

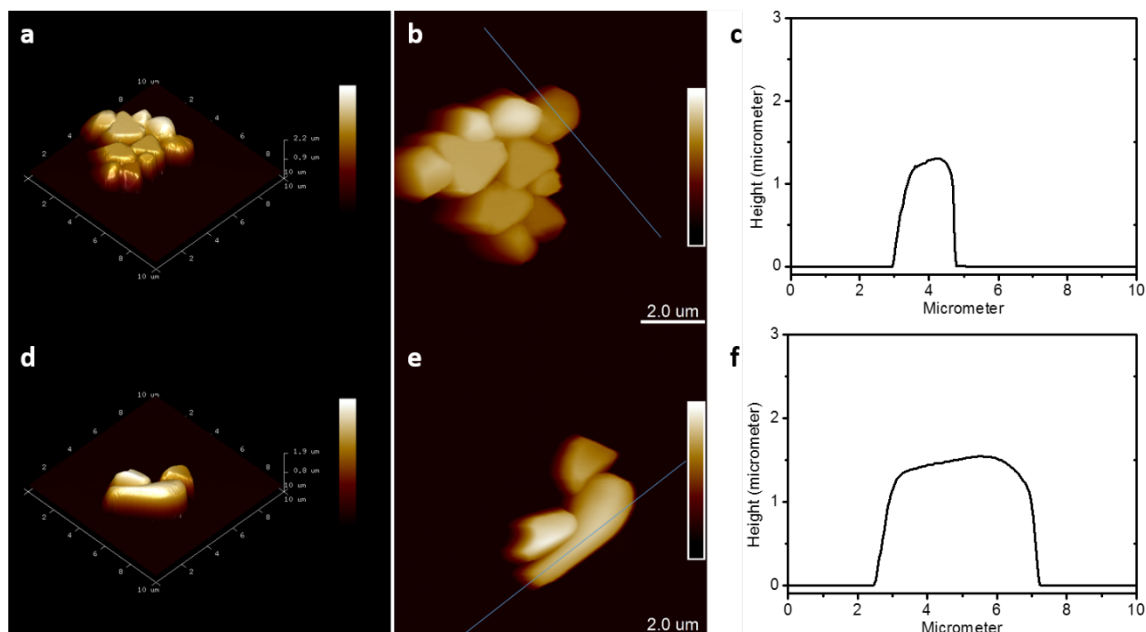
**Supplementary Table 1 | Single crystal X-ray diffraction data of 1D bulk single crystal perovskites.**

Empirical formula	C4 H14 Br4 N2 Pb	
Formula weight	617.00	
Temperature	103(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	I m m a	
Unit cell dimensions	a = 14.6230(16) Å	$\alpha = 90^\circ$ .
	b = 6.1041(7) Å	$\beta = 90^\circ$ .
	c = 14.4070(16) Å	$\gamma = 90^\circ$ .
Volume	1286.0(2) Å <sup>3</sup>	
Z	4	
Density (calculated)	3.187 Mg/m <sup>3</sup>	
Absorption coefficient	25.504 mm <sup>-1</sup>	
F(000)	1096	
Crystal size	0.280 x 0.095 x 0.060 mm <sup>3</sup>	
Theta range for data collection	1.984 to 28.488°.	
Index ranges	-18<=h<=19, -7<=k<=8, -19<=l<=19	
Reflections collected	7142	
Independent reflections	895 [R(int) = 0.0481]	
Completeness to theta = 25.242°	99.9 %	
Absorption correction	Empirical	
Max. and min. transmission	0.2163 and 0.07491	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	895 / 0 / 39	
Goodness-of-fit on F <sup>2</sup>	1.137	
Final R indices [I>2sigma(I)]	R1 = 0.0182, wR2 = 0.0416	
R indices (all data)	R1 = 0.0185, wR2 = 0.0417	
Extinction coefficient	0.00168(9)	
Largest diff. peak and hole	1.068 and -1.161 e.Å <sup>-3</sup>	

