

Supplementary materials:
**TiO₂-based hybrid nanocomposites modified by phosphonate
molecules as selective PAH adsorbents**

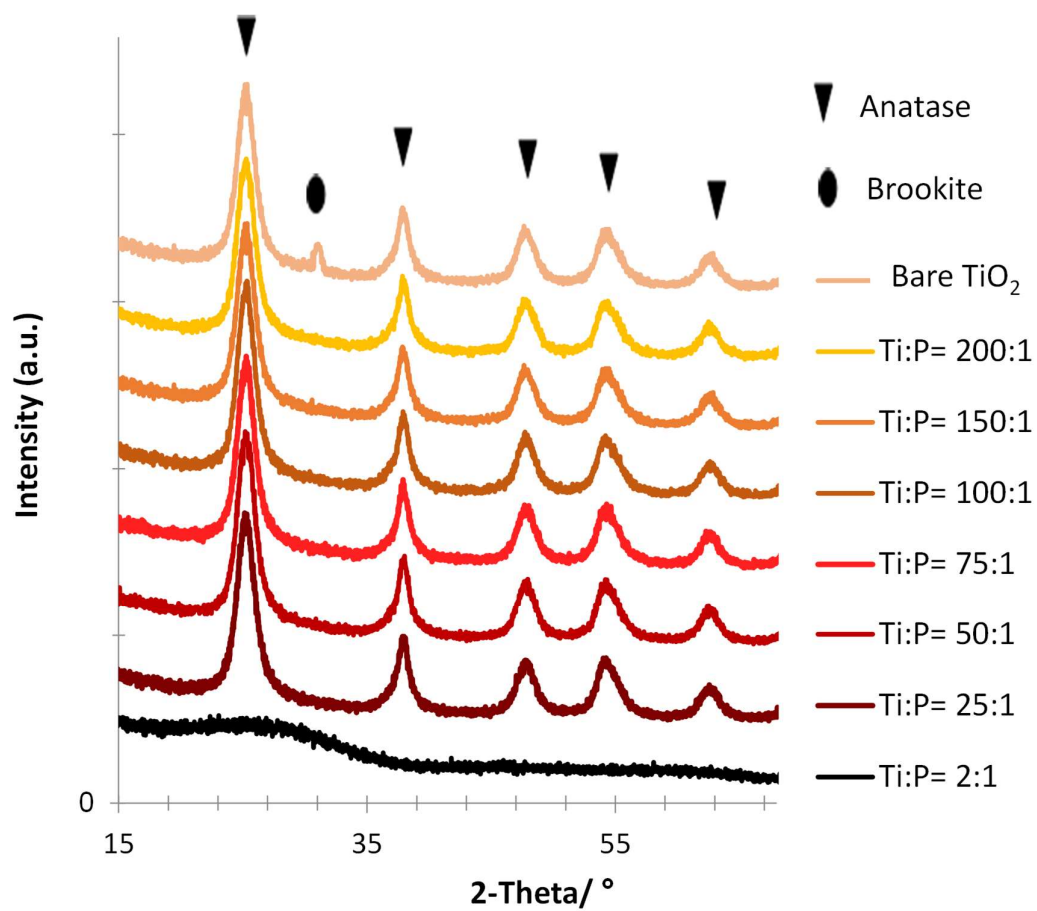


Figure S1: Powder XRD of samples (TiO₂)_x(VPA) (x=2-200) and bare TiO₂

