materials

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*Article**

Synthesis and Characterization of TiO_2 Nanoparticles for the Reduction of Water Pollutants

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Received: 30 August 2017; Accepted: 18 October 2017; Published: date

Supplementary Materials

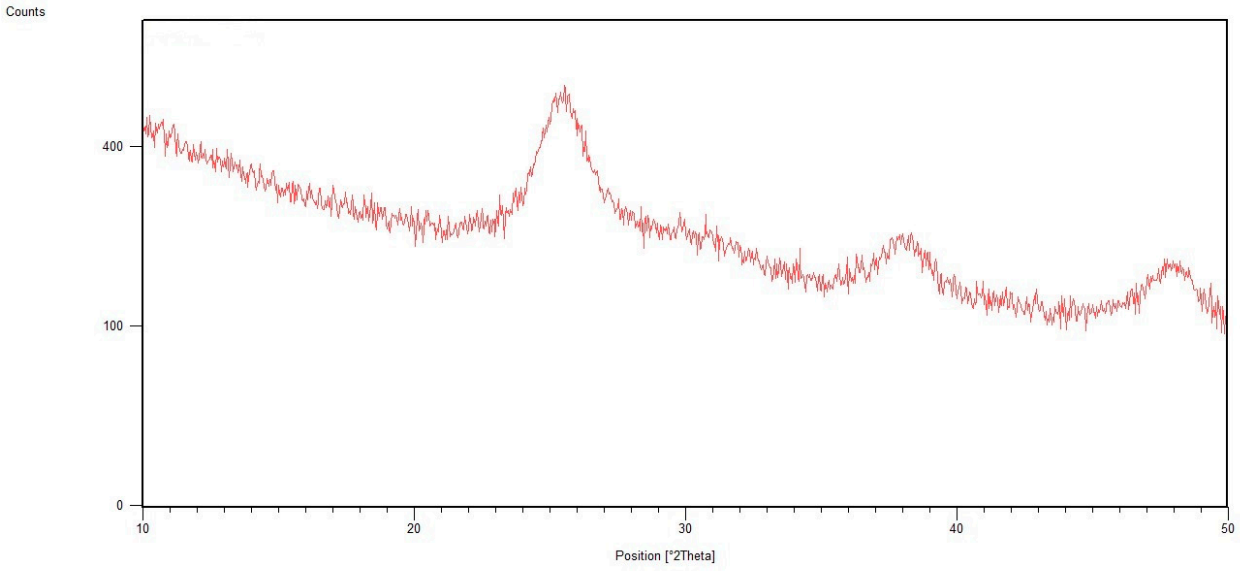


Figure S1. XRPD pattern of sample of Synthesis 1.

