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Supporting Information

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A Biominerization Strategy for “Net”-Like Interconnected TiO₂ Nanoparticles Conformably Covering Reduced Graphene Oxide with Reversible Interfacial Lithium Storage

*Qiang Zhang, Yong Yan, and Ge Chen**

Supporting Information

Title

Bio-mineralization Strategy toward “Net”-like Interconnected TiO₂ Nanoparticles Conformably Covering Reduced Graphene Oxide with Reversible Interfacial Lithium Storage

*Qiang Zhang, Yong Yan[‡] and Ge Chen**

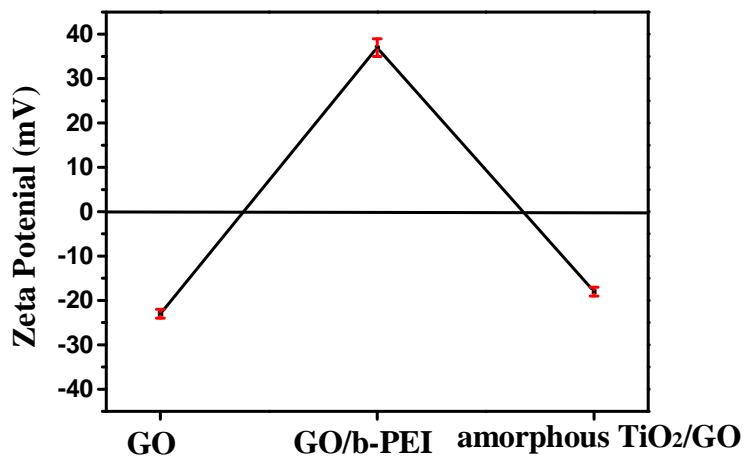


Figure S1. Zeta potentials of GO, GO/b-PEI, and amorphous TiO₂/GO composites.

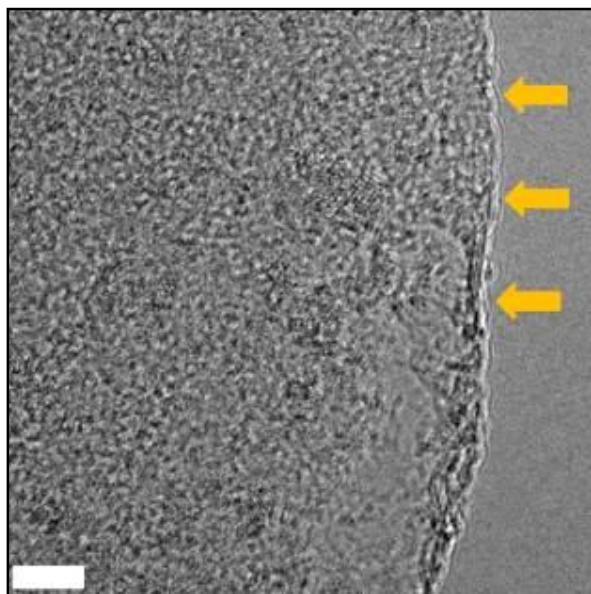


Figure S2. HRTEM image of amorphous TiO_2 /GO composite; the edge of GO marked by the orange arrow indicates the ultrathin nature (3–4 layers), bar 5 nm.

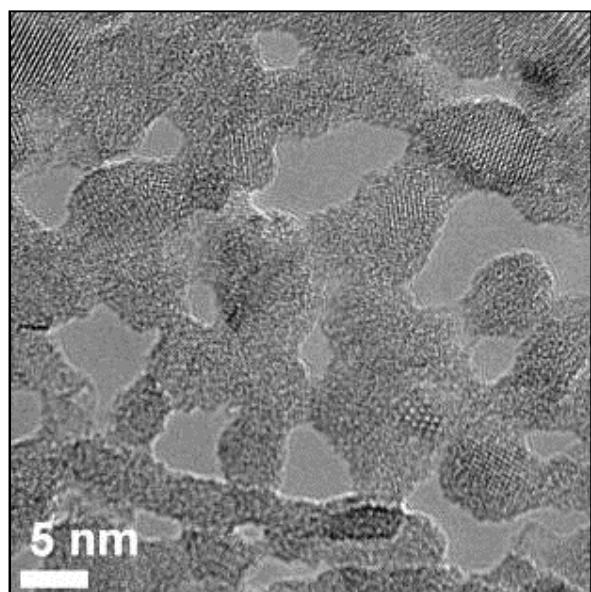


Figure S3. HRTEM image of anatase TiO_2 /RGO composite, indicating the "net"-like structure of TiO_2 conformably covering RGO, bar 5 nm.

