Supplemental data

Figure captions

Figure S1. Photographs, conceptual diagram, SEM and EDX images of Fe-(oxyhydr)oxide chimneys collected from Urashima site. Conceptual diagram and photographs of Fe-rich chimney structure at Urashima site are shown in (A).

SEM images (B, C, E, F, H, I) and EDS spectrums (D, G, J) of the Fe-(oxyhydr)oxide samples collected from chimney of Urashima site are shown. Mint Green crosses indicate the EDS analysis areas in (C), (F) and (I). Only one spot (spot no. 001) data of each EDS analysis are shown in (D), (G) and (J). (B), (E) and (H) are same view. Scale bars, 10 μm. (C), (F) and (I) are same view. Scale bars, 20 μm.

Figure S2. Rarefaction curves from 16S rRNA gene clone libraries.

Phylotypes were defined at 97% sequence similarity. Rarefaction analysis (Heck et al., 1975 (1)) was performed using the DOTUR program (Schloss and Handelsman, 2005 (2)) to evaluate the phylotype richness of microbial communities in the iron mats.



Figure S1



Figure S2

Genus	Mariprofundus	Gallionella				
Class	Zetaproteobacteria	Betaproteobacteria				
Species	M. ferrooxydans	G. ferruginea,	G. capsiferriformans			
Feature of Strains:	PV-1 (3, 4)	(5-8)	ES-2 (9)			
Isolate source	Deep-sea Hydrothermal sediment	Well water, Groundwater	Groundwater			
Morphology	Curved rods	Bean-shapes or curved rods	Bean-shapes or curved rods			
Stalks	+	+	+			
Length (µm)	1.0 - 4.0	1.5-2.5	~0.8			
Width (µm)	0.3 - 0.6	0.3-1.0	ND			
Motility	+	+	+			
Temperature range (optimum) for growth (°C)	10-30 (25-30)	5-25(20)	4-30(ND)			
pH range (optimum) for growth	5.5-7.2 (6.2-6.5)	5.0-6.5 (ND)	5.5-7.5 (6.0-6.5)			
Salinity range (optimum) for growth (% NaCl)	1.5-4.5 (2.5)	ND	ND			
Energy sources:						
Fe(II)	+	+	+			
Thiosulfate, sulfide	_	+	_			
Other lithotrophy	_	_	_			
Organic matter	_	+	_			
Doubling time (h)*	12	10	12			
G+C content (mol%)	54	35.2-37.6	52.7			

TABLE S1. A summary of the differences between genus Mariprofundus and genus Gallionella.

ND: not determined

*Doubling time: In the proliferation when using iron as an energy source.

Sample Element	TMm	NMS	SAM	SIM Mean % Weight	UCSW	UCM*	UCI**
С	4.2	9.6	6.9	10.1	5.6	9.1	5.1
0	18.9	21.3	36.3	42.1	14.1	39.4	19.0
Na	12.6	3.0	5.7	3.4	5.4	3.4	13.3
Mg	0.0	0.9	0.7	0.5	0.2	0.4	0.7
Al	0.4	0.1	0.0	0.1	0.1	0.1	0.1
Si	6.8	7.4	21.8	22.8	5.4	14.2	20.2
S	1.0	0.4	0.0	0.0	0.0	0.0	0.0
Cl	12.5	5.2	10.6	2.5	3.6	2.5	11.7
Ca	1.6	1.1	0.6	0.4	0.3	0.2	0.2
Fe	41.7	51.1	17.4	18.2	38.3	30.9	29.7
Ag	0.4	0.0	0.0	0.0	27.1	0.0	0.0

TABLE S2. Elemental compositions of iron mat samples determined by SEM-EDS analysis.

Showed relative element concentrations for the iron mat samples.

