Using Machine Learning to Predict Suitable Conditions for Organic Reactions

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Supplementary Information

Computational methods

Neural network models are developed in Keras (Version 2.0.2) and trained using the Theano backend (Version 1.0.0). All trainings are performed on a single NVIDIA GeForce GTX 1080 GPU. The code is developed in Python 2.7.

Frequency vs rank plots for catalysts, reagents and solvents

Supplementary Figures 1-3 are the frequency vs. rank plots for catalyst, solvent and reagent in Reaxys. It can be seen some duplicated records of the same chemical exists (e.g. palladium diacetate as catalyst and sodium hydroxide as reagent). We kept these as different classes as there is not a good way to systematically identify and curate this issue as it is not clear what is the pattern of these duplicated chemicals with different ids. However, during the training, the model learns that they are very similar entities because one or another of these duplicated chemicals are used in the same type of reactions.



Supplementary Figure 1. Frequency vs. rank plot for catalyst in Reaxys (with the top ones labeled in yellow)



Supplementary Figure 2. Frequency vs. rank plot for solvent in Reaxys (with the top ones labeled in yellow)



Supplementary Figure 3. Frequency vs. rank plot for reagent in Reaxys (with the top ones labeled in yellow)

Comparison of prediction accuracies with a null model

A null model is defined to always give the same prediction of top ten combinations chosen based on the frequencies of the catalysts, solvents and reagents. The top 10 combinations of the null model are listed below in Supplementary Table 1.

Supplementary Table 1. The top-ten combinations used in the null model

Rank	Catalyst	Solvent 1	Solvent 2	Reagent 1	Reagent 2
1					
2		DCM ^a			
3		THF			
4				TEA	
5				K_2CO_3	
6		DCM		TEA	
7		THF		TEA	
8		DCM		K_2CO_3	
9		THF		K ₂ CO ₃	
10	Pd on activated				
	charcoal				

^a: DCM: dichloromethane, THF: tetrahydrofuran, TEA: triethylamine

The comparison of prediction accuracies are shown in Supplementary Table 2. The accuracy values shown are for top-three exact matches. In general, it can be seen that the accuracy values are much lower for the null model. The accuracy of c, s2 and r2 predictions for the null model are high, since a majority of reactions do not use a catalyst, a second solvent or a second reagent, but the trained model is still better than the null model by a large margin.

Supplementary Table 2. Comparison of accuracy for the true context to be in the top-3 predictions to a null model

	Trained neural network model	Null model
С	93.6%	87.3%
s1	75.8%	49.4%
s2	90.1%	85.6%
r1	73.2%	22.0%
r2	89.3%	82.3%
c, s1, r1	57.3%	5.7%
c, s1, s2, r1, r2	50.1%	4.7%

^a c, s1, s2, r1, r2 refer to catalyst, solvent 1, solvent 2, reagent 1 and reagent 2, respectively;

For temperature prediction, we analyzed a baseline model that predicts the most frequently used temperature for all reactions. Supplementary Figure 4 shows the distribution of temperature for reactions in the test set. The most frequently used temperature is the room temperature (20 °C) which covers a majority of reaction, and the accuracy of the predicted temperature by the baseline model (which is always 20 °C) being within the $\pm 10^{\circ}$ C or $\pm 20^{\circ}$ C range of the recorded temperature are 40.0% and 49.4%. In the meantime the distribution spans a wide range. Simply predicting the room temperature (20 °C) will result in a mean absolute error of 35.3 °C, which is significantly larger than prediction given by the trained model and would be misleading for reactions that require high or low temperatures.



Supplementary Figure 4. Temperature distribution for reactions in the test set

Full list of evaluation of reaction examples

Supplementary Table 3. 62 reactions from eleven reaction types randomly chosen from the test dataset

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature °C	Top Predicted Temperature	Closest Predicted Temperature	Reaction type
5301921	$\bigcirc \mathcal{I}_{\mathcal{I}} \xrightarrow{\mathcal{I}}_{\mathcal{I}} \bigcirc \longrightarrow \bigcirc \mathcal{I}_{\mathcal{I}} \xrightarrow{\mathcal{I}}_{\mathcal{I}} \bigcirc \bigcirc \bigcirc \mathcal{I}_{\mathcal{I}} \xrightarrow{\mathcal{I}}_{\mathcal{I}} \bigcirc $	—04 —0 Ka+	040 Ka+	040 Na+	20.0	19.3	19.3	Hydrolysis
8712792	$\neg \neg \downarrow_{s-\bigcirc} - \bigcirc \longrightarrow **-\bigcirc - \bigcirc \frown \bigcirc$	∕он к+ но.	К+ НО-	—сн к+ но.	N/A	104.4	59.7	Hydrolysis
5261303	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	∕он Ком-Мо2 №3	∕o ₄ →₩2	∕о _м кан-тиа	EN/A	92.9	\$8.2	Hydrolysis
8655717	Jool, ← Jool,	С — он Nat но.	j −01 № H0.	С) — ОН Кан ИО.	NA	50.2	30.2	Hydrolysis
5302180	$och + b \rightarrow och + b$		-or c=0-	-он о= <u><u></u> он о=<u></u> он о=<u></u></u>	57.5	35.1	35.1	Hydrolysis
53162	$-\alpha_1 \circ \beta_1 \circ \beta_0 \longrightarrow \circ \beta_1 \circ \beta_0 \circ \beta$	~o~	a∼a	~o~	0.0	12.2	5.5	Esterification
5185761	- " " l l l l l l l l l l l l l l l l l		¹ y [™] t ² W		10.0	15.7	11.2	Esterification
8698819	$\sim \sim \qquad \qquad$	L-t-Y	°~~° "	ĊO	95.0	51.6	59.7	Esterification
8669084	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	1 326 * 7			NA	15.0	15.0	Esterification
5336683	$\phi^{\dagger}\phi + h^{\dagger}\phi + \phi^{\dagger}\phi \rightarrow + h^{\dagger}\phi + \phi^{\dagger}\phi $	Y~0	\uparrow	Y*•	20.0	548.2	48.2	Esterification
5220300	$\dot{f}_{0}^{0} \rightarrow \dot{f}_{0}^{0}$	© ∽∘∽ ĭ	~~~ "~~"	Rearys Name lithium bronide	-78.0	5.7	13.6	Alkylation
2820838	$\sim \rightarrow \sim \sim$	elles	or to	orten	54.5	71.6	71.6	Alkylation
38568283	$\sim\sim 2 + + + + + + + + + + + + + + + + + +$	Y O	°~° ⟨)~-(°~° √)(N/A	18.4	18.4	Alkylation
8738792	o or o 3 - Souro	_=x<	°∼° →≺	—≡× _}⊷<	20.0	23.1	47.6	Alkylation
8574982	20 0 tox - Stat	∽ ₀₁ ю ю	∽ ₀₁ to 10	∕о _в ка но.	40.0	28.2	28.2	Alkylation
8591626	$\sim \sim $	—Он №+ но. но—он	—он №+ но- но-он	—он №+ но. но-он	10.0	15.1	15.1	Epoxidation
38647861	$\neg \partial \langle - \longrightarrow \gamma \rangle \langle - \neg \neg \gamma \rangle \langle - \neg \rangle \langle - \neg \gamma \rangle \langle - \neg \gamma \rangle \langle - \neg \rangle \rangle \langle - \neg \rangle \langle - \neg \rangle \langle - $	Б.У.У. В. но−он	L K	~~L	0.0	5.4	47.8	Epoxidation
8655813	$\underset{i=1}{\overset{i=1}{$	ON 1043	ок миз	ок вка	20.0	22.5	22.5	Epoxidation
8720054	$\xrightarrow{+}_{\mathbb{P}} \xrightarrow{\mathbb{P}} $			a~a HO-O	20.0	15.2	15.2	Epoxidation
8560867	$\cdot \circ \circ$	$O \downarrow $ * "	D L Y "	DLY «	0.0	20.7	20.7	Epoxidation
8598928	$\neg \uparrow \uparrow \uparrow \downarrow \downarrow \rightarrow \neg \uparrow \uparrow \uparrow \downarrow \uparrow \downarrow \downarrow$	$\label{eq:alpha} \alpha ~ \sigma ~ \begin{tabular}{c} \sigma & \sigma & \\ & \\ & \\ & \\ & \\ & \\ & \\ & \\ &$	to to a with	Ranys Name softum bydråle	0.0	12.7	20.9	Epoxidation
3229646	$\swarrow^{\alpha}_{\alpha} \overset{\alpha}{\longrightarrow} \overset{\alpha}{\overset{\alpha}{\longrightarrow} \overset{\alpha}{\longrightarrow} \overset{\alpha}{\longrightarrow} \overset{\alpha}{\longrightarrow} \overset{\alpha}{\longrightarrow} \overset{\alpha}{\overset{\alpha}{\longrightarrow} \overset{\alpha}{\overset}{$	a~a C	OrO Ka+	ONO Na+	0.0	235.8	35.8	Epoxidation
10069687	$\mathcal{J}_{\mathcal{F}}^{\mathcal{F}} \mathcal{J}_{\mathcal{F}}^{\mathcal{F}} \longrightarrow \mathcal{J}_{\mathcal{F}}^{\mathcal{F}} \mathcal{J}_{\mathcal{F}}^{\mathcal{F}}$	Kijarty lactoas() oor waalim	L 100 He for the real L an	Rearys Name bis(certylacetonate):novunatium	25.0	15.2	48.5	Epoxidation
9925682	$\mathcal{A}_{\alpha} \xrightarrow{\mathcal{C}_{\alpha}} \xrightarrow{\mathcal{C}_{\alpha}} \longrightarrow \xrightarrow{\mathcal{C}_{\alpha}} \mathcal{C$	Image: Construction of the second s	and one was non-of-the	Rearys Name bis(acrtylacetonate)oxovanadium	N/A	6.0	34.3	Epoxidation

