

Using Machine Learning to Predict Suitable Conditions for Organic Reactions

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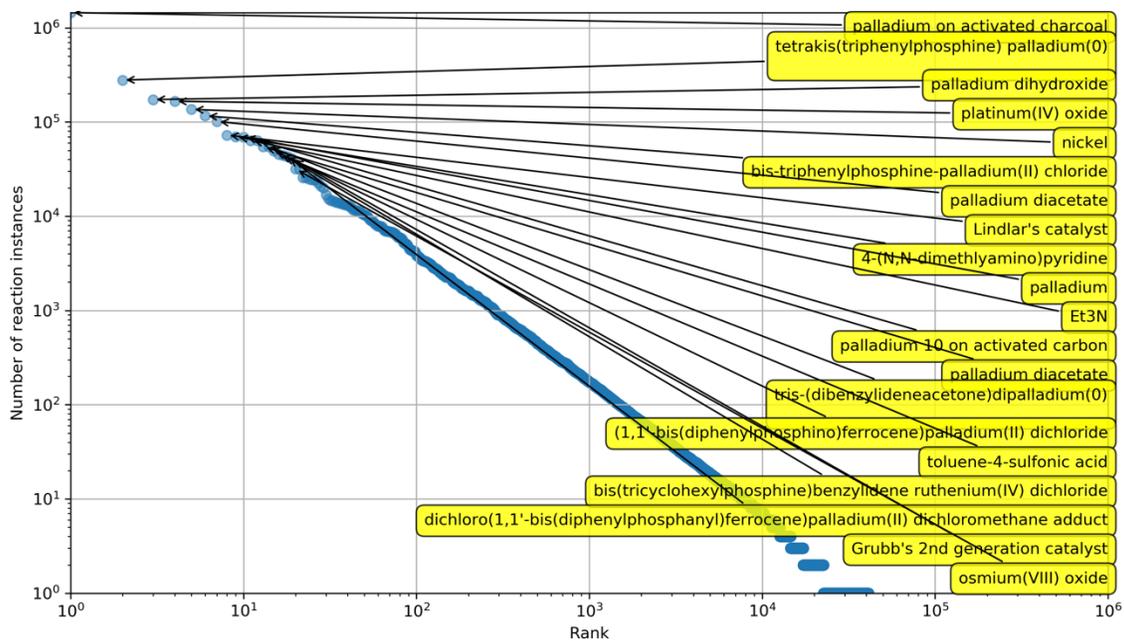
Supplementary Information

Computational methods

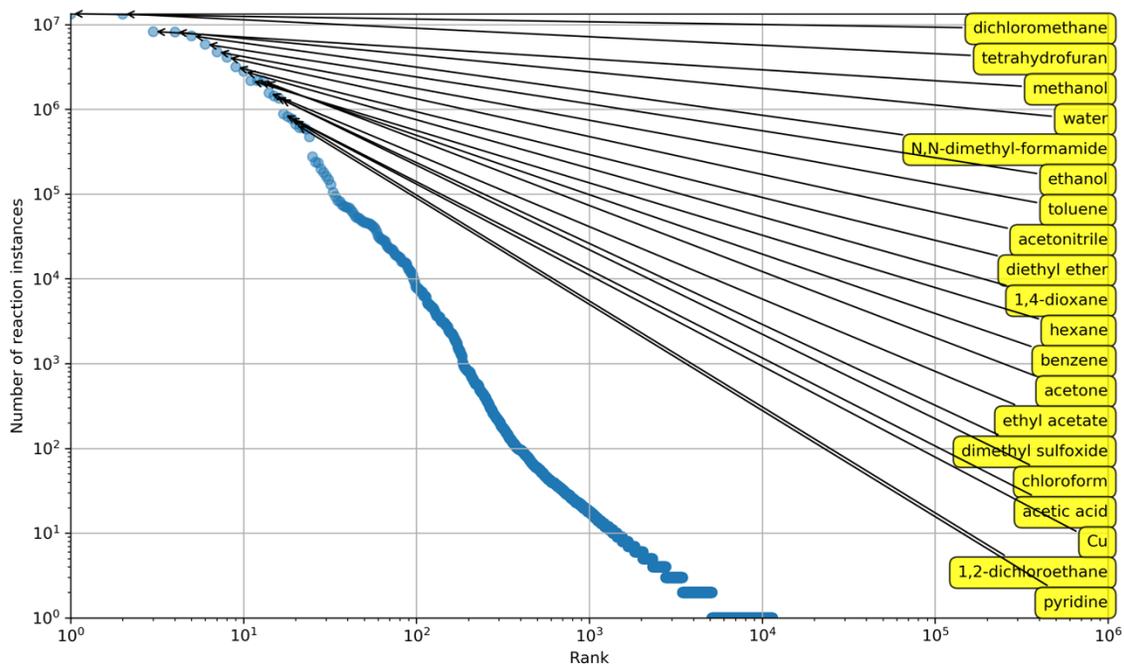
Neural network models are developed in Keras (Version 2.0.2) and trained using the Theano backend (Version 1.0.0). All trainings are performed on a single NVIDIA GeForce GTX 1080 GPU. The code is developed in Python 2.7.

