Supporting Information

Site-Selective Aliphatic C–H Chlorination Using N-Chloroamides Enables a Synthesis of Chlorolissoclimide

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General Methods

All reactions were performed in oven-dried ($120\,^{\circ}$ C) or flame-dried glassware under an atmosphere of dry argon unless otherwise noted. Reaction solvents including dichloromethane (CH₂Cl₂, Fisher, HPLC Grade), hexanes (Fisher, HPLC Grade), diethyl ether (Et₂O, Fisher, BHT stabilized, HPLC Grade), benzene (C₆H₆, Fisher, HPLC Grade), tetrahydrofuran (THF, Fisher, HPLC Grade), and toluene (PhCH₃, Fisher, HPLC Grade) were dried by percolation through a column packed with neutral alumina and a column packed with Q5 reactant, a supported copper catalyst for scavenging oxygen, under a positive pressure of argon.

Solvents for workup and chromatography were: acetone (Fisher, ACS grade), hexanes (Fisher or EMD, ACS Grade), EtOAc (EtOAc, Fisher, ACS Grade), dichloromethane (CH_2Cl_2 , Fisher, ACS Grade), and methanol (MeOH, Fisher, ACS Grade). Column chromatography was performed using EMD Millipore 60 Å (0.040–0.063 mm) mesh silica gel (SiO_2). Analytical thin-layer chromatography was performed on Merck silica gel 60 F254 TLC plates. Visualization was accomplished with UV (254 or 210 nm), and p-anisaldehyde, or ceric ammonium molybdate and heat as developing agents.

Chloroform-*d* (CDCl₃, D 99.8%, DLM-7) and methanol-*d*₄ (CD₃OD, D 99.8%, DLM-24) and dichloromethane-*d*₂ (CD₂Cl₂, D 99.9%, DLM-23) were purchased from Cambridge Isotope Laboratories. Citric acid (ACS grade, anhydrous, Fisher), K₂CO₃ (anhydrous, 99%, Alfa Aesar), NaHCO₃ (ACS grade, Fisher), NaOH (ACS grade, Macron or Fisher), Na₂S₂O₃ (ACS grade, Fisher), *p*-anisaldehyde (99%, Acros Organics), trimethylaluminum (Al(CH₃)₃, Sure Pack™reagent grade, Sigma Aldrich), benzoyl peroxide

(BPO, reagent grade, Sigma Aldrich), *tert*-butylhydroperoxide (TBHP, 5.5 M in decanes, Sigma Aldrich), di-*iso*-butylaluminum hydride (*i*Bu₂AlH, Sure Pack™reagent grade, Sigma Aldrich), ammonia (2.0 M in methanol, Acros Organics), selenium (IV) oxide (SeO₂, 99.8%, Acros Organics), dimthyl sulfoxide (DMSO, extra dry with molecular sieves, water <50 ppm, Acros Organics) were used without further purification.

The following reagents were distilled from the indicated drying agents under argon prior to use: triethylamine (Et₃N, EMD, CaH₂), pyridine (Alfa Aesar, CaH₂), 2,4,6-collidine (Alfa Aesar, CaH₂), and *N*,*N*-diisopropylethylamine (*i*-Pr₂NEt, Alfa Aesar, CaH₂). Oxalyl chloride ((COCl)₂, Sigma Aldrich) trimethylsilyl triflate (TMSOTf, Alfa Aesar), dibutylboron triflate (*n*Bu₂BOTf, Acros Organics), and thionyl chloride (SOCl₂, Sigma Aldrich), were fractionally distilled prior to use.

Infrared (IR) spectra were obtained using a Jasco 260 Plus Fourier transform infrared spectrometer. GC Spectra were obtained using a Shimadzu GC-2010 gas chromatograph with a Shimadzu AOC-20s Autosampler, and Shimadzu SHRXI-5MS GC column. The results of the kinetic isotope study were analyzed using an Agilent Gas Chromatograph- Mass Spectrometer with a 6850 series GC system and a 5973 Network Mass Selective Detector. Proton and carbon magnetic resonance spectra (¹H NMR and ¹³C NMR) were recorded at 298K on a Bruker GN500 (500 MHz, ¹H; 125 MHz, ¹³C), Bruker CRYO500 (500 MHz, ¹H; 125 MHz, ¹³C), Bruker AVANCE600 (600 MHz, ¹H), Bruker model DRX 400, DRX 500, or a Bruker AVANCE III 600 CryoProbe (¹H NMR at 400, 500 or 600 MHz and ¹³C NMR at 100, 126 or 151 MHz) spectrometer with solvent resonance as the internal standard (¹H NMR: CHCl₃ at 7.28* ppm, C₆D₆ at 7.16 ppm, CDH₂SOCD₃ at 2.49 ppm, CDHCl₂ at 5.32 ppm; ¹³C NMR: CDCl₃ at 77.0 ppm). ¹H NMR data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, ddd = doublet of doublets, td = triplet of doublets, tdd = triplet of doublet of doublets, qd = quartet of doublets, m = multiplet, br. s. = broad singlet), coupling constants (Hz), and integration. Mass spectra for the methods development were obtained using a Thermo LTqFT mass spectrometer with electrospray introduction and external calibration. Mass spectrometry data for the synthesis were obtained from the University of California, Irvine Mass Spectrometry Facility. Highresolution mass spectra (HRMS) were recorded on a Waters LCT Premier spectrometer using ESI-TOF (electrospray ionization-time of flight) and data are reported in the form of (m/z). Melting points (mp) were recorded on a Laboratory Devices Mel-Temp II melting point apparatus and are uncorrected. Thin layer chromatography (TLC) was performed on SiliaPlate 250µm thick silica gel plates provided by Silicycle. Visualization was accomplished with short wave UV light (254 nm), aqueous basic potassium permanganate solution, or ethanolic acidic p-anisaldehyde solution followed by heating. chromatography was performed using SiliaFlash P60 silica gel (40-63 µm) purchased from Silicycle. *CHCl₃ at 7.26 ppm for compounds S8-S16 and 39-46.

Compound Preparation

Amide Synthesis

N-(**tert-butyl**)-3,5-bis(**trifluoromethyl**)benzamide was synthesized by reacting tert-butylamine with the corresponding acid chloride in THF. To a 0 °C solution of 10.0g carboxylic acid (38.75 mmol) in DCM (150 mL) and DMF (100 uL) was added 6.5 mL oxalyl chloride (71.50 mmol) dropwise under an argon atmosphere. The solution was stirred at 0 °C for 15 min. then warmed to room temperature for 1.5 hours.

The resultant solution was evaporated almost to dryness under reduced pressure to remove the DCM. The reaction mixture was dissolved in dry THF and cooled to 0 °C. Then 10.1 mL of tertbutylamine (71.5 mmol) was added dropwise. The solution was allowed to warm to room temperature and stirred overnight. The reaction was diluted with Et₂O, washed with a 2.5M NaOH solution, 3x 1M HCl solution, 1x Brine, dried with magnesium sulfate and concentrated under reduced pressure. The product was purified via single solvent recrystallization from benzene to give the product (11.1 g, 35.5 mmol, 92% yield) as a white crystalline solid. Physical and spectral data were in accordance with literature data.¹

N-Chlorination with Sodium Hypochlorite

General Procedure: To a 250 mL foil wrapped flask under N₂, amide (3.50 g, 11.2 mmol) was added and dissolved in a mixture of ethyl acetate (20 mL) and *tert*-butanol (0.85 mL, 11.2 mmol). 50 mL of a sodium hypochlorite solution was then added (~1.5M in H₂O Sigma-Aldrich), followed by acetic acid (6.4 mL, 112 mmol). The reaction was stirred at room temperature for 3-4 hours. When the reaction was complete as judged by TLC analysis (3 hours usually sufficient) the reaction was diluted with Et₂O, washed three times with saturated sodium bicarbonate solution, once with water, and once with brine. The organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The crude material usually contains traces of chlorinated ethyl acetate. The crude product was purified via column chromatography (1-5% EtOAc/Hexane) to give the choloroamide product (3.8g, 10.9 mmol, 98% yield) as a clear oil. *General storage: N-Chloro reagents were stored in foil-wrapped vials in the freezer when not in use. The reagents can be weighed out on the bench top without risk of decomposition*.

Analytical data for **Chloroamide**: ¹**H NMR** ¹H NMR (600MHz ,CHLOROFORM-d) δ = 8.09 (s, 2 H), 7.97 (s, 1 H), 1.62 (s, 9 H) ppm; ¹³**C NMR** (CHLOROFORM-*d*, 125 MHz) 172.2, 138.3, 131.9, 131.7, 131.5, 131.3, 128.6, 125.6, 124.4, 124.3, 124.2, 124.1, 123.8, 122.0, 120.2, 64.7, 27.9 ppm; **IR** (thin film, cm⁻¹) 2985.3, 2939.9, 1459.9, 1385.6, 1369.2, 1182.2, 1139.7, 906.4, 702.9, 681.7; **HRMS** (ESI) Calcd. for $[C_{13}H_{12}NClF_6O_3+Na]^+ = 370.04$, Found = 370.04.

S3

Substrate Synthesis

Note: Hydrocarbon substrates were obtained commercially and used without further purification unless otherwise noted.

Benzenesulfonyl Pentanol (21). To a 0° solution of pentanol (2.46 mL, 22.7 mmol) and trimethylamine (4.16 mL, 29.49 mmol) in DCM (25 mL) benzenesulfonyl chloride was added dropwise. The reaction was allowed to warm to room temperature and stir overnight. The reaction was diluted with Et₂O (100 mL) and the organic layer was washed twice with 100 mL of 1N HCL solution, and once with brine. The organic layer was then dried with magnesium sulfate and concentrated under reduced pressure. The crude material was purified using column chromatography (5-10% EtOAc/Hexanes) to isolate benzenesulfonyl pentanol (4.77 g, 20.9 mmol, 92% yield) as a clear liquid.

Analytical data for **Benzensulfonyl Pentanol:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 7.92 (d, J = 8.3 Hz, 2 H), 7.67 (s, 1 H), 7.61 - 7.53 (m, 2 H), 4.05 (t, J = 6.6 Hz, 2 H), 1.69 - 1.59 (m, 2 H), 1.33 - 1.20 (m, 4 H), 0.85 (t, J = 7.2 Hz, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 136.2, 133.7, 129.2, 127.8, 70.9, 28.5, 27.4, 22.0, 13.8 ppm; **IR** (thin film, cm⁻¹) 2959.2, 2934.1, 2870.5, 1449.2, 1360.3,1186.9, 1097.3, 958.4, 914.1, 826.4, 755.9, 590.1; **HRMS** (ESI) Calcd. for [C₁₁H₁₆SO₃+Na]⁺ = 251.07, Found = 251.07.

N-Pentyl Phthalimide (17): was synthesized via an alkylation reaction of iodopentane with phthalimide potassium salt: To a room temperature solution of iodopentane (2.5 mL, 19.14 mmol) in DMF (200 mL), the phthalimide salt (7.09 g, 38.3 mmol) was added. The reaction was headed to 80 °C overnight. The reaction was cooled to room temperature and diluted with Et₂O and water. The organic layer was separated and washed 8 times with 100 mL of water to remove the residual phthalimide salt. The organic layer was then dried with magnesium sulfate and concentrated under reduced pressure. The crude material was purified using column chromatography (25% EtOAc/Hexanes) to isolate N-Phth pentane (3.45 g, 15.5 mmol, 82% yield) as a clear, yellowish liquid.

Analytical data for *N*-Pentyl Phthalimide: ¹H NMR (600MHz ,CHLOROFORM-d) δ = 7.88 - 7.75 (m, 1 H), 7.73 - 7.59 (m, 2 H), 3.74 - 3.56 (m, 2 H), 1.75 - 1.57 (m, 2 H), 1.31 (br. s., 4 H), 0.94 - 0.79 (m, 3 H); ¹³C NMR (125MHz ,CHLOROFORM-d) δ = 168.4, 133.8, 132.1, 123.1, 37.9, 28.9, 28.2, 13.9 ppm; **IR** (thin film, cm⁻¹) 3468.4, 3061.4, 2934.2, 1773.2, 1712.5, 1614.13, 1465.6, 1397.2, 1367.3, 1186.0, 1058.7, 980.6880.3, 793.5, 719.3, 620.0, 530.3; **HRMS** (ESI) Calcd. for [C₁₃H₁₅NO₂+H]⁺ = 218.11, Found = 218.04.

2-Phthalimidyl Hexane (**S1**). Phthalimide substrate **S1** was prepared via Mitsunobu reaction from the corresponding alcohol. To a 250 mL flame-dried round bottom flask 2-hexanol (1.2 mL, 9.8 mmol) was added and dissolved in THF. Phthalimide (2.16 g, 14.7 mmol) and triphenylphosphine (3.85 g, 14.7 mmol) were then added and the reaction was cooled to 0°C. Diisopropyl azodicarboxylate (2.97 mL, 14.7 mmol) was then added dropwise over 5 min. The reaction was allowed to warm to room temperature and stir overnight. When the reaction was complete by TLC the reaction was concentrated under reduced pressure and purified directly via column chromatography (20% EtOAc/Hexane) to give the product (1.7 g, 7.2 mmol, 73% yield) as clear oil. Physical and spectral data were in accordance with literature data.²

S2. Sulfonate substrate **S2** was prepared via a condensation reaction from the corresponding alcohol in an identical manner as described above for **benzenesulfonyl pentanol.** The crude material was purified using column chromatography (5-10% EtOAc/Hexanes) to isolate **S2** (1.5 g, 6.1 mmol, 94% yield) as a clear yellowish liquid. Physical and spectral data were in accordance with literature data.³

S3. Phthalimide substrate **S3** was prepared via Mitsunobu reaction from the corresponding alcohol in an identical manner as described above for substrate **S3**. The crude reaction mixture was purified directly via column chromatography (20% EtOAc/Hexane) to give the product (1.6 g, 6.9 mmol, 70% yield) as clear oil.

Analytical data for **S3:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 7.92 - 7.82 (m, 2 H), 7.77 - 7.70 (m, 2 H), 3.64 - 3.45 (m, 2 H), 2.07 - 1.95 (m, 1 H), 1.51 - 1.40 (m, 1 H), 1.40 - 1.26 (m, 2 H), 1.23 - 1.09 (m, 1 H), 0.95 - 0.86 (m, 6 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 168.7, 133.8, 132.1, 123.2, 44.4, 36.6, 32.4, 19.9, 17.4, 14.3 ppm; **IR** (thin film, cm⁻¹) 2959.2, 2930.3, 1773.3, 1711.5, 1465.6, 1435.7, 1398.1, 1186.97, 1061.6, 723.2; **HRMS** (ESI) Calcd. for [C₁₄H₁₇NO₂+Na]⁺ = 254.12, Found = 254.12.

S4 was synthesized via the following two step protocol. To a flame-dried 25 mL flask 1.0 g (7.6 mmol) of DL-norleucene was added, followed by phthalic anhydride (1.7 g, 11.4 mmol). The two solids were then dissolved in DMF (15 mL) and heated to 155° C for 12 hours. The reaction was cooled to room temperature and poured into 50 mL of 1N HCl, and cooled to 0° C for 20 minutes (if precipitation does not occur extract with Et₂O to isolate). The phthalimide product precipitates as a colorless solid which was collected (2. g 7.6 mmol, *quant*) via vacuum filtration and dried under vacuum. The phthalimide was used in the next step without further purification. Phthalimide protected norleucene (2.0 g, 7.6 mmol) was added to a flask fitted with a reflux condenser and dissolved in DCM (25 mL). Dimethylformamide-dimethylacetal (3.1 mL, 22.9 mmol) was then added dropwise and the reaction was refluxed for 4h. When TLC analysis revealed the reaction was complete, the reaction was cooled to room temperature, diluted with Et₂O, washed twice with saturated sodium bicarbonate solution, four times with saturated ammonium chloride solution, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified via column chromatography (30-40% EtOAc/Hexanes gradient) to give **S4**

(1.98 g, 7.2 mmol, 96% yield) as a yellowish oil. Physical and spectral data were in accordance with literature data.⁴

S5. Phthalimide substrate **S5** was prepared via Mitsunobu reaction from the corresponding alcohol in an identical manner as described above for substrate **S1**. The crude reaction mixture was purified directly via column chromatography (25% EtOAc/Hexane) to give the product (2.1 g, 7.1 mmol, 74% yield) as clear oil.

Analytical data for **S5:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 7.84 - 7.80 (m, 2 H), 7.72 - 7.68 (m, 2 H), 7.58 (d, J = 7.7 Hz, 2 H), 7.35 (t, J = 7.7 Hz, 2 H), 7.28 (t, J = 1.0 Hz, 1 H), 5.47 - 5.21 (m, 1 H), 2.72 - 2.46 (m, 1 H), 2.36 - 2.25 (m, 1 H), 1.47 - 1.29 (m, 4 H), 0.91 (t, J = 7.2 Hz, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 168.4, 139.9, 133.9, 131.9, 128.5, 128.2, 127.8, 123.2, 55.1, 30.7, 29.2, 22.3, 14.0 ppm; **IR** (thin film, cm⁻¹) 3062.4, 3031,5, 2958.3, 2929.3, 2867.6, 1773.2, 1711.5, 1465.6, 1387.5, 1068.9, 880.3, 721.2; **HRMS** (ESI) Calcd. for [C₁₉H₁₉NO₂+Na]⁺ = 316.13, Found = 316.13.

S6. Phthalimide substrate **S6** was prepared via Mitsunobu reaction from the corresponding alcohol in an identical manner as described above for substrate **S1**. The crude reaction mixture was purified directly via column chromatography (20% EtOAc/Hexane) to give the product (1.4 g, 5.7 mmol, 67% yield) as clear oil.

Analytical data for **S6:** ¹**H NMR** (600MHz ,CHLOROFORM-d) d = 7.87 - 7.78 (m, 2 H), 7.75 - 7.63 (m, 2 H), 5.71 (dddd, J = 5.9, 8.4, 10.1, 17.1 Hz, 1 H), 5.03 (d, J = 18.3 Hz, 0 H), 4.94 (d, J = 10.3 Hz, 1 H), 4.29 (tt, J = 5.2, 10.2 Hz, 1 H), 2.81 (td, J = 9.2, 14.2 Hz, 1 H), 2.55 - 2.46 (m, 1 H), 1.75 (tdd, J = 5.4, 10.3, 13.8 Hz, 1 H), 1.39 - 1.17 (m, 4 H), 0.86 (t, J = 7.3 Hz, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) $\delta = 168.1$, 135.9, 133.9, 131.9 123.2, 117.3, 54.2, 31.7, 28.6, 22.2, 13.9 ppm; **IR**

(thin film, cm⁻¹) 2961..2, 2931.2, 2874.4, 1465.6, 1450.2, 1360.5, 1186.1, 1097.3, 966.2, 820.6, 756.9, 689.4, 590.1; **HRMS** (ESI) Calcd. for $[C_{15}H_{17}NO_2+Na]^+=266.12$, Found = 266.10.

S7 was synthesized via the following two step protocol starting from 1-adamantanecarboxaldehyde (accessed via a protocol reported by Leadbeater⁵). To a 0° solution of aldehyde (3.92 g, 16.2 mmol) in THF (80 mL) allylmagnesiumchloride solution (21 mL, 1.0M in THF) was added dropwise. The reaction was allowed to warm to room temperature and stir overnight. The reaction was quenched with saturated ammonium chloride solution and extracted three times with ether. The combined organic layers were washed with brine, dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified via column chromatography (10% EtOAc/Hexane) to give the product (2.9 g, 14.1 mmol, 87% yield) as colorless solid. The adamantanol (500 mg, 2.4 mmol) was then dissolved in pyridine (8 mL) and cooled to 0° C. Mesyl chloride was then added dropwise and the reaction was allowed to warm to warm to room temperature and stirred for 6 hours. When the reaction was complete judging by TLC analysis the crude reaction mixture was diluted with Et₂O, and washed three times with 1N HCl and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified via column chromatography (10% EtOAc/Hexane) to give the product (655 mg, 2.3 mmol, 96% yield).

Analytical data for **S7:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 5.95 - 5.85 (m, 1 H), 5.23 - 5.12 (m, 2 H), 4.40 (dd, J = 2.9, 9.2 Hz, 1 H), 3.03 (s, 3 H), 2.54 (tdd, J = 1.4, 5.3, 15.0 Hz, 1 H), 2.44 - 2.35 (m, 1 H), 2.04 (br. s., 3 H), 1.77 - 1.70 (m, 3 H), 1.70 - 1.63 (m, 6 H), 1.59 (dd, J = 1.5, 12.1 Hz, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 135.4, 118.2, 91.6, 39.3, 38.2, 36.8, 36.7, 33.9, 28.1 ppm; **IR** (thin film, cm⁻¹)2906.2, 2851.2, 1642.1, 1451.2, 1333.5, 1171.5, 905.4, 791.4, 732.8, 539.0; **HRMS** (ESI) Calcd. for [C₁₅H₂₄SO₃+Na]⁺ = 307.13, Found = 307.14.

N-Phth Rimantadine (30). The hydrochloride salt (1.44 g, 6.7 mmol) was dissolved in DMF (15 mL) and cooled to 0 °C. Neat sodium hydride (161.3 mg, 6.7 mmol) was then added with gas evolution and the reaction was stirred at 0 °C for 10 min. The reaction was then allowed to warm to room temperature over 30 min. Phthalic Anhydride (1.48 g, 10.4 mmol) was then added and the reaction was heated to reflux overnight. The reaction was then diluted with Et₂O and 1N HCl. The organic layer was separated and washed twice more with 1N HCl, and once with brine. The organic layer was then dried with magnesium sulfate and concentrated under reduced pressure. The crude material was purified using column chromatography (10% EtOAc/Hexanes) to isolate *N*-Phth Rimantadine (1.53 g, 4.95 mmol, 74% yield) as a white solid.

Analytical data for *N*-Phth Rimantadine (30): ¹H NMR (600MHz ,CHLOROFORM-d) δ = 7.93 - 7.79 (m, 2 H), 7.78 - 7.65 (m, 2 H), 4.05 (q, J = 7.3 Hz, 1 H), 1.99 (br. s., 3 H), 1.73 - 1.57 (m, 12 H), 1.51 (d, J = 7.3 Hz, 3 H); ¹³C NMR (125 MHz ,CHLOROFORM-d) δ = 169.5, 169.4, 133.9, 133.7, 132.3, 131.6, 123.2, 122.9, 56.8, 39.5, 37.8, 36.8, 28.4, 11.8 ppm, IR (thin film, cm⁻¹) 2095.3, 2849.3, 2255.3, 1774.2, 1451.2, 1373, 1170.6, 1095.3, 912.2, 726.1, 531.3; HRMS (ESI) Calcd. for [C₂₀H₂₃NO₂+H]⁺ = 332.16, Found = 332.16.

Synthesis of Chloride Standards

Note: Chloride standards were obtained commercially and used without further purification unless otherwise noted. Commercially obtained chloride include: chlorocyclohexane, 1-chloroadamantane, 2-chloroadamantane, 1-chlorohexane, 2-chlorohexane, 3-chlorohexane.

Methyl 5-chlorohexanoate was synthesized via the following two-step procedure: The first step of the procedure is adapted from the method reported by Wolf et. al. The lactone (2.0 g, 17.5 mmol) was added to a flask containing a solution of 33% HBr in AcOH (5 mL) and fitted with a reflux condenser. The reaction was heated to 75 °C for 4 hours then cooled to room temperature, at which point methanol (8.0 mL) was added and the mixture was stirred at room temperature overnight. The reaction was then partially concentrated under reduced pressure, taken up in EtOAc, washed three times with a saturated aqueous solution of sodium bicarbonate, brine, and the organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (5% EtOAc/Hexane) to isolate methyl 5-bromohexanoate 72% yield (2.05 g) as a clear liquid. Methyl 5-bromohexanoate (2.0 mL, 9.6 mmol) was then dissolved in DMF (15 mL) and lithium chloride was added in one portion (1.22 g, 28.9 mmol). The reaction was heated to 90° C and stirred overnight. When the reaction was complete as judged my CG-MS analysis, the reaction was diluted with Et₂O, washed three times with 1N HCl, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The product required no purification. Physical and spectral data were in accordance with literature data.

N-4-chloropentyl phthalimide was synthesized via a four-step protocol. *N*-4-bromopentyl phthalimide was previously synthesized in our laboratory, for synthetic details regarding its synthesis see our previous publication. *N*-4-bromopentyl phthalimide (500 mg, 1.6 mmol) was then dissolved in DMF (5 mL) and lithium chloride was added in one portion (214 mg, 5.1 mmol). The reaction was heated to 90° C and stirred overnight. When the reaction was complete as judged my CG-MS analysis, the reaction was diluted with Et₂O, washed three times with 1N HCl, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (15% EtOAc/Hexane) to isolate *N*-4-chloropentyl phthalimide (354 mg, 1.4 mmol, 88% yield) as a clear oil. Physical and spectral data were in accordance with literature data.⁸

1,4-Dichloropentane was prepared via substitution of the corresponding dibromide. 1,4-dibromopentane (1.0 g, 4.3 mmol) was dissolved in DMF (15 mL) and lithium chloride was added in one portion (1.1 g, 26.2 mmol). The reaction was heated to 90° C and stirred overnight. When the reaction was complete as judged my CG-MS analysis, the reaction was diluted with Et₂O, washed three times with 1N HCl, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (2% EtOAc/Hexane) to isolate 1,4-Dichloropentane (467 mg, 3.3 mmol, 77% yield) as a yellowish liquid.

Analytical data for **1,4-Dichloropentane:** ¹**H NMR** ¹H NMR (600MHz ,CHLOROFORM-d) $\delta = 4.12$ - 4.01 (m, 1 H), 3.65 - 3.55 (m, 2 H), 2.11 - 2.00 (m, 1 H), 2.00 - 1.87 (m, 2 H), 1.87 - 1.77 (m, 1 H), 1.60 - 1.52 (m, 3 H); ¹³**C NMR** (125 MHz ,CHLOROFORM-d) $\delta = 57.9$, 44.5, 37.4, 29.6, 25.5 ppm; **IR** (thin film, cm⁻¹)2956.0, 2929.3, 1725.1, 1445.4, 1379.8, 1287.3, 773.3, 729.9, 652.8, 611.3; **LRMS** Calcd. for $[C_5H_{10}Cl_2]^+ = 140.02$, Found = 140.09

2-Chloro-5-Cyanopentane was prepared via a two-step protocol. 2-Bromo-5-Cyanopentane was previously synthesized in our laboratory, for synthetic details regarding its synthesis see our previous publication. Paramonet 2-Bromo-5-Cyanopentane (1.8 mL, 10.2 mmol) was then dissolved in DMF (15 mL) and lithium chloride was added in one portion (1.3 g, 30.6 mmol). The reaction was heated to 90° C and stirred overnight. When the reaction was complete as judged my CG-MS analysis, the reaction was diluted with Et₂O, washed three times with 1N HCl, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (10% EtOAc/Hexane) to isolate 2-Chloro-5-Cyanopentane (1.1 g, 8.6 mmol, 84% yield) as a yellowish liquid.

Analytical data for **2-Chloro-5-Cyanopentane:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 4.11 - 3.99 (m, 1 H), 2.47 - 2.37 (m, 2 H), 2.00 - 1.87 (m, 2 H), 1.86 - 1.75 (m, 2 H), 1.58 - 1.49 (m, 3 H); ¹³**C NMR** (125 MHz ,CHLOROFORM-d) δ = 119.3, 57.4, 38.8, 25.4, 22.6, 16.8 ppm; **IR** (thin film, cm⁻¹) 2972.2, 2932.3, 2874.4, 2246.6, 1454.1, 1429.9, 1379.8, 1255.4, 1123.3, 612.3; **HRMS** (ESI) Calcd. for $[C_6H_{10}CIN+Na]^+$ = 154.04, Found = 154.04.

4-Chloropentyl Acetate was synthesized via the following four-step protocol. The first step of the procedure is adapted from the method reported by Wolf et. al.⁶ The lactone (2.0 g, 17.5 mmol) was added to a flask containing a solution of 33% HBr in AcOH (5 mL) and fitted with a reflux condenser. The reaction was heated to 75 °C for 4 hours then cooled to room temperature, at which point methanol (8.0 mL) was added and the mixture was stirred at room temperature overnight. The reaction was then partially concentrated under reduced pressure, taken up in EtOAc, washed three times with a saturated aqueous solution of sodium bicarbonate, brine, and the organic layer was dried with magnesium sulfate and concentrated under reduced pressure. The crude product (3.6 g, 17.5, quant.) was used in the next step without purification. The crude bromoester (2.0 g, 14.2 mmol) was dissolved in DMF (15 mL) and lithium chloride (1.8 g, 42.5 mmol) was added in one portion. The reaction was heated to 90° C and stirred overnight. When the reaction was complete as judged my CG-MS analysis, the reaction was diluted with Et₂O, washed three times with 1N HCl, and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude chloroester (1.9 g, 13.1 mmol, 92% yield) was used in the next step without purification. The chloroester (2.0 g, 13.8 mmol) was added dropwise to a 0° solution of lithium aluminum hydride (675 mg, 20.7 mmol) in THF (30 mL). The reaction was stirred for 30 min at 0° then allowed to warm to room temperature and stirred for an additional 2 hours. The reaction was cooled back to 0° and quenched slowly by the sequential addition of water (2 mL), NaOH solution (4 mL, 2.5M), and water (10 mL). The reaction was filtered over celite, and concentrated under reduced pressure. The crude reaction mixture was diluted with Et₂O, washed with saturated sodium bicarbonate, and brine, dried with magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (15% EtOAc/Hexane) to isolate the chloro-alcohol (1.31g, 10.72 mmol, 78% yield) as a clear liquid. Lastly, the chloro-alcohol (500 mg, 4.1 mmol) was acetylated in in neat acetic anhydride (1 mL) using catalytic DMAP (24 mg, 0.2 mmol). The reaction was complete after stirring overnight at room temperature. The reaction was diluted with Et₂O, washed twice with saturated sodium bicarbonate, once with water, dried over magnesium sulfate and concentrated under reduced pressure. The product (660 mg, 4.0 mmol, 98% yield) did not require purification.