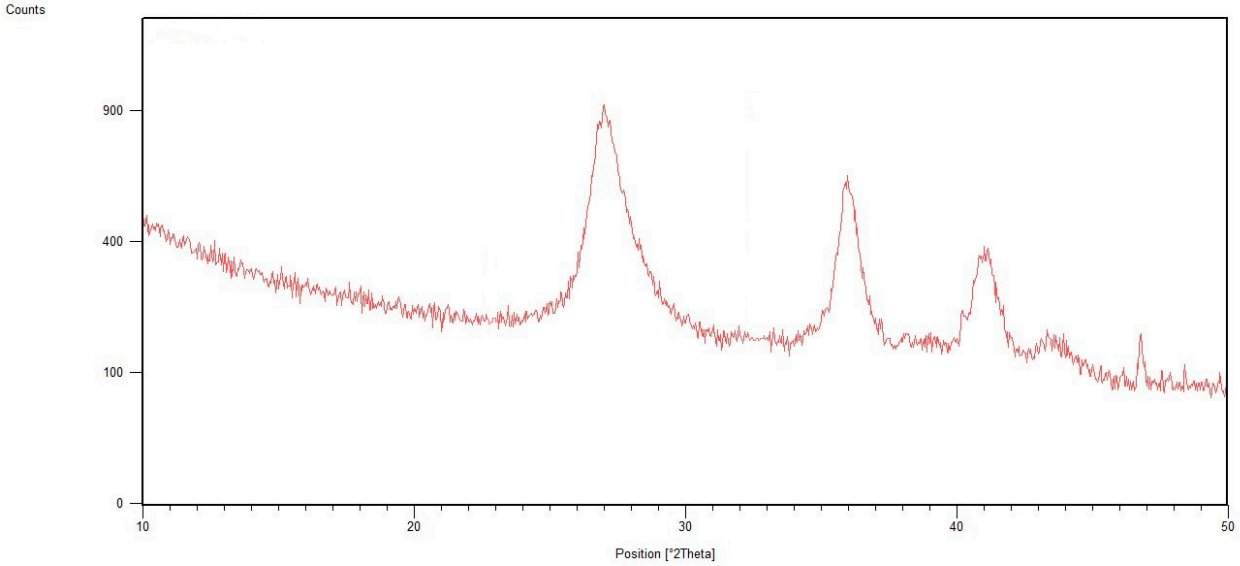


Figure S2. XRPD pattern of sample of Synthesis 2.

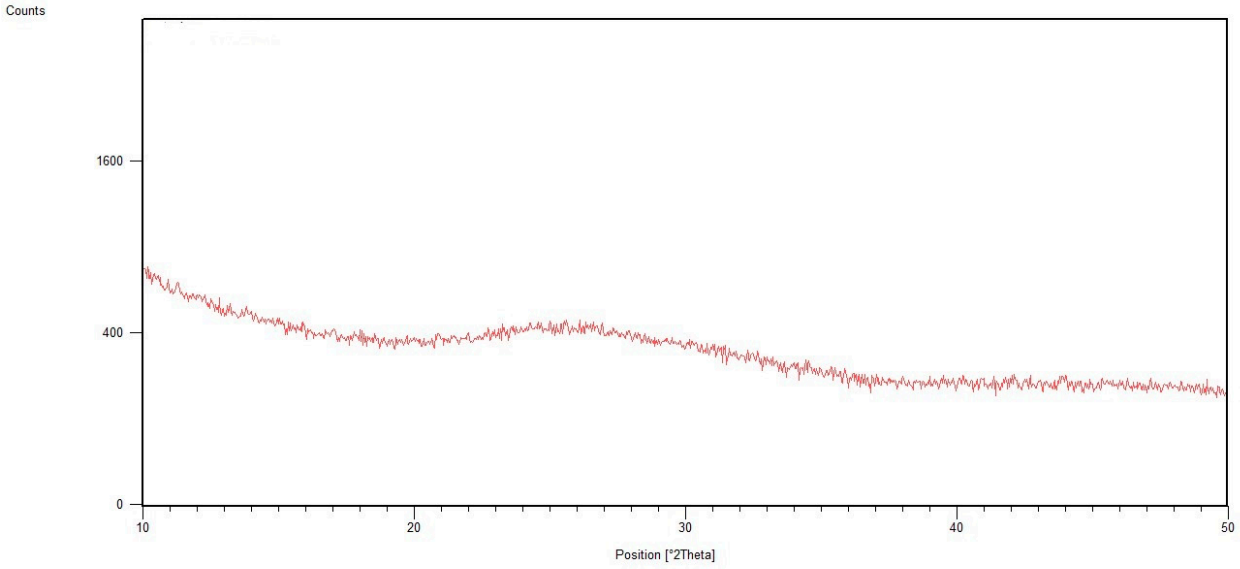


Figure S3. XRPD pattern of sample of Synthesis 3.

