

**Original Article**

**A self-assembling prodrug approach emerges as the most efficient strategy to produce nanoparticles with high payloads of pipemidic acid, a poorly soluble crystalline antibiotic**

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## Supplementary Information

List:

Figure S1

Figure S2

Nanoemulsion method - PIP release upon dilution

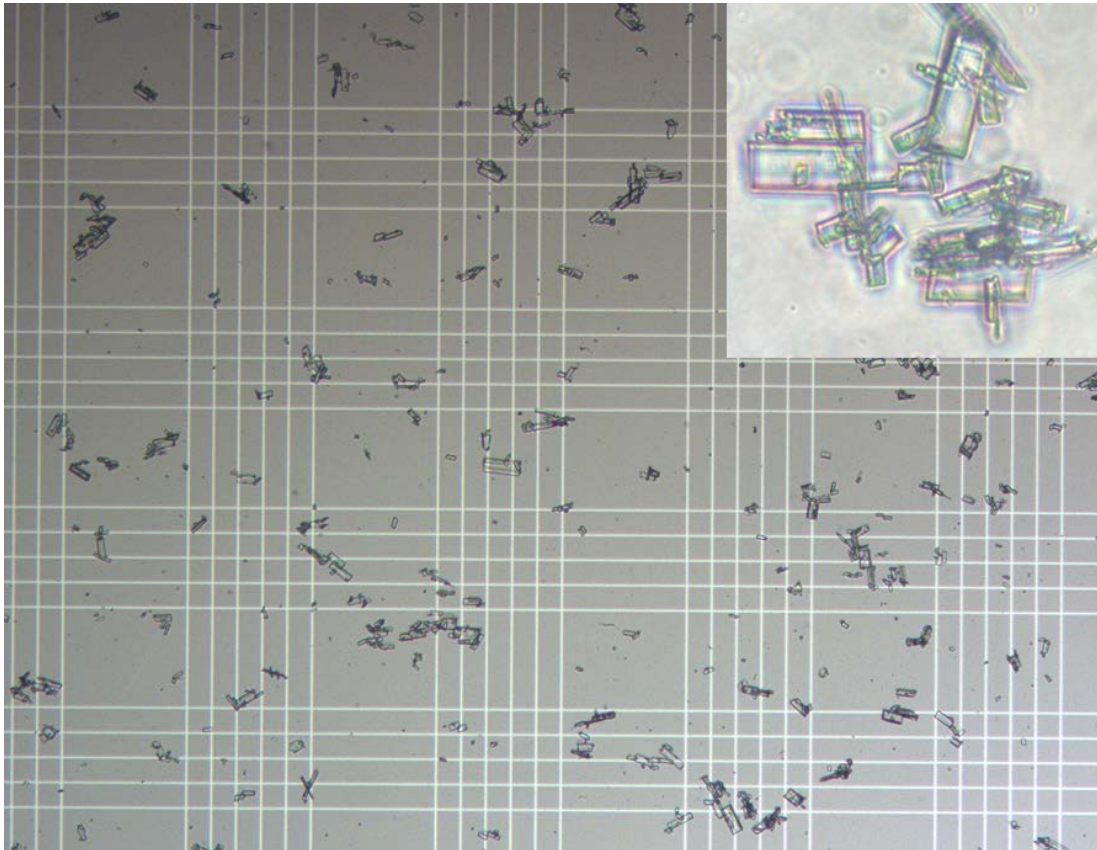
Nanoprecipitation method supplementary information

Figure S3

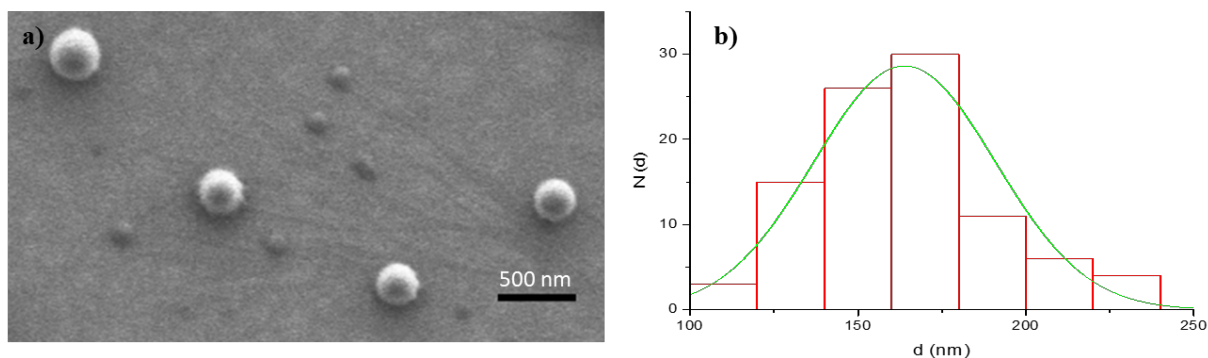
Figure S4

Figure S5

Figure S6



**Figure S1.** Optical image of PIP crystals in PLA/PLGA NPs. Typical example of PIP crystals formed in NPs suspension observed with 10 times and 40 times (zoomed panel) magnification in an inverted optical microscope over a Malassez chamber.



