

Synthesis of novel inhibitors of α -amylase based on thiazolidine-4-one skeleton containing pyrazole moiety and their configurational studies

Parvin Kumar^{a*}, Meenakshi Duhan^a, Kulbir Kadyan^a, Jayant Sindhu^b, Sunil Kumar^c, Hitender Sharma^c

^aDepartment of Chemistry, Kurukshetra University Kurukshetra-136119

^bS D (PG) College, Panipat-132103,

^cInstitute of Pharmaceutical Science, Kurukshetra University Kurukshetra-136119

*E.mail: parvinjangra@gmail.com; parvinchem@kuk.ac.in

Synthesis of 2-(*p*-tolylimino)thiazolidin-4-one (3) [1]

It was prepared by a three component reaction of 1-*p*-tolylthiourea (1) (1.0 mmol), ethyl bromoacetate (1.0 mmol) and sodium acetate (2.0 mmol) in glacial acetic acid under the refluxing condition for 2 h. After completion of reaction, reaction was quenched using ice and the solid so formed was filtered under suction and recrystallized from ethanol. Yield 80%; M.pt: 167-170°C.

Synthesis of 5-((3-(aryl)-1-phenyl-1*H*-pyrazol-4-yl)methylene)-2-(*p*-tolylimino)thiazolidin-4-one (5a-g) [1]

2-Arylimino-thiazolidin-4-ones **1** (0.5 mmol) and 1-phenyl-3-(*p*-substituted phenyl)-1*H*-pyrazole-4-carbaldehydes **2** (0.6 mmol) were dissolved in absolute ethanol. Piperidine (0.5 mmol) was added to the reaction mixture and the reaction mixture was stirred for 8 h at 60°C until precipitate formed. Then the mixture was cooled to room temperature, and the precipitates formed were filtered and washed with absolute ethanol to yield the final compound **5a-5g** in good to excellent yield.

Table S1: Percentage of peaks of *E* and *Z* isomers present in **5a-5g** in DMSO-*d*₆

Compound	Pyr-H (H ₁₇)		H _{10/12}		H _{28/32}		C-CH ₃ (C-14)	
	<i>E</i> -isomer	<i>Z</i> -isomer	<i>Z</i> -isomer	<i>E</i> -isomer	<i>E</i> -isomer	<i>Z</i> -isomer	<i>Z</i> -isomer	<i>E</i> -isomer
5a	8.61 (39.2%)	8.52 (61.7%)	^b 7.19- 7.21 (59.1%)	6.97 (40.8%)	7.96 (40.7%)	7.92 (59.3%)	2.31 (60.3%)	2.30 (39.6%)
5b	8.64 (39.8%)	8.54 (60.1%)	^b 7.19- 7.21 (59.1%)	6.96 (40.8%)	7.97 (40.6%)	7.93 (59.3%)	2.31 (60.0%)	2.30 (39.9%)
5c	8.30 (38.8%)	8.18 (61.1%)	^b 6.93- 6.98	6.74 (40%)	7.72 (37.7%)	7.66 (62.2%)	2.13 (62.9%)	2.11 (37.1%)
5d	8.58 (37.1%)	8.47 (62.8%)	^a -	^a -	7.93 (37.0%)	7.88 (62.2%)	2.31 (62.4%)	2.30 (37.5%)
5e	8.54 (28.1%)	8.42 (71.8%)	^a -	^a -	^a -	^a -	2.31 (62.6%)	2.30 (37.3%)
5f	8.56 (39.1%)	8.44 (60.8%)	^b 7.11- 7.18	6.93 (40%)	7.92 (42.3%)	7.87 (57.6%)	2.31 (61.2%)	2.30 (38.7%)
5g	8.61 (42.0%)	8.49 (58.4%)	^a 7.14- 7.16	6.91 (40%)	^b 7.93- 7.96	-	2.31 (61.2%)	2.30 (38.7%)

^a Signals are not separated^b Mixed with multiplet**Table S2:** DFT Calculations for different configuration of **5a**

Configuration	<i>2Z,5Z</i>	<i>2E,5E</i>	<i>2Z,5E</i>	<i>2E,5Z</i>
Dipole moment (Debye)	3.97	3.15	1.24	1.49
Difference in energy(kcal/mole)	0	71.04	60.14	2.08

Fig: S1

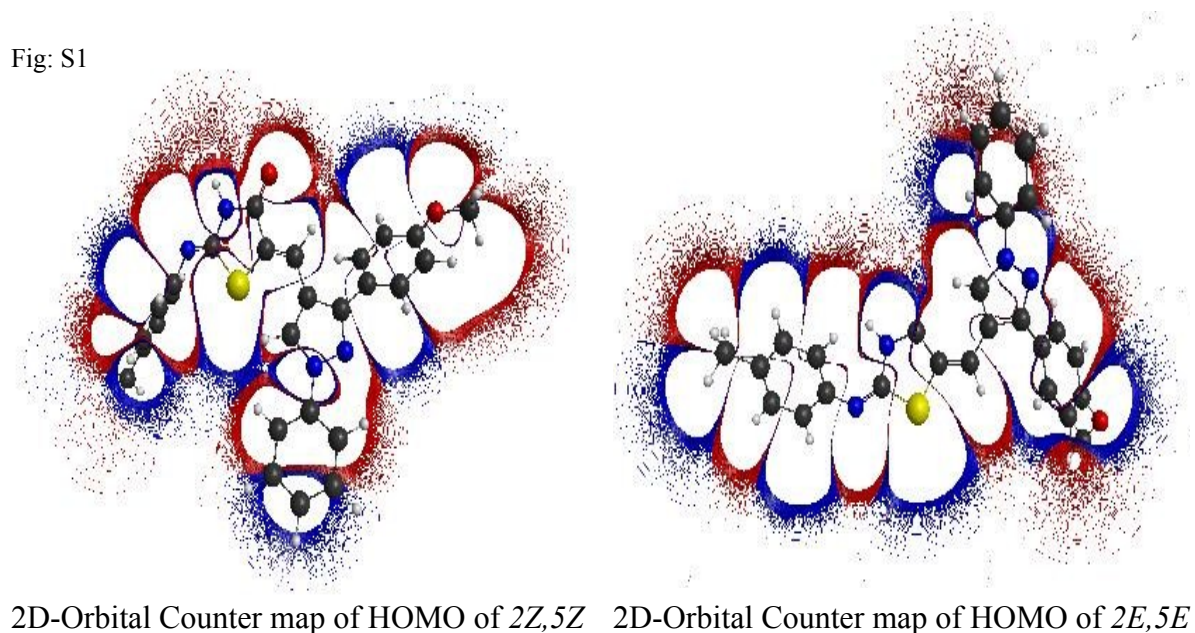
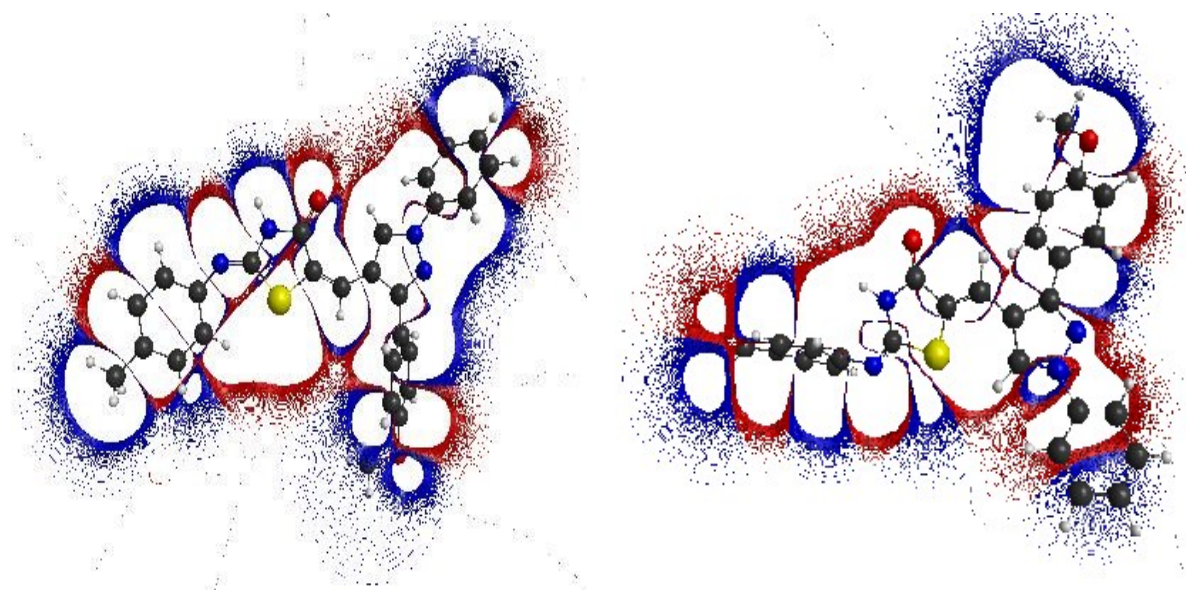


