- **1** Supplemental Materials
- 2

3 Supplemental Methods

4

5 *Plant material*

Experiments were performed on three 8-year-old individuals of *M. glyptostroboides*, grown
under glasshouse conditions for 10 weeks following a dormant overwintering period outside.
Growth conditions were 16 h days, with supplemental light from sodium vapour lambs over
the morning and evening providing a minimum of 300 µmol m⁻² s⁻¹ PAR at the leaf surface.
Day/night temperatures were 23°C/15°C respectively. Plants were grown in 20 L pots of
8:2:1 medium of composted pine bark, course river sand and peat moss and watered daily.
Once a week, plants were supplemented with liquid fertiliser (Aquasol; Hortico Ltd).

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14 Determination of the g_s vs. Ψ_l relationship

15 In total, six branches were excised from the three individuals and allowed to slowly desiccate on a laboratory bench. At intervals, initially every 20-30 min but at longer intervals as 16 branches dried out, g_s of a short shoot was measured using an infrared gas analyser (LI-6400; 17 LI-COR Biosciences). Chamber conditions were set at a light intensity of 1000 µmol m⁻² s⁻¹ 18 PAR (above light saturation for g_s in *M. glyptostroboides*), chamber temperature of 22°C and 19 20 D of 1.5 kPa, the same as external conditions. Short shoots outside the chamber were illuminated with a customised fibre optic light shower. At the same time as g_s was measured, 21 22 leaf water potential of an excised neighbouring short shoot was measured using a Scholander 23 pressure chamber. The maximum duration of desiccation did not exceed 4.5 h to avoid both 24 excessive loss of hydraulic conductivity and synthesis of ABA, the latter in M. 25 glyptostroboides typically synthesised after 6 h of desiccation (McAdam and Brodribb, 2014). 26

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28 Determination of the g_s vs. [ABA] relationship

29 The data used here for the g_s vs. [ABA] relationship in leaves of *M. glyptostroboides* is identical to that in McAdam and Brodribb (2014) and so only a brief recap of the method will 30 31 be included here. Stomatal sensitivity to ABA was determined by four independent methods: (1) feeding ABA into the transpiration stream of fully hydrated, excised shoots; (2) 32 33 rehydrated excised shoots previously allowed to slowly bench dry up to 24 h to stimulate ABA synthesis; (3) rehydrated excised shoots of plants undergoing drought stress; (4) in vitro 34 35 response of stomatal aperture to ABA in solution and g_s calculated using the formula of Parlange and Waggoner (1970). In all methods bar the last one, stomatal conductance was 36 determined by gas exchange using an infrared gas analyser. For full details of all the methods 37 and ABA sampling, extraction, purification and quantification see McAdam and Brodribb 38 (2014). 39

40

41 Shoot excision in air followed by rehydration by recutting underwater

42 Dynamic traces of g_s to short term changes in plant water status caused by excision in air to disrupt the hydraulic supply, followed by recutting underwater to reconnect hydraulic supply 43 were identical to those in McAdam and Brodribb (2014), but will be described again here as 44 the method is important for interpreting the model. Three branches were excised from the 45 plants and after removing the periderm around the cut end of the shoots to avoid xylary 46 47 blockages by resin, the branches were recut under resin-filtered deionised water. Leaves from a short shoot approximately halfway along the branch were enclosed in the chamber of an 48 infrared gas analyser, with chamber conditions at a light intensity of 1500 μ mol m⁻² s⁻¹ PAR, 49 D of approximately 1.2 kPa and chamber temperature of 22°C. Gas exchange was 50 automatically logged at intervals of 1 min. Leaves outside the chamber were illuminated with 51 a customised fibre optic light shower, providing a minimum of 300 μ mol m⁻² s⁻¹ PAR at the 52 leaf surface. Once a steady-state was reached (defined as less than 3% change in g_s over 8 53 min), the cut end of the branch was removed from the water and excess water around the cut 54 55 dried with paper towel to remove hydraulic supply to the branch. The branch was allowed to dehydrate and stomata close to approximately 50% of the initial g_s , at which point the branch 56 57 was rehydrated by recutting the branch underwater to reconnect hydraulic supply. Samples for ABA quantification were taken at the initial steady-state, at the minimum g_s and at the 58 final steady-state following rehydration on neighbouring short shoots. 59

