www.sciencemag.org/content/361/6398/171/suppl/DC1



Supplementary Materials for

Deconstructive fluorination of cyclic amines by carbon-carbon cleavage

Jose B. Roque,* Yusuke Kuroda,* Lucas T. Göttemann, Richmond Sarpong*

*These authors contributed equally to this work. †Corresponding author. Email: rsarpong@berkeley.edu

Published 13 July 2018, *Science* **361**, 171 (2018) DOI: 10.1126/science.aat6365

This PDF file includes:

Materials and Methods Figs. S1 to S3 NMR Spectra References

Table of Contens

1. General Considerations	
2. Experimental Procedures for Preparation of Starting Materials	
3. Experimental Procedures for the Silver-Mediated Fluorination	S12
4. Mechanistic Studies	
5. NMR Studies	
6. Electrochemical Measurement	S26
7. NMR Spectra	S28
8. References and Notes	

1. General Considerations

1.1. Solvents and Reagents

Tetrahydrofuran (THF) and triethylamine (Et₃N) were sparged with argon and dried by passing through alumina columns using argon in a Glass Contour solvent purification system. Dichloromethane (CH₂Cl₂) was freshly distilled over calcium hydride under a N₂ atmosphere prior to each use. N-Boc-piperidine (**1c**), N-methyl-2-pyrrolidinone (**1p**) and N-methyl-2-piperidinone (**1n**) were obtained from commercial vendors and used as received. Reagents for the fluorination reaction were purchased from commercial vendors as follows: Silver tetrafluoroborate (AgBF₄, 99%) was purchased from Oakwood Chemicals and stored in a glovebox. Selectfluor[®] was purchased from Matrix Scientific. Acetone (HPLC) was purchased from Fisher Scientific. Water (HPLC) was purchased from Fisher Scientific.

1.2. Experimental Procedures

Unless otherwise noted in the experimental procedures, reactions were carried out in flame or oven-dried glassware under a positive pressure of N₂ in anhydrous solvents using standard Schlenk techniques. Reaction temperatures above room temperature (22–23 °C) were controlled by an IKA® temperature modulator and monitored using liquid-in-glass thermometers. Reaction progress was monitored using a combination of LC/MS analysis (via a Shimadzu LCMS-2020 (UFLC) equipped with the LC-20AD solvent delivery system, a SPD-20AV prominence UV/Vis detector (SPD-M20A Photo Diode Array), and a Thermo Scientific Hypersil GOLD HPLC column (5 μ m particle size, 4.6 × 50 mm)), and thin-layer chromatography (TLC) on SiliCycle Siliaplates (glass backed, extra hard layer, 60 Å, 250 μ m thickness, F254 indicator). Flash column chromatography was performed with either glass columns using Silicycle silica gel (40–63 μ m particle size) or with a Yamazen Smart Flash EPCLC W-Prep 2XY (dual channel) automated flash chromatography system on prefilled, premium, universal columns using ACS grade solvents. Preparative thin layer chromatography was performed on SiliCycle Siliaplates (glass backed, extra hard layer, 60 Å, 250 μ m thickness, F254 indicator).

1.3. Analytical Instrumentation

¹H NMR and ¹³C NMR data were recorded on Bruker AVQ-400, AVB-400, RDX-500, AV-600 and AV-700 spectrometers using CDCl₃ as solvents, typically at 20–23 °C. Chemical shifts (δ) are reported in ppm relative to the residual solvent signal (δ 7.26 for ¹H NMR, δ 77.16 for ¹³C NMR in CDCl₃, δ 3.31 for ¹H NMR, δ 49.00 for ¹³C NMR in CD₃OD). The ¹⁹F NMR spectra were acquired on an AVQ-400 spectrometer and internally referenced to CFCl₃ (δ 0.00). Data for ¹H, ¹³C and ¹⁹F NMR spectroscopy are reported as follows; chemical shift (δ ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quint = quintet, hept = heptet, m = multiplet, br = broad), coupling constant (Hz), integration. Melting points were determined using a MEL-TEMPTM apparatus and are uncorrected. Optical rotations were measured on a Perkin-Elmer 241

polarimeter. High-resolution mass spectra (HRMS) were obtained from the Catalysis Facility of the Lawrence Berkeley National Laboratory (supported by the Director, Office of Science, of the US Department of Energy under contract no. DE-AC02-05CH11231) using a PerkinElmer AxION 2 TOF-MS.

2. Experimental Procedures for Preparation of Starting Materials

2.1. Preparation of N-Protected Cyclic Amines

Phenyl(piperidin-1-yl)methanone (1a) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (44).

