## Supporting Information

## Brominated luciferins are versatile bioluminescent probes

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**Figure S1.** Optical analyses of luciferin analogs. (A) Fluorescence spectra of **D**–Luc and brominated analogs using 365 nm excitation light (pH 7.6). (B) Bioluminescence spectra of **D**–Luc and the brominated analogs at pH 7.6. (C) Bioluminescence spectra of **5'**–**BrLuc** at different pH values.



**Figure S2.** Representative <sup>1</sup>H NMR spectrum of a luciferin phenyl ester prepared as described in the *General Experimental Procedures* section.



**Figure S3.** Representative luminescence data following the addition of base to the phenyl ester of **D–Luc** (as described in the *General Experimental Procedures* section). The shaded area denotes the region used for trapezoidal integration.



**Figure S4.** Luciferin analog **5'–BrLuc** (100  $\mu$ L of a 1 mM solution) was administered (i.v.) into a luciferase transgenic mouse, and bioluminescence images were acquired. Photon flux values were integrated over the entire animal, and plotted as total flux (shown). Data are representative of n = 3 independent experiments.

### **General Experimental Procedures**

### **Expression and purification of Fluc**

Firefly luciferase was expressed and purified as previously described.<sup>[1]</sup>

### **Bioluminescence kinetic measurements**

Measurements were acquired on a Tecan F200 Pro injection port luminometer with a neutral density filter. Reactions were performed in black 96-well flat-bottom plates (Grenier). Bioluminescence buffer<sup>[2]</sup> (93.5  $\mu$ L of 20 mM Tris-HCl pH 7.6, 2 mM MgSO<sub>4</sub>, 2 mM ATP, 0.1 mM EDTA, 1 mM TCEP, 0.5 mg/mL BSA) was added to each well, followed by coenzyme A (0.5  $\mu$ L of a 100 mM solution) and luciferin substrate (1  $\mu$ L of a 0.01-100 mM solution in DMSO). The luminescence from each well was measured for 30 s prior to the addition of Fluc (5  $\mu$ L of a 1 mg/mL solution in bioluminescence buffer). Luminescence was then recorded every 0.1 s over a 1-min period. Samples were analyzed in triplicate and multiple runs were performed. The emission maxima were determined by averaging the largest photon outputs from five independent runs.  $K_m$  and relative  $k_{cat}$ values were determined using nonlinear regression analyses and robust fit outlier removal in GraphPad Prism (version 6.0f for Macintosh, GraphPad Software).

### **Bioluminescence imaging (in vitro)**

Imaging was performed using an IVIS Lumina (Xenogen) system equipped with a cooled CCD camera. Reactions were performed in black 96-well flat-bottom plates (Grenier). Bioluminescence buffer (93.5  $\mu$ L) was added to each well, along with coenzyme A (0.5  $\mu$ L of a 100 mM solution) and luciferin substrate (1  $\mu$ L of a 0.5-100

mM solution in DMSO). To initiate photon production, Fluc (5  $\mu$ L of a 1 mg/mL solution in bioluminescence buffer) was added to each well. The plate was then briefly agitated and placed in the IVIS instrument. The bioluminescent output was recorded every 5-30 s over a 45-75 min time period. Measurements were performed in triplicate.

### **Bioluminescence imaging (in cellulo)**

HEK293 cells stably expressing Fluc (provided by the Contag Lab, Stanford) were grown in DMEM supplemented with fetal bovine serum (FBS, 10%) penicillin (10 U/mL), and streptomycin (10  $\mu$ g/mL). The cells were cultured in a water-saturated CO<sub>2</sub> (5%) incubator at 37 °C. Imaging was performed using an IVIS Lumina (Xenogen) system equipped with a heated stage (37 °C) and a cooled CCD camera. Reactions were performed in black 96-well flat-bottom plates (Grenier) with 100,000 cells per well. Luciferin (50  $\mu$ L of 2X stock in PBS, pH 7.4) was added to each well, and bioluminescence images were acquired as above.

### Bioluminescence imaging (in vivo)

Pathogen free luciferase-expressing transgenic mice (B6;FVB-*Ptprc<sup>a</sup>* Tg(CAG-luc,-GFP)L2G85Chco *Thy1<sup>a</sup>*/J) or FVB mice were obtained from the Jackson Laboratory. The mice were housed in University of California, Irvine's animal care facility and provided access to food and water *ad libitum*. All procedures were approved by the Institutional Animal Care and Use Committee at UC Irvine (protocol #2011-2987 to J.A.P.). FVB mice were innoculated with luciferase-epressing DB7 cells (10<sup>6</sup>) in the right flank. Luciferin solutions were formulated using the potassium salt of the desired

luciferin and sterile PBS (Dulbecco's Phosphate-Buffered Saline, ThermoFisher). Mice were anesthetized with isoflurane (2% in 1 L/min of  $O_2$ ), and were injected i.v. (tail vein) or i.p. (intraperitoneal) with 100  $\mu$ L of luciferin solutions. Bioluminescent images were acquired using the IVIS Lumina system (PerkinElmer). Images were acquired every minute for 15 min (10 second exposure per image). Images were analyzed using Living Image software.

### **Bioluminescence** emission spectra

Emission spectra for D-luciferin and all analogs were recorded on a FluoroMax-4 spectrometer (Horiba Jobin-Yvon). Luciferin (10  $\mu$ L of an 10 mM solution in bioluminescence buffer, pH 7-9) and Fluc (10  $\mu$ L of a 1 mg/mL solution in bioluminescence buffer) along with coenzyme A (5  $\mu$ L of a 100 mM solution) were placed in a 10 mm path length quartz cuvette (1 mL total volume). Emission data were collected over 450-750 nm (1 nm intervals) at room temperature. The acquisition times were 0.1 s/wavelength. The spectra were then normalized to D-luciferin and plotted.

#### General chemiluminescence procedure

Phenyl esters of each luciferin analog were prepared following the basic procedure of Kim *et al.*<sup>[3]</sup> In brief, the potassium salt of each luciferin (6.0  $\mu$ mol) was added to an oven-dried, two-dram vial containing a small stir bar. Deuterated dimethylsulfoxide (0.55 mL) containing a mesitylene internal standard (0.275  $\mu$ L) was then added, and the luciferin was dissolved with stirring (5 min). Phenylchloroformate (0.76  $\mu$ L, 6.0  $\mu$ mol) was subsequently added, and a brief color change was observed in

most cases. The solutions were stirred for an additional 5 min. A portion of each solution (5  $\mu$ L) was reserved, and the remainder was added to an NMR tube for analysis. The NMR sample was kept at ambient temperature until luminometer measurements were acquired (see below). At that point, the NMR sample was frozen (-73 °C) to preserve the contents of the tube. At a later time, the tube was thawed and a <sup>1</sup>H NMR spectrum was immediately acquired (2 scans, 20 s relaxation delay). The concentration of the luciferin phenyl ester was determined via comparison to the internal standard (Figure S2).

The reserved portion of each luciferin ester solution was diluted to 0.5 mL with anhydrous DMSO, and 50  $\mu$ L of this solution was added to six wells of a black 96-well flat-bottom plate (Greiner). Chemiluminescence values were acquired on a Tecan Infinite F200 PRO plate-reading luminometer. Data were acquired for 1.5 s prior to injection of potassium phenoxide solution (50  $\mu$ L of a of 0.1 M solution). The phenoxide solution was prepared via dissolution of potassium *tert*-butoxide (112 mg) and phenol (94 mg) in anhydrous dimethylsulfoxide (10 mL) with stirring (30 min). The total volume in each well was 100  $\mu$ L. After the addition of base, luminescence data were collected for an additional 50 s (100 ms integration times were used). Relative luminescence yields were determined via trapezoidal integration of the data (Figure S2).

