

Supporting Information for;

Co-operative Transitions of Responsive-Polymer Coated Gold Nanoparticles; Precision Tuning and Direct Evidence for Co-operative Aggregation

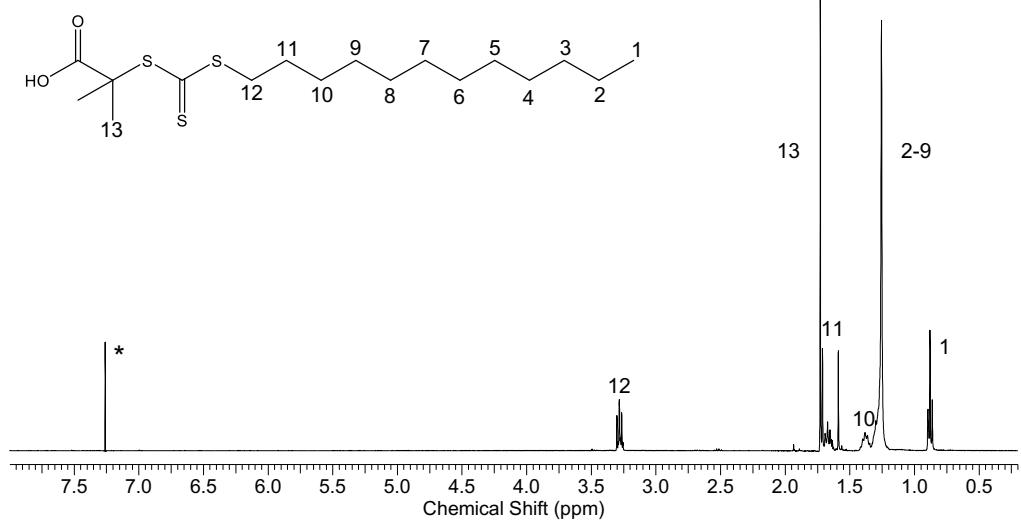
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Additional Characterisation Data

A



B

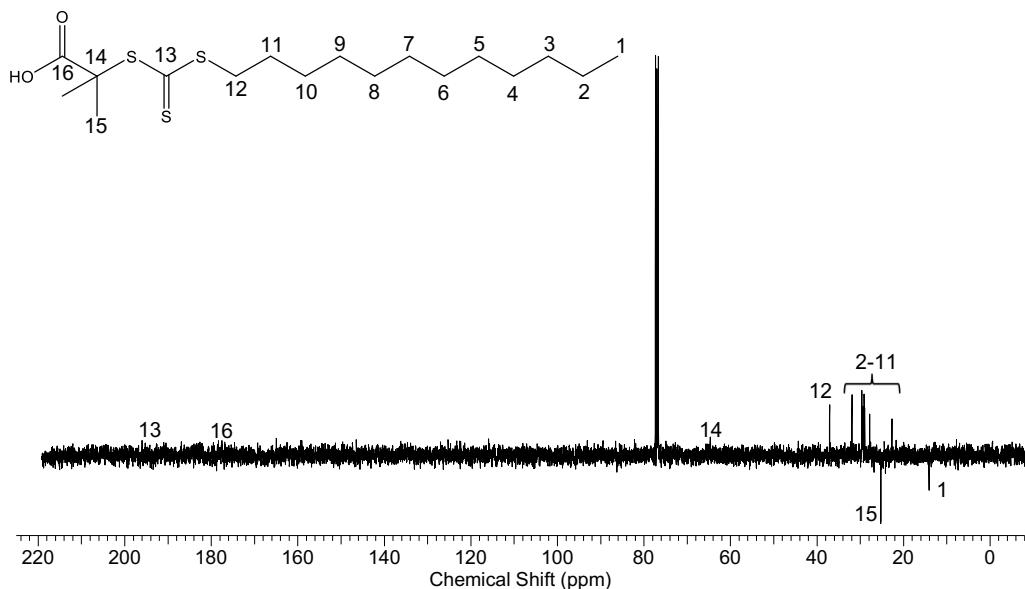


Figure S1. ^1H NMR and ^{13}C NMR spectra of 2-(dodecylthiocarbonothioylthio)-2-methylpropanoic acid in CDCl_3 . * = solvent

^1H NMR (400 MHz, CDCl_3) δ_{ppm} : 3.28 (2H, t, $J_{12-11} = 7.54$ Hz, H¹²); 1.71 (6H, s, H¹³); 1.67 (2H, p, $J_{11-10, 11-12} = 7.53$ Hz, H¹¹); 1.38 (2H, p, $J_{10-9, 10-11} = 7.78$ Hz, H¹⁰); 1.26 (16H, m, H²⁻⁹); 0.88 (3H, t, $J_{1-2} = 7.03$ Hz, H¹).

