



Supporting Information

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The Alkaline Stability of Anion Exchange Membrane for Fuel Cell Applications: The Effects of Alkaline Media

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Supporting Information

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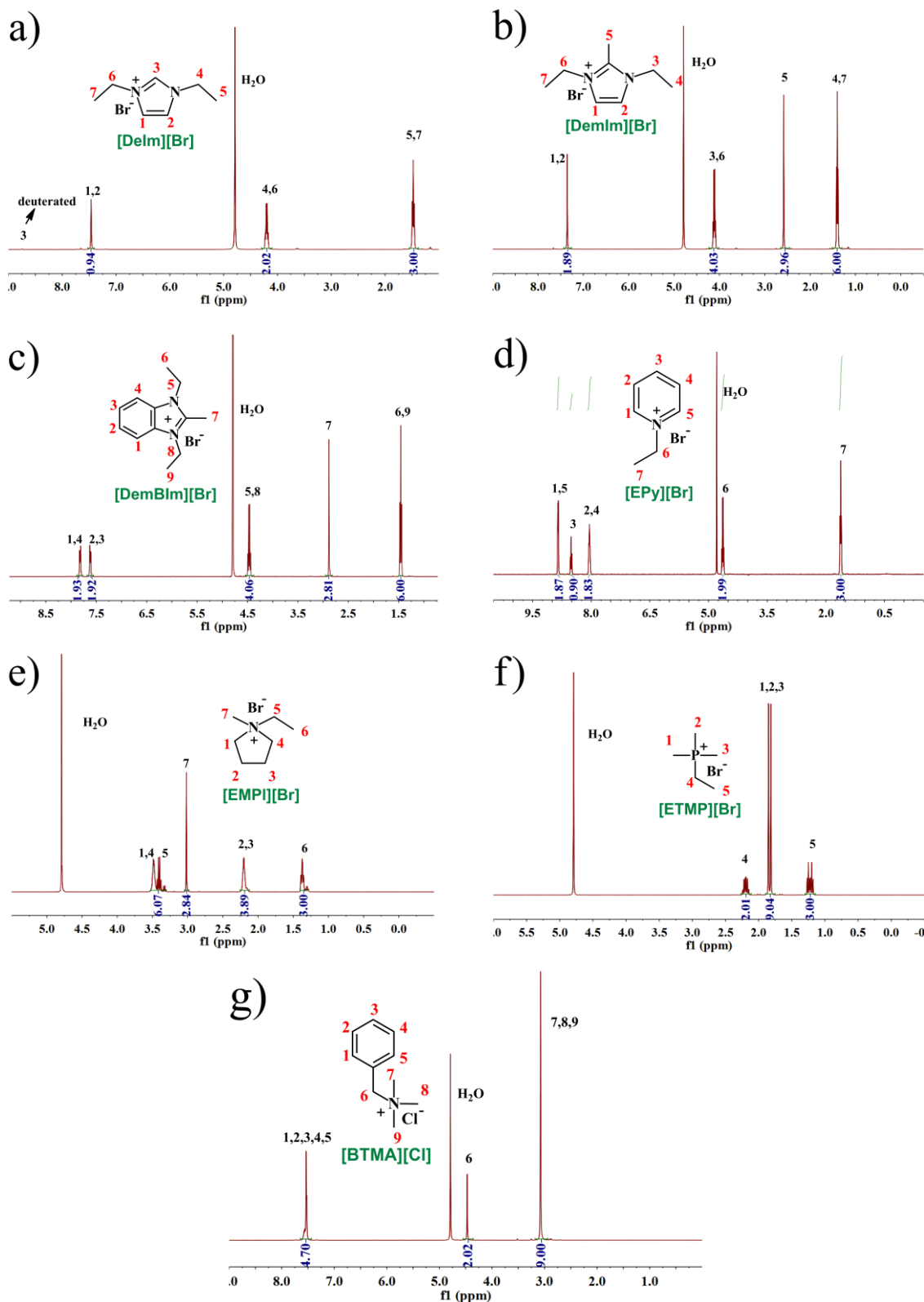


Figure S1. ^1H NMR spectra for various organic cations studied in this work including: (a) [DeIm][Br], (b) [DemIm][Br], (c) [DemBIm][Br], (d) [EPy][Br], (e) [EMPI][Br], (f) [ETMP][Br] and (g) [BTMA][Cl].

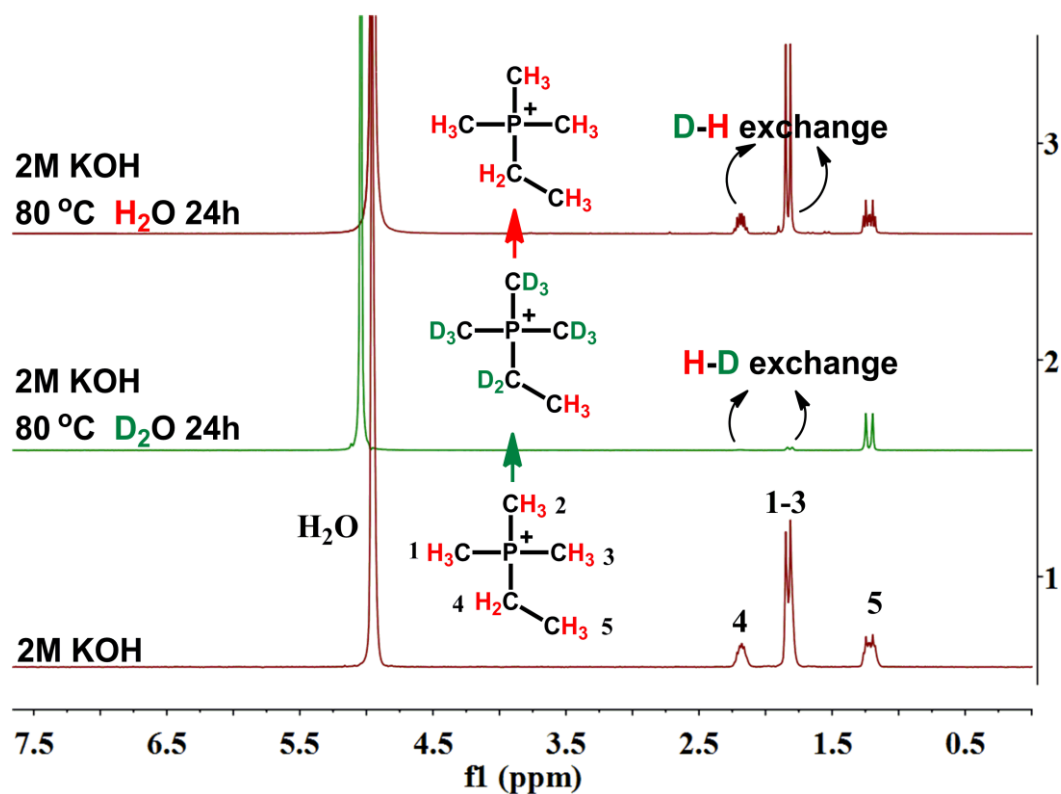


Figure S2. ^1H NMR spectra of $[\text{ETMP}]^+$ in 2 M KOH/ D_2O at $80\text{ }^\circ\text{C}$ for 24 h, before and after a second storage period in 2 M KOH/ H_2O for 24 h at $80\text{ }^\circ\text{C}$. The green and red arrows indicate the disappearance and reappearance of $\alpha\text{-C}$ protons after the interconversion of hydrogen and deuterium, respectively.

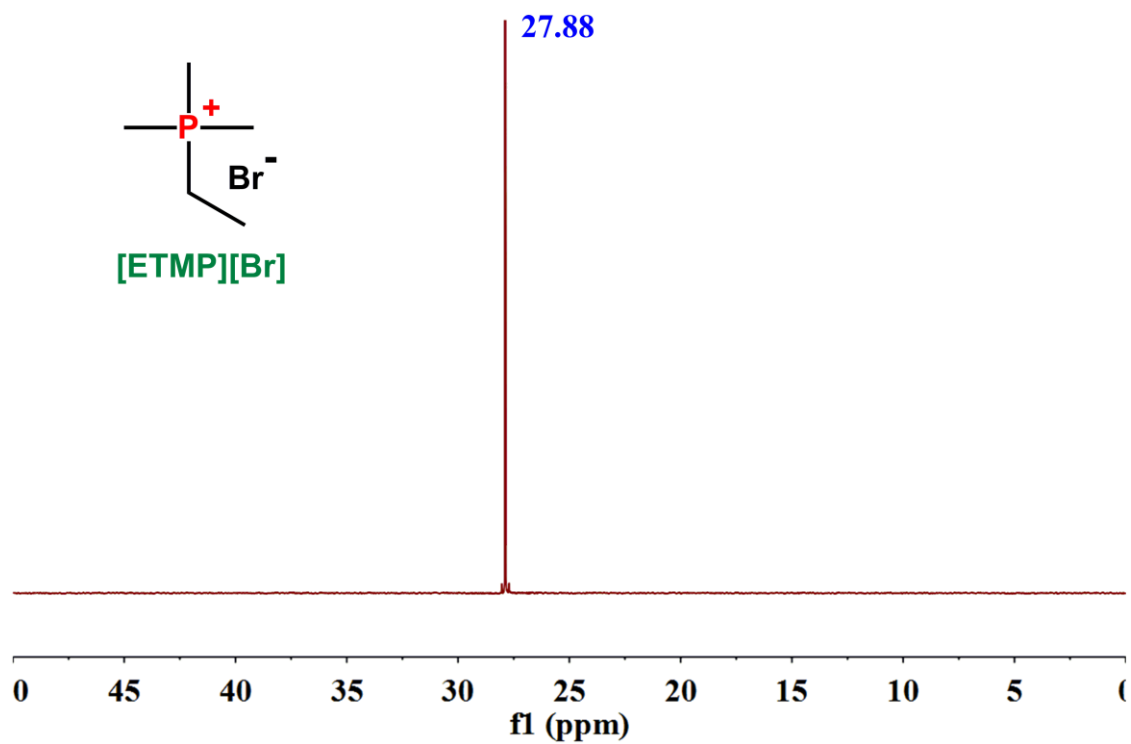


Figure S3. ^{31}P NMR spectra for [ETMP][Br].