**Figure S2** PLGA PIP-loaded NPs observed by SEM. a: SEM micrograph. b: Histogram of the size distribution of the NPs showing a mean diameter  $d = 164 \pm 27$  nm.  $N(d)$  refers to the total counts and the scale bar corresponds to 500 nm.

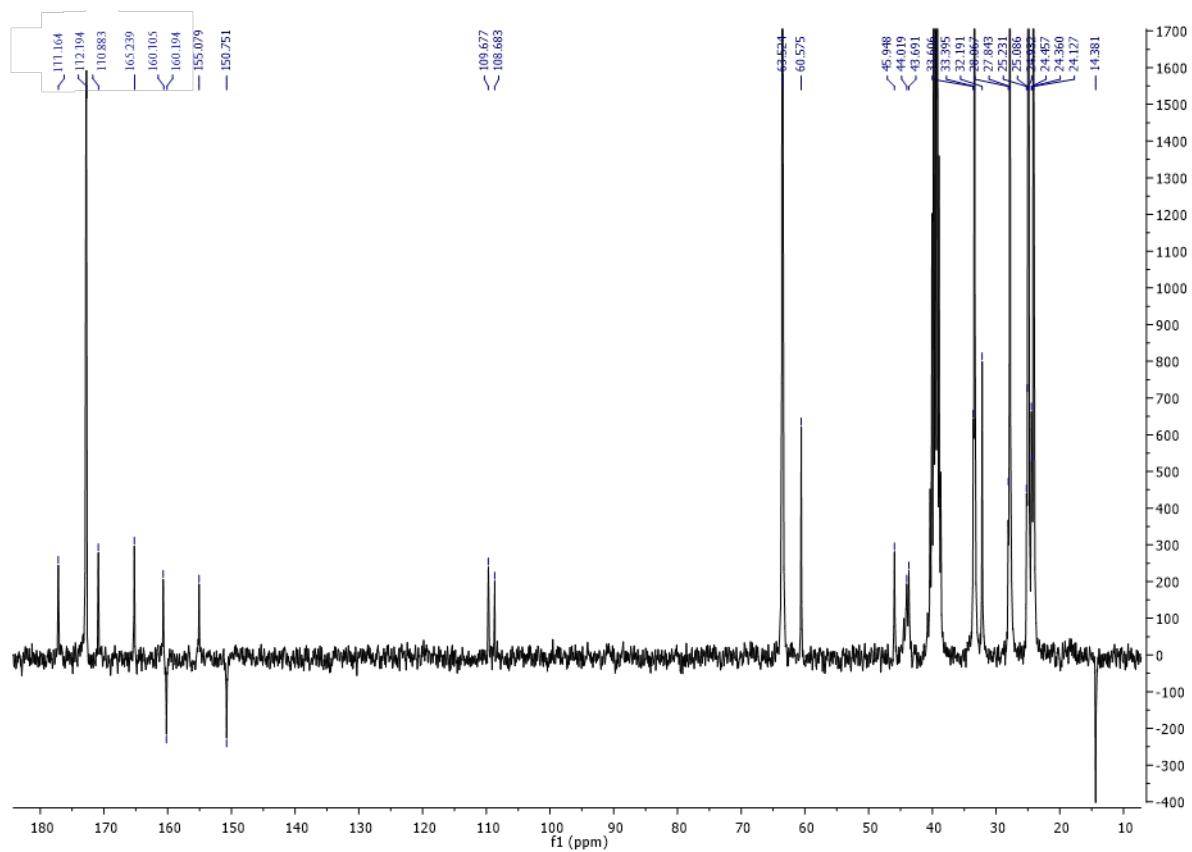
### Nanoemulsion method - PIP release upon dilution

The release of PIP from NPs was thoroughly studied after progressive dilution in water and PIP content was lost within less than 5 min. Interestingly, there was a correlation between the amount of burst-released drug and the dilution factor. For example, the DL of NPs made using polymer P33 (see Main text - method section 4.1) decreased from 8.7% to 4.1%, 2.4% and 1.7% ( $w/w$ ) after 0, 10, 20 and 50 times dilution, respectively.

