

Supporting Information

**Quantum Refinement Does Not Support Dinuclear Copper Sites in
Crystal Structures of Particulate Methane Monooxygenase**

*Lili Cao, Octav Caldararu, Amy C. Rosenzweig, and Ulf Ryde**

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Methods

Quantum refinement

Quantum refinement is standard crystallographic refinement supplemented by QM calculations for a small, but interesting part of the protein.^[1,2] Crystallographic refinement software change the protein model (coordinates, occupancies, B factors, etc.) to improve the fit of the observed and calculated structure-factor amplitudes (usually estimated as the residual disagreement, the R factor). Owing to the limited resolution normally obtained for biomolecules, the experimental data are supplemented by some chemical information, usually in the form of a MM force field.^[3] Then, the refinement takes the form of a minimization or simulated annealing calculation by molecular dynamics using an energy function of the form

$$E_{\text{cryst}} = w_A E_{\text{Xray}} + E_{\text{MM}} \quad (1)$$

where E_{Xray} is a penalty function that describes how well the model agrees with the experimental data (we have used a maximum-likelihood refinement target using amplitudes, MLF^[4,5]). E_{MM} is an empirical (molecular mechanics, MM) energy function with bond, angle, dihedral and non-bonded terms, and w_A is a weight factor, which is necessary because E_{MM} is in energy units whereas E_{Xray} is unit-less. It determines the relative importance of the crystallographic raw data and the MM force field for the final structure.

Quantum chemistry can be introduced in this function by replacing the MM potential for a small region of the protein (system 1) by QM calculations, yielding a QM energy for system 1, E_{QM1} . To avoid double counting, we must then subtract the MM energy of system 1, E_{MM1} ,

$$E_{\text{cqx}} = w_A E_{\text{Xray}} + E_{\text{MM12}} + w_{\text{QM}} E_{\text{QM1}} - E_{\text{MM1}} \quad (2)$$

Thereby, we introduce an accurate energy function for the system of interest. Such an energy function is implemented in the software ComQum-X,^[1] which is a combination of the QM software Turbomole^[6] and the crystallography and NMR system (CNS),^[7,8] version 1.3. The factor w_{QM} in Eq. 5 is another weight, which is needed because the CNS MM force field is based on a statistical analysis of crystal structures.^[9] Therefore, the force constants are not energy-derived, as is the QM term, but they are in arbitrary statistical units. Experience has shown that the CNS force constants are typically three times larger than energy-based force constants,^[9] and $w_{\text{QM}} = 3$ has therefore been used throughout this work.^[1] Crystallographic refinement is traditionally performed without any electrostatic interactions, because hydrogen atoms are not discerned in the structure. We have followed this custom and excluded electrostatics and hydrogen atoms from all crystallography and MM calculations (but they are of course included in the QM calculations).

Two crystal structures were studied in this investigation, viz. the revised pMMO structure from *M. capsulatus* (Bath) (3RGB) at 2.8 Å resolution^[10] and the 2.68-Å structure of pMMO from *Methylocystis sp.* strain M (3RFR).^[10] Coordinates, occupancies, B factors and structure factors were obtained from the protein databank files. From these files, we also obtained the space group, unit-cell parameters, resolution limits, R factors and the test set used for the evaluation of the R_{free} factor.

The full protein was used in all calculations, including all crystal water molecules. In each cycle of the geometry optimization, the surrounding protein was allowed to relax by one cycle of crystallographic minimization and one cycle of individual B-factor refinement. However, the new coordinates and B factors were accepted only if the R factor was reduced. The w_A factor was determined by CNS in the first iteration of the refinement and then kept fixed. It was 4.91 for 3RGB and 4.93 for 3RFR. Electron-density maps were generated using

phenix.maps.^[11]

We used the standard CNS MM force field for all atoms (protein_rep.param, water_rep.param, and ion.param). For the junctions, where a bond is broken between the QM and MM regions, we employed the hydrogen link-atom approach i.e. the QM systems were truncated by H atoms (called HL below), replacing a C atom in the protein (called CL). The force field for the QM system was constructed automatically from that of the protein. For the HL atoms, we employed the same angle and dihedral parameters as for the CL atoms. For the XL–HL bond (where XL is the atom in the QM system bound to HL), we used as the equilibrium bond length ($r_{\text{XL-HL}}^0$) that obtained by geometry optimisation with the TPSS-D3/def2-SV(P) approach for a model of the isolated ligand (e.g. imidazole). The force constant ($k_{\text{XL-HL}}$) was obtained from the corresponding force constant and equilibrium bond length of the XL–CL bond according to:

$$k_{\text{XL-HL}} = k_{\text{XL-CL}} \left(\frac{r_{\text{XL-HL}}^0}{r_{\text{XL-CL}}^0} \right)^2 \quad (3)$$

Van der Waals parameters of 0.0188 kJ/mol and 2.2272 Å were used for all junction atoms (taken from the HA atom type).

The quality of the models was compared using the real-space difference density Z-score^[12] (RSZD), calculated by EDSTATS in the CCP4 suite,^[13] which measures the local accuracy of the model. The maximum of the absolute values of the positive and negative RSZD (combined RSZD) for the Cu ion and all the first-sphere ligands was taken as the quality metric (individual values for each of the groups are given in the supplementary material). It is called RSZD in the main text and RSZD_{max} in this supplementary material (to differ it from RSZD for the individual residues and atoms, which also will be discussed). RSZD is typically < 3.0 for a good model. In addition, the strain energy of the active site (ΔE_{str}) was calculated, i.e. the difference in the QM energy of the QM region when optimised in the crystal or freely in vacuum.^[14]

QM calculations

All QM calculations were performed with the Turbomole software, version 7.1.^[6] We employed the density-functional theory (DFT) method TPSS^[15] and the def2-SV(P)^[16] basis set. The calculations were sped up by expanding the Coulomb interactions in an auxiliary basis set, the resolution-of-identity (RI) approximation.^[17,18] Empirical dispersion corrections were included with the DFT-D3 approach^[19] and Becke–Johnson damping,^[20] as implemented in Turbomole.

The QM system consisted of one or two Cu ions, two imidazole groups modelling the sidechains of His-137 and 139, and H₂N–(CH₂)₂–imidazole, as a model of His-33 (sidechain and backbone). Sometimes, one or two water molecules were added or models of Glu-35, Asp-79C (modelled by an acetate groups), Tyr-374 (modelled by phenol) or His-137 was extended to imidazole–(CH₂)₂–CO–NH–CH₃, including the backbone towards the next residue. The Cu ions were always modelled in both the Cu⁺ and Cu²⁺ states, in the latter case with unrestricted formalism. For dimeric Cu sites, we employed three oxidation states: fully reduced, fully oxidised and the mixed-valence Cu⁺–Cu²⁺ state. After some test calculations, we decided to use the ferromagnetically coupled triplet state for the fully oxidised state, which was lower in energy than the antiferromagnetically coupled open-shell singlet state (obtained with the broken-symmetry approach), owing to the lack of any strong bridging ligand.

For the study of the reaction mechanism (Figure 6 in the main article), structures with the same QM model (plus a methane molecule) were optimised in gas phase with the TPSS-D3/def2-SV(P) approach. Then, more accurate single-point energies were calculated with the def2-TZVPD^[21,22] basis set and both the TPSS-D3 and B3LYP-D3^[23–25] methods. These calculations were run in a COSMO continuum solvent.^[26,27] The default optimised COSMO radii (and a water solvent radius of 1.3 Å),^[28] whereas a radius of 2 Å was used for the metals.^[29] The dielectric constant was 4, whereas we used default values for all other parameters. Frequency calculations were performed at the TPSS-D3/def2-SV(P) level of theory and from these, zero-point energies were calculated and it was confirmed that transition states had one and minima had no imaginary frequencies.

Details of quantum refined structures of protomer 1 of the 3RGB structure

We started by performing quantum refinement of site A in the first protomer of the 3RGB structure using a dinuclear Cu site (called C2W0 in the tables to indicate that it contains two Cu ions and no extra water molecules). This gave structures quite similar to the starting crystal geometry as can be seen in Figures S1a and S1b: The two Cu ions are on top of each other, but in variance to the starting structure (Figure 1a in the main article), all three imidazole groups are directed towards the Cu ions with the ND1 (His-33 and His-137; called N_A and N_B in the following and in the Tables) or NE2 (His-139; N_C) atoms binding. One of the Cu ions (Cu_A) coordinates to all four N atoms, whereas the other (Cu_B) coordinates to only one or two of them (N_A and possibly also N_C). The distance to the His-33 backbone amide group (N_D) is typically longer than that to the imidazole groups. The detailed distances depend on the oxidation states of the two Cu ions and are listed in Table S1. In the (fully) reduced state, the two Cu_B–N distances (to N_A and N_C) are short (2.02–2.03 Å), whereas the four Cu_A–N distances are longer and much more varying (2.07–2.63 Å). This reflects the preference of Cu⁺ for low coordination numbers. However, in the mixed-valence (MV) and (fully) oxidised states, the Cu_A–N distances to the imidazole groups are short, 2.05–2.15 Å, that to N_D is somewhat longer, 2.24–2.30 Å, whereas only the Cu_B–N_A distance is short (2.08–2.13 Å) and the Cu_B–N_C bond is essentially broken, 2.89–2.93 Å. The Cu–Cu distance is 2.21–2.25 Å. The spin densities show that Cu_B is actually in the reduced state also in the fully oxidised structure. Instead, the spin is spread over Cu_A (0.34 *e*) and the ligands. Also in the MV state, there is no spin on Cu_B and only 0.41 *e* spin on Cu_A.

If we set $w_A = 0$ in Eqn. 2, i.e. turn off the restraint towards the experimental data, thereby obtaining a QM/MM calculation with the CNS MM force field, the MV and especially the oxidised structures reorganise significantly, as can be seen in Table S1 (method Q). However, the reduced structure remains more intact, with both Cu ions two-coordinate with Cu–N distances of 1.89–2.08 Å and a Cu–Cu distance of 2.46 Å. This reflects the intrinsic preferences of the metals and the ligands in the protein.

Further information about the preferences of the metal can be obtained by freely optimising the metals and the four ligands in vacuum, i.e. ignoring the covalent and steric influence of the protein (besides the covalent link between N_A and N_D). For the reduced site, this gave little changes from the QM/MM structure (cf. Table S1, method V). For the MV and oxidised states, the differences were larger, but mainly related to Cu_B, which bound with side-on interactions with the imidazole rings, because all N atoms were occupied in a square-planar coordination to Cu_A.

To obtain structures that better fitted the experimental data, we attempted to model the site in other ways. First, we tried to use the model suggested by Yoshizawa and coworkers in their QM/MM studies of the reaction mechanism,^[30–32] i.e. a dimeric model with the two Cu ions and the four N ligands in the same plane (Figure S2a). We could successfully obtain such a structure by model building and then refining the structure with $w_A = 0$. The structures are

described in Table S1 (called Y2W0a). Cu_A binds to His-33 (both sidechain and backbone amide) and His-137 with Cu_A-N = 1.97–2.09 Å. The latter group bridges to Cu_B, which also binds to His-139 with distances of 1.86–1.99 Å. The Cu–Cu distance is 2.35–2.45 Å, actually increasing with the oxidation state of the Cu ions. However, if we turned on the restraint towards the experimental data, the site went back to something similar to the original QM-refined structures, i.e. with the four N atoms mainly coordinating to Cu_A.

A prime problem with site A, especially in the higher oxidation states, is that it contains too few ligands for two metal ions. Therefore, we tried to enhance the site with two water molecules. After many attempts, we were able to obtain some reasonable structures and the best (in terms of RSZD_{max}) are shown in Tables S1 and S2 (called Y2W2), and in Figure 2b. In the MV and oxidised structures, Cu_A is still coordinated by all four N atoms in a square plane (Cu–N distances of 1.98–2.13 Å) and also by one of the water molecules in an axial position (2.52–2.53 Å). N_B and N_C bridges to Cu_B but with a quite poor geometry (2.07–2.19 Å). Cu_B also binds to the other water molecule (2.02–2.04 Å). The RSZD_{max} scores are higher for the reduced structure, but appreciably better for the MV and oxidised states (1.5) than the structures started from the original crystal structure. The strain energies are also appreciably better, 149–159 kJ/mol. However, the $2F_o - F_c$ electron-density maps in Figure 3a in the main article show that there is no electron density around the water molecules.

Consequently, we also optimised the corresponding structures without the water molecules (Y2W0 in Tables S1 and S2). They are more similar for all three oxidation states (except that N_C does not bind to Cu_A in the reduced state) with Cu_A-N distances of 1.91–2.19 Å and Cu_B-N distances of 1.90–2.28 Å to N_B and N_C. They gave slightly better RSZD_{max} scores (1.3 for the MV and oxidised states), but somewhat higher strain energies (170–220 kJ/mol).

Yoshizawa and coworkers have also sometimes suggested structures in which nearby residues coordinate to site A.^[30–32] Therefore, we tried to enhance the QM system with models of the sidechains Glu-35, Asp-79C or Tyr-374 and then repeated the QM refinement. However, it turned out that these residues were too far away in the crystal structure to coordinate to any of the Cu ions and therefore the structures were closely similar to those obtained without any added residues (Table S1). However, if the QM region was extended with the carbonyl group of His-137, this group was found to coordinate to Cu_B with Cu–O distances of 2.30–2.52 Å in all three oxidation states as can be seen in Figure S2c. This led to a shortening of the Cu_A-N_D bond (2.16–2.19 Å) and a slight elongation of the Cu_A-N_B bond (2.11–2.23 Å), but otherwise the structures were similar to the original QM-refined structures. Yet, these structures reproduced the crystallographic data quite poorly, with RSZD_{max} scores of 2.6–2.7 (Table S2). The strain energies were also nearly as high as for the original C2W0 structures, 178–226 kJ/mol.