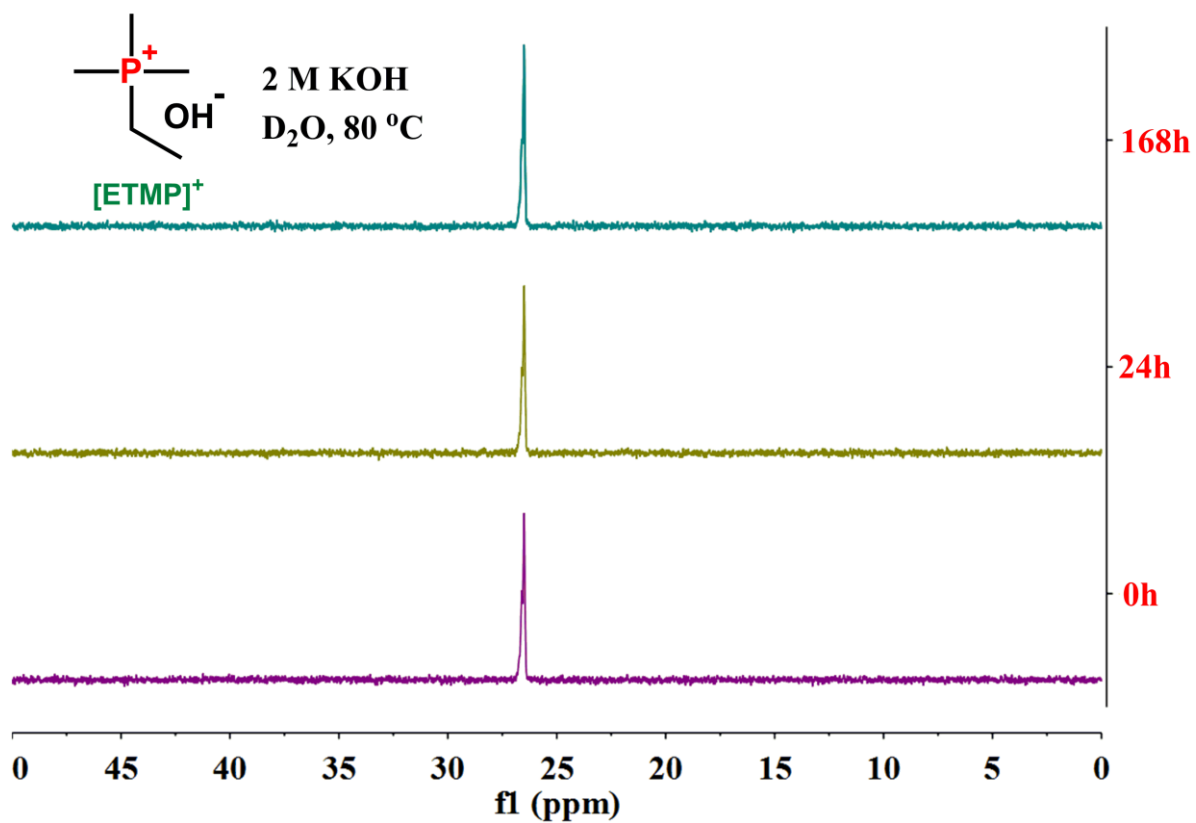


Figure S4. ^{31}P NMR spectra for [ETMP] $^+$ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) D_2O solution at 80 °C for 0, 24 and 168h, respectively.

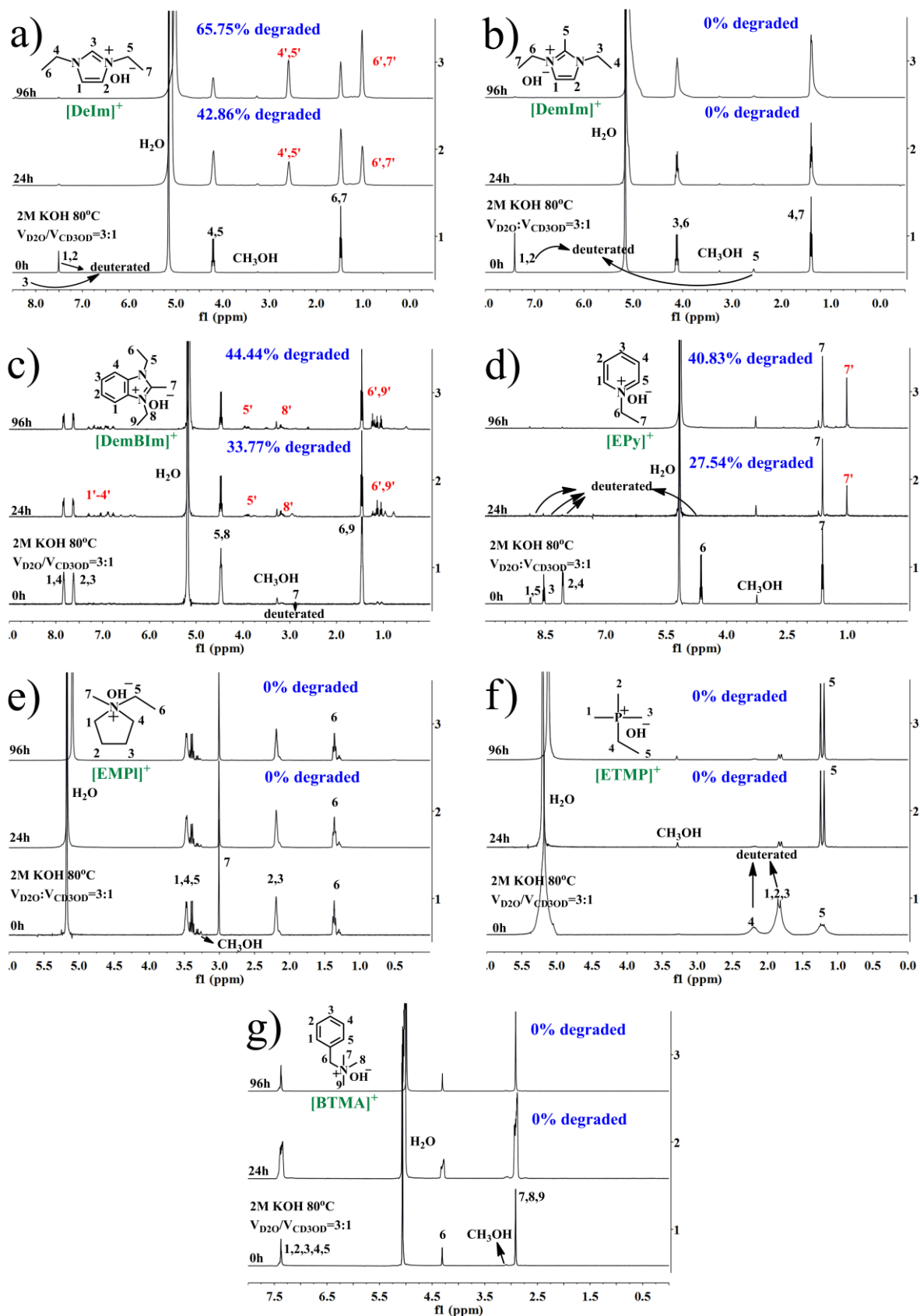


Figure S5. ^1H NMR spectra for (a) [DeIm] $^+$, (b) [DemIm] $^+$, (c) [DemBIm] $^+$, (d) [EPy] $^+$, (e) [EMPI] $^+$, (f) [ETMP] $^+$ and (g) [BTMA] $^+$ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) mixed solution ($\text{V}_{\text{D}_2\text{O}}/\text{V}_{\text{CD}_3\text{OD}}=3:1$) at 80 °C for 24 and 96h, respectively.

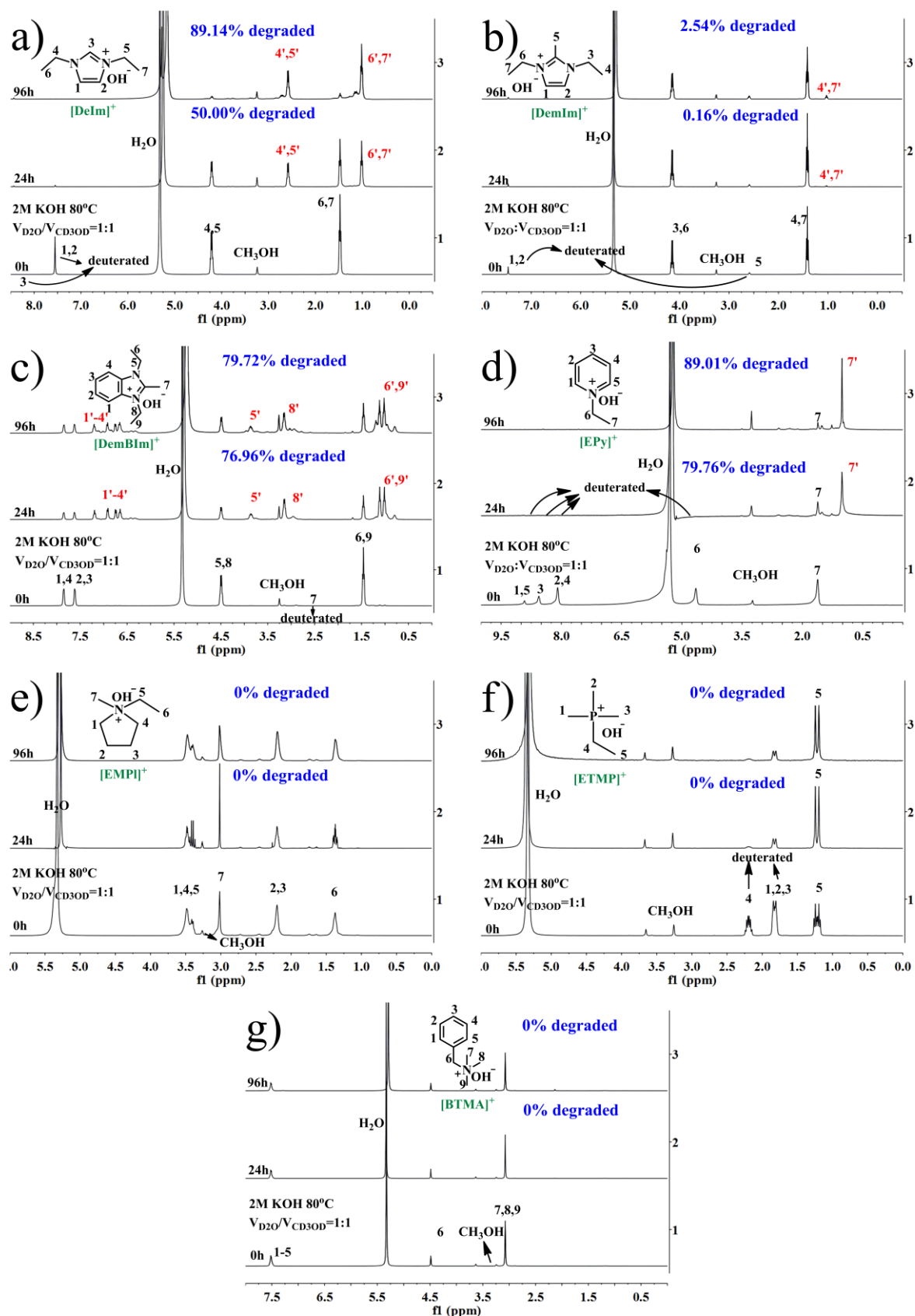


Figure S6. ^1H NMR spectra for (a) $[\text{DeIm}]^+$, (b) $[\text{DemIm}]^+$, (c) $[\text{DemBIm}]^+$, (d) $[\text{EPy}]^+$, (e) $[\text{EMPI}]^+$, (f) $[\text{ETMP}]^+$ and (g) $[\text{BTMA}]^+$ in 2 M KOH ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) mixed solution ($\text{V}_{\text{D}_2\text{O}}/\text{V}_{\text{CD}_3\text{OD}}=1:1$) at 80°C for 24 and 96h, respectively.

