

The Hydrogen Atom Transfer Reactivity of Sulfinic Acids

Markus Griesser, Jean-Philippe R. Chauvin and Derek A. Pratt*

*Department of Chemistry and Biomolecular Sciences, University of Ottawa,
Ottawa, Ontario, CANADA K1N 6N5*

*dpratt@uottawa.ca

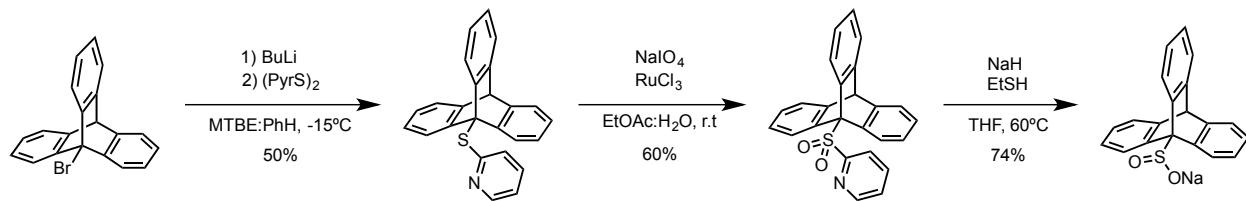
Supporting Information

Table of Contents

Synthetic Procedures	S2
Procedure to Determine BDEs Through REqEPR	S5
LFP – Details and Results	S8
Co-autoxidations in Organic Solutions	S14
Summary of Rate Constants for Peroxyl Radicals	S16
Naphthyl-Neophyl Clock to Measure Alkyl Radical Kinetics	S17
NMR Spectra	S18
Computational Data	S23
References	S44

Synthetic Procedures

Synthesis of 9-trptycenesulfinic acid



2-(9,10-[1,2]benzenoanthracen-9(10H)-ylthio)pyridine.

9-Bromotriptycene (4.0 g, 12.0 mmol) was dissolved in dry MTBE (95 mL) and benzene (145 mL) and cooled to -15 °C under nitrogen. After stirring for 15 minutes, BuLi (5.8 mL, 14.5 mmol) was added dropwise. The solution was warmed to 0 °C over 30 minutes and 2,2'-dithiopyridine was added all at once and the resulting yellow solution was stirred overnight. The reaction was quenched with water (100 mL) and added to EtOAc (150 mL) in a separatory funnel. The organic phase was washed with water and brine, dried over MgSO₄, filtered and concentrated in vacuo. The crude solid was purified by column chromatography using 10% EtOAc/hexanes as the eluent to yield the product as a white solid (2.10 g, 50%). ¹H NMR (400 MHz; CDCl₃): δ 8.49 (ddd, *J* = 4.8, 1.8, 0.8 Hz, 1H), 7.51-7.43 (m, 6H), 7.29-7.25 (m, 1H), 7.07-6.94 (m, 7H), 6.67 (d, *J* = 8.2 Hz, 1H), 5.46 (s, 1H). ¹³C NMR (101 MHz; CDCl₃): δ 158.5, 149.3, 145.4, 142.9, 135.8, 125.9, 125.0, 124.6, 123.8, 123.6, 119.7, 62.1, 54.1. HRMS (EI, [M⁺]): m/z calcd for C₂₅H₁₇NS: 363.1082, found 363.1079.

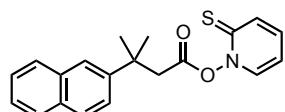
2-(9,10-[1,2]benzenoanthracen-9(10H)-ylsulfonyl)pyridine.

2-(9,10-[1,2]benzenoanthracen-9(10H)-ylthio)pyridine (1.50 g, 4.30 mmol) was added to a 1:1 mixture of EtOAc and water (16 mL) and cooled to 0 °C under N₂. RuCl₃ (44.4 mg, 0.21 mmol) was added and the solution was stirred for 10 minutes before NaIO₄ (5.50 g, 26.0 mmol) was added portion wise and the mixture was vigorously stirred at room temperature overnight. The solution was diluted with ethyl acetate (100 mL) and washed with water and brine. The organic layer was dried over magnesium sulfate, filtered and concentrated in vacuo. The crude yellow solid was purified by column chromatography using a gradient of 10-20% EtOAc/hexanes as the eluent to yield the product as a white

solid (1.10 g, 60%). ^1H NMR (400 MHz; CD₃CN): δ 8.41 (dt, J = 8.1, 1.0 Hz, 1H), 8.36 (ddd, J = 4.7, 1.7, 0.9 Hz, 1H), 8.24-8.18 (m, 2H), 7.80 (dt, J = 7.9, 0.5 Hz, 2H), 7.64 (ddd, J = 7.7, 4.7, 1.1 Hz, 1H), 7.53-7.49 (m, 3H), 7.13-7.08 (m, 4H), 6.97-6.93 (m, 2H), 5.60 (s, 1H). ^{13}C NMR (101 MHz; acetone-*d*₆): δ 162.4, 149.8, 147.3, 146.4, 141.6, 140.8, 140.2, 128.8, 127.2, 126.85, 126.66, 125.35, 125.24, 124.99, 124.91, 124.2, 122.8, 78.8, 54.9. HRMS (EI, [M⁺]): m/z calcd for C₂₅H₁₇NO₂S: 395.0980, found 395.0985.

Sodium 9,10-[1,2]benzenoanthracene-9(10H)-sulfinate.

2-(9,10-[1,2]benzenoanthracen-9(10H)-ylsulfonyl)pyridine (1.00 g, 2.53 mmol) was dissolved in dry THF (12 mL), cooled to 0 °C and purged with N₂ in a sealed tube. NaH (30.4 mg, 7.59 mmol) was added portion wise followed by EtSH (violent gas evolution) and the tube was sealed quickly. The solution was stirred at 70 °C overnight after which the reaction was quenched with methanol (5 mL) and concentrated. The crude product was purified by column chromatography using a gradient of 10-20% MeOH:DCM to yield the product as a off white solid (63.0 mg, 74%). ^1H NMR (400 MHz; CD₃OD): δ 8.26 (br, 3H), 7.36-7.34 (m, 3H), 7.01-6.94 (m, 6H), 5.32 (s, 1H). ^{13}C NMR (151 MHz; CD₃OD): δ 148.1, 143.9, 126.5, 126.2, 125.5, 123.8, 75.9, 55.8. HRMS (ESI, [M-H⁺]): m/z calculated for C₂₀H₁₃O₂S: 317.0642, found 317.0652.



2-thioxopyridin-1(2H)-yl 3-methyl-3-(naphthalen-2-yl)butanoate. 3-methyl-3-(naphthalen-2-yl)butanoic acid^{1,2} (0.600 g, 2.60 mmol) was dissolved in dry benzene (15 mL) in an over-dried round bottom flask and purged with N₂ for 5 minutes. DMF (1 drop) and thionyl chloride (0.283 mL, 3.9 mmol) were added dropwise and the solution was heated at reflux for 2 hours. Excess thionyl chloride was removed by rotary evaporation and the crude acyl chloride was dissolved in benzene (2 mL) and added dropwise to a stirred suspension of 2-mercaptopypyridine sodium salt (0.500 g, 3.38 mmol) and DMAP (0.050 g, 0.40 mmol) in benzene (10 mL). The solution was stirred at room temperature for 1 hour in the dark after which it was concentrated and the crude yellow solid was purified by column chromatography using 50% EtOAc/hexanes as the

eluent to yield the product as a yellow solid (0.53 g, 61%). ^1H NMR (600 MHz; CDCl_3): δ 7.86-7.81 (m, 4H), 7.60 (ddd, J = 8.7, 4.8, 1.8 Hz, 2H), 7.50-7.45 (m, 2H), 7.07 (ddd, J = 8.7, 7.0, 1.6 Hz, 1H), 6.75 (d, J = 7.0 Hz, 1H), 6.34 (td, J = 6.9, 1.7 Hz, 1H), 3.15 (s, 2H), 1.68 (s, 6H). ^{13}C NMR (151 MHz; CDCl_3): δ 175.9, 166.6, 144.2, 137.5, 137.3, 133.5, 133.3, 132.1, 128.2, 128.2, 127.5, 126.4, 126.0, 124.5, 124.2, 112.5, 45.2, 37.9, 29.1.

N-Hydroxy-N-methylbenzamide.³ A suspension of N-methylhydroxylamine hydrochloride (2.02 g, 24 mmol) in THF/ H_2O (40 mL, 10:1) at 0 °C was treated with NaHCO_3 (4.06 g, 48 mmol) and stirred for 10 min, then benzoyl chloride (2.81 mL, 24 mmol) was added dropwise over 10 min. The reaction was stirred overnight without cooling. After dilution with 50 mL of DCM and H_2O the solution was extracted three times with DCM and washed with brine. The yellow oil was purified by column chromatography (20% EtOAc/hexanes to 100% EtOAc) to give a pale yellow oil (2.50 g, 69%). ^1H NMR (400 MHz; DMSO-d_6): δ 9.99 (s, 1H), 7.63-7.56 (m, 2H), 7.48-7.37 (m, 3H), 3.24 (s, 3H).

2,4-Dimethyl-2,4-diphenyl-3-pentanone.⁴ 1,3-diphenylpropan-2-one (2.00 g, 9.5 mmol), dissolved in 30 mL of dry THF was added dropwise over 15 min to a suspension of NaH (2.28 g, 95 mmol) in 30 mL of dry THF under N_2 at 0 °C. After 30 min of stirring, MeI (14.2 g, 100 mmol) was added dropwise and after 30 min the reaction was heated to reflux for 3 h. After cooling and addition of water the mixture was extracted three times with EtOAc, washed with brine, and dried over MgSO_4 and concentrated in vacuo. To also convert remaining trimethylated product the procedure was repeated once more with a third of the reagents. Finally the product was recrystallized from hexanes to give white needles (1.18 g, 46%). ^1H NMR (400 MHz; CDCl_3): δ 7.30-7.27 (m, 2H), 7.26-7.24 (m, 2H), 7.23-7.17 (m, 6H), 1.29 (s, 12H).

Procedure to Determine BDEs Through REqEPR

Electron paramagnetic resonance (EPR) spectra were recorded on a Bruker EMXplus (X-band) spectrometer equipped with an ER 4119HS cavity. The (composite) spectra were fitted and the radical concentration was determined using the quantitative EPR package of the Bruker Xenon software. To generate the radicals 5% (v/v) di-*tert*-butyl peroxide was added to the solutions in benzene (with or without 10% v/v t-BuOH) under nitrogen and the EPR cavity irradiated with a Hamamatsu LC5 Hg-Xe lamp (150 W) via a 3.5 mm quartz light guide. REqEPR experiments with sulfonyl radicals were performed under continuous irradiation.

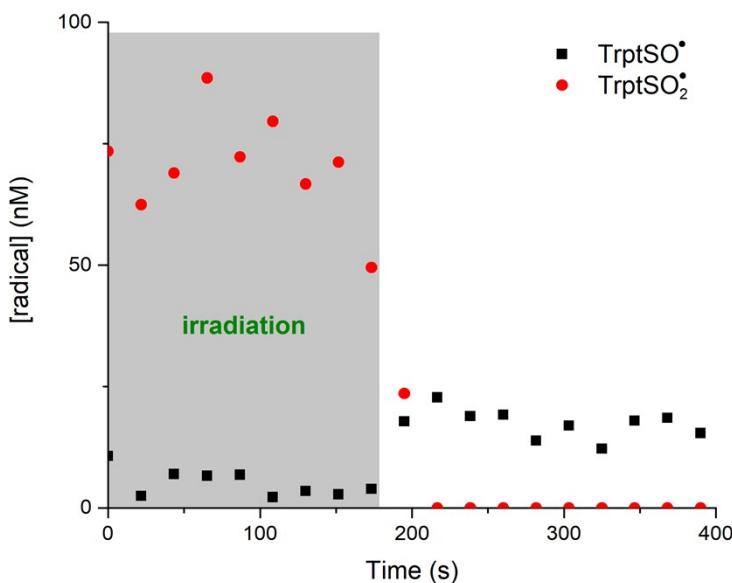


Figure S1. Concentrations of sulfinyl and sulfonyl radicals, determined during and after the irradiation of TrptSO₂H (4 mM) and 5% (v/v) di-*tert*-butyl peroxide in benzene with 10% (v/v) t-BuOH at 25 °C.

During the REqEPR experiments the concentrations of nitroxide and sulfonyl radical were recorded over time under continuous irradiation, which was necessary due to the decay of the sulfonyl radicals. Data after reaching “stable” conditions were processed and averaged. In total 5 concentrations (54 data points) were averaged for TrptSO₂H and 4 concentrations (75 data points) were averaged for PhSO₂H.

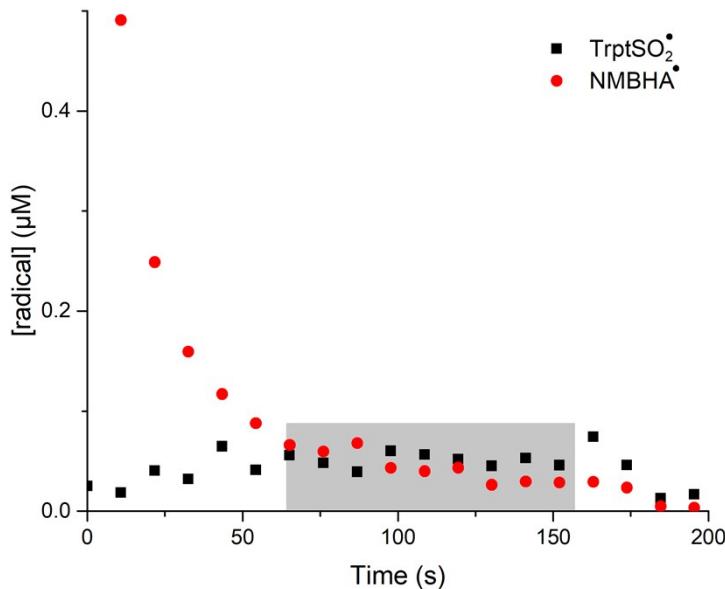


Figure S2. Concentrations of nitroxide and sulfonyl radical, determined during the irradiation of TrptSO₂H (4 mM) and NMBHA (0.5 mM) and 5% (v/v) di-*tert*-butyl peroxide in benzene with 10% (v/v) *t*-BuOH at 25 °C. The grey area represents the region of useful data.

Assuming a negligible reaction entropy for the H-atom exchange and given the known BDE of NMBHA, the BDE for sulfinic acid can be determined. As all involved compounds are relative strong hydrogen bond donors it is important to take this influence on the apparent BDE into account. For this purpose we also re-examined the BDE of NMBHA as at the time no correction had been made.⁵ The important properties to know is the hydrogen bond acidity α_2^H of the analytes as well as the hydrogen bond basicity β_2^H of the solvent system. This influence has been quantified previously⁶ and can be expressed by the following equation:

$$\Delta H \left(\frac{kcal}{mol} \right) = -20.56 \alpha_2^H \beta_2^H + 0.59$$

The β_2^H for benzene is 0.14 and for the mixture of benzene/*t*-BuOH 9:1, which was necessary to solubilize NMBHA and TrptSO₂H, was determined as 0.34. This was achieved via the ¹⁹F NMR method we have described previously.⁷ The α_2^H values

for phenols have been studied previously.⁷ We employed a value of $\alpha_2^H = 0.24$ for BHT,⁸ while NMBHA was determined to have $\alpha_2^H = 0.45$ (via ¹H NMR shifts in CDCl₃ and DMSO-d₆).⁹

Using these values we obtained a new, solvent corrected, BDE for NMBHA of 78.0 ± 0.2 kcal/mol. Referencing this value and employing an $\alpha_2^H = 0.64$ for sulfinic acid (see manuscript) we can determine an O-H BDE of 77.6 kcal/mol for 9-trptycenesulfinic acid. For phenylsulfinic acid a similar correction (in benzene) is necessary, finally yielding a BDE of 78.3 kcal/mol. Both values are in very good agreement with the CBS-QB3 calculations. The direct experimental error from the processed data is 0.2 kcal/mol, although including the error for the reference NMBHA and the solvent effect a more appropriate error is ± 1.0 kcal/mol.