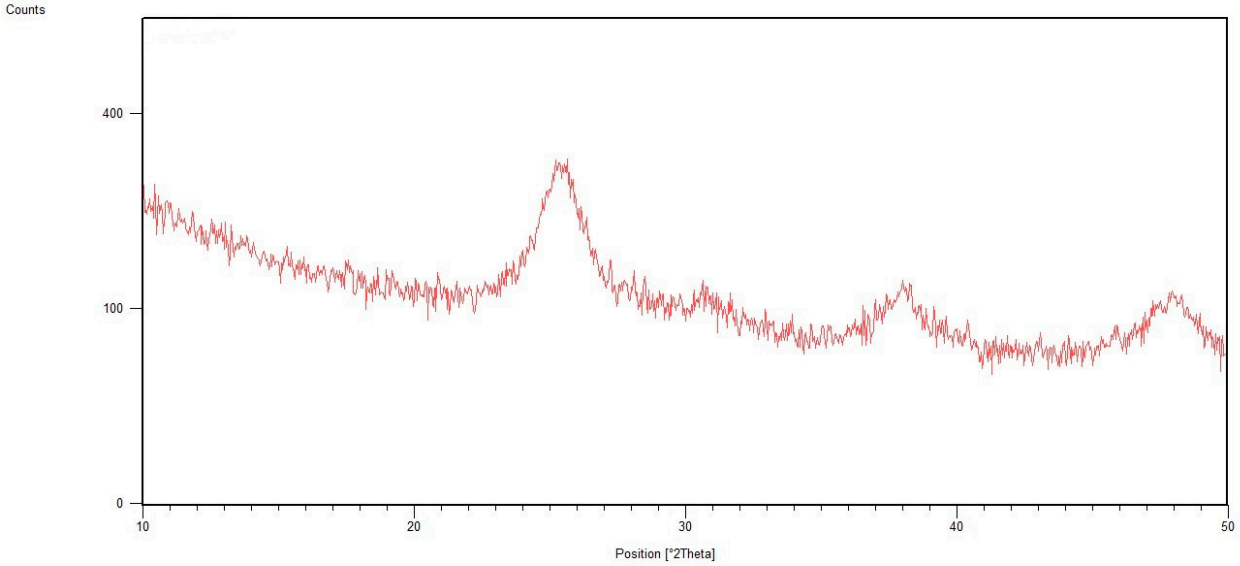


Figure S4. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 10:1: 0, 50 °C.

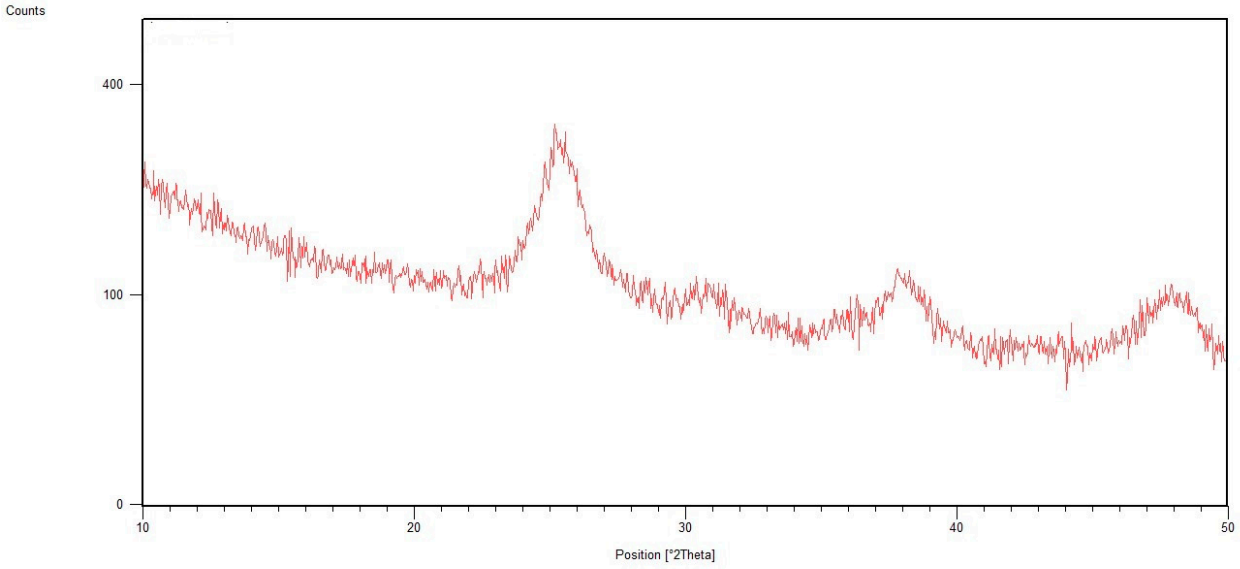


Figure S5. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 2:1: 0, 50 °C.

