

IUCrJ

Volume 4 (2017)

Supporting information for article:

Atom interaction propensities of oxygenated chemical functions in crystal packings

Christian Jelsch and Yvon Bibila Mayaya Bisseyou

Supplementary Materials.

Atom interaction propensities of oxygenated chemical functions in crystal packings.

IUCr J.

Christian Jelsch^a & Yvon Bibila Mayaya Bisseyou^b.

^aCRM2, CNRS, Institut Jean Barriol, Université de Lorraine, Vandoeuvre les Nancy cedex, France,

^b Laboratoire de Cristallographie et Physique Moléculaire, UFR SSMT, Université Félix Houphouët-Boigny, 22 BP 582 Abidjan 22, Côte d'Ivoire.

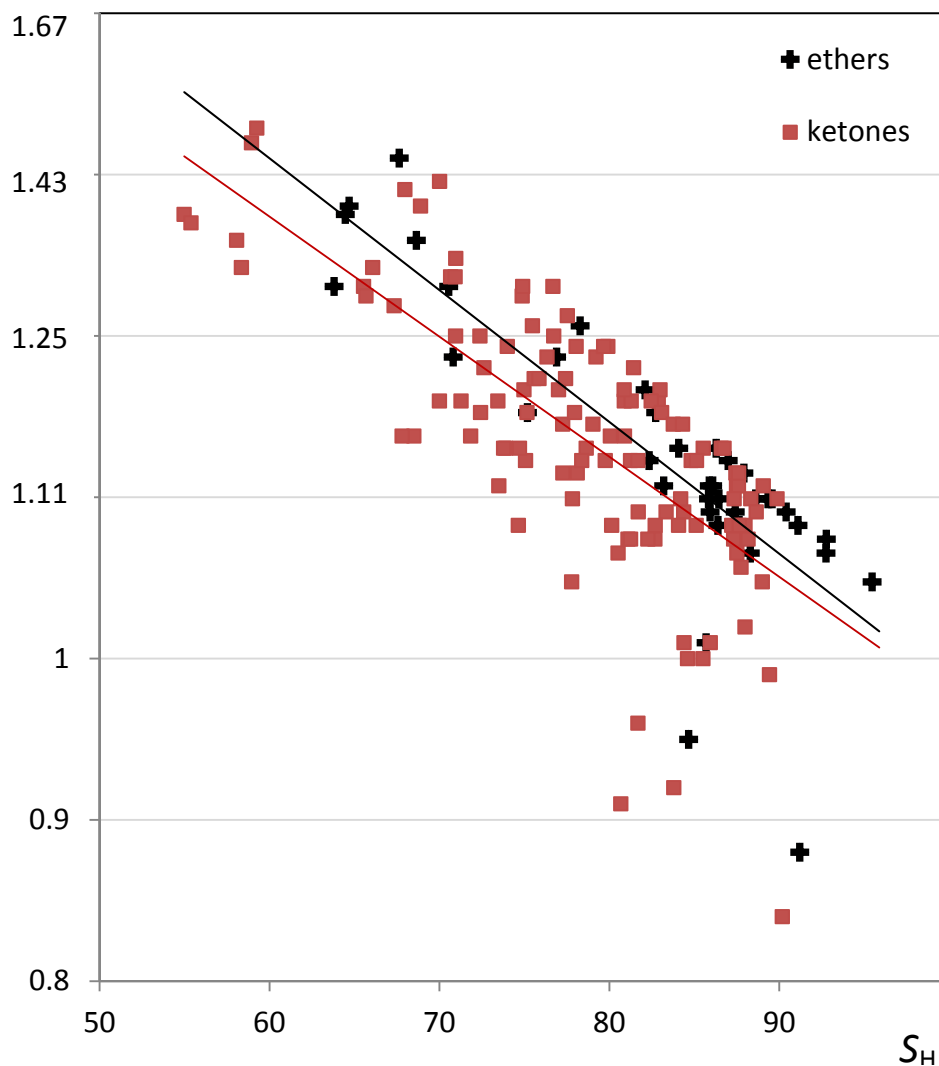


Figure SUP1. Comparison of enrichment ratios E_{OH} for ketones and ethers as a function of hydrogen content on the surface.

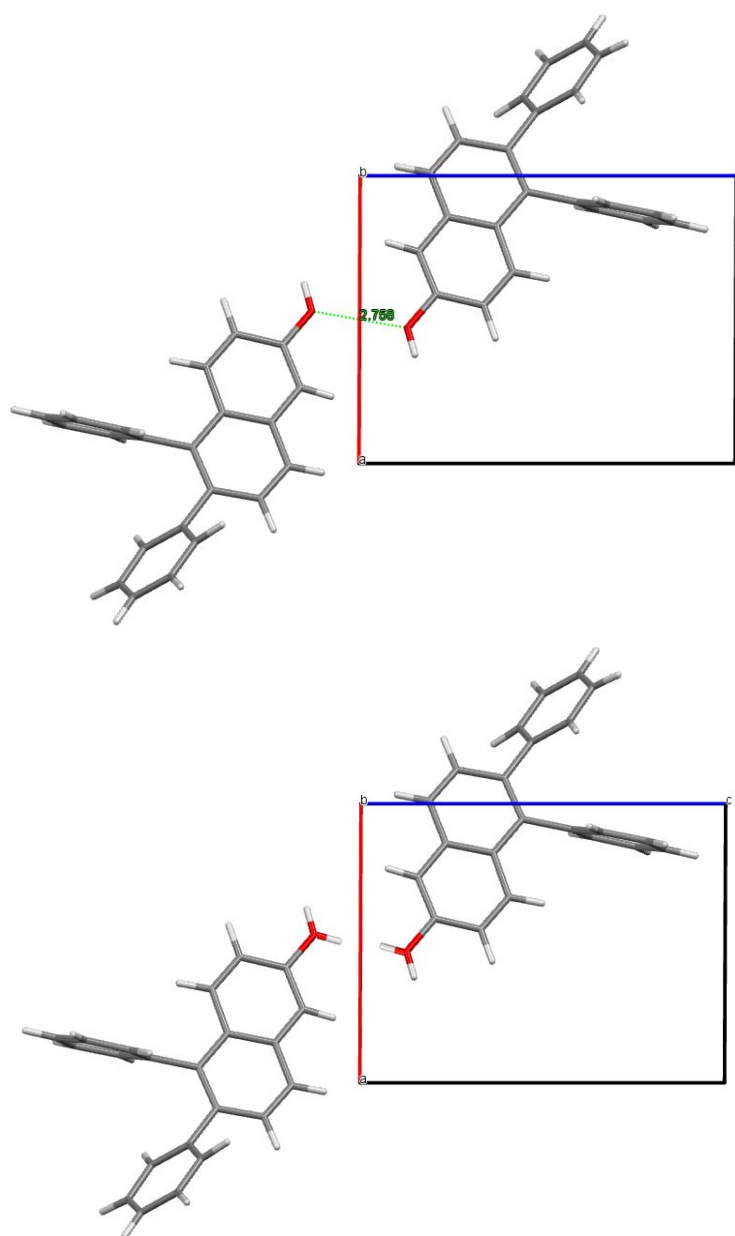


Figure Sup2

View of a dimer of KAZHAU molecule, which is an outlier among phenols with a large $E(O,O) = 11.8$ enrichment ratio. The middle of the O...O short interaction is an inversion center. Top view: the published structure. Bottom view: proposed correct structure with a disordered hydroxyl hydrogen atom forming a O-Ho...O hydrogen bond (occupancies $\frac{1}{2}$). Reference: Stanciu, C., Olmstead, M. M., Phillips, A.D., Stender, M. & Power, P. P., (2003), *Eur. J. Inorg. Chem.*, 3495-3500.

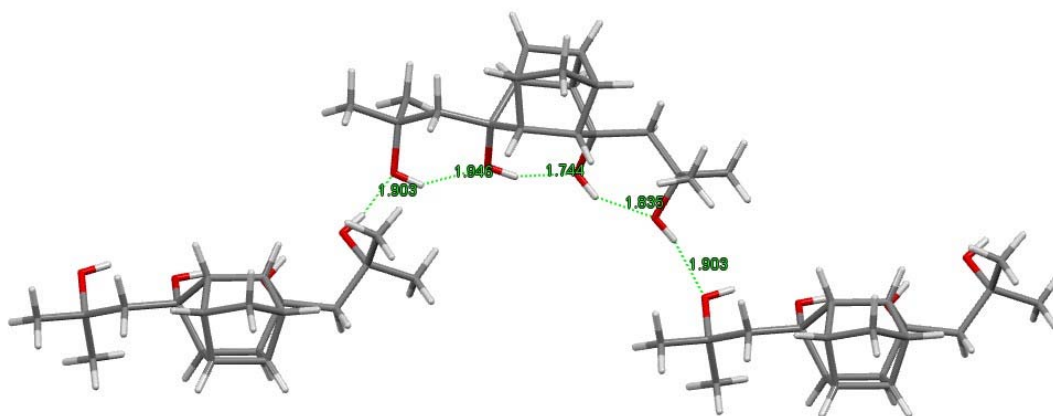


Figure SUP3

Example of an alcohol with an only moderately enriched O...Ho contacts which appears as an outlier in the $[S_o, E(O,HO)]$ scatterplot at position $S_o=7.6$ and $E(O,HO)=2.25$. The molecule is peculiar at it has three intramolecular H-bonds.