Finally, we instead tried to model site A with a single Cu ion (C1W0 in Tables S1 and S2). This gave structures that were similar to that of Cu_A in the dinuclear sites. The geometry was distorted square planar with a rather small N_B-Cu-N_D angle of 117° (reduced state) to 133° (oxidised state), as can be seen in Figure S3a and S3b. Otherwise, the geometry was quite regular. In the oxidised state, the Cu–N distances were 1.85–2.22 Å, with the longest bond to N_D. In the reduced state, the three bonds to the sidechains were similar (1.84–2.23 Å), but the bond to N_D increased to 2.62 Å. If the same structures were optimised with $w_A = 0$, the structures became more square-planar with N_B-Cu-N_D angles of 139° and 165°. The Cu–N distances become more similar, 1.97–2.10 Å for the oxidised state and 1.96–2.12 (N_A-N_C) and 2.33 Å (N_D) for the reduced state. If the structures were instead optimised in vacuum, the reduced state became tetrahedral with all four Cu–N distances of 1.99–2.05 (N_A-N_C) and 2.14 Å (N_D), whereas the oxidised state became almost perfectly square planar (N_B-Cu-N_D angle of 169°) with Cu–N distances of 1.99–2.02 (N_A-N_C) and 2.07 Å (N_D). In all three oxidised

structures, the spin of the Cu ion was $\sim 0.5 e$.

There remains some positive difference density on top of the Cu ion in this structure, which was interpreted as an additional Cu ion in the original crystal structure. Yet, it is too close to the Cu ion and with a too low density to be another metal. Therefore, we tried to model it as a water molecule instead (C1W1 in Tables S1 and S2). In the reduced state, the water molecule did not coordinate to Cu (as could be expected from the preference of Cu^+ for low coordination numbers). Therefore, the structure did not change significantly from that obtained without the water molecule. However, for the oxidised state, the water molecule coordinated with a Cu–O distance of 2.20 Å as can be seen in Figure S4. As a consequence of the increased coordination number, the distance to the other ligands increased by 0.03–0.07 Å. If optimised with QM/MM or in vacuum, the Cu–O distance increased somewhat (2.25–2.31 Å) and the Cu–N distances became more similar.

The other protomers of the 3RGB structure

Next, we studied site A in the other two protomers of the 3RGB structure. As mentioned in the main text, the structure of these two sites is different from that in the first protomer. This was also reflected in the quantum-refined structures of this site, started from the crystal structure with two Cu ions (C2W0): For all three oxidation states, strongly distorted structures were obtained with Cu_A coordinating to His-137 and bidentately to His-33 as can be seen in Figure S5a. His-137 bridges to Cu_B , which also interacts with the sidechain of His-139, but with a far-from-ideal, side-on geometry. The corresponding distances are listed in Table S1. The RSZD_{max} scores were worse than in protomer 1, 2.8–2.9, whereas the strain energies were similar, 178–247 kJ/mol (Table S2). Protomer 3 gave similar structures (also described in Tables S1 and S2). However, the RSZD_{max} scores were somewhat smaller, 2.0–2.8, as was the strain energies (160–224 kJ/mol).

To check whether this striking difference in the metal-site geometry is really dictated by the crystallographic data, we used the trifold non-crystallographic symmetry to copy the structure of protomer 1 to site A in protomer 2 and 3 also. However, the structures converged back to the original structures again.

Next, we also optimised Yoshizawa-type dinuclear structures in the other two protomers (Y2W2). They are also presented in Tables S1, S2 and Figure S5b, and it can be seen that the structures are rather similar to those in protomer 1, but with some variation. For protomer 2, these structures fitted the crystallographic data better than the other dinuclear structures with RSZD_{max} scores of 1.9–2.2 and strain energies of 110–144 kJ/mol. However, for protomer 3, they were much worse in terms of RSZD_{max} (2.9–3.2), but the strain energies were still lower (104–160 kJ/mol).

We also optimised the corresponding structures without water molecules (Y2W0). From the results in Tables S1 and S2, it can be seen that they were similar to the corresponding structures with water, except that for Cu_B in protomer 3, N_D was replaced by N_A . In protomer 3, these structures gave lower RSZD_{max} scores than the structures with water, but higher strain energies, whereas both measures were worse for protomer 2.

Next, we tried to model a single Cu ion in both protomers. This gave structures that were similar to the corresponding structures in protomer 1 (Figure S5c). The Cu–N distances (shown in Table S1) were also rather similar. The electron-density maps were also better than those of the dinuclear Cu model, with RSZD_{max} of 1.1–1.7 for protomer 2 and 1.5–2.0 for protomer 3. The strain energies were also much lower than for the dinuclear structures, 37–76 kJ/mol.

Finally, we also tried a single Cu ion with an additional water molecule. It gave results similar to that of protomer 1, with the difference that the water molecule did not coordinate to

Cu in protomer 3. However, in both protomers, the structures without the axial water molecule gave lower $RSZD_{\max}$ scores and strain energies (Table S2).

The 3RFR crystal structure

Finally, we considered the 3RFR crystal structure. As mentioned in the introduction, it contains a dinuclear site A only in protomer 1, whereas this site is mononuclear in the other two protomers. Therefore, we performed quantum-refinement also of that site, using the 3RFR crystallographic data. With two Cu ions in the oxidised state (C2W0), the dinuclear site gave a structure similar to that of the first protomer in the 3RGB crystal (Table S1). However, for the reduced state, the structure was distorted and less similar. The side chains of His-137 and 139 coordinate primarily to Cu_A , whereas both N_A and N_D coordinate primarily to Cu_B , although all atoms except N_B are within 3 Å also of the other Cu ion. The Cu–Cu distance is 2.24 Å. These structures fitted the crystallographic raw data quite poorly with $RSZD_{\max}$ scores of 1.6–1.9 and very high strain energies, 292–357 kJ/mol.

However, much more reasonable structures were obtained with a single Cu ion in this site. Both the reduced and oxidised states gave almost perfectly square-planar structures with N_B –Cu– N_D angles of 171–177°. The Cu–N distances were similar to those of the 3RGB structure, 1.94–2.40 Å for the reduced state and 1.96–2.15 Å for the oxidised state (Table S1). These structures also fitted the experimental data better, with $RSZD_{\max} = 1.3$ –1.4 and strain energies of 32 (oxidised state) or 88 kJ/mol (reduced state). Structures with a coordinated water molecule could also be obtained for both oxidation states (Cu–O distances of 2.43–2.50 Å). However, they gave larger $RSZD_{\max}$ scores of 2.3 and 2.6.

For the other two protomers, the original crystal structure contained mononuclear sites. This was confirmed by the quantum-refinement, which gave structures similar to the other mononuclear sites in the 3RGB or 3RFR structures (Tables S1 and S2). However, no structures were obtained with a bound water molecule, not even for the oxidised state. The structures fitted the experimental data appreciably better than for protomer 1, with $RSZD_{\max} = 0.4$ –0.9 without the water molecule. On the other hand, the strain energies were similar, 37–47 kJ/mol for the oxidised state and 89–101 kJ/mol for the reduced state.

Discussion of different $RSZD$ scores

In this study, we have used quantum refinement to re-refine the putative dinuclear Cu site A in two crystal structures of pMMO. We have tried several different dinuclear structures, as well as a mononuclear structure in all three protomers in both enzymes and also tried to add one or two extra water molecules. We have observed some variation among the two crystal structures and the three protomers.

In terms of the strain energies, our results are quite conclusive (Table 1 in the main article): The mononuclear models always give lower strain energies than the dinuclear models (37–85 kJ/mol, compared to 104–300 kJ/mol). For the same crystal and protomer, the best mononuclear structure always give a 63–85 kJ/mol lower strain energy than the best dinuclear structure. This confirms the impression that the dinuclear structures are not reasonable, even after the quantum refinement.

The $RSZD_{\max}$ results are slightly less conclusive (Tables 1 and S2). For the 3RFR structure and protomers 2 and 3 of the 3RGB structure, the mononuclear sites still give the lowest scores (1.3 compared to 1.6 in 3RFR, 1.1 compared to 1.9 in protomer 2 and 1.5 compared to 1.8 for protomer 3 of 3RGB). However, for protomer 1 in 3RGB, the lowest scores are actually obtained for the Y2W0 structure (1.3, compared to 1.5 for the best mononuclear site). On the other hand, it is not entirely clear how to compare $RSZD$ scores

between sites of different composition. The RSZD scores are calculated for individual residues (viz. for His-33, 137 and 139, as well as for the Cu ions and water molecules; the individual RSZD results are shown in Table S2). These are not the same for all systems and they do not cover the same space in the various structures. Moreover, it is unclear how to combine 4–7 RSZD results to a single score in an unambiguous manner: We have used the maximum value ($RSZD_{max}$, called simply RSZD in the main article), but other measures are also conceivable. Table S2 shows two additional scores, obtained from the sum or the average. In fact, the mononuclear site gives the lowest RSZD score if the sum is considered (3.1, compared to 3.4), whereas the Y2W2 structure gives the best results if the average is considered (0.5 compared to 0.7).

A third argument against a dinuclear site is the large variation of the dinuclear structures in the various protomers and crystals: The geometries vary extensively (cf. Table S1) and the best one also varies between the crystals and protomers (the Y2W0 structure is best in protomer 1, the Y2W2 in protomer 2, whereas C2W0 is best in protomer 3 of the 3RGB structures).

Taken all results together, we find no conclusive evidence that the crystal structures of pMMO show a dinuclear Cu site. Instead, a mononuclear site fits the experimental data at least as well and it also gives chemically much more reasonable structures that are similar in both crystals and all protomers. Such a site is more similar to the mononuclear Cu site in the lytic polysaccharide monooxygenases, although it contains an additional histidine residue and lacks the nearby tyrosine residue.^[33] The main evidence for a dinuclear site comes from EXAFS investigations, showing a striking reproducibility of the Cu–Cu interaction in pMMOs from four different methanotrophs.^[10,34–36] One possibility is that the dicopper site detected by EXAFS is disrupted upon crystallization. On the other hand, EXAFS fits are known to give multiple local minima, so that different structures can give equally good fits.^[37,38] This will be examined in future studies by a similar combination of computational and EXAFS methods.^[39]

Reaction mechanism

To provide a further check of the suggested mononuclear Cu site, we studied the reaction of the suggested mononuclear active site with a molecule of methane. Yoshizawa and Shiota have already studied with QM method putative reaction mechanisms of mono- and dinuclear Cu sites in pMMO.^[30] However, their mononuclear model came from site B in the protein and therefore contained two His side chains and an acetate group and no His brace or amino group. Therefore, we repeated their study using our preferred mononuclear site in Figure 4 in the main article, i.e. with three His side chains, one of which is connected to a terminal amino group, as is shown in Figure 6 in the main article. Geometries were optimised at the TPSS-D3/def2-SV(P) level of theory in gas phase. Then, more accurate energies were calculated with the def2-TZVDP basis set with both the TPSS-D3 and B3LYP-D3 methods in a COSMO continuum solvent with a dielectric constant of 4. All energies are given in Table S3. Reported energies in Figure 6 and in the text are B3LYP-D3/def2-TZVPD energies, including the zero-point energy, obtained from frequencies at the TPSS-D3/def2-SV(P) level of theory.

Following the previous study, we started with a Cu(III) ion with an oxo ligand (or equivalently Cu(II)–O^{•-}; net charge +1; called RS in Figure 6). All structures were studied in both the singlet and triplet states. In both cases, there was 0.5 spin on the Cu ion, 0.05–0.06 spins on each His ligand and the remainder on O (0.7 for the singlet and 1.3 for the triplet). In the singlet, the O spin was opposite to that of the other atoms and groups. Unlike the study by Yoshizawa and Shiota,^[30] the two states were not degenerate; instead the triplet was 85 kJ/mol more stable than the singlet. Next, a methane molecule was added to the complex, forming a weak hydrogen bond to the oxo group (2.58 Å CH–O distance; R). It had nearly the

same energy as the RS structure (plus a free methane molecule).

A transition state (TS1) for the transfer of one of the hydrogen atoms of the methane molecule could be located. It had a CH–O distance of 1.17 Å and was 22 kJ/mol higher in energy than RS for the triplet. It is characterised by a single imaginary frequency of $1475i\text{ cm}^{-1}$, similar to that found by Yoshizawa and Shiota ($1445i\text{ cm}^{-1}$).^[30] This transition state leads to an intermediate with Cu–OH interacting with a methyl radical (Im1). It had a O–H distance of 0.99 Å and a OH–C distance of 2.18 Å and it was 16 kJ/mol more stable than RS. The Cu ion still had 0.5 spin, whereas 0.3 spin resided on OH and there was 0.95 spin on the methyl radical.

This intermediate could rearrange essentially barrierless to a second intermediate (Im2) in which the methyl radical also binds to the Cu ion. In this intermediate, the spins on Cu and OH have increased slightly, whereas that on the methyl radical has decreased strongly to 0.6. For the triplet state, Im2 is 10 kJ/mol less stable than Im1. However, for the singlet-state structures (which up to Im1 has been 75–150 kJ/mol less stable than the triplet states), Im2 is strongly stabilised and actually becomes 46 kJ/mol more stable than the triplet state (and 52 kJ/mol more stable than ³RS). Therefore, we suggest, in accordance with the study of Yoshizawa and Shiota,^[30] that the spin state changes from triplet to singlet at this point. ¹Im2 is a closed-shell singlet, indicating a Cu(III)–OH[–]–CH₃[–] complex.

Finally, the methanol product (P) can form by a recombination of OH and the methyl group. The corresponding transition state (TS2) has a rather complicated structure because both groups are still bound to Cu. The C–O distance is 1.91 Å and the imaginary frequency is $396i\text{ cm}^{-1}$, slightly smaller than that obtained by Yoshizawa and Shiota ($473i\text{ cm}^{-1}$).^[30] It is still a closed-shell species and it is 53 kJ/mol higher in energy than ¹Im2. In the product, methanol has dissociated from the Cu(I) ion and is 217 kJ/mol more stable than the starting point (³RS).

