A sensitive electrochemical sensor for *in vitro* detection of parathyroid hormone based on MoS₂-graphene composite

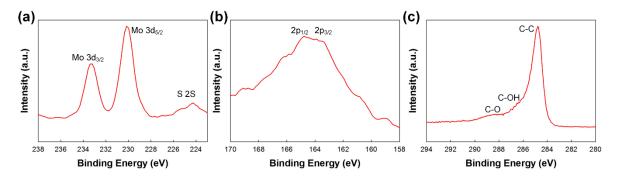
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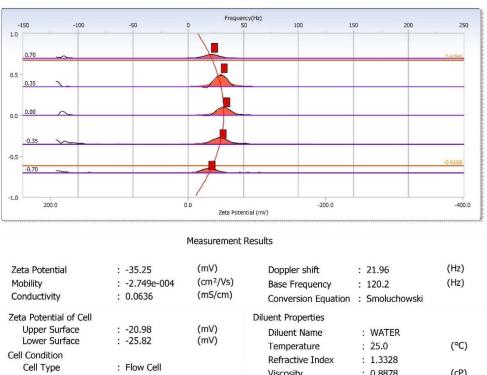
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Supplementary Fig. S1 High resolution XPS spectra corresponding to MG composite (a) Mo 3d, S 2p (b) 2p and (c) C 1s



(V/cm)

(mA)

Viscosity

Dielectric Constant

: 0.8878

: 78.3

(cP)

EOS Plot

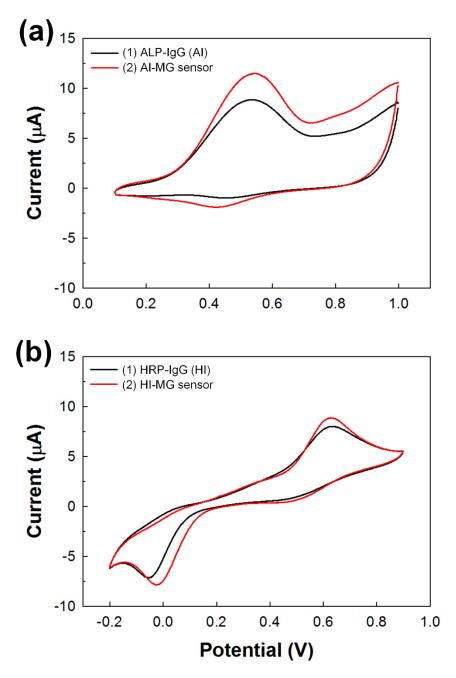
Supplementary Fig. S2 Zeta potential of MG composite

: -16.54

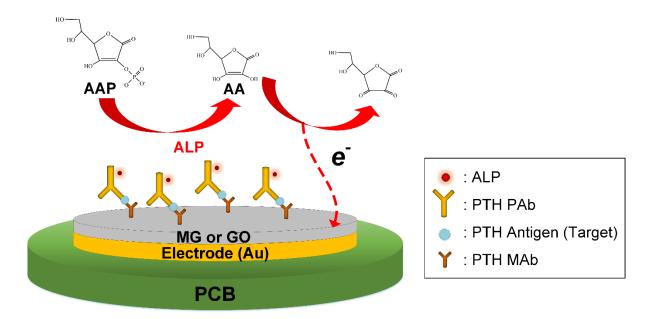
: -0.05

Avg. Electric Field

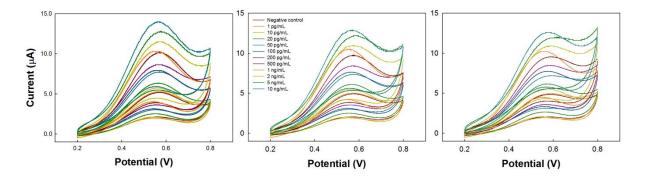
Avg. Current



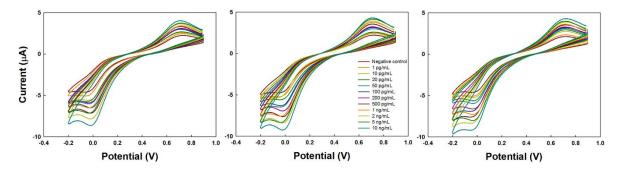
Supplementary Fig. S3 Cyclic voltammogram (CV) measurement of modified electrodes (a) (1) AI electrode (2) AI-MG electrode at scan rate 100 mV/s (b) (1) HI electrode (2) HI-MG electrode at scan rate 100 mV/s, respectively