¹³C NMR (400 MHz, CDCl₃) δ_{ppm}: 196.00 (C¹³); 177.56 (C¹⁶); 64.69 (C¹⁴); 37.09 (C¹²); 31.92, 29.71, 29.63, 29.56, 29.45, 29.35, 29.12, 22.70 (C²⁻⁹); 28.97 (C¹⁰); 27.81 (C¹¹); 25.25 (C¹⁵); 14.13 (C¹).

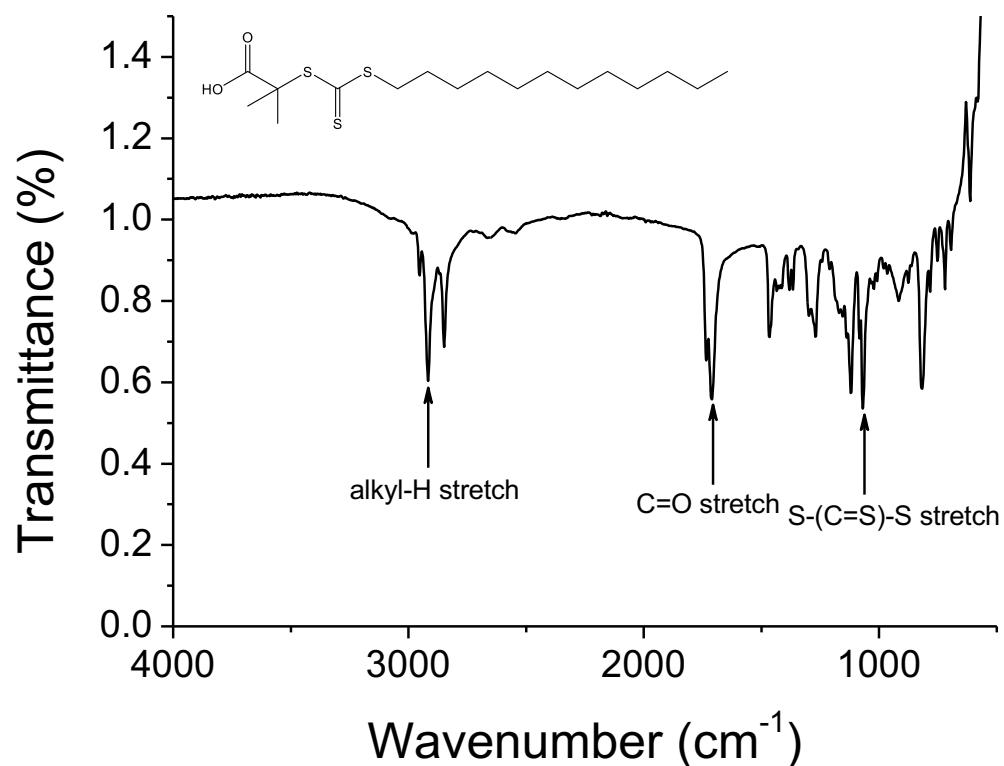


Figure S2. Infrared spectra of 2-(dodecylthiocarbonothioylthio)-2-methylpropanoic acid.

IR cm⁻¹: 2917 (alkyl-H stretch); 1712 (C=O stretch); 1070(S-(C=S)-S stretch).