Bz

1-(Piperidin-1-yl)ethan-1-one (1b) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (45).

2,2-Dimethyl-1-(piperidin-1-yl)propan-1-one (1d) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (*46*).

Azetidin-1-yl(phenyl)methanone (1e) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (44).

Phenyl(pyrrolidin-1-yl)methanone (1f) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (*44*).







Azepan-1-yl(phenyl)methanone (1g) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (47).



Azocan-1-yl(phenyl)methanone (1h): A 25 mL round-bottomed flask was charged with a solution of azocane (300 mg, 3.02 mmol) and Et₃N (0.57 mL, 4.1 mmol) in CH₂Cl₂ (5.0 mL) and cooled to 0 °C. Benzoyl chloride (0.320 mL, 2.75 mmol) was added dropwise over 5 min and the resulting mixture was warmed to room temperature. After 24 h, the reaction mixture was quenched with 1 M HCl aq. (5.0 mL) and the phases were separated. The aqueous phase was extracted with CH₂Cl₂ (10 mL \times 3). The combined organic layers were washed with brine (2.0 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude residue was purified by column chromatography (SiO₂, 50% EtOAc/hexanes) to provide the title compound (400 mg, 66%) as a yellow oil.

¹**H NMR** (400 MHz, CDCl₃): δ 7.41–7.34 (m, 5H), 3.62 (t, *J* = 6.1 Hz, 2H), 3.31 (br, 2H), 1.86 (br, 2H), 1.61–1.59 (m, 8H);

¹³C NMR (101 MHz, CDCl₃) δ 171.5, 137.7, 129.0, 128.5, 126.4, 51.2, 46.7, 27.0, 26.5, 26.4, 25.6, 24.2; HRMS (ESI): Calc'd for C₁₄H₂₀NO [M+H]⁺: 218.1539, found: 218.1535.



(4-Methylpiperidin-1-yl)(phenyl)methanone (1i) was prepared from 4-methylpiperidine using a procedure analogous to that for the preparation of 1h. Spectral data were in full agreement with the reported literature values (48).





data were in full agreement with the reported literature values (49).

(2-Ethylpiperidin-1-yl)(phenyl)methanone (1k) was prepared from 2-ethylpiperidine using a procedure analogous to that used for the synthesis of 1h. The title compound was obtained as a colorless oil (525 mg, 81%).

¹**H NMR** (600 MHz, CDCl₃, *ca.* 1:1 mixture of rotamers): δ 7.38–7.33 (m, 5H), 4.81 (br, 0.5H), 4.57 (br, 0.5H), 3.67 (br, 0.5H), 3.49 (br, 0.5H), 3.02 (br, 0.5H), 2.78 (br, 0.5H), 1.81–1.76 (m, 1H), 1.74–1.31 (m, 7H), 0.95 (br, 1.5H), 0.73 (br, 1.5H);

¹³C NMR (151 MHz, CDCl₃, mixture of rotamers): δ 170.9, 137.3, 129.1, 128.5, 126.6, 56.1, 49.8, 43.3, 37.0, 28.8, 28.0, 26.4, 25.9, 22.8, 19.2, 10.8 (*One* ¹³C signal is overlapping with others due to amide rotation);
HRMS (ESI): Calc'd for C₁₄H₂₀NO [M+H]⁺: 218.1538, found: 218.1540.



(3,5-Dimethylpiperidin-1-yl)(phenyl)methanone (11) was prepared from 3,5-dimethylpiperidine (mixture of *cis* and *trans*) using a procedure analogous to that for the preparation of 1h. Spectral data were in full agreement with the literature values (50).



(Octahydroquinolin-1(2*H*)-yl)(phenyl)methanone (1m) was prepared from *trans*-decahydroquinoline using a procedure analogous to that used for the synthesis of 1h. The title compound was obtained as a yellow oil (276 mg, 53%).

¹**H NMR** (400 MHz, CDCl₃): δ 7.36 (br, 5H), 3.50 (td, *J* = 10.7, 3.2 Hz, 1H), 2.43–2.28 (m, 2H), 2.29–2.24 (m, 1H), 1.79–1.52 (m, 7H), 1.49–1.16 (m, 4H), 1.13–1.03 (m, 1H);

¹³C NMR (101 MHz, CDCl₃): 171.6, 137.8, 129.3, 128.4, 126.9, 61.3, 42.4, 38.2, 33.2, 30.5, 26.7, 26.3, 25.6, 23.7;

HRMS (ESI): Calc'd for C₁₆H₂₂NO [M+H]⁺: 244.1696, found: 244.1697.

