

Supplementary Information for

Metastable quasicrystal-induced nucleation in a bulk glass-forming liquid

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Figs. S1 to S6 Table S1 References for SI reference citations



Fig. S1. Diffraction pattern of Mg₅₁Zn₂₀-type phase. Selected area diffraction patterns of the microstructure in as-cast Mg₆₄Zn₂₇Yb₉ shown in Figure 2c along (*A*) [201], (*B*) [250] and (*C*) [034] zone axes of Mg₅₁Zn₂₀-type phase (orthorhombic, Immm, a = 14.083 Å, b = 14.486 Å, c = 14.025 Å) (1) with modified lattice parameters (a = 13.707 Å, b = 14.819 Å, c = 14.047 Å).



Fig. S2. Mg–Zn–Yb compositions. Mg–Zn–Yb ternary diagram indicating compositions for: QC and metallic glass-forming alloys found in this study; QC-forming alloys investigated by Mitani and Ishimasa (2); binary approximant phases: Mg₅₁Zn₂₀ (3, 4), Zn₁₇Yb₃ (5); and constructed cluster lines binding Zn₁₇Yb₃ to Mg and eutectic composition Mg_{71.1}Zn_{28.9} (6) to Yb. The approximant phase Mg₂₉Zn₆₀Yb₁₁ "3" and the iQC phase Mg₃₈Zn₅₄Yb₈ "2" follow the cluster line between Zn₁₇Yb₃ and Mg. The Mg₆₉Zn₂₇Yb₄ alloy composition (see arrow) is located near the intersection of these two cluster lines and forms a BMG.



Fig. S3. XRD pattern of a Mg₆₉Zn₂₇Yb₄ bulk metallic glass. XRD pattern taken with Cu K α radiation from the cross-sectional surface of a Mg₆₉Zn₂₇Yb₄ rod cast to 1 mm diameter.



Fig. S4. XRD pattern of stable equilibrium state. XRD pattern taken with Cu K α radiation for a Mg₆₉Zn₂₇Yb₄ alloy solidified at a cooling rate of 20 K/min using DSC. Diffraction peaks are indexed as Mg (hexagonal, P63/mmc, space group 194, a=b=3.2107 Å, c=5.2061 Å) and Mg_{13.2}Zn_{29.3}Gd_{3.5}-type Mg₂₉Zn₆₀Yb₁₁ phase (hexagonal, P63/mmc, space group 194, a=b=14.8053 Å, c=8.8256 Å). Rietveld analysis was performed using reference Mg crystal structure (ICDD reference code 04-003-5224, a=b=3.2040 Å, c=5.2070 Å) and Mg_{13.2}Zn_{29.3}Gd_{3.5} crystal structure (ICDD reference code 04-009-2178, a=b=14.6330 Å, c=8.7610 Å).



Fig. S5. XRD pattern of metastable state. XRD pattern taken with Mo K α radiation for a Mg₆₉Zn₂₇Yb₄ alloy solidified at a cooling rate of 100 K/s using FDSC. Diffraction peaks are indexed as hexagonal Mg and primitive-type iQC; see Table S1.



Fig. S6. Temperature profiles for studying the metastable-to-stable transition. The effect of temperature and time on the metastable-to-stable phase transition was investigated by designing FDSC experiments using the illustrated temperature profiles. $Mg_{69}Zn_{27}Yb_4$ melt is solidified at a cooling rate of 100 K/s, which results in formation of metastable QC and Mg phases. Cooling is interrupted at 240°C and the temperature of the metastable solid is increased to isothermal treatment temperatures (305, 315, 325°C) at a rate of 1,000 K/s. The metastable solid is then held at these temperatures for various time intervals (0–70 s), which generates a metastable-to-stable phase transition. Finally, the alloy is melted at a rate of 1,000 K/s.

Table S1. XRD peak positions for metastable state. Peak positions (2 θ , *d*) and normalized peak intensities of hexagonal Mg and Mg–Zn–Yb primitive-type iQC formed during solidification in FDSC at a rate of 100 K/s, determined from XRD data (see Fig. S5). Peaks for the QC phase are indexed following the work of Elser (7) and distances are calculated using the six-dimensional lattice parameter $a_{6D} = 7.46$ Å.

Index (QC)	Index (Mg)	2θ (Mo)	d _{measured} (Å)	d _{calculated} (Å)	∆ <i>d</i> (Å)	Intensity
(111000)		9.23	4.416	4.425	0.009	7
(111100)		10.49	3.888	3.833	0.055	7
	(100)	14.74	2.771			9
	(002)	15.78	2.589			13
(211111)		16.41	2.490	2.491	0.001	55
	(101)	16.72	2.444			69
(221001)		17.23	2.372	2.369	0.003	100
(222100)		19.46	2.103	2.088	0.015	9
(311111)		20.40	2.007	2.015	0.008	9
(222121)	(102)	21.62	1.895	1.882	0.013	5
	(110)	25.66	1.600			4
(332002)	(103)	28.02	1.468	1.464	0.004	23
(333101)	(200)	29.72	1.386	1.390	0.004	7
(422211)	(112)	30.19	1.365	1.368	0.003	8
	(201)	30.92	1.339			5
(422212)	(004)	31.96	1.297	1.307	0.010	2
(333222)		32.99	1.252	1.243	0.009	3
(443110)		35.00	1.182	1.182	0.000	1
(444000)		37.39	1.109	1.106	0.002	2

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