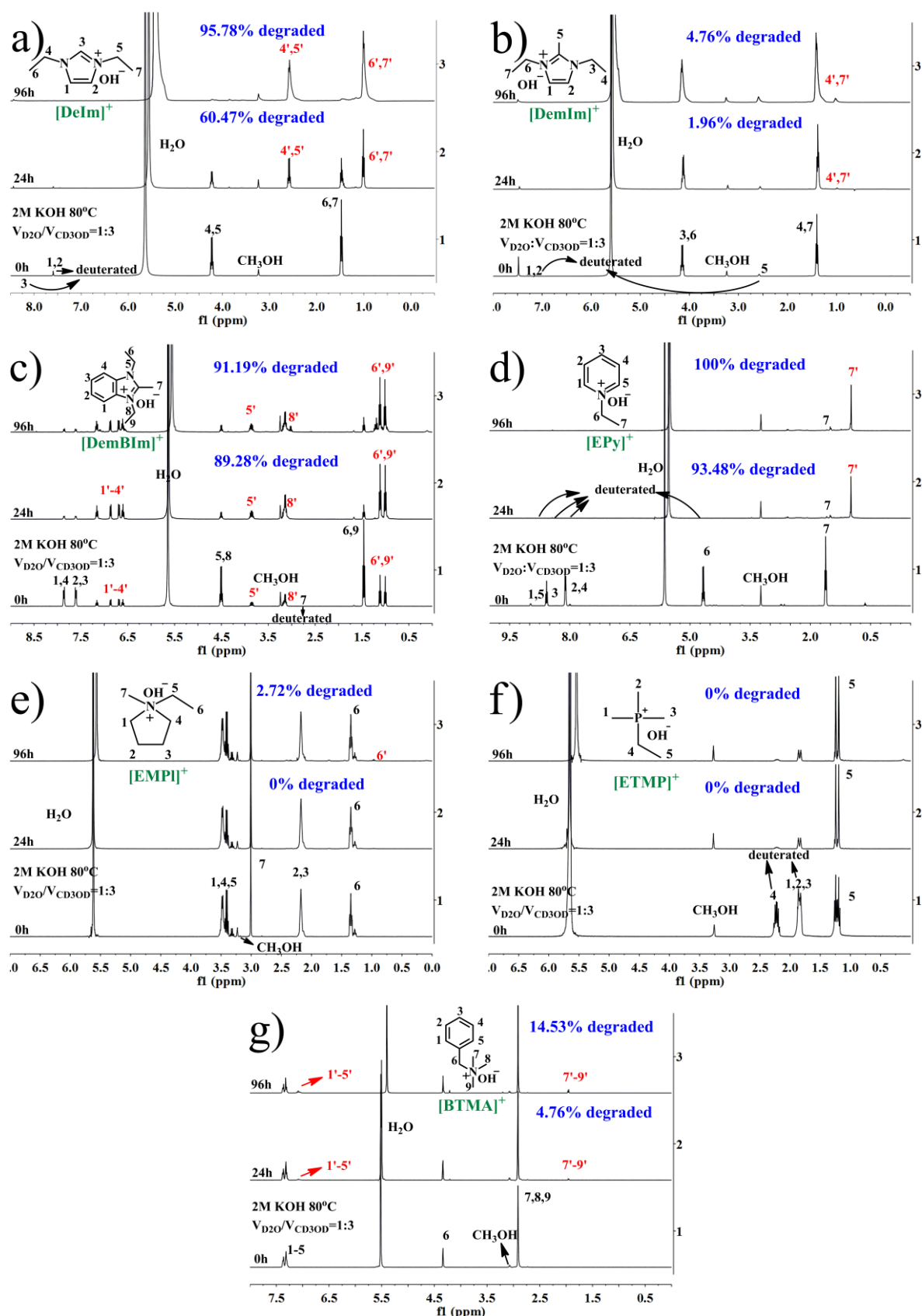


Figure S7. ^1H NMR spectra for (a) $[\text{DeIm}]^+$, (b) $[\text{DemIm}]^+$, (c) $[\text{DemBIm}]^+$, (d) $[\text{EPy}]^+$, (e) $[\text{EMPI}]^+$, (f) $[\text{ETMP}]^+$ and (g) $[\text{BTMA}]^+$ in 2 M KOH ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) mixed solution ($V_{\text{D}_2\text{O}}/V_{\text{CD}_3\text{OD}}=1:3$) at 80 °C for 24 and 96h, respectively.

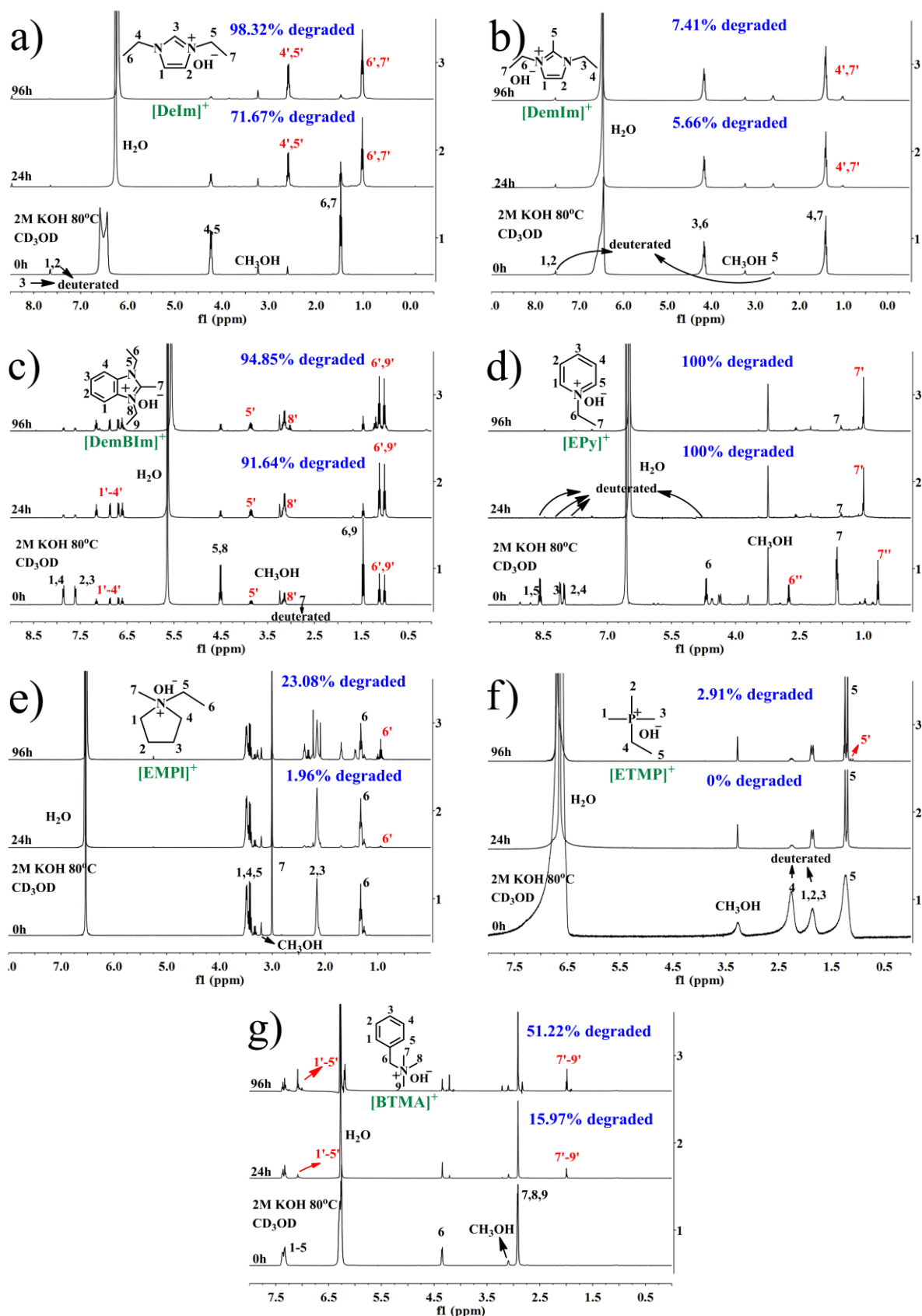


Figure S8. ^1H NMR spectra for (a) [DeIm] $^+$, (b) [DemIm] $^+$, (c) [DemBIm] $^+$, (d) [EPy] $^+$, (e) [EMPI] $^+$, (f) [ETMP] $^+$ and (g) [BTMA] $^+$ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) CD_3OD solution at 80 °C for 24 and 96h, respectively.

