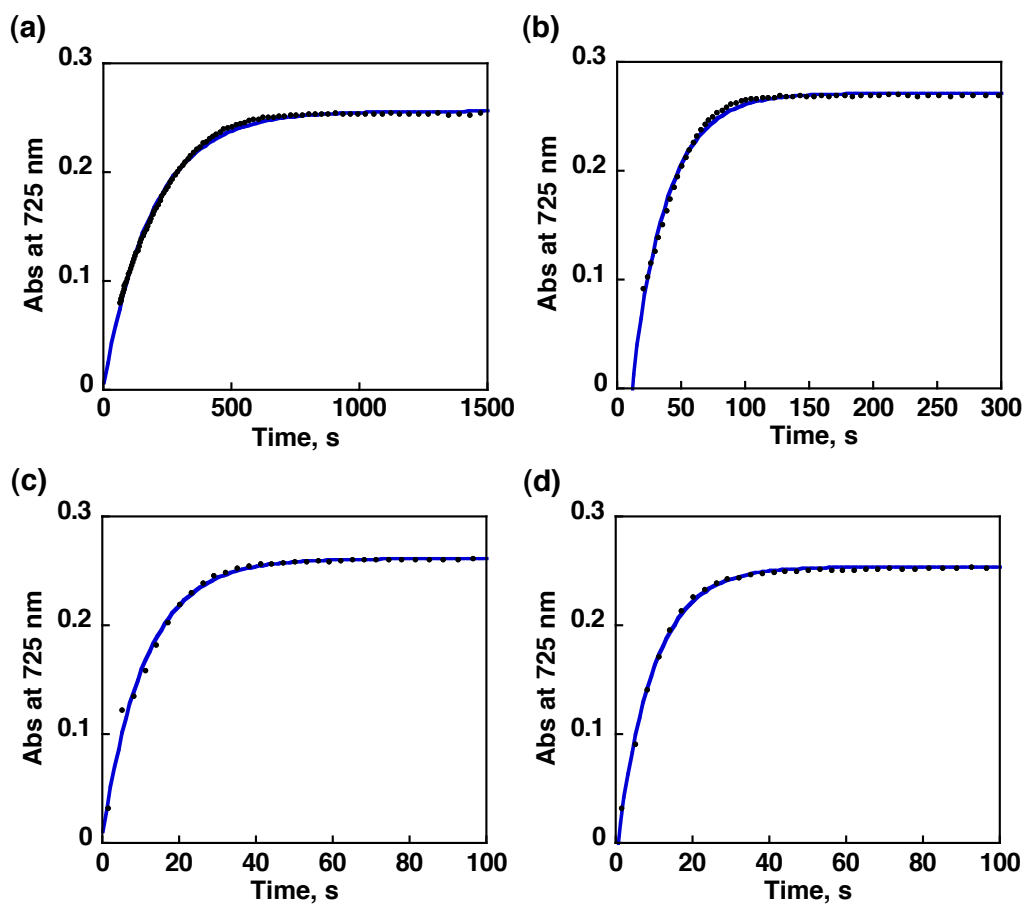


Figure S11. Plots of formed $\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))$ ($\lambda_{\text{max}} = 725 \text{ nm}$) versus time and best-fit lines to give pseudo-first-order rate constant (k_{obs}) of the reduction reaction of $[\text{Mn}^{\text{IV}}(\text{OH})(\text{TBP}_8\text{Cz}^+)](\text{OTf})$ ($1.0 \times 10^{-5} \text{ M}$) with HMB ($2.0 \times 10^{-2} \text{ M}$) in PhCN containing (a) $20 \mu\text{M}$, (b) $40 \mu\text{M}$, (c) $80 \mu\text{M}$, and (d) $90 \mu\text{M}$ of HOTf.

