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Supporting Information

Efficient Capture and Release of Circulating Tumor Cells by Temperature-Sensitive Polymer-Graphene Oxide Composite

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Chemicals Acryloyl chloride (Fluka), piperidine (Aldrich), triethylamine (Aldrich), and basic alumina (Aldrich) were used without further purification. AIBN (Aldrich) was recrystallized from methanol before use. Anisole (Aldrich) used was anhydrous grade (99.7%). *N,N*diethylacrylamide was procured from TCI America. Dichloromethane, hexanes, and ethyl acetate used were solvent grade and were used without purification.

Polymer Synthesis

N-acryloyl piperidine was synthesized through reaction between acryloyl chloride and piperidine (Figure 1).^[1] In short, 0.11 mol of piperidine and 0.12 mol of triethylamine were dissolved in 100 mL of dichloromethane maintained at 0-5°C. A solution of acryloyl chloride (0.10 mol) in 15 mL of dichloromethane was added drop-wise to the above solution over 2 hours under constant stirring. After complete addition, the reaction mixture was stirred at room temperature for 24 hours and was extracted with water and purified by column chromatography (hexane:ethyl acetate, 1:1) to yield colorless to light yellow liquid. *N,N*-diethylacrylamide was passed through a basic alumina column prior to polymerization. AIBN was recrystallized from

methanol before use. In a typical polymerization reaction, the required amount of monomers was dissolved in anisole and 0.3 mol% (of total monomer content) of AIBN was added to the solution. The reaction flask was completely sealed and the solution was purged with Argon for 20 minutes. The reaction was carried out at 65°C for 20 hours. After the reaction, all the solvent was evaporated at high temperature under vacuum to obtain white solid residue. The residue was re-dissolved in chloroform and then twice precipitated in ethyl acetate to obtain white solid mass. The precipitate was recovered and dried at 60°C under vacuum for 2-3 days.

Polymer Characterization

The synthesized polymers were characterized by gel permeation chromatography (GPC, Waters Inc., 1515 Isocratic HPLC pump and 2414 RI detector) using 3 Styragel columns- HR2, HR3 and HR4 in series maintained at 35° C with chloroform as eluent (flow rate- 1 mL/min, total elution time- 40 min). The instrument was calibrated with polystyrene standards. LCST was ascertained by measuring UV-vis transmittance (Varian Cary 50 Bio) of a 0.1 wt.% aqueous solution of polymers as a function of temperature. A thermocouple was used for real-time measurement of temperature, with the metal junction dipped in the cuvette during the measurement. For effective measurement, the polymer solution was cooled down to 2-3°C along with the metal cuvette holder to slow down the heating up of sample in ambient condition. CaCl² was placed inside the UV-vis spectrophotometer chamber to ensure humidity-free environment. This was necessary to prevent atmospheric water vapor from condensing on the cold cuvette walls. UV-vis spectrum was measured from 200-800 nm at every 0.2-0.5°C with more frequent measurements near the transition temperature. Transmittance at 400 nm was

plotted against temperature and the temperature for 50% transmittance was noted as the LCST. Molecular weights and LCSTs of different polymer batches are noted below (Supplemental Table 1).

Table S1. Molecular weights, PDI, and LCST of different batches of synthesized polymers used in the study

Polymer	M_n (kDa)	$M_w(kDa)$	PDI	$LCST$ ($°C$)
P1	209.246	308.086	.47	13.6
P ₂	151.332	253.380	1.67	1າ 7
P3	175.085	255.778	l.46	12.0
P4	173.019	303.009	. 75	

Materials Kapton polyimide tape was purchased from Cole Parmer. Ethanol, acetone, chloroform, and isopropanol were solvent grade and were used without further purification. Surface modifying agents – (Heptadecafluoro-1,1,2,2-tetrahydrodecyl)trichlorsilane (HFTCS) and 2-methoxy(polyethyleneoxy)propyltrimethoxysilane (PEG-silane) – were purchased from Gelest Inc. Microscope glass slides were purchased from Fisher.

Device Fabrication

The glass slides were sequentially washed with chloroform, acetone, and isopropanol via sonication for 5 minutes each. The glass slides were then air dried and treated in a UV-ozone generator for 30 minutes to remove any carbon contamination and to obtain a high density of surface hydroxyl groups. The cleaned substrates were patterned using Kapton tape by masking the active device area. Kapton tape was chosen for its impermeability to silane vapors and good stability at high temperatures. The patterned substrates were then cleaned with wipes dipped in

ethanol to remove any adhesive residue and treated with HFTCS via vapor phase surface modification at 100°C for 30 minutes. HFTCS treatment results in hydrophobic fluoroalkyl groups on the unmasked peripheral regions of the substrates which prevent the use of any physical confining barrier to pattern the device with polymer-GO film by drop-casting method. After HFTCS treatment, the Kapton tape mask was removed and the glass slides were washed with copious amounts of ethanol to remove any physisorbed silane as well as any adhesive residue. The second surface modification was done in liquid phase by immersing the glass slides in 3.35mM of PEG-silane in ethanol for 12-15 hours. Subsequently, the glass slides were again washed with ethanol to remove any physisorbed silane. A polymer-GO blend solution containing 10 mg/mL of polymer in 975 μL DMF and 25 μL of GO-PEG solution was then drop-casted in requisite amount on the surface modified glass substrates and allowed to dry at 60°C in an oven. The PDMS chamber was assembled on the glass substrate with polymer-GO composite film through corona discharge to produce a microfluidic device (Figure S1).

