

1 **Supplementary Table 1.** Molar mass (M), O:C and T_g for SOA products in volatility
 2 bins.

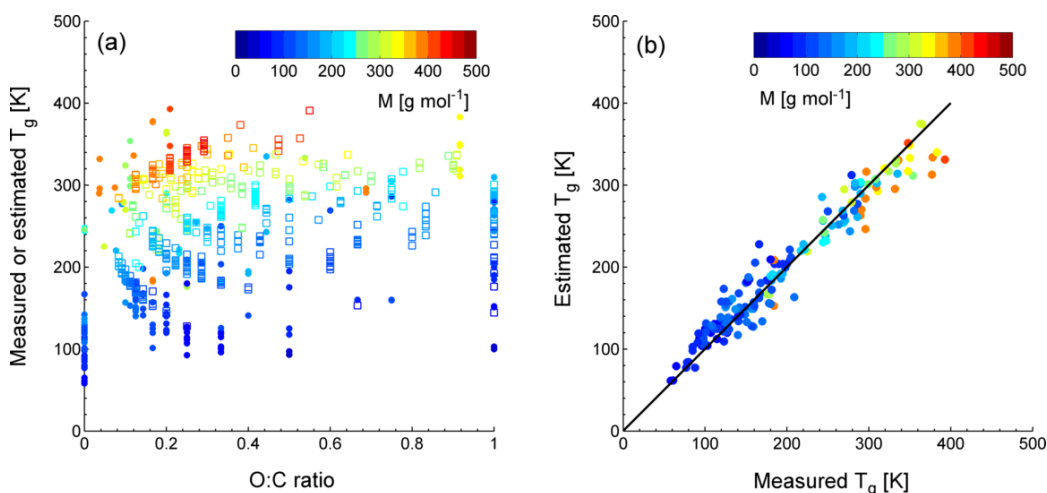
| Volatility ($\mu\text{g m}^{-3}$) | Anthropogenic SOA | | | Biogenic SOA | | |
|--|-----------------------------|------|-----------|-----------------------------|------|-----------|
| | M (g mol^{-1}) | O:C | T_g (K) | M (g mol^{-1}) | O:C | T_g (K) |
| 1 | 362 | 0.6 | 326 | 293 | 0.4 | 298 |
| 10 | 249 | 0.4 | 276 | 203 | 0.24 | 234 |
| 100 | 227 | 0.3 | 256 | 185 | 0.14 | 211 |
| 1000 | 205 | 0.25 | 236 | 167 | 0.1 | 192 |

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5 **Supplementary Table 2.** Molar mass distributions for sensitivity studies of T_g
 6 prediction.

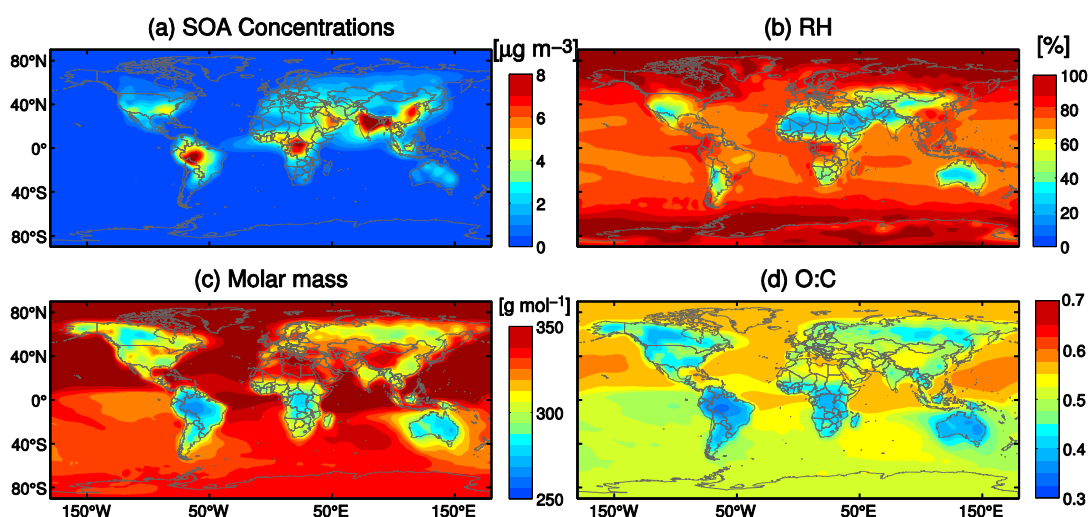
| Volatility ($\mu\text{g m}^{-3}$) | Anthropogenic SOA | | Biogenic SOA | |
|--|-------------------|----------|--------------|----------|
| | High case | Low case | High case | Low case |
| 1 | 434 | 290 | 352 | 234 |
| 10 | 299 | 199 | 244 | 162 |
| 100 | 272 | 182 | 222 | 148 |
| 1000 | 246 | 164 | 200 | 134 |