61 *Rehydration following drought*

An individual plant was droughted by withholding water and branches sampled at 6, 10, 14 62 and 21 days post cessation of watering. Leaves from a short shoot were enclosed in an 63 infrared gas analyser chamber with chamber conditions at a light intensity of 1000 µmol m⁻² 64 65 s⁻¹ PAR, D of approximately 1.2 kPa and chamber temperature of 22°C. Chamber conditions 66 and gas exchange were automatically logged at intervals of 1 min. Prior to rehydration, a tissue sample from a neighbouring short shoot was taken for ABA quantification. The branch 67 was rehydrated by recutting the branch underwater in resin-filtered deionised water to 68 69 reconnect hydraulic supply instantaneously. ABA extraction, purification and quantification were as described in McAdam and Brodribb (2014). 70

71

72 *Model fitting and data analysis*

Values for hydraulic parameters K and C used in the model were the mean values obtained by 73 74 Martins et al. (2016), except for the rehydration kinetic after 21 days drought, where the leaf water potential was sufficiently low for significant hydraulic loss to have occurred (Table 1). 75 76 In this case, K was calculated from the vulnerability curve in McAdam and Brodribb (2014). 77 Values for stomatal conductance sensitivity to turgor pressure χ were estimated from the linear region of g_s vs. Ψ_l relationships from excised leaves (slope of the relationship should be 78 χ ; Supplemental Fig. S1; Table 1). These values were then used to fit eqn 8 to g_s vs. [ABA] 79 data (Fig. 1), allowing M, d and [ABA]₀ to be fitted to minimise sum of squares. All 80 81 parameters were used unaltered to calculate dynamic solutions for the cases of excision-82 rehydration (eqn 12a during dehydration, eqn 12b following rehydration) and rehydration 83 following drought using the ABA concentrations obtained from experiment (eqn 13). In modelling the response to rehydration following drought, reconnection of hydraulic supply 84 was taken to occur at time t = 0 s. 85

- Fitting the Hill equation variant of ABA dependence (eqn S19) to g_s vs. [ABA] data used K,
- 87 *C* and χ as above and allowed *d*, *M*, *K*_{*A*}, *k*₃/*k*₁ and *n* to be unconstrained while minimising the
- sum of squares. Fitting of the Tardieu and Davies model to g_s vs. [ABA] data required more
- constraints. As the Tardieu and Davies model (eqn S20) requires both [ABA] and Ψ_l , an
- 90 effective Ψ_l was reconstructed from the g_s vs. [ABA] data using the known values of g_s , D
- 91 and P_{atm} and the mean value for K using eqn S26. Initially g_{min} , α , β and δ were allowed to be
- 92 fitted to minimise sum of squares; however, the best fit produced $\delta > 0$, resulting in wrong-

way dynamics to changes in Ψ_l . Fitted g_{min} also tended to be higher than g_s observed in some of the drier rehydration kinetics. Instead, g_{min} was set at 0.005 mol m⁻² s⁻¹ and the Tardieu and Davies model fitted to the steady state g_s vs. [ABA] for different values of δ . Simulations with these parameters using eqns S20, S24, S25 and S26 were then compared with the observed excision-rehydration kinetics until a good fit was obtained. Once a good parameter set was obtained, this set was used to simulate rehydration kinetics using eqns S20, S24, S25 and S27.