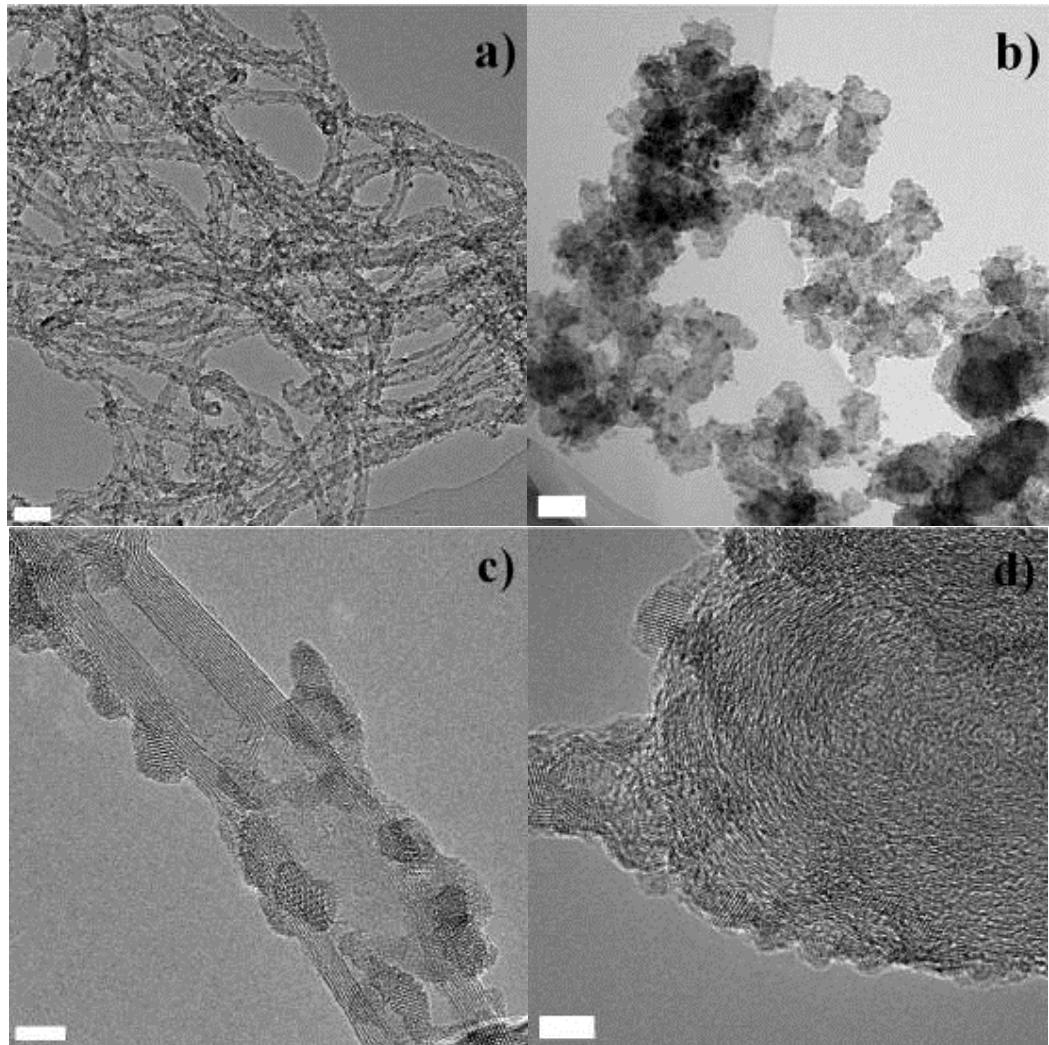


Figure S4. a) TEM image of anatase TiO₂/CNTs, bar 50 nm. b) TEM image of anatase TiO₂/carbon spheres (XC-72), bar 50 nm. c) HRTEM image of anatase TiO₂/CNTs, bar 5 nm. d) HRTEM image of anatase TiO₂/carbon spheres (XC-72), bar 5 nm.

