Electronic Supplementary Information For

Rational Design of Substituted Maleimide Dyes with Tunable Fluorescence and Solvafluorochromism

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Contents

1. Materials and apparatus	S3
2. Chemical structures of maleimides and synthetic protocols.	S4
Chemical structures of maleimides	S4
Experimental procedure for di-halogen maleimides prepara	tion . S5
General procedure of aminomaleimides synthesis SError! Bo not defined.	okmark
General procedure for alkoxymaleimides synthesis	S8
General procedure for thioaminomaleimide synthesis:	S10
3. Fluorophore characterization	S15
4. UV-Vis Spectrum of maleimides	S17
5. Solvatochromic determination	S19
6. Photographs in different solvents under UV light	S21
7. DSE effect of second thiol substituted amino maleimides	S22
8. NMR spetra of substituted maleimides	S23
9. Computational Details	S44
10. References	S67

1. MATERIALS AND APPARATUS

All chemicals and reagents were purchased from either: Sigma Aldrich, Fisher Chemicals, Acros Chemicals or Alfa Aesar and used as received. Dithiolmaleimides was synthesized according to a previous paper.¹ Solvents were purchased from Fisher Scientific and used as received. Dry solvents were used directly from a drying and degassing solvent tower delivery system.

NMR spectra were recorded on a Bruker Avance 300, a Bruker Avance III HD 400 or a Bruker Avance III HD 500 spectrometer at 298k and 300, 400 and 500 MHz, respectively. Shifts are quoted in δ in parts per million and quoted relative to the internal standard trimethylsilane (TMS). High Resolution Mass Spectra (HR-MS) were conducted on a Bruker UHR-Q-ToF MaXis spectrometer with electrospray ionization. Infrared spectra were recorded (neat) on a PerkinElmer 100 FT-IR, UV-Vis spectroscopy was carried out on a Perkin Elmer Lambda 365 UV/vis spectrometer or an Agilent Cary 60 UV-Vis Spectrometer at room temperature. Fluorescence spectra were recorded using an Agilent Cary Eclipse Fluorescence spectrophotometer. Quartz cells with four polished sides (Starna) were used for fluorescence and UV-vis measurements. Fluorescence lifetime and absolute fluorescence quantum yield were performed on an Edinburgh Instruments FLS920 steady-state spectrometer fitted with an integrating sphere. Experiments were carried out in solution using 1 cm path length quartz cuvettes with four transparent polished faces.

2. CHEMICAL STRUCTURES OF MALEIMIDES AND SYNTHETIC

PROTOCOLS

Chemical structures of maleimides



Experimental procedure for di-halogen maleimides preparation



3,4-diiodo-1-methyl-1H-pyrrole-2,5-dione (1c):



This compound was prepared by the method from the previous literature.² 3,4-dibromo-N-methylmaleimide (0.5 g, 1.859 mmol) and sodium iodide (1.115 g, 7.437 mmol) in acetic acid (20 ml) was heated at refluxing temperature for 2 h. Then the solution was added to water. The obtained yellow precipitate was filtered,

washing with water and drying at 60 °C to obtain the 3,4-diiodo-1-methyl-1H-pyrrole-2,5-dione (0.527 g, 78.3%). Matches literature data.²

TLC conditions: $R_f = 0.53$ (petroleum ether: ethyl acetate= 4:1), Yield =78.3%. ¹H NMR (400 MHz, CDCl₃, ppm) δ = 3.15 (1H, s, H1).¹³C NMR (101 MHz, CDCl₃, ppm) δ =166.2 (C2), 117.1 (C3), 26.0 (C1). HR-MS (MaXis) m/z found 358.8141 (M+Na)⁺, calculated 385.8145; FTIR (neat) v_{max} / cm⁻¹ 1695 and 1560 (C=O of maleimide), 1426 and 1361 (C=C of maleimide).

3,4-dichloro-1-methyl-1H-pyrrole-2,5-dione (1a):



This compound was prepared by the method from the previous literature.³ 3,4-dichloromaleic anhydride (1 g, 5.98 mmol), methylamine hydrochloride (0.605 g, 8.97 mmol), and potassium acetate (1.24 g, 8.97 mmol) were added to a solution of acetic acid (10 ml), and stirred for 4 hrs under reflux. The reaction become

yellow, and was cooled to 25-30 °C, and poured slowly onto chilled sodium bicarbonate solution (1 M, 50 ml) until no effervesce was observed, followed by stirring for 1 hour. The solids were filtered, washed with *n*-hexane (2 x 30 ml) and dried under vacuum to obtain 3,4-dichloro-1-methyl-1H-pyrrole-2,5-dione (0.732 g, 68%). Matches literature data.³

TLC conditions: $R_f = 0.61$ (petroleum ether: ethyl acetate= 4:1), Yield =68.0%. ¹H NMR (400 MHz, CDCl₃) δ = 3.12 (1H, s, H1); ¹³C NMR (101 MHz, CDCl₃, ppm) δ = 162.8 (C2), 133.51 (C3), 25.1 (C1); FTIR (neat) v_{max} / cm⁻¹ 1708 and 1617 (C=O of maleimide), 1436 and 1381 (C=C of maleimide).

General procedure of aminomaleimides synthesis



Reactions were performed according to the protocol established by a previous paper.⁴ All reactions were performed in THF (20 mL) at room temperature with 2,3-dibromomaleimide/2,3-dibromomethylmaleimide (1 eq.), sodium carbonate (2.5 eq.) and a small excess of amine (1.05-1.1 eq.). Consumption of 2,3-dibromomaleimide was monitored by TLC, and was complete within 30 min to 2 h. The solvent was then evaporated under reduced pressure and the residue was taken up with 150 mL of CH₂Cl₂. The resultant mixture was washed with water (2 × 150 mL), dried with sodium sulfate and purified *via* column chromatography on silica gel with petroleum ether/ethyl acetate.

3-chloro-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (2a):



TLC conditions: $R_f = 0.60$ (petroleum ether: ethyl acetate= 2:1), Yield =60.5%. ¹H NMR (400 MHz, CDCl₃, ppm) δ =5.30 (1H, s, H2), 3.76 – 3.58 (2H, m, H3), 3.00 (3H, s, H1), 1.30 (3H, t, ³J_{H-H} = 7.2 Hz, H4); ¹³C NMR (101 MHz, CDCl₃, ppm) δ =38.1 (C3), 24.2 (C1), 16.1 (C4); HR-MS (MaXis) m/z found 211.0247

 $(M+Na)^+$, calculated 211.0245; FTIR (neat) v_{max} / cm^{-1} 3344 (H-N of amine), 1694 and 1629 (C=O of maleimide), 1436 and 1372 (C=C of maleimide).

3-bromo-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (2b):



TLC conditions: $R_f = 0.60$ (petroleum ether: ethyl acetate= 2:1) Yield =54.3%. ¹H NMR (400 MHz, CDCl₃, ppm) δ =5.35 (1H, s, H2), 3.78 – 3.63 (2H, m, H3), 3.02 (3H, s, H1), 1.29 (3H, t, ³J_{H-H} = 7.2 Hz, H4); ¹³C NMR (101 MHz, CDCl₃, ppm) δ =38.0 (C3), 24.4 (C1), 15.8 (C4); HR-MS (MaXis) m/z found 256.9720 $(M+Na)^+$, calculated 256.9719; FTIR (neat) v_{max} / cm^{-1} 3332 (H-N of amine), 1707 and 1655 (C=O of maleimide), 1449 and 1372 (C=C of maleimide).

3- iodo-4-(ethylamino)- 1-methyl-1H-pyrrole-2,5-dione (2c):



TLC conditions: $R_f = 0.60$ (petroleum ether: ethyl acetate= 2:1); Yield = 32.1%; ¹H NMR (400 MHz, CDCl₃, ppm) δ =5.35 (1H, s, H2), 3.78 – 3.63 (2H, m, H3), 3.02 (3H, s, H1), 1.29 (3H, t, ³J_{H-H} 7 Hz, H4). ¹³C NMR (101 MHz, CDCl₃, ppm) δ =38.0 (C3), 24.4 (C1), 15.8 (C4); HR-MS (MaXis)

m/z found 302.9604 (M+Na)⁺, calculated 302.9601. FTIR (neat) v_{max} / cm⁻¹ 3332 (H-N of amine), 1604 (C=O of maleimide), 1449 (C=C of maleimide).

3-(butylamino)-4-chloro-1-methyl-1H-pyrrole-2,5-dione (2d):



TLC conditions: $R_f = 0.50$ (petroleum ether: ethyl acetate= 2:1), Yield = 58.7%. ¹H NMR (400 MHz, CDCl₃, ppm) δ =7.08 (1H, s, H1), 5.31 (1H, s, H2), 3.62 (2H, td, ³J_{H-H} = 14, 7 Hz, H3), 1.74 – 1.53 (2H, m, ³J_{H-H} = 7 Hz, H4), 1.48 – 1.34 (2H, sex, ³J_{H-H} = 7 Hz, H5), 0.97 (3H, t, ³J_{H-H} = 7 Hz, H6). ¹³C

NMR (101 MHz, CDCl₃, ppm) δ =42.8 (C3), 32.8 (C4), 19.6 (C5), 13.6 (C6); HR-MS (MaXis) m/z found 225.0404 (M+Na) ⁺, calculated 225.0401; FTIR (neat) v_{max} / cm⁻¹ 3319 (H-N of amine), 3138 (H-N of maleimide) 1772 and 1707 (C=O of maleimide), 1404 and 1359 (C=C of maleimide).