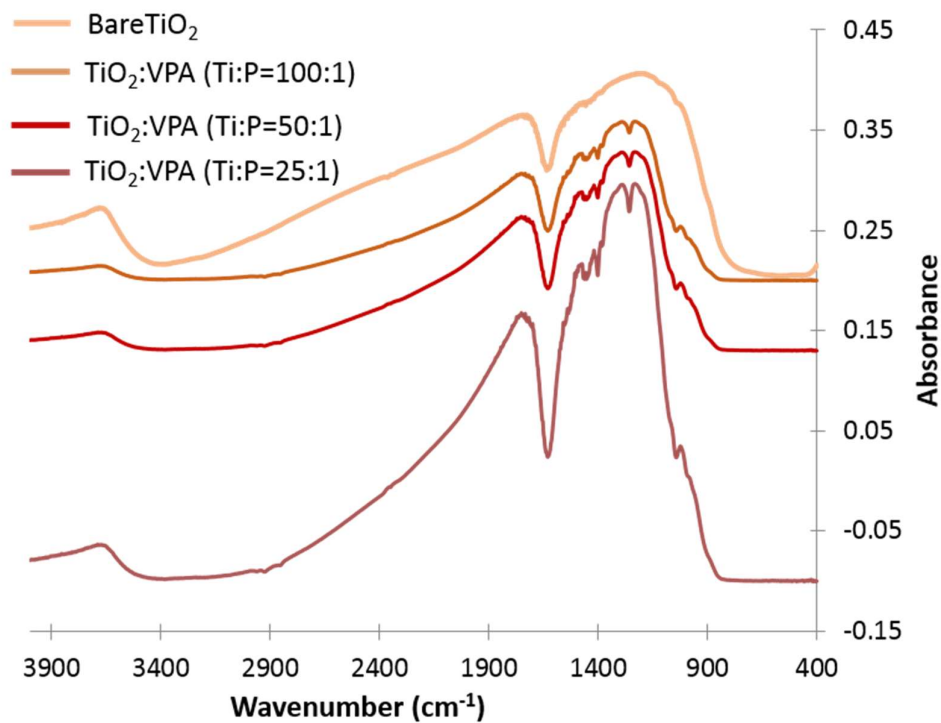


Figure S2: FT-IR spectra of samples (TiO₂)_x(VPA) (x=25, 50, 100) and bare TiO₂

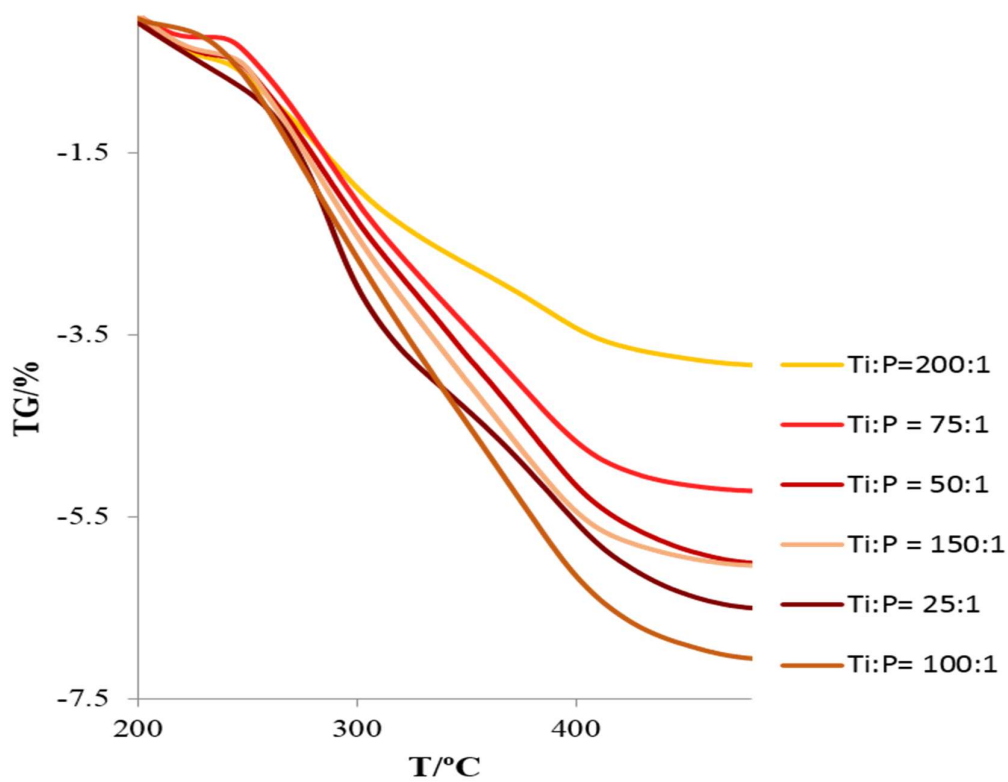


Figure S3: TGA of (TiO₂)_x(VPA) samples

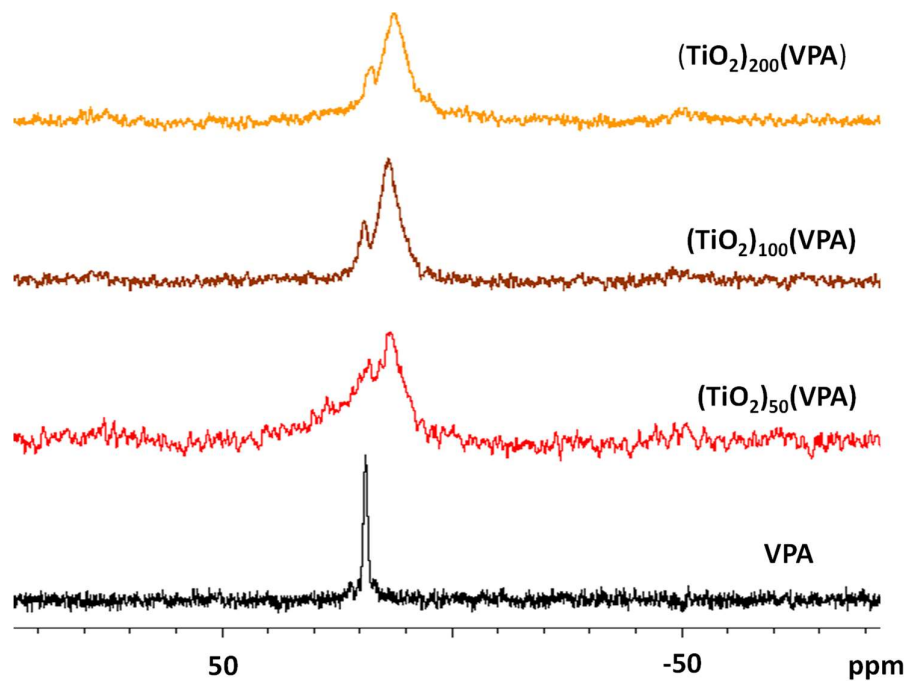