Reaxys Name sodium hydride Reaxys Name sodium hydride O→ ¹/₂³/₂⁴, ⁴/₂⁴ \bigcirc \bigcirc $\operatorname{raise} \operatorname{cont}_{\mathbb{P}} \operatorname{cont}_{\mathbb$ Reaxys Name sodium hydride -=n -=n (N Reavys Name lithium chloride Reaxys Name sodium hydride $\mathcal{Y}_{209(10)} \xrightarrow{\circ} \mathcal{Y}_{1}^{\circ} \xrightarrow{\circ} \mathcal{Y}_{2}^{\circ} \xrightarrow{\circ} \mathcal{Y}_{2}$ Reaxys Name sedium hydride Reaxys Name sodium hydride Reaxys Name sodium hydride a∕a \bigcirc u∕a $_{2876112}$ $^{\circ}$ $^$ \bigcirc D a∼a \bigcirc С) ° с ° к+ \square NH2- №+ \bigcirc \checkmark K+ $\begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & &$ HAT OF HAT `Y~~ __ `n~°° n⊕ $||_{\mathcal{H}^{(n)}} = \left\{ \begin{array}{c} & & \\ &$ zn... [] _____ zn··· 💭 🛴 zn... 💭 🛴 a∼a No b b b $|+_{\mathcal{Y}_{n}} - \mathcal{Y}_{n} - \mathcal{Y}$ a∼a ≫t+t a∼a Notert $_{351204} \xrightarrow{\circ}_{0} \xrightarrow{\circ}_{0}$ C HCI С на С на AH7- Li+ Altr- Li+ AlH7- U+ $26961 \left| \begin{array}{c} 0 \\ + 1 \\ + 1 \\ - 0 \end{array} \right|^{-0} \left| \begin{array}{c} 0 \\ -$ $_{265062}$ $_{a}$ $\xrightarrow{}$ $_{a}$ $\xrightarrow{}$ $_{a}$ $\xrightarrow{}$ $_{a}$ $\xrightarrow{}$ $_{a}$ Site Site Name palladium on activated charces 124.2 AH7- U+ $519786 \xrightarrow{H2N} \xrightarrow{A} \xrightarrow{D} \xrightarrow{H2N} \xrightarrow{A} \xrightarrow{D} \xrightarrow{D}$ AH7- Li+ CH OCTO Reaxys Name palladium on activated charco ·· Zh·· HCl ~ 1.0 a∼a °cso vo $= 0 \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}_{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}_{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}} (\cdots \to \mathcal{A} \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}) \xrightarrow{\mathcal{A}}$ а но-он ада во-он 3.1 3.1 Reaxys Name jones reagent Reaxys Name jones reagent Reaxys Name jones reagent $\bigcirc = \circ \longrightarrow \bigcirc = \circ$ HO-OFF OFF _____он но-он $\overset{\circ}{\rightarrow}$ Reavys Name Jone's reasent Reavys Name silica eel Reaxys Name Jone's reagent Reaxys Name Jone's reagent 176.7 130.2 $77260 \qquad \bigcirc NH \qquad \bigcirc N \rightarrow \qquad \bigcirc N \rightarrow \bigcirc (N \rightarrow)) (N \rightarrow) (N \rightarrow)) (N \rightarrow) (N$ ° X K+ 0-04 \square

1126041	$ \sim \circ \circ \rightarrow \circ \circ$	Shorten		Gon a C + C + C + C + C + C + C + C + C + C	30.0	949	34.7	Buchwald- Hartwig
1107702	$ \bigcirc \bigcirc \bigcirc \bigcirc \bigcirc \frown \bigcirc \frown \frown \frown \frown \frown \frown \frown \bigcirc \frown \bigcirc \frown \bigcirc $	De la construcción discoste			120.0 ye	120.5	120.5	Bochwald- Hartwig
2604617	$ (\mathcal{A}_{\mathcal{A}}) \to (\mathcal{A}) \to ($	C ⁺ ₀ ^a [→] ₀ [−] C ⁺ Sincerate Reavys Name pollution	C ² ⁽²) ⁽²) ⁽²⁾ C ² ⁽²⁾	Constraint of the second secon	900	104.3	104.3	Buchwald- Hartwig
3382323	$\frac{1}{2} 0 + 0 - 0 \rightarrow \frac{1}{2} 0 0 + 0 - 0 \rightarrow 0 + 0 + 0 + 0 + 0 + 0 + 0 + 0 + 0 +$	0 N+2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 № 2	$ \begin{array}{c} & & \\ & & $	80.0	113.5	113.5	Buchwald- Hartwig
9603192	$\uparrow \sim \sim$			- For the and	19.0	31.5	39.1	Grubbs
3603226	$\mathcal{G}^{+}_{\mathcal{A}} \to \mathcal{G}^{+}_{\mathcal{A}} \to$		c√a	~~ (Sa 22)	40.0	35.6	22.6	Grubbs
2584073	$ = \int_{\mathbb{C}} \int_{\mathbb{C}$	The first state of the state of	Harry Liberatoria	The line show	100.0	101.8	101.8	Suzuki
4033276	$\mathbb{E}_{\mathcal{O}}^{-1} (\mathbb{O}_{\mathcal{O}}^{0}) \xrightarrow{\mathbb{O}}_{\mathcal{O}}^{0} \longrightarrow \mathbb{O}_{\mathcal{O}}^{-1} (\mathbb{O}_{\mathcal{O}}^{0}) \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \mathbb{O}_{\mathcal{O}}^{0} \longrightarrow \mathbb{O}_{\mathcal{O}}^{0} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \mathbb{O}_{\mathcal{O}}^{0} \longrightarrow \mathbb{O}_{\mathcal{O}}^{0} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \mathbb{O}_{\mathcal{O}}^{0} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \mathbb{O}_{\mathcal{O}}^{0} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}}^{0}} \xrightarrow{\mathbb{O}_{\mathcal{O}^{0}}} \xrightarrow{\mathbb{O}_{$	$-\equiv N \xrightarrow{\circ} \int_{0}^{\circ} K_{*}$ Reaxys Name palladium diacetate	~_он н≥о °~с к+	H20 0 0 Na+	8.0	51.5	81.1	Suzuki
9208506	$\neg \beta \gamma \gamma \gamma \gamma \gamma \gamma \rightarrow \neg \beta \gamma \gamma$	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	~~~~ NO ~~ No		N∕A	84.1	84.1	Suzuki
2888571	$\cdot \cdot $	γ γ ο γ γ κ. Reaxys Name (1,1)- biddirbury/hboahim/ferrorane/hallafilm(Π) dirbheide			N∕A	95.3	95.3	Suzuki
1004052	$ p \not = p \not = p \rightarrow p \rightarrow p \not = p \rightarrow p \rightarrow p \not = p \rightarrow p$	Resy's Name palladium discrate Ressys	Resty Name pilladium diacetae. Resty	Image: Constraint of the second sec	20.0	52.2	52.2	Suzuki