3-bromo-4-(butylamino)-1H-pyrrole-2,5-dione (2e):



TLC conditions: $R_f = 0.50$ (petroleum ether: ethyl acetate= 2:1), Yield = 69.2%; ¹H NMR (400 MHz, CDCl₃, ppm) δ =7.17 (1H, s, H1), 5.41 (1H, s, H2), 3.65 (2H, dd, J = 7 Hz, H3), 1.70 – 1.58 (2H, quin, ³J_{H-H} = 7 Hz, H4), 1.42 (2H, sex, ³J_{H-H} = 7 Hz, H5), 0.97 (3H, t, ³J_{H-H} = 7 Hz, H6). ¹³C NMR

(101 MHz, CDCl₃) δ =42.9 (C3), 32.7 (C4), 19.5 (C5), 13.7 (C6); HR-MS (MaXis) m/z found 268.9898 (M+Na) ⁺, calculated 268.9896. FTIR (neat) v_{max} / cm⁻¹ 3344 (H-N of amine), 3138 (H-N of maleimide) 1617 and 1707 (C=O of maleimide), 1462 and 1333 (C=C of maleimide);

General procedure for alkoxymaleimides synthesis



Reactions were performed according to the protocol established by a previous paper.⁵ Sodium alkoxide solution was freshly prepared by the addition of sodium (2 mmol, 1 equiv) to the desired alcohol. This solution was then added dropwise to a cooled solution of the maleimide (2 mmol, 1 equiv) in the same alcohol (10 mL). This was kept at 0 °C in an ice bath. Consumption of 2, 3-dibromo/chloro maleimide was monitored by TLC, and was complete within 30 min to 1 h. Then the reaction mixture was concentrated in vacuo and separated between ethyl acetate and aqueous ammonium chloride solution. The aqueous layer was washed with ethyl acetate (50 mL×3), then the combined organic extracts washed with water (50 mL×3), brine, and dried with Na₂SO₄. Concentration *in vacuo*, followed by purification *via* column chromatography on silica gel with petroleum ether/ethyl acetate, afforded the products.

3-chloro-4-ethoxy-1-methyl-1H-pyrrole-2,5-dione (3a):



TLC conditions: $R_f = 0.38$ (petroleum ether: ethyl acetate= 4:1). Yield = 55.1%. ¹H NMR (400 MHz, CDCl₃, ppm) δ =4.72 (2H, q, ³J_{H-H} = 7 Hz, H2), 3.05 (3H, s, H1), 1.49 (3H, t, ³J_{H-H} = 7 Hz, H3). ¹³C NMR (101 MHz, CDCl₃, ppm) δ =68.8 (C2), 23.9 (C1), 15.1 (C3). HR-MS (MaXis) m/z found 212.0084 (M+Na) ⁺,

calculated 212.0085; FTIR (neat) v_{max} / cm⁻¹ 1707 and 1655 (C=O of maleimide), 1436 and 1384 (C=C of maleimide), 1255 (C-O of alcohol).

3-bromo-4-ethoxy-1-methyl-1H-pyrrole-2, 5-dione (3b):



TLC conditions: $R_f = 0.38$ (petroleum ether: ethyl acetate= 4:1). Yield = 42.5%.; ¹H NMR (400 MHz, CDCl₃, ppm) δ =4.70 (2H, q, ³J_{H-H} = 7 Hz, H2), 3.03 (3H, s, H1), 1.47 (3H, t, ³J_{H-H} = 7 Hz, H3); ¹³C NMR (101 MHz, CDCl₃) δ 68.7 (C2), 24.1 (C1), 15.4 (C3); FTIR (neat) v_{max} / cm⁻¹ 1720 and 1640 (C=O of maleimide), 1436 and 1382 (C=C of maleimide), 1268 (C-O of alcohol).

3-(benzyloxy)-4-chloro-1-methyl-1H-pyrrole-2,5-dione (3c):



TLC conditions: $R_f = 0.41$ (petroleum ether: ethyl acetate= 4:1). Yield =43.7%; ¹H NMR (400 MHz, DMSO, ppm) δ =7.60 – 7.39 (5H, m, H3), 5.71 (2H, s, H2), 2.94 (3H, s, H3); ¹³C NMR (101 MHz, DMSO, ppm) δ =131.24 – 125.94 (m, C3), 73.83 (C2), 24.42 (C1); HR-MS (MaXis) *m/z* found 274.0243, calc. 274.0241 ([M+Na] ^{+,}100%); FTIR (neat)

 v_{max} / cm⁻¹ 1720 and 1642 (C=O of maleimide), 1449 and 1397 (C=C of maleimide), 1255 (C-O of alcohol), 752 and 701 (C-H of aromatic).

3-(benzyloxy)-4-bromo-1-methyl-1H-pyrrole-2,5-dione (3d):



TLC conditions: $R_f = 0.41$ (petroleum ether: ethyl acetate= 4:1). Yield = 32.1%; ¹H NMR (400 MHz, DMSO, ppm) δ =7.55 – 7.35 (5H, m, H3), 5.69 (2H, s, H2), 2.90 (3H, s, H3); ¹³C NMR (101 MHz, DMSO, ppm) δ =130.4 – 126.2 (m, C3), 73.8 (C2), 24.5 (C1); HR-MS (MaXis) *m/z* found

317.9731, calc. 317.9736 ([M+Na] ^{+,}100%); FTIR (neat) v_{max} / cm⁻¹ 1707 and 1642 (C=O of maleimide), 1423 and 1382 (C=C of maleimide), 1255 (C-O of alcohol), 740 and 688 (C-H of aromatic);

3-chloro-1-methyl-4-phenoxy-1H-pyrrole-2,5-dione (3e):



TLC conditions: $R_f = 0.41$ (petroleum ether: ethyl acetate= 4:1); Yield =55.1%; ¹H NMR (400 MHz, MeOD, ppm) δ =7.43 (2H, t, J = 8 Hz, H2), 7.28 (2H, t, J = 7 Hz, H3), 7.23 (1H, d, J = 8 Hz,), 3.03 (3H, s, H1). ¹³C NMR (101 MHz, MeOD, ppm) δ =129.43 (C2), 125.47 (C3), 118.70 (C4),

22.87 (C1); HR-MS (MaXis) m/z found 260.0085 (M+Na)⁺, calculated 260.0085; FTIR (neat) v_{max} / cm⁻¹ 1720 (C=O of maleimide), 1433 and 1392 (C=C of maleimide), 1270 (C-O of alcohol), 774 and 695 (C-H of aromatic).

3-bromo-1-methyl-4-phenoxy-1H-pyrrole-2,5-dione (3f):



TLC conditions: $R_f = 0.41$ (petroleum ether: ethyl acetate= 4:1); Yield =51.0%. ¹H NMR (400 MHz, DMSO, ppm) δ =7.43 (2H, t, J = 7.5 Hz, H2), 7.28 (3H, dd, J = 14, 7 Hz, H2,3), 2.93 (3H, s, H1); ¹³C NMR (101 MHz, DMSO, ppm) δ =130.2 (C2), 125.9 (C3), 119.1 (C4), 24.7 (C1). HR-MS

(MaXis) m/z found 303.9586 (M+Na) ⁺, calculated 303.9580; FTIR (neat) v_{max} / cm⁻¹ 1720 (C=O of maleimide), 1436 and 1359 (C=C of maleimide), 1165 (C-O of alcohol), 830 and 714 (C-H of aromatic).

General procedure for thioaminomaleimide synthesis:



Aminobromomaleimide (2 mmol, 1 equiv.) was dissolved in DMF (30 mL) followed by addition of NaOH (3 mmol, 1.5 equiv). The mixture was stirred for 5 minutes and then thiol (2 mmpl, 1 equiv.) was added dropwise. Upon complete addition, the solution was allowed to warm to 80 °C and left to stir for 20 hours. Then the reaction mixture was concentrated in vacuo and dissolved in ethyl acetate. The organic solution washed with water, brine and dried over anhydrous Na₂SO₄. The solution was filtered and concentrated *in vacuo*, and the crude product was purified by column chromatography (petroleum ether: ethyl acetate = 10:1-3:1) to yield the yellow product.

3-(butylthio)-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (4a):



 $\begin{array}{ll} TLC \ \ conditions: \ R_{f} \ = \ 0.63 \ \ (petroleum \ ether: \ ethyl \\ acetate= 2:1); \ Yield = 36.8\%; \ ^{1}H \ NMR \ (400 \ MHz, \ CDCl_{3}, \\ ppm) \ \delta = 5.54 \ (1H, \ s, \ H2), \ 3.88 \ - \ 3.79 \ (2H, \ m, \ H3), \ 3.01 \\ (3H, \ s, \ H1), \ 2.71 \ - \ 2.65 \ (2H, \ t, \ ^{3}J_{H-H} = 7 \ Hz, \ H5), \ 1.61 \ - \\ 1.51 \ (2H, \ quin, \ ^{3}J_{H-H} = 7 \ Hz, \ H6), \ 1.41 \ (2H, \ sex, \ ^{3}J_{H-H} = 7 \\ Hz, \ H7), \ 1.30 \ (3H, \ t, \ ^{3}J_{H-H} = 7 \ Hz, \ H4 \), \ 0.91 \ (3H, \ t, \ ^{3}J_{H-H} \ - 7 \ Hz, \ H4 \), \ 0.91 \ (3H, \ t, \ ^{3}J_{H-H} \ + 10 \ Hz, \ (3H, \ t, \) \) \ (3H, \ t, \) \ (3H, \ t, \) \) \ (3H, \ t, \) \ (3H, \ t, \) \) \) \ (3H, \ t, \)$

= 7 Hz, H8); ¹³C NMR (101 MHz, CDCl₃, ppm) δ =38.4 (C3), 35.7 (C5), 31.9 (C6), 24.1 (C1), 21.8 (C7), 15.4 (C4), 13.6 (C8); HR-MS (MaXis) *m/z* found 265.0983, calc. 265.0981 ([M+Na]^{+,} 100%); FTIR (neat) v_{max} / cm⁻¹ 3307 (H-N of amine), 1708 and 1617 (C=O of maleimide), 1436 and 1371 (C=C of maleimide).

3-(ethylamino)-1-methyl-4-(octadecanethio)-1H-pyrrole-2,5-dione (4b):



TLC conditions: $R_f = 0.39$ (petroleum ether: ethyl acetate= 2:1); Yield =18.2 %; ¹H NMR (400 MHz, CDCl₃, ppm) δ =5.53 (1H, s, H2), 3.82 (2H, p, ³J_{H-H} = 7 Hz, H3), 3.00 (3H, s, H1), 2.69 – 2.64 (2H, m, H5), 1.32 – 1.19 (34H, m, H6+H4), 0.88 (3H, t, *J* = 7 Hz, H7); ¹³C NMR (1 MHz, CDCl₃, ppm) δ =38.5 (C3), 36.2 (C5), 31.9- 29.8 (C6), 24.2 (C4), 22.9 (C6), 15.3 (C7), 14.0

(C1); HR-MS (MaXis) m/z found 461.3171, calc. 461.3172 ([M+Na] ^{+,} 100%); FTIR (neat) v_{max} / cm^{-1} 3320 (H-N of amine), 1700 and 1621 (C=O of maleimide), 1463 and 1391 (C=C of maleimide).