Analytical data for **4-Chloropentyl Acetate:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 4.15 - 4.01 (m, 3 H), 2.06 (s, 3 H), 1.94 - 1.70 (m, 4 H), 1.54 (d, J = 6.6 Hz, 3 H); ¹³**C NMR** (125 MHz, CHLOROFORM-d) δ = 171.1, 63.8, 58.2, 36.7, 25.9, 25.4, 20.9 ppm; **IR** (thin film, cm⁻¹) 2967.9, 1740.4, 1446.3, 1366.3, 1240.9, 1038.5, 608.4; **HRMS** (ESI) Calcd. for [C₇H₁₃ClO₂+Na]⁺ = 187.04, Found = 187.04.

4-Chlorpentyl Benzenesulfonate was synthesized from the same chloroalcohol used to synthesize 4-Chloropentyl Acetate above. The chloroalcohol (500 mg, 4.1 mmol) was dissolved in DCM (8 mL) and triethyl amine (0.74 mL, 5.3 mmol) was added. The reaction was cooled to 0° and benzene sulfonyl chloride (0.57 mL, 4.5 mmol) was added dropwise. The reaction was allowed to warm to room temperature and was stirred overnight. When the reaction was complete as judged via TLC analysis the reaction was diluted with Et₂O, washed twice with 1N HCl and once with brine. The organic layer was dried over magnesium sulfate and concentrated under reduced pressure. The crude product was purified using column chromatography (10% EtOAc/Hexane) to isolate 4-Chlorpentyl Benzenesulfonate (667 mg, 2.5 mmol, 62% yield).

Analytical data for **4-Chlorpentyl Benzenesulfonate:** ¹**H NMR** ¹H NMR (600MHz ,CHLOROFORM-d) $\delta = 8.00 - 7.86$ (m, 2 H), 7.74 - 7.64 (m, 8 H), 7.63 - 7.49 (m, 2 H), 4.16 - 4.04 (m, 2 H), 4.02 - 3.91 (m, 1 H), 1.97 - 1.86 (m, 1 H), 1.86 - 1.74 (m, 2 H), 1.74 - 1.63 (m, 1 H), 1.49 (d, J = 6.6 Hz, 3 H); ¹³**C NMR** (125 MHz ,CHLOROFORM-d) $\delta = 135.9$, 133.9, 129.3, 127.9, 70.1, 57.8, 35.9, 26.1, 25.4 ppm; **IR** (thin film, cm⁻¹) 3067.2, 2972.7, 1447.3, 139.6, 1186.0, 1097.3, 969.3, 917.5, 826.7, 751.3, 689.4, 589.1; **HRMS** (ESI) Calcd. for $[C_{11}H_{15}ClSO_3+Na]^+ = 385.03$, Found = 385.03.

1-Chloro-1-Methylcyclohexane was synthesized according to literature procedures and used without purification. Physical and spectral data were in accordance with literature data.¹⁰

1-Chloro-1,2-Dimethylcyclohexane was synthesized according to literature procedures referenced above and used without purification. Physical and spectral data were in accordance with literature data. Both NMR and GC analysis show both the *trans* and *cis* products in a 87:13 ratio respectively. The selectivity is consistent with literature precedent.¹¹

C-H Chlorination Procedures with N-Chloro Amides

Chlorination of Cycloalkanes (Table 1)

$$\begin{array}{c|c} & & & & \\ & & & & \\ \hline & & & & \\ \hline & \\ \hline & & \\ \hline & & \\ \hline & \\ \hline & & \\ \hline & & \\ \hline & & \\ \hline & \\ \hline & & \\ \hline & & \\ \hline & \\ \hline & & \\$$

A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (50.0 mg, 0.143 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μL). Cyclohexane (15.5 μL, 0.143 mmol), and cesium carbonate (46.2 mg, 0.143 mmol) were then added. *Note: cycloalkane was added as a stock solution in benzene to improve the reproducibility of the results.* The reaction was then sealed with teflon tape and taken out of the glovebox, and irradiated with two 23W compact fluorescent light bulbs for 90 minutes at 55° C. A white, semi-soluble precipitate forms as the reaction reaches completion. Upon competition, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 71.6% GC yield of cyclohexylchloride. A small amount of dichloride products (2% GC yield) are also formed in the reaction.

Reactions in Table 1 were performed in an identical fashion to the procedure described above.

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$$+ \bigcirc_{D_{12}} \xrightarrow{F_3C} \bigcirc_{CF_3} \qquad \bigcirc_{CF_3} \qquad \bigcirc_{D_{11}} \qquad \bigcirc_{D_{12}} \qquad \bigcirc_{CS_2CO_3, \text{ PhH, } hv, 55^\circ} \qquad \bigcirc_{D_{11}} \qquad \bigcirc_{$$

A flame-dried, 1 dram vial was charged with a stir bar and fitted with a PTFE lined screw cap. Chloroamide (20 mg, 0.058 mmol) was added to the vial in the absence of ambient light, and the reaction was taken into a glovebox, and dissolved in 200 uL of dry, freeze-pump-thawed benzene. Cyclohexane (27.6 uL, 0.255 mmol) and Cyclohexane- d_{12} (27.5 uL, 0.255 mmol), and cesium carbonate (18.9 mg, 0.058 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 45 minutes at 55° C. The reaction was then diluted with DCM and analyzed using an Agilent Gas Chromatograph- Mass Spectrometer with a 6850 series GC system and a 5973 Network Mass Selective Detector to determine the ratio of non-deuterated to deuterated product (Ratio = $4.9 = K_{H/D}$).

A flame-dried, 1 dram vial was charged with a stir bar and fitted with a PTFE lined screw cap. Bromoamide (20 mg, 0.058 mmol) was added to the vial in the absence of ambient light, and the reaction was taken into a glovebox, and dissolved in 200 uL of dry, freeze-pump-thawed benzene. Cyclohexane (27.6 uL, 0.255 mmol) and Cyclohexane- d_{12} (27.5 uL, 0.255 mmol), and cesium carbonate (18.9 mg, 0.058 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 45 minutes at 55° C. The reaction was then diluted with DCM and analyzed using an Agilent Gas Chromatograph- Mass Spectrometer with a 6850 series GC system and a 5973 Network Mass Selective Detector to determine the ratio of non-deuterated to deuterated product (Ratio = $4.9 = K_{H/D}$).

Chlorination of Methylcyclohexane (Table 2)

Methylcyclohexane: A flame-dried, 1 dram vial was charged with a stir bar and. chloroamide (50.0 mg, 0.143 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (140 μ L). Methylcyclohexane (18.2 μ L, 0.143 mmol), and cesium carbonate (46.2 mg, 0.143 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 2 hours at 55° C. Upon competition, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The approximate yield was also verified with NMR analysis using 2,5 dimethylfuran as an internal standard. 74% GC yield of combined bromide products. *The secondary chloride products were assigned via analogy to our C-H bromination chemistry*.

Figure S1.

Overlaid Chromatogram: Chlorination of MeCyclohexane with Authentic 3° Product

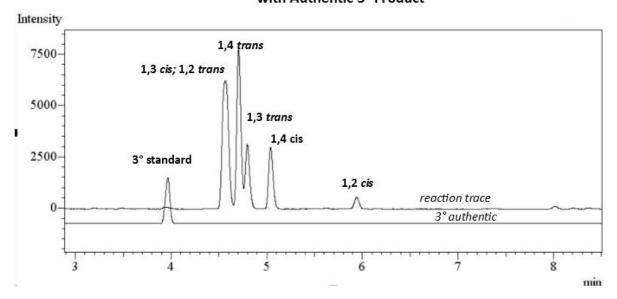


Table S1.

Chlorination of Methylcyclohexane: Selectivity			
Product	Retention Time	Percent Area	
3° chloride	4.00	1.51	
1,3-cis; 1,2-trans	4.58	36.19	
1,4-trans	4.75	31.58	
1,3-trans	4.81	13.65	
1,4- <i>cis</i>	5.05	14.65	
1,2- <i>cis</i>	5.95	Trace	

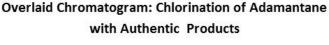
Bromination of Complex, Cyclic Alkanes (Table 2)

Norbornane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.29 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (0.29 mL). Norbornane (27.9 mg, 0.29 mmol), and cesium carbonate (94.2 mg, 0.29 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The reaction gives only the 2-exo product. 53.3% GC yield of 2-exo-chloronorbornane. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate 2-exo-chloronorbornane (18.2 mg, 0.14 mmol, 54% yield) as an off-white solid. Physical and spectral data were in accordance with literature data.¹²

Adamantane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (50.0 mg, 0.143 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (140 μ L).

Adamantane (19.5 mg, 0.143 mmol), and cesium carbonate (46.7 mg, 0.143 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 2 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The reaction gives both 1-and 2-Chloroadamante in a 19 to 1 ratio by GC respectively. 69.3% GC yield of combined chloroadamantanes.

Figure S2.



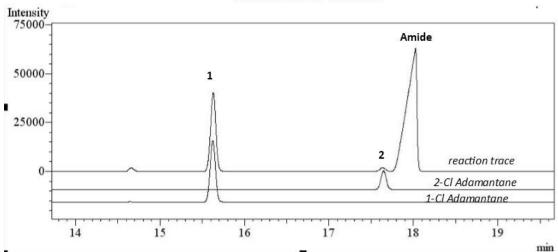


Table S2.

Chlorination of Adamantane: Selectivity			
Product Retention Time Percent Area			
1-Chloroadamantane	15.63	94.91	
2-Chloroadamantane	17.65	5.09	

1,2-trans-dimethylcyclohexane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide **1** (50.0 mg, 0.143 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (140 μ L). 1,2-trans-dimethylcyclohexane (30.5 μ L, 0.143 mmol), and cesium carbonate (46.7 mg, 0.058)

mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 2 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. Upon completion, the reaction mixture was concentrated under reduced pressure and dissolved in pentanes. The resulting suspension was run through a plug of silica and concentrated a second time. The reaction was analyzed by NMR using 2,5-dimethylfuran as an internal standard to determine the yield of secondary chloride products. GC analysis was used to determine the relative amount and yield, of the tertiary chloride products. 79.0% NMR yield of combined chloride products.

By analogy to our bromination chemistry, none of the cis tertiary product is observed in the reaction trace. Bromination of the trans isomer of starting material gives only 0.9% functionalization at a tertiary position to give only the trans isomer. Additionally, formation of the cis product from the trans isomer of starting material involves a thermodynamically unfavorable trapping step.

Figure S3.

Overlaid Chromatogram: Chlorination of 1,2 trans dimethylcyclohexane with Authentic 3° Products

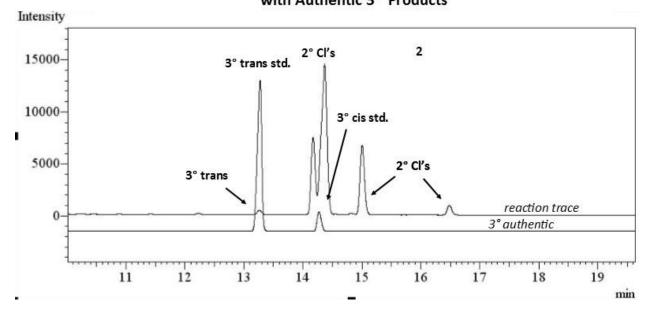


Table S3.

Chlorination of 1,2 trans-dimethylcyclohexane: Selectivity			
Product	Retention Time	Percent Area	
3° trans	13.25	0.91	
	14.17	20.32	
	14.36	56.71	
	15.07	19.41	
	16.48	2.63	

1,2-cis-dimethylcyclohexane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (50.0 mg, 0.143 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (140 μL). 1,2-cis-dimethylcyclohexane (30.5 μL, 0.143 mmol), and cesium carbonate (46.6 mg, 0.143 mmol) were then added.. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 2 hours at 55°. Upon completion, the reaction mixture was concentrated under reduced pressure and dissolved in pentanes. The resulting suspension was run through a plug of silica and concentrated a second time. The reaction was analyzed by NMR using 2,5-dimethylfuran as an internal standard to determine the yield of secondary bromide products. GC analysis was used to determine the relative amount and yield, of the tertiary bromide products. 72.9% NMR yield of combined chloride products.

Figure S4.

Overlaid Chromatogram: Chlorination of 1,2 cis dimethylcyclohexane with Authentic 3° Products

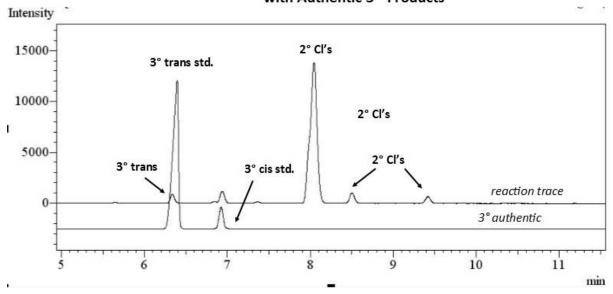


Table S4.

Chlorination of 1,2 <i>cis</i> -dimethylcyclohexane: Selectivity		
Product	Retention Time	Percent Area
3° trans	6.34	3.32
3° cis	6.94	4.81
	8.04	85.56
	8.52	4.30
	9.42	2.98

Chlorination of Electron Withdrawing Alkanes (Table 3)

MeO
$$Cs_2CO_3$$
, PhH, hv , 55° MeO Cs_2CO_3

Methyl hexanoate: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (560 μ L). Methyl hexanoate (21.2 μ L, 0.143 mmol), and cesium carbonate (93.2 mg, 0.286 mmol) were then added. *Note: alkane was added as a stock solution in benzene to improve the reproducibility of the results.* The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W

compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 83.2% GC yield of combined chloride products. *Only the major isomer was synthesized independently. The secondary chloride products were assigned via analogy to our C-H bromination chemistry*.

Figure S5.

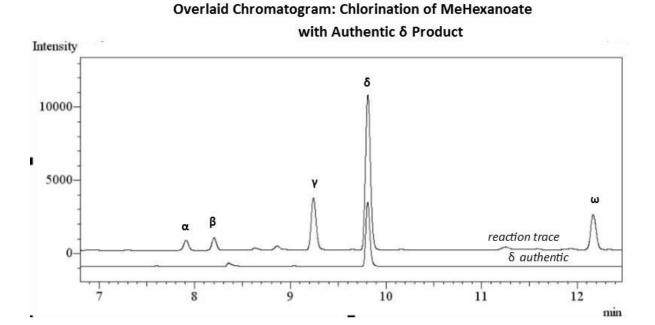


Table S5.

Chlorination of Methylhexanoate: Selectivity			
Product	Retention Time	Percent Area	
α	7.91	3.64	
β	8.20	4.62	
γ	9.24	19.67	
δ	9.81	57.63	
ω	12.16	14.41	

The response factors for all isomers of chloro-Methylhexanoate were calculated to be nearly identical; therefore, the relative percent area of the product peaks reflects the relative amounts of the different isomers formed in the reaction.

N-Penthylphthalimide: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μL). *N*-Pentylphthalimide (31.0 mg, 0.143 mmol), and cesium carbonate (93.4 mg, 0.143 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 79.7% GC yield of combined chloride products.

Figure S6.

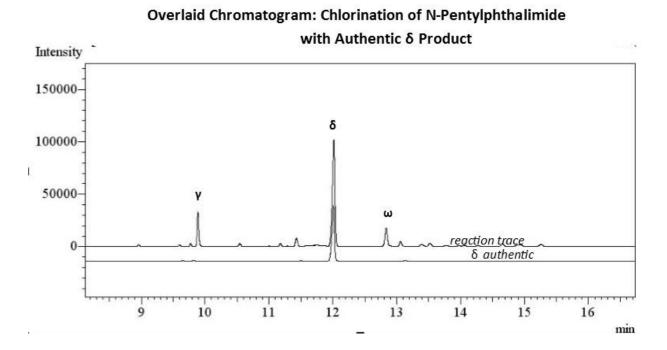


Table S6.

Chlorination of N-Pentylphthalimide: Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ	11.43	4.77	
δ	12.02	81.35	
ω	12.84	13.87	

NC
$$Cs_2CO_3$$
, PhH, hv , 55° NC Cl

Hexanenitrile: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μL). Hexanenitrile (17.2 μL, 0.143 mmol), and cesium carbonate (93.8 mg, 0.286 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 78.6% GC yield of combined chloride products.

Figure S7.

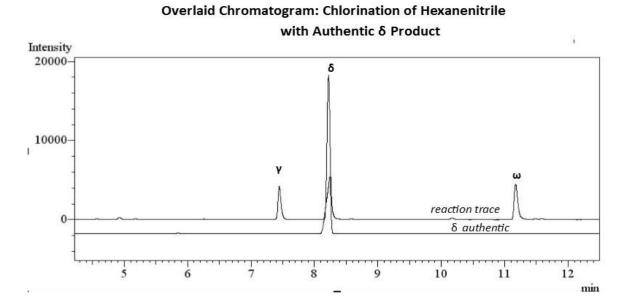


Table S7.

Chlorination of Hexane Nitrile: Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ	11.17	15.32	
δ	12.81	73.93	
ω	14.66	10.74	

$$CI$$
 Cs_2CO_3 , PhH, hv , 55°
 CI
 Cs_2CO_3

1-Chlorohexane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μ L). 1-Chlorohexane (17.3 μ L, 0.143 mmol), and cesium carbonate (93.0 mg, 0.286 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 74.0% GC yield of combined chloride products.

Figure S8.



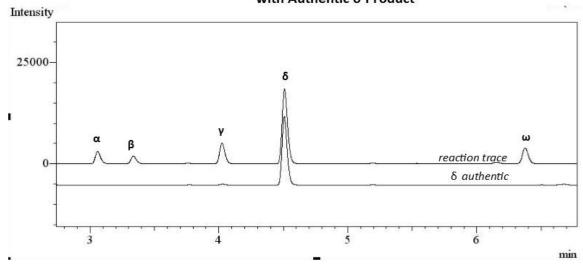


Table S8.

Chlorination of 1-Chlorohexane: Selectivity			
Product	Retention Time	Percent Area	
α	3.06	9.03	
β	3.37	5.54	
γ	4.02	15.29	
δ	4.51	57.23	
ω	6.37	12.91	

$$\begin{array}{c} & & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\$$

Amyl Acetate: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μ L). Amyl acetate (21.3 μ L, 0.143 mmol), and cesium carbonate (93.1 mg, 0.286 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 89.4% GC yield of combined chloride products.

Only the major isomer was synthesized independently. The secondary chloride products were assigned via analogy to our C-H bromination chemistry.

Figure S9.

Overlaid Chromatogram: Chlorination of Amylacetate with Authentic δ Product

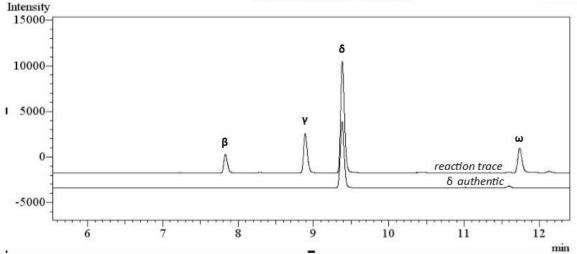


Table S9.

Chlorination of Amyl Acetate: Selectivity			
Product	Retention Time	Percent Area	
α			
β	7.83	8.9	
γ	8.88	19.69	
δ	9.83	57.85	
ω	11.73	13.47	

Pentyl Benzenesulfonate: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (100.0 mg, 0.286 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (280 μL). Pentyl benzenesulfonate (32.6 mg, 0.143 mmol), and cesium carbonate (92.5 mg, 0.286 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 86.5% GC yield of combined chloride products.

Only the major isomer was synthesized independently. The secondary chloride products were assigned via analogy to our C-H bromination chemistry.

Figure S10.

Overlaid Chromatogram: Chlorination of Pentyl Benzene Sulfonate with Authentic δ Product

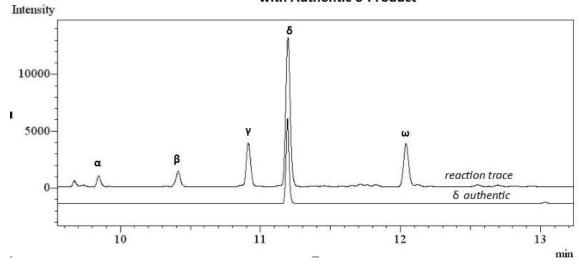


Table S10.

Chlorination of Pentyl Benzene Sulfonate: Selectivity			
Product	Retention Time	Percent Area	
α	9.84	3.48	
β	10.41	6.05	
γ	10.91	16.13	
δ	11.19	55.76	
ω	12.04	18.57	

n-Hexane: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (40.0 mg, 0.116 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (120 μ L). n-Hexane (15.2 μ L, 0.116 mmol), and cesium carbonate (37.8 mg, 0.116 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. 70.4% GC yield of combined chloride products.

Figure S11.

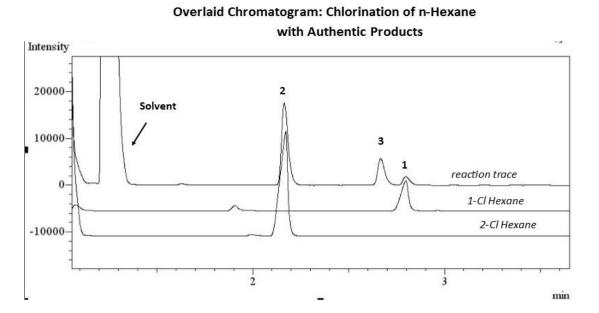


Table S11.

Chlorination of n-Hexane: Selectivity		
Product	Retention Time	Percent Area
2-Chlorohexane	2.16	69.51
3-Chlorohexane	2.66	23.38
1-Chlorohexane	2.79	7.10

Chlorination of in the Presence of More Reactive C-H Bonds and Substrate Unsaturation (Table 4)

$$\begin{array}{c} & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

S1: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 µL). The phthalimide substrate (50.2 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (88%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S1:** ¹**H NMR** ¹H NMR (600MHz ,CHLOROFORM-d) $\delta = 7.91 - 7.80$ (m, 2 H), 7.79 - 7.66 (m, 2 H), 4.44 - 4.28 (m, 1 H), 4.13 - 3.94 (m, 0.8 H), 3.51 (t, J = 6.6 Hz, 0.4 H), 2.33 (dtd, J = 4.4, 9.9, 14.0 Hz, 0.5 H), 2.25 - 2.07 (m, 1 H), 2.07 - 1.95 (m, 0.5 H), 1.89 (tdd, J = 5.4, 10.7, 13.8 Hz, 0.5 H), 1.84 - 1.69 (m, 1.5 H), 1.69 - 1.55 (m, 1.5 H), 1.55 - 1.43 (m, 6 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) $\delta = 168.5$, 134.0, 133.9, 131.9, 131.8, 123.2, 123.1, 58.3, 57.9, 47.3, 47.2, 46.7, 44.8, 37.6, 37.2, 32.9, 32.1, 31.1, 30.7, 25.4, 25.3, 24.1, 18.8, 18.7 ppm; **IR** (thin film, cm⁻¹) 2954.1, 2928.2, 1771.2, 1713.2, 1455.4, 1435.8, 1400.0, 1189.4, 1031.1, 723.2; **HRMS** (ESI) Calcd. for $[C_{14}H_{16}CINO_2+Na]^+ = 288.08$, Found = 288.08

Figure S12.

Overlaid Chromatogram: Chlorination of Substrate S1 with Isolated Products

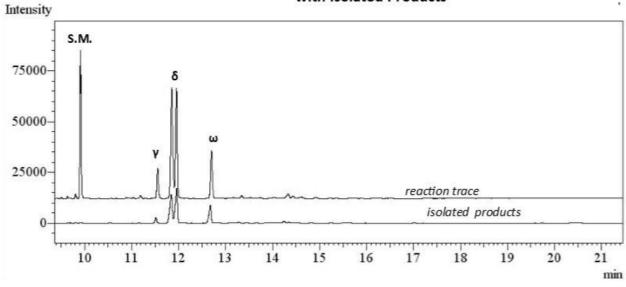


Table S12.

Chlorination of Phthalimide Substrate S1 : Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ	11.51	8.72	
δ	11.81	42.75	
δ	11.91	32.74	
ω	12.04	18.57	

Chlorination of S2: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 μ L). The sulfonate substrate (52.9 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard.

The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (76%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S2:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 7.96 - 7.91 (m, 2 H), 7.70 - 7.64 (m, 1 H), 7.60 - 7.53 (m, 2 H), 4.76 - 4.62 (m, 1 H), 3.98 - 3.85 (m, 0.8 H), 3.45 (t, J = 6.6 Hz, 0.2 H), 1.90 - 1.50 (m, 0.8 H), 1.47 - 1.43 (m, 3 H), 1.42 - 1.31 (m, 2 H), 1.30 - 1.25 (m, 3 H) ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 137.3, 133.7, 133.6, 129.4, 129.3, 129.2, 129.1, 127.9, 127.8, 127.7, 127.6, 80.5, 79.77, 37.7, 35.8, 35.1, 33.8, 33.3, 31.9, 25.4, 25.3, 21.0, 20.9 ppm; **IR** (thin film, cm⁻¹) 3010.3, 2926.0, 1777.3, 1710.3, 1448.9, 1452.0, 1400.8, 1387.4, 1190.2, 1033.0, 750.2; **HRMS** (ESI) Calcd. for [C₁₂H₁₇ClSO₃+Na]⁺ = 288.05, Found = 299.05

Overlaid Chromatogram: Chlorination of Substrate S2

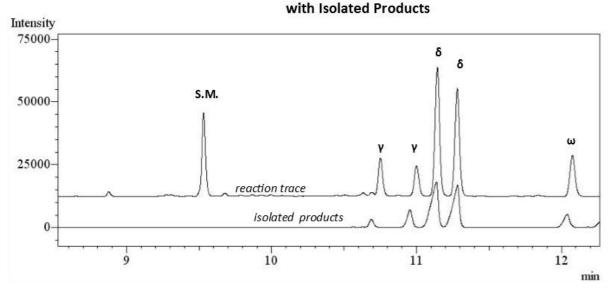


Table S13.

Figure S13.

Chlorination of Sulfonate Substrate S2: Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ	10.75	9.41	
γ	11.00	8.90	
δ	11.14	37.05	
δ	11.28	31.23	
ω	12.08	13.36	

Chlorination of S3: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 µL). The phthalimide substrate (50.3 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (69%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S3:** ¹**H NMR** (600MHz ,CHLOROFORM-d) $\delta = 7.90 - 7.81$ (m, 2 H), 7.78 - 7.69 (m, 13 H), 4.33 - 4.19 (m, 0.3 H), 4.11 (ddd, J = 3.5, 6.6, 10.5 Hz, 0.3 H), 3.70 - 3.63 (m, 0.3 H), 3.63 - 3.48 (m, 2 H), 2.33 (dtd, J = 3.5, 6.9, 13.6 Hz, 0.5 H), 2.28 - 2.20 (m, 0.3 H), 2.08 - 1.96 (m, 0.3 H), 1.96 - 1.85 (m, 3 H), 1.85 - 1.74 (m, 0.7 H), 1.74 - 1.66 (m, 1 H), 1.58 - 1.44 (m, 3 H), 1.01 - 0.91 (m, 3 H); ¹³C NMR (125MHz ,CHLOROFORM-d) $\delta = 168.8$, 168.7, 168.6, 134.1, 134.0, 133.9, 133.8, 132.0, 131.9, 131.8, 123.3, 123.2, 56.2, 56.1, 45.2, 45.1, 44.7, 43.9, 43.8, 43.0, 32.2, 31.5, 30.8, 30.7, 30.6, 29.9, 26.1, 25.2, 18.1, 17.4, 16.7 ppm; **IR** (thin film, cm⁻¹) 3013.0, 2926.0, 1780.7, 1703.1, 1458.2, 1452.9, 1382.8, 1374.2, 1210.0, 1058.1, 701.2; **HRMS** (ESI) Calcd. for $[C_{14}H_{16}CINO_2+Na]^+ = 288.07$, Found = 288.08

Figure S14.

Overlaid Chromatogram: Chlorination of Substrate S3 with Isolated Products

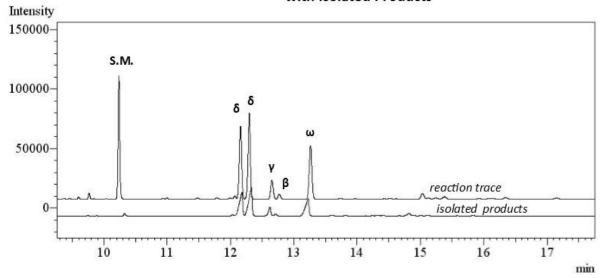


Table S14.

