

Analytical and Bioanalytical Chemistry

Electronic Supplementary Material

Two-dimensional mass defect matrix plots for mapping genealogical links in mixtures of lignin depolymerisation products

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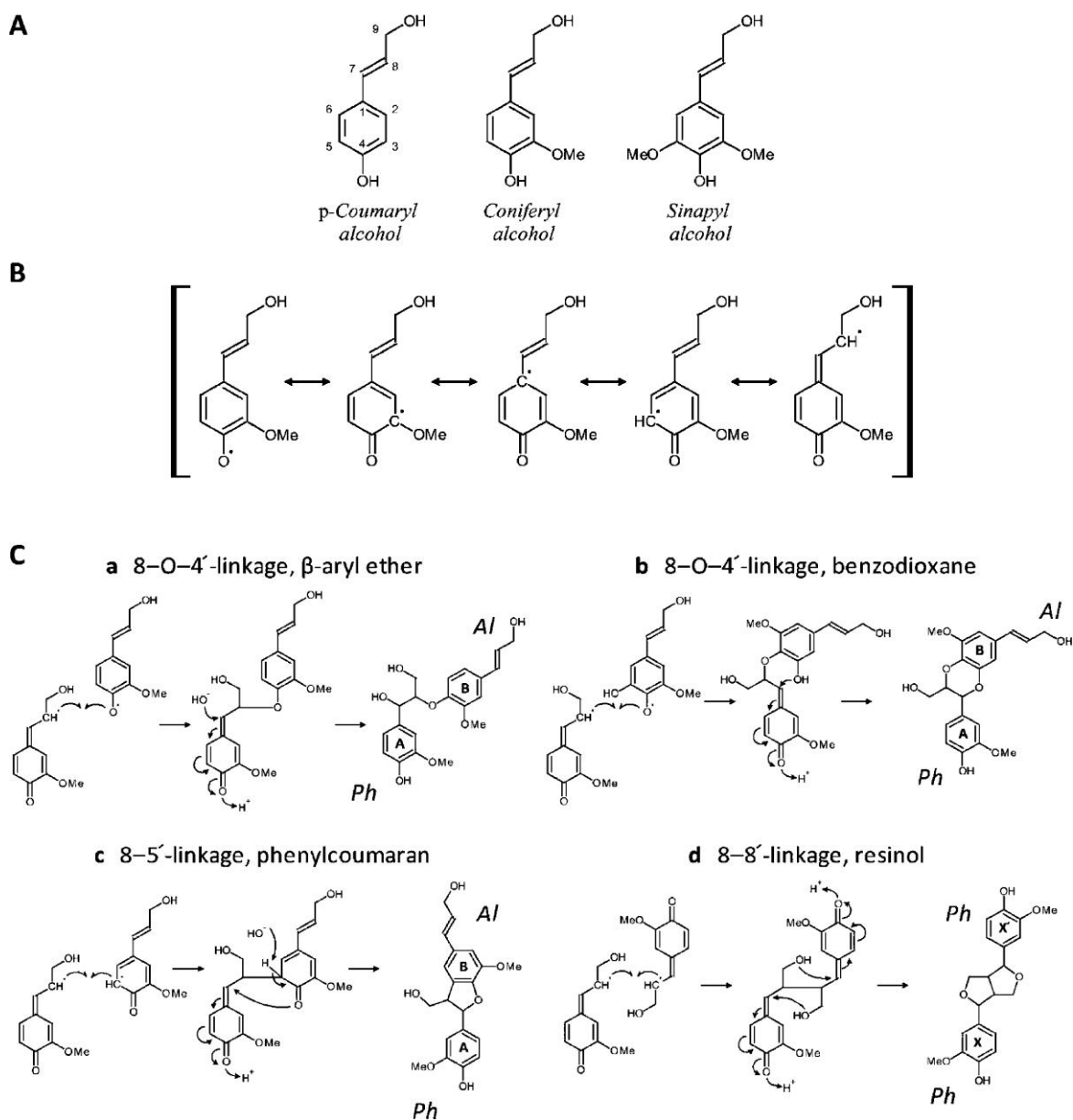


