

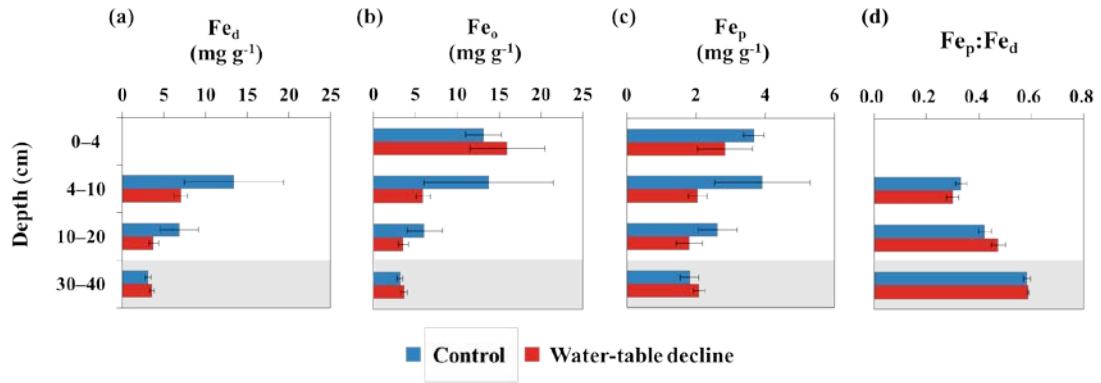
File name: Supplementary Information

Description: Supplementary Figures, Supplementary Tables and Supplementary References

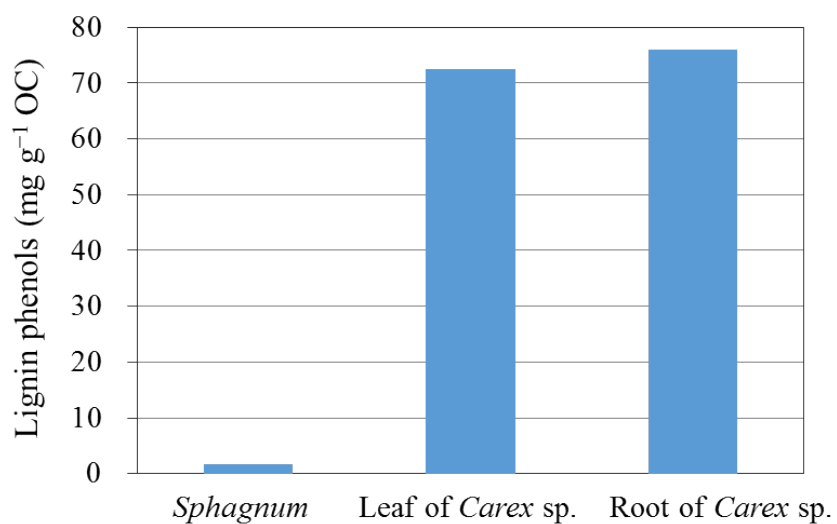
Supplementary Table 1: Summary of published results on extracellular enzyme activities in response to water-table decline (WTD) or drought.