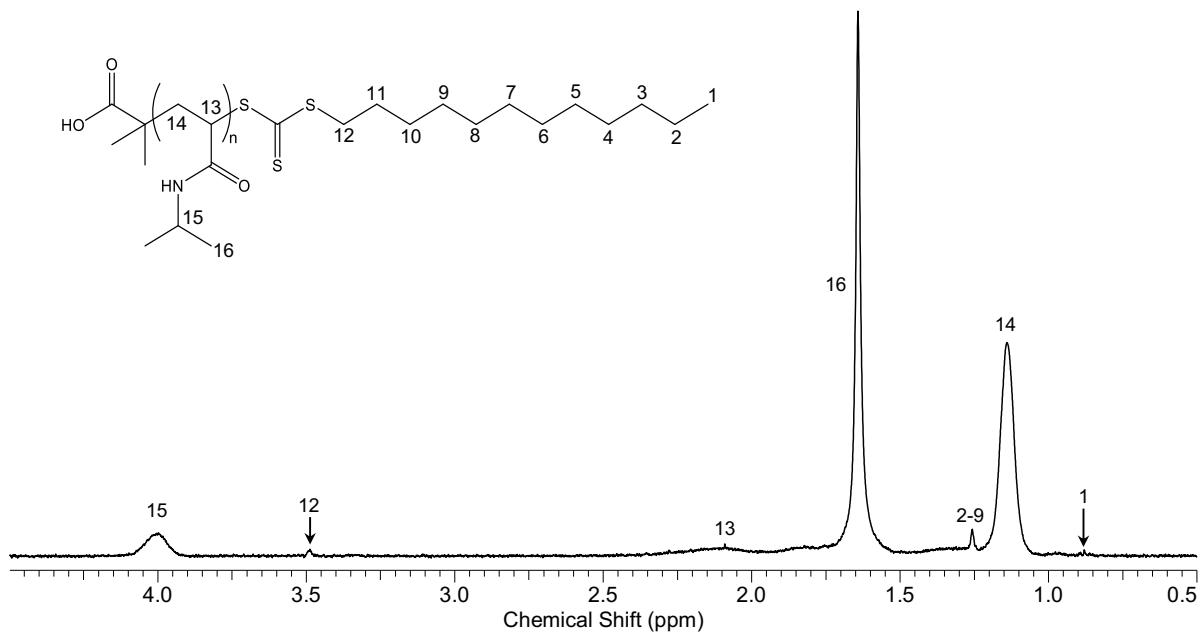


Figure S3. ^1H NMR spectra of poly(N-isopropylacrylamide).

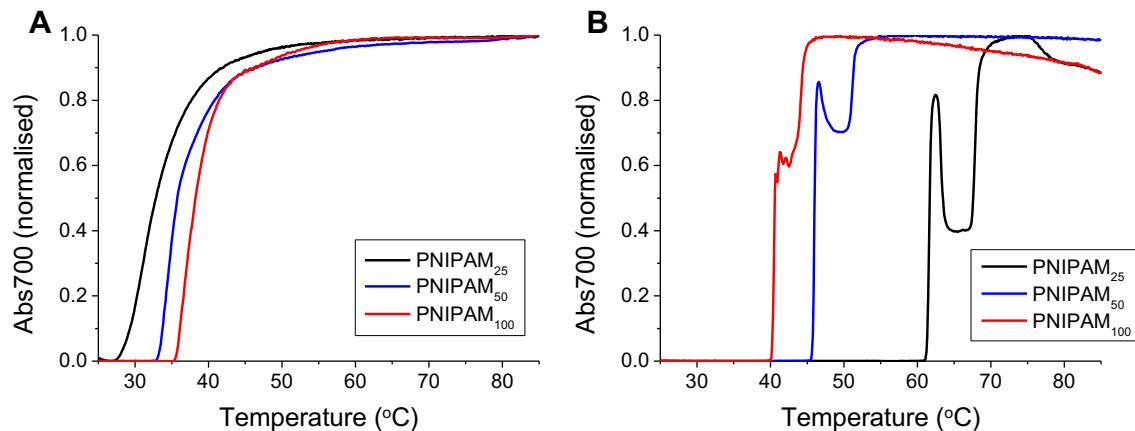


Figure S4. Turbidimetry scans (absorbance at 700 nm) of homopolymer in a) pure water and b) PBS. In all cases the total polymer concentration of the solutions was 2.5 mg mL⁻¹.