LFP – Details and Results

Table S1. Summary of rate constants (k_H) for HAT by cumyl radicals, determined via photolysis of dicumylketone.

Solvent	k_H (M ⁻¹ s ⁻¹)	
	PhSO ₂ H	TrptSOH
isooctane	-	(1.2 ± 0.2) × 10 ⁷
chlorobenzene	(1.7 ± 0.2) × 10 ⁶	(5.0 ± 0.4) × 10 ⁶
benzene	(1.0 ± 0.2) × 10 ⁶	(2.9 ± 0.4) × 10 ⁶
anisole	(2.3 ± 0.3) × 10 ⁵	(6.1 ± 0.8) × 10 ⁵

Table S2. Summary of rate constants (k_H) for HAT by benzophenone triplet.

Solvent	k_H (M ⁻¹ s ⁻¹)		
	PhSO ₂ H	TrptSO ₂ H	TrptSOH
chlorobenzene	(2.3 ± 0.6) × 10 ⁹	-	-
benzene	(1.7 ± 0.3) × 10 ⁹	-	(2.3 ± 0.1) × 10 ⁹
acetonitrile	(6.5 ± 0.1) × 10 ⁷	(1.8 ± 0.1) × 10 ⁸	(8.6 ± 0.1) × 10 ⁸
ethyl acetate	(2.0 ± 0.1) × 10 ⁷	-	(5.0 ± 0.1) × 10 ⁸

Table S3. Summary of rate constants (k_H) for HAT by cumyloxyl radicals, determined via photolysis of dicumylperoxide.

Solvent	k_H (M ⁻¹ s ⁻¹)	
	PhSO ₂ H	TrptSOH
isooctane	-	(2.3 ± 0.5) × 10 ⁹
chlorobenzene	-	(1.6 ± 0.1) × 10 ⁹
benzene	(6.9 ± 1.3) × 10 ⁸	(1.4 ± 0.2) × 10 ⁹
anisole	-	(6.9 ± 0.2) × 10 ⁸
acetonitrile	-	(3.6 ± 0.1) × 10 ⁸
ethyl acetate	-	(2.3 ± 0.3) × 10 ⁸
dimethyl sulfoxide	-	(5.2 ± 1.0) × 10 ⁷

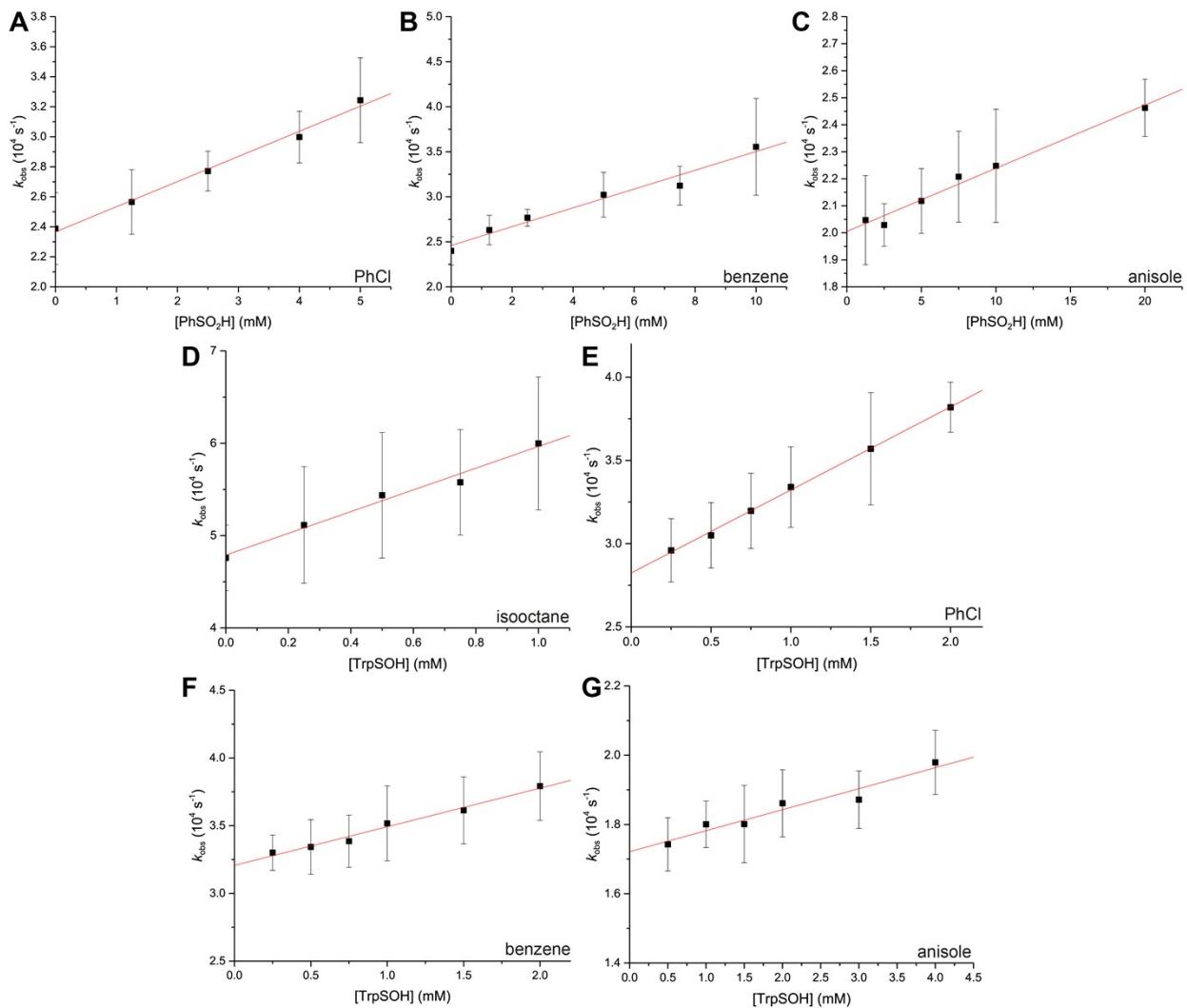


Figure S3. Dependence of the pseudo-first-order rate on the concentration of PhSO_2H (A-C) and TrpSOH (D-G) upon photolysis of dicumylketone (2 mM) at 25 °C in PhCl (A, E), benzene (B, F), anisole (C, G), and isooctane (D).

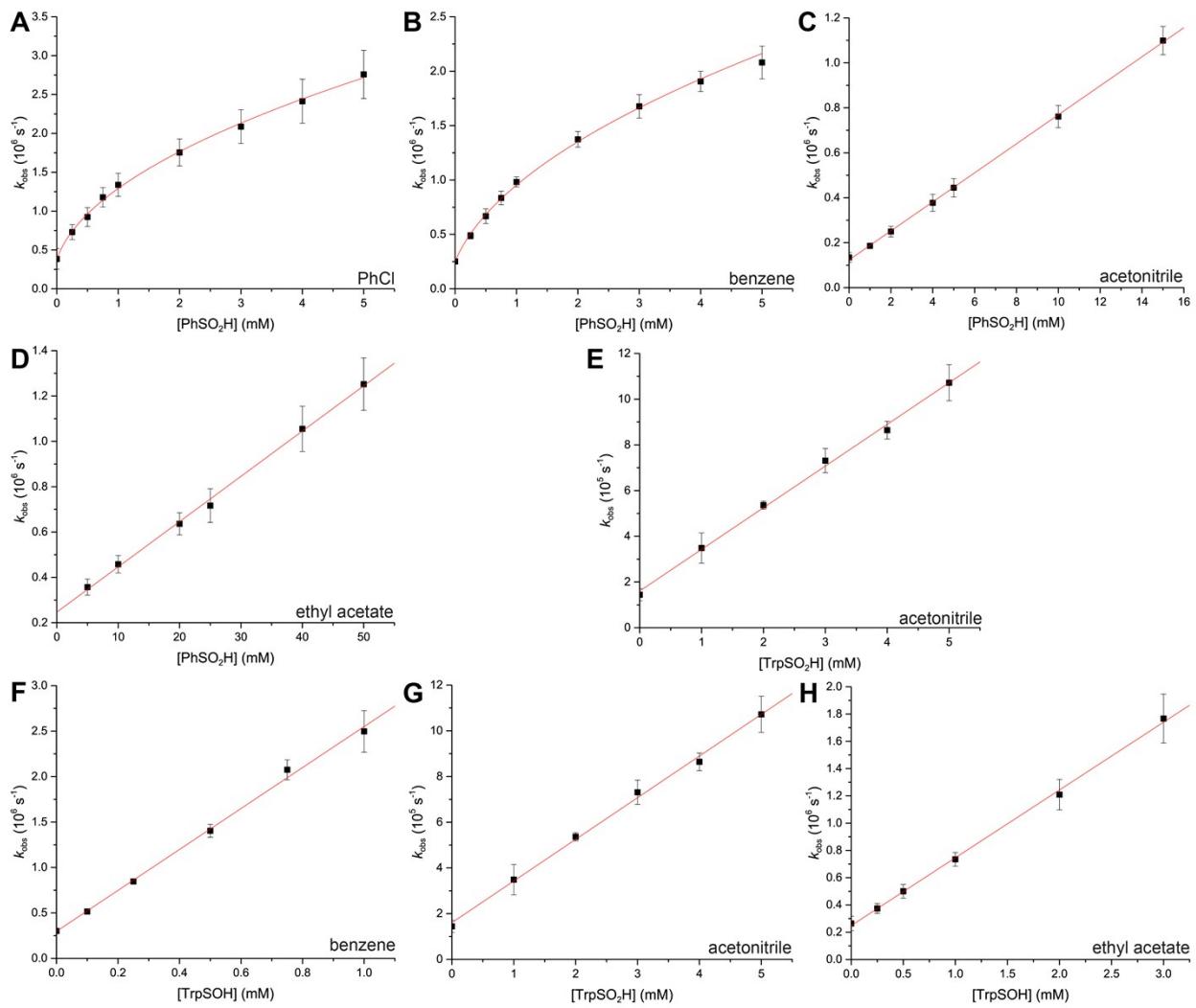


Figure S4. Dependence of the pseudo-first-order rate on the concentration of PhSO_2H (A-D), TrptSO_2H (E), and TrpSOH (F-H) upon photolysis of benzophenone (2 mM) at 25 °C in chlorobenzene (A), benzene (B, F), acetonitrile (C, E, G), and ethyl acetate (D, H).

Generation of cumylperoxyl radicals

LFP experiments (308 nm) were performed under oxygen atmosphere in benzene with between 1 and 3 mM of dicumylketone and the appearance of sulfonyl radical (from sulfinic acid) was monitored at 350 nm. The concentration of PhSO₂H was varied between 1 and 20 mM. No appearance of signal could be observed (also at longer timescales eg. 250 µs). A higher concentration of dicumylketone led to a decrease in radical production because of optical density.

Performing equivalent experiments under nitrogen atmosphere were successful in determining HAT kinetics of cumyl radicals indirectly in a concentration range of 1-5 mM of PhSO₂H.

Transient Absorption Experiments with PhSSPh

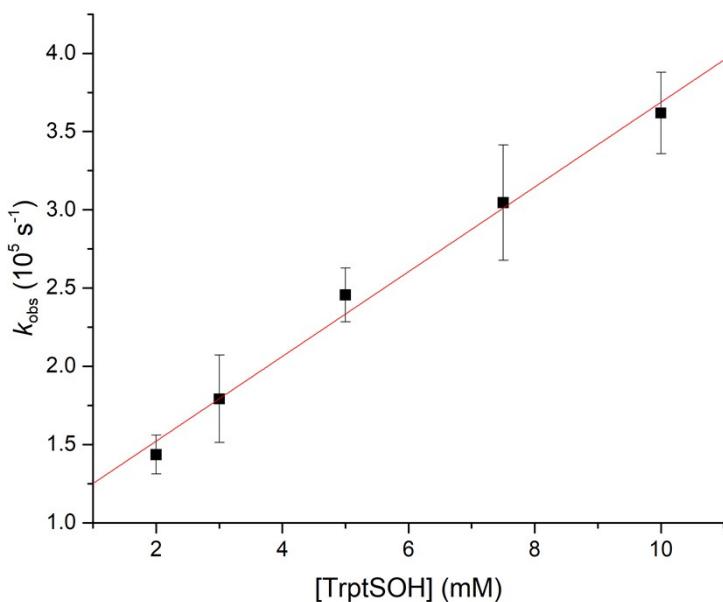


Figure S5. Dependence of the pseudo-first-order rate on the concentration of TrptSOH upon photolysis of diphenyl disulfide in acetonitrile at 25 °C.

As the phenylthiyl radical has a visible absorption (observed at 460 nm) and is easily produced by photolysis of diphenyldisulfide (0.2 mM), it is a convenient probing molecule. While for TrptSOH HAT activity with PhS^\bullet can be observed ($k = (2.7 \pm 0.4) \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$ in acetonitrile – single experiment), no reactivity could be discerned for PhSO_2H . This is also corroborated when calculating the corresponding TS via CBS-QB3. For PhSO_2H $\Delta G^\ddagger = 15.1 \text{ kcal/mol}$ corresponding to a rate constant of $k = 1.2 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$, in addition the reaction is only exergonic by 4.6 kcal/mol. In contrast for sulfenic acids $\Delta G^\circ \sim -14 \text{ kcal/mol}$ and the rate constant predicted near the diffusion limit.

Estimation of the Kinetics of Addition of the Tosyl Radical to Oxygen

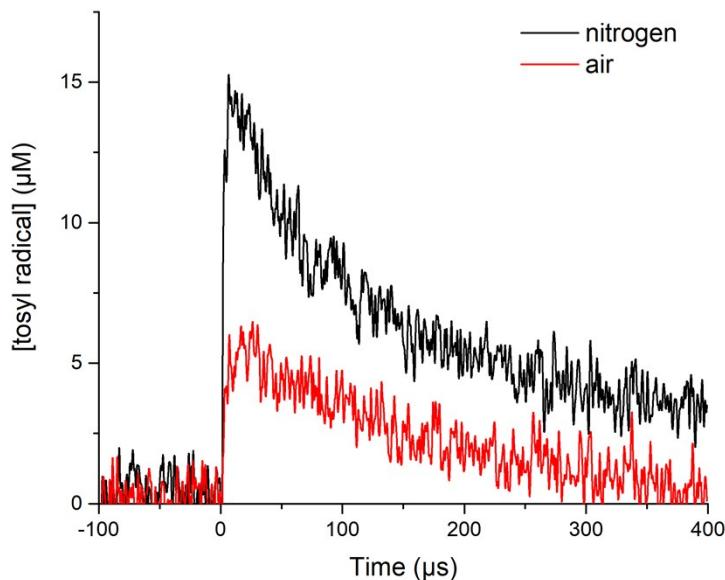


Figure S6. Decays for the tosyl radical (335 nm, $\epsilon = 1200 \text{ M}^{-1} \text{ cm}^{-1}$)¹⁰ under nitrogen (black) and air (red) generated from a solution of 0.2 mM p-toluenesulphonyl iodide¹¹ in benzene at 25 °C.