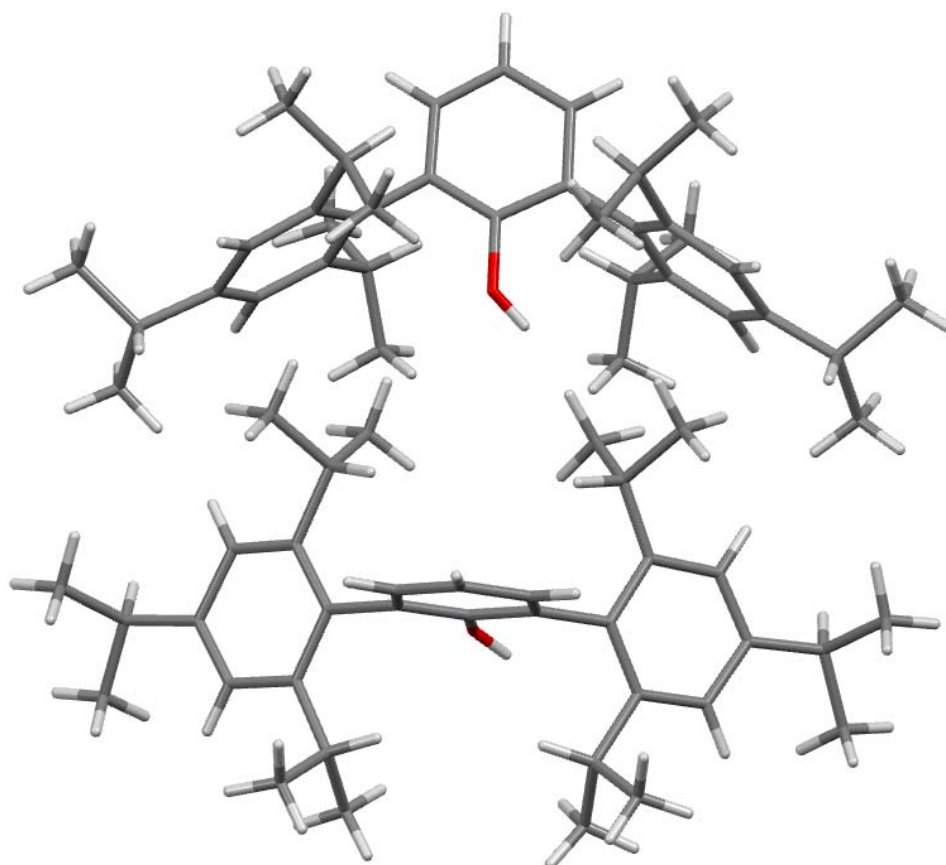


Figure Sup4.

Example of an outlier phenol molecule (refcode AKUSEE) without intermolecular O-H...O hydrogen bond due to the bulky environment around the hydroxyl group (view of a dimer).

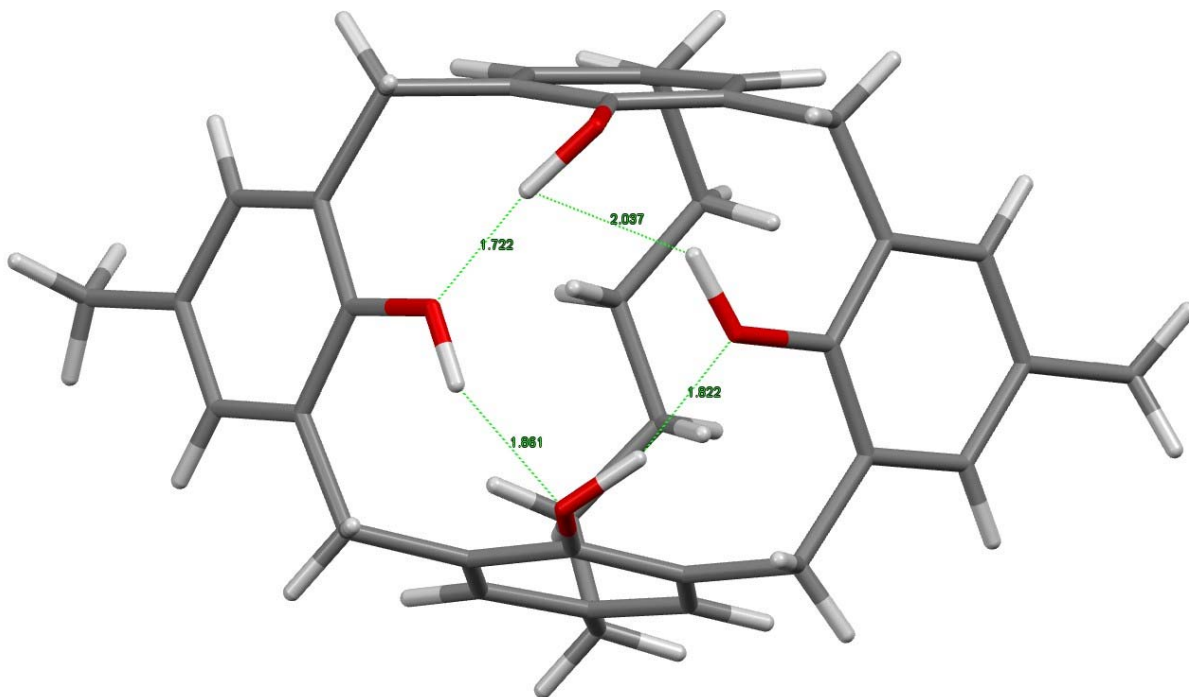


Figure Sup5.

Example of an outlier phenol molecule (refcode JABVUD) without intermolecular O-H...O hydrogen bond but with intramolecular H-bonds.

Figure Sup6.

Example of an outlier phenol molecule (refcode TUJYAZ) without intermolecular O-H...O hydrogen bond but with intramolecular C-H... π hydrogen bonds

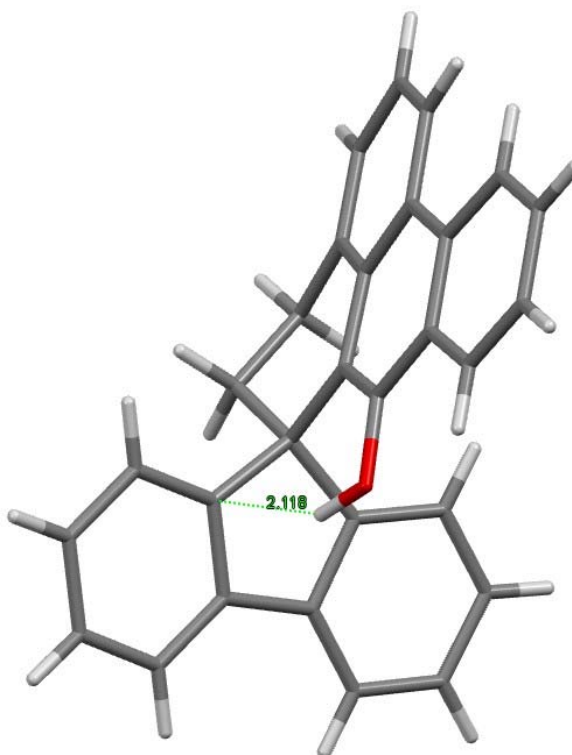


Figure Sup7.

Repartition of Hirshfeld surface content as function of S_O oxygen content in phenols.

