# Phenol Derivatives as Coupling Partners with Alkylsilicates in Photoredox/Nickel Dual Catalysis

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#### Key to Abbreviated Terms:

bpy: 2,2'-bipyridyl CFL: Compact fluorescent light LED: Light-emitting diode dtbbpy: 4,4'-di-*tert*-butyl-2,2'-dipyridyl

### **General Considerations**

All chemical transformations requiring inert atmospheric conditions utilized Schlenk line techniques with a 5-port dual-bank manifold. Argon was used to provide such an atmosphere. NMR Spectra (<sup>1</sup>H, <sup>13</sup>C {<sup>1</sup>H}, <sup>19</sup>F) were performed at 298 K. <sup>1</sup>H NMR spectra obtained in CDCl<sub>3</sub> were referenced to residual non-deuterated chloroform ( $\delta$  7.26) in the deuterated solvent. <sup>13</sup>C {<sup>1</sup>H} NMR spectra obtained in CDCl<sub>3</sub> were referenced to chloroform ( $\delta$  77.2). Reactions were monitored by GC/MS, HPLC, <sup>1</sup>H NMR, and/or by TLC on silica gel plates (60 Å porosity, 250 µm thickness). TLC analysis was performed using hexanes/EtOAc as the eluant and visualized using permanganate stain and/or UV light. Silica plugs utilized flash silica gel (60 Å porosity, 32-63 µm). Flash chromatography was accomplished using an automated system (visualizing at 254 nm, monitoring at 280 nm) with silica cartridges (60 Å porosity, 20-40 µm). Solvents were purified either by distillation over sodium or CaH<sub>2</sub> or by use of drying cartridges through a solvent delivery system. Irradiation of reaction vessels was accomplished either using standard 26 W CFLs or LEDs (blue or white). The choice of light source did not appear to have any effect on reaction success. A fan was employed to ensure reactions remained at or near rt when using either CFLs or LEDs.

#### **Chemicals:**

Deuterated NMR solvents, MgSO<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>, pentane, hexanes, and Et<sub>2</sub>O were used as purchased. Et<sub>3</sub>N and *i*-Pr<sub>2</sub>NH were purchased from commercial suppliers and distilled from CaH<sub>2</sub> prior to use. THF was purchased from commercial suppliers and dried via a solvent delivery system. Catechol was purchased and recrystallized from refluxing hexanes or heptanes. DMF (99.8%, extra dry) was stored over 4 Å molecular sieves. NiCl<sub>2</sub>•dme (min. 97%) and RuCl<sub>3</sub>•3H<sub>2</sub>O were purchased commercially. Phenol derivatives were either purchased from commercial suppliers or prepared in-house. The aryl triflates 2a, 2b, 2c, 2d, and 2e were purchased from commercial suppliers and used without further purification. The following aryl triflates, -tosylates, and mesylates prepared according procedures: 4-cyanophenyl were to literature

trifluoromethanesulfonate (2f),<sup>1</sup> 4-(trifluoromethyl)phenyl trifluoromethanesulfonate (2g),<sup>2</sup> 3trifluoromethanesulfonate  $(2h)^{1}$ 4-fluorophenvl (trifluoromethyl)phenyl (2n),<sup>2</sup> 4-formylphenyl trifluoromethanesulfonate  $(20)^3$  4trifluoromethanesulfonate benzoylphenyl trifluoromethanesulfonate (2p),<sup>4</sup> 3-cyanophenyl trifluoromethanesulfonate (2q),<sup>1</sup> 3-fluorophenyl trifluoromethanesulfonate  $(2\mathbf{r})$ ,<sup>5</sup> 4-bromophenyl trifluoromethanesulfonate  $(2\mathbf{s})$ ,<sup>6</sup> 4-acetylphenyl 4-methylbenzenesulfonate (2i),<sup>7</sup> naphthalen-1-yl 4-methylbenzenesulfonate (2i),<sup>6</sup> 3-(trifluoromethyl)phenyl 4-methylbenzenesulfonate (2k),<sup>8</sup> 4-acetylphenyl methanesulfonate (21),<sup>6</sup> naphthalen-1-yl methanesulfonate (2m).<sup>6</sup> Ru(bpy)<sub>3</sub>(PF<sub>6</sub>)<sub>2</sub> was prepared in-house by the procedure outlined in our previous reports.<sup>9</sup> Silicates were prepared from their corresponding alkyltrimethoxysilanes. Information (preparation protocols, characterization etc.) for all silicates can be found in our previous reports.<sup>10</sup>

<sup>&</sup>lt;sup>1</sup> Lee, D-Y.; Hartwig, J. F. Org. Lett, 2005, 7, 1169.