3-(ethylamino)-1-methyl-4-(phenylthio)-1H-pyrrole-2,5-dione (4c):



TLC conditions: $R_f = 0.42$ (petroleum ether: ethyl acetate= 2:1); Yield =50.9%; ¹H NMR (400 MHz, CD₃CN, ppm) δ = 7.30 (2H, dd, ³J_{H-H} = 10, 4 Hz, H5), 7.23 (2H, dd, ³J_{H-H} = 5, 3 Hz, H6), 7.19 – 7.14 (1H, m, H7), 3.70 (2H, p, ³J_{H-H} = 7 Hz, H3), 2.96 (3H, s, H1), 1.16 (t, ³J_{H-H} = 3H, 7 Hz, H4). ¹³C NMR (101 MHz, CD₃CN, ppm) δ =129.6 (C5), 126.0 (C6), 125.8 (C7), 38.3 (C3),

24.1 (C1), 15.4 (C4). HR-MS (MaXis) m/z found 285.0667, calc. 285.0668 ([M+Na]^{+,} 100%); FTIR (neat) v_{max} / cm^{-1} 3313 (H-N of amine), 1696 and 1614 (C=O of maleimide), 1437 and 1390 (C=C of maleimide), 792 and 683 (C-H of aromatic).

3-(ethylamino)-1-methyl-4-(o-tolylthio)-1H-pyrrole-2,5-dione (4d):



TLC conditions: $R_f = 0.47$ (petroleum ether: ethyl acetate= 2:1); Yield =55.6%; ¹H NMR (400 MHz, CD₃CN, ppm) δ =7.17 (1H, t, ³J_{H-H} = 8 Hz,H5), 7.12 (1H, d, ³J_{H-H} = 7.5 Hz, H6), 7.06 (2H, t, ³J_{H-H} = 6 Hz,H7,H8), 3.65 (2H, p, ³J_{H-H} = 7 Hz, H3), 2.97 (3H, s,H1), 2.38 (3H, s, H9), 1.13 (1H, t, ³J_{H-H} = 7 Hz,H4); ¹³C NMR (101 MHz, CD₃CN, ppm) δ =130.6 (C5), 127.1 (C7), 125.4 (C6),

125.2 (C8), 38.3 (C3), 24.1 (C1), 19.3 (C9), 15.4 (C4); HR-MS (MaXis) m/z found 299.0827, calc. 299.0825 ([M+Na] ^{+,}100%); FTIR (neat) v_{max} / cm⁻¹ 3327 (H-N of amine),

1703 and 1602 (C=O of maleimide), 1436 and 1383 (C=C of maleimide), 797 and 744 (C-H of aromatic).

3-(benzylthio)-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (4e):



TLC conditions: $R_f = 0.39$ (petroleum ether: ethyl acetate= 2:1); Yield =11.9%; ¹H NMR (400 MHz, DMSO) δ 7.23 (5H, ddd, J = 26, 16, 7 Hz, H 4-6), 3.78 (2H, s, H3), 3.27 (2H, dd, J = 13, 6 Hz, H7), 2.88 (3H, s, H1), 0.90 (3H, t, J = 7 Hz, H8); ¹³C NMR (101 MHz, DMSO, ppm) δ 129.2 (C4), 128.7, 127.3 (C5, C6), 37.8 (C7), 24.1 (C1), 15.0 (C8); HR-MS (MaXis) *m/z* found

299.0825, calc. 299.0825 ([M+Na]^{+,} 100%); FTIR (neat) v_{max} / cm⁻¹ 3306 (H-N of amine), 1707 and 1604 (C=O of maleimide), 1345 (C=C of maleimide, 757 and 641 (C-H of aromatic).

3-(butylamino)-1-methyl-4-(phenylthio)-1H-pyrrole-2,5-dione (4f):



TLC conditions: $R_f = 0.23$ (petroleum ether: ethyl acetate= 2:1). Yield =42.3%. ¹H NMR (400 MHz, DMSO, ppm) δ =7.28 (2H, t, *J* = 7 Hz, H7), 7.13 (3H, dd, *J* = 14, 7 Hz, H8-9), 3.52 (2H, d, *J* = 6 Hz, H3), 2.91 (3H, s, H1), 1.52 - 1.33 (2H, m, H4), 1.17 (2H, dd, *J* = 14, 7 Hz, H5), 0.71 (3H, t, *J* = 7 Hz, H6); ¹³C NMR (101 MHz,

DMSO) δ 129.5 (C7), 125.4 (C8), 125.3 (C9), 42.6 (C3), 32.2 (C4), 24.5 (C1), 19.6 (C5), 13.9 (C6); HR-MS (MaXis) *m/z* found 289.1015, calc. 289.1016 ([M+Na]^{+,} 100%); FTIR (neat) v_{max} / cm⁻¹ 3332 (H-N of amine), 1681 and 1604 (C=O of maleimide), 1436 and 1384 (C=C of maleimide), 740 and 688 (C-H of aromatic).

1-methyl-3-(phenylamino)-4-(phenylthio)-1H-pyrrole-2,5-dione (4g):



TLC conditions: $R_f = 0.44$ (petroleum ether: ethyl acetate= 2:1); Yield =44.2%; ¹H NMR (400 MHz, DMSO, ppm) δ = 10.10 (1H, s, H2), 7.18 – 6.99 (8H, m, H4-8), 6.91 (H2, d, J = 7 Hz, H3), 2.97 (3H, s, H1); ¹³C NMR (101 MHz, DMSO, ppm) δ 129.1 (C5), 128.0 (C8), 126.3 (C4), 125.8 (C7), 125.5 (C5), 124.8 (C6), 24.6

(C1); HR-MS (MaXis) m/z found 333.0678, calc. 333.0668 ([M+Na] ^{+,} 100%); FTIR (neat) v_{max} / cm⁻¹ 3281 (H-N of amine), 1694 and 1604 (C=O of maleimide), 1449 (C=C of maleimide), 740 and 688 (C-H of aromatic).

3-(butylamino)-4-(phenylthio)-1H-pyrrole-2,5-dione (4h):



TLC conditions: $R_f = 0.23$ (petroleum ether: ethyl acetate= 2:1); Yield = 50.1%; ¹H NMR (400 MHz, DMSO, ppm) δ =10.75 (1H, s, H1), 8.32 (1H, s, H2), 7.29 (2H, t, ³J_{H-H} = 7 Hz, H7), 7.18 – 7.06 (3H, m, H8, H9), 3.51 (2H, d, ³J_{H-H} = 7 Hz, H3), 1.49 – 1.33 (2H, m, H4), 1.22 – 1.08 (2H, m, H5), 0.72 (3H, t, ³J_{H-H} = 7 Hz, H6); ¹³C NMR (101 MHz,

DMSO, ppm) δ 129.5 (C7), 125.3 (C8, C9), 42.4 (C3), 32.2 (C4), 19.6 (C5), 13.9 (C6); HR-MS (MaXis) *m/z* found 299.0872, calc. 299.0825 ([M+Na]^{+,} 100%); FTIR (neat) v_{max} / cm⁻¹ 3216 (H-N of amine), 1720 and 1617 (C=O of maleimide), 1345 (C=C of maleimide), 740 and 688 (C-H of aromatic).

3-(butylthio)-4-(ethylamino)-1H-pyrrole-2,5-dione (4i):



TLC conditions: $R_f = 0.39$ (petroleum ether: ethyl acetate= 2:1); Yield =18.9%; ¹H NMR (400 MHz, CDCl₃, ppm) δ =7.2 (1H, s, H1), 5.5 (1H, s, H2), 3.8 (2H, p, ³J_{H-H} = 7 Hz, H3), 2.7 – 2.5 (2H, m, H5), 1.6 – 1.5 (2H, m, H6), 1.4 – 1.3 (2H, m, H7), 1.2 (3H, ³J_{H-H} = 7 Hz, H4), 0.8 (3H, ³J_{H-H} = 7 Hz, H8). ¹³C NMR (101 MHz, CDCl₃, ppm) δ

171.1, 166.3 (C=O), 38.4 (C3), 35.6 (C5), 31.9 (C6), 21.9 (C7), 15.4(C4), 13.6 (C8); HR-MS (MaXis) *m/z* found 229.1007, calc. 229.1005 ($[M+H]^{+}$, 100%) FTIR (neat) v_{max} / cm⁻¹ 3319 (H-N of amine), 3164 (H-N of maleimide), 1707 and 1617 (C=O of maleimide), 1359 (C=C of maleimide).

3-(benzylamino)-4-(butylthio)-1H-pyrrole-2,5-dione (4j):



TLC conditions: $R_f = 0.42$ (petroleum ether: ethyl acetate= 2:1). Yield =26.4%. ¹H NMR (400 MHz, MeOD, ppm) δ =7.23 (5H, ddd, ³J_{H-H} = 25, 16, 7; H 4, 5, 6), 3.8 (2H, s, H3), 3.25 (2H, t, ³J_{H-H} = 7 Hz, H7), 1.49 – 1.19 (4H, m, H8, 9), 0.9 (3H, t, ³J_{H-H} = 7 Hz, H10); ¹³C NMR (101 MHz, MeOD, ppm) δ 128. 7, 127.8, 126.7 (C6, 5, 4), 42.0 (C7), 39.3 (C8), 31.2 (C9), 19.4 (C10);

HR-MS (MaXis) m/z found 313.0981, calc. 313.0981 ([M+Na]^{+,}100%); FTIR (neat) v_{max} / cm⁻¹ 3315 (H-N of amine), 1711 and 1620 (C=O of maleimide), 1359 (C=C of maleimide, 765 and 705 (C-H of aromatic).