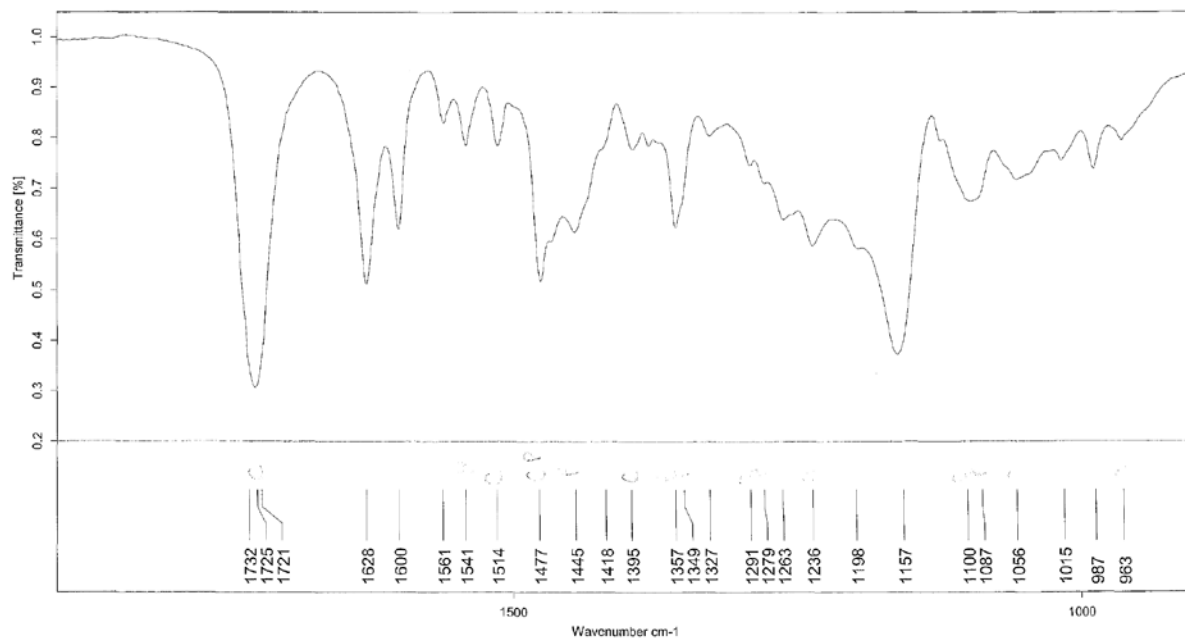
### Nanoprecipitation method supplementary information

The P19 NPs were formulated in presence and in absence of PIP. Interestingly, the size measurement of the obtained particles highlighted a 50–60 nm difference in the mean size of empty and loaded NPs. For example, in the case of a final P19 concentration of 5 mg/mL, the empty NPs displayed a size of  $170 \pm 1.3$  nm while the loaded ones of only  $117 \pm 1.6$  nm (PIP/P19 ratio =1:10). The same behavior was evidenced also when increasing P19 final concentration to 10 mg/mL (or higher) (empty NPs=  $212 \pm 2.3$  nm, PIP NPs=  $150 \pm 2$  nm). The same tendency was evidenced also when formulating the NPs by using other polymers.