Frequency vs rank plots for catalysts, reagents and solvents

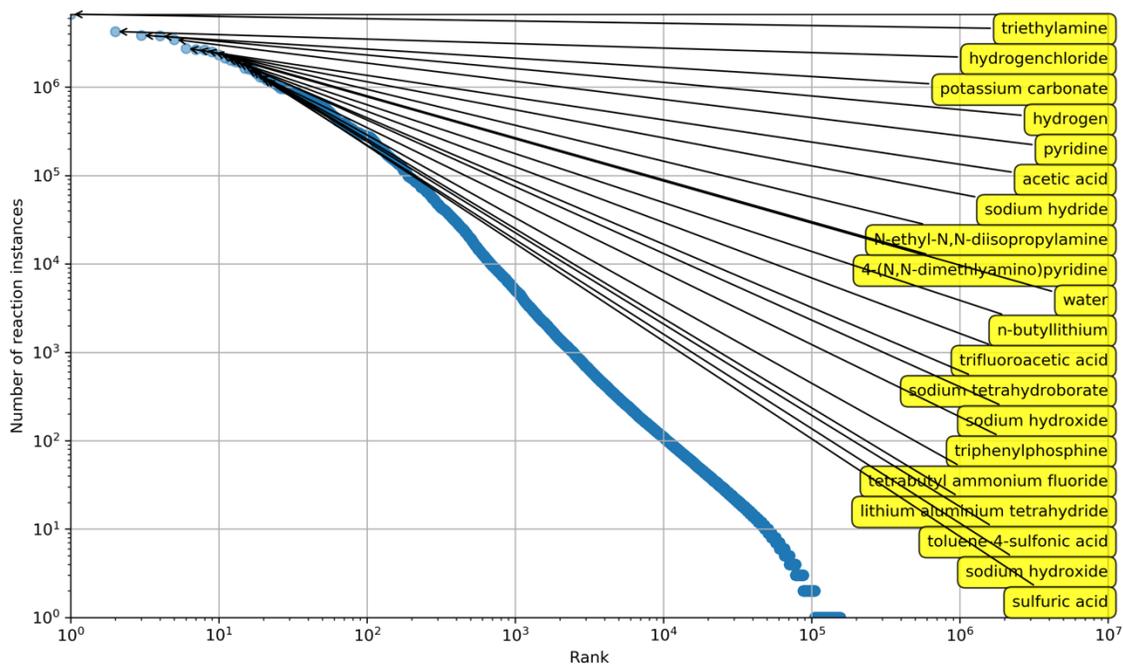
Supplementary Figures 1-3 are the frequency vs. rank plots for catalyst, solvent and reagent in Reaxys. It can be seen some duplicated records of the same chemical exists (e.g. palladium diacetate as catalyst and sodium hydroxide as reagent). We kept these as different classes as there is not a good way to systematically identify and curate this issue as it is not clear what is the pattern of these duplicated chemicals with different ids. However, during the training, the model learns that they are very similar entities because one or another of these duplicated chemicals are used in the same type of reactions.



Supplementary Figure 1. Frequency vs. rank plot for catalyst in Reaxys (with the top ones labeled in yellow)



Supplementary Figure 2. Frequency vs. rank plot for solvent in Reaxys (with the top ones labeled in yellow)



Supplementary Figure 3. Frequency vs. rank plot for reagent in Reaxys (with the top ones labeled in yellow)

Comparison of prediction accuracies with a null model

A null model is defined to always give the same prediction of top ten combinations chosen based on the frequencies of the catalysts, solvents and reagents. The top 10 combinations of the null model are listed below in Supplementary Table 1.

Supplementary Table 1. The top-ten combinations used in the null model

Rank	Catalyst	Solvent 1	Solvent 2	Reagent 1	Reagent 2
1					
2		DCM ^a			
3		THF			
4				TEA	
5				K ₂ CO ₃	
6		DCM		TEA	
7		THF		TEA	
8		DCM		K ₂ CO ₃	
9		THF		K ₂ CO ₃	
10	Pd on activated charcoal				

