


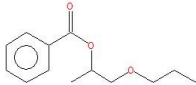



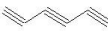






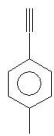
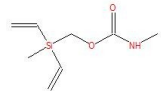
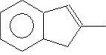
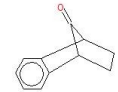

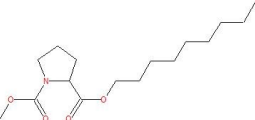
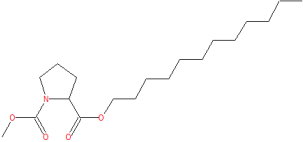
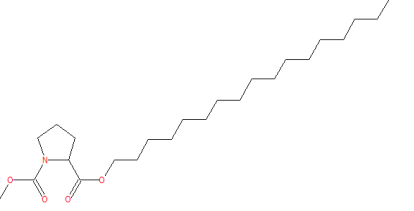
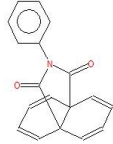
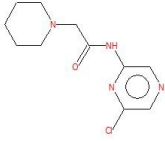


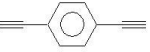
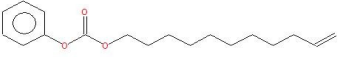
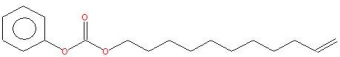


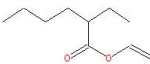
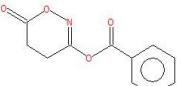
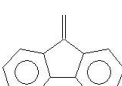
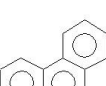


Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [% / %]
Methyl alcohol	CH ₄ O		43.5	0.28	80.63 / n.a.
Methane	CH ₄		21.1		
3-Methyl-1,2-diazirine	C ₂ H ₄ N ₂		54.6	0.28	100 / n.a.
Acetylene	C ₂ H ₂		96.5	0.30	88.47 / 100
1-Buten-3-yne	C ₄ H ₄		95.2	0.31	11.91 / n.a.
1,3-Cyclopentadiene, 1-methyl	C ₆ H ₈		35.3	0.33	5.00 / 2.71
Benzene	C ₆ H ₆		77.3	0.43	7.08 / 26.08
1,3-Butadiyne	C ₄ H ₂		98.7	0.45	8.28 / n.a.
Hepta-4,6-diyne-2-ol	C ₇ H ₈ O		67.8	0.48	5.22 / n.a.
Toluene	C ₇ H ₈		47.7	0.59	4.42 / 38.89
Water	H ₂ O		92.1	0.70	18.76 / n.a.
Hepta-4,6-diyne-2-ol	C ₇ H ₈ O		60.2	0.71	12.34 / 39.62
Ethylbenzene	C ₈ H ₁₀		61.0	0.85	4.29 / 3.91
2,4-Hexadiyne	C ₆ H ₆		43.6	0.91	4.04 / 4.49
1,5-Hexadien-3-yne	C ₆ H ₆		21.2		
Bicyclo[4.1.0]octa-1,3,5-triene-7,8-dione	C ₈ H ₄ O ₂		48.8	1.07	3.14 / 27.35
Oxiranecarboxamide, 3-phenyl-	C ₉ H ₉ NO ₂		54.5	1.10	3.85 / n.a.

Bicyclo[4.1.0]hepta-1,3,5-triene	C_7H_6		88.8	1.32	3.19 / 7.23
Benzene, 1,2,3-trimethyl-	C_9H_{12}		12.4	1.37	3.07 / 2.94
Benzene, (1-methylethyl)-	C_9H_{12}		8.27		
1-Propoxypropan-2-yl benzoate	$C_{13}H_{18}O_3$		8.27		
Styrene	C_8H_8		30.6	1.61	3.44 / 9.49
Bicyclo[4.1.0]octa-1,3,5-triene	C_8H_8		28.3		
1,3,5,7-Cyclooctatetraene	C_8H_8		26.1		
Hexa-1,3,5-triyne	C_6H_2		96.7	2.56	2.66 / 3.53
Phenylethyne	C_8H_6		89.4	2.82	2.75 / 7.02
1H-Indene, 1-chloro-2,3-dihydro-	C_9H_9Cl		29.1	4.21	2.04 / 2.86
Indene	C_9H_8		26.9		
Benzene, 1-propynyl	C_9H_8		24.7	4.54	1.72 / 6.64
2-Methylphenylacetylene	C_9H_8		23.8		
2-Thiazolidinecarboxylic acid	$C_4H_7NO_2S$		32.9	4.68	2.14 / 2.86