S7



Methyl (*S*)-1-Benzoylpiperidine-2-carboxylate (1n) was prepared from L-pipecolic acid methyl ester hydrochloride using a procedure analogous to that used for the preparation of 1h. Spectral data were in full agreement with the reported literature values (*51*).



Methyl Benzoyl-*L***-prolinate (10)** was prepared from L-proline methyl ester hydrochloride using a procedure analogous to that used for the synthesis of **1h**. Spectral data were in full agreement with the reported literature values (*52*).



1-Pivaloylpiperidin-2-one (1r) was prepared from 2-piperidinone and pivaloyl chloride using a procedure analogous to that used for the synthesis of **1h**. The title compound was obtained as a colorless oil (456 mg, 83%).

¹H NMR (500 MHz, CDCl₃): δ 3.50 (br s, 2H), 2.46 (t, J = 5.8 Hz, 2H), 1.85–1.84 (m, 4H), 1.28 (s, 9H); ¹³C NMR (126 MHz, CDCl₃): δ 190.0, 173.2, 47.2, 43.7, 34.0, 27.7, 22.7, 21.5; HRMS (ESI): Calc'd for C₁₀H₁₈NO₂ [M+H]⁺: 184.1332, found: 184.1333.



1-Benzoylpiperidine-3-carboxylic acid (1s) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (53).



1-Benzoylpiperidine-4-carboxylic acid (1t) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (54).

2.2. Preparation of Peptides

Representative Procedure for Methyl Ester Hydrolysis



A 100 mL round-bottom flask was charged with a solution of **1p** (1.24 g, 5.00 mmol) in 3:1 THF: H₂O (10 mL) and cooled to 0 °C. LiOH•H₂O (210 mg, 25.0 mmol) was added and the resulting mixture was warmed to room temperature. After 13 h, the reaction mixture was cooled to 0 °C and acidified with 1 M HCl aq. (10 mL) to pH <2. The solution was then diluted with EtOAc (10 mL) and the aqueous layer was extracted with EtOAc (10 mL × 3). The combined organic layers were washed with brine (5.0 mL), dried over MgSO₄, filtered and concentrated under reduced pressure to afford **S1**, which was used in the next step without further purification.

Representative Procedure for Condensation Reaction



Methyl ((*S*)-1-Benzoylpiperidine-2-carbonyl)-*L*-valinate (3a): A 100 mL round-bottomed flask was charged with a solution of *L*-valine methyl ester hydrochloride (922 mg, 5.50 mmol) in CH₂Cl₂ (45 mL) and cooled to 0 °C. *i*Pr₂NEt (0.96 mL, 5.5 mmol) was added dropwise over 5 min and the resulting mixture was stirred at 0 °C for 10 min. To this solution were added the crude **S1**, hydroxybenzotriazole (HOBt: 676 mg, 5.00 mmol) followed by 1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDC: 1.05 g, 5.50 mmol) and the resulting mixture was warmed to room temperature. After 19 h, the reaction mixture was cooled to 0 °C and quenched with 1 M HCl aq. (10 mL). The phases were separated and the aqueous phase was extracted with CH₂Cl₂ (10 mL × 3). The combined organic layers were washed with brine (10 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude residue was purified by column chromatography (SiO₂, 25% to 50% EtOAc/ hexanes) to provide the title compound (1.23 g, 71% over 2

steps) as a white amorphous solid.

Optical Rotation: $[\alpha]_{D}^{22} = -100 (c \ 0.770, \text{CHCl}_3);$

¹**H NMR** (600 MHz, CDCl₃, *ca*. 3:1 mixture of rotamers): δ 7.44 (br s, 5H), 7.10 (d, *J* = 6.6 Hz, 0.75H), 6.60 (br, 0.25H), 5.29 (s, 0.75H), 4.79 (br, 0.25H), 4.64 (br, 0.25H), 4.52 (s, 0.75H), 4.42 (br, 0.25H), 3.76 (s, 3H), 3.72 (s, 0.75H), 3.04 (t, *J* = 12.4 Hz, 0.75H), 2.86 (br, 0.25H), 2.35–2.30 (m, 1H), 2.27–2.20 (m, 1H), 1.93–1.84 (m, 0.75H), 1.76–1.74 (m, 1.25H), 1.68–1.47 (m, 3H), 0.93 (d, *J* = 6.9 Hz, 6H);

¹³C NMR (151 MHz, CDCl₃, peaks of major rotamer are listed): δ 172.5, 172.0, 170.9, 135.2, 130.2, 128.6, 127.1, 57.1, 52.6, 52.1, 46.1, 30.9, 25.5, 25.3, 20.5, 19.2, 17.7;

HRMS (ESI): Calc'd for $C_{19}H_{27}N_2O_4 [M+H]^+$: 347.1965, found: 347.1959.