^a: DCM: dichloromethane, THF: tetrahydrofuran, TEA: triethylamine

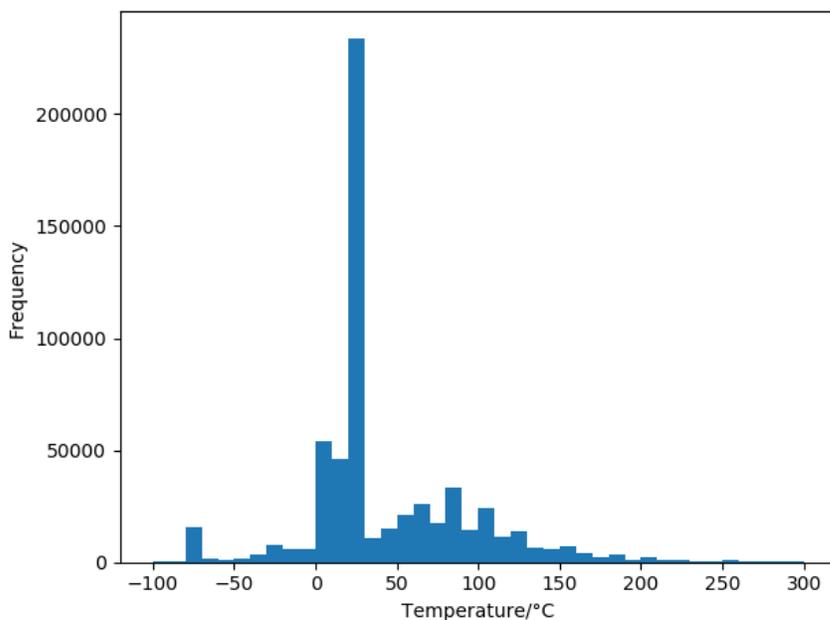
The comparison of prediction accuracies are shown in Supplementary Table 2. The accuracy values shown are for top-three exact matches. In general, it can be seen that the accuracy values are much lower for the null model. The accuracy of c, s2 and r2 predictions for the null model are high, since a majority of reactions do not use a catalyst, a second solvent or a second reagent, but the trained model is still better than the null model by a large margin.

Supplementary Table 2. Comparison of accuracy for the true context to be in the top-3 predictions to a null model

	Trained neural network model	Null model
c	93.6%	87.3%
s1	75.8%	49.4%
s2	90.1%	85.6%
r1	73.2%	22.0%
r2	89.3%	82.3%
c, s1, r1	57.3%	5.7%
c, s1, s2, r1, r2	50.1%	4.7%

^a c, s1, s2, r1, r2 refer to catalyst, solvent 1, solvent 2, reagent 1 and reagent 2, respectively;

For temperature prediction, we analyzed a baseline model that predicts the most frequently used temperature for all reactions. Supplementary Figure 4 shows the distribution of temperature for reactions in the test set. The most frequently used temperature is the room temperature (20 °C) which covers a majority of reaction, and the accuracy of the predicted temperature by the baseline model (which is always 20 °C) being within the $\pm 10^{\circ}\text{C}$ or $\pm 20^{\circ}\text{C}$ range of the recorded temperature are 40.0% and 49.4%. In the meantime the distribution spans a wide range. Simply predicting the room temperature (20 °C) will result in a mean absolute error of 35.3 °C, which is significantly larger than prediction given by the trained model and would be misleading for reactions that require high or low temperatures.



Supplementary Figure 4. Temperature distribution for reactions in the test set

Full list of evaluation of reaction examples

Supplementary Table 3. 62 reactions from eleven reaction types randomly chosen from the test dataset

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature / °C	Top Predicted Temperature / °C	Closest Predicted Temperature / °C	Reaction type
5301921		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	20.0	19.3	19.3	Hydrolysis
8712792		<chem>—OH K+ HO-</chem>	<chem>K+ HO-</chem>	<chem>—OH K+ HO-</chem>	N/A	104.4	99.7	Hydrolysis
5261303		<chem>—OH H2N—NH2 H2O</chem>	<chem>—OH —NH2</chem>	<chem>—OH H2N—NH2</chem>	N/A	92.9	88.2	Hydrolysis
8655717		<chem>—OH Na+ HO-</chem>	<chem>—OH Na+ HO-</chem>	<chem>—OH Na+ HO-</chem>	N/A	30.2	30.2	Hydrolysis
5302180		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	57.5	35.1	35.1	Hydrolysis
531162		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	0.0	12.2	5.5	Esterification
5185761		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	10.0	15.7	11.2	Esterification
8698819		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	95.0	51.6	99.7	Esterification
8669084		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	N/A	15.0	15.0	Esterification
5336683		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	20.0	48.2	48.2	Esterification
5220300		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	-78.0	5.7	13.6	Alkylation
2820838		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	54.5	71.6	71.6	Alkylation
8568283		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	N/A	18.4	18.4	Alkylation
8738792		<chem>—OH —O— Na+</chem>	<chem>Cl—Cl</chem>	<chem>—OH —O— Na+</chem>	20.0	23.1	47.6	Alkylation
8574992		<chem>—OH K+ HO-</chem>	<chem>—OH K+ HO-</chem>	<chem>—OH K+ HO-</chem>	40.0	28.2	28.2	Alkylation
8591626		<chem>—OH Na+ HO- HO—OH</chem>	<chem>—OH Na+ HO- HO—OH</chem>	<chem>—OH Na+ HO- HO—OH</chem>	10.0	15.1	15.1	Epoxidation
8647861		<chem>—OH Na+ HO- HO—OH</chem>	<chem>—OH Na+ HO- HO—OH</chem>	<chem>—OH Na+ HO- HO—OH</chem>	0.0	5.4	47.8	Epoxidation
8655813		<chem>—OH NH3</chem>	<chem>—OH NH3</chem>	<chem>—OH NH3</chem>	20.0	22.5	22.5	Epoxidation
8720054		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	20.0	15.2	15.2	Epoxidation
8560887		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	0.0	20.7	20.7	Epoxidation
8598828		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	0.0	12.7	20.9	Epoxidation
5229646		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	0.0	35.8	35.8	Epoxidation
10096807		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	25.0	15.2	48.5	Epoxidation
9925682		<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	<chem>—OH —O— Na+</chem>	N/A	6.0	34.3	Epoxidation