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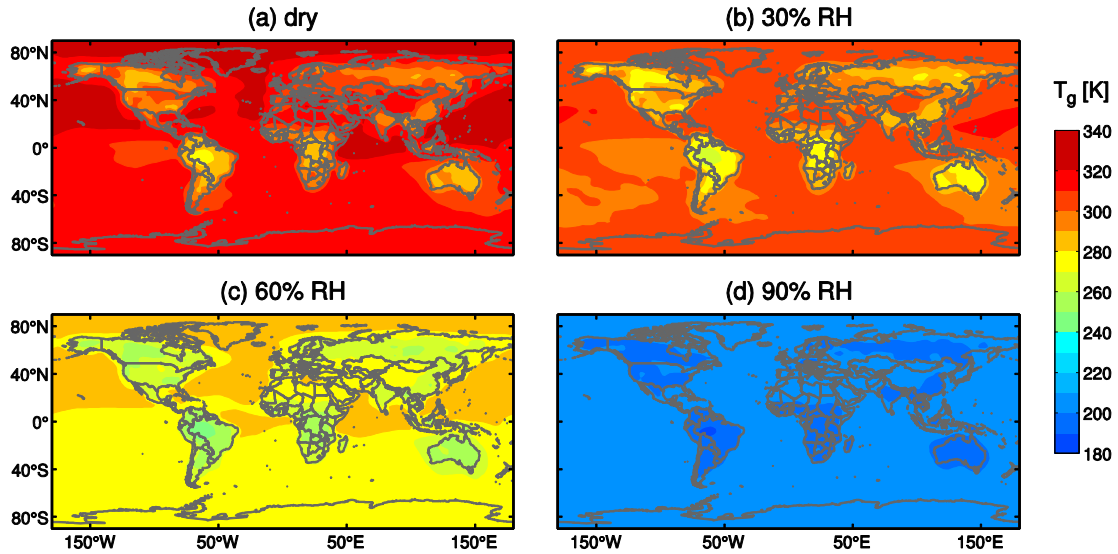


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 9 **Supplementary Figure 1. Characteristic relations between molecular O:C ratio**
 10 **and glass transition temperature of organic compounds.** (a) Measured (circles)
 11 and estimated (squares) glass transition temperature (T_g) of organic compounds as a
 12 function of O:C ratio. Organic compounds with measured T_g are from Koop et al.
 13 (2011)¹ and Dette et al. (2014)². Those with estimated T_g are 654 SOA components
 14 from Shiraiwa et al. (2014)³. (b) Comparison of measured and estimated T_g for 179
 15 organic compounds^{1,2}. The markers are color-coded by molar mass. The correlation
 16 coefficient is 0.97, demonstrating that the estimation method of T_g using EPI and the
 17 Boyer-Kauzmann rule is adequate.

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 20 **Supplementary Figure 2. Global modeling of SOA.** Modeled annual averages of (a)
 21 SOA concentrations, (b) relative humidity, (c) molar mass and (d) O:C at the surface
 22 during the years 2005-2009.