The following rate constants were determined for the (apparent) disproportionation reaction in second order:

$$\text{nitrogen: } k_{\text{RSO}_2} = (7.3 \pm 0.5) \times 10^8 \text{ M}^{-1} \text{ s}^{-1}$$

$$\text{air: } k_{\text{obs}} = (1.0 \pm 0.2) \times 10^9 \text{ M}^{-1} \text{ s}^{-1}$$

Under the assumption that the difference in second order rate is caused by the reaction with oxygen the following equation can be constructed:

$$k_{\text{obs}}[\text{RSO}_2]^2 = k_{\text{RSO}_2}[\text{RSO}_2]^2 + k_{\text{O}_2}[\text{RSO}_2][\text{O}_2]$$

As $[\text{RSO}_2^\cdot] \sim 6 \mu\text{M}$ at the begin of the decay we can estimate the rate constant $k_{\text{O}_2} = (9 \pm 4) \times 10^5 \text{ M}^{-1} \text{ s}^{-1}$.

Co-autoxidations in Organic Solutions

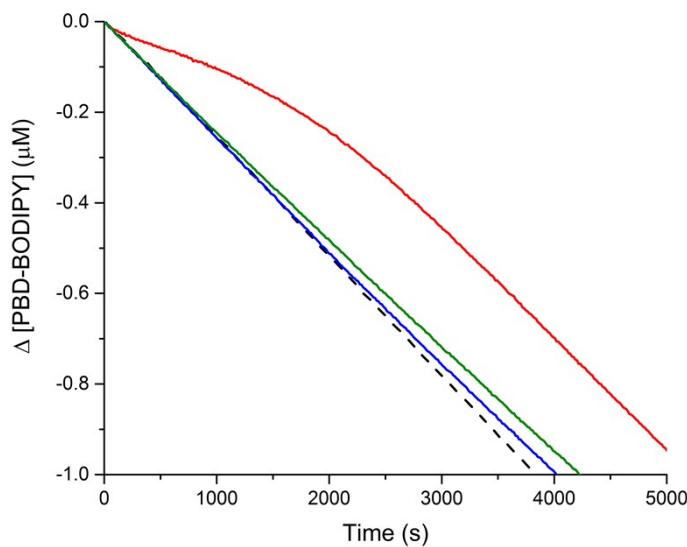


Figure S7. Thermally initiated (AIBN, 6 mM) co-autoxidations of THF (3.1 M) and PBD-BODIPY (10 μ M) at 37 °C in PhCl inhibited by 10 μ M of TrptSOH (red), TrptSO₂H (blue) or PhSO₂H (green). Reaction progress was monitored via absorbance at 588 nm ($\epsilon = 128$ 100 M⁻¹ cm⁻¹).¹²

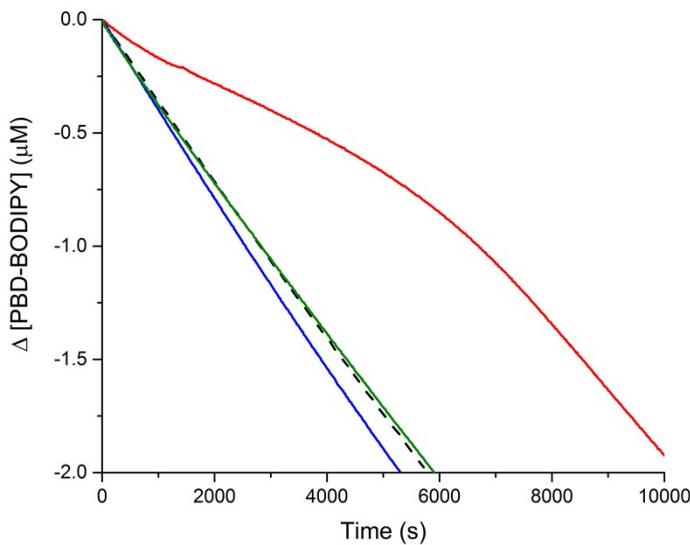


Figure S8. Thermally initiated (AIBN, 6 mM) co-autoxidations of dioxane (2.9 M) and PBD-BODIPY (10 μ M) at 37 °C in PhCl inhibited by 10 μ M of TrptSOH (red), TrptSO₂H (blue) or PhSO₂H (green). Reaction progress was monitored via absorbance at 587 nm ($\epsilon = 123\,000$ M⁻¹ cm⁻¹).¹²

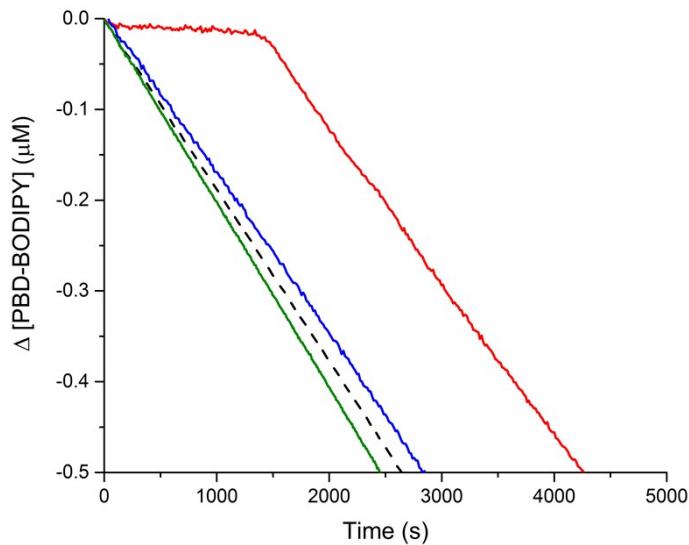


Figure S9. Thermally initiated (AIBN, 6 mM) co-autoxidations of styrene (4.3 M) and PBD-BODIPY (10 μ M) in PhCl at 37 °C inhibited by 10 μ M of TrptSOH (red), TrptSO₂H (blue) or PhSO₂H (green). Reaction progress was monitored via absorbance at 591 nm ($\epsilon = 139\ 000\ M^{-1}\ cm^{-1}$).¹³

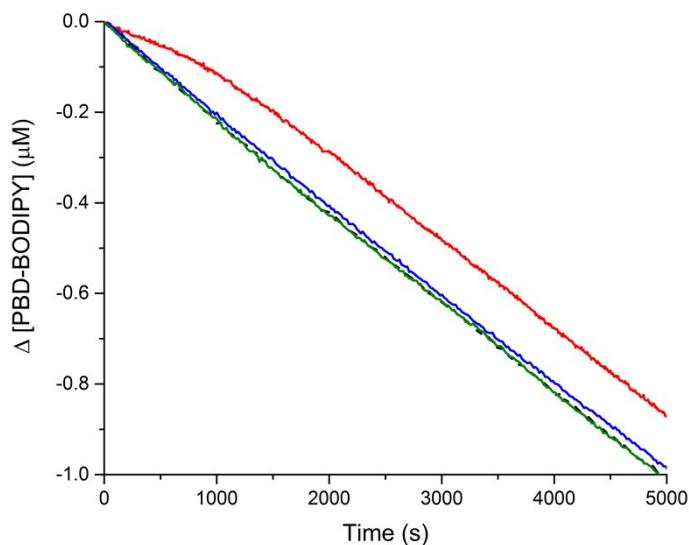


Figure S10. Thermally initiated (AIBN, 6 mM) co-autoxidations of styrene (4.3 M) and PBD-BODIPY (10 μ M) in MeCN at 37 °C inhibited by 10 μ M of TrptSOH (red), TrptSO₂H (blue) or PhSO₂H (green). Reaction progress was monitored via absorbance at 585 nm ($\epsilon = 127\ 860\ M^{-1}\ cm^{-1}$).¹³

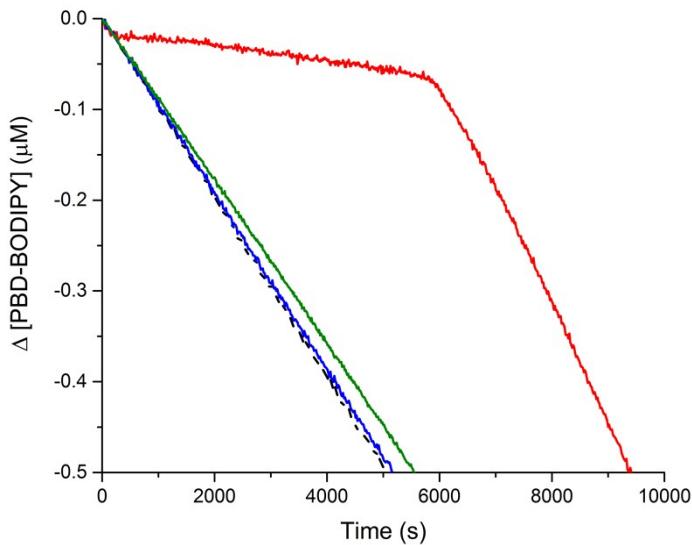


Figure S11. Thermally initiated (AIBN, 6 mM) co-autoxidations of hexadecene (2.8 M) and PBD-BODIPY (10 μ M) at 37 °C in PhCl inhibited by 10 μ M of TrptSOH (red), TrptSO₂H (blue) or PhSO₂H (green). Reaction progress was monitored via absorbance at 588 nm ($\epsilon = 148\ 700\ M^{-1}\ cm^{-1}$).¹²

Summary of Rate Constants for Peroxyl Radicals

Table S4. Summary of rate constants (k_{inh}) and stoichiometry (n) during inhibited co-autoxidations (6 mM AIBN) at 37 °C.

Substrate/Solvent	TrptSOH	
	$k_{inh}\ (M^{-1}\ s^{-1})$	n
THF/PhCl	$(1.2 \pm 0.2) \times 10^5$	0.6 ± 0.1
Dioxane/PhCl	$(2.7 \pm 0.4) \times 10^5$	0.7 ± 0.1
Styrene/PhCl	$(3.6 \pm 0.7) \times 10^6$	0.5 ± 0.05
Styrene/MeCN	$(2.5 \pm 0.6) \times 10^5$	0.5 ± 0.03
Hexadecene/PhCl	$(1.8 \pm 0.4) \times 10^6$	0.7 ± 0.1

Naphthyl-Neophyl Clock to Measure Alkyl Radical Kinetics

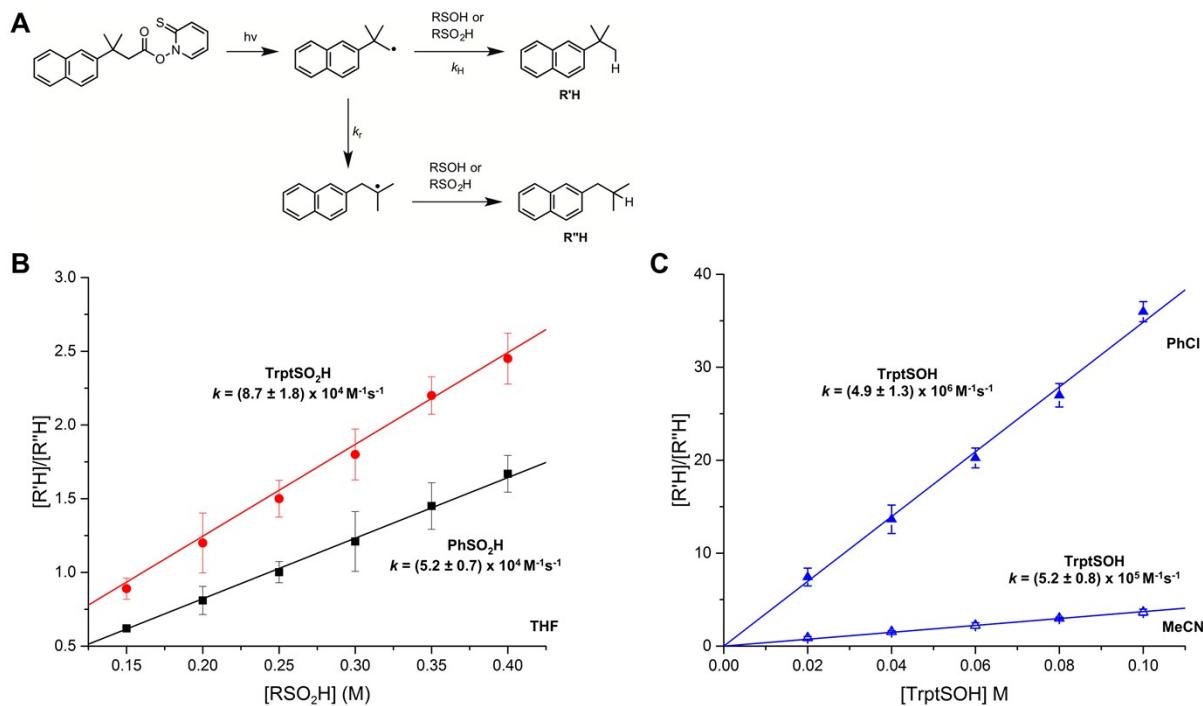
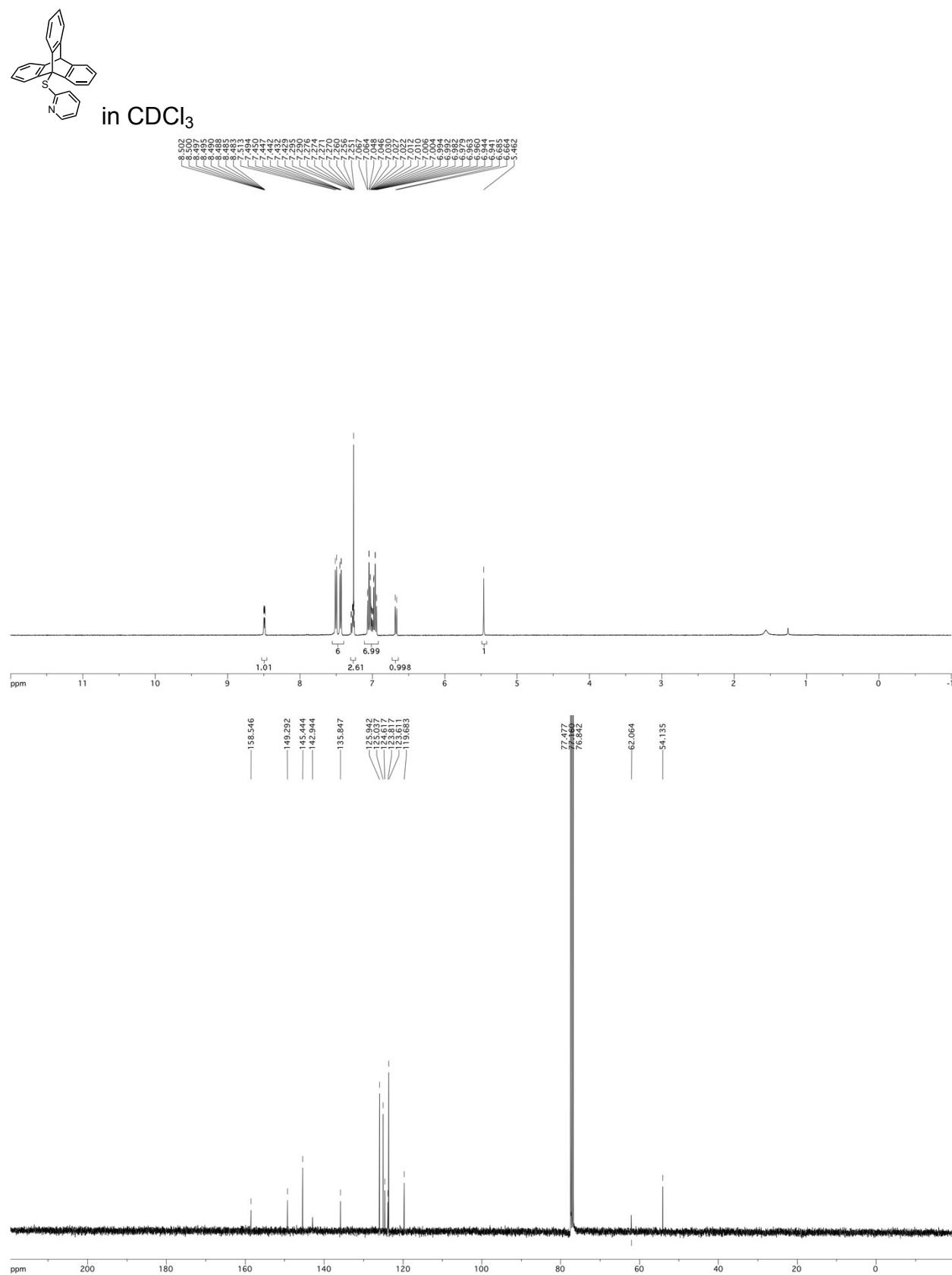
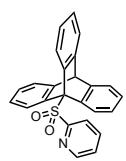