From the energy components in Table S3, it can be seen that the effect of an increase in the basis set from def2-SV(P) to def2-TZVPD (for the TPSS-D3 method) is quite small for the triplet state (up to 17 kJ/mol, in general reducing the relative energies), whereas it is larger for the R, TS1, and Im1 states (up to 67 kJ/mol), increasing the energies. Likewise, the effect of changing the DFT functional from TPSS to B3LYP has a relatively small effect on the triplet energies (up to 24 kJ/mol, increasing the energies), but a much larger effect on the singlet energies (again up to 67 kJ/mol), although it is mainly an effect of decreasing the relative energy of all open-shell singlet states by 50–67 kJ/mol. Thus, the singlet and triplet states are much closer in energy at the TPSS level of theory. The zero-point energy has a rather small effect on all energies, up to 17 kJ/mol with a varying sign (but nearly the same energy for the singlet and triplet states).

References

- [1] U. Ryde, L. Olsen, K. Nilsson, *J. Comput. Chem.* **2002**, *23*, 1058–1070.
- [2] U. Ryde, K. Nilsson, *J. Am. Chem. Soc.* **2003**, *125*, 14232–14233.
- [3] G. J. Kleywegt, T. A. Jones, *Acta Crystallogr. Sect. D Biol. Crystallogr.* **1998**, *54*, 1119–1131.
- [4] N. S. Pannu, R. J. Read, *Acta Crystallogr. Sect. A Found. Crystallogr.* **1996**, *A52*, 659–668.
- [5] P. D. Adams, N. S. Pannu, R. J. Read, A. T. Brünger, *Proc. Natl. Acad. Sci. U. S. A.* **1997**, *94*, 5018–5023.
- [6] F. Furche, R. Ahlrichs, C. Hättig, W. Klopper, M. Sierka, F. Weigend, *Wiley Interdiscip. Rev. Comput. Mol. Sci.* **2014**, *4*, 91–100.
- [7] A. T. Brünger, P. D. Adams, G. M. Clore, W. L. Delano, P. Gros, R. W. Grosse-kunstleve, J. Jiang, J. Kuszewski, M. Nilges, N. S. Pannu, et al., *Acta Crystallogr. Sect.*

- D Biol. Crystallogr.* **1998**, *54*, 905–921.
- [8] A. T. Brunger, *Nat. Protoc.* **2007**, *2*, 2728–33.
- [9] R. A. Engh, R. Huber, *Acta Crystallogr. Sect. A* **1991**, *47*, 392–400.
- [10] S. M. Smith, S. Rawat, J. Telser, B. M. Hoffman, T. L. Stemmler, A. C. Rosenzweig, *Biochemistry* **2011**, *269*, 26344–26348.
- [11] P. V. Afonine, W. Ralf, J. J. Headd, C. Thomas, *Acta Crystallogr. Sect. D Biol. Crystallogr.* **2012**, *68*, 352–367.
- [12] I. J. Tickle, *Acta Crystallogr. Sect. D Biol. Crystallogr.* **2012**, *68*, 454–467.
- [13] M. D. Winn, C. Charles, K. D. Cowtan, E. J. Dodson, A. G. W. Leslie, A. McCoy, J. Stuart, N. Garib, H. R. Powell, J. Randy, *Acta Crystallogr. Sect. D Biol. Crystallogr.* **2011**, *4449*, 235–242.
- [14] U. Ryde, *Dalton Trans.* **2007**, 607–625.
- [15] J. Tao, J. P. Perdew, V. N. Staroverov, G. E. Scuseria, *Phys. Rev. Lett.* **2003**, *91*, 146401.
- [16] A. Schäfer, H. Horn, R. Ahlrichs, *J. Chem. Phys.* **1992**, *97*, 2571–2577.
- [17] K. Eichkorn, O. Treutler, H. Öhm, M. Häser, R. Ahlrichs, *Chem. Phys. Lett.* **1995**, *240*, 283–289.
- [18] K. Eichkorn, F. Weigend, O. Treutler, R. Ahlrichs, *Theor. Chem. Acc.* **1997**, *97*, 119–124.
- [19] S. Grimme, J. Antony, S. Ehrlich, H. Krieg, *J. Chem. Phys.* **2010**, *132*, 154104 (19 pages).
- [20] S. Grimme, S. Ehrlich, L. Goerigk, *J. Comput. Chem.* **2011**, *32*, 1456–1465.
- [21] F. Weigend, R. Ahlrichs, *Phys. Chem. Chem. Phys.* **2005**, *7*, 3297–305.
- [22] D. Rappoport, F. Furche, *J. Chem. Phys.* **2010**, *133*, 134105 (11 pages).
- [23] A. D. Becke, *Phys. Rev. A* **1988**, *38*, 3098–3100.
- [24] C. Lee, W. Yang, R. G. Parr, *Phys. Rev. B* **1988**, *37*, 785–789.
- [25] A. D. Becke, **2005**, *5648*, 5648–5652.
- [26] A. Klamt, G. Schüürmann, *J. Chem. Soc. - Perkin Trans. 2* **1993**, 799–805.
- [27] A. Schäfer, A. Klamt, D. Sattel, J. C. W. Lohrenz, F. Eckert, *Phys. Chem. Chem. Phys.* **2000**, *2*, 2187–2193.
- [28] A. Klamt, V. Jonas, T. Bürger, J. C. W. Lohrenz, *J. Phys. Chem. A* **1998**, *102*, 5074–5085.
- [29] E. Sigfridsson, U. Ryde, *J. Comput. Chem.* **1998**, *19*, 377–395.
- [30] K. Yoshizawa, Y. Shiota, *J. Am. Chem. Soc.* **2006**, *128*, 9873–9881.
- [31] Y. Shiota, K. Yoshizawa, *Inorg. Chem.* **2009**, *48*, 838–845.
- [32] S. Itoyama, K. Doitomi, T. Kamachi, Y. Shiota, K. Yoshizawa, *Inorg. Chem.* **2016**, *55*, 2771–2775.
- [33] W. T. Beeson, V. V. Vu, E. A. Span, C. M. Phillips, M. A. Marletta, *Annu. Rev. Biochem.* **2014**, *84*, 923–946.
- [34] R. L. Lieberman, D. B. Shrestha, P. E. Doan, B. M. Hoffman, T. L. Stemmler, A. C. Rosenzweig, *Proc. Natl. Acad. Sci. U. S. A.* **2003**, *100*, 3820–3825.
- [35] A. S. Hakemian, K. C. Kondapalli, J. Telser, B. M. Hoffman, T. L. Stemmler, A. C. Rosenzweig, *Biochemistry* **2008**, *47*, 6793–6801.
- [36] R. L. Lieberman, K. C. Kondapalli, D. B. Shrestha, A. S. Hakemian, S. M. Smith, J. Telser, J. Kuzelka, R. Gupta, A. S. Borovik, S. J. Lippard, et al., *Inorg. Chem.* **2006**, *45*, 8372–8381.
- [37] X. Li, P. E. M. Siegbahn, U. Ryde, *Proc. Natl. Acad. Sci. U. S. A.* **2015**, *112*, 3979–84.
- [38] Y. W. Hsiao, U. Ryde, *Inorganica Chim. Acta* **2006**, *359*, 1081–1092.
- [39] Y. W. Hsiao, Y. Tao, J. E. Shokes, R. A. Scott, U. Ryde, *Phys. Rev. B - Condens. Matter Mater. Phys.* **2006**, *74*, 1–17.

Table S1. Cu–ligand distances (Å) in the various complexes. The results for different crystal structures and protomers are presented in separated sections of the table. The structures are described with a four-letter code, giving the number of Cu ions (C2 and C1; Y2 represent Yoshizawa-like structures with two Cu ions, such as those in Figure S2) and the number of water molecules, W2, W1 or W0 (the Glu35, Asp79, Tyr374 and His137 structures of 3RGB, subunit A all represent C2W0 structures, but with enhanced QM regions). The methods are quantum refinement (R), QM/MM (Q; i.e. quantum refinement with $w_A = 0$) and vacuum optimisations (V). State is the oxidation state of the Cu ion(s), reduced (Red; one or two Cu^I), oxidised (Ox; one or two Cu^{II}) or mixed valence (MV; Cu^I+Cu^{II}). Cu–ligand distances are reported only if they were shorter than 3.0 Å. N_A, N_B and N_C are the coordinating sidechain atoms of His-33, His-137 and His-139, respectively. N_D is the backbone (amino-terminal) N atom or His-33. O₁ and O₂ are the O atoms of the water molecules that are present in some structures. In the structure with the QM region enhanced with the backbone of His137, O₁ represents the carbonyl oxygen.

Structure	Method	State	Distance to Cu _A						Distance to Cu _B							
			Cu _B	N _A	N _B	N _C	N _D	O ₁	O ₂	N _A	N _B	N _C	N _D	O ₁	O ₂	
3RGB, protomer 1																
C2W0	R	Red	2.21	2.34	2.07	2.63	2.25				2.02		2.03			
		MV	2.25	2.07	2.10	2.06	2.24				2.08		2.89			
		Ox	2.25	2.05	2.15	2.06	2.30				2.13		2.93			
	Q	Red	2.46	2.76	2.01		2.08				1.89		1.89			
		MV	2.69	2.15	2.03	2.00	2.09				1.94					
		Ox	5.49	2.05	2.13	2.05	2.09									
	V	Red	2.54	2.72	1.91		1.96				1.90		1.89			
		MV	3.63	2.05	2.06	2.00	2.07				2.67	2.64				
		Ox	5.67	2.09	2.06	2.09	2.07									
Y2W0a	Q	Red	2.35	1.97	2.07		2.09				1.97	1.87				
		MV	2.43	1.97	2.08		2.07				1.99	1.86				
		Ox	2.45	2.01	2.07		2.08				1.99	1.87				
	R	Red	2.27	2.01	2.01	1.93					2.09		1.98			
		MV	2.59	1.91	2.17	2.09	2.02					2.15	2.11			
		Ox	3.54	1.98	2.03	2.03	2.02									
	V	Red	2.41	2.00	2.02		2.01					1.97	1.90			
		Red	2.51	2.01	2.06	1.99	2.99	2.23			2.17			2.02	2.08	2.07
		MV	2.51	1.94	2.07	2.12	1.98	2.53				2.19	2.09			2.04
Y2W2	R	Ox	2.51		2.19	2.09		2.04			1.94	2.07	2.12	1.98		2.53
		Ox,alt	2.60	1.98	2.06	2.13	1.98	2.52				2.19	2.07			2.02
		Red	2.56	1.98	2.08	1.96					2.50			1.95		1.94
	V	MV	2.46		2.15	1.96		2.06	2.18		1.98	2.01		2.00		2.89
		Ox	3.66			2.08		2.00	2.09		2.06	2.00	2.39	2.02		
		Red	2.18	1.94	2.19		1.95					1.90	1.83			
	Y2W0	R	MV	2.42	1.91	2.08	2.13	2.00				2.21	2.11			
			Ox	2.47	1.98	2.05	2.13	2.00				2.28	2.11			
			Red	2.41	2.00	2.02		2.01				1.97	1.90			
V		MV	3.93	1.98	2.10	2.10	2.06				2.85	2.83				
		Ox	3.89	2.12	2.07	2.08	2.06				2.89	2.83				
		Red	2.21	2.45	2.09	2.61	2.15				2.01		2.01	2.94		
Glu35		MV	2.21	2.34	2.07	2.62	2.17				2.03		2.03	2.94		
		Ox	2.20	2.26	2.07	2.57	2.16				2.05		2.08	2.95		
		Red	2.21	2.34	2.07	2.64	2.25				2.01		2.02			
Asp79	MV	2.21	2.34	2.07	2.64	2.25				2.01		2.02				
	Ox	2.25	2.09	2.10	2.09	2.25				2.06		2.79				
	Ox	2.25	2.09	2.10	2.09	2.25				2.06		2.79				

Tyr374	R	Red	2.21	2.35	2.06	2.64	2.24		2.01	2.02			
		MV	2.21	2.35	2.06	2.64	2.24		2.01	2.02			
		Ox	2.21	2.27	2.06	2.62	2.24		2.03	2.04			
His137	R	Red	2.25	2.23	2.11	2.49	2.19		2.09	2.19	2.52		
		MV	2.30	2.01	2.21	2.03	2.16		2.14	2.97	2.30		
		Ox	2.28	2.01	2.23	2.05	2.18		2.15	2.94	2.39		
	V	Red	2.49	2.90	1.92		1.98		1.93	1.93	2.13		
		MV	2.42		1.97	2.02	2.03	2.52	1.88		2.91	1.89	
		Ox	2.55		1.99	2.07	2.05	2.31	1.91		2.90	1.90	
C1W0	R	Red		1.84	2.23	1.90	2.62						
		Ox		1.85	2.13	1.94	2.22						
	Q	Red		1.96	2.12	2.02	2.33						
		Ox		1.97	2.06	2.01	2.10						
	V	Red		2.01	1.99	2.05	2.14						
		Ox		1.99	2.02	2.02	2.07						
C1W1	R	Red		1.87	2.25	1.93	2.55	3.09					
		Ox		1.89	2.16	1.99	2.29	2.20					
	Q	Ox		2.00	2.08	2.03	2.09	2.25					
		Red		2.28	2.01	2.02	2.06						
	V	Ox		2.02	2.03	2.04	2.08	2.31					
3RGB, protomer 2													
C2W0	R	Red	2.29	1.85		2.12	2.18		1.97	2.27			
		MV	2.36	1.84		2.12	2.13		1.95	2.23			
		Ox	2.40	1.86		2.13	2.13		1.96	2.25			
	V	Red	2.42	1.93		2.01	2.09		1.90	1.97			
		MV	2.55	1.94		2.04	2.07		1.94	1.96			
		Ox	3.62	2.00		2.10	2.07		1.94	2.02			
	Y2W2	R	Red	2.51	1.92	2.06	2.03	2.34	2.83	2.88	2.47	1.95	1.94
			MV	2.96	1.89	2.07	2.05	2.18	2.28	2.31	2.14	1.93	
			Ox	3.00	1.92	2.07	2.07	2.19	2.30	2.28	2.14	1.93	
V		Red	2.59	2.34	1.93		1.99	2.96		1.91	2.16	1.95	
		MV	2.47	1.93		1.97	2.08	2.83	1.93	2.49	2.12	2.03	
		Ox	2.96	2.07	2.10	2.24	2.07		2.21	2.07	2.04	2.05	
Y2W0	R	Red	2.45	1.87	2.26	1.96	2.30		2.39	2.09			
		MV	2.92	1.87	2.07	2.01	2.15		2.45	2.24			
		Ox	3.12	1.93	2.05	2.01	2.13		2.59	2.36			
	V	Red	2.41	1.92		2.01	2.10		2.75	2.02			
		MV	3.89	1.98	2.10	2.10	2.07		2.84	2.82			
		Ox	3.90	2.12	2.07	2.09	2.06		2.90	2.83			
C1W0	R	Red		1.88	2.13	1.95	2.49						
		Ox		1.89	2.04	2.01	2.20						
	V	Red		2.01	1.98	2.05	2.14						
		Ox		1.99	2.02	2.02	2.07						
C1W1	R	Ox		1.91	2.02	2.02	2.21	2.37					
	V	Ox		2.02	2.03	2.04	2.08	2.34					
3RGB, protomer 3													
C2W0	R	Red	2.29	1.98		2.24	2.03		1.95	2.04			
		MV	2.30	1.95		2.17	2.00		1.93	2.06			
		Ox	2.33	1.98		2.15	2.00		1.93	2.08			
	V	Red	2.41	1.93		2.01	2.09		1.90	1.97			
		MV	2.56	1.94		2.03	2.07		1.89	1.95			
		Ox	3.62	2.00		2.10	2.07		1.94	2.02			
	Y2W2	R	Red	2.47	2.21	2.04	2.01	2.21	2.89	2.93	2.65	1.94	1.92
			MV	3.94	2.00	2.02	2.01	2.11			2.72	1.89	
			Ox	3.93	2.07	2.01	2.01	2.11			2.74	1.92	