<sup>&</sup>lt;sup>2</sup> Mori, A.; Mizusaki, T.; Ikawa, T.; Maegawa, T.; Monguchi, Y.; Sajiki, H. Chem. Eur. J. 2007, 13, 1432.

<sup>&</sup>lt;sup>3</sup> Maegawa, T.; Kitamura, Y.; Sako, S.; Udzo, T.; Sakurai, A.; Tanaka, A.; Kobayashi, Y.; Endo, K.; Bora, U.; Kurita, T.; Kozaki, A.; Monguchi, Y.; Sajiki, H. *Chem. Eur. J.* **2007**, *13*, 5937.

<sup>&</sup>lt;sup>4</sup> Zhao, B.; Lu, X. Org. Lett. 2006, 8, 5987.

<sup>&</sup>lt;sup>5</sup> Stang, P. J.; Anderson, A. G. J. Org. Chem. 1976. 41, 781.

<sup>&</sup>lt;sup>6</sup> Chung, C. W. Y.; Toy, P. H. Tetrahedron 2005, 61, 709.

<sup>&</sup>lt;sup>7</sup> Kuroda, J.; Inamoto, K.; Hiroya, K.; Doi, T. Eur. J. Org. Chem. 2009, 2251.

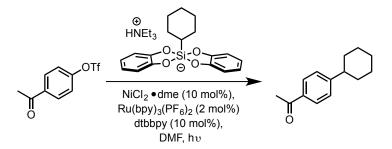
<sup>&</sup>lt;sup>8</sup> Munday, R. H.; Martinelli, J. R.; Buchwald, S. L. J. Am. Chem. Soc. 2008, 130, 2754.

<sup>&</sup>lt;sup>9</sup> Mabrouk, P.A.; Wrighton, M. S. Inorg. Chem. 1986, 25, 526.

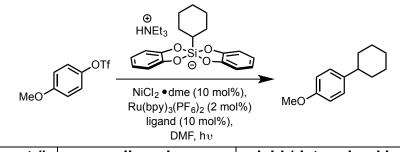
<sup>&</sup>lt;sup>10</sup> (a) Jouffroy, M.; Primer, D.; Molander, G. A. J. Am. Chem. Soc., **2016**, 138, 475. (b) Patel, N. R., Kelly, C. B., Jouffroy, M., Molander, G. A. Org Lett. **2016**, 18, 764.

# Reaction Optimization and Control Studies for Aryl Triflate Cross-Coupling

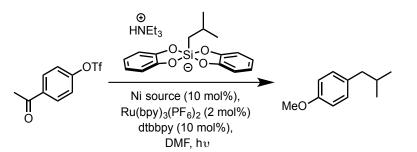
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experiment #	conditions	conversion (determined by HPLC)		
1	5 mol % NiCl <sub>2</sub> ·dme, 5 mol % dtbbpy 1.2 equiv alkylsilicate (not preforming Ni-ligand complex) (24 h)	84%		
2	5 mol% NiCl <sub>2</sub> ·dme, 5 mol % dtbbpy 1.2 equiv alkylsilicate (preforming Ni-ligand complex) (24 h)	87%		
3	10 mol % NiCl <sub>2</sub> ·dme, 10 mol % dtbbpy 1.2 equiv alkylsilicate (24 h)	95%		
4	5 mol % NiCl <sub>2</sub> ·dme, 5 mol % dtbbpy 1.5 equiv alkylsilicate (24 h)	87%		
5	10 mol % NiCl <sub>2</sub> ·dme, 10 mol % dtbbpy 1.5 equiv alkylsilicate (24 h)	98%		
6	10 mol % NiCl <sub>2</sub> ·dme, 10 mol % dtbbpy 1.5 equiv alkylsilicate (36 h)	100%		



experiment #	conditions	yield (determined by HPLC)
1	no light	0%
2	no Ru photocatalyst	10%
3	no NiCl₂·dme	0%