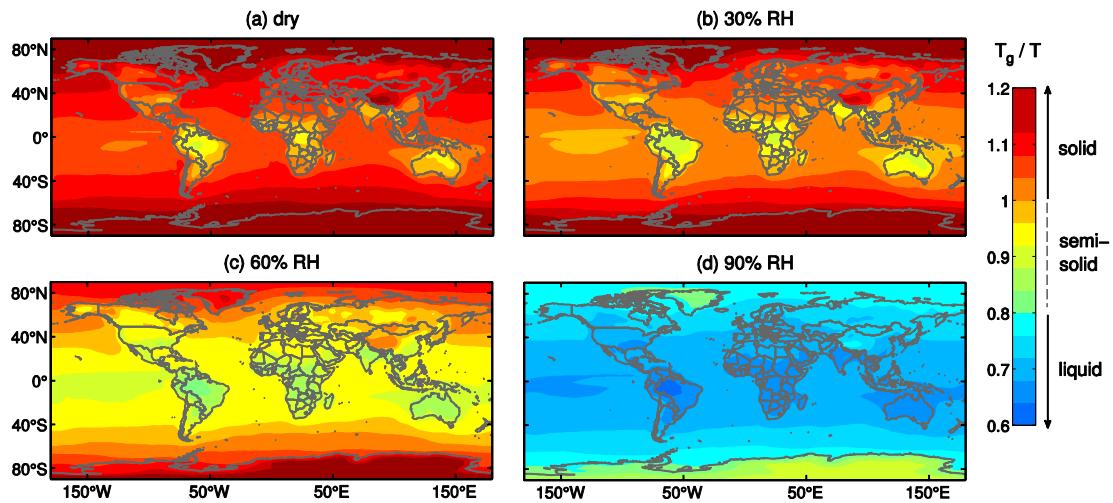
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24 **Supplementary Figure 3. Modeled annual averages of T_g of SOA particles.** (a)

25 Dry condition, (b) 30% RH, (c) 60% RH, and (d) 90% RH at the surface during the

26 years 2005-2009.

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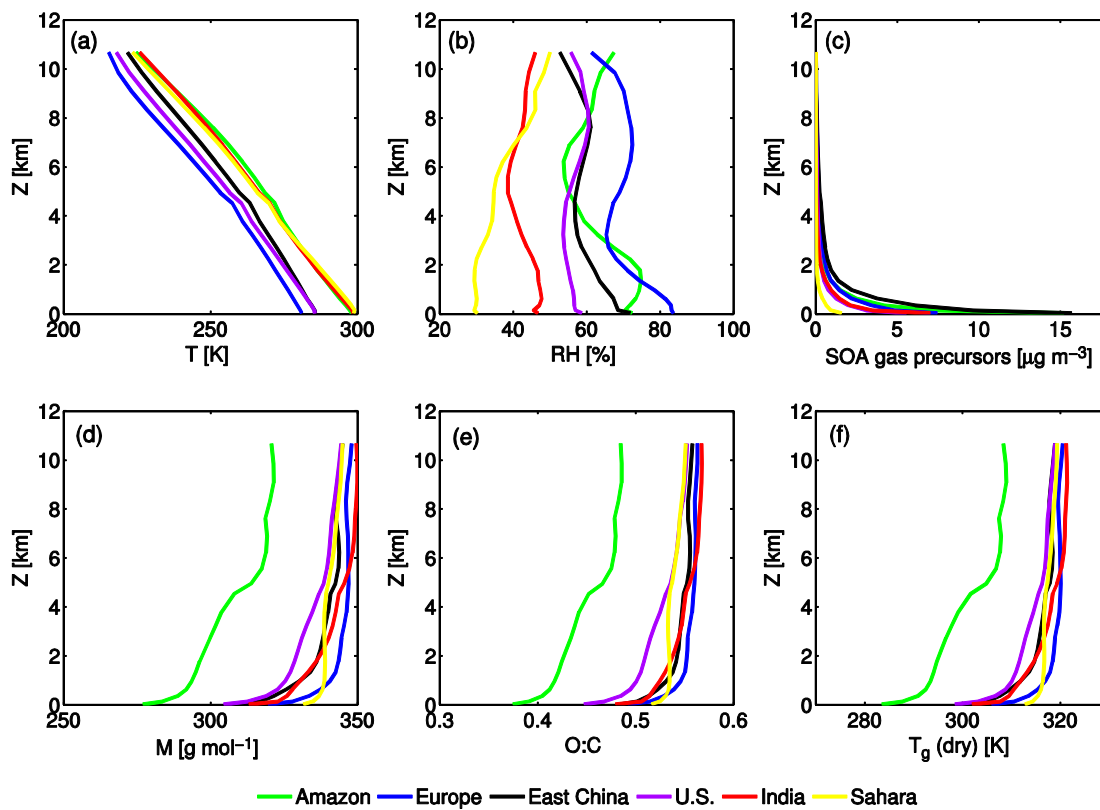
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29 **Supplementary Figure 4. Modeled annual averages of the inverse ambient**