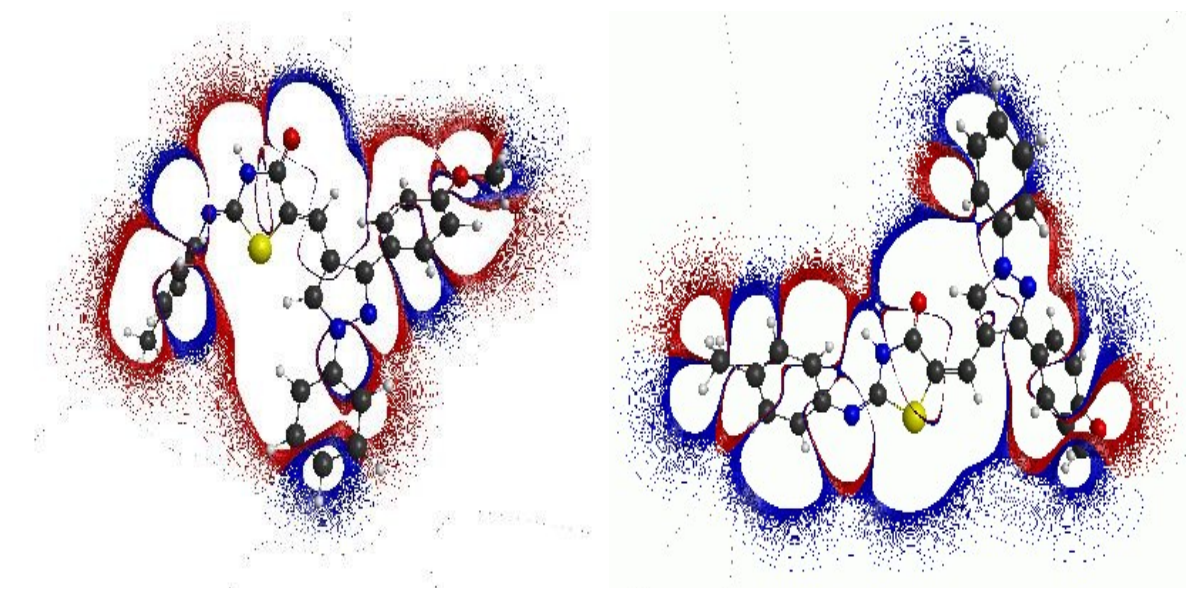
Fig: S2



2D-Orbital Counter map of HOMO of 2Z,5E

2D-Orbital Counter map of HOMO of 2E,5Z

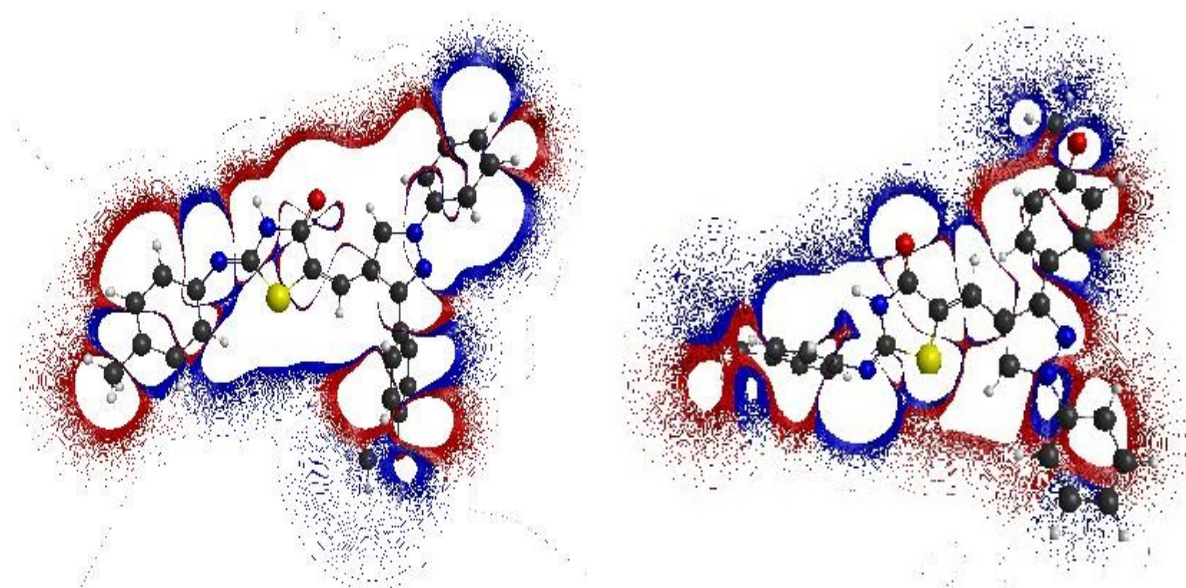
Fig: S3



2D-Orbital Counter map of LUMO of 2Z,5Z

2D-Orbital Counter map of LUMO of 2E,5E

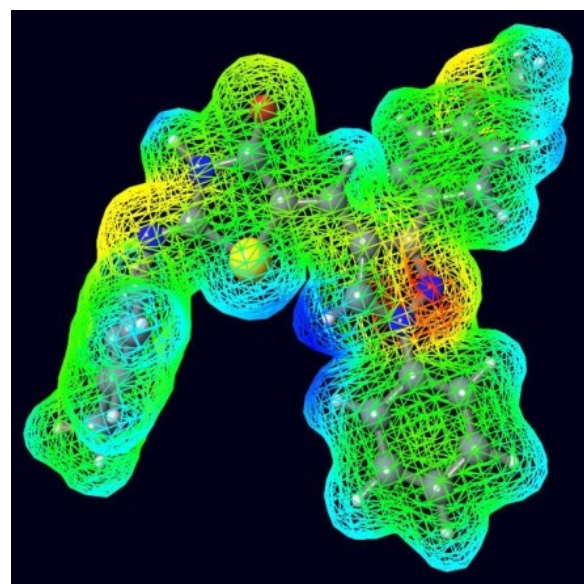
Fig: S4



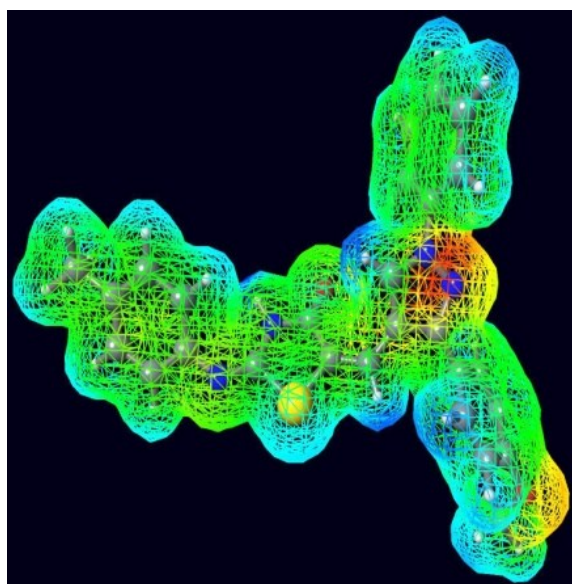
2D-Orbital Counter map of LUMO of 2Z,5E

2D-Orbital Counter map of LUMO of 2E,5Z

Fig: S5

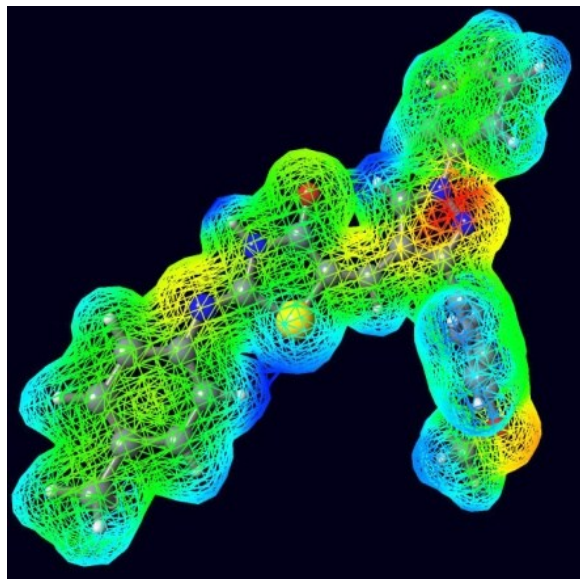


Molecular electrostatic potential surface at 0.01 of 2Z,5Z

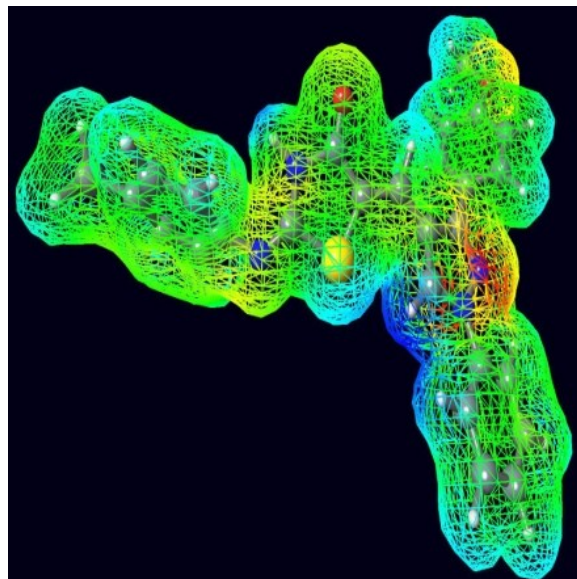


Molecular electrostatic potential surface at 0.01 of 2E,5E

Fig: S6

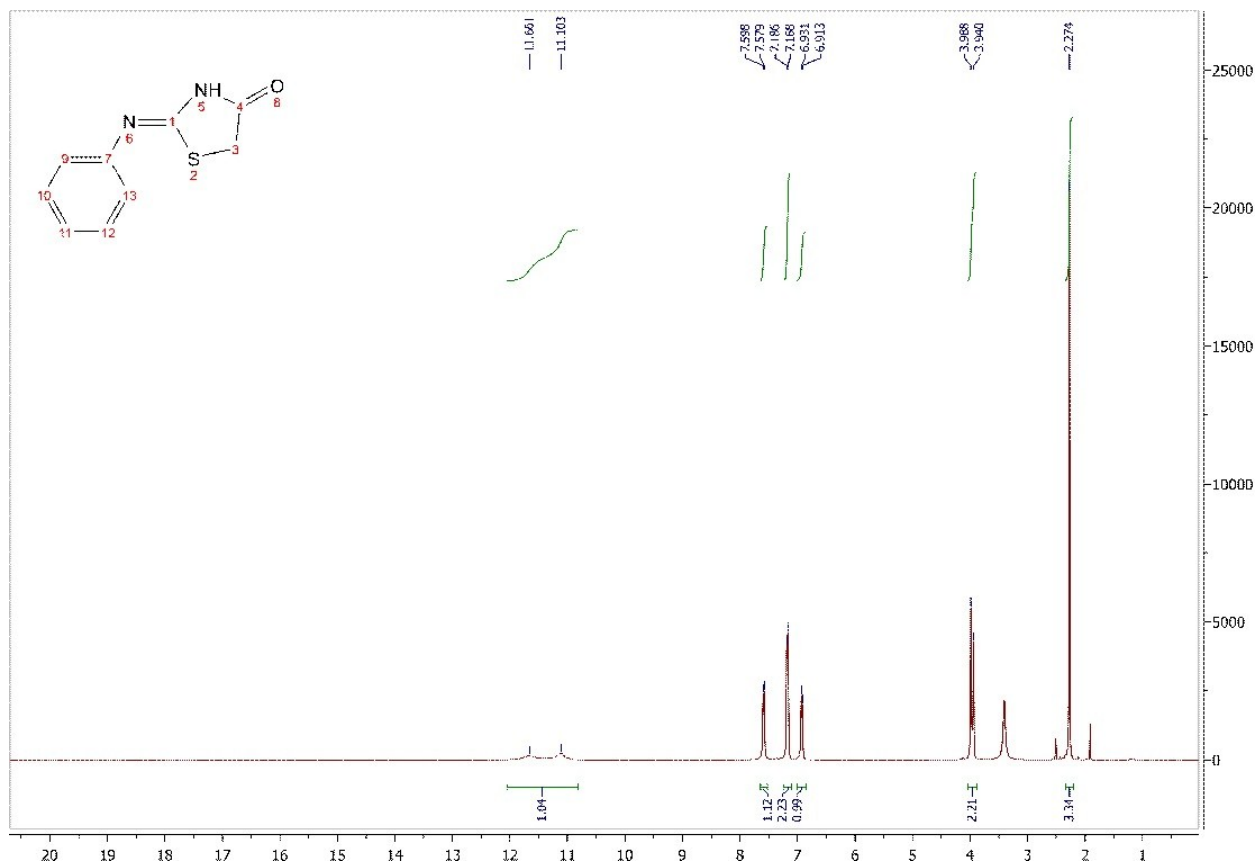


Molecular electrostatic potential surface at 0.01 of
2Z,5E

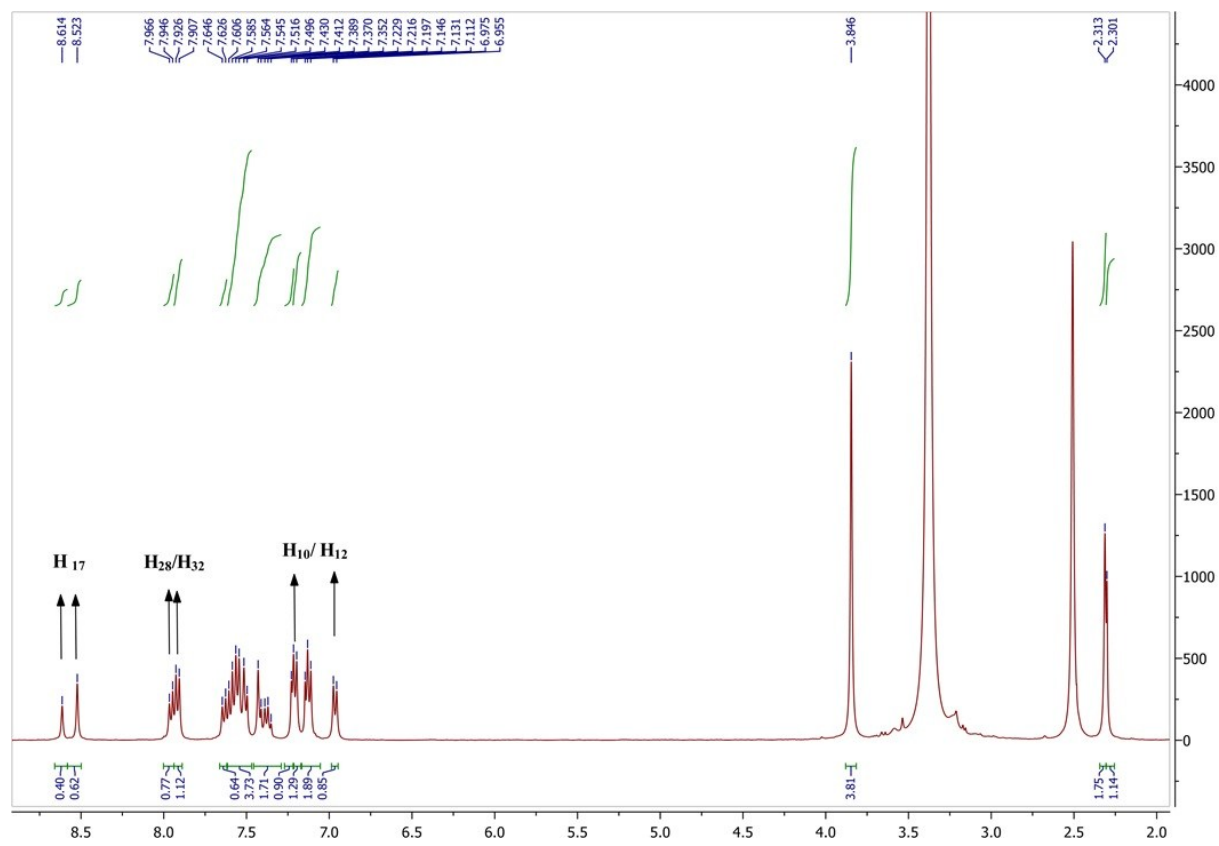


Molecular electrostatic potential surface at 0.01 of
2E,5Z

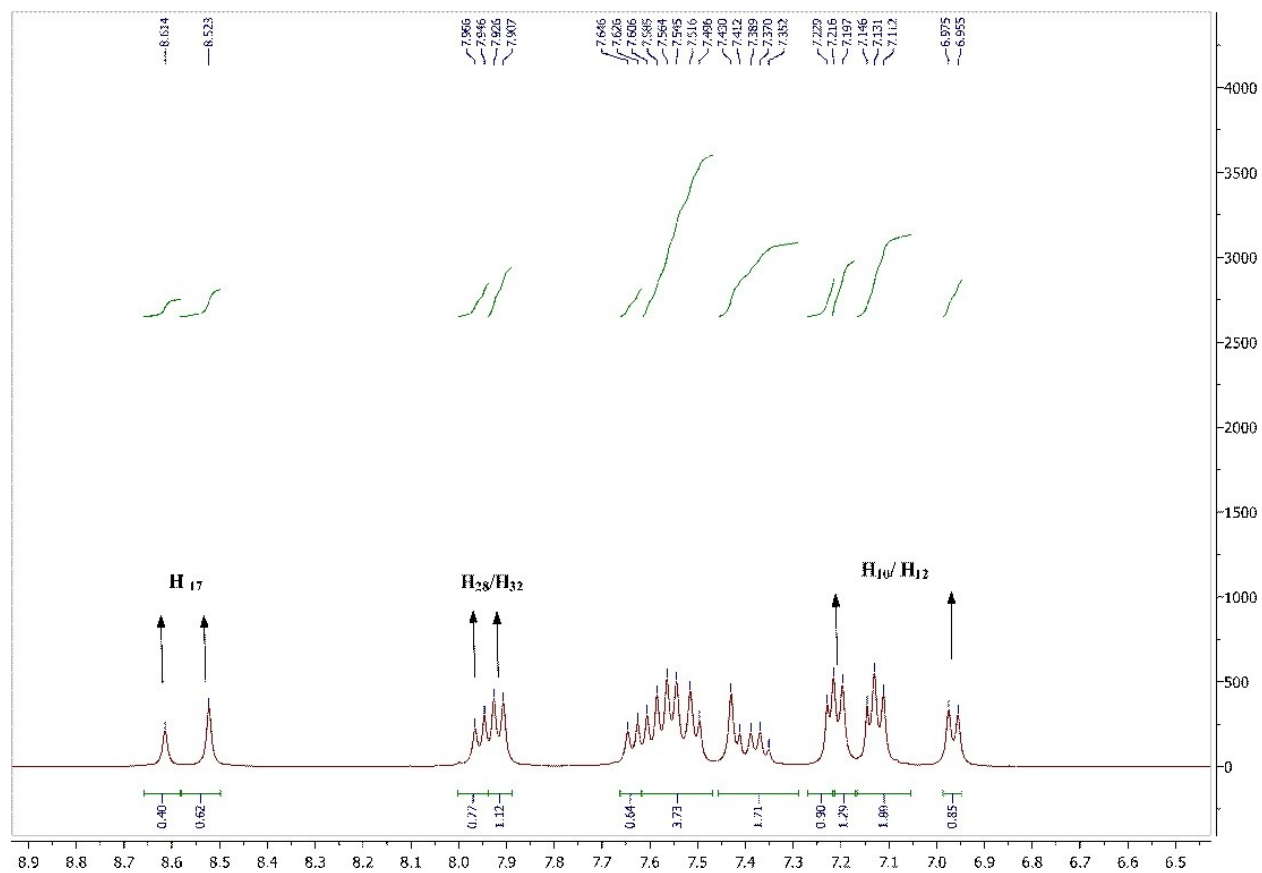
^1H NMR spectrum of 3



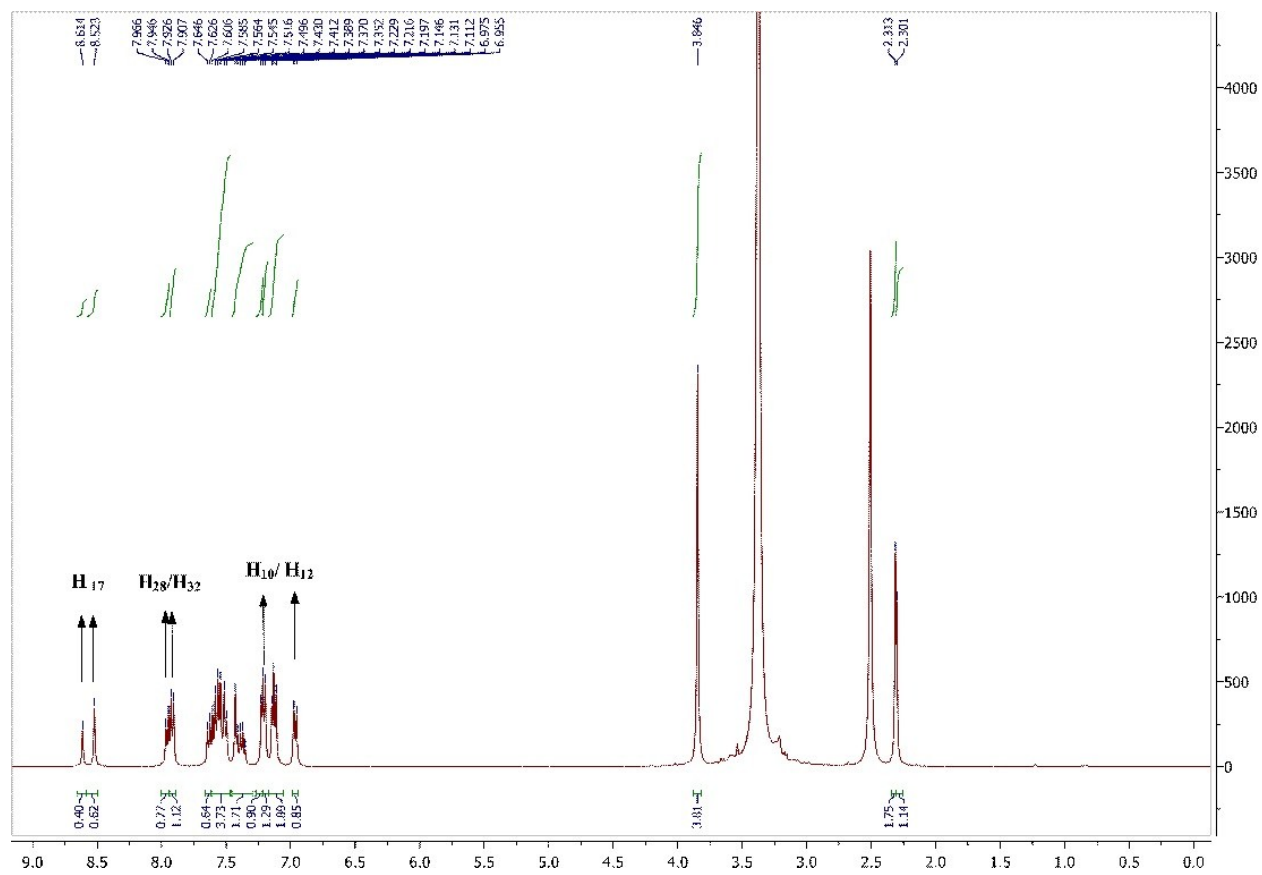
^1H NMR spectrum of **5a(1)**(DMSO)