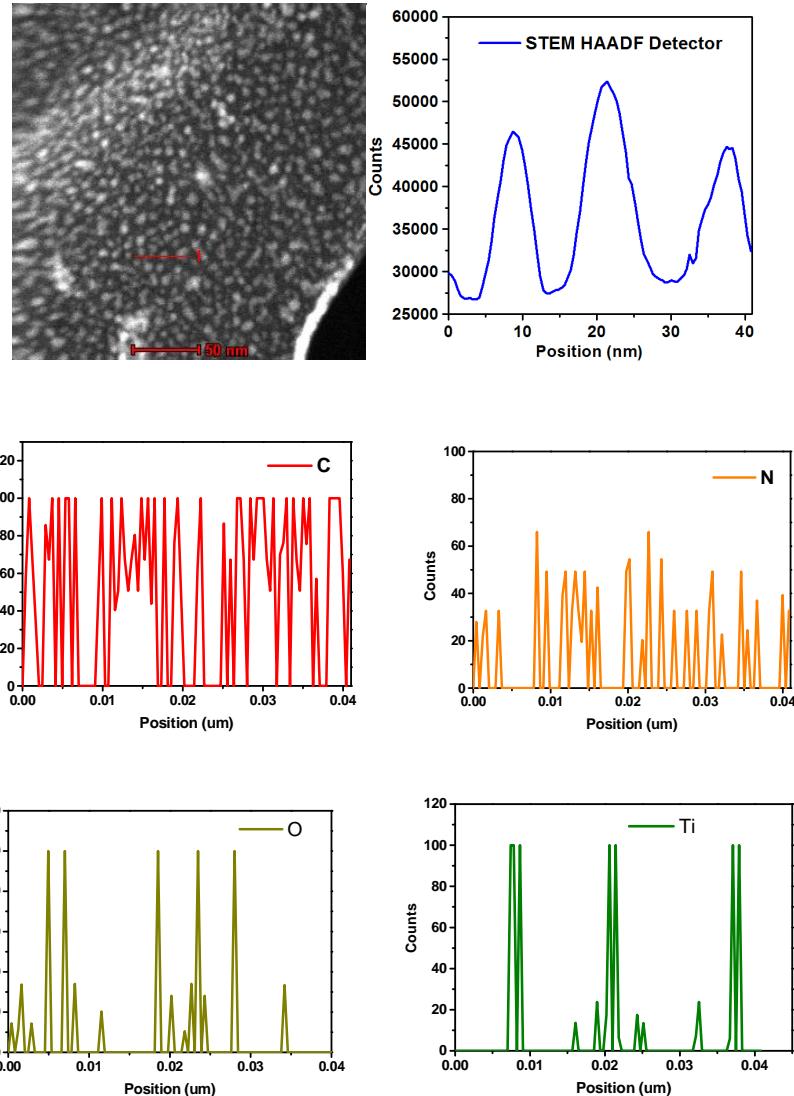


Figure S5. Energy-dispersive X-ray spectroscopy (EDS) line scan analysis based on the STEM-HADDF model demonstrating the distribution of C, N, O, and Ti species; while the Ti and O species are dispersed on the particles, the C and N species are dispersed along the line.