Supplementary Table 4. 100 reactions randomly chosen from the test dataset

Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperatu °C	Top Predicted Temperatur °C	Closest Predicted e/Temperature/ *C
1383215	$\overset{_{\mathrm{H}}}{\rightarrow} \overset{_{\mathrm{L}}}{\rightarrow} $	√⊃ o ^l a	ord _a	،	N/A	40.3	32.1
10061846	$\sim \sim $	r →r 0 □ □ □ □ □ □ □ □ □ □ □ □ □ □ □ □ □ □	$ \begin{array}{c} & f \\ $	r → x+3 G √ G	60.0	29.4	23.8
44110175	$\begin{array}{cccc} X_{p^*} & Y^{\circ} & \longrightarrow & \begin{array}{c} & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ \end{array} \end{array}$	$\uparrow^{\circ}\mathcal{A}^{\circ}$	Y*Y ⊔∗	D D Y*Y u*	-78.0	-65.6	-66.6
10261736	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		= Reaxys Name annuonium cerium(IV) nitrate	— \equiv H Reaxys Name annovaium cerium(IV) nitrate	0.0	15.1	15.1
40887565	$\overset{m_{n-1}}{\longrightarrow} \overset{m_{n-1}}{\longrightarrow} $	\bigcirc	\bigcirc	↔	60.0	54.3	54.3
36028195	$ _{-} _{+} $	Cu+ Cu+ Reaxys Name potassium fluoride	Cu+ Cu+ Reaxys Name potassium fluccide	Cu+ Cu+ Reaxys Name potassium fluoride	60.0	49.1	49.1
2176355	$+\stackrel{\circ}{O} \longrightarrow +\stackrel{\circ}{O}$	— OH BH4- Ns+ Reaxys Name cerium(III) chloride	AH7. ∐+	— OH BH4- Na+ Reaxys Name cerium(III) chloride	0.0	-18.7	-4.2
2069147	$\mathcal{A}_{\mathcal{C}} \mathcal{A}_{\mathcal{C}} \mathcal{A}_{\mathcal{C}} \mathcal{A}_{\mathcal{C}}$	↓ ₁₀		L _H . Li+	N/A	42.6	-13.1
1720472	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Ť	~ y ~o	` ↑~•	N/A	54.0	54.0
3204380	${}^{b}\mathcal{D}_{a}^{c}\mathcal{D}_{a}^{c}\mathcal{D}_{a}^{c} \longrightarrow {}^{b}\mathcal{D}_{a}^{c}\mathcal{D}_{a}^{c}\mathcal{D}_{a}^{c}$	—он к+ но-		ON K+ HO	N/A	26.2	24.5
4822624	$200+00 \rightarrow 0+00$	°~° ,)~	a~a J~	°~°),√	40.0	5.7	5.7
34927996	$ \begin{array}{cccc} & & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ &$	a~a _}~<	$\mathfrak{p} \not\prec f$	°~° , , , , , , , , , , , , , , , , , ,	N/A	14.5	12.1
1013620	$\sim \backsim \sim \sim$	-0 Na+			N/A	263.4	263.4
3928353	${}_{o}\mathcal{A}^{+}_{wa}, {}_{o}^{\perp}{}_{o}^{+}{}_{o}^{a} \longrightarrow {}_{i}\mathcal{F}^{w}_{i}{}_{o}^{\perp}{}_{o}^{a}$	\bigcirc	\bigcirc	\bigcirc	60.0	66.7	66.7
26314583	$\begin{array}{c} (1) \\ (1) \\ (2) \\$	\bigcirc	Ð	Ð	N/A	25.8	225.8
38789242	$\neg \uparrow \neg \neg$	OH H2O BrBr HBr	— СН H2O Br-Br HBr	—он н2о в/-вг наг	25.0	36.3	36.3
25852492	$\sim \mathcal{O}_{\mathcal{O}}}}}}}}}}$				90.0	88.2	98.2
3665938	$\overset{\text{\tiny ND}}{\longrightarrow}\overset{\text{\tiny ND}}{\longrightarrow}\overset{\text{\tiny ND}}{\longrightarrow}\overset{\text{\tiny ND}}{\longrightarrow}\overset{\text{\tiny ND}}{\longleftarrow}\overset{\text{\tiny ND}}{\longleftarrow}\overset{\text{\tiny ND}}$	Сн к+ нс-	С — он нго Reaxys Name lithium hydroxide hydraie	—он к+ но-	N/A	28.7	36.7
3958431	$a \stackrel{\circ}{\underset{\alpha}{\overset{\alpha}{\overset{\alpha}{\overset{\alpha}}{\overset{\alpha}{\overset{\alpha}}{\overset{\alpha}{\alpha$	ω	Cu	cu	190.0	254.5	254.5
2215973	$-\mathcal{O}^*\!\!\times \longrightarrow \mathcal{O}^{***}\mathcal{O}$	o≠La ∽∽∽u		a~s~a ~~~n	N∕A	-15.5	-47.1
4940666	$\overset{\circ}{\longrightarrow} \overset{\circ}{\longleftarrow} \overset{\circ}{\longrightarrow} \overset{\circ}{\to} \overset{\circ}$	r − r o= − c ou+2 ∽on		Си	100.0	85.4	60.6
5310995	$ \begin{array}{c} \prod_{j=2}^{p_{i2}} & & \\ \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i1}} & \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i1}} & \\ \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \\ \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} & \\ \prod_{i=1}^{p_{i2}} & \prod_{i=1}^{p_{i2}} &$	√₀ va₀	∽œ.		70.0	42.1	344.6
25900877	$ \overset{u_{2}}{\overset{u_{2}}}}}{\overset{u_{2}}{u_{u$	\square	\square	\square	65.0	46.3	46.3
35559718	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	a~a artical production area area area area area area area are	°∼° ^d osd _{on} √	°~° ° ² t ^{2°} ∕	30.0	21.7	21.7
42570631	$\mathbb{CC} \mathbb{C} $	0 0 	d in the second	0 ^{0H} ⊢1	120.0	104.7	104.7

 $\left| \int_{\mathbb{R}^{n}} \int_{\mathbb{R}^{n}}$ \square \square $(1427)e \sim (1) \int_{0}^{\infty} (1 - 1) \int_{0}^{\infty} (1 \begin{array}{c} u_{12}^{HC} & u_{20} & u_{0}^{-1} - \int_{-H}^{0} \int_{-H}^{0} u_{1} & u_{1+} \\ u_{10}^{HC} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} - \int_{-H}^{0} \int_{-H}^{0} \int_{-H}^{0} \int_{-H}^{0} \int_{-H}^{0} u_{10}^{HC} & u_{10}^{-1} & u_{10}^{-1} \\ u_{10}^{HC} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} & u_{10}^{-1} \\ u_{10}^{HC} & u_{10}^{-1} & u_{1$ J. $\operatorname{SSSSITZ} - \operatorname{Olog} - \operatorname{Olog} + \operatorname{Olog} + \operatorname{Olog} - \operatorname{Olog} + \operatorname$
 Image: Strategy Name borns trifluoride diethyl
 Image: Strategy Name borns trifluoride diethyl effective
 Image: Strategy Name borns trifluoride diethyl effective
 Image: Strategy Name borns trifluoride diethyl effective
 N/NO a~a v a~a v $0^{100} \bigcirc 0^{-1} \bigcirc$ 0=9=0 0-1 0-0-0-1 0-1 0-1 0-Ni 🚫 ____ ____ Cl_Cl_0= 0, Na+ HO-O (1) = (1 + 1) + (1 + 1)₩⁰ ⁰ ⁰ K+ ₩~0 ~~ 0 к+ ₩ 0 -0 0 K+ J~ +1.3% J № нго уч но. $_{3834}$ $\underset{Mg}{\longrightarrow} \longrightarrow$ P CH Reaxys Name silica ge °=s Reaxys ID 19278319 $\left|_{212201}\right|_{\mathcal{H}} \stackrel{0}{\longrightarrow} \stackrel{0}{\rightarrow} \stackrel{0}{\rightarrow}$ Со Li+ но-Со ціт но- $\text{serverse} \xrightarrow{\mu_{i}} \overset{\mu_{i}}{\longrightarrow} \overset{\mu_{i$ D H20 J D H20 J $31607336 \longrightarrow \stackrel{h_{0}}{\longrightarrow} 0 \longrightarrow 0 \longrightarrow 0$ Reaxys Name bor $d \sim d \sim s \sim s \sim s \sim s \sim s$ ~^{Cl} Cl _S(_N=N+=N- _Na+ N+ D Br ОН — он Н Reaxys Name palladium он activated CH OH H Reaxys Name palla 2H_A,^{2H} Li* 0 2H 4. U+ 2H, 2H U+

 $\mathcal{P}_{\mathsf{SGED}} \bigcirc \mathcal{O}_{\mathcal{F}} \bigcirc \longrightarrow \ \mathcal{O}_{\mathcal{F}} \bigcirc \mathcal{O}_{\mathcal{F}})$ () — " Со нсі нсі н20 _____ —он мнз $\operatorname{constraint} (\mathcal{A}_{\mathcal{A}}) \xrightarrow{\circ} \mathcal{A}_{\mathcal{A}} \xrightarrow{\circ} \mathcal{A} \xrightarrow$ \sim^{α} _____N | ____N a~a $\operatorname{unu} + \operatorname{f}^{\circ} \operatorname{u}^{\circ} \operatorname{f}^{\circ} \operatorname$ \bigcirc \bigcirc $\begin{array}{ccc} & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ &$ $\bigcirc - \bigcirc$ \bigcirc к+ но- $\underset{m_{0}}{\overset{m_{0}}{\rightarrow}} \overset{m_{0}}{\rightarrow} \overset{m_{0$ d y d a – a d Å d $= 0, \quad 1 \to 0, \quad 1 \to$ ₩⁰ ⁰ ⁰ к+ ₩⁰ ⁰ ⁰ к+ and the and total ~ L' → NI C L K+ Control Contro 3816753 $0 \xrightarrow{M^{H}}_{\leftarrow 1} a^{HO}_{\leftarrow} \xrightarrow{MO}_{M^{H}} \longrightarrow \underset{HO}{\overset{M}}_{H^{H}} \xrightarrow{MO}_{H^{H}} \xrightarrow{MO}_{H^{H}}$ ₩⁰ ⁰ ⁰ к+ $\operatorname{SIMTZ} \bigoplus_{s \to t} \operatorname{\operatorname{Suff}} (\mathcal{O}_{-st} \xrightarrow{s_{s}} (\mathcal{O}_{-s} \xrightarrow{s_{s}} \mathcal{O}_{-s} \xrightarrow{s_{$ $\lim_{n\to\infty} \left\{ \int_{\Omega} \int_{\Omega}$ $13901 \xrightarrow{i} \mathcal{J}^{i} \mathcal{O} \xrightarrow{i} \mathcal{O}^{i} \xrightarrow{i} \longrightarrow \xrightarrow{i} \mathcal{O}^{i} \overset{i}{\mathcal{I}} \overset{i$ ^{OH} → Na+ OH → Na+ CH O. Na+ D J F D S F ↓ N* F- H20 ∽он ∽он ____он $\underset{\mu_{\sigma} \leftarrow \sigma}{ } \xrightarrow{\mu_{\sigma}} \xrightarrow{$ ava N r 21.0 $(\mathbb{C}^{n} / \mathbb{C}^{n}) \to (\mathbb{C}^{n} / \mathbb{C}^{n}) \to (\mathbb{$ 22880779 HO HO HF HO HFa~a ^oed a~a of a a~a °er а~а а~а ц -он