Chlorination of Phthalimide Substrate S3: Selectivity			
Product	Retention Time	Percent Area	
α			
β	12.77	2.05	
γ	12.65	8.17	
δ	12.96	29.96	
δ	11.28	36.36	
ω	13.27	25.49	

PhthN
$$CO_2Me$$
 S4 Cs_2CO_3 , PhH, hv , 55° CO_2Me CO_3Me

Chlorination of S4: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 μ L). The phthalimide substrate (58.2 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The crude reaction mixture was purified using column chromatography (10-20% EtOAc/Hexane) to isolate

a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (66%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S4:** 1 **H** (600MHz ,CHLOROFORM-d) δ = 7.96 - 7.86 (m, 2 H), 7.83 - 7.72 (m, 2 H), 4.92 - 4.81 (m, 1 H), 4.13 - 3.97 (m, 0.8 H), 3.79 - 3.72 (m, 3 H), 3.52 (t, J = 6.6 Hz, 0.4 H), 2.59 - 2.49 (m, 0.5 H), 2.47 - 2.38 (m, 0.8 H), 2.38 - 2.23 (m, 1 H), 1.88 - 1.76 (m, 4 H), 1.74 - 1.65 (m, 1 H), 1.54 - 1.45 (m, 3 H); 13 **C NMR** (125MHz ,CHLOROFORM-d) δ = 169.5, 169.4, 167.7, 167.6, 167.5, 134.4, 134.3, 131.7, 123.7, 123.6, 123.5, 57.8, 57.4, 52.9, 52.8, 51.9, 51.4, 44.5, 37.0, 36.7, 36.6, 31.7, 28.0, 26.4, 26.1, 25.4, 25.3, 23.6 ppm; **IR** (thin film, cm⁻¹) 3102.6, 2987.4, 2929.2, 1779.3, 1751.0, 1710.5, 1526.6, 1446.9, 1342.3, 1309.3, 1275.3, 882.9, 758.4, 719.4; **HRMS** (ESI) Calcd. for [C₂₅H₁₆ClNO₄+Na]⁺ = 332.07, Found = 332.06

Figure S15.

Overlaid Chromatogram: Chlorination of Substrate S4 with Isolated Products

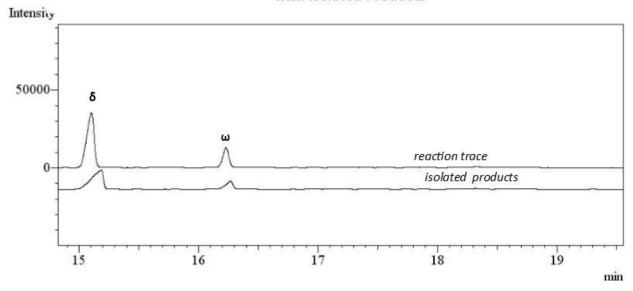


Table S15.

Chlorination of Phthalimide Substrate S4: Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ			
δ	15.14	77.54	
ω	16.23	22.45	

$$\begin{array}{c} & & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

Chlorination of S5: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 µL). The phthalimide substrate (64.1 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (81%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S5**: ¹**H NMR** (600MHz ,CHLOROFORM-d) $\delta = 7.85 - 7.80$ (m, 2 H), 7.74 - 7.68 (m, 2 H), 7.60 - 7.54 (m, 2 H), 7.39 - 7.33 (m, 2 H), 7.32 - 7.25 (m, 1 H), 5.40 - 5.29 (m, 1 H), 4.19 - 4.02 (m, 0.8 H), 3.53 (t, J = 6.6 Hz, 0.4 H), 2.88 - 2.74 (m, 0.3 H), 2.74 - 2.55 (m, 1 H), 2.55 - 2.41 (m, 0.4 H), 1.88 (qd, J = 7.0, 13.9 Hz, 0.3 H), 1.83 - 1.70 (m, 0.5 H), 1.56 - 1.49 (m, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) $\delta = 168.4$, 168.3, 168.2, 139.5, 139.3, 129.2, 134.1, 124.0, 133.9, 131.8, 131.7, 128.8, 128.7, 128.6, 128.2, 128.1, 128.0, 127.9, 127.8, 123.4, 123.3, 123.2, 58.2, 57.9, 54.9, 54.8, 45.4, 44.7, 37.7, 37.8, 37.6, 34.7, 32.1, 31.6, 30.3, 28.4, 28.1, 25.4, 25.3, 24.4, 22.7, 14.2 ppm; **IR** (thin film, cm⁻¹) 3022.4, 2959.2, 2929.3, 2859.4, 1769.3, 1710.9, 1463.7, 1397.1, 1072.3, 884.1, 757.5, 721.2; **HRMS** (ESI) Calcd. for [C₁₉H₁₈CINO₂+Na]⁺ = 350.09, Found = 350.10

Figure S16.

Overlaid Chromatogram: Chlorination of Substrate S5 with Isolated Products

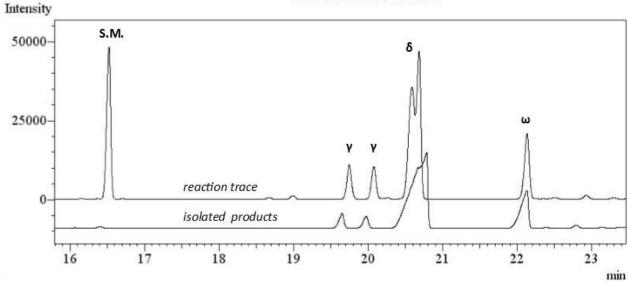


Table S16.

Chlorination of Phthalimide Substrate S5: Selectivity			
Product	Retention Time	Percent Area	
α			
β			
γ	19.74	8.06	
γ	20.07	7.68	
δ	20.58	34.94	
δ	20.68	32.90	
ω	22.13	16.41	

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ &$$

Chlorination of S6: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (150.0 mg, 0.44 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (440 μ L). The phthalimide substrate (52.8 mg, 0.22 mmol), and cesium carbonate (144.8 mg, 0.44 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard.

The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a mixture of major chloride products for characterization. 1D selective TOCSY NMR experiments were used to determine the identity of the major products. Reaction yield (78%) was calculated from crude NMR spectra using 2,5 dimethylfuran as an internal standard.

Analytical Data for mixture of chloride products formed from **S6**: ¹**H NMR** (600MHz ,CHLOROFORM-d) $\delta = 7.87 - 7.83$ (m, 2 H), 7.77 - 7.71 (m, 2 H), 6.33 - 6.19 (m, 1 H), 5.33 - 5.17 (m, 2 H), 4.81 - 4.68 (m, 1 H), 4.13 - 3.96 (m, 1 H), 3.52 (t, J = 6.6 Hz, 0.5 H), 2.39 - 2.26 (m, 0.7 H), 2.26 - 2.03 (m, 3 H), 2.03 - 1.88 (m, 0.5 H), 1.88 - 1.70 (m, 3 H), 1.70 - 1.60 (m, 2 H), 1.55 - 1.43 (m, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) $\delta = 168.1$, 135.4, 135.3, 134.1, 134.0, 131.8, 123.3, 123.2, 118.0, 112.9, 58.0, 57.9, 53.9, 53.8, 53.5, 44.7, 37.2, 37.0, 32.0, 31.2, 29.3, 29.2, 25.4, 25.3, 23.8 ppm; **IR** (thin film, cm⁻¹) 2938.4, 2864.1, 1463.1, 1448.2, 1363.5, 1127.1, 1007.2, 980.4, 962.2, 824.4, 751.8, 684.9, 591.0; **HRMS** (ESI) Calcd. for $[C_{15}H_{16}CINO_2+Na]^+ = 300.07$, Found = 300.08

Figure S17.

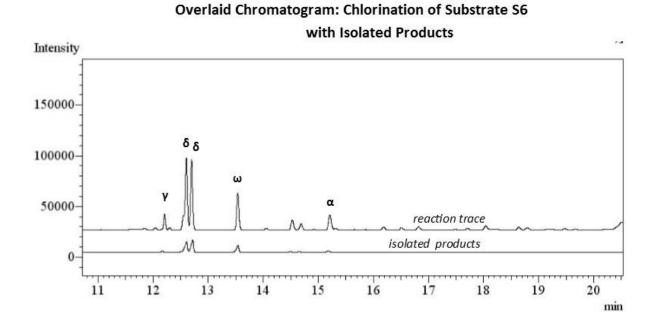


Table S17.

Chlorination of Phthalimide Substrate S6: Selectivity					
Product	Retention Time	Percent Area			
α	15.14	8.99			
β					
γ	12.13	36.32			
δ	12.53	28.77			
δ	12.63	19.28			
ω	13.46	8.99			

Chlorination of Complex Substrates

MsO
$$CF_3$$
 CF_3 CS_2CO_3 , PhH, hv , 55°

Chlorination of S7: A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (160.0 mg, 0.46 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (950 μL). The adamantane substrate (130.4 mg, 0.46 mmol), and cesium carbonate (150.s mg, 0.46 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a single isomer of product (95 mg, 0.29 mmol, 65% yield) as a white solid.

Analytical Data for single chloride product formed from **S7:** ¹**H NMR** (600MHz ,CHLOROFORM-d) δ = 5.96 - 5.81 (m, 1 H), 5.26 - 5.09 (m, 2 H), 4.47 (dd, J = 2.9, 9.4 Hz, 1 H), 3.09 - 2.98 (m, 3 H), 2.61 - 2.49 (m, 1 H), 2.41 (td, J = 8.9, 15.1 Hz, 1 H), 2.34 - 2.20 (m, 2 H), 2.17 - 2.10 (m, 2 H), 2.10 - 1.94 (m, 4 H), 1.76 - 1.49 (m, 7 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 134.7, 118.7, 89.4, 67.8, 47.6, 46.9, 46.8, 41.1, 39.3, 36.9, 36.5, 34.7, 34.1, 31.0, 30.9 ppm; **IR** (thin film, cm⁻¹) 2933.2, 2858.1, 2254.4, 1774.2, 1710.5, 1612.2, 1451.2, 1355.7, 1170.6, 1108.9, 1036.5, 912.2, 879.4, 835.0, 724.1, 531.3; **HRMS** (ESI) Calcd. for [C₁₅H₂₃ClSO₃+Na]⁺ = 341.10, Found = 341.09

Rimantadine (30): A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (300.0 mg, 0.88 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (950 μL). The phthalimide substrate (272.3 mg, 0.22 mmol), and cesium carbonate (286.1 mg, 0.88 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. The crude reaction mixture was purified using column chromatography (5% EtOAc/Hexane) to isolate a single isomer of product (199 mg, 0.58 mmol, 66% yield) as a white solid. NMR of the crude reaction mixture showed a small amount of dichloride formed (10:1 mono:di).

Analytical Data for **31:** ¹**H NMR** ¹H NMR (600MHz ,CHLOROFORM-d) δ = 7.89 - 7.79 (m, 2 H), 7.77 - 7.62 (m, 2 H), 4.11 (q, J = 7.5 Hz, 1 H), 2.26 - 2.16 (m, 2 H), 2.11 - 1.90 (m, 6 H), 1.70 - 1.53 (m, 6 H), 1.51 (d, J = 7.3 Hz, 3 H); ¹³**C NMR** (125MHz ,CHLOROFORM-d) δ = 169.2, 169.1, 134.1, 133.9, 132.1, 131.4, 123.4, 123.1, 68.6, 55.6, 49.4, 46.9, 46.8, 42.1, 37.8, 37.6, 34.8, 31.4, 31.3, 11.96 ppm; **IR** (thin film, cm⁻¹) 2935.1, 2888.9, 1709.5, 1642.1, 1333.5, 1172.5, 905.4, 835.9, 791.6, 736.7, 539.0; **HRMS** (ESI) Calcd. for [C₂₀H₂₂BrNO₂+H]⁺ = 366.12, Found = 366.12

$$R = \frac{1}{C} + \frac{1}{C} +$$

Chlorination of Cholestane (32): A flame-dried, 1 dram vial was charged with a stir bar and chloroamide (80.0 mg, 0.23 mmol) in the dark (overhead lights in the laboratory turned off), fitted with a PTFE lined screw cap and the reaction was taken into a glovebox, and dissolved in of dry, freeze-pump-thawed benzene (800 μ L). The alkane substrate (88.3 mg, 0.23 mmol), and cesium carbonate (72.3 mg, 0.23 mmol) were then added. The reaction was then sealed with teflon tape and taken out of the glovebox

and irradiated with two 23W compact fluorescent light bulbs for 4 hours at 55°. Upon completion, the reaction mixture was diluted with DCM (3 mL) and directly analyzed by gas chromatography using dodecane as an internal standard. NMR of the crude reaction mixture was compared to existing literature data to determine the yield (75%) and selectivity of the reaction. The analytical data was consistent with the literature.¹³

Synthesis of Chlorolissoclimide: Experimental Procedures

2-Chlorosclareolide 39.

(+)-Sclareolide 37 (600 mg, 2.39 mmol, 1.0 equiv), Cs₂CO₃ (780 mg, 2.39 mmol, 1.0 equiv), and chloroamide 1 (1.74 g, 5.0 mmol, 2.1 equiv) were dissolved in degassed benzene in the glovebox (5 mL). The mixture was stirred and irradiated with 2 - 23W CFL bulbs (1650 lumens each) at 55 °C for 36 h (when ¹H NMR revealed ~95% conversion). The mixture was filtered and concentrated under reduced pressure. Direct chromatographic purification (SiO₂, 5-10% EtOAc in hexanes) afforded a clear amorphous solid (535 mg, 3.9 mmol, 78% yield) Another 22mg were present in the mixed fractions to give a total of 557mg, 81.6% yield. (The reaction can be performed without the use of freeze pump thawed PhH or a glove box by degassing the reaction mixture with Ar for 15 minutes prior to heating to achieve 73% yield.) Alternative Procedure Using Benzoyl Peroxide as the Radical Initiator: (+)-Sclareolide 37 (1.00 g, 3.99 mmol, 1.0 equiv), benzoyl peroxide (165 mg, 0.68 mmol, 0.17 equiv), K₃PO₄ (144 mg, 0.68 mmol, 0.17 equiv), and chloroamide 1 (3.5 g, 10.0 mmol, 2.5 equiv) were dissolved in benzene (13 mL) and the mixture was degassed for 20 min with an Ar balloon bearing a 18 gauge needle. A reflux condenser was attached to the flask and the mixture was heated at reflux for 24 h. A 30 µL aliquot revealed a ~50% conversion (1H NMR). The reaction mixture was cooled to ambient temperature and more benzoyl peroxide (100 mg, 0.41 mmol, 0.10 equiv) and K₃PO₄ (87 mg, 0.41 mmol, 0.10 equiv) were added. The solution was degassed in the same manner and heated for 16 h after which aliquot NMR revealed ~75% conversion. This process was repeated again and after another 15 h no sclareolide remained. The mixture was concentrated under a stream of air. Direct chromatographic purification (SiO₂, 10% EtOAc in hexanes) afforded a clear amorphous solid (941 mg, 3.31 mmol, 83% yield). Our data match those previously reported by Groves. 13

¹H NMR (500 MHz, CDCl₃) δ 4.22 (tt, J = 12.2, 4.2 Hz, 1H), 2.43 (dd, J = 16.2, 14.8 Hz, 1H), 2.29 (dd, J = 16.2, 6.5 Hz, 1H), 2.13 (dt, J = 12.0, 3.3 Hz, 1H), 1.98–2.07 (m, 2H), 1.91 (dq, J = 14.4, 3.3 Hz, 1H), 1.70 (td, J = 12.4, 4.3 Hz, 1H), 1.54 (t, J = 12.7 Hz, 1H), 1.32–1.42 (m, 2H), 1.33 (s, 3H), 1.13 (dd, J = 12.7, 2.7 Hz, 1H), 0.972 (s, 3H), 0.968 (s, 3H), 0.90 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 176.0, 85.8, 58.6, 55.7, 53.8, 52.4, 49.7, 38.4, 38.2, 35.8, 32.9, 28.6, 21.6, 21.3, 20.1, 15.8; IR (film) 2951, 2871, 1773, 1387, 1197 cm⁻¹; HRMS (ES+) m/z calc d for C₁₆H₂₅O₂Cl [M]⁺: 284.1543; found 284.1548; [α]²⁵_D +48.2° (c = 1.00, CHCl₃).

(Note: The following two-step sequence to make **40** is based on the procedures for the non-chlorinated variant reported by Boukouvalas¹⁴ with slight modifications.)

Amide S8.

Al(CH₃)₃ (2.0 M in toluene, 3.50 mL, 6.99 mmol, 2.1 equiv) was added to a suspension of N,O-dimethylhydroxylamine hydrochloride (647 mg, 6.66 mmol, 2.0 equiv) in CH₂Cl₂ (12 mL) at 0°C. The white suspension became a clear solution. The ice bath was removed and after 2 h a solution of **39** (950 mg, 3.33 mmol, 1.0 equiv in 12 mL CH₂Cl₂) was added dropwise over ~5 min at ambient temperature. After stirring for 2 hours at 23 °C, the flask was cooled in an ice bath and 5 mL of 10% aq. H₂SO₄ was added.

The ice bath was removed and the crude mixture was diluted with CH_2Cl_2 (40 mL), H_2O (10 mL), and 1M HCl (30 mL). The organic phase was separated and the aqueous phase was extracted with CH_2Cl_2 (2 x 40 mL). The combined organic layers were dried over MgSO₄, filtered and concentrated under reduced pressure. The crude product was purified by flash chromatography (SiO₂, 70% \rightarrow 80% EtOAc in hexanes) to afford amide S8 as a thick clear oil (1.05 g, 3.04 mmol, 91% yield).

¹H NMR (600 MHz, CDCl₃) δ 4.17 (tt, J = 12.3, 4.0 Hz, 1H), 3.74 (s, 1H), 3.20 (s, 1H), 2.58 (d, J = 16.1 Hz, 1H), 2.48 (dd, J = 16.4, 6.7 Hz, 1H), 2.08 (ap d, J = 10.9, 1H), 2.03 (dd, J = 6.5, 3.6 Hz, 1H), 1.92–1.98 (m, 2H), 1.69 (dq, J = 13.8, 2.4 Hz, 1H), 1.50 (t, J = 12.7 Hz, 1H), 1.47 (dd, J = 13.3, 3.9 Hz, 1H), 1.32 (t, J = 12.2 Hz, 1H), 1.26 (qd, J = 13.2, 3.2 Hz, 1H) 1.16 (s, 1H), 1.09 (dd, J = 12.3, 2.0 Hz, 1H), 0.95 (s, 3H), 0.88 (s, 3H), 0.85 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 175.5, 72.7, 61.3, 56.3, 55.8, 55.1, 54.9,

51.9, 49.6, 44.1, 41.0, 35.9, 33.1, 27.0, 23.4, 21.8, 20.1, 16.5; **IR** (film) v 3500, 2966, 2927, 1655, 1454, 1340, 1158 cm⁻¹; **HRMS** (ES+) m/z calc'd for $C_{18}H_{32}O_3CINa$ [M + Na]⁺: 368.1968; found 368.1963; $[\alpha]^{25}D_1$ +32.9° (c = 1.00, CHCl₃).

Alkene 40.

Alcohol **S8** (510 mg, 1.46 mmol, 1.0 equiv) was dissolved in CH_2Cl_2 (10 mL). Pyridine (236 μ L, 2.92 mmol, 2.0 equiv) was added, and the solution was cooled to -78 °C. Pyridine (969 μ L, 12.0 mmol, 8.25 equiv) was added to a conical flask containing $SOCl_2$ (530 μ L, 7.30 mmol, 5 equiv) and CH_2Cl_2 (15 mL) and this clear mixture was transferred via Teflon® cannula¹⁵ dropwise over 15 min to the flask containing alcohol **S8**.

The reaction was complete after 45 min at –78 °C and sat. aq. NaHCO₃ (20 mL) was added. The product mixture was warmed to ambient temperature, the phases were separated and the aqueous phase was extracted (2 x 30 mL CH₂Cl₂). The combined organic extracts were dried (MgSO₄), filtered, and concentrated under reduced pressure. The pale yellow crude oil was purified by flash column chromatography (SiO₂, 11% EtOAc in hexanes) to afford alkene **40** as a thick oil (364 mg, 1.11 mmol, 76% yield).

¹**H NMR** (600 MHz, CDCl₃) δ 4.78 (s, 1H), 4.48 (s, 1H), 4.17 (tt, J = 12.3, 3.9 Hz, 1H), 3.73 (s, 3H), 3.16 (s, 3H), 2.72 (dd, J = 15.0, 10.7 Hz, 1H), 2.58 (d, J = 10.1 Hz, 1H), 2.37–2.42 (m, 2H), 2.09–2.18 (m, 2H), 1.98 (ddd, J = 12.8, 3.7, 2.2 Hz, 1H), 1.70–1.76 (m, 1H), 1.55 (t, J = 12.6 Hz, 1H), 1.50 (t, J = 12.2 Hz, 1H), 0.97 (s, 3H), 0.85 (s, 3H), 0.79 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 174.1, 148.3, 107.0, 61.4, 55.8, 53.9, 52.0, 51.4, 49.3, 41.2, 37.1, 36.1, 33.3, 32.5, 24.1, 23.4, 22.2, 15.3; IR (film) v 2939, 2852, 1776, 1667, 1461, 1385, 1004 cm⁻¹; **HRMS** (ESI) m/z calc'd for C₁₈H₃₀O₂CINNa [M + Na]⁺ 350.1863, found 350.1866; [α]²⁵_D +7.38° (c = 1.00, CHCl₃).

Allylic Alcohol S9.

A solution of *t*-butyl hydroperoxide (133 μL, 5.5 M in decane, 730 μmol, 4.4 equiv) was added to an ice cold stirring suspension of selenium dioxide (5.5 mg, 49.8 μmol, 0.3 equiv) in CH₂Cl₂ (2.5 mL). After 30 min, a solution of amide **40** (54.4 mg, 166 μmol, 1.0 equiv) was added dropwise and the solution was allowed to warm to room temperature. TLC analysis indicated complete consumption of starting material after 8 h. The excess peroxide was quenched with sat. aq. Na₂SO₃. The biphasic solution was diluted with water (5 mL) and CH₂Cl₂ (5 mL) and the aqueous phase was extracted with CH₂Cl₂ (3 x 20 mL). The combined organic phases were dried (MgSO₄), and concentrated *in vacuo*. The crude oil was purified by column chromatography (SiO₂, 80% EtOAc in hexanes) to give alcohol **S9** as a white amorphous solid (36.4 mg, 105 μmol, 63% yield).

¹H NMR (500 MHz, CDCl₃) δ 5.01 (s, 1H), 4.61 (s, 1H), 4.37 (s, 1H), 4.17 (tt, J = 12.3, 3.9 Hz, 1H), 3.74 (s, 3H), 3.16 (s, 3H), 3.06 (d, J = 10.8 Hz, 1H), 2.68 (dd, J = 15.8, 10.9 Hz, 1H), 2.42 (dd, J = 16.1, 3.3 Hz, 1H), 2.13 (ap d, J = 12.1 Hz, 1H) 2.00 (ddd, J = 12.8, 3.8, 2.1 Hz, 1H), 1.91 (dt, J = 13.9, 2.7 Hz, 1H), 1.84 (dd, J = 12.9, 2.9 Hz, 1H), 1.60 (t, J = 12.6 Hz, 1H), 1.56 (t, J = 12.2 Hz, 1H), 1.48 (td, J = 13.4, 3.1 Hz, 1H), 0.98 (s, 3H), 0.87 (s, 3H), 0.76 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 177.5, 149.6, 110.2, 73.1, 61.4, 55.7, 52.0, 49.0, 46.2, 46.0, 41.3, 35.7, 32.4, 33.0, 29.8, 26.7, 22.1, 14.4; IR (film) v 3405, 2960, 2939, 2897, 1645, 1460, 1436, 1389, 766 cm⁻¹; HRMS (ES+) m/z calc'd for C₁₈H₃₀O₃ClNNa [M + Na]⁺: 366.1812, found 366.1818; $[\alpha]^{25}p$ –25.7° (c = 0.60, CHCl₃).

Enone 41.

A solution of oxalyl chloride (16.7 μ L, 192 μ mol, 1.2 equiv) in CH₂Cl₂ (800 μ L) was cooled to – 78 °C and DMSO (27 μ L, 384 μ mol, 2.4 equiv) was added. After 10 min alcohol **S9** (55 mg in 800 μ L CH₂Cl₂, 160 μ mol, 1.0 equiv) was added dropwise over 10 min. The solution was stirred for 20 min and Et₃N (111.5 μ L, 800 μ mol, 5 equiv) was added. The suspension was allowed to warm to 0 °C. After 40 min

at 0 °C no starting material was present by TLC analysis and the mixture was diluted with hexanes (5 mL) and sat. aq. NH₄Cl (5 mL). The biphasic solution was further diluted with 1:1 hexanes in EtOAc (15 mL) and water. The organic phase was washed sequentially with brine (10 mL) and water (10 mL), dried (MgSO₄), filtered, and concentrated under reduced pressure. Flash column chromatography (SiO₂, 35% EtOAc in hexanes) gave **41** as a thick oil (53 mg, 154 μmol, 96% yield).

¹**H NMR** (500 MHz, CDCl₃) δ 6.00 (d, J = 2.5 Hz, 1H), 5.11 (d, J = 2.4 Hz, 1H), 4.18 (tt, J = 12.4, 3.8 Hz, 1H), 3.74 (s, 3H), 3.19 (s, 3H), 2.97–3.03 (m, 1H), 2.58–2.72 (m, 3H), 2.28 (dd, J = 17.9, 14.1 Hz, 1H), 2.23 (ap d J = 12.5 Hz, 1H), 2.05 (ddd, J = 12.9, 3.7, 2.2 Hz, 1Hz), 1.73 (dd, J = 14.0, 4.6 Hz, 1H), 1.60 (t, J = 12.7 Hz, 1H), 1.56 (t, J = 12.3 Hz, 1H), 0.96 (s, 3H), 0.94 (s, 3H), 0.91 (s, 3H); ¹³**C NMR** (125 MHz, CDCl₃) δ 200.6, 146.8, 121.1, 72.1, 61.5, 54.6, 51.6, 49.8, 49.6, 48.9, 39.5, 37.5, 36.0, 32.7, 32.3, 28.9, 21.4, 15.1; **IR** (film) v 2963, 1693, 1662, 1608, 1461, 1415, 1388, 1264 cm⁻¹; **HRMS** (ES+) m/z calc'd for C₁₈H₂₈O₃ClNNa [M + Na]⁺: 364.1655, found 364.1658; [α]²²_D –36.3° (c = 0.51, CHCl₃).

CI...

A1

$$i$$
-Bu₂AlH

 Et_2O , $-78 \rightarrow -25^{\circ}C$
 i -Bu₂AlH

 i -B

Aldehyde S10.

Enone **41** (72 mg, 205 μ mol, 1.0 equiv) was dissolved in Et₂O (4 mL, 0.05 M) and the solution was cooled to -78° C, forming a suspension. A solution of iBu_2AlH in toluene (1.0 M, 615 μ L, 615 μ mol, 3.0 equiv) was added slowly along the side of the flask and the reaction mixture became homogenous. After 40 min at -78° C, the flask was warmed to -25° C over 1 h then stirred at that temperature. After 1.5 h acetone (50 μ L) was added followed by 1M HCl (1 mL). The biphasic solution was warmed to ambient temperature and 1M HCl (3 mL) and water (3 mL) were added. The aqueous phase was extracted (3 x 6 mL 1:1 EtOAc in hexanes) and the combined organic extracts were dried (MgSO₄), filtered and concentrated. The crude oil was purified by chromatography (SiO₂, 22% \rightarrow 27% EtOAc in hexanes) to afford **S10** as a thick oil (44 mg, 154 μ mol, 75% yield).

¹**H NMR** (500 MHz CDCl₃) δ 9.66 (d, J = 3.0 Hz, 1H) 5.23 (s, 1H), 4.62 (s, 1H), 4.16 (tt, J = 12.3, 4.0 Hz, 1H), 4.10 (br s, 1H), 2.63 (ddd, J = 17.0, 11.1, 3.0 Hz, 1H), 2.48 (dd, J = 17.0, 3.5 Hz, 1H), 2.41 (ap d, J = 11.0 Hz, 1H), 2.07–2.14 (m, 3H), 2.02 (ddd, J = 13.0, 3.9, 2.1 Hz, 1H), 1.86 (d, J = 4.1 Hz, 1H), 1.55 (t, J = 12.7 Hz, 1H), 1.41 (t, J = 12.2 Hz, 1H), 1.23–1.32 (m, 2H), 1.00 (s, 3H), 0.88 (s, 3H), 0.75 (s, 3H); ¹³C

NMR(125 MHz, CDCl₃) δ 201.6, 149.3, 105.9, 72.97, 55.0, 51.84, 51.76, 49.4, 48.5, 40.7, 39.3, 35.9, 33.2, 32.7, 22.1, 15.1; **IR** (film) ν 3400, 2960, 2853, 2725, 1720, 1647, 1459, 1391 cm⁻¹; **HRMS** (ES+) m/z calc'd for C₁₆H₂₅O₂ClNH₄ [M + NH₄]⁺: 302.1887, found 302.1882. [α]²²_D -18.1° (c = 1.0, CHCl₃).

CI...
$$CH_2Cl_2$$
, -78 °C H $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$ $OTMS$

TMS ether 42.