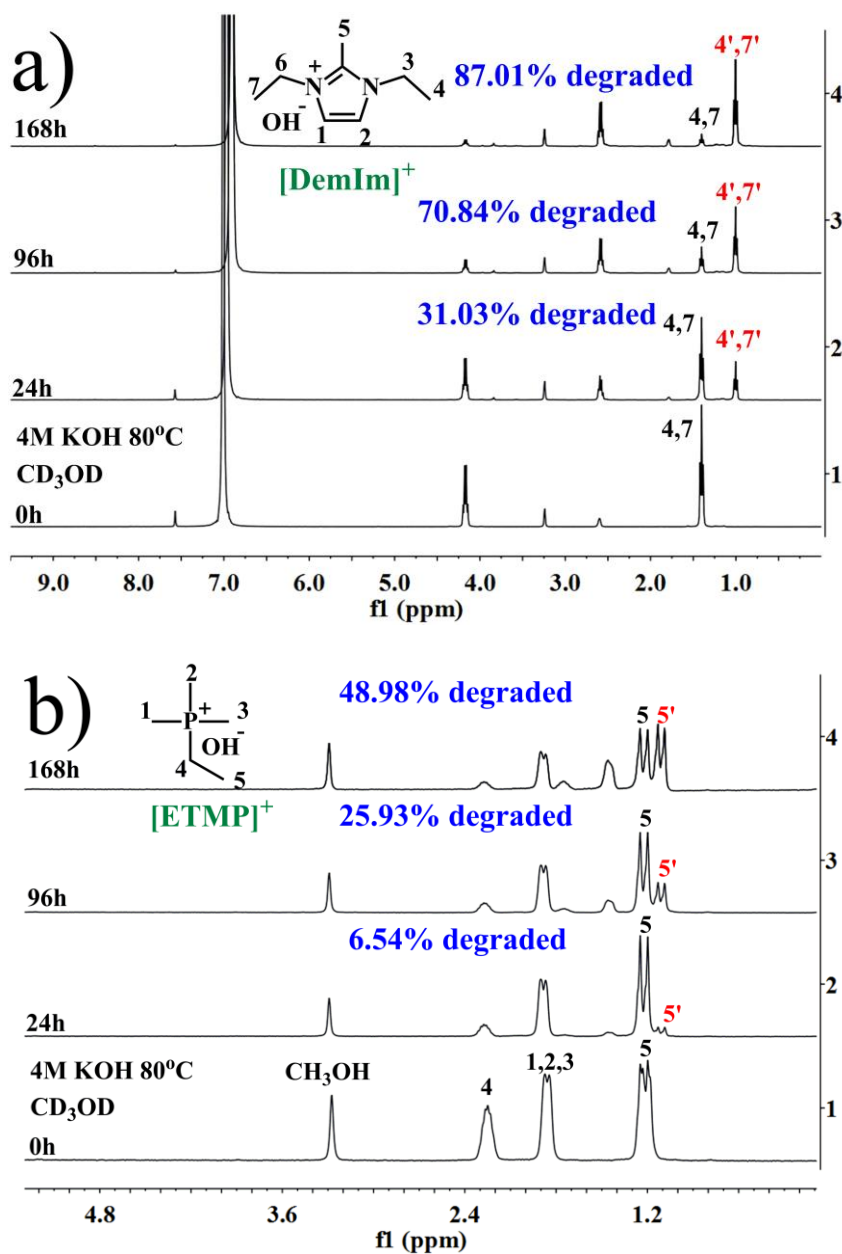


Figure S9. ^1H NMR spectra for (a) $[\text{DemIm}]^+$, (b) $[\text{ETMP}]^+$ in 4 M KOH CD_3OD solutions ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) at 80 °C for 24, 96 and 168h, respectively.

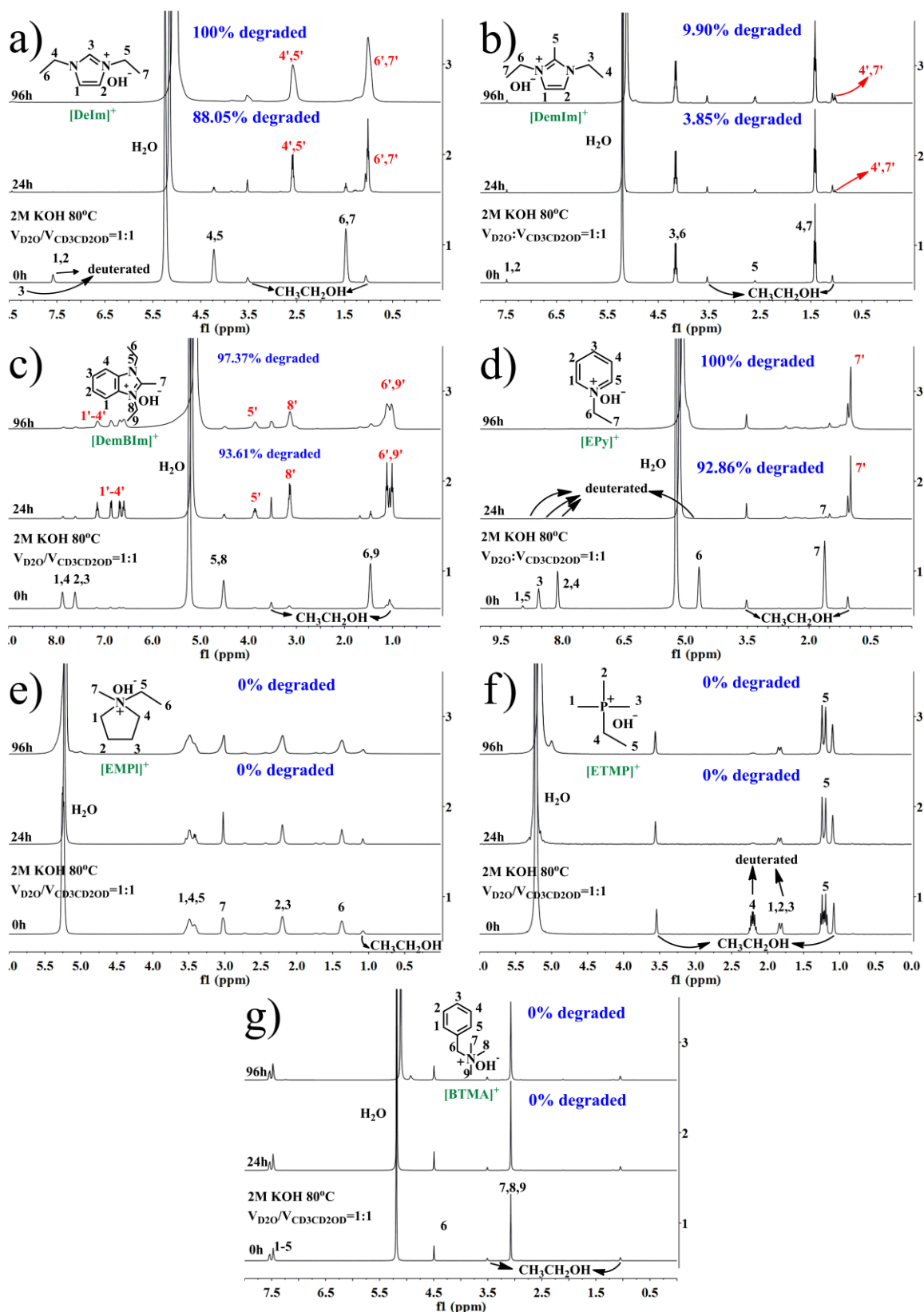


Figure S10. ^1H NMR spectra for (a) $[\text{DeIm}]^+$, (b) $[\text{DemIm}]^+$, (c) $[\text{DemBIm}]^+$, (d) $[\text{EPy}]^+$, (e) $[\text{EMPI}]^+$, (f) $[\text{ETMP}]^+$ and (g) $[\text{BTMA}]^+$ in 2 M KOH ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) mixed solution ($\text{V}_{\text{D}_2\text{O}}/\text{V}_{\text{CD}_3\text{CD}_2\text{OD}} = 1:1$) at 80°C for 24 and 96h, respectively.

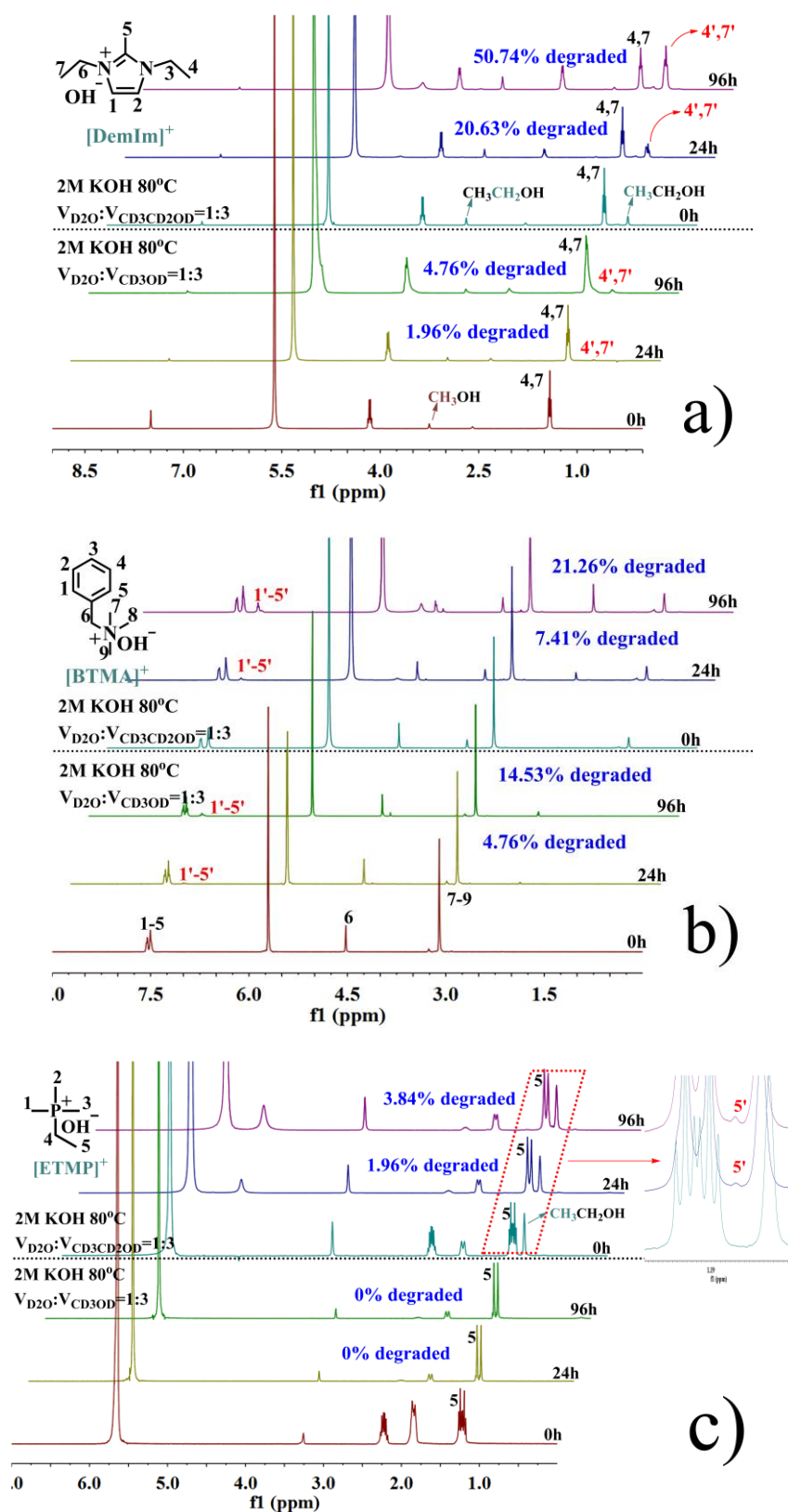


Figure S11. ¹H NMR spectra for (a) [DemIm]⁺, (b) [BTMA]⁺ and (c) [ETMP]⁺ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) $V_{D_2O}/V_{CD_3OD}=1:3$ and $V_{D_2O}/V_{CD_3CD_2OD}=1:3$ solution at 80 °C for 24 and 96h, respectively.