Fluorescent biotin assay

To verify the ability to immobilize biotinylated antibody to the polymer-GO film surface, surface coverage by a fluorescently labeled biotin (Biotin (5-fluorescein) conjugate, Sigma Aldrich) was assessed (Figure S4a). Three polymer-GO films underwent the entirety of the conjugation chemistry (i.e. treatment with the GMBS cross-linker and NeutrAvidin; termed "Condition") with fluorescent biotin addition as the terminal step. To account for non-specific binding, three polymer-GO films were treated only with the fluorescent biotin to serve as a control in an analogous fashion to an isotype control (termed "Control"). ImageJ was used to

quantify the fluorescence. This technique showed a statistically significant increase in fluorescence intensity relative to the control (Figure S4b).

Cell labeling for optimization experiments

Cells were stained with CellTracker™ Green CMFDA Dye (Life Technologies/ThermoFisher Scientific) according to the manufacturer's protocol. The staining process takes approximately two hours and was performed in parallel with device preparation.

Figure S1. Schematic for CTC device fabrication.

(on composite film surface)

Figure S2. Schematic of graphene oxide functionalization chemistry

Figure S3. Fluorescence microscopy images of polymer-GO films incubated with FSE dye before and after being dipped in either cold (5°C) or room temperature (20°C) water for the specified time durations. Scale bar: 200 μm.

Figure S4. a) Schematic represents fluorescent biotin assay and negative control. b) The full conjugation chemistry features statistically higher fluorescence than the negative control as assessed via optical density ($p = 0.019$).

Table S2. Experimental results from Live/Dead assay (MCF-7 cell line).

Table S3. Comparison of CTC isolation technologies. Both commercially available and recently developed CTC isolation technologies are compared across multiple relevant metrics and abilities, showing the high versatility and performance of the technology put forth in this work. [2-13]

References

[1] Y. S. Jo, A. J. van der Vlies, J. Gantz, S. Antonijevic, D. Demurtas, D. Velluto, J. A. Hubbell, Macromolecules 2008, 41, 1140.

[2] S. Riethdorf, H. Fritsche, V. Müller, T. Rau, C. Schindlbeck, B. Rack, W. Janni, C. Coith, K. Beck, F. Jänicke, Clinical Cancer Research 2007, 13, 920. [3] M. G. Krebs, R. Sloane, L. Priest, L. Lancashire, J.-M. Hou, A. Greystoke, T. H. Ward, R. Ferraldeschi, A. Hughes, G. Clack, Journal of clinical oncology 2011, 29, 1556. [4] M. Wendel, L. Bazhenova, R. Boshuizen, A. Kolatkar, M. Honnatti, E. H. Cho, D. Marrinucci, A. Sandhu, A. Perricone, P. Thistlethwaite, Physical biology 2012, 9, 016005. [5] C. Ruiz, J. Li, M. S. Luttgen, A. Kolatkar, J. T. Kendall, E. Flores, Z. Topp, W. E. Samlowski, E. McClay, K. Bethel, Physical biology 2015, 12, 016008. [6] A. H. Talasaz, A. A. Powell, D. E. Huber, J. G. Berbee, K.-H. Roh, W. Yu, W. Xiao, M. M. Davis, R. F. Pease, M. N. Mindrinos, Proceedings of the National Academy of Sciences 2009, 106, 3970. [7] M. G. Krebs, J.-M. Hou, R. Sloane, L. Lancashire, L. Priest, D. Nonaka, T. H. Ward, A. Backen, G. Clack, A. Hughes, Journal of Thoracic Oncology 2012, 7, 306. [8] G. Vona, A. Sabile, M. Louha, V. Sitruk, S. Romana, K. Schütze, F. Capron, D. Franco, M. Pazzagli, M. Vekemans, The American journal of pathology 2000, 156, 57. [9] E. Ozkumur, A. M. Shah, J. C. Ciciliano, B. L. Emmink, D. T. Miyamoto, E. Brachtel, M. Yu, P. I. Chen, B. Morgan, J. Trautwein, A. Kimura, S. Sengupta, S. L. Stott, N. M. Karabacak, T. A. Barber, J. R. Walsh, K. Smith, P. S. Spuhler, J. P. Sullivan, R. J. Lee, D. T. Ting, X. Luo, A. T. Shaw, A. Bardia, L. V. Sequist, D. N. Louis, S. Maheswaran, R. Kapur, D. A. Haber, M. Toner, Sci Transl Med 2013, 5, 179ra47. [10] L. Zhang, L. D. Ridgway, M. D. Wetzel, J. Ngo, W. Yin, D. Kumar, J. C. Goodman, M. D. Groves, D. Marchetti, Science translational medicine 2013, 5, 180ra48. [11] I. Baccelli, A. Schneeweiss, S. Riethdorf, A. Stenzinger, A. Schillert, V. Vogel, C. Klein, M. Saini, T. Bäuerle, M. Wallwiener, Nature Biotechnology 2013, 31, 539. [12] Z. Zhang, H. Shiratsuchi, J. Lin, G. A. Chen, R. M. Reddy, E. Azizi, S. Fouladdel, A. C. Chang, L. Lin, H. Jiang, M. Waghray, G. Luker, D. M. Simeone, M. S. Wicha, D. G. Beer, N. Ramnath, S. Nagrath, Oncotarget 2014, 5, 12383.

[13] H. J. Yoon, T. H. Kim, Z. Zhang, E. Azizi, T. M. Pham, C. Paoletti, J. Lin, N. Ramnath, M. S. Wicha, D. F. Hayes, D. M. Simeone, S. Nagrath, Nature Nanotechnology 2013, 8, 735.