## Synthetic experimental procedures

All reactions were performed in flame- or oven-dried glassware under positive of nitrogen or argon unless otherwise pressure noted. Dichloromethane, dimethylacetamide, N.N-dimethylformamide, triethylamine, and toluene were dried by columns packed with activated alumina on a solvent purification system. Anhydrous pyridine and DMSO were purchased from Acros Organics in AcroSeal<sup>TM</sup> bottles. All reagents were used as purchased without further purification. 4,5-Dichloro-1,2,3dithiazol-1-ium chloride (Appel's salt) was synthesized according to a published procedure<sup>[4]</sup> and stored in a desiccator. Thin layer chromatography (TLC) was performed on Merck 60 F<sub>254</sub> pre-coated silica gel plates, and TLC plates were visualized with UV light and ninhydrin stain when appropriate. Flash-column chromatography was performed using silica gel (60 Å, 230-240 mesh, Merck KGA). NMR spectra were recorded with Bruker Advanced spectrometers using deuterated solvents. <sup>1</sup>H NMR spectra were recorded at 400 or 500 MHz as indicated. <sup>13</sup>C NMR spectra were recorded at 125 MHz. <sup>1</sup>H NMR data are reported in the following order: chemical shift (δ ppm), multiplicity, coupling constant (Hz), and integration. <sup>13</sup>C NMR data are reported in terms of chemical shift. Infrared spectra were recorded using a Thermo Scientific iD5 ATR infrared spectrophotometer. High-resolution mass spectra were obtained from the UC Irvine Mass Spectrometry Facility. The abbreviations used can be found in the document JOC Standard Abbreviations and Acronyms.

### Synthetic procedures

## **3-Bromo-4-isopropoxyaniline (S1)**



Following the general method of Shen and Driver,<sup>[5]</sup> to a flask containing 2-bromo-1-isopropoxy-4-nitrobenzene<sup>[6]</sup> were added iron filings (0.15 g, 560 μmol), acetone (3 mL) and water (10 mL). Glacial acetic acid (1 mL)

was then added, and the mixture was heated at reflux for 3 h. The mixture

was then diluted with ethyl acetate (20 mL) and washed with saturated sodium carbonate (2 x 20 mL), ammonium chloride (2 x 20 mL), and brine (1 x 20 mL). The organic layer was then dried with MgSO<sub>4</sub>, filtered, and concentrated *in vacuo*. The residue was purified *via* flash-column chromatography (eluting with 8:2 hexanes:ethyl acetate) to yield **S1** (72 mg, 57%) as a brown oil. The spectra matched those reported previously.<sup>[7]</sup>

### 2,6-Dibromo-4-isopropoxyaniline (1)



Following the general procedure of Popeney and Guan,<sup>[8]</sup> to a solution of 4-isopropoxy aniline (1.20 g, 7.90 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (66 mL) and methanol (22 mL) was added calcium carbonate (3.03 g, 30.0 mmol),

followed by benzyltrimethylammonium tribromide (6.53 g, 16.0 mmol). The reaction was stirred at room temperature for 2 h. The reaction was then quenched with 1 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and washed with 1 M Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (2 x 100 mL), water (2 x 100 mL) and brine (1 x 100 mL). The organic layer was dried with MgSO<sub>4</sub>, then filtered and concentrated *in vacuo*. The crude material was purified by flash-column chromatography (eluting with 8:2 hexanes:ethyl acetate) to afford **1** (1.0 g, 43%) as an orange oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.03 (s, 2H), 4.34 (septet, *J* = 6.1 Hz, 1H), 1.29 (d, *J* = 6.1

Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  150.2, 136.4, 120.9, 109.0, 72.0, 22.0; HRMS (CI) *m/z* calcd for C<sub>9</sub>H<sub>12</sub>Br<sub>2</sub>NO [M+H]<sup>+</sup> 309.9265, found 309.9274.

## Representative procedure for the synthesis of Appel's salt adducts (2, 4, 6)

## (Z)-2,6-Dibromo-N-(4-chloro-5H-1,2,3-dithiazol-5-ylidene)-4-isopropoxyaniline (2)



Following the general method of Michaelidou and Koutentis,<sup>[9]</sup> to a flask of **1** (0.94 g, 3.3 mmol) under argon was added Appel's salt (0.83 g, 4.0 mmol), followed by anhydrous  $CH_2Cl_2$  (15 mL) and anhydrous pyridine (0.59 mL, 7.3 mmol). The reaction mixture was stirred at room temperature for 2 h, then loaded onto silica gel and

purified *via* flash-column chromatography (eluting with 8:2 hexanes:ethyl acetate). Compound **2** (1.3 g, 91%) was isolated as a brown oil. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  7.33 (s, 2H), 4.66 (septet, *J* = 6.0 Hz, 1H), 1.23 (d, *J* = 6.0 Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  164.8, 155.8, 146.7, 142.8, 120.3, 113.6, 71.3, 22.0; HRMS (ESI-TOF)<sup>+</sup> *m/z* calcd for C<sub>11</sub>H<sub>10</sub>Br<sub>2</sub>ClN<sub>2</sub>OS<sub>2</sub> [M+H]<sup>+</sup> 442.8290, found 442.8295.

### (Z)-3-Bromo-N-(4-chloro-5H-1,2,3-dithiazol-5-ylidene)-4-isopropoxyaniline (4)



Compound **4** was isolated as a brown oil (0.92 g, 87%). <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ )  $\delta$  7.52, (s, 1H), 7.28 (s, 2H), 4.72 (septet, J =4.8 Hz, 1H), 1.34 (d, J = 4.8 Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ 159.2, 152.5, 147.5, 144.5, 125.2, 121.0, 116.6, 113.3, 72.1, 22,3; HRMS (CI) m/z calcd for C<sub>11</sub>H<sub>11</sub>BrClN<sub>2</sub>OS<sub>2</sub> [M+H]<sup>+</sup> 364.9185,

found 364.9189.

### (Z)-N-(4-chloro-5H-1,2,3-dithiazol-5-ylidene)-4-isopropoxyaniline (6)



found 287.0084.

# Representative procedure for the fragmentation and cyclization of Appel's salt adducts

(3, 5, 8)



flushed with dry nitrogen and charged with  $CH_2Cl_2$  (5 mL). The flask was cooled to 0 °C in an ice bath and DBU (0.29 mL, 1.9 mmol) was added. The reaction mixture was stirred for 5 min, then adsorbed to silica gel. The adsorbed material was rinsed with hexanes, and then eluted with 7:3 hexanes:ethyl acetate. The isolated thioamide was used immediately in the next reaction. (Note: this compound degrades quickly).

Following the general procedure of Inamoto and coworkers,<sup>[10]</sup> to a flask containing (2,6-dibromo-4-isopropoxyphenyl)carbamothioyl cyanide (0.24 g, 640  $\mu$ mol based on crude yield from previous step), was added palladium(II) chloride (11 mg, 64  $\mu$ mol), copper(I) iodide (60 mg, 320  $\mu$ mol), and tetrabutyl ammonium bromide (0.43 g,

1.3 mmol). The flask was flushed with dry nitrogen, and then DMF (8 mL) and DMSO (8 mL) were added. The reaction was heated at 125 °C for 2 h. The mixture was then diluted with ethyl acetate (40 mL) and washed with 1 M NaHSO<sub>4</sub> (1 x 40 mL), water (3 x 40 mL), ammonium chloride (1 x 40 mL) and brine. The organic layers were combined and then dried with MgSO<sub>4</sub>, and concentrated in vacuo. The concentrate was purified via flash-column chromatography (eluting with 9:1 hexanes: ethyl acetate) to yield 3 (33 mg, 6.7% over two steps) as a brown solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.45, (d, J = 2.3Hz, 1H), 7.29 (d, J = 2.3 Hz, 1H), 4.64 (septet, J = 6.0 Hz, 1H), 1.40 (d, J = 6.0 Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>) δ 159.0, 145.2, 137.7, 133.6, 122.9, 119.2, 112.8, 104.3, 71.7, 21.9; HRMS (ESI-TOF)<sup>+</sup> m/z calcd for C<sub>11</sub>H<sub>10</sub>BrN<sub>2</sub>OS [M+H]<sup>+</sup> 350.9779, found 350.9783.

