

Electronic Supporting Information: Adsorption of milk proteins (β -casein and β -lactoglobulin) and BSA onto hydrophobic surfaces

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Materials and Methods

In the experimental part, we used several buffered solutions. The details of the composition of the buffers are presented below:

Table S1. Composition of buffered solutions

pH	Component	Salt concentration (mM)	Ionic strength (mM)
4	AcH	13.50	2.01
5	AcH	3.15	2.00
6	NaH ₂ PO ₄	1.79	2.00
7	NaH ₂ PO ₄	1.13	2.01
8	H ₃ BO ₃	15.00	0.82
9	H ₃ BO ₃	5.14	1.90
10	H ₃ BO ₃	2.20	1.98

All solutions were prepared with double distilled and deionized (Milli-Q) water. The desired pH was achieved by addition of few droplets of a NaOH solution (this component is taken into account for the calculation of ionic strength). In the case of the buffer for preparing β -casein solution, we used 10 mM of Bis-Tris and the pH was adjusted with dilute HCl (2.40 mM of ionic strength).

Results

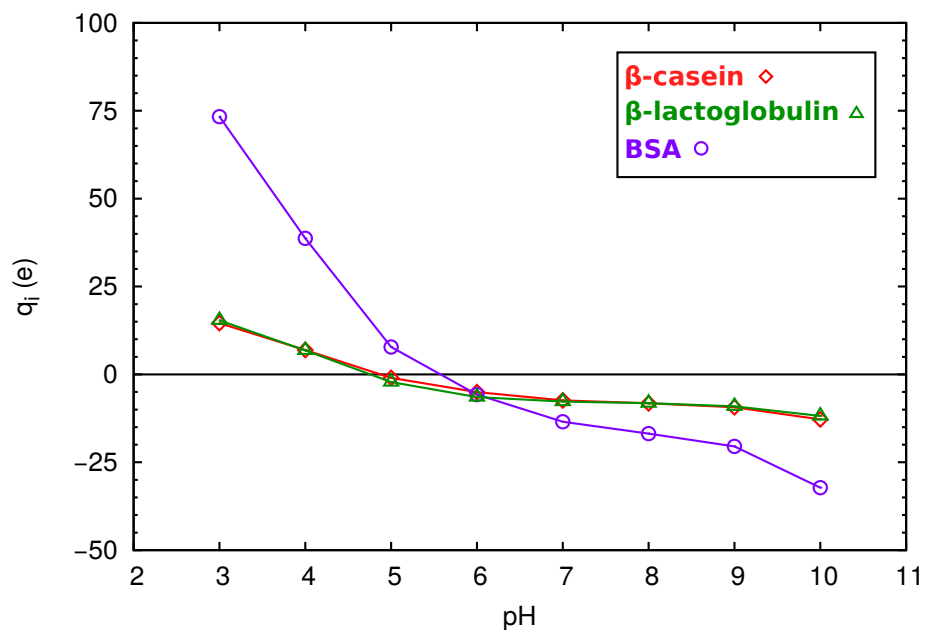


Figure S1. Charges (in units of electronic charge) for each protein molecule as a function of pH. The electrostatic charge of the proteins has been calculated from the 3D structures equilibrated in implicit water (PropKa calculation).

Table S2. Secondary structure (β -sheet and α -helix content) of β -casein protein obtained in MD simulations in bulk or adsorbed onto surfaces of different charge and different pH

Simulation	β -sheet	α -helix
Bulk (pH=7)	77%	8%
Neutral surface (pH=7)	40%	43%
Negative surface (pH=7)	37%	44%
Negative surface (pH=4)	36%	45%

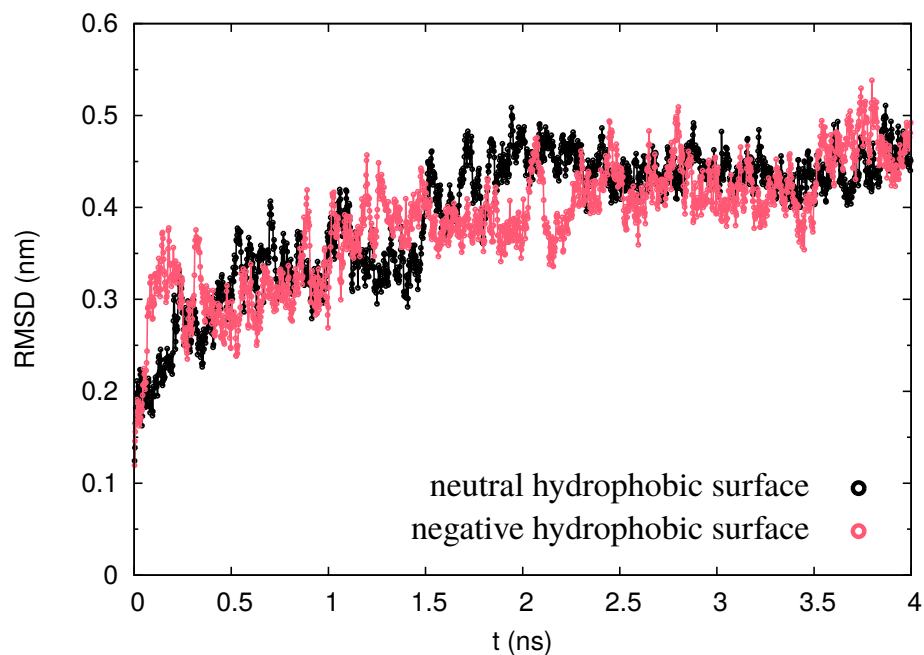


Figure S2. Time evolution of the root mean squared deviation (RMSD) between the β -casein structure in solution and adsorbed onto a neutral hydrophobic surface or a negative hydrophobic surface.