30 **temperature ($1/T$) scaled by the glass transition temperature (T_g) of SOA (T_g/T).**

31 (a) Dry condition, (b) 30% RH, (c) 60% RH, and (d) 90% RH at the surface during

32 the years 2005-2009.

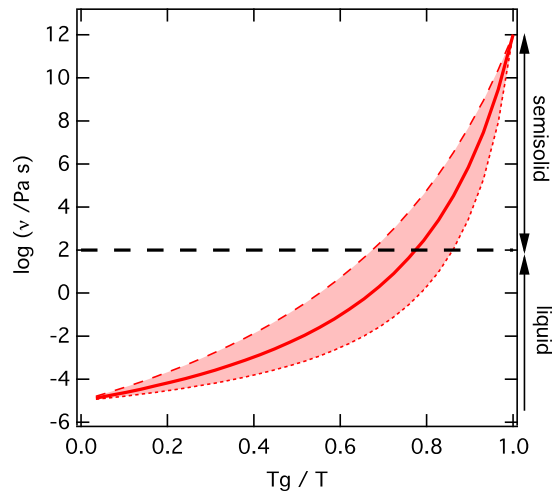


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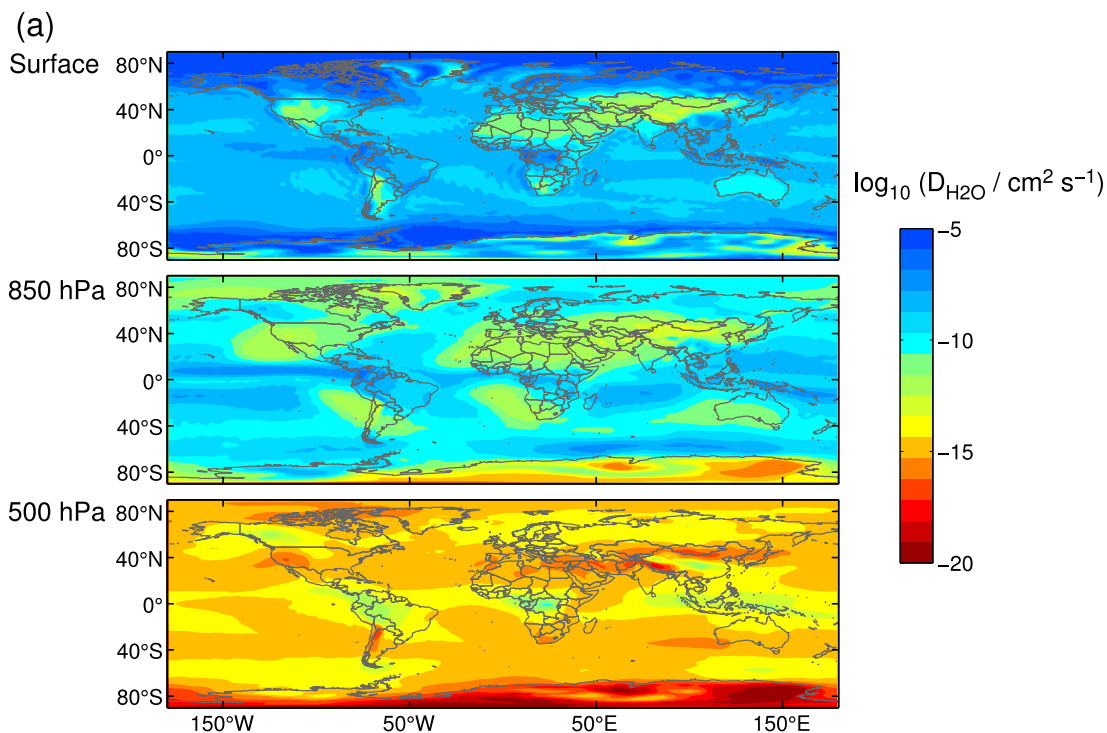
35 **Supplementary Figure 5. Modeled mean vertical profiles.** (a) ambient temperature,
 36 (b) relative humidity, (c) concentrations of SOA gas precursors, (d) molar mass, (e)
 37 O:C, and (f) T_g of dry SOA. The simulation grids covered by the Amazon basin,
 38 Europe, East China, U.S., India and Sahara are shown in Fig. 2(a).