The observed rehydration kinetics were compared with exponential dynamics by fitting 100 exponential curves of the form $g_s = A + B\exp(-t/\tau)$, fitting by minimising the sum of squares. 101 Fitted steady state g_s was obtained as the parameter A, while the fitted halftime was 102 calculated from τ as $t_{1/2} = \tau ln2$. Bounds were placed on the expected range of halftimes by 103 selecting the minimum and maximum values for C/K observed within branches and using the 104 corresponding C in those branches, while γ and D were kept constant. Values for C and K in 105 these cases were: 797.2 mmol m⁻² MPa⁻¹ and 3.61 mmol m⁻² s⁻¹ MPa⁻¹ respectively for the 106 minimum case; 1525 mmol m⁻² MPa⁻¹ and 4.51 mmol m⁻² s⁻¹ MPa⁻¹ respectively for the 107 maximum case. Note that these do not correspond to the extremum of C and K. Halftime 108 bounds were then calculated using eqn 9. Steady state g_s obtained by fitting the exponential 109 curves was compared with the modelled steady state g_s using average plant parameters by 110 performing a two-tailed paired t-test on the residuals to see whether the mean was 111 112 significantly different from 0. The expected range in steady state g_s was modelled using the observed range in K and eqn 8. Maximum observed K was 4.51 mmol m⁻² s⁻¹ MPa⁻¹, while 113 minimum observed was 2.47 mmol m⁻² s⁻¹ MPa⁻¹. 114

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117 Supplemental Model Development

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119 The ABA hydraulic model

121A full derivation of the ABA hydraulic model is provided here. As mentioned in the text, the122hydraulic core of the model at the leaf level was based on inward and outward fluxes of123water, following Jones (1982):124 $\frac{dW}{dt} = J - E$ eqn 1

where *W* is the water content of the leaf per leaf area (mol m⁻²), *t* is time (s), *J* is the flux of liquid water entering the leaf (mol m⁻² s⁻¹) and *E* is the flux of water vapour lost from the leaf by transpiration (mol m⁻² s⁻¹).

eqn S1

128 Using an Ohm's Law approximation, the flux of water into the leaf can be expressed as

 $J = K(\Psi_{\rm s} - \Psi_{\rm l})$

129

130 where *K* is the hydraulic conductivity of the leaf (mol m⁻² s⁻¹ Pa⁻¹),
$$\Psi_s$$
 is the source water
131 potential and Ψ_l is the leaf water potential (both in units of Pa).

Assuming well-mixed conditions, i.e. a negligible boundary layer resistance, the flux of waterout of the leaf can be expressed as

134
$$E = \frac{g_s D}{P_{atm}}$$
 eqn S2

where g_s is stomatal conductance (mol m⁻² s⁻¹), *D* is the vapour pressure difference (Pa) and P_{atm} is atmospheric pressure (Pa).

By definition of capacitance as the change in leaf water content per change in leaf water
potential and the chain rule, the rate of change of water content can be expressed in terms of
water potential as

140
$$\frac{dW}{dt} = \frac{dW}{d\Psi_l} \frac{d\Psi_l}{dt} = C \frac{d\Psi_l}{dt}$$
 eqn S3

141 Combining eqns 1, S1, S2 and S3 and diving through by *C*, eqn 2 is obtained:

142
$$\frac{d\Psi_l}{dt} = \frac{K}{C} (\Psi_s - \Psi_l) - \frac{g_s D}{CP_{atm}}$$
eqn 2

143 A more in depth expression for g_s is now required. In general, stomatal pore area and by

extension stomatal conductance is a function of both guard cell and epidermal turgor

145 pressures (Franks et al., 1998; Franks and Farquhar, 2007), with guard cell turgor (P_g) acting

- to open stomata while epidermal turgor pressure acting to reduce pore aperture. In
- 147 angiosperms the control of stomatal aperture can be dominated by the epidermis and is

148 known as mechanical advantage, occasionally leading to transient wrong way responses of stomata to changes in plant water status (Franks et al., 1998; Buckley et al., 2003; Buckley, 149 150 2005). No mechanical advantage or wrong-way response has been observed in ferns (Franks and Farquhar, 2007; Brodribb and McAdam, 2011), while conifers appear to also exhibit no 151 152 wrong-way response (McAdam and Brodribb, 2012; McAdam and Brodribb, 2014; Martins 153 et al., 2016). It has been suggested the reduced influence of epidermal turgor on stomatal 154 aperture in ferns, lycophytes and gymnosperms is due to greater lignification of the dorsal walls of guard cells in these lineages compared with angiosperms (McAdam and Brodribb, 155 2014). Ignoring the influence of the epidermis, stomatal conductance was assumed to be a 156 linear function of Pg (Cowan, 1972; Dewar, 1995; 2002; Buckley et al., 2003; Buckley, 2005) 157

158
$$g_s = \chi (P_g - P_0) \qquad \text{eqn 3}$$

159 where P_0 is the guard cell turgor pressure where stomata fully close and χ is the constant of 160 proportionality.