Fig. S1 Radical-radical dimerisation of monolignols. (A) Monolignols. (B) Radical delocalisation following monolignol oxidation, illustrated for the coniferyl alcohol radical. (C) Main monolignol dimerisation reactions: *Ph*, phenolic end group; *Al*, aliphatic end group. For all compounds except the resinols, the phenolic and aliphatic end groups correspond to the A and B rings, respectively. Resinols contain two phenolic end groups corresponding to the X and X' rings. Reprinted with permission from Morreel K, Kim H, Lu F, et al (2010) Mass spectrometry-based fragmentation as an identification tool in lignomics. *Anal Chem* 82:8095–105

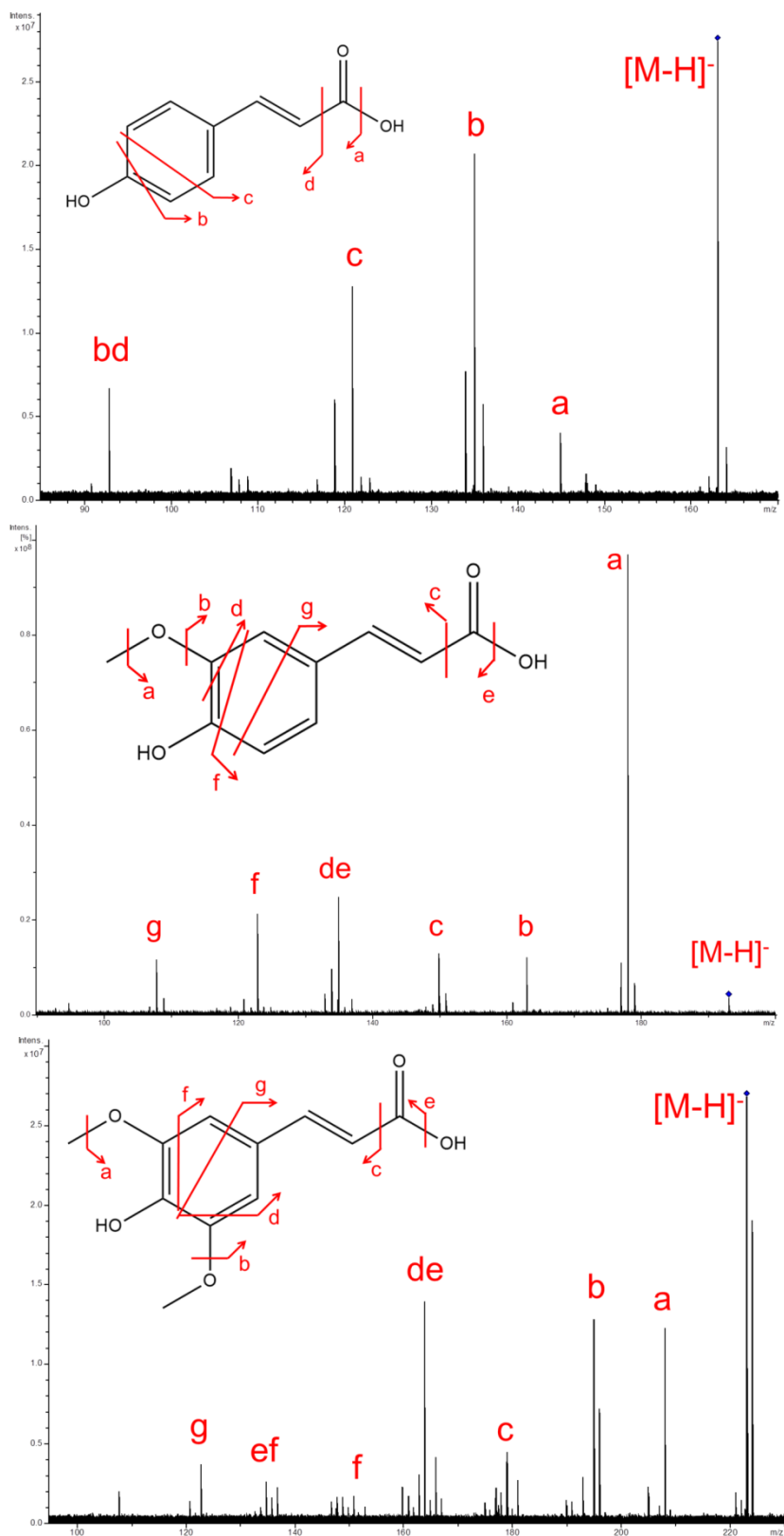


Fig. S2 CID-MS/MS spectra for the deprotonated $[M-H]^-$ ion of coumaryl (*top*), coniferyl (*middle*), and sinapyl acids (*bottom*), along with CID fragmentation schemes. The detailed peak assignments can be found in Table S2