Benzeno, 1-ethynyl-4-methyl-	C_9H_8		50.0	5.04	1.44 / 2.08
Divinyl(N-methylcarbamoyloxy methyl)silane	$C_8H_{15}NO_2Si$		34.6	5.70	1.58 / 3.22
2-Methylindene	$C_{10}H_{10}$		21.5	5.76	1.86 / 2.36
1,4-Methanonaphthalene-9-one, 1,2,3,4-tetrahydro-	$C_{11}H_{10}O$		16.5		
1H-Indene, 3-methyl-	$C_{10}H_{10}$		10.6		
d-Proline, N-methoxycarbonyl-, nonyl ester	$C_{16}H_{29}NO_4$		7.44	7.79	1.41 / 4.75
d-Proline, N-methoxycarbonyl-, dodecyl ester	$C_{19}H_{35}NO_4$		5.84		
d-Proline, N-methoxycarbonyl-, octadecyl ester	$C_{25}H_{47}NO_4$		4.94		
4a,8a-(Methaniminomethano)naphthalene-9,11-dione, 10-phenyl-	$C_{18}H_{13}NO_2$		46.90	7.87	1.42 / 5.30

Acetamide, 2-(1-piperidyl)-N-(6-chloropyrazin-2-yl)-	$C_{11}H_{15}ClN_4O$		29.3	8.08	1.69 / 5.89
Naphthalene	$C_{10}H_8$		49.5	8.32	1.10 / n.a.
Azulene	$C_{10}H_8$		37.9		
1,4-Diethynylbenzene	$C_{10}H_6$		47.9	10.50	1.23 / 3.96
Carbonic acid, phenyl undec-10-enyl ester	$C_{18}H_{26}O_3$		22.3	15.36	0.69 / 3.93
Carbonic acid, pentadecyl phenyl ester	$C_{22}H_{36}O_3$		16.6		
Acenaphthylene	$C_{12}H_8$		64.5	15.45	0.84 / 4.94
Caprolactam	$C_8H_{11}NO$		78.0	16.00	0.60 / n.a.
Vinyl 2-ethylhexanoate	$C_{10}H_{18}O_2$		30.4	18.37	0.75 / 3.38
1,2(4H)-Oxazine-3-ol, 5,6-dihydro-6-oxo-, benzoate	$C_{11}H_9NO_4$		19.1	19.62	0.63 / 1.48
9H-Fluorene, 9-methylene	$C_{14}H_{10}$		28.9	22.33	0.70 / 4.00
Phenanthrene	$C_{14}H_{10}$		25.6		

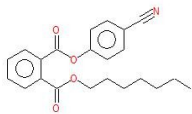
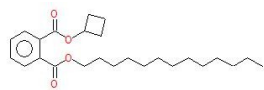
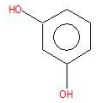
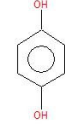



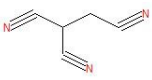


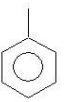

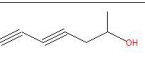
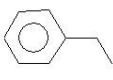
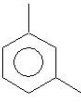
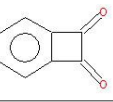
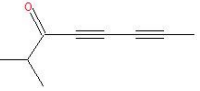



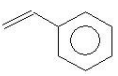


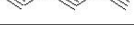
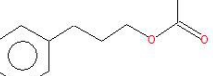
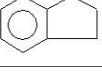
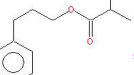
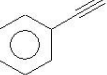