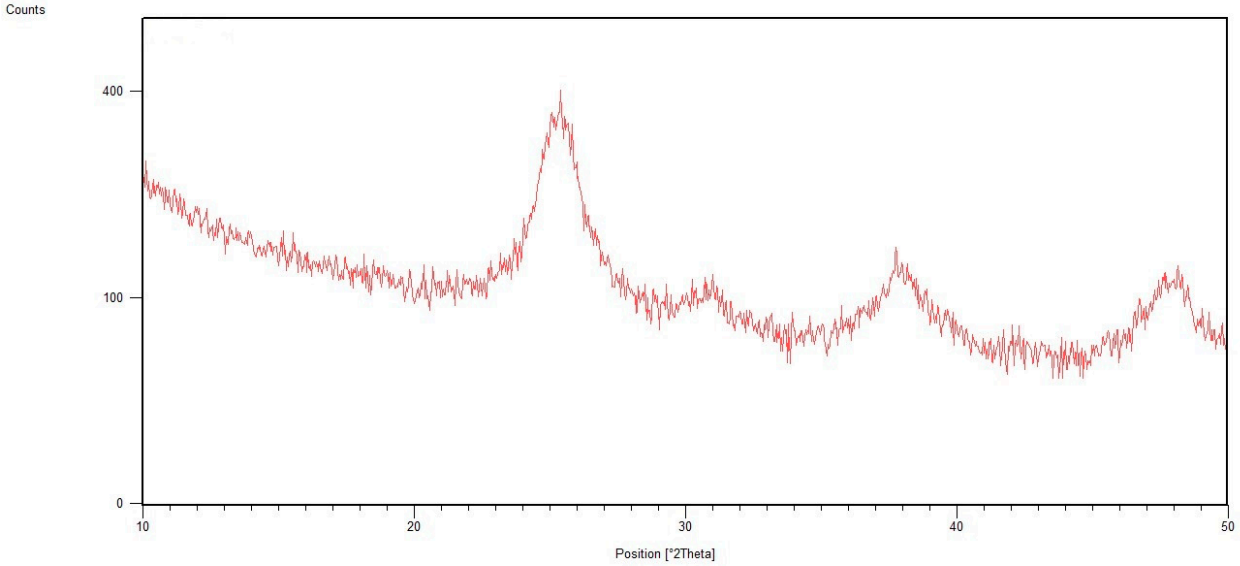


Figure S6. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 10:1: 0, r.t.