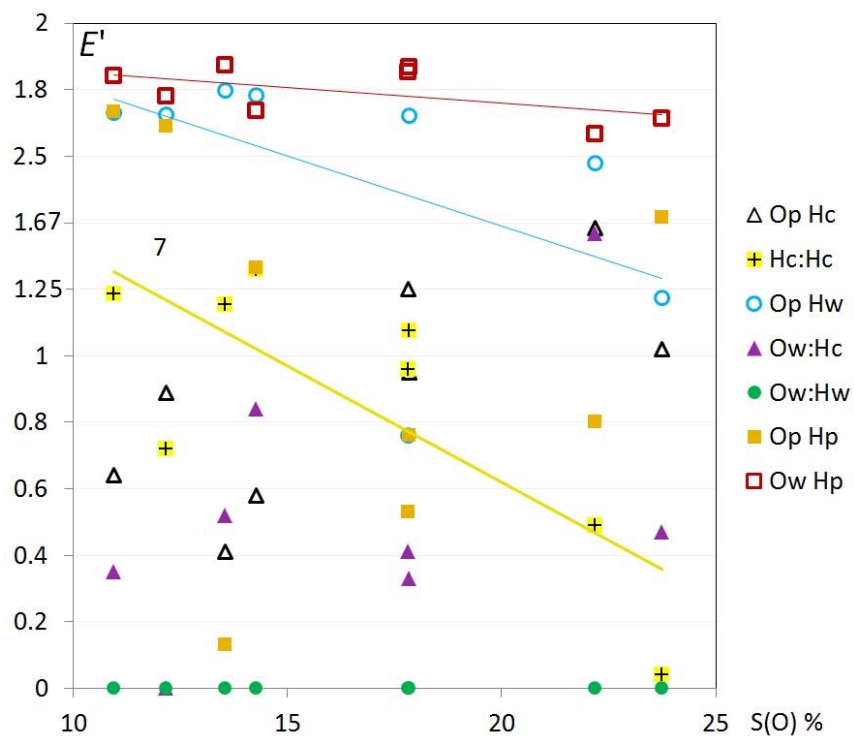
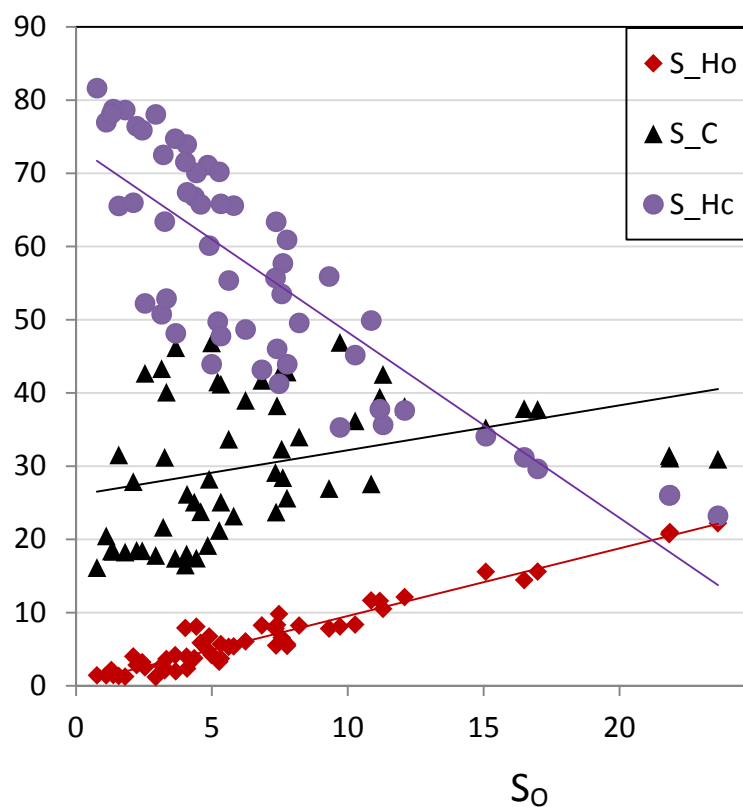


Figure Sup 8. Enrichment of the oxygen...hydrogen contacts in crystals of phenol-monohydrates as a function of total oxygen content on the molecular surface. "p" and "w" suffixes refer to phenol and water, respectively. The E ratios of the Hc...Hc hydrophobic interactions are also displayed.

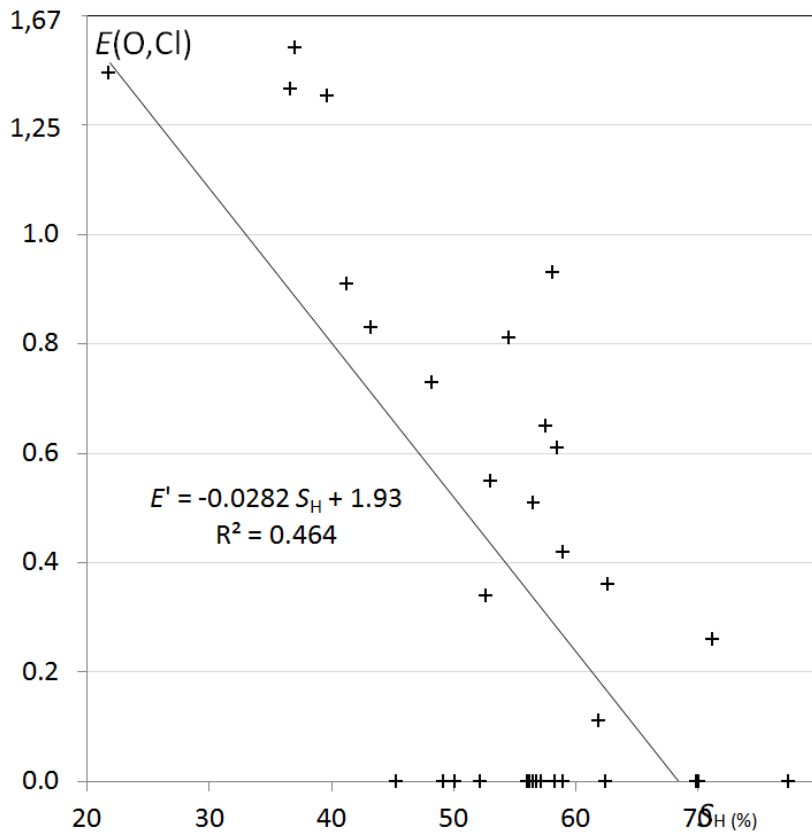


Fig. Sup9. Enrichment $E(O,Cl)$ in chloro-ethers as a function of S_H .

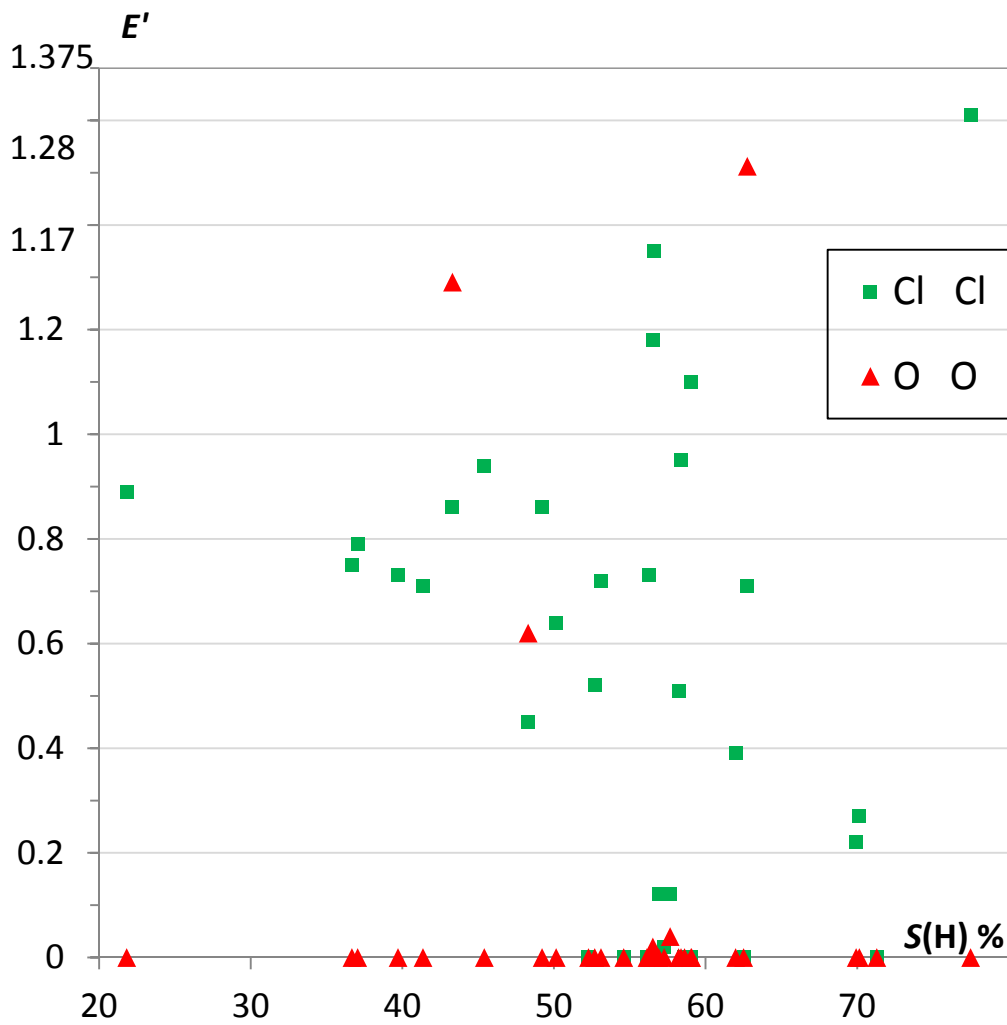


Figure Sup10. Enrichment ratios between chlorine and oxygen atoms in chloro hydrocarbons containing an ether group.