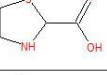

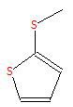

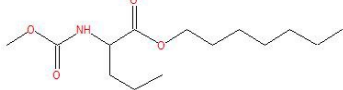
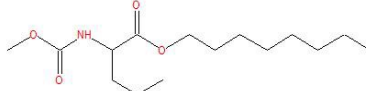
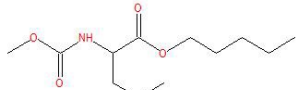
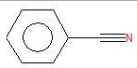
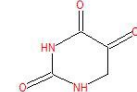
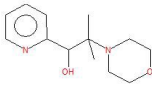
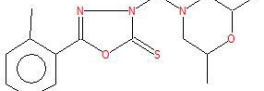
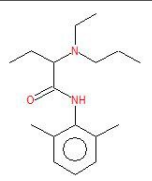
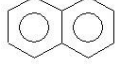
Phthalic acid, 4-cyanophenyl heptyl ester	$C_{22}H_{23}NO_4$		12.4	23.24	0.48 / n.a.
Phthalic acid, cyclobutyl tridecyl ester	$C_{25}H_{38}O_4$		8.26		
Resorcinol	$C_6H_6O_2$		18.9	27.51	0.68 / 3.53
Hydroquinone	$C_6H_6O_2$		18.9		

Table 1: Detected molecules from *A3.1_{gas}* as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. Molecules containing elements other than *C*, *H*, *N* and *O* may represent mismatches in the database, or a contamination from the cleaning solution (*Deconex*, Borer).

Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [%]
3-Methyl-1,2-diazirine	$C_2H_3N_2$		66.1	0.30	100 / 0.28
1-Buten-3-yne	C_4H_4		95.2	0.31	99.55 / 100
Ammonia	H_3N		98.0	0.35	12.68 / 10.62
2-Cyanosuccinonitrile	$C_5H_3N_3$		59.4	0.44	5.10 / 0.62
1,3-Butadiyne	C_4H_2		91.2	0.46	9.11 / 6.34
Sulfur dioxide	O_2S		69.0	0.49	6.46 / 1.81
Toluene	C_7H_8		50.2	0.61	2.59 / n.a.
Water	H_2O		94.7	0.71	13.06 / n.a.
Hepta-4,6-diyne-2-ol	$C_7H_{10}O$		46.5	0.72	10.81 / n.a.
Ethylbenzene	C_8H_{10}		43.1	0.88	2.12 / 0.27
Benzene, 1,3-dimethyl-	C_8H_{10}		27.0		
Bicyclo[4.2.0]octa-1,3,5-triene-7,8-dione	$C_8H_6O_2$		49.2	1.08	2.00 / 0.72
4,6-Octadiyn-3-one, 2-methyl-	$C_{10}H_{10}O$		77.0	1.18	1.90 / 0.36
Bicyclo[4.1.0]hepta-1,3,5-triene	C_7H_8		83.5	1.33	1.46 / n.a.
Acetic acid, sodium salt	$C_2H_3NaO_2$		90.8	1.40	1.52 / n.a.
1,3,5,7-Cyclooctatetraene	C_8H_8		34.6		

Styrene	C_8H_8		26.5	1.68	1.50 / 0.29
Bicyclo[4.2.0]octa-1,3,5-triene	C_8H_8		24.5		
Bicyclo[4.1.0]hepta-1,3,5-triene	C_7H_8		56.5	2.32	1.41 / 0.46
Hexa-1,3,5-triyne	C_6H_2		96.8	2.58	1.22 / 0.51
3-Phenyl-1-propanol, acetate	$C_{11}H_{14}O_2$		22.5	2.66	1.29 / 0.30
Indane	C_9H_{10}		14.6		
Propanoic acid, 2-methyl-, 3-phenylpropyl ester	$C_{13}H_{18}O_2$		11.2		
Phenylethyne	C_8H_6		88.8	2.85	1.26 / n.a.
Indene	C_9H_8		23.1	4.26	1.06 / n.a.
Benzene, 1-propynyl-	C_8H_6		16.3		
Benzene, 1-ethynyl-4-methyl-	C_9H_8		13.8		
2-Thiazolidinecarboxylic acid	$C_3H_5NO_2S$		35.3	4.69	1.06 / 0.48
Benzene, 1-ethynyl-4-methyl-	C_9H_8		83.9	5.02	1.14 / n.a.