Publication	Eco-type	Dominant vegetation	Soil type	Soil pH	SOC (%)	Fe(II) (mg g ⁻¹ soil)	Result
Freeman et al. ¹	Peatland	<i>Sphagnum</i> and <i>Juncus</i> species	Peat	4.68–5.02	na	na	Phenol oxidative activity increased with O ₂ (WTD or drought)
Freeman et al. ²	Riparian gully mire	<i>Sphagnum</i>	Peat	4.5	na	na	
Fenner and Freeman ³	Ombrotrophic peatland	<i>Sphagnum</i>	Peat	4–4.8	~50*	na	
Romanowicz et al. ⁴	Oligotrophic bog	<i>Sphagnum</i> , <i>Polytrichum</i> species, sedges and ericaceous shrubs	Peat	4	na	na	
Hall and Silver ⁵	Humid tropical forest	Trees (<i>Dacryodes excelsa</i> , <i>Prestoea montana</i> , <i>Cyrtilla racemiflora</i>)	Ultisols, Oxisols, Inceptisols	4.3–5.3	na	0.05–7.39	Phenol oxidative or hydrolytic enzyme activity decreased with O ₂ (WTD or drought)
Hall et al. ⁶				4.3–5.3	2.8–16.7	0.11–4.11	
Liu et al. ⁷	Rice paddy field	Rice (<i>Oryza sativa</i> L.)	Silt-clay Ultisol	5.9	1.1	0.2–1.7	
Toberman et al. ⁸	Boreal mire peatland	Sedge, <i>Sphagnum</i> and <i>Juncus</i> species	Peat	4.0–4.7	na	na	
Toberman et al. ⁹	Upland heathland	Shrub (<i>Calluna vulgaris</i> L.)	Peaty Podsol	3.9	~45*	na	
Toberman et al. ¹⁰	Boreal mire peat	Sedge, trees, moss	Peat	4.0–5.5	na	na	
This study	Alpine wetland	Sedge (<i>Carex</i>)	Organic layers of silty clay Mat-Cryic Cambisol	6–8	7.5–18.5	0.68–7.20	

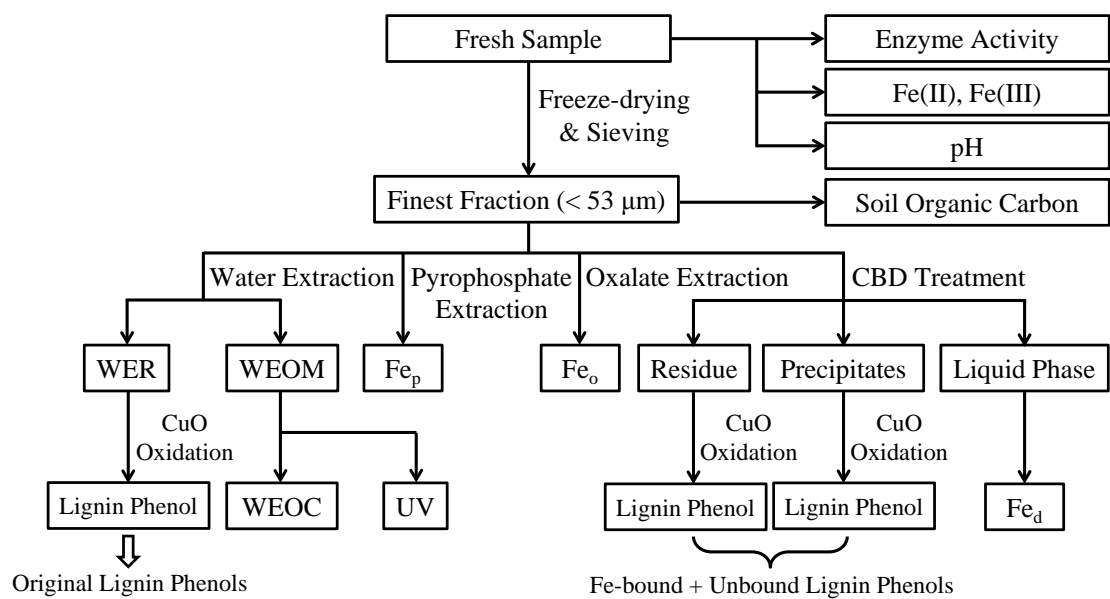
* Estimated as 50% of organic matter content. SOC: soil organic carbon; Fe(II): ferrous iron.