2255356		H H	1	H	299.9	263.1	263.1
4667439	$\xrightarrow{-n_{p-1}} \bigcirc \downarrow \bigcirc$				NA	-17.4	-17.4
39121363	$\mathcal{B}^{\mathcal{C}} \to \mathcal{O}^{\mathcal{D}}$	<i>—</i>	Reazys Name sodium bis(2-methoxyethoxy)duminium	Reaxys Name sodium bis(2 methoxyethoxy)aluminium tiltyytride	NA	65.8	65.8
42279161	$\mathcal{C} \xrightarrow{\mathcal{C}} \mathcal{C} \xrightarrow{\mathcal{C}} $	a, a o=i → , , , , , , , , , , , , , , , , , ,	a∼a o= <u>0</u> +{{}-{}- n{}-	a∼a o=j_ou	20.0	22.3	19.9
1821342	$\mathcal{A} \mathcal{O}^{2} \mathcal{O}^{-} \mathcal{O}^$	H20 K+ H0-	AM7- Li+	К+ НО-	100.0	10.5	103.5
2170049	$ \overset{\circ}{\sim} \overset{\circ}{\sim} \overset{\circ}{\circ} \overset{\circ}{\sim} \circ$	60			30.0	35.0	35.0
28774608	$\int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \int_{-\infty}^{\infty$		Претисти пон		-78.0	56.2	16.7
1320851	$\gamma \circ \circ \circ \circ \to \circ \uparrow \circ \circ$	a a a	` †∽₀	a y a	N/A	51.6	32.7
9372184		The way in the second s	Ч [~] 0 ~ыл ксуди	↓≪ο ~m~ nr∕fn nd	20.0	21.2	21.2
40224906	${}^{\mathrm{Hill}} \mathcal{O} \to {}^{\mathrm{holl}} \mathcal{O} \to {}^{$	G, G G, G , G , G , G , G , G , G , G ,	a, a a, a fine reaction of the second	G, G G, G , G Reaxys Name O-(benzotriazol- 1-y)-NNN N-tetramethyluronium tetralluoroborate	27.5	21.1	21.1
29311031	$f_{\gamma}^{2}(\mathcal{G}^{\alpha}) \longrightarrow f_{\gamma}^{2}(\mathcal{G}^{\alpha})$	°+C° № °+C	ort a	Ø→(120.0	98.3	97.4
40807619	$ \begin{tabular}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & \\ & & & & \\ & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & $	¥~°° ² ¥° K+	₩~° ² χ° K+	¥~° °ξ° κ,	120.0	64.6	54.6
27931253	$\bigcirc \stackrel{0}{\longrightarrow} \stackrel{0}{\rightarrow} $	O Reaxys Name sodium hydride Reaxys Name sodiu	a∼a)∽	Rexxys Name sodium hydride	-20.0	10.4	94
39062813	$\sum_{j=0}^{n} (j \to j) = (j$	a∼a v⊘ w	a المربع الم	a∽a 10 №	0.0	19.8	6.1
4410590	$ \qquad \qquad$	— СН 1843	Ah17- Li+	—он к+ но-	NA	12.1	36.3
27978248	$ \overset{\circ \wedge}{\to} \overset{\circ}{\to} \overset$	—он н20	——————————————————————————————————————	−oH ci∕a	N/A	56.1	45.2
27784122	$ \overset{\circ}{\overset{\circ}{\underset{m_{q}}}} \overset{\circ}{\underset{m_{q}}} }{\underset{m_{q}}} {\underset{m_{q}}} }{\underset{m_{q}}} }{\underset{m_{q}}$	~_0H	Ð	~он	130.0	47.3	76.0
28190848	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	ÎI Çind			20.0	39.7	39.7
24898802	$ \underset{\mu}{\bigcirc} \xrightarrow{\operatorname{res}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\longrightarrow} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{res}} \underset{\mu}{\overset{\operatorname{res}}}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\operatorname{res}}{\overset{\operatorname{res}}{\longrightarrow}} \underset{\mu}{\overset{\tau}}{\overset{\tau}} \underset{\mu}{\overset{\tau}}{\overset{\operatorname{res}}} \underset{\mu}{\overset{\operatorname{res}}{\overset{\tau}}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}{\overset{\tau}} \underset{\mu}}{\overset{\tau}} \underset{\mu}} \underset{\mu}}{\overset{\tau}} \underset{\mu}} \underset{\mu}}{\overset{\tau}} \underset{\mu}}{\overset{\tau}} \underset{\mu}}{\overset{\tau}} \underset{\mu}} $	$\bigcirc \not \rightarrowtail \overset{\circ}{\sim} \star^*$	°, , , , , , , , , , , , , , , , , , ,	∑) ² ∕ K*	70.0	87.0	387.0
609622	$\neg \langle \mathcal{O} \land \to \land \neg \langle \mathcal{O} \rangle \rangle$		Ка+ НО-	Ка+ НО-	180.0	41.4	41.4
884394	$ \ \ \ \ \ \ \ \ \ \ \ \ \$	7~	`¥∽₀	7~0	70.0	62.9	62.9
7050517	$\sim \sim $		a_µco	a~keo	255.0	49.4	49.4
5021035	${}^{*}\!$	ή~ °ς, κ.	тр~о °~с° к+	Υ∽° °ς° κ+	80.0	49.3	49.3
26014672	$\overleftarrow{\hspace{-0.5ex} \leftarrow}^{\prime} \longrightarrow \overleftarrow{\hspace{-0.5ex} \leftarrow}^{\circ}$	CR60 C C C C C C C C C C C C C C C C C C C	0-0	С ж ⁰ а-а	N/A	180.7	113.6
37452339	$+\!$	Þ	Ð	Ŕ	140.0	125.2	125.2

$\mathcal{H}_{\mathcal{A}_{\mathcal{O}}}^{\mathcal{A}_{\mathcal{O}}} \xrightarrow{\mathcal{A}_{\mathcal{O}}} \mathcal{H}_{\mathcal{O}}^{\mathcal{A}}$	E C C C C C C C C C C C C C C C C C C C	↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓	D	20.0 17.	.1 1	17.1
$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ $	D X us	₩ u.	D X u	40.0 -25	5.5	25.5

Supplementary Table 5. 40 reactions from eleven reaction types randomly chosen from the test dataset
that has the least number of correctly or similarly predicted elements

