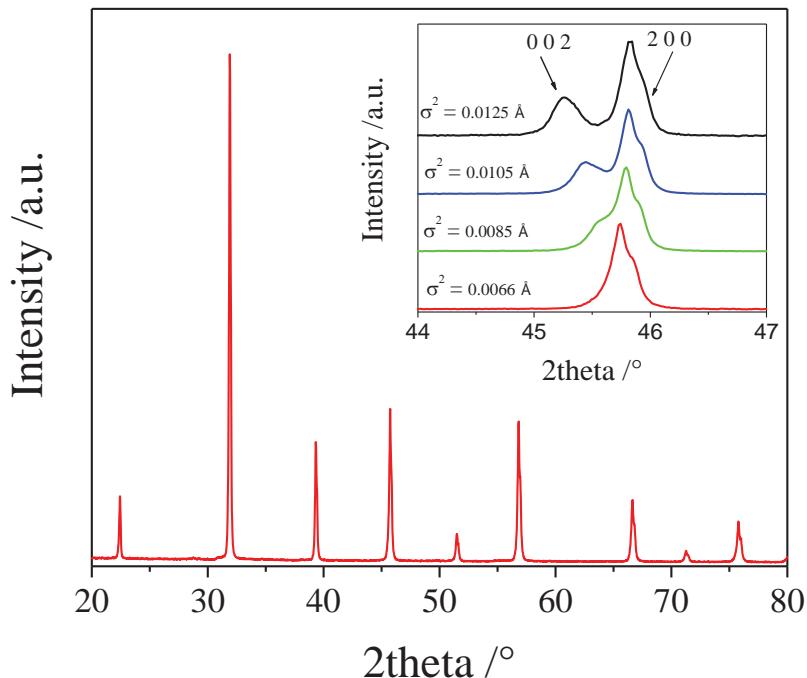


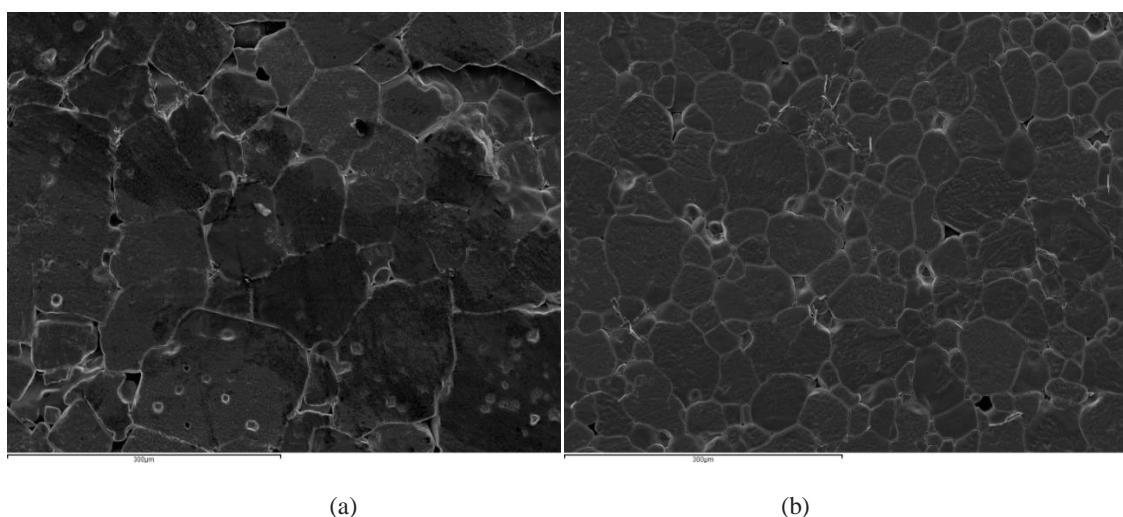
# Effect of ionic radii on the Curie temperature in $\text{Ba}_{1-x-y}\text{Sr}_x\text{Ca}_y\text{TiO}_3$ compounds