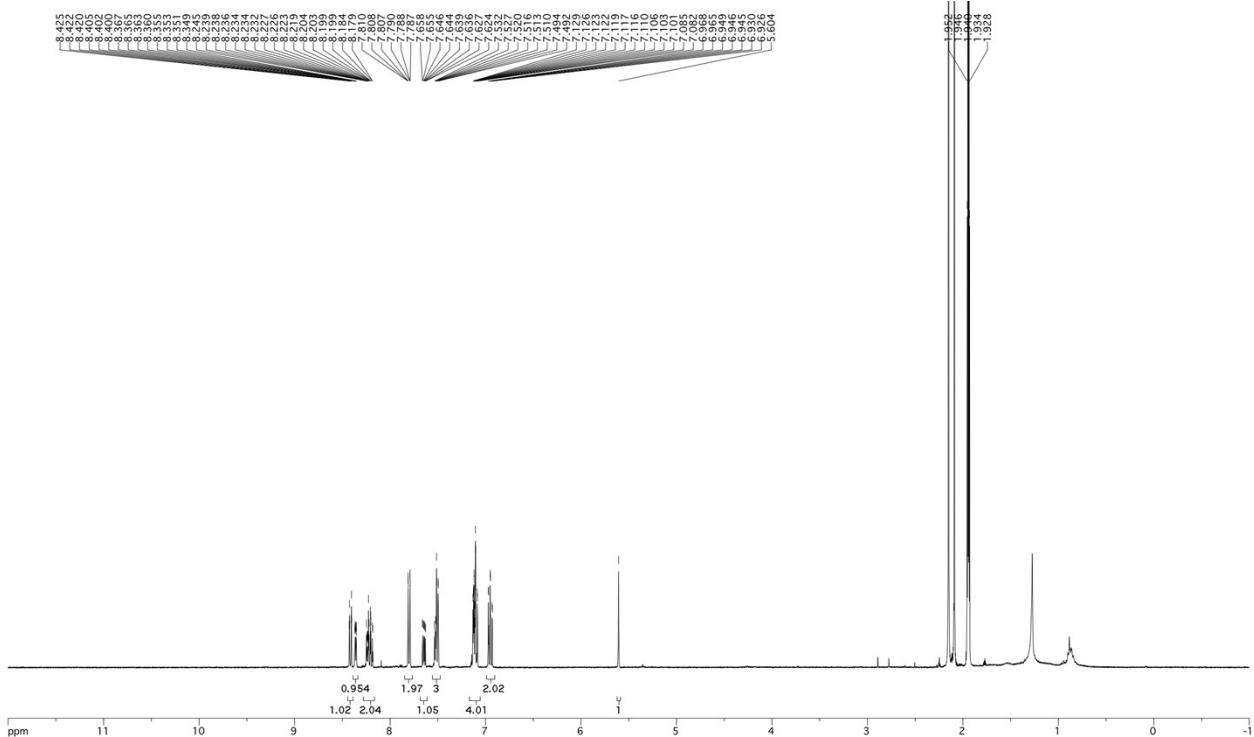
Figure S12. Photolytic generation of the 2-methyl-2-(2-naphthyl)-1-propyl radical as a radical clock to obtain the rate constant for HAT from sulfinic acids and sulfenic to alkyl radicals (A). Ratio of reduced ($R'H$) to rearranged and reduced ($R''H$) products as a function of the concentration of TrptSO_2H (red) or PhSO_2H (blue) during photocleavage of the radical precursor in THF at 25 °C (B). Ratio of reduced ($R'H$) to rearranged and reduced ($R''H$) products as a function of the concentration of TrptSOH during photocleavage of the radical precursor in chlorobenzene or acetonitrile at 25 °C (C).

¹H and ¹³C NMR Spectra

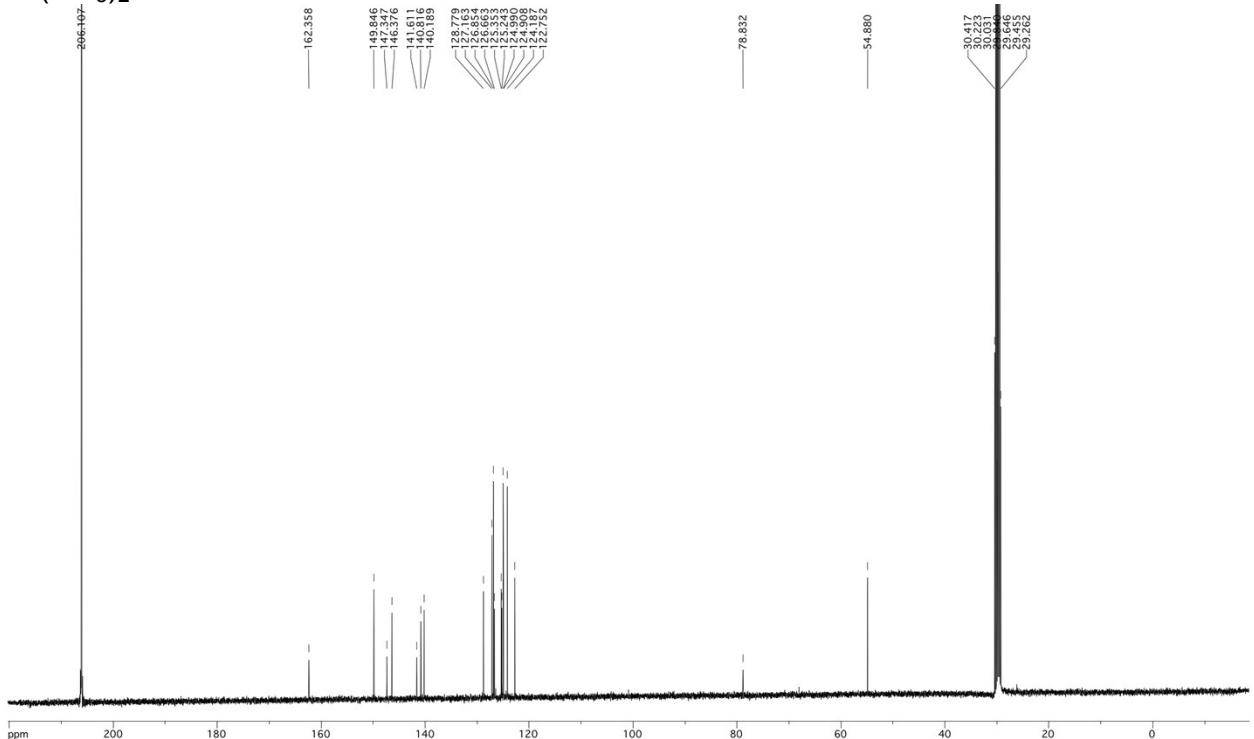


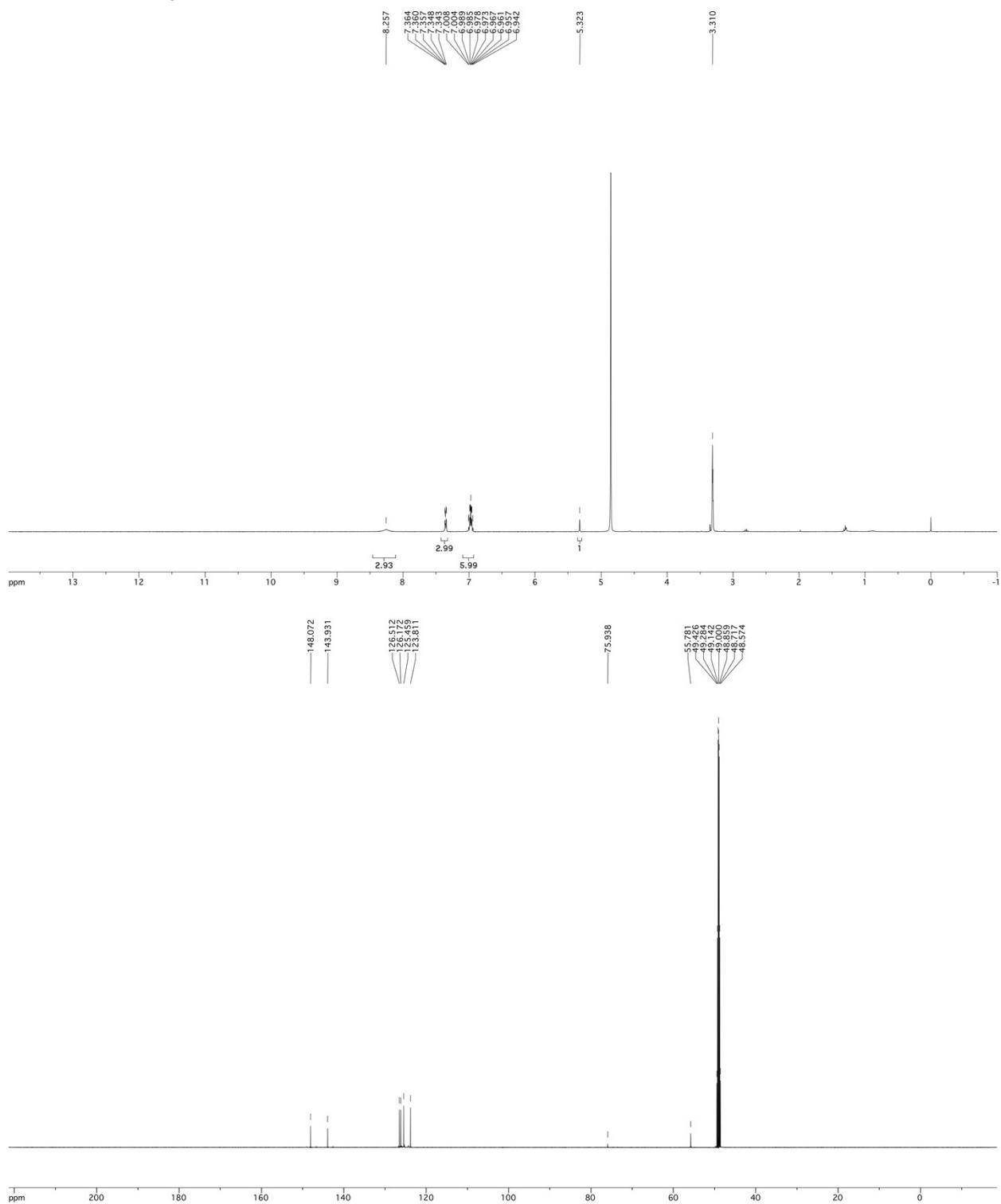
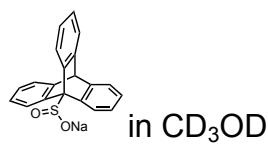