Table S1 Peak assignments for Figure 2 (bottom)

No.	Compound	Calc. <i>m/z</i>	Exp. <i>m/z</i>	mDa error	ppm error
1	C ₁₅ H ₁₆ O ₁₀	355.06707	355.066919	-0.15	-0.43
2	C ₁₉ H ₁₆ O ₇	355.082326	355.08233	0.00	0.01
3	C ₁₆ H ₂₀ O ₇ S	355.085697	355.08572	0.02	0.06
4	C ₁₆ H ₂₀ O ₉	355.103456	355.10336	-0.10	-0.27
5	C ₂₀ H ₂₀ O ₆	355.118712	355.11868	-0.03	-0.09
6	C ₂₃ H ₁₉ O ₃ ¹³ C	355.129498	355.12973	0.23	0.65
7	C ₁₇ H ₂₄ O ₈	355.139841	355.13984	0.00	0.00
8	C ₂₁ H ₂₄ O ₅	355.155097	355.15505	-0.05	-0.13
9	C ₂₄ H ₂₂ NO ₂	355.157777	355.158027	0.25	0.70
10	C ₁₈ H ₂₈ O ₇	355.176227	355.17615	-0.08	-0.22
11	C ₂₂ H ₂₈ O ₄	355.191483	355.19148	0.00	-0.01
12	C ₁₉ H ₃₂ O ₄ S	355.194854	355.19482	-0.03	-0.10
13	C ₂₅ H ₂₇ O ¹³ C	355.202269	355.20271	0.44	1.24
14	C ₁₉ H ₃₂ O ₆	355.212612	355.21257	-0.04	-0.12
15	C ₂₂ H ₃₁ O ₃ ¹³ C	355.223398	355.22371	0.31	0.88
16	C ₂₃ H ₃₂ O ₃	355.227868	355.22788	0.01	0.03
17	C ₂₀ H ₃₆ O ₅	355.248998	355.24902	0.02	0.06
18	C ₂₃ H ₃₅ O ₂ ¹³ C	355.259784	355.26019	0.41	1.14
19	C ₂₁ H ₄₀ O ₄	355.285383	355.28544	0.06	0.16

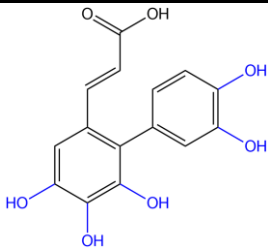
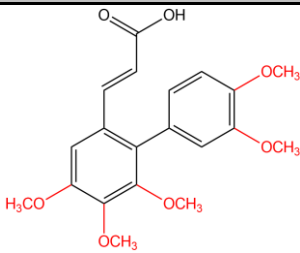
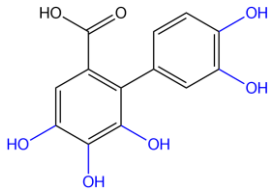
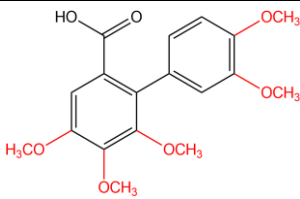
Table S2 CID fragments of coumaryl, coniferyl, and sinapyl acid