3. FLUOROPHORE CHARACTERIZATION

Table S1 Structures of substituted maleimides and associated fluorescence properties



Sample	R ₁	\mathbf{R}_2	R ₃	Solvent	Фf ^a %	Emax M ⁻¹ cm ⁻¹	λ ^b abs nm	λ ^b ex nm	λ ^b em nm	Δλ ^c nm
				Diethyl Ether	37	3669	239, 369	365	475	236,106
2	CU		CI	Cyclohexane	56	3198	238, 365	364	482	244,117
2a	CH ₃	NHCH ₂ CH ₃	Cl	Methanol	1	3355	239, 379	380	550	311, 171
				Water	< 0.1	2442	241, 398	397	580	339, 182
				Diethyl Ether	30	3733	239, 370	371	474	235, 104
21	CU		D	Cyclohexane	37	3496	239, 366	364	472	233, 106
20	CH ₃	NHCH ₂ CH ₃	BL	Methanol	1	4811	241, 384	379	548	307, 164
				Water	< 0.1	3645	241, 400	397	583	342, 183
				Diethyl Ether	8	4419	237, 370	368	487	250, 117
2.	CU	NUCU CU	т	Cyclohexane	11	3483	239, 372	363	472	233, 100
20	СП3	NHCH ₂ CH ₃	1	Methanol	1	3405	238, 388	383	560	322, 172
				Water	< 0.1	2925	238, 404	396	580	342, 176
				Diethyl Ether	42	5073	226, 359	355	461	235, 102
24	TT	NH(CH ₂) ₃ CH ₃	Cl	Cyclohexane	65	4220	226, 356	352	442	216, 72
20	п		CI	Methanol	1	4313	229, 374	370	512	283, 142
_				Water	< 0.1	2867	228, 387	387	550	322, 180
				Diethyl Ether	31	5531	226, 358	358	461	235, 103
20	п	NH(CH.).CH.	D.	Cyclohexane	47	4489	226, 355	351	441	215, 86
20	п	NH(CH ₂) ₃ CH ₃	DI	Methanol	2	4451	229, 374	367	513	284, 139
				Water	< 0.1	2768	228, 387	387	547	319, 160
				Diethyl Ether	12	770	230, 335	332	458	228, 123
3a	CH_3	OCH ₂ CH ₃	Cl	Cyclohexane	24	875	237, 330	332	461	224, 131
				Methanol	2	825	236, 333	330	486	250, 153
				Water	< 0.1	548	236, 342	341	505	269, 163
				Diethyl Ether	10	666	230, 336	332	458	228, 122
2h	CH.		B r	Cyclohexane	24	668	238, 333	332	461	223, 128
50	C113	001120113	DI	Methanol	2	629	237, 335	332	486	249, 151
				Water	< 0.1	432	236, 343	344	501	265, 158
3c	CH ₃	OCH ₂ Ph	Cl	Diethyl Ether	12	1267	232, 335	334	464	232, 129
3d	CH ₃	OCH ₂ Ph	Br	Diethyl Ether	9	907	233, 340	329	465	232, 125
3e	CH ₃	OPh	Cl	Diethyl Ether	< 0.1	586	238, 333			
3f	CH ₃	OPh	Br	Diethyl Ether	< 0.1	821	243, 325			

Sample	R1	\mathbf{R}_2	R 3	Solvent	Фf ^a %	€max M ⁻¹ cm ⁻¹	λ ^b abs nm	λ ^b ex nm	λ ^b em nm	Δλ ^c nm
4a	CH_3	NHCH ₂ CH ₃	S(CH ₂) ₃ CH ₃	Diethyl Ether	4	1917	240, 379	377	558	318, 165
4b	CH_3	NHCH ₂ CH ₃	S(CH ₂) ₁₇ CH ₃	Diethyl Ether	6	2790	232, 383	377	561	329, 160
4c	CH_3	NHCH ₂ CH ₃	SPh	Diethyl Ether	12	1761	230, 370	364	545	315, 160
4d	CH_3	NHCH ₂ CH ₃	SPhCH ₃	Diethyl Ether	12	2595	236, 370	367	547	311, 161
4e	CH_3	NHCH ₂ CH ₃	SCH_2Ph	Diethyl Ether	4	2130	238, 381	378	564	326, 164
4f	CH_3	NH(CH ₂) ₃ CH ₃	SPh	Diethyl Ether	9	3898	246, 368	359	544	298, 160
4g	CH_3	NHPh	SPh	Diethyl Ether	< 0.1	4202	230, 398			
				Diethyl Ether	23	2685	230, 355	355	526	296, 157
46			CDh	Cyclohexane	37	4743	241, 355	347	513	272, 147
411	п	$N\Pi(C\Pi_2)_3C\Pi_3$	SPII	Methanol	< 0.1	4269	241, 370	364	593	352, 223
				Water						
4i	Н	NHCH ₂ CH ₃	S(CH ₂) ₃ CH ₃	Diethyl Ether	13	4363	227, 378	369	535	308, 147
4j	Н	NHCH ₂ CH ₃	SCH ₂ Ph	Diethyl Ether	17	2276	224, 370	367	532	308, 152

^a Relative fluorescent quantum yield measured using quinine sulfate as reference ($\Phi_f = 59\%$, in 0.105 M HClO₄).⁶

 b The absorption, excitation and emission wavelength were recorded at 10 μM in diethyl ether, slit width ex 5 nm, em 5 nm.

 c Stokes shift in diethyl ether solution (10 $\mu M).$

4. UV-VIS SPECTRUM OF MALEIMIDES



Figure S1. a) Excitation, emission and absorption spectra of **2a** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **2a** in diethyl ether at 10 μ M. Peaks at $\lambda_{ex} = \lambda_{em}$ are due to Rayleigh scattering from the sample solution.



Figure S2. a) Excitation, emission and absorption spectra of **2b** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **2b** in diethyl ether at 10 μ M. Peaks at $\lambda_{ex} = \lambda_{em}$ are due to Rayleigh scattering from the sample solution.



Figure S3. a) Excitation, emission and absorption spectra of **2c** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **2c** in diethyl ether at 10 μ M. Peaks at $\lambda_{ex} = \lambda_{em}$ are due to Rayleigh scattering from the sample solution.



Figure S4. a) Excitation, emission and absorption spectra of **2e** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **2e** in diethyl ether at 10 μ M. Peaks at λ ex = λ em are due to Rayleigh scattering from the sample solution.



Figure S5. a) Excitation, emission and absorption spectra of **3a** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **3a** in diethyl ether at 10 μ M. Peaks at $\lambda ex = \lambda em$ are due to Rayleigh scattering from the sample solution.



Figure S6. a) Excitation, emission and absorption spectra of **4h** in diethyl ether at 10 μ M; b) 2D excitation-emission spectra (with a 5 nm step) of **4h** in diethyl ether at 10 μ M. Peaks at $\lambda ex = \lambda em$ are due to Rayleigh scattering from the sample solution.

5. SOLVATOCHROMIC DETERMINATION

The emission spectra of each compound in different solvents was recorded using the corresponding absorption peak as the excitation wavelength. (CH= cyclohexane, DE= diethyl ether, THF=tetrahydrofutan, MeOH=Methanol, DMF= Dimethylformamide,)



Figure S7. Absorption (a) and emission (b) spectra of 2a in different solvents at 10 μ M.



Figure S8. Absorption (a) and emission (b) spectra of 2b in different solvents at 10 μ M.



Figure S9. Absorption (a) and emission (b) spectra of 2c in different solvents at 10 μ M.



Figure S10. Absorption (a) and emission (b) spectra of 3a in different solvents at 10 μ M.



Figure S11. Absorption (a) and emission (b) spectra of **3b** in different solvents at 10 μ M.



Figure S12. Absorption (a) and emission (b) spectra of **4h** in different solvents at 10 μ M.

6. PHOTOGRAPHS IN DIFFERENT SOLVENTS UNDER UV LIGHT



Figure S13. Photographs of ABM (2e) and ATM (4h) in different solvents under the 365 nm hand-held UV lamp. 1-8: water, methanol, DMF, dioxane, THF, diethyl ether, CH_2Cl_2 and cyclohexane.



Figure S14. Photographs of OBM (**3b**) and OCM (**3a**) in different solvents under the 365 nm hand-held UV lamp. 1-8: water, methanol, DMF, dioxane, THF, diethyl ether, CH_2Cl_2 and cyclohexane.





Figure S15. Fluorescent lifetimes of ATM (**4h**) and ABM (**2e**) in solid and solution states (10 μ M in dioxane).



Figure S16. a) Fluorescence spectra of thiophenol substituted aminomaleimide (**4h**) in different ratios of hexane/CH₂Cl₂ solution (0-100% hexane); b) Intensity and wavelength changes in different ratios of hexane/CH₂Cl₂ solutions; (insert images are **4h** under UV light in the solutions); (20 μ M, λ_{ex} 358 nm, slit width ex:5 nm, em:5 nm); c) CIE diagram of the fluorescent wavelength shift for **4h** upon changing the solvent quality. d) Fluorescence spectra of aminobromomaleimide (**2e**) in different ratios of hexane/CH₂Cl₂ solution (0-100% hexane); e) Intensity and wavelength changes in different ratios of hexane/CH₂Cl₂ solutions; (insert images are **2e** under UV light in the solutions); (20 μ M, λ_{ex} 362 nm, slit width ex:5 nm, em:5 nm); f) CIE diagram of fluorescent wavelength shift for **2e** upon changing the solvent quality.