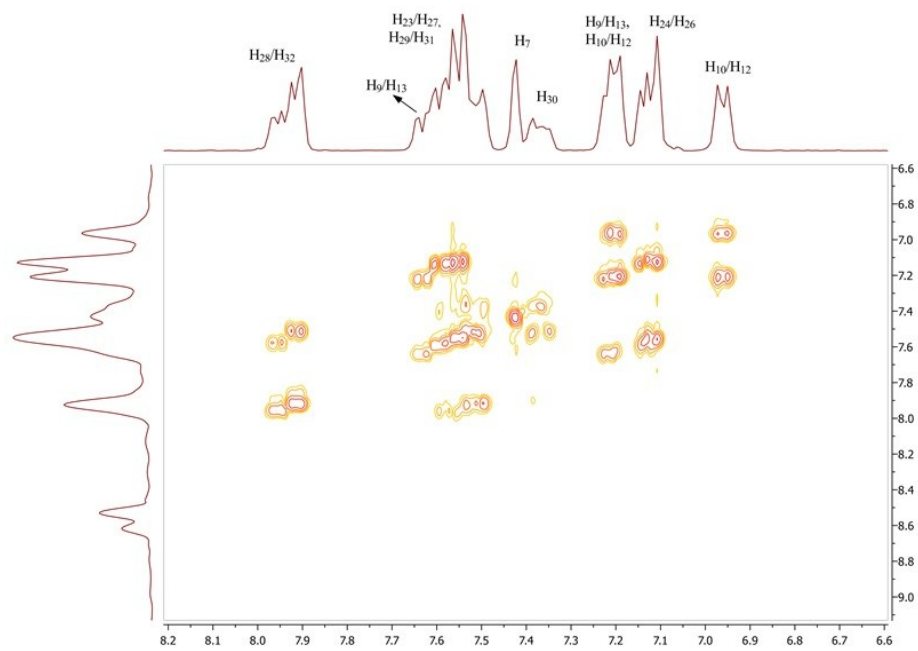
¹H NMR spectrum of **5a(2)**(DMSO)



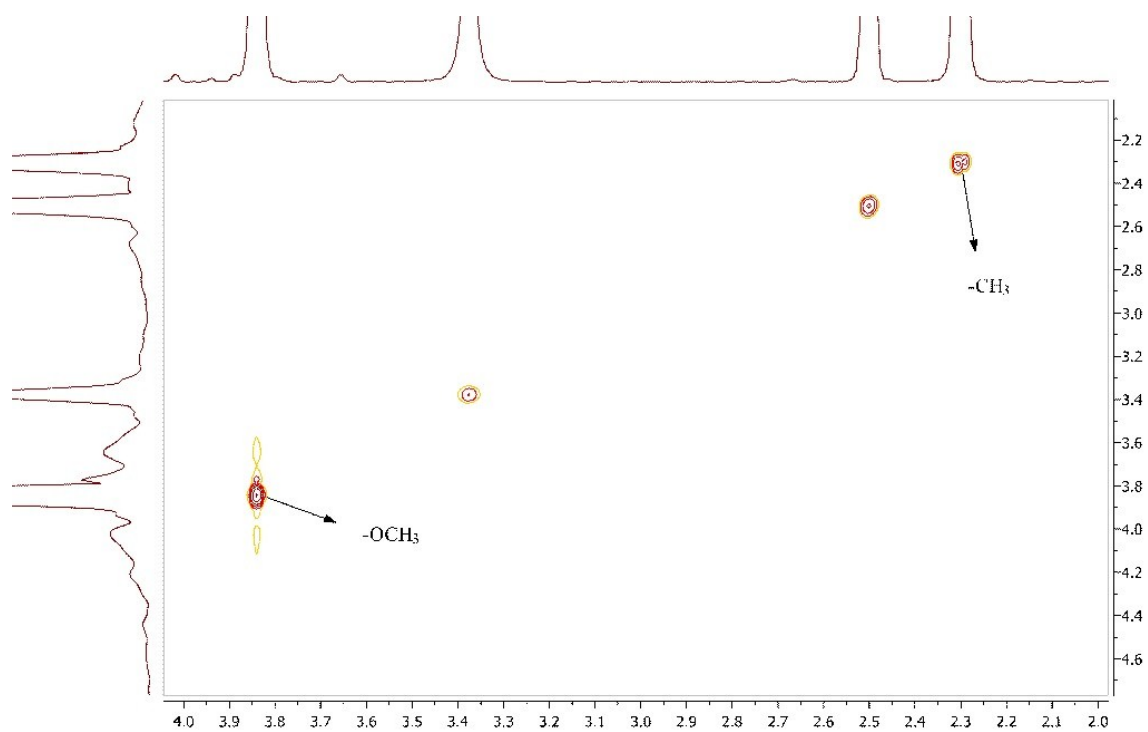
¹H NMR spectrum of 5a(3)(DMSO)



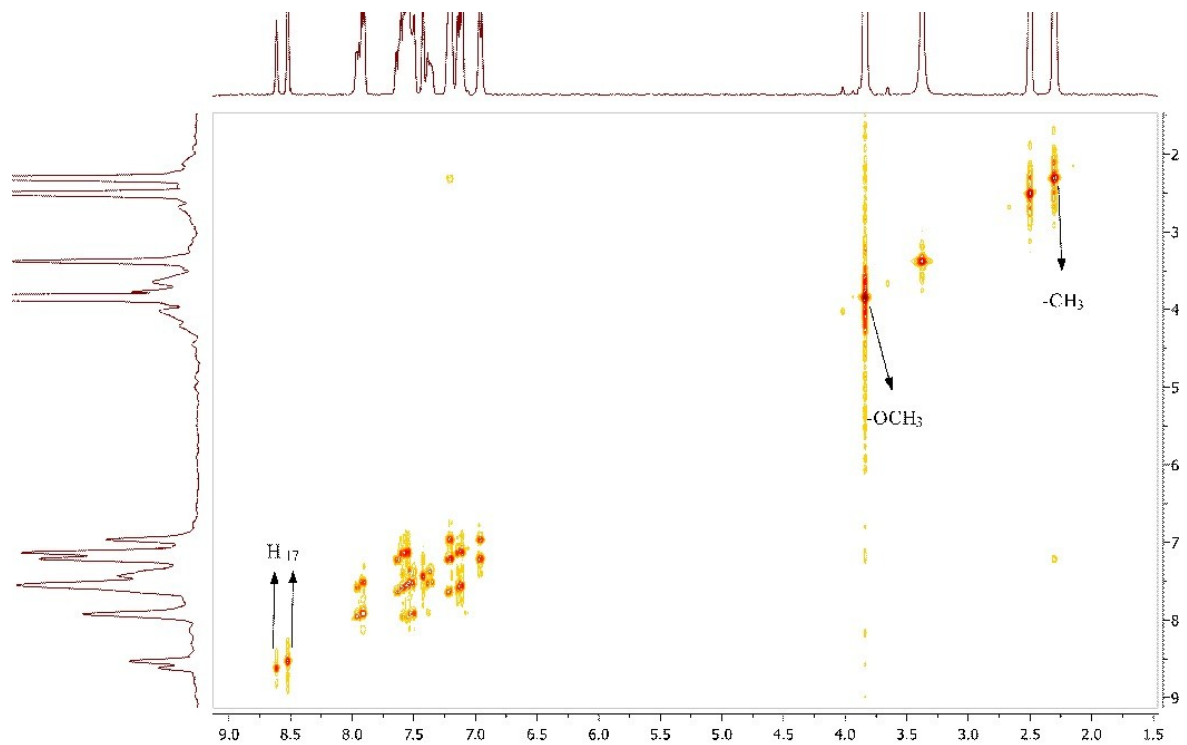
^1H - ^1H 2D-1 spectrum of 5a(DMSO)



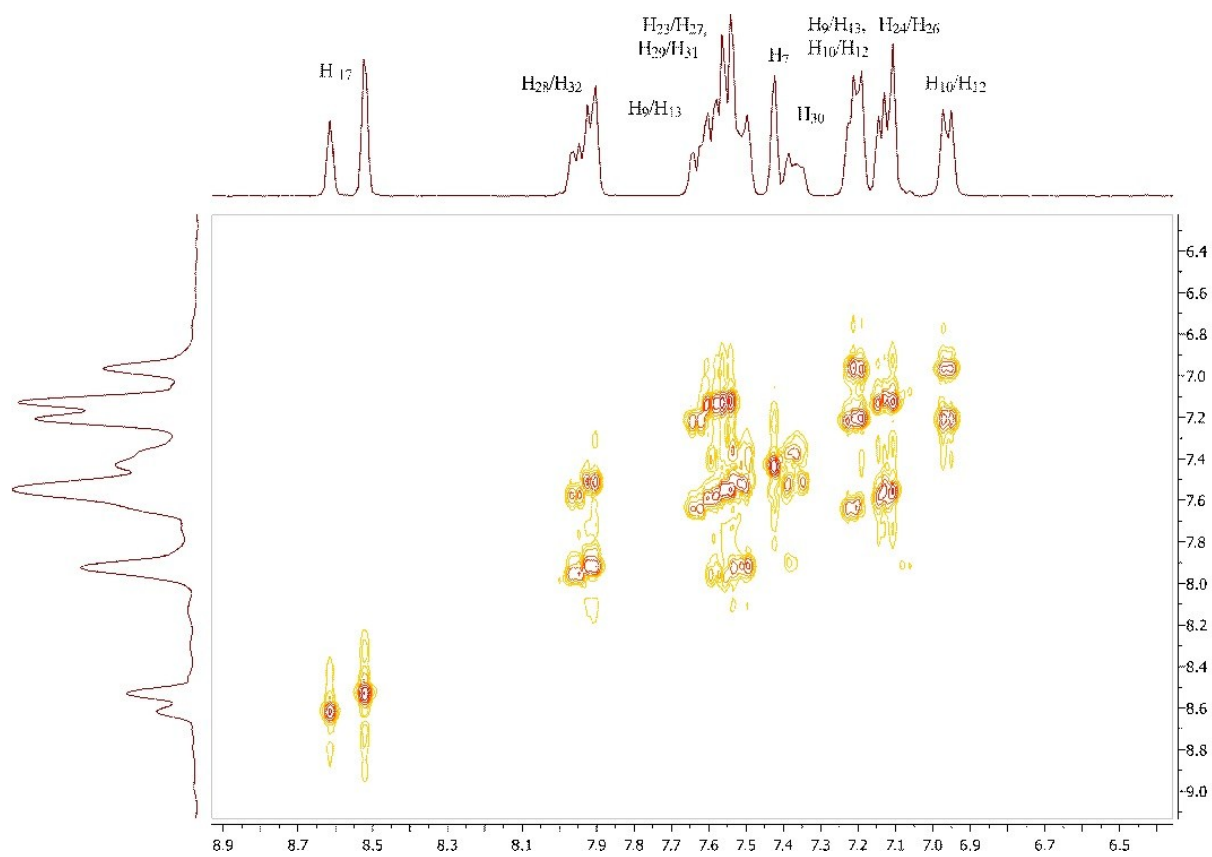
^1H - ^1H 2D-2 spectrum of 5a (DMSO)



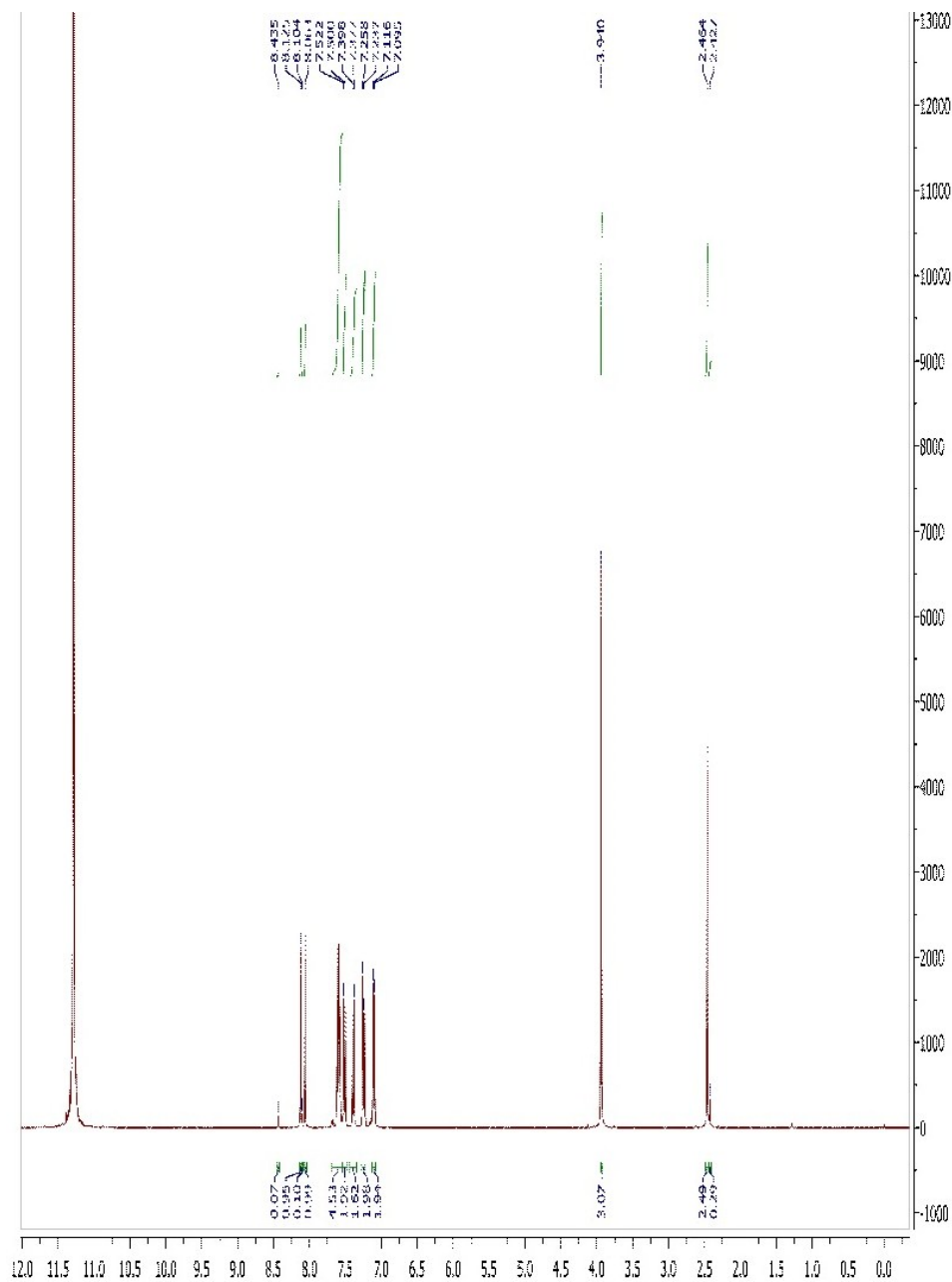
^1H - ^1H 2D-3 spectrum of 5a (DMSO)