4053563					20.0	15.4	15.4	Wittig
1558712					100.0	44.0	80.1	Wittig
41951909					9.0	10.7	9.5	Wittig
9299119					N/A	16.6	16.6	Wittig
28171183					10.0	47.6	19.9	Wittig
28476112					20.0	53.2	53.2	Wittig
9240363					20.0	30.3	26.4	Wittig
2287762					20.0	42.3	9.4	Wittig
9659689					20.0	21.5	18.4	Deprotection
5357436					30.0	17.8	17.8	Deprotection
9341851					N/A	16.9	16.9	Deprotection
9588894					N/A	15.9	15.9	Deprotection
5351204					N/A	20.0	20.0	Deprotection
9297919					N/A	9.7	9.7	Reduction
31266961					78.0	16.2	16.2	Reduction
5265062					30.0	124.2	16.9	Reduction
8619786					N/A	20.8	20.9	Reduction
5261651					20.0	75.4	75.4	Reduction
28315206					10.0	19.5	18.2	Oxidation
5307293					25.0	3.1	3.1	Oxidation
9228133					30.0	1.7	1.7	Oxidation
116049					N/A	46.2	46.2	Oxidation
5081868					20.0	20.6	20.6	Oxidation
330468					9.0	12.2	12.2	Oxidation
497226					80.0	176.7	130.2	Blackwald-Hartwig
34039114					95.0	42.6	70.1	Blackwald-Hartwig

11260415					90.0	94.9	94.7	Buchwald-Hartwig
11077029					120.0	120.5	120.5	Buchwald-Hartwig
2646174					90.0	104.3	104.3	Buchwald-Hartwig
3382327					90.0	113.5	113.5	Buchwald-Hartwig
9601192					19.0	31.5	39.1	Erubbe
3603266					40.0	35.6	22.6	Erubbe
2584073					100.0	101.8	101.8	Suzuki
4033276					N/A	51.5	81.1	Suzuki
9208506					N/A	84.1	84.1	Suzuki
2888714					N/A	95.3	95.3	Suzuki
10440528					20.0	52.2	52.2	Suzuki

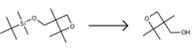
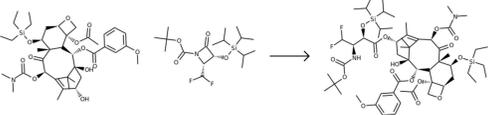
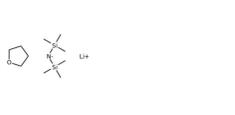
Supplementary Table 4. 100 reactions randomly chosen from the test dataset

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature °C	Top Predicted Temperature °C	Closest Predicted Temperature °C
138215					N/A	40.3	32.1
10061846					60.0	29.4	23.8
44110175					-78.0	-65.6	-66.6
10261736					0.0	15.1	15.1
4087565					50.0	54.3	54.3
36028195					60.0	49.1	49.1
2176355					0.0	-18.7	-4.2
2069147					N/A	-42.6	-13.1
1720472					N/A	54.0	54.0
3234380					N/A	26.2	24.5
4822624					60.0	5.7	5.7
34927996					N/A	34.5	32.1
1013620					N/A	263.4	263.4
3928353					60.0	66.7	66.7
26314583					N/A	25.8	25.8
38789242					25.0	36.3	36.3
24821402					60.0	88.2	88.2
3665928					N/A	28.7	36.7
3928431					190.0	254.5	254.5
2215973					N/A	15.5	47.1
4940666					100.0	85.4	60.6
5310995					70.0	42.1	44.6
25900877					65.0	46.3	46.3
35559718					30.0	21.7	21.7
42570631					120.0	104.7	104.7

9136462			—OH		20.0	32.6	33.0
41425736			Reaxys ID 13154734		Reaxys ID 13154734	Reaxys ID 13154734	20.0 33.5 33.5
38554472			etherate Reaxys Name boron trifluoride diethyl		etherate Reaxys Name boron trifluoride diethyl etherate	etherate Reaxys Name boron trifluoride diethyl	N/A 123.1 123.1
9671652							N/A 189.9 152.5
27763214							20.0 7.2 7.2
5052676							0.0 -7.8 -7.8
207394							N/A 78.1 77.4
281216							200.0 259.9 233.1
518600							N/A 36.4 36.4
24973931							N/A 16.9 16.9
27979401							N/A 149.4 149.4
1654607							N/A 123.7 123.7
8541633							N/A -48.9 26.0
23314056							N/A 37.4 39.0
6963834							N/A -12.7 -12.7
9899404			Reaxys Name silica gel		Reaxys ID 19278319		N/A 105.5 96.6
5127331							0.0 27.4 30.9
38768506							70.0 45.6 34.0
31607336							20.0 18.9 22.8
1822491			Name lithium bromide		Reaxys	Reaxys Name potassium bromide	98.0 83.4 169.9
11221187							20.0 -2.0 12.5
2383147							20.0 42.3 42.3
4492881			baracol Reaxys Name palladium on activated		baracol Pd+2 Reaxys Name palladium on activated	baracol Reaxys Name palladium on activated	60.0 18.8 21.1
4233762							N/A 2.5 11.3