Figure S4: Solid state $^{31}\text{P}\{^1\text{H}\}$ MAS NMR spectra of VPA and hybrid samples $(\text{TiO}_2)_x(\text{VPA})$ with $x = 50, 100$ et 200

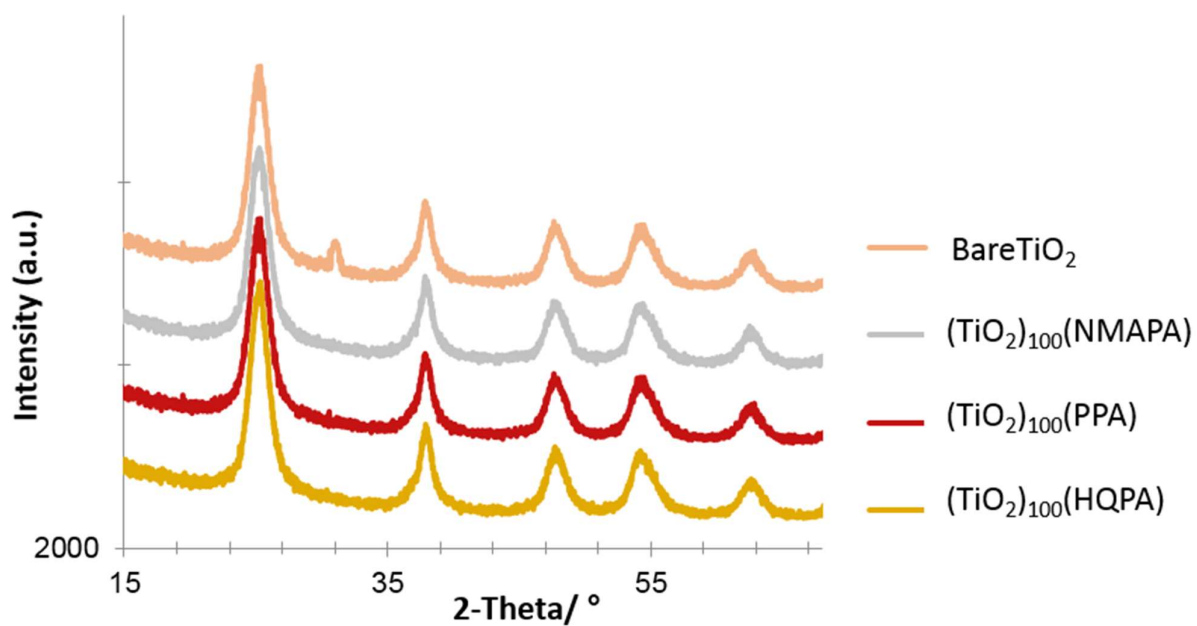


Figure S5: Powder XRD of samples $(\text{TiO}_2)_{100}(\text{PA})$ and bare TiO_2