Reaxys ID	Restion	True Context	Top Prediction	Closest Prediction	True Temperature *C	Top Predicted Temperature	Closest Predicted e/Temperatury	e Reaction type
8542026	$\downarrow \longrightarrow \downarrow \frown \frown \frown$	$\square \xrightarrow{\mu} \downarrow_{ar} \xrightarrow{\mu} \downarrow_{o}$	i Reaxys Name Lindur's catalyst	~_l l	-5.0	°C 25.0	°C 29.1	Reduction, Hydrolysis
5342174	$\overset{\mathcal{H}_{a}}{}_{f_{a}} \overset{\mathcal{H}_{a}}{} \overset{\mathcal{H}_{a}}{\longrightarrow} \overset{\mathcal{H}_{a}}{} \overset{\mathcal{H}_{a}}{$	a∼a H50 NO + + + + + + + + + + + + + + + + + +	864 Na+ Reaxys Name onne	264 Na+ Reaxys Name conne	20.0	-65.7	-65.7	Hydrolysis
5240460	$ \begin{array}{cccc} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & &$	-он а на одна страна к+	100 ° 0 No+	идо ⁰ -с ⁰ ка+	INA	48.4	48.4	Hydrolysis
8538458	$\sim {} }{} }{ }{ }{} }{ }{ }{ }{} }{ } }{ }{ }{ }{ } \stackrel$	—Он — Он Ю—Он Еслаун Nume phosphate buffer	a~a _}~	a~a _}~	39.0	14,4	14.4	Condensation, Hydrolysis, Oxidation
8651198	$\mathcal{X}^{\mathcal{A}} \xrightarrow{\mathcal{A}} \mathcal{A} \longrightarrow \overset{\mathcal{A}}{\to} \mathcal{A}^{\mathcal{A}}$		— он на	—он на	NA	25.5	27.1	Hydrolysis
10059687	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} $	Image: Contract of the second secon	L 1420 of of of the set of the set	Retays Name His(sert/facetonate/seconatalium	25.0	15.2	48.5	Epoxidation
9925682	$\mathcal{A}_{ar} \xrightarrow{f}_{ar} \longrightarrow \xrightarrow{f}_{ar} $	Difacetylacetonaticsovanadium	0~0 °+0 NH NO-0+0	Razys Name biskertylaestonate/our-wandlum	N/A	6.0	34.3	Sharpless
29729359	$\overleftarrow{\mathcal{O}}^{-n} \overleftarrow{\mathcal{O}}^{-n} \longrightarrow \overleftarrow{\mathcal{O}}^{-n} \overleftarrow{\mathcal{O}}^{-n}$	$\sum_{i=1\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\$	and a a a	$ \begin{array}{c} & & \\ & & $	10.0	25.7	60.8	Friedel-Crafts alkylation
8529950	$10^{-10} \xrightarrow{\mu} 10^{-10} \xrightarrow{\mu} 10^{-10}$		Reaxys Name copper(l) chloride	Reaxys Name copper(1) chloride	20.0	52.8	7.6	Alkylation, allylindation, Bromination
8526713	$\downarrow_{i} \uparrow \bigcirc \longrightarrow \bigcirc \frown \downarrow_{i} \frown \bigcirc \bigcirc \frown \downarrow_{i}$		D)**~ u.	D Jun un	-78.0	-14.9	14.9	deprotonation, Addition, Alkylation
8592120	$ \overset{\bigcirc}{\longrightarrow} \overset{\times}{\longrightarrow} \overset{\bigcirc}{\longrightarrow} \overset{\sim}{\longrightarrow} \overset{\sim}{\longrightarrow} \overset{\sim}{\longrightarrow} \overset{\circ}{\longrightarrow} \overset{\circ}{\to} \circ$	Сн ноо одоодо к+	Д на	Г. на	90.0	28.3	28.3	dealkylation
3629688	$\sim \ \circ f_{abc}^{+} f_{abc}^{+} \longrightarrow \ \circ f_{abc}^{+} f_{abc}^{+}$	a~a 100 № H2 / N Br.	Reaxys Name sodiam hydride	рани страниции на страниции н	20.0	13.4	24.6	Alkylation
34446920	$\mathcal{E}_{\mathcal{T}} \mathcal{O} \longrightarrow \mathcal{O}_{\mathcal{U}}^{\mathbb{L}} \mathcal{O}$	Сн −он /1 но.	Rexxys Name samarhum diiodide	°Z Z v	20.0	-9.7	26.5	[2,3]-Wittig Rearrangement
33420713	$ \rightarrow \rightarrow$		Reavys Name cozone		78.0	-58.5	70.5	Wittig reaction
28838907	$\bigcirc {}_{\mathcal{C}} \bigcirc \bigcirc \longrightarrow \bigcirc {}_{\mathcal{C}} \bigcirc $	α, σ α, σ κατ σ, το δ Reasys Name silica gel	Ð	αα Reavys Name 4 A molecular sieve	20.0	21.2	8.3	Wittig reaction
29349495	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$-\equiv_{N}$ $a_{\checkmark} \sigma$ $c_{\checkmark} \sigma$ Reaxy Name Ithium chloride	Rexeys Name uselium hydride	Reacy) Nune solium hydride	20.0	7.4	7.4	Wittig-Horner reaction
35093977	$\sim 0.88. \rightarrow 0.00$		-04 -0 Na+	-04 -5 Na+	20.0	17.6	17.6	Wittig Olefination
5266589	$\mathcal{F} \subseteq \mathbb{Q}_{p_{1}}^{n} \longrightarrow \mathbb{Q}_{p_{1}}^{n}$	— <u>—</u> я нао <u>сі</u>	Reaxys Nume silica gel Reaxys ID 13273590	Reverse Name field), charide Reverse Name silica nel	N/A	46.2	40.0	deprotection
8780005	$ \begin{array}{c} \overset{a}{\leftarrow} \overset{c}{\leftarrow} \overset{c}{\leftarrow$	—————————————————————————————————————	<u>الْم</u> 140		22.0	32.3	38.4	deprotection
3585244	$\mathcal{F}_{\mathcal{F}}^{\mathcal{F}} \longrightarrow \mathcal{F}_{\mathcal{F}}^{\mathcal{F}} \mathcal{D}$	ت اس Roxyy Name tin(1), shloride	04 H Reaxys Name pailadium 10 on activated carbon	—Ot II Reavys Name palladium 10 on activated carbon	20.0	24.8	24.8	Reduction
8692625	$ (\bigcup_{i=1}^{NQ} (\bigcup$	CH HZO H	H0~~~0	∼osi H Reaxys Name palladium on activated charcoal	100.0	87.6	42.0	Formylation, reduction
29313445		H20 H20 IN/3 LH	UH	100 ЦН	-60.0	3.6	-60.5	Birch reduction
42522603	$ \begin{array}{c} & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & $	₩0 LH	CM NH NG	Г. NO LH	63.0	-68.8	-58.8	Birch
38646319	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	10 H20 H2 H2 - H2 - H1	—он М	-04 J. HS. 54	20.0	25.3	25.3	Reduction

1025531	$\cdot \odot \overset{\sim}{\leftarrow} \overset{\sim}{\leftarrow} \overset{\sim}{\rightarrow} \overset{\sim}{\rightarrow} \overset{\sim}{\rightarrow} \overset{\circ}{\rightarrow} $	о ноо но-он Reasys Name potassium fluoride	D Y T T	Reasys Name cesium flooride	25.0	19,0	31.9	Tamao oxidation
5351285	$H_{0} \rightarrow \downarrow \uparrow \downarrow \downarrow$	→ Na+ o-Cl ++++++++++++++++++++++++++++++++++++	Na+ 0-Br	Na+ 0-8r	55.0	17.0	17.0	Oxidation
8520314	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	○ ○ ○ → H25e	~~~ , H25e	~~~ , H35e	20.0	20.5	20.5	Substitution, Oxidation
9273620	$\mathcal{A} \to \mathcal{A} \to $	$a \rightarrow a$ $b \rightarrow b \rightarrow b$ $a \rightarrow b \rightarrow b \rightarrow b \rightarrow b$ $a \rightarrow b \rightarrow $	a ho-	a ho-o c c c c c c c c c c c c c c c c c c	20.0	4.6	4.6	Baeyer- Villiger oxidation
3627506	$f_{\mathcal{F}}^{\mathcal{F}} \to \mathcal{O}^{\mathcal{F}} f_{\mathcal{F}}^{\mathcal{F}}$	المعنى المعنى (di-o-usivjphosphino/benzy)dipalladium(li)	100	-OH HO	80.0	89.1	44,4	Oxidation
2600766	$\hat{\beta}_{i}^{i} - \hat{\gamma}_{j}^{i} \longrightarrow \hat{\beta}_{j} \hat{\gamma}_{i}^{j} \hat{\rho}$	ни но с с с с с с с с с с с с с с с с с с	() () () () () () () () () () () () () (() () () () () () () () () () () () () (60.0	95.2	95.2	Suruki coupling
1034216	$\circ \to \to$	pathalism diacente Ferry Name	the second secon	$ \int_{-\infty}^{\infty} 100 = \int_{-\infty}^{\infty} \int_{-\infty}^{\infty} \frac{\kappa_{\pi}}{Rearcy Near (1.7)} $ Note (1.1) hits diplomy theory theorem (2.1) dash briefs	100.0	92.5	90.7	Suzuki- Miyaura cross- coupling
4286679	$\downarrow > \downarrow >$	() 1/20 0 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	~~~~~ KO ~~~~~ K	() K) K) K) K	50.0	102.9	97.7	Suzuki coupling
2952248	$\underset{i \in \mathcal{I}}{\overset{i}{\longrightarrow}} - \underset{i \in \mathcal{I}}{\overset{i \in \mathcal{I}}{\longrightarrow}} \xrightarrow{i \in \mathcal{I}} - \underset{i \in \mathcal{I}}{\overset{i \in \mathcal{I}}{\longrightarrow}} \xrightarrow{i \in \mathcal{I}} - \underset{i \in \mathcal{I}}{\overset{i \in \mathcal{I}}{\longrightarrow}} \xrightarrow{i \in \mathcal{I}} \xrightarrow$		0 √0 M+2 ↓ H0 ↓ 0 Kt		80.0	50.3	58.1	Suzuki- Miyaura cross- coupling reaction
9645904	$\mathbb{Y} \leftarrow \mathbb{Y} \longrightarrow \mathbb{Y}$	" " " " " " " " " " " " " " " " " " "	Recy New solim hydroxie		130.0	60.2	78.5	Suzuki reaction
1115219	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		ni~on K		110.0	134.9	111.2	Buchwald amidation
2356733	$\circ \mathcal{G}_{\mathcal{G}}^{\mathcal{G}}(\mathcal{A}) \to \mathcal{G}_{\mathcal{G}}^{\mathcal{G}}(\mathcal{A})$	\mathcal{L}_{0} NH2 \longrightarrow $+ \mathcal{X}$ \mathcal{X} NH4	\square		100.0	73.7	100.1	Buchwald- Hartwig coupling
9293801	() () () () () () () () () (~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~		}• ~ ~ (){€}-() × ~ ~	100.0	126.4	95.9	Buchwald- Hartwig coupling
2589752	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	он ноок ок к* Кекку ID 11400916	$\forall \sim \forall \leftarrow$	$\downarrow \sim \downarrow \prec$	90.0	92.9	92.9	Buchwald reaction
2593601	$\begin{array}{cccccccccccccccccccccccccccccccccccc$		}• ≈ ~~~ ~ ~ -{}}-0	}- ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	80.0	67.1	67.1	Bachwald- Hartwig reaction
3608758	\sim \downarrow	O and for the second	c√a	*	NA	-3.3	14,7	Grubbs Olefin Metathesis