	V	Red	2.59	2.32	1.93	1.99	2.95					
		MV	4.76	1.99	2.02	2.10	2.04			1.91	2.17	1.95
		Ox	4.93	2.08	2.05	2.10	2.04				1.93	
Y2W0	R	Red	2.45	2.04	2.35	1.95	2.19			2.38	2.06	
		MV	2.77	2.01	2.08	2.00	2.11			2.36	2.23	
		Ox	2.86	2.09	2.06	1.99	2.11			2.47	2.29	
	V	Red	2.42	1.93		2.01	2.09			1.90	1.97	
		MV	3.90	1.98	2.11	2.10	2.07			2.84	2.82	
		Ox	3.90	2.12	2.07	2.08	2.06			2.90	2.83	
C1W0	R	Red		2.03	2.19	1.92	2.27					
		Ox		2.00	2.02	1.97	2.12					
	V	Red		2.01	1.99	2.05	2.14					
		Ox		1.99	2.02	2.02	2.07					
C1W1	R	Ox		2.02	2.01	1.98	2.12	3.05				
	V	Ox		2.02	2.03	2.04	2.08	2.34				
3RFR, protomer 1												
C2W0	R	Red	2.24	2.73	2.06	1.92	2.93			2.09	2.79	2.34
		MV	2.34	2.23	2.07	1.92	2.54			2.36	3.00	2.46
		Ox	2.32	2.34	2.11	1.93	2.70			2.32	2.94	2.17
C1W0	R	Red		2.12	2.26	1.94	2.39					
		Ox		2.05	2.09	1.96	2.15					
C1W1	R	Red		2.06	2.19	1.92	2.76	2.43				
		Ox		2.04	2.08	1.96	2.14	2.50				
3RFR, protomer 2												
C1W0	R	Red		2.12	2.08	2.06	2.56					
		Ox		2.01	2.01	2.03	2.30					
C1W1	R	Ox		2.02	2.01	2.03	2.28					
3RFR, protomer 3												
C1W0	R	Red		2.00	2.31	1.96	2.34					
		Ox		2.00	2.10	1.96	2.16					
C1W1	R	Ox		2.01	2.08	1.96	2.15					

Table S2. RSZD scores for each ligand in the active site, together with the maximum, sum and average. The last two columns give R_{free} and R of the structures. PM, Struct and State are the protomer, the structure and the oxidation state of site A.

PM	Struct	State	RSZD									R_{free}	R		
			H33	H137	H139	Cu _A	Cu _B	Wat ₁	Wat ₂	Max	Sum			Av	
3RGB crystal structure															
1	start		1.0	2.0	0.8	4.7					4.7	8.5	2.1		
	C2W2	Red	0.2	2.0	0.8	2.1	1.5				2.1	6.6	1.3	0.2831	0.2627
		MV	0.4	1.8	0.8	0.9	1.9				1.9	5.8	1.2	0.2832	0.2627
		Ox	0.5	1.9	0.9	0.8	2.1				2.1	6.2	1.2	0.2832	0.2627
	Y2W2	Red	1.3	1.6	0.4	0.0	0.8	0.4	3.2	3.2	7.7	1.1	0.2879	0.2437	
		MV	1.0	1.5	0.5	0.1	0.6	0.2	0.3	1.5	4.2	0.6	0.2878	0.2437	
		Ox	0.6	1.5	0.4	0.1	0.4	0.1	0.3	1.5	3.4	0.5	0.2878	0.2437	
		Ox _{,alt}	3.0	2.5	4.3	1.4	4.2	0.7	2.0	4.3	18.1	2.6	0.2879	0.2439	
	Y2W0	Red	1.8	1.5	1.8	2.7	0.9			2.7	8.7	1.7	0.2881	0.2439	
		MV	0.7	1.3	0.4	1.2	0.9			1.3	4.5	0.9	0.2879	0.2438	
		Ox	0.8	1.3	0.6	1.1	0.6			1.3	4.4	0.9	0.2879	0.2438	
	His137	Red	0.3	2.7	1.4	2.5	1.4			2.7	8.3	1.7	0.2831	0.2627	
		MV	1.5	2.7	0.8	1.9	2.6			2.7	9.5	1.9	0.2831	0.2627	
		Ox	0.4	2.3	0.7	1.5	2.6			2.6	7.5	1.5	0.2832	0.2627	
	C1W0	Red	0.2	1.7	0.3	0.9				1.7	3.1	0.8	0.2824	0.2627	
		Ox	0.3	1.6	0.4	1.2				1.6	3.5	0.9	0.2823	0.2626	
	C1W1	Red	0.3	1.7	0.2	0.7		0.6		1.7	3.5	0.7	0.2827	0.2626	
		Ox	0.3	1.5	0.3	0.7		0.5		1.5	3.3	0.7	0.2822	0.2626	
2	C2W2	Red	1.3	1.5	1.4	2.0	2.8			2.8	9.0	1.8	0.2832	0.2626	
		MV	1.3	1.5	1.6	2.1	2.9			2.9	9.4	1.9	0.2832	0.2626	
		Ox	1.3	1.5	1.6	2.1	2.9			2.9	9.4	1.9	0.2832	0.2626	
	Y2W2	Red	1.7	0.3	1.6	0.7	0.5	0.8	2.2	2.2	7.8	1.1	0.2832	0.2626	
		MV	1.2	0.7	1.3	1.6	0.4	0.5	1.9	1.9	7.6	1.1	0.2822	0.2626	
		Ox	1.2	0.6	1.5	1.5	0.3	0.5	2.1	2.1	7.7	1.1	0.2823	0.2626	
	Y2W0	Red	1.7	0.4	0.8	2.5	0.4			2.5	5.8	1.2	0.2823	0.2625	
		MV	1.2	0.6	0.6	2.7	0.8			2.7	5.9	1.2	0.2823	0.2625	
		Ox	1.4	0.7	0.7	2.8	1.0			2.8	6.6	1.3	0.2823	0.2625	
	C1W0	Red	1.1	0.5	0.4	0.7				1.1	2.7	0.7	0.2823	0.2625	
		Ox	1.7	0.5	0.4	0.3				1.7	2.9	0.7	0.2823	0.2625	
	C1W1	Ox	1.1	0.4	1.1	0.4		2.4		2.4	5.4	1.1	0.2822	0.2625	
3	C2W2	Red	0.3	2.0	1.4	0.1	1.9			2.0	5.7	1.1	0.2830	0.2627	
		MV	0.5	1.9	1.5	0.5	2.6			2.6	7.0	1.4	0.2830	0.2627	
		Ox	0.5	1.8	1.6	0.5	2.8			2.8	7.2	1.4	0.2830	0.2627	
	Y2W2	Red	0.3	1.5	0.0	0.1	1.4	0.2	3.0	3.0	6.5	0.9	0.2828	0.2627	
		MV	0.3	1.5	0.2	0.9	1.6	0.0	3.2	3.2	7.7	1.1	0.2821	0.2627	
		Ox	0.5	1.5	0.2	0.9	1.6	0.0	2.9	2.9	7.6	1.1	0.2821	0.2627	
	Y2W0	Red	0.3	1.5	0.3	1.0	1.8			1.8	4.9	1.0	0.2827	0.2627	
		MV	0.2	1.5	0.4	0.9	2.1			2.1	5.1	1.0	0.2819	0.2627	
		Ox	0.2	1.6	0.5	0.9	2.2			2.2	5.4	1.1	0.2819	0.2627	
	C1W0	Red	0.1	1.5	0.3	1.0				1.5	2.9	0.7	0.2822	0.2626	
		Ox	0.4	1.6	0.0	1.3				1.6	3.3	0.8	0.2829	0.2627	
	C1W1	Ox	0.4	1.6	0.0	2.0		1.6		2.0	5.6	1.1	0.2829	0.2627	
3RFR crystal structure															
1	C2W2	Red	0.0	0.1	0.1	0.6	2.7			2.7	3.5	0.7	0.2763	0.2483	
		MV	0.0	0.0	0.3	0.5	1.6			1.6	2.4	0.5	0.2763	0.2483	
		Ox	0.0	0.4	0.4	0.6	1.9			1.9	3.3	0.7	0.2763	0.2483	
	C1W0	Red	0.4	0.0	0.1	1.4				1.4	1.9	0.5	0.2765	0.2483	
		Ox	0.1	0.0	0.1	1.3				1.3	1.5	0.4	0.2765	0.2483	
	C1W1	Red	0.6	0.0	0.1	1.3		2.6		2.6	4.6	0.9	0.2767	0.2484	
		Ox	0.0	0.0	0.1	0.9		2.3		2.3	3.3	0.7	0.2764	0.2483	
2	C1W0	Red	0.2	0.0	0.4	0.1				0.4	0.7	0.2	0.2765	0.2485	
		Ox	0.4	0.1	0.4	0.2				0.4	1.1	0.3	0.2765	0.2485	
	C1W1	Ox	0.3	0.1	0.4	0.5		1.3		1.3	2.6	0.5	0.2764	0.2484	
3	C1W0	Red	0.0	0.3	0.2	0.9				0.9	1.4	0.4	0.2762	0.2484	
		Ox	0.0	0.3	0.2	0.9				0.9	1.4	0.4	0.2762	0.2484	
	C1W1	Ox	0.0	0.2	0.2	1.0		0.5		1.0	1.9	0.4	0.2760	0.2483	

Table S3. Energy components of the various reactant states in the reaction with methanol. Five energies are given for each reactant in both the triplet ($S = 1$) and singlet state ($S = 0$): TPSS-D3/def2-SV(P) optimised geometries, TPSS-D3/def2-TZVPD and B3LYP-D3/def2-TZVPD single point energies, obtained in a COSMO continuum solvent with a dielectric constant of 4, zero-point energies (ZPE) from TPSS-D3/def2-SV(P) frequencies and the total energies, which are the sum of the ZPE and the B3LYP-D3/def2-TZVPD energies. The reference energy is always the triplet RS state.

Energy Method	QM		QM+COSMO		QM+COSMO		ZPE		Total	
	TPSS-D3		TPSS-D3		B3LYP-D3		TPSS-D3		B3LYP-D3	
Basis	def2-SV(P)		def2-TZVPD		def2-TZVPD		def2-SV(P)		def2-TZVPD	
S	1	0	1	0	1	0	1	0	1	0
RS	0.0	19.4	0.0	33.0	0.0	85.2	0.0	-0.3	0.0	84.8
R	-21.0	-2.5	-11.7	24.6	-1.2	74.2	2.9	2.9	1.7	77.1
TS1	24.0	28.1	20.7	61.1	35.1	124.4	-13.2	-12.6	21.9	111.8
Im1	6.9	6.8	-10.0	74.2	-8.8	141.2	-6.8	-7.0	-15.6	134.2
Im2	-20.1	-87.0	-32.2	-100.0	-8.1	-60.8	1.9	9.0	-6.2	-51.8
TS2	125.6	-30.9	113.2	-34.7	132.4	-5.5	-8.3	6.2	124.2	0.7
P	-32.9	-237.1	-16.5	-226.9	9.0	-233.6	13.7	16.9	22.6	-216.7

Figure S1. QM-refined structures of site A in the first protomer of the 3RGB structure, modelled by two Cu ions in the (a) fully reduced and (b) fully oxidised states.

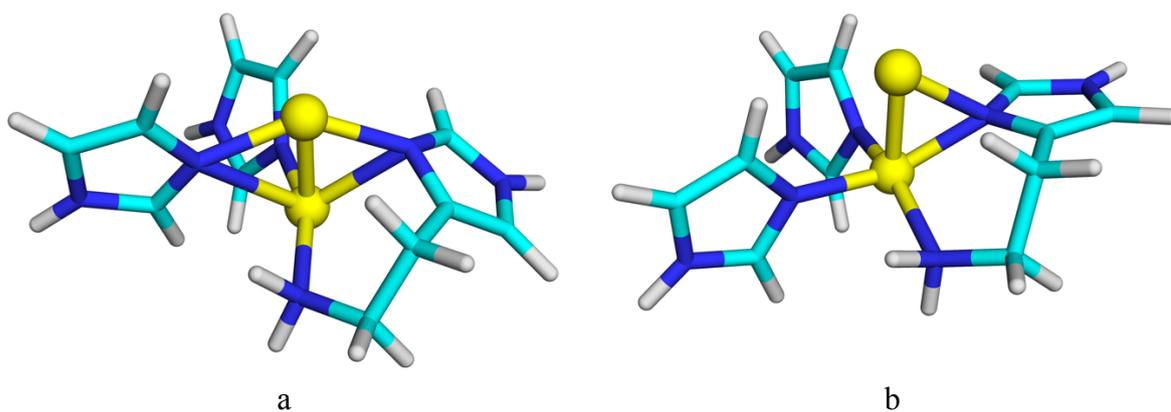


Figure S2. (a) The Yoshizawa-type structure of pMMO (first protomer of the 3RGB structure, Y2W0a) obtained by QM/MM (i.e. QM-refinement with $w_A = 0$) for the reduced state. (b) A quantum-refined structure with two extra water molecules for the MV state (Y2W2). (c) The quantum-refined structure of site A with the backbone carbonyl group coordinating to Cu_B (reduced state).