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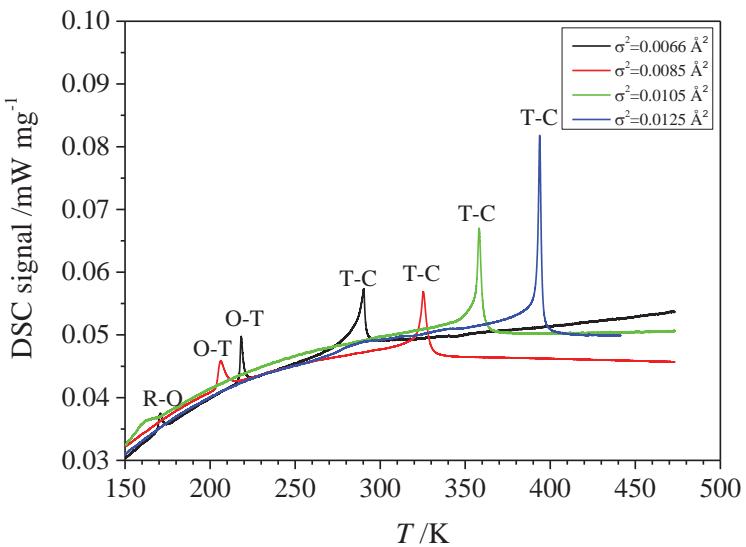
<sup>a</sup> Department of Materials, Imperial College London, London, SW7 2AZ, UK.



Supplementary Figure S1. XRD pattern of  $\text{Ba}_{0.65}\text{Sr}_{0.35}\text{TiO}_3$ . The inset shows tetragonal splitting of (200) peak in series A compounds.



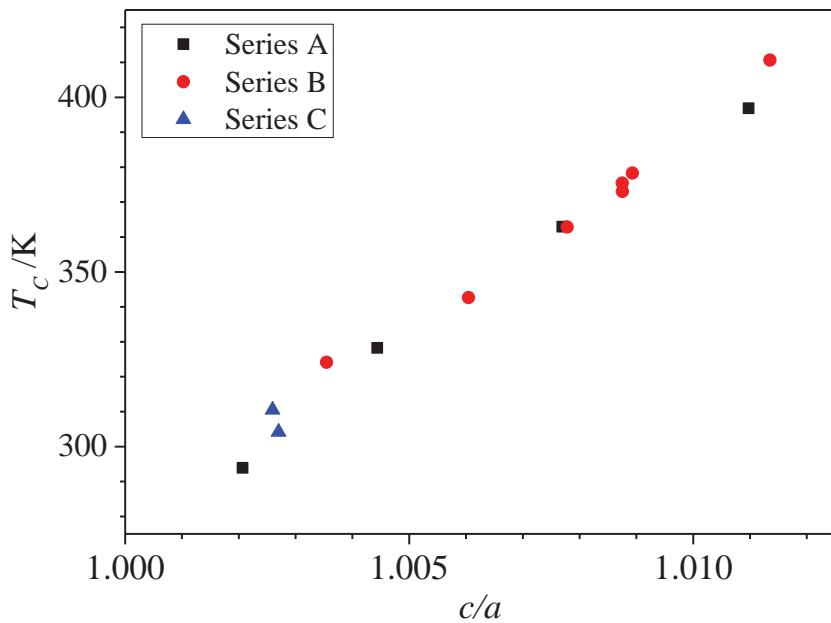
Supplementary Figure S2. SEM images of  $\text{Ba}_{0.65}\text{Sr}_{0.35}\text{TiO}_3$  (a) and  $\text{Ba}_{0.78}\text{Ca}_{0.22}\text{TiO}_3$  (b) pellets thermally etched at 1400 °C for 0.5 hrs