Methyl (*S*)-1-(Benzoyl-*L*-alanyl)piperidine-2-carboxylate (3b) was prepared from methyl (*S*)-piperidine-2-carboxylate (55) and *N*-benzoyl-*L*-alanine according to the representative procedure. The title compound was obtained as a colorless foam (1.37 g, 86% over 2 steps).

Optical Rotation: $[\alpha]^{22}{}_{D} = -54 (c \ 1.7, \text{CHCl}_{3});$

¹**H NMR** (600 MHz, CDCl₃, *ca*. 4:4:1:1 mixture of rotamers): δ 7.78 (d, *J* = 7.5 Hz, 2H), 7.45 (t, *J* = 7.5 Hz, 1H), 7.42–7.41 (m, 1H), 7.38 (t, *J* = 7.5 Hz, 2H), 5.35 (d, *J* = 5.3 Hz, 0.4H), 5.29 (d, *J* = 5.3 Hz, 0.4H), 5.14 (quint, *J* = 6.8 Hz, 0.4H), 5.10 (quint, *J* = 6.8 Hz, 0.4H), 5.04 (quint, *J* = 6.8 Hz, 0.1H), 4.94 (quint, *J* = 6.8 Hz, 0.1H), 4.87 (d, *J* = 4.0 Hz, 0.1H), 4.62 (d, *J* = 4.9 Hz, 0.1H), 4.54 (d, *J* = 12.8 Hz, 0.1H), 4.48 (d, *J* = 13.9 Hz, 0.1H), 3.86 (d, *J* = 12.8 Hz, 0.4H), 3.80 (d, *J* = 13.3 Hz, 0.4H), 3.75 (s, 0.3H), 3.70 (s, 1.2H), 3.68 (s, 1.2H), 3.59 (s, 0.3H), 3.28–3.21 (m, 0.8H), 2.77–2.69 (m, 0.2H), 2.33 (d, *J* = 13.6 Hz, 0.1H), 2.27–2.23 (m, 0.9H), 1.72–1.71 (m, 2H), 1.65–1.59 (m, 1H), 1.51–1.28 (m, 3H), 1.42 (d, *J* = 6.8 Hz, 1.5H), 1.41 (d, *J* = 6.8 Hz, 1.5H);

¹³C NMR (151 MHz, CDCl₃, peaks of 2 major rotamers are listed): δ 172.6, 172.4, 171.3, 171.2, 166.3, 166.2, 134.2, 134.2, 131.5, 128.5, 128.5, 127.1, 127.0, 52.5, 52.4, 52.3, 52.3, 46.0, 45.8, 43.5, 43.4, 26.6, 26.4, 25.2, 25.1, 20.9, 19.6, 18.2 (*Two* ¹³C signals are overlapping with others);

HRMS (ESI): Calc'd for $C_{17}H_{22}N_2O_4Na[M+Na]^+$: 341.1472, found: 341.1471.





Optical Rotation: $[\alpha]^{22}_{D} = -105 (c \ 2.01, \text{CHCl}_{3});$

¹**H NMR** (600 MHz, CDCl₃, *ca*. 4:1 mixture of rotamers): δ 7.43 (br s, 5H), 7.17 (d, *J* = 7.3 Hz, 0.8H), 6.79 (br, 0.2H), 6.62 (d, *J* = 5.0 Hz, 0.8H), 6.46 (br, 0.2H), 5.29 (s, 0.8H), 4.79 (br, 0.2H), 4.58 (quint, *J* = 7.2 Hz, 1H), 4.35 (br, 0.2H), 4.33–4.30 (m, 1H), 3.74 (s, 3H), 3.71 (br, 0.8H), 3.05 (t, *J* = 12.8 Hz, 0.8H), 2.88 (br, 0.2H), 2.32–2.12 (m, 2H), 1.85–1.52 (m, 5H), 1.41 (d, *J* = 7.2 Hz, 3H), 0.95 (d, *J* = 6.7 Hz, 6H);

¹³C NMR (151 MHz, CDCl₃, peaks of major rotamer are listed): δ 173.2, 172.7, 171.3, 170.5, 135.2, 130.4, 128.7, 127.3, 58.4, 53.0, 52.5, 48.2, 46.2, 30.8, 25.7, 25.5, 20.8, 19.5, 18.2, 17.9;

HRMS (ESI): Calc'd for C₂₂H₃₁N₃O₅Na [M+Na]⁺: 440.2156, found: 440.2151.