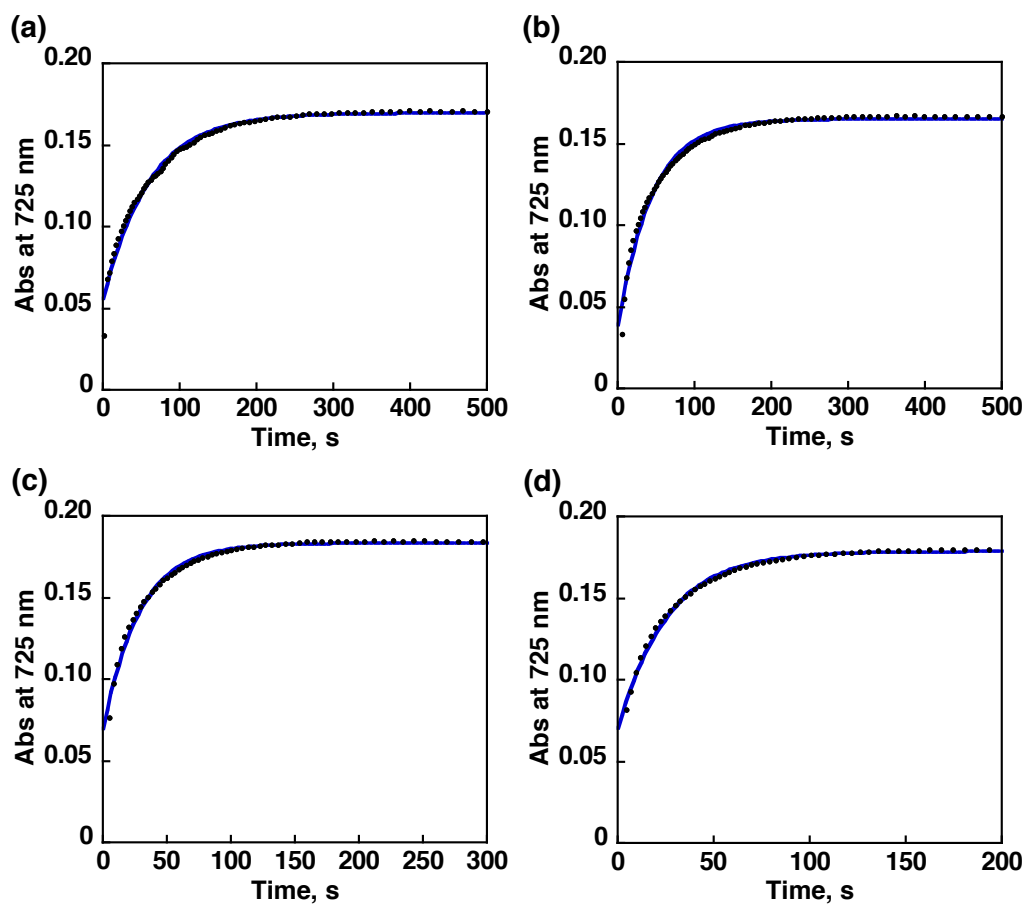


Figure S12. Plots of formed $\text{Mn}^{\text{III}}(\text{OTf})(\text{TBP}_8\text{Cz}(\text{H}))$ ($\lambda_{\text{max}} = 725 \text{ nm}$) versus time and best-fit lines to give pseudo-first-order rate constant (k_{obs}) of the reduction reaction of $[\text{Mn}^{\text{IV}}(\text{OH})(\text{TBP}_8\text{Cz}^+)](\text{OTf})$ ($1.0 \times 10^{-5} \text{ M}$) with HOTf ($3.0 \times 10^{-5} \text{ M}$) in PhCN containing (a) 2 mM, (b) 4 mM, (c) 6 mM, and (d) 8 mM of HMB.

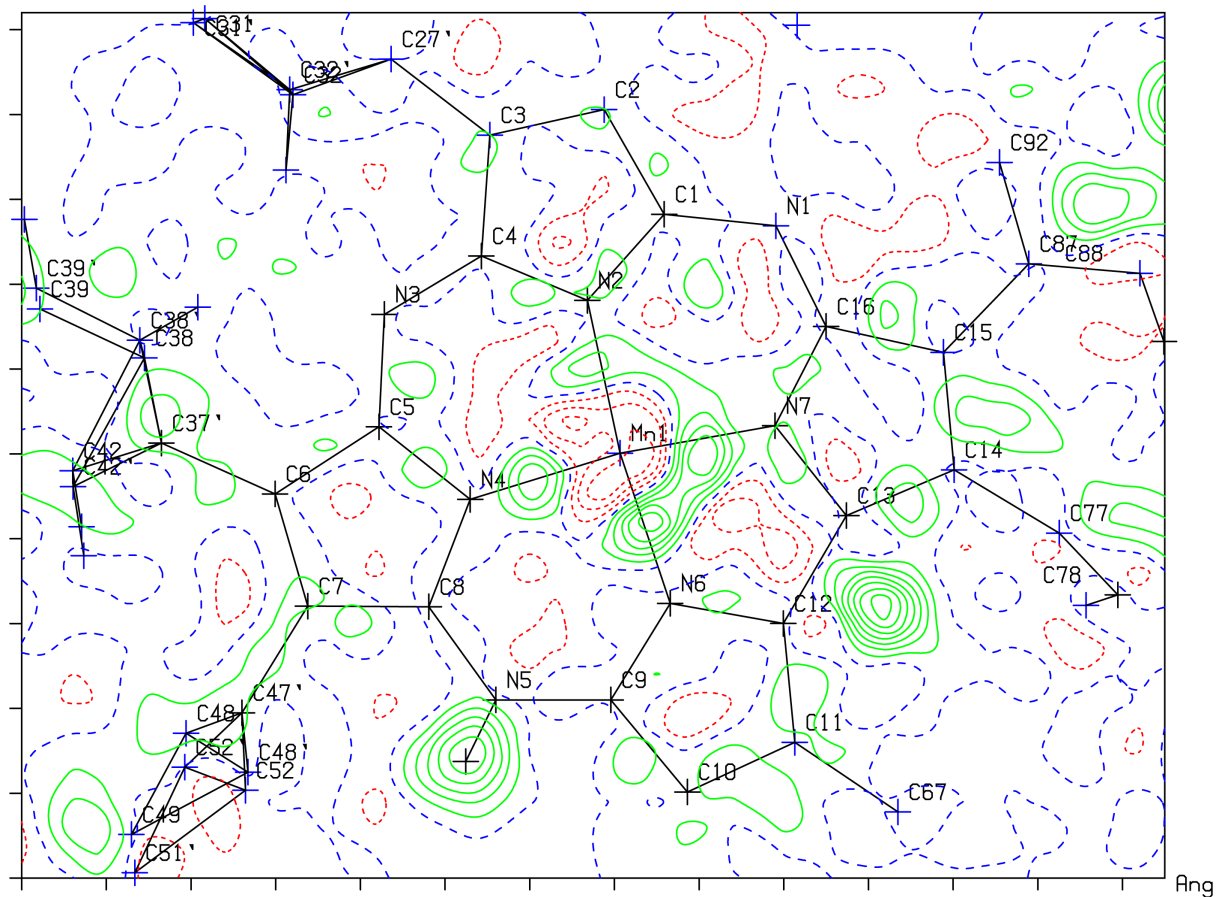


Figure S13. Difference Fourier map of Mn^{III}(OTf)(TBP₈Cz(H)) at 110(2) K drawn on the plane of the corrolazine core. N5 is the identified protonated site.