coumaryl alcohol				
Fragment cleavage	Proposed formula	Calc. <i>m/z</i>	Exp. <i>m/z</i>	ppm error
[M-H] ⁻	C ₉ H ₇ O ₃	163.04007	163.04040	2.04
a	C ₉ H ₅ O ₂	145.02841	145.02841	0.00
b	C ₈ H ₇ O ₂	135.04406	135.04400	-0.41
c	C ₇ H ₅ O ₂	121.02841	121.02830	-0.88
bd	C ₆ H ₅ O	93.03349	93.03360	1.17
coniferyl alcohol				
Fragment cleavage	Proposed formula	Calc. <i>m/z</i>	Exp. <i>m/z</i>	ppm error
[M-H] ⁻	C ₁₀ H ₉ O ₄	193.05063	193.05050	-0.68
a	C ₉ H ₆ O ₄	178.02606	178.02600	-0.34
b	C ₉ H ₇ O ₃	163.03897	163.03890	-0.44
c	C ₈ H ₆ O ₃	150.03115	150.03100	-0.97
de	C ₈ H ₇ O ₂	135.04406	135.04400	-0.41
f	C ₇ H ₇ O ₂	123.04406	123.04400	-0.46
g	C ₆ H ₄ O ₂	108.02058	108.02050	-0.75
sinapyl alcohol				
Fragment cleavage	Proposed formula	Calc. <i>m/z</i>	Exp. <i>m/z</i>	ppm error
[M-H] ⁻	C ₁₁ H ₁₁ O ₅	223.06120	223.06110	-0.43
a	C ₁₀ H ₈ O ₅	208.03663	208.03670	0.36
b	C ₁₀ H ₁₁ O ₄	195.06519	195.06510	-0.44
c	C ₁₀ H ₇ O ₃	175.03897	175.03890	-0.41
de	C ₉ H ₈ O ₃	164.04680	164.04670	-0.59
f	C ₈ H ₇ O ₃	151.03897	151.03890	-0.47
ef	C ₈ H ₇ O ₂	135.04406	135.04400	-0.41
g	C ₇ H ₇ O ₂	123.04406	123.04400	-0.46

Table S3 CID fragments of compound C₂₇H₂₈O₉ (Figure 5)

Fragment cleavage	Proposed formula	Calc. <i>m/z</i>	Exp. <i>m/z</i>	ppm error
[M-H]-	C ₂₇ H ₂₇ O ₉	495.16606	495.16600	-0.11
a	C ₂₇ H ₂₅ O ₈	477.15439	477.15460	0.43
ab	C ₂₇ H ₂₃ O ₇	459.14383	459.14370	-0.28
c	C ₂₆ H ₂₅ O ₇	449.15948	449.15960	0.27
d	C ₂₀ H ₂₁ O ₇	373.12818	373.12810	-0.21
e	C ₁₈ H ₁₇ O ₇	345.09688	345.09680	-0.23
e-H₂O	C ₁₈ H ₁₇ O ₆	329.10197	329.10190	-0.20
f	C ₁₈ H ₁₉ O ₅	315.12270	315.12280	0.32
g	C ₁₈ H ₁₉ O ₄	299.12779	299.12780	0.05
h	C ₁₅ H ₁₃ O ₅	273.07575	273.07570	-0.18
i	C ₉ H ₉ O ₄	181.04954	181.04960	0.36
j	C ₉ H ₉ O ₂	149.05971	149.05980	0.63

Table S4 Chemical formulae and peak areas for the compounds plotted in Figure 6

blue dots ●		
Compound formula	Peak area	Proposed structures
$C_{15}H_{12}O_2$	3422	
$C_{15}H_{12}O_3$	23039	
$C_{15}H_{12}O_4$	35326	
$C_{15}H_{12}O_5$	110062	
$C_{15}H_{12}O_6$	47377	
$C_{15}H_{12}O_7$	5019	
red dots ●		
Compound formula	Peak area	Proposed structures
$C_{15}H_{12}O_2$	3422	
$C_{16}H_{14}O_3$	21978	
$C_{17}H_{16}O_4$	37415	
$C_{18}H_{18}O_5$	177968	
$C_{19}H_{20}O_6$	662431	
$C_{20}H_{22}O_7$	1222609	
blue triangles ▲		
Compound formula	Peak area	Proposed structures
$C_{13}H_{10}O_2$	3010	
$C_{13}H_{10}O_3$	18431	
$C_{13}H_{10}O_4$	29498	
$C_{13}H_{10}O_5$	20760	
$C_{13}H_{10}O_6$	1777	
$C_{13}H_{10}O_7$	1716	
red triangles ▲		
Compound formula	Peak area	Proposed structures
$C_{13}H_{10}O_2$	3010	
$C_{14}H_{12}O_3$	32076	
$C_{15}H_{14}O_4$	269771	
$C_{16}H_{16}O_5$	313164	
$C_{17}H_{18}O_6$	317250	
$C_{18}H_{20}O_7$	212027	