36324962					35.0	98.8	101.2
1986429					0.0	30.6	-1.9
23851486					N/A	42.9	68.6
5337948					100.0	23.6	23.6
10509025					-40.0	75.7	75.7
37060180					20.0	30.9	35.3
5327471					N/A	8.7	13.5
545454					N/A	115.6	122.6
461707					N/A	35.1	146.1
11242664					N/A	29.7	29.7
24909005					N/A	80.0	80.0
30084900					20.0	16.7	16.7
4594557					N/A	2.8	2.8
30102370					110.0	95.8	95.8
3816753					N/A	82.7	87.5
5210772					0.0	11.4	11.4
2176468					N/A	99.7	98.1
5139901					N/A	100.6	100.6
29034611					20.0	22.4	22.4
1387456					N/A	62.1	62.1
10149615					20.0	21.0	13.9
29237461					20.0	15.6	15.6
22880776					20.0	14.6	14.6
32792568					-30.0	52.5	16.2

225356	$\text{C}=\text{O} \longrightarrow \text{CH}_4$	Ni	Ni	Ni	299.9	263.1	263.1
4667439					N/A	-17.4	-17.4
39121363					N/A	45.8	45.8
42279161					20.0	22.3	19.9
1821342			AH7: Li+		100.0	103.5	103.5
2170849					30.0	35.0	35.0
2874608					-78.0	56.2	16.7
1320851					N/A	31.6	32.7
9372184					20.0	21.2	21.2
40224906					27.5	21.1	21.1
29311031					120.0	98.3	97.4
40807619					120.0	64.6	64.6
27911253					-20.0	10.4	9.4
39062813					0.0	19.8	6.1
4410590			AH7: Li+		N/A	12.1	16.3
27978248					N/A	56.1	45.2
27784122					130.0	47.3	76.0
28190848					20.0	39.7	39.7
24898802					70.0	37.0	37.0
609622			Na+ HO-		180.0	41.4	41.4
884394					70.0	62.9	62.9
7008517					255.0	49.4	49.4
5021033					80.0	49.3	49.3
26014672					N/A	180.7	113.6
37452339					140.0	125.2	125.2

9905540		 F ⁻	 F ⁻	20.0	37.1	37.1
28176521		 Li ⁺	 Li ⁺	-40.0	-25.5	-25.5

Supplementary Table 5. 40 reactions from eleven reaction types randomly chosen from the test dataset that has the least number of correctly or similarly predicted elements

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature/°C	Top Predicted Temperature/°C	Closest Predicted Temperature/°C	Reaction type
8542026					5.0	20.0	20.1	Reduction, Hydrolysis
5342174					20.0	65.7	65.7	Hydrolysis
5240460					N/A	48.4	48.4	Hydrolysis
8538458					39.0	34.4	34.4	Combustion, Hydrolysis, Oxidation
8631198					N/A	25.5	27.1	Hydrolysis
10069687					25.0	35.2	48.5	Epoxylation
9925682					N/A	6.0	34.3	Sharpless epoxidation
29729396					10.0	25.7	30.8	Friedel-Crafts Alkylation
8529950					20.0	52.8	7.6	Alkylation, allylation, Bromination
8530713					78.0	14.9	14.9	Deprotection, Addition, Alkylation
8592120					90.0	28.3	28.3	Alkylation
8629688					20.0	13.4	24.6	Alkylation
34446926					20.0	9.7	26.5	1,2-Wittig Rearrangement
33420713					78.0	58.5	70.5	Wittig reaction
28838907					20.0	21.2	8.3	Wittig reaction
29349499					20.0	7.4	7.4	Wittig-Horner reaction
15993977					20.0	17.6	17.6	Wittig alkylation
5266589					N/A	46.2	40.0	deprotection
8780005					22.0	32.3	38.4	deprotection
8585244					20.0	24.8	24.8	Reduction
8692625					100.0	37.6	42.0	Formylation, reduction
29313469					40.0	3.6	60.5	Hydro-reduction
42522603					63.0	68.8	58.8	Hydro-reduction
8646019					20.0	25.3	25.3	Reduction

