Supporting Information

Improved charge carrier lifetime in planar perovskite solar cells by bromine doping

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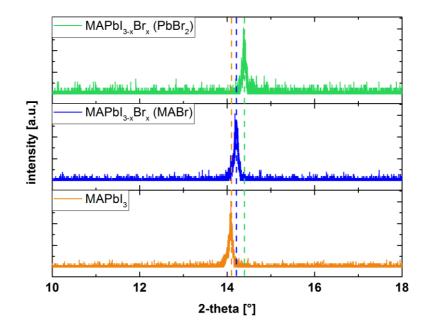


Figure S1. XRD spectra of mixed halide perovskite layers using PbI₂ + MAI (orange), PbI₂ + MAI/MABr (blue) and PbBr₂ + MAI (green) as precursor salts. Introducing bromine in the perovskite crystal leads to a decreased lattice constant.

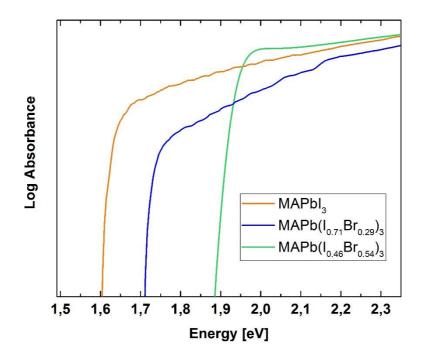


Figure S2. Optical absorption spectra for all studied perovskite films. According to Noh et al. and Gil-Escrig et al. we determined the band gap and estimated the bromine-to-iodine ratio for the MAPb($I_{1-x}Br_x$)₃ layers^{1,2}. For the PbI₂ + MAI/MABr approach we calculated x to be 0.29. In case of PbBr₂ + MAI the bromine content was increased up to 0.54.

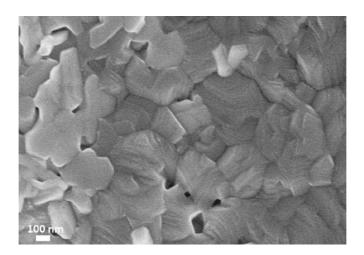


Figure S3. Surface SEM image of the resulting perovskite layer using NH₄Cl and DIO as additives in the PbI₂-solution. The additives lead to an increased crystal domain size without changing the lattice structure by slowing down the chemical reaction.

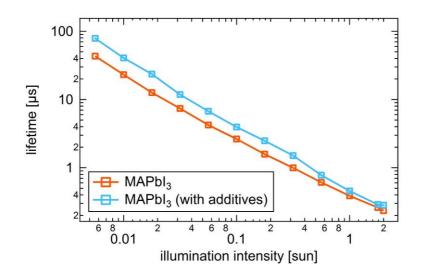


Figure S4. Impact of crystal domain size (varied by adding NH₄Cl and DIO to the PbI₂solution) on charge carrier lifetimes measured with TPV. An increase in τ with crystal domain size by a factor of 1.18 at 1 sun illumination can be observed.

References

[1] Noh, J. H., Im, S. H., Heo, J. H., Mandal, T. N. & Seok, S. I. Chemical Management for Colorful, Efficient, and Stable Inorganic–Organic Hybrid Nanostructured Solar Cells. Nano Lett. **13**, 1764-1769 (2013)

[2] Gil-Escrig, L., Miquel-Sempere, A., Sessolo, M. & Bolink, H.J. Mixed Iodide-Bromide Methylammonium Lead Perovskite-based Diodes for Light Emission and Photovoltaics. J.
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