### 5-Bromo-6-isopropoxybenzo[d]thiazole-2-carbonitrile (5)

Compound 5 was isolated as a brown solid (79 mg, 4%). <sup>1</sup>H Br√ NMR (400 MHz, CDCl<sub>3</sub>) δ 8.40 (s, 1H), 7.36 (s, 1H), 4.69 (septet, J = 6.0 Hz, 1H), 1.47, (d, J = 6.0 Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  155.2, 146.9, 136.1, 134.4, 129.3, 115.9, 113.0, 104.4, 73.0, 21.8; HRMS (CI) m/z calcd for  $C_{11}H_{10}BrN_2OS [M+H]^+ 296.9697$ , found 296.9694.

#### 6-Isopropoxybenzo[d]thiazole-2-carbonitrile (7)



Compound 7 was isolated as a brown solid (48 mg, 3%).  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  8.07 (d, J = 9.1 Hz, 1H), 7.34 (d, J= 2.4 Hz, 1H), 7.20 (dd, J = 9.1, 2.5 Hz, 1H), 4.66 (septet, J = 5.3 Hz, 1H), 1.41 (d, J = 5.3 Hz, 1H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  158.9, 146.7, 137.5, 126.0, 119.6, 113.3, 104.9, 71.1, 21.9; HRMS (CI) *m/z* calcd for C<sub>11</sub>H<sub>11</sub>N<sub>2</sub>OS [M+H]<sup>+</sup> 236.0858, found 236.0864.

### 7-Bromo-6-isopropoxybenzo[d]thiazole-2-carbonitrile (8)

To a flask of 7 (0.13 g, 60 µmol) was added *N*bromosuccinimide (0.16 g, 92 µmol), followed by CH<sub>3</sub>CN (15 mL). The reaction mixture was stirred for 12 h, then extracted with ethyl acetate (30 mL). The combined organic layers were washed with water (3 x 60 mL) and brine (1 x 60 mL), then dried with MgSO<sub>4</sub>, filtered, and concentrated *in vacuo* to yield **8** (0.14 g, 75%). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.22 (d, *J* = 11.4 Hz, 1H), 7.60 (d, *J* = 11.5 Hz, 1H), 4.88 (septet, *J* = 6.0 Hz, 1H), 1.31 (d, *J* = 6.0 Hz, 6H); <sup>13</sup>C NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  155.3, 145.7, 134.2, 124.7, 116.4, 113.0, 103.3, 73.7, 22.2; HRMS (CI) *m/z* calcd for C<sub>11</sub>H<sub>10</sub>BrN<sub>2</sub>OS [M+H]<sup>+</sup> 296.9697, found 296.9696.

# Representative procedure for the formation of brominated luciferins (7'–BrLuc, 4'– BrLuc, 5'–BrLuc)

(S)-2-(7-Bromo-6-hydroxybenzo[d]thiazol-2-yl)-4,5-dihydrothiazole-4-carboxylic acid (7'-BrLuc)



mixture was stirred at room temperature under nitrogen for 24 h. The reaction was then

quenched with a saturated solution of ammonium chloride (10 mL) and extracted with ethyl acetate (30 mL). The organic layers were combined and washed with saturated ammonium chloride (2 x 30 mL) and brine (1 x 30 mL). The organic layer was then dried with MgSO<sub>4</sub>, filtered, and concentrated *in vacuo*. The concentrate was purified via flashcolumn chromatography (eluting with 1:1 hexanes:ethyl acetate) to yield the deprotected intermediate which was used immediately in the following step.

The isolated material (33 mg, 130 µmol, based on the crude isolated yield) was dissolved in degassed methanol (2 mL) and D-cysteine (24 mg, 140 µmol) in degassed 0.05 M phosphate buffer (pH 8.0) was added. The mixture was stirred at room temperature under nitrogen, overnight. The mixture was then acidified with 1 M NaHSO<sub>4</sub> (10 mL) and extracted with ethyl acetate (20 mL). The combined organic layers were washed with saturated ammonium chloride (2 x 20 mL) and brine (1 x 20 mL), then dried over MgSO<sub>4</sub>, filtered, and concentrated *in vacuo* to yield **7'–BrLuc** (41 mg, 65% over two steps) as a yellow solid. Note: this compound was treated with anhydrous K<sub>2</sub>CO<sub>3</sub> (1.0 equiv.) in water and lyophilized for chemiluminescence assays. <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O)  $\delta$  7.56 (d, *J* = 8.9 Hz, 1H), 6.89 (d, *J* = 8.9 Hz, 1H), 5.17 (m, 1H), 3.76 (m, 1H), 3.55 (m, 1H); <sup>13</sup>C NMR (500 MHz, D<sub>2</sub>O)  $\delta$  180.6, 168.4, 164.4, 157.0, 144.6, 142.9, 125.7, 123.4, 103.7, 82.7, 39.1. Note: high resolution mass spectrometry was not successful due to multiple fragmentation pathways.<sup>[11]</sup>

## (S)-2-(4-Bromo-6-hydroxybenzo[d]thiazol-2-yl)-4,5-dihydrothiazole-4-carboxylic acid (4'-BrLuc)



# (S)-2-(5-Bromo-6-hydroxybenzo[d]thiazol-2-yl)-4,5-dihydrothiazole-4-carboxylic acid (5'-BrLuc)

Br HO S N GOH Compound 5'-BrLuc was isolated as a yellow solid (16 mg, 36% over two steps). <sup>1</sup>H NMR (500 MHz, D<sub>2</sub>O)  $\delta$ 7.81 (s, 1H), 7.12 (s, 1H), 5.20 (apparent t, J = 8.6 Hz, 1H), 3.81 (m, 1H), 3.60 (m, 1H); <sup>13</sup>C NMR (500 MHz, D<sub>2</sub>O)  $\delta$  180.0, 168.2, 160.4, 158.8, 147.4, 139.0, 129.1, 116.2, 110.1, 82.8, 39.1. Note: high resolution mass spectrometry was not successful due to multiple fragmentation pathways.<sup>[11]</sup> Scheme S1: Synthesis of 6'-deoxyLuc.



### Phenylcarbamothioyl cyanide (S2)

A dry, nitrogen-purged round bottom flask containing a stir bar and anhydrous tetrahydrofuran (30 mL) was charged with aniline (0.91 mL, 10 mmol), and Appel's salt (2.19 g, 10.5 mmol). The resulting solution was stirred at room temperature for 40 min and pyridine (1.66 mL, 20.5 mmol) was subsequently added. When the starting material was completely consumed (by TLC), a solution of sodium thiosulfate pentahydrate (3.2 g, 20 mmol) in 15 mL water and CH<sub>3</sub>CN (15 mL) was added. The solution was stirred, and when complete consumption of the intermediate was observed by TLC, the reaction mixture was diluted with ethyl acetate, washed with saturated NaHSO<sub>4</sub>, and dried with MgSO<sub>4</sub>. The mixture was filtered, concentrated in vacuo and purified by flash column chromatography (eluting with 10% ethyl acetate in hexanes) to yield S2 (0.75 g, 46%) as a brown solid. Note: high resolution mass spectrometry was not obtained for this compound due to its multiple fragmentation pathways. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, mixture of tautomers)  $\delta$  9.51 (s, 1H), 7.78 (d, J = 8.3 Hz, 1H), 7.57 – 7.28 (m, 4H), 1.75 (s, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 165.8, 162.0, 137.0, 136.9, 130.2, 129.5, 129.0, 128.5, 122.9, 122.6, 113.7, 112.1.