Aldehyde **S10** (12.5 mg, 43 μ mol, 1.0 equiv) was dissolved in CH₂Cl₂ (450 μ L) and cooled to -78 °C. To the reaction mixture were sequentially added 2,4,6-collidine (57 μ L, 430 μ mol, 10 equiv) and TMSOTf (47 μ L, 258 μ mol, 6.0 equiv). After 1 h at -78 °C, triethylamine (50 μ L) and methanol (100 μ L) were added and the flask was warmed to ambient temperature. The solution was diluted with 10% EtOAc in hexanes (5 mL) and washed consecutively with water, citric acid (10% aq.), and brine (4 mL each). The organic extract was dried over MgSO₄, filtered, and concentrated under reduced pressure. The residue was purified by flash column chromatography to give aldehyde **42** a thin clear film (14.5 mg, 40.6 μ mol, 97% yield).

¹H NMR (500 MHz, CDCl₃) δ 9.64 (d, J = 2.8 Hz, 1H), 5.25 (s, 1H), 4.58 (s, 1H), 4.15 (tt, J = 12.2, 3.8 Hz, 1H), 4.02 (ap dd, J = 10.1, 5.2 Hz, 1H), 2.62 (ddd, J = 16.8, 10.9, 3.1 Hz, 1H), 2.46 (dd, J = 16.8, 3.5 Hz, 1H), 2.39 (ap d, J = 10.6 Hz, 1H), 2.08 (ap d, J = 12.1 Hz, 1H), 2.03 (ddd, J = 12.9, 3.9, 2.0 Hz, 1H), 1.93 (ddd, J = 11.9, 4.2, 1.7 Hz, 1H), 1.54 (t, J = 12.7 Hz, 1H), 1.39 (t, J = 11.9 Hz, 1H), 1.32 (ap t, J = 11.1 Hz, 1H), 1.29 (t, J = 15.1 Hz, 1H), 0.98 (s, 3H), 0.87 (s, 3H), 0.75 (s, 3H), 0.13 (s, 9H); ¹³C NMR(125 MHz, CDCl₃) δ 201.8, 148.5, 107.0, 73.5, 55.1, 51.9, 51.8, 49.5, 48.6, 40.7, 39.4, 35.8, 33.6, 33.1, 22.1, 15.1, -0.1; IR (film) v 2956, 2851, 2718, 1726, 1650, 1460, 1393, 1251 cm⁻¹; HRMS (ES+) m/z calc'd for C₁₉H₃₃O₂ClSiNH₄ [M + NH₄]⁺: 374.2282, found 374.2285; [α]_D²² -8.60° (c = 1.10, CHCl₃).

Chlorolissoclimide 34.

Imide 43¹⁶ (14.0 mg, 48 μmol, 1.3 equiv) was dried by azeotropic distillation from of benzene (20 μL) in a 1 dram vial and dissolved in CH₂Cl₂ (250 μL, 0.1 M). The solution was cooled to –78 °C and *n*Bu₂BOTf (1.0 M in CH₂Cl₂, 52 μL, 52 μmol, 1.4 equiv)¹⁷ was added along the side of the vial. The clear solution was stirred at –78 °C for 30 min. At that temperature, *i*Pr₂NEt (1.0 M in CH₂Cl₂, 67 μL, 67 μmol, 1.8 equiv) was added and the mixture was stirred for 20 min. To ensure complete enolization, the vial was warmed to 0 °C for 30 sec and cooled back to –78 °C. The solution often becomes pale yellow upon warming but remained clear. Finally, a solution of aldehyde 42 (13.2 μg in 100 μL CH₂Cl₂, 37 μmol, 1.0 equiv) was added at –78 °C and the reaction vial was warmed over 1.5 hours to –25 °C and a temperature between –30 and –25 °C was maintained for 20 h. The reaction mixture was quenched with methanol (50 μL) warmed to ambient temperature, and water (3 mL) and CH₂Cl₂ (3 mL) were added. The phases were separated and the aqueous phase was extracted with CH₂Cl₂ (2 x 3 mL). The combined organic extracts were dried (MgSO₄), filtered, and concentrated *in vacuo* to yield crude product 44 a yellow oil.

The crude aldol product was transferred to a 1 dram vial, NH₃ (2.0 M in methanol, 800 μL) was added, and the reaction was stirred at 23 °C for 24 h. The solution was concentrated *in vacuo* to afford a mixture of chlorolissoclimide (34), and the primary amide of methyl ester 44. To complete succinimide formation, the crude material was dissolved in dry THF (400 μL) and NaH (55% suspension in mineral oil, 3.5 mg, 80 μmol) was added at ambient temperature. The solution evolved bubbles for ~10 min. Water (200 μL) was added followed by 1M HCl (100 μL). The solution was quickly added to a mixture of pH 7 buffer (2 mL), brine (1 mL), and water (1 mL) and extracted with CH₂Cl₂ (4 x 4 mL CH₂Cl₂). The combined organic extracts were dried (MgSO₄) and concentrated under reduced pressure. Flash chromatrography (SiO₂, 60%→70% EtOAc in hexanes) afforded a modestly (~85%) pure product. To further purify the product, this mixture was taken up in CH₂Cl₂ (250 μL) and hexanes was added until a white precipitate formed (~300 μL). The suspension was centrifuged for 1 min and the solvent was decanted with a pipette, leaving pure chlorolissoclimide (34) as a white film (7.8 mg, 20.3 μmol, 55% yield).

The data obtained for our synthetic sample of **34** matched those reported by Biard. For ¹H NMR and ¹³C NMR data comparison to natural **34** in CD₂Cl₂ see Table S18, below. ¹⁸

¹H NMR (500 MHz CDCl₃) δ 8.88 (br s, 1H), 5.30 (s, 1H), 4.90 (s, 1H), 4.29 (br s, 1H), 4.17 (tt, J = 12.2, 3.9 Hz, 1H), 4.00 (s, 1H), 3.03 (d, J = 3.2 Hz, 1H), 2.88–2.95 (m, 1H), 2.85 (dd, J = 18.0, 5.0 Hz, 1H), 2.68 (dd, J = 18.0, 9.2 Hz, 1H), 2.61 (br s, 1H), 2.20 (, J = 10.5 Hz, 1H), 2.09 (dd, J = 10.3, 4.8 Hz, 1H), 2.02 (d, J = 10.9 Hz, 1H), 1.82 (ddd, J = 13.2, 11.6, 5.3 Hz, 1H), 1.69 (s, 1H), 1.55–1.63 (m, 1H), 1.52 (t, J = 12.7 Hz, 1H), 1.23–1.28 (m, 2H), 1.20 (t, J = 14.2 Hz, 1H), 0.99 (s, 3H), 0.87 (s, 3H), 0.74 (s, 3H); ¹³C NMR(125 MHz, CDCl₃) δ 178.4, 176.0, 149.6, 105.5, 73.3, 68.9, 55.0, 52.2, 51.9, 51.7, 49.4, 46.7, 41.6, 35.9, 33.2, 33.1, 29.3, 29.0, 22.0, 15.0; IR (film) v 2972, 2951, 1773, 1709, 1647, 1538, 1461, 1368 cm⁻¹; HRMS (ES+) m/z calc'd for C₂₀H₃₀O₄ClNH₄ [M + Na]⁺: 406.1761, found 406.1756. [α]²²_D +50° (c = 0.11, CH₃OH), [lit.: [α]_D²⁵ +119.3° (c = 1.59, CH₃OH)].

Alcohol S12.

Weinreb amide $S11^3$ (115 mg, 392 µmol) was subjected to the reaction conditions described above for S9 to provide alcohol S12 (81.4 mg, 263 µmol, 67% yield) as a thick oil after column chromatography (SiO₂, 80% EtOAc in hexanes).

¹H NMR (600 MHz, CDCl₃) δ 4.96 (s, 1H), 4.56 (s, 1H), 4.36 (t, J = 2.3 Hz, 1H), 3.72 (s, 3H), 3.15 (s, 3H), 2.98 (d, J = 10.7 Hz, 1H), 2.64 (t, J = 5.3 Hz, 1H), 2.45 (dd, J = 16.3, 3.4 Hz, 1H), 1.89 (dt, J = 13.9, 2.8 Hz, 1H), 1.77 (dd, J = 13.2, 2.8 Hz, 1H), 1.59–1.47 (m, 4H), 1.43 (d, J = 11.4 Hz, 1H) 1.27–1.18 (m, 2H), 0.90 (s, 3H), 0.82 (s, 3H), 0.71 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 174.4, 150.7, 109.1, 73.5, 61.3, 47.2, 46.2, 41.9, 39.0, 38.6, 33.2, 33.0, 32.4, 30.4, 27.8, 21.6, 19.3, 13.8; IR (film) v 3412, 2928, 1644, 1459, 1441, 1387 cm⁻¹; HRMS (ES+) m/z cale'd for C₁₈H₃₁NO₃Na [M + Na]⁺: 332.2202; found 332.2202; [α]²⁵_D –29.9° (c = 0.46, CHCl₃).

Enone S13.

Alcohol **S12** (140 mg, 450 μ mol) was subjected to the reaction conditions described for **41** to provide alcohol **S13** (137 mg, 445 μ mol, 98% yield) as an oil after flash column chromatography (SiO₂, 40% EtOAc in hexanes).

¹H NMR (600 MHz, CDCl₃) δ 5.94 (s, 1H), 5.05 (s, 1H), 3.71 (s, 3H), 3.15 (s, 3H), 2.92 (s, 1H), 1.71–1.66 (m, 2H), 1.60, (dt, J = 13.6, 3.2 Hz, 1H), 1.57 (m, 1H), 1.48 (d, J = 13.4 Hz, 1H), 1.27–1.19 (m, 3H), 0.89 (s, 6H), 0.85 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 202.0, 173.6, 147.9, 119.7, 61.4, 60.3, 50.7, 50.1, 41.5, 38.5, 38.1, 37.1, 33.4, 32.5, 29.3, 28.7, 20.9, 18.8, 14.4; **IR** (film) $_{\rm V}$ 2927, 1693, 1664, 1607, 1460.1, 1414, 1387 cm⁻¹; **HRMS** (ES+) m/z calc d for C₁₈H₂₉NO₃Na [M + Na]⁺: 330.2045; found 330.2038; [α]²³_D –41.8° (c = 0.45, CHCl₃).

Aldehyde S14.

Enone S13 (20 mg, 65 μ mol, 1 equiv) was azeotroped in a 1 dram vial from benzene (30 μ L) then taken up in Et₂O (650 μ L, 0.1 M). The clear solution was cooled to -78 °C to give a cloudy white suspension. A solution of LiAlH₄ (1M in Et₂O, 130 μ mol, 130 μ L, 2 equiv) was added dropwise down the side of the vial. After 3 h at -78 °C, no starting material remained by TLC analysis and 50 μ L of EtOAc was added. The reaction mixture was warmed to -10 °C and water (10 μ L) and 1M aq. NaOH (10 μ L) were added sequentially followed by warming to room temperature. The reaction mixture was poured into ice water (5 mL) and extracted with (3 x 4 mL Et₂O). The combined organic phases were dried over MgSO₄, filtered, and concentrated. The crude substance was purified by flash column chromatography (SiO₂, 25% EtOAc in hexanes) to aldehyde give S14 as a clear oil (10.8 mg, 43 μ mol, 66% yield).

¹**H NMR** (500 MHz, CDCl₃) δ 5.18 (s, 1H), 4.58 (s, 1H), 4.09 (m, 1H), 2.56 (ddd, J = 16.8, 10.6, 3.1 Hz, 1H), 2.48 (dd, J = 16.8, 4.0 Hz, 1H), 2.14–2.11 (m, 1H), 1.77 (d, J = 5.3 Hz, 1H), 1.76–1.59 (m, 1H), 1.54–

1.50 (m, 1H), 1.45 (d, J = 13.0 Hz, 1H), 1.34–1.26 (m, 2H), 1.22 (td, J = 13.1, 4.0 Hz, 1H), 1.06 (td, J = 12.6, 3.9 Hz, 1H), 0.93 (s, 3H), 0.83 (s, 3H), 0.70 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 202.8, 150.5, 104.8, 73.4, 52.8, 49.0, 41.8, 39.6, 39.1, 38.6, 33.45, 33.42, 21.6, 19.1, 14.5; **IR** (film) v 3422, 2924, 2846, 2721, 1722, 1649, 1460, 1388 cm⁻¹; **HRMS** (ES+) m/z calc'd for C₁₆H₂₆O₂Na [M + Na]⁺: 273.1830; found 273.1827; $[\alpha]^{22}_{D} - 15.3^{\circ}$ (c = 0.2, CHCl₃)

TMSOTf, collidine
$$CH_2Cl_2, -78 °C$$

$$80\%$$
S14
$$S15$$

TMS ether S15.

Aldehyde **S14** (9.0 mg, 36 μ mol) was subjected to the reaction conditions described for **42** and afforded TMS ether **S15** as a thin film (9.2 mg, 28 μ mol, 80% yield) after flash column chromatrography (SiO₂, 5% EtOAc in hexanes).

¹H NMR (600 MHz, CDCl₃) δ 9.63 (s, 1H), 5.21 (s, 1H), 4.54 (s, 1H), 4.02 (m, 1H), 2.53 (ddd, J = 19.9, 10.5, 3.12 Hz, 1H), 2.45 (dd, J = 16.7, 4.0 Hz, 1H), 2.28 (d, J = 10.5, 1H) 1.92 (app. d, J = 10, 1H) 1.59–1.56 (m, 1H), 1.55–1.48 (m, 2H), 1.44 (d, J = 13.2 Hz, 1H), 1.36 (q, J = 12.6 Hz, 1H), 1.25–1.16 (m, 2H), 1.04 (t, J = 12.9 Hz, 1H), 0.89 (s, 3H), 0.82 (s, 3H), 0.69 (s, 3H), 0.13 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 203.0, 149.7, 105.9, 73.9, 52.9, 49.2, 41.8, 39.6, 39.2, 38.6, 34.2, 33.4, 29.5, 21.6, 19.2, 14.5, –0.10; IR (film) v 2951, 2845, 2712, 1727, 1649, 1459, 1389, 1250 cm⁻¹; HRMS (ES+) m/z calc'd for C₁₉H₃₄O₂SiNa [M + Na]*: 345.2226; found 345.2229; [α]₀²⁵ -10.4° (c = 0.38, CHCl₃).

Haterumaimide Q 36.

Aldehyde S15 (9.3 mg, 30 μ mol) was subjected to the reaction conditions described for 34 and afforded haterumaimide Q (36) as a thin film (5.5 mg, 18 μ mol, 52% yield) after flash column

chromatography (SiO₂, 60% EtOAc in hexanes). The data obtained for our synthetic sample of **36** matched those reported by Ueda.¹⁹

For ¹H NMR and ¹³C NMR data comparison to natural **3** in (CD₃)₃SO see Table S19, below.

¹H NMR (500 MHz, CDCl₃) δ 7.78, (s, 1H), 5.30 (s, 1H), 4.86 (s, 1H), 4.36 (t, J = 7.0 Hz, 1H), 3.99 (dd, J = 11.1, 5.4 Hz, 1H), 2.92–2.89 (m, 1H, H₁₃), 2.87 (dd, J = 16.5, 5.5 Hz, 1H), 2.67 (dd, J = 16.5, 7.6 Hz, 1H), 2.15–2.09 (m, 1H), 1.93 (s, 1H), 1.84–1.76 (m, 1H), 1.59–1.52 (m, 2H), 1.50 (d, J = 11.3 Hz, 1H), 1.45 (d, J = 13.1 Hz, 1H), 1.17 (td, J = 13.1, 4.1 Hz, 1H), 1.14 (td, J = 13.1, 2.4 Hz, 1H), 0.95–0.85 (m, 1H), 0.89 (s, 3H), 0.81 (s, 3H), 0.68 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 178.6, 176.2, 150.8, 104.3, 73.8, 69.1, 53.3, 52.0, 46.8, 41.9, 39.5, 39.1, 33.8, 33.6, 33.5, 29.4, 29.0, 21.6, 19.3, 14.5; IR (film) cm⁻¹; HRMS (ES+) m/z calc'd for C₂₀H₃₁O₄NNa [M + Na]⁺: 372.2151, found 373.2145; [α]_D²⁵ +31.1° (c = 0.040, CHCl₃) [lit.: [α]_D²⁵ +36.0° (c = 0.19, CHCl₃)]

Aldehyde S16.

Alkene **40** (22 mg, 67.2 μ mol, 1.0 equiv) was dissolved in Et₂O (1.3 mL, 0.05 M) and the solution was cooled to -78° C. A solution of iBu_2AlH in toluene (1.0 M, 134 μ L, 134 μ mol, 2.0 equiv) was added slowly along the side of the flask. After 2 h at -78° C, the flask was warmed to -20° C and quenched with 50 μ L of acetone followed by water (200 μ L). The biphasic solution was warmed to ambient temperature and diluted with more 1M HCl (1 mL) and water (2 mL). The aqueous phase was extracted with Et₂O (3 X 5 mL) and the combined organic extracts were dried (MgSO₄), filtered and concentrated under reduced pressure. The crude oil was purified by chromatography (SiO₂, 5% EtOAc in hexanes) and afforded **S16** a clear oil (17.2 mg, 64 μ mol, 95% yield).

¹H NMR (600 MHz, CDCl₃) δ 9.65 (d, J = 3.2 Hz, 1H), 4.87 (s, 1H), 4.42 (s, 1H), 4.16 (tt, J = 12.6, 3.9 Hz, 1H), 2.54 (ddd, J = 17.2, 11.8, 3.3 Hz, 1H), 2.41–2.47 (m, 3H), 2.07–2.13 (m, 2H), 2.00 (ddd, J = 12.8, 3.8, 2.2 Hz, 1H), 1.74–1.79 (m, 1H), 1.56 (t, J = 12.6 Hz, 1H), 1.45 (t, J = 12.2 Hz, 1H), 1.32 (qd, J = 12.7, 4.1 Hz, 1H), 1.26 (dd, J = 12.6, 2.3 Hz, 1H), 1.24 (s, 3H), 0.88 (s, 3H), 0.76 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 202.4, 147.1, 109.2, 55.4, 54.1, 52.0, 50.5, 49.3, 41.0, 39.6, 37.1, 36.1, 33.3, 23.3, 22.2, 15.1; IR

(film) v 2960, 2853, 2714, 1726, 1345, 1460, 1390, 1339 cm⁻¹; **HRMS** (ES+) m/z calc'd for C₁₆H₂₅OCINH₄ [M + NH₄]⁺: 286.1938, found 286.1945; $[\alpha]_{\mathbf{p}}^{25}$ –1.0° (c = 0.52, CHCl₃).

7-Deoxychlorolissoclimide 45.

Aldehyde **S16** (13 mg, 48.4 μmol) was subjected to the reaction conditions described for **34** to yield **45** (13.1 mg, 35.0 μmol, 73% yield) after column chromatography (SiO₂, 55% EtOAc in hexanes) as a clear oil.

¹H NMR (600 MHz, CDCl₃) δ 7.80 (s, 1H), 4.97 (s, 1H), 4.36 (s, 1H), 4.16 (tt, J = 12.2, 3.9 Hz, 1H), 2.89–2.94 (m, 1H), 2.89 (dd, J = 17.4, 5.3 Hz, 1H), 2.70 (dd, J = 17.3, 8.6 Hz, 1H), 2.45 (ap d, J = 13.0 Hz, 1H), 2.23 (d, J = 11.4 Hz, 1H), 1.96–2.03 (m, 2H), 1.92 (d, J = 3.6 Hz, 1H), 1.72–1.80 (m, 2H), 1.68 (d, J = 10.8 Hz, 1H), 1.59 (dd, J = 14.0, 7.9 Hz, 1H), 1.52 (t, J = 12.8 Hz, 1H), 1.30 (qd, J = 15.1, 4.3 Hz, 1H), 1.28 (t, J = 11.8 Hz, 1H), 1.13 (dd, J = 12.5, 2.5 Hz, 1H), 0.96 (s, 3H), 0.86 (s, 3H), 0.75 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 178.6, 176.2, 147.7, 108.6, 77.2, 69.3, 55.4, 54.6, 53.6, 52.0, 49.6, 46.8, 42.0, 37.8, 36.1, 33.3, 29.4, 29.2, 23.7, 22.1, 15.0; IR (film) v 3500, 2958, 2936, 2853, 1776, 1712, 1463, 1388, 1186 cm⁻¹; HRMS (ES+) m/z calc'd for C₂₀H₃₀ClNO₃Na [M + Na]⁺: 390.1812, found 390.1820; [α]_D²⁵ +64.3 (c = 0.19, CHCl₃).

7-Deoxyhaterumaimide Q 46.

Known aldehyde 38^3 (11.7 mg, 50.0 µmol) was subject to the reaction conditions described for 34 and chromatographically purified (SiO₂, 50% EtOAc in hexanes) to yield 46 (11.0 mg, 33 µmol, 63% yield) as a clear oil. The characterization data were consistent with previously reported data.²⁰

¹**H NMR** (500 MHz, CDCl₃) δ 7.97 (s, 1H), 4.91 (s, 1H), 4.67 (s, 1H), 4.35 (ap t, J = 6.9 Hz, 1H), 2.93–2.98 (m, 1H), 2.87 (dd, J = 17.7, 5.3 Hz, 1H), 2.67 (dd, J = 17.7, 9.0 Hz, 1H), 2.42 (ddd, J = 12.9, 4.1, 2.5

Hz, 1H), 2.03–1.95 (m, 2H), 1.79–1.66 (m, 3H), 1.65–1.55 (m, 2H), 1.52 (tt, J = 13.0, 3.6 Hz, 1H), 1.41 (ap d, J = 13.3 Hz, 1H), 1.33 (qd, J = 12.9, 4.4 Hz, 1H), 1.17 (td, J = 13.4, 3.8 Hz, 1H), 1.08 (dd, J = 12.6, 2.7 Hz, 1H), 0.94 (td, J = 12.6, 4.2 Hz, 1H), 0.89–0.84 (m, 1H), 0.88 (s, 3H), 0.80 (s, 3H), 0.69 (s, 3H).

13C NMR (125 MHz, CDCl₃) δ 178.9, 176.5, 149.0, 107.4, 69.5, 55.7, 53.8, 46.8, 42.0, 39.8, 39.2, 38.2, 33.6, 33.5, 29.4, 29.1, 24.3, 21.6, 19.3, 14.4; IR (film) v 3441, 3224, 3079, 2933, 2845, 1774, 1713, 1459, 1364, 1188 cm⁻¹; HRMS [ES+] m/z calc'd for C₂₀H₃₁O₃NNa [M + Na]⁺: 356.2202, found 356.2203.

[α]_D²² +51.1° (c = 0.35, CHCl₃) [lit.: [α]_D²⁰ +40.6° (c = 0.71, CHCl₃)].

Comparison Tables for chlorolissoclimide (1) and haterumaimide Q (2).

Table 1: Synthetic and natural chlorolissoclimide (1) in CD₂Cl₂.

Atom	δ C nat.	δ C syn.	Δ	δ H nat.	Multiplicity, J	δ H syn.	Multiplicity, J	Δ
	(ppm)	(ppm)		(ppm)	(Hz)	(ppm)	(Hz)	
1	50.1	50.1	-	2.35	tdd (12.5, 4.0, 2.1)	2.23	ap d (12.1)	0.12^{21}
				1.30	t (12.5)	1.28	t (12.0)	-0.02
2	56.4	56.5	0.1	4.20	tt (12.5, 4.0)	4.20	tt (12.2, 3.9)	-
3	52.8	52.8	-	2.01	ddd (12.5, 4.0, 2.1)	2.01	ddd (12.8, 3.9, 2.0)	-
				1.53	t (12.5)	1.53	t (12.6)	
4	36.8	36.7	-0.1	-	-	-	-	-
5	52.9	52.9	-	1.20	S	1.20	S	-
6	34.0	34.0	-	2.10	S	2.09	m	0.01 -
				1.22	S	1.22	S	
7	74.1	74.1	-	4.00	d (9.1)	4.00	br s	-
8	150.9	150.9	-	-	-	-	-	-
9	52.3	52.3	-	1.65	dd (11.3, 7.8)	1.63	m (overlapping)	-0.02
10	42.4	42.4	-	-	-	-	-	-
11	29.9	30.0	0.1	1.80	ddd (14.6, 11.3, 5.6)	1.82	ddd (15.2, 11.2, 5.6)	0.02
				1.60	dt (14.6, 7.8)	1.60	m (overlapping)	-
12	69.7	69.7	-	4.32	dddd (7.8, 5.6, 4.0, 2.2)	4.32	m	-
13	47.5	47.5	-	2.90	ddd (9.1, 5.2, 2.2)	2.89	ddd (9.0, 5.1, 2.2)	0.01
14	30.1	30.2	0.1	2.85	dd (17.7, 5.2)	2.83	dd (17.8, 5.1)	0.02
				2.65	dd (17.7, 9.1)	2.66	dd (17.8, 9.2)	0.01
15	180.0	180.3	0.3	-	-	-	-	-
16	177.5	177.8	0.3	-	-	-	-	-
17	105.8	105.9	0.1	5.31	d (1.6)	5.32	S	0.01 -
				4.92	d (1.6)	4.91	S	0.01
18	33.8	33.8	-	1.00	S	0.98	S	-0.02
19	22.6	22.7	0.1	0.85	S	0.87	S	0.02
20	15.5	15.6	0.1	0.75	S	0.74	S	-0.01
NH				8.31	br s	8.35	br s	0.04
7-OH				N.O.	-	2.14	d (4.3)	-
12-OH				2.37	br s	2.42	d (2.8)	0.05

Table 2: Synthetic and natural haterumaimide Q (3) in $(CD_3)_3SO$.

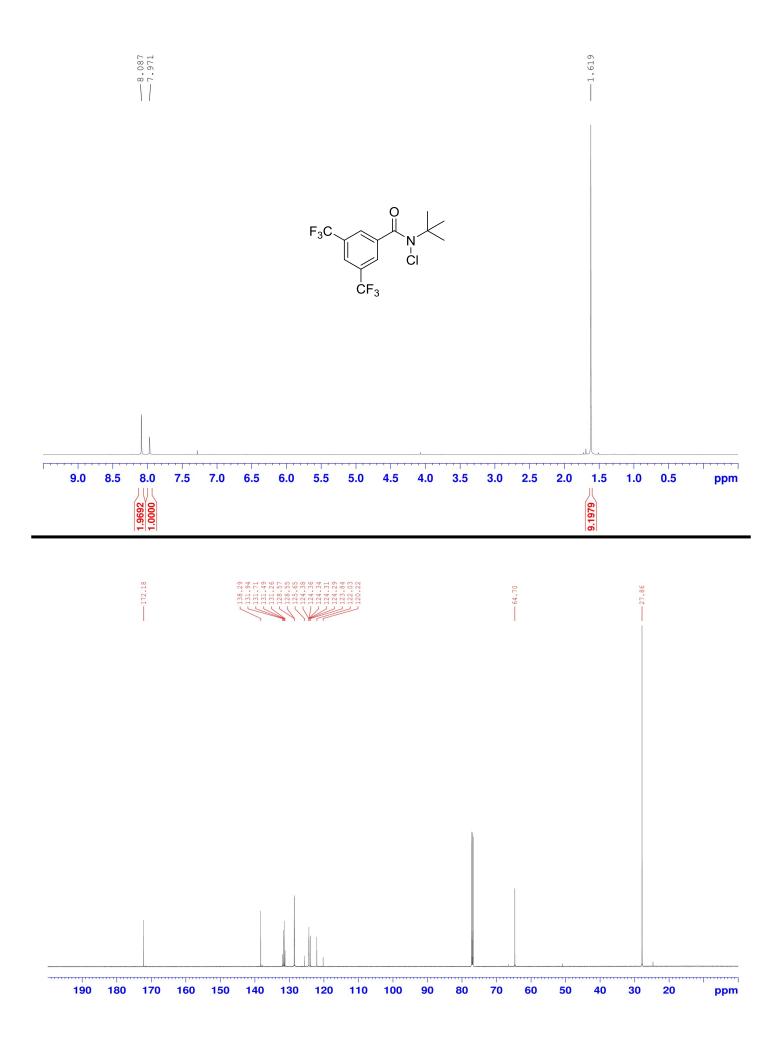
Atom	δ C nat.	δ C syn.	Δ	δ H nat.	Multiplicity, J	δ H syn.	Multiplicity, J	Δ
	(ppm)	(ppm)		(ppm)	(Hz)	(ppm)	(Hz)	
1	38.0	38.0	-	1.60	ddd (13.5, 5.5, 3.0)	1.60	m	-
				0.91	dd (13.5, 4.0)	0.91	td (10.7, 3.7)	-0.02
2	18.9	18.9	-	1.50	m	1.50	m	-
				1.42	m		m	
3	41.5	41.5	-	1.36	ddd (13.5, 5.5, 3.0)	1.36	d (13.0)	-
				1.11	m	1.53	m	
4	33.1	33.1	-	-	-	-	-	-
5	52.2	52.2	-	1.12	dd (12.0, 3.5)	1.20	m	-
6	33.5	33.5	-	1.87	ddd (12.0, 4.4, 3.5)	1.87	m	0.01
				1.13	m	1.22	m	-
7	72.0	71.9	-0.1	3.75	ddd (11.0, 5.5, 4.5)	3.76	dd (9.9, 4.8)	0.01
8	151.1	151.1	-	-	-	-	-	-
9	49.8	49.8	-	1.44	dd (10.0, 5.5)	1.63	dd (12.1)	-0.02
10	38.7	38.6	0.1	-	-	-	-	-
11	29.5	29.5	-	1.58	m	1.58	m	-
				1.40	m		m	
12	68.8	66.8	2.0	3.99	dddd (9.0, 6.5, 5.0, 2.0)	3.99	m	-
13	45.3	45.3	-	2.80	ddd (8.5, 5.0, 2.0)	2.80	ap t (6.3 Hz)	-
14	28.9	28.9	-	2.52	dd (17.5, 5.0)	2.45 -	m	-
				2.46	dd (17.5, 8.5)	2.61		
15	181.1	181.1	-	-	-	-		-
16	178.8	178.8	-	-	-	-		-
17	103.6	103.6	-	5.18	S	5.19	S	0.01
				4.76	S	4.76	S	-
18	33.3	33.3	-	0.85	S	0.85	S	-
19	21.5	21.5	-	0.75	S	0.75	S	-
20	14.2	14.2	-	0.57	S	0.58	S	0.01
NH	_			10.99	br s	10.98	br s	-0.01
7-OH	_			4.91	d (4.5)	4.89	S	-0.02
12-OH				4.92	d (5.0)	4.90	S	-0.02

Cell Viability Assays.