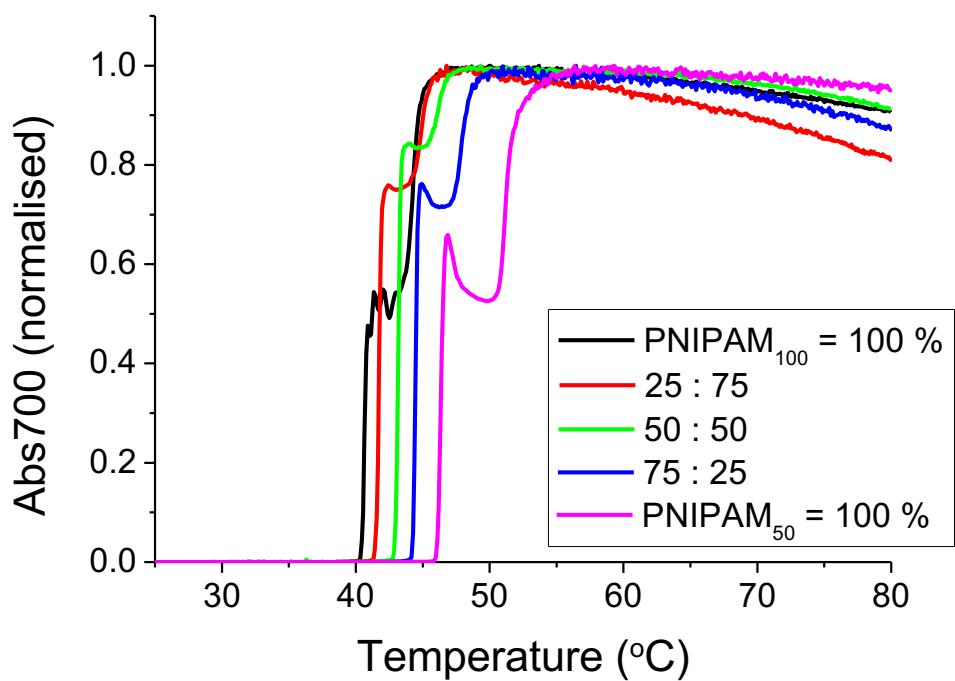


Figure S5. Turbidimetry analysis of blends of PNIPAM₅₀ and PNIPAM₁₀₀ at 2.5 mg mL⁻¹.