### **Benzothiazole-2-carbonitrile (S3)**

A dry, nitrogen-purged round bottom flask containing a stir bar, anhydrous DMSO (100 mL), and anhydrous DMF (100 mL) was charged with **S2** (720 mg, 4.5 mmol), palladium(II) chloride (64 mg, 0.45 mmol), copper(I) iodide (420 mg, 2.2 mmol), and tetrabutylammonium bromide (2.9 g, 9.0 mmol). The flask was fitted with a condenser and heated at 120 °C under nitrogen for 8 h. The solution was then cooled and quenched with water. Ethyl acetate was added and the layers were separated. The aqueous layer was extracted twice with ethyl acetate. The organic layers were then combined and washed with water (6 x 150 mL) and brine (2 x 100 mL). The organics were combined, and dried with MgSO<sub>4</sub>. The mixture was filtered, concentrated *in vacuo* and purified by flash column chromatography (eluting with 10% ethyl acetate in hexanes) to yield **S3** (228 mg, 32%) as a taupe solid. Spectra matched those previously reported.<sup>[12]</sup>

#### 6'-deoxyLuc



S N OH M OH M OH M S (222 mg, 1.38 mmol) was dissolved in CH<sub>3</sub>CN (6 mL) and a solution of K<sub>2</sub>CO<sub>3</sub> (228 mg, 1.70 mmol), D-cysteine hydrochloride monohydrate (252 mg, 1.50 mmol), and water

(3 mL) was added dropwise with stirring. After 45 min, the volatile organics were removed *in vacuo*, and hydrochloric acid (1 M) was added until the solution was acidic. A cream colored precipitate was observed. The solid was collected by filtration, washed with water, and dried under high vacuum to afford **6'-deoxyLuc** (319 mg, 88%) as an off

white solid. <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  8.16 (ddd, *J* = 28.6, 7.7, 0.5 Hz, 1H), 7.57 (dtd, *J* = 19.7, 7.4, 1.2 Hz, 1H), 4.94 (app t, *J* = 9.0 Hz, 1H), 3.77 (dd, *J* = 10.4, 8.2 Hz, 1H), 3.51 (t, *J* = 10.0 Hz, 1H). <sup>13</sup>C NMR (126 MHz, DMSO-*d*<sub>6</sub>)  $\delta$  169.9, 162.2, 159.4, 152.7, 135.1, 126.9, 126.9, 123.8, 122.7, 84.0, 36.3; HRMS (ESI<sup>+</sup>) calcd for C<sub>11</sub>H<sub>8</sub>N<sub>2</sub>O<sub>2</sub>S<sub>2</sub>Na [M + Na]<sup>+</sup> 324.9484, found 324.9479.

### Scheme S2: Synthesis of 6'–NHMeLuc and 6'–NMe<sub>2</sub>Luc.



## 6-(Methylamino)benzo[d]thiazole-2-carbonitrile (S4) and 6-

### (Dimethylamino)benzo[d]thiazole-2-carbonitrile (S5)



carbonitrile (88 mg, 0.50 mmol) and

NaBH<sub>4</sub> (29 mg, 0.77 mmol) were dissolved in THF (10 mL). In a separate dried flask, paraformaldehyde (23 mg, 0.77 mmol) was suspended in dry THF (10 mL), and H<sub>2</sub>SO<sub>4</sub> (40  $\mu$ L, 0.77 mmol) was added. The two solutions were stirred separately at room temperature for 15 min. The carbonitrile solution was then added to the paraformaldehyde solution dropwise via syringe. The resulting mixture was stirred under nitrogen for 1 h, then an additional equivalent of NaBH<sub>4</sub> was added. After 1 h, the reaction mixture was basified with a 1.8 M solution of KOH (20 mL) and ethyl acetate (20 mL) was added. The layers were separated and the aqueous layer was extracted with additional ethyl acetate (2 x 10 mL). The combined organic layers were washed with

brine (1 x 20 mL), and dried over MgSO<sub>4</sub>. The mixture was then filtered and concentrated *in vacuo*. The crude material ws purified by flash column chromatography (eluting with 30% ethyl acetate in hexanes) to yield a mixture of **S4** (31 mg, 33%), and **S5** (17 mg, 16%) as orange-red solids. Spectra matched those previously reported.<sup>[2]</sup>

6'-NHMeLuc



6'-NMe<sub>2</sub>Luc



**6'–NMe<sub>2</sub>Luc** was prepared from **S5** as previously reported.<sup>[2]</sup> Spectra matched those previously reported.<sup>[2]</sup>

## Stille cross-coupling of 5 with phenyltributylstannane

A 15 mL pressure tube containing lithium chloride (9 mg, 200  $\mu$ mol) was flame dried under vacuum. After cooling, **5** (55.0  $\mu$ mol) was flame dried under vacuum. After cooling, **5** (55.0 mg, 168  $\mu$ mol), and palladium tetrakis (43 mg, 37  $\mu$ mol) were added. The flask was evacuated and flushed with nitrogen, and dioxane (2.5 mL) was added against positive flow, followed by phenyltributylstannane (63  $\mu$ L, 200  $\mu$ mol). The mixture was heated at 120 °C for 6 h. The mixture was cooled and washed with water (3 times, 3 mL). The organics were then dried over magnesium sulfate, filtered, and

concentrated via rotary evaporation. The crude mixture was purified via preparative TLC (eluting with 30% ethyl acetate in hexanes) to provide **9** (17 mg, 31%). <sup>1</sup>H NMR (500 MHz, CDCl3)  $\delta$  7.82 – 7.77 (m, 2H), 7.55 – 7.49 (m, 2H), 7.49 – 7.43 (m, 1H), 7.31 (dd, J = 17.7, 3.1 Hz, 1H), 4.71 (dt, J = 15.1, 7.5 Hz, 1H), 1.44 (d, J = 7.6 Hz, 6H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  158.8, 144.8, 139.4, 138.6, 137.4, 132.5, 129.6, 128.6, 128.5, 119.2, 113.5, 104.0, 71.1, 22.0. HRMS (ESI<sup>+</sup>) calcd. for C<sub>17</sub>H<sub>14</sub>N<sub>2</sub>OSNa [M + Na]<sup>+</sup> 317.0724, found 317.0731.

### **Computational Details**

Density functional theory (DFT) structural optimizations of the singlet ground state  $S_0$  and the first singlet excited state ( $S_1$ ) were performed using the hybrid-GGA functional, PBE0<sup>[14]</sup> in the gas phase. Basis sets of double-zeta quality with polarization and diffuse functions<sup>[15]</sup> (def2-SVPD) were necessary to bind the additional electron of the anionic structures. Analytical force constant calculations<sup>[16]</sup> for the ground state and numerical force constant calculations were performed to verify minima by the absence of imaginary vibrational modes. Constrained excited state geometry optimizations were performed by fixing the outof-plane bending angle of the carbon-carbon single bond connecting the two thiazoline rings to 0° and adjusting the torsional angle between the nitrogens in each ring. All calculations were performed with the quantum chemistry package TURBOMOLE.<sup>[17,18]</sup>

### Molecular orbitals involved in the emission of luciferin analogs.

For each luciferin analogue, we report the molecular orbitals involved in the emission "de-excitation" from the  $S_1$  excited state geometry. All orbitals are plotted with a contour value of 0.05au. The primary contribution to the excitation is a HOMO to LUMO transition that tends to have pi-pi character originating on the anionic oxygen attached to the benzothiazole with the transition dipole moment pointing towards the thiazoline motif. In the case of luciferin that are electron poor at the 6' position of the benzothiazole, e.g. **6'-deoxyLuc**, no such character is observed and corroborates the notion that an electron rich moiety is necessary for strong emission.