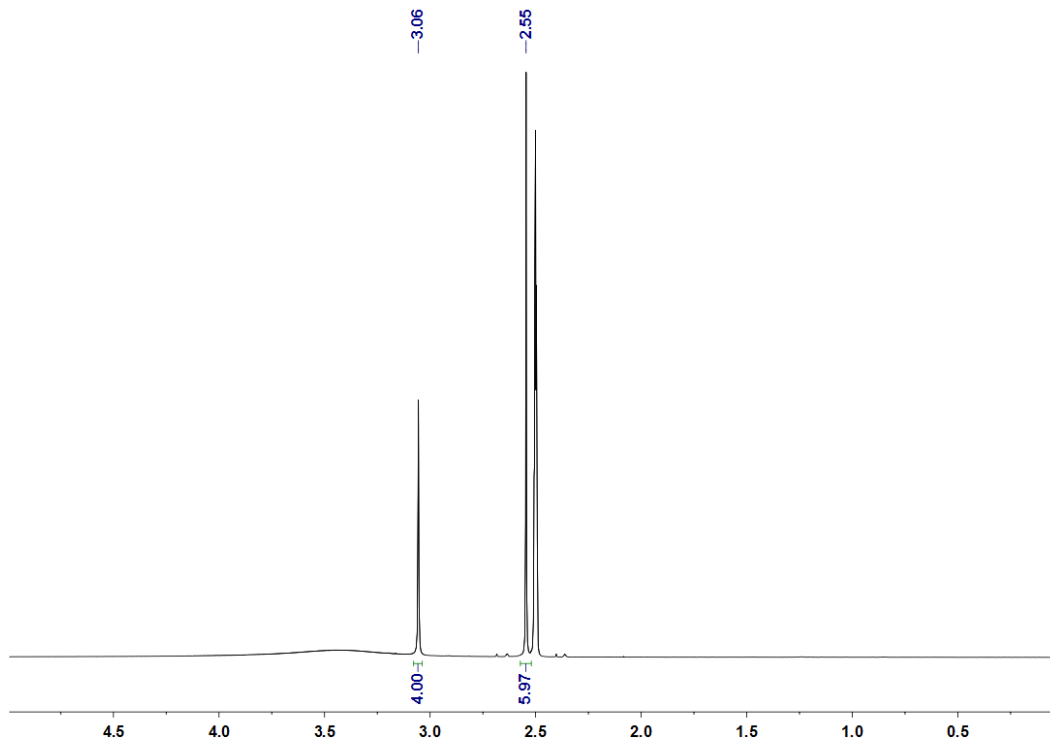
**Supplementary Table 2 | Chemical yield and composition of the component molar ratio in 1D microscale perovskites according to TGA and  $^1\text{H}$  NMR data.**

Product	Reagent	Starting amount	Molar ratio <sup>a</sup>	Theoretical ratio <sup>b</sup>	Yield
	Octanoic Acid	1.00 mmol	0.00	0.00	
Microscale Perovskites	N, N'-dimethylethylenediamine	0.30 mmol	1.02	1.00	95 %
	PbBr <sub>2</sub>	0.10 mmol	1.00	1.00	

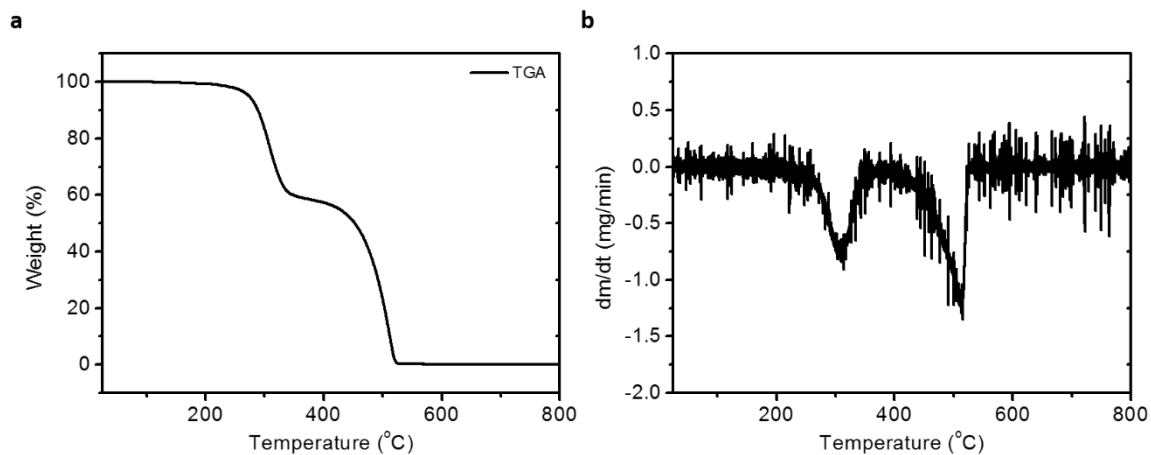
<sup>a</sup> Based on  $^1\text{H}$  NMR and TGA data; <sup>b</sup> based on the chemical formula:  $\text{C}_4\text{N}_2\text{H}_{14}\text{PbBr}_4$ .