8. NMR SPETRA OF SUBSTITUTED MALEIMIDES 3-chloro-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (2a):



3-bromo-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (2b):



3- iodo-4-(ethylamino)- 1-methyl-1H-pyrrole-2,5-dione (2c):



3-(butylamino)-4-chloro-1-methyl-1H-pyrrole-2,5-dione (2d):



3-bromo-4-(butylamino)-1H-pyrrole-2,5-dione (2e)



3-chloro-4-ethoxy-1-methyl-1H-pyrrole-2,5-dione (3a)



S28

3-bromo-4-ethoxy-1-methyl-1H-pyrrole-2, 5-dione (3b)



3-(benzyloxy)-4-chloro-1-methyl-1H-pyrrole-2,5-dione (3c):



3-(benzyloxy)-4-bromo-1-methyl-1H-pyrrole-2,5-dione (3d)



3-chloro-1-methyl-4-phenoxy-1H-pyrrole-2,5-dione (3e):



S32

3-bromo-1-methyl-4-phenoxy-1H-pyrrole-2,5-dione (3f):



3-(butylthio)-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione(4a)



3-(ethylamino)-1-methyl-4-(octadecylthio)-1H-pyrrole-2,5-dione (4b)





3-(ethylamino)-1-methyl-4-(phenylthio)-1H-pyrrole-2,5-dione(4c)



3-(ethylamino)-1-methyl-4-(o-tolylthio)-1H-pyrrole-2,5-dione(4d)

3-(benzylthio)-4-(ethylamino)-1-methyl-1H-pyrrole-2,5-dione (4e)



3-(butylamino)-1-methyl-4-(phenylthio)-1H-pyrrole-2,5-dione (4f):





1-methyl-3-(phenylamino)-4-(phenylthio)-1H-pyrrole-2,5-dione (4g)

3-(butylamino)-4-(phenylthio)-1H-pyrrole-2,5-dione (4h)



S41

3-(butylthio)-4-(ethylamino)-1H-pyrrole-2,5-dione (4i)



3-(benzylamino)-4-(butylthio)-1H-pyrrole-2,5-dione (4j)



9. COMPUTATIONAL DETAILS

The geometries for all studied systems in the ground state have been fully optimized with the CAM-B3LYP⁷ functional and the 6-311G(d,p) basis set.^{8,9} The dispersion effects and the solvent were included in the optimization process. The D3-Grimme's dispersion¹⁰ with Becke-Johson damping factor^{11,12} was used to evaluate the dispersion effects. The solvent was considered using the polarization continuum model (PCM)^{13, 14} and the dielectric constant of diethyl ether ($\varepsilon = 4.24$). The harmonic vibrational frequencies were also calculated at the same level of theory in order to verify that all the stationary points are minima of their potential energy surface. The time-dependent density functional theory (TD-DFT)¹⁵⁻²³ was applied to compute the absorption and emission (fluorescence) spectra at the ground and the first singlet excited state, respectively. The charge distribution was obtained following the natural bond orbital (NBO) partition scheme by Weinhold and co-workers.²⁴ All the calculations were performed using the Gaussian09 program package.²⁵

Compound	λ_{exc} ^c	f exc ^d	λ _{em} ^e	f em ^d	$\Delta\lambda^{\rm f}$
1 a	302.4	0.023	468.4	0.011	166.0
1b	309.2	0.040	465.6	0.022	156.4
1c	323.2	0.078	463.7	0.058	140.5
2a	338.6	0.087	460.2	0.070	121.6
2b	339.0	0.087	458.9	0.073	119.9
2c	339.9	0.088	458.5	0.078	118.6
2d	330.3	0.121	435.3	0.104	104.9
2e	331.3	0.119	435.4	0.105	104.2
3 a	307.3	0.035	442.7	0.028	135.4
3 b	309.0	0.038	442.1	0.032	133.1
3c	308.0	0.031	449.7	0.022	141.7
3d	310.2	0.035	449.8	0.026	139.5
3e	306.3	0.033	445.5	0.024	139.2
3 f	308.9	0.036	459.8	0.035	151
4 a	334.0	0.090	551.5	0.084	217.5
4 b	333.8	0.090	552.9	0.085	219.0
4 c	328.3	0.067	510.4	0.065	182.1
4d	328.1	0.068	511.5	0.065	183.4
4e	336.2	0.095	550.6	0.089	214.5
4 f	336.8	0.064	510.0	0.065	173.2
4 g	373.5	0.088	508.2	0.098	134.7
4h	316.7	0.087	491.4	0.084	174.8
4i	327.2	0.115	528.8	0.106	201.5
4 j	326.3	0.104	537.6	0.105	211.3

Table S2. TD-DFT excitation and emission energies and oscillator strengths of the maleimide derivatives studied in this work.^{a,b}

^a All excitation and emission energies correspond to the first excited state ($\pi \rightarrow \pi^*$ transition). ^b All values computed at CAM-B3LYP-D3BJ/6-311G(d,p) level of theory in solution using the PCM with $\varepsilon = 4.24$. ^c Excitation wavelength, in nm. ^d Oscillator strength. ^e Emission wavelength, in nm. ^f Stokes shift (λ_{em} - λ_{exc}).

Compound	q _{C2}	q c3	Δq
1a	-0.13	-0.13	0.00
1b	-0.20	-0.20	0.00
1c	-0.32	-0.32	0.00
2a	-0.24	0.15	0.39
2b	-0.32	0.15	0.47
2c	-0.44	0.15	0.59
2d	-0.25	0.14	0.39
2e	-0.32	0.14	0.47
3 a	-0.23	0.29	0.52
3b	-0.31	0.29	0.6
3c	-0.22	0.28	0.50
3d	-0.29	0.28	0.57
3e	-0.21	0.28	0.49
3 f	-0.29	0.28	0.57
4 a	-0.40	0.18	0.58
4 b	-0.40	0.18	0.58
4 c	-0.42	0.19	0.61
4d	-0.42	0.19	0.61
4 e	-0.40	0.18	0.59
4f	-0.42	0.19	0.61
4 g	-0.38	0.16	0.53
4h	-0.42	0.19	0.61
4 i	-0.40	0.17	0.57
4j	-0.42	0.19	0.60

Table S3. Natural population analysis of the Carbon atoms adjacent to the electron withdrawing and donating groups (labeled as C_2 and C_3 , respectively).^a

^a All values computed at CAM-B3LYP-D3BJ/6-311G(d,p) level of theory in solution using the PCM with $\varepsilon = 4.24$.



Figure S17. HOMO and LUMO Kohn-Sham orbitals of **1a**, **1b**, **1c** and **2a** maleimide derivatives.



 2b HOMO
 2b LUMO
 2c HOMO
 2c LUMO

 Image: Comparison of the state o

Figure S18. HOMO and LUMO Kohn-Sham orbitals of **2b**, **2c**, **2d** and **2e** maleimide derivatives.



Figure S19. HOMO and LUMO Kohn-Sham orbitals of **3a**, **3b**, **3c** and **3d** maleimide derivatives.



Figure S20. HOMO and LUMO Kohn-Sham orbitals of **3e**, **3f**, **4a**, and **4b** maleimide derivatives.



Figure S21. HOMO and LUMO Kohn-Sham orbitals of 4c, 4d, 4e and 4f maleimide derivatives.



Figure S22. HOMO and LUMO Kohn-Sham orbitals of 4g, 4h, 4i and 4j maleimide derivatives.

Table S4. Cartesian coordinates (in Å) of the stationary points in the ground state optimized at the CAM-B3LYP-D3BJ/6-311G(d,p) level of theory in solution using the PCM with $\varepsilon = 4.24$.

С	-0.569939	0.677136	-0.000135
С	-0.586832	-0.650148	0.002315
С	0.836264	-1.146833	0.001007
Ν	1.631825	-0.014532	-0.001932
0	1.265302	2.267340	-0.004657
0	1.214277	-2.282938	0.002625
С	3.082796	-0.039832	-0.001653
Η	3.452921	-0.529994	0.898060
Η	3.450648	-0.568592	-0.879992
Η	3.427360	0.991100	-0.024105
Cl	-1.861360	1.776889	-0.000987
Cl	-1.905628	-1.716690	0.005352
1b			
С	-1.425260	-1.090979	0.114405
С	-0.069890	-0.456563	-0.079915
С	-0.231534	0.856101	-0.203217
С	-1.704211	1.168563	-0.098332
Ν	-2.336708	-0.047679	0.089115
0	-1.673575	-2.253254	0.260474
0	-2.233835	2.240925	-0.161487
С	-3.771339	-0.203314	0.241265
Η	-4.120668	0.345845	1.114815
Η	-4.287323	0.161564	-0.646027
Η	-3.972711	-1.263790	0.371304
Br	1.474886	-1.476326	-0.117349
Br	1.023108	2.192326	-0.461961
1c			
С	-1.346667	-1.340713	0.135454
С	-0.256196	-0.312673	-0.061393
С	-0.824711	0.881465	-0.204375
С	-2.323235	0.709406	-0.110805
Ν	-2.540771	-0.641430	0.090799
0	-1.219077	-2.521045	0.296806
0	-3.167947	1.555468	-0.191632
С	-3.854055	-1.240512	0.236768
Η	-4.370715	-0.812824	1.095157
Н	-4.447251	-1.074885	-0.661834
Н	-3.711425	-2.307473	0.389530
Ι	1.731796	-0.871452	-0.074456
Ι	0.008964	2.745927	-0.507641
2a			
С	1.476204	-0.119859	-0.842751
С	2.851062	0.283966	-1.150410
С	3.251094	1.222803	-0.262051
С	2.086949	1.455194	0.686551
N	1 076971	0.638168	0 283784