Comparison with the neighbor approach

The nearest-neighbor method searches the training set for the reaction with the maximum similarity to the reaction for prediction. Cosine similarity of the reaction fingerprint¹ is used to quantify reaction similarity. A thorough search is performed for the 62 reactions in Supplementary Table 3. The top-ten most similar reactions are retrieved, and the top-one prediction and the prediction within top-ten that has the maximum elements matching the true condition are listed in Supplementary Table 6. Good suggestions are found for a majority of the cases, and the overall accuracy for these 62 reactions is comparable to the neural network model. However, the search for one reaction takes ~40 minutes on a single intel® Xeon(R) CPU E5-2690 0 @2.90GHz, as compared to ~100ms for the neural network approach running on the same machine, which is over 10,000 times faster. Further, the nearest neighbor method proposes conditions by simply copy and paste, which does not have the ability to infer missing information.

Supplementary Table 6. Nearest neighbor predictions for the 62 reaction examples in Supplementary

Reaxys	kratin	True Context	Top Prediction	Closest Prediction	True Temperature °C	Top Predicted Temperature	Closest Predicted
5301921	$ \begin{array}{c} & & \\ & & $		—он а,уа —о. №+	040 Na+	20.0	20.0	12.0
8712792	$\neg \neg (-\bigcirc -\bigcirc \longrightarrow -\bigcirc -\bigcirc$	∽_он к+ но.	°°+°° K+	∽_он к+ но.	NA	N/A	80.0
5261303	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	лан ная-тыр нар	~он ~~ ₁₆₂	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	-D	IVA	NA
8655717	fool → fool	() — Он Кан Но-	↓ U+ H0-	С — оч ноо Reasy Nine likim hydruside moodhytene	NA	20.0	20.0
5302180	$\circ \stackrel{\circ}{\longleftarrow} \uparrow \circ \rightarrow \circ \stackrel{\circ}{\longleftarrow} \uparrow \circ$				57.5	N/A	NA
53162	$\xrightarrow{-\infty} \mathscr{Y}_{-\infty}^{c} \longrightarrow \mathscr{Y}_{-\infty}^{c}$	~o~	α∼α	α∼α	0.0	0.0	0.0
5185761	- " " f f h a a f a f a f a f a f a f a f a f		art a	-La	10.0	N/A	NA
8698819	$\overset{\circ}{\longrightarrow}\overset{\circ}{\to}{\to}\overset{\circ}$		°c+o₁	$\downarrow^{\succ}_{\downarrow^{\circ}} \uparrow$	95.0	NA	NA
8669084	$\begin{cases} -\infty & \swarrow^{(n+1)} + \swarrow^{(n+1)} \\ \searrow^{(n+1)} + \swarrow^{(n+1)} & \longrightarrow^{(n+1)} + \swarrow^{(n+1)} + \swarrow^{(n+1)} \\ \longrightarrow^{(n+1)} + \swarrow^{(n+1)} + (1+1) + ($	1 326 * 7			N/A	NA	N/A
5336683	$\mathcal{O}^{1}\mathcal{O}, \mathcal{H}^{-1}\mathcal{O}_{\mathcal{O}}\mathcal{G}^{*} \longrightarrow \mathcal{H}^{-1}\mathcal{O}_{\mathcal{O}}\mathcal{G}^{*}$	Y~	₩	Y~0	220.0	30.0	380.0
5220300	$\dot{\mathcal{F}}^{0} \rightarrow \dot{\mathcal{F}}^{0}$	D ~~ I	D ~~ 1	j> ∽∘∽ ĭ	78.0	78.0	-78.0
2820838	$\sim \downarrow \Box \rightarrow \Box \Box \sim$	HO OFFON	NO EN CON	NO EN CON	54.5	95.0	95.0
8568283	$\sim \sim $	Y~ 13	∞~∘ √≺	°~° _∕∕	NA	90.0	90.0
8738792	0 0 9 9 - 8 0 - 0	_=xv	-== x ° √ ° ↔	—Ξ« ⁶ ζ ^ο α·	20.0	35.0	35.0
8574982	$\rightarrow 0$ of $\phi \phi \rightarrow \gamma \phi \phi$	∽он кт но.	∽ _{0К} (4 но	∽ок €+ НО.	40.0	40.0	40.0
8591626	\sim	—ОН №+ НО- НО- ОН		—он као Na+ но- но-он	10.0	0.0	0.0
8647861		() Ул () но но-сн	б Харанан но но-он	() ус. () но но-он	0.0	0.0	0.0
8655813	$\lim_{n\to\infty} \mathcal{F}_{\mathcal{F}_{n}}^{(n)} \xrightarrow{\mu_{\mathcal{F}_{n}}} \longrightarrow \mathcal{F}_{\mathcal{F}_{n}}^{(n)} \xrightarrow{\mu_{n}}$	OH 143	OH	04 M3	20.0	NA	20.0
8720054	$\xrightarrow{++}$	a~a ^{NO-0} −	o~a →	o∼a ^{bo-o}	20.0	220.0	20.0
8560867	$\gamma \circ \circ$	$O \downarrow \chi *$	Reaxys Name rodium hydride	$O \downarrow \chi *$	0.0	NA	0.0
8598928	$\neg (\uparrow \uparrow \uparrow \downarrow \uparrow \rightarrow \neg (\uparrow \uparrow $	a contraction of the second se	°~° "→→→⊂∑	a∼a _{NO-} ∂ NO-∂	0.0	NA	NA
5229646	$\overset{\sim}{} \overset{\sim}{} \overset{\sim}$	a~a (C)	0 G= (41=1+236 N3+	a~a Hqfo	0.0	110.0	N/A
10069687	$ \begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	Control of the second s	a~a wo-o	$\stackrel{\scriptstyle \succ}{ \stackrel{\scriptstyle \sim}{ \stackrel{\scriptstyle \sim}{ \stackrel{\scriptstyle \sim}}}} \stackrel{\scriptstyle \circ}{ \stackrel{\scriptstyle \sim}{ \stackrel{\scriptstyle \sim}}} \stackrel{\scriptstyle \circ}{ \stackrel{\scriptstyle \sim}{ \stackrel{\scriptstyle \sim}}}$	25.0	-10.0	N/A