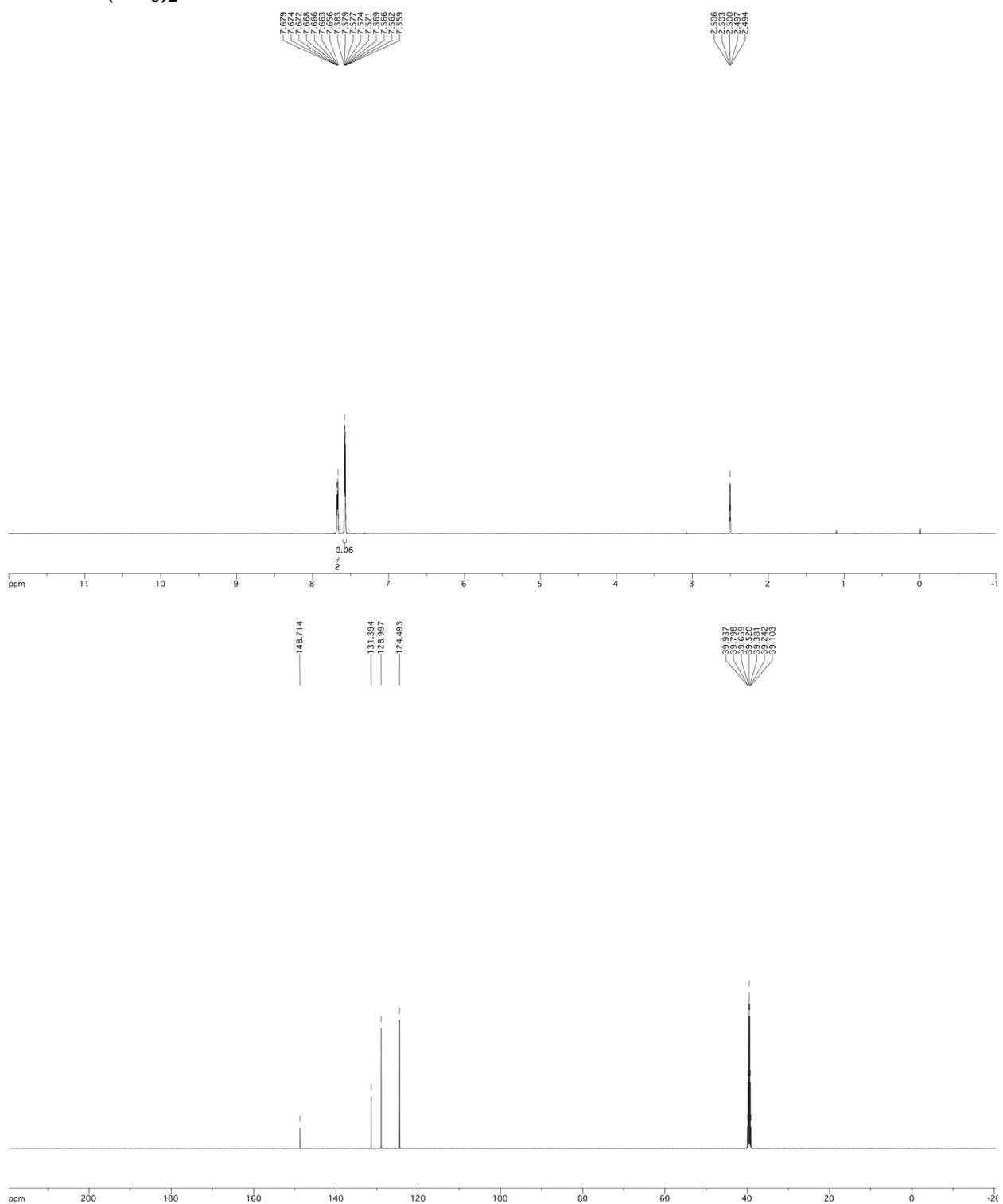
in CD₃CN

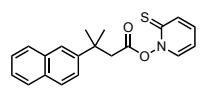


in $(CD_3)_2CO$

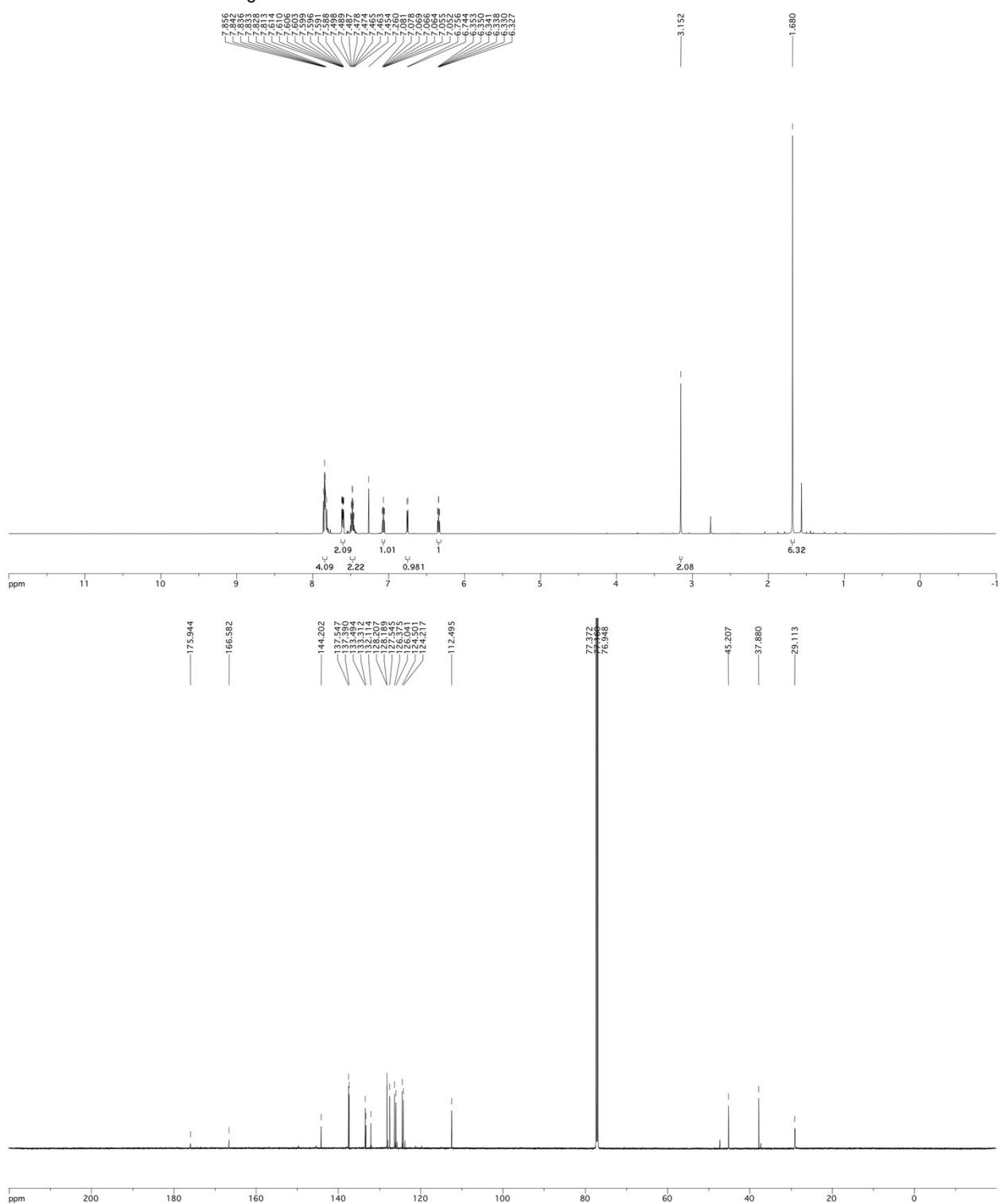








in CDCl_3



Computational Data

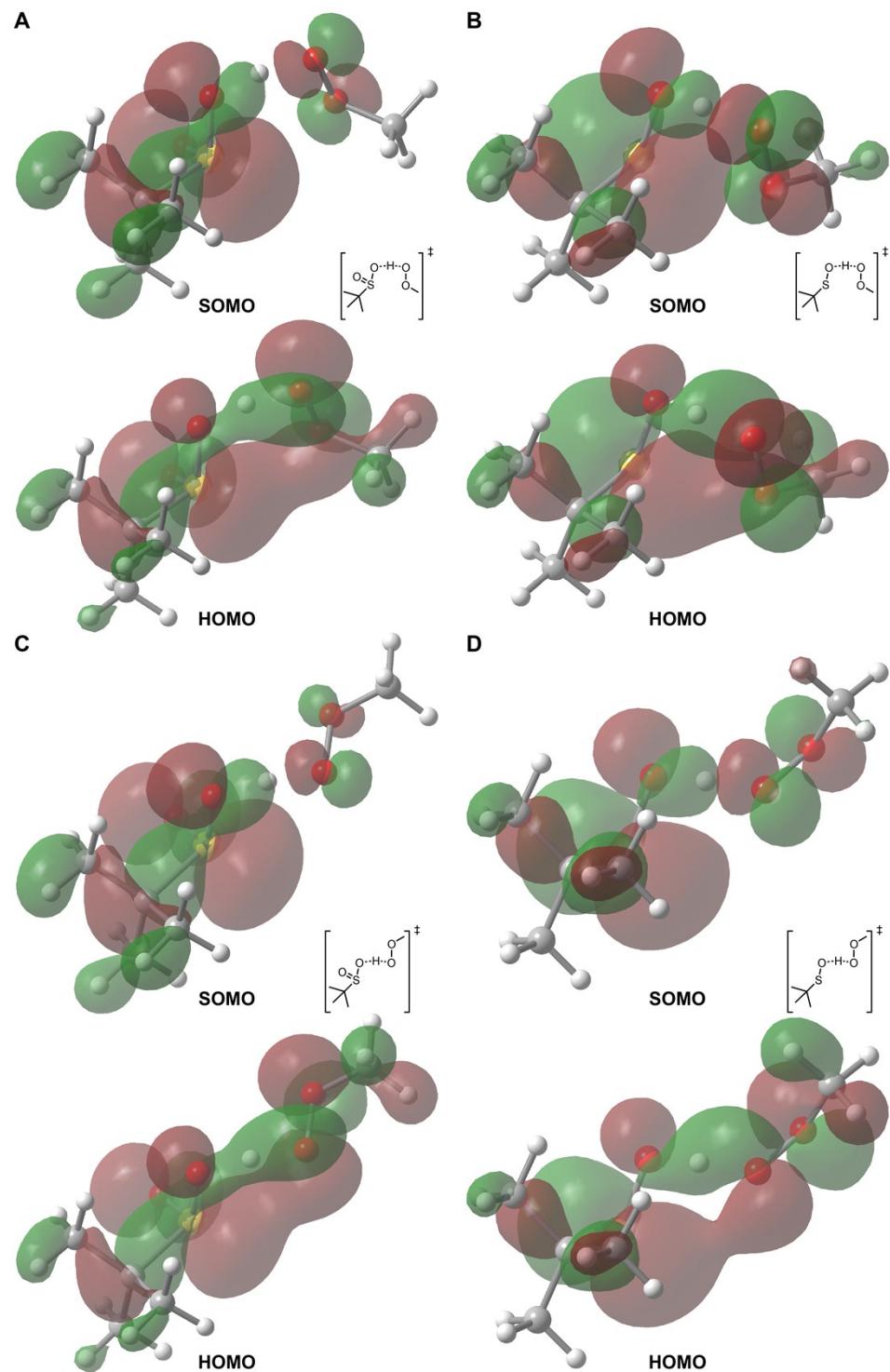


Figure S13. Calculated (CBS-QB3) TS structures for HAT between model sulfenic and sulfinic acid and model peroxy radical (MeOO[·]) and associated free energy barriers and rate constants estimated using transition state theory, SOMO and HOMO visualized.

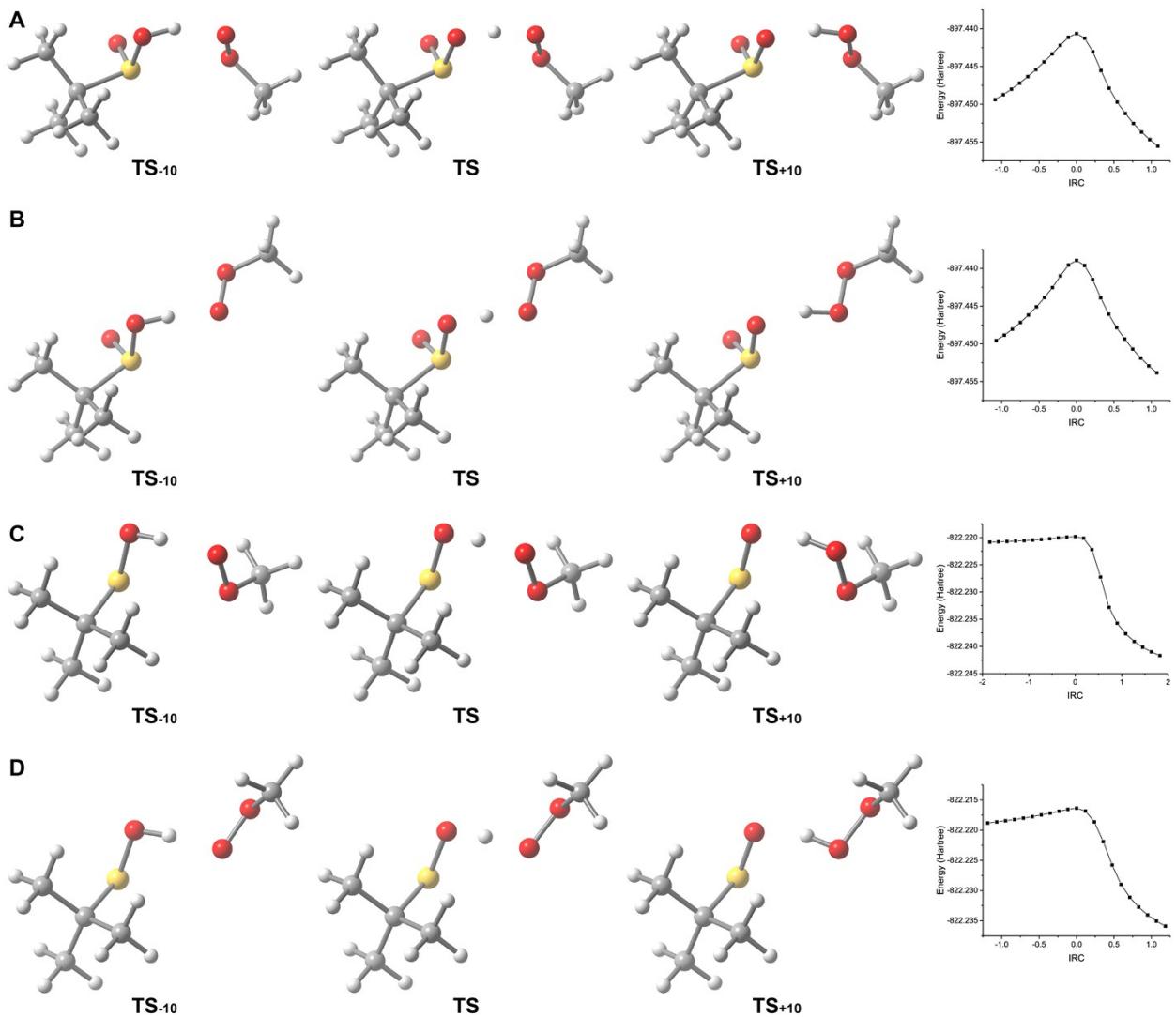
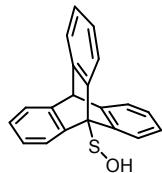


Figure S14. Calculated (CBS-QB3) structures for HAT between model sulfenic and sulfinic acid and model peroxy radical (MeOO^{\cdot}) as well as corresponding IRC energy diagrams. Visualized are the TS as well as structures preceding and following the TS by 10 steps (stepsize = 0.1 Bohr).