MTS assays were performed for cell viability as described by the supplier (Promega; Madison, WI). Briefly, 5000 cells/well for solid tumor cell lines were seeded in 96-well plates, incubated overnight at 37°C in 5% (v/v) CO₂ and exposed to compounds in a dose-dependent manner for 48 h. For assays using blood tumor cells, 10000 cells/well were seeded in 96-well plates, followed by treating cells with compounds in a dose-dependent manner for 48h. Dimethyl sulfoxide (DMSO) was used as the vehicle control. Viable cells were determined by tetrazolium conversion to its formazan dye. Absorbance was monitored at 490 nm using an automated ELISA plate reader. IC₅₀ values were determined using CalcuSyn software.

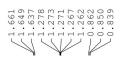
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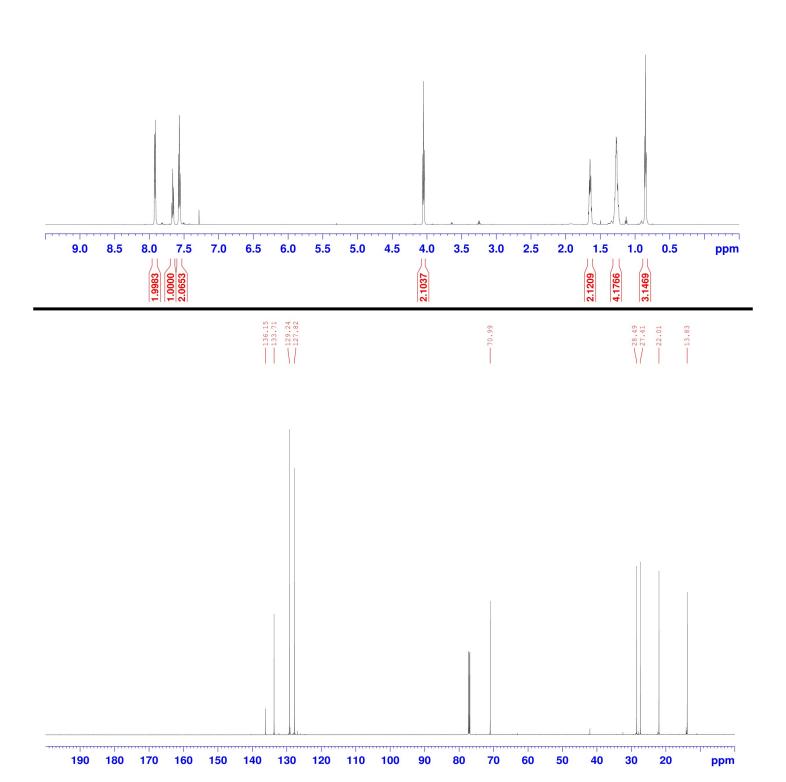


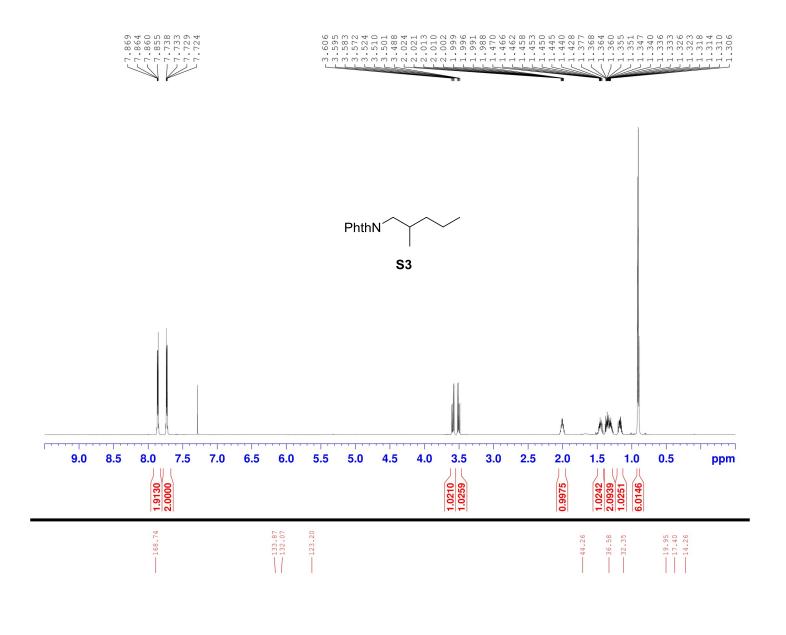


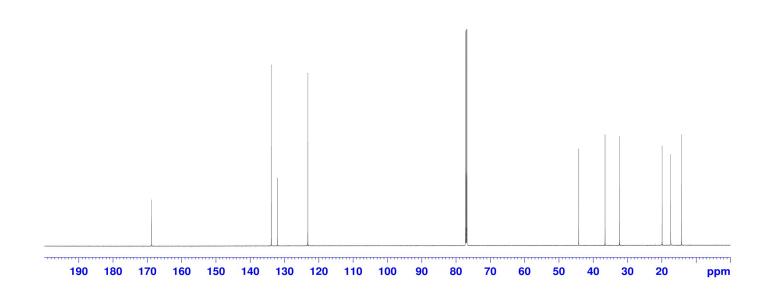


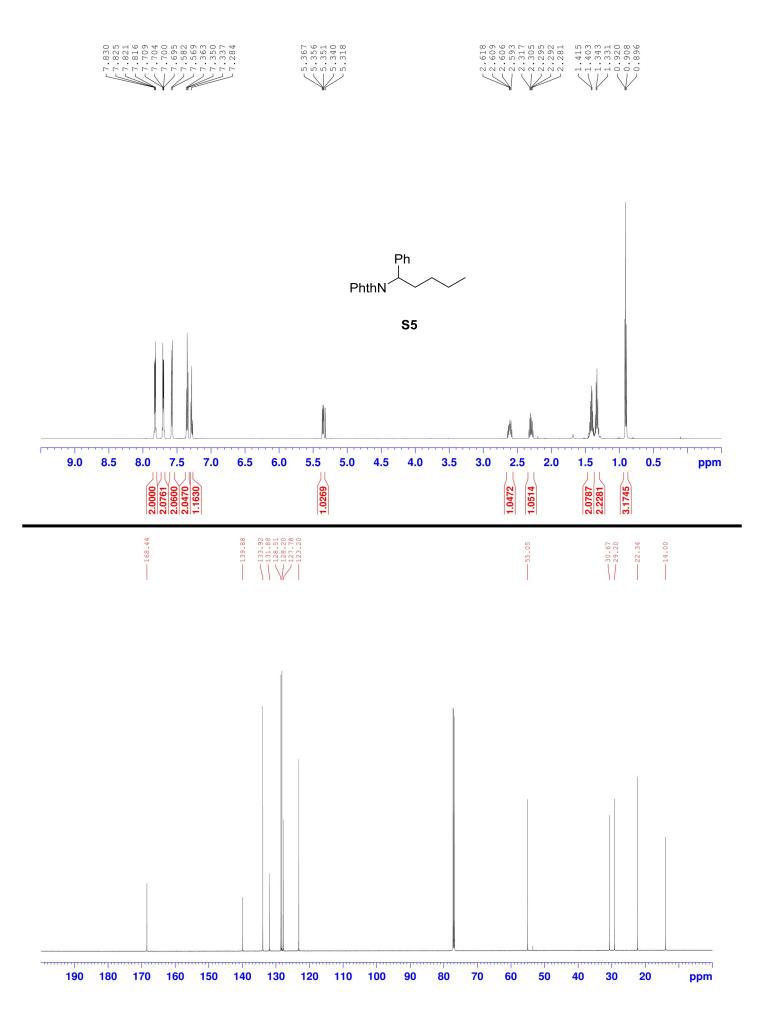


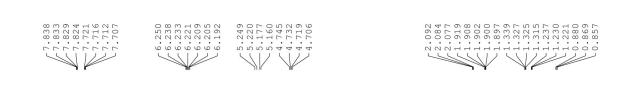


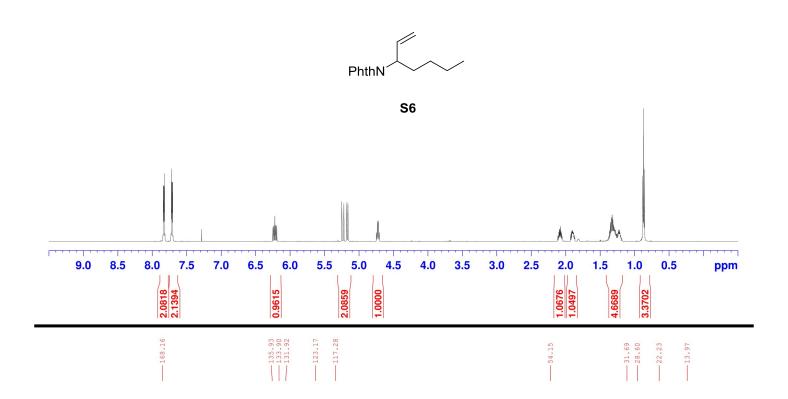


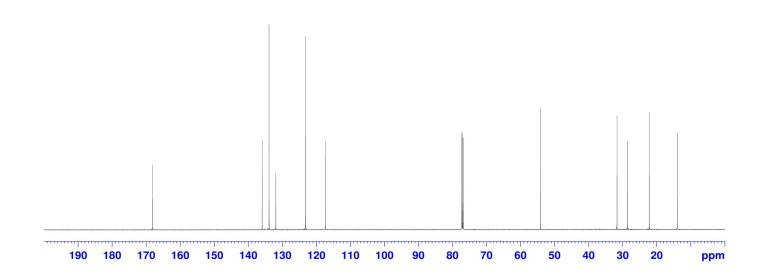


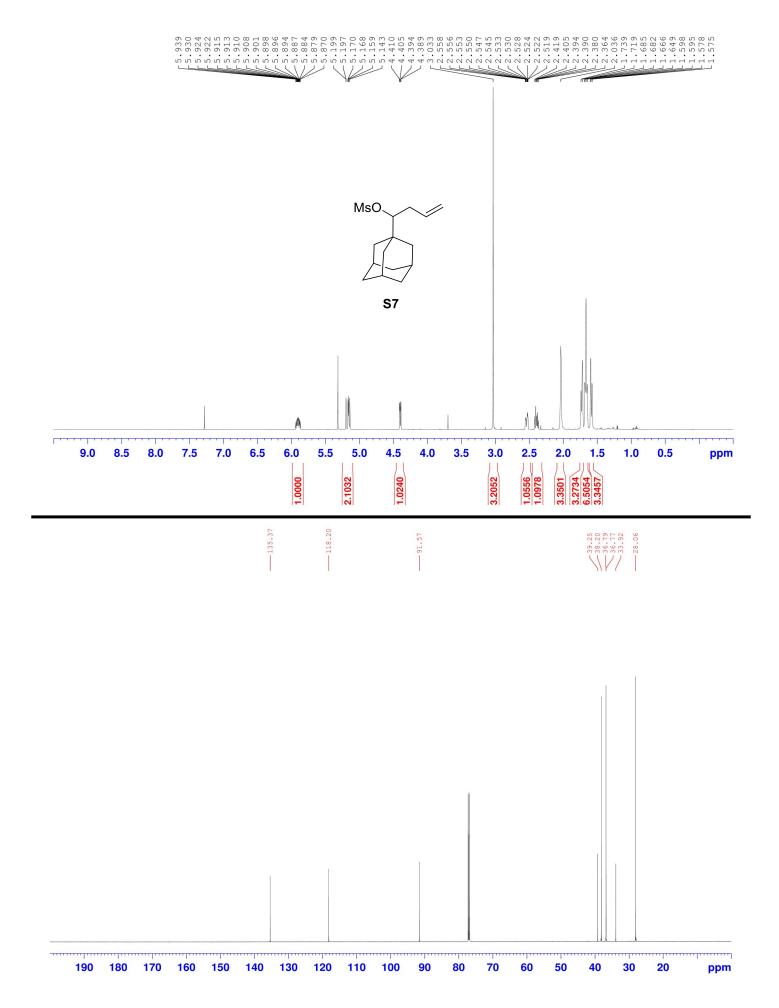


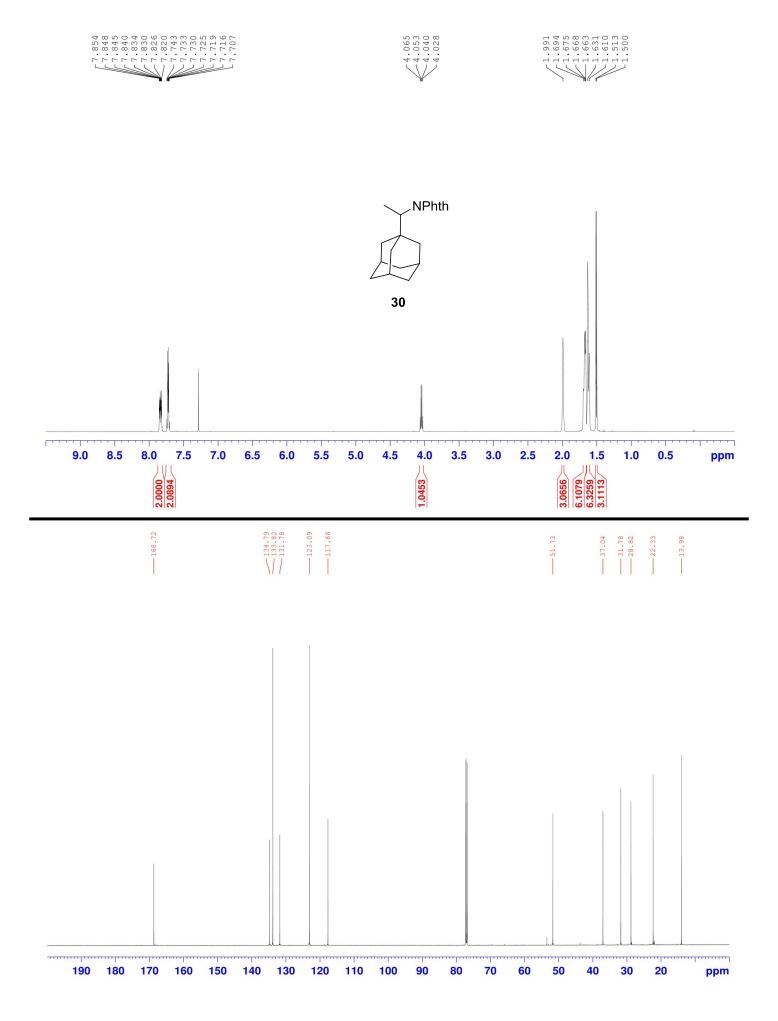


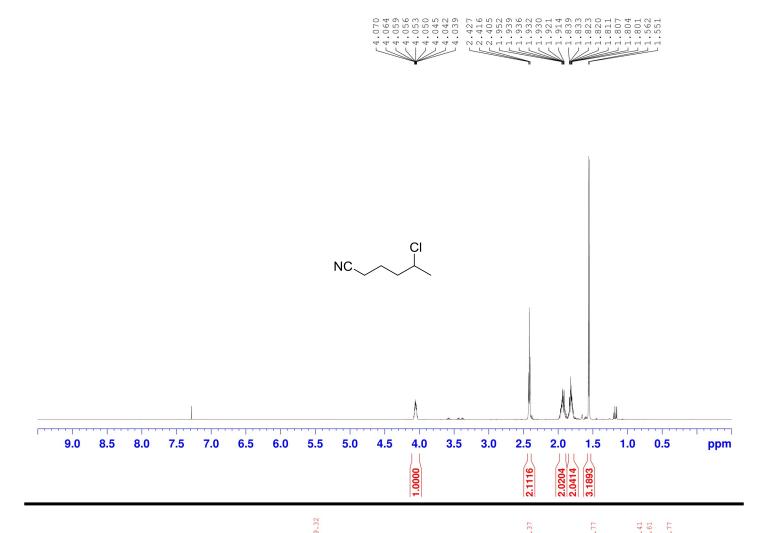


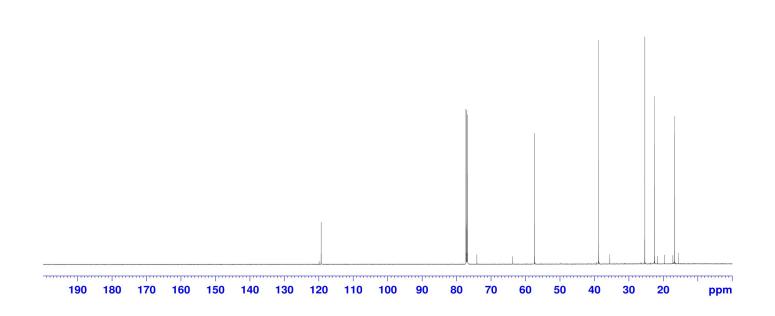


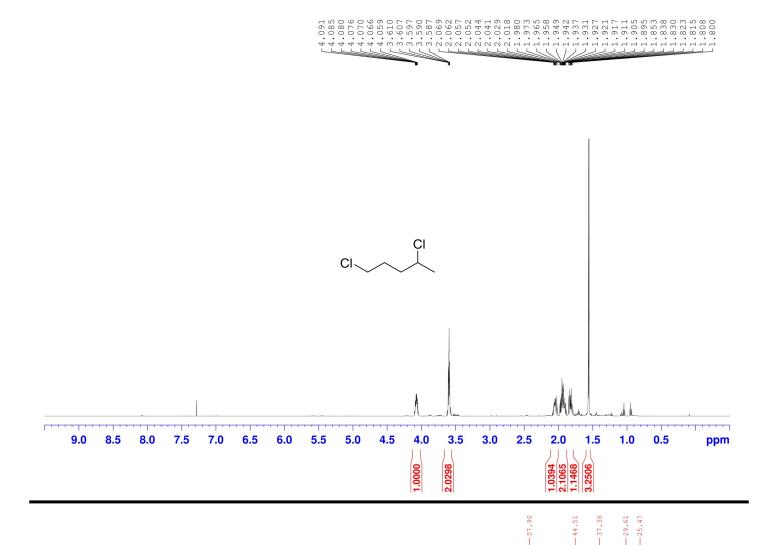


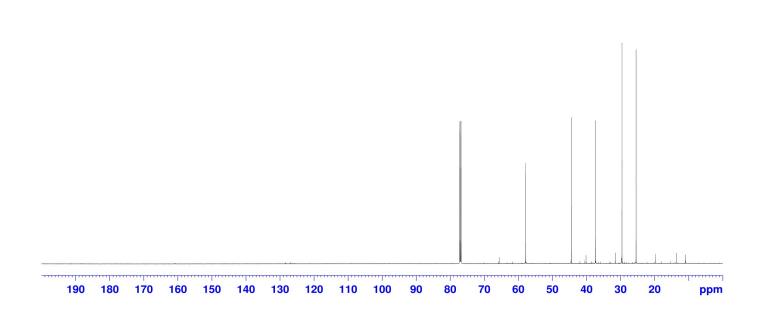


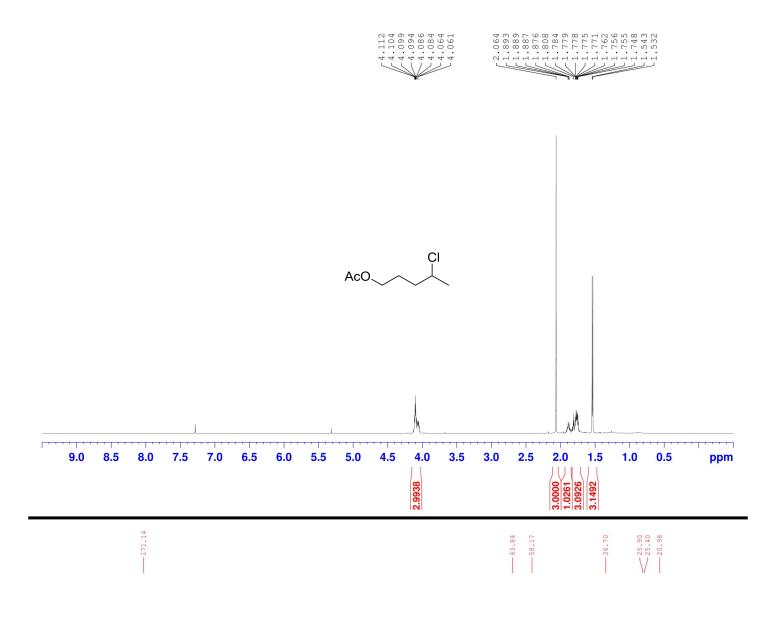


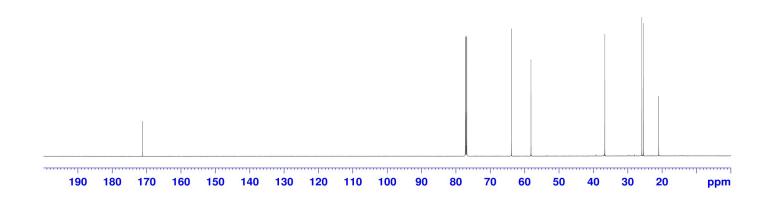




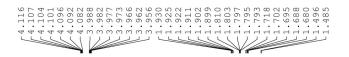


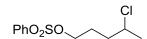


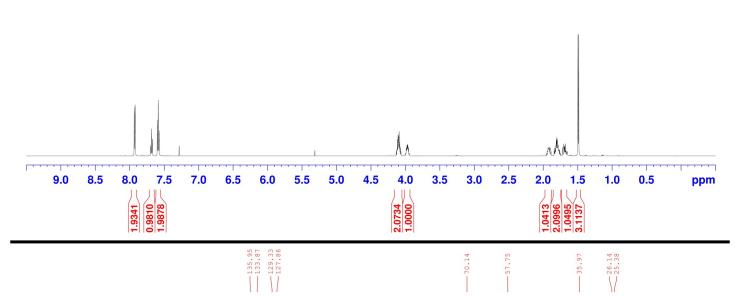


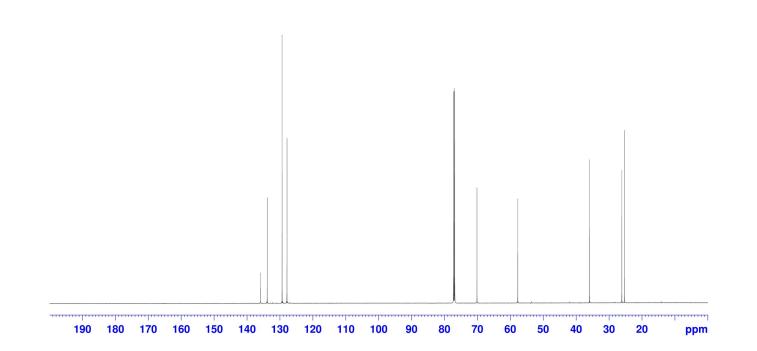


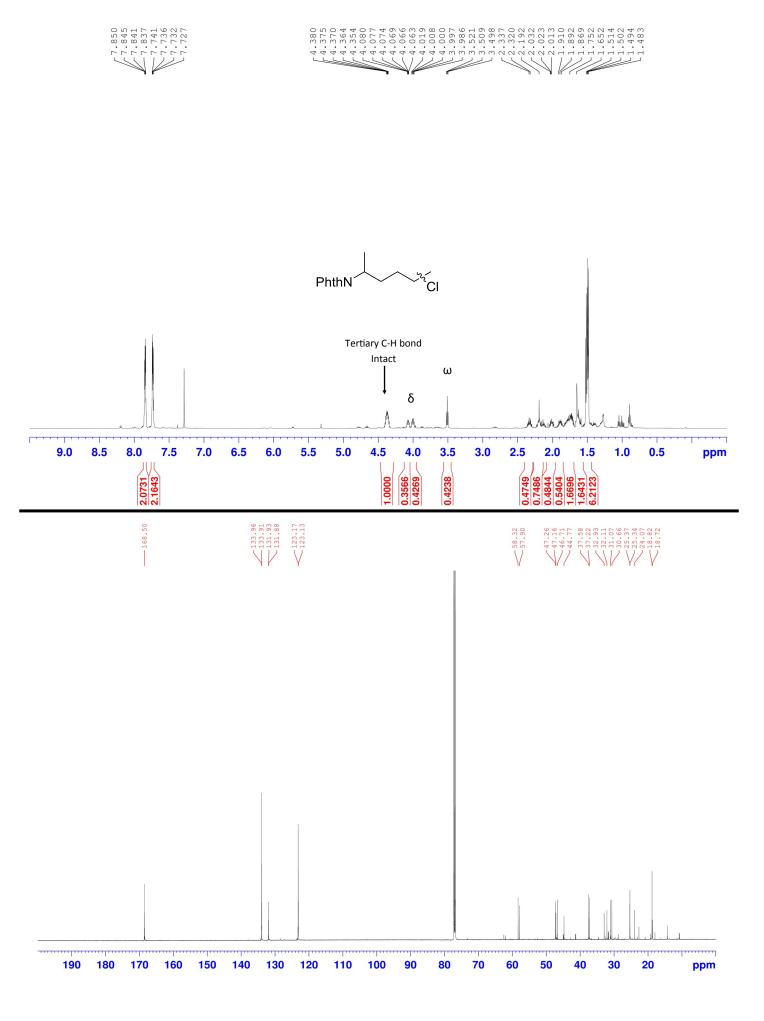


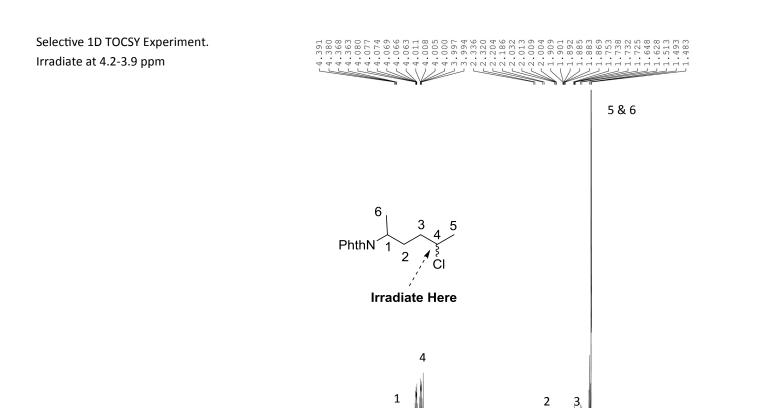












9.0

8.5

8.0

7.5

7.0

6.5

6.0

5.5

5.0

4.5

4.0

3.5

3.0

2.5

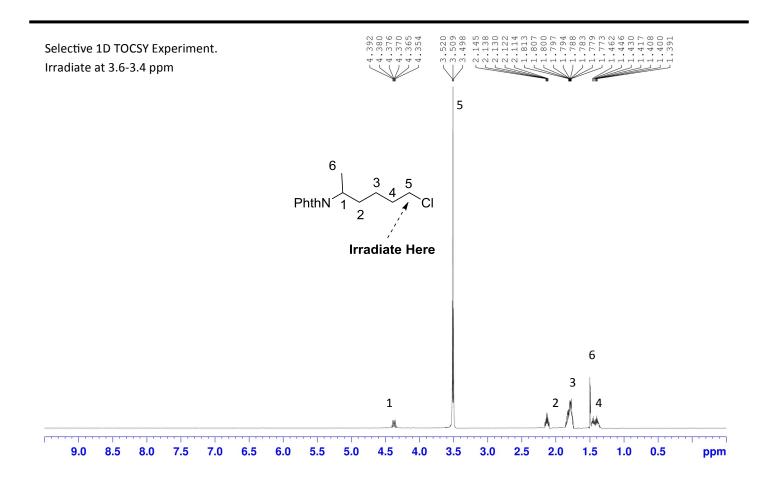
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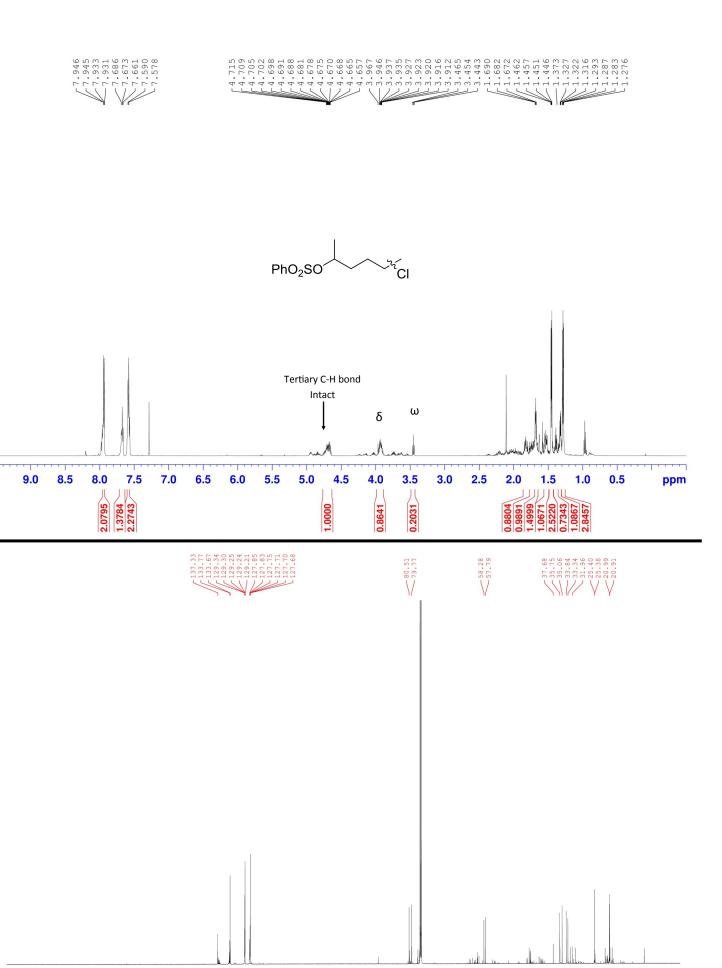
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1.0

