SUPPLEMENTARY INFORMATIONS

A cinnamon-derived procyanidin compound displays anti-HIV-1 activity by blocking heparan sulfate- and co-receptor- binding sites on gp120 and reverses T cell exhaustion via impeding Tim-3 and PD-1 upregulation

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Figure A

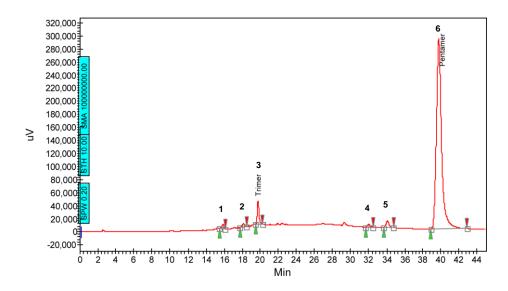


Figure A: HPLC analysis of IND02. The IND02, isolated from cinnamon extract, was analysed by HPLC, and followed by UV detection at 280 nm, demonstrating the presence of procyanidin trimer (~ 4 % of the eluted material) and pentamer polyphenols (~ 92 % of the eluted material). Black triangles indicate the beginnings and ends of each peak (see Table A). Unknown material (peak 1, 2, 4 and 5) represents 4.1 % of the total UV trace.

Table A

Peak N°	Identity	Start	Time	End	Height	Area	Area
		(min)	(min)	(min)	(uV)	(uV.Min)	(%)
1	unknown	15.48	15.86	16.11	6623.4	1643.5	0.784
2	unknown	17.74	18.13	18.49	5099.6	1525.8	0.728
3	Trimer	19.43	19.72	20.24	38086.6	8250.9	3.937
4	unknown	31.65	31.98	32.53	4307.3	1415.4	0.675
5	unknown	33.66	34.07	34.78	10499.4	4075.1	1.945
6	Pentamer	38.92	39.82	42.93	293862.9	192653.9	91.931
Total					358479.1	209564.5	100.000

The table shows the chromatographic characteristics of the cinnamon-isolated IND02. Peak numbers referred to Figure A.

Figure B

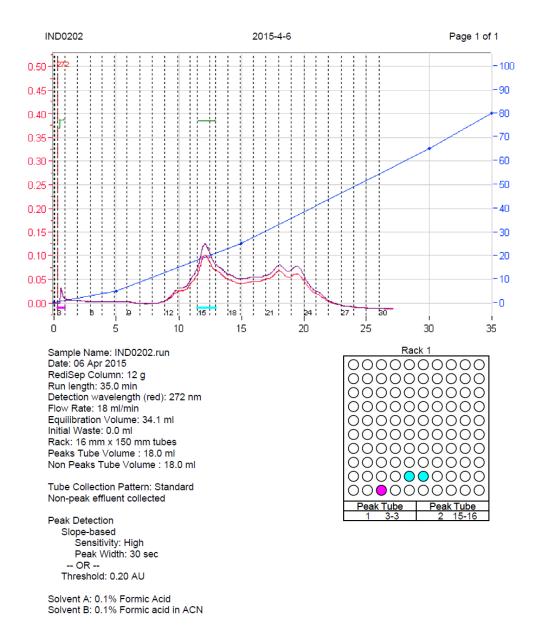


Figure B: HPLC analysis of IND02-trimer. The sample (blue fraction on the chromatogram) was analyzed for Optical rotation, λmax in UV spectrum, and NMR spectroscopy on 600 MHz (Bruker Biospin Instrument). All the experiments were performed at 25° C, using 10 mg sample for ¹H1 NMR and 50 mg sample for ¹³C NMR. The observed values were compared with literature reported values of cinnamtannin B1 (Table B below). The structure predicted using observed values matched with that of literature reported Cinnamatanin B1 [1,2] and indicated the purity of the trimer to be 99%. This was also confirmed using Quantum Mechanics and Neuronal network calculations.

Table B

Ring	No	13C		1H		
		Observed Reported		Observed	Reported	
(upper unit) C	C-2	100.2	99.95		i	
	C-3	67.2	67.17	3.18	3.29	
	C-4	28.86	28.86	4.06	4.15	
A	C-5	154.2	156.74	1.00		
	C-6	98.3	98.33	5.87	5.97	
	C-7	157.9	157.8			
	C-8	96.6	96.6	5.92	6.02	
	C-9	154.2	154.14	1 217 2		
	C-10	105	104.9			
В	C -1'	132.5	132.44			
	C-2'	119.9	119.90	6.93	7.03	
	C-3'	145.56	145.5			
	C-4'	146.6	146.59			
	C-5'	115.8	116.18	6.75	6.84	
	C-6'	115.9	115.75	6.76	6.86	
Middle unit F	C-2	78.9	78.86	5.60	5.70	
	C-3	72.6	72.55	4.02	4.13	
	C-4	38.3	38.27	4.45	4.56	
D	C-5	155.8	155.76			
	C-6	96.1	96.09	5.70	5.80	
	C-7	151.1	151.08			
	C-8	106.8	106.42			
	C-9	151.8	151.78			
	C-10	106.8	106.73			
Е	C-1'	131.8	131.76			
	C-2'	116.8	116.72	7.21	7.32	
	C-3'	145.9	145.89			
	C-4'	146.3	146.27			
	C-5'	116.2	115.79	6.73	6.82	
	C-6'	121.4	121.36	7.10	7.19	
Lower unit	C-2	80.4	80.27	4.28	4.38	
I	C-3	67.6	65.71	3.76	3.85	
	C-4	29.9	29.84	2.74	2.83	
G	C-5	155.6	155.99			
	C-6	96.5	96.51	6.01	6.10	
	C-7	155.6	155.53			
	C-8	108.9	108.85			
	C-9	156.1	155.78			
	C-10	100.1	100.08			
Н	C-1'	132.2	133.17			
	C-2'	115.5	115.48	6.72	6.82	
	C-3'	145.4	145.31			
	C-4'	145.8	145.74			
	C-5'	116	116.03	6.64	6.76	
	C-6'	119.5	119.45	6.63	6.72	

The table shows the comparison of the NMR output (600 MHz) for ¹³C and ¹H in the observed experiment versus the reported literature, confirming the identity of the compound to be Cinnamatanin B1.

Note:

- 1. While prediction methodology relied on standard solvent chemical shift to predict the chemical shifts, the observed chemical shifts were referenced to TMS [(CH₃)₄Si], the standard deviation between the predicted and the observed chemical shifts is 0.095 ppm.
- 2. IND02-trimer has fourteen -OH groups and all of them appear at 4.52 ppm as a broad peak. They are not listed in the above table.

Supplementary reference

- 1. Ben Amor N, Bouaziz A, Romera-Castillo C, Salido S, Linares-Palomino PJ, et al. (2007) Characterization of the intracellular mechanisms involved in the antiaggregant properties of cinnamtannin B-1 from bay wood in human platelets. J Med Chem 50: 3937-3944.
- 2. Kamiya K, Watanabe C, Endang H, Umar M, Satake T (2001) Studies on the constituents of bark of Parameria laevigata Moldenke. Chem Pharm Bull (Tokyo) 49: 551-557.