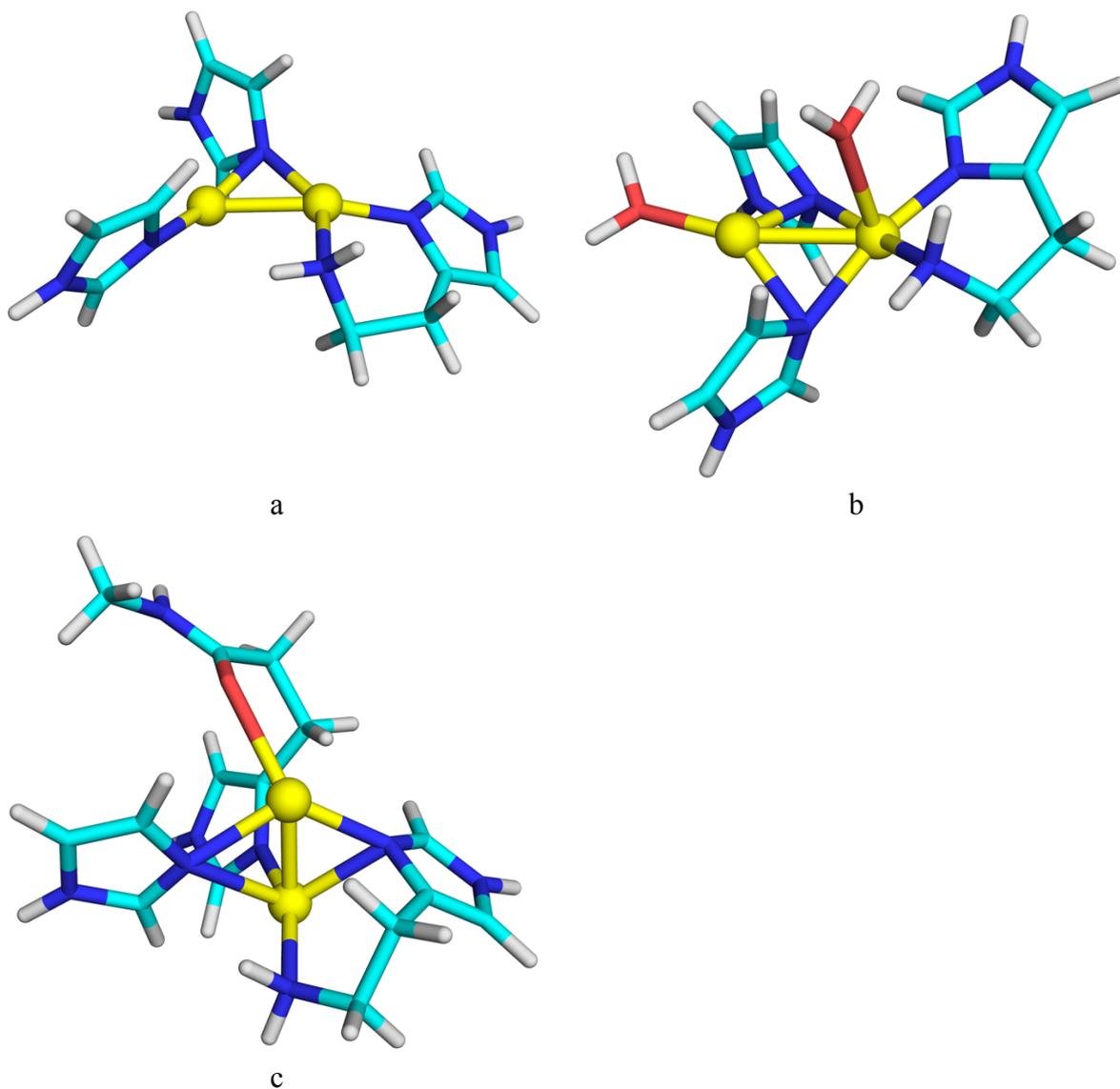


Figure S3. QM-refined structures of site A, modelled with a single Cu ion (C1W0) in the (a) reduced and (b) oxidised states.

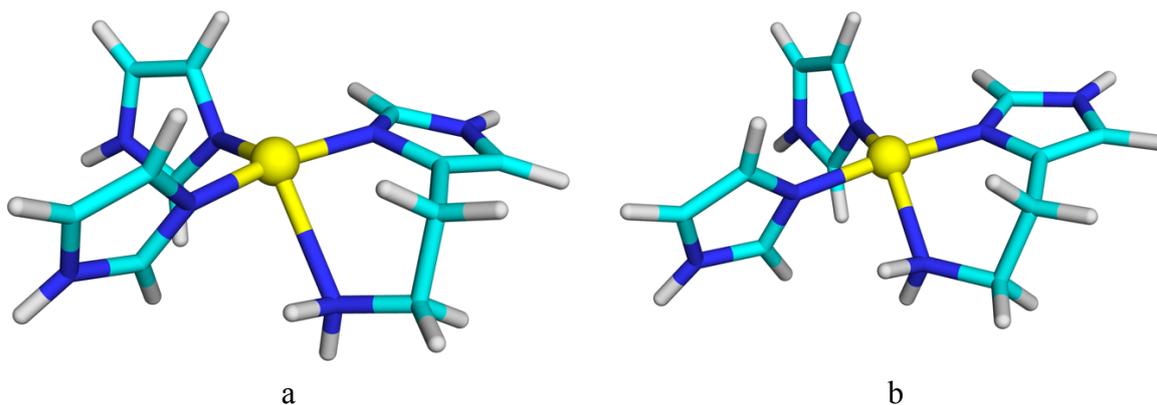


Figure S4. QM-refined structures of site A, modelled with a single Cu ion and the water molecule in the (a) oxidised state (C1W1). Part (b) shows the electron-density maps of the oxidised state. The $2mF_o - DF_c$ maps are contoured at 1.0σ and the $mF_o - DF_c$ maps are contoured at $+3.0 \sigma$ (green) and -3.0σ (red).

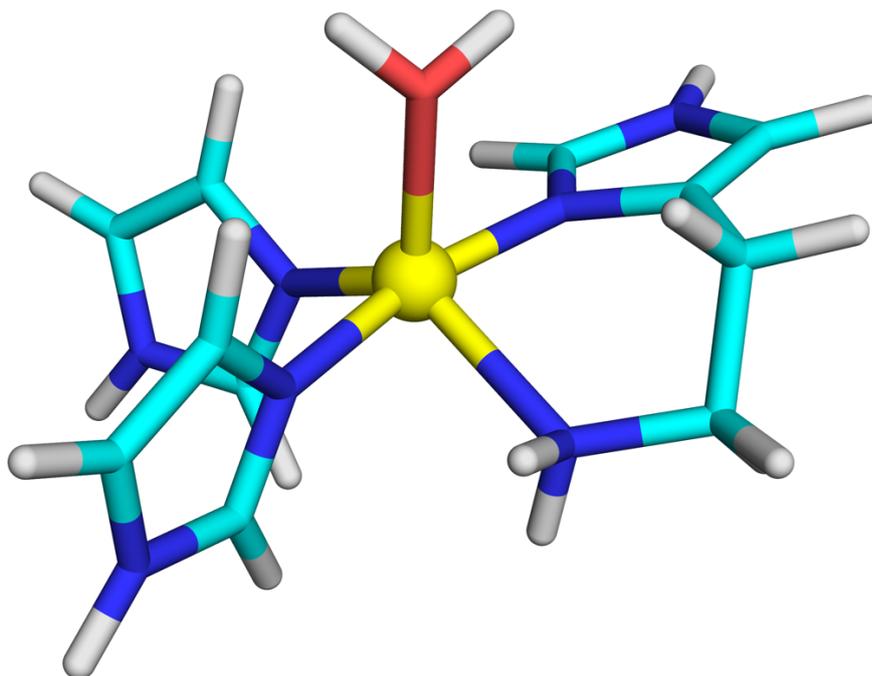


Figure S5. QM-refined structures of site A in the second protomer of the 3RGB structure, modelled (a) with a two Cu ions (C2W0) in the reduced state, (b) the Yoshizawa-type structure (Y2W2) or (c) with a single Cu ion (still in the reduced state).

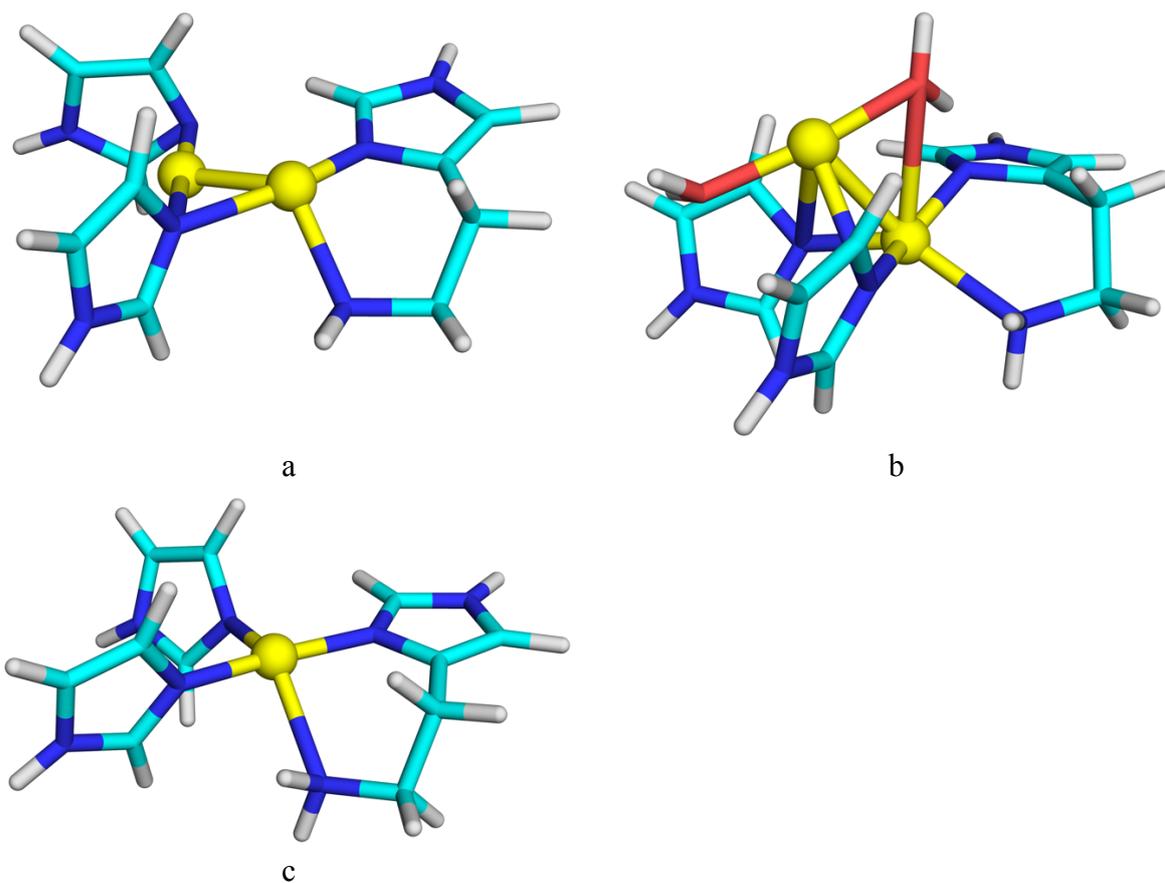


Table S4. Coordinates of the QM system for the mononuclear site in all optimised structures.