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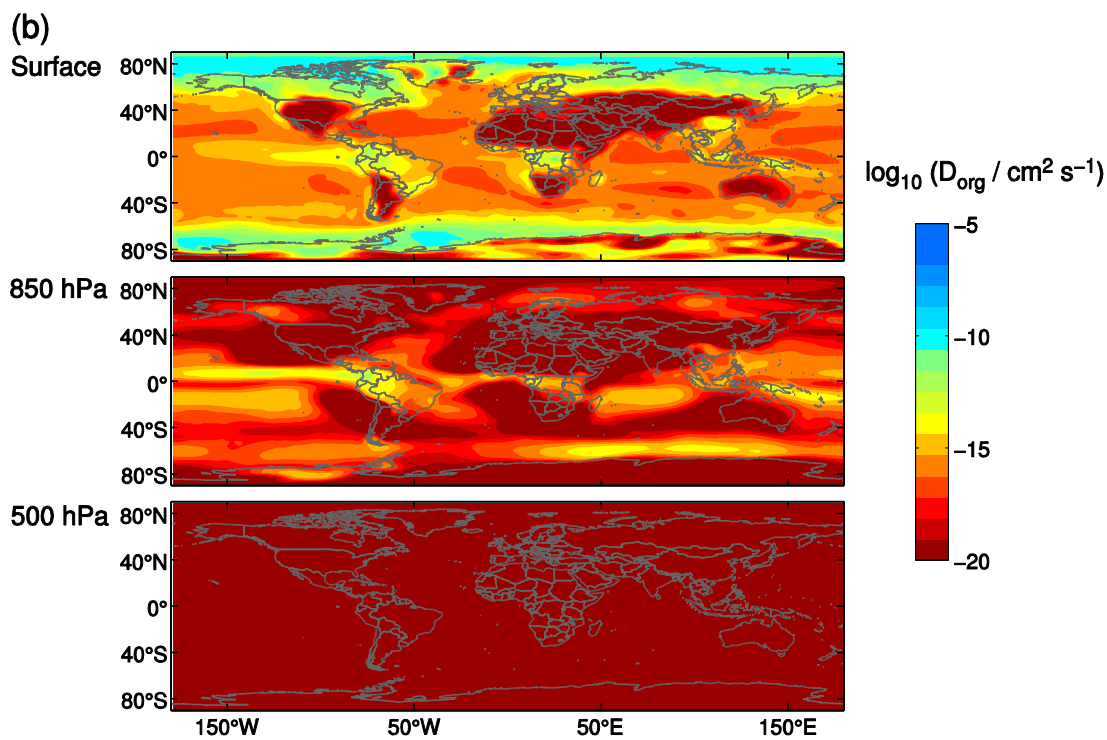


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41 **Supplementary Figure 6. The Angell plot of viscosity vs. T_g / T .** The lines represent
 42 different fragility (D) of $D = 10$ (the solid line) as the base case for this study as well
 43 as $D = 5$ (the dotted line) and $D = 20$ (the dashed line) representing the possible range
 44 for SOA. The black dashed line at viscosity of 10^2 Pa s indicates the threshold of
 45 liquid and semisolid states.