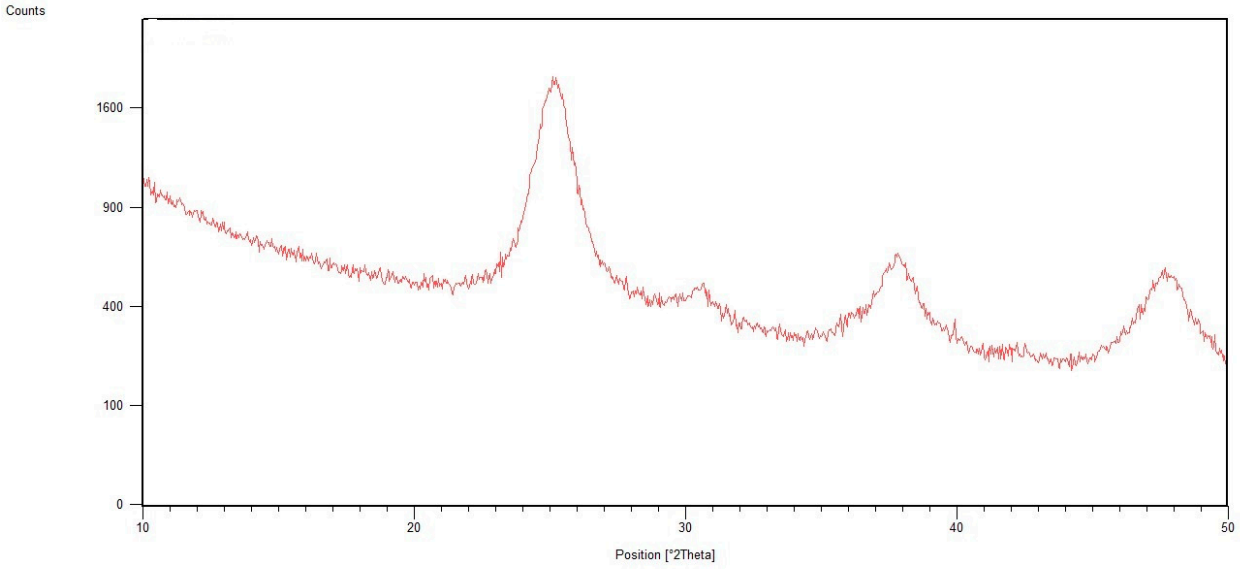


Figure S7. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 2:1: 0, r.t.

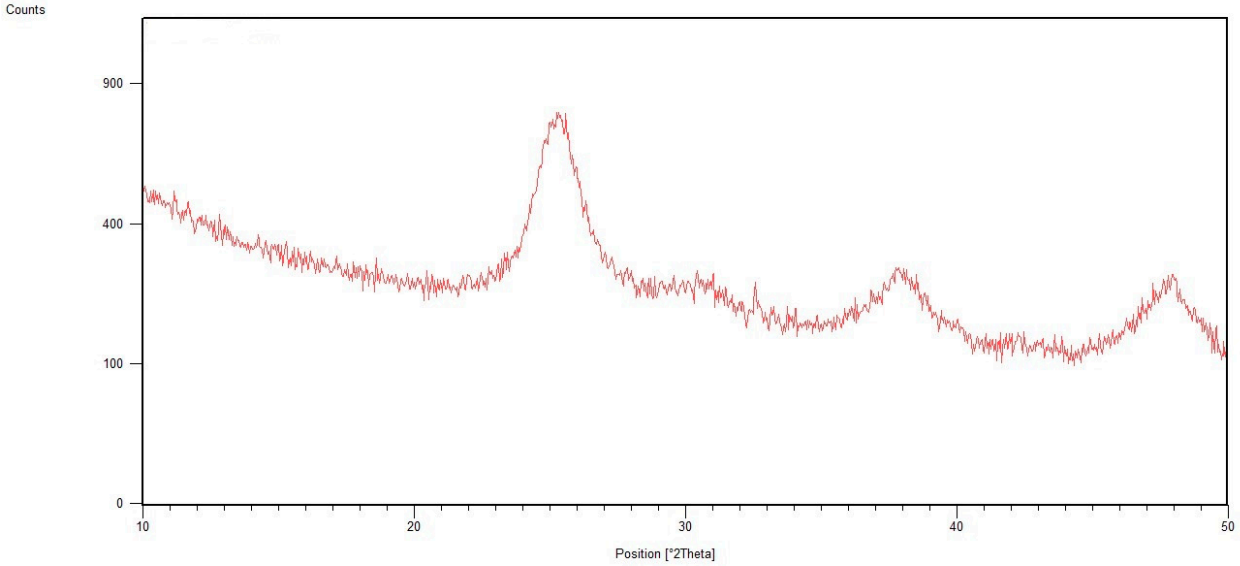


Figure S8. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 10:1: 0.52, 50 °C.

