	Atom	N^{\dagger}	$R(\text{\AA})$	$\sigma^2(\text{\AA}^2)$			
Sediment B	S	2	2.24	0.0029	-		
$(\chi^2 = 1.37)$	As	1	2.58	0.0031			
	As	2.5	3.49	0.0043	-		
	As	0.75	3.63	0.0114			
	S	2.5	3.65	0.0064			
Reference compounds:					Crystallographic values		
	Atom	N^{\dagger}	<i>R</i> (Å)	σ^2 (Å ²)	Atom	N	$R(\text{\AA})$
Realgar [‡]	S	2	2.24	0.0034	S	2	2.225-2.251
(AsS)	As	1	2.58	0.0048	As	1	2.565-2.579
$(\chi^2 = 0.54)$					S	1	3.415-3.513
	As	2.5	3.47	0.0057	As	2.5	3.439-3.499
					As	0.5	3.562
	As	0.75	3.63	0.0118	As	0.75	3.624-3.627
					S	0.75	3.570-3.617
	S	2.5	3.67	0.0078	S	2.5	3.657-3.689
Orpiment [§]	S	3	2.28	0.0040	S	3	2.243-2.307
(As_2S_3)	As	1	3.19	0.0097	As	1	3.191
$(\chi^2 = 0.24)$	S	1	3.22	0.0150	S	1	3.219, 3.291
	S	1	3.47	0.0168	S	1	3.475, 3.514
	As	1	3.52	0.0068	As	1	3.521
	As	1	3.57	0.0111	As	1	3.568
Arsenolite	0	3	1.80	0.0040	0	3	1.787
(As ₂ O ₃)	As	3	3.24	0.0035	As	3	3.223
$(\chi^2 = 0.57)$	As	6	3.88	0.017	As	6	3.946

Table 2. Arsenic K-edge extended x-ray absorption fine-structure (EXAFS) fit results^{*}

*EXAFS fit results for arsenic spectra shown in Fig. 1*b*; *N* is the number of backscattering atoms at distance (*R*); σ^2 , the Debye–Waller term, is the absorber-backscatterer mean-square relative displacement. *R* and σ^2 were varied in fits; *N* was fixed on crystallographic values. χ^2 is a reduced least-squares goodness-of-fit parameter [= (*F*-factor)/(# of points – # of variables)]. Based on empirical fits to reference

compounds, error in *R* is ± 0.02 Å; error in σ^2 is $\pm 25\%$ (1). Crystallographic values for reference compounds from independent x-ray diffraction analyses were taken from the literature.

[†]Fixed parameter.

^{*}Realgar structure from ref. 2; average of four unique arsenic sites. Disordered sulfur shells at 3.415-3.513 Å and 3.570-3.617 Å cancel and do not contribute to EXAFS amplitudes.

[§]Orpiment crystal structure from ref. 2; average of two unique arsenic site.

[¶]Arsenolite crystal structure from ref. 3; additional oxygen shells at 3.049 Å and 3.450 Å scatter too weakly to contribute to EXAFS amplitudes.

1 O'Day, P. A., Rehr, J. J., Zabinsky, S. I. & Brown, G. E., Jr. (1994) J. Am. Chem. Soc. 116, 2938–2949.

2. Mullen, D. J. E. & Nowacki, W. (1972) Z. Kristallogr. 136, 48-65.

3. Pertlik, F. (1978) Czech J. Phys. 28, 170-176.