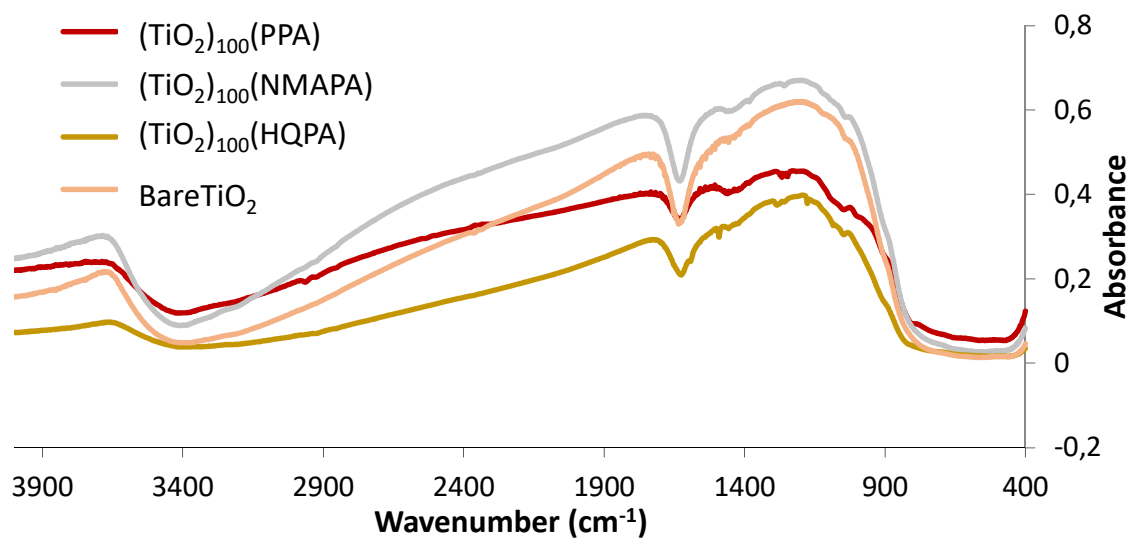


Figure S6 : FT-IR spectra of bare TiO_2 and hybrid samples $(\text{TiO}_2)_{100}(\text{PPA})$, $(\text{TiO}_2)_{100}(\text{NMAPA})$ and $(\text{TiO}_2)_{100}(\text{HQPA})$

