

Figure S1. Bound nickel ion identified by anomalous difference map and electron density maps of selected regions of Ni/GMPPNP-bound *Kp*UreG.

In (A) & (B), anomalous difference electron density map, contoured at 7σ , was generated by PHENIX.REFINE using diffraction data collected at the nickel peak wavelength (SI Appendix, Table S2). (A) A nickel ion, coordinated in a square-planar geometry by Cys66 and His68 of the CPH motif, is located at the dimer interface of the Ni/GMPPNP-bound *Kp*UreG dimer. Residues from the opposite protomer are indicated by apostrophes. (B) A nickel ion was found to occupy the magnesium binding site in the nucleotide binding pocket of UreG, coordinated in an octahedral geometry by Thr15 (G1 motif), Asp43 (G2 motif), Glu98 (G3 motif), β/γ phosphate groups of GMPPNP, and a water. It is noted that the conserved residue Glu98 of the G3 motif is responsible for binding an active site magnesium ion in the G3E family of GTPases (1). Two protomers of *Kp*UreG are colored in light and dark grey. (C) 2Fo-Fc electron density map contoured at 1.5 σ showing the bound nickel at the dimeric interface. (D) Fo-Fc electron density map (before the GMPPNP was modeled) contoured at 3.0 σ .



Figure S2. Structural comparison of the GDP-bound *Hp*UreG and Ni/GMPPNP-bound *Kp*UreG

(A) Reorganization of hydrogen bonding network in the metal binding motif. In the GDP-bound state, His68 forms hydrogen bond with the conserved Asn103 (G3) and is pointing away from the nickel binding site. In the GTP-bound state, this hydrogen bond is broken, and Asp102 flips up to form hydrogen bonds with His68, Asn103 and Asn103' from the opposite protomer. Residues from the opposite protomer are indicated by apostrophes. (B) Charge-charge repulsion between the γ -phosphate group and Asp37 elicits conformational changes that induce helix-2 to tilt towards the nucleotide-binding site and bring Glu42 to form a salt-bridge with Arg130 of the opposite protomer. (C) The opposite protomer of UreG (green) undergoes a rigid-body movement towards the bound GMPPNP, forming two extra hydrogen bonds (yellow dotted lines) as indicated.



Figure S3. Dissociation of UreG from the $UreG_2F_2H_2$ complex in the presence of GTP γ S was greatly reduced in the *H. pylori* UreG D37A/E42A double mutant.

Protein samples of 15 μ M of UreG₂F₂H₂ complex (wild-type or mutant) were mixed with 45 μ M nickel ion and 300 μ M GTP γ S (red lines) or GDP (cyan lines), and were analyzed by SEC/SLS. In the presence of GTP γ S, wild-type UreG dissociated from UreG₂F₂H₂ to form a UreG dimer, while the dissociation was greatly reduced in the UreG D37A/E42A mutant.



Figure S4. GTP-dependent formation of the $UreE_2G_2$ complex was greatly reduced in *H. pylori* UreG D37A/E42A mutant.

Protein samples of 30 μ M UreG (wild type or mutant) were mixed with 30 μ M UreE, 45 μ M nickel ion and 300 μ M GTP γ S (red lines) or GDP (cyan lines), and were analyzed by SEC/SLS. In the presence of GTP γ S, wild type UreG interacted with UreE to form a UreE₂G₂ complex. The ability to form the UreE₂G₂ complex in the presence of GTP γ S was greatly reduced in the D37A/E42A mutant.



Figure S5. Preparation of the nickel-charged UreE dimer (UreE₂/Ni) and the nickel-charged UreE₂G₂ complex (UreE₂G₂/Ni).

To prepare UreE₂/Ni, 200 μ M purified UreE₂ sample was mixed with 1 mM NiSO₄. Excess nickel was removed by a HiTrap Desalting column (GE Healthcare). To prepare the UreE₂G₂/Ni complex, equal molar ratio (~100 μ M) of nickel-charged UreE dimer and UreG was mixed in the presence of 2 mM MgSO₄ and 1 mM GTP, followed by gel filtration chromatography using a Superdex 200 Increase 10/300 gel filtration column (GE Healthcare). To prepare the UreE₂G₂ complex without the bound nickel (UreE₂G₂), apo-UreE₂ and UreG were mixed instead. The nickel content in UreE₂, UreE₂/Ni, UreE₂G₂, UreE₂G₂/Ni complexes prepared were determined by atomic absorption spectroscopy, which confirmed that one nickel ion was bound per UreE₂/Ni or UreE₂G₂/Ni. The protein complexes were also found to be of correct molecular weight by SEC/SLS (SI Appendix, Figure S8).



Figure S6. Dissociation of UreE₂G₂ to UreE₂G upon GTP Hydrolysis.

15 μ M UreE₂G₂/Ni complex (injection 1) in 1 mM GTP, 0.2 mM TCEP, 100 mM NaCl, 20 mM HEPES, pH 7.2 was incubated with (injection 2) or without (injection 3) 10 mM KHCO₃ at 37 °C for 120 min. Protein samples were then analyzed by SEC/SLS. Our results showed that the UreE₂G₂ complex dissociated into UreE₂G and a monomeric UreG (injection 2) upon activation of GTP hydrolysis by KHCO₃.



Figure S7. The switching of UreG from the $UreG_2F_2H_2$ complex to the $UreE_2G_2$ complex is dependent on GTP but not on nickel.