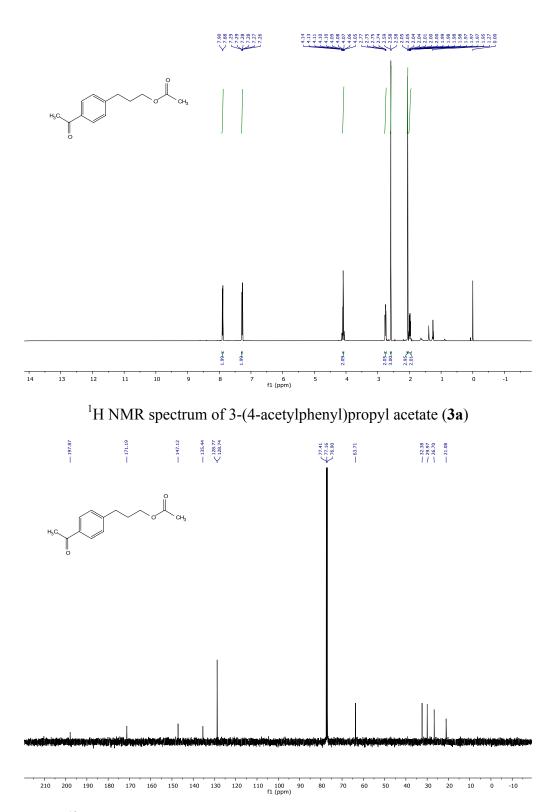


experiment #	ligand	yield (determined by HPLC)
1	2,2':6',2"-terpyridine	0%
2	2,2'-bis(2-oxazoline)	0%
3	neocuproine	0%

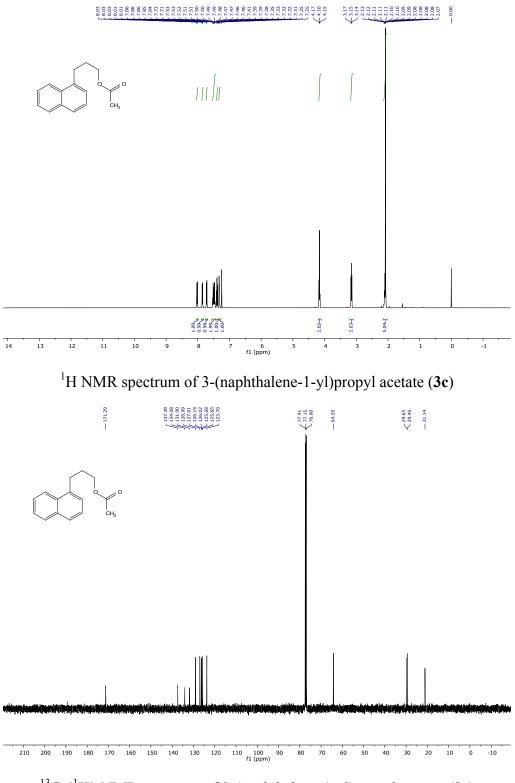


experiment #	Ni source	yield
1	Ni(COD) <sub>2</sub>	60% (isolated yield)
2	Ni(OTf) <sub>2</sub>	0% (determined by HPLC)

