

Supplementary Materials for Progressive fuzzy cation- π assembly of biological catecholamines

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Fig. S6. Surface morphologies of the NMe₄⁺-intercalated polydopamine functionalized surfaces at St-Bc with 0.1 M NMe₄⁺ (top) and 1 M NMe₄⁺ (bottom) by SEM and atomic force microscopy analyses.

Other Supplementary Material for this manuscript includes the following:

(available at advances.sciencemag.org/cgi/content/full/4/9/eaat7457/DC1)

Movie S1 (.mp4 format). Stability of polydopamine coating in NaOH, DDW, and KOH solution.

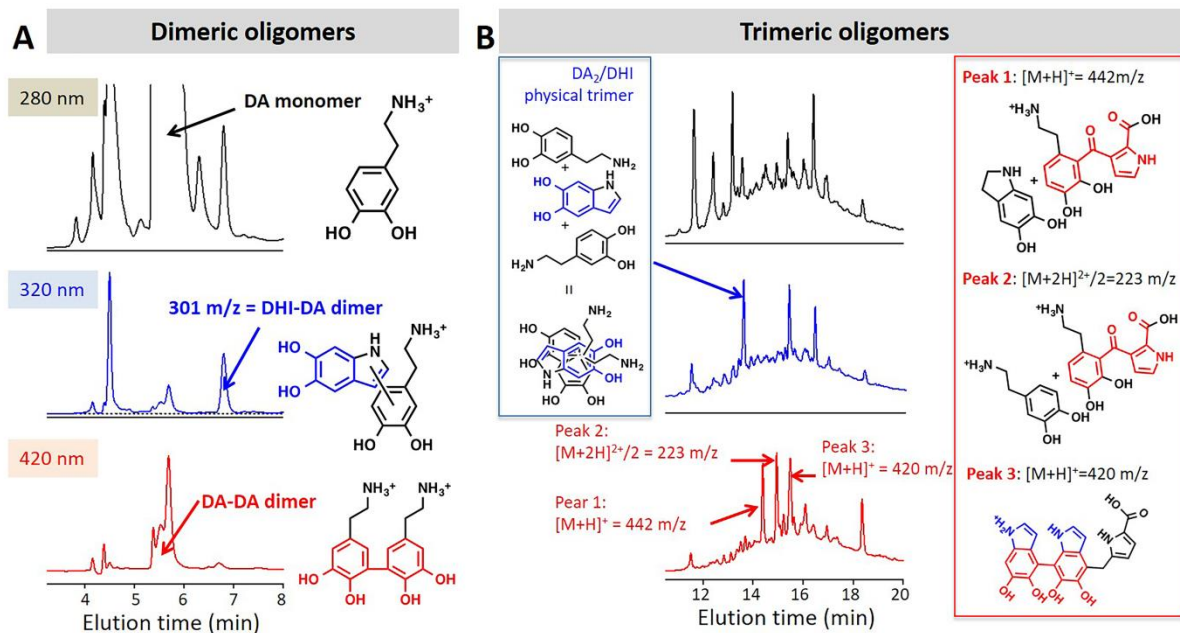


Fig. S1. Chemical structure of oligomers formed during polydopamine coating under St-Bc pH conditions analyzed by HPLC-MS with a C18 column. (A) Dimeric intermediates detected at approximately 4–8 min of elution time identified by both mass spectra and UV absorbance at 280 nm (black line), 320 nm (blue line), and 420 nm (red line). **(B)** Trimeric oligomers including the pre-defined (dopamine)₂/DHI (DA₂/DHI) physical trimer (blue box) and newly identified ones assigned as trimers containing partially opened/degraded DHI subunits (red box).