Optimized Gaussian Structures and CBS-QB3 Energies

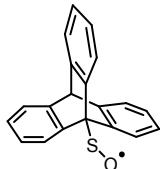
TrptSOH



CBS-QB3 Enthalpy= -1241.996261 CBS-QB3 Free Energy= -1242.056530

0	1		
S	0.62950700	0.07043500	2.57309600
O	2.31774400	0.10013200	2.49552800
C	0.18417100	0.01794200	0.80103000
C	-0.58333200	-0.05865100	-1.68524300
H	-0.90862800	-0.09175900	-2.72516000
C	-1.33885200	-0.17097800	0.64442600
C	-1.74034100	-0.20605800	-0.70049600
C	-3.07736700	-0.36929100	-1.02886500
H	-3.38326500	-0.39786500	-2.06941700
C	-4.02689600	-0.49852000	-0.01063700
H	-5.07359500	-0.62644600	-0.26221500
C	-3.63067900	-0.46494500	1.32120000
H	-4.36740800	-0.56717100	2.10966600
C	-2.28213100	-0.30130800	1.65516100
H	-1.99224500	-0.28249500	2.69952700
C	0.50504600	1.31133700	0.02091700
C	0.10693200	1.25057700	-1.32303500
C	0.82078900	-1.14043300	0.00109100
C	0.40189700	-1.16754400	-1.33709000
C	1.08827800	2.47062500	0.51394200
H	1.33497600	2.55934200	1.56617100
C	0.32046700	2.32699800	-2.17138600
H	0.000594300	2.27436400	-3.20846900
C	1.30741000	3.55548200	-0.34126700
H	1.76359300	4.45942700	0.04602100
C	0.93337400	3.48247500	-1.67830300
H	1.10484900	4.32615800	-2.33705700
C	1.69997800	-2.10693900	0.47193600
H	2.03811600	-2.08730500	1.49822200
C	0.86156900	-2.14957600	-2.20268400
H	0.52872400	-2.16625500	-3.23531800
C	1.74969200	-3.11869800	-1.72983200
H	2.11446100	-3.88959500	-2.39931300
C	2.16295400	-3.09522600	-0.40188900
H	2.85123900	-3.84875500	-0.03600900
H	2.59142500	1.00672800	2.30572400

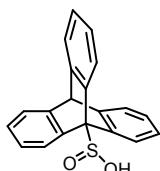
TrptSO[•]



CBS-QB3 Enthalpy= -1241.385689 CBS-QB3 Free Energy= -1241.446231

O 2			
S	-2.62746500	-0.46387600	0.00000000
O	-2.81075600	-1.96475200	0.00000000
C	-0.81432600	-0.13434900	0.00000000
C	1.70533700	0.47371900	0.00000000
H	2.76551300	0.72692800	0.00000000
C	-0.56202800	1.38782100	0.00000000
C	0.80517600	1.70304600	0.00000000
C	1.22313300	3.02510300	0.00000000
H	2.28188100	3.26184400	0.00000000
C	0.27110700	4.04862000	0.00000000
H	0.59171400	5.08415400	0.00000000
C	-1.08397100	3.73983700	0.00000000
H	-1.82199200	4.53363500	0.00000000
C	-1.50767700	2.40667100	0.00000000
H	-2.56920300	2.19131500	0.00000000
C	-0.06641500	-0.69006300	1.22993600
C	1.29453600	-0.34709400	1.21861000
C	-0.06641500	-0.69006300	-1.22993600
C	1.29453600	-0.34709400	-1.21861000
C	-0.58430500	-1.45402400	2.26717000
H	-1.62388800	-1.75324500	2.27088500
C	2.13020600	-0.75456100	2.24770400
H	3.18141100	-0.48619800	2.23540100
C	0.26133300	-1.86307500	3.30340000
H	-0.14050800	-2.46053600	4.11366800
C	1.60736600	-1.51498500	3.29743300
H	2.25521400	-1.83752600	4.10460800
C	-0.58430500	-1.45402400	-2.26717000
H	-1.62388800	-1.75324500	-2.27088500
C	2.13020600	-0.75456100	-2.24770400
H	3.18141100	-0.48619800	-2.23540100
C	1.60736600	-1.51498500	-3.29743300
H	2.25521400	-1.83752600	-4.10460800
C	0.26133300	-1.86307500	-3.30340000
H	-0.14050800	-2.46053600	-4.11366800

TrptSO₂H

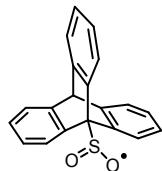


CBS-QB3 Enthalpy= -1317.140039 CBS-QB3 Free Energy= -1317.202009

O 1			
S	0.96445100	-0.58511300	-2.28114900

O	1.36085500	-2.17197500	-1.93965800
O	2.18928900	0.22440800	-2.41747700
C	0.23429900	-0.14940200	-0.61500600
C	-0.87240700	0.47640400	1.64251800
H	-1.33342200	0.74362200	2.59334700
C	-0.99857200	-1.02197800	-0.28257000
C	-1.58399200	-0.67177800	0.94334400
C	-2.71141000	-1.33242600	1.40834800
H	-3.15211400	-1.05367500	2.35971000
C	-3.27587400	-2.35212500	0.63888400
H	-4.15799400	-2.87279300	0.99380500
C	-2.70692600	-2.69560000	-0.58203600
H	-3.14527200	-3.48491300	-1.18184700
C	-1.56655000	-2.03499000	-1.04913100
H	-1.13418300	-2.33418000	-1.99330900
C	1.20879900	-0.26049100	0.57604300
C	0.58250100	0.04950900	1.79478900
C	-0.27324200	1.30839800	-0.58633500
C	-0.87254900	1.63019100	0.64247700
C	2.56672400	-0.56980100	0.55754800
H	3.09569400	-0.69553300	-0.37778300
C	1.29073400	-0.00520400	2.98573100
H	0.79910900	0.23727800	3.92200300
C	3.27524000	-0.62926000	1.76296500
H	4.33176100	-0.87057600	1.74524600
C	2.64213600	-0.36119900	2.97025000
H	3.19956400	-0.40675600	3.89880500
C	-0.19183300	2.26542900	-1.59032500
H	0.29917900	2.05124100	-2.53055900
C	-1.39636500	2.89494600	0.86088400
H	-1.85687000	3.13747200	1.81282700
C	-1.32492600	3.85356100	-0.15392700
H	-1.73272100	4.84451000	0.01002600
C	-0.72578500	3.53905600	-1.36786300
H	-0.66221400	4.28571800	-2.15097300
H	1.81708700	-2.23850200	-1.08535800

TrptSO₂:

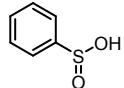


CBS-QB3 Enthalpy= -1316.520376 CBS-QB3 Free Energy= -1316.582936

0 2			
S	2.25674300	1.17282800	0.00000000
O	2.35399200	1.89950000	1.27051500
O	2.35399200	1.89950000	-1.27051500
C	0.60037700	0.26659700	0.00000000
C	-1.66770600	-0.94409700	0.00000000
H	-2.62921400	-1.45697200	0.00000000
C	0.43061600	-0.64159900	1.23131200
C	-0.81379600	-1.29276600	1.21432900
C	-1.16250200	-2.16924300	2.23051400
H	-2.12556700	-2.66821900	2.21175800
C	-0.26389000	-2.40592700	3.27392600
H	-0.53058500	-3.08992900	4.07145100
C	0.96908000	-1.76460800	3.28932100

H	1.66504900	-1.94586500	4.10012400
C	1.32361700	-0.87648400	2.26903900
H	2.27691600	-0.36802000	2.31166100
C	-0.58115800	1.25322600	0.00000000
C	-1.80930400	0.57335200	0.00000000
C	0.43061600	-0.64159900	-1.23131200
C	-0.81379600	-1.29276600	-1.21432900
C	-0.54686200	2.64024100	0.00000000
H	0.39442500	3.17499700	0.00000000
C	-3.00024200	1.28458400	0.00000000
H	-3.94833800	0.75729700	0.00000000
C	-1.74979400	3.35181700	0.00000000
H	-1.72606500	4.43521100	0.00000000
C	-2.96797600	2.68125400	0.00000000
H	-3.89569600	3.24193900	0.00000000
C	1.32361700	-0.87648400	-2.26903900
H	2.27691600	-0.36802000	-2.31166100
C	-1.16250200	-2.16924300	-2.23051400
H	-2.12556700	-2.66821900	-2.21175800
C	-0.26389000	-2.40592700	-3.27392600
H	-0.53058500	-3.08992900	-4.07145100
C	0.96908000	-1.76460800	-3.28932100
H	1.66504900	-1.94586500	-4.10012400

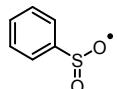
PhSO₂H



CBS-QB3 Enthalpy= -779.808964 CBS-QB3 Free Energy= -779.851846

0 1			
C	-2.07554500	1.24041300	0.27880600
C	-2.84886600	0.09026500	0.13171700
C	-2.24532000	-1.13054500	-0.17478000
C	-0.86459100	-1.20441300	-0.33275700
C	-0.10237500	-0.04702700	-0.17603200
C	-0.69234900	1.17745900	0.11875200
S	1.70094700	-0.11651200	-0.42011900
O	2.17140900	1.26122600	-0.18832200
O	2.05481700	-1.07129600	0.91512800
H	-2.54840500	2.18743200	0.51257100
H	-3.92503700	0.14400100	0.25071800
H	-2.85058300	-2.02218400	-0.29117000
H	-0.38533800	-2.14888800	-0.56613700
H	-0.06244400	2.05396400	0.21599900
H	1.72113300	-0.62648200	1.71122800

PhSO₂•



CBS-QB3 Enthalpy= -779.188467 CBS-QB3 Free Energy= -779.230967

0 2			
S	-0.14498200	-1.71918400	0.00000000
O	0.38866500	-2.18270100	1.28520400

O	0.38866500	-2.18270100	-1.28520400
C	-0.08871400	0.09140600	0.00000000
C	-0.09682600	0.76414100	1.21976900
C	-0.09682600	2.15630100	1.21042300
C	-0.09903000	2.84916700	0.00000000
C	-0.09682600	2.15630100	-1.21042300
C	-0.09682600	0.76414100	-1.21976900
H	-0.08335800	0.20566300	2.14670100
H	-0.09050400	2.69852600	2.14865000
H	-0.10092700	3.93303700	0.00000000
H	-0.09050400	2.69852600	-2.14865000
H	-0.08335800	0.20566300	-2.14670100

tert-BuSOH



CBS-QB3 Enthalpy= -630.958987 CBS-QB3 Free Energy= -631.000322

0 1			
S	0.96312700	-0.81679200	-0.01806700
O	2.06273300	0.47277900	0.11035400
H	2.31178300	0.71140400	-0.79044800
C	-0.65925900	0.08175600	0.00081900
C	-0.75607400	1.02495600	-1.20368700
H	-1.71881200	1.54723100	-1.19937300
H	0.02878200	1.78461900	-1.16728400
H	-0.66674800	0.47658900	-2.14438100
C	-0.79151300	0.85864900	1.31598000
H	-1.72416300	1.43351100	1.32524400
H	-0.79347400	0.18097800	2.17218900
H	0.04178200	1.55396700	1.43455700
C	-1.70311000	-1.04069600	-0.09297000
H	-2.70742000	-0.60761700	-0.05936000
H	-1.61009400	-1.60079900	-1.02733100
H	-1.61379600	-1.74144000	0.74157800

tert-BuSO[•]



CBS-QB3 Enthalpy= -630.352256 CBS-QB3 Free Energy= -630.393921

0 2			
S	0.51173100	-1.16329100	0.00000000
O	-0.78578900	-1.94371900	0.00000000
C	0.03808900	0.64692500	0.00000000
C	-0.78578900	0.91613000	1.26438000
H	-1.13698900	1.95312400	1.26798400
H	-1.65600000	0.25765300	1.30066700
H	-0.19433900	0.75403800	2.16921800
C	-0.78578900	0.91613000	-1.26438000
H	-1.13698900	1.95312400	-1.26798400
H	-0.19433900	0.75403800	-2.16921800
H	-1.65600000	0.25765300	-1.30066700
C	1.36361000	1.41533800	0.00000000

H	1.16776500	2.49183800	0.00000000
H	1.96238800	1.18689600	0.88633300
H	1.96238800	1.18689600	-0.88633300

tert-BuSO₂H



CBS-QB3 Enthalpy= -706.108573 CBS-QB3 Free Energy= -706.152201

0 1			
S	-0.92589900	-0.16834000	-0.53899500
O	-1.47732300	-1.28369500	0.26062200
O	-1.56236000	1.26196700	0.07162600
H	-1.70818900	1.12632100	1.02202500
C	0.85752800	0.03255100	0.04508900
C	0.87109900	0.13797800	1.56877000
H	0.28738500	-0.67173600	2.01321600
H	1.89778700	0.06280900	1.93883500
H	0.47051700	1.09625400	1.90999000
C	1.55490200	-1.24739200	-0.43428900
H	1.47295400	-1.36727600	-1.51914800
H	2.61937300	-1.20092500	-0.18577300
H	1.12656400	-2.13130500	0.04145100
C	1.42301900	1.27520200	-0.64535300
H	2.47867100	1.39634500	-0.38317400
H	1.36001100	1.19037700	-1.73436100
H	0.88748900	2.17637000	-0.34241800

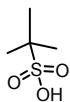
tert-BuSO₂[•]



CBS-QB3 Enthalpy= -705.490439 CBS-QB3 Free Energy= -705.534391

0 2			
S	1.02117400	0.00000000	-0.35884500
O	1.55160700	-1.27340100	0.14666100
O	1.55160800	1.27340100	0.14666100
C	-0.88766000	0.00000000	0.02506900
C	-0.98454500	-0.00001400	1.55053300
H	-2.04220000	-0.00001700	1.83515600
H	-0.51389300	-0.88752900	1.97638200
H	-0.51389300	0.88749100	1.97639800
C	-1.44106600	-1.27237200	-0.60617000
H	-2.51428700	-1.33387100	-0.39801000
H	-1.31040800	-1.27367600	-1.69111300
H	-0.95957500	-2.16115400	-0.19728000
C	-1.44106100	1.27238700	-0.60614500
H	-1.31040300	1.27371100	-1.69108800
H	-2.51428200	1.33388600	-0.39798500
H	-0.95956800	2.16115800	-0.19723700

tert-BuSO₃H



CBS-QB3 Enthalpy= -781.277017 CBS-QB3 Free Energy= -781.321987

	0	1	
S	0.83821800	-0.16226800	-0.07101000
O	1.30530200	0.09337000	-1.41537600
O	1.34359500	1.10444600	0.83944700
H	1.82015700	1.70201900	0.24558700
C	-0.99040200	0.03742300	0.00632300
C	-1.35656500	1.39438000	-0.60614500
H	-0.98518100	1.48007800	-1.62827800
H	-2.44561500	1.49048600	-0.62685200
H	-0.95688000	2.21886000	-0.01330300
C	-1.55925800	-1.12319500	-0.82674700
H	-1.26571000	-2.08836300	-0.41134800
H	-2.65063900	-1.06065700	-0.81692900
H	-1.22042500	-1.07074200	-1.86287700
C	-1.41991400	-0.06326900	1.47533800
H	-2.50915600	0.01980300	1.52957800
H	-1.12375700	-1.01931900	1.90782200
H	-0.98447500	0.73912000	2.07279100
O	1.20948300	-1.35869500	0.63184800

tert-BuSO₃[•]



CBS-QB3 Enthalpy= -780.608433 CBS-QB3 Free Energy= -780.654740

	0	2	
S	-0.85925100	0.00002500	-0.00000300
O	-1.32610700	1.34424800	-0.38155600
O	-1.32606800	-1.00255800	-0.97334600
C	0.97380000	-0.00000500	0.00001000
C	1.42934400	0.35866600	-1.42175300
H	1.07283900	1.34717900	-1.71600300
H	2.52218400	0.36574800	-1.44962700
H	1.07290100	-0.37205200	-2.14964300
C	1.42939400	1.05193400	1.02151300
H	1.07272900	0.81261300	2.02468300
H	2.52223800	1.07238100	1.04173200
H	1.07314300	2.04770400	0.75253200
C	1.42929900	-1.41065000	0.40026200
H	2.52213200	-1.43840800	0.40786500
H	1.07301900	-1.67559500	1.39710200
H	1.07257000	-2.15971500	-0.30859800
O	-1.32616900	-0.34167900	1.35487800

tert-butyl radical



CBS-QB3 Enthalpy= -157.425524 CBS-QB3 Free Energy= -157.462444

0	2			
C		0.00001200	-0.00003700	-0.14928700
C		-1.13936400	-0.95436600	0.01501900
C		1.39621300	-0.50948600	0.01473600
H		-1.36709400	-1.14542200	1.07940200
H		-0.92133500	-1.92825200	-0.43495800
H		-2.06000800	-0.56867600	-0.43453000
H		1.67570700	-0.61107100	1.07907900
H		1.52256800	-1.49965800	-0.43479600
H		2.13046400	0.16629100	-0.43542600
C		-0.25685800	1.46386100	0.01484300
H		0.53744900	2.06834700	-0.43473500
H		-1.20929200	1.76198800	-0.43512700
H		-0.30847700	1.75661200	1.07922500