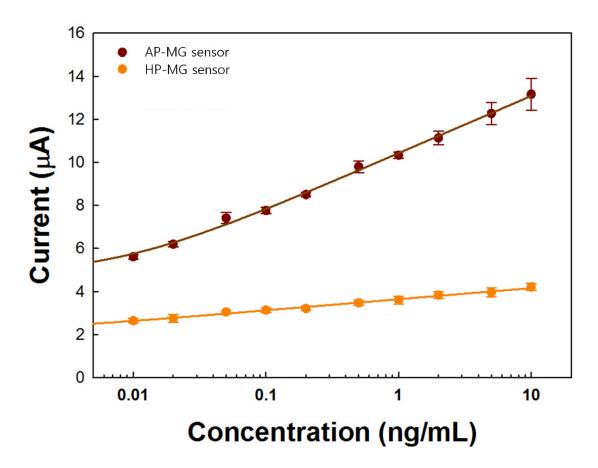
Supplementary Fig. S4 Mechanism of AP-GO and AP-MG sensors with AAP substrate



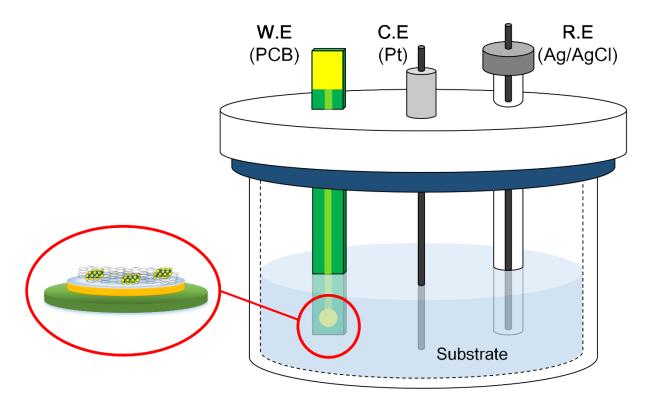
Supplementary Fig. S5 Three cycling of AP-MG sensor with different concentration of PTH



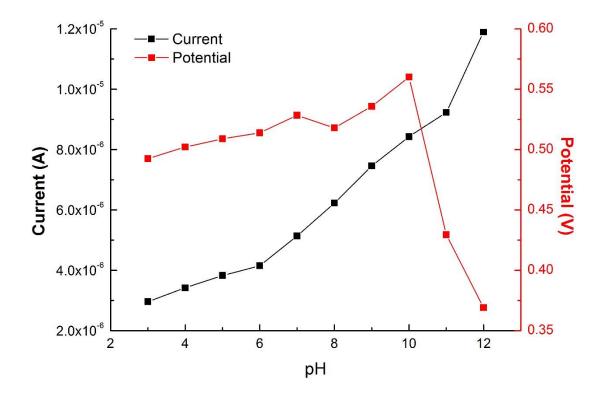
Supplementary Fig. S6 Three cycling of HP-MG sensor with different concentration of PTH



Supplementary Fig. S7 PTH calibration curve obtained at PTH concentration ranging from 1 pg/mL to 10 ng/mL by both AP-MG and HP-MG sensors



Supplementary Fig. S8 Schematic representation of experimental setup wherein Au-PCB as working electrode, Pt as counter electrode and Ag/AgCl (Sat'd 3M KCl) as reference electr ode, respectively. MG is deposited on Au-PCB electrode



Supplementary Fig. S9 Effect of (AAP solution in 50 mM Tris-HCl + 10 mM MgCl₂) pH on A P-MG sensor. The maximum potential (0.55 V) is observed at pH 9.6