O 0.773511 -0.925263 -1.393003

1a

0	2.096949	2.225381	1.614741
С	-0.220254	0.539015	0.918006
Η	-0.116698	0.219920	1.955216
Η	-0.736506	1.498574	0.886608
Н	-0.794244	-0.200975	0.365200
Ν	4.342220	1.948013	-0.043250
Н	4 293207	2 534341	0 779516
C	5 620905	1 810330	-0 727777
ч	6 157647	2 747047	0.583801
C	6 4 4 2 2 0 5	2.141941	-0.383891
с ц	0.44 <i>3293</i> 5.017806	0.030477	-0.212929
п	3.917890	-0.500008	-0.374416
п	1.39/998	0.388203	-0.739900
н	6.644130	0.742228	0.854720
H	5.433594	1./09206	-1./96206
CI	3.695361	-0.421379	-2.478955
2b			
С	1.457789	-0.113997	-0.851235
С	2.835552	0.276277	-1.165084
С	3.247637	1.206088	-0.270407
С	2.089608	1.443871	0.687368
Ν	1.071550	0.639400	0.284636
0	0.739171	-0.905930	-1.400741
0	2.112819	2.208693	1.620030
С	-0.223704	0.548011	0.924008
Н	-0.117931	0.223520	1.959324
Н	-0.732366	1 511821	0 899069
н	-0.805920	-0.185039	0.370639
Br	3 726674	-0.500921	-2 622747
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4c C C C C N O O C H H H N H C	-1.642071 -0.295102 -0.374019 -1.814840 -2.504086 -2.025718 -2.225264 -3.915880 -4.135486 -4.507636 -4.164719 0.456752 0.036830 1.892835	-1.115150 -0.719095 0.556030 1.021472 -0.011230 -2.151753 2.109402 0.006043 0.783285 0.182188 -0.967447 1.419791 2.313860 1.259120	0.689053 0.252710 -0.229137 -0.095977 0.449044 1.163730 -0.422815 0.766148 1.498531 -0.132245 1.181964 -0.785472 -1.008615 -0.977296
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4 C C C C C N O O C H H H N H C H C	-1.642071 -0.295102 -0.374019 -1.814840 -2.504086 -2.025718 -2.225264 -3.915880 -4.135486 -4.507636 -4.164719 0.456752 0.036830 1.892835 2.172265 2.694204	-1.115150 -0.719095 0.556030 1.021472 -0.011230 -2.151753 2.109402 0.006043 0.783285 0.182188 -0.967447 1.419791 2.313860 1.259120 1.905493 1.609196	0.689053 0.252710 -0.229137 -0.095977 0.449044 1.163730 -0.422815 0.766148 1.498531 -0.132245 1.181964 -0.785472 -1.008615 -0.977296 -1.809339 0.267947
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C C C N	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407	$\begin{array}{c} 1.460135\\ 0.798054\\ 0.052234\\ 0.334084 \end{array}$
C C C N O	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174	$\begin{array}{c} 1.460135\\ 0.798054\\ 0.052234\\ 0.334084\\ 1.559002 \end{array}$
C C C N O O	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434
C C C N O O N	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924	$\begin{array}{c} 1.460135\\ 0.798054\\ 0.052234\\ 0.334084\\ 1.559002\\ -0.639434\\ 0.659502 \end{array}$
C C C N O O N H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657
C C C C N O O N H S	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693
C C C C N O O N H S H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277
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C C C N O O N H S H C C H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994
C C C N O O N H S H C C H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038
C C C N O O N H S H C C H H C	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604
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C C C N O O N H S H C C H H C H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253	-2.618948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565
C C C N O O N H S H C C H H C H H C	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490
C C C N O O N H S H C C H H C H H C H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067
C C C N O O N H S H C C H H C H H C H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579
C C C C N O O N H S H C C H H C H H C H H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219 1.527587	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147 4.563694	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579 0.478835
C C C C N O O N H S H C C H H C H H C H H H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219 1.527587 1.549300	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147 4.563694 2.92455	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579 0.478835 -0.176305
C C C C N O O N H S H C C H H C H H C H H H H H	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219 1.527587 1.549300 0.574519	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147 4.563694 2.924555 4.176000	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579 0.478835 -0.176305 -0.955758
C C C C N O O N H S H C C H H C H H C H H H H H C	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219 1.527587 1.549300 0.574519 0.986055	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147 4.563694 2.924555 4.176000 -0.281683	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579 0.478835 -0.176305 -0.955758 1.506458
C C C C N O O N H S H C C H H C H H C H H H H H H C C	-1.825504 -1.652685 -2.639511 -3.493218 -2.951782 -1.177824 -4.441915 -3.008683 -3.796609 -0.424312 -3.295404 -2.299713 -1.139479 -1.935234 -3.028314 -0.269807 -0.523453 -1.531253 0.912682 -0.876775 0.098219 1.527587 1.549300 0.574519 0.986055 2.004676	-2.018948 -1.184862 -0.508100 -1.524210 -2.737407 -3.560174 -1.252773 0.752924 0.890541 -0.597072 -3.618004 1.921885 2.332690 1.696139 2.728786 3.412582 1.454532 2.677446 3.792511 4.299155 3.049147 4.563694 2.924555 4.176000 -0.281683 0 501601	1.460135 0.798054 0.052234 0.334084 1.559002 -0.639434 0.659502 0.038657 2.558693 -0.018277 1.163248 0.261869 2.163994 1.251038 0.893604 0.056122 -0.700565 0.010490 1.106067 1.858579 0.478835 -0.176305 -0.955758 1.506458 2.042423

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II U	2.055590	-3.337873	-3.469307
II U	J.084038	-3.800007	-3.043901
11 Ч	4.031799	-4.787341	-2.339877
и Н	-3 328884	-0.007940	0.178246
11	-5.52000+	-0.0+2213	0.170240
4j			
Č	-1.056031	-2.800211	0.948325
С	-0.766376	-1.363568	1.049129
С	-1.872781	-0.683903	0.622595
С	-2.941774	-1.701525	0.251050
Ν	-2.381378	-2.920567	0.464042
0	-0.353045	-3.747272	1.189623
0	-4.044043	-1.430361	-0.154883
Ν	-2.221474	0.578953	0.451098
Н	-3.172925	0.707629	0.126775
S	0.782091	-0.782151	1.619858
Н	-2.841547	-3.803728	0.303874
C	-1.418262	1.767668	0.690809
C	-1.719412	2.834569	-0.348227
Н	-0.370266	1.482607	0.648259
Н	-1.615065	2.145124	1.700093
C	-0.890562	4.095097	-0.131968

Η	-1.512523	2.423214	-1.339484
Η	-2.785993	3.086874	-0.319389
С	-1.128020	5.141161	-1.214049
Η	-1.120379	4.519470	0.850752
Η	0.169449	3.824144	-0.111579
Η	-0.528119	6.036764	-1.041494
Η	-0.866311	4.747391	-2.199608
Η	-2.177910	5.443639	-1.246081
С	1.815730	-1.000872	0.107702
Η	2.821175	-0.728658	0.429401
Η	1.811377	-2.060995	-0.140010
С	1.360642	-0.159264	-1.042545
С	1.798993	1.156473	-1.182143
С	0.456825	-0.665955	-1.973138
С	1.346181	1.946556	-2.227530
Η	2.501589	1.562982	-0.462877
С	-0.003915	0.125033	-3.016803
Η	0.117715	-1.691267	-1.883102
С	0.439689	1.433284	-3.147046
Η	1.700621	2.965558	-2.326184
Η	-0.707148	-0.283964	-3.732195
Η	0.084477	2.050529	-3.963284

Table S5. Cartesian coordinates (in Å) of the stationary points in the first excited state optimized at the CAM-B3LYP-D3BJ/6-311G(d,p) level of theory in solution using the PCM with $\varepsilon = 4.24$.

1a

С	0.829009	1 213256	-0.002687
C	0.029009	0.726026	0.002007
C	-0.499578	0.720020	0.000470
С	-0.510834	-0.698857	0.001399
С	0.810344	-1.205222	-0.000611
Ν	1.608053	-0.003504	-0.003239
0	1 327860	2 322275	-0.004021
ŏ	1 205070	2 320448	0.000868
C	1.293979	-2.520440	-0.000808
C	3.028490	-0.024280	-0.002670
Н	3.380997	-0.552303	0.890557
Η	3.380674	-0.601743	-0.864034
Η	3.394210	0.997859	-0.028043
C1	-1 857944	1 725027	0.002515
	1 88/015	1.676104	0.002313
CI	-1.004915	-1.070104	0.005571
1b			
С	-1.381911	-1.154479	0.117363
С	-0.129822	-0.513612	-0.068133
С	-0 309262	0 894625	-0.200135
č	-1 685825	1 22/310	-0.105358
N	2 210215	0.054095	0.105550
N	-2.510515	-0.034983	0.088270
0	-1.723/10	-2.312326	0.271560
0	-2.313386	2.266220	-0.156927
С	-3.717363	-0.210252	0.236802
Н	-4.057805	0.378524	1.094932
н	-4 221529	0 197637	-0 645413
ц	3 030304	1 265250	0.367550
п	-3.939394	-1.203230	0.307330
Br	1.459612	-1.431646	-0.120765
Br	1.001647	2.154653	-0.456656
1c			
С	-1.293653	-1.377218	0.137900
Ĉ	-0 288438	-0 382091	-0.050023
c	0.002524	0.807702	0.000020
C	-0.902334	0.897793	-0.205192
C	-2.320216	0.759872	-0.11/312
Ν	-2.510883	-0.635595	0.089440
0	-1.243471	-2.583706	0.306757
0	-3.239395	1.558216	-0.188181
С	-3.803436	-1.228167	0.232977
н	-4 324287	-0 768560	1 077694
ц	4 208447	1.020214	0.662050
п	-4.396447	-1.029314	-0.003030
Н	-3.6/5614	-2.295892	0.386557
I	1.693608	-0.852586	-0.075769
Ι	-0.014524	2.706507	-0.501419
2a			
Ĉ	1 456802	-0 155909	-0 850718
č	2 817009	0.2/085/	_1 112852
c	2.017070	1 240001	0.162102
C	3.220943	1.249801	-0.102193
C	2.10/9/3	1.4/869/	0./10/02
Ν	1.093319	0.617000	0.250893
0	0.744868	-0.976911	-1.426405

O C H H H N H C H C H H H H C I	$\begin{array}{c} 2.025941\\ -0.202439\\ -0.110275\\ -0.634183\\ -0.825780\\ 4.349722\\ 4.324202\\ 5.598732\\ 6.125105\\ 6.462666\\ 5.948803\\ 7.391389\\ 6.711208\\ 5.374484\\ 3.710714 \end{array}$	2.257151 0.543207 0.142890 1.542720 -0.109564 1.925449 2.557111 1.794004 2.741329 0.643081 -0.312406 0.601105 0.780956 1.671917 -0.415136	$\begin{array}{c} 1.668702\\ 0.871247\\ 1.883428\\ 0.938015\\ 0.265969\\ -0.002163\\ 0.791476\\ -0.736565\\ -0.623861\\ -0.230460\\ -0.342888\\ -0.802296\\ 0.822664\\ -1.795873\\ -2.378085\end{array}$
2b	1 40 47 45	0 105550	0.067000
C	1.434745	-0.137752	-0.867999
C	2.799422	0.254799	-1.137448
C	3.219666	1.236028	-0.172494
C	2.115528	1.460686	0.713629
N	1.088536	0.618232	0.250390
0	0.706096	-0.939940	-1.449098
C	2.051624	2.225180	1.084524
C H	-0.202391	0.546692	0.881625
н	-0.089095	0.245344	1.924007
H	-0.681030	1.52/589	0.860189
H Da	-0.799185	-0.180035	0.337010
DI	3.732114	-0.477903	-2.338007
	4.349383	1.909112	-0.002230
н С	4.319830	2.525499	0.803533
С П	5.002945	1.795175	-0.729239
п	6.1100/3	2.747083	-0.010000
С Ц	0.4/9139	0.033003	-0.213008
п ц	J.9/0/10 7 410700	-0.307293	-0.329410
п ц	6717926	0.024727	-0.779743
п п	5 3 8 5 8 3 7	1 66/352	1 780177
11	5.565657	1.004332	-1.709177
2c			
С	1.429934	-0.140736	-0.872836
С	2.797253	0.243041	-1.156680
С	3.221135	1.222753	-0.188423
С	2.120691	1.452552	0.701399
N	1.088309	0.615560	0.244452
0	0.695093	-0.942165	-1.447963
U C	2.063938	2.220772	1.669966
C	-0.200053	0.549139	0.882100
H	-0.089462	0.202198	1.911435
H	-0.654483	1.541162	0.903274
H	-0.816659	-0.142328	0.314113
IN	4.348676	1.901064	-0.010905