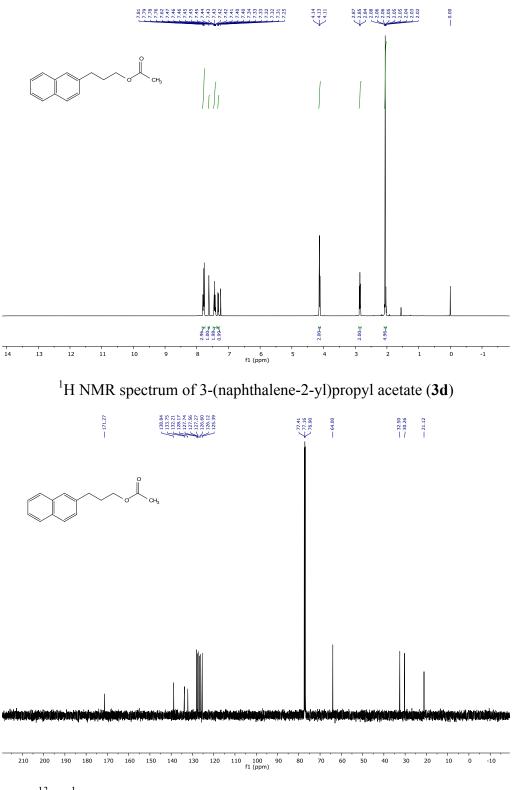




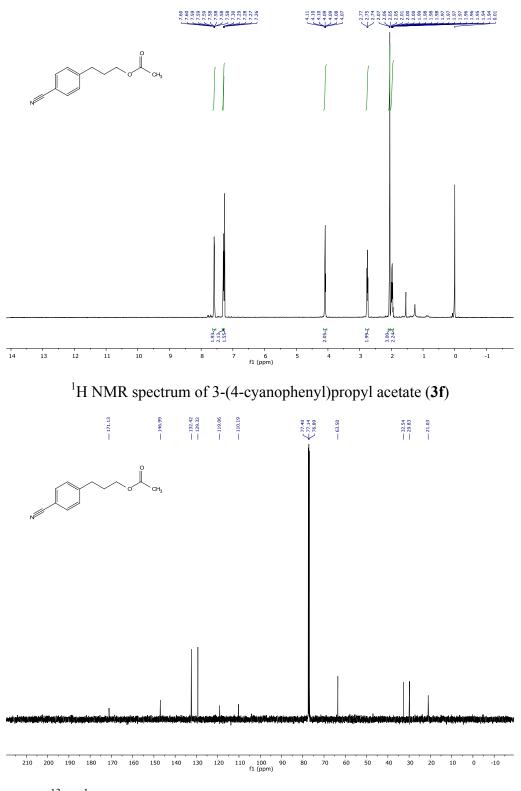
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(4-acetylphenyl)propyl acetate (3a)



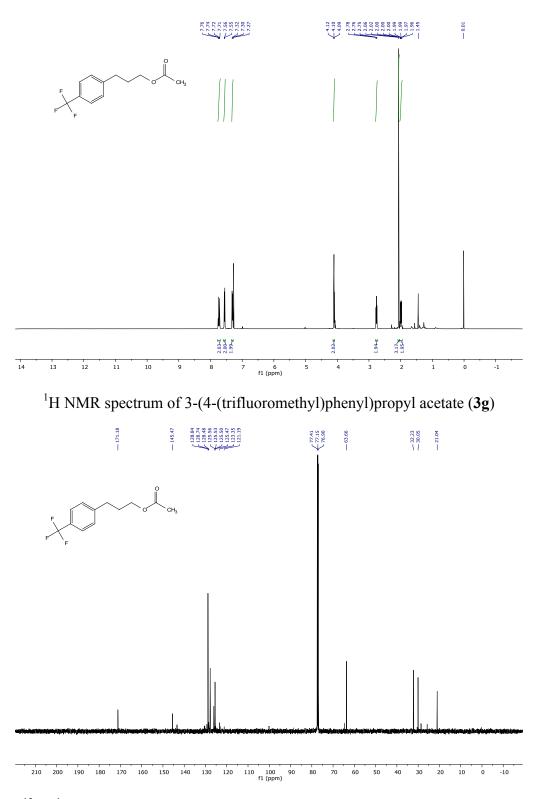
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(naphthalene-1-yl)propyl acetate (**3c**)



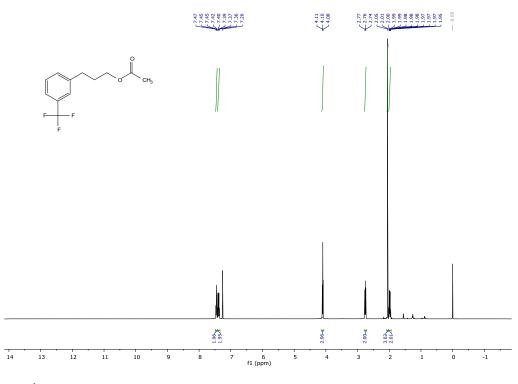
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(naphthalene-2-yl)propyl acetate (3d)



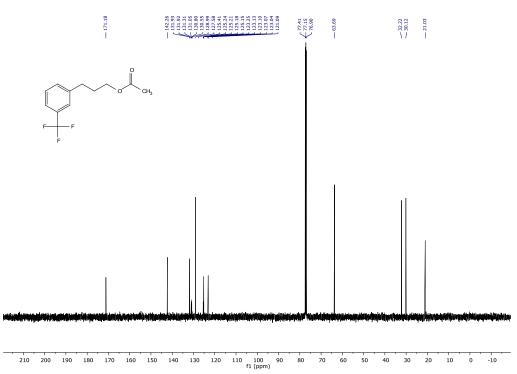
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(4-cyanophenyl)propyl acetate (**3f**)