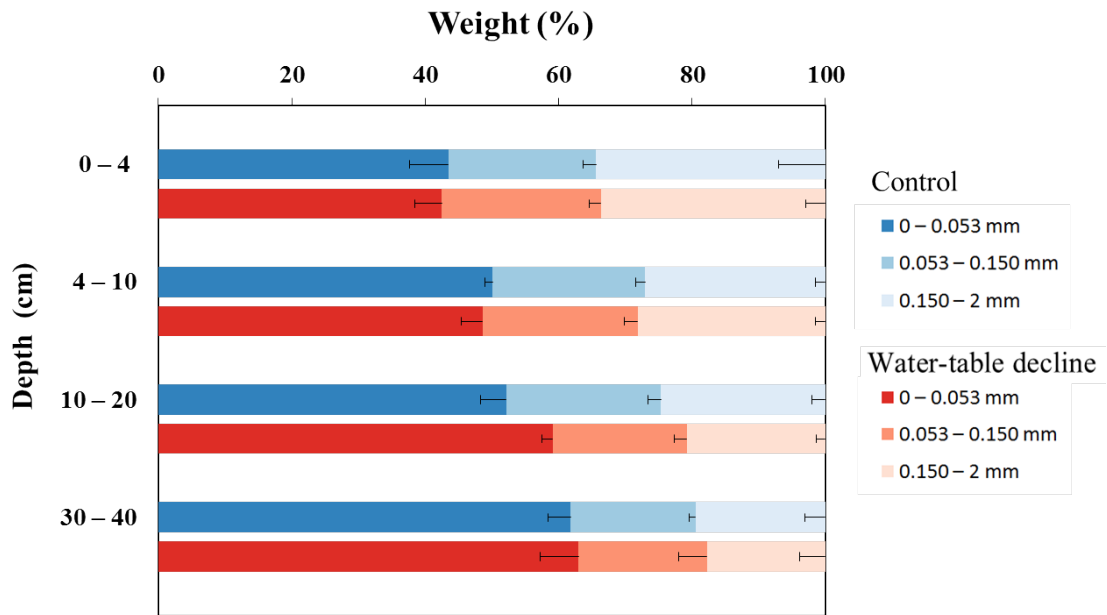
Supplementary Figure 1: Contents of different forms of extractable iron (Fe) in the wetland soils (< 53 μm). (a) Dithionite-extractable Fe (Fe_d); (b) oxalate-extractable Fe (Fe_o); (c) pyrophosphate-extractable Fe (Fe_p); (d) ratio of Fe_p:Fe_d indicating degree of Fe complexation with organic matter. Error bars represent standard error of mean (s.e.m.; n = 4). Upper- and lower-case letters denote difference among soil depths in the control and water-table decline treatment, respectively (*p* < 0.05). Fe_d was not measured for 0–4 cm due to the limited sample size.



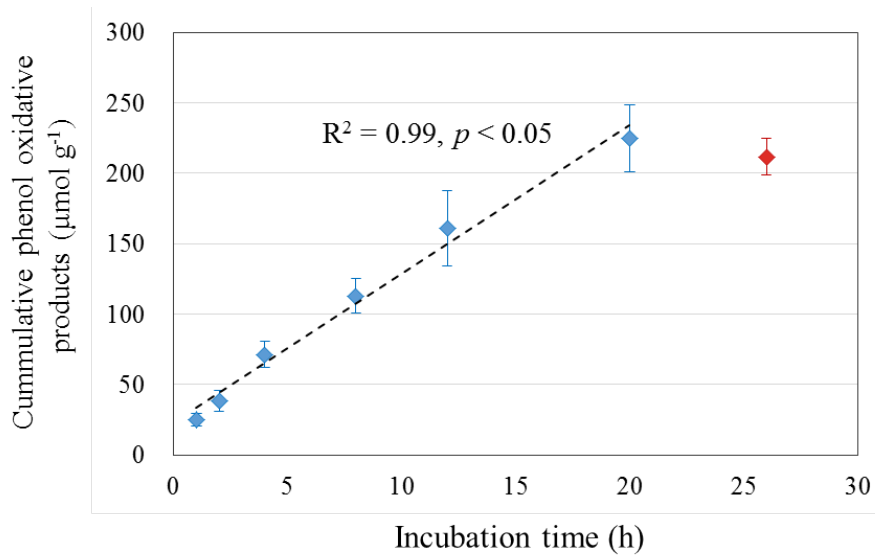
Supplementary Figure 2: Organic carbon (OC)-normalized lignin phenol concentrations in *Sphagnum* and sedge (*Carex* sp.) tissues collected from wetlands in southern China and the Haibei Station.