Η	4.308515	2.515740	0.795876
С	5.608608	1.795052	-0.724983
Η	6.111882	2.755510	-0.614231
С	6.494469	0.670060	-0.197940
Η	6.001045	-0.297467	-0.297972
Η	7.427194	0.638599	-0.764226
Η	6.733903	0.828803	0.854385
Η	5.403484	1.658305	-1.786519
Ι	3.774641	-0.598123	-2.731723
2d			
C	2.491447	1.082513	0.158517
Č	1.141999	0.746916	-0.265850
C	1.003257	-0.681918	-0.343445
C	2.265507	-1.260293	0.036984
N	3.086902	-0.159363	0.317056
0	3.006288	2.181016	0.341727
0	2.559695	-2.460254	0.102384
Ν	-0.009289	-1.459301	-0.692000
Η	0.211248	-2.448583	-0.628016
С	-1.363548	-1.070993	-1.044192
Η	-1.785512	-1.890161	-1.626630
Η	-1.325374	-0.194388	-1.692579
С	-2.238008	-0.789265	0.178852
С	-3.635181	-0.330062	-0.222010
Η	-1.760545	-0.022042	0.794767
Η	-2.297982	-1.694737	0.789493
С	-4.521558	-0.044892	0.984636
Η	-3.553940	0.570535	-0.839474
Η	-4.102796	-1.094552	-0.850914
Η	-4.643136	-0.937895	1.602567
Η	-4.088284	0.737680	1.612365
Cl	-0.005209	1.933341	-0.590351
Η	4.048459	-0.259594	0.615724
Н	-5.514983	0.285961	0.676439
2e			
C	2.518821	0.759623	0.342597
Č	1.167388	0.487702	-0.123110
C	1.021427	-0.922940	-0.365139
C	2.280117	-1.548916	-0.052475
Ν	3.107764	-0.495166	0.356280
0	3.041980	1.823857	0.657219
0	2.566099	-2.750355	-0.128988
Br	-0.049789	1.840627	-0.322898
Ν	0.012525	-1.659319	-0.805261
Η	0.241204	-2.647636	-0.855223
С	-1.345886	-1.246810	-1.105582
Η	-1.760366	-1.992413	-1.784426
Η	-1.318922	-0.295941	-1.640128
С	-2.220201	-1.127419	0.144034
С	-3.619631	-0.625681	-0.192248
Η	-1.744131	-0.443767	0.852473
Η	-2.276176	-2.103690	0.634169
С	-4.505137	-0.503379	1.042210
H	-3.542473	0.347704	-0.687758
H	-4.085365	-1.303689	-0.914769
H	-4.620797	-1.469480	1.539637
H	-4.074456	0.193470	1.765553
Н	4.067148	-0.635571	0.645028

Н	-5.500976	-0.140724	0.781107
2.			
sa C	1 311612	0 992948	0 117095
C	-0.074331	0.755941	-0.124005
č	-0.314533	-0.656470	-0.300193
С	0.934439	-1.340117	-0.183817
Ν	1.872826	-0.299231	0.070923
0	1.945408	2.024325	0.315546
0	1.229169	-2.527972	-0.257281
С	3.262537	-0.553823	0.294511
Н	3.432825	-0.784179	1.352440
H	3.559388	-1.419472	-0.295341
H	3.824366	0.341506	0.03/542
С и	-2./3399/	-0.804047	-0.44/948
п С	-3.342981	-1.339130	-0.95/001
н	-2 520375	0.116265	1 504623
Н	-4.173433	-0.296226	1.043841
Н	-3.051750	-1.576008	1.533700
Н	-2.797630	0.121204	-1.017464
Cl	-1.165143	2.040688	-0.229139
0	-1.393832	-1.343959	-0.557675
3b	1 257240	1 022595	0 167211
C	-1.237240	-1.052383	0.10/211
C	-0 174603	0.996903	-0.320891
C	-1.570737	1.291616	-0.226261
N	-2.168793	0.036466	0.073869
0	-1.568461	-2.194980	0.406431
0	-2.194114	2.341128	-0.342885
С	-3.574869	-0.115169	0.293523
Η	-3.810553	0.072142	1.347204
Η	-4.104480	0.618032	-0.312263
Н	-3.854702	-1.138195	0.051199
Br	1.525242	-1.449286	-0.178556
С и	2.089003	1.8/1/10	-0.443035
п	2.433183	2.730302	-0.938323
н	2.472190	1.013429	1.548601
Н	3.560344	1.896854	1.103445
Н	2.082175	2.786786	1.502894
Н	2.455215	0.987513	-0.961389
0	0.653671	1.964789	-0.608377
30			
эс С	1 015268	0 991457	0 223531
C	-0.296936	0.769896	-0.282724
č	-0.521016	-0.635254	-0.488164
Č	0.661308	-1.343365	-0.122303
Ν	1.554911	-0.315717	0.299368
0	1.625223	2.010887	0.522694
0	0.925475	-2.539263	-0.085024
С	2.854511	-0.591713	0.825877
H	2.787120	-0.778580	1.904173
H	3.244543	-1.487974	0.346483
H	3.484832	0.280651	0.665270
U U	-2.913548	-0.81/146	-0.6/21/8
п	-3.330499	-1.02/3/9	-1.0338/6

Η	-3.111659	0.079303	-1.257911
Cl	-1.319201	2.051151	-0.682105
0	-1.567274	-1.288314	-0.933091
Ċ	-3.143376	-0.586935	0.795859
Ċ	-3.924398	0.480807	1.219749
Ĉ	-2.590172	-1.450810	1.738042
Ĉ	-4 160200	0.681250	2 572995
н	-4 345016	1 164097	0.490951
$\hat{\mathbf{C}}$	2 815888	1 2/1805	3 080618
н	-1.97/527	-2 281117	1 / 12233
C	3 603/80	0 177456	3 500471
с u	4 768200	1 517404	2 805141
и П	-4.708233	1.01/1/494	2.893141
11 11	-2.378028	-1.914145	J.01/12/
п	-3.776366	-0.013703	4.303939
3d			
С	-0.922253	-0.960612	0.230426
С	0.255168	-0.349709	-0.293151
С	0.030185	1.055267	-0.499287
Ċ	-1.309824	1.364222	-0.116735
N	-1.837374	0.115018	0.315655
0	-1.188540	-2.117506	0.533160
0	-1.924989	2.423128	-0.070471
Č	-3.153029	-0.021334	0.858952
Н	-3.131485	0.161131	1.939634
Н	-3 801445	0 720951	0 397016
Н	-3.493713	-1.040652	0.688129
Br	1 713018	-1 361438	-0 768344
C	2 234473	2 009294	-0.668869
н	2.556853	2.009291	-1.066377
н	2.556055	1 216278	-1 229858
0	0.813079	2.005085	-0.952909
c	2 502867	1 806305	0.806240
c	3 591198	1.62981	1 261749
c	1 67/375	2 530044	1.201742
C	3 857482	1 077822	2 621553
с u	1 220620	0.640738	0.552733
II C	4.230020	0.049738	0.332733
с u	0.910906	2.442301	1 272550
пС	2.027615	1 714280	2 522028
с u	3.027013	0.502272	2.069599
н ц	4.700793	0.302372	2.700300
п u	1.202003	2.940219	5.700992 4.504514
11	5.220951	1.039093	4.394314
3e		o 5	
С	-2.151285	-0.777447	0.723442
С	-0.858597	-0.516399	0.185973
С	-0.808284	0.823748	-0.335006
С	-2.074845	1.443944	-0.142929
Ν	-2.846305	0.432648	0.501044
0	-2.634594	-1.759434	1.259596
0	-2.496458	2.552867	-0.419089
Cl	0.357896	-1.681773	0.214300
0	0.159790	1.500437	-0.910787
С	-4.201732	0.635949	0.905583
Н	-4.720859	-0.320308	0.882252
Η	-4.227227	1.017956	1.933151
Η	-4.654072	1.378010	0.250659
С	1.429810	0.943428	-1.121830

С	1.677377	0.296130	-2.316166
С	2.398124	1.119666	-0.153417
С	2.946428	-0.220173	-2.533404
Η	0.889191	0.197703	-3.051208
С	3.662307	0.597473	-0.385063
Н	2.159317	1.648984	0.759798
С	3.935008	-0.075467	-1.568822
Η	3.161374	-0.736428	-3.460332
Η	4.435320	0.718574	0.363067
Η	4.923305	-0.481192	-1.743728
3f			
C	-2.270792	0.997230	-0.448480
С	-0.909769	0.683685	-0.748471
С	-0.649007	-0.681349	-0.404154
С	-1.842678	-1.281418	0.096239
Ν	-2.784941	-0.216148	0.067828
0	-2.930674	2.018047	-0.589330
0	-2.088923	-2.406946	0.507646
0	0.461250	-1.382988	-0.575062
С	-4.120888	-0.350471	0.563964
Н	-4.754961	0.364829	0.043593
Н	-4.447132	-1.378692	0.416950
Н	-4.150848	-0.126019	1.636105
С	1.685401	-0.816249	-0.265491
С	2.761878	-1.186321	-1.056871
С	1.825319	0.040168	0.817788
С	4.006244	-0.652106	-0.773333
Н	2.603895	-1.866836	-1.882801
С	3.078067	0.566278	1.086332
Н	0.972739	0.274938	1.441983
С	4.166137	0.229261	0.291443
Н	4.855794	-0.921054	-1.387880
Н	3.205239	1.235356	1.927668
Н	5.142447	0.643405	0.508095
Br	0.168948	1.830840	-1.698796
49			
ча С	-1.165162	-1.035499	-0.241640
Č	0.101284	-0.334239	-0.352253
Č	-0.160440	1.071515	-0.211270
Č	-1.575294	1.249673	-0.050575
Ň	-2.117834	-0.034197	-0.092109
0	-1.412049	-2.250102	-0.232154
0	-2.197260	2.300212	0.078151
C	-3.522742	-0.297928	0.089430
Н	-3.841375	0.007616	1.087990
Н	-4.100969	0.266271	-0.643261
Н	-3.678991	-1.365937	-0.038335
N	0.624223	2.142693	-0.298221
Н	0.111287	3.014073	-0.226411
C	2.073089	2.187412	-0.254957
Ĥ	2.367291	3.179024	-0.596901
C	2.641561	1.914940	1.134943
Ĥ	2.353824	0.925730	1.494324
Н	3.731764	1.960286	1.105294
Н	2.282088	2.656897	1.849339
Н	2.468443	1.464353	-0.973629
S	1.630302	-1.059495	-0.571787
Ĉ	1.280422	-2.553000	-1.562718