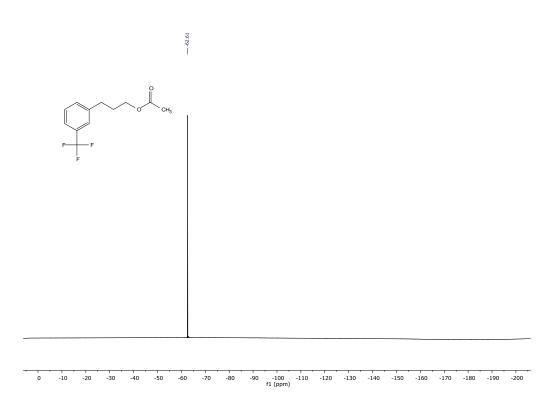
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(4-(trifluoromethyl)phenyl)propyl acetate (**3g**)



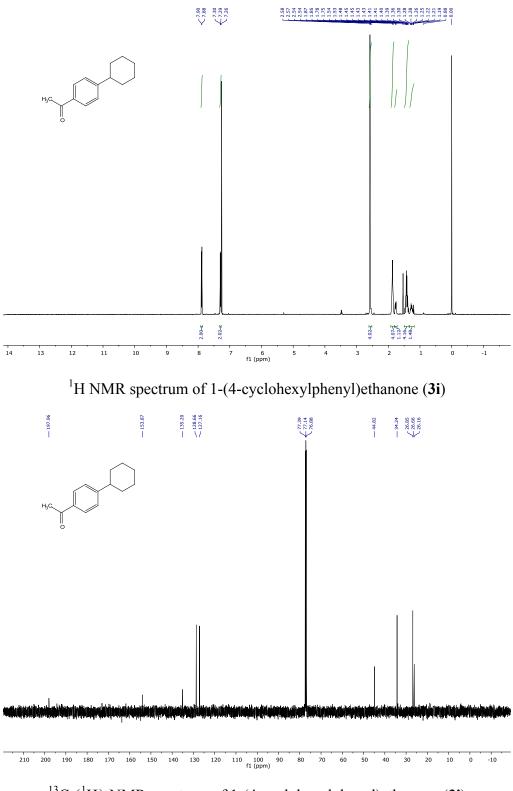
<sup>1</sup>H NMR spectrum of 3-(3-(trifluoromethyl)phenyl)propyl acetate (**3h**)



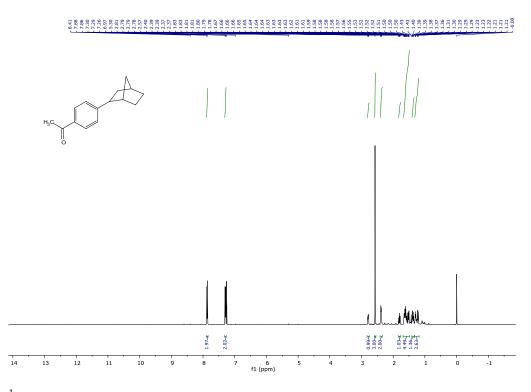
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 3-(3-(trifluoromethyl)phenyl)propyl acetate (3h)



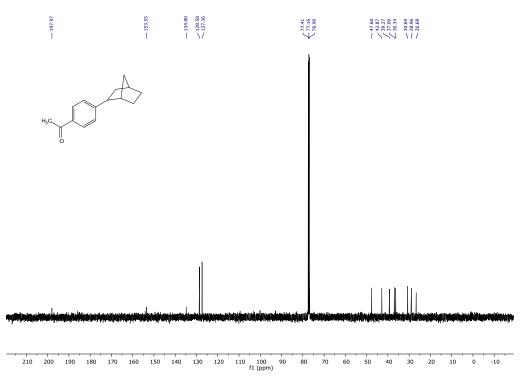
<sup>19</sup>F NMR spectrum of 3-(3-(trifluoromethyl)phenyl)propyl acetate (**3h**)



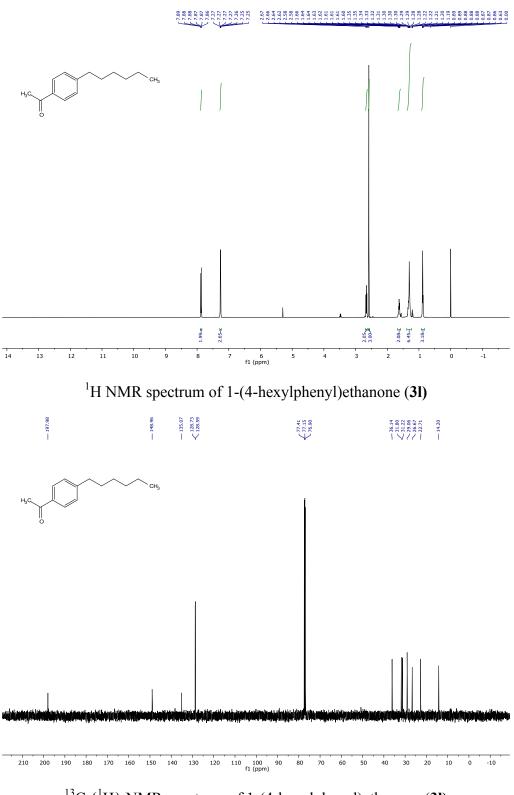
 $^{13}\text{C}$  {<sup>1</sup>H} NMR spectrum of 1-(4-cyclohexylphenyl)ethanone (**3i**)