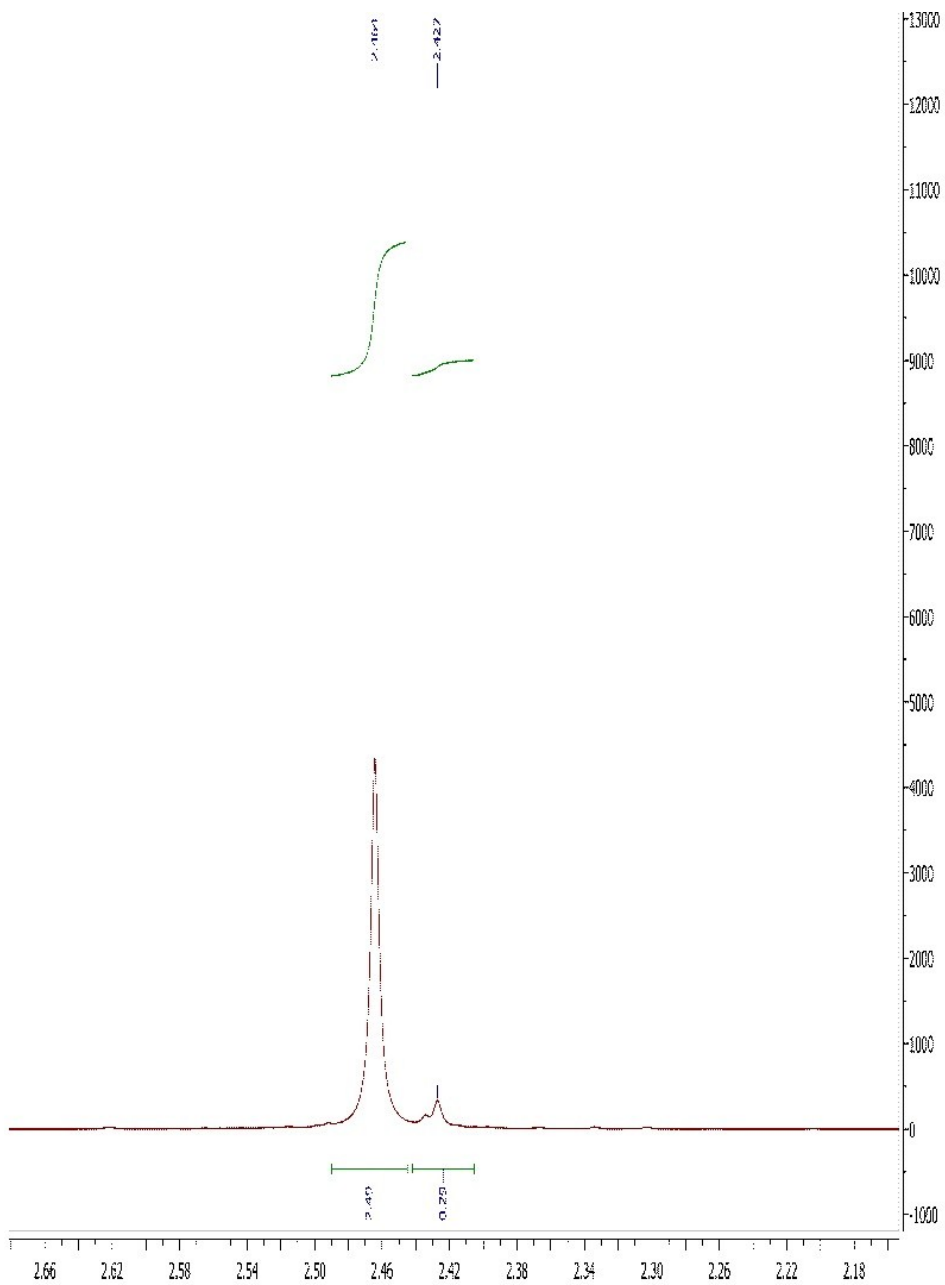
^1H - ^1H 2D-4 spectrum of 5a (DMSO)



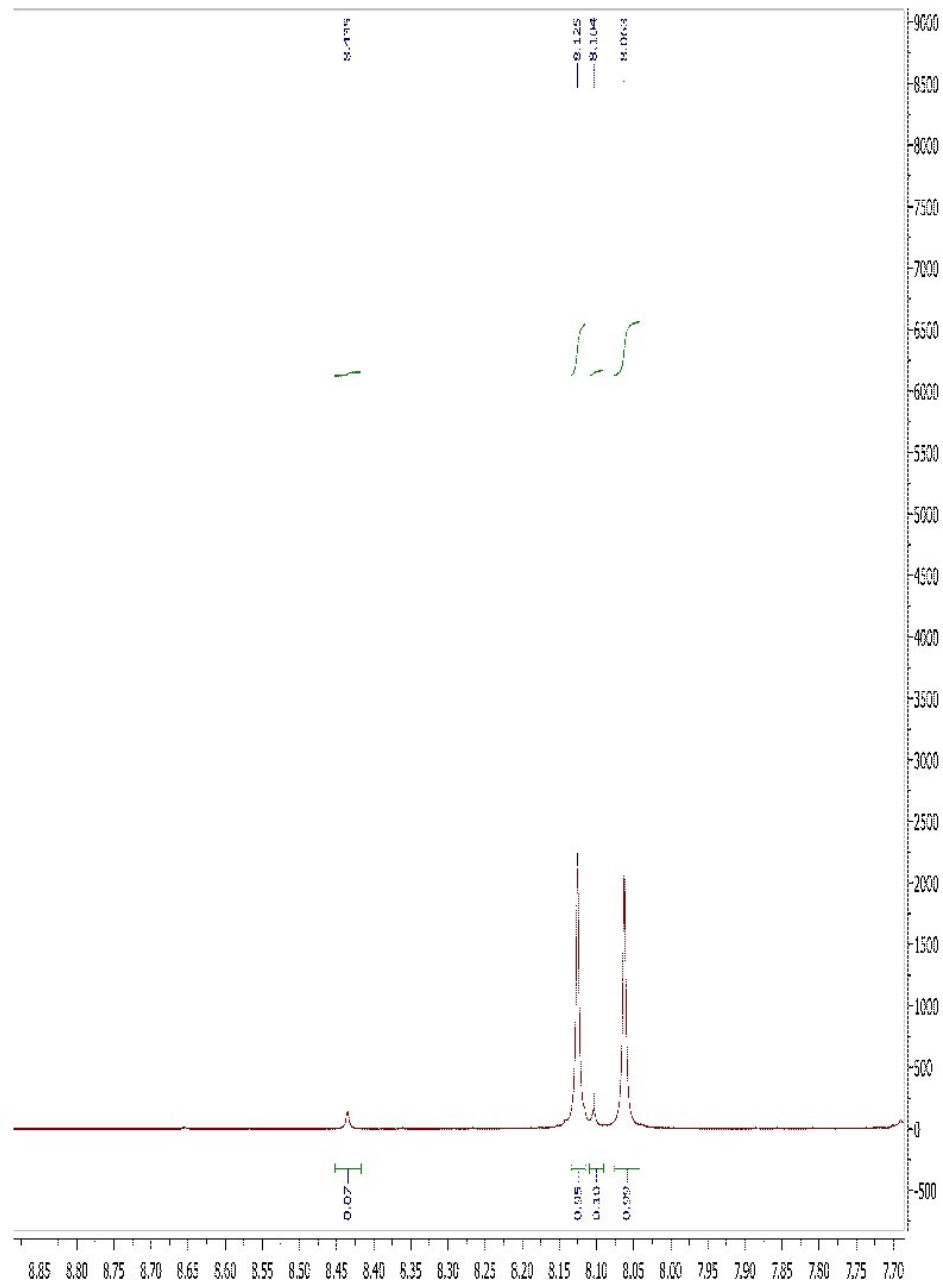
¹H NMR spectrum of 5a(1) (TFA)



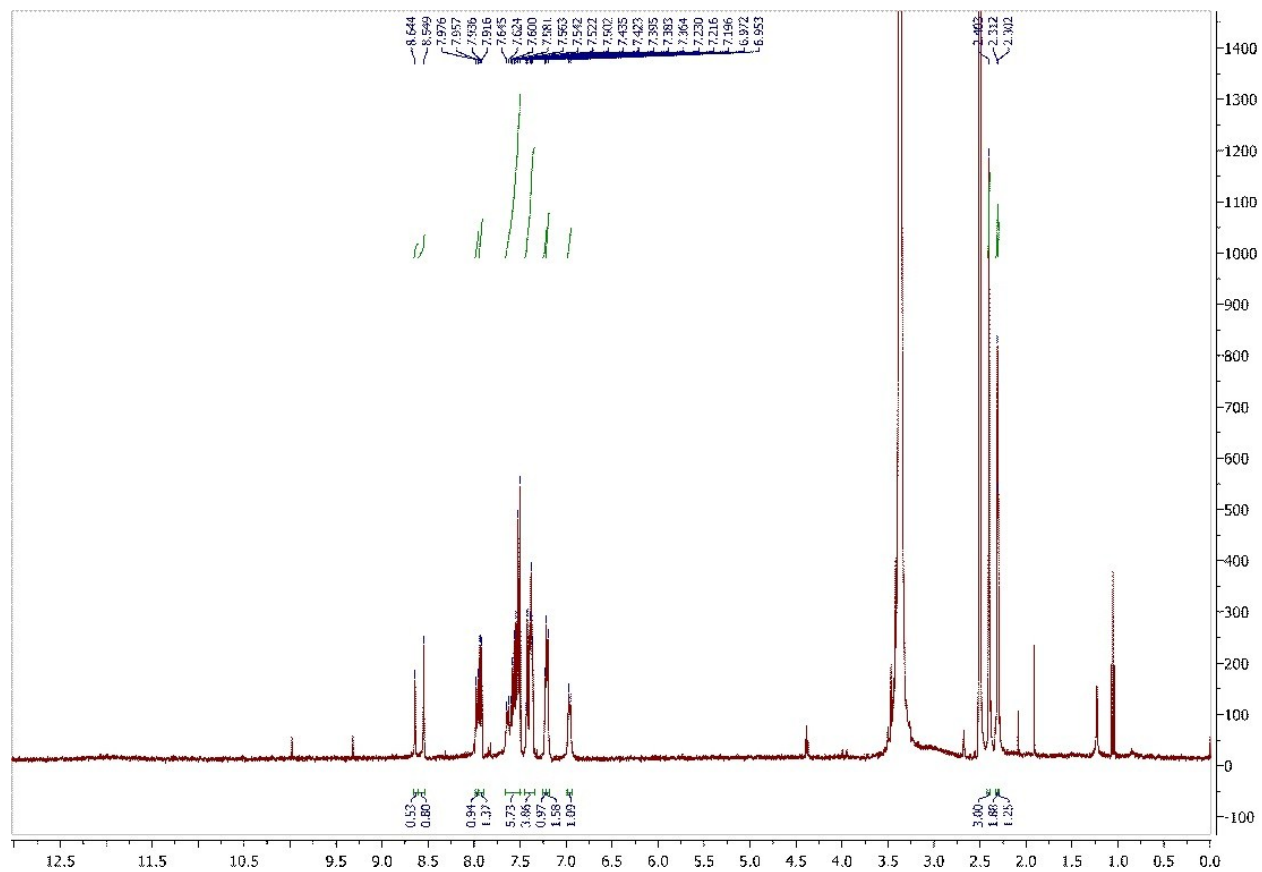
¹H NMR spectrum of 5a(2) (TFA)



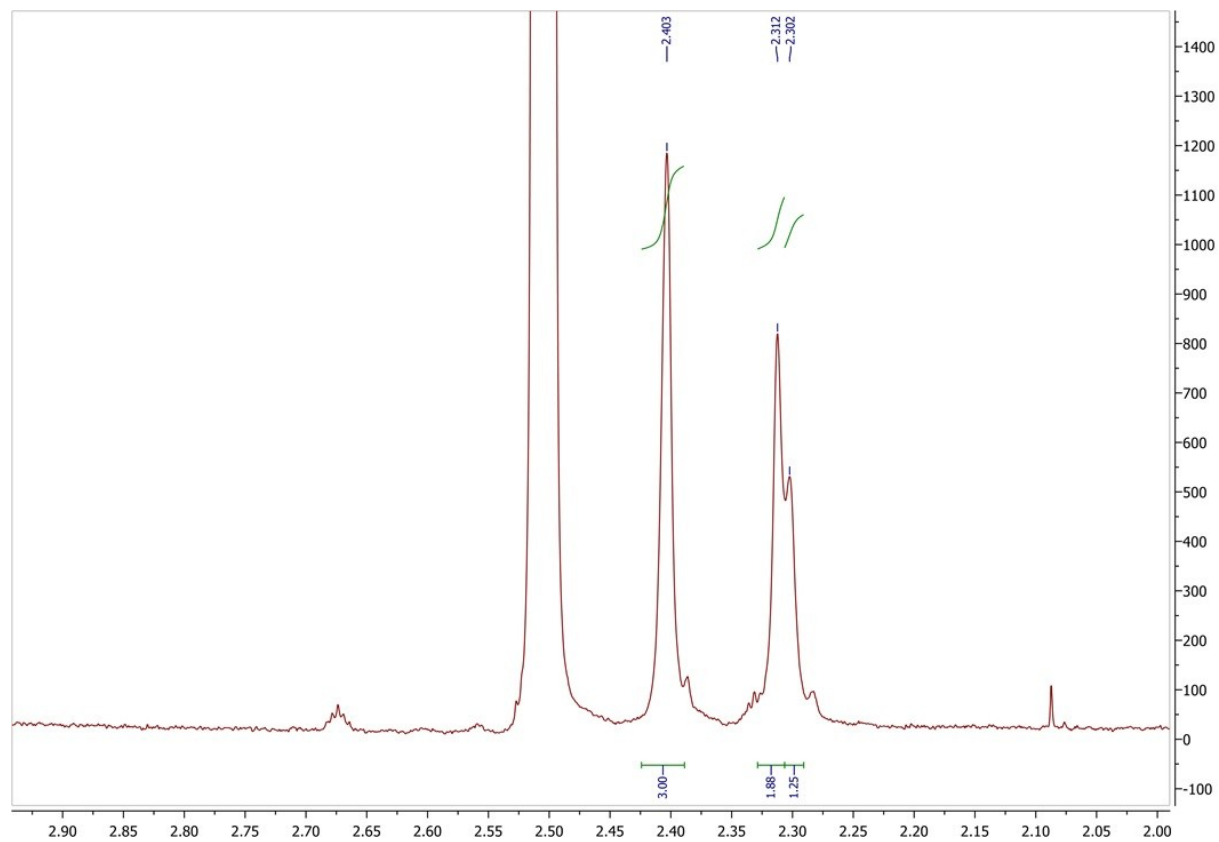
¹H NMR spectrum of 5a(3) (TFA)