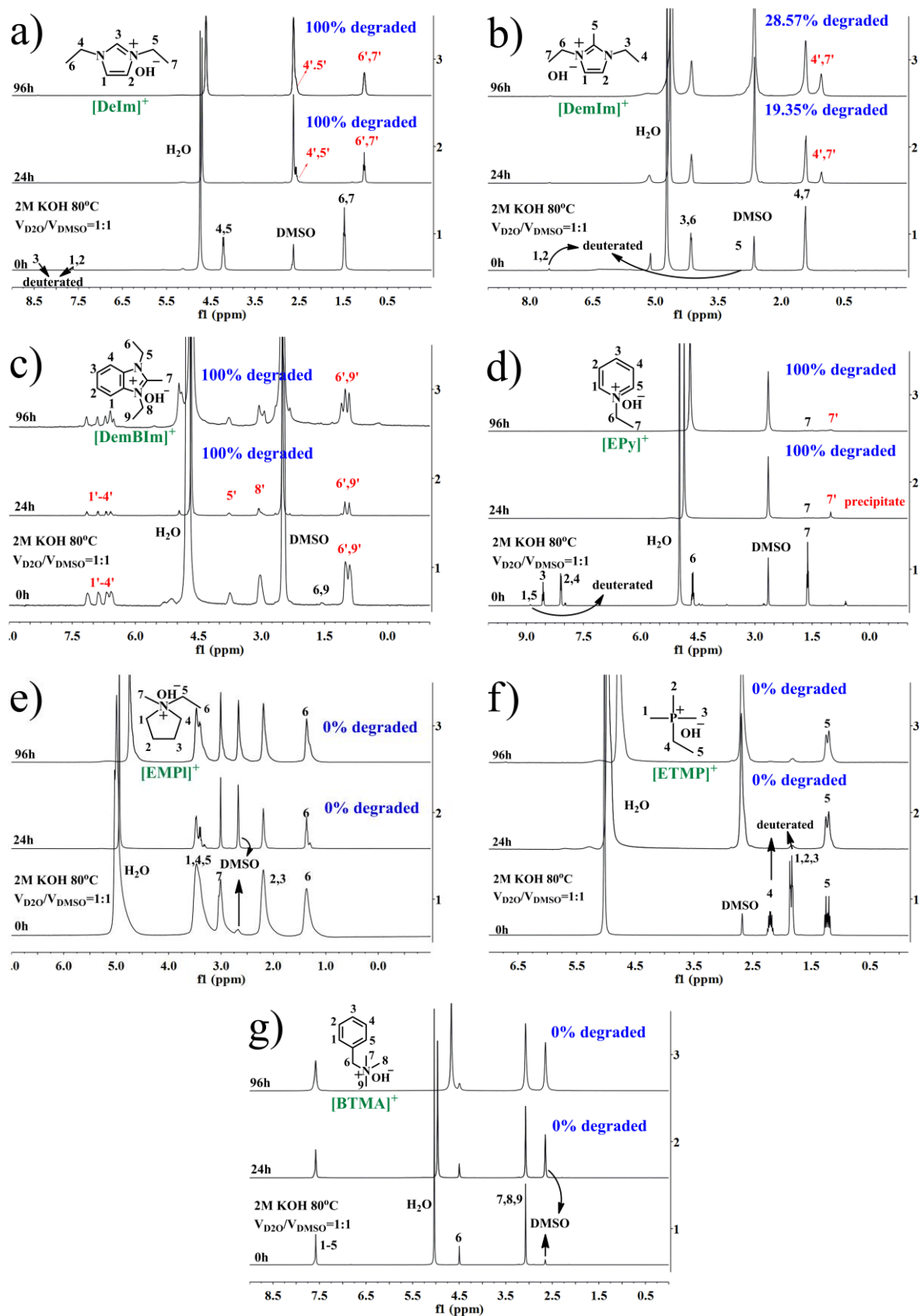


Figure S12. ^1H NMR spectra for (a) [DeIm]⁺, (b) [DemIm]⁺, (c) [DemBIm]⁺, (d) [EPy]⁺, (e) [EMPI]⁺, (f) [ETMP]⁺ and (g) [BTMA]⁺ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) mixed solution ($V_{\text{D}_2\text{O}}/V_{\text{DMSO}}=1:1$) at 80 °C for 24 and 96h, respectively.

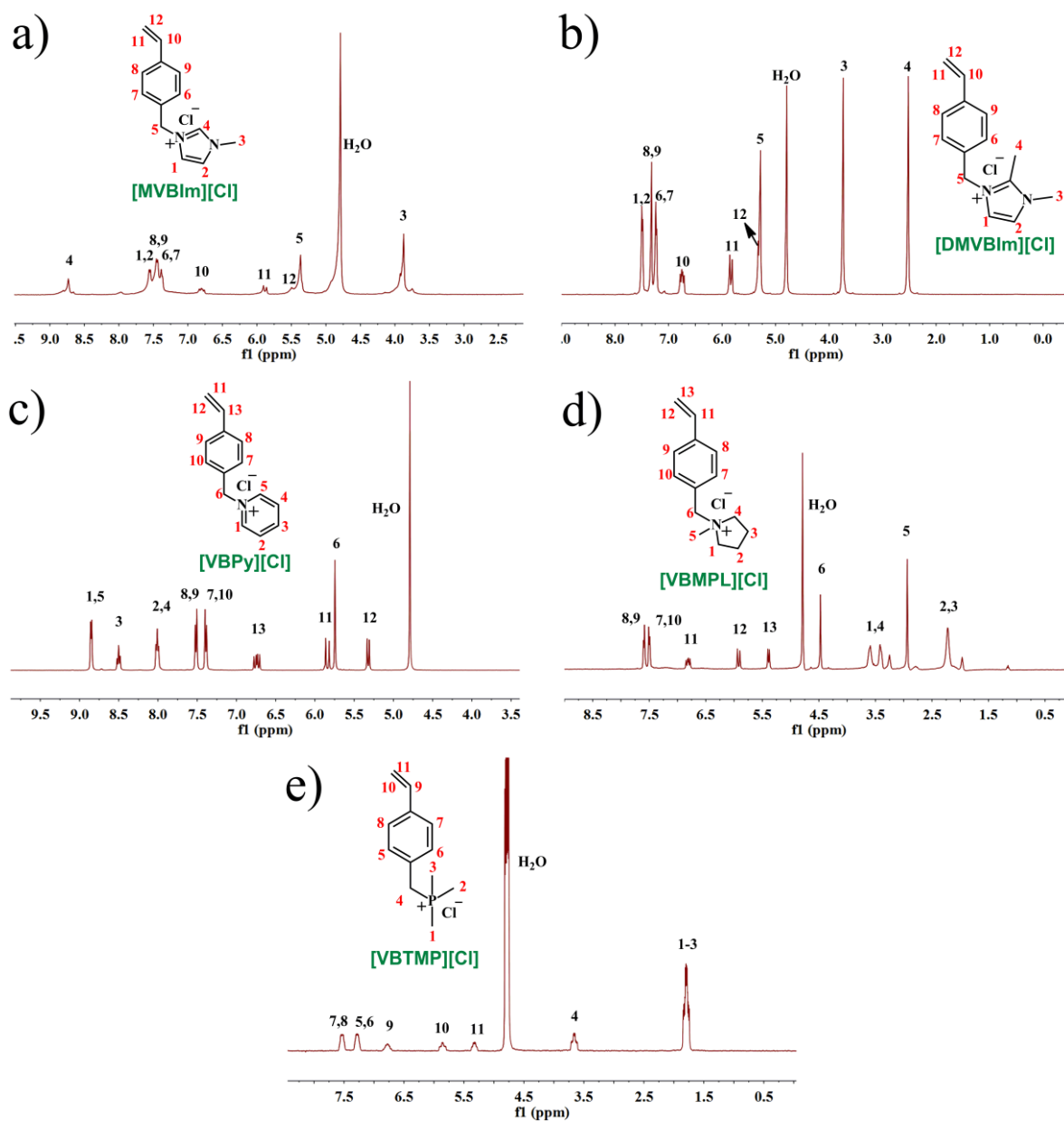


Figure S13. ^1H NMR spectra for the vinyl monomer synthesized in this work including: (a) $[\text{MVBIm}][\text{Cl}]$, (b) $[\text{DMVBIm}][\text{Cl}]$, (c) $[\text{VBPy}][\text{Cl}]$, (d) $[\text{VBMPL}][\text{Cl}]$ and (e) $[\text{VBTMP}][\text{Cl}]$.