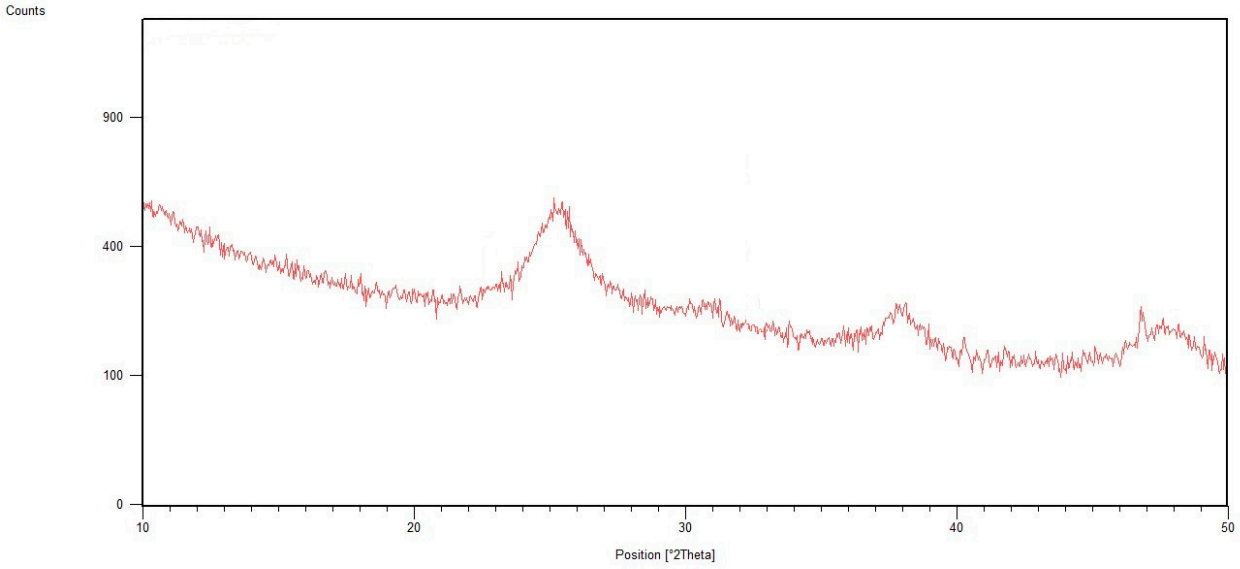
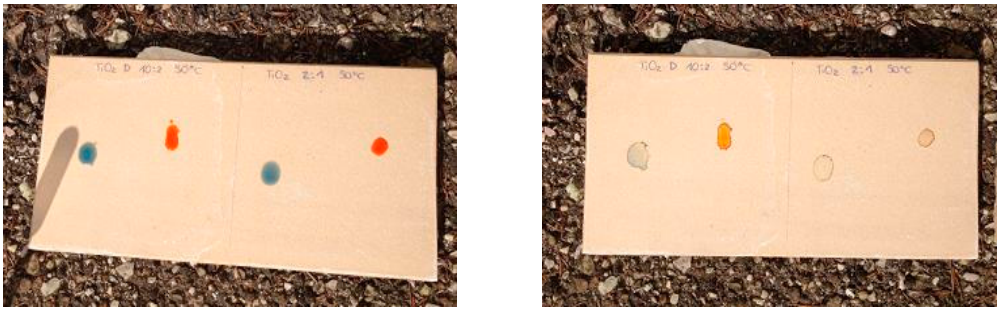


Figure S9. XRPD pattern of sample with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 2:1: 0.52, 50 °C.

(a) (b)

Figure S10. Ceramic titles with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$ 10:1 (left part of image) and 2:1 (right part of image), 50 °C, before (a) and after (b) ½ hour of sunlight exposure.



(a) (b)

Figure S11. Ceramic titles with molar ratio $C_{12}H_{28}O_4Ti$:$CO(NH_2)_2$:NH_4Cl 10:1:0.52 (left part of image) and 2:1: 0.52 (right part of image), 50 °C, before (a) and after (b) ½ hour of sunlight exposure.