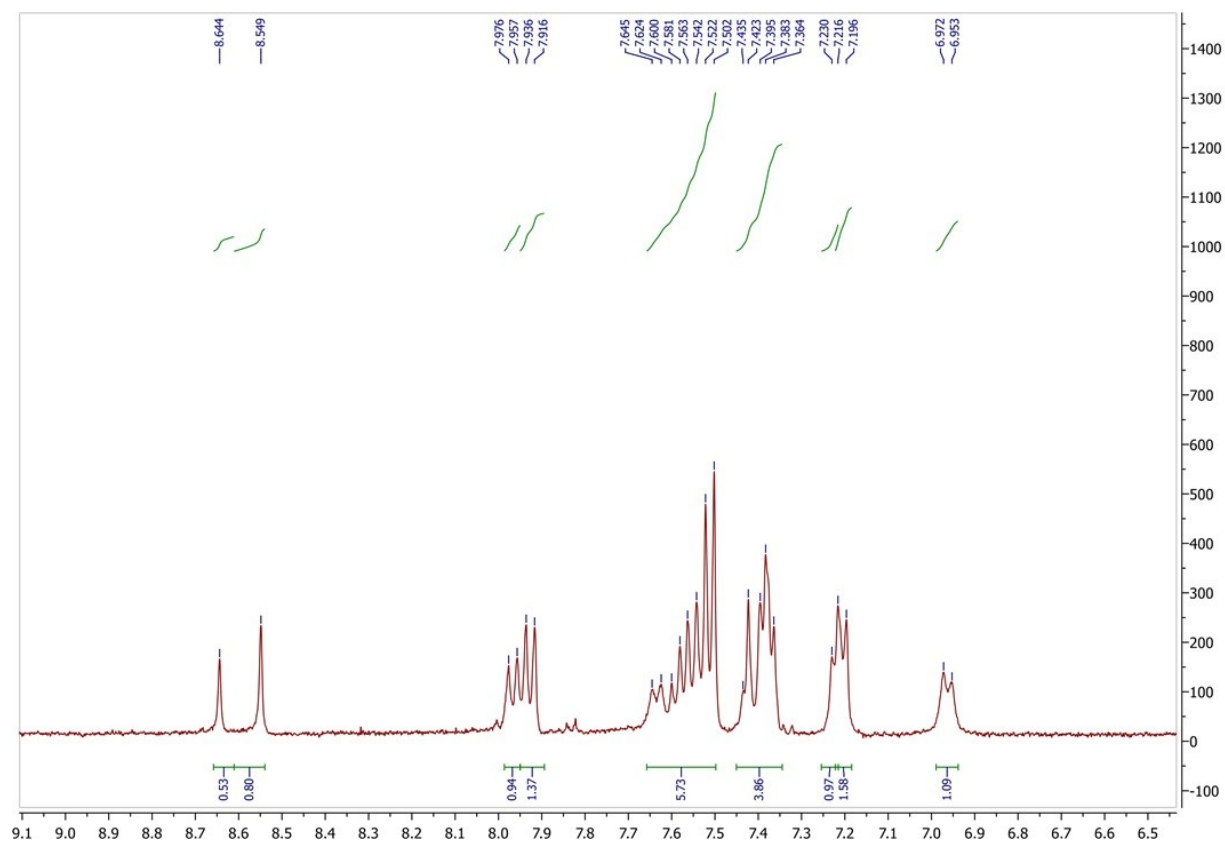
¹H NMR spectrum of 5b(1)



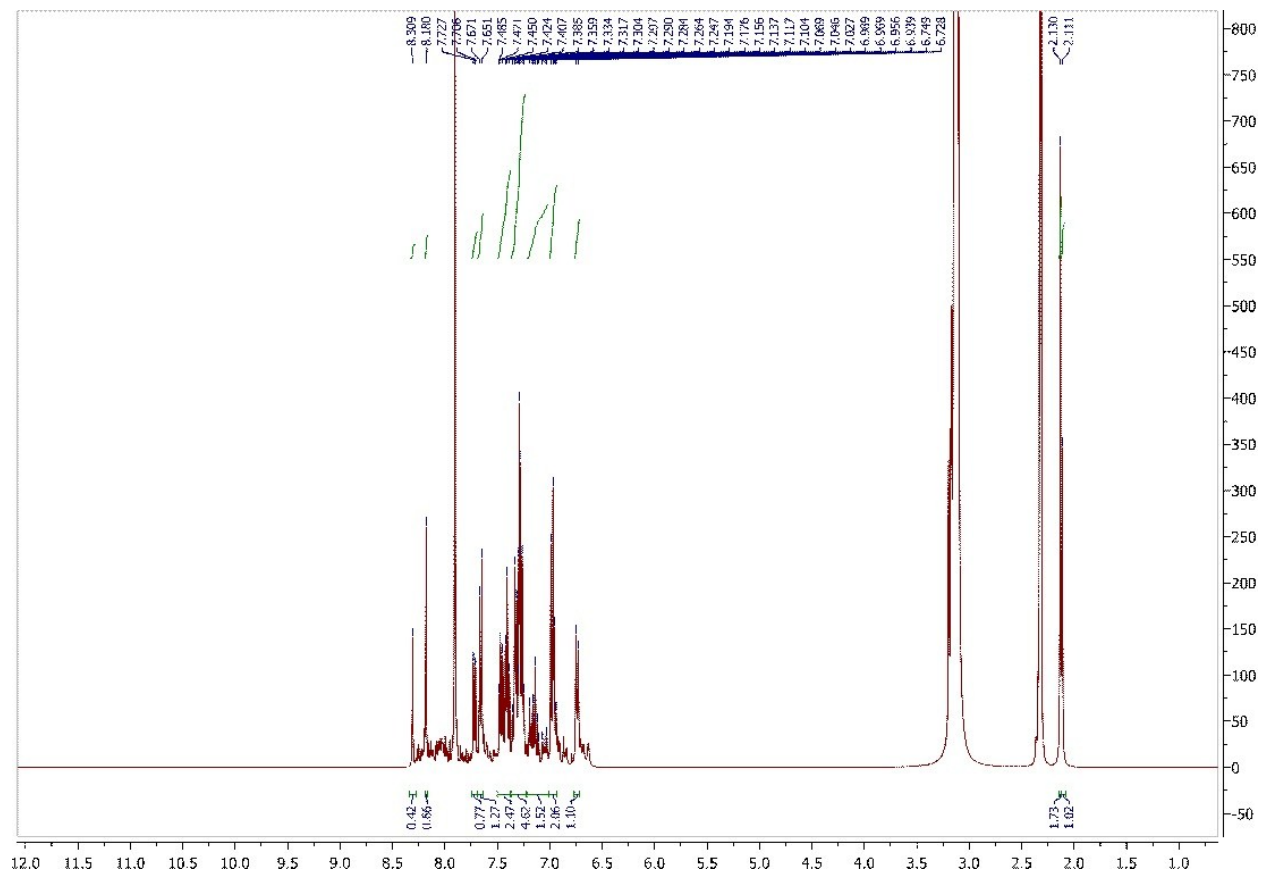
¹H NMR spectrum of 5b(2)



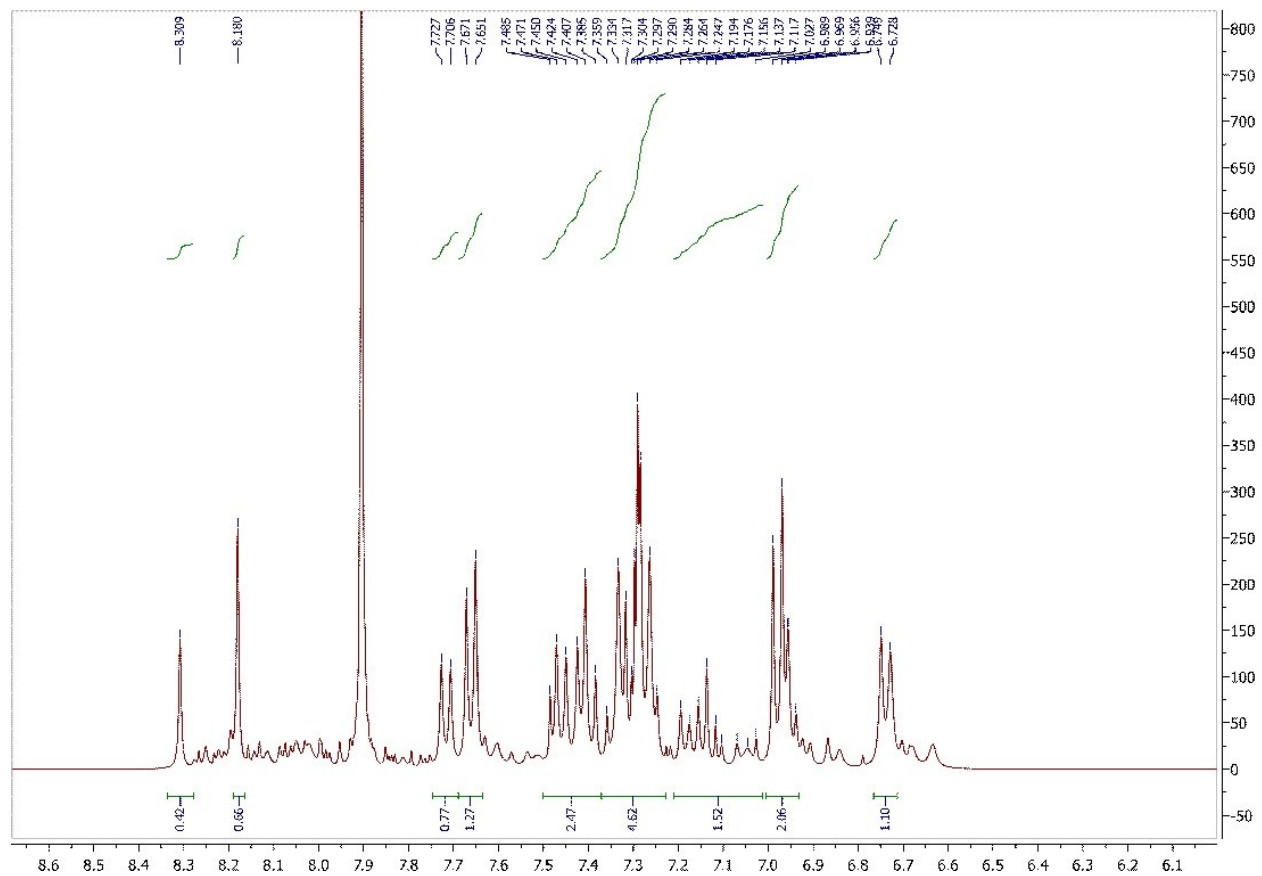
¹H NMR spectrum of 5b(3)



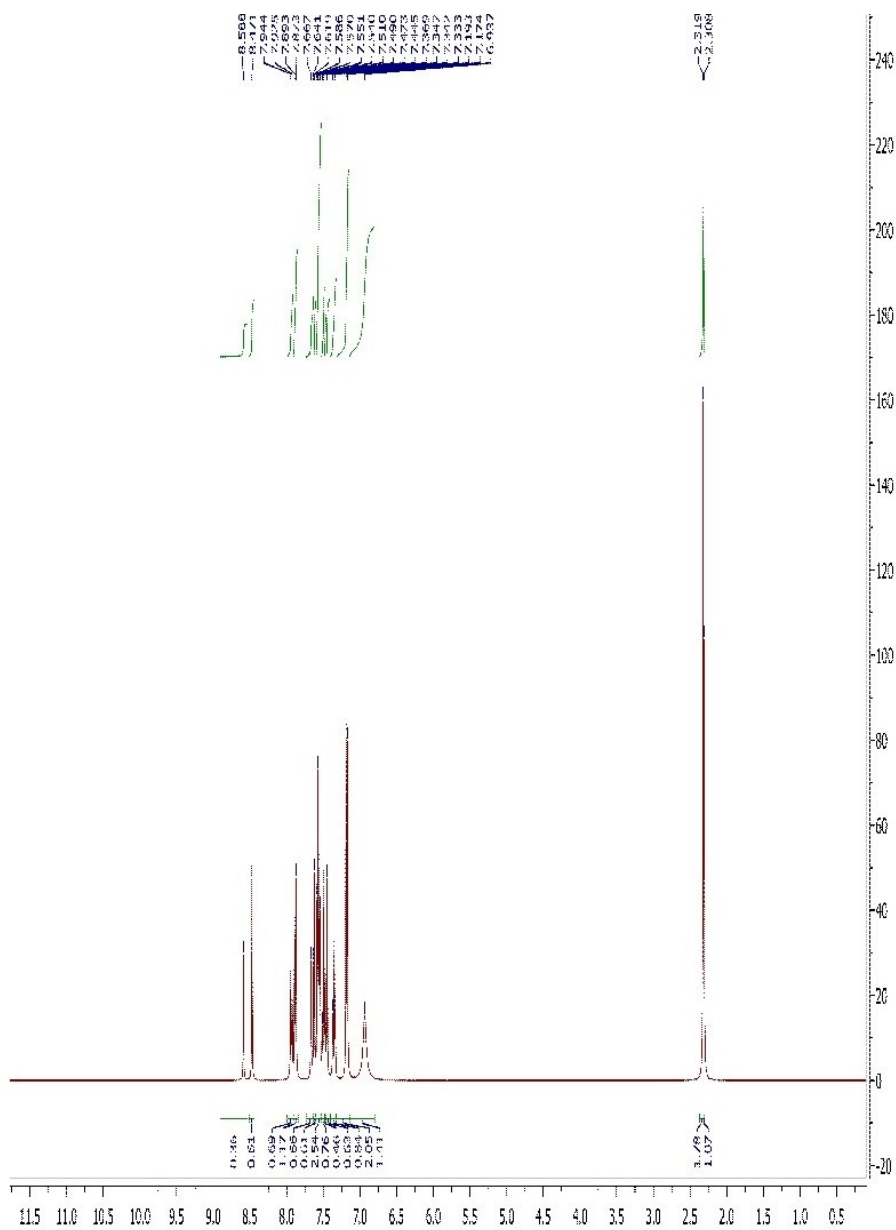
¹H NMR spectrum of 5c(1)