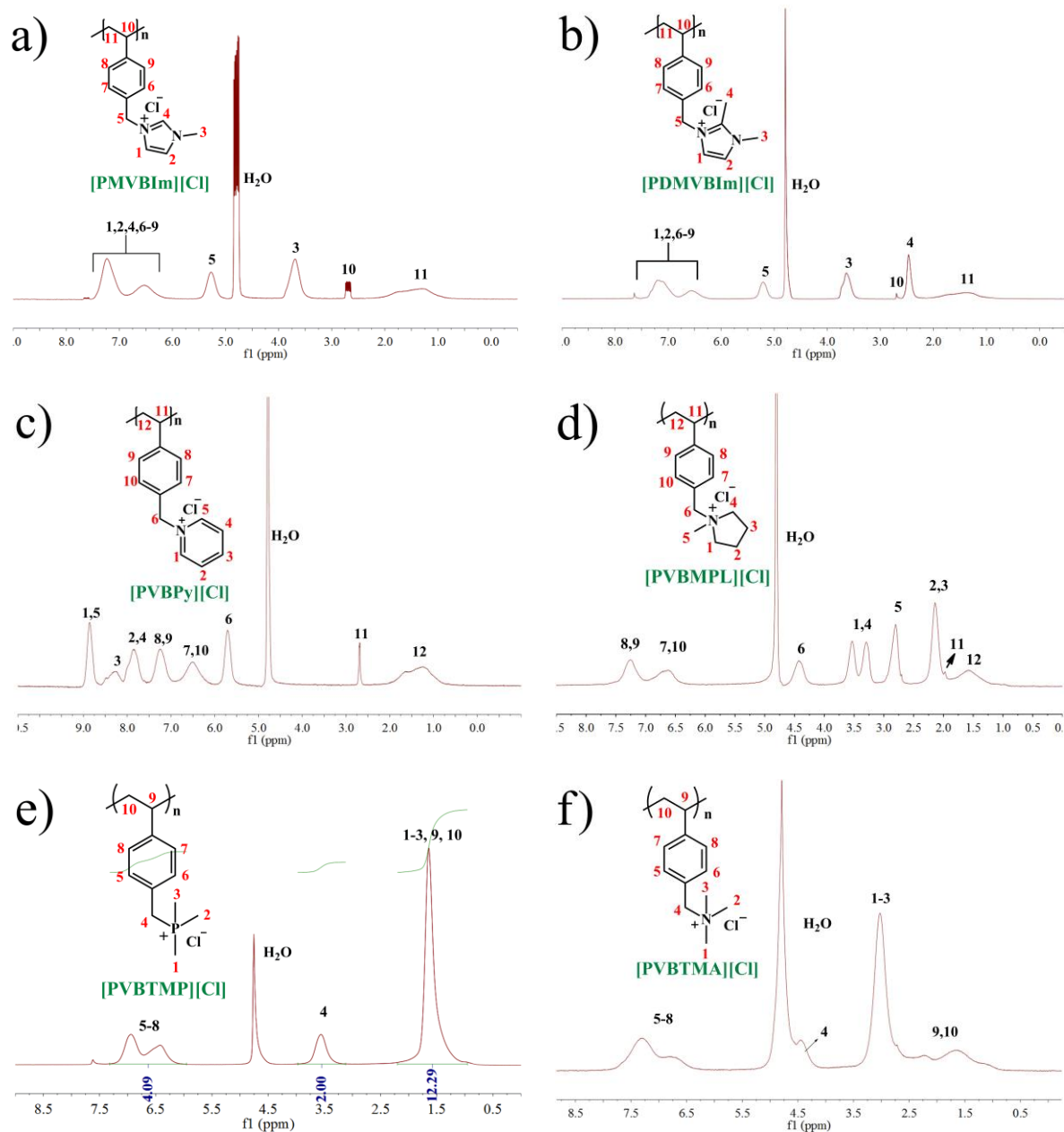


Figure S14. ^1H NMR spectra for various cationic polymers studied in this work including: (a) [PMVBIIm][Cl], (b) [PDMVBIIm][Cl], (c) [PVBPy][Cl], (d) [PVBMPIL][Cl], (e) [PVBTMP][Cl] and (f) [PVBTMA][Cl].

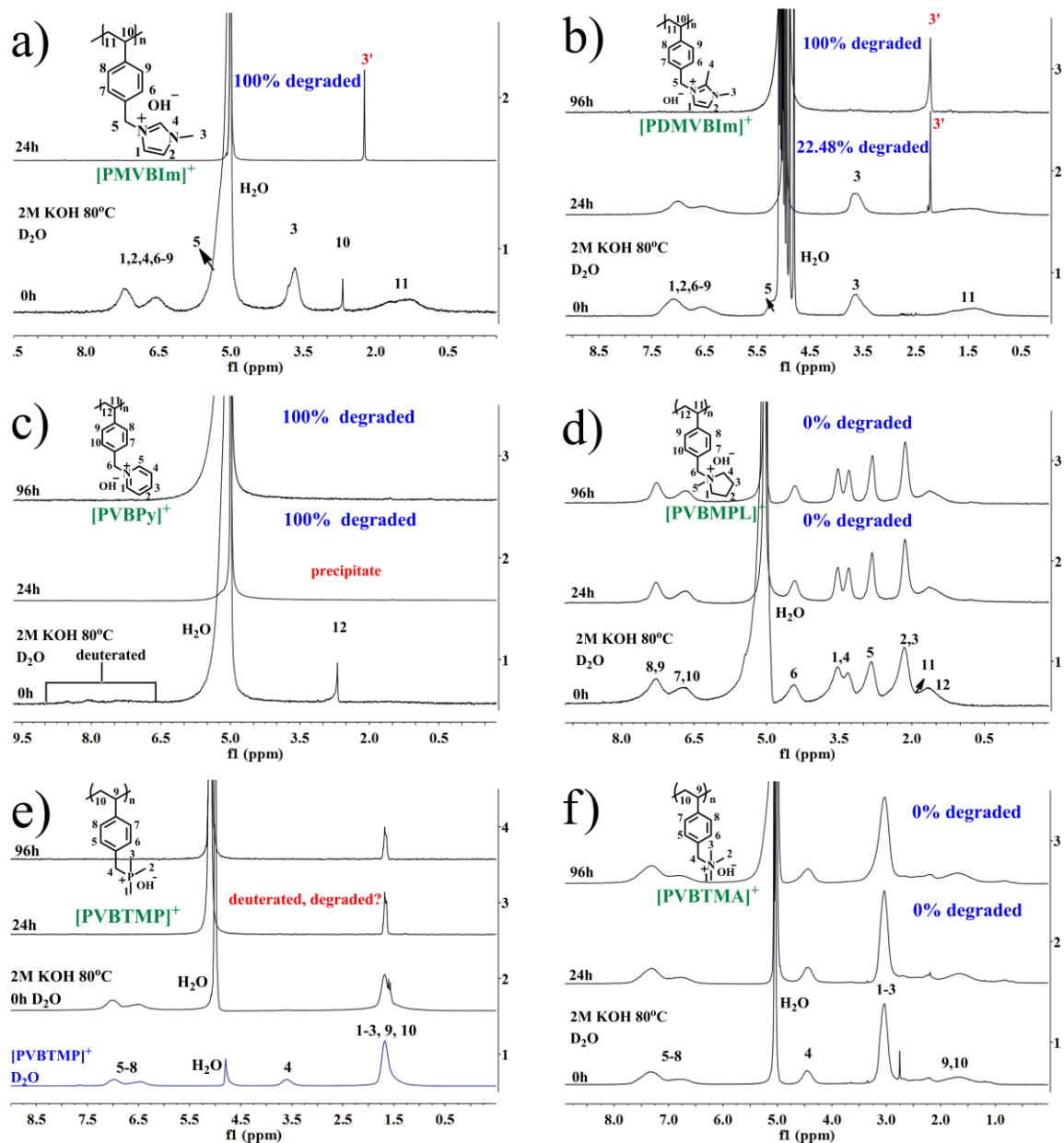


Figure S15. ^1H NMR spectra for (a) [PMVBIm]⁺, (b) [PDMVBIm]⁺, (c) [PVBPy]⁺, (d) [PVBMP1]⁺, (e) [PVBTMP]⁺ and (f) [PVBTMA]⁺ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) D₂O solution at 80 °C for 24 and 96 h, respectively.