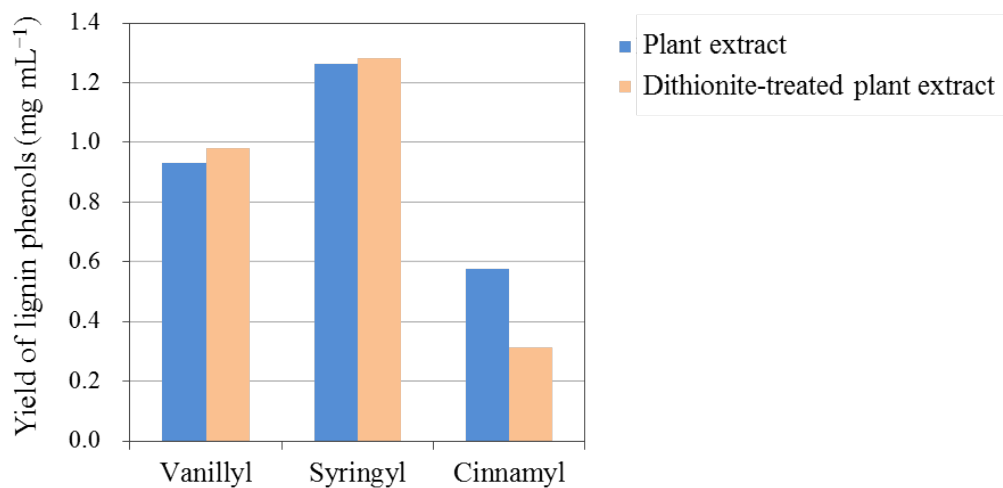
Supplementary Figure 3: Scheme of sample treatments and analyses. WER: water-extracted residue; WEOM: water-extractable organic matter; WEOC: water-extractable organic carbon; CBD: citrate-bicarbonate-dithionite method; Fe(II): ferrous iron; Fe(III): ferric iron; Fe_p: pyrophosphate-extractable iron; Fe_o: oxalate-extractable iron; Fe_d: dithionite-extractable iron.



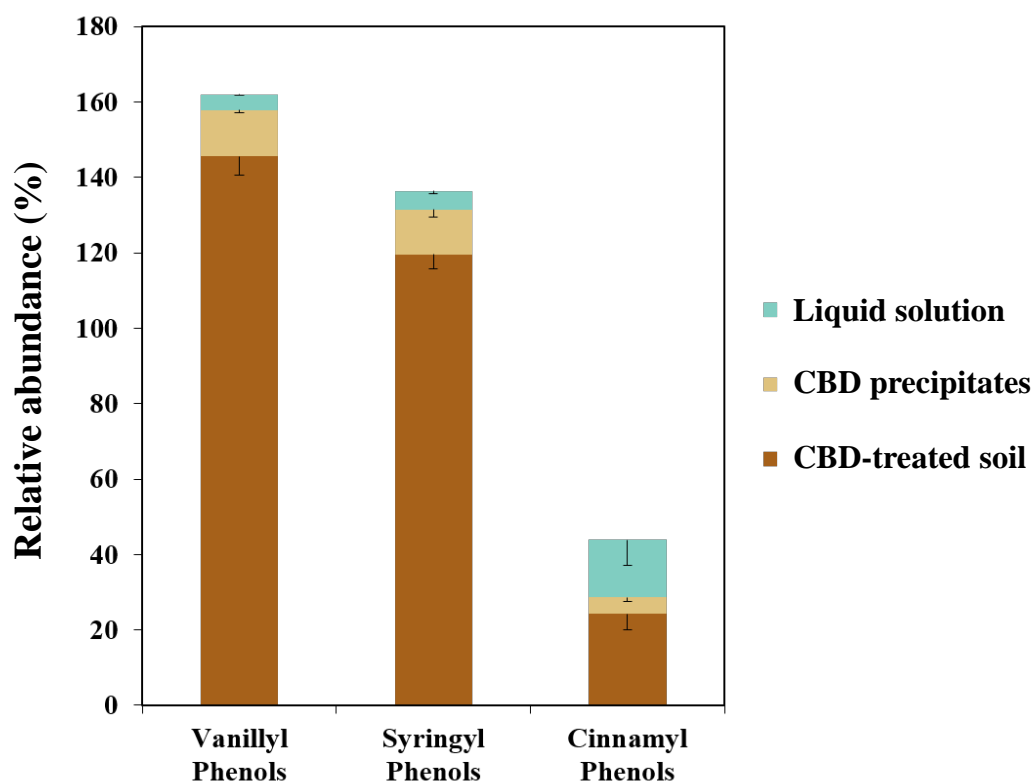
Supplementary Figure 4: Weight percentage of each grain size fraction in the control and water-table decline treated soils. Error bars represent standard error of mean (s.e.m.; n = 4).