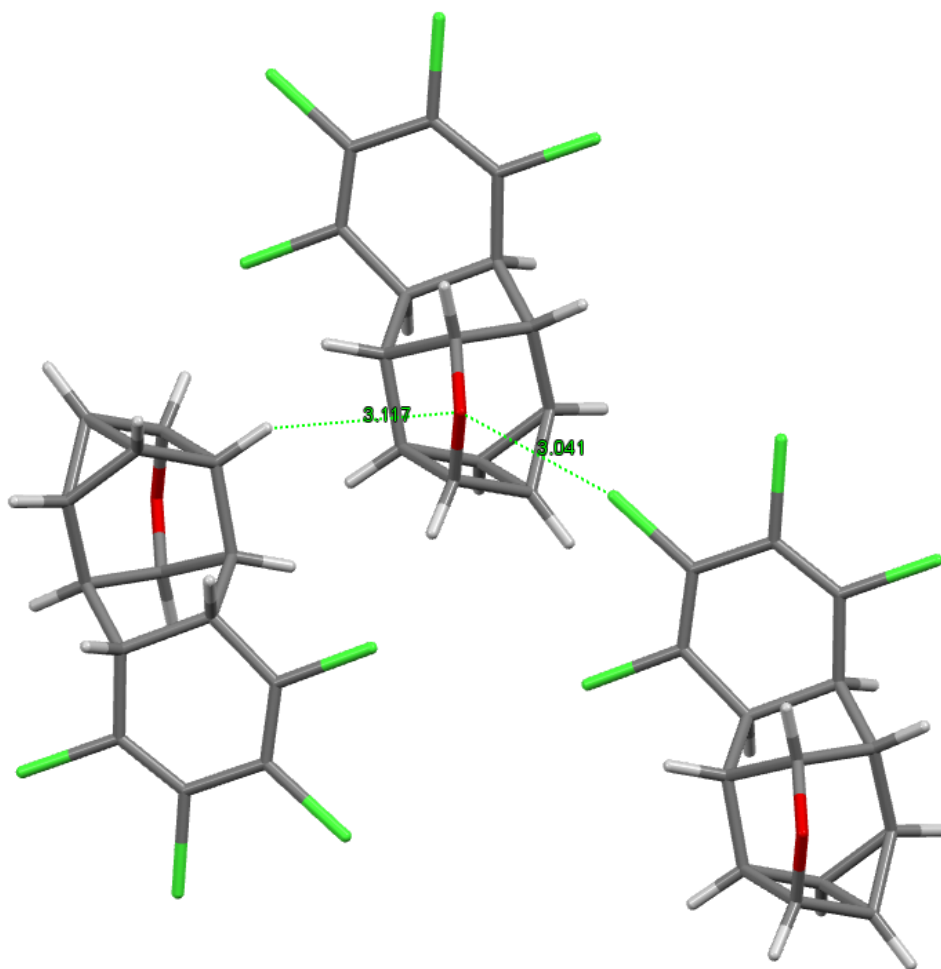


Figure Sup11 Example of a chloro-ether compound with high $E(O,Cl)= 1.34$. The unique ether oxygen atoms forms a $O\dots Cl$ halogen bond and a $C-H\dots O$ weak hydrogen bond. The angle $C-Cl\dots O$ is 158.2° . Compound refcode FUDBAI: Etzkorn, M., Amado-Sierra, M. del R. I., Smeltz, S. D., Gerken, M. (2009), *Tetrahedron Lett.* 50, 2991-2993.

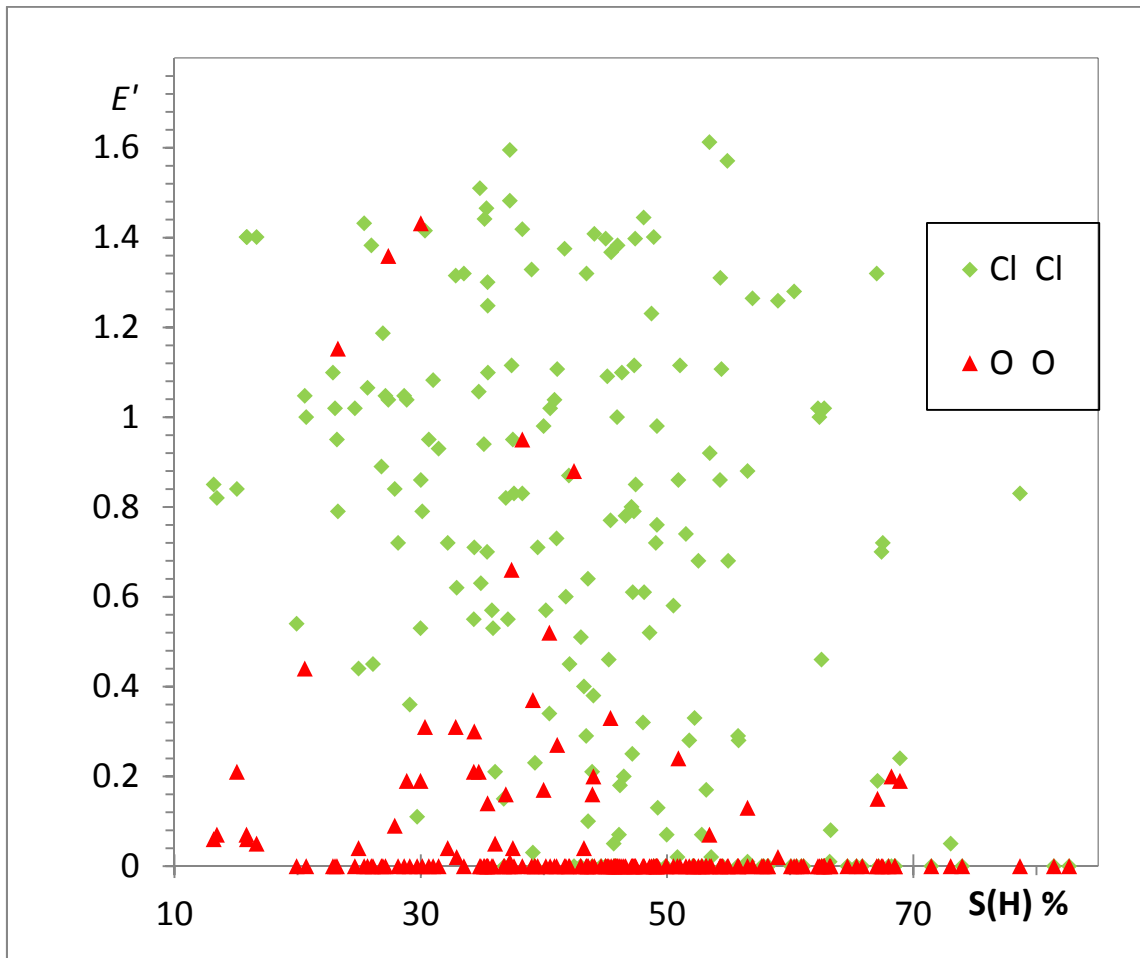


Figure Sup12.

Enrichment ratios between of chlorine...chlorine and of oxygen...oxygen contacts in chlorinated ketones. .