С	2.519735	-3.431777	-1.606961
Η	0.996319	-2.206212	-2.557362
Η	0.425015	-3.049829	-1.110074
С	2.290243	-4.674060	-2.464027
Η	3.369201	-2.867750	-2.006468
Η	2.790236	-3.734816	-0.591008
С	3.516853	-5.576678	-2.520685
Η	1.438150	-5.232428	-2.064510
Η	2.010841	-4.366932	-3.476886
Η	3.331569	-6.457855	-3.137634
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н 4c С С	-1.849411 -0.590306 -0.576240	-1.031692 -0.938442 0.309728	1.006387 0.291124 -0.432432
н 4c С С С С	-1.849411 -0.590306 -0.576240 -1.850351	-1.031692 -0.938442 0.309728 0.944082	1.006387 0.291124 -0.432432
H 4c C C C C C N	-1.849411 -0.590306 -0.576240 -1.850351 2.561278	-1.031692 -0.938442 0.309728 0.944082 0.103130	1.006387 0.291124 -0.432432 -0.222306 0.636755
H 4c C C C C C C N O	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269
4c C C C C C C N O O	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269
4c C C C C C N O O C	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 3.902501	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525
H 4c C C C C C C C N O O C H	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 3.944991	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1 378322	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1 523775
4cCCCCCCNOOCHH	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 4.599309	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645
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H 4c C C C C C N O O C H H H N H	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 0.008347	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 1.388258
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H 4c C C C C C N O O C H H H N H C H	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 2.014652
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H 4CCCCCNOOCHHHNHCHCH	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131380	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0 738328
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H 4CCCCNOOCHHHNHCHCHHHH	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137 3.625323 2.461794	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131389 0.689469 1.860772 -0.231147	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0.738328 -0.213134 0.431590 -1.744845
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H 4CCCCNOOCHHHNHCHCHHHHSC	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137 3.625323 2.461794 1.911140 0.541263 1 199721	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131389 0.689469 1.860772 -0.231147 -2.197365 -2.452320	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0.738328 -0.213134 0.431590 -1.744845 0.486041 -1.138647
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H 4CCCCCNOOCHHHNHCHCHHHHSCCCC	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137 3.625323 2.461794 1.911140 0.541263 1.199721 2.537742 0.397199 3.074440	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131389 0.689469 1.860772 -0.231147 -2.197365 -2.452339 -2.824471 -2.331480 3.060597	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0.738328 -0.213134 0.431590 -1.744845 0.486041 -1.138647 -1.252621 -2.271851 2.507750
H 4CCCCCNOOCHHHNHCHCHHHHSCCCCC	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137 3.625323 2.461794 1.911140 0.541263 1.199721 2.537742 0.397199 3.074449	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131389 0.689469 1.860772 -0.231147 -2.197365 -2.452339 -2.824471 -2.331480 -3.060587 2.914277	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0.738328 -0.213134 0.431590 -1.744845 0.486041 -1.138647 -1.252621 -2.271851 -2.507759 0.265225
H 4CCCCCNOOCHHHNHCHCHHHHHSCCCCCHC	-1.849411 -0.590306 -0.576240 -1.850351 -2.561278 -2.259454 -2.231414 -3.902501 -3.944991 -4.599309 -4.169584 0.344592 -0.008347 1.763263 2.108269 2.569014 2.247137 3.625323 2.461794 1.911140 0.541263 1.199721 2.537742 0.397199 3.07449 3.152922 0.940702	-1.031692 -0.938442 0.309728 0.944082 0.103130 -1.918149 2.039349 0.381626 1.378322 0.346636 -0.372903 0.996479 1.906892 0.748684 1.487135 0.865178 0.131389 0.689469 1.860772 -0.231147 -2.197365 -2.452339 -2.824471 -2.331480 -3.060587 -2.914377 2 558146	1.006387 0.291124 -0.432432 -0.222306 0.636755 1.757269 -0.659180 1.082525 1.523775 0.242645 1.818107 -1.111451 -1.388358 -1.291789 -2.014652 -0.001079 0.738328 -0.213134 0.431590 -1.744845 0.486041 -1.138647 -1.252621 -2.271851 -2.507759 -0.365825 3.521408
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C 0.162453 2.326154 -0 H 0.379223 2.982055 -1 C -0.180293 3.156354 0 H -0.411547 2.520813 1 H 0.670155 3.787237 0 H 1.057479 1.730543 -0 S 0.167774 -0.356951 1 C 1.599174 -0.247871 0 C 2.723563 0.445346 1 C 1.606211 -0.877984 -0 C 3.837422 0.501013 0 C 2.726358 -0.795167 -1 H 0.736088 -1.433209 -0 C 3.843293 -0.100953 -0 C 3.843293 -0.100953 -0 H 4.716085 1.034736 0 H 4.725107 -0.033358 -1 C 2.733780 1.117665 2 H 1.957239 1.882393 2 H 2.556275 0.402560 3 H 3.696681 1.593041 2 4e C -1.204544 -1.032662 -0 C -0.144886 1.010355 -0 C -1.564068 1.231846 -0 N -2.136721 -0.018887 -1 O -1.488575 -2.225419 -4 O -2.170461 2.296819 0 C -3.559135 -0.247413 -0 H -4.055012 0.629568 -0	1.840317
H 0.379223 2.982055 -1 C -0.180293 3.156354 0 H -0.411547 2.520813 1 H 0.670155 3.798073 0 H 1.057479 1.730543 -0 S 0.167774 -0.356951 1 C 1.599174 -0.247871 0 C 2.723563 0.445346 1 C 1.606211 -0.877984 -0 C 3.837422 0.501013 0 C 2.726358 -0.795167 -1 H 0.736088 -1.433209 -0 C 3.843293 -0.100953 -0 H 2.729384 -1.281369 -2 H 4.725107 -0.033358 -2 H 4.725107 -0.033358 -2 H 1.957239 1.882393 2 H 1.957239 1.882393 2 H 3.696681 1.593041 2 H 2.556275 0.402560 3 H 3.696681 1.231846 -0 C -1.204544 -1.032662 -0 <	0.645638
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H -1.040895 3.798073 (0) H 1.057479 1.730543 $-(0)$ S 0.167774 -0.356951 1 C 1.599174 -0.247871 (0) C 2.723563 0.445346 1 C 1.606211 -0.877984 $-(0)$ C 3.837422 0.501013 (0) C 2.726358 -0.795167 -1 H 0.736088 -1.433209 $-(0)$ C 3.843293 -0.100953 $-(0)$ H 4.716085 1.034736 (0) H 4.725107 -0.033358 $-(0)$ H 4.725107 -0.033358 $-(2)$ H 2.556275 0.402560 33 H 2.556275 0.402560 33 H 3.696681 1.593041 2 4e $-(2).1204544$ -1.032662 $-(2)$ C -0.144886 1.010355 $-(2)$ C -1.204544 -1.032662 $-(2)$ C -1.488575 -2.225419 $-(2)$ O -2.170461 2.296819 (2) O -2.170461 2.296819 (2) O -3.559135 -0.247413 $-(2)$ H -3.922115 -0.408151 (2) H </td <td>0.855041</td>	0.855041
H 1.057479 1.730543 -0.5913 S 0.167774 -0.356951 1 C 1.599174 -0.247871 0 C 2.723563 0.445346 1 C 1.606211 -0.877984 -0 C 3.837422 0.501013 0 C 2.726358 -0.795167 -1 H 0.736088 -1.433209 -0 C 3.843293 -0.100953 -0 H 4.716085 1.034736 0 H 2.729384 -1.281369 -2 H 4.725107 -0.033358 -7 C 2.733780 1.117665 2 H 1.957239 1.882393 2 H 2.556275 0.402560 3 H 3.696681 1.593041 2 4eC -1.204544 -1.032662 -0 C 0.087662 -0.380559 -0 C -1.564068 1.231846 -0 N -2.136721 -0.018887 -0 C -3.559135 -0.247413 -0 H -3.922115 -0.408151 0 H -4.055012 0.629568 -0	0.395473
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	0.469351
C 1.599174 -0.247871 (C C 2.723563 0.445346 1 C 1.606211 -0.877984 -(C C 3.837422 0.501013 0 C 2.726358 -0.795167 -1 H 0.736088 -1.433209 -(C C 3.843293 -0.100953 -(C H 4.716085 1.034736 0 H 2.729384 -1.281369 -2 H 4.725107 -0.033358 -1 C 2.733780 1.117665 2 H 1.957239 1.882393 2 H 2.556275 0.402560 3 H 3.696681 1.593041 2 4e C -1.204544 -1.032662 -(C C 0.087662 -0.380559 -(C C -0.144886 1.010355 -(C C -1.564068 1.231846 -(C N -2.136721 -0.018887 -(C) N -2.136721 -0.018887 -(C) O -1.488575 -2.225419 -(C) C -3.559135 -0.247413 -(C) H -3.922115 -0.408151 (C) H -4.055012 0.629568 -(C)	1.792192
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C 1.606211 -0.877984 -(C 3.837422 0.501013 0 C 2.726358 -0.795167 -1 H 0.736088 -1.433209 -(C 3.843293 -0.100953 -(H 4.716085 1.034736 0 H 2.729384 -1.281369 -2 H 4.725107 -0.033358 -2 C 2.733780 1.117665 2 H 1.957239 1.882393 2 H 2.556275 0.402560 3 H 3.696681 1.593041 2 4e C -1.204544 -1.032662 -(C 0.087662 -0.380559 -(C -0.144886 1.010355 -(C -1.564068 1.231846 -(N -2.136721 -0.018887 -(O -1.488575 -2.225419 -(O -2.170461 2.296819 0 C -3.559135 -0.247413 -(H -3.922115 -0.408151 0 H -4.055012 0.629568 -(1.222317
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.494958
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