Methyl Benzoyl-L-prolyl-L-valinate (3d) was prepared from *N*-benzoyl-*L*-proline (56) and *L*-valine methyl ester hydrochloride according to the representative procedure. The title compound was obtained as a white solid (538 mg, 81%).

Melting Point: 104–106 °C;

Optical Rotation: $[\alpha]_{D}^{22} = -142$ (*c* 0.960, CHCl₃);

¹**H NMR** (600 MHz, CDCl₃): δ 7.45–7.38 (m, 5H), 7.33 (br, 1H), 4.80 (s, 1H), 4.49 (t, *J* = 6.4 Hz, 1H), 3.71 (s, 3H), 3.51 (s, 1H), 3.44 (s, 1H), 2.43 (s, 1H), 2.18–2.16 (m, 1H), 2.02 (s, 2H), 1.81 (s, 1H), 0.92 (d, *J* = 6.4 Hz, 3H), 0.89 (d, *J* = 6.4 Hz, 3H);

¹³C NMR (151 MHz, CDCl₃): δ172.2, 171.1, 171.0, 136.4, 130.3, 128.5, 127.0, 59.8, 57.6, 52.1, 50.4, 31.1, 27.1, 25.5, 19.2, 17.8;

HRMS (ESI): Calc'd for C₁₈H₂₄N₂O₄Na [M+Na]⁺: 355.1628, found: 355.1627.

2.3. Preparation of Enamides



(3,4-Dihydropyridin-1(2*H*)-yl)(phenyl)methanone (10a) was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (57).



100 Α mL round-bottomed flask with solution of tert-butyl was charged а 4-methyl-2-oxopiperidine-1-carboxylate (58) (2.00 g, 9.38 mmol) in dioxane (40 mL) and cooled to 0 °C. HCl (4.0 M solution in dioxane, 7.00 mL, 28.1 mmol) was added and the resulting mixture was warmed to room temperature. After 16 h, the reaction mixture was concentrated under reduced pressure to afford a white solid, which was used in the next step without further purification. The solid was dissolved in MeCN (30 mL) and the resulting solution was cooled to 0 °C. To this solution was added Et₃N (3.92 mL, 28.1 mmol), DMAP (115 mg, 0.938 mmol) and BzCl (1.31 mL, 11.3 mmol) and the reaction mixture was heated to 70 °C. After 12 h, H₂O (1.0 mL) was added and the reaction mixture was allowed to continue to stir at 70 °C for an additional 1 h. The solution was then allowed to cool to room temperature, poured into a separatory funnel, and washed with sat. NaHCO₃ aq. (20 mL). The aqueous phase was extracted with EtOAc (10 mL \times 2). The combined organic layers were washed with 1 M HCl (10 mL) and brine (10 mL), dried over MgSO₄, filtered and concentrated under reduced pressure. The crude residue was purified by column chromatography (SiO₂, 5% to 15% EtOAc/ hexanes) to provide S2 (1.47 g, 73% over 2 steps) as a white solid.

Melting Point: 74–77 °C;

¹**H NMR** (500 MHz, CDCl₃): δ 7.54 (d, J = 7.4 Hz, 2H), 7.46 (t, J = 7.4 Hz, 1H), 7.38 (t, J = 7.4 Hz, 2H), 3.95 (dt, J = 12.9, 4.3 Hz, 1H), 3.64 (ddd, J = 12.9, 11.5, 4.0 Hz, 1H), 2.59 (ddd, J = 16.4, 4.6, 1.9 Hz, 1H), 2.23 (dd, J = 16.4, 10.9 Hz, 1H), 2.13–2.05 (m, 2H), 1.65–1.57 (m, 1H), 1.10 (d, J = 6.4 Hz, 3H); ¹³**C NMR** (126 MHz, CDCl₃): δ 174.8, 173.4, 136.2, 131.6, 128.2, 128.0, 45.4, 42.9, 30.9, 28.8, 21.3; **HRMS** (ESI): Calc'd for C₁₃H₁₆NO₂ [M+H]⁺: 218.1176, found: 218.1183.



(4-Methyl-3,4-dihydropyridin-1(2*H*)-yl)(phenyl)methanone (10b) was prepared from S2 according to a published procedure (57). The title compound was obtained as a colorless oil (323 mg, 74%).