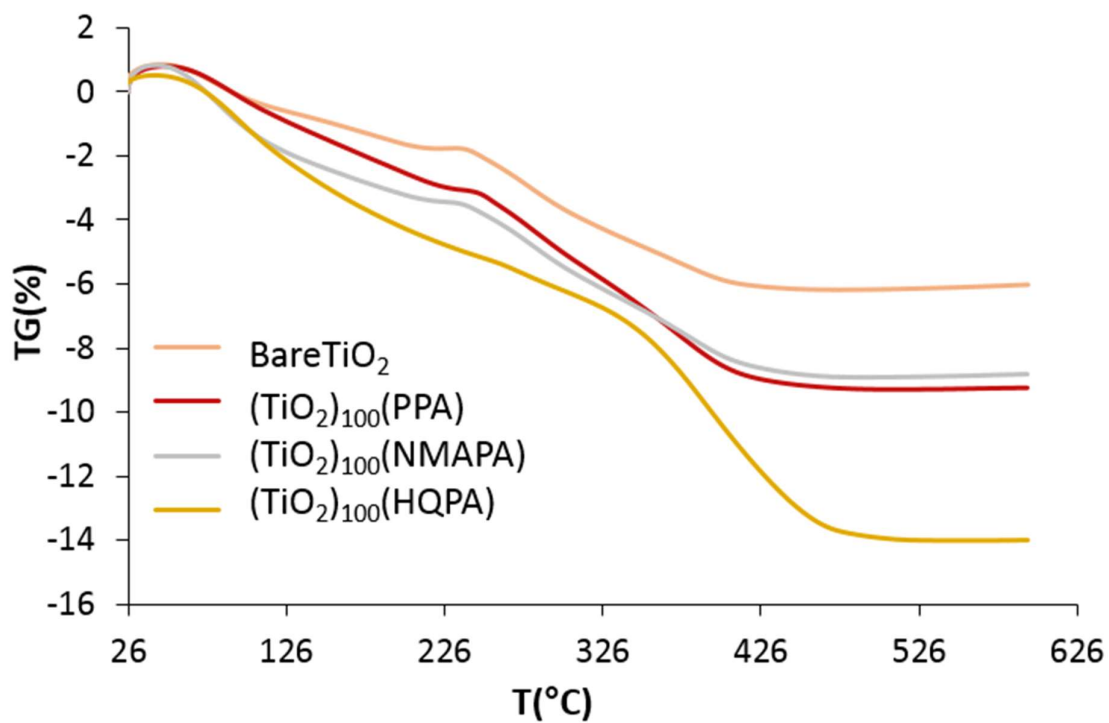


Figure S7: TGA patterns of bare TiO_2 and hybrid samples $(\text{TiO}_2)_{100}(\text{PPA})$, $(\text{TiO}_2)_{100}(\text{NMAPA})$ and $(\text{TiO}_2)_{100}(\text{HQPA})$

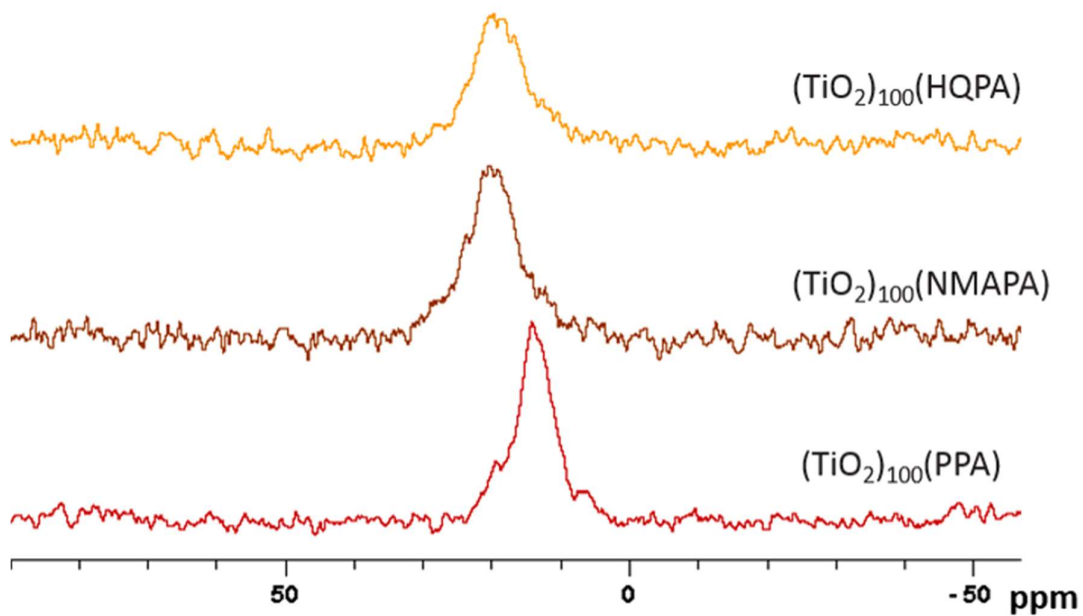


Figure S8: ^{31}P $\{^1\text{H}\}$ MAS NMR spectra of hybrid samples $(\text{TiO}_2)_{100}(\text{PPA})$, $(\text{TiO}_2)_{100}(\text{NMAPA})$ and $(\text{TiO}_2)_{100}(\text{HQPA})$

