

Supplementary Information

Gas Protection of Two-Dimensional Nanomaterials from High-Energy Impacts

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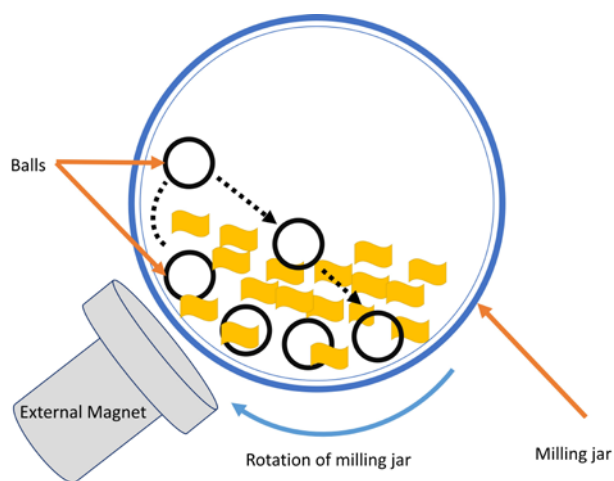


Figure S1. Schematic of high-energy ball milling device

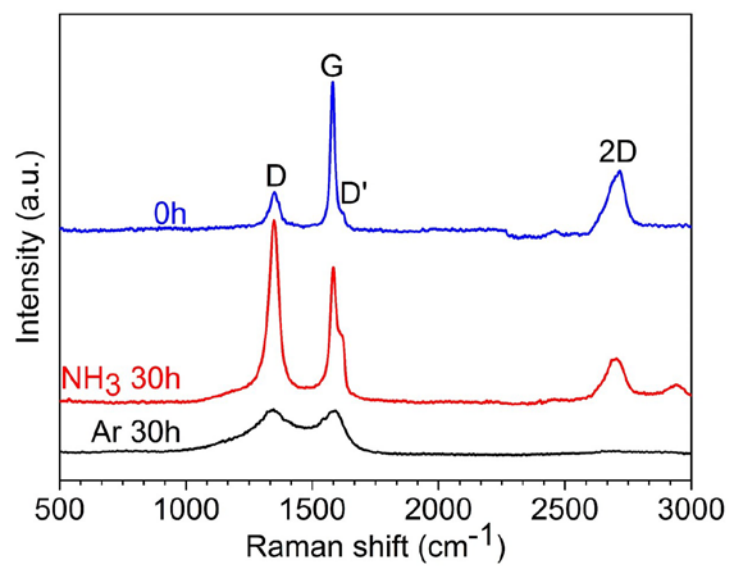


Figure S2. Raman spectra of graphite with and without ball milling treatment.

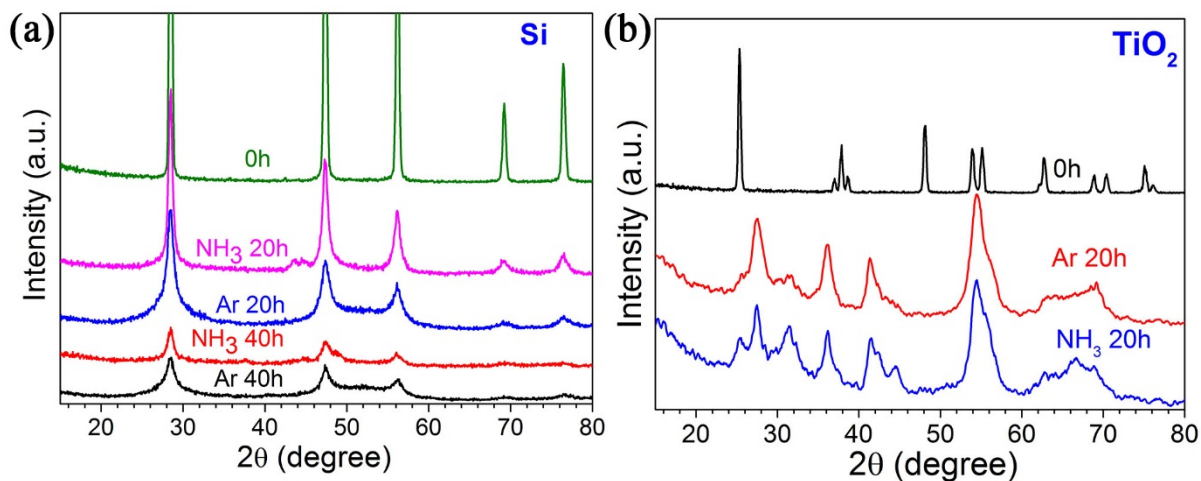


Figure S3. XRD patterns of Si (a) and TiO₂ (b) milled in different gases. The similar XRD patterns indicate the same structures produced in different gases.