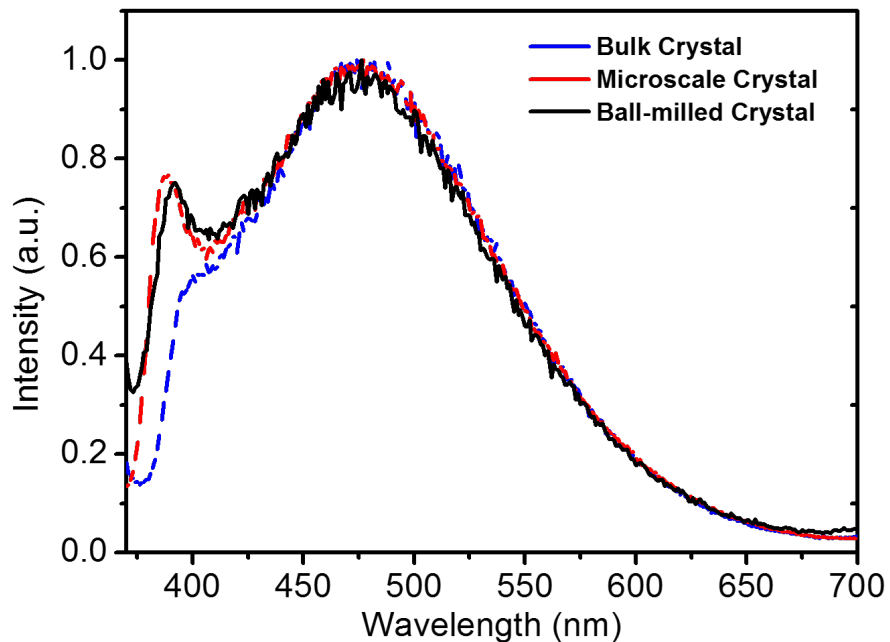
**Supplementary Figure 1 | AFM characterization of the 1D microscale perovskites. a,d, 3D AFM images of  $\text{C}_4\text{N}_2\text{H}_{14}\text{PbBr}_4$  microscale perovskites. b,e, 2D AFM images of the as-prepared microscale perovskites. c,f, Height profile along the blue line in 2D AFM images (b) and (e).**



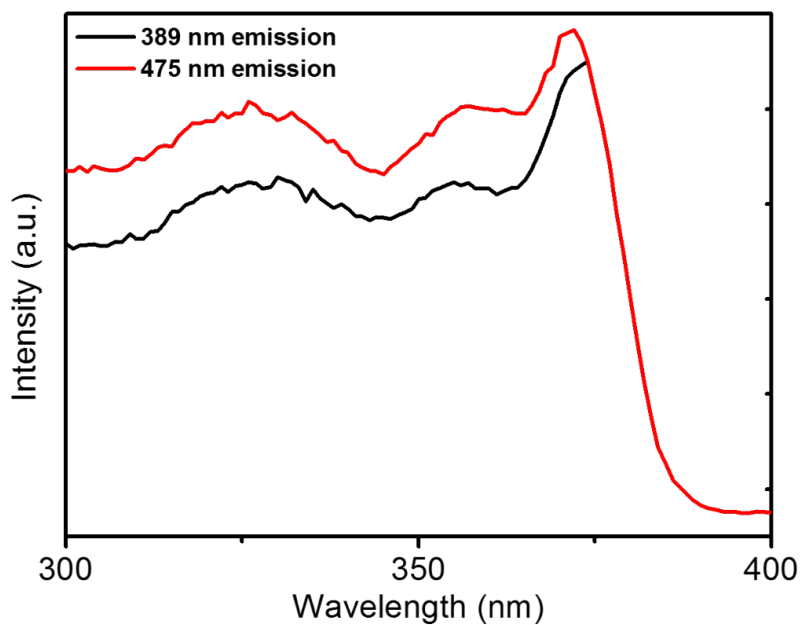
**Supplementary Figure 2 | Proton NMR spectrum of the 1D microscale perovskites in DMSO-d<sub>6</sub>.**



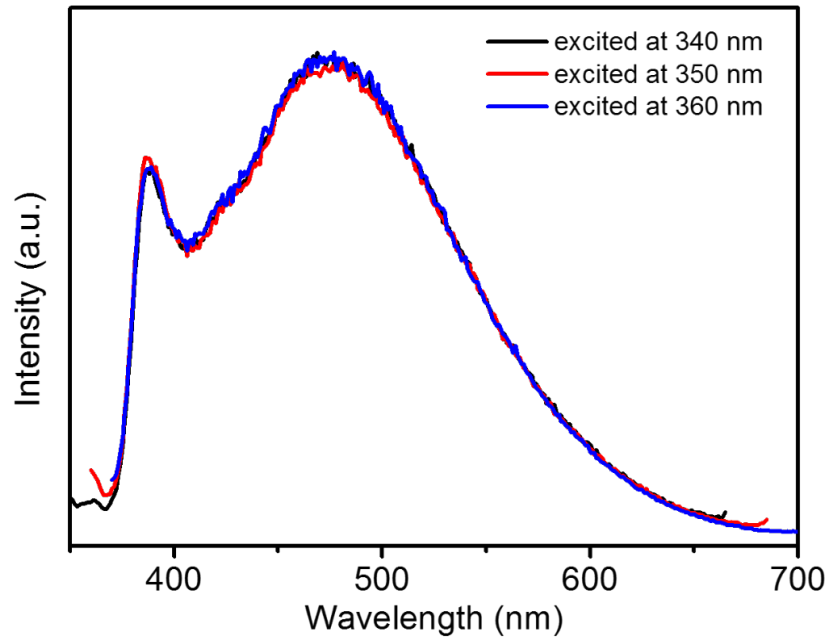
**Supplementary Figure 3 | TGA study of the 1D microscale perovskites. a,** TGA heating curves of  $C_4N_2H_{14}PbBr_4$  microscale perovskites. **b,** The corresponding 1st derivative.



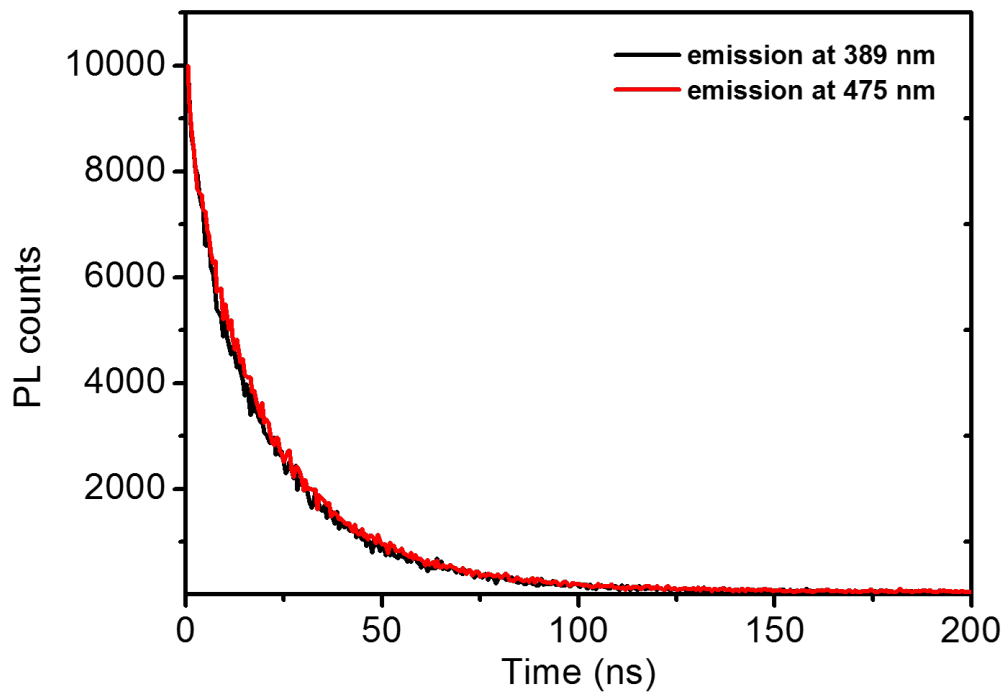
**Supplementary Figure 4 | Emission (excited at 360 nm) spectra of ball-milled bulk crystals, as well as bulk and microscale crystals at room temperature.**



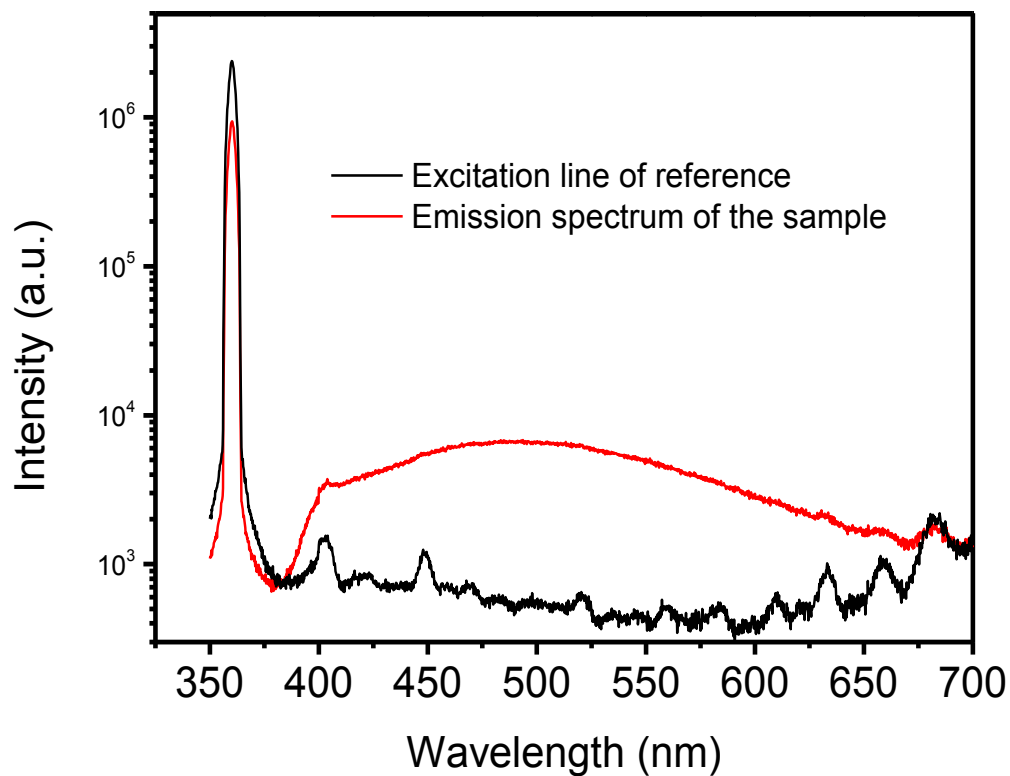
**Supplementary Figure 5 | Excitation spectra for the 389 and 475 nm emissions of the 1D microscale perovskites.**



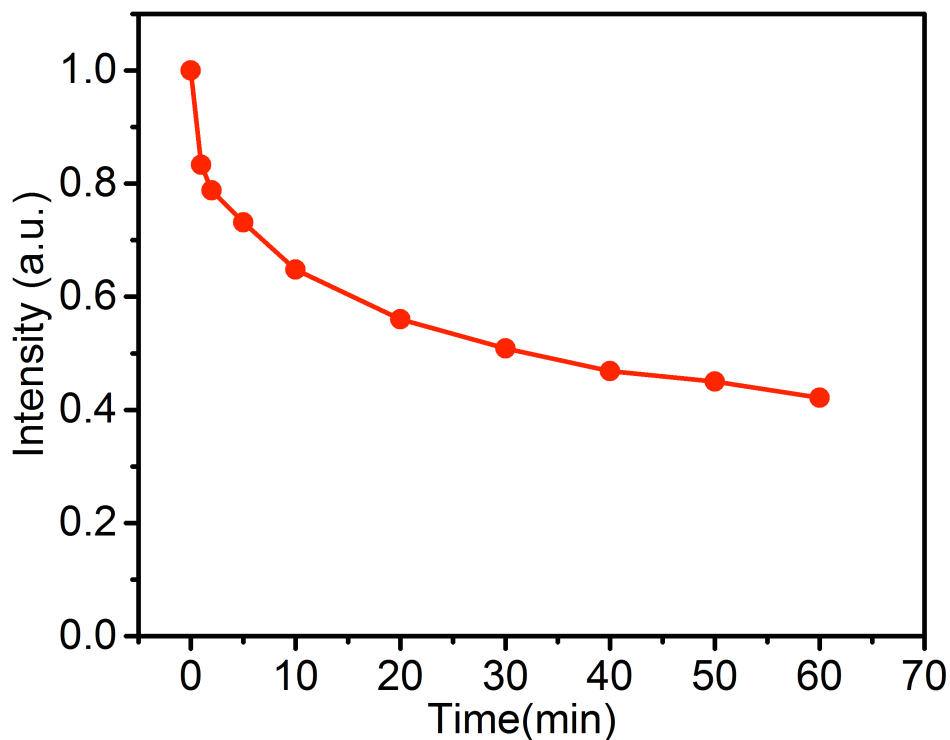
**Supplementary Figure 6 | Emission spectra of the 1D microscale perovskites at different excitation wavelengths.**



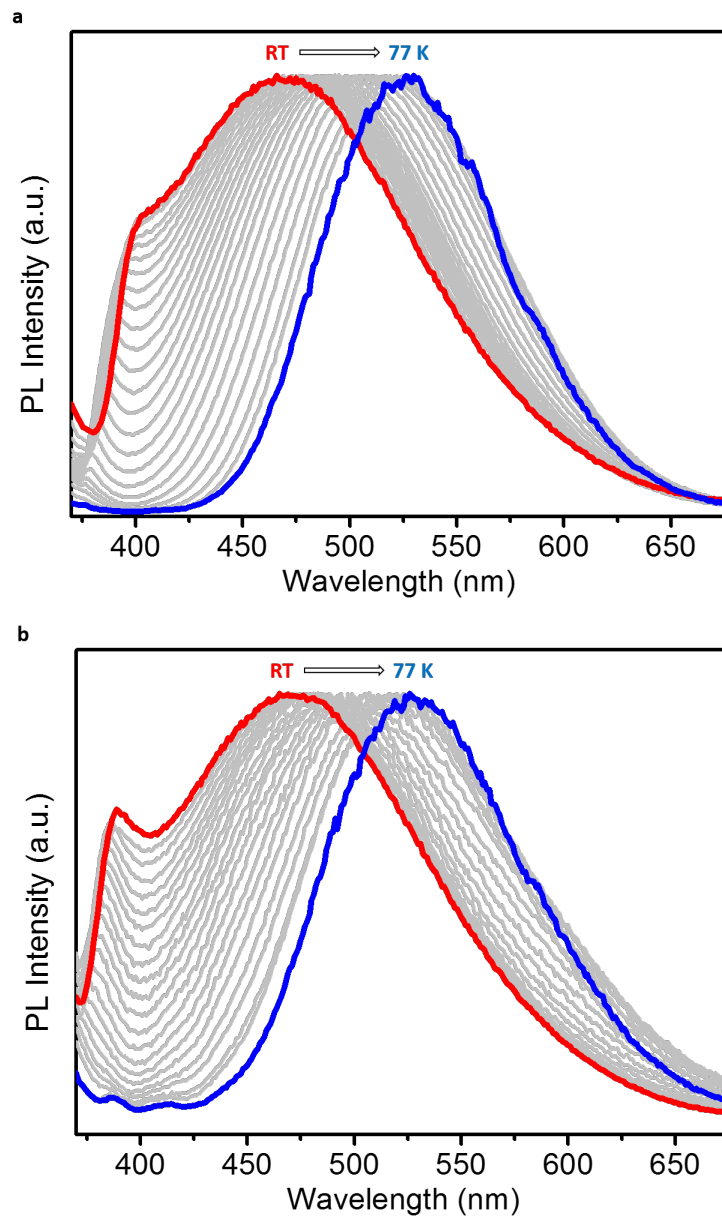
**Supplementary Figure 7 | Decay curves monitored at 389 nm and 475 nm for microscale crystals at room temperature.**



**Supplementary Figure 8 | Excitation line of reference and emission spectrum of 1D Pb bromide perovskite bulk crystals by an integrating sphere.**



**Supplementary Figure 9 | Photoluminescence intensity changes after continuous illumination in air using Hg lamp.**



**Supplementary Figure 10 | Temperature dependent emission spectra of bulk (a) and microscale (b) crystals from room temperature to 77 K (one scan per 10 K, spectra are normalized for clarity).**