Compound	НОМО	LUMO
D-Luc		
6'-aminoLuc		
6'- MeNHLH <sub>2</sub>		
6'-Me <sub>2</sub> NLH <sub>2</sub>		
CycLuc1	:	
4'–BrLuc		



### Ordering of relevant spin and geometric states

In order to reconcile the differences in experimental and computational results for the brominated luciferin series, and **6'-methoxyLuc**, the ground and excited electronic states were studied in both the planar and twisted geometric states of the neutral and anionic species; both singlet and triplet states were included to help rationalize possible intruder states that may be responsible for the quenching of the chemiluminescence. Electronic states were computed at the geometry of the first singlet excited state, which was confirmed to be a

minimum by numerical frequency analysis. In the case of neutral **4'–BrLuc**, no twisted  $S_1$  minimum was found. This picture gives relative energetics of the different electronic states at the assumed emissive nuclear configuration.

For native **D-luc**, both the twisted and the planar S1 state are relatively isolated from other states. Given that our model can reasonably predict the emission strength of native luciferin, the isolated nature of the S<sub>1</sub> state helps to establish a baseline by demonstrating that it is less probable that there is some other electronic state lower in energy that can be accessed. In the case of the anionic 7'-BrLuc, one observes a very small separation between the planar S<sub>1</sub> state and the twisted triplet ground state, which could account for the considerable decrease in experimentally measured emission. The planar S<sub>1</sub> state for the other brominated luciferins seems to be modestly isolated, but there could be a distribution of protonated and deprotonated luciferin whose different accessible electronic states could cause interference. Considering 6'-methoxyLuc, there seem to be no nearby electronic states, but there is a low-lying planar triplet state that could act as a channel for nonradiative decay. The overall environment in solution is complex compared to our model and many factors besides those described here could be in effect; however, electronic states in both the planar and twisted configurations are definitely accessible under photoexcitation and provide a reasonable explanation for the quenching of chemiluminescence.









## D-Luc

S0	Energy $= -1$	438.843074	343
С	-1.2813373	4.2620165	0.9672691
С	-0.5639216	3.0090061	1.0785824
С	0.8064982	3.0073406	1.1731258
С	1.5866350	4.2210546	1.1679940
С	0.8995049	5.4641204	1.0601380
С	-0.4599930	5.4801449	0.9650070
S	1.9038716	1.6557876	1.3136942
С	3.2596334	2.7958787	1.3508706
N	2.9068526	4.0653135	1.2658870
С	4.5972422	2.3599630	1.4611518
S	5.8813591	3.5980542	1.4854097
Ν	4.9702370	1.1080088	1.5469792
С	6.3234754	0.9454358	1.6461948
С	7.1002788	2.2773794	1.6302748
0	6.9126056	-0.1140081	1.7399920
0	-2.5220026	4.3248825	0.8775392
Η	-1.0078312	6.4213743	0.8805400
Η	1.4886068	6.3838958	1.0560990
Η	-1.1523024	2.0906698	1.0827860
Η	7.7967781	2.2782298	0.7806943
Η	7.6821334	2.3670627	2.5575835
<b>S</b> 1	Energy $= -1$	438.837364	170
С	-1.3077364	4.2411522	0.9742593
С	-0.5251260	3.0144489	1.0853816
С	0.8732071	3.0787400	1.1717981
С	1.5811853	4.2982977	1.1571969
С	0.8343348	5.4909973	1.0477456
С	-0.5449927	5.4702381	0.9599381
S	1.9763984	1.7519350	1.3077014
С	3.3014484	2.9069091	1.3384698
Ν	2.9533467	4.1697306	1.2536181
С	4.6384353	2.4025237	1.4553801
S	5.9626301	3.5791753	1.4991833
N	4.9516785	1.1310834	1.5337424
С	6.2915119	0.9031962	1.6420835
С	7.1250283	2.2021387	1.6445183
0	6.8391147	-0.1862002	1.7311552
0	-2.5568109	4.1882638	0.8988577
Η	-1.1121422	6.3987922	0.8751294
Η	1.3781167	6.4389700	1.0335322
Η	-1.0664676	2.0675599	1.0986238

Н 7.8317080 2.1872338 0.8022538	
Н 7.7034552 2.2664248 2.5772437	
6'-aminoLuc	
S0 Energy = -1419.531486458	
C -4.4714246 0.6994008 -0.0531430	
C -3.3692188 1.5614488 -0.0974595	
C -4.2658215 -0.7069461 0.0061684	
C -3.0009735 -1.2492622 0.0244840	
C -2.0921667 1.0063512 -0.0803884	
C 0.2163665 0.2382510 -0.0579573	
S -0.5584490 1.8136621 -0.1235692	
C 1.6607835 0.1366511 -0.0622214	
C 3.7786309 0.8331079 -0.1041887	
C 4.0076765 -0.6855881 -0.0437758	
S 2.3821488 -1.4681121 -0.0009116	
Н 4.5888990 -0.9326787 0.8545634	
Н 4.5767526 -1.0051463 -0.9269342	
N 2.4274703 1.1729047 -0.1094693	
N -0.5788406 -0.7855947 -0.0088251	L
C -1.8823847 -0.3974068 -0.0192836	)
Н -3.5184629 2.6409029 -0.1456867	
Н -2.8488274 -2.3279379 0.0711795	
Н -5.1385883 -1.3622747 0.0351306	
O 4.6868831 1.6208778 -0.1420452	
N -5.7511204 1.1900916 -0.1014747	
H -5.9101560 2.1741553 0.0461581	
H -6.5173909 0.5846211 0.1462701	
S1 Energy = -1419.527675377	
C -4.4803789 0.6851627 -0.0178088	
C -3.3266056 1.5285508 -0.0610676	
C -4.3249978 -0.7152743 0.0290246	
C -3.0598503 -1.2789556 0.0336751	
C -2.0683707 0.9377477 -0.0550975	
C 0.2198206 0.1264484 -0.0523619	
S -0.5154815 1.7232744 -0.1001641	
C 1.6501356 0.0722123 -0.0638300	
C 3.7302659 0.8922127 -0.1104362	
C 4.0432301 -0.6194627 -0.0588742	
S 2.4601755 -1.4950325 -0.0164827	
Н 4.6396163 -0.8437868 0.8364763	
Н 4.6276427 -0.9073218 -0.9438741	
N 2.3864681 1.1573965 -0.1079582	
N -0.6099309 -0.9016371 -0.0076986	5

C -1.9050211 -0.4652489 -0.0080012 Н -3.4498270 2.6122753 -0.0984732 H -2.9332101 -2.3617129 0.0698932 Н -5.2111955 -1.3504931 0.0610567 O 4.6238033 1.7152787 -0.1476353 N -5.7079766 1.2648930 -0.0238185 H -5.8180548 2.2669642 -0.0557480 Н -6.5484712 0.7079876 0.0058242