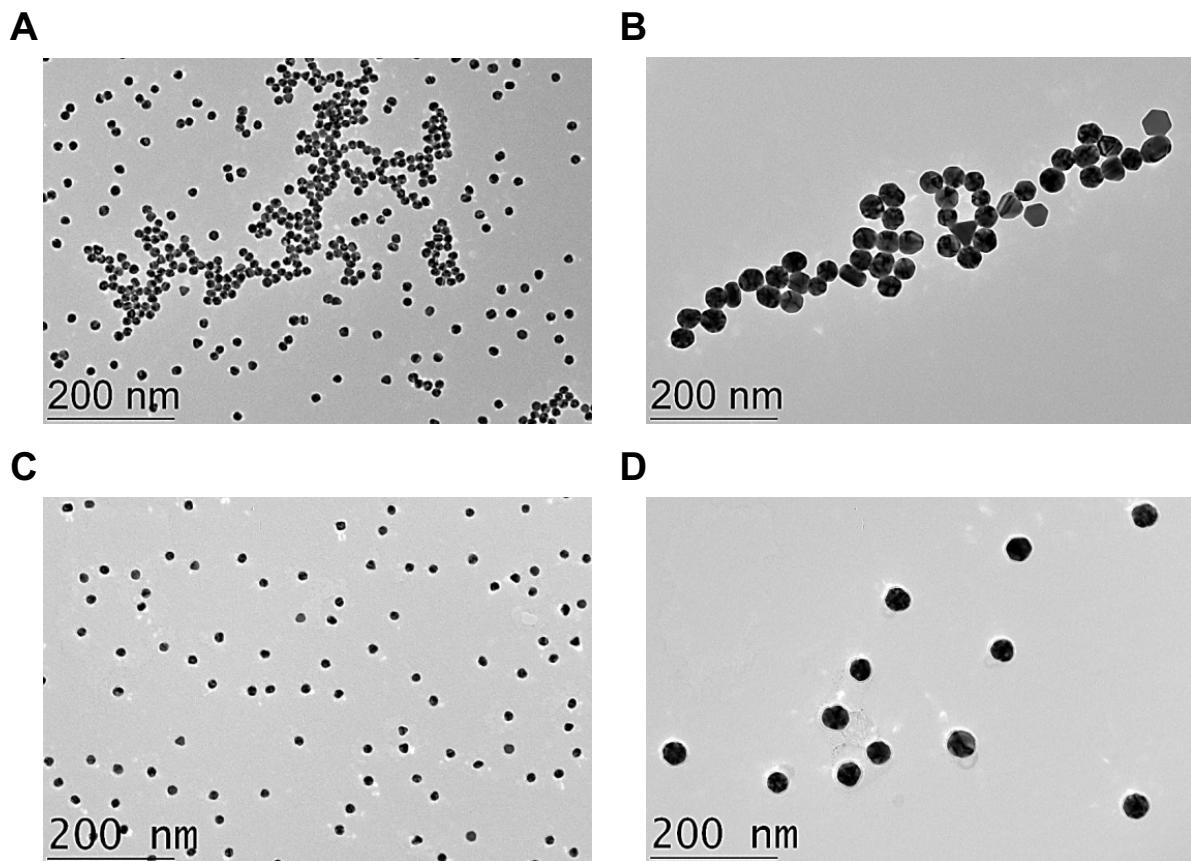


Figure S6. TEM images of uncoated a) 15nm and b) 40nm sized bare gold nanoparticles and PNIPAM₁₀₀ coated c) 15nm and d) 40nm sized gold nanoparticles (higher colloidal stability).

Table S1. XPS elemental ratios for PNIPAM/gold hybrid nanoparticles.

| Sample | Au 4f [%] | C 1s [%] | O 1s [%] | N 1s [%] | Cu 2p [%] | S 2p [%] |
|---|-----------|----------|----------|----------|-----------|----------|
| Bare Au₁₅ | 2.54 | 45.24 | 41.33 | 0.07 | 10.83 | - |
| PNIPAM₁₀₀@Au₁₅ | 1.66 | 75.38 | 17.79 | 3.80 | 1.36 | 0.02 |
| PNIPAM₅₀@Au₁₅ | 2.6 | 70.35 | 19.83 | 4.83 | 2.36 | 0.02 |
| PNIPAM₂₅@Au₁₅ | 3.53 | 70.68 | 18 | 4.79 | 2.99 | 0.01 |
| Cu foil reference | - | - | - | 59.94 | 40.06 | - |

| Sample | Au 4f [%] | C 1s [%] | O 1s [%] | N 1s [%] | Cu 2p [%] | S 2p [%] |
|---|-----------|----------|----------|----------|-----------|----------|
| Bare Au₄₀ | 1.00 | 17.81 | 16.27 | 0.03 | 4.26 | - |
| PNIPAM₁₀₀@Au₄₀ | 1.00 | 45.41 | 10.72 | 2.29 | 0.82 | 0.012 |
| PNIPAM₅₀@Au₄₀ | 1.00 | 27.06 | 7.63 | 1.86 | 0.91 | 0.008 |
| PNIPAM₂₅@Au₄₀ | 1.00 | 20.02 | 5.10 | 1.36 | 0.85 | 0.003 |
| Cu foil reference | - | - | - | 59.94 | 40.06 | - |