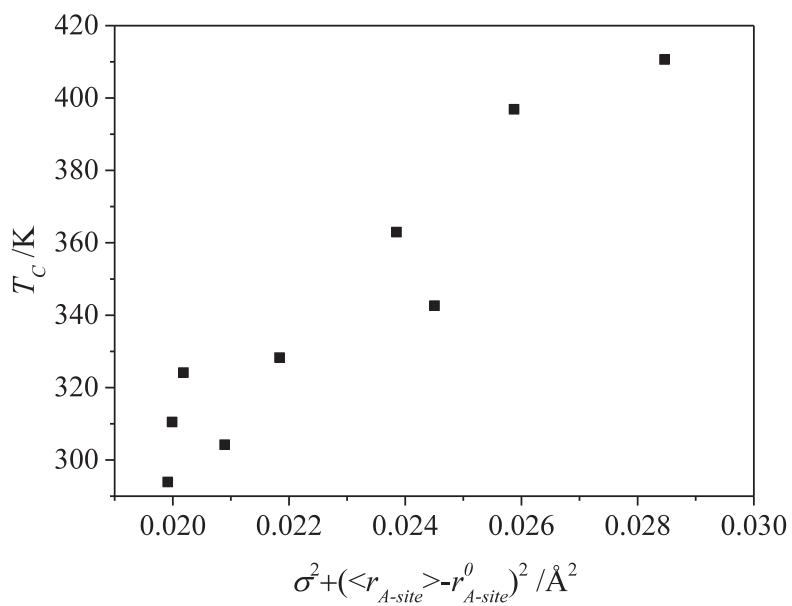
Supplementary Figure S3. Temperature dependences of DSC signal for compounds from series A.

Supplementary Table S4 ICP and DSC results of studied BSCT samples.

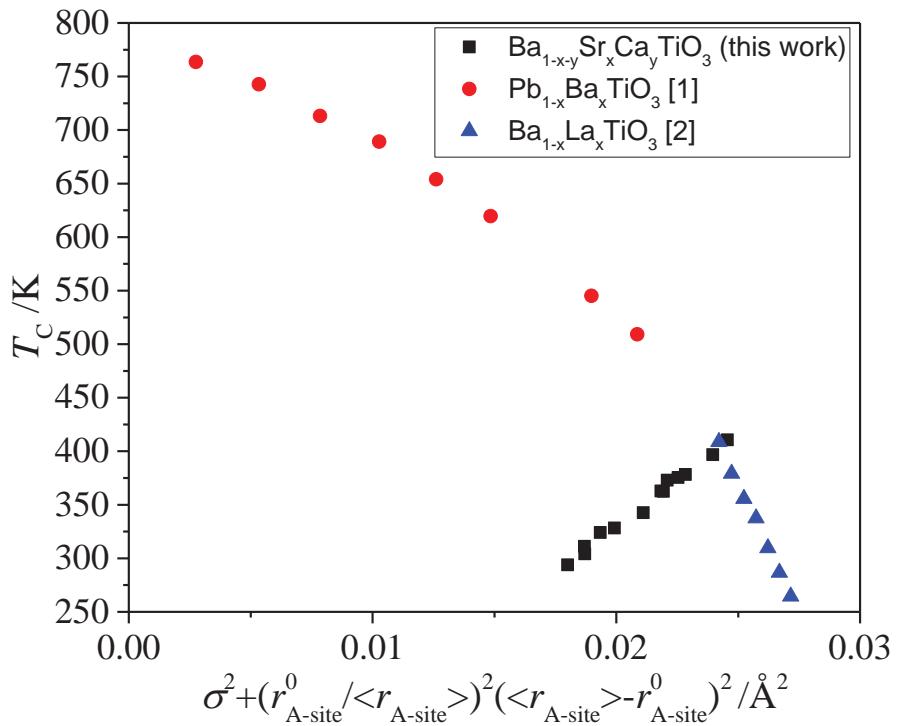
Intended composition	Analysed composition	$T_{R-O}$ /K	$T_{O-T}$ /K	$T_C$ /K
<b>Series A</b>				
Ba <sub>0.65</sub> Sr <sub>0.35</sub> TiO <sub>3</sub>	Ba <sub>0.655(2)</sub> Sr <sub>0.349(1)</sub> TiO <sub>3</sub>	171	218	291
Ba <sub>0.69</sub> Sr <sub>0.24</sub> Ca <sub>0.07</sub> TiO <sub>3</sub>	Ba <sub>0.704(2)</sub> Sr <sub>0.237(1)</sub> Ca <sub>0.076(1)</sub> TiO <sub>3</sub>	145	206	325
Ba <sub>0.74</sub> Sr <sub>0.12</sub> Ca <sub>0.15</sub> TiO <sub>3</sub>	Ba <sub>0.742(14)</sub> Sr <sub>0.120(2)</sub> Ca <sub>0.144(1)</sub> TiO <sub>3</sub>	NA	163	358
Ba <sub>0.78</sub> Ca <sub>0.22</sub> TiO <sub>3</sub>	Ba <sub>0.798(9)</sub> Ca <sub>0.206(1)</sub> TiO <sub>3</sub>	NA	NA	394
<b>Series B</b>				
Ba <sub>0.8</sub> Sr <sub>0.2</sub> TiO <sub>3</sub>	Ba <sub>0.796(3)</sub> Sr <sub>0.197(1)</sub> TiO <sub>3</sub>	187	251	341
Ba <sub>0.6</sub> Sr <sub>0.2</sub> Ca <sub>0.2</sub> TiO <sub>3</sub>	Ba <sub>0.607(6)</sub> Sr <sub>0.202(3)</sub> Ca <sub>0.198(1)</sub> TiO <sub>3</sub>	NA	NA	320
Ba <sub>0.9</sub> Ca <sub>0.1</sub> TiO <sub>3</sub>	Ba <sub>0.887(1)</sub> Ca <sub>0.113(2)</sub> TiO <sub>3</sub>	139	224	402
Ba <sub>0.85</sub> Sr <sub>0.1</sub> Ca <sub>0.05</sub> TiO <sub>3</sub>	Ba <sub>0.854(2)</sub> Sr <sub>0.100(1)</sub> Ca <sub>0.046(1)</sub> TiO <sub>3</sub>	173	248	376