## 6'-MeNHLH<sub>2</sub>

S0 Energy = -1458.761653271
C -4.4872976 0.6696379 -0.0489409
C -3.3824353 1.5327374 -0.1473878
C -4.2720884 -0.7304882 0.0981780
C -2.9995231 -1.2591416 0.1461942
C -2.1052672 0.9900004 -0.0985360
C 0.2091789 0.2348838 -0.0417864
S -0.5742369 1.8011763 -0.1969441
C 1.6531610 0.1391059 -0.0510849
C 3.7681740 0.8399187 -0.1370161
C 4.0035162 -0.6744100 -0.0183228
S 2.3812831 -1.4597322 0.0727991
Н 4.5957110 -0.8835664 0.8824064
Н 4.5639884 -1.0267934 -0.8945450
N 2.4162632 1.1749752 -0.1482248
N -0.5804386 -0.7877926 0.0730776
C -1.8861072 -0.4084055 0.0475822
Н -3.5376057 2.6067408 -0.2618215
Н -2.8411273 -2.3321072 0.2603661
Н -5.1294213 -1.3979398 0.1745117
O 4.6738481 1.6285687 -0.2099286
N -5.7542279 1.1765487 -0.0942315
C -6.9503067 0.3858882 -0.0248385
Н -5.8519107 2.1722670 -0.2066495
Н -7.0154443 -0.3411765 -0.8521188
Н -7.8189738 1.0492230 -0.0928716
Н -7.0241598 -0.1697600 0.9249636
S1 Energy = -1458.758090104
C -4.4940452 0.6138937 -0.0458721
C -3.3500759 1.4627083 -0.1364094
C -4.3265676 -0.7776759 0.0912730

	2.1271213	1.5777570 0.1715117
0	4.6738481	1.6285687 -0.2099286
Ν	-5.7542279	1.1765487 -0.0942315
С	-6.9503067	0.3858882 -0.0248385
Н	-5.8519107	2.1722670 -0.2066495
Η	-7.0154443	-0.3411765 -0.8521188
Н	-7.8189738	1.0492230 -0.0928716
Н	-7.0241598	-0.1697600 0.9249636
S1	Energy $= -1$	458.758090104
S1 C	Energy = -1 -4.4940452	458.758090104 0.6138937 -0.0458721
S1 C C	Energy = -1 -4.4940452 -3.3500759	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094
S1 C C C	Energy = -1 -4.4940452 -3.3500759 -4.3265676	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094 -0.7776759 0.0912730
S1 C C C C	Energy = -1 -4.4940452 -3.3500759 -4.3265676 -3.0503608	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094 -0.7776759 0.0912730 -1.3241779 0.1393935
S1 C C C C C C	Energy = -1 -4.4940452 -3.3500759 -4.3265676 -3.0503608 -2.0836445	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094 -0.7776759 0.0912730 -1.3241779 0.1393935 0.8894190 -0.0874074
S1 C C C C C C C C	Energy = -1 -4.4940452 -3.3500759 -4.3265676 -3.0503608 -2.0836445 0.2132737	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094 -0.7776759 0.0912730 -1.3241779 0.1393935 0.8894190 -0.0874074 0.1040229 -0.0300020
S1 C C C C C C C S	Energy = -1 -4.4940452 -3.3500759 -4.3265676 -3.0503608 -2.0836445 0.2132737 -0.5432916	458.758090104 0.6138937 -0.0458721 1.4627083 -0.1364094 -0.7776759 0.0912730 -1.3241779 0.1393935 0.8894190 -0.0874074 0.1040229 -0.0300020 1.6872872 -0.1817513

C 1.6443727 0.0696163 -0.0445610
C 3.7108312 0.9173472 -0.1637966
C 4.0492132 -0.5815949 -0.0108331
S 2.4807752 -1.4770004 0.1024370
Н 4.6543218 -0.7338128 0.8936643
Н 4.6333114 -0.9197773 -0.8779575
N 2.3633530 1.1606597 -0.1676050
N -0.6032185 -0.9254865 0.0826763
C -1.9065876 -0.5057360 0.0515846
Н -3.4831241 2.5410358 -0.2431797
Н -2.9132055 -2.4009690 0.2454444
Н -5.1955703 -1.4295292 0.1595461
O 4.5903845 1.7502028 -0.2662265
N -5.7219068 1.2035500 -0.0986606
C -6.9633152 0.4926620 -0.0284835
Н -5.7527764 2.2078640 -0.1987465
Н -7.0532381 -0.2347449 -0.853072
Н -7.7923929 1.2032438 -0.0971100
Н -7.0519632 -0.0626493 0.9204859

## 6'-Me<sub>2</sub>NLH<sub>2</sub>

S0 Energy = -1497.990421989
C -4.4861974 0.7491604 -0.0910281
C -3.3611606 1.5968276 -0.1029254
C -4.2810704 -0.6653136 -0.0660703
C -3.0199584 -1.2164225 -0.0494197
C -2.0926843 1.0261819 -0.0865954
C 0.2120281 0.2425126 -0.0606940
S -0.5521773 1.8244503 -0.0965605
C 1.6546717 0.1319701 -0.0515171
C 3.7772472 0.8162629 -0.0572887
C 3.9971041 -0.7046679 -0.0201194
S 2.3667505 -1.4782345 -0.0106258
Н 4.5657277 -0.9700488 0.8810927
Н 4.5755926 -1.0121816 -0.9014106
N 2.4291049 1.1640647 -0.0717082
N -0.5910980 -0.7771687 -0.0435917
C -1.8908190 -0.3788969 -0.0578753
Н -3.4715258 2.6777862 -0.1257107
Н -2.8823335 -2.2979416 -0.0301953
Н -5.1408846 -1.3315481 -0.0599715
O 4.6912395 1.5987041 -0.0718205
N -5.7553110 1.2633106 -0.1035751
C -6.9004261 0.3869323 -0.1106124
Н -6.9294967 -0.2594303 0.7820583
Н -6.9190932 -0.2614487 -1.0026626

H -7.8138926 0.9887326 -0.1156257 C -5.9470514 2.6908924 -0.1349602 H -5.5047395 3.1446984 -1.0383141 H -5.4994370 3.1821005 0.7455521 H -7.0177921 2.9136213 -0.1359343 S1 Energy = -1497.987259200C -4.4966845 0.7060184 -0.0951344 C -3.3379064 1.5324563 -0.1030681 C -4.3377266 -0.7006779 -0.0714968 C -3.0750048 -1.2718974 -0.0554425 C -2.0812256 0.9327865 -0.0856622 C 0.2064057 0.1197652 -0.0591166 S -0.5299475 1.7200821 -0.0907584 C 1.6380950 0.0676805 -0.0498289 C 3.7165129 0.8948966 -0.0549151 C 4.0342412 -0.6157478 -0.0211218 S 2.4534342 -1.4965669 -0.0141354 H 4.6189543 -0.8505449 0.8791625 H 4.6322186 -0.8879359 -0.9019022 N 2.3720938 1.1552380 -0.0672201 N -0.6200908 -0.9069333 -0.0462647 C -1.9188294 -0.4680740 -0.0610814 H -3.4222004 2.6165555 -0.1228028 H -2.9600356 -2.3561791 -0.0384609 H -5.2079783 -1.3521217 -0.0662337 O 4.6074021 1.7221357 -0.0680827 N -5.7431889 1.2882624 -0.1110534 C -6.9170887 0.4558623 -0.1048516 Н -6.9465014 -0.1815247 0.7945456 Н -6.9379717 -0.2073221 -0.9857761 H -7.8124491 1.0821284 -0.1182652 C -5.8948431 2.7218790 -0.1370610 H -5.4332203 3.1604881 -1.0369026 H -5.4356909 3.1923887 0.7476793 H -6.9584563 2.9718076 -0.1428581 CvcLuc1 S0 Energy = -1496.809364595C -4.5332049 0.8700070 0.1884754 C -3.4533613 1.7437545 0.1266118 C -4.3651522 -0.5455850 0.2133999 C -3.1167863 -1.1115332 0.1776570 C -2.1843145 1.1564577 0.0896171 C 0.1120877 0.3549146 0.0035415