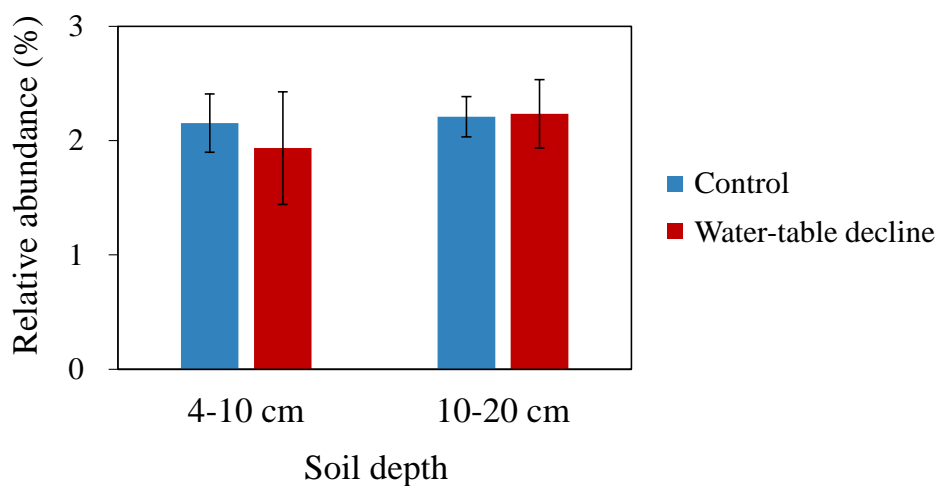
Supplementary Figure 5: Changes of cumulative phenol oxidative products in the surface wetland soil with increasing incubation time for the assay of phenol oxidative activities (mean \pm s.e.m.; n = 3). Note that phenol oxidative products increased linearly with time within 20 h.



Supplementary Figure 6: Lignin phenol yields of untreated and dithionite-treated water extracts from oak leaves in the preliminary experiment.



Supplementary Figure 7: Relative abundance of lignin phenols in different phases after the citrate-bicarbonate-dithionite (CBD) treatment compared with that in the original control soil at 30–40 cm. Error bars represent standard error of mean (s.e.m.; $n = 4$). Lignin in the liquid solution was concentrated using C_{18} solid phase extraction cartridges as described in the preliminary experiment. Note that cinnamyl phenols were relatively more abundant in the solution than vanillyl and syringyl phenols, indicating a high solubility and potential to loss during the CBD treatment.



Supplementary Figure 8: Relative abundance of lignin phenols in the liquid solution after the citrate-bicarbonate-dithionite (CBD) treatment compared with that in the original soil at 4–10 and 10–20 cm in both control and water-table decline treatments. Errors represent standard error of mean (s.e.m.; n = 4).

Supplementary References

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