1025318					25.0	19.0	31.9	Tamao oxidation
5351285					35.0	17.0	17.0	Oxidation
8520314					20.0	20.5	20.5	Substitution, Oxidation
9271620					20.0	4.6	4.6	Bayer-Villiger oxidation
8627906					80.0	89.1	44.4	Oxidation
2600760					80.0	98.2	95.2	Suzuki coupling
10342109					100.0	92.5	90.7	Suzuki-Miyaura cross-coupling
42866700					80.0	102.9	97.7	Suzuki coupling
2922445					80.0	80.3	88.1	Suzuki-Miyaura cross-coupling reaction
9645984					130.0	90.2	78.5	Suzuki reaction
11152198					110.0	134.9	111.2	Buchwald amidation
23567330					100.0	73.7	100.1	Buchwald-Hartwig coupling
9292801					100.0	126.4	95.9	Buchwald-Hartwig coupling
25897520					90.0	92.9	92.9	Buchwald reaction
25936145					80.0	87.1	87.1	Buchwald-Hartwig reaction
3608781					N/A	-3.3	14.7	Grubbs Olefin Metathesis

Comparison with the neighbor approach

The nearest-neighbor method searches the training set for the reaction with the maximum similarity to the reaction for prediction. Cosine similarity of the reaction fingerprint¹ is used to quantify reaction similarity. A thorough search is performed for the 62 reactions in Supplementary Table 3. The top-ten most similar reactions are retrieved, and the top-one prediction and the prediction within top-ten that has the maximum elements matching the true condition are listed in Supplementary Table 6. Good suggestions are found for a majority of the cases, and the overall accuracy for these 62 reactions is comparable to the neural network model. However, the search for one reaction takes ~40 minutes on a single intel® Xeon(R) CPU E5-2690 0 @2.90GHz, as compared to ~100ms for the neural network approach running on the same machine, which is over 10,000 times faster. Further, the nearest neighbor method proposes conditions by simply copy and paste, which does not have the ability to infer missing information.

Supplementary Table 6. Nearest neighbor predictions for the 62 reaction examples in Supplementary Table 3

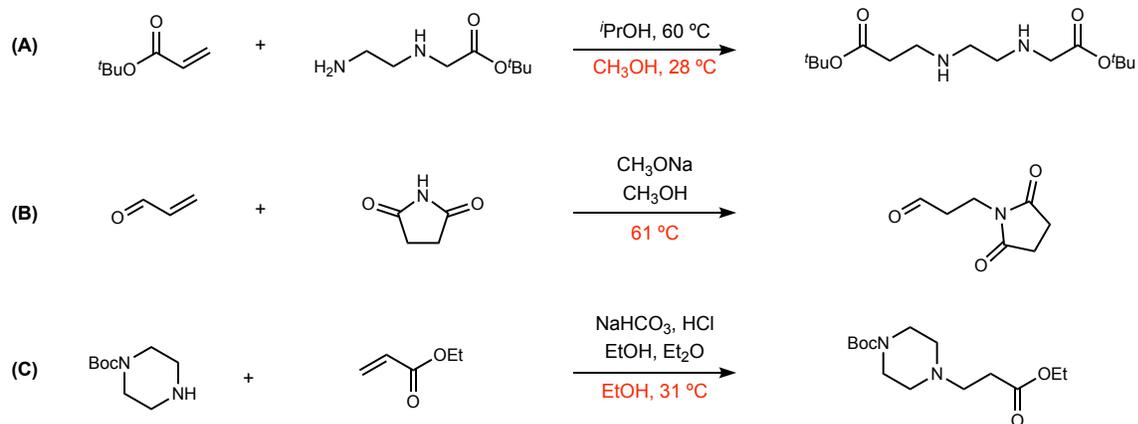
Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature °C	Top Predicted Temperature °C	Closest Predicted Temperature °C
5301921		<chem>-OH -O Na+</chem>	<chem>-OH Cl-Cl -O Na+</chem>	<chem>-OH -O Na+</chem>	20.0	20.0	12.0
8712792		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>-OH K+ HO-</chem>	N/A	N/A	30.0
5261300		<chem>-OH H2N-NH2 H2O</chem>	<chem>-OH -CH2-CH2-NH2</chem>	<chem>-OH -CH2-CH2-NH2</chem>	N/A	N/A	N/A
8655717		<chem>-OH Na+ HO-</chem>	<chem>O=C=O Li+ HO-</chem>	<chem>-OH H2O</chem> Reaxys Name lithium hydroxide monohydrate	N/A	20.0	20.0
5302180		<chem>-OH O=C=O K+</chem>	<chem>-OH O=C=O K+</chem>	<chem>-OH O=C=O K+</chem>	57.5	N/A	N/A
53162		<chem>-OH -O Na+</chem>	<chem>Cl-Cl</chem>	<chem>Cl-Cl</chem>	0.0	0.0	0.0
5185761		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	10.0	N/A	N/A
3608819		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	35.0	N/A	N/A
8609084		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	N/A	N/A	N/A
5336680		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	20.0	30.0	30.0
5220300		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	78.0	78.0	78.0
2820858		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	54.5	55.0	55.0
8568283		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	N/A	0.0	0.0
8738792		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	20.0	35.0	35.0
8570982		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	40.0	40.0	40.0
8591626		<chem>-OH Na+ HO- HO-OH</chem>	<chem>-OH H2O Na+ HO- HO-OH</chem>	<chem>-OH H2O Na+ HO- HO-OH</chem>	10.0	0.0	0.0
8647861		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	0.0	0.0	0.0
8655813		<chem>-OH Na+</chem>	<chem>-OH -O Na+</chem>	<chem>-OH Na+</chem>	20.0	N/A	20.0
8720054		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	20.0	20.0	20.0
850807		<chem>-OH K+ HO-</chem>	<chem>O=C=O K+</chem>	<chem>O=C=O K+</chem>	0.0	N/A	0.0
8598928		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	0.0	N/A	N/A
5229546		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	0.0	110.0	N/A
1096987		<chem>-OH -O Na+</chem>	<chem>O=C=O Na+</chem>	<chem>O=C=O Na+</chem>	25.0	10.0	N/A