¹**H NMR** (500 MHz, CDCl₃, *ca*. 3:1 mixture of conformers): δ 7.45–7.35 (m, 5H), 7.22 (br, 0.25H), 6.37 (d, *J* = 7.9 Hz, 0.75H), 5.08 (br, 0.25H), 4.68 (d, *J* = 7.9 Hz, 0.75H), 4.03 (br, 0.75H), 3.59–3.55 (m, 1H), 3.47 (br, 0.25H), 2.34 (br, 1H), 2.02 (br, 0.75H), 1.83 (br, 0.25H), 1.55 (br, 0.75H), 1.40 (br, 0.25H), 1.02 (d, *J* = 7.0 Hz, 3H);

¹³C NMR (126 MHz, CDCl₃, peaks of major conformer are listed): δ 169.3, 135.1, 130.2, 128.4, 128.2, 126.3, 113.8, 39.9, 30.0, 27.4, 21.3;

HRMS (ESI): Calc'd for C₁₃H₁₆NO [M+H]⁺: 202.1226, found: 202.1226.



Phenyl(2,3,4,5-tetrahydro-1*H***-azepin-1-yl)methanone (10c)** was prepared according to a published procedure. Spectral data were in full agreement with the reported literature values (59).

3. Experimental Procedures for the Silver-Mediated Fluorination

3.1. Representative Procedure for the Silver-Mediated Monofluorination of Cyclic Amines



To a 1-dram vial was added sequentially **1a** (18.9 mg, 0.100 mmol), AgBF₄ (77.9 mg, 0.400 mmol), Selectfluor[®] (142 mg, 0.400 mmol) and 1:9 acetone: H₂O (0.5 mL) The resulting mixture was heated to 40 °C and held at this temperature. After 1 h, the reaction mixture was partitioned with EtOAc (0.5 mL) and H₂O (0.5 mL) and the phases were separated. The aqueous phase was extracted with EtOAc (1.5 mL × 3) and the combined organic layers were concentrated under reduced pressure. The crude residue was purified by preparative thin-layer chromatography (50% EtOAc/hexanes) to provide *N*-(4-fluorobutyl)-*N*-formylbenzamide (**2a**) (18.0 mg, 81%) as a pale yellow oil.

¹**H** NMR (600 MHz, CDCl₃): δ 8.93 (s, 1H), 7.57 (t, *J* = 7.2 Hz, 1H), 7.53–7.48 (m, 4H), 4.48 (dt, *J* = 47.6, 5.6 Hz, 2H), 3.92 (t, *J* = 7.1 Hz, 2H), 1.82–1.72 (m, 4H);

¹³**C NMR** (151 MHz, CDCl₃): 172.5, 164.3, 133.7, 132.3, 129.1, 128.9, 83.6 (d, *J* = 165.2 Hz), 40.2, 28.0 (d, *J* = 20.2 Hz), 24.2 (d, *J* = 5.0 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.5 – –217.9 (m, 1F);

HRMS (ESI): Calc'd for C₁₂H₁₄FNO₂Na [M+Na]⁺: 246.0906, found: 246.0906.



N-(4-Fluorobutyl)-*N*-formylacetamide (2b): The title compound was prepared according to the representative procedure using 1b. Purification by preparative thin-layer chromatography (25% EtOAc/hexanes) provided the title compound (7.2 mg, 45%) as a colorless oil.

¹**H NMR** (600 MHz, CDCl₃): δ 9.16 (s, 1H), 4.45 (dt, *J* = 47.0, 5.7 Hz, 2H), 3.73 (t, *J* = 7.3 Hz, 2H), 2.41 (s, 3H), 1.74–1.62 (m, 4H);

¹³C NMR (151 MHz, CDCl₃): δ 171.2, 162.8, 83.6 (d, *J* = 165.3 Hz), 39.6, 27.9 (d, *J* = 20.0 Hz), 24.3 (d, *J* = 4.3 Hz), 23.0;

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.8 (tt, *J* = 47.9, 24.4 Hz, 1F);

HRMS (EI): Calc'd for $C_7H_{13}FNO_2 [M+H]^+$: 162.0925, found: 162.0933.



tert-Butyl (4-Fluorobutyl)(formyl)carbamate (2c): The title compound was prepared according to the representative procedure using 1c. Purification by preparative thin-layer chromatography (25% EtOAc/hexanes) provided the title compound (8.6 mg, 39%) as a colorless oil.