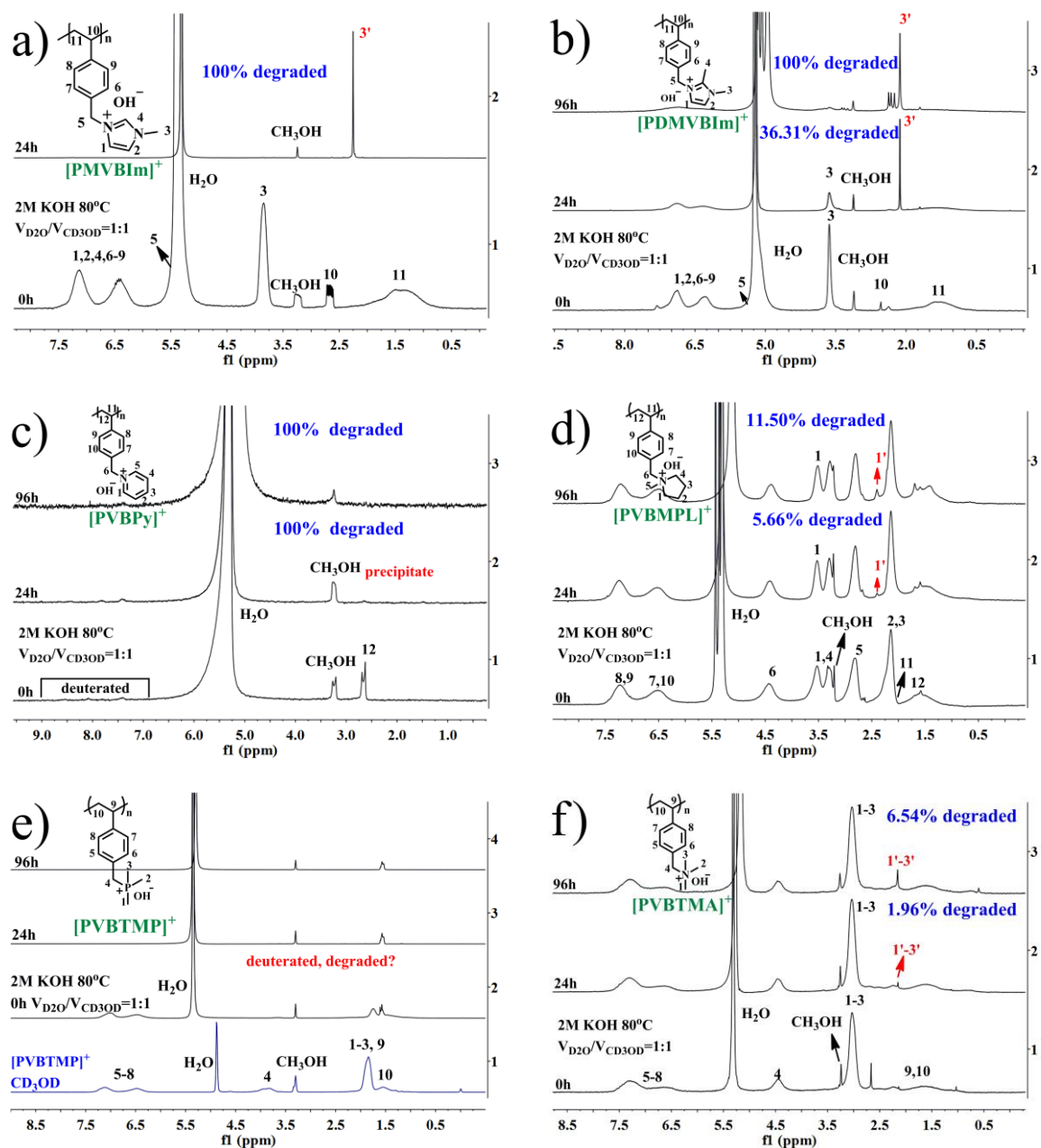


Figure S16. ^1H NMR spectra for (a) [PMVBI] $^m_+$, (b) [PDMVBI] $^m_+$, (c) [PVBPy] $^m_+$, (d) [PVBMP]L $^m_+$, (e) [PVBTTMP] $^m_+$ and (f) [PVBTTMA] $^m_+$ in 2 M KOH ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) mixed solution ($\text{V}_{\text{D}_2\text{O}}/\text{V}_{\text{CD}_3\text{OD}}=1:1$) at 80 °C for 24 and 96 h, respectively.

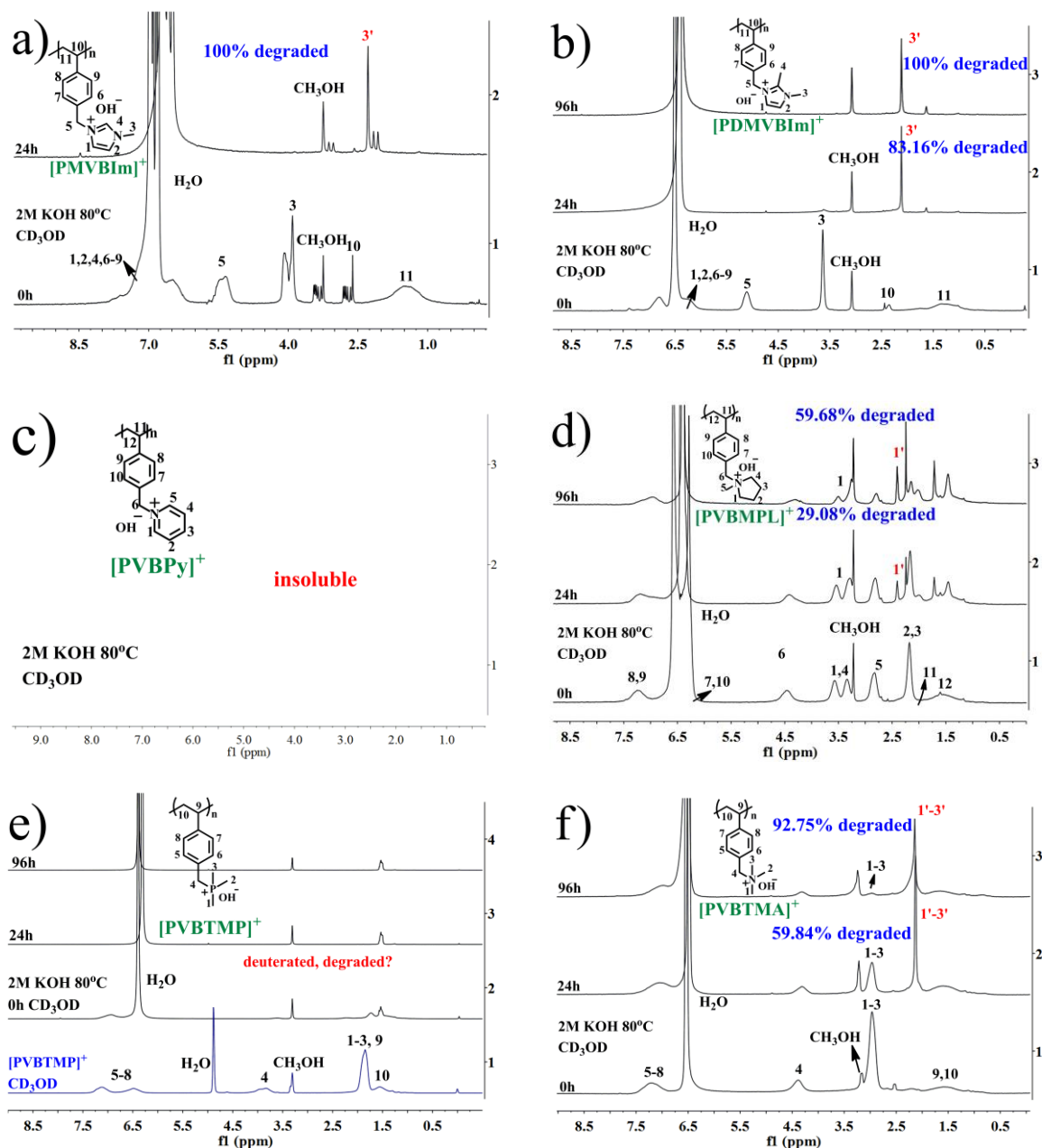


Figure 17. 1H NMR spectra for (a) $[PMVBI]_m^+$, (b) $[PDMVBI]_m^+$, (c) $[PVBPy]^+$, (d) $[PVBMP]_m^+$, (e) $[PVBTMP]^+$ and (f) $[PVBTMA]_m^+$ in 2 M KOH ($[KOH]/[cation] = 15/1$, molar ratio) CD_3OD solution at 80 °C for 24 and 96h, respectively.

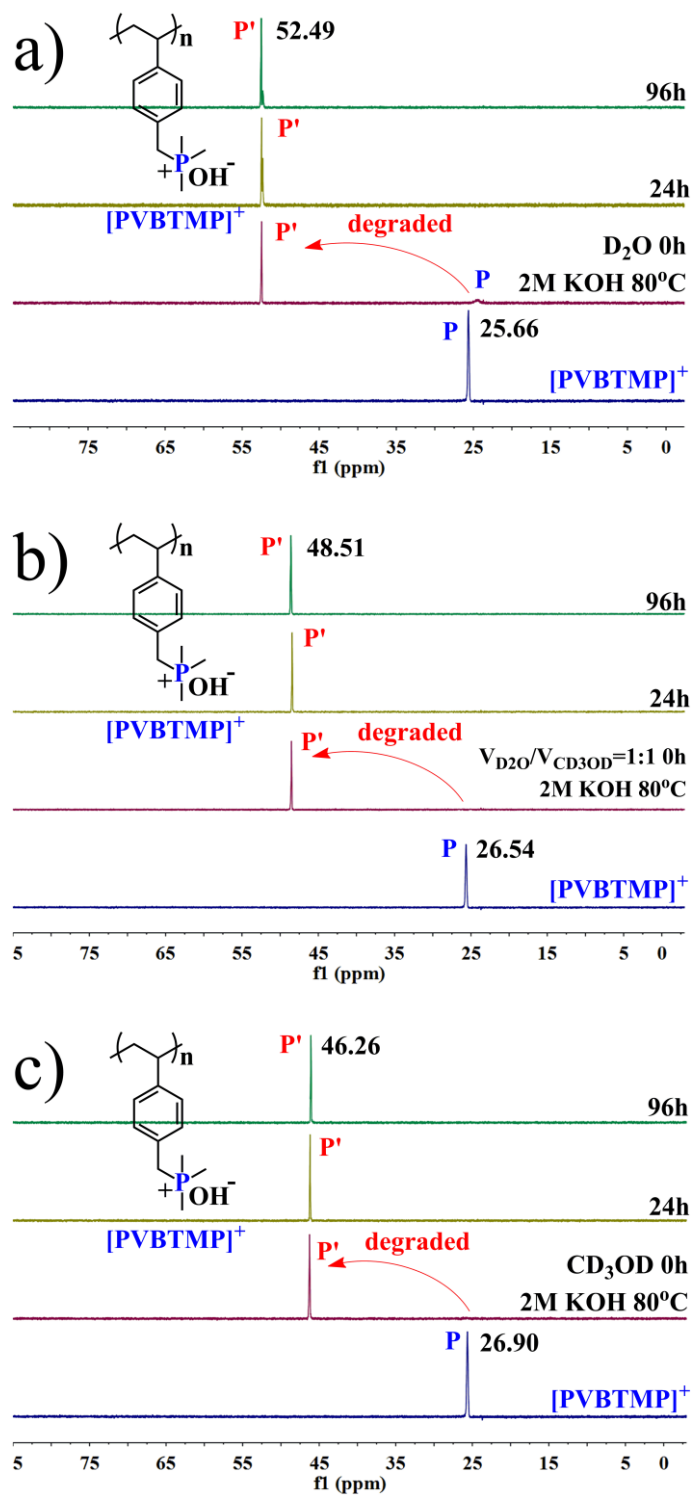


Figure S18. ³¹P NMR spectra for [PVBTMP]⁺ and [PVBTMP]⁺ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) (a) D₂O, (b) V_{D₂O}/V_{CD₃OD}=1:1 and (c) CD₃OD solutions at 80 °C for 0, 24 and 96h, respectively.