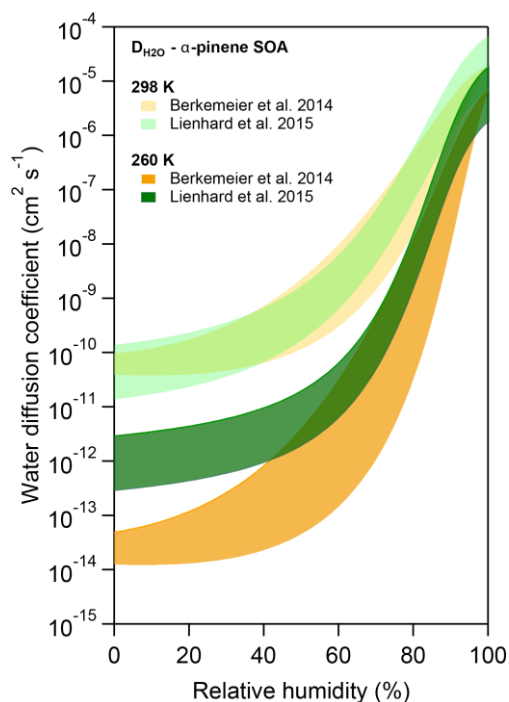


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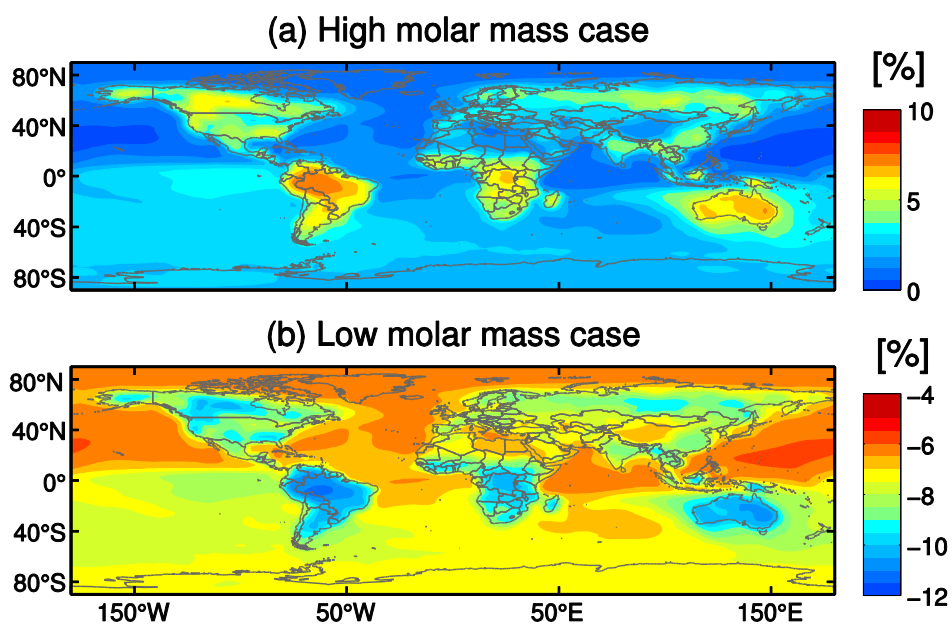
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48 **Supplementary Figure 7. Global maps of estimated bulk diffusivities. Bulk**
 49 **diffusivities ($\text{cm}^2 \text{s}^{-1}$) of (a) water and (b) organic molecules in SOA particles at the**
 50 **surface, 850 hPa, and 500 hPa.**



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52 **Supplementary Figure 8. Water diffusivity in α-pinene SOA.** Comparison of water
 53 diffusion coefficients in α-pinene SOA (O:C ratio = 0.54) obtained from water uptake
 54 experiments in an electrodynamic balance⁴ (green shaded areas) with values obtained
 55 with the semi-empirical estimation method presented in Berkemeier et al. (2014)⁵
 56 (orange shaded areas) at 260 K and 298 K. Water diffusivities were calculated with
 57 O:C = 0.5, $T_{g,SOA} = 278.5 \pm 7$ K, $\kappa_{org} = 0.12 \pm 0.02$, $k_{GT} = 1.5 \pm 1$, $\rho_{org} = 1.4$ g cm⁻³.