161 Expressing P_g as

162
$$P_g = \Psi_g + \pi_g$$
 eqn S4

163 where Ψ_g and π_g are guard cell water potential and osmotic pressure respectively, eqn 3 can 164 be expressed in terms of water potential as

165
$$g_s = \chi (\Psi_g + \pi_g - P_0)$$
eqn S5

A treatment of guard cell water relations is now required. Although it has been suggested the 166 167 guard cell hydraulic connection with the rest of the plant occurs via the vapour phase (Peak and Mott, 2011), a liquid phase hydraulic connection was favoured in the model. Meidner 168 (1975) suggested a major proportion of total evaporative loss of water into the atmosphere 169 occurred through the guard cells, a process known as peristomatal transpiration. This 170 assumption has been used in the models of Dewar (1995; 2002), while Buckley et al. (2003) 171 allowed for the division of evaporation between the epidermis and guard cells. Provided some 172 173 transpiration occurs directly from guard cells, a water potential gradient will occur between 174 the guard cells and the rest of the leaf. However, guard cells possess thick cuticles on the 175 exterior surface and in the throat of the pore, and it would appear unfavourable for the plant to be losing most of its water through peristomatal transpiration. It was therefore assumed 176 177 that most evaporation occurred within the mesophyll. Moreover, it was assumed there was

negligible hydraulic resistance between the rest of the leaf and the guard cells. Both

- assumptions are consistent with a previous iterative hydraulic model that successfully
- 180 predicted dynamics to changes in water status in ferns and conifers (Brodribb and McAdam,
- 181 2011; McAdam and Brodribb, 2014; Martins et al., 2016). The assumptions of a negligible
- resistance between the guard cell and the rest of the leaf and negligible transpiration
- 183 occurring directly from the guard cells leads to

184
$$\Psi_g = \Psi_l$$
 eqn S6

In the light, the guard cell turgor pressure is higher than the turgor pressure of the epidermis 185 186 and mesophyll cells due to the active accumulation of solutes in the guard cells, such as potassium and malate (Kollist et al., 2014). In the model, the osmotic pressure in the guard 187 188 cells was assumed to be composed of two components: the first consisted of the background 189 osmotic pressure of the leaf and was assumed unaffected by active processes (π_l); the second consisted of a component that could be actively regulated by metabolic processes in the guard 190 191 cell, such as light-induced build up of osmolytes or the ABA-induced efflux of solutes (π_a). It 192 was also assumed changes in volume of the guard cell are small so that the osmotic pressure 193 of the guard cell do not change appreciably with changes in volume. This last assumption is probably violated in angiosperms as guard cell volume can greatly change between closed 194 195 and fully open states (Raschke, 1975), although volume change is expected to be smaller in 196 conifers due to larger guard cell size.

197 Equation S5 then becomes

198 $g_s = \chi(\Psi_l + \pi_a + \pi_l - P_0)$

199 and letting $d = \pi_l - P_0$ gives eqn 4.

200 The active metabolic control of stomatal conductance occurs through π_a , while hydropassive 201 control occurs through Ψ_l . In general, a description of inward and outward fluxes of 202 osmolytes in terms of ion channel behaviour is difficult (Hills et al., 2012), while the role of 203 osmolyte synthesis in the guard cells is still unclear (Lawson, 2009). Instead, simple 204 expressions were used to describe the inward flux or accumulation of osmolytes in the guard cells in the light, and the outward flux of osmolytes triggered by ABA. Although both inward 205 and outward fluxes of solutes are often driven by changes in guard cell membrane potential 206 (Hills et al., 2012), here it was assumed both processes occurred independently. However, it 207 will be shown that this assumption can be identical mathematically to a case where the light 208

eqn S7

driven influx of solutes is reduced by the presence of ABA, as expected for depolarisation ofthe membrane.