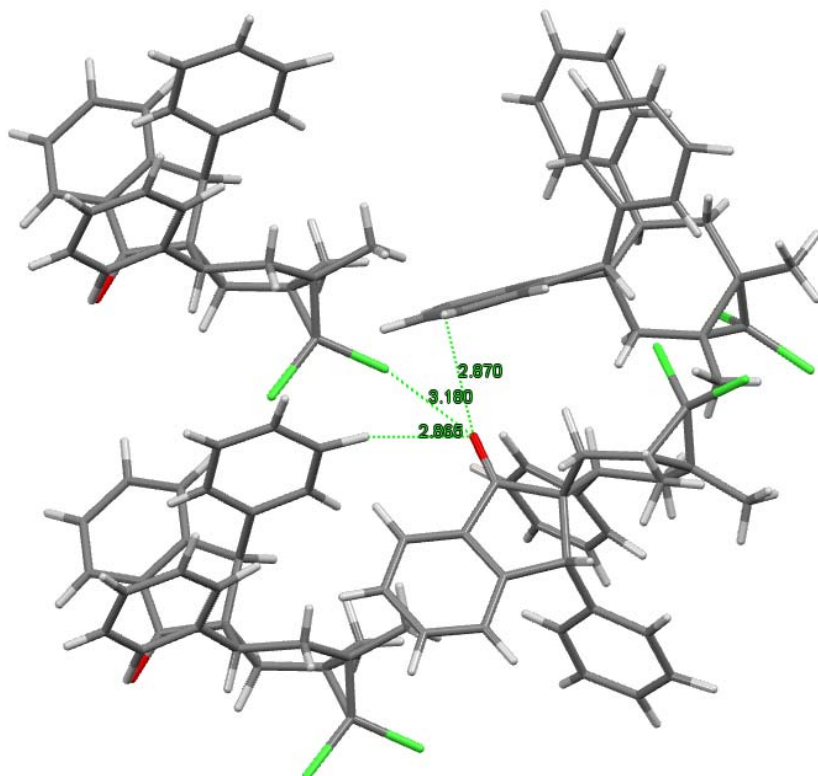


Figure Sup13

Example of a chloro-ketone compound VIKLIK with high $E(O,Cl)=2.58$. The unique oxygen atom forms an $O\dots Cl$ halogen bond and two $C-H\dots O$ weak hydrogen bonds. The angle $C-Cl\dots O$ is 155.7° .

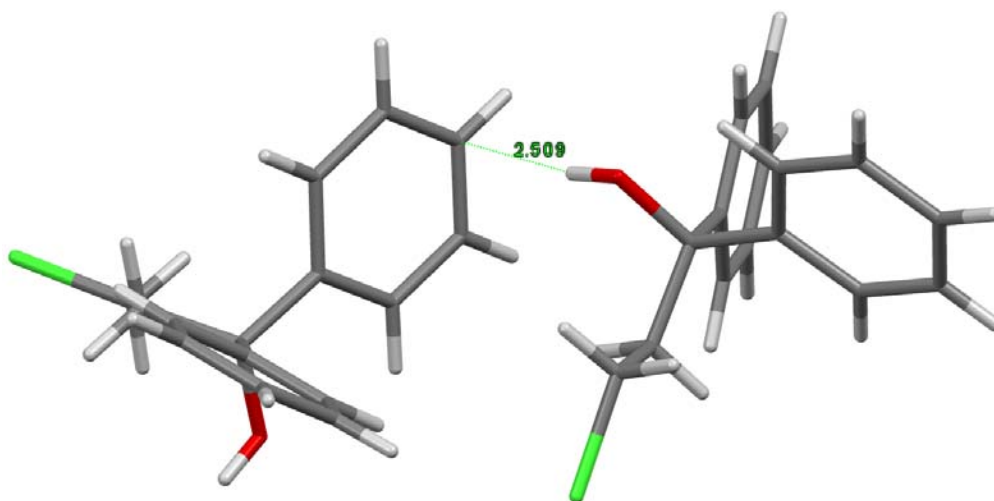


Figure Sup14. Example of an outlier among chloro-alcohol compounds with absence of strong hydrogen $O-H\dots O$ and of $O-H\dots Cl$ contact. The unique hydroxyl hydrogen atoms forms an $O-H\dots \pi$ hydrogen bond with a phenyl ring.

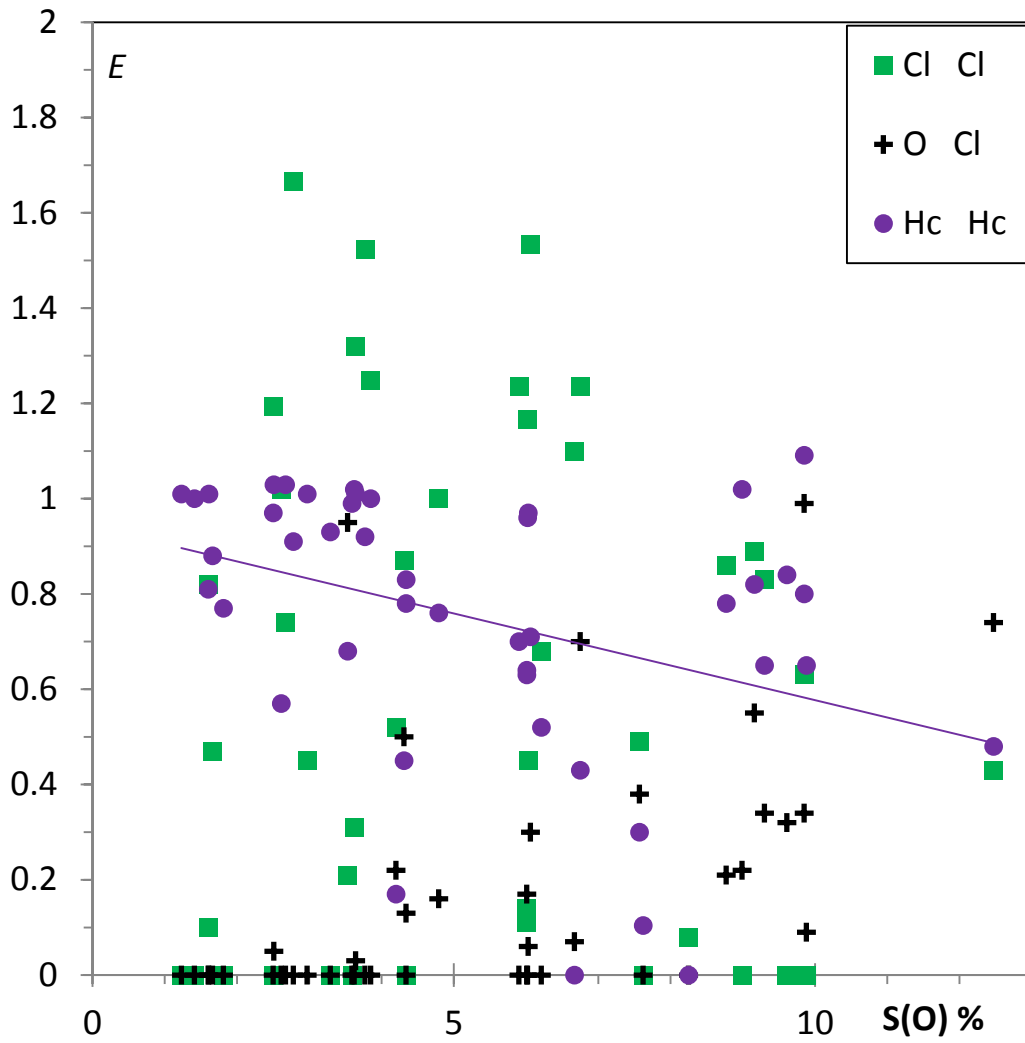


Figure Sup15

Enrichment ratios of some contact types in chloro-alcohols.

Table SUP1

List of oxygenated hydrocarbon compounds retrieved from the Cambridge Structural Database for each family of analyzed molecules.