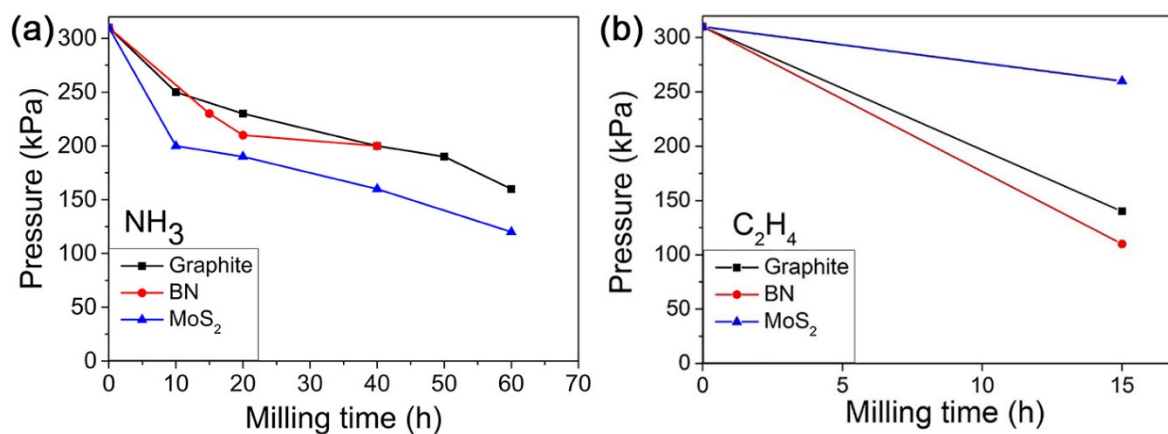


Figure S4. (a) NH₃ and (b) C₂H₄ pressure change during milling of different materials. The pressure reduction in the sealed milling chamber indicates gas absorption into the milled materials continuously over the milling process.

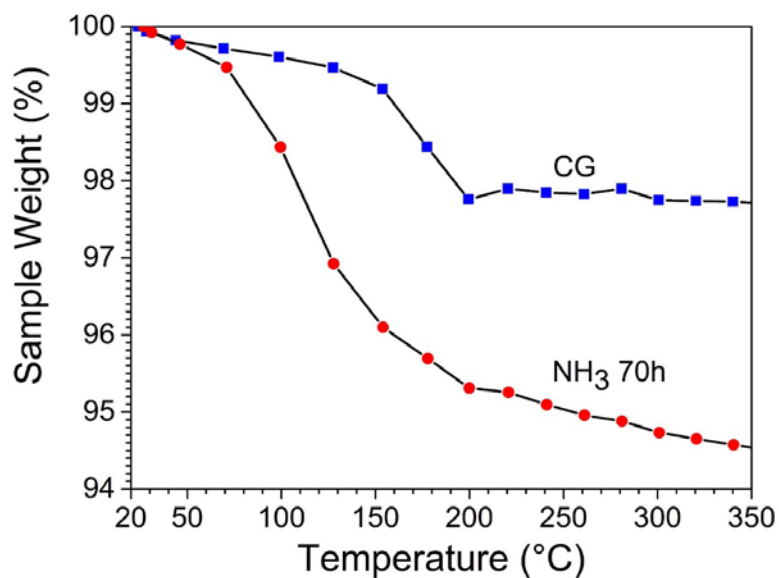


Figure S5. Thermal gravimetric (TG) curves of commercial graphite (CG) and the graphite milled for 70 h in NH₃ showing release of the adsorbed gases in different temperature ranges due to their different gas adsorption (physisorption or chemisorption) nature