*Chimney interior part at Urashima site

**Chimney most interior parts at Urashima site

	TMm	NMI	NMS	SAM	SIM	UCSW
Aciditiobacillus	1		3		1	
Alteromonadales	5	6	1	1	1	1
BD-8 marine group			1			
Chromatiales	1	14	3	2		
Escherichia shigella			1			
Coxiellaceae				1		1
Methylococcales				5	1	19
Halomonadaceae		1				1
Pseudospirillum		2				
Salinicola		1				
Actinobacter	3	2				
Piscirickettsiaceae		16	1			
Leucothrix			1			
Other gamma	3		9		3	
Vibrio		1	1			
Sinobacteraceae	1		1	2	6	
Total	14	43	22	11	12	23

TABLE S3. Further taxonomic affiliation of gammaproteobacterial 16S rRNA gene clones from iron mat samples in different deep-sea hydrothermal systems.

TABLE S4. Further taxonomic affiliation of deltaproteobacterial 16S rRNA gene clones from iron mat samples in different deep-sea hydrothermal systems.

	TMm	NMI	NMS	SAM	SIM	UCSW
43F-1404R	1			2	1	
Bacteriovoraceae		1			1	1
Bdellovibrionaceae				1	1	1
Desulfobulbaceae				1		7
Desulfurellaceae				2		3
DTB120	21		1	1	4	1
GR-WP33-30	5		1		2	
Myxococcales			4	2	4	2
SAR324 clad			2			
Sh765B-TzT-29			2	1		
Total	33	1	12	12	14	16

TABLE S5. Further taxonomic affiliation of alphaproteobacterial 16S rRNA gene clones from iron mat samples in different deep-sea hydrothermal systems.

	TMm	NMI	NMS	SAM	SIM	UCSW
Caulobacteraceae	1	3				
MNG3					1	
Uncultured					3	
Parvulaculaceae					1	
Rhodobacteraceae	2		1	1		5
Rhodospirillaceae	2	7	6	1	8	1
Rickettsiaceae	1					1
S26-47			1			
SAR11 crade		2		1	2	
SB1-18					1	
Sphingomonadacea				2		
Total	6	12	8	5	16	7

TABLE S6. Chemical compositions^I of hydrothermal fluid and/or pore-water of iron mat samples in the Fox site of Tarama knoll, the Yellow Top site and Champagne vent site of NW Eifuku field, the Snail and Urashima site of the southern Mariana Trough.

Sampling site	Sampling	T _{max} ^{II}	pН	Mg	Ca	$\mathrm{NH_4}^+$	SO_4^{2-}	H_2S	H_2	CO_2	CH_4
	year	°C		(mM)	(mM)	(µM)	(mM)	(mM)	(µM)	(mM)	(µM)
Fox site "	2010	24	6.4	48.8	12.0	98.2	25.6	ND	2.9	12.7	5.2
Yellow Top site	2010	2.8	7.4	ND^{IV}	ND	<100	ND	0.026	ND	ND	ND
Champagne vent site ^v	2004	103	ND	43.0	ND	ND	ND	12.5	< 0.2	3000	< 0.2
Champagne vent site ^v	2005	108	ND	43.0	ND	ND	ND	30.0	< 0.2	800	< 0.2
Urashima site ^{VI}	2010	76.0	5.7	39.2	13.3	13.1	22.2	0.0	0.5	11.4	4.9
Snail site ^{VI}	2010	64.5	6.9	42.5	11.1	8.9	25.3	0.0	0	3.6	0.0

^I Measured at 25°C

^{II} T_{max}: Maximum temperature

^{III} Gas data was collected in 2011 (NT11-18 cruise; Makita et al. (10))

^{IV} ND: No data

^v Estimated end-member composition by Lupton et al. (11), based on extrapolating to Mg concentration 43

mM. The Champagne vent fluids had pH ranging from 3.4 to 4.8. The low end of measured pH of Champagne vent fluids was consistent with CO_2 buffering in the end-member fluid.

^{VI} A detailed discussion of Snail and Urashima site can be found in Toki et al. (12).

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