Thiophene, 2-(methylthio)-	$C_6H_6S_2$		48.5		
1H-Phosphole, 2,3-dihydro-1,4-dimethyl-, 1-oxide	$C_7H_{11}OP$		32.3		
l-Norvaline, N-methoxycarbonyl-, heptyl ester	$C_{15}H_{29}NO_4$		26.2		
l-Norvaline, N-methoxycarbonyl-, octyl ester	$C_{16}H_{31}NO_4$		16.9	5.85	0.92 / n.a.
l-Norvaline, N-methoxycarbonyl-, pentyl ester	$C_{12}H_{23}NO_4$		16.3		
Benzonitrile	C_7H_5N		73.2	6.29	0.86 / 0.52
2,4,5-Trihydroxypyrimidine	$C_4H_4N_2O_3$		25.6		
1-[2-Pyridyl]-2,2-dimethyl-2-morpholino ethanol	$C_{12}H_{18}N_2O_2$		20.1	6.74	0.89 / 0.39
1,3,4-Oxadiazole-2(3H)-thione, 3-(2,6-dimethylmorpholinomethyl)-5-(2-methylphenyl)-	$C_{18}H_{21}N_3O_2S$		19.3		
Etidocaine	$C_{17}H_{28}N_2O$		41.0	7.83	0.65 / n.a.
Naphthalene	$C_{10}H_8$		58.0	8.30	0.52 / n.a.

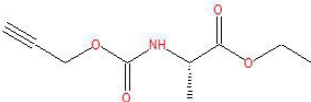
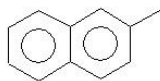
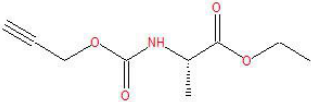
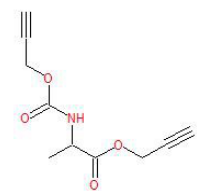
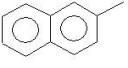
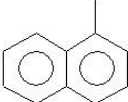

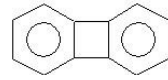
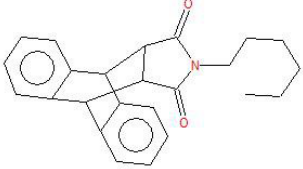
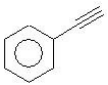

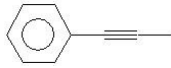
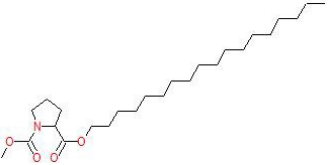
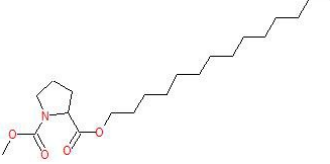
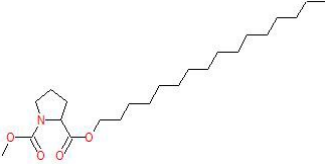
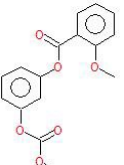
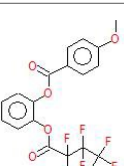
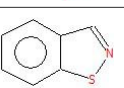
L-Alanine, n-propargyloxycarbonyl-, ethyl ester	$C_9H_{13}NO_4$		31.4	8.87	0.51 / n.a.
Naphthalene, 2-methyl-	$C_{11}H_{10}$		76.5	10.15	0.58 / 0.23
L-Alanine, n-propargyloxycarbonyl-, ethyl ester	$C_9H_{13}NO_4$		21.1	10.48	0.57 / 0.30
D-Alanine, N-propargyloxycarbonyl-, propargyl ester	$C_{10}H_{11}NO_4$		14.9		
Naphthalene, 2-methyl-	$C_{11}H_{10}$		39.5	10.72	0.41 / n.a.
Naphthalene, 1-methyl-	$C_{11}H_{10}$		20.4		
Acenaphthylene	$C_{12}H_8$		48.1	15.46	0.33 / n.a.
Biphenylene	$C_{12}H_8$		31.1		
Benzo[<i>f</i>]isoindol-1,3-(1 <i>H</i> ,3 <i>H</i>)-dione, 3 <i>a</i> ,4,9,9 <i>a</i> -tetrahydro-4,9- <i>O</i> -benzo-2-hexyl-	$C_{24}H_{29}NO_2$		22.3	22.59	0.29 / n.a.