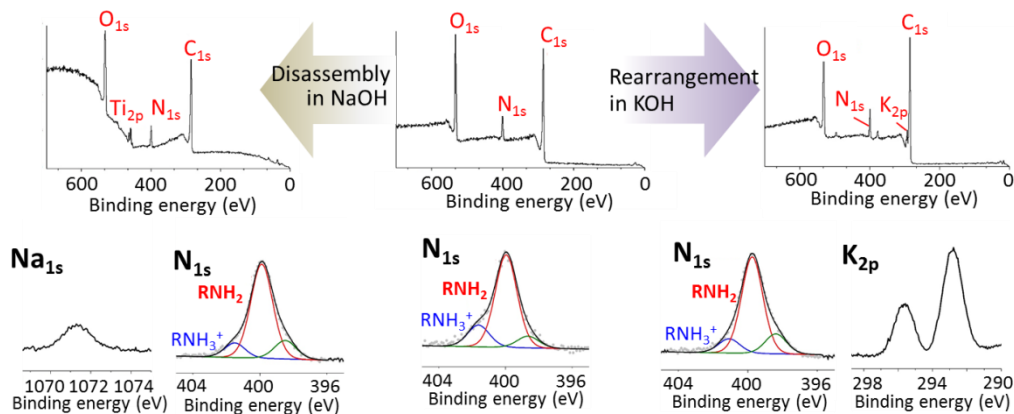


Fig. S2. XPS analyses of polydopamine-coated TiO₂ surfaces after 1 hour of strong base treatment. (i.e., 0.1 N NaOH (pH 11) and 0.1 N KOH (pH 11)). 1st row: survey scan and 2nd row: cation moieties (Na_{1s} and K_{1s}) and nitrogen (-NH₃⁺ in blue, -NH₂ in red, and -NH- in green) species. The protonated amines (-NH₃⁺) in the polydopamine-coated TiO₂ surface under Md-Bc existed above 35% (401.6-401.8 eV), whereas most protonated amines disappeared after both base treatments (below 8%).

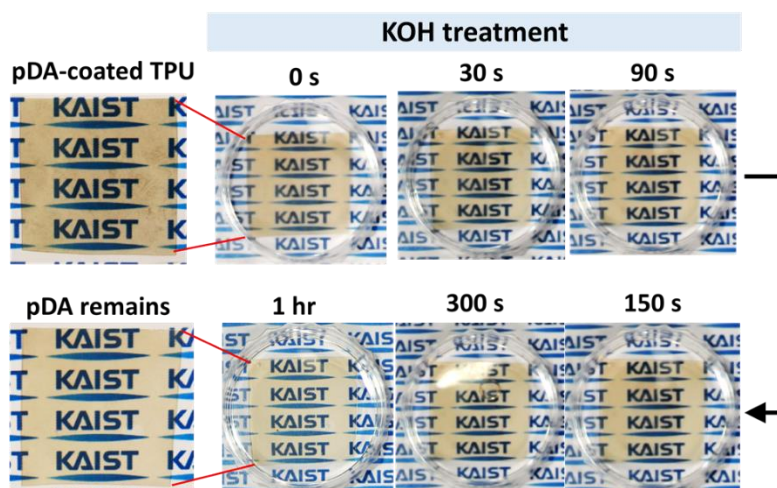


Fig. S3. Maintenance of polydopamine coating on a TPU film in 0.1 N KOH (pH 12) solution for 1 hour.