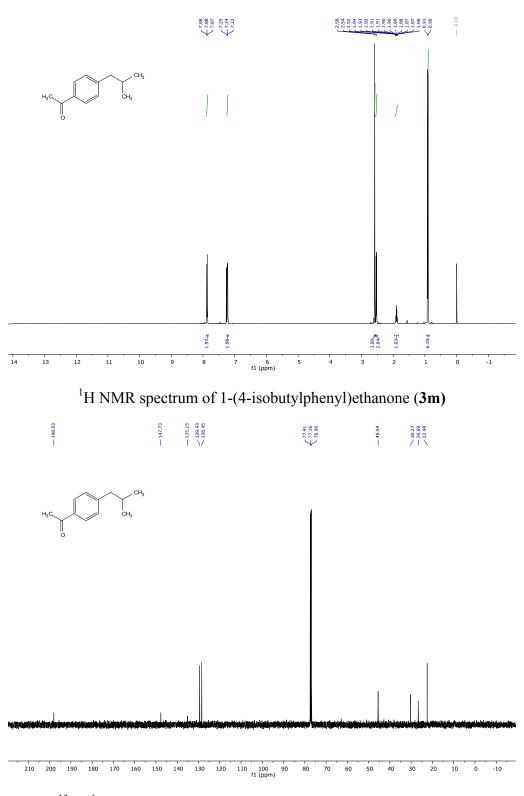
<sup>1</sup>H NMR spectrum of (±)-1-(4-(bicyclo[2.2.1]heptan-2-yl)phenyl)ethanone (**3j**)



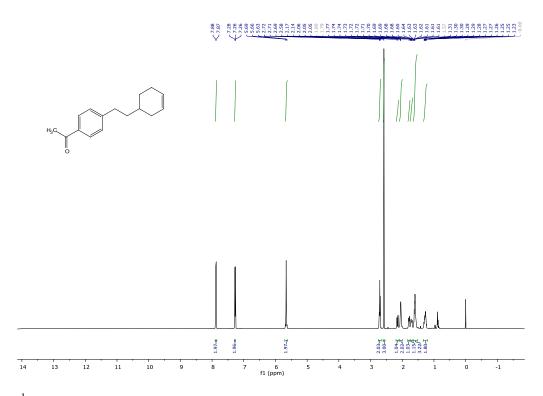
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of (±)-1-(4-(bicyclo[2.2.1]heptan-2-yl)phenyl)ethanone (**3j**)



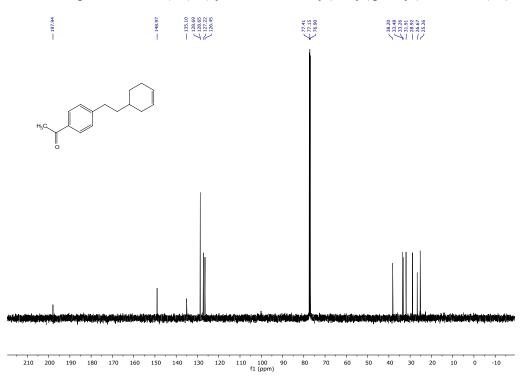
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 1-(4-hexylphenyl)ethanone (**3**I)