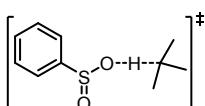
tert-Butane



CBS-QB3 Enthalpy= -158.077849 CBS-QB3 Free Energy= -158.111186

0	1			
C		0.00000000	0.00000000	0.37225800
C		0.00000000	1.46129200	-0.09577300
C		1.26551600	-0.73064600	-0.09577300
H		0.00000000	0.00000000	1.47022700
H		0.00000000	1.51929200	-1.18993700
H		0.88410000	1.99527900	0.26499300
H		-0.88410000	1.99527900	0.26499300
H		1.31574600	-0.75964600	-1.18993700
H		2.17001200	-0.23198600	0.26499300
H		1.28591200	-1.76329300	0.26499300
C		-1.26551600	-0.73064600	-0.09577300
H		-1.28591200	-1.76329300	0.26499300
H		-2.17001200	-0.23198600	0.26499300
H		-1.31574600	-0.75964600	-1.18993700

PhSO₂H H-abstraction by *tert*-butyl radical TS

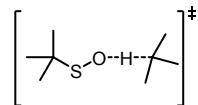


TS freq.: -1582.62 cm⁻¹
 CBS-QB3 Enthalpy= -937.237245 CBS-QB3 Free Energy= -937.297834

0	2			
C		-2.93675900	-0.45203400	1.25594700
C		-3.34070400	-1.31830400	0.24160100
C		-2.72517000	-1.26993100	-1.01007300
C		-1.70354900	-0.35489600	-1.25031800
C		-1.30213900	0.49897300	-0.22364500

C	-1.91530200	0.46839400	1.02483200
S	-0.02185600	1.76016400	-0.54525800
O	0.34844300	2.28621400	0.78648400
O	1.11881000	0.87432600	-1.20512700
H	-3.41687900	-0.49068700	2.22723500
H	-4.13837500	-2.02960900	0.42343800
H	-3.04164600	-1.94396600	-1.79801100
H	-1.21423000	-0.31175200	-2.21650600
H	-1.57947100	1.15237300	1.79473600
H	1.64398100	0.00931100	-0.48337200
C	3.73882200	-0.23887400	0.12817900
H	3.75356300	0.74437700	0.60334800
H	4.07528000	-0.12816600	-0.90546100
C	2.36330300	-0.87019500	0.20473100
C	2.20263300	-2.15389400	-0.58282700
H	2.79898000	-2.95380900	-0.11973100
H	1.16278500	-2.48901900	-0.60026900
H	2.54928800	-2.04169000	-1.61318000
C	1.74564700	-0.87903700	1.58873700
H	1.72390200	0.12242000	2.02189500
H	0.72544800	-1.26888800	1.57376800
H	2.33823300	-1.52572300	2.25244700
H	4.47011600	-0.87333600	0.64995900

tert-BuSOH H-abstraction by *tert*-butyl radical TS



TS freq.: -1692.93 cm⁻¹
 CBS-QB3 Enthalpy= -788.388783 CBS-QB3 Free Energy= -788.448835

0 2			
S	-0.91188100	-1.15284800	0.30074000
H	1.19318700	-0.36829100	-0.37122600
C	-2.10567600	0.23927600	-0.01670400
C	2.45681500	0.15291200	0.05056400
C	2.24565100	0.50784700	1.50306900
H	1.53056900	1.32530200	1.62405800
H	1.88732200	-0.34773000	2.08222900
O	0.27452500	-0.89759100	-0.79545000
C	-1.43814700	1.58819600	0.26892200
H	-0.57078500	1.73493000	-0.37821900
H	-2.14183100	2.40525400	0.07520600
H	-1.11162700	1.65662400	1.30915000
C	-2.57647400	0.16245100	-1.47381200
H	-3.11133100	-0.77050900	-1.66483300
H	-3.24656200	0.99903600	-1.70094700
H	-1.72305000	0.20958600	-2.15306200
C	-3.26023100	-0.02377800	0.95986400
H	-2.92510900	0.01822200	1.99990200
H	-4.03458400	0.73779300	0.82570700
H	-3.71877300	-1.00076600	0.78440900
C	2.69235300	1.32036600	-0.88098600
H	3.68493500	1.76197200	-0.70072000
H	2.66141400	1.01178800	-1.92942400
H	1.95508500	2.11418400	-0.73440700
C	3.34047900	-1.04997300	-0.19512000
H	3.03851800	-1.90327500	0.41743900
H	3.31710700	-1.35869700	-1.24339500

H	4.38657300	-0.81691000	0.05682400
H	3.19422000	0.83400100	1.95829600

tert-BuO[•]

CBS-QB3 Enthalpy= -232.559991 CBS-QB3 Free Energy= -232.597221

O 2			
O	-0.51504300	1.35749900	0.00000000
C	-0.06471700	0.05489400	0.00000000
C	-0.51504300	-0.67961200	1.27723900
H	-0.18332600	-0.13225500	2.16166200
H	-1.60556800	-0.74508500	1.30469300
H	-0.11054400	-1.69502700	1.31475700
C	-0.51504300	-0.67961200	-1.27723900
H	-0.18332600	-0.13225500	-2.16166200
H	-0.11054400	-1.69502700	-1.31475700
H	-1.60556800	-0.74508500	-1.30469300
C	1.48764400	0.22435400	0.00000000
H	1.94050400	-0.77022900	0.00000000
H	1.81084000	0.76741200	-0.88873800
H	1.81084000	0.76741200	0.88873800

tert-Butanol

CBS-QB3 Enthalpy= -233.227909 CBS-QB3 Free Energy= -233.264675

O 1			
O	-0.48653800	1.36731600	0.00000000
C	0.00169900	0.01261900	0.00000000
C	-0.48653800	-0.70694000	1.26469500
H	-0.16787500	-0.15928600	2.15447800
H	-1.58016300	-0.76856600	1.27614400
H	-0.09557500	-1.72686300	1.32338500
C	-0.48653800	-0.70694000	-1.26469500
H	-0.16787500	-0.15928600	-2.15447800
H	-0.09557500	-1.72686300	-1.32338500
H	-1.58016300	-0.76856600	-1.27614400
C	1.52451200	0.14733300	0.00000000
H	2.00503600	-0.83417700	0.00000000
H	1.85226800	0.69819500	-0.88444200
H	1.85226800	0.69819500	0.88444200
H	-1.44885700	1.33225500	0.00000000

MeOO[•]

CBS-QB3 Enthalpy= -189.954731 CBS-QB3 Free Energy= -189.985244

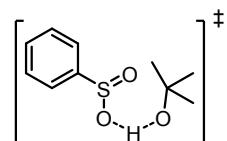
O 2			
C	1.09605800	-0.18318200	0.00000000
H	1.87467600	0.57860400	0.00000800
H	1.14883700	-0.80072600	0.89698200
H	1.14884100	-0.80070700	-0.89699400
O	-0.15733600	0.54388500	0.00000000
O	-1.18625200	-0.27864500	0.00000000

MeOOH

CBS-QB3 Enthalpy= -190.589542 CBS-QB3 Free Energy= -190.620447

0	1			
C		1.12961700	-0.22363900	0.02672600
H		1.97292200	0.47126800	0.02466300
H		1.14400300	-0.82582300	0.94203000
H		1.18954600	-0.87767300	-0.84877800
O		-0.01619300	0.60684200	-0.03138200
O		-1.16412700	-0.28550700	-0.09072700
H		-1.64161500	0.00338800	0.69860100

PhSO₂H + *tert*-BuO[•] TS

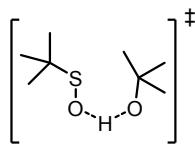


TS freq.: -1156.36 cm⁻¹

CBS-QB3 Enthalpy = -1012.376400 CBS-QB3 Free Energy= -1012.438625

0	2			
S		-0.53043300	1.85129200	-0.05562700
O		0.11774400	1.85327800	1.27683700
O		0.45755600	1.41803500	-1.21298200
H		0.98165100	0.45531200	-1.09817600
O		1.54325400	-0.69223800	-0.83259800
C		2.65377100	-0.70706300	0.00650900
C		2.20281200	-0.86694300	1.47377200
H		1.63265700	0.01624800	1.77020400
H		1.56451200	-1.74674700	1.57655900
H		3.06966600	-0.97658800	2.13116100
C		3.41648700	-1.99735200	-0.43978500
H		3.72301500	-1.90940300	-1.48270700
H		4.30163500	-2.11160100	0.19050800
H		2.77466800	-2.87152400	-0.32798400
C		3.54115600	0.53058800	-0.18710600
H		4.44884300	0.45849300	0.41750300
H		3.82794700	0.63347000	-1.23634700
H		3.00132400	1.43077000	0.11352900
C		-1.68545800	0.44595600	-0.03361500
C		-2.05440900	-0.09973700	1.19211200
C		-2.24085500	-0.00465300	-1.23044700
C		-2.98239500	-1.13916700	1.21504900
H		-1.60563900	0.28018500	2.10128400
C		-3.16653600	-1.04298900	-1.19479500
H		-1.93805900	0.43567200	-2.17289000
C		-3.53724200	-1.60834700	0.02588700
H		-3.27015100	-1.58183400	2.16179100
H		-3.59654600	-1.41241000	-2.11861700
H		-4.26102100	-2.41506100	0.04867400

tert-BuSOH + *tert*-BuO[•] TS

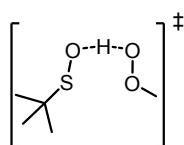


TS freq.: -396.84 cm⁻¹

CBS-QB3 Enthalpy = -863.528387 CBS-QB3 Free Energy= -863.590026

O 2			
S	1.06762400	-0.69445100	-0.76224100
H	-0.42902300	-1.01373400	0.63449400
C	2.37939700	0.31328300	0.08533600
O	-1.24679900	0.17107200	0.45724600
C	-2.56719100	0.15974400	-0.00120000
O	0.42849600	-1.58161600	0.43735300
C	1.75761800	1.14455000	1.21150000
H	1.29726200	0.50029300	1.96214800
H	2.54004700	1.73431600	1.70084900
H	0.98765000	1.81748100	0.83319800
C	3.43573900	-0.65564900	0.63531500
H	3.91993200	-1.21606400	-0.16780100
H	4.20518200	-0.09534900	1.17701100
H	2.97881700	-1.36749700	1.32526300
C	2.95246400	1.20237100	-1.02649200
H	2.20113900	1.89514800	-1.41399500
H	3.78080900	1.79461400	-0.62669700
H	3.34069400	0.60934300	-1.85974100
C	-3.46532300	-0.00238200	1.25181000
H	-4.52052900	0.01990800	0.96461100
H	-3.25885700	-0.95478900	1.74545300
H	-3.27026700	0.80543100	1.95916900
C	-2.81059600	-0.99377300	-0.98923900
H	-2.16530200	-0.88153100	-1.86447000
H	-2.58227000	-1.95414000	-0.51968400
H	-3.84959200	-1.01790500	-1.33041600
C	-2.84597300	1.52291300	-0.66545700
H	-2.66211600	2.33144100	0.04487600
H	-2.18396600	1.65992700	-1.52367800
H	-3.88197500	1.58234000	-1.01096500

tert-BuSOH + MeOO[•] TS syn



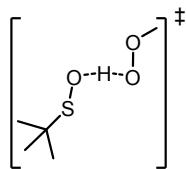
TS freq.: -473.63 cm⁻¹

CBS-QB3 Enthalpy = -820.915623 CBS-QB3 Free Energy= -820.969459

O 2			
O	-0.35654700	-1.58688300	-0.17904800
H	-1.14468600	-1.20584200	0.41768100
C	-3.29704000	0.63272600	-0.59088600
H	-3.24670300	1.59073400	-1.10871300
H	-3.16032600	-0.19548400	-1.28849300
H	-4.24438400	0.52601500	-0.05895900

O	-2.18533500	-0.53080400	1.04044100
O	-2.22105300	0.65183800	0.36158000
C	1.64871300	0.25602800	0.13058800
S	0.34045600	-0.39484100	-1.02045200
C	2.35463300	1.35068200	-0.68130100
H	1.66983500	2.16099900	-0.94568400
H	3.16406500	1.77987900	-0.08351000
H	2.79392300	0.95320100	-1.60055100
C	2.60272600	-0.89345400	0.47769400
H	3.35890300	-0.54675200	1.18990700
H	2.05587000	-1.72108000	0.93296200
H	3.11354400	-1.26834200	-0.41224700
C	0.99137400	0.82939900	1.39134000
H	1.76531200	1.19895500	2.07243600
H	0.31646900	1.65175800	1.14830400
H	0.41192100	0.06791900	1.91570300

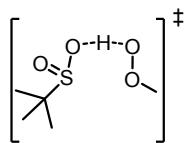
tert-BuSOH + MeOO[•] TS anti



TS freq.: -1094.21 cm⁻¹
 CBS-QB3 Enthalpy = -820.905338 CBS-QB3 Free Energy= -820.961411

O 2			
S	0.85329400	-1.14391900	-0.29519200
O	-0.37562900	-0.90302400	0.71434500
O	-2.10820900	0.21398500	-0.50441100
O	-3.32674300	-0.36825100	-0.28455400
H	-1.21261300	-0.45398000	0.16873100
C	-4.22179100	0.59850200	0.27321500
H	-5.17531500	0.08034600	0.38044300
H	-3.85289300	0.93161100	1.24659200
H	-4.31942400	1.44974000	-0.40427200
C	1.95055700	0.32438500	0.01250300
C	3.16032400	0.10363400	-0.90489500
H	2.86777700	0.07725200	-1.95807900
H	3.87066900	0.92560000	-0.77575600
H	3.68143000	-0.82792500	-0.66718200
C	2.35524200	0.33266800	1.49163000
H	2.95968900	1.21972900	1.70870000
H	1.47002300	0.35407400	2.12984800
H	2.94072300	-0.55357400	1.74704600
C	1.18878500	1.60285700	-0.35817000
H	1.81291000	2.47917100	-0.15306900
H	0.91819200	1.61153100	-1.41622700
H	0.27208800	1.69517900	0.22756200

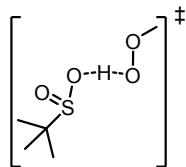
tert-BuSO₂H + MeOO[•] TS syn



TS freq.: -1606.69 cm⁻¹
 CBS-QB3 Enthalpy = -896.048552 CBS-QB3 Free Energy= -896.105357

0	2			
S	-0.18329600	-0.45328400	0.16239700	
O	-0.41877000	-1.90748200	0.23301400	
O	0.31384000	0.03535400	-1.24679400	
H	1.46744600	0.07642900	-1.27120400	
C	-1.92046400	0.36068800	0.12539200	
O	2.68315700	0.24095800	-1.06948400	
O	2.79187100	-0.28896100	0.19345000	
C	3.63953000	0.55012600	0.98583000	
H	3.66265800	0.08631800	1.97171200	
H	4.63837200	0.57401100	0.54530100	
H	3.22415800	1.55842400	1.03420500	
C	-1.70393000	1.86743300	0.01674200	
H	-1.19900200	2.12922400	-0.91324000	
H	-2.67496200	2.37297600	0.03812000	
H	-1.11326100	2.24964600	0.85429700	
C	-2.65854100	-0.22683400	-1.07630600	
H	-2.69160400	-1.31597600	-1.01479500	
H	-3.68624600	0.15137500	-1.08538000	
H	-2.17422000	0.05522200	-2.01168900	
C	-2.56686500	-0.03904800	1.45191200	
H	-2.00322700	0.34427100	2.30718400	
H	-3.57635500	0.38108600	1.50057200	
H	-2.64017700	-1.12359600	1.54365400	

tert-BuSO₂H + MeOO[•] TS anti



TS freq.: -1742.44 cm⁻¹
 CBS-QB3 Enthalpy = -896.042882 CBS-QB3 Free Energy= -896.101106