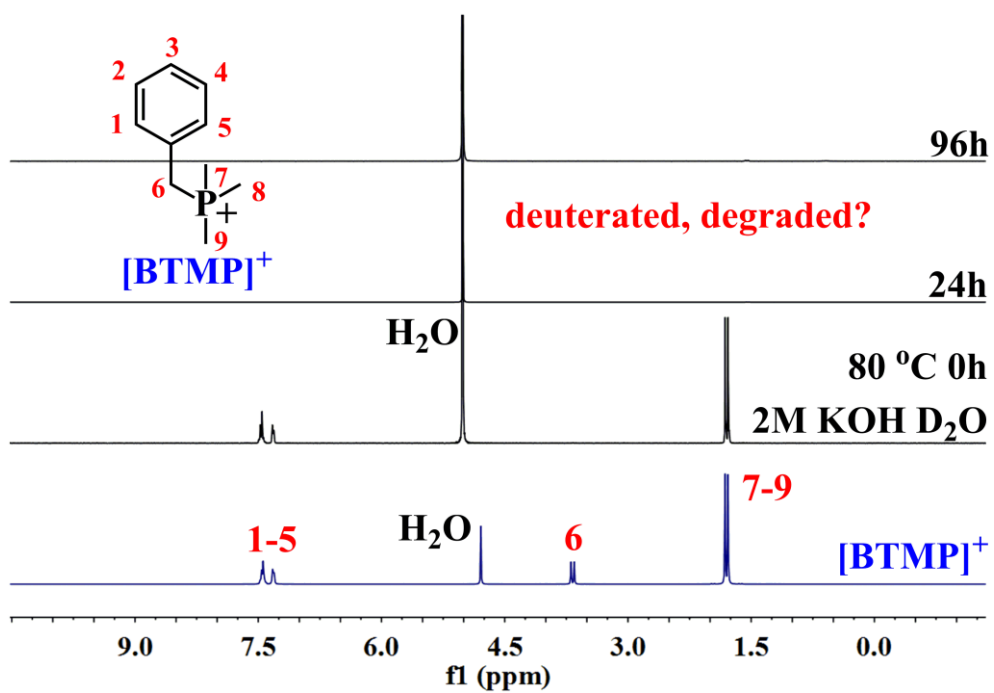


Figure S19. ^1H NMR spectra for $[\text{BTMP}]^+$ and $[\text{BTMP}]^+$ in 2 M KOH ($[\text{KOH}]/[\text{cation}] = 15/1$, molar ratio) D_2O solution at 80 °C for 0, 24 and 96h, respectively.

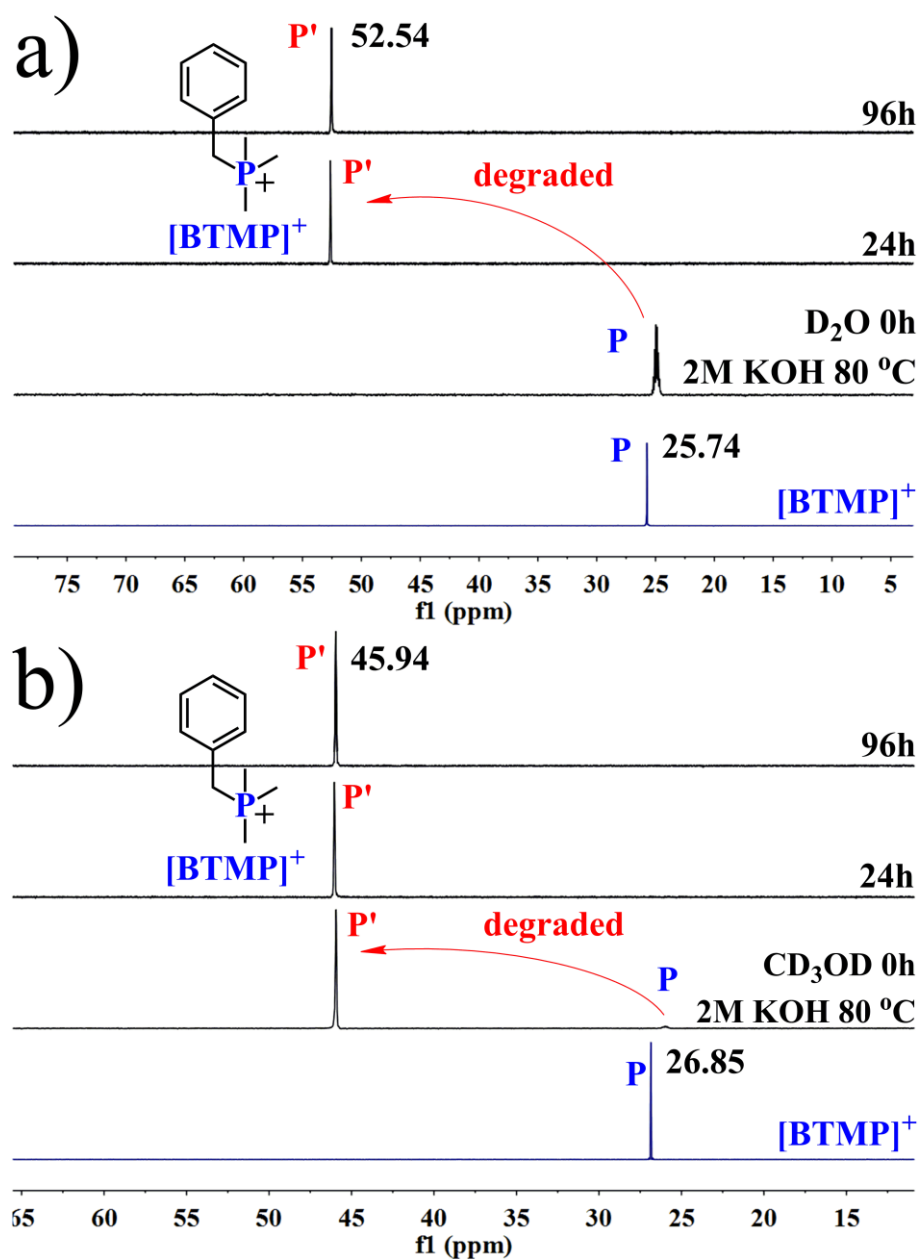


Figure S20. ^{31}P NMR spectra for $[\text{BTMP}]^+$ and $[\text{BTMP}]^+$ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) (a) D_2O and (b) CD_3OD solutions at 80 °C for 0, 24 and 96h, respectively.

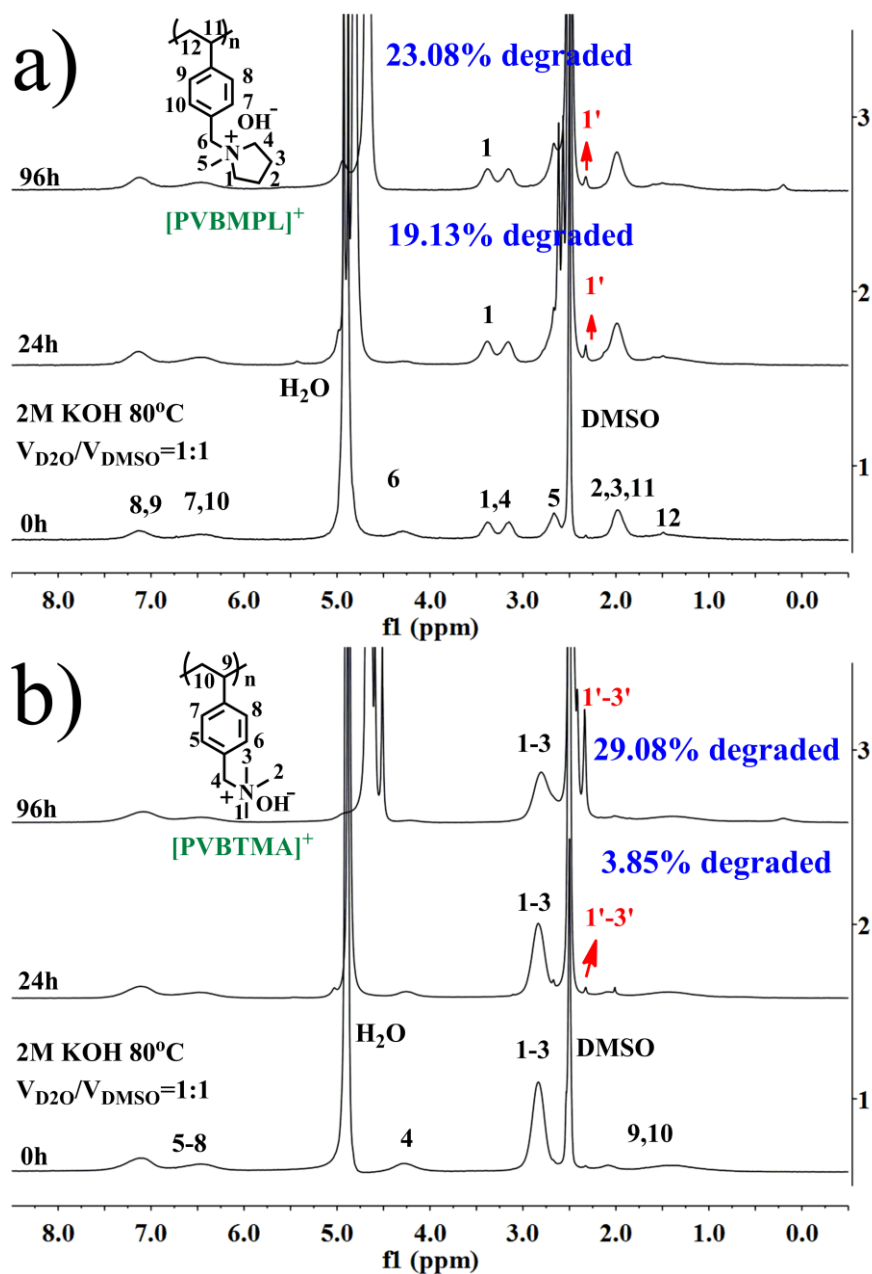


Figure S21. ¹H NMR spectra for (a) [PVBMP]⁺ and (b) [PVBMTMA]⁺ in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) mixed solution (V_{D₂O}/V_{DMSO}=1:1) at 80 °C for 0, 24 and 96h, respectively.

Table S1. The IEC values of cationic polymers in 2 M KOH ([KOH]/[cation] = 15/1, molar ratio) solutions at 80 °C for 96 h.

Cationic Polymers	IEC (mmol g ⁻¹)				
	Initial value	D ₂ O (2 M KOH 96h)	V _{D₂O} /V _{CD₃OD} =1:1 (2 M KOH 96h)	CD ₃ OD (2 M KOH 96h)	V _{D₂O} /V _{DMSO} =1:1 (2 M KOH 96h)
[PVB MPL][OH]	4.566	4.566	4.230	2.396	3.858
[PVB TMA][OH]	5.181	5.181	4.941	0.524	4.033
[PDMVB Im][OH]	4.348	0.000	0.000	0.000	\
[PMVB Im][OH]	4.630	0.000	0.000	0.000	\
[PVB Py][OH]	4.695	0.000	0.000	0.000	\
[PVB TMP][OH]	4.762	0.000	0.000	0.000	\