Table 2: Detected molecules from $A3_2_{gas}$ as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. Molecules containing elements other than *C*, *H*, *N* and *O* may represent mismatches in the database, or a contamination from the cleaning solution (*Deconex*, Borer).

Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [%]
3-Methyl-1,2-diazirine	C ₂ H ₄ N ₂		60.8	0.28	100 / 100
1-Buten-3-yne	C ₆ H ₄		96.7	0.30	24.48 / 9.30
Ammonia	H ₃ N		98.5	0.34	12.68 / 2.73
Benzene	C ₆ H ₆		69.6	0.44	10.81 / 3.91
1,3-Butadiyne	C ₄ H ₂		97.9	0.46	12.79 / 6.22
Hepta-4,6-diyne-2-ol	C ₇ H ₈ O		72.4	0.51	9.19 / 0.62
Methyl isocyanide	C ₂ H ₃ N		42.0	0.56	7.65 / 1.99
Acetonitrile	C ₂ H ₃ N		33.0		
Toluene	C ₇ H ₈		45.1	0.58	8.62 / 2.29
Water	H ₂ O		98.8	0.69	19.92 / 32.18
Hepta-4,6-diyne-2-ol	C ₇ H ₈ O		48.1	0.71	16.54 / 7.62
Ethylbenzene	C ₈ H ₁₀		29.7	0.85	7.01 / 1.35
Benzene, 1,3-dimethyl-	C ₈ H ₁₀		25.1		
Bicyclo[4.2.0]octa-1,3,5-triene-7,8-dione	C ₈ H ₄ O ₂		44.8	1.05	6.53 / 0.68
4,6-Octadiyn-3-one, 2-methyl-	C ₉ H ₁₀ O		87.3	1.14	5.48 / 2.27
Bicyclo[4.1.0]hepta-1,3,5-triene	C ₇ H ₆		45.3	1.29	5.56 / 0.18
Styrene	C ₈ H ₈		31.0	1.64	5.17 / 0.85
Bicyclo[4.2.0]octa-1,3,5-triene	C ₈ H ₈		23.7		
Hexa-1,3,5-triyne	C ₆ H ₂		92.1	2.54	4.66 / 0.71

Phenylethyne	C ₈ H ₆		82.3	2.83	4.58 / 0.43
Benzene, 1-ethynyl-4-methyl-	C ₉ H ₈		33.7	4.25	3.63 / 0.90
Benzene, 1-propynyl-	C ₉ H ₈		21.8		
d-Proline, N-methoxycarbonyl-, octadecyl ester	C ₂₅ H ₄₇ NO ₄		9.55	8.38	2.50 / 0.90
d-Proline, N-methoxycarbonyl-, tetradecyl ester	C ₂₁ H ₃₉ NO ₄		5.98		
d-Proline, N-methoxycarbonyl-, hexadecyl ester	C ₂₃ H ₄₃ NO ₄		5.98		
1,3-Benzenediol, o-methoxycarbonyl-, o'-(2-methoxybenzoyl)-	C ₁₆ H ₁₄ O ₆		14.9	9.50	1.53 / n.a.
1,2-Benzenediol, o-(4-methoxybenzoyl)-o'-(2,2,3,3,4,4,4-heptafluorobutyl)-	C ₁₈ H ₁₁ F ₇ O ₅		10.2		
1,2-Benzisothiazole	C ₇ H ₅ NS		44.2		