- 211 The flux of solutes into the guard cell was assumed to be proportional to the difference
- between the current osmotic pressure and a target osmotic pressure (M) obtainable in the
- absence of ABA, set by environmental conditions such as light, temperature and carbon
- dioxide concentration (Kirschbaum et al., 1988; Haefner et al., 1997):

215 *inward flux* =
$$k_1(M - \pi_a)$$
 eqn S8

216 where k_1 is the rate constant for the inward flux.

The flux of solutes out of the guard cell was assumed to be dependent on the level of ABA within the leaf ([ABA]) and π_a by simple mass action. This assumption is analogous to the activation of outward channels being proportional to [ABA] and the resulting loss of solutes occurring by simple collision kinetics. Although in practice efflux kinetics would be much more complex, the simplest case was used as a first approximation. This gave

222 *outward flux* =
$$k_2 \pi_a$$
[ABA] eqn S9

- 223 where k_2 is the rate constant for the outward flux.
- 224 The equation for osmotic pressure of the guard cell becomes

225
$$\frac{d\pi_a}{dt} = k_1(M - \pi_a) - k_2\pi_a[ABA]$$
 eqn S10

As mentioned earlier, it could be argued that ABA reduces the osmotic pressure of the guard
cell through either reducing the maximum osmotic pressure obtainable, equivalent to
depolarising the membrane, or through activating efflux channels, or a combination of both.
However, upon rearranging eqn S10,

230
$$\frac{d\pi_a}{dt} = (k_1 + k_2 [ABA]) \left(\frac{k_1}{k_1 + k_2 [ABA]} M - \pi_a\right)$$
 eqn S11

This is equivalent to ABA reducing the maximum possible osmotic pressure for the guard cell, while also increasing the rate at which equilibrium is achieved. Thus the two interpretations are effectively equivalent. In testing the model, inward and outward fluxes and thus the net flux balance was considered constant for the short-term dynamics in plant water status. At steady state the flux balance gives

237
$$k_1(M - \pi_a) = k_2 \pi_a$$
[ABA] eqn 5

238 which upon rearrangement gives

239
$$\pi_a = \frac{M}{1 + \frac{[ABA]}{[ABA]_0}}$$
eqn 6

where $[ABA]_0$ is the ratio of inward and outward rate constants and is the [ABA] where π_a is half the maximum value.

The goal now is to express eqn 2 in terms of g_s . Expressing eqn 4 in terms of Ψ_l and

substituting into eqn 2, noting that π_a is constant over the short-term dynamics in plant water status gives

245
$$\frac{dg_s}{dt} = \frac{\chi K}{C} \left(\frac{M}{1 + \frac{[ABA]}{[ABA]_0}} + \Psi_s + d \right) - \frac{1}{C} \left(K + \frac{\chi D}{P_{atm}} \right) g_s$$
eqn 7

Steady-state g_s is obtained by letting $\frac{dg_s}{dt} = 0$. For a plant with hydraulic supply connected, this gives

248
$$g_s^* = \frac{\chi}{1 + \frac{\chi D}{KP_{atm}}} \left[\frac{M}{1 + \frac{[ABA]}{[ABA]_0}} + \Psi_s + d \right]$$
eqn 8

For the case of where hydraulic supply is not connected, such as when a leaf is excised in air, K = 0 and by inspection of eqn 7 gives a steady state of $g_s^* = 0$.

For constant plant parameters, eqn 7 is a linear ordinary differential equation for g_s and gives exponentials as analytical solutions

253
$$g_s = g_1 e^{-\frac{t}{C} \left(K + \frac{\chi D}{P_{atm}}\right)} + g_s^*$$
eqn S12

where g_1 is a constant with units of conductance, dependent on initial conditions.

From eqn S12, it can be seen that the halftime for a dynamic is dependent on two component

- 256 halftimes characteristic of hydraulic and evaporative processes respectively. The total
- 257 halftime

258
$$t_{1/2 total} = \frac{C \ln 2}{\left(K + \frac{\chi D}{P_{atm}}\right)}$$
eqn 9

is the halftime for a step change in hydraulic supply or demand where the plant is both hydraulically connected and transpiring. If the leaf is not transpiring (D = 0), such as in the rehydration method for determining *K* and *C* (Blackman and Brodribb, 2011), the halftime becomes

263
$$t_{1/2 hydraulic} = \frac{c}{\kappa} ln 2$$
 eqn 10

and is denoted here as the hydraulic halftime as it depends only on the hydraulic properties ofthe leaf.