0.5

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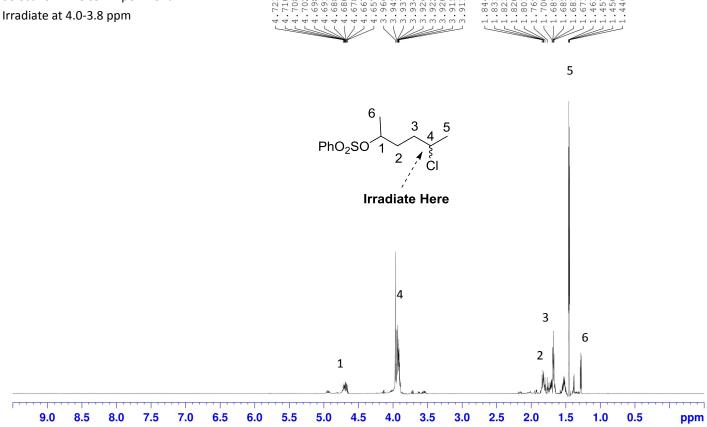


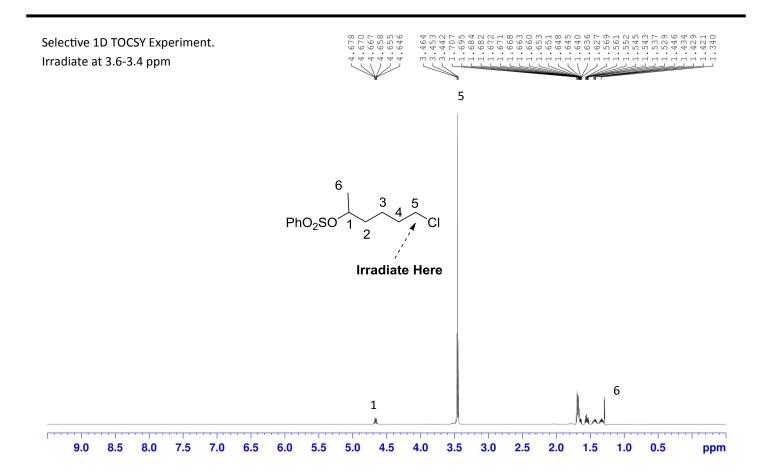


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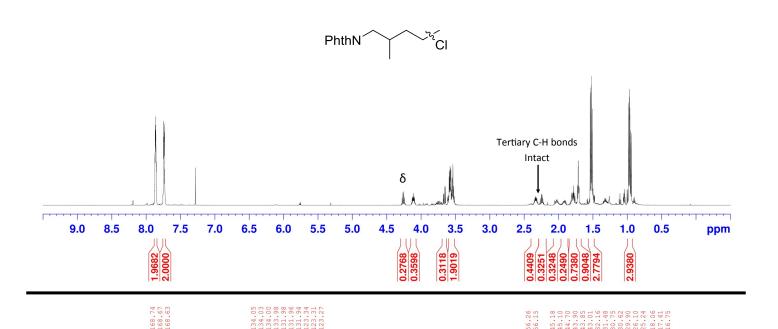
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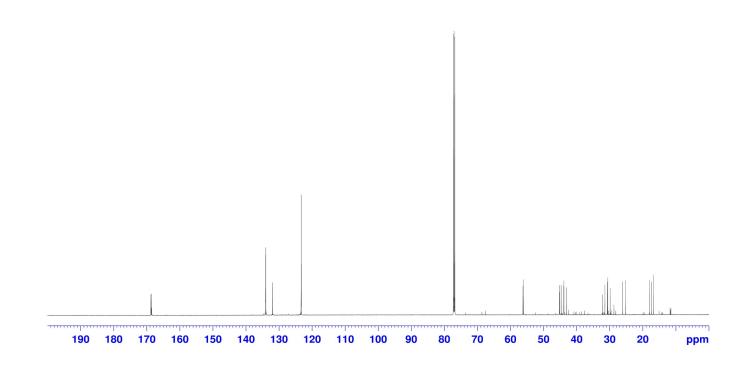
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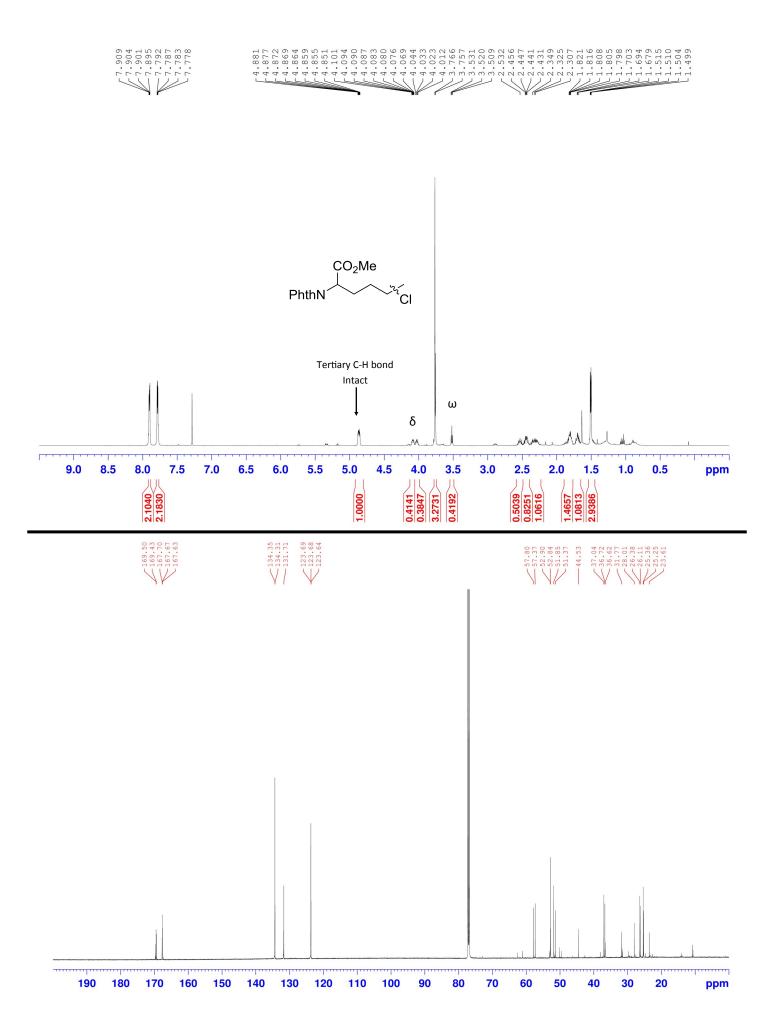


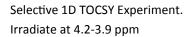




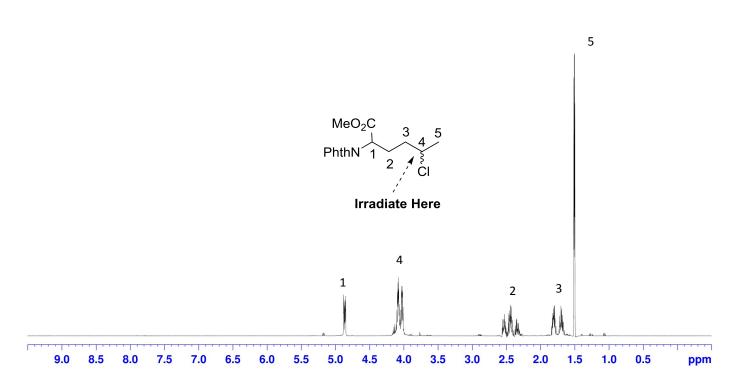


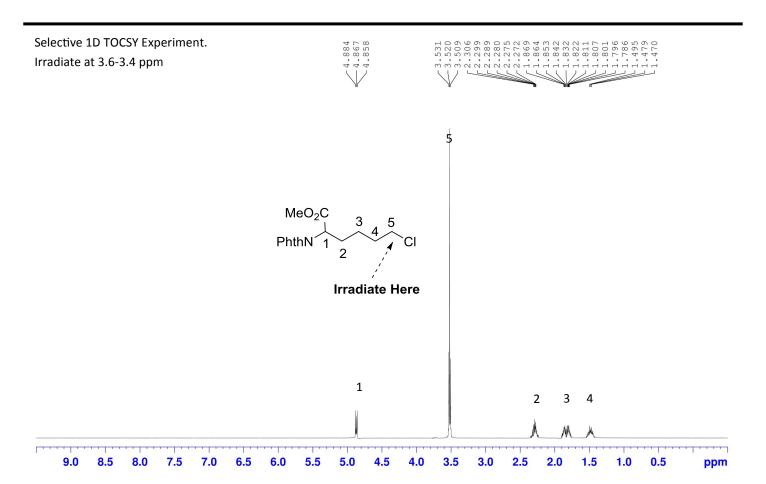


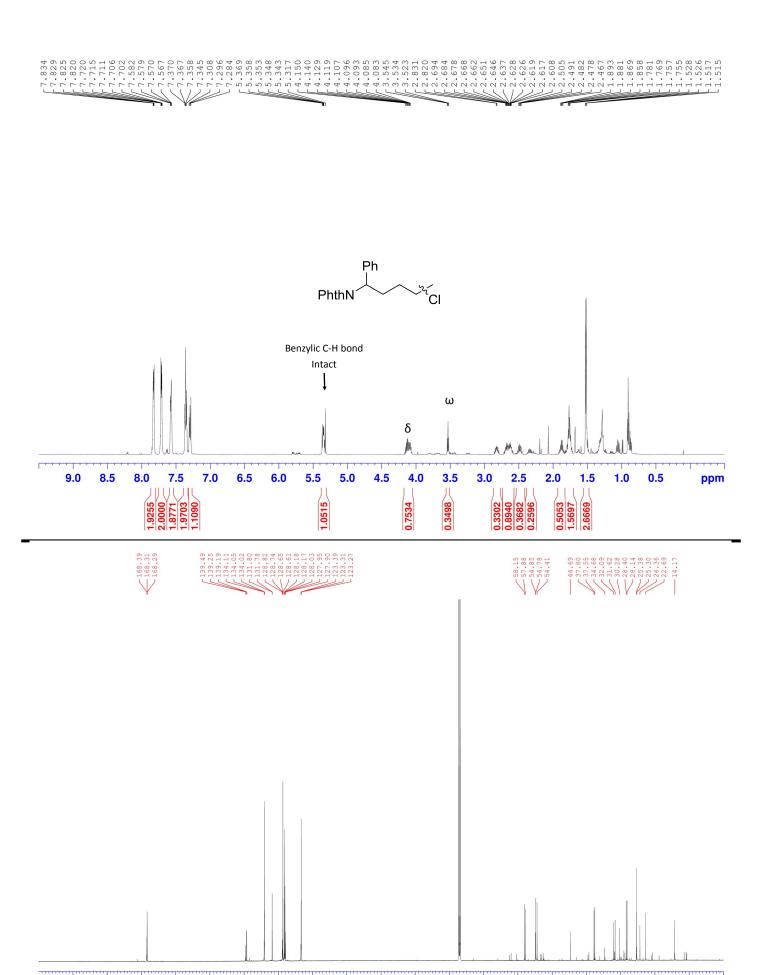






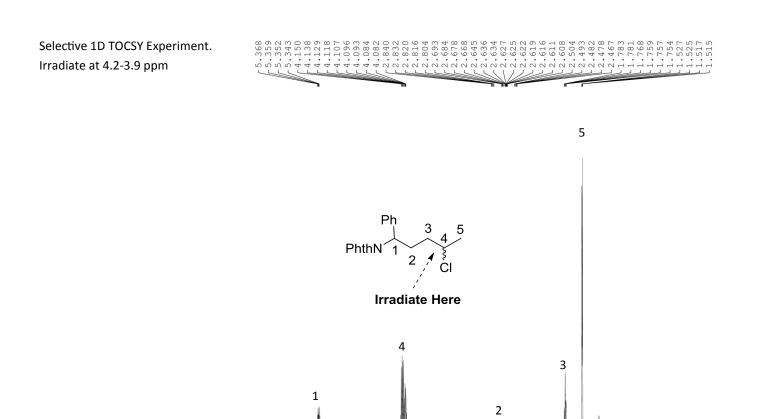


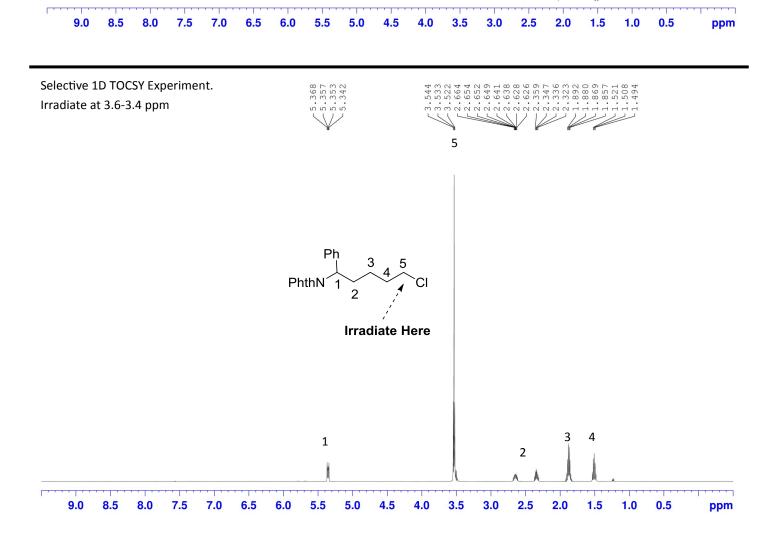


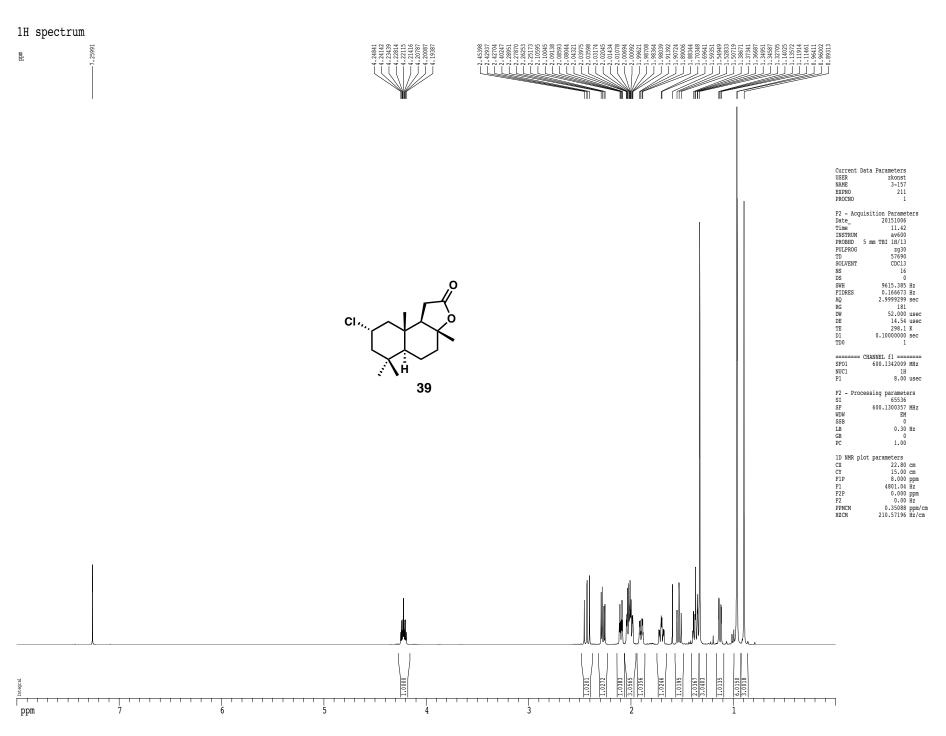


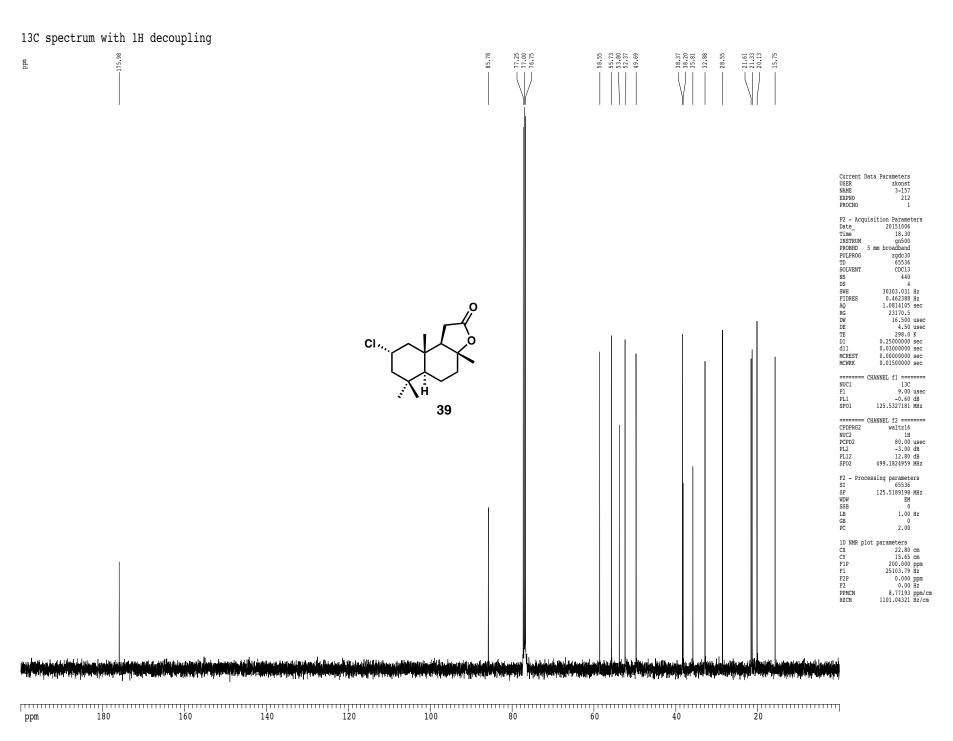
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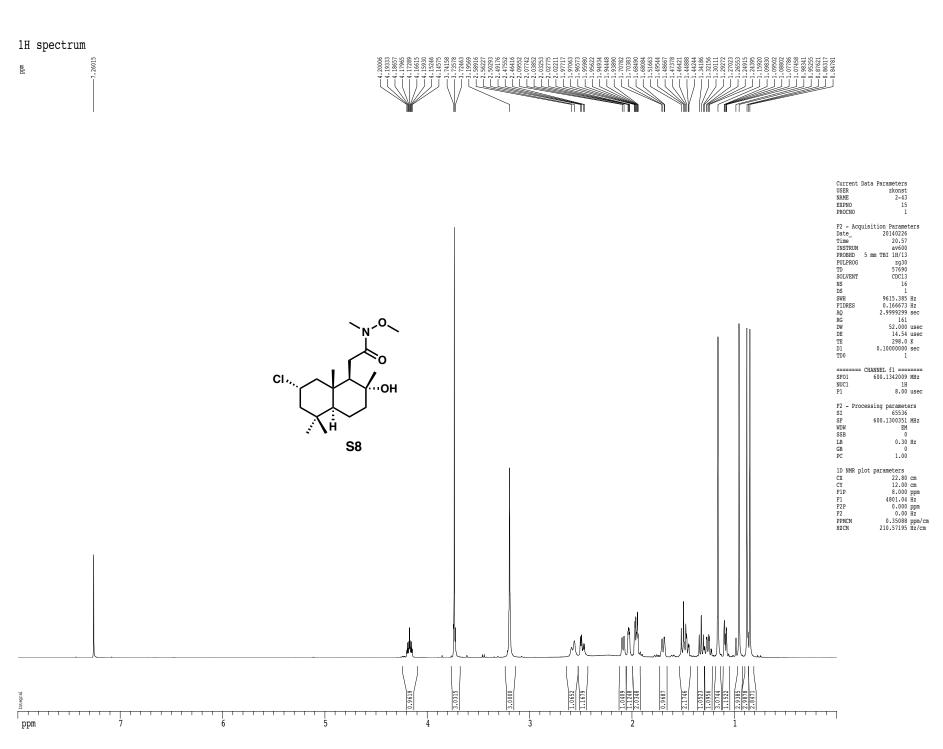
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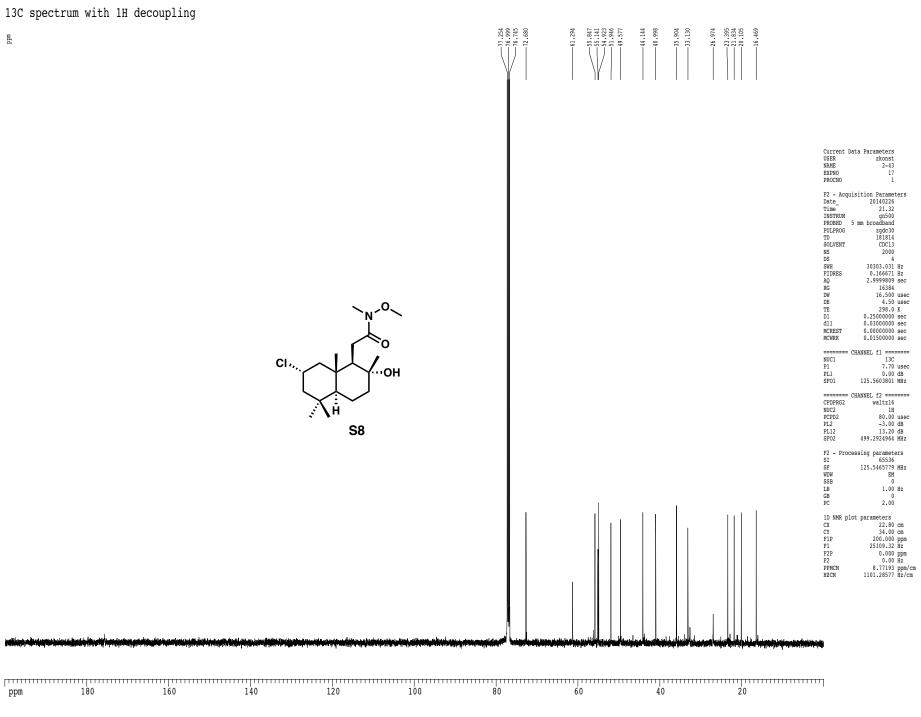