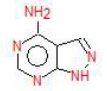
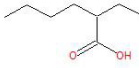
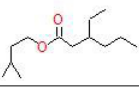
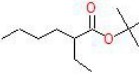


1H-Pyrazolo[3,4-d]pyrimidin-4-amine	C ₅ H ₅ N ₅		37.4	11.99	1.32 / n.a.
2-Ethyl-hexoic acid	C ₈ H ₁₆ O ₂		58.4	12.56	1.61 / 0.37
3-Methylbutyl 2-ethylhexanoate	C ₁₃ H ₂₆ O ₂		18.4	13.56	1.10 / n.a.
Hexanoic acid, 2-ethyl-, 1,1-dimethylethyl ester	C ₁₂ H ₂₄ O ₂		16.2		
Iodoacetylene	C ₂ HI		84.7	15.46	1.31 / 0.60
Carbon disulfide	CS ₂		63.0	20.11	0.93 / n.a.

Table 3: Detected molecules from *A3.3_{gas}* as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. Molecules containing elements other than *C*, *H*, *N* and *O* may represent mismatches in the database, or a contamination from the cleaning solution (*Deconex*, Borer). Fluorine may originate from the teflon used for sealing.


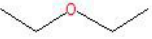


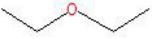

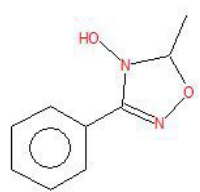




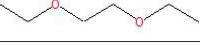
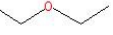


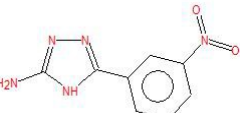
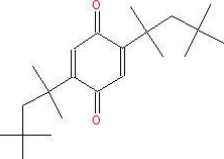
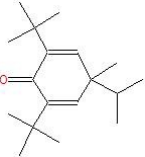
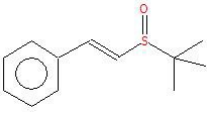
Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [% / %]	Concentration [(± 1.026) mol/l]
Pentane	C ₅ H ₁₂		80.8	0.12	14.97 / 8.76	0.401
Ethyl ether	C ₄ H ₁₀ O		87.7	0.13	13.68 / 7.83	0.389
Cyclohexane	C ₆ H ₁₂		76.0	0.14	9.48 / 5.75	0.232
Pentane	C ₅ H ₁₂		53.5	0.29	100 / 100	3.031
Ethyl ether	C ₄ H ₁₀ O		73.5	0.30	62.80 / 62.79	2.614
Cyclohexane	C ₆ H ₁₂		74.4	0.32	54.29 / 73.32	3.090
4-Hydroxy-5-methyl-3-phenyl-δ(2)-1,2,4-oxadiazoline	C ₉ H ₁₀ N ₂ O ₂		37.0	22.60	0.28 / 0.35	0.007

Table 4: Detected molecules from *A3.1liquid* as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. For estimating the molar concentration, we compared the areas and used the known solvent concentration as reference (printed in bold).

Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [% / %]	Concentration [(± 1.911) mol/l]
1-Propanol, 2-methyl-	C ₄ H ₁₀ O		62.2	0.23	19.45 / 33.27	2.073
Cyclohexane	C ₆ H ₁₂		78.9	0.25	12.04 / 13.06	1.032
Butane, 2-methyl-	C ₅ H ₁₂		36.1	0.28	100 / 87.44	3.162
Pentane	C ₅ H ₁₂		34.7			3.039
Ethane, 1,2-diethoxy-	C ₆ H ₁₄ O ₂		52.5	0.28	83.33 / 34.71	1.825
Ethyl ether	C ₄ H ₁₀ O		40.3			1.401
2H-Pyran, 3,4-dihydro	C ₆ H ₈ O		47.3	0.30	71.30 / 100	4.738
Cyclohexane	C ₆ H ₁₂		35.4	0.32	32.68 / 87.14	3.090
1,2,4-Triazol-3-amine, 5-(3-nitrophenyl)-	C ₈ H ₇ N ₅ O ₂		20.1	11.78	0.33 / n.a.	n.a.
p-Benzoquinone, 2,5-bis(1,1,3,3-tetramethylbutyl)-	C ₂₂ H ₃₆ O ₂		17.7			n.a.
2,6-Bis(1,1-dimethylethyl)-4-methyl-4-isopropylcyclohexa-2,5-dien-1-one	C ₁₈ H ₃₀ O		12.9			n.a.
(E)-tert-Butylsulfinyl-2-phenylethene	C ₁₂ H ₁₆ OS		50.1			15.43

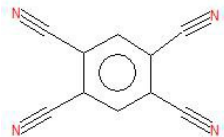
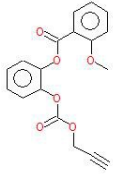
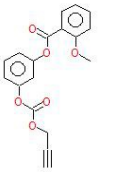
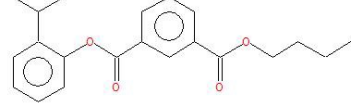
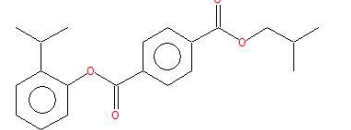
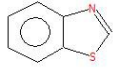
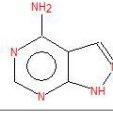
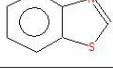
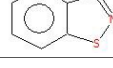

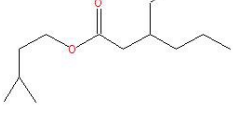
1,2,4,5- Benzenetetracarbonitrile	$C_{10}H_2N_4$		84.7	22.57	0.16 / n.a.	n.a.
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Table 5: Detected molecules from *A3-2_{liquid}* as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. For estimating the molar concentration, we compared the areas and used the known solvent concentration as reference (printed in bold). Molecules containing elements other than *C*, *H*, *N* and *O* may represent mismatches in the database, or a contamination from the cleaning solution (*Deconex*, Borer).

Name	Molecular formula	Structure	Probability [%]	Delay time [min]	Relative height/ area [% / %]	Concentration [(± 1.980) mol/l]
1-Propanol, 2-methyl-	C ₄ H ₁₀ O		63.5	0.23	4.03 / 2.36	0.240
Cyclohexane	C ₆ H ₁₂		80.7	0.25	4.14 / 0.97	0.126
Pentane	C ₅ H ₁₂		29.2	0.28	100 / 68.64	3.212
2-Oxetanone, 4-methyl-	C ₄ H ₆ O ₂		20.6			2.266
Acetic acid, (2-ethoxy)ethoxy-, methyl ester	C ₇ H ₁₄ O ₄		38.6	0.28	85.19 / 36.79	2.276
Allyl isocyanate	C ₄ H ₅ NO		72.1	0.30	70.83 / 100	11.553
Cyclohexane	C ₆ H ₁₂		60.5	0.31	50.00 / 31.88	3.090
Carbon monoxide	CO		74.2	0.57	1.78 / 0.24	0.028
Water	H ₂ O		82.9	0.67	2.70 / 3.83	0.509
Di-tert-butyl peroxide	C ₈ H ₁₈ O ₂		58.9	3.38	0.81 / 0.16	0.014
Oxalic acid, butyl propyl ester	C ₉ H ₁₆ O ₄		17.6	9.73	0.42 / 0.16	0.005
di-tert-Butyl dicarbonate	C ₁₀ H ₁₈ O ₅		10.1			0.003
Oxalic acid, isobutyl propyl ester	C ₉ H ₁₆ O ₄		7.95			0.002
1,2-Benzenediol, o-(4-methoxybenzoyl)-o'-(2,2,3,3,4,4,4-heptafluorobutyl)-	C ₁₈ H ₁₁ F ₇ O ₅		8.07			0.002