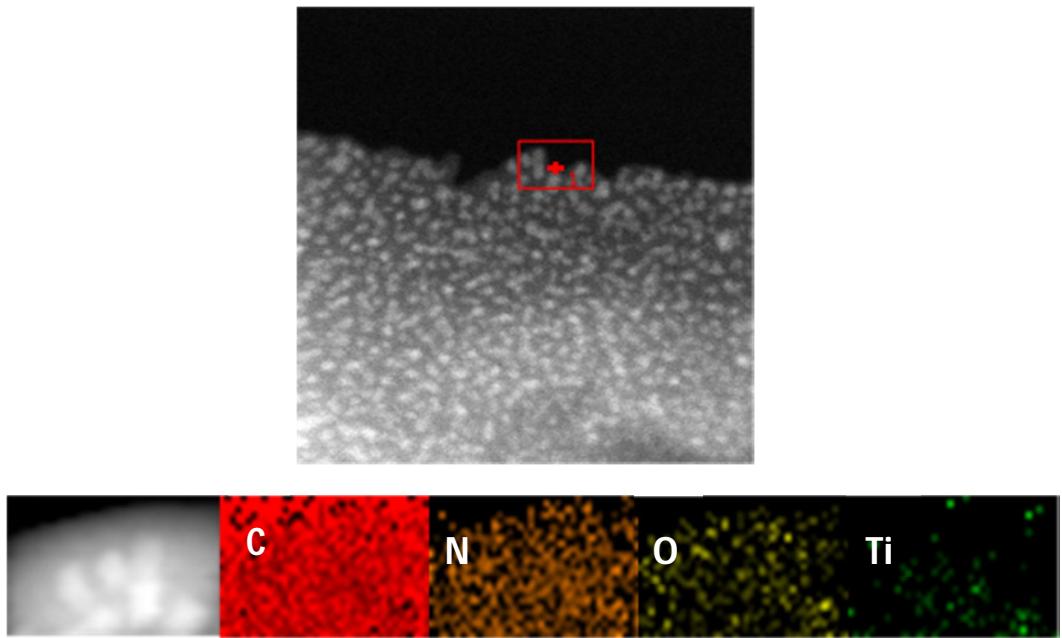


Figure S6. EDS elemental mapping of C, N, O, and Ti species for the anatase TiO_2 /RGO composite.

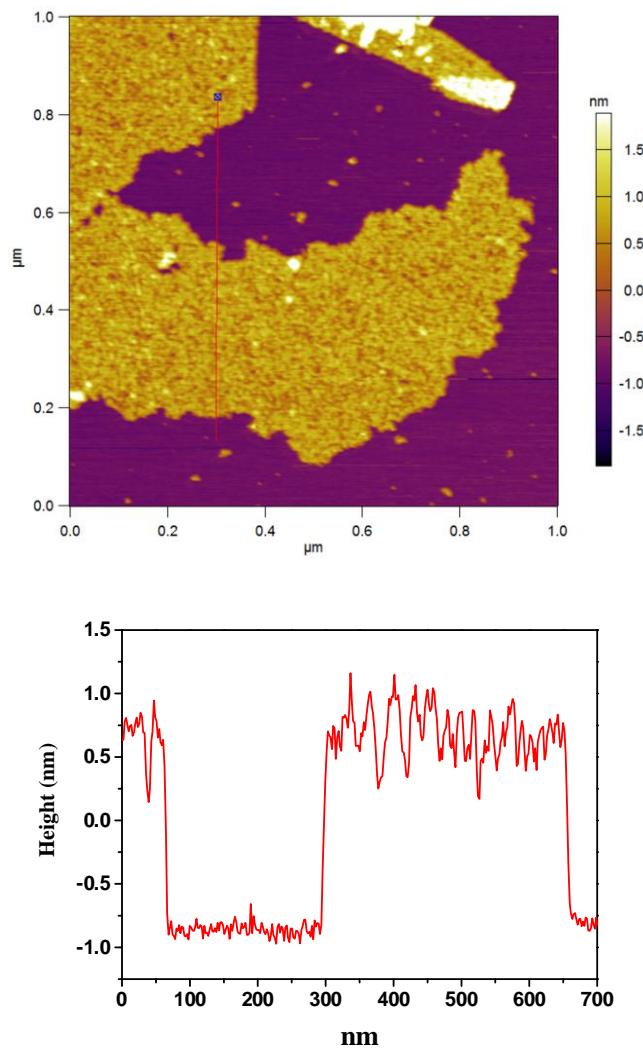


Figure S7. Topography image of GO on a mica substrate and the corresponding height-profile analysis along the line.