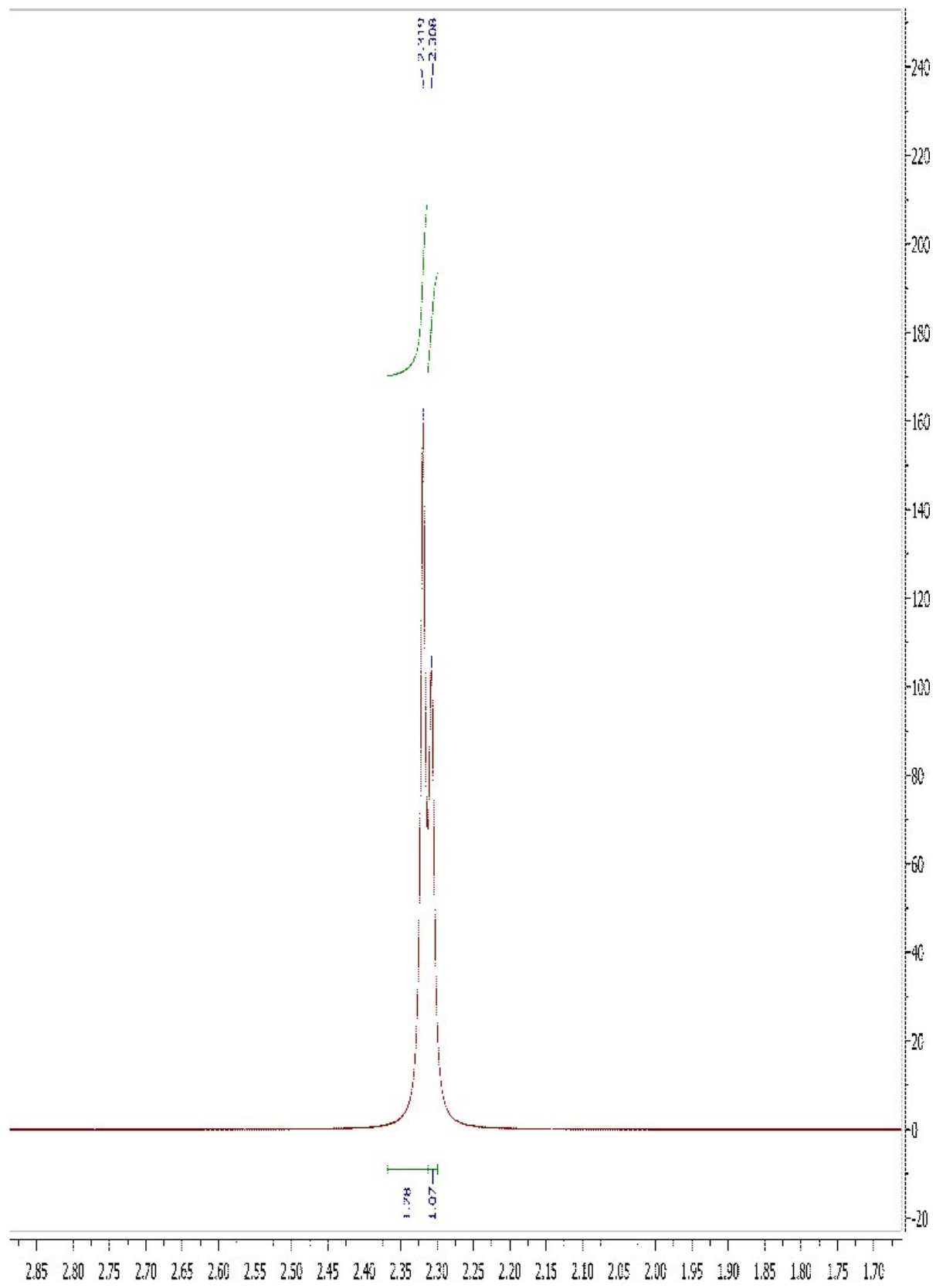
¹H NMR spectrum of 5c(2)



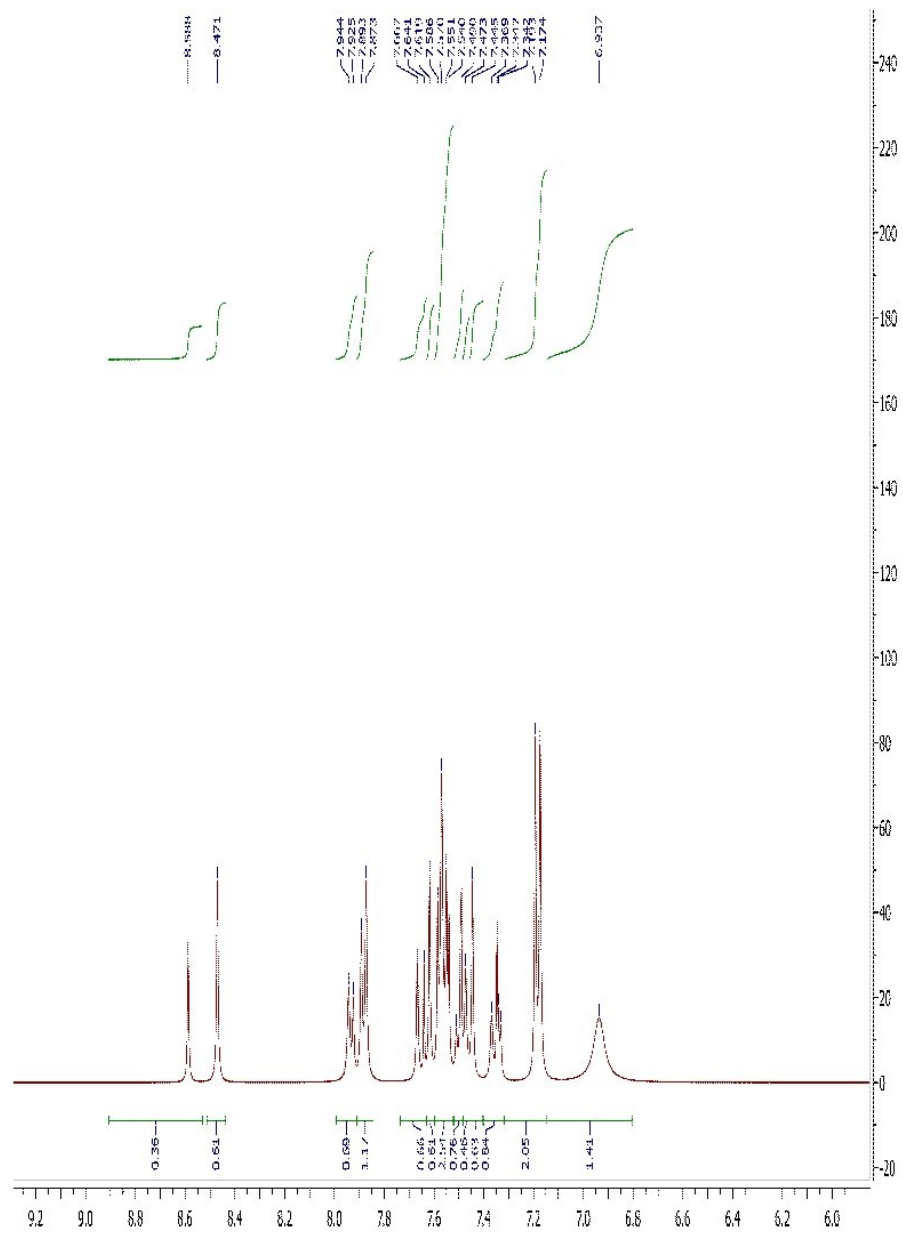
¹H NMR spectrum of 5d(1)



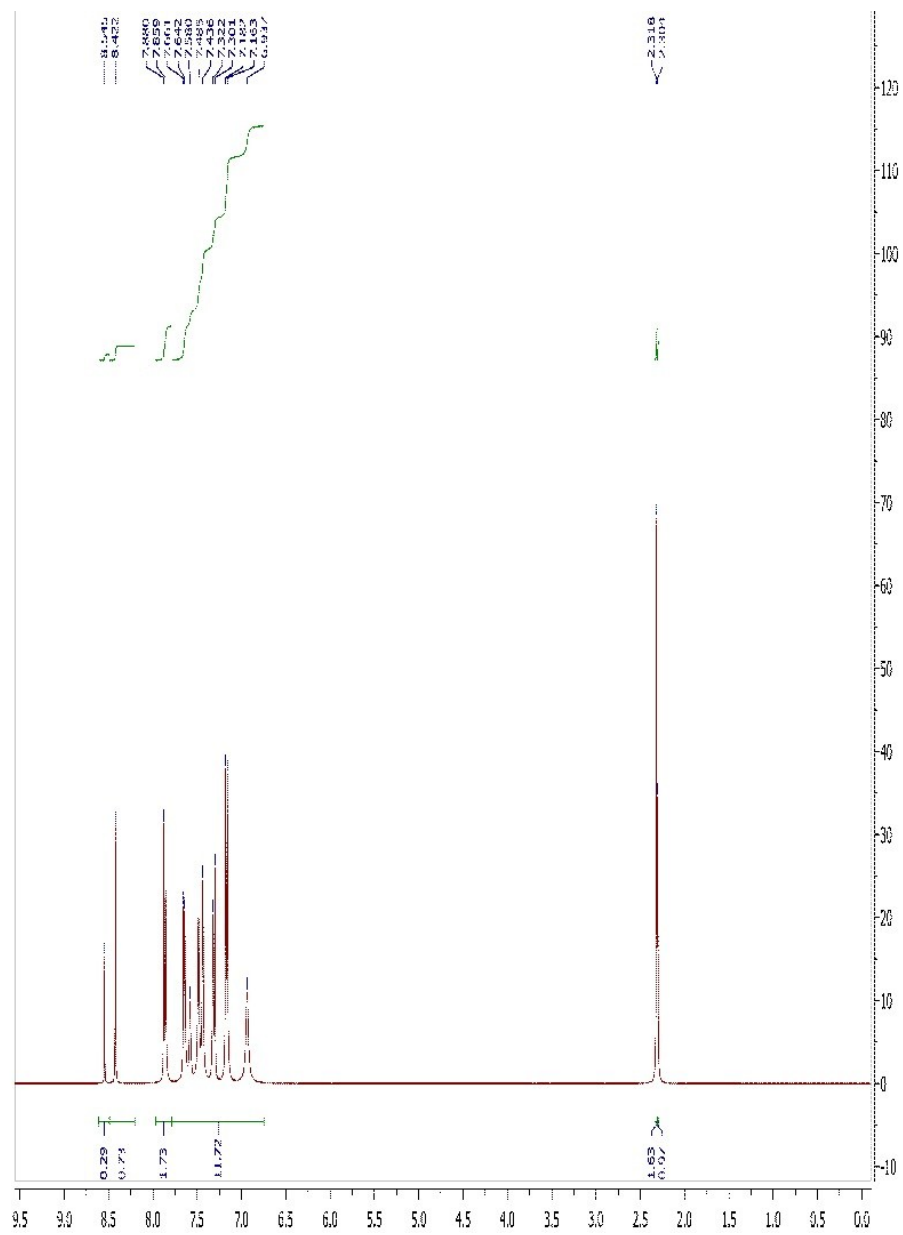
NMR spectrum of 5d(2)



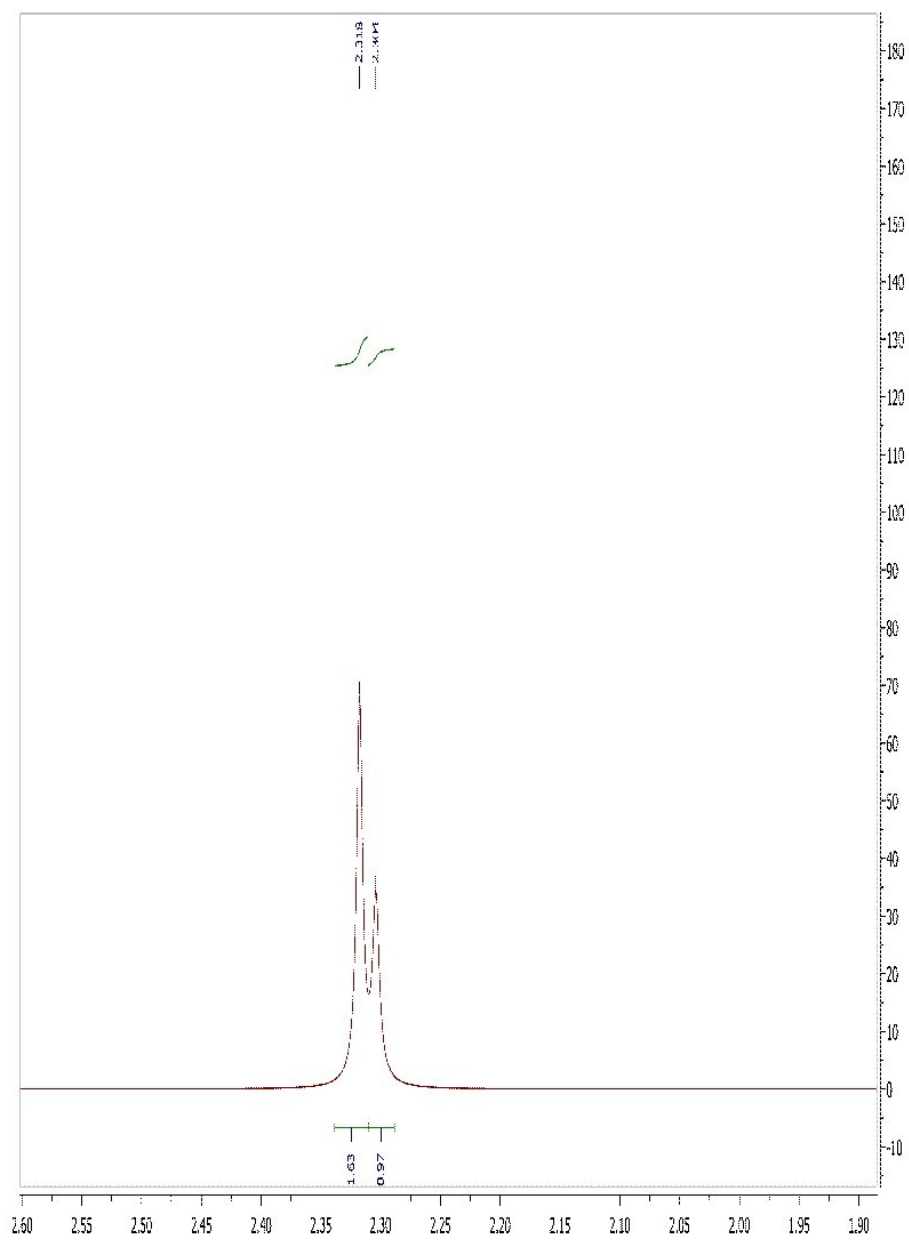
^1H NMR spectrum of 5d(3)