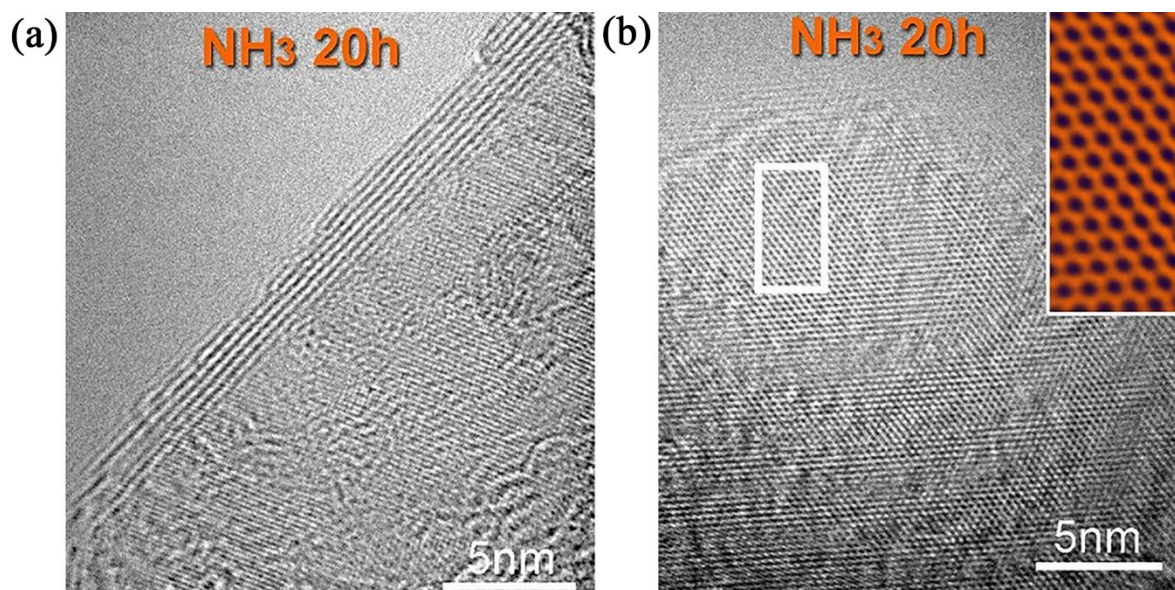


Figure S6. HR-TEM images of BN nanosheets produced by ball milling in NH₃ for 20 h.

(a) The edges of BN nanosheets; (b) in-plane structure. The inset shows the hexagonal rings from the reversed FFT image of part of (b).

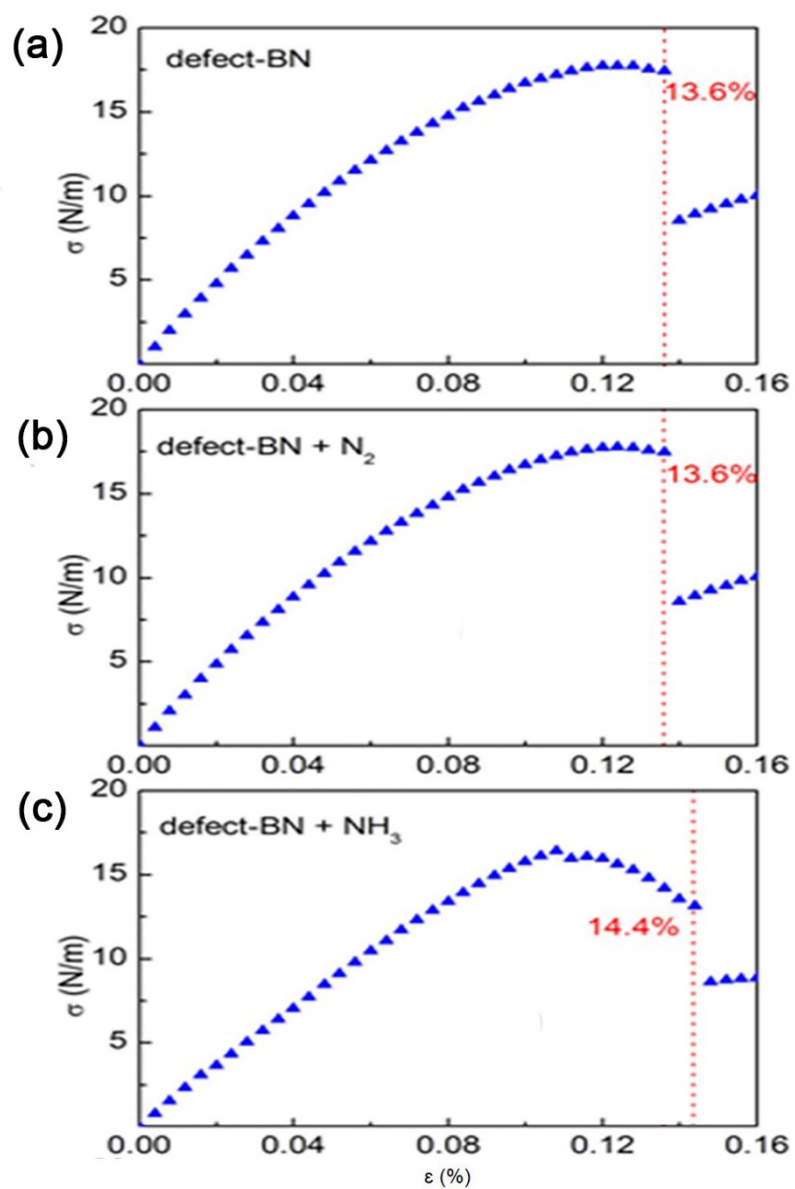


Figure S7. Simulated stress-strain curves for (a) a defected BN monolayer and a defected BN monolayer in the presence of (b) N_2 and (c) NH_3 attachment.