266 If the leaf is transpiring but has no hydraulic supply (K = 0), the halftime becomes

267
$$t_{1/2 \ evaporative} = \frac{CP_{atm}}{\chi D} ln 2$$
 eqn 11

and is denoted here as the evaporative halftime as it depends on the evaporative properties ofthe leaf. The total halftime can thus be represented as

270
$$t_{1/2 \ total} = \frac{1}{\left(\frac{1}{t_{1/2} \ hydraulic} + \frac{1}{t_{1/2} \ evaporative}\right)}}$$
 eqn S13

From eqn S13 it can be seen that the total halftime will be less than the two componenthalftimes.

The model was tested under two scenarios. In the first, a fully hydrated branch was excised in air, cutting off hydraulic supply. As the leaves dried out, stomata closed, before the branch was recut underwater to reconnect hydraulic supply. If the initial excision occurred at t = 0, then K = 0 up until recutting underwater at $t = t_r$. For $t < t_r$, eqn S12 becomes

277
$$g_s = g_s^* e^{-\left(\frac{\chi D}{CP_{atm}}\right)t}$$
eqn 12a

For $t > t_r$, the full form of eqn S12 applies. Matching the boundary condition at $t = t_r$ gives

279
$$g_1 = g_s^* \left(e^{\frac{\kappa}{c}t_r} - e^{\left(\frac{\kappa}{c} + \frac{\chi D}{CP_{atm}}\right)t_r} \right) = -g_s^* e^{\frac{t_r}{c}\left(\kappa + \frac{\chi D}{P_{atm}}\right)} \left(1 - e^{-\left(\frac{\chi D}{CP_{atm}}\right)t_r}\right) \quad \text{eqn S14}$$

280 Substituting into eqn S12 gives, for $t > t_r$

281
$$g_s(t) = g_s^* \left\{ 1 - \left[1 - e^{-\left(\frac{\chi D}{CP_{atm}}\right)t_r} \right] e^{-\left(\frac{K}{C} + \frac{\chi D}{CP_{atm}}\right)(t-t_r)} \right\}$$
eqn 12b

In the second scenario, branches of droughted plants were rehydrated by excision underwater. If the initial g_s was g_{s0} and cutting underwater occurred at t = 0, then

284
$$g_1 = g_{s0} - g_s^*$$
 eqn S15

285 Substituting into eqn S12 then gives

286
$$g_s(t) = (g_{s0} - g_s^*)e^{-\left(\frac{K}{C} + \frac{\chi D}{CP_{atm}}\right)t} + g_s^*$$
 eqn 13

287

288 Hill equation kinetics variant for ABA sensitivity

289

To compare the form used above for g_s sensitivity to ABA against other similar alternative forms, the steady state model was modified to use a general Hill equation form for ABAdriven efflux of solutes from the guard cell. In this scenario, the activation of ion channels was seen to follow Hill equation kinetics, while the loss of solutes still followed simple mass action. Influx of solutes into the guard cell was kept unchanged from the original model. Using Hill equation kinetics, the outward flux of solutes can be represented as

296 *outward flux* =
$$\frac{k_3 \pi_a}{\left(\frac{K_A}{[ABA]}\right)^n + 1}$$
 eqn S16

where k_3 is the rate constant for efflux, K_A is the level of ABA where half the channels are active and *n* is the Hill coefficient. Combining with eqn S8, the equation for balance of inward and outward fluxes of solutes becomes

300
$$k_1(M - \pi_a) = \frac{k_3 \pi_a}{\left(\frac{K_A}{[ABA]}\right)^n + 1}$$
 eqn S17

301 Rearranging, eqn S17 becomes

302
$$\pi_a = \frac{M}{1 + \frac{k_3/k_1}{\left(\frac{K_A}{[ABA]}\right)^n + 1}}$$
 eqn S18

303 Equation S18 was then substituted in place of the ABA dependence of eqn 8 to yield

304
$$g_s^* = \frac{\chi}{1 + \frac{\chi D}{KP_{atm}}} \left[\frac{M}{1 + \frac{k_3/k_1}{\left(\frac{K_A}{[ABA]}\right)^n + 1}} + \Psi_s + d \right]$$
 eqn S19