REMARK PDB files of the mononuclear copper site in pMMO obtained by quantum refinement
REMARK L. Cao, O. Caldararu, A. C. Rosenzweig, U. Ryde
REMARK Quantum refinement does not support dinuclear copper sites in crystal structures
REMARK of particulate methane monooxygenase
REMARK Angew. Chemie 2017
REMARK Supplementary material
REMARK 18 structures are given:
REMARK 2 Crystal structures: 3RGB or 3RFR
REMARK 3 subunits
REMARK Reduced, Cu(I), oxidised, Cu(II), or oxidised state with an extra water molecule
REMARK Only heavy atoms in the QM system are included; other atoms are not moved
REMARK

REMARK 3RGB, Subunit 1, Reduced state

REMARK DATE:04-Nov-2016 15:59:39 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	33	39.558	130.222	81.449	1.00	73.31	A	C	0
ATOM	2	CG	HIS	33	39.536	130.282	82.943	1.00	72.87	A	C	0
ATOM	3	CD2	HIS	33	40.433	130.882	83.811	1.00	72.68	A	C	0
ATOM	4	ND1	HIS	33	38.383	129.977	83.651	1.00	72.38	A	N	0
ATOM	5	CE1	HIS	33	38.558	130.382	84.908	1.00	72.38	A	C	0
ATOM	6	NE2	HIS	33	39.798	130.929	85.043	1.00	72.37	A	N	0
ATOM	7	C	HIS	33	39.582	131.594	79.449	1.00	73.42	A	C	0
ATOM	8	N	HIS	33	37.660	131.761	81.212	1.00	73.34	A	N	0
ATOM	9	CA	HIS	33	39.101	131.627	80.915	1.00	73.37	A	C	0
ATOM	10	CB	HIS	137	35.988	128.077	85.420	1.00	43.65	A	C	0
ATOM	11	CG	HIS	137	35.055	129.223	85.194	1.00	43.97	A	C	0
ATOM	12	CD2	HIS	137	33.793	129.437	85.714	1.00	43.67	A	C	0
ATOM	13	ND1	HIS	137	35.445	130.309	84.427	1.00	44.90	A	N	0
ATOM	14	CE1	HIS	137	34.440	131.182	84.497	1.00	44.62	A	C	0
ATOM	15	NE2	HIS	137	33.425	130.694	85.267	1.00	43.95	A	N	0
ATOM	16	CB	HIS	139	33.507	127.696	78.902	1.00	49.60	A	C	0
ATOM	17	CG	HIS	139	34.226	128.599	79.774	1.00	50.45	A	C	0
ATOM	18	CD2	HIS	139	35.368	128.398	80.520	1.00	51.09	A	C	0
ATOM	19	ND1	HIS	139	33.792	129.883	80.036	1.00	50.61	A	N	0
ATOM	20	CE1	HIS	139	34.668	130.428	80.938	1.00	51.73	A	C	0
ATOM	21	NE2	HIS	139	35.625	129.541	81.267	1.00	51.75	A	N	0
ATOM	22	CU+1	CU1	416	36.806	129.802	82.728	1.00	64.05	A	CU	1

END

REMARK 3RGB, Subunit 1, Oxidised state

REMARK DATE:04-Nov-2016 15:59:39 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	33	39.911	130.476	81.572	1.00	74.85	A	C	0
ATOM	2	CG	HIS	33	39.672	130.484	83.055	1.00	74.33	A	C	0
ATOM	3	CD2	HIS	33	40.522	130.877	84.072	1.00	74.26	A	C	0
ATOM	4	ND1	HIS	33	38.438	130.187	83.638	1.00	73.84	A	N	0
ATOM	5	CE1	HIS	33	38.532	130.410	84.955	1.00	73.92	A	C	0
ATOM	6	NE2	HIS	33	39.789	130.817	85.246	1.00	74.06	A	N	0
ATOM	7	C	HIS	33	39.578	131.593	79.450	1.00	75.04	A	C	0
ATOM	8	N	HIS	33	37.650	131.186	81.051	1.00	75.22	A	N	0
ATOM	9	CA	HIS	33	39.071	131.617	80.906	1.00	75.00	A	C	0
ATOM	10	CB	HIS	137	36.012	128.058	85.421	1.00	43.37	A	C	0
ATOM	11	CG	HIS	137	35.116	129.201	85.158	1.00	43.47	A	C	0
ATOM	12	CD2	HIS	137	33.902	129.524	85.725	1.00	42.88	A	C	0
ATOM	13	ND1	HIS	137	35.514	130.203	84.285	1.00	44.35	A	N	0
ATOM	14	CE1	HIS	137	34.554	131.143	84.346	1.00	44.08	A	C	0
ATOM	15	NE2	HIS	137	33.575	130.760	85.205	1.00	43.00	A	N	0
ATOM	16	CB	HIS	139	33.513	127.691	78.906	1.00	47.68	A	C	0
ATOM	17	CG	HIS	139	34.225	128.591	79.782	1.00	48.49	A	C	0
ATOM	18	CD2	HIS	139	35.316	128.377	80.597	1.00	49.04	A	C	0
ATOM	19	ND1	HIS	139	33.795	129.881	80.000	1.00	48.66	A	N	0
ATOM	20	CE1	HIS	139	34.602	130.431	80.945	1.00	49.68	A	C	0
ATOM	21	NE2	HIS	139	35.525	129.530	81.346	1.00	49.68	A	N	0
ATOM	22	CU+1	CU1	416	36.881	129.891	82.686	1.00	64.20	A	CU	1

END

REMARK 3RGB, Subunit 1, Oxidised state with an extra water molecule

REMARK DATE:08-Nov-2016 15:55:59 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	A	33	39.883	130.479	81.582	1.00	71.86	A	C	0
ATOM	2	CG	HIS	A	33	39.690	130.528	83.075	1.00	71.34	A	C	0
ATOM	3	CD2	HIS	A	33	40.598	130.819	84.076	1.00	71.43	A	C	0
ATOM	4	ND1	HIS	A	33	38.442	130.380	83.686	1.00	70.70	A	N	0
ATOM	5	CE1	HIS	A	33	38.584	130.569	84.998	1.00	70.88	A	C	0
ATOM	6	NE2	HIS	A	33	39.886	130.832	85.265	1.00	71.25	A	N	0
ATOM	7	C	HIS	A	33	39.579	131.595	79.450	1.00	72.08	A	C	0
ATOM	8	N	HIS	A	33	37.642	131.281	81.056	1.00	72.17	A	N	0
ATOM	9	CA	HIS	A	33	39.079	131.634	80.904	1.00	72.04	A	C	0

ATOM	10	CB	HIS	A	137	36.006	128.065	85.426	1.00	42.08	A	C	0
ATOM	11	CG	HIS	A	137	35.102	129.208	85.170	1.00	42.46	A	C	0
ATOM	12	CD2	HIS	A	137	33.873	129.505	85.718	1.00	41.85	A	C	0
ATOM	13	ND1	HIS	A	137	35.503	130.233	84.327	1.00	43.52	A	N	0
ATOM	14	CE1	HIS	A	137	34.532	131.157	84.383	1.00	43.48	A	C	0
ATOM	15	NE2	HIS	A	137	33.541	130.746	85.214	1.00	42.16	A	N	0
ATOM	16	CB	HIS	A	139	33.514	127.692	78.904	1.00	46.16	A	C	0
ATOM	17	CG	HIS	A	139	34.206	128.617	79.773	1.00	47.19	A	C	0
ATOM	18	CD2	HIS	A	139	35.312	128.435	80.569	1.00	47.74	A	C	0
ATOM	19	ND1	HIS	A	139	33.747	129.896	80.001	1.00	47.61	A	N	0
ATOM	20	CE1	HIS	A	139	34.555	130.464	80.937	1.00	48.85	A	C	0
ATOM	21	NE2	HIS	A	139	35.502	129.588	81.320	1.00	48.45	A	N	0
ATOM	22	O	HOH	A	416	37.843	127.924	82.467	1.00	52.69	A	O	0
ATOM	23	CU+1	CUI	A	417	36.893	129.893	82.710	1.00	66.97	A	CU	1

END

REMARK 3RGB, Subunit 2, Reduced state

REMARK DATE:11-Apr-2017 12:41:20 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	33	35.739	157.866	32.437	1.00	62.03	E	C	0
ATOM	2	CG	HIS	E	33	35.413	159.178	31.782	1.00	62.12	E	C	0
ATOM	3	CD2	HIS	E	33	36.188	160.006	30.997	1.00	62.91	E	C	0
ATOM	4	ND1	HIS	E	33	34.150	159.748	31.878	1.00	61.28	E	N	0
ATOM	5	CE1	HIS	E	33	34.133	160.860	31.154	1.00	61.46	E	C	0
ATOM	6	NE2	HIS	E	33	35.362	161.058	30.613	1.00	63.11	E	N	0
ATOM	7	C	HIS	E	33	35.696	155.460	32.461	1.00	61.00	E	C	0
ATOM	8	N	HIS	E	33	33.609	156.748	32.014	1.00	62.27	E	N	0
ATOM	9	CA	HIS	E	33	35.054	156.648	31.755	1.00	61.93	E	C	0
ATOM	10	CB	HIS	E	137	31.830	162.236	33.284	1.00	41.72	E	C	0
ATOM	11	CG	HIS	E	137	30.823	161.393	32.545	1.00	41.82	E	C	0
ATOM	12	CD2	HIS	E	137	29.476	161.607	32.358	1.00	41.88	E	C	0
ATOM	13	ND1	HIS	E	137	31.184	160.218	31.909	1.00	42.05	E	N	0
ATOM	14	CE1	HIS	E	137	30.077	159.703	31.394	1.00	41.53	E	C	0
ATOM	15	NE2	HIS	E	137	29.020	160.518	31.646	1.00	41.95	E	N	0
ATOM	16	CB	HIS	E	139	30.178	156.685	37.123	1.00	45.17	E	C	0
ATOM	17	CG	HIS	E	139	30.622	157.008	35.779	1.00	44.56	E	C	0
ATOM	18	CD2	HIS	E	139	31.601	157.862	35.329	1.00	45.27	E	C	0
ATOM	19	ND1	HIS	E	139	30.063	156.436	34.661	1.00	44.86	E	N	0
ATOM	20	CE1	HIS	E	139	30.674	156.975	33.565	1.00	45.65	E	C	0
ATOM	21	NE2	HIS	E	139	31.617	157.843	33.947	1.00	45.55	E	N	0
ATOM	22	CU+1	CUI	E	416	32.678	158.940	32.726	1.00	59.68	E	CU	1

END

REMARK 3RGB, Subunit 2, Oxidised state

REMARK DATE:11-Apr-2017 12:41:20 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	33	35.959	157.841	32.100	1.00	59.51	E	C	0
ATOM	2	CG	HIS	E	33	35.457	159.187	31.618	1.00	59.86	E	C	0
ATOM	3	CD2	HIS	E	33	36.194	160.201	31.044	1.00	60.67	E	C	0
ATOM	4	ND1	HIS	E	33	34.138	159.651	31.698	1.00	59.11	E	N	0
ATOM	5	CE1	HIS	E	33	34.073	160.871	31.162	1.00	59.44	E	C	0
ATOM	6	NE2	HIS	E	33	35.308	161.233	30.767	1.00	60.99	E	N	0
ATOM	7	C	HIS	E	33	35.677	155.458	32.469	1.00	58.40	E	C	0
ATOM	8	N	HIS	E	33	33.763	156.952	32.592	1.00	59.37	E	N	0
ATOM	9	CA	HIS	E	33	34.999	156.659	31.802	1.00	59.12	E	C	0
ATOM	10	CB	HIS	E	137	31.850	162.244	33.294	1.00	40.56	E	C	0
ATOM	11	CG	HIS	E	137	30.875	161.388	32.559	1.00	40.50	E	C	0
ATOM	12	CD2	HIS	E	137	29.579	161.657	32.186	1.00	40.24	E	C	0
ATOM	13	ND1	HIS	E	137	31.205	160.120	32.112	1.00	40.70	E	N	0
ATOM	14	CE1	HIS	E	137	30.120	159.610	31.522	1.00	39.97	E	C	0
ATOM	15	NE2	HIS	E	137	29.126	160.521	31.555	1.00	40.28	E	N	0
ATOM	16	CB	HIS	E	139	30.183	156.693	37.124	1.00	42.83	E	C	0
ATOM	17	CG	HIS	E	139	30.610	157.015	35.777	1.00	42.30	E	C	0
ATOM	18	CD2	HIS	E	139	31.508	157.937	35.293	1.00	42.78	E	C	0
ATOM	19	ND1	HIS	E	139	30.095	156.358	34.689	1.00	42.57	E	N	0
ATOM	20	CE1	HIS	E	139	30.635	156.896	33.567	1.00	43.27	E	C	0
ATOM	21	NE2	HIS	E	139	31.505	157.859	33.911	1.00	43.06	E	N	0
ATOM	22	CU+1	CUI	E	416	32.717	158.882	32.685	1.00	60.13	E	CU	1

END

REMARK 3RGB, Subunit 2, Oxidised state with an extra water molecule

REMARK DATE:28-Apr-2017 22:41:54 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	33	35.886	157.873	32.189	1.00	59.80	E	C	0
ATOM	2	CG	HIS	E	33	35.398	159.200	31.646	1.00	60.12	E	C	0
ATOM	3	CD2	HIS	E	33	36.154	160.227	31.120	1.00	61.04	E	C	0
ATOM	4	ND1	HIS	E	33	34.064	159.621	31.606	1.00	59.31	E	N	0
ATOM	5	CE1	HIS	E	33	34.010	160.833	31.058	1.00	59.75	E	C	0
ATOM	6	NE2	HIS	E	33	35.265	161.230	30.758	1.00	61.48	E	N	0
ATOM	7	C	HIS	E	33	35.679	155.459	32.468	1.00	58.72	E	C	0
ATOM	8	N	HIS	E	33	33.688	156.900	32.464	1.00	59.81	E	N	0
ATOM	9	CA	HIS	E	33	35.003	156.662	31.792	1.00	59.53	E	C	0

ATOM	10	CB	HIS	E	137	31.848	162.241	33.292	1.00	39.65	E	C	0
ATOM	11	CG	HIS	E	137	30.869	161.386	32.558	1.00	39.95	E	C	0
ATOM	12	CD2	HIS	E	137	29.577	161.680	32.189	1.00	39.80	E	C	0
ATOM	13	ND1	HIS	E	137	31.171	160.110	32.113	1.00	40.08	E	N	0
ATOM	14	CE1	HIS	E	137	30.074	159.621	31.530	1.00	39.60	E	C	0
ATOM	15	NE2	HIS	E	137	29.098	156.551	31.562	1.00	40.03	E	N	0
ATOM	16	CB	HIS	E	139	30.186	156.696	37.125	1.00	41.73	E	C	0
ATOM	17	CG	HIS	E	139	30.582	157.002	35.765	1.00	40.91	E	C	0
ATOM	18	CD2	HIS	E	139	31.427	157.943	35.234	1.00	41.13	E	C	0
ATOM	19	ND1	HIS	E	139	30.090	156.276	34.711	1.00	41.47	E	N	0
ATOM	20	CE1	HIS	E	139	30.600	156.792	33.563	1.00	42.20	E	C	0
ATOM	21	NE2	HIS	E	139	31.427	157.802	33.856	1.00	41.45	E	N	0
ATOM	22	O	HOH	E	416	33.826	159.548	34.573	1.00	48.02	E	O	0
ATOM	23	CU+1	CUI	E	417	32.656	158.842	32.632	1.00	62.64	E	CU	1