15 μ M UreG₂F₂H₂ complex (grey solid lines) and 15 μ M apo-UreE dimer (grey dotted lines) was added to 2 mM MgSO₄, 0.2 mM TCEP, 100 mM NaCl, 20 mM HEPES, pH 7.2 buffer with (red lines) or without (cyan lines) 300 μ M GTP γ S. The protein samples were analyzed by SEC/SLS. In the absence of GTP γ S (left panel), only small amount of UreG dissociated from the UreG₂F₂H₂ complex to form the UreE₂G complex. In the presence of GTP γ S (right panel), wild-type UreG completely dissociated from the UreG₂F₂H₂ complex to form the UreE₂G₂ complex with the apo-UreE dimer. For the UreG D37A/E42A mutant, the GTP-dependent swapping of protein binding partners was greatly abolished.



Expected MW

Figure S8. Size-exclusion-chromatography/static-light-scattering (SEC/SLS) analyses of urease accessory proteins and their complexes.

Protein samples of urease accessory proteins and their complexes were prepared and analyzed by SEC/SLS using a Superdex 200 Increase 10/300 column as described in the Materials and Methods. The elution volumes of (1) $\text{UreG}_2\text{F}_2\text{H}_2$, (2) UreF_2H_2 , (3) nickel-charged UreE_2G_2 , (4) apo- UreE_2G_2 , (5) nickel-charged UreE dimer, (6) apo-UreE dimer, (7) nickel-charged UreG dimer, and (8) UreG monomer, were 11.8, 12.4, 13.3, 13.3, 14.9, 14.9, 14.5, and 15.5 mL, respectively. The molecular weights estimated by static-light-scattering were indicated at the elution peak, and they agreed well with the expected molecular weights.



Figure S9. Circular dichroism spectra of wild-type (solid lines) and D37A/E42A (dotted lines) *H. pylori* UreG.

The spectra collected were similar at 25°C, 37°C and 42°C, suggesting that both wild-type and the mutant were folded and stable at these temperatures.



Figure S10. Protein-protein interactions are essential for in-vitro urease activation.

To test if protein-protein interactions are essential to the urease activation, urease accessory proteins (UreE₂/Ni, UreG₂/Ni, apo-UreE₂, apo-UreG, UreG₂F₂H₂, UreF₂H₂) and apo-urease were added to either side of a dialysis membrane (dotted lines) in a two-chamber dialyzer (solid lines). The concentrations and the location of the proteins added are indicated schematically in the left panel. The buffer in both chambers contained 20 mM HEPES pH 7.5, 200 mM NaCl, 1 mM TCEP, 2 mM MgSO₄ and 300 µM GTP. The dialysis membrane allows diffusion of nickel ions but prevents direct protein-protein interactions across the membrane. After equilibration at 4 °C for 16 hr, 10 mM KHCO₃ was added to both chambers to activate the GTP hydrolysis required for urease activation. (A) & (B). Urease was activated only when UreE₂/Ni or UreG₂/Ni were added to the right chamber where other urease accessory proteins and apo-urease were located (A1 and B1). On the other hand, urease activation was greatly abolished when UreE₂/Ni or UreG₂/Ni were added in the left chamber and were separated from apo-urease and other urease accessory proteins by the dialysis membrane (A2 and B2). Urease activation was nickel dependent as addition of apo-UreE₂ (A3) or apo-UreG (B3) failed to activate urease. (C) We added 20 μ M NiSO₄ to the left chamber as indicated, and showed that free nickel ions, if presence, could diffuse across the dialysis membrane and activate urease in the right chamber with $UreG_2F_2H_2$ (C1), but not with $UreF_2H_2$ (C3). However, activation of urease was inhibited by addition of apo-Ure E_2 (C2) in the left chamber.

Data collection	
Wavelength (Å)	1.54
Space group	I2
Cell dimensions	
a, b, c (Å)	83.55, 49.32, 129.70
α, β, γ (°)	90.0, 103.17, 90.0
Resolution (Å)	19.43-1.80 (1.84-1.80)
R _{merge}	0.060 (0.462)
Ι/σΙ	13.1 (2.6)
Completeness (%)	99.7 (99.8)
Redundancy	3.5 (3.4)
Refinement	
Resolution (Å)	19.43 - 1.80
No. reflections	47824 (2812)
R _{work} / R _{free}	0.1563 / 0.1905
No. atoms	
Protein	2982
Ligand/ion	67
Water	488
B-factors (Å ²)	
Protein	21
Ligand/ion	13
Water	33
R.m.s. deviations	
Bond lengths (Å)	0.014
Bond angles (°)	1.315
Ramachandran	
Favored (%)	98
Allowed (%)	2
Outliers (%)	0

Table S1. Data collection and refinement statistics.

*Values in parentheses are for highest-resolution shell.

Data collection	Ni-peak
Wavelength (Å)	1.4853
Space group	P21
Cell dimensions	
a, b, c (Å)	84.06, 49.38, 129.61
α, β, γ (°)	90.0, 102.39, 90.0
Resolution (Å)	30.32-1.80 (1.83-1.80)
R _{merge}	0.042 (0.288)
Ι/σΙ	29.9 (6.7)
Completeness (%)	99.8 (99.1)
Redundancy	6.4 (6.1)

Table S2. Data collection statistics for nickel anomalous data.

*Values in parentheses are for highest-resolution shell.

SI Reference

1. Wittinghofer A, Vetter IR (2011) Structure-function relationships of the G domain, a canonical switch motif. *Annu Rev Biochem* 80:943–71.