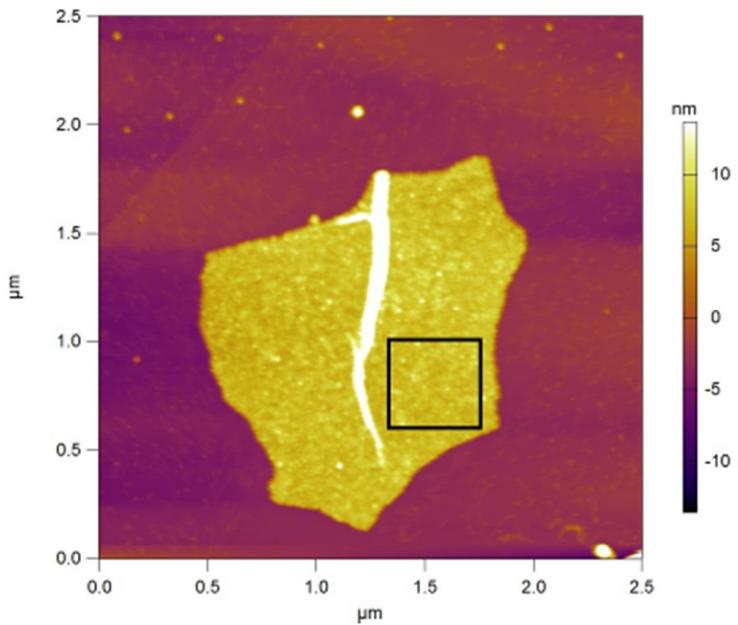


Figure S8. Topography image of anatase TiO₂/RGO on the HOPG substrate; the black square region is ready for current image collecting.

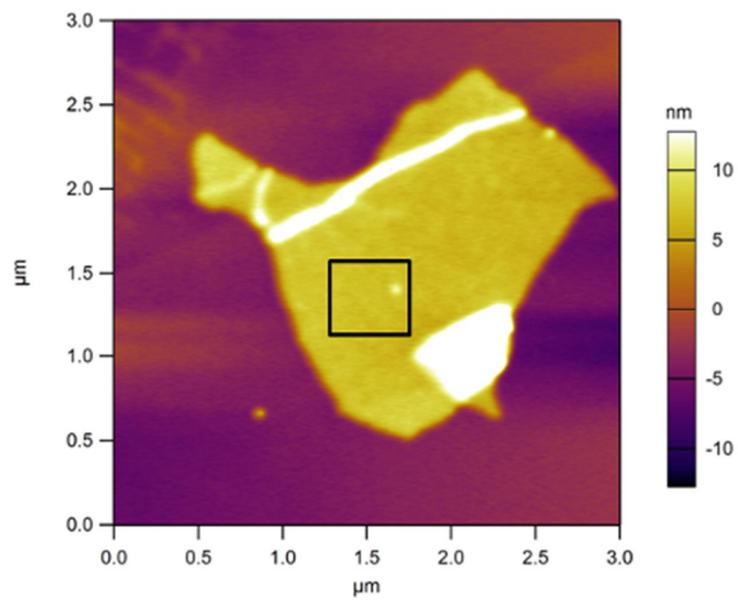


Figure S9. Topography image of amorphous TiO₂/GO on the HOPG substrate; the black square region is ready for current image collecting.

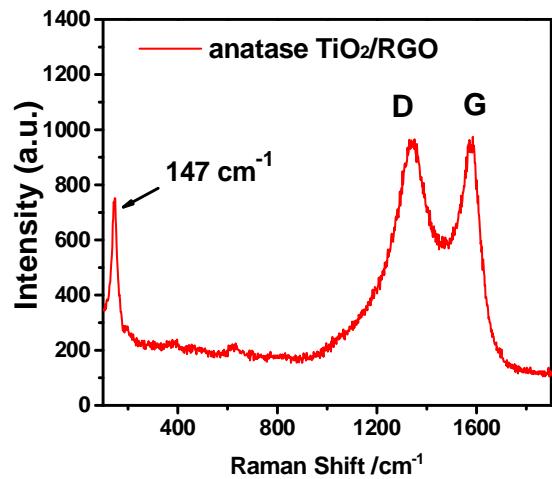


Figure S10. Raman spectrum of the anatase TiO_2 /RGO composite.

