Nucleoside-Lipid-Based Nanocarriers for Sorafenib Delivery

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Supporting Information

HPLC analysis of SLNs

A reversed phase UHPLC method was developed to study SLN composition and chemical stability.

The separation was carried out with the column Syncronis C18 50 x 2.1 mm, 1.7 μ m. The mobile phase consisted of 70/30 MeOH/26.5 mM ammonium acetate (pH = 7.4) (A) and 25 mM ammonium acetate in MeOH (pH 7.9) (B). A flow rate of 0.2 mL/min was used and the gradient profile was 0-2 min, 0-100% B; 2-20 min, 100% B. The column temperature was set

at 25°C. The detection was performed at 267 nm for Sorafenib and di C_{16} dT, and 257 nm for DOTAU. The injected volume was 1.0 μ L leading to quantitation limits of 0.6 ng for Sorafenib and 15 ng for both nucleolipids, DOTAU and di C_{16} dT.

Standard curves for Sorafenib, DOTAU, and diC₁₆dT in ethanol are shown in Figures SI1, SI2, and SI3, respectively. The HPLC analyses of both formulations are presented in Figure SI4.



Figure SI1 Standard curve Sorafenib: A = f(C) (UV 267 nm)



Figure SI2 Standard curve DOTAU: A = f(C) (UV 257 nm)



Figure SI3 Standard curve diC₁₆dT: A = f(C) (UV 267 nm)



Figure SI4. Simultaneous analysis of Sorafenib and nucleolipids: diC₁₆dT (A) and DOTAU (B) in formulations.

Stability studies of Sorafenib/diC16dT formulations

A similar chemical stability study was carried out on Sorafenib and $diC_{16}dT$ in SLNs at 4°C and 37°C. As shown on figure SI5, Sorafenib and $diC_{16}dT$ remain stable over a period of one month at both temperatures.



→ diC16dT 4°C ···· diC16dT 37°C – ★ – Sorafenib 4°C ···· ★··· Sorafenib 37°C

Figure SI5. Chemical stability of Sorafenib and diC₁₆dT in SLNs versus time at 4°C and 37° C.

The influence of dissolution solvent and temperature on DOTAU and diC_{16}dT was investigated.

Then, the impact of the nature of dissolution solvent was investigated. Figure SI6 A & B presents the stability of DOTAU and diC₁₆dT, respectively, in various solvents at 40°C. DOTAU was significantly more stable in ACN (only 10% of decrease over 3 days) than in H₂O (50% of decrease over 3 days), MeOH (up to 90% of decrease over 3 days) or EtOH (up to 85% of decrease over only 1 day) (Figure SI6.A). On the contrary, diC16dT appeared stable over time whatever the solvent nature (Figure SI6.B).



Figure SI6. Chemical studies of DOTAU (A) and diC16dT (B) at 40°C in different solvents during time.



Figure SI7. Average Size of SLN⁻ measure by DLS



Figure SI8. EDX spectra at SLN- (up) and on the copper grid (down)



Figure SI9. Colloidal stability of SLN⁻ measuring by DLS experiments



Figure SI10. Zeta potential after 7 days (left) and 30 days (right) at 4°C and 37°C for SLN⁺ (up) and SLN⁻ (down)



Figure SI11. Cell morphology assessed by phase contrast microscopy on carcinoma cell lines in the absence of Sorafenib (control), with 5 μ M of free Sorafenib, with SLN⁻ at 4 μ M and SLN⁺ at 2.8 μ M upon 4 days of treatment.



Figure SI12. Representative dot plots of cell granulosity (SSC-A) against PE-Texas Red-A detection after 4 days of HuH7 incubation with SLN+ loaded with Sorafenib at different concentrations (0, 1, 5, 10, 25, 50 and 100 μ M). The percentage of PE-Texas Red positive cells were quantified by FACS. Control corresponds to HuH7 without Ethidium homodimer-1 dye.



Figure SI13. HuH7 cells lines with SLN^+ at 50 μ M. Calcein were used to stain live cells and ethidium stain dead cells.