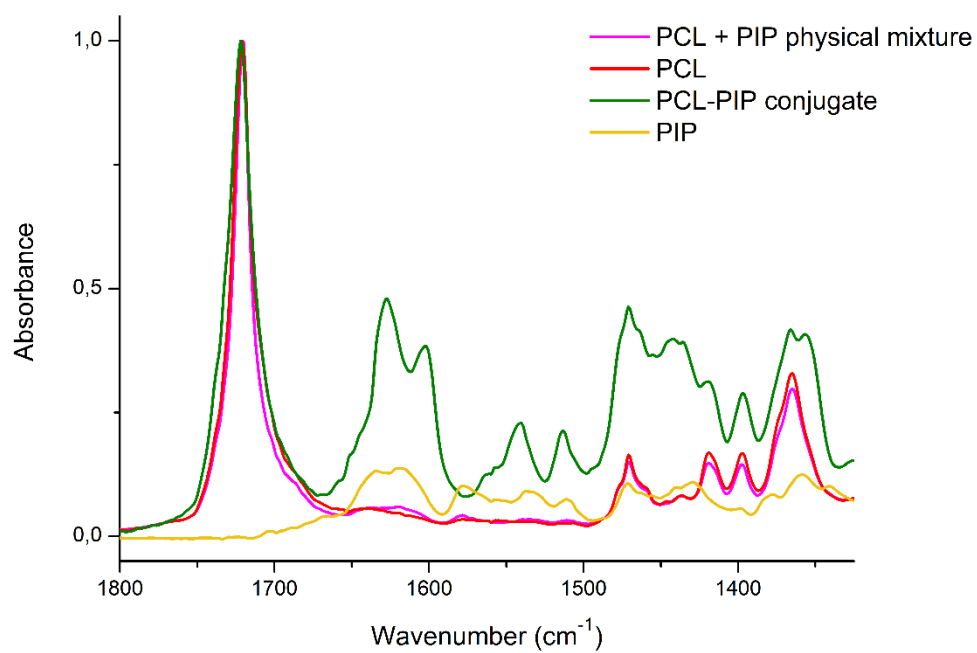
## Supplementary Information



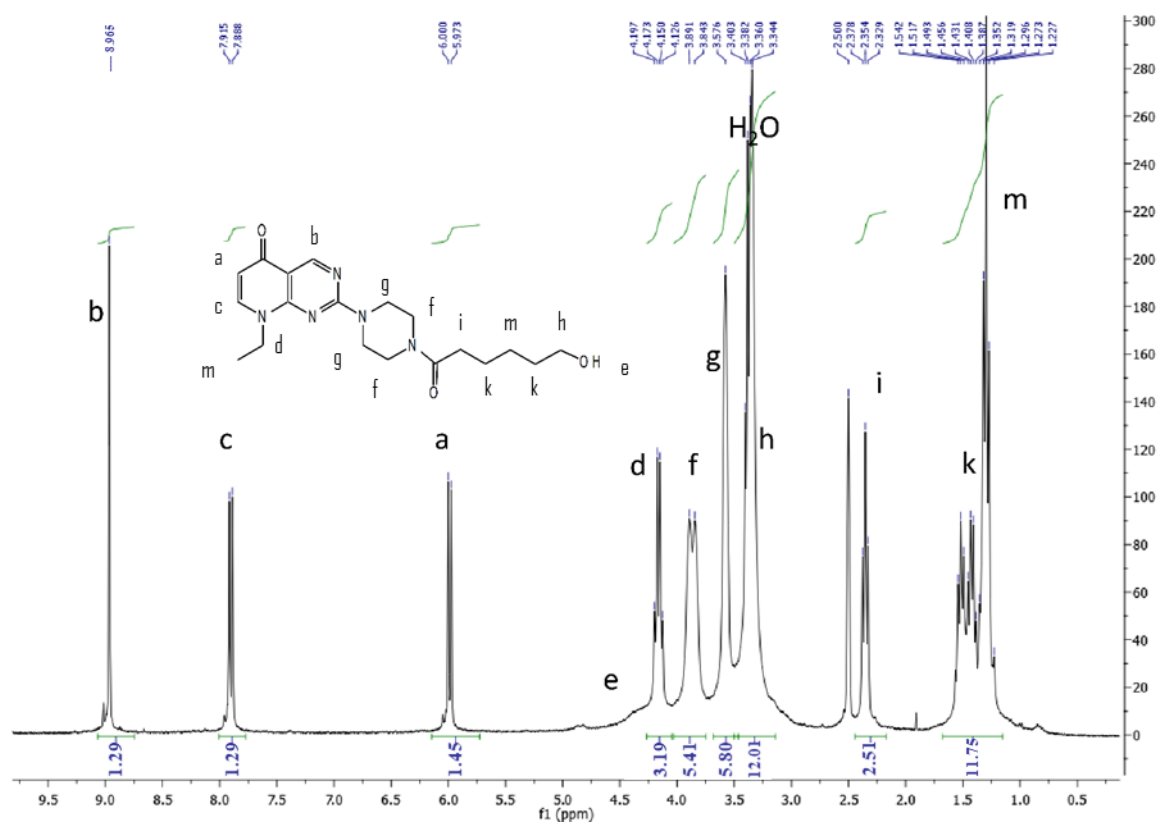
**Figure S3**  $^{13}\text{C}$  NMR spectrum of PCL-PIP in  $\text{DMSO-}d_6$ .



**Figure S4** IR spectrum of PCL-PIP.



**Figure S5** IR spectra of PCL, PIP, PCL-PIP conjugate and a physical mixture of PCL and PIP.



**Figure S6** <sup>1</sup>H NMR of decarboxylated caprolactone adduct (**4**) (see Fig. 2 main manuscript).

## Supplementary Information