END

REMARK 3RGB, Subunit 3, Reduced state

REMARK DATE:26-Jun-2017 15:10:27 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	I	33	34.208	101.593	33.639	1.00	65.19	I	C	0
ATOM	2	CG	HIS	I	33	33.709	100.358	32.950	1.00	63.32	I	C	0
ATOM	3	CD2	HIS	I	33	34.393	99.184	32.704	1.00	62.66	I	C	0
ATOM	4	ND1	HIS	I	33	32.362	100.165	32.640	1.00	62.85	I	N	0
ATOM	5	CE1	HIS	I	33	32.236	98.899	32.241	1.00	61.69	I	C	0
ATOM	6	NE2	HIS	I	33	33.443	98.278	32.260	1.00	61.82	I	N	0
ATOM	7	C	HIS	I	33	34.334	102.858	35.741	1.00	67.59	I	C	0
ATOM	8	N	HIS	I	33	32.195	102.113	34.878	1.00	66.92	I	N	0
ATOM	9	CA	HIS	I	33	33.615	101.692	35.082	1.00	66.67	I	C	0
ATOM	10	CB	HIS	I	137	29.516	99.980	30.220	1.00	57.16	I	C	0
ATOM	11	CG	HIS	I	137	28.741	99.823	31.519	1.00	56.37	I	C	0
ATOM	12	CD2	HIS	I	137	27.384	99.728	31.741	1.00	56.56	I	C	0
ATOM	13	ND1	HIS	I	137	29.413	99.669	32.729	1.00	55.16	I	N	0
ATOM	14	CE1	HIS	I	137	28.481	99.502	33.657	1.00	55.62	I	C	0
ATOM	15	NE2	HIS	I	137	27.235	99.536	33.102	1.00	55.52	I	N	0
ATOM	16	CB	HIS	I	139	28.174	106.072	33.340	1.00	48.56	I	C	0
ATOM	17	CG	HIS	I	139	28.627	104.733	33.712	1.00	48.95	I	C	0
ATOM	18	CD2	HIS	I	139	29.466	103.864	33.053	1.00	49.86	I	C	0
ATOM	19	ND1	HIS	I	139	28.257	104.090	34.875	1.00	49.19	I	N	0
ATOM	20	CE1	HIS	I	139	28.859	102.855	34.888	1.00	49.72	I	C	0
ATOM	21	NE2	HIS	I	139	29.588	102.693	33.784	1.00	49.20	I	N	0
ATOM	22	CU+1	CUI	I	417	30.789	101.270	33.302	1.00	71.43	I	CU	1

END

REMARK 3RGB, Subunit 3, Oxidised state

REMARK DATE:26-Jun-2017 15:10:27 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	I	33	34.380	101.309	33.784	1.00	63.58	I	C	0
ATOM	2	CG	HIS	I	33	33.727	100.174	33.039	1.00	61.90	I	C	0
ATOM	3	CD2	HIS	I	33	34.320	99.009	32.604	1.00	61.51	I	C	0
ATOM	4	ND1	HIS	I	33	32.350	100.069	32.782	1.00	61.73	I	N	0
ATOM	5	CE1	HIS	I	33	32.130	98.862	32.244	1.00	60.88	I	C	0
ATOM	6	NE2	HIS	I	33	33.303	98.208	32.119	1.00	61.20	I	N	0
ATOM	7	C	HIS	I	33	34.319	102.861	35.740	1.00	65.57	I	C	0
ATOM	8	N	HIS	I	33	32.308	102.351	34.484	1.00	64.60	I	N	0
ATOM	9	CA	HIS	I	33	33.570	101.702	35.050	1.00	64.69	I	C	0
ATOM	10	CB	HIS	I	137	29.528	99.989	30.215	1.00	54.82	I	C	0
ATOM	11	CG	HIS	I	137	28.775	99.851	31.514	1.00	53.92	I	C	0
ATOM	12	CD2	HIS	I	137	27.490	99.400	31.715	1.00	54.16	I	C	0
ATOM	13	ND1	HIS	I	137	29.356	100.079	32.760	1.00	52.36	I	N	0
ATOM	14	CE1	HIS	I	137	28.437	99.793	33.682	1.00	52.88	I	C	0
ATOM	15	NE2	HIS	I	137	27.299	99.382	33.081	1.00	53.20	I	N	0
ATOM	16	CB	HIS	I	139	28.178	106.071	33.333	1.00	47.84	I	C	0
ATOM	17	CG	HIS	I	139	28.630	104.730	33.706	1.00	48.14	I	C	0
ATOM	18	CD2	HIS	I	139	29.488	103.855	33.079	1.00	48.88	I	C	0
ATOM	19	ND1	HIS	I	139	28.249	104.112	34.877	1.00	48.41	I	N	0
ATOM	20	CE1	HIS	I	139	28.858	102.896	34.950	1.00	48.99	I	C	0
ATOM	21	NE2	HIS	I	139	29.614	102.716	33.861	1.00	48.33	I	N	0
ATOM	22	CU+1	CUI	I	417	30.869	101.290	33.339	1.00	73.67	I	CU	1

END

REMARK 3RFR, Subunit 1, Reduced state

REMARK DATE:11-Apr-2017 17:59:04 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	A	29	49.904	34.452	46.113	1.00	78.48	A	C	0
ATOM	2	CG	HIS	A	29	50.430	35.676	46.800	1.00	78.54	A	C	0
ATOM	3	CD2	HIS	A	29	49.880	36.941	46.776	1.00	78.53	A	C	0
ATOM	4	ND1	HIS	A	29	51.518	35.666	47.672	1.00	78.55	A	N	0
ATOM	5	CE1	HIS	A	29	51.621	36.906	48.148	1.00	78.05	A	C	0
ATOM	6	NE2	HIS	A	29	50.649	37.701	47.635	1.00	79.24	A	N	0
ATOM	7	C	HIS	A	29	49.125	32.190	46.334	1.00	78.25	A	C	0
ATOM	8	N	HIS	A	29	51.510	32.812	46.840	1.00	79.15	A	N	0
ATOM	9	CA	HIS	A	29	50.085	33.184	46.967	1.00	78.41	A	C	0

ATOM	10	CB	HIS	A	133	55.147	36.089	48.320	1.00	49.64	A	C	0
ATOM	11	CG	HIS	A	133	54.624	36.076	49.695	1.00	53.16	A	C	0
ATOM	12	CD2	HIS	A	133	55.244	36.654	50.785	1.00	52.66	A	C	0
ATOM	13	ND1	HIS	A	133	53.443	35.493	50.085	1.00	55.37	A	N	0
ATOM	14	CE1	HIS	A	133	53.328	35.721	51.388	1.00	55.22	A	C	0
ATOM	15	NE2	HIS	A	133	54.400	36.430	51.852	1.00	54.43	A	N	0
ATOM	16	CB	HIS	A	135	55.203	29.470	50.334	1.00	53.71	A	C	0
ATOM	17	CG	HIS	A	135	54.332	30.604	50.095	1.00	53.64	A	C	0
ATOM	18	CD2	HIS	A	135	54.265	31.598	49.142	1.00	53.89	A	C	0
ATOM	19	ND1	HIS	A	135	53.299	30.837	50.978	1.00	53.37	A	N	0
ATOM	20	CE1	HIS	A	135	52.639	31.952	50.561	1.00	55.61	A	C	0
ATOM	21	NE2	HIS	A	135	53.217	32.453	49.466	1.00	54.70	A	N	0
ATOM	22	CU+2	CU2	A	419	52.513	34.050	48.621	1.00	79.73	A	CU	2

END

REMARK 3RFR, Subunit 1, Oxidised state

REMARK DATE:11-Apr-2017 17:59:04 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	A	29	50.208	34.382	45.978	1.00	79.02	A	C	0
ATOM	2	CG	HIS	A	29	50.621	35.625	46.705	1.00	78.80	A	C	0
ATOM	3	CD2	HIS	A	29	50.048	36.876	46.632	1.00	78.59	A	C	0
ATOM	4	ND1	HIS	A	29	51.581	35.637	47.721	1.00	78.91	A	N	0
ATOM	5	CE1	HIS	A	29	51.585	36.876	48.235	1.00	78.27	A	C	0
ATOM	6	NE2	HIS	A	29	50.669	37.636	47.602	1.00	79.12	A	N	0
ATOM	7	C	HIS	A	29	49.122	32.198	46.333	1.00	79.01	A	C	0
ATOM	8	N	HIS	A	29	51.420	32.758	47.297	1.00	80.48	A	N	0
ATOM	9	CA	HIS	A	29	50.050	33.231	46.983	1.00	79.19	A	C	0
ATOM	10	CB	HIS	A	133	55.148	36.089	48.284	1.00	49.30	A	C	0
ATOM	11	CG	HIS	A	133	54.589	36.034	49.630	1.00	52.69	A	C	0
ATOM	12	CD2	HIS	A	133	55.082	36.685	50.738	1.00	51.36	A	C	0
ATOM	13	ND1	HIS	A	133	53.475	35.302	49.989	1.00	55.17	A	N	0
ATOM	14	CE1	HIS	A	133	53.275	35.525	51.296	1.00	54.77	A	C	0
ATOM	15	NE2	HIS	A	133	54.231	36.362	51.770	1.00	53.16	A	N	0
ATOM	16	CB	HIS	A	135	55.205	29.470	50.334	1.00	53.53	A	C	0
ATOM	17	CG	HIS	A	135	54.334	30.603	50.099	1.00	53.54	A	C	0
ATOM	18	CD2	HIS	A	135	54.293	31.613	49.163	1.00	53.64	A	C	0
ATOM	19	ND1	HIS	A	135	53.285	30.818	50.965	1.00	53.36	A	N	0
ATOM	20	CE1	HIS	A	135	52.627	31.931	50.572	1.00	55.24	A	C	0
ATOM	21	NE2	HIS	A	135	53.234	32.452	49.495	1.00	54.11	A	N	0
ATOM	22	CU+2	CU2	A	419	52.498	34.050	48.630	1.00	81.48	A	CU	2

END

REMARK 3RFR, Subunit 1, Oxidised state with an extra water molecule

REMARK DATE:11-Apr-2017 17:45:49 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	A	29	50.553	34.233	45.910	1.00	75.20	A	C	0
ATOM	2	CG	HIS	A	29	50.819	35.534	46.606	1.00	75.45	A	C	0
ATOM	3	CD2	HIS	A	29	50.157	36.737	46.466	1.00	75.65	A	C	0
ATOM	4	ND1	HIS	A	29	51.642	35.620	47.728	1.00	75.53	A	N	0
ATOM	5	CE1	HIS	A	29	51.484	36.847	48.242	1.00	75.15	A	C	0
ATOM	6	NE2	HIS	A	29	50.588	37.537	47.505	1.00	75.75	A	N	0
ATOM	7	C	HIS	A	29	49.116	32.206	46.325	1.00	75.32	A	C	0
ATOM	8	N	HIS	A	29	51.127	32.728	47.746	1.00	76.22	A	N	0
ATOM	9	CA	HIS	A	29	50.006	33.265	46.969	1.00	75.23	A	C	0
ATOM	10	CB	HIS	A	133	55.148	36.084	48.289	1.00	50.55	A	C	0
ATOM	11	CG	HIS	A	133	54.598	36.053	49.642	1.00	53.60	A	C	0
ATOM	12	CD2	HIS	A	133	55.109	36.707	50.741	1.00	52.24	A	C	0
ATOM	13	ND1	HIS	A	133	53.488	35.330	50.018	1.00	56.05	A	N	0
ATOM	14	CE1	HIS	A	133	53.304	35.559	51.325	1.00	55.02	A	C	0
ATOM	15	NE2	HIS	A	133	54.270	36.392	51.785	1.00	53.81	A	N	0
ATOM	16	CB	HIS	A	135	55.210	29.462	50.331	1.00	52.38	A	C	0
ATOM	17	CG	HIS	A	135	54.333	30.586	50.113	1.00	52.71	A	C	0
ATOM	18	CD2	HIS	A	135	54.285	31.610	49.191	1.00	52.23	A	C	0
ATOM	19	ND1	HIS	A	135	53.292	30.784	50.991	1.00	52.42	A	N	0
ATOM	20	CE1	HIS	A	135	52.632	31.904	50.616	1.00	54.19	A	C	0
ATOM	21	NE2	HIS	A	135	53.233	32.440	49.542	1.00	52.77	A	N	0
ATOM	22	CU+2	CU2	A	419	52.508	34.052	48.698	1.00	77.21	A	CU	2
ATOM	23	O	HOH	A	420	53.618	33.361	46.562	1.00	36.62	A	O	0

END

REMARK 3RFR, Subunit 2, Reduced state

REMARK DATE:28-Nov-2016 12:00:25 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	29	16.569	-12.883	54.958	1.00	78.25	E	C	0
ATOM	2	CG	HIS	E	29	15.439	-12.937	55.949	1.00	78.11	E	C	0
ATOM	3	CD2	HIS	E	29	14.085	-13.056	55.712	1.00	78.62	E	C	0
ATOM	4	ND1	HIS	E	29	15.669	-13.011	57.321	1.00	76.42	E	N	0
ATOM	5	CE1	HIS	E	29	14.484	-13.218	57.890	1.00	77.56	E	C	0
ATOM	6	NE2	HIS	E	29	13.500	-13.248	56.952	1.00	79.31	E	N	0
ATOM	7	C	HIS	E	29	18.769	-12.489	54.377	1.00	79.15	E	C	0
ATOM	8	N	HIS	E	29	17.642	-11.035	56.072	1.00	79.37	E	N	0
ATOM	9	CA	HIS	E	29	17.897	-12.410	55.592	1.00	78.91	E	C	0