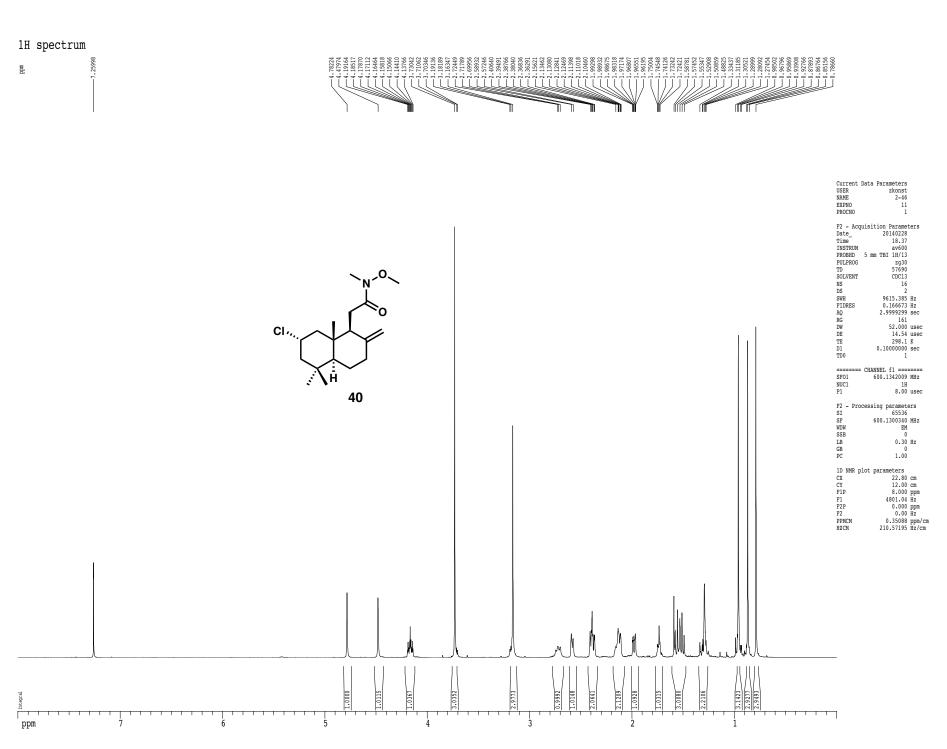


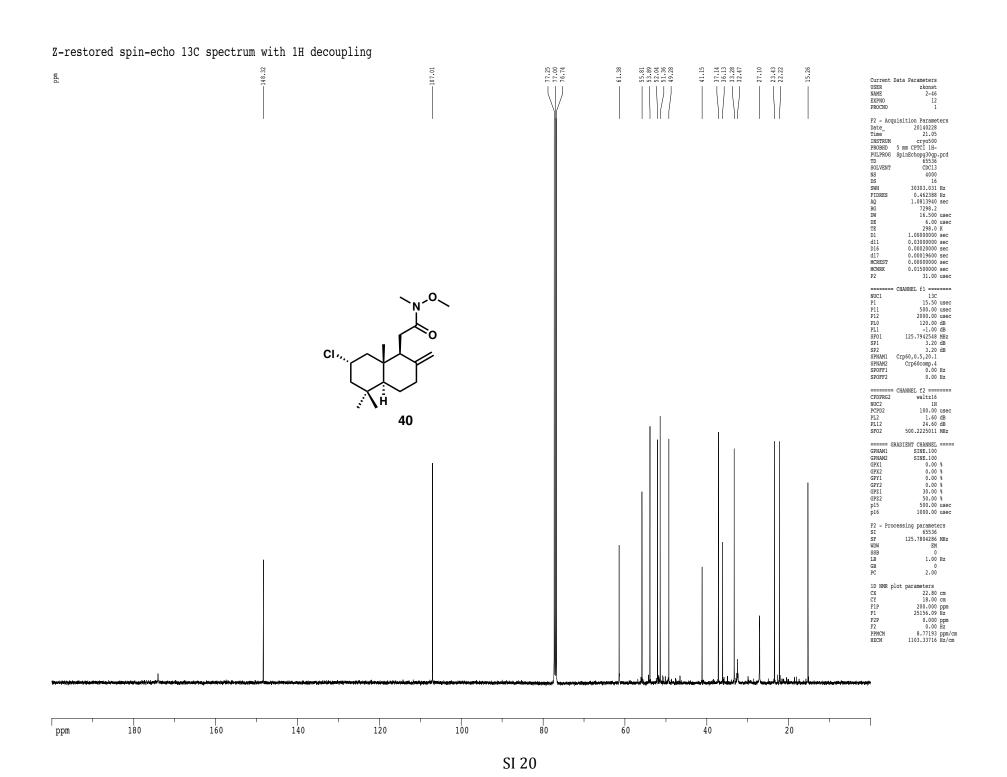


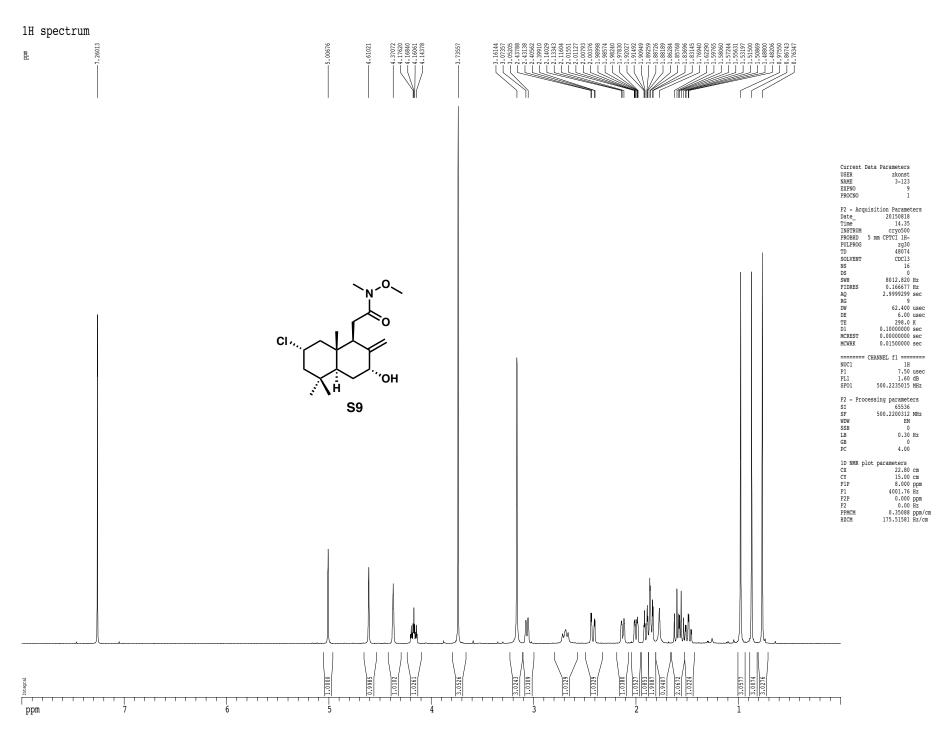


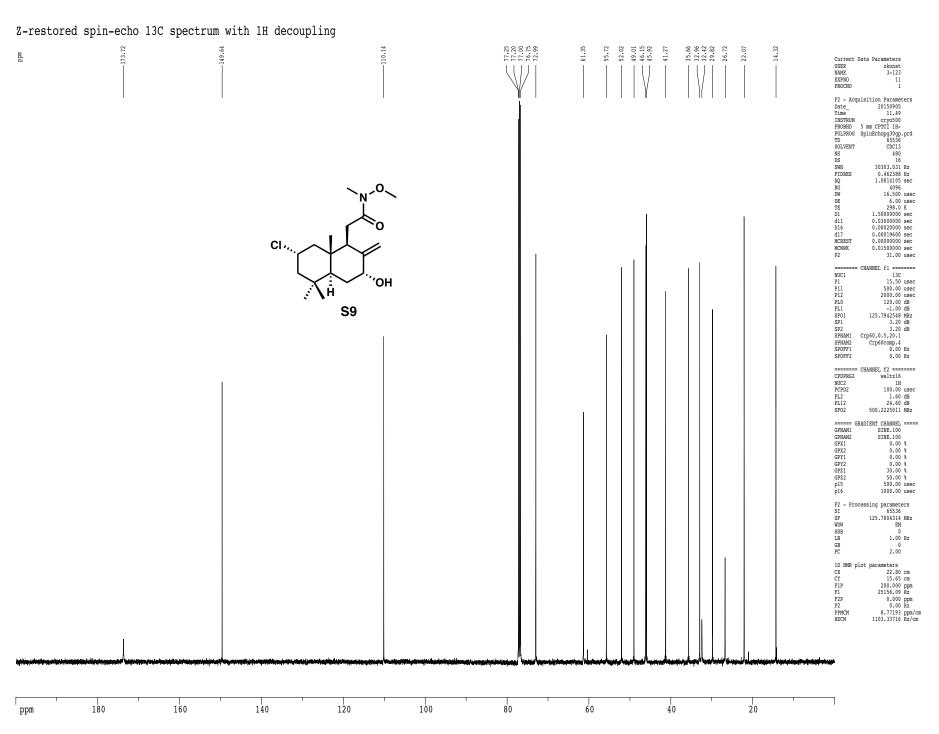


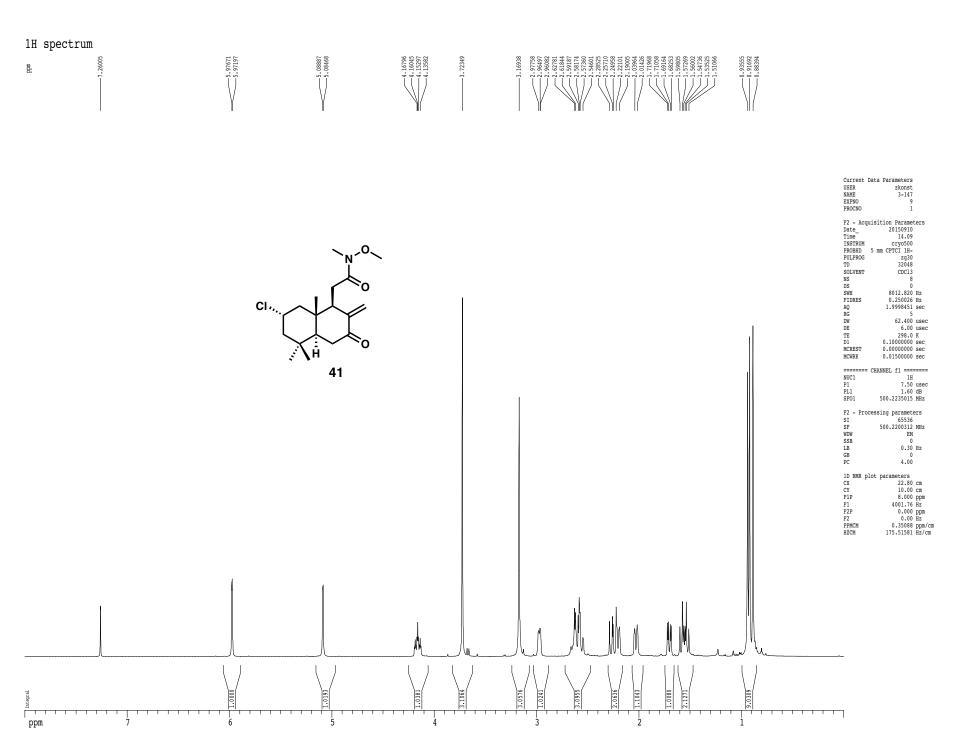


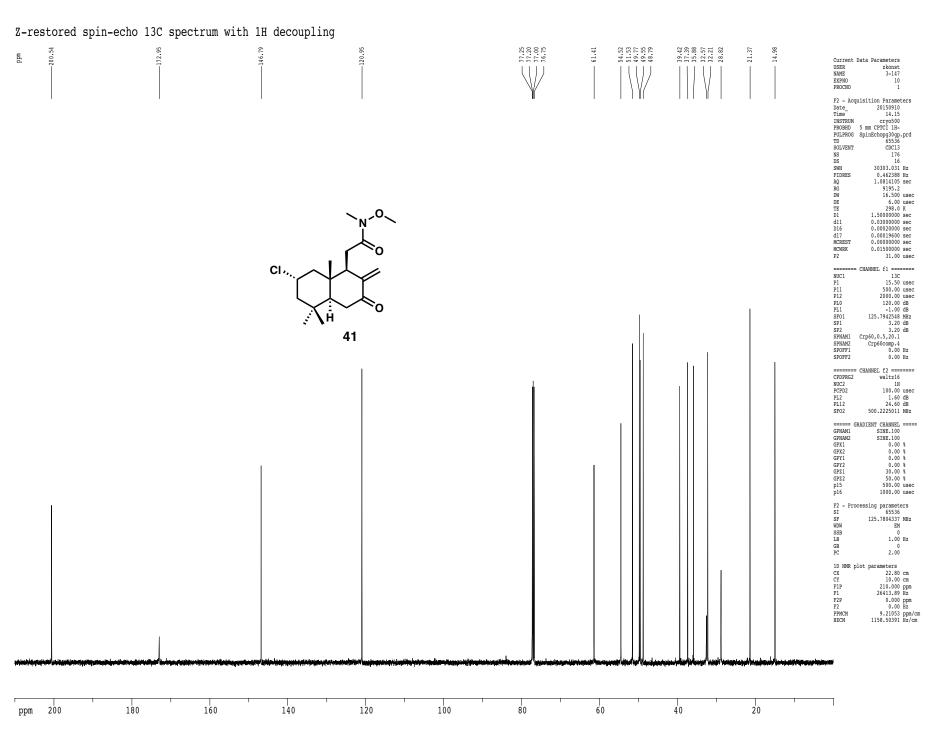


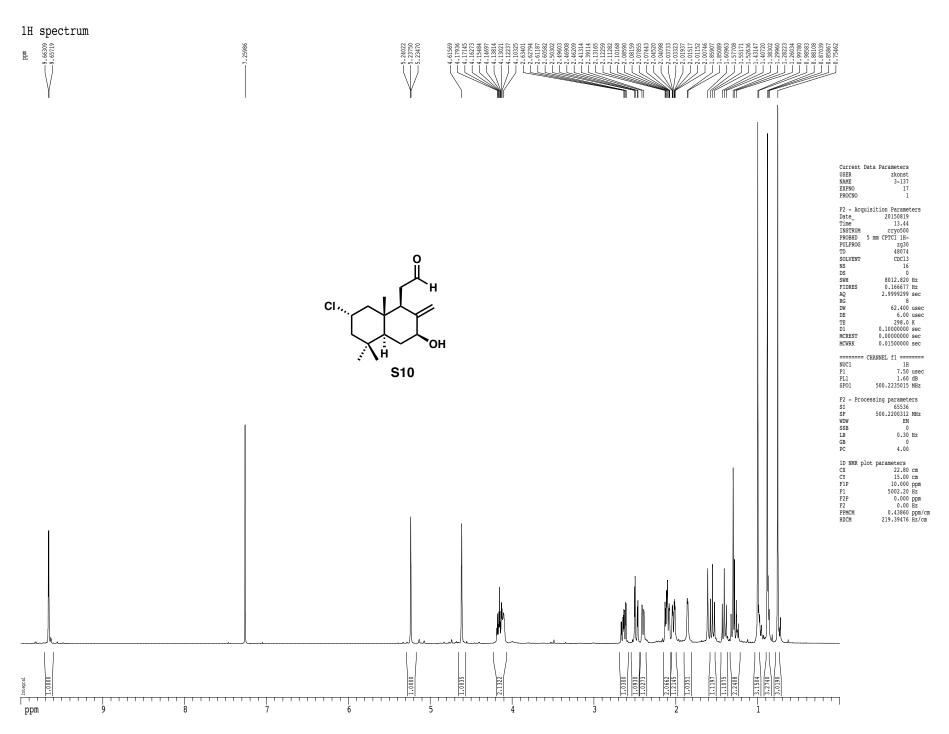


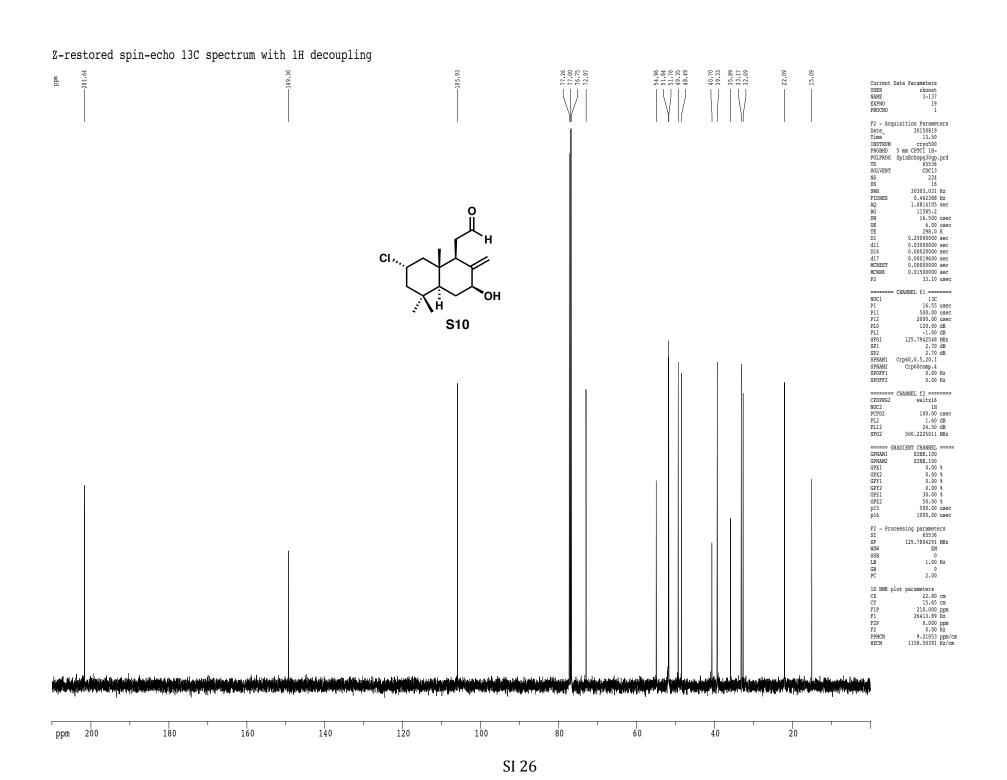


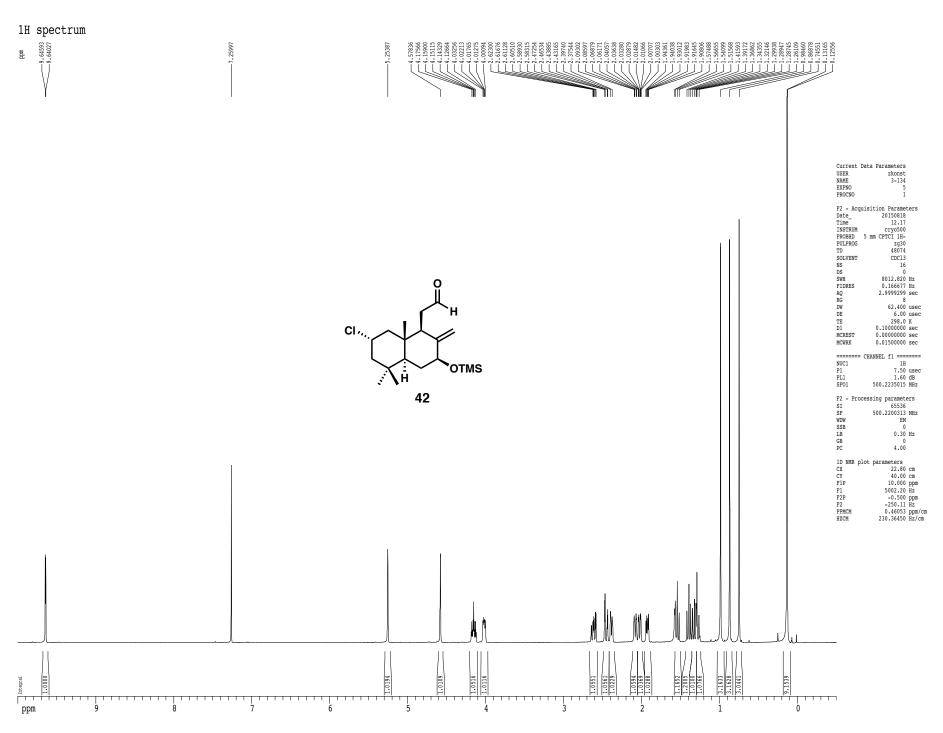


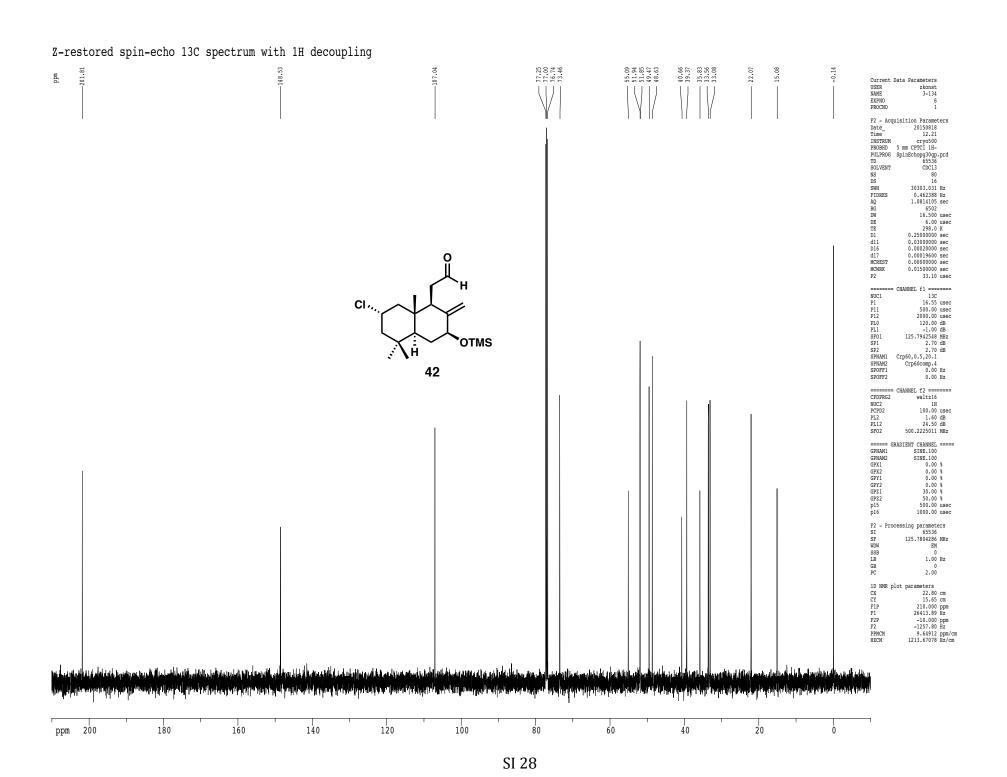


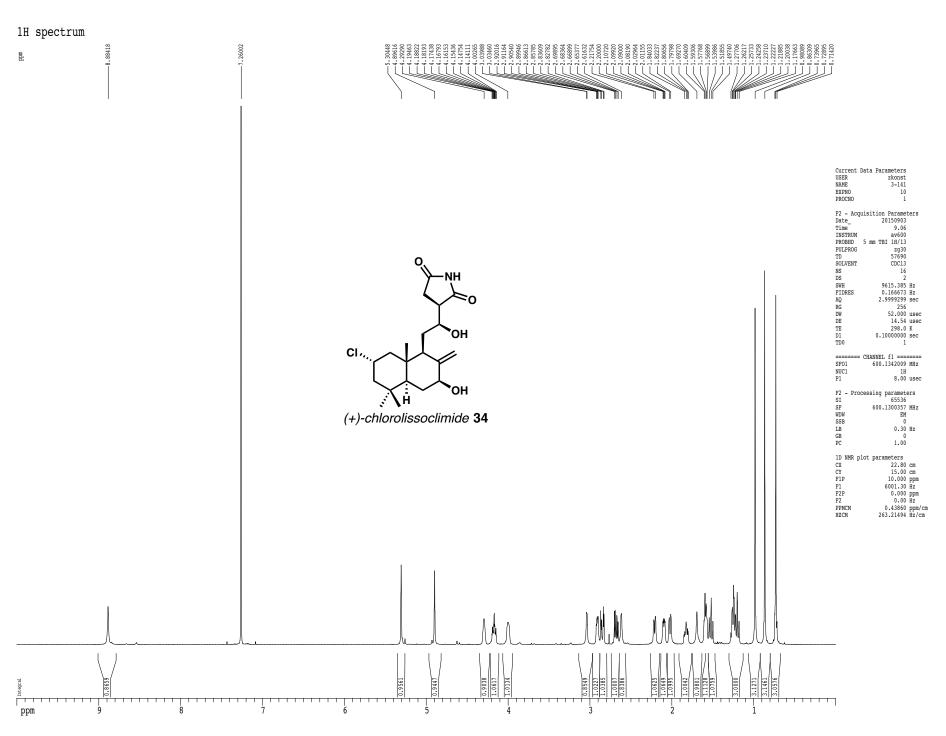


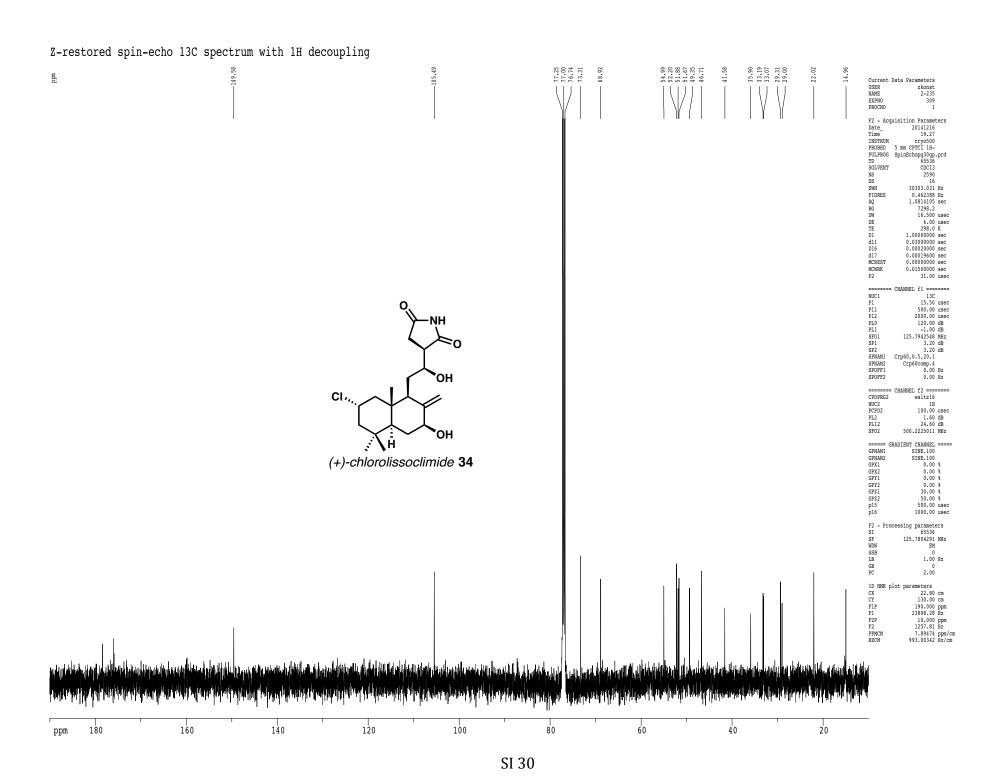


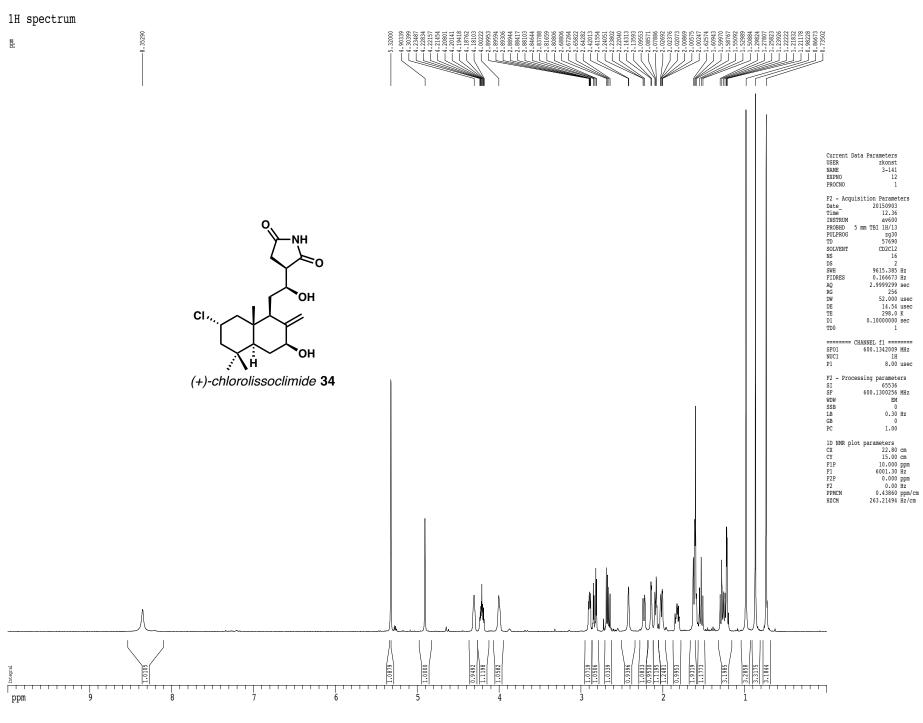


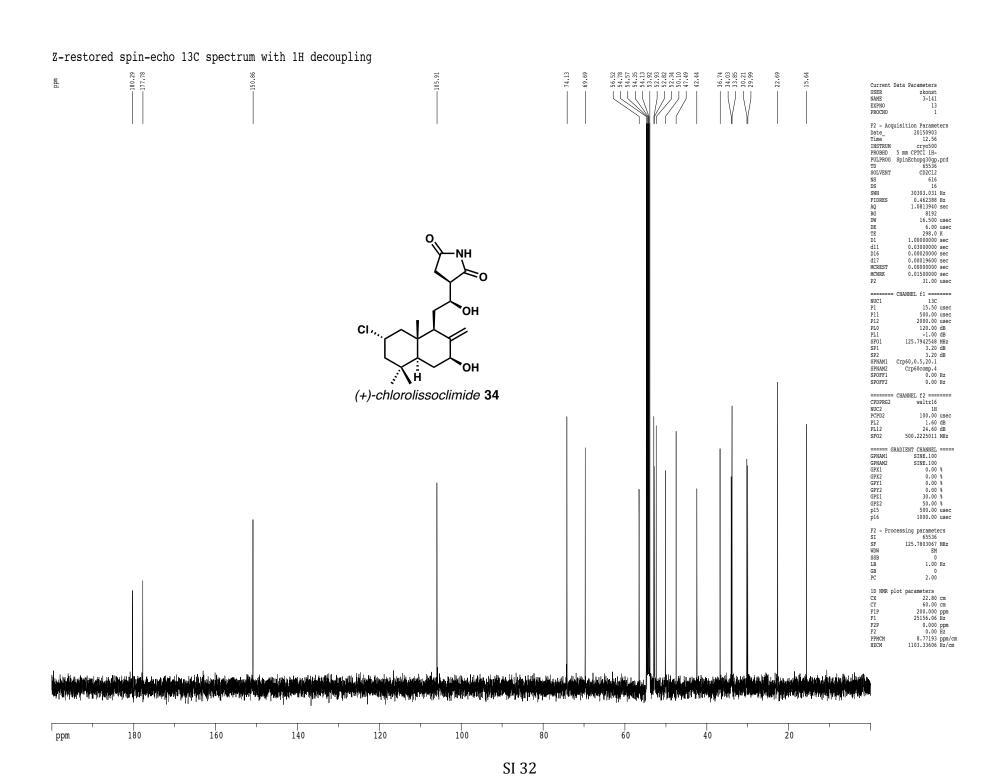