Ba <sub>0.78</sub> Sr <sub>0.1</sub> Ca <sub>0.12</sub> TiO <sub>3</sub>	Ba <sub>0.794(2)</sub> Sr <sub>0.103(1)</sub> Ca <sub>0.103(1)</sub> TiO <sub>3</sub>	NA	NA	369
Ba <sub>0.75</sub> Sr <sub>0.1</sub> Ca <sub>0.15</sub> TiO <sub>3</sub>	Ba <sub>0.754(2)</sub> Sr <sub>0.101(1)</sub> Ca <sub>0.145(1)</sub> TiO <sub>3</sub>	NA	NA	363
Ba <sub>0.7</sub> Sr <sub>0.1</sub> Ca <sub>0.2</sub> TiO <sub>3</sub>	Ba <sub>0.704(1)</sub> Sr <sub>0.102(1)</sub> Ca <sub>0.195(1)</sub> TiO <sub>3</sub>	NA	NA	357
<b>Series C</b>				
Ba <sub>0.68</sub> Sr <sub>0.32</sub> TiO <sub>3</sub>	Ba <sub>0.692(3)</sub> Sr <sub>0.321(6)</sub> TiO <sub>3</sub>	176	227	302
Ba <sub>0.62</sub> Sr <sub>0.28</sub> Ca <sub>0.1</sub> TiO <sub>3</sub>	Ba <sub>0.625(3)</sub> Sr <sub>0.283(4)</sub> Ca <sub>0.098(1)</sub> TiO <sub>3</sub>	NA	180	307



Supplementary Figure S5.  $T_c$  as a function of  $c/a$  for the studied compounds.



Supplementary Figure S6.  $T_c$  as a function of  $\sigma^2 + (\langle r_{A-site} \rangle - r_{A-site}^0)^2 / \text{\AA}^2$  for samples studied in this work.



Supplementary Figure S7.  $T_c$  as a function of  $\sigma^2 + \left( \frac{r_{A\text{-site}}^0}{\langle r_{A\text{-site}} \rangle} \right)^2 (\langle r_{A\text{-site}} \rangle - r_{A\text{-site}}^0)^2 / \text{\AA}^2$  for titanates with iso electronic ( $\text{Ba}_{1-x-y}\text{Sr}_x\text{Ca}_y\text{TiO}_3$  this work) and non-iso electronic ( $\text{Pb}_{1-x}\text{Ba}_x\text{TiO}_3$ <sup>1</sup>,  $\text{Ba}_{1-x}\text{La}_x\text{TiO}_3$ <sup>2</sup>) doping on the A site.

<sup>1</sup> T. Ikeda, J. Phys. Soc. Jap. **14** 1286 (1959)

<sup>2</sup> L. Ben and D. C. Sinclair, Appl. Phys. Lett., **98** 092907 (2011)