LIST of ketones

ADETAE ANDION11 ATCDEO BEHZIX BEHZOD BNORSO10 BOJVUS BUCHAI
CASJUB CEDDUK01 CEVVOO CEVXAC CFSOB COQCUG CUFHAM CUJXOU
CUJXUA CUJYAH CYDECO DANJEH DAZVEF DELZID DELZOJ DIWLOK DIWMEB
DIWPIL DIWPOO DUSDOL DUWBEC EBAVAE EJEFOO FONCOB FURGII10
FURGOO10 GEBRAG GETNUO GILZUW HABZAL HEBBEV HEJGUY HIXHIF05
HMTCDO INUHII JAPBIL JAPBOR JAPDAF JIBFOP JOPLOP JULJUV KEPBAJ
KETTUY01 KETVAG10 KIKYOS KIMBOX KOQHON LAVDAN LAVHIZ LAVHOF
LAVHUL LIHMIA LIHMOG LIXMOV LIZNOY LIZNUE LULHIJ MABEST MDHPNO
MICRAT03 MICRAT04 MUSMUI NAFBOL NCUBEB10 NOYWAZ OFOQIL OMCHDO
OXCHHX PADCAY PCDODO PROPTO PUKJUA PUSDEM QEYFAB QIFHESQOYTAZ
QULNIV QULYOL RASWEN RIJGAS ROCMAY SAZVAQ SEJFOC SEJPOM SIHXUE
SIHYAL SILMIJ TABNON TBUCHO TCDODO TCPDON TCTDTP02 TETMOU TETNIP
TETNIP01 TMCYNO TOHZUL UNDECO UYUGAX VAMKEZ01 VARCUM VAXRES
VOQRAU WIHJAZ XEQWOG XONXON YABCOT ZAKNUU ZZZMZM01 ZZZQAI02
ZZZWGK01

LIST of aliphatic ethers

CANNIO CEZSEF CMPANL01 CUKCIU01 DCHXCR01 DCHXCS01 DCHXCS02
DCHXCT DCHXCT01 FOBHIN GASLUI GEFJIK GEFJOQ HAZKAV HOXOCD02
LAKDEH MECZID MOFPAY NABREN NAKWUR OBIBOQ PUTHOB QEGSOL
SEMKOK SIPSAL SISMUC SISNAJ TANMAT TEMCAP TIVHOW TIVJAK TIVJOY
TIVJUE TOXCDP TOXCTD VEQKEH ZZZWGK03 YAGCEQ ZONRAV ZOSSEF

LIST of esters

ADUQIZ AGAJEX AGAJIB ALUVIM BEDNAZ BEFJUR BEKJIK BOBCIE BOBCIE10
BOJZUV BOPVIN BOSPUO CBTCME CINGAH CMEOXE01 CORNIH CORZAK
DADQIK DEVBIP DODCHD DOJFOZ DOJFUF DOJGAM DUFMIA ECOTDA FAFDAS
FAFDAS01 FAFDAT FATRUN FAWLEU FEJKOV FEMLEO FICYUM FITGEU
FOLVUX FUMJON GEJLAK GEQQIC GOGXUV GOKMOK HEGSAN01 HEGSAN10
HEMKOZ HIFWOI HINHUI HINJEV HOSNAE HOYTOF IWUFAH JAXMIF JEDZAT
JEGYIE JIRXIS JORJEF JORJEF01 JORSIS KAMRAS KAXPOP KEJXII KEJXOO
KEMFOY KUZVIN LEDNOX LEDPIT LITLIK MAJYAX MAMNAC MCMSCA10
MESQOR MTCDEC NACCID NETDOG NIFCOU NIFCUA NOSLOW NUQPIZ OMIDOD
PAPVOR PATCUI PEHWAA PEKKEV PERCEN PESFUL PIJQOO PILLEB PIYYIG
QETVUG QIKBER QOYKIY SEDLUI SEJWOT SIQQEP SUKXAX SUNHIS SURCEN
SURCIR TELREH TIHMUS TIHNAZ TIVCOR TORMAP UCOJOM VANLEB VARPEJ
VEBYEG01 VESDAY VESDAZ VEXRAR VODFOJ VOHNAH VUPVOR WECCAI
WECCUC01 WELSUD WIRPAO XAGKEW XEGKID XODHAZ YARZUN03 YIXTUV
ZAMVAK ZILQAM ZORKUM

LIST of nitro ether

ACENUS BAWJIT DADBAM DARGAF EFAXOY FALHUW FEFPEM FESPEY
IPATOJ JUJSIR JULGAY MEMDIS MUBLUR NILNOL NNAPMP NUBZIU
QEDYOP RUJBAA RUJSEV SILRUB SINJIJ VALTEJ YAWPAP ZIKYIB

LIST of alcohols

ACAVIJ AFILEI ASOCER ASOCIV ASOCOB ASOTOR BAJNAB BAVLOZ BCHMOL
BESWON BEVBAF BILMIU BOGKIR BOMKUJ CAVDEI1 CEJXIY CERMOB
CERMUH COYFEB CUVFII CYOCDL01 DABKOG ECEPEJ EHI XOI FECCOF01
FEPKUG FERTAY FEXKOI FIFRIV01 FUNCIA GIGYAW GOJKEV GOMQUV
GOXMAH GUHBIU GURBUQ HANPAN HEBDIB HIYHAY HOCCAD HUKLIJ
HXCHEX ICAHID IQAFAI JAJSOC JAXLUQ JEDBUQ JEQZUB JONLIH JONLON
KAPNAQ KEKPEW KEPSON KEZTIU KEZTOA KOFFUH KOPKUV KUKYAR
KUXFIT LOBQOK LOJXIS LONGAX LUdTUA MAMGEN MATLIC MEGJAJ
MEGJEN MENVIL MINFEV MURABD MUTYOP NAYZOD NEPGCL01 NEZ FON
NIWNUC NULZEA NUMTUK PELBIR PERYTO04 PICYAB PILRAE PIMARB PIRBOJ
PIRLOT PIWXIK PIYXOL PMGLYC POHYEQ POJGUQ POJYIX POMWUK PUFJIJ
QATTEK QATTOU QATVEM QAZQAJ QIMCUK QURRUQ QUTFOA RAJDIP
RIHMAX RIQMAG RIWTIB01 RIYZEF ROGNIL RUKNER RUSKEW RUVWAG
SEVPAL SEVPEP SILGEZ SIQGUU SUPMIA SUPMOG TEZQIA THXCHO10 TIQHAC
TIXXED UDEMEV VAHXAD VEKMED VIDXIR VIWHAL VIWTAX WEFWAH
WEJMAA WERTUI WESTET WESTIX01 WOYKOK WUTMAA XEWKER XEYMIX
XIDQUW XIJPEL YAJYIS YEJNIK YEYLUK ZEPAD ZINWAU ZZZFFY01
ZZZGSC01 ZZZKPE01 ZZZPSA01

Alcohols molecules discarded, due to wrong hydroxyl orientation.

Detected by high $E(O,O)$

BCUNDL CUVFII EHIYEZ GUHBIU GEBSUB NEMXUY QIVYIF
RIQMAG SETJIK SOTBEI TEZQIA UDEMEV VIDXIR

Detected by low $E(O,HO)$

EPFRED02 SETJIK EHIYEZ NEMXUY SOTBEI QIVYIF

LIST of phenols

ABUCOP AXEKOD BECNII BECPAC BEZRIJ BIRKOC01 CARJIQ CARKOX
CATCOL12 CEYKUM CONNID CUFVII CUPDUM DENSOL01 DOGWOL DOGWUR
DOHDPH02 EJEVOE ESTILO03 FEVCAL FIPYAG GALCAY HEGFAB HESKOF
HESKOF01 HYQUIN08 ILABAZ JABWAK KAPVIH LERFET01 LESVEL MEBMIP
NAPHOL01 NIYQUH NIYRAO NPHLDL01 NUTSUQ OTASUY PHGLOL01
PIDZIM PYRGAL02 PYRGAL03 RAGLEQ RASCEU RAWDOI RESORA03
RESORA13 SAGSOJ SIJHOJ SOPXUQ SUZXAN UCOFIC UDEMAR UKILAC
UKILAC01 VOGRUE WAJJOG01 WANNII WETFAD01 WIKFIF WIZNAV XAXSIY
YOTXUB ZZZGWU01

Phenol molecules (retained) with $E(O, Ho)=0$

ACAYIN AKUSEE AKUSII BERKEQ BIPXEF DPPHOL01 EFEMOT FECYUI GANVAS
JABVUD KANJUF KAZHAU LIDJAK MBPHOL01 TUJYAZ XILZEX XMBPPR

Phenol molecules discarded, due to wrong hydroxyl orientation.