1,2-Benzenediol, o-(2-methoxybenzoyl)-o-(propargyloxycarbonyl)-	$C_{16}H_{14}O_6$		6.81	10.22	0.40 / 0.17	0.002
1,3-Benzenediol, o-(2-methoxybenzoyl)-o-(propargyloxycarbonyl)-	$C_{16}H_{14}O_6$		6.28			0.002
Isophthalic acid, butyl 2-isopropylphenyl ester	$C_{21}H_{24}O_4$		23.7			0.007
Terephthalic acid, isobutyl 2-isopropylphenyl ester	$C_{21}H_{24}O_4$		9.38	11.46	0.33 / 0.18	0.003
Benzothiazole	C_6H_5NS		34.0			0.015
1H-Pyrazolo[3,4-d]pyrimidin-4-amine	$C_5H_5N_5$		21.3	11.87	0.35 / 0.27	0.009
Benzothiazole	C_6H_5NS		48.7			n.a.
1,2-Benzisothiazole	C_6H_5NS		26.6	11.97	0.23 / n.a.	n.a.
2-Ethyl-hexoic acid	$C_{11}H_{20}O_2$		73.2	12.53	0.27 / n.a.	n.a.
3-Methylbutyl 2-ethylhexanoate	$C_{19}H_{36}O_2$		30.2			n.a.
				13.52	0.36 / n.a.	

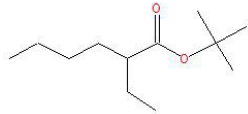
Hexanoic acid, 2-ethyl-, 1,1-dimethylethyl ester	$C_{12}H_{24}O_2$		18.3		n.a.
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Table 6: Detected Molecules from *A3_3liquid* as obtained using the NIST database as a reference. For low-percentage matches several possible molecules are listed. The highest peak, or the largest peak area in the chromatogram are used as references. Heights and areas of other peaks from other substances are given as a percentage relative to the respective reference and taken as their relative abundances. Not available (n.a.) area means that the area could not be determined due to asymmetric or overlapping peaks. Note that both estimates (height vs. area) can differ more or less, depending on the precise shape of a peak. For estimating the molar concentration, we compared the areas and used the known solvent concentration as reference (printed in bold). Molecules containing elements other than *C*, *H*, *N* and *O* may represent mismatches in the database, or a contamination from the cleaning solution (*Deconex*, Borer). Fluorine may originate from the teflon used for sealing.

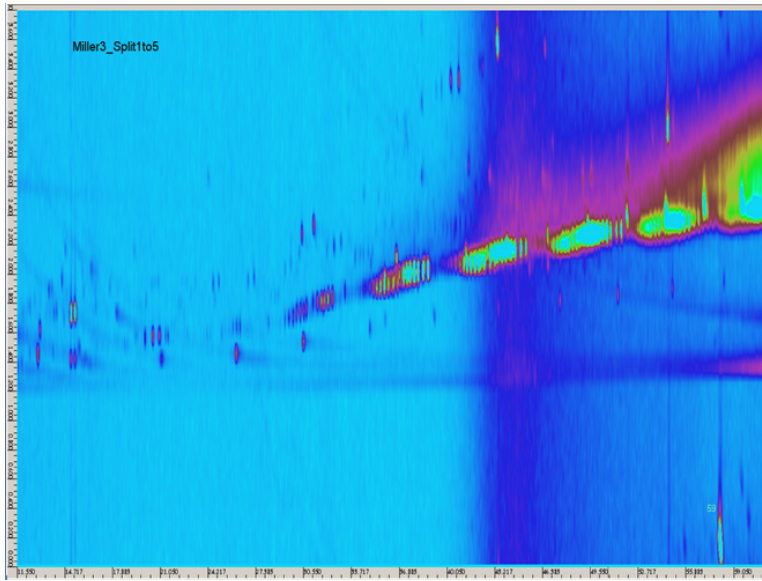


Fig. 1: Contour plot of the procedural blank.