306 *Tardieu and Davies model*

307

For a comparison of the model with the currently used model for ABA dependence of g_s , the model of Tardieu and Davies (1993) was modified to fit with the experimental test conditions. The model of Tardieu and Davies uses an empirical form to relate [ABA] and Ψ_l to g_s :

312
$$g_s = g_{min} + \alpha exp(\beta[ABA]exp(\delta \Psi_l))$$
 eqn S20

313 where g_{min} is the minimum value for g_s , α is the maximum difference from g_{smin} in the 314 absence of ABA, β is the sensitivity to [ABA] and δ is the sensitivity to leaf water potential. 315 Whereas the original model of Tardieu and Davies took [ABA] to be the concentration of 316 ABA in the xylem sap, here it was taken to be the level of ABA in the leaf.

The original formulation of the Tardieu and Davies model (Tardieu and Davies, 1993) was aimed at steady-state or a quasi-steady state condition. The level of ABA was a flux balance of ABA coming into the leaf from the xylem sap, ABA lost through the transpiration stream and ABA catabolised:

321
$$[ABA] = \frac{J_{ABA}}{I+b}$$
 eqn S21

where J_{ABA} is the flux of ABA synthesised in the roots, J (= E at steady state) is the flux of water through the leaf and *b* is the flux catabolised.

The Tardieu and Davies model was modified to be applicable to the short-term changes in plant water relations used to test the hydraulic ABA model. The modified model still described the [ABA] in the leaf as a result of a flux balance of ABA transported into the leaf by the liquid flux and the loss of ABA through transpiration, but in the non-steady state the fluxes were not equal. For short-term changes in plant water relations, changes in plant water status were of too short a duration to significantly affect ABA biosynthesis or catabolism, thus the rate of catabolism was set to zero. Moreover, as biosynthesis was assumed to be negligible over a similar timeframe, the [ABA] in the xylem sap was assumed to be the sameas the bulk leaf tissue. The rate of change of [ABA] in the leaf was given by

333
$$\frac{d[ABA]}{dt} = \frac{18}{FWA} (J[ABA]_x - E[ABA])$$
 eqn S22

where $[ABA]_x$ is the level of ABA in the xylem sap and FWA is the fresh weight of leaves per unit area (g m⁻²). The factor 18 (g mol⁻¹) is to convert the molar water fluxes into mass fluxes.

Modelling plant water relations for the Tardieu and Davies model began at eqn 2. As g_s in the Tardieu and Davies model is no longer a linear function of Ψ_l , a numerical solution is required. Crudely discretising eqn 2 and letting $\Psi_s = 0$ as was the case for the model tests leads to

341
$$\frac{\Psi_{lt+1} - \Psi_{lt}}{\Delta t} = \frac{1}{c} \left(-K \Psi_{lt} - \frac{g_s([ABA], \Psi_{lt})D}{P_{atm}} \right)$$
eqn S23

where $\Psi_{l\,t+l}$ and $\Psi_{l\,t}$ are the next and current values of Ψ_l respectively, Δt is the time step between calculations and g_s has the Tardieu and Davies form (eqn S20) and is evaluated at Ψ_l t. Rearranging gives the iterative solution for Ψ_l as

345
$$\Psi_{l\,t+1} = \Psi_t + \frac{\Delta t}{c} \left(-K\Psi_{l\,t} - \frac{g_s([\text{ABA}], \Psi_{l\,t})D}{P_{atm}} \right) \qquad \text{eqn S24}$$

346 Discretising eqn S22 leads to

347
$$[ABA]_{t+1} = \frac{18\Delta t}{FWA} J[ABA]_x - \left(\frac{18\Delta t}{FWA}E - 1\right) [ABA]_t \qquad \text{eqn S25}$$

348 where J and E were calculated using eqn S1 and S2.