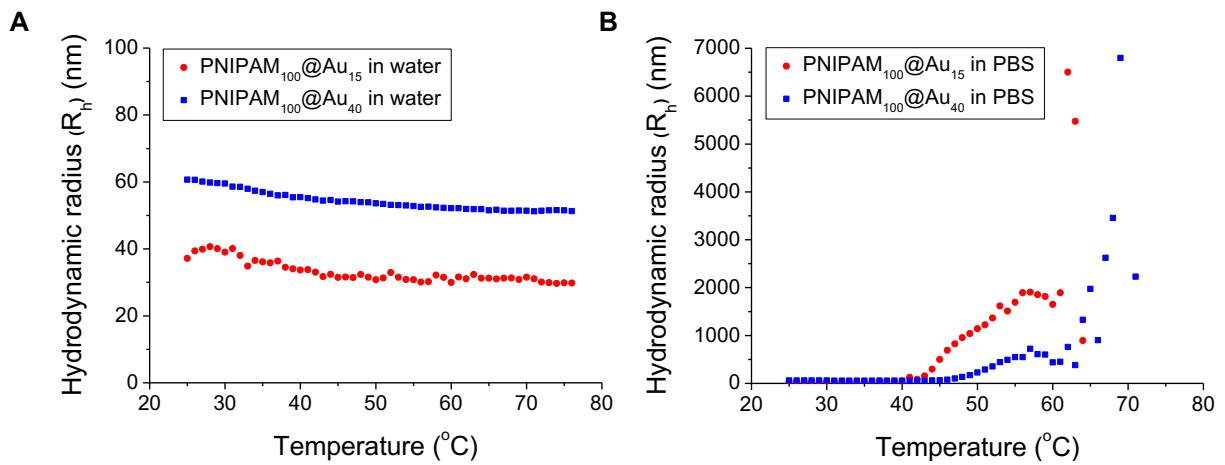


Figure S7. Hydrodynamic radius (R_h) for the PNIPAM coated 15 nm and 40 nm gold nanoparticles in a) pure water and b) PBS buffer as a function of temperature.

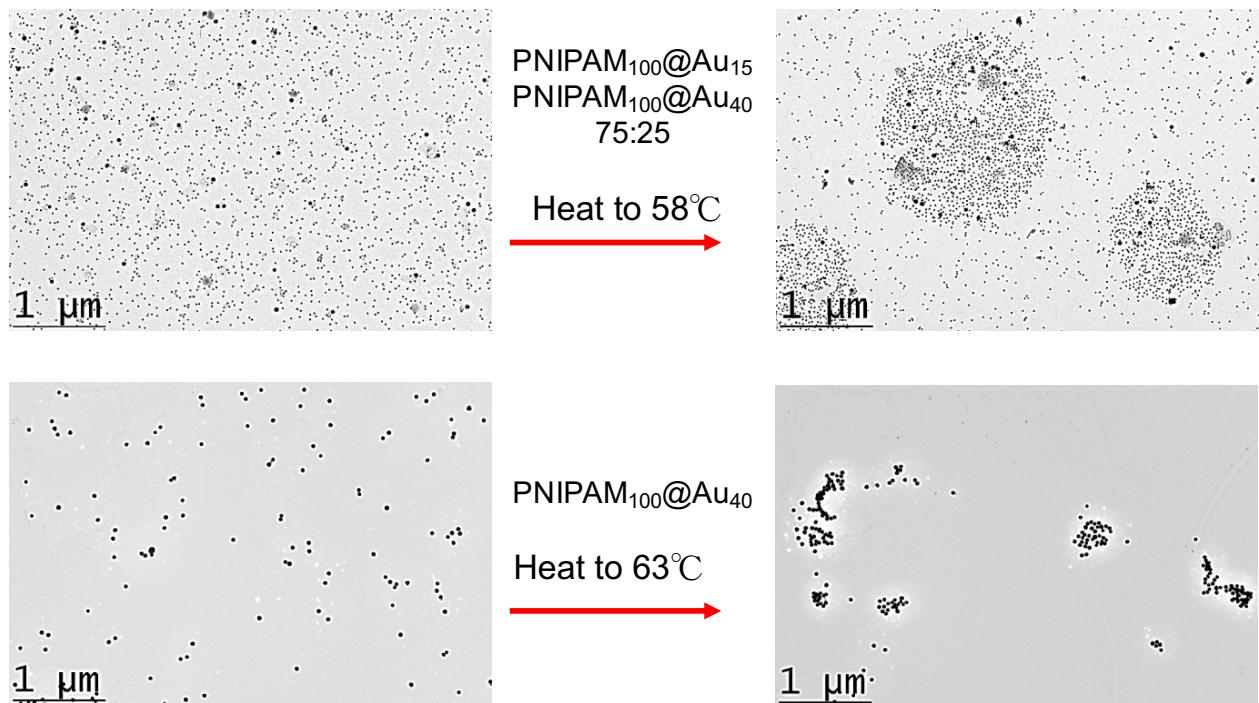


Figure S8. Additional TEM analysis of co-operative particle aggregation. Left hand column shows nanoparticles at 25 °C (below their cloud point). Right hand column shows the same particles which were prepared above cloud point of the 15 nm particles (58 °C) and 40 nm particles (63 °C).