Supporting information for:

Comparison of rhenium-porphyrin dyads for CO₂ photoreduction: photocatalytic studies and charge separation dynamics studied by time-resolved IR spectroscopy

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Table of contents

NMR spectra and assignments	S 1
Mass spectrometry of [Dyad 3 pic]OTf	S 8
Cyclic voltammetry	S 9
Crystallographic data	S10
Emission spectra	S13
TON plot for [Dyad 2 pic]OTf and Dyad 2 Br	S15
Characterisation of hydrogenated porphyrin	S16
IR spectra of ligand substitution	S17
Dynamics in UV/vis spectra during catalysis	S20



Figure S 1. ¹H NMR assignment of Dyad 3 Br.





Figure S 3. 1 H- 1 H NOESY spectrum of **Dyad 3 Br** (THF- d_{8} 400 MHz).



Figure S 4. ¹³C NMR assignment of Dyad 3 Br.



Figure S 6. 1 H- 13 C HMBC NMR spectrum of Dyad 3 Br (THF- d_{8} 400 MHz).

8.5

8.0

9.0

10.0

9.5

ppm

7.5



Figure S 7. ¹H (top) and ¹³C (bottom) assignment of [Dyad 3 pic]OTf.





Figure S 11. ¹H-¹³C HMBC NMR spectrum of [Dyad 3 pic]OTf (THF-d₈ 400 MHz).



Figure S 12 ESI-MS of **[Dyad 3 pic]OTf**: Top is full mass range, bottom left is expansion of M⁺, bottom right is the calculated isotope pattern.







Figure S 14. Cyclic voltammogram of Dyad 3 Br



Figure S 15. X-ray crystal structure of **[Dyad 1 pic]PF**₆. Hydrogen atoms and 2 porphyrin phenyls, PF_6 counterion and solvent of crystallization omitted for clarity. Selected bond length (Å): Zn(1)-O(1): 2.216(4).



Figure S 16. X-ray crystal structure of 5-[4-[(2-methoxy-4-nitro-phenylcarbonyl)-amino]phenyl]-10,15,20-triphenyl porphyrin. Selected hydrogen atoms and solvent of crystallization omitted for clarity. H(5) was located. Thermal ellipsoids shown with probability of 50%. Selected bond length (Å): O(2)-N(5): 2.658(4).

Identification code	rnp1309	
Empirical formula	$C_{67.5}H_{49}Cl_5F_6N_8O_4PReZn$	
Formula weight	1609.94	
Temperature/K	110.00(10)	
Crystal system	triclinic	
Space group	P-1	
a/Å	12.9474(5)	
b/Å	15.8804(5)	
c/Å	17.3277(6)	
$\alpha/^{\circ}$	98.752(3)	
β/°	92.264(3)	
$\gamma/^{\circ}$	106.821(3)	
Volume/Å ³	3357.4(2)	
Z	2	
$\rho_{calc} mg/mm^3$	1.592	
m/mm^{-1}	2.452	
F(000)	1602.0	
Crystal size/mm ³	$0.1612 \times 0.091 \times 0.0831$	
2Θ range for data collection	5.7 to 54.2°	
Index ranges	$\textbf{-16} \leq h \leq 16, \textbf{-20} \leq k \leq 20, \textbf{-22} \leq l \leq 18$	
Reflections collected	25307	
Independent reflections	14802[R(int) = 0.0307]	
Data/restraints/parameters	14802/25/880	
Goodness-of-fit on F ²	1.073	
Final R indexes [I>= 2σ (I)]	$R_1 = 0.0659, wR_2 = 0.1868$	
Final R indexes [all data]	$R_1 = 0.0809, wR_2 = 0.1986$	
Largest diff. peak/hole / e $Å^{-3}$ 4.12/-1.78		
CCDC number	1406001	