Initial steady-state Ψ_l for excision-rehydration kinetics was calculated by iteration using eqn S20 and

351
$$\Psi_l = \frac{-g_s D}{K P_{atm}}$$
 eqn S26

352 Initial Ψ_l for rehydration kinetics were calculated by inverting eqn S20:

353
$$\Psi_l = \frac{1}{\delta} ln \left(\frac{1}{\beta [ABA]} ln \left(\frac{g_s - g_{min}}{\alpha} \right) \right)$$
 eqn S27

At the start of both excision-rehydration and rehydration kinetics, $[ABA]_x$ and [ABA] were assumed to be in equilibrium.

- Combining eqn S24, S25, the relevant initial condition and evaluating g_s using eqn S20 at
- ach time step produced the modelled dynamics for the Tardieu and Davies model, noting
- that during dehydration kinetics hydraulic supply was disconnected, i.e. K = 0.
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412 Supplemental Figure S1. The linear range of dependence of stomatal conductance on leaf water
413 potential for *M. glyptostroboides*. *χ* was estimated as the slope of the linear line of best fit for stomatal
414 conductance vs. leaf water potential. Within this range of water potentials, the assumption that
415 stomatal conductance was a linear function of turgor pressure appeared valid.











422 Supplemental Figure S2. Exponential fits to observed kinetics of stomatal conductance recovery

423 upon recutting underwater, following drought to increase leaf ABA levels. Exponentials fitted the

424 observed kinetics well at most ABA levels (A, 175 ng g^{-1} , $R^2 = 0.99$; B, 604 ng g^{-1} , $R^2 = 0.99$; C, 1835

425 ng g^{-1} , $R^2 = 0.97$; A, 5951 ng g^{-1} , $R^2 = 0.81$).

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430 **Supplemental Figure S3.** Tardieu and Davies model (eqn S20) fitted to data from McAdam and 431 Brodribb (2014), showing the dependence of fully hydrated stomatal conductance on leaf ABA levels 432 ($R^2 = 0.80$). A stomatal sensitivity to leaf water potential ($\delta = -1.2$ MPa⁻¹) was selected to best fit 433 excision-rehydration kinetics. This fit was later used for modelling rehydration kinetics. Other 434 parameter values for the fit were: $g_{min} = 0.005$ mol m⁻² s⁻¹, $\alpha = 0.242$ mol m⁻² s⁻¹, $\beta = -0.00064$ g ng⁻¹.

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442 **Supplemental Figure S4.** Stomatal conductance recovery upon recutting underwater, following

443 drought to increase leaf ABA levels, with modelled recovery using the ABA hydraulic model (solid

444 grey line) and the Tardieu and Davies model (grey dashed line).

445



447 **Supplemental Figure S5.** Plotted dependence of stomatal conductance on leaf water potential for the 448 Tardieu and Davies model with $\delta = -1.2 \text{ MPa}^{-1}$, during the simulated excision-rehydration and 449 rehydration following drought kinetics at different ABA levels. Within a simulated dynamic using the 450 Tadieu and Davies model, [ABA] tended to change by a relatively small amount, while simulated Ψ_l

- 451 changed substantially. The Tardieu and Davies model tended to correspond well with the ABA
- 452 hydraulic model when the g_s vs. Ψ_l relationship was approximately linear with the same slope as used

- 453 in the ABA hydraulic model (i.e. 175 ng g^{-1} , 430 ng g^{-1} and 604 ng g^{-1} cases). Fitting the simulated g_s
- 454 vs. Ψ_l relationship from the Tardieu and Davies model with a linear line of best fit gave: 175 ng g⁻¹: χ
- 455 = 0.0755 mol m⁻² s⁻¹ MPa⁻¹, R² = 0.994; 430 ng g⁻¹: χ = 0.0949 mol m⁻² s⁻¹ MPa⁻¹, R² = 0.996; 604 ng
- 456 g^{-1} : $\chi = 0.105 \text{ mol m}^{-2} \text{ s}^{-1} \text{ MPa}^{-1}$, $R^2 = 0.999$; 1835 ng g^{-1} : $\chi = 0.0503 \text{ mol m}^{-2} \text{ s}^{-1} \text{ MPa}^{-1}$, $R^2 = 0.928$;
- 457 5951 ng g⁻¹: $\chi = 0.0144$ mol m⁻² s⁻¹ MPa⁻¹, R² = 0.995.