9925682		 Reaxys Name	 Reaxys Name bis(acetylacetonate)oxovanadium	 Reaxys Name bis(acetylacetonate)oxovanadium	N/A	N/A	N/A
40853683		 Reaxys Name lithium bromide	 Reaxys Name lithium bromide	 Reaxys Name lithium bromide	20.0	20.0	20.0
1558712					100.0	80.0	110.0
41951969		 Reaxys Name sodium hydride	 Reaxys Name lithium chloride	 Reaxys Name sodium hydride	9.0	20.0	20.0
9299119		 Reaxys Name sodium hydride	 Reaxys Name sodium hydride	 Reaxys Name sodium hydride	N/A	N/A	15.0
28171183					10.0	10.0	10.0
28476112					20.0	20.0	20.0
9240863		 K+	 K+	 K+	20.0	20.0	20.0
2287782					20.0	N/A	20.0
9609689					20.0	N/A	20.0
5357436		 Zn ⁺⁺	 Zn ⁺⁺	 Zn ⁺⁺	20.0	20.0	20.0
5304351					N/A	N/A	N/A
8588894					N/A	20.0	20.0
5351204		 HCl	 HCl	 HCl	N/A	N/A	N/A
5297919					N/A	N/A	N/A
3126961		 Reaxys Name cerium(III) chloride heptahydrate	 Reaxys Name cerium(III) chloride heptahydrate	 Reaxys Name cerium(III) chloride heptahydrate	78.0	78.0	78.0
5259092					30.0	150.0	25.0
8619786					N/A	35.0	N/A
5261651					20.0	5.0	5.0
28315200					10.0	N/A	20.0
5307293					25.0	N/A	N/A
9228133		 Reaxys Name Jones reagent	 Reaxys Name Jones reagent	 Reaxys Name Jones reagent	20.0	20.0	20.0
116049					N/A	20.0	12.5

5081868				20.0	0.0	0.0	
33448		Reaxys Name Jones's reagent.Reaxys Name silica gel	Reaxys Name Jones's reagent	Reaxys Name Jones's reagent	0.0	N/A	N/A
497260		 Reaxys ID 21565323		 Reaxys ID 21565323	80.0	N/A	90.0
34039114					95.0	105.0	105.0
11280415		 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	90.0	100.0	85.0
1107029		 Name palladium diacetate Reaxys	 Name palladium diacetate Reaxys	 Name palladium diacetate Reaxys	120.0	120.0	120.0
26046174		 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	90.0	90.0	90.0
33823231		 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	80.0	80.0	80.0
9603192		 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	19.0	19.0	19.0
36052206		 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	 diacetate Reaxys Name palladium	80.0	80.0	80.0
25840737		 Name lithium chloride Reaxys	 Name lithium chloride Reaxys	 Name lithium chloride Reaxys	100.0	100.0	100.0
4033276		 Reaxys Name palladium diacetate	 Reaxys Name palladium diacetate	 Reaxys Name palladium diacetate	8.0	120.0	120.0
9208506		 Name lithium chloride Reaxys	 Name lithium chloride Reaxys	 Name lithium chloride Reaxys	N/A	N/A	N/A
2888514		 bis(diphenylphosphino)ferrocene/palladium(II) dichloride Reaxys Name (1,1'-bis(diphenylphosphino)ferrocene)palladium(II) dichloride	 bis(diphenylphosphino)ferrocene/palladium(II) dichloride Reaxys	 bis(diphenylphosphino)ferrocene/palladium(II) dichloride Reaxys Name (1,1'-bis(diphenylphosphino)ferrocene)palladium(II) dichloride	N/A	N/A	N/A
1040528		 Name potassium fluoride Reaxys Name palladium diacetate Reaxys	 Name potassium fluoride Reaxys Name palladium diacetate Reaxys	 Name potassium fluoride Reaxys Name palladium diacetate Reaxys	20.0	20.0	30.0