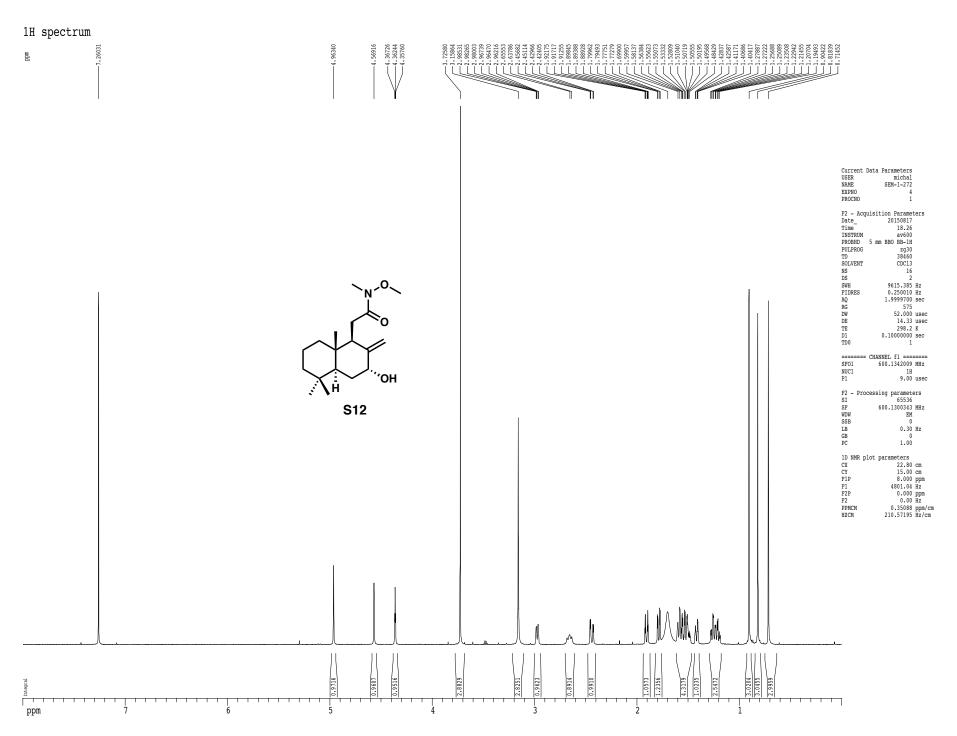


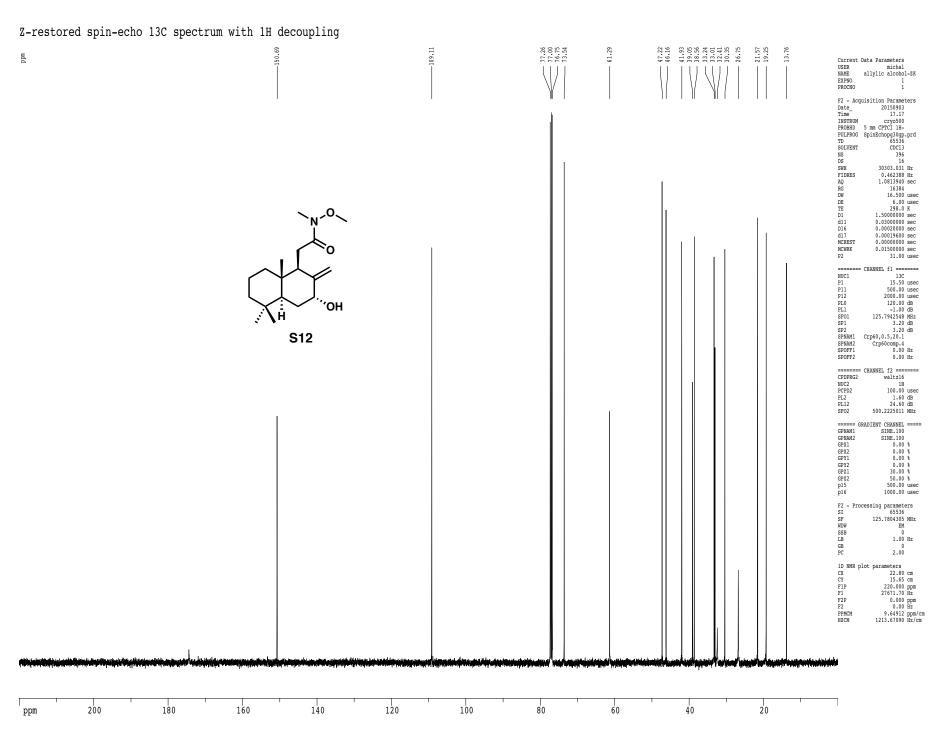




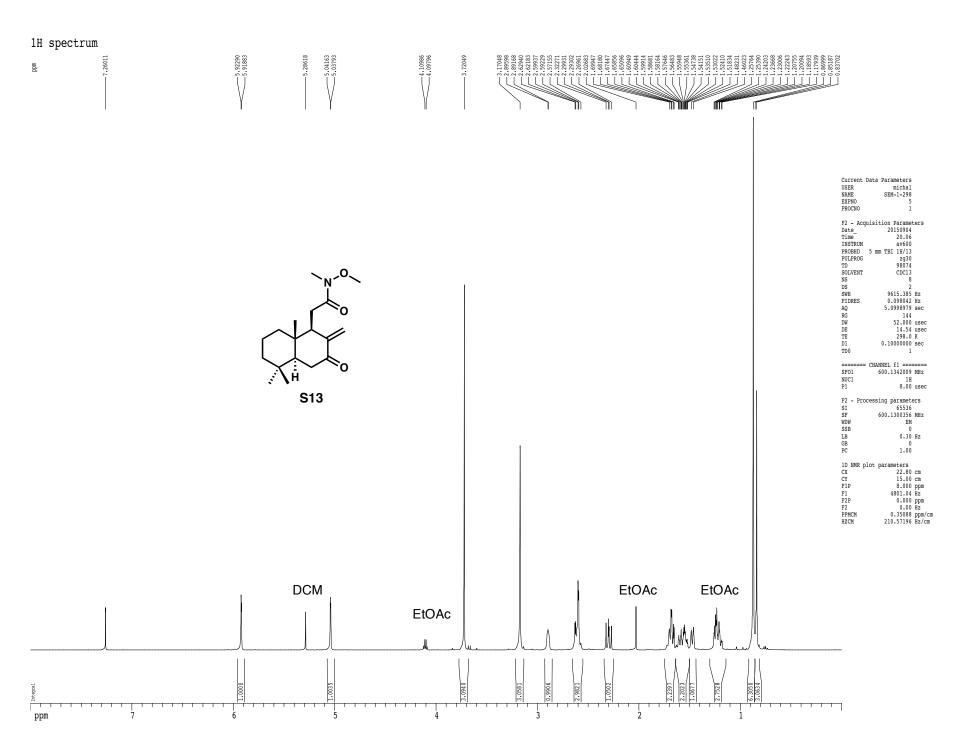


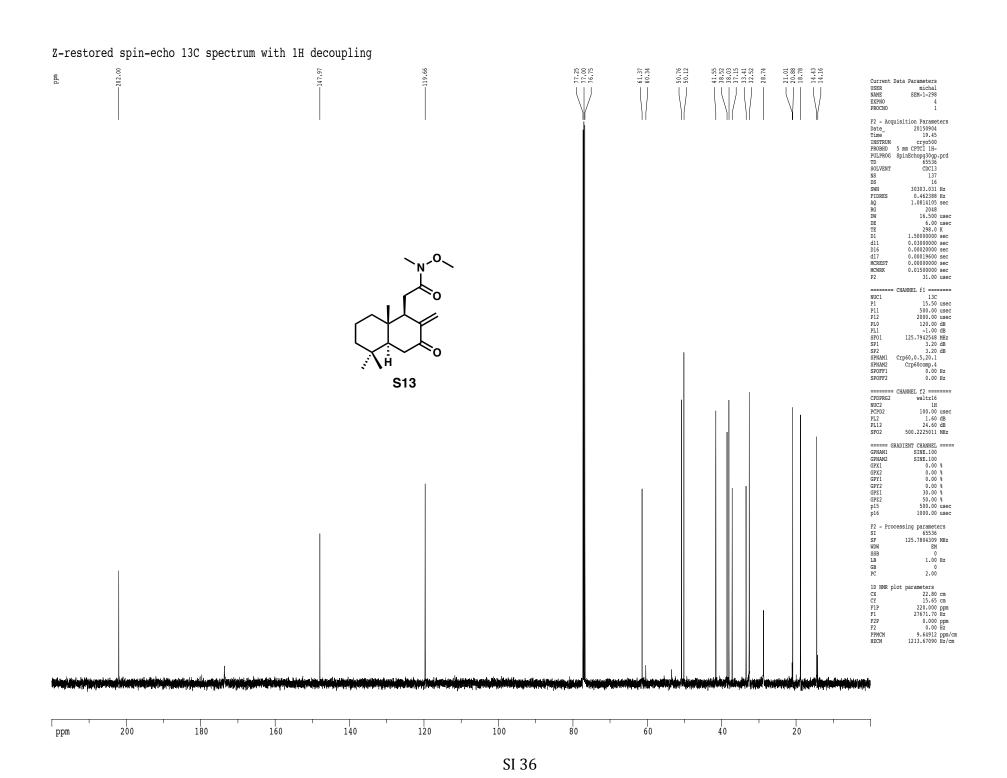


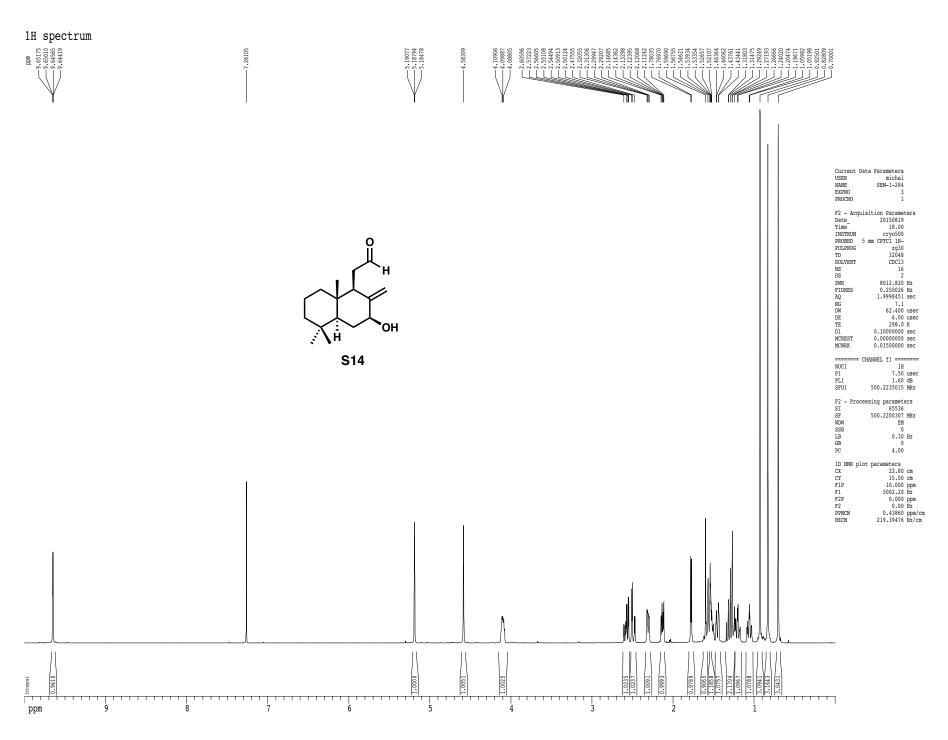


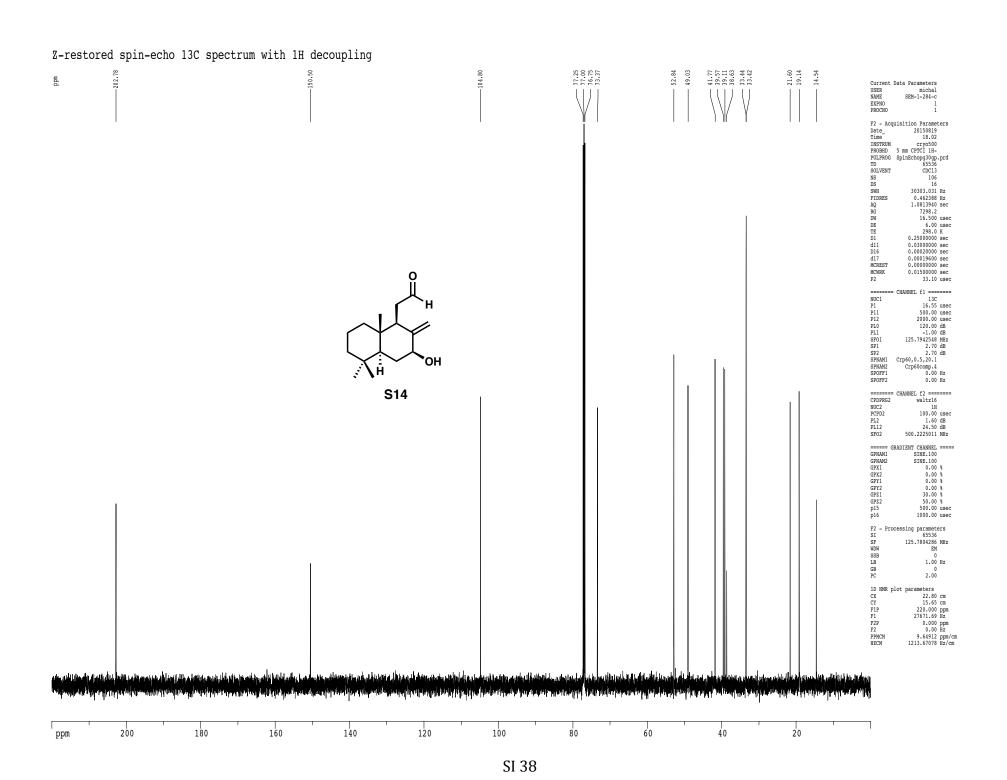


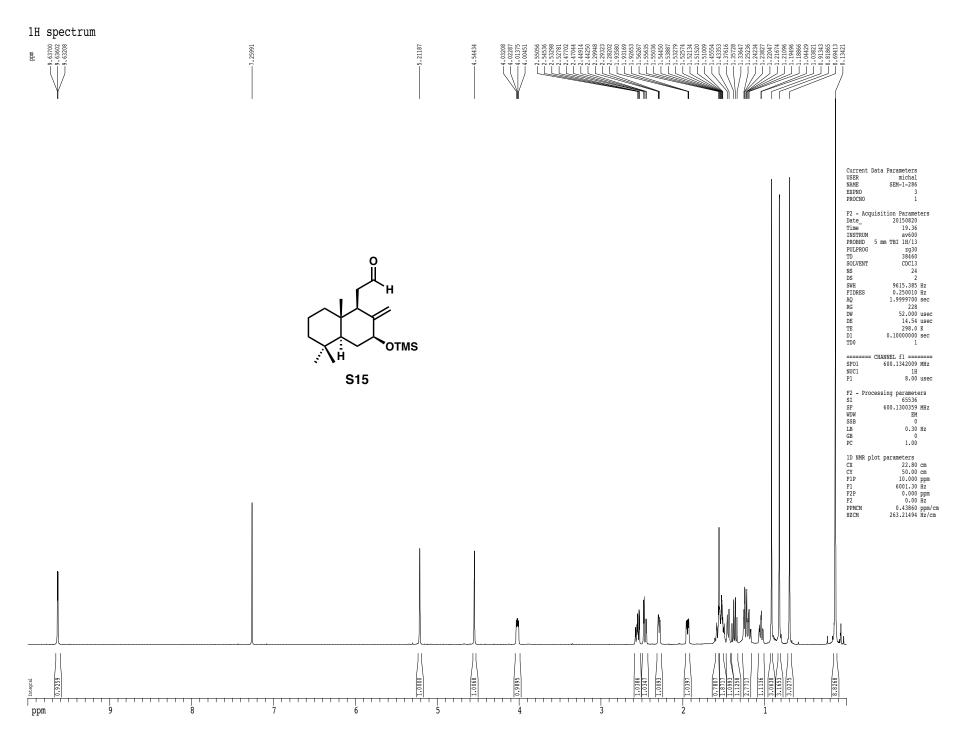
SI 34

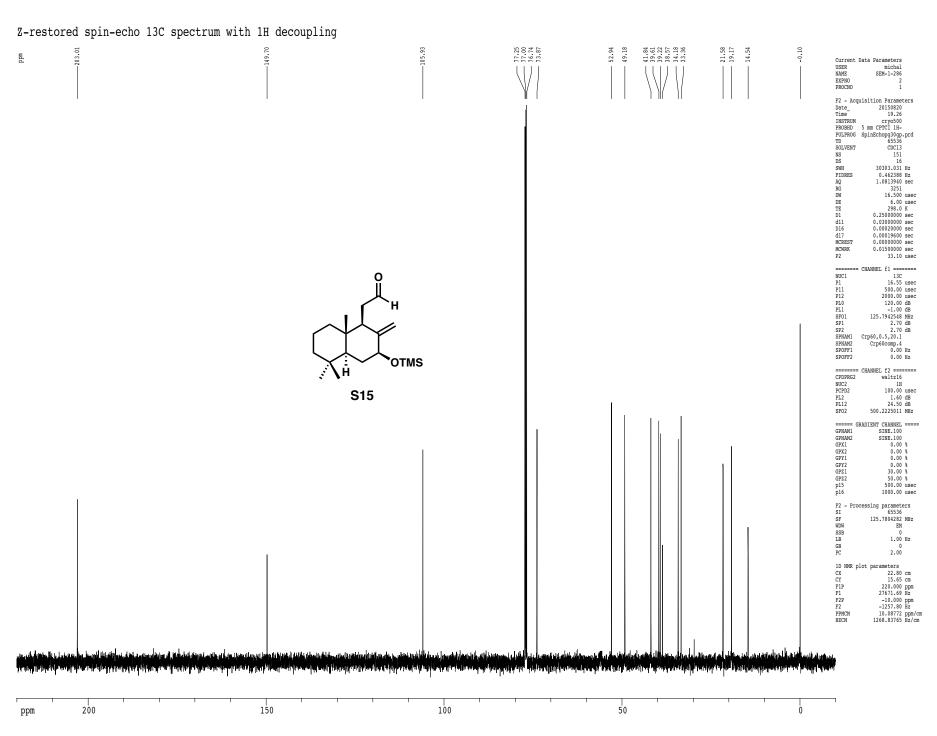


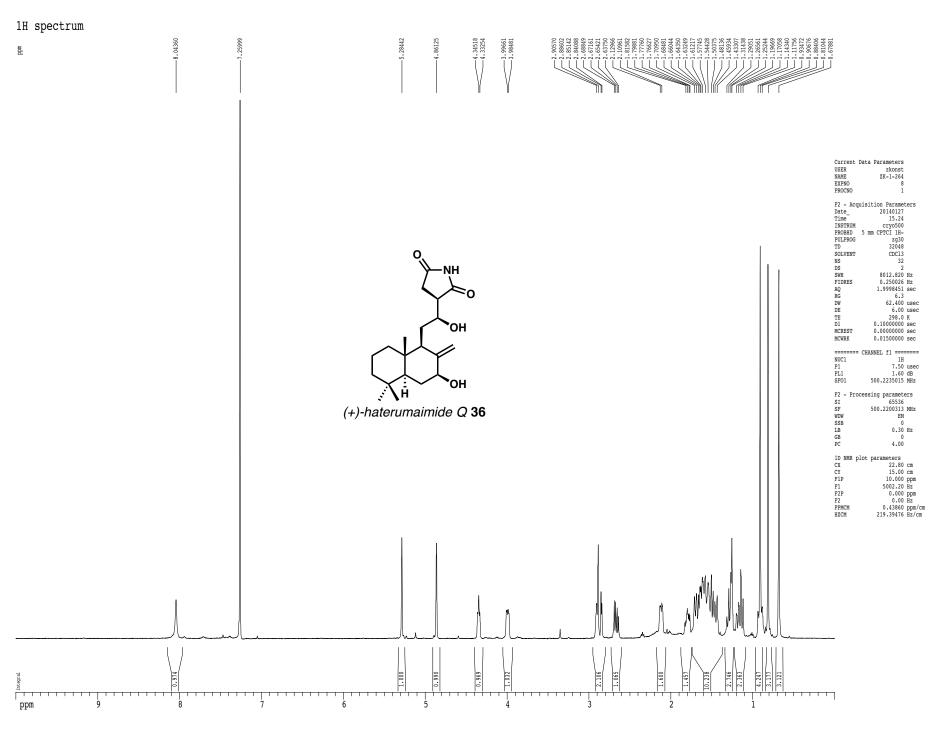


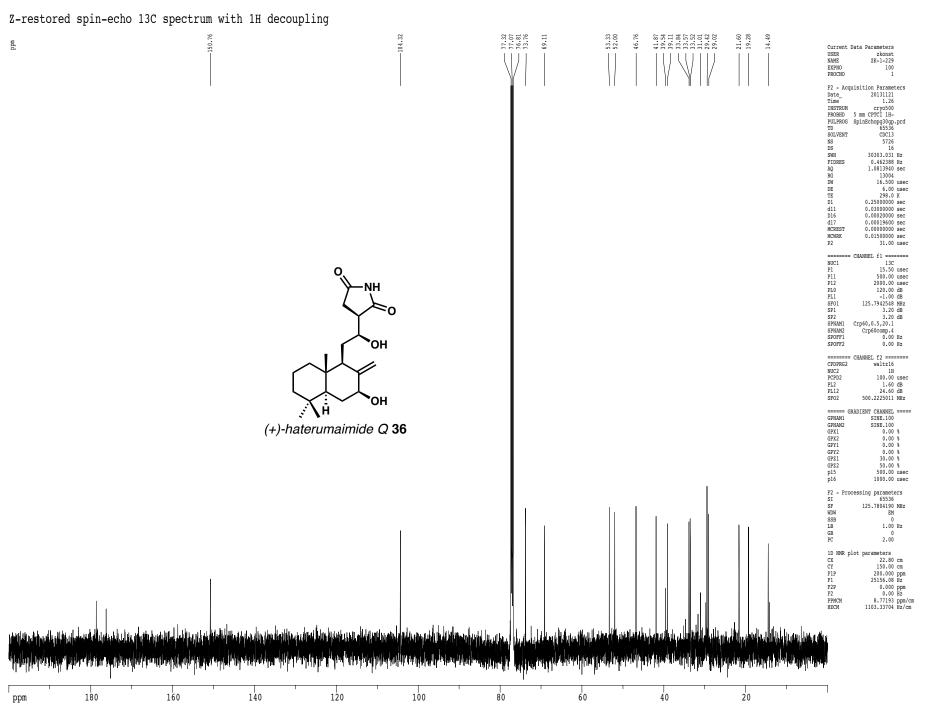




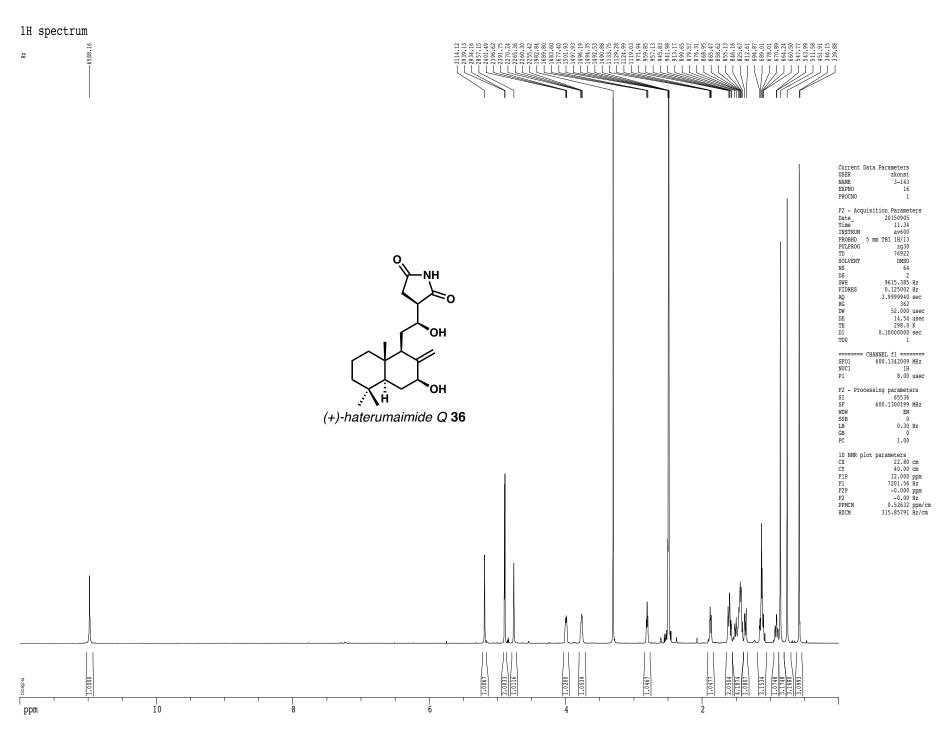


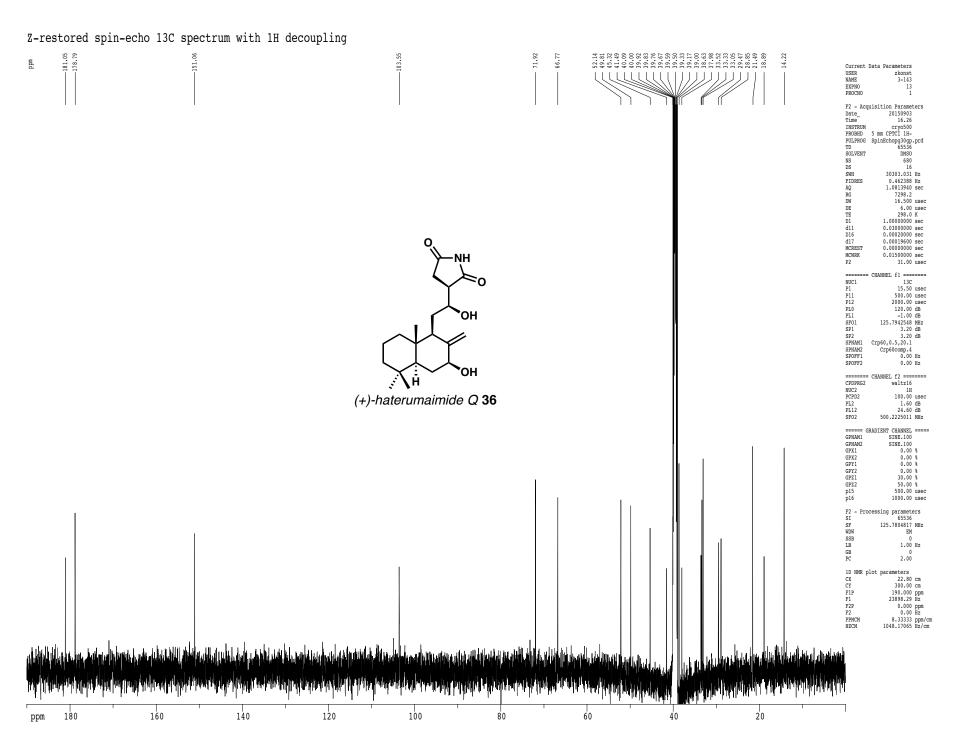




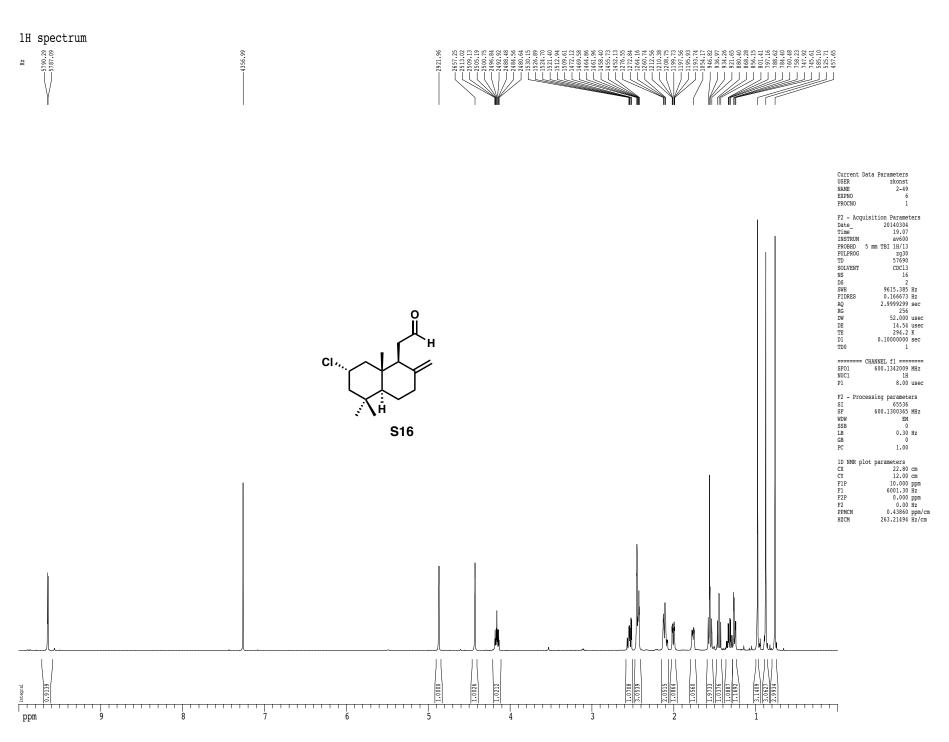


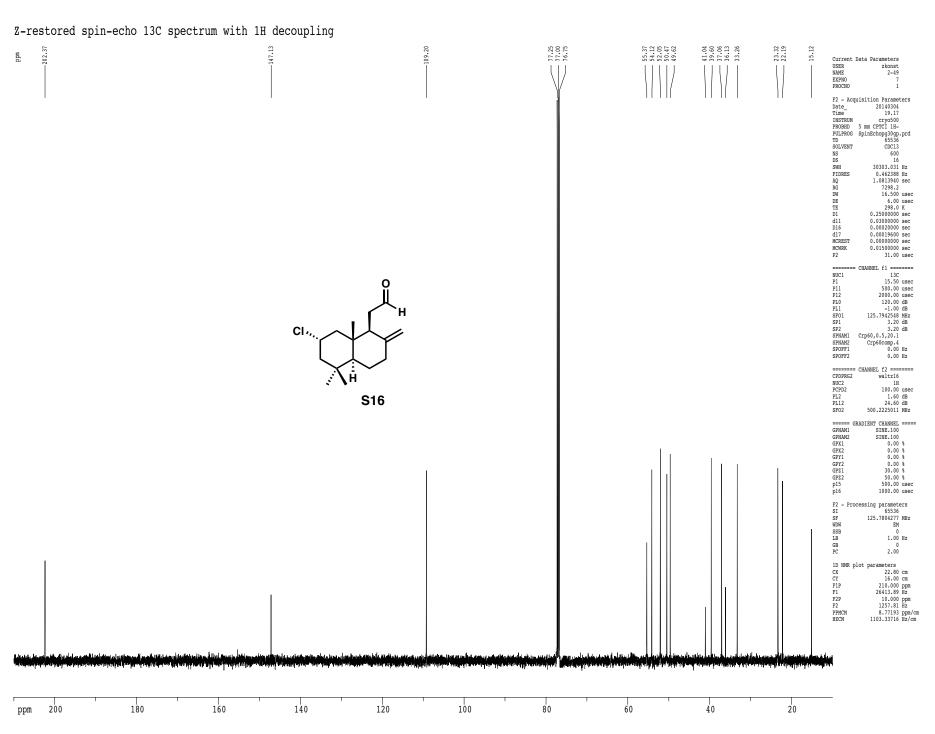
SI 42





SI 44





SI 46