KAZHAU

LIST of alcohols-phenols

EHAVAL ENOJOH ENOKAU EWUGUY FEGFAZ GAYKEW GOLMOL IQESEC
IQESIG IQESOM JAGQOX KEKTIE KOHJOG NURXED SAWHON SIPROY
TITHOU UPAHUP XIJFEB XURHUO XUYRUF

Alcohol-phenol molecule discarded

MERHIB

LIST of hydroxyl-carbonyls

ATEPOE BALGAW BASYID BERJIR BEXGEQ BOBVIZ BOKVUS BOVMUU
CUCHOX DAWQUO DESVIG DHPRTO02 DICREM DIYMOO DUNHOK DUZPUJ
DXTMPO EADSUX EFEPOW EGUCOA EHALUU EHNPRG EJABUN ELOPUQ
EYNAPO10 FAFSOV FAGYER01 FAVBAH FICPOY FITYUC FIZLIK FUXMOA01
GELQOD GERRAX GIVROT GODWAZ HIHTIB HIRROQ HMEPRG HMMAZO
HMNPRG HNOBCH HONSUY HPRGDO HTCDDO HXPRED ICUQAZI CUQAZ01
IQOCUM IVOFUU JAZCIX JIWPEL JIWPEL01 JOHMOK KEFCED KIFGIP KUCDER
KUMZEX LAVJAT LESNEE LIQCEU LIXFIH LIXFON LIXFUT LOBPID LOCJOD
LOFZIQ LOSKAG MERBUH MILVUZ MILWAG MIMREG MTHPRG NAMMU
KNETIN DNETIND01 NETIND02 NORGES01 NORGES10 NUBYUF OHMEDC10
OHUNEL PAGLEO PEZFEG01 PIHHAR POKQOX POSJAI PRGDOL02 PUVLOI
QAQVUA QEQMII QEXLUB QINFOI QUYTUA RIHJEY RIMQUB RONTET SEBNUJ
SEHCAK SIYWON TANBEP TEGWAD TETNUB TETPAJ TETPEN TIFNIG TIFNUS
TIMVAM TOLNEO TOLNIS TONWIE UKUBAD ULEMEE UQOHOY URAJON
UZEVEB VABQAS VEXKEO VIDVIP VIHCAR VIHCEV VOHTAP WAMVIQ
WAVWOF WAVXIA WAVYAT WEFWEL WELXIU WEQMAG WOBKAZ YEGZOZ
YOFWOG ZUDQAQ

LIST of hydroxyl-carbonyls dimers

BARYAU EGUCUG EJAYEU EYHENO EZOQUF GAHJEE KESKAV MACKAD1
NOLFIF RAKZOT RHODON SOFJEC VAPROU XETXIF XMPRYO YABCEJ
ZZZPNG01

LIST of alcohols-monohydrate

BABYUA COQCOA ENIVON ESEDMH EVARUP FOZWEX GATVED GIXFID
GIYJAA HXVITD IRIDE JAPZAB MYTOLD01 NAYKUT PECXIE TAGSAV01
TILZUL TIVBOP XAXKEO

Structure discarded : UBUGON

LIST of alcohols-ketones-monohydrate

ACOWAQ BEBFOD BOQSUV CAMTUG EGOTOL GOFUCB ITEJEW JIFYEC
KEFCIH LIBWUQ NIFFEN NUSPEX PIRBUP SEZBAB TOVTED UQABIY
VIDVAH VOQWEE WINLIP XABMAQ XEPGAC YAFQIG YAHQED

Structures discarded: ITEJEW YAHQED

LIST of chloro-ethers

AXAGUC AZOPEL BONDEP EDIZEW FAMFOO FUDBAI FUYJUF HEWJIE HUGBAN
IDEFIG JEHJOV LINPIH LOCBEM MERQIK MERQOQ MERQUW MERRAD MERREH
MILPUT MONGAY MUWROM NAVGAU ODOTUW ODOVEI OSOKEN PODJIC
PUYSIL REBKOY TITQAO UDANAP VAFBEJ WINRUG YAWLIT

LIST of chloro-ketones

ABEFOD ACOQIU AGAQUW AGOLEN AXAREX AZIBER BAGZIS BAMYUJ
BEMMEL BIMVAV BIMVEZ BOKJIW BZQDCL10 BZQDCL11 CBCYHO CEKZEX
CIGGUU CIQFAK CLBENQ CLMANO CLPMCY01 CLTCOT CODHAE CORCUH
CORDAO COXBST CPCPRO10 DALKIM DASMIU DASNOA DCLBQN DCLBQN01
DCLBQN03 DCLBZQ20 DIKYUR DISWEJ DISZIO DITYAG DITYAG10 DIYYUG
DOXQIQ DUJPUU DUVFOP20 DUXTUL DUZSIA DUZSUM DUZTAT DUZTIB
EBEQOR ECECUL EFUFER EKUYUF ELEWEX EROFAT EVIKAW EXODAW
EXODIE EXODOK EYISIO FABGUK FABRAB10 FABRAB11 FAKKAD FAKKEH
FAKKOR FOMSAB FUWNAM GATRAU GAVBEL GICMUA GIHGAG GIKNUJ
HACJOK HENBEH HILFUF HIZGOO HOQTAJ HUDHAQ HUFZAK HUGQOQ IBEP
IDIRIY IFINES IFORIG IJOWUA INEHAK IPUVOF JEQVUX01 JIZQEQ JIZQEQ01
JORGEC JORGIG KAFXIY KAVDOA KAVDUG KIDBUU KOSKIM KUMKUY
KUMKUY10 KUNRAM LAPGIS LAPHAL LAPROL LEBGUU LEGXOM LEJDAH
LEJDEL LEJDIP LEJDOV LEJYUU LUGMUW MATDUG MEGHUD MEHGAI MIGXAC
MIYCED MIYLAH MODSUU MUCONC10 MUDBUJ MUDREJ MUQZED NAFDAY
NAFQEQ NORBAY OGOROR OHIBOW02 OKEPUP OMIVAH OMIZUF PABZAT
PAMPEY PAQJUO PEJHOD PEJHOD01 PEJJOF PEJJOF01 PEJKAS PEKDUF
PIQLOQ PIQSEN PIRDEB PIWRES POJWUG POJXAN PONPEO PUKJEL PUQSOJ
PUYWUB QAVGID QEYAN QEWXIA QEWXIA01 QIQBEZ QOWNAR RISHOR
ROFVIS ROGPEK ROMVEV RUFGOP RUPSIE SENZAO SIMTUE SUQDAJ
SUQFAL SUQFIT SUQGEQ TAJBUB TAWLAE TCOXHN TIFZOX TIGSAE
TIMNUA TIVBEG TOMZIG TURBAK UBIKEV UCOXUG UNUVAB UREKEI
UVEKOW VEDXEH VIDHOF VIKLIK VUKVEC WAHNOI WAKWOW WEDMOH
WIRCUV WOBBUL WOJGOS XEGLEA XEZHEP XEZNEX XUGVOK XUGYAZ
XUGZEE YAKHOH YIFDUN YIVGIU YUDGAG ZEPVOH ZOBVAN

LIST of chloro-alcohols

AKEKEH ANALEH CUGPEA CUGPIE DADZAK DADZIS DAWMOE EJIWOJ
EXOPEM GAKNAH HHCLMI HOYCOO HOYCUU ICUREE ICUROO ICUSAB ICUSOP
IDOGEN IDURII IKUYUJ KIRBAQ LAKFOS LEPUY LESPOQ MUDZUH NESVAJ
NETBAQ OCENOA OCENUG QEVRIQ QEWYUN RAJFIT SIFXEL TEFSAA TIMROW
TIMRUC UFEQON ULEPOR VIWNAS WECPEB WECPIF WEGGOF WEGHAS WIXYEI
XEHKIE XEHLEB01 XULSOM

Table Sup 2.

Statistics on E and E' values for all families of compounds.

Definition used here for "Mean value E' " : $\langle E \rangle = \langle E' \rangle$ if $\langle E' \rangle$ is smaller than 1.

$\langle E \rangle = 2 - 1 / \langle E' \rangle$ if $\langle E' \rangle >$ is larger than 1.

ssd E' : sample standard deviation

Std err Mean E' : Standard deviation of Mean E'

No. structures : Number of structures

Ketones

Contact	H...H	O...H	O...O
Mean value E'	0,97	1,14	0,02
Mean value E	0,97	1,16	0,02
ssd E'	0,04	0,08	0,07
Std err Mean E'	0,004	0,01	0,01
Std err Mean E	0,004	0,01	0,01
No. structures	119		

Ethers

Contact	H...H	C...H	O...H
Mean value E'	0,97	1,04	1,12
Mean value E	0,97	1,05	1,14
ssd E'	0,04	0,08	0,09
Std err Mean E'	0,01	0,01	0,01
Std err Mean E	0,01	0,01	0,01
No. structures	39		

Nitro-Ethers

Contact	H...H	H...Occ	H...On
Mean value E'	0,84	1,03	1,31
Mean value E	0,86	1,03	1,45
Ssd E'	0,23	0,49	0,11
Std err Mean E'	0,05	0,10	0,02
Std err Mean E	0,05	0,10	0,02
No. structures	24		

Esters

Contact	H...H	H...Occ	H...O=c
Mean value E'	0,92	1,07	1,19
Mean value E	0,93	1,07	1,23
ssd E'	0,07	0,12	0,10
Std err Mean E'	0,01	0,01	0,01
No. structures	123		

