Supporting Information

Functionalized AIE nanoparticles with efficient deep-red emission, mitochondria specificity, cancer cell selectivity and multiphoton susceptibility

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Figure S1. HRMS spectrum of TPE-TETRAD.



Figure S2. ¹H NMR spectrum of TPE-TETRAD. ¹H NMR (400 MHz, CDCl₃), δ (ppm): 7.56–7.43 (m, 20H), 7.39–7.34 (m, 16H), 7.33–7.0 (m, 66H), 6.67 (d, J = 16 Hz, 2H; pyran –CH=), 6.63 (s, 2H; pyran H).



¹³C NMR spectrum of TPE-TETRAD. ¹³C NMR (75 MHz, CDCl₃), δ (ppm): 159.5, Figure S3. 150.0, 145.9, 143.8, 142.7, 141.2, 140.5, 139.1, 136.2, 131.6, 131.2, 128.7, 127.7, 126.2, 125.7, 122.1, 116.5, 115.2, 106.4, 53.4.

Table S1. Optical Properties of TPA-DCM and TPE-TETRAD							
	λ _{em} (nm) ^c						
	λ _{ab} (nm) ^a	E _g (eV) ^b	solution	aggregate	Stokes shift ^d		
TPA-DCM TPE-TETRAD	465 500	2.72 2.56	620 668	670 675	155 168		

^{a)}Absorption maxima (λ_{ab}) in THF(10 μ M); ^{b)}HOMO-LUMO band gap (E_g) derived from theoretical DFT calculations; c)emission maxima (λ_{em}) for solution (in THF,10 μ M) and aggregate (in THF/water 1:9 v/v, 10 μ M); d)Stokes shift derived from subtracting λ_{em} from λ_{ab} .



Figure S4. HOMO and LUMO energy levels of TPA-DCM and TPE-TETRAD, respectively. Molecular orbital amplitude plots of HOMO and LUMO energy levels of TPA-DCM and TPE-TETRAD were calculated using density functional theory (DFT) in Gaussian 09 program using B3LYP/6-31G(d) basis set.

	$\begin{array}{c} NC CN \\ P3 P2 P1 \\ P3 P2 P1 \\ O P3 P2 P1 O O P1 O O O P1 O O O O P1 O O O O O O O O$	
Angles between planes		-
P_1-P_2	~0	-
P_2-P_3	~67	
P ₂ -P ₄	~67	
P ₃ -P ₄	~76	

Figure S5. Summary of dihedral angles (°) for TPA-DCM. The dihedral angles between any two phenyl rings of TPA in TPA-DCM are from $\sim 67^{\circ}$ to $\sim 76^{\circ}$.

	P3 $P2$ $P1$ $P1$
	P4
	P5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5 5
Dihedral angles (°)	
C5-C6-C7-C8	-48.4
C3-C4-C7-C8	-49.9
C2-C1-C7-C8	-50.0
C9-C10-C8-C7	-49.7
Angles between pla	ines
P ₁ -P ₂	~0
$P_2 - P_3$	~68
P_2-P_4	~67
P ₃ -P ₄	~75
P ₄ -P ₅	~35

Figure S6. Summary of dihedral angles (°) for TPE-TETRAD. In the TPE component of TPE-TETRAD, the dihedral angles between any phenyl rings are \sim 50°. The dihedral angles between any two phenyl rings of TPA in TPE-TERAD are from \sim 67° to \sim 75°.



Figure S7. Fluorescence lifetime of (A) TPE-TETRAD thin film solid, (B) TPE-TETRAD nanoaggregates (10 μ M) in water, (C) TPE-TETRAD@Biotin NPs (6 μ g/mL) in water and (D) TPE-TETRAD (10 μ M) in THF.



Figure S8. Relative ROS generation profile for TPE-TETRAD using DCFH-DA cellular reactive oxygen species detection assay kit. Change in fluorescent intensity at 534 nm for TPE-TETRAD, DCFH-DA indicator and their mixture in PBS upon white light irradiation for various time intervals. Excitation wavelength: 485 nm. [TPE-TETRAD] = 10 μ M; [DCFH-DA] = 1 μ M.



Colocalization Statistics				
Pearson's Correlation	0.919			
Overlap Coefficient	0.9288			
Colocalization Rate	94.53%			
Colocalization Area	306.53 µm²			
Area Image	5302.86 µm²			
Area Foreground	324.27 µm²			
Area Background	4978.70 µm²			

Figure S9. CLSM colocalization statistics for HeLa cervical cancer cells stained with 25 nM MitoTracker Green for 10 min and 6 μ g/mL TPE-TETRAD@Biotin NPs for 1 h at 37 °C.



Figure S10. Schematic illustration of the experimental setup for two-photon excited fluorescence.





Calculation of TPE-TETRAD@Biotin AIE NPs Concentration

The density of TPE-TETRAD@Biotin can be estimated as 1 g/cm³ due to its high water stability. The average size of the AIE NPs as verified by TEM and DLS was 100 nm. 0.25 mg of powder was obtained after freeze-drying 1 mL of the filtered stock solution. The TPE-TETRAD@Biotin concentration in a 1 mL stock suspension can be calculated from the following equation:

$$\frac{Total \ volume \ of \ TPE - TETRAD@Biotin \ NP}{Average \ volume \ of \ each \ NP} = \frac{\frac{0.25 \ x \ 10^{-3}}{1 g/mL}}{\frac{4}{3} \pi \ x \ (50 \ x \ 10^{-7})^3 mL} = 4.78 \ x \ 10^{11}, \ Total$$
number of NP in 1mL.

$$[\text{TPE-TETRAD}@\text{Biotin}] = \frac{\frac{4.78 \times 10^{11}}{6.02 \times 10^{23} \text{ mol}^{-1}}}{1 \times 10^{-3} L} = 0.8 \text{ nM}, \text{ final concentration in stock solution}$$
$$\frac{0.8nM \times 50\mu L}{2000 \text{ N}}$$

Working Concentration = $2000\mu L$ = 20 pM, final working concentration for cell imaging

Or alternatively we can calculate the mass concentration:

Working Concentration = $\frac{250\mu g/mL \times 50\mu L}{2000\mu L}$ = 6.25 µg/mL, final working concentration for cell imaging.