0	2			
S	0.44641200	-0.49084900	-0.30578500	
O	0.62089700	-1.91651500	0.04108900	
O	-0.39130200	0.30779900	0.74526700	
H	-1.45449700	0.44654900	0.25466200	
C	2.15862900	0.32573400	-0.02215100	
O	-2.48428400	0.61729300	-0.37642800	
O	-3.36814500	-0.31350800	0.11284100	
C	1.99594900	1.80904800	-0.34091800	
H	1.28534900	2.28361200	0.33615400	
H	2.96440500	2.30832900	-0.23228100	
H	1.65529900	1.96414300	-1.36848400	
C	2.52808900	0.06399100	1.43703300	
H	2.53602300	-1.00660900	1.64828800	
H	3.52902100	0.46422600	1.63020000	
H	1.82241000	0.54787500	2.11330300	
C	3.10056200	-0.38053800	-0.99648600	
H	2.79489000	-0.22649700	-2.03502400	
H	4.10983200	0.02685900	-0.88001500	
H	3.13711100	-1.45323800	-0.80175300	
C	-4.69626900	0.18964900	-0.03659700	

H	-4.90736400	0.37372000	-1.09264900
H	-5.34330200	-0.59516400	0.35484300
H	-4.81086600	1.11193000	0.53788100

PhS[•]

CBS-QB3 Enthalpy= -628.910084 CBS-QB3 Free Energy= -628.947151

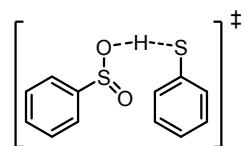
O 2			
C	0.00000000	0.00000000	-2.23137611
C	0.00000000	-1.21271391	-1.53451120
C	0.00000000	-1.21800602	-0.14939220
C	0.00000000	-0.00000000	0.57708589
C	-0.00000000	1.21800602	-0.14939220
C	-0.00000000	1.21271391	-1.53451120
H	0.00000000	0.00000000	-3.31546411
H	0.00000000	-2.14988087	-2.07920928
H	0.00000000	-2.14899207	0.40354772
H	-0.00000000	2.14899207	0.40354772
H	-0.00000000	2.14988087	-2.07920928
S	0.00000000	0.00000000	2.29996189

PhSH

CBS-QB3 Enthalpy= -629.536732 CBS-QB3 Free Energy= -629.575407

O 1			
S	2.28881889	-0.08321411	0.00000000
C	0.50622890	0.00087900	0.00000000
C	-0.19632603	1.20936605	0.00000000
C	-0.20018418	-1.20648195	0.00000000
C	-1.58857803	1.20509913	0.00000000
H	0.33831503	2.15231701	0.00000000
C	-1.59144418	-1.19999687	0.00000000
H	0.33672376	-2.14853499	0.00000000
C	-2.29350710	0.00374718	0.00000000
H	-2.12183897	2.14923917	0.00000000
H	-2.12710924	-2.14266783	0.00000000
H	-3.37704410	0.00536924	0.00000000
H	2.51271497	1.24002788	0.00000000

PhSO₂H + PhS[•] TS

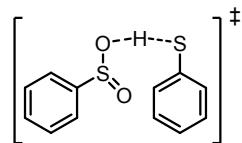


TS freq.: -1468.84 cm⁻¹
 CBS-QB3 Enthalpy = -1408.712426 CBS-QB3 Free Energy= -1408.774886

O 2			
S	-1.27027700	1.43625400	-1.02284800
O	-0.55912600	2.20640600	0.02392200
O	-0.40594300	0.42302600	-1.79647700
H	0.24624700	-0.49631700	-1.06459000
C	-2.46895600	0.38556200	-0.13208300
C	-2.76304800	0.67483000	1.19646400

C	-3.12165500	-0.63285700	-0.82441900
C	-3.71972100	-0.09585800	1.85422200
H	-2.23504900	1.47510000	1.69921700
C	-4.07698700	-1.39325300	-0.15604200
H	-2.87384200	-0.83600900	-1.85936200
C	-4.37543700	-1.12465700	1.18017500
H	-3.94931500	0.10658700	2.89405900
H	-4.58588700	-2.19593900	-0.67707600
H	-5.12136900	-1.71860500	1.69572600
S	1.06894000	-1.62425400	-0.46272600
C	2.50509900	-0.72519100	-0.01974000
C	3.74830200	-1.39060000	-0.03197800
C	2.44592900	0.62135000	0.39635000
C	4.89609600	-0.73360000	0.38498700
H	3.79352700	-2.41969100	-0.36745900
C	3.60541000	1.27072700	0.79510600
H	1.50062700	1.15028500	0.39628900
C	4.83086000	0.59998300	0.79663600
H	5.84641500	-1.25501600	0.37983100
H	3.55352400	2.30768400	1.10641300
H	5.73172000	1.11583800	1.10850300

tert-BuSOH + PhS[·] TS



TS freq.: -1458.08 cm⁻¹
 CBS-QB3 Enthalpy = -1259.878512 CBS-QB3 Free Energy= -1259.940940

O	2		
S	-1.46102300	0.04483700	-1.11628100
H	-0.21925500	1.90205900	-0.32136600
C	-2.49973300	-0.60302300	0.28553600
S	1.16986900	2.24024800	0.44217300
O	-1.26966300	1.58137700	-0.78845600
C	1.90928300	0.66678900	0.17196300
C	2.33855600	-0.11339600	1.26390200
C	2.11753300	0.17136800	-1.13138200
C	2.96237100	-1.33660900	1.05633900
H	2.18000200	0.26115800	2.26789700
C	2.72950200	-1.06174900	-1.33071000
H	1.81599700	0.77608300	-1.97791500
C	3.15706400	-1.81813800	-0.24076400
H	3.29665900	-1.92044800	1.90689000
H	2.88353400	-1.42713700	-2.33990600
H	3.64295800	-2.77428600	-0.39844100
C	-1.72536100	-0.44785100	1.59891700
H	-1.47869800	0.59839500	1.78917200
H	-2.34137900	-0.80720200	2.42987300
H	-0.79619900	-1.02121900	1.58549700
C	-3.81229900	0.18965400	0.31851500
H	-4.38679900	0.04851700	-0.60001100
H	-4.424462200	-0.14573900	1.16184700
H	-3.61430800	1.25606200	0.44003900
C	-2.73350600	-2.07894700	-0.06370300
H	-1.79410400	-2.63629700	-0.11074600
H	-3.35634800	-2.53920800	0.70875000

H -3.25212000 -2.19170500 -1.01989100

O₂

CBS-QB3 Enthalpy= -150.161251 CBS-QB3 Free Energy= -150.184523

0 3
O 0.00000000 0.00000000 0.60281700
O 0.00000000 0.00000000 -0.60281700

MeSO₂•

CBS-QB3 Enthalpy= -587.802595 CBS-QB3 Free Energy= -587.836703

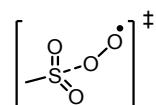
0 2
S 0.28431400 0.19915100 0.00000000
O -0.17133300 0.74686800 1.28110900
O -0.17133300 0.74686800 -1.28110900
C -0.17133300 -1.57887100 0.00000000
H 0.24068000 -2.02352900 -0.90329600
H 0.24068000 -2.02352900 0.90329600
H -1.26105500 -1.61602600 0.00000000

MeSO₂H

CBS-QB3 Enthalpy= -588.424080 CBS-QB3 Free Energy= -588.458321

0 1
S -0.16471300 -0.16721300 -0.42035100
O -0.71241600 -1.27207800 0.38146200
O -0.61990400 1.32150500 0.25986500
H -1.51766200 1.50064100 -0.05045200
C 1.54171300 0.07527200 0.14625900
H 2.09576500 -0.82342800 -0.12483200
H 1.96255500 0.95676500 -0.33657100
H 1.50303200 0.19439000 1.22929700

MeSO₂• + O₂ TS

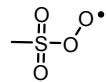


TS freq.: -132.52 cm⁻¹

CBS-QB3 Enthalpy = -737.961367 CBS-QB3 Free Energy= -738.003697

0 2
S -0.54950500 -0.20719400 -0.00217200
O -1.30163900 -0.94762200 -1.00536600
O -0.31018800 -0.71738000 1.33884800
C -1.26988200 1.45689600 0.13510700
H -0.63671300 2.03443400 0.80517800
H -1.30358300 1.88370300 -0.86515900
H -2.27284600 1.33235300 0.54570400
O 1.69730800 0.34239500 -0.67325800
O 2.49258200 -0.01198900 0.18207400

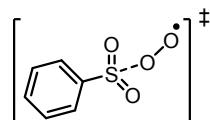
MeS(O)₂OO[•]



CBS-QB3 Enthalpy = -737.986159 CBS-QB3 Free Energy= -738.025794

0	2			
S		-0.48743500	-0.13071900	0.06624100
O		-1.35886000	-0.75359100	-0.88375000
O		-0.47448000	-0.44289200	1.46346700
C		-0.41549800	1.63837100	-0.16576000
H		0.46795200	2.00084600	0.35605600
H		-0.37378500	1.83902700	-1.23423800
H		-1.32593400	2.04041500	0.27894000
O		1.16111200	-0.55152200	-0.58629500
O		2.11269100	0.04563000	0.07332200

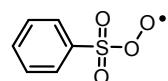
PhSO₂[•] + O₂ TS



TS freq.: -104.32 cm⁻¹
CBS-QB3 Enthalpy = -929.348716 CBS-QB3 Free Energy= -929.399501

0	2			
S		1.09898100	-0.61592600	0.36174900
O		1.27686800	-2.03971700	0.10173200
O		1.61399600	0.04769600	1.55206300
O		2.22178000	0.76015200	-1.32859500
O		2.94228200	1.57875100	-0.78536500
C		-0.63304700	-0.19452500	0.15727300
C		-1.43823000	-1.01670600	-0.62944900
C		-1.11114700	0.97920400	0.73830700
C		-2.76849100	-0.65547300	-0.82080500
H		-1.03075900	-1.92145500	-1.06127800
C		-2.44394800	1.32414300	0.53597000
H		-0.45569700	1.58910800	1.34613800
C		-3.26822700	0.51103900	-0.24202800
H		-3.41451800	-1.28696000	-1.41907600
H		-2.83879100	2.22582800	0.98887900
H		-4.30479200	0.78713900	-0.39692500

PhS(O)₂OO[•]

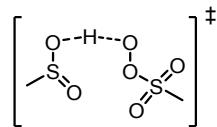


CBS-QB3 Enthalpy = -929.371766 CBS-QB3 Free Energy= -929.420032

0	2			
S		1.28451300	0.36554600	-0.38180500
O		1.66127300	1.74334100	-0.25873100
O		1.78538500	-0.52680100	-1.38538600
O		1.87252300	-0.30821900	1.21939900

O	1.70628300	-1.59264400	1.30535800
C	-0.46160200	0.16037300	-0.19182200
C	-1.19826200	1.18979700	0.39488700
C	-1.05184300	-1.02957200	-0.61790400
C	-2.56909100	1.01715700	0.55237900
H	-0.70439300	2.10092000	0.70587500
C	-2.42387900	-1.18524900	-0.44795500
H	-0.44711600	-1.80156200	-1.07362800
C	-3.17750100	-0.16664300	0.13445500
H	-3.16155600	1.80669000	0.99861000
H	-2.90355300	-2.10059900	-0.77279400
H	-4.24624000	-0.29478200	0.26146100

MeSO₂H + MeS(O)₂OO[•] TS

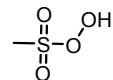


TS freq.: -746.25 cm⁻¹

CBS-QB3 Enthalpy = -1259.878512 CBS-QB3 Free Energy= -1259.940940

0 2			
S	1.90283600	-0.04266700	0.25925300
O	3.00700200	0.32778600	1.15246700
O	1.62853400	1.02935800	-0.88048100
H	0.69779200	1.50320200	-0.69243000
C	2.50656600	-1.37094300	-0.83400200
H	2.79988300	-2.19676900	-0.18808100
H	1.68213000	-1.64248900	-1.48995600
H	3.35799400	-0.97273700	-1.38517900
O	-0.59550400	1.84978400	-0.27267700
O	-0.82555500	0.92534800	0.71334100
S	-1.69345100	-0.44656200	0.09297400
O	-0.98369600	-0.94248200	-1.05500900
O	-1.91332300	-1.24284500	1.26388800
C	-3.18400800	0.37302100	-0.43809400
H	-3.77768300	-0.37545200	-0.96241600
H	-3.69904700	0.74272600	0.44609500
H	-2.88623300	1.18111100	-1.10332200

MeS(O)₂OOH



CBS-QB3 Enthalpy = -738.637045 CBS-QB3 Free Energy= -738.675689

0 1			
H	-2.23753300	-0.52397100	0.59254300
O	-2.06088600	0.16116300	-0.07246900
O	-0.94729500	-0.42223100	-0.78937500
S	0.46200700	-0.15083900	0.07481400
O	0.20788900	-0.49967500	1.44553400
O	1.44700000	-0.81477600	-0.72551800
C	0.61906500	1.61813500	-0.06626400
H	1.47613300	1.90836400	0.54113200
H	0.77864500	1.86099300	-1.11475400
H	-0.29740800	2.06337100	0.31626000

References

1. M. Ghosal, S. Das and D. Mukherjee, *Synth. Commun.*, 1989, **19**, 3275-3282.
2. A. K. Saha, S. Das and D. Mukherjee, *Synth. Commun.*, 1993, **23**, 1821-1828.
3. G. X. Ortiz, B. N. Hemric and Q. Wang, *Org. Lett.*, 2017, **19**, 1314-1317.
4. L. Valgimigli, J. T. Banks, J. Lusztyk and K. U. Ingold, *J. Org. Chem.*, 1999, **64**, 3381-3383.
5. R. Amorati, M. Lucarini, V. Mugnaini, G. F. Pedulli, F. Minisci, F. Recupero, F. Fontana, P. Astolfi and L. Greci, *J. Org. Chem.*, 2003, **68**, 1747-1754.
6. P. Mulder, H.-G. Korth, D. A. Pratt, G. A. DiLabio, L. Valgimigli, G. F. Pedulli and K. U. Ingold, *J. Phys. Chem. A*, 2005, **109**, 2647-2655.
7. L. A. Farmer, E. A. Haidasz, M. Griesser and D. A. Pratt, *J. Org. Chem.*, 2017, **82**, 10523-10536.
8. G. Litwinienko and K. U. Ingold, *J. Org. Chem.*, 2003, **68**, 3433-3438.
9. M. H. Abraham, R. J. Abraham, J. Byrne and L. Griffiths, *J. Org. Chem.*, 2006, **71**, 3389-3394.
10. H. T. Ho, O. Ito, M. Iino and M. Matsuda, *J. Phys. Chem.*, 1978, **82**, 314-319.
11. G. L. Edwards, C. A. Muldoon and D. J. Sinclair, *Tetrahedron*, 1996, **52**, 7779-7788.
12. J.-P. R. Chauvin, M. Griesser and D. A. Pratt, *J. Am. Chem. Soc.*, 2017, **139**, 6484-6493.
13. E. A. Haidasz, A. T. M. Van Kessel and D. A. Pratt, *J. Org. Chem.*, 2016, **81**, 737-744.