ATOM	10	CB	HIS	E	133	14.307	-10.420	60.056	1.00	64.38	E	C	0
ATOM	11	CG	HIS	E	133	15.150	-11.339	60.881	1.00	66.46	E	C	0
ATOM	12	CD2	HIS	E	133	15.146	-11.545	62.247	1.00	67.43	E	C	0
ATOM	13	ND1	HIS	E	133	16.102	-12.150	60.303	1.00	66.98	E	N	0
ATOM	14	CE1	HIS	E	133	16.676	-12.835	61.284	1.00	65.96	E	C	0
ATOM	15	NE2	HIS	E	133	16.124	-12.498	62.481	1.00	66.39	E	N	0
ATOM	16	CB	HIS	E	135	20.623	-7.700	60.115	1.00	47.31	E	C	0
ATOM	17	CG	HIS	E	135	19.864	-8.938	59.747	1.00	48.28	E	C	0
ATOM	18	CD2	HIS	E	135	18.658	-9.176	59.124	1.00	48.99	E	C	0
ATOM	19	ND1	HIS	E	135	20.391	-10.189	59.998	1.00	48.89	E	N	0
ATOM	20	CE1	HIS	E	135	19.512	-11.126	59.553	1.00	49.05	E	C	0
ATOM	21	NE2	HIS	E	135	18.449	-10.538	59.021	1.00	48.87	E	N	0
ATOM	22	CU+2	CU2	E	420	16.767	-11.565	58.421	1.00	82.94	E	CU	2

END

REMARK 3RFR, Subunit 2, Oxidised state

REMARK DATE:28-Nov-2016 12:00:25 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	29	16.481	-12.623	54.981	1.00	80.85	E	C	0
ATOM	2	CG	HIS	E	29	15.406	-12.855	56.012	1.00	80.62	E	C	0
ATOM	3	CD2	HIS	E	29	14.133	-13.330	55.797	1.00	81.09	E	C	0
ATOM	4	ND1	HIS	E	29	15.595	-12.757	57.396	1.00	79.64	E	N	0
ATOM	5	CE1	HIS	E	29	14.478	-13.195	57.985	1.00	80.54	E	C	0
ATOM	6	NE2	HIS	E	29	13.577	-13.540	57.043	1.00	81.64	E	N	0
ATOM	7	C	HIS	E	29	18.765	-12.495	54.379	1.00	81.43	E	C	0
ATOM	8	N	HIS	E	29	17.839	-11.183	56.363	1.00	82.06	E	N	0
ATOM	9	CA	HIS	E	29	17.879	-12.461	55.607	1.00	81.34	E	C	0
ATOM	10	CB	HIS	E	133	14.298	-10.410	60.035	1.00	64.90	E	C	0
ATOM	11	CG	HIS	E	133	15.154	-11.333	60.826	1.00	66.47	E	C	0
ATOM	12	CD2	HIS	E	133	15.085	-11.667	62.161	1.00	67.06	E	C	0
ATOM	13	ND1	HIS	E	133	16.213	-12.014	60.257	1.00	66.88	E	N	0
ATOM	14	CE1	HIS	E	133	16.789	-12.742	61.218	1.00	66.30	E	C	0
ATOM	15	NE2	HIS	E	133	16.125	-12.555	62.381	1.00	66.13	E	N	0
ATOM	16	CB	HIS	E	135	20.627	-7.695	60.116	1.00	47.65	E	C	0
ATOM	17	CG	HIS	E	135	19.882	-8.934	59.755	1.00	48.47	E	C	0
ATOM	18	CD2	HIS	E	135	18.636	-9.171	59.224	1.00	48.90	E	C	0
ATOM	19	ND1	HIS	E	135	20.453	-10.177	59.929	1.00	48.82	E	N	0
ATOM	20	CE1	HIS	E	135	19.573	-11.126	59.543	1.00	49.03	E	C	0
ATOM	21	NE2	HIS	E	135	18.456	-10.538	59.117	1.00	48.70	E	N	0
ATOM	22	CU+2	CU2	E	420	16.855	-11.559	58.403	1.00	86.08	E	CU	2

END

REMARK 3RFR, Subunit 2, Oxidised state with an extra water molecule

REMARK DATE:12-Apr-2017 14:20:10 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	E	29	16.503	-12.684	54.985	1.00	81.59	E	C	0
ATOM	2	CG	HIS	E	29	15.410	-12.873	56.005	1.00	81.34	E	C	0
ATOM	3	CD2	HIS	E	29	14.109	-13.263	55.779	1.00	81.73	E	C	0
ATOM	4	ND1	HIS	E	29	15.605	-12.809	57.390	1.00	80.25	E	N	0
ATOM	5	CE1	HIS	E	29	14.461	-13.186	57.970	1.00	81.19	E	C	0
ATOM	6	NE2	HIS	E	29	13.540	-13.459	57.022	1.00	82.26	E	N	0
ATOM	7	C	HIS	E	29	18.771	-12.496	54.383	1.00	82.28	E	C	0
ATOM	8	N	HIS	E	29	17.794	-11.169	56.352	1.00	82.90	E	N	0
ATOM	9	CA	HIS	E	29	17.887	-12.464	55.627	1.00	82.16	E	C	0
ATOM	10	CB	HIS	E	133	14.300	-10.410	60.034	1.00	64.98	E	C	0
ATOM	11	CG	HIS	E	133	15.159	-11.324	60.816	1.00	66.53	E	C	0
ATOM	12	CD2	HIS	E	133	15.113	-11.656	62.154	1.00	67.04	E	C	0
ATOM	13	ND1	HIS	E	133	16.202	-12.017	60.231	1.00	67.00	E	N	0
ATOM	14	CE1	HIS	E	133	16.792	-12.747	61.181	1.00	66.38	E	C	0
ATOM	15	NE2	HIS	E	133	16.150	-12.551	62.356	1.00	66.09	E	N	0
ATOM	16	CB	HIS	E	135	20.624	-7.703	60.117	1.00	47.70	E	C	0
ATOM	17	CG	HIS	E	135	19.873	-8.929	59.759	1.00	48.53	E	C	0
ATOM	18	CD2	HIS	E	135	18.628	-9.156	59.220	1.00	49.10	E	C	0
ATOM	19	ND1	HIS	E	135	20.431	-10.180	59.932	1.00	48.90	E	N	0
ATOM	20	CE1	HIS	E	135	19.547	-11.123	59.537	1.00	48.90	E	C	0
ATOM	21	NE2	HIS	E	135	18.438	-10.522	59.106	1.00	48.78	E	N	0
ATOM	22	O	HOH	E	419	18.763	-14.119	58.597	1.00	76.39	E	O	0
ATOM	23	CU+2	CU2	E	420	16.863	-11.579	58.389	1.00	84.92	E	CU	2

END

REMARK 3RFR, Subunit 3, Reduced state

REMARK DATE:28-Nov-2016 12:00:25 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	I	29	71.009	-19.302	34.919	1.00	76.64	I	C	0
ATOM	2	CG	HIS	I	29	72.127	-20.295	35.141	1.00	76.90	I	C	0
ATOM	3	CD2	HIS	I	29	72.990	-20.827	34.204	1.00	76.91	I	C	0
ATOM	4	ND1	HIS	I	29	72.545	-20.746	36.406	1.00	77.70	I	N	0
ATOM	5	CE1	HIS	I	29	73.631	-21.503	36.216	1.00	76.48	I	C	0
ATOM	6	NE2	HIS	I	29	73.923	-21.588	34.891	1.00	76.21	I	N	0
ATOM	7	C	HIS	I	29	70.053	-17.100	35.655	1.00	78.29	I	C	0
ATOM	8	N	HIS	I	29	70.122	-19.079	37.154	1.00	78.47	I	N	0
ATOM	9	CA	HIS	I	29	70.843	-18.324	36.121	1.00	77.51	I	C	0

ATOM	10	CB	HIS	I	133	72.019	-23.668	38.446	1.00	68.19	I	C	0
ATOM	11	CG	HIS	I	133	73.118	-23.008	39.295	1.00	69.03	I	C	0
ATOM	12	CD2	HIS	I	133	74.179	-23.625	39.951	1.00	69.37	I	C	0
ATOM	13	ND1	HIS	I	133	73.222	-21.629	39.434	1.00	68.64	I	N	0
ATOM	14	CE1	HIS	I	133	74.313	-21.403	40.158	1.00	68.41	I	C	0
ATOM	15	NE2	HIS	I	133	74.932	-22.579	40.489	1.00	69.35	I	N	0
ATOM	16	CB	HIS	I	135	68.328	-19.987	42.685	1.00	43.79	I	C	0
ATOM	17	CG	HIS	I	135	69.341	-19.969	41.642	1.00	43.91	I	C	0
ATOM	18	CD2	HIS	I	135	69.639	-20.796	40.577	1.00	43.76	I	C	0
ATOM	19	ND1	HIS	I	135	70.149	-18.862	41.521	1.00	43.89	I	N	0
ATOM	20	CE1	HIS	I	135	70.893	-19.016	40.385	1.00	44.83	I	C	0
ATOM	21	NE2	HIS	I	135	70.614	-20.184	39.803	1.00	43.38	I	N	0
ATOM	22	CU+2	CU2	I	420	71.598	-20.589	38.156	1.00	82.47	I	CU	2

END

REMARK 3RFR, Subunit 3, Oxidised state

REMARK DATE:28-Nov-2016 12:00:25 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	I	29	71.003	-19.318	34.918	1.00	75.89	I	C	0
ATOM	2	CG	HIS	I	29	72.119	-20.308	35.143	1.00	76.17	I	C	0
ATOM	3	CD2	HIS	I	29	73.003	-20.823	34.220	1.00	76.17	I	C	0
ATOM	4	ND1	HIS	I	29	72.531	-20.771	36.411	1.00	77.04	I	N	0
ATOM	5	CE1	HIS	I	29	73.635	-21.520	36.230	1.00	75.75	I	C	0
ATOM	6	NE2	HIS	I	29	73.932	-21.579	34.915	1.00	75.55	I	N	0
ATOM	7	C	HIS	I	29	70.058	-17.102	35.657	1.00	77.52	I	C	0
ATOM	8	N	HIS	I	29	70.266	-19.125	37.215	1.00	77.77	I	N	0
ATOM	9	CA	HIS	I	29	70.880	-18.337	36.116	1.00	76.79	I	C	0
ATOM	10	CB	HIS	I	133	71.993	-23.684	38.411	1.00	69.80	I	C	0
ATOM	11	CG	HIS	I	133	73.028	-22.960	39.244	1.00	70.43	I	C	0
ATOM	12	CD2	HIS	I	133	74.103	-23.447	39.972	1.00	70.29	I	C	0
ATOM	13	ND1	HIS	I	133	73.043	-21.567	39.284	1.00	70.58	I	N	0
ATOM	14	CE1	HIS	I	133	74.097	-21.207	40.035	1.00	70.14	I	C	0
ATOM	15	NE2	HIS	I	133	74.761	-22.317	40.452	1.00	69.87	I	N	0
ATOM	16	CB	HIS	I	135	68.323	-19.992	42.681	1.00	43.95	I	C	0
ATOM	17	CG	HIS	I	135	69.341	-19.971	41.639	1.00	44.15	I	C	0
ATOM	18	CD2	HIS	I	135	69.649	-20.779	40.562	1.00	43.82	I	C	0
ATOM	19	ND1	HIS	I	135	70.143	-18.857	41.551	1.00	44.25	I	N	0
ATOM	20	CE1	HIS	I	135	70.897	-18.964	40.431	1.00	45.19	I	C	0
ATOM	21	NE2	HIS	I	135	70.622	-20.123	39.816	1.00	43.42	I	N	0
ATOM	22	CU+2	CU2	I	420	71.602	-20.533	38.166	1.00	83.83	I	CU	2

END

REMARK 3RFR, Subunit 3, Oxidised state with an extra water molecule

REMARK DATE:12-Apr-2017 14:54:54 created by user: lili

REMARK CNS VERSION:1.3

ATOM	1	CB	HIS	I	29	70.997	-19.319	34.928	1.00	77.06	I	C	0
ATOM	2	CG	HIS	I	29	72.114	-20.311	35.138	1.00	77.31	I	C	0
ATOM	3	CD2	HIS	I	29	72.969	-20.855	34.203	1.00	77.29	I	C	0
ATOM	4	ND1	HIS	I	29	72.552	-20.752	36.405	1.00	78.09	I	N	0
ATOM	5	CE1	HIS	I	29	73.641	-21.518	36.213	1.00	76.86	I	C	0
ATOM	6	NE2	HIS	I	29	73.908	-21.606	34.891	1.00	76.67	I	N	0
ATOM	7	C	HIS	I	29	70.060	-17.099	35.661	1.00	78.66	I	C	0
ATOM	8	N	HIS	I	29	70.286	-19.111	37.236	1.00	78.87	I	N	0
ATOM	9	CA	HIS	I	29	70.893	-18.334	36.127	1.00	77.97	I	C	0
ATOM	10	CB	HIS	I	133	71.995	-23.681	38.415	1.00	70.54	I	C	0
ATOM	11	CG	HIS	I	133	73.017	-22.958	39.248	1.00	71.12	I	C	0
ATOM	12	CD2	HIS	I	133	74.087	-23.432	39.993	1.00	70.93	I	C	0
ATOM	13	ND1	HIS	I	133	73.038	-21.564	39.263	1.00	71.41	I	N	0
ATOM	14	CE1	HIS	I	133	74.088	-21.189	40.012	1.00	70.91	I	C	0
ATOM	15	NE2	HIS	I	133	74.745	-22.295	40.457	1.00	70.55	I	N	0
ATOM	16	CB	HIS	I	135	68.328	-19.991	42.678	1.00	43.89	I	C	0
ATOM	17	CG	HIS	I	135	69.339	-19.974	41.642	1.00	44.04	I	C	0
ATOM	18	CD2	HIS	I	135	69.644	-20.780	40.561	1.00	43.68	I	C	0
ATOM	19	ND1	HIS	I	135	70.146	-18.862	41.551	1.00	44.25	I	N	0
ATOM	20	CE1	HIS	I	135	70.902	-18.969	40.429	1.00	44.79	I	C	0
ATOM	21	NE2	HIS	I	135	70.619	-20.126	39.816	1.00	43.21	I	N	0
ATOM	22	O	HOH	I	419	73.631	-17.917	38.640	1.00	60.97	I	O	0
ATOM	23	CU+2	CU2	I	420	71.621	-20.507	38.169	1.00	83.12	I	CU	2

END