S -0.6418683 1.9392121 0.0022762

C 1.5514649 0.2322211 -0.0585663
C 3.6773770 0.8961032 -0.1735707
C 3.8849595 -0.6267986 -0.1410293
S 2.2501354 -1.3848422 -0.0472983
Н 4.4949910 -0.8940398 0.7320955
Н 4.4162729 -0.9430566 -1.0485217
N 2.3340119 1.2567174 -0.1214294
N -0.6993542 -0.6578692 0.0621271
C -1.9952502 -0.2539636 0.1118390
Н -3.5905528 2.8247393 0.1037956
Н -2.9647913 -2.1914208 0.1969222
O 4.5971562 1.6694522 -0.2374958
N -5.8720096 1.1721768 0.2532084
C -6.6725631 -0.0121654 -0.0344857
Н -6.2017164 2.0817157 -0.0338749
C -5.7307858 -1.1718301 0.3314280
Н -6.9422776 -0.0552014 -1.1051920
Н -7.6014056 -0.0146429 0.5505426
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Н -5.9041856 -1.4993389 1.3695036
S1 Energy = -1496.805965042
C -4.5188303 0.8448987 0.0649123
C -3.4057626 1.7038943 -0.0120338
C -4.3933374 -0.5645303 0.1453505
C -3.1441156 -1.1508470 0.1515579
C -2.1554606 1.0885034 -0.0049445
C 0.1249134 0.2489555 -0.0108181
S -0.5914225 1.8572849 -0.0896190
C 1.5596392 0.1717704 -0.0422211
C 3.6502853 0.9576060 -0.1443735
C 3.9407244 -0.5561388 -0.0564846
S 2.3452184 -1.4049256 0.0351680
Н 4.5484552 -0.7654240 0.8349124
Н 4.5062645 -0.8768773 -0.9425104
N 2.3087500 1.2431367 -0.1275050
N -0.7045317 -0.7689543 0.0710004
C -2.0034705 -0.3225669 0.0759388
Н -3.5234551 2.7860514 -0.0737316
Н -3.0096882 -2.2313761 0.2125619
O 4.5534279 1.7670832 -0.2187222
N -5.8340155 1.1946670 0.0757037
C -6.7191482 0.0498699 0.1600770
Н -6.1648151 2.1465673 0.0261097
C -5.7705014 -1.1687788 0.2125729
Н -7.3916321 0.0217248 -0.7129199

Η	-7.3574459	0.1257051	1.0554184
Η	-5.9546457	-1.8571504	-0.6255961
Η	-5.9152757	-1.7511800	1.1346954

## 4'-BrLuc

S0 Energy = -4011.790393777
O -6.8448625 3.5066263 0.0000277
C -5.7550987 2.8962422 -0.0001237
C -4.4743511 3.5689170 -0.0003715
C -3.3128202 2.8399157 -0.0005227
C -3.2717813 1.3958513 -0.0004524
C -4.5269237 0.7361945 -0.0002101
C -5.7085744 1.4350660 -0.0000525
S -1.6680617 3.4417652 -0.0008231
C -1.0904728 1.7475828 -0.0008238
N -2.0679365 0.8190991 -0.0006187
C 0.2372386 1.4631867 -0.0010187
S 0.7687657 -0.3190377 -0.0010166
N 1.2835062 2.2978614 -0.0012224
C 2.4933628 1.7402297 -0.0013884
C 2.4823412 0.1899125 -0.0013115
Н 3.0104887 -0.1843149 0.8905541
Н 3.0102020 -0.1844092 -0.8933036
O 3.5877124 2.3056246 -0.0015908
Н -4.4739261 4.6604285 -0.0004295
Н -6.6651359 0.9136911 0.0001339
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236 S 0.8186614 -0.3320233 -0.0011378
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236 S 0.8186614 -0.3320233 -0.0011378 N 1.2191917 2.2868683 -0.0011642
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236 S 0.8186614 -0.3320233 -0.0011378 N 1.2191917 2.2868683 -0.0013885
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236 S 0.8186614 -0.3320233 -0.0011378 N 1.2191917 2.2868683 -0.0013885 C 2.5314880 0.2469192 -0.0013424
H -6.6651359 0.9136911 0.0001339 Br -4.5378820 -1.1629725 -0.0001056 S1 Energy = -4011.710603849723 O -6.7829107 3.6274877 -0.0000102 C -5.7500840 2.9238810 -0.0001261 C -4.4317689 3.5471242 -0.0003866 C -3.2788420 2.7497812 -0.0005274 C -3.3110194 1.3393784 -0.0004453 C -4.5919987 0.7416342 -0.0001902 C -5.7608189 1.4779501 -0.0000238 S -1.6406106 3.3008724 -0.0008041 C -1.1150913 1.6245661 -0.0007993 N -2.0771422 0.7305339 -0.0006072 C 0.2923759 1.3577569 -0.0010236 S 0.8186614 -0.3320233 -0.0011378 N 1.2191917 2.2868683 -0.0013885 C 2.5314880 0.2469192 -0.0013424 H 3.0703789 -0.1065825 0.8892589

O 3.5235144 2.4402342 -0.0015856 H -4.3849815 4.6365284 -0.0004720 H -6.7312604 0.9838975 0.0001866 Br -4.6826752 -1.1525073 -0.0000731

## 5'-BrLuc

J-DI Luc
S0 Energy = -4011.801518284
O -6.7891639 3.5566622 -0.0032265
C -5.7123659 2.9455578 -0.0020496
C -4.4263046 3.6087163 -0.0048395
C -3.2568236 2.8885796 -0.0034073
C -3.2398164 1.4495888 0.0009261
C -4.4786221 0.7552349 0.0037855
C -5.6411938 1.4692353 0.0023466
S -1.6061438 3.4546740 -0.0061729
C -1.0610640 1.7704864 -0.0014388
N -2.0314378 0.8797553 0.0018765
C 0.3135098 1.4331837 -0.0013169
S 0.7500242 -0.2931721 0.0036376
N 1.2894669 2.3021346 -0.0046127
C 2.5322110 1.7265635 -0.0037090
C 2.4881321 0.1859963 0.0010092
Н 3.0086956 -0.1874802 0.8932456
Н 3.0074937 -0.1929026 -0.8896401
O 3.5923193 2.3188898 -0.0062399
H -4.4424913 4.6989112 -0.0081132
Br -7.2975886 0.5503267 0.0061108
Н -4.4737861 -0.3346915 0.0070588
S1 Energy = -4011.796387194
O -6.7437212 3.6772546 -0.0031658
C -5.7186254 2.9697383 -0.0018696
C -4.3954213 3.5818131 -0.0044834
C -3.2377910 2.7950392 -0.0030318
C -3.2846579 1.3868124 0.0009242
C -4.5476541 0.7623570 0.0035154
C -5.7058689 1.5174540 0.0022117
S -1.5935900 3.3370327 -0.0058902
C -1.0840590 1.6554509 -0.0015344
N -2.0491522 0.7671677 0.0016384

C 0.3244271 1.3753295 -0.0014447 S 0.8310340 -0.3204403 0.0041211 N 1.2604039 2.2930692 -0.0052328 C 2.5258156 1.7815085 -0.0041678 C 2.5505003 0.2384339 0.0013366 H 3.0856052 -0.1180970 0.8930933

```
H 3.0847145 -0.1245141 -0.8883520
O 3.5668260 2.4206560 -0.0070476
H -4.3611156 4.6716393 -0.0076101
Br -7.3876280 0.6456411 0.0056586
H -4.5949914 -0.3270960 0.0065609
```