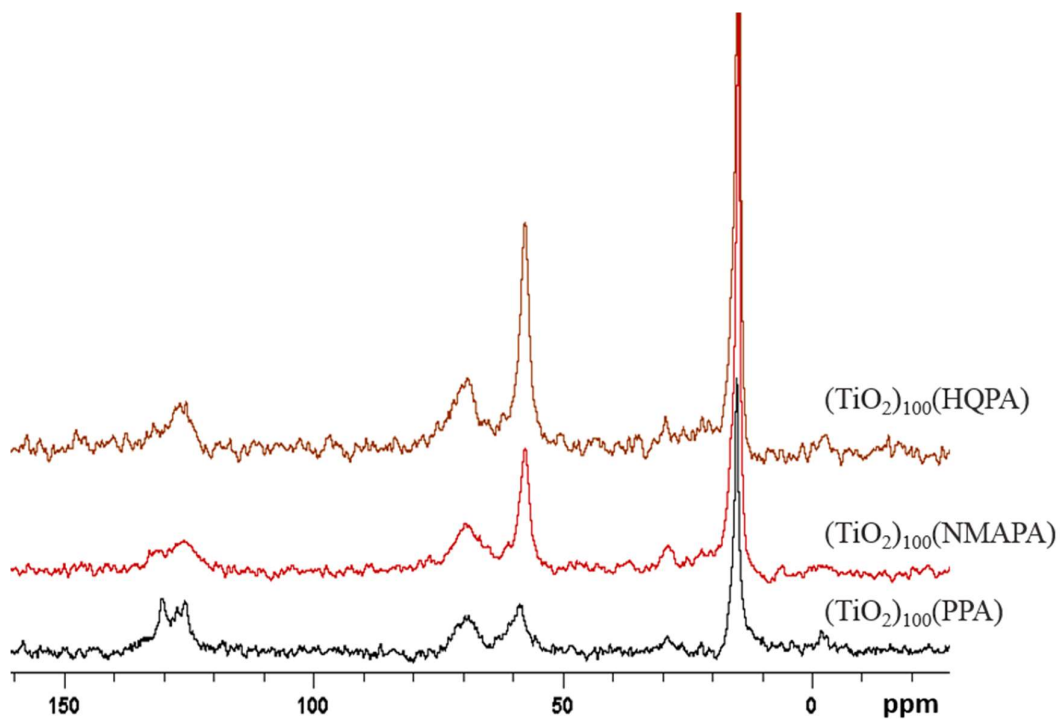


Figure S9: ^{13}C $\{^1\text{H}\}$ MAS NMR spectra of hybrid samples $(\text{TiO}_2)_{100}(\text{PPA})$, $(\text{TiO}_2)_{100}(\text{NMAPA})$ and $(\text{TiO}_2)_{100}(\text{HQPA})$

Table S1: Abbreviations and structures of the EPA 16 PAHs




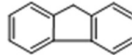
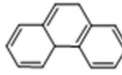
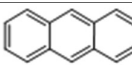
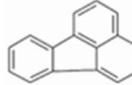
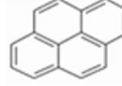
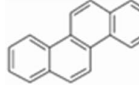
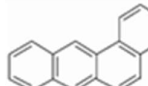
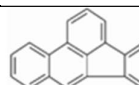
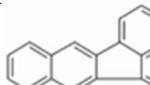
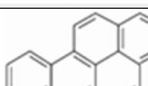
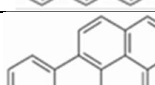
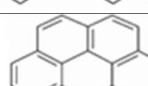
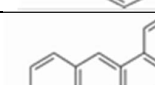
PAH	Abbreviation	Structure
Naphtalene	NAP	
Acenaphthylene	ACY	
Acenaphthene	ACP	
Fluorene	FLR	
Phenanthrene	PHE	
Anthracene	ANT	
Fluoranthene	FLT	
Pyrene	PYR	
Chrysene	CHR	
Benzo[a]Anthracene	BaA	
Benzo[b]fluoranthene	BbF	
Benzo[k]fluoranthene	BkF	
Benzo[a]pyrene	BaP	
Indeno[1.2.3-cd] pyrene	IcP	
Dibenz[a,h]anthracene	DbA	
Benzo[ghi]perylene	BgP	

Table S2: BET surface and porous volume of bare TiO₂ and hybrid (TiO₂)₁₀₀(O₃P-R) nanomaterials

Material	Porous volume (cm³ / g)	BET surface area (m² / g)
TiO ₂	0.49	250
(TiO ₂) ₁₀₀ (VPA)	0.39	269
(TiO ₂) ₁₀₀ (PPA)	0.50	285
(TiO ₂) ₁₀₀ (NMAPA)	0.51	225
(TiO ₂) ₁₀₀ (HQPA)	0.50	322

Table S3: Summary of kinetic parameters adsorption of (TiO₂)₁₀₀(VPA) toward Benzo(a)pyrene

Model	Parameters
Pseudo-first-order	
k_1 (min ⁻¹)	0.00207
q_e (mg g ⁻¹)	0.00855
R^2	0.1403
Pseudo-second-order	
k_2 (g mg ⁻¹ min ⁻¹)	1.383
q_e (mg g ⁻¹)	0.850
R^2	1.000