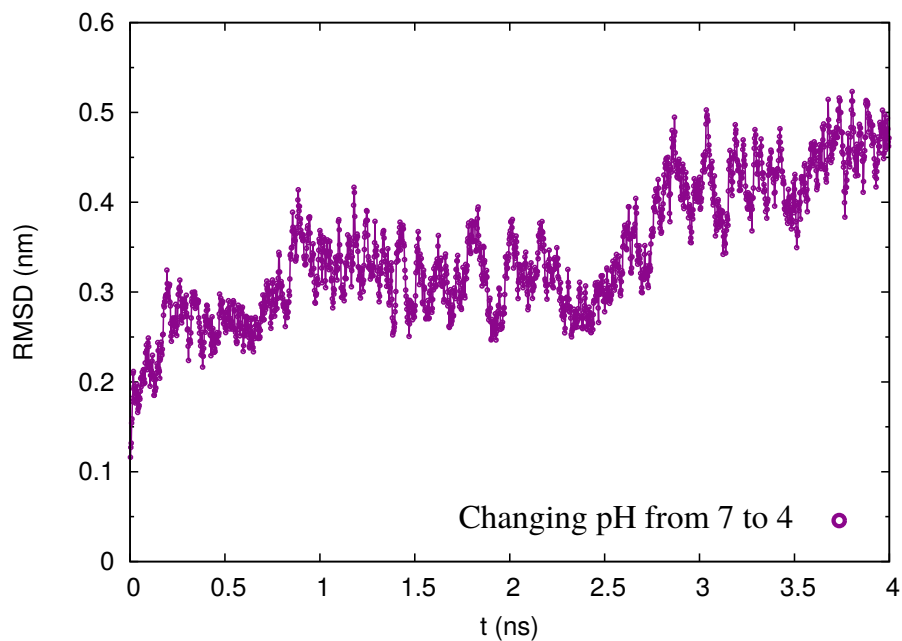


Figure S3. Time evolution of the root mean squared deviation (RMSD) of the β -casein structure adsorbed onto a negative hydrophobic surface from pH 7 to pH 4.

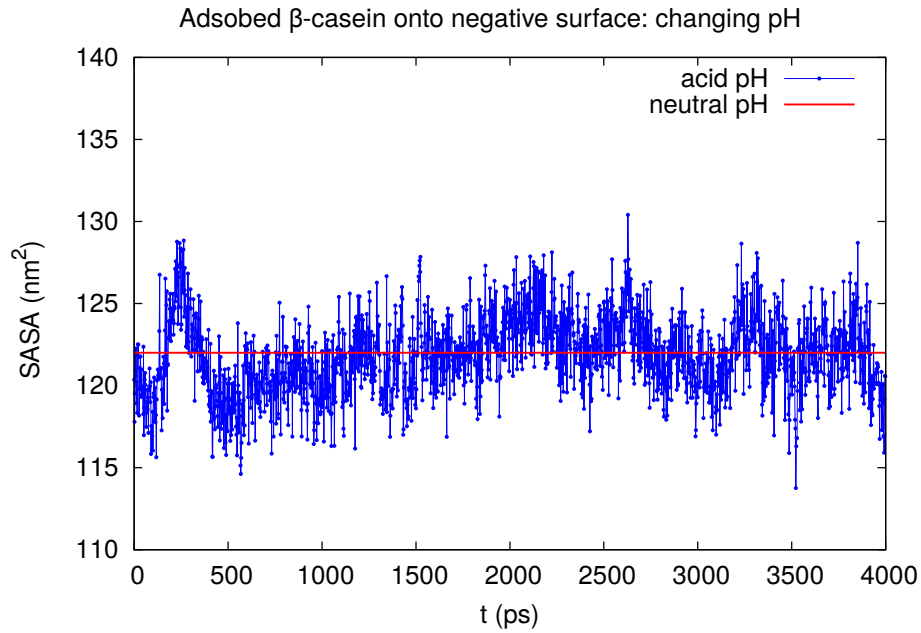


Figure S4. Time evolution of the solvent-accessible surface area (SASA) between the β -casein structure adsorbed onto a negative hydrophobic surface from pH 7 to pH 4.

Table S3. Values of critical charge density (σ_{crit}) using Manning model, eqn (11), and Bocquet *et al.* model, eqn (12), calculated for spherical colloids ($a_{anionic}=80$ nm) and ($a_{cationic}=255$ nm) immersed in different buffered solutions

pH	Ionic concentration (mM)	Anionic latex		Cationic latex	
		Manning $ \sigma_{crit} $ (e/nm ²)	Bocquet <i>et al.</i> $ \sigma_{crit} $ (e/nm ²)	Manning $ \sigma_{crit} $ (e/nm ²)	Bocquet <i>et al.</i> $ \sigma_{crit} $ (e/nm ²)
3	1.00	0.068	0.052	0.063	0.048
4	2.01	0.080	0.071	0.076	0.067
5	2.00	0.080	0.071	0.076	0.067
6	1.90	0.080	0.071	0.076	0.067
7	1.57	0.080	0.071	0.076	0.067
8	0.82	0.064	0.048	0.059	0.044
9	1.90	0.079	0.069	0.075	0.066
10	1.98	0.071	0.074	0.076	0.067