() ----- + Reaxys Name $\mathsf{with} = \left\{ \begin{array}{c} \mathcal{A}_{\mathrm{orb}} \\ \mathcal{A}_{\mathrm{or$ CPH CPH Reaxys Name bis(acetylacetonine)suovanadium NA NA NA -=N -=N C -=n → N C Reaxys Name lithium bromide $(558712) \qquad \neg \mathcal{L}_{\ll} w^{26} \qquad \bigcirc = \circ \qquad \longrightarrow \qquad \neg \mathcal{L}_{\checkmark} \bigcirc$ \bigcirc - \bigcirc - $(1200) \longrightarrow (10^{-1}) \longrightarrow (10^{-1$ Reaxys Name sodium hydride -=N -=N __N_____Reaxys Name lithium chloride Reaxys Name sodium hydride П на но Reaxys Name sodium hydride \bigcirc \bigcirc $\stackrel{\circ}{\searrow}$ $\overset{\circ}{\times}$ ** a∼a α∕α a∼a **~** \bigcirc -∫∫ a√a 20.0 20.0 20.0 20.0 N/A 20.0 NH2- Na+ NH2 Na+ 20.0 N/A 20.0 \sum^{NH} $= \sum_{i=1}^{n} \sum_{j=1}^{n} \frac{1}{n_{ij}} \sum_{j=1}^{n} \frac{1}{n_{ij}}} \sum_{j=1}^{n} \frac{1}{n_{ij}} \sum_{$ zn... [) 🛴 zn··· 💭 🙏 alfo - alfo CI OF F Cl → Cl → F F α∕α → ^k ^k □ ∽ □ → HO F F CI CI HO F d∕d No k $\operatorname{row} \xrightarrow{\circ} \operatorname{form} \operatorname{form}$ С на √__О на С на $\mathcal{F}_{\mathcal{F}}^{\mathcal{F}} \longrightarrow \mathcal{F}_{\mathcal{F}}^{\mathcal{F}} \mathcal{F}_{\mathcal{F}}^{\mathcal{F}}$ Altr. Li+ Altr. Li+ Altr- Li+ $\xrightarrow{}_{++-\circ} \xrightarrow{}_{\times} \longrightarrow \xrightarrow{}_{\times} \xrightarrow{}_{\times}$ \$0.0 **[**150.0 **[**25.0 $\operatorname{a}_{\operatorname{constraint}} \xrightarrow{\operatorname{a}} \xrightarrow{\operatorname{a}} \operatorname{a}_{\operatorname{constraint}} \operatorname{constraint} \operatorname{a}_{\operatorname{constraint}} \operatorname{a}_{\operatorname{constraint}} \operatorname{constraint} \operatorname{a}_{\operatorname{constraint}} \operatorname{constraint}} \operatorname{constraint} \operatorname{constraint} \operatorname{constraint}} \operatorname{constraint} \operatorname{constraint} \operatorname{constraint} \operatorname{constraint}} \operatorname{constraint}$ Reaxys Name palladium-barham carbonate Reaxys Name xyl AH7- U+ 5.0 5.0 ∽₀∽ ^{Ci}sahi^C Ci AH7· Li+ CH ϕ^{+0} μ^{-0} μ^{-1} Reaxys Name palladium on activated charcoa ∼₀∼ ^{Cl}, AH3 AH7. Li+ $\bigcirc \neg \neg \bigcirc \checkmark \rightarrow \bigcirc \neg \neg \bigcirc \checkmark$ a-a Josto and the ада нго урада но-он HO-OH HO-OH Reaxys Name chloroform-dl -25.0 N/A N/A $|22113| \xrightarrow{\circ} (f_{1}^{*}) \xrightarrow{\circ} (f_{1}^{$ Reaxys Name jones reagent 20.0 20.0 20.0 Reaxys Name jones reagent Reaxys Name jones reagent HONG IN OF A ~_____ № _____ №№ но-сн о с но−он №+ с но

5081868	${\sim}{\leftarrow}{\sim}{\sim}{\sim}{\sim}{\rightarrow}{$	у ⁴ 20 —он но-он	огде ^ю нин — он но-он	огде ^о №+ —он но—он	20.0	90.0	0.0
339468	$\overset{\sim}{=}\overset{\sim}{\to}\overset{\to}$	Reaxys Name Jone's reagent Rearys Name silica gel	Reaxys Nume Jone's reagent	Reaxys Name Jone's reagent	0.0	N/A	N/A
4977260	$\bigcirc^{**} \bigtriangledown^{=}_{-} \longrightarrow \bigcirc^{-}_{-} \bigtriangledown$			C → + K+ Reaxys ID 21555523	80.0	N/A	90.0
3403911	$\circ, \mathcal{A}^{\mathcal{L}}_{\mathcal{L}} \circ \circ \to \circ \circ \mathcal{A}^{\mathcal{L}}_{\mathcal{L}} \circ$				95.0	105.0	105.0
1126041.	$M (\mathcal{O} ' \mathcal{H} \mathcal{H} \mathcal{H} \mathcal{O} \mathcal{O}) \longrightarrow \mathcal{O} (\mathcal{O} \mathcal{O} \mathcal{O})$		→ NI (C) 0 ² / ₂ × <		290.0	100.0	585.0
1107702	$\bigcirc \bigcirc \bigcirc \frown \frown$	C C C C C C C C C C C C C C C C C C C	Contraction discrite Reasys	C C C C C C C C C C C C C C C C C C C	120.0	120.0	120.0
2604617	$ (\mathcal{G}_{\mathcal{G}}^{i}) \to (\mathcal{G}_{\mathcal{G}}^{i}$	C ¹ ₀ ⁰ C ² C∗ tioente	Construction of the second sec	Control Contro	30.0	390.0	30 .0
3382323	$\frac{1}{2} - 0^{+} \stackrel{\text{\tiny (D-O)}}{\rightarrow} \text{\tiny$	° 14+2 √° ¢¢° C++	C 18+2 C C+	↓ NH+2 ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓ ↓	80.0	80.0	80.0
9603192	$\uparrow^{\sim} \sim \downarrow^{\sim} \bigcirc \rightarrow \downarrow^{\sim} \downarrow^{\sim} \bigcirc \bigcirc$				19.0	19.0	19.0
3603226	$3^{+}_{+}^{+}_{0} \rightarrow 3^{+}_{-}^{+}_{-}^{0}_{-$				540.0	40.0	40.0
2584073	$ = \int_{-\infty}^{\infty} \int_{-\infty}$	Nume latencies	Num lifting should Kary	Varie literatoria	100.0	100.0	100.0
4033276	$ { \hspace{1cm} = \hspace{1cm} } { \end{array} } { \hspace{1cm} } { \end{array} } { \hspace{1cm} } { \end{array} { \end{array} } { \end{array} } { \end{array} } { \end{array} } { \end{array} { } { \end{array} } { } { \end{array} } { } $	$-\equiv u \bigcup_{0}^{d_{i}} \int_{0}^{0} \varepsilon_{*}$ Reavy Name pollation discette	-= v ~~ ·	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	8.0	120.0	120.0
9208506	$\xrightarrow{-n} \xrightarrow{-n} -n$	~~~~ ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	°+' 10	N/A	NA	NA
2888571	$\overset{n}{\rightarrow} - \overset{n}{\rightarrow} $	۲۰۰۰ ۲۰۰۰ ۲۰۰۰ ۲۰۰۰ ۲۰۰۰ ۲۰۰۰ ۲۰۰۰ ۲۰۰		مريح د K+ dishlorde د بريم د	NA	NA	NA
1004052	$ \rho^{i} \varphi \ \rho_{i} \longrightarrow \rho \phi \varphi $	C C C C C C C C C C C C C C C C C C C	°-4° к.	∫ 0 ³ + 0 ⁰ K+ Reaxys ID 23/64488	20.0	20.0	50.0

Evaluation of the Michael additions used by Marcou et al.²

Among the 52 reactions, 34 of them are found in the final dataset, and we have detailed information about these reactions for evaluation. The model is retrained with these reactions excluded from the training set. The top-ten accuracy of these 34 reactions for similar matches is 47.0%, and the accuracy for similarly matching catalyst, solvent 1 and reagent 1 is 55.9%, lower than the overall accuracy of the entire dataset, yet a significant improvement over literature results.² Furthermore, it has been pointed out that many of these reactions can occur under different conditions, meaning some predictions are not necessarily wrong, even when they differ from the recorded context. For example, Supplementary Figure 5 shows reactions where the exact recorded condition is not predicted. The first example has a different solvent predicted, but it is similar to the recorded solvent (both are alcohols).³ For the second example, although the model does not suggest the correct solvent and reagent in the first choice, it recognizes the need for basic conditions in the subsequent suggestions, with the second suggestion being piperidine and the third being triethylamine.⁴ Supplementary Figure 5(C) is another example of data quality that complicates the analysis. Two reagents NaHCO₃ and HCl are not commonly used in the same reactions but in separate steps which is the case in the reported procedure.⁵ Additionally, the true solvent used for the reaction is ethanol where diethyl ether is used as an extraction solvent in workup. The full prediction results are presented in Supplementary Table 7.



Supplementary Figure 5. Examples of Michael additions where none of the recorded c, s1 or r1 are predicted. A) Methanol is predicted, which is similar to the recorded solvent (isopropanol); B) the top prediction is incorrect, but subsequent predictions suggest bases as reagents (the second suggestion is piperidine and the third is triethylamine); C) the reaction has incompatible reagents that are used in different stages but not reflected in the reaction record.