Alcohols

Contact	O...Ho	O...O	C...O	Hc...Ho	O...Hc	Hc...Hc	C...Hc
Mean value E'	1,80	0,10	0,25	0,54	0,56	1,09	1,11
Mean value E	5,08	0,10	0,25	0,54	0,56	1,10	1,13
ssd E'	0,11	0,37	0,30	0,16	0,16	0,06	0,14
Std err Mean E'	0,01	0,03	0,03	0,01	0,01	0,00	0,01
No. structures	144						

Phenols

Contact	Hc...Ho	O...Hc	Hc...Hc	C...O	C...Hc	O...Ho	O...O
Mean value E'	0,60	0,88	0,92	0,37	1,14	1,41	0,17
Mean value E	0,60	0,88	0,92	0,37	1,17	1,69	0,17
ssd E'	0,29	0,28	0,22	0,34	0,16	0,76	0,56
Std err Mean E'	0,03	0,03	0,02	0,04	0,02	0,08	0,06
No. structures	81						

Alcohol-Phenol

Contact	C...Ha	C...Hp	C...C	Hc...Ha	Hc...Hp	Hc...Oa	Hc...Op
Mean value E'	0,81	0,22	0,69	0,64	0,69	0,61	0,94
Mean value E	0,81	0,22	0,69	0,64	0,69	0,61	0,94
ssd E'	0,61	0,32	0,33	0,32	0,30	0,28	0,25
Std err Mean E'	0,14	0,07	0,07	0,07	0,07	0,06	0,06
No. structures	20						

Contact	C...Hc	Hc...Hc	Hp...Oa	Hp...Op	Ha...Oa	Ha...Op
Mean value E'	1,15	1,00	1,63	0,18	0,90	1,20
Mean value E	1,18	1,00	2,69	0,18	0,90	1,25
ssd E'	0,10	0,08	0,70	0,54	0,78	0,91
Std err Mean E'	0,02	0,02	0,16	0,12	0,17	0,20

Alcohol-ketone

Contact	Hc...Ha	Hc...Hc	Hc...Oa	Hc...O=c	C...Hc	Ha...Oa	Ha...O=c	C...Ha
Mean value E'	0,65	0,99	1,08	0,96	1,06	0,43	1,55	0,50
Mean value E	0,65	0,99	1,08	0,96	1,07	0,43	2,22	0,50
ssd E'	0,23	0,06	0,24	0,22	0,11	0,73	0,64	0,44
Std err Mean E'	0,02	0,01	0,02	0,02	0,01	0,06	0,06	0,04
No. structures	133							

Alcohol monohydrate

contact	Oa...Ha	Ow...Ha	Oa...Hw	Ow...Hw	Oa...Hc	Ow...Hc
Mean value E'	1,32	1,80	1,75	0,40	0,46	0,61
Mean value E	1,47	5,12	3,99	0,40	0,46	0,61
ssd E'	0,70	0,12	0,17	0,66	0,25	0,28
Std err Mean E'	0,16	0,03	0,04	0,15	0,06	0,07
No. structures	19					

contact	Hc...C	Hc...Hc	Hw...Ha	Hc...Ha	Hw...Hw	Hw...Hc	Ha...Ha
Mean value E'	1,14	1,15	1,04	0,42	0,48	0,69	0,78
Mean value E	1,16	1,17	1,04	0,42	0,48	0,69	0,78
ssd E'	0,26	0,14	0,50	0,21	0,51	0,16	0,52
Std err Mean E'	0,06	0,03	0,11	0,05	0,12	0,04	0,12

Alcohol-ketone monohydrate

Contact	Ha...Oa	Hw...Oa	Ha...Ow	Hw...Ow	Ha...O=c	Hw...O=c
Mean value E'	0,59	1,48	1,86	0,13	0,53	1,46
Mean value E	0,71	1,93	7,01	0,53	0,68	1,84
ssd E'	0,81	0,62	0,05	0,39	0,81	0,60
Std err Mean E'	0,17	0,13	0,01	0,08	0,17	0,13
No. structures	22					

Contact	Hc...O=c	Hc...Oa	Hc...Ow
Mean value E'	0,94	0,81	0,82
Mean value E	0,95	0,84	0,85
ssd E'	0,23	0,37	0,27
Std err Mean E'	0,05	0,08	0,06

Chloro-ether

Contact	Cl...Cl	Cl...Hc	Hc...Hc	C...Cl	C...Hc	C...C	O...Cl	O...Hc
Mean value E'	0,55	1,17	0,84	0,95	1,00	0,77	0,40	1,24
Mean value E	0,69	1,20	0,86	0,95	1,00	0,81	0,62	1,32
ssd E'	0,44	0,10	0,10	0,35	0,25	0,44	0,46	0,27
Std err Mean E'	0,08	0,02	0,02	0,06	0,04	0,08	0,08	0,05
No. structures	33							
Contact	O...C	O...O						
Mean value E'	0,35	0,11						
Mean value E	0,61	0,53						
ssd E'	0,43	0,35						
Std err Mean E'	0,08	0,06						

Chloro ketones

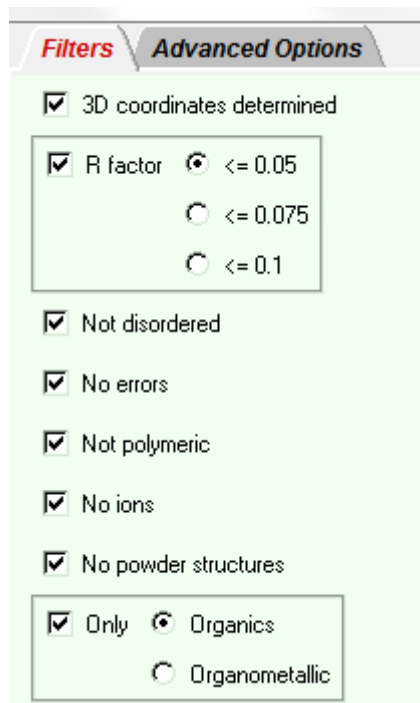
Contact	Cl...Cl	Cl...H	H...H	C...Cl	C...H	C...C	O...Cl	O...H
Mean value E'	0,64	1,16	0,78	0,87	0,89	0,99	0,44	1,36
Mean value E	0,74	1,19	0,82	0,88	0,90	0,99	0,64	1,55
ssd E'	0,51	0,20	0,23	0,34	0,27	0,44	0,45	0,16
Std err Mean E'	0,04	0,01	0,02	0,02	0,02	0,03	0,03	0,01
No. structures	203							
Contact	O...C	O...O						
Mean value E'	0,63	0,06						
Mean value E	0,73	0,52						
ssd E'	0,52	0,20						
Std err Mean E'	0,04	0,01						

Chloro-alcohols

Contact	Cl...Cl	Cl...Ho	Ho...Ho	C...Cl	C...Ho	C...C	O...Cl	O...C
Mean value E'	0,56	0,55	0,49	0,85	0,64	0,82	0,17	0,63
Mean value E	0,70	0,69	0,66	0,87	0,73	0,85	0,55	0,73
ssd E'	0,53	0,62	0,60	0,36	0,54	0,49	0,26	2,67
Std err Mean E'	0,08	0,09	0,09	0,05	0,08	0,07	0,04	0,40
No. structures	45							
Contact	O...Ho	O...O	Cl...Hc	Hc...Ho	C...Hc	O...Hc	Hc...Hc	
Mean value E'	1,48	0,06	1,18	0,60	1,07	0,90	0,75	
Mean value E	1,91	0,52	1,21	0,71	1,08	0,91	0,80	
ssd E'	0,75	0,27	0,23	0,28	0,25	0,43	0,29	
Std err Mean E'	0,11	0,04	0,03	0,04	0,04	0,06	0,04	

Protocol to search molecules in the CSD.

To find molecules in the CSD , the following options were always applied:



Filters **Advanced Options**

- 3D coordinates determined
- R factor
 - ≤ 0.05
 - ≤ 0.075
 - ≤ 0.1
- Not disordered
- No errors
- Not polymeric
- No ions
- No powder structures
- Only
 - Organics
 - Organometallic

To search, for example, the chloro-ether molecules in the CSD with Conquest, the following steps were applied:

- 1) Molecules containing C H O Cl were searched
- 2) Molecules containing an ether C-O-C group were searched
- 3) Molecules containing a O-H group were searched
- 4) Molecules containing a C=O group were searched
- 5) A combination of : must have (1) & (2) must not have (3) & (4) was done
- 6) Remaining molecules were inspected on screen and selected if suitable. Molecules with missing hydrogen, errors, with solvent molecules or multimers were discarded.