### 7'-BrLuc

```
S0 Energy = -4011.759897560
O -6.8117381 3.4595796 0.0029335
C -5.7216478 2.8707448 0.0013385
C -4.4348731 3.5452103 0.0045764
C -3.2531015 2.8403188 0.0026474
C -3.2340551 1.4008789 -0.0027133
C -4.4737326 0.7034700 -0.0060358
C -5.6422345 1.4048733 -0.0040951
S -1.6157756 3.4241695 0.0058077
C -1.0583751 1.7446135 -0.0001276
N -2.0207710 0.8437951 -0.0040662
C 0.3182677 1.4165633 -0.0005116
S 0.7643403 -0.3082045 -0.0066373
N 1.2897944 2.2900858 0.0033428
C 2.5355001 1.7211731 0.0020153
C 2.4996733 0.1802397 -0.0037318
H 3.0212415 -0.1961605 0.8866843
H 3.0223923 -0.1895294 -0.8962514
O 3.5926833 2.3185410 0.0048960
Br -4.4145662 5.4330358 0.0116693
H -4.4573940 -0.3881333 -0.0100862
H -6.6069278 0.8930852 -0.0065650
S1 Energy = -4011.795056788
O -6.7843515 3.5880324 0.0033296
C -5.7447304 2.9044070 0.0014106
C -4.4133815 3.5294764 0.0045355
C -3.2306306 2.7536740 0.0023402
C -3.2794805 1.3475690 -0.0029263
C -4.5444281 0.7235825 -0.0060107
C -5.7124689 1.4607122 -0.0039673
S -1.5979460 3.3016099 0.0053605
C -1.0783132 1.6198572 -0.0006018
N -2.0425238 0.7318175 -0.0044872
C 0.3286271 1.3500736 -0.0007245
S 0.8529965 -0.3429053 -0.0066833
N 1.2604692 2.2750099 0.0033161
C 2.5278859 1.7751634 0.0022345
C 2.5671773 0.2317636 -0.0034626
```

H 3.1050927 -0.1254000 0.8863331 H 3.1061519 -0.1188499 -0.8952208 O 3.5658623 2.4214073 0.0053212 Br -4.3248798 5.3856081 0.0115190 H -4.5783246 -0.3683564 -0.0101170 H -6.6841035 0.9640973 -0.0064089

### 6'-deoxyLuc

S0 Energy = -1364.262556246
C -3.8887853 0.5411853 0.0180407
C -3.9813083 -0.8641396 0.0333510
C -2.8472751 -1.6632313 0.0353918
Н -4.9657325 -1.3351363 0.0428777
C -2.6586667 1.1731746 0.0050638
H -4.8026377 1.1368910 0.0154811
C -1.4949478 0.3864087 0.0077234
Н -2.5681963 2.2595490 -0.0081928
C -1.6014809 -1.0284963 0.0224105
Н -2.9301269 -2.7502593 0.0459045
N -0.2139126 0.8614673 -0.0042680
S -0.0171715 -1.7323308 0.0190287
C 0.6414379 -0.1116984 -0.0006608
C 2.0829149 0.0893377 -0.0129804
S 2.6897696 1.7383480 -0.0324891
N 2.9128232 -0.8948249 -0.0102107
C 4.3652767 1.0675144 -0.0392240
C 4.2416239 -0.4637484 -0.0226020
H 4.9265511 1.3996161 0.8443366
Н 4.9118806 1.3813354 -0.9385757
O 5.1979637 -1.1909620 -0.0204063
S1 Energy = -1364.230020612
C -3.8559532 0.5712146 0.0087144
C -3.9801107 -0.8276326 -0.0767330
C -2.8552328 -1.6466446 -0.0950343
Н -4.9723293 -1.2785667 -0.1279337
C -2.6136292 1.1765583 0.0796964
Н -4.7565146 1.1873864 0.0202942
C -1.4550570 0.3728865 0.0672685
Н -2.5053782 2.2595641 0.1460202
C -1.6012735 -1.0435992 -0.0229363
Н -2.9574920 -2.7302281 -0.1616933
N -0.1826011 0.8234438 0.1372677
S -0.0334010 -1.7961254 -0.0189254
C 0.6835525 -0.1825684 0.0994154
C 2.0575145 0.0104766 0.1765614

2.7004163	1.6696698 0.3979463
3.0050565	-0.9447297 0.1866544
4.2720697	1.0967426 -0.1928708
4.2224603	-0.4725259 0.1487035
5.1188655	1.6225679 0.2616548
4.3679470	1.0850764 -1.2904330
5.3410903	-0.9529663 0.2563626
	$\begin{array}{c} 2.7004163\\ 3.0050565\\ 4.2720697\\ 4.2224603\\ 5.1188655\\ 4.3679470\\ 5.3410903 \end{array}$

### 6'-methoxyLuc

S0 Energy = -1478.585905861				
С	-1.9457970	4.8111594	0.6845139	
С	-0.6014930	5.0305282	0.4054565	
С	-2.5278098	3.6288309	0.2233707	
С	-1.7700652	2.6829162	-0.5062453	
С	-0.4334305	2.9108234	-0.7785140	
С	0.1743274	4.0905754	-0.3264205	
Η	-2.2352423	1.7647298	-0.8604312	
S	0.3912284	6.3856428	0.8378426	
Ν	1.4763897	4.4471802	-0.5197342	
С	1.7296041	5.5970213	0.0209326	
С	3.0327088	6.2358887	-0.0217056	
Ν	3.2560264	7.3808234	0.5256109	
S	4.3402938	5.4041219	-0.8533971	
С	5.4209665	6.7867227	-0.4324256	
С	4.5770564	7.7980662	0.3590192	
Н	5.8164506	7.2677286	-1.3369697	
Н	6.2664107	6.4530192	0.1838765	
0	5.0296732	8.8297438	0.7783827	
Н	0.1572773	2.1868500	-1.3402255	
Η	-2.5554427	5.5199750	1.2436690	
0	-3.8308072	3.4632503	0.5189803	

C -4.4987251 2.3008076 0.0942555 H -5.5301848 2.3930159 0.4477822 H -4.0472212 1.3953547 0.5305478 H -4.4999112 2.2162513 -1.0043316 S1 Energy = -1478.580085938 C -1.8830714 4.7861949 0.6530733 C -0.5388785 4.9784218 0.3602200 C -2.5226898 3.5843639 0.1975980 C -1.8064591 2.6305781 -0.5233447 C -0.4547536 2.8521572 -0.8025455 C 0.2029577 4.0230497 -0.3696471 Н -2.2836720 1.7170192 -0.8717882 S 0.4740507 6.3254358 0.7825550 N 1.5078183 4.3379954 -0.5857000 C 1.8023033 5.5113490 -0.0448518 C 3.0585498 6.1795059 -0.0584263 N 3.2276935 7.3498519 0.5172052 S 4.4434269 5.4342124 -0.8657924 C 5.4314980 6.8726580 -0.3890814 C 4.5033062 7.8273186 0.3954081 Н 5.8282793 7.3874346 -1.2750250 Н 6.2774763 6.5748132 0.2457066 O 4.9219653 8.8806493 0.8346122 Н 0.1181310 2.1142760 -1.3659913 Н -2.4780628 5.5069822 1.2143548 O -3.8099637 3.4991991 0.5308661 C -4.5550478 2.3653572 0.1431859 H -5.5683465 2.5186056 0.5234362 H -4.1262099 1.4509222 0.5814733 H -4.5780176 2.2726754 -0.9536606

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![](_page_69_Figure_0.jpeg)