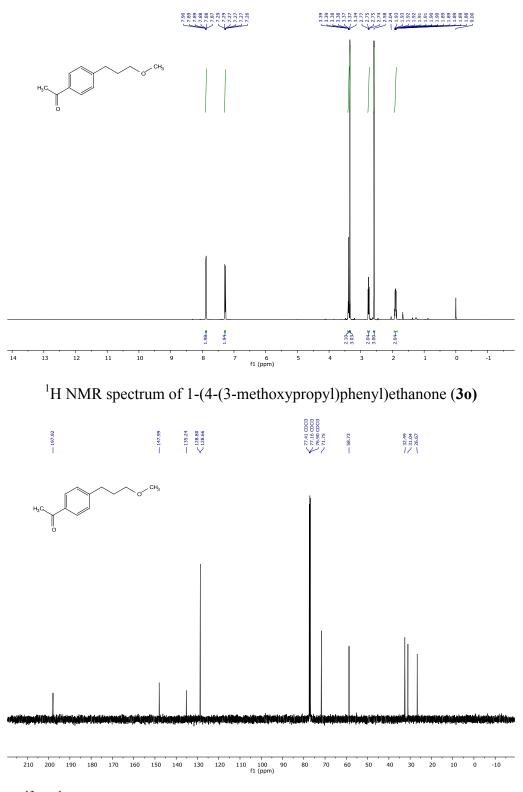
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 1-(4-isobutylphenyl)ethanone (**3m**)



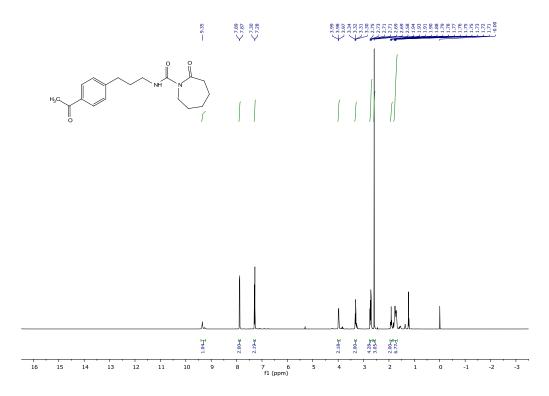
<sup>1</sup>H NMR spectrum of 1-(4-(2-(cyclohex-3-en-1-yl)ethyl)phenyl)ethanone (**3n**)



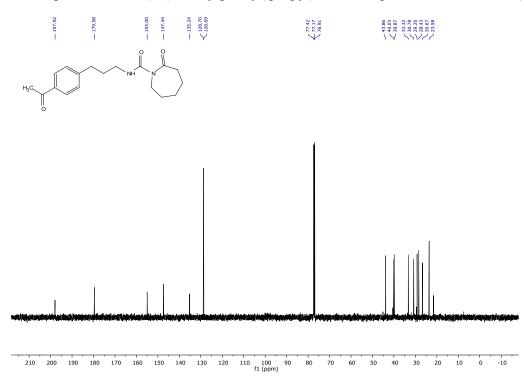
 $^{13}\text{C}$  {<sup>1</sup>H} NMR spectrum of 1-(4-(2-(cyclohex-3-en-1-yl)ethyl)phenyl)ethanone (**3n**)



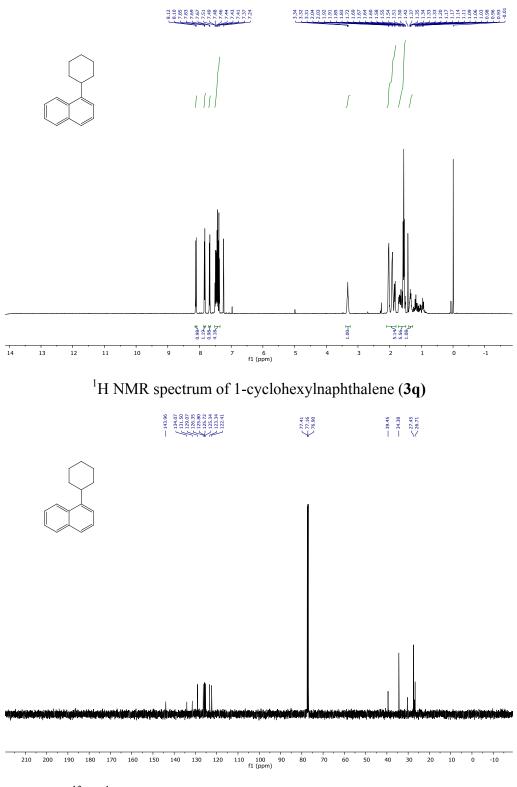
 $^{13}C$  { $^{1}H$ } NMR spectrum of 1-(4-(3-methoxypropyl)phenyl)ethanone (30)