¹**H NMR** (600 MHz, CDCl₃): δ 9.17 (s, 1H), 4.45 (dt, *J* = 47.0, 5.7 Hz, 2H), 3.63 (t, *J* = 7.1 Hz, 2H), 1.73–1.63 (m, 4H), 1.54 (s, 9H);

¹³C NMR (151 MHz, CDCl₃): δ 163.2, 152.6, 84.2, 83.6 (d, *J* = 165.2 Hz), 40.2, 28.2, 27.8 (d, *J* = 20.0 Hz), 24.4 (d, *J* = 5.0 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.9 (tt, *J* = 48.1, 25.7 Hz, 1F);

HRMS (ESI): Calc'd for C₁₀H₁₉FNO₃ [M+H]⁺: 220.1343, found: 220.1351.



N-(4-Fluorobutyl)pivalamide (2d): The title compound was prepared according to the representative procedure using 1d. Purification by preparative thin-layer chromatography (25% EtOAc/hexanes) provided the title compound (12.3 mg, 70%) as a white solid.

Melting Point: 67–69 °C;

¹**H** NMR (600 MHz, CDCl₃): δ 5.70 (br, 1H), 4.47 (dt, J = 47.2, 5.8 Hz, 2H), 3.29 (t, J = 6.8 Hz, 2H), 1.76–

1.61 (m, 4H), 1.19 (s, 9H);

¹³C NMR (151 MHz, CDCl₃): δ 178.6, 83.9 (d, *J* = 164.6 Hz), 39.1, 38.8, 27.9 (d, *J* = 19.9 Hz), 27.7, 25.9 (d, *J* = 4.4 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –219.0 (tt, *J* = 47.4, 25.8 Hz, 1F);

HRMS (ESI): Calc'd for C₉H₁₉FNO [M+H]⁺: 176.1445, found: 176.1442.



N-(2-Fluoroethyl) benzamide (2e): The title compound was prepared according to the representative procedure using 1e. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (6.7 mg, 40%) as a colorless oil.

¹**H NMR** (400 MHz, CDCl₃): δ 7.80 (d, *J* = 7.5 Hz, 2H), 7.51 (t, *J* = 7.5 Hz, 1H), 7.43 (t, *J* = 7.5 Hz, 2H), 6.63 (br, 1H), 4.60 (dt, *J* = 47.4, 4.9 Hz, 2H), 3.77 (dq, *J* = 28.3, 4.9 Hz, 2H);

¹³**C NMR** (101 MHz, CDCl₃): δ 167.8, 134.2, 131.8, 128.7, 127.1, 83.0 (d, *J* = 166.4 Hz), 40.6 (d, *J* = 19.7 Hz);

¹⁹F NMR (376 MHz, CDCl₃): δ –223.0 – –223.5 (m, 1F);

HRMS (ESI): Calc'd for C₉H₁₁FNO [M+H]⁺: 168.0819, found: 168.0825.



N-(3-Fluoropropyl) benzamide (2f): The title compound was prepared according to the representative procedure using 1f. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (6.0 mg, 33%) as a pale yellow waxy solid.

¹**H** NMR (400 MHz, CDCl₃): δ 7.76 (d, J = 6.9 Hz, 2H), 7.51 (t, J = 6.9 Hz, 1H), 7.43 (t, J = 6.9 Hz, 2H),

6.44 (br, 1H), 4.61 (dt, J = 47.3, 6.0 Hz, 2H), 3.63 (q, J = 6.0 Hz, 2H), 2.04 (dquint, J = 28.2, 6.0 Hz, 2H);

¹³C NMR (101 MHz, CDCl₃): δ 167.8, 134.6, 131.6, 128.7, 127.0, 83.0 (d, *J* = 163.9 Hz), 37.4 (d, *J* = 4.1 Hz), 30.3 (d, *J* = 19.2 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –218.9 – –219.3 (m, 1F);

HRMS (ESI): Calc'd for C₁₀H₁₃FNO [M+H]⁺: 182.0976, found: 182.0976.



N-(5-Fluoropentyl)-N-formylbenzamide (2g): The title compound was prepared according to the representative procedure using 1g. Purification by preparative thin-layer chromatography (50%)

EtOAc/hexanes) provided the title compound (15.6 mg, 67%) as a colorless oil.