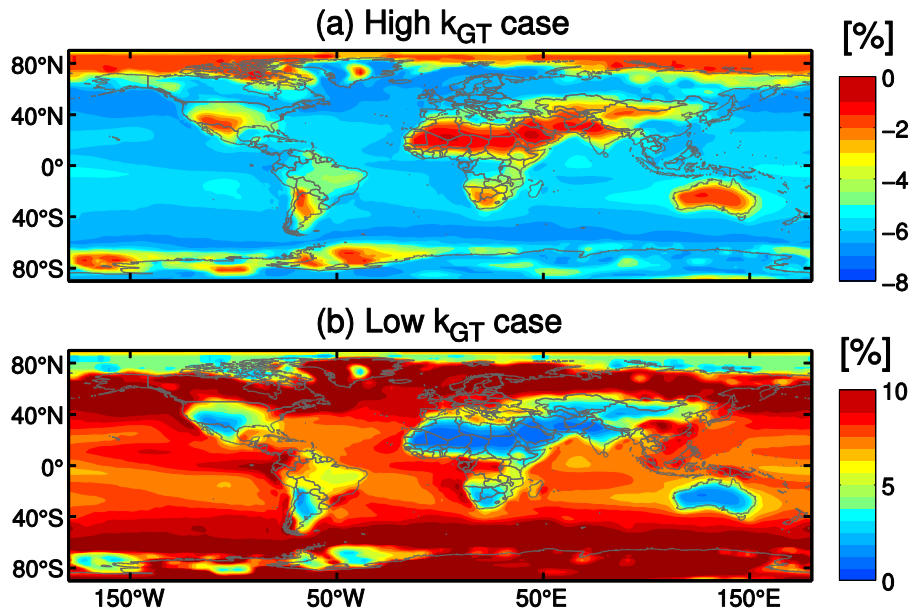


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59 **Supplementary Figure 9. Sensitivity of molar mass on T_g .** Modeled annual mean
 60 differences of T_g for SOA (without water) between a base case (Table 1) and cases
 61 with (a) high and (b) low molar mass values assigned in volatility bins
 62 ([Supplementary Table 2](#)), respectively.

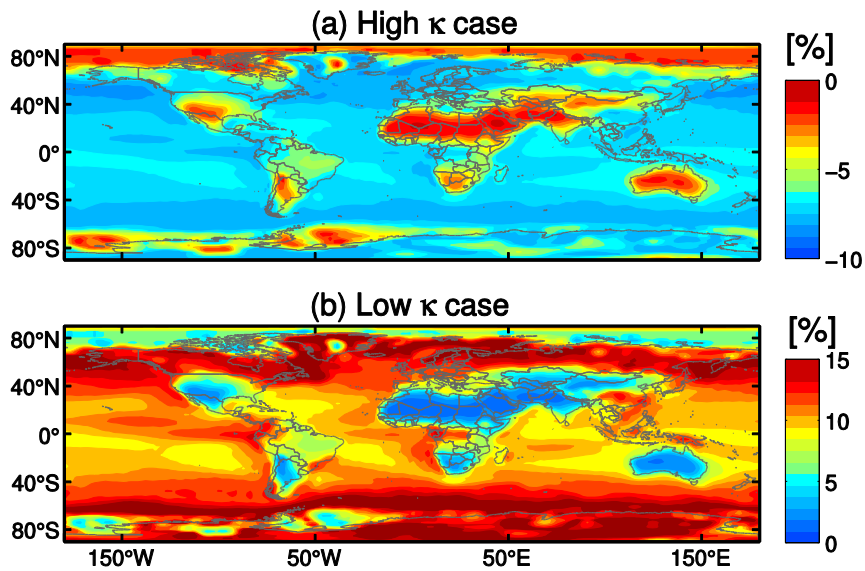
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66 **Supplementary Figure 10. Sensitivity of k_{GT} on T_g .** Modeled annual mean
 67 differences of T_g for SOA between a base case with the Gordon-Taylor constant of
 68 k_{GT} equaling to 2.5 and cases with (a) a higher k_{GT} of 3.5 and (b) a lower k_{GT} of 1.5,
 69 respectively.



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71 **Supplementary Figure 11. Sensitivity of κ on T_g .** Modeled annual mean differences
 72 of T_g for SOA between a base case with the hygroscopicity parameter of κ equaling to
 73 0.1 and cases with (a) a higher κ of 0.15 and (b) a lower κ of 0.05, respectively.

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75 **Supplementary references.**

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77 compounds: dependency on molecular properties and implications for secondary organic
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