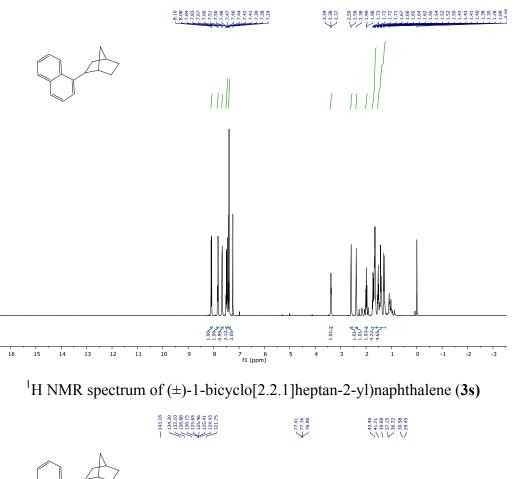
<sup>1</sup>H NMR spectrum of N-(3-(4-acetylphenyl)propyl)-2-oxoazepane-1-carboxamide (**3p**)

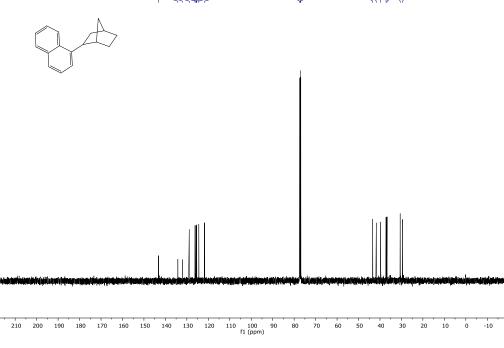


 $^{13}$ C { $^{1}$ H} NMR spectrum of N-(3-(4-acetylphenyl)propyl)-2-oxoazepane-1-carboxamide (**3p**)

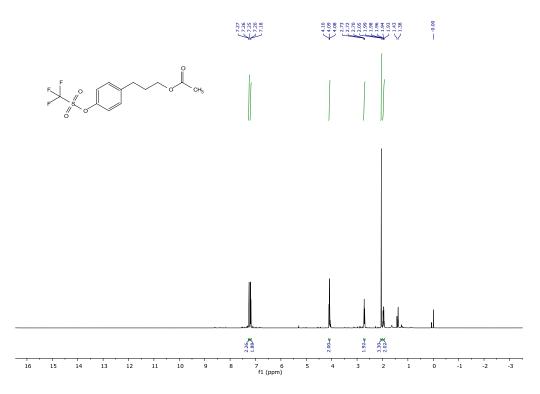


 $^{13}C$  {<sup>1</sup>H} NMR spectrum of 1-cyclohexylnaphthalene (**3q**)

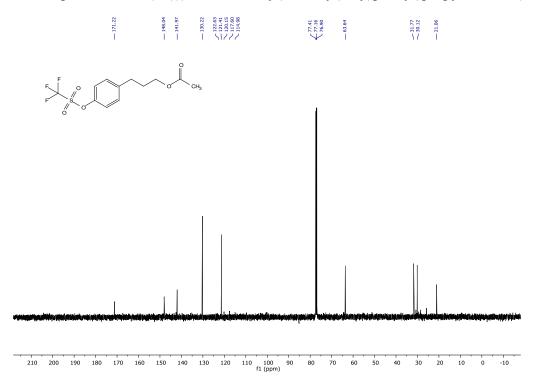




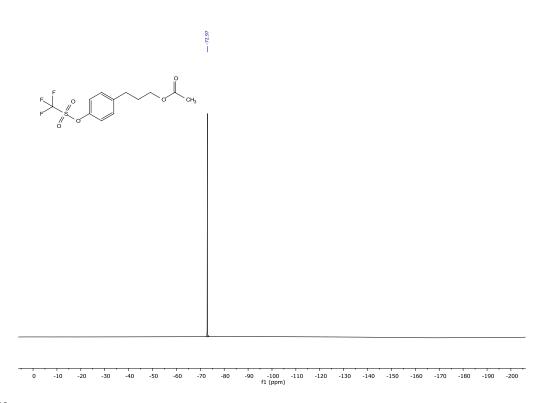
 $^{13}C$  {<sup>1</sup>H} NMR spectrum of (±)-1-bicyclo[2.2.1]heptan-2-yl)naphthalene (**3s**)



<sup>1</sup>H NMR spectrum of 3-(4-(((trifluoromethyl)sulfonyl)oxy)phenyl)propyl acetate (**3t**)



 $^{13}C$  { $^{1}H$ } NMR spectrum of 3-(4-(((trifluoromethyl)sulfonyl)oxy)phenyl)propyl acetate (**3t**)



<sup>19</sup>F NMR spectrum of 3-(4-(((trifluoromethyl)sulfonyl)oxy)phenyl)propyl acetate (**3t**)