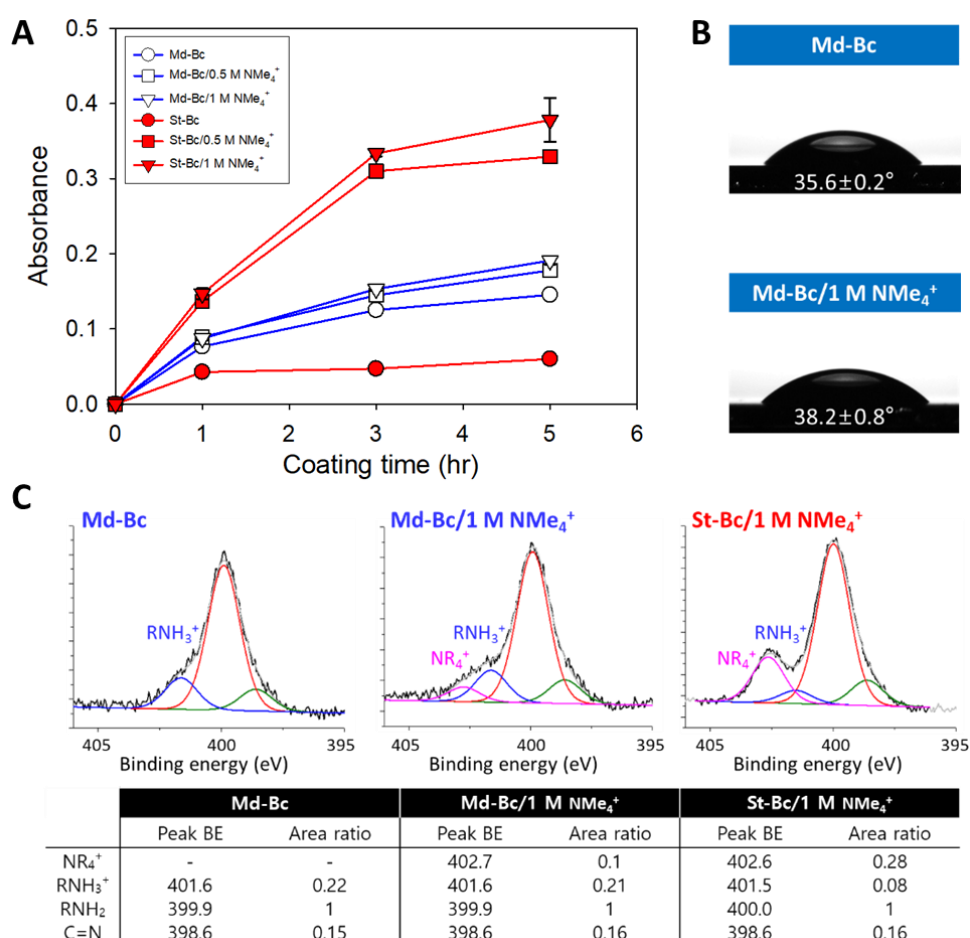


Fig. S4. Intercalation of NMe₄⁺ on polydopamine coating with pH variations. (A) The initial polydopamine film growth rate monitored by UV absorption at 400 nm on a transparent glass substrate as the surface color becomes dark brown progressively (Blue lines: Md-Bc, Red lines: St-Bc / circle: no NMe₄⁺, square: with 0.5 M NMe₄⁺, reverse triangle: with 1 M NMe₄⁺). (B) Wettability of polydopamine coated substrates in Md-Bc, with (bottom) and without (top) presence of NMe₄⁺. (C) XPS N_{1s} high resolution analysis of polydopamine coating in Md-Bc (left), Md-Bc with NMe₄⁺ (middle), and St-Bc with NMe₄⁺ (right): pink line for NMe₄⁺, blue line for R- NH₃⁺, red line for R-NH₂, and green line for C=N. Peak binding energy (denoted as peak BE) and the area ratio of each nitrogen peak are summarized in a table below the XPS N_{1s} graphs.

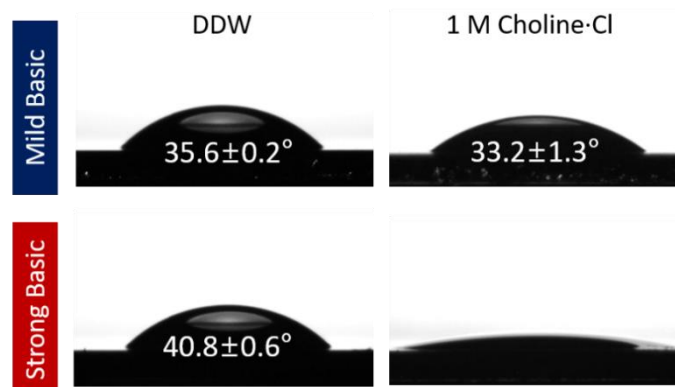


Fig. S5. Wettability of polydopamine-coated substrates in Md-Bc and St-Bc conditions in the presence and absence of 1 M choline, a precursor to neurotransmitter ACh.

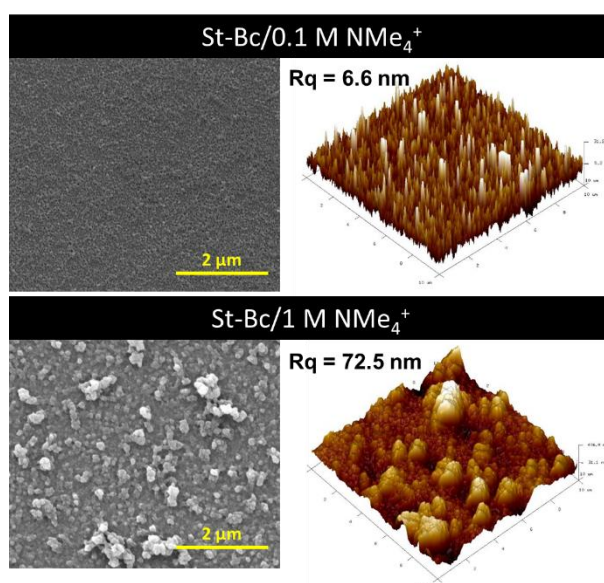


Fig. S6. Surface morphologies of the NMe_4^+ -intercalated polydopamine functionalized surfaces at St-Bc with 0.1 M NMe_4^+ (top) and 1 M NMe_4^+ (bottom) by SEM and atomic force microscopy analyses.