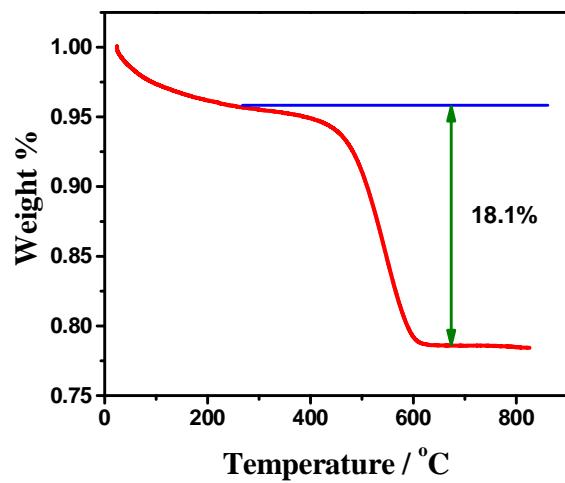


Figure S11. TGA analysis of the anatase TiO_2 /RGO composite in air.

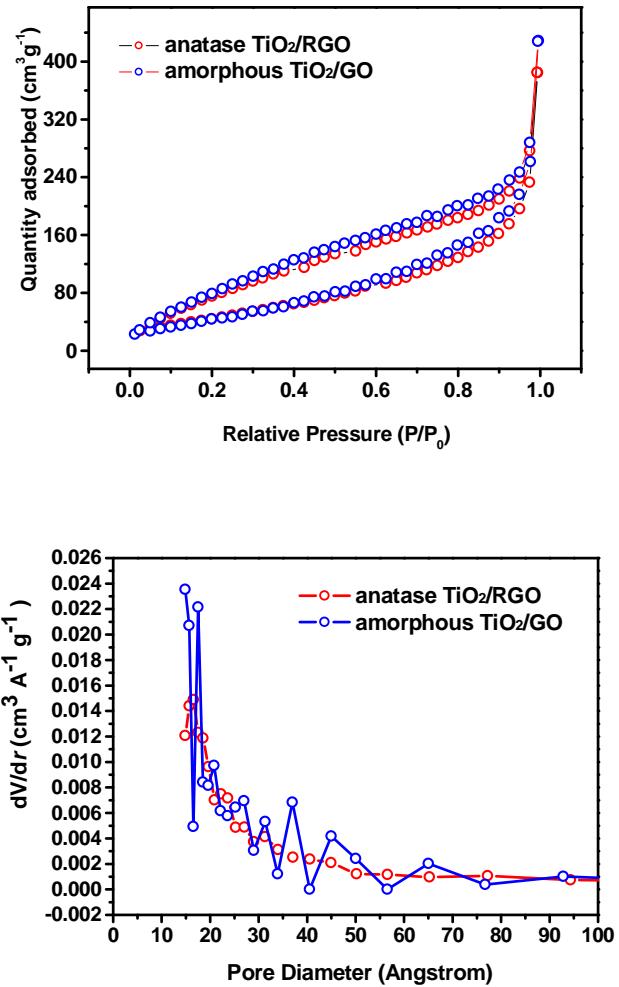


Figure S12. N₂ adsorption/desorption isotherms of amorphous TiO₂/GO and anatase TiO₂/RGO composites and corresponding pore size distribution results analyzed by the BJH method.

Table S1 Comparison of electrochemical performance of obtained anatase TiO₂/RGO with other relative anodes materials.

Materials	Discharge density	Capacity (mAh g ⁻¹)	Cycle numbers	Reference
Anatase TiO ₂ /RGO	5C	151	50	Current study
Anatase TiO ₂ -FGS	1C	160	100	[9]
TiO ₂ -RGO	5C	175	30	[11]
TiO ₂ -RGO	5.9C	143	35	[16]
TiO ₂ /GAs	5.9C	130	40	[18]
Graphene-TiO ₂	5C	130	100	[10]
TiO ₂ -RGO	5C	152	100	[71]
TiO ₂ -GNS -CNT	10C	111	100	[72]
Graphene-TiO ₂ -C	0.59C	95	20	[73]