Supplementary Table 7. 34 examples of Michael additions that are used for evaluation in the work of

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Reaxys ID	Reaction	True Context	Top Prediction	Closest Prediction	True Temperature °C	Top Predicted Temperature/ °C	Closest Predicted Temperature/ °C
1493719	$f_{\mathcal{O}} \land f^{\wedge} \longrightarrow \mathcal{O} \sim f_{\mathcal{O}}$	<u>~</u>			20.0	23.7	23.7
635835	~ 104 $er \rightarrow r \rightarrow$	\bigcirc	\bigcirc	\bigcirc	N/A	34.6	52.9
1288921		—он	—он	—он	20.0	29.9	29.9
24352854	$ \mathbb{C}_{\mathcal{A}} = \mathbb{C}_{\mathcal{A}} \xrightarrow{\mathcal{A}}_{\mathcal{A}} \xrightarrow{\mathcal{A}} \xrightarrow{\mathcal{A}}_{\mathcal{A}} \xrightarrow{\mathcal{A}} \xrightarrow{\mathcal{A}}_{\mathcal{A}} \xrightarrow{\mathcal{A}} \xrightarrow{\mathcal{A}}_{\mathcal{A}} \xrightarrow{\mathcal{A}} $	₩ ¹ ~0 H20 ² y ⁰ K+	°- с К+ С Reaxys Name potassium iodide	``\ ^ ~````````````````````````````````````	N/A	67.4	57.3
8861928	$\bigcirc = \stackrel{i}{\longrightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} \stackrel{i}{\rightarrow} $	_= _N ⁻⁰ ↓ ⁰ κ+	—он	-=N °↓ Na+	N/A	18.6	29.6
23748241	$\sim \sim $	—он	— он	—он	10.0	26.3	26.3
25669226	$+ ^{\flat} \odot - \bigcirc ~ \sim ^{\flat} \sim \rightarrow + ^{\flat} \odot - \bigcirc ~ \sim ^{\circ} \sim$	Сон	∽он	∽он	N/A	57.2	57.2
23227051	$\sim \mathcal{L} \mathcal{L} \rightarrow + \mathcal{L} \mathcal{L} \rightarrow \mathcal{L}$	~о~ ОН ОН Na+ HCI	∽он	∽он	N/A	30.7	30.7
24273953	$\overset{\text{HZN}}{\longrightarrow} \overset{\circ}{\longrightarrow} \overset{\circ}{\to} \overset{\circ}{\to}$	∽он	∽он	∽он	N/A	36.3	36.3
24717084	$\sim 0^{**} \sim 1^{*} \longrightarrow \sim 0^{*} \sqrt{1^{*}}$	∽он	∽он	∽он	N/A	34.9	34.9
24971546	\sim	тон ул	си	~сн	N/A	46.9	50.0
694909	$\longrightarrow \qquad \qquad$	Лон Ли	Nt D HO-	—он »т О но-	30.0	35.6	26.5
46547	$ \circ \mathcal{A}_{\mathrm{hol}} \sim \circ \sim \circ \sim \circ \circ$	∕~он ∕~о. №+		∽oHN	N/A	60.8	60.8
1456140	$\bigcirc NH NS / \longrightarrow NS / $				20.0	37.4	37.4
24879120	$\times \mathbb{I}^{\mathrm{pr}} \xrightarrow{\sim} \mathbb{I}_{\mathrm{s}} \xrightarrow{\sim} \mathbb{I}^{\mathrm{s}} \xrightarrow{\sim} \mathbb{I}^{\mathrm{s}}$	\sim_{\circ}	∽он	Сн	N/A	79.5	79.5
2924844	$_{HO}^{O}+_{HO} \longrightarrow _{HO}^{O}+_{HO}^{HH}$	H2O Na+ HO-		H2O Na+ HO-	0.0	51.9	16.7
22879782	${}^{\mathrm{H}}_{\mathrm{C}} \rightarrow {}^{\mathrm{OH}} \xrightarrow{\times} {}^{\mathrm{OH}}_{\mathrm{C}} \xrightarrow{\mathbb{C}} \xrightarrow{\mathbb{C}} {}^{\mathrm{OH}}_{\mathrm{C}} \xrightarrow{\mathbb{C}} \xrightarrow{\mathbb{C}} {}^{\mathrm{OH}}_{\mathrm{C}} \xrightarrow{\mathbb{C}} \xrightarrow{\mathbb{C}} {}^{\mathrm{OH}}_{\mathrm{C}} \xrightarrow{\mathbb{C}} \xrightarrow{\mathbb{C}} \xrightarrow{\mathbb{C}} {}^{\mathrm{OH}}_{\mathrm{C}} \xrightarrow{\mathbb{C}} \mathbb$	=N	-=N ⁰ ↓ ⁰ K+	— <u>—</u> »	N/A	43.4	48.4
24889073	$\overset{``}{\longrightarrow} \overset{``}{\longrightarrow} \overset{`'}{\longrightarrow} \overset{''}{\longrightarrow} \overset{'''}{\longrightarrow} \overset{''}{\longrightarrow} $	a~a			N/A	43.7	43.7
1284594	$\overset{\text{Hom}}{\longrightarrow} \overset{\text{op}}{\longrightarrow} \text$	р н₂о ∕∽ н₂о	Br Br.	Br J St Br.	20.0	14.3	14.3
24576080	$\sim \times \times \times \circ \downarrow_{\mathrm{ret}} _{\mathrm{ret}} \sim \circ \downarrow_{\mathrm{ret}} \longrightarrow + \circ \downarrow_{\mathrm{ret}} _{\mathrm{ret}} $	Reaxys Name ammonium chloride	\square	\square	N/A	3.6	3.6
3363418	$\downarrow_{q^{\circ-}} \longrightarrow \uparrow_{s_{s}}^{l}$	— <u>—</u> »		<u>=</u> n	N/A	61.1	29.8
24576110	$\lambda_{\eta^{\circ}} \longrightarrow \gamma^{1} \gamma^{1}$	····· ······			N/A	73.3	73.3
701198	$\overset{\circ}{\longrightarrow} \overset{\circ}{\longrightarrow} \overset{\circ}{\to} \overset{\circ}$	H20 Na+ -0-B 0-B-0 0-B-0		H20 Na+ -0	20.0	47.0	27.6
24581594	\ll_{HCSO} \sim_{CSO} \longrightarrow \sim_{CSO}	-0- Na+ 0=	— ОН — О- Na+	—0- Na+	N/A	19.6	36.2
649317	$\overset{\circ}{\rightarrow}\overset{\circ}{\downarrow}^{s_{s_{s_{s_{s_{s_{s_{s_{s_{s_{s_{s_{s_{$	Ret .	\sum	<u> </u>	20.0	25.9	25.9
4051	$\bigcirc \uparrow \uparrow \\ \frown \rightarrow \uparrow \\ \frown \bigcirc$	arta ava			20.0	28.6	28.6
1297143		~ он	—он	—он	10.0	24.0	24.0

4046	$\bigcirc H \xrightarrow{h \leq h} \longrightarrow h \leq h > h > h > h > h > h > h > h > h >$				90.0	33.8	33.8
3598744	$128_{\text{OH}} \xrightarrow{\text{N}}_{\text{H}} \xrightarrow{\text{N}}_{\text{H}} \xrightarrow{\text{H}}_{\text{H}} \xrightarrow{\text{H}}_{\text{H}}$				50.0	57.8	57.8
2472218	$\mathcal{I}_{0} \mathcal{I}_{0} \mathcal$	\square	∽он	\square	N/A	32.4	26.5
2450353:	$\bigcirc ````````````````````````````````````$	~~~	H2O	H20	N/A	62.4	62.4
2356774	$\operatorname{M}_{\mathcal{M}} (\mathcal{M}_{\mathcal{M}} \to \mathcal{M}_{\mathcal{M}}) \to \mathcal{M}_{\mathcal{M}} (\mathcal{M}_{\mathcal{M}})$	H20 No Br	—он	H2O Na+ HO-	80.0	38.3	41.3
2496066:	$ \longrightarrow \longrightarrow$	$\supset \sim$		$\supset \sim$	N/A	66.9	44.4
2366507	$\chi_{o} h_{\omega} \chi_{o} h_{\mu\nu} \longrightarrow \chi_{o} h_{\mu\nu} h_{\mu} \ell_{o}$	CH	—он	—он	60.0	28.4	28.4

Feature definition used in the Morgan fingerprints

Invariants	SMARTS
Hydrogen bond donor	[[N;!H0;v3],[N;!H0;+1;v4],[O,S;H1;+0],[n;H1;+0]]
Hydrogen bond	[\$([O,S;H1;v2]-[!\$(*4[O,N,P,S])]),[O,S;H0;v2],
acceptor	[O,S;-],\$([N&v3H1,H2]-[!\$(*4[O,N,P,S])]), [N;v3;H0],[n,o,s;+0],F]
Basic group	[([N;H2&+0][[C,a];!([C,a](4O))]),([N;H1&+0]([[C,a];!([C,a](4O))])[[C,a];!)]
	([C,a](4O))], ([N;H0&+0]([C;!(C(4O))])([C;!(C(4O))])[C;!(C(4O))]),
	[N,n;X2;+0]]
Hydrophobic group	[\$([C;H2,H1](!4*)[C;H2,H1][C;H2,H1][[C;H1,H2,H3];
	!\$(C4*)]),\$(C([C;H2,H3])([C;H2,H3])[C;H2,H3])]
Acidic group	[O;H1]-[C,S](4[O,S,P])
Halogen	[F,Cl,Br,I]
Attachment point to	[\$([A;D3](@*)(@*)~*)]
aliphatic ring	
Attachment point to	$[(a;D3](@^*)(@^*)^*)]$
aromatic ring	
Any unusual atom (not	[!#1;!#6;!#7;!#8;!#9;!#16;!#17;!#35;!#53]
H,C,N,O,F,S,Cl,Br,I)	

Supplementary Table 8. Feature definition as defined by Gobbi et al.⁶

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