¹**H NMR** (700 MHz, CDCl₃): δ 8.92 (s, 1H), 7.57 (t, *J* = 7.5 Hz, 1H), 7.53 (d, *J* = 7.5 Hz, 2H), 7.49 (t, *J* = 7.5 Hz, 2H), 4.45 (dt, *J* = 47.3, 6.1 Hz, 2H), 3.88 (t, *J* = 7.5 Hz, 2H), δ 1.81–1.68 (m, 4H), 1.49 (quint, J = 7.5 Hz, 2H);

¹³C NMR (176 MHz, CDCl₃): δ 172.5, 164.4, 133.7, 132.3, 129.1, 128.9, 83.9 (d, *J* = 164.7 Hz), 40.5, 30.1 (d, *J* = 19.8 Hz), 27.7, 22.8 (d, *J* = 5.3 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.9 (tt, *J* = 47.3, 25.5 Hz, 1F);

HRMS (ESI): Calc'd for C₁₃H₁₇FNO₂ [M+H]⁺: 238.1238, found: 238.1238.



N-(6-Fluorohexyl)-N-formylbenzamide (2h): The title compound was prepared according to the representative procedure using 1h. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (12.4 mg, 49%) as a colorless oil.

¹**H** NMR (700 MHz, CDCl₃) δ 8.92 (s, 1H), 7.60 (t, *J* = 7.2 Hz, 1H), 7.53–7.48 (m, 4H), 4.47 (dt, *J* = 47.3, 6.0 Hz, 2H), 3.87 (t, *J* = 7.4 Hz, 2H), 1.73–1.63 (m, 4H), 1.47–1.41 (m, 4H);

¹³C NMR (176 MHz, CDCl₃) δ 172.5, 164.4, 133.8, 132.3, 129.1, 128.9, 84.2 (d, *J* = 164.4 Hz), 40.6, 30.4 (d, *J* = 19.7 Hz), 28.0, 26.7, 25.0 (d, *J* = 5.2 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.5 (tt, *J* = 48.9, 24.5 Hz, 1F);

HRMS (ESI): Calc'd for C₁₄H₁₈FNO₂Na [M+Na]⁺: 274.1214, found: 274.1216.



N-(4-Fluoro-3-methylbutyl)-*N*-formylbenzamide (2i): The title compound was prepared according to the representative procedure using 1i. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (14.0 mg, 59%) as a colorless oil.

¹**H NMR** (400 MHz, CDCl₃): δ 8.92 (s, 1H), 7.60–7.47 (m, 5H), 4.32 (ddd, *J* = 47.5, 8.9, 5.7 Hz, 1H), 4.29 (ddd, *J* = 47.5, 8.9, 5.9 Hz, 1H), 3.94 (t, *J* = 7.6 Hz, 2H), 1.98–1.77 (m, 2H), 1.58–1.47 (m, 1H), 1.05 (dd, *J* = 6.7, 0.9 Hz, 3H);

¹³**C NMR** (101 MHz, CDCl₃): δ 172.4, 164.3, 133.7, 132.4, 129.1, 128.9, 88.0 (d, *J* = 169.7 Hz), 38.8, 32.4 (d, *J* = 18.5 Hz), 31.1 (d, *J* = 5.1 Hz), 15.8 (d, *J* = 6.8 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –221.2 (td, *J* = 47.4, 19.4 Hz, 1F);

HRMS (ESI): Calc'd for C₁₂H₁₆FNONa [M–CO+Na]⁺: 232.1108, found: 232.1107.



N-(5-Fluoropentan-2-yl)benzamide (2j): The title compound was prepared according to the representative procedure using 1j. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (17.0 mg, 81%) as a colorless oil.

¹**H NMR** (400 MHz, CDCl₃): δ 7.75 (d, *J* = 7.3 Hz, 2H), 7.49 (t, *J* = 7.3 Hz, 1H), 7.42 (d, *J* = 7.3 Hz, 2H), 5.96 (br, 1H), 4.47 (dddd, *J* = 47.2, 9.0, 6.4, 3.7 Hz, 2H), 4.26 (hept, *J* = 6.6 Hz, 1H), 1.86–1.60 (m, 4H), 1.27 (d, *J* = 6.6 Hz, 3H);

¹³C NMR (101 MHz, CDCl₃): δ 167.1, 134.9, 131.5, 128.7, 126.9, 83.9 (d, *J* = 164.8 Hz), 45.5, 33.0 (d, *J* = 4.3 Hz), 27.3 (d, *J* = 19.9 Hz), 21.3;

¹⁹F NMR (376 MHz, CDCl₃): δ –217.6 – –218.0 (m, 1F);

HRMS (ESI): Calc'd for C₁₂H₁₆FNONa [M+Na]⁺: 232.1108, found: 232.1111.



N-(6-Fluorohexan-3-yl) benzamide (2k): The title compound was prepared according to the representative procedