## **Supplementary Materials: Refinement of Magnetite Nanoparticles by Coating with Organic Stabilizers**

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Figure S1. FTIR spectrum of 3a (Fe<sub>3</sub>O<sub>4</sub> covered with O-phosphoryl ethanolamine).



Figure S2. FTIR spectrum of 3c (Fe<sub>3</sub>O<sub>4</sub> covered with phospho-L-ascorbic acid).

15

16

17

18

**3g-**3 h

**3g-**24 h

**3h**-3 h

**3h**-24 h

Entry	MNP	Description	XRPD (Average Crystallite Size-Scherrer)	VSM (Magnetization)/em u/g	<b>Crystallite Size and Strain</b> $\beta_{tot} \cos \theta = C \epsilon \sin \theta + K \lambda / L$	
					Crystallite Size	Strain (ε. 10 <sup>-4</sup> )
1	1	Naked Fe <sub>3</sub> O <sub>4</sub> , 3 h	11.5 nm	68.5	11.0 nm	30
2	1	Naked Fe <sub>3</sub> O <sub>4</sub> , 24 h	14.1 nm	65.7	14.2 nm	9.5
3	<b>3a-</b> 3 h	Fe3O4 covered with O-phosphoryl ethanolamine, 3 h	25.7 nm ª	69.6 ª	20.1 nm	15
4	<b>3a</b> -24 h	Fe <sub>3</sub> O <sub>4</sub> covered with O-phosphoryl ethanolamine, 24 h	13.6 nm ª	67.8 ª	13.4 nm	20
5	<b>3b</b> -3 h	Fe <sub>3</sub> O <sub>4</sub> covered with glycerol phosphate, 3 h	25 nm	68.8	20.1 nm	14
6	<b>3b</b> -24 h	Fe <sub>3</sub> O <sub>4</sub> covered with glycerol phosphate, 24 h	27.9 nm	70.8	22.6 nm	16
7	<b>3c-</b> 3 h	Fe3O4 covered with phospho-L-ascorbic acid, 3 h	17.1 nm	68.2	16.5 nm	10
8	<b>3c-</b> 24 h	Fe <sub>3</sub> O <sub>4</sub> covered with phospho-L-ascorbic acid, 24 h	13.3 nm	59.8	13.0 nm	3
9	<b>3d</b> -3 h	Fe <sub>3</sub> O <sub>4</sub> covered with phospho-D,L-serine, 3 h	24.6 nm <sup>a</sup>	64.6	20.6 nm	12
10	<b>3d</b> -24 h	Fe3O4 covered with phospho-D,L-serine, 24 h	27.2 nm	65.4	27.0 nm	8
11	<b>3e</b> -3 h	Fe <sub>3</sub> O <sub>4</sub> covered with glycolic acid, 3 h	32.0 nm	74.0	27.0 nm	15
112	<b>3e</b> -24 h	Fe <sub>3</sub> O <sub>4</sub> covered with glycolic acid, 24 h	39.0 nm <sup>a</sup>	80	34.7 nm	10
13	<b>3f-</b> 3 h	Fe <sub>3</sub> O <sub>4</sub> covered with lactic acid, 3 h	25 nm	45.5 ª	16.4 nm	17
14	<b>3f</b> -24 h	Fe <sub>3</sub> O <sub>4</sub> covered with lactic acid, 24 h	20 nm	47.0 ª	12.8 nm	23

31.7 nm

30.7 nm

15.2 nm

15.3 nm

71.9

68.1

74.6

73

23.6 nm

25.6 nm

14.6 nm

14.4 nm

20

19

3

15

**Table S1.** Average crystallite sizes and saturation magnetization of MNP 1 and 3 including crystallite sizes and strains obtained by applying Williamson-Hall method (right column).

<sup>a</sup> average value of two samples.

Fe<sub>3</sub>O<sub>4</sub> covered with malic acid, 3 h

Fe<sub>3</sub>O<sub>4</sub> covered with malic acid, 24 h

Fe<sub>3</sub>O<sub>4</sub> covered with mandelic acid, 3 h

Fe<sub>3</sub>O<sub>4</sub> covered with mandelic acid, 24 h



Figure S3. FTIR spectrum of 3d (Fe<sub>3</sub>O<sub>4</sub> covered with phospho-D,L-serine).



Figure S4. FTIR spectrum of 3e (Fe<sub>3</sub>O<sub>4</sub> covered with glycolic acid).



Figure S5. FTIR spectrum of 3f (Fe<sub>3</sub>O<sub>4</sub> covered with lactic acid).



Figure S6. FTIR spectrum of 3g (Fe<sub>3</sub>O<sub>4</sub> covered with malic acid).



Figure S7. FTIR spectrum of 3h (Fe<sub>3</sub>O<sub>4</sub> covered with mandelic acid).





Figure S8. XPS spectra of 3d-3 h.



Figure S9. XPS spectra of 3f-3 h.



Figure S10. XPS spectra of 3g-3 h.







Figure S12. TEM image of 3c-3 h (Fe<sub>3</sub>O<sub>4</sub> covered with phospho-L-ascorbic acid).



Figure S13. TEM image of 3e-24 h (Fe<sub>3</sub>O<sub>4</sub> covered with glycolic acid).



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