Evaluation of the Michael additions used by Marcou et al. ²

Among the 52 reactions, 34 of them are found in the final dataset, and we have detailed information about these reactions for evaluation. The model is retrained with these reactions excluded from the training set. The top-ten accuracy of these 34 reactions for similar matches is 47.0%, and the accuracy for similarly matching catalyst, solvent 1 and reagent 1 is 55.9%, lower than the overall accuracy of the entire dataset, yet a significant improvement over literature results.² Furthermore, it has been pointed out that many of these reactions can occur under different conditions, meaning some predictions are not necessarily wrong, even when they differ from the recorded context. For example, Supplementary Figure 5 shows reactions where the exact recorded condition is not predicted. The first example has a different solvent predicted, but it is similar to the recorded solvent (both are alcohols).³ For the second example, although the model does not suggest the correct solvent and reagent in the first choice, it recognizes the need for basic conditions in the subsequent suggestions, with the second suggestion being piperidine and the third being triethylamine.⁴ Supplementary Figure 5(C) is another example of data quality that complicates the analysis. Two reagents NaHCO₃ and HCl are not commonly used in the same reactions but in separate steps which is the case in the reported procedure.⁵ Additionally, the true solvent used for the reaction is ethanol where diethyl ether is used as an extraction solvent in workup. The full prediction results are presented in Supplementary Table 7.



Supplementary Figure 5. Examples of Michael additions where none of the recorded c, s1 or r1 are predicted. A) Methanol is predicted, which is similar to the recorded solvent (isopropanol); B) the top prediction is incorrect, but subsequent predictions suggest bases as reagents (the second suggestion is piperidine and the third is triethylamine); C) the reaction has incompatible reagents that are used in different stages but not reflected in the reaction record.

Supplementary Table 7. 34 examples of Michael additions that are used for evaluation in the work of Marcou et al.

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature/°C	Top Predicted Temperature/°C	Closest Predicted Temperature/°C
1493719					20.0	23.7	23.7
605835					N/A	34.6	32.9
1288921					20.0	29.9	29.9
24352854			 Reaxys Name potassium iodide		N/A	67.4	57.3
8861928					N/A	18.6	29.6
23748241					10.0	26.3	26.3
25669226					N/A	57.2	57.2
23227051					N/A	30.7	30.7
24279555					N/A	36.3	36.3
24717084					N/A	34.9	34.9
24971546					N/A	46.9	50.0
694909					30.0	35.6	26.5
46547					N/A	60.8	60.8
1456140					20.0	37.4	37.4
24879120					N/A	79.5	79.5
2924844					0.0	51.9	16.7
22879782					N/A	43.4	48.4
24889073					N/A	43.7	43.7
1284594					20.0	14.3	14.3
24576080			 Reaxys Name ammonium chloride		N/A	-3.6	-3.6
3363418					N/A	61.1	29.8
24576110					N/A	73.3	73.3
701198					20.0	47.0	27.6
24581594					N/A	19.6	36.2
649317					20.0	25.9	25.9
4051					20.0	28.6	28.6
1297143					10.0	24.0	24.0

4046							90.0	33.8	33.8
35987442							50.0	57.8	57.8
24722182							N/A	32.4	26.5
24503535			H ₂ O	H ₂ O			N/A	62.4	62.4
22567748		H ₂ O	 Bi-	 Bi-			80.0	38.3	41.3
24960665							N/A	66.9	44.4
23665071							60.0	28.4	28.4

Feature definition used in the Morgan fingerprints

Supplementary Table 8. Feature definition as defined by Gobbi et al. ⁶

Invariants	SMARTS
Hydrogen bond donor	<chem>[N;!H0;v3],[N;!H0;+1;v4],[O,S;H1;+0],[n;H1;+0]</chem>
Hydrogen bond acceptor	<chem>[\$([O,S;H1;v2]-[!\$(*4[O,N,P,S]))],[O,S;H0;v2],[O,S;-],[N&v3;H1,H2]-[!\$(*4[O,N,P,S]))],[N;v3;H0],[n,o,s;+0],F]</chem>
Basic group	<chem>[\$([N;H2&+0][C,a];!\$([C,a](4O))),\$([N;H1&+0]([C,a];!\$([C,a](4O))))[C,a];!\$([C,a](4O))),\$([N;H0&+0]([C;!\$C(4O)))([C;!\$C(4O))][C;!\$C(4O))],[N,n;X2;+0]</chem>
Hydrophobic group	<chem>[\$([C;H2,H1](14*)[C;H2,H1][C;H2,H1][C;H1,H2,H3];!\$C4*),\$C([C;H2,H3])([C;H2,H3])[C;H2,H3])]</chem>
Acidic group	<chem>[O;H1]-[C,S](4[O,S,P])</chem>
Halogen	<chem>[F,Cl,Br,I]</chem>
Attachment point to aliphatic ring	<chem>[\$([A;D3](@*)(@*)~*)]</chem>
Attachment point to aromatic ring	<chem>[\$([a;D3](@*)(@*)*)]</chem>
Any unusual atom (not H,C,N,O,F,S,Cl,Br,I)	<chem>[!#1;!#6;!#7;!#8;!#9;!#16;!#17;!#35;!#53]</chem>

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