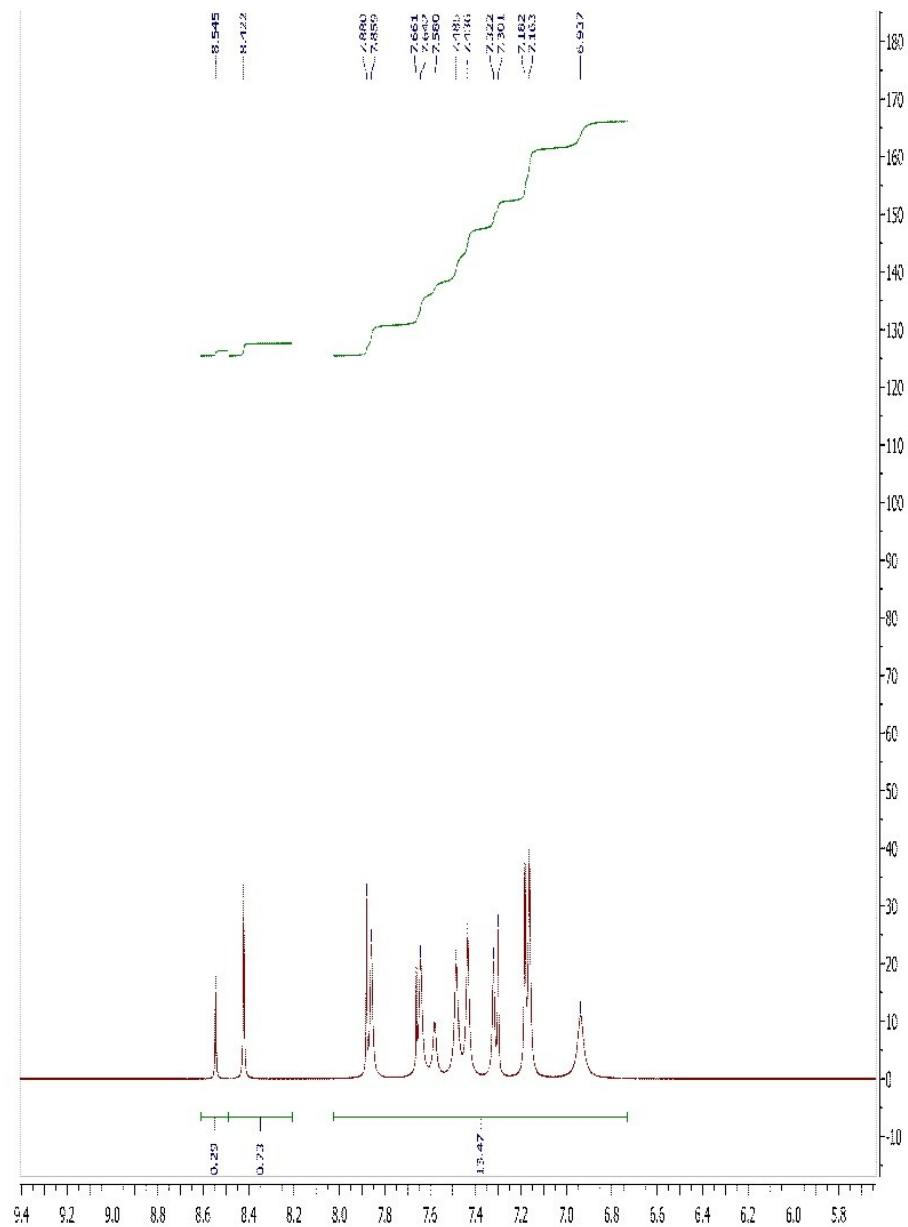
^1H NMR spectrum of 5e(1)



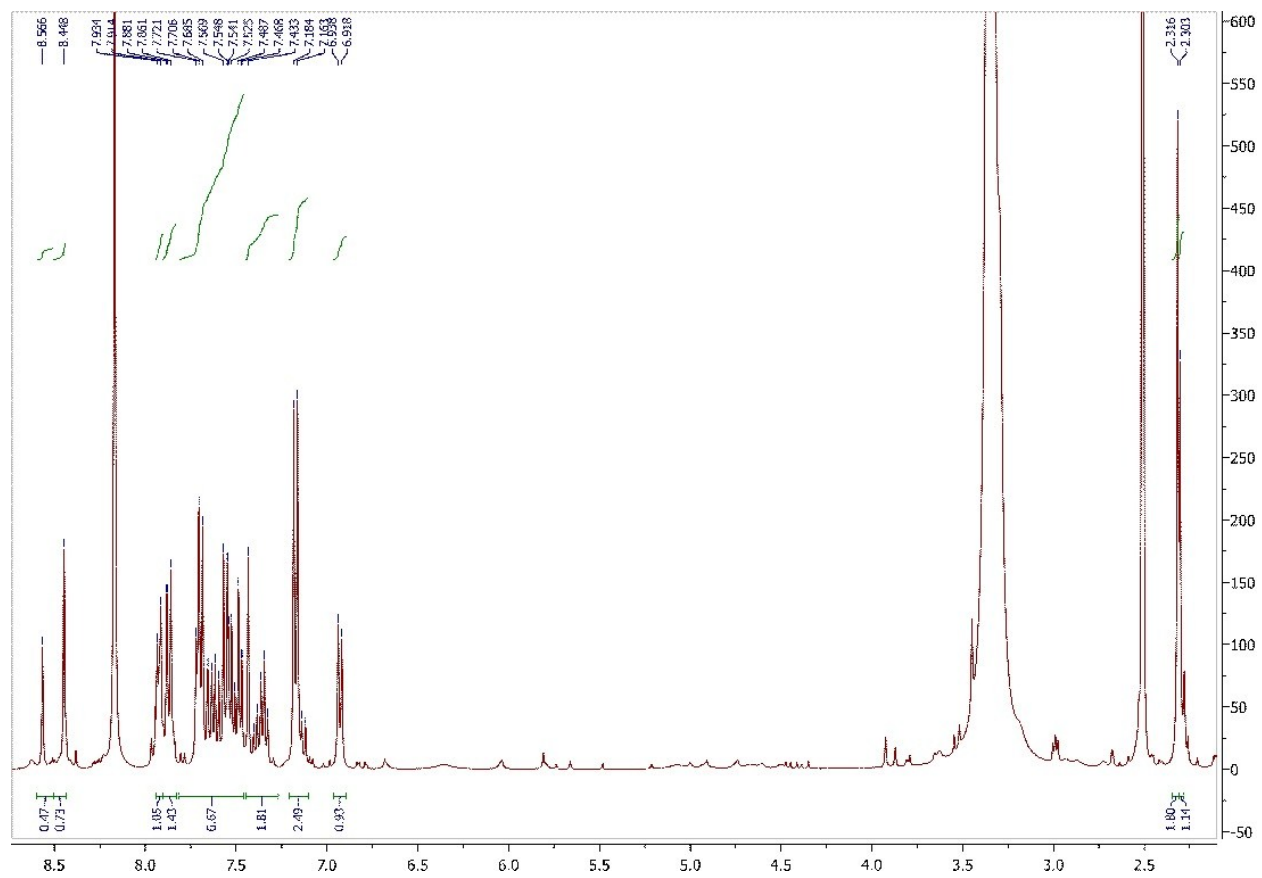
^1H NMR spectrum of 5e(2)



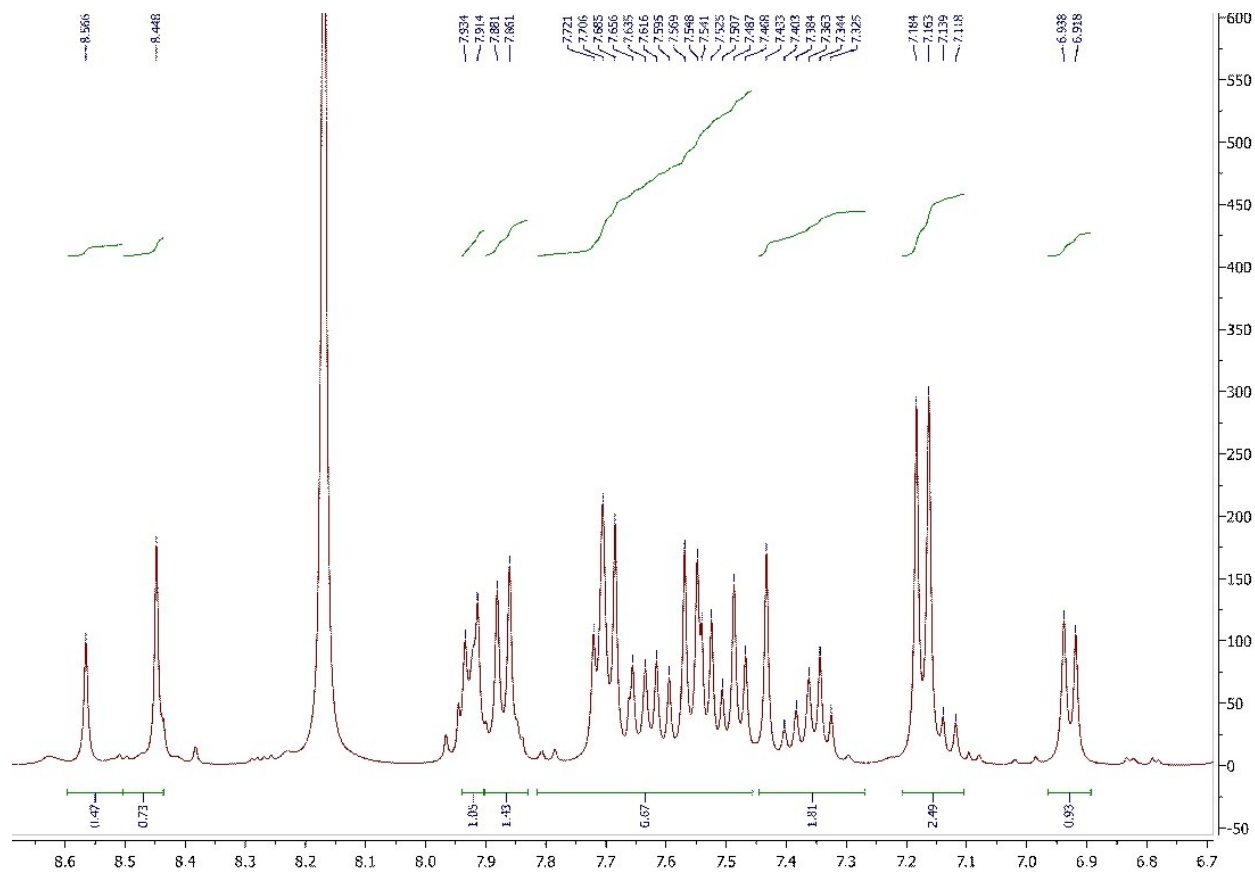
^1H NMR spectrum of 5e(3)



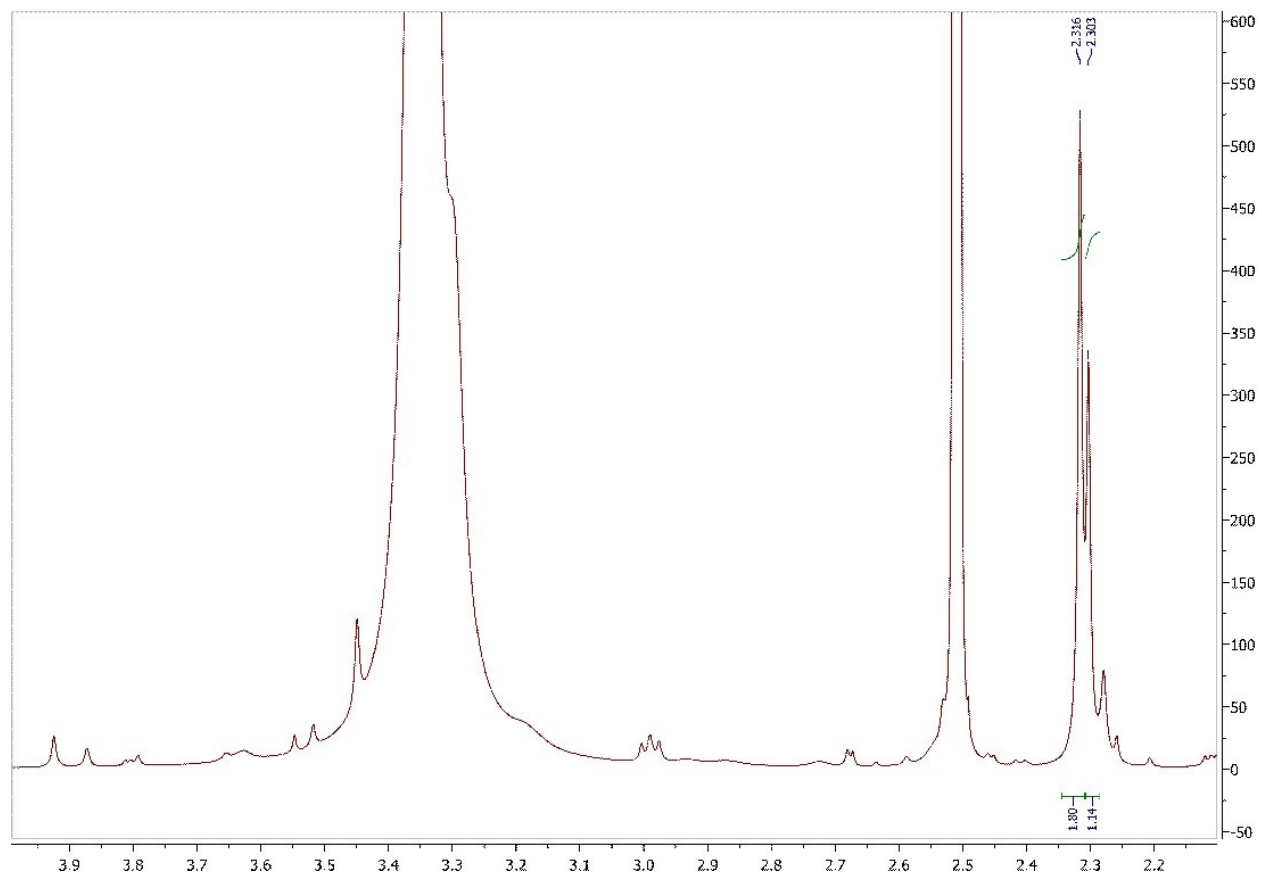
^1H NMR spectrum of 5f(1)