Table S 1. Crystallographic data for [Dyad 1 pic][PF_6] 2.5 CH_2Cl_2

Table S 2. Crystallographic data for 5-[4-[(2-methoxy-4-nitro-phenylcarbonyl)-amino]phenyl]-10,15,20-triphenyl porphyrin 0.595 CH2Cl2

Empirical formula	$C_{52.59375}H_{37.1875}Cl_{1.1875}N_6O_4$	
Formula weight	859.29	
Temperature/K	110.00(10)	
Crystal system	orthorhombic	
Space group	Fddd	
a/Å	23.5964(4)	
b/Å	37.3470(6)	
c/Å	40.0789(6)	
α/°	90.00	
β/°	90.00	
γ/°	90.00	
Volume/Å ³	35319.7(9)	
Z	32	
$\rho_{calc}mg/mm^3$	1.293	
m/mm ⁻¹	0.152	
F(000)	14303.4	
Crystal size/mm ³	$0.1818 \times 0.1617 \times 0.1159$	
2Θ range for data collection	5.76 to 50.7°	
Index ranges	$-25 \le h \le 28, -44 \le k \le 44, -45 \le l \le 48$	
Reflections collected	26924	
Independent reflections	8081[R(int) = 0.0340]	
Data/restraints/parameters	8081/6/606	
Goodness-of-fit on F^2	1.035	
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0578, wR_2 = 0.1587$	
Final R indexes [all data]	$R_1 = 0.0712, wR_2 = 0.1693$	
Largest diff. peak/hole / e Å ⁻³ 0.85/-1.07		
CCDC number	1406000	



Figure S 17. Emission spectra of **[Dyad 2 pic]OTf** and the Zn porphyrin-bridgebipyridine ligand of Dyad 2.



Figure S 18. Emission spectra of **[Dyad 3 pic]OTf** and the Zn porphyrin-bridgebipyridine ligand of Dyad 3.



Figure S 19. Emission time scan (black) and fitting (green) for [Dyad 3 pic]OTf. λ_{ex} = 560 nm and λ_{em} = 605 nm.



Figure S 20. Emission spectra of ZnTPP, Dyad 1 Br, Dyad 2 Br and Dyad 3 Br in THF.



Figure S 21. Emission time scan (black) and fitting (green) for Dyad 3 Br. $\lambda_{ex} = 560$ nm and $\lambda_{em} = 605$ nm.



Figure S 22. TON_{CO} for Dyad 2 Br and [Dyad 2 pic]OTf. In DMF:TEOA 5:1 and with $\lambda > 520$ nm irradiation.



Figure S 23 ¹H NMR spectra. Above: authentic zinc tetraphenylchlorin, below: photoreaction mixture of ZnTPP in DMF:TEOA 5:1.



Figure S 24. Left: ESI-MS of ZnTPP in DMF/TEOA after irradiation with $\lambda > 520$ nm, right: calculated isotope pattern for ZnTPP, Zn chlorin and Zn isobacteriochlorin superimposed on top of one another in a 1:1:1 ratio.



Figure S 25. IR spectra of [Dyad 2 pic]OTf following dissolution in DMF at times given.



Figure S 26. IR spectra of Dyad 2 Br following dissolution in DMF at times given.



Figure S 27. IR spectra of **[Dyad 2 pic]OTf** in DMF after addition of TEOA (DMF:TEOA 5:1): a) absorption spectra at times after addition of TEOA, b) difference spectra at same times.



Figure S 28. IR difference spectra of Dyad 2 Br in DMF after addition of TEOA (DMF:TEOA 5:1) and with $\lambda > 520$ nm irradiation relative to before addition of TEOA: a) under N₂, b) same sample following bubbling of CO₂.



Figure S 29. Absorbance changes in the UV/vis spectra of Dyad 1 Br and [Dyad 1 pic]PF₆ during CO₂ reduction: a) changes in signal intensity at 560 nm and turnovers of CO, b) changes in signal intensity at 610 nm, c) changes in signal intensity at 630 nm.