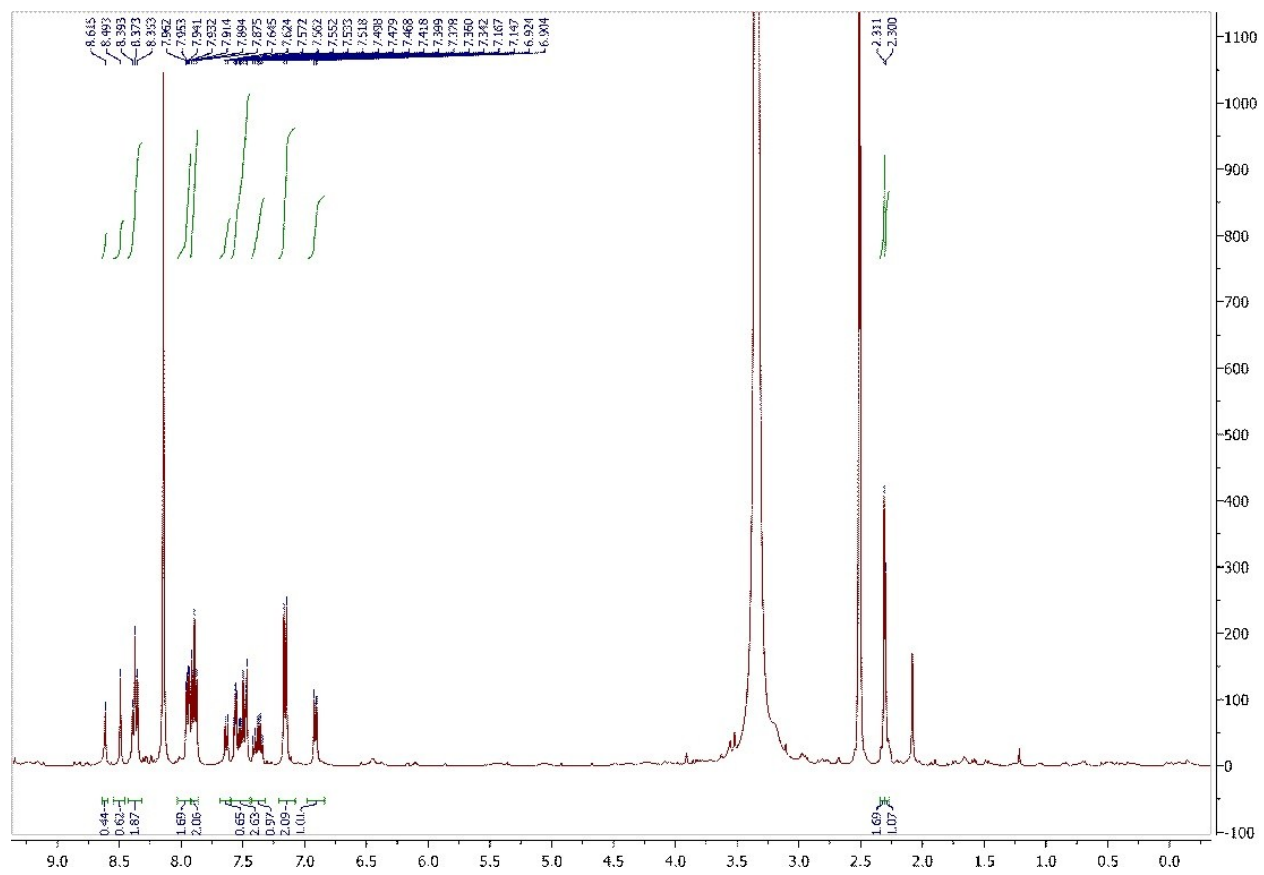
^1H NMR spectrum of 5f(2)



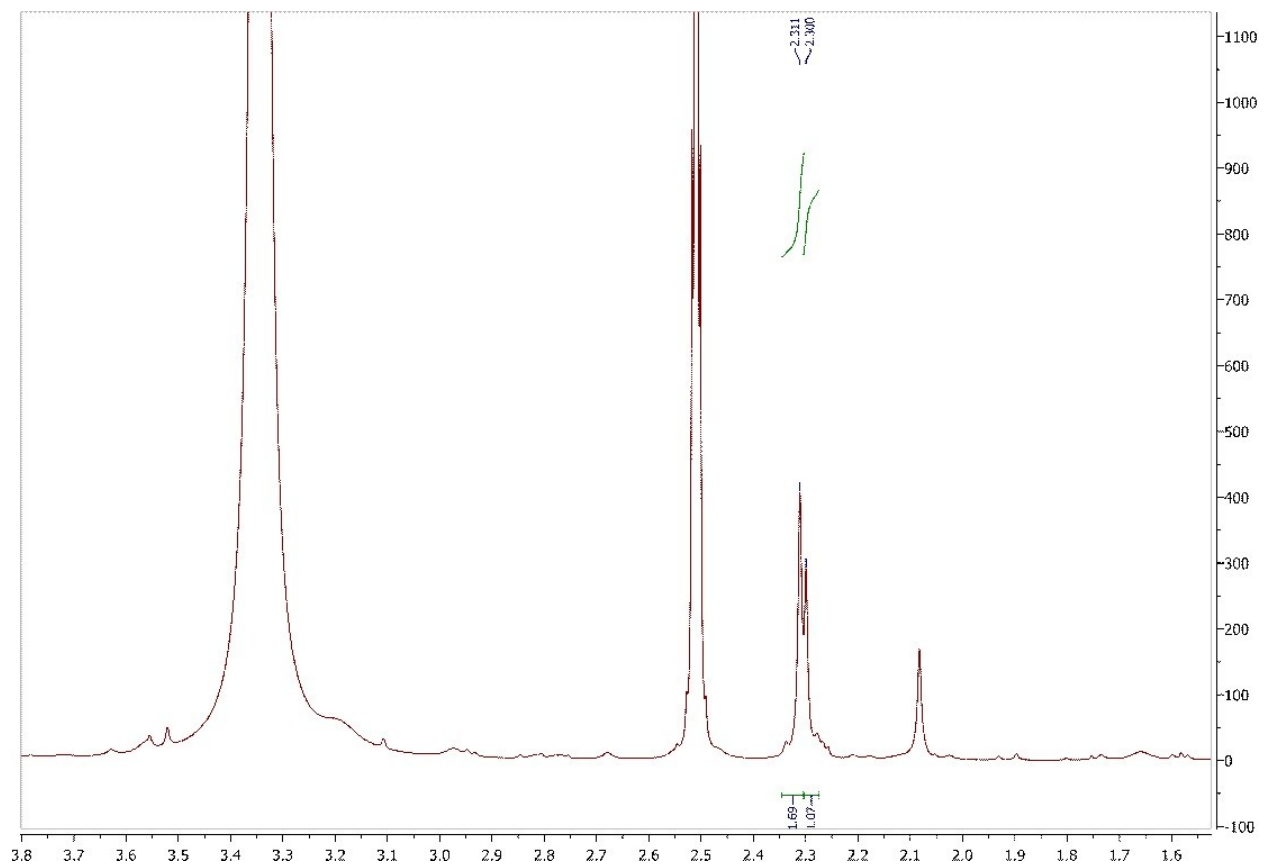
¹H NMR spectrum of 5f(3)



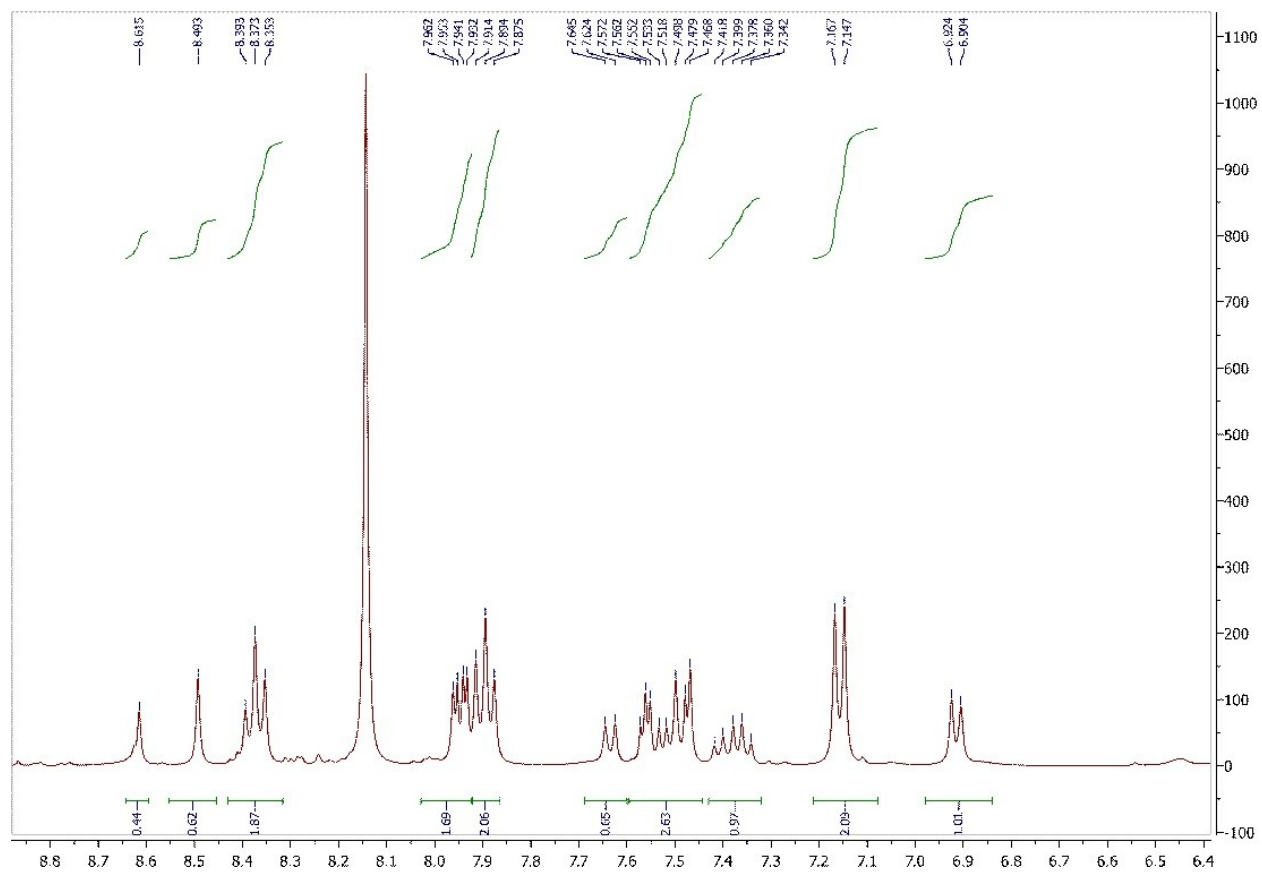
^1H NMR spectrum of 5g(1)



¹H NMR spectrum of 5g(2)



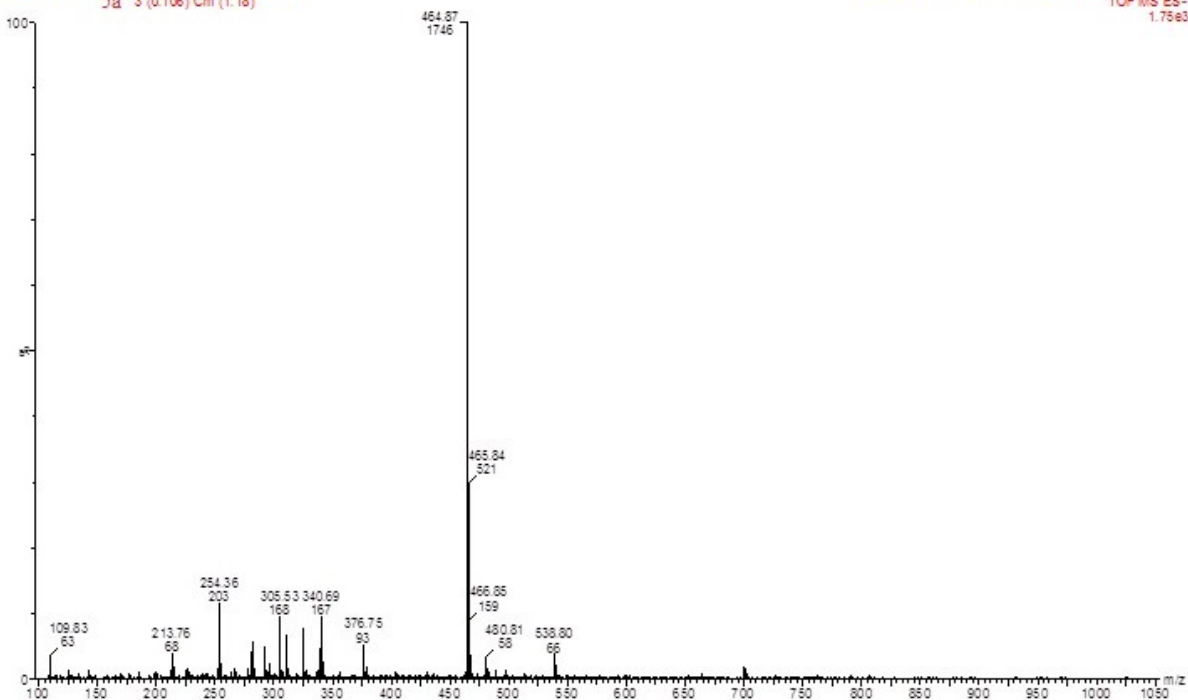
^1H NMR spectrum of 5g(3)



ESI-MS (m/z) of 5a(1)

WATERS, Q-TOF MICROMASS (ESI-MS)
5a 3 (0.106) Cm (1:18)

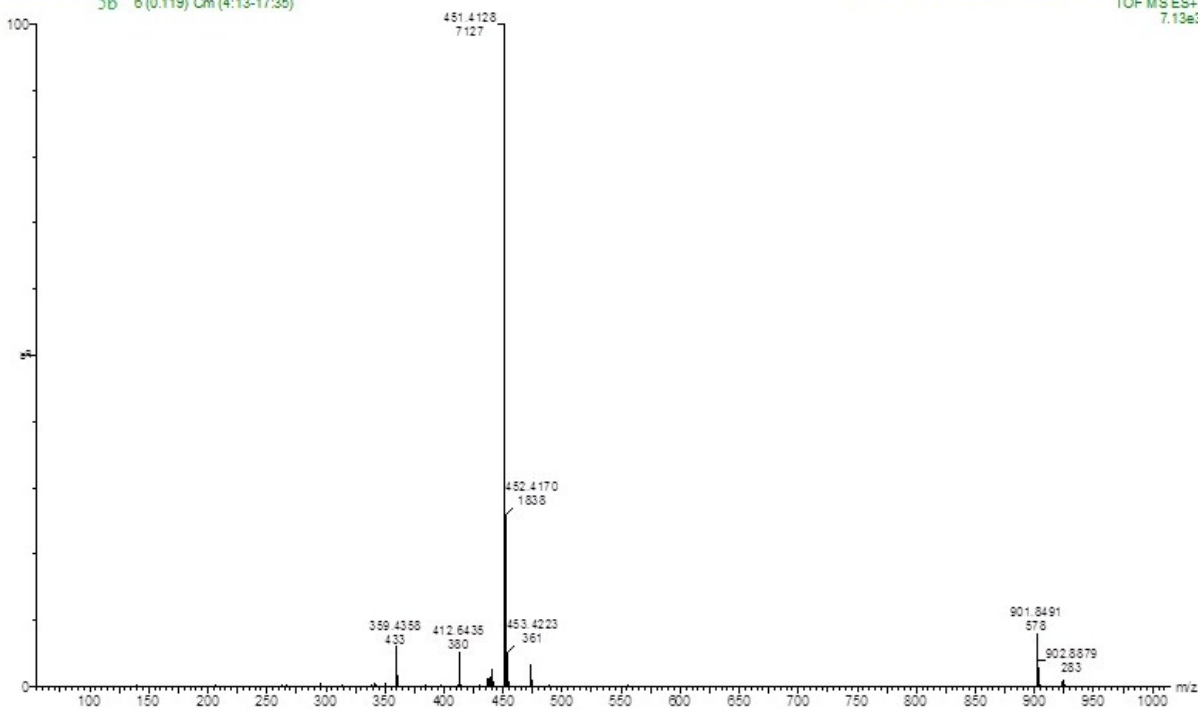
SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES-
1.75e3



ESI-MS (m/z) of 5b(1)

WATERS, Q-TOF MICROMASS (ESI-MS)
5b 6 (0.119) Cm (4:13-17:35)

SAIF/CIL,PANJAB UNIVERSITY,CHANDIGARH
TOF MS ES+
7.13e3



Reference

1. H. Zhou, S. Wu, S. Zhai, A. Liu, Y. Sun, R. Li, Y. Zhang, S. Ekins, P. W. Swaan, B. Fang, B. Zhang and B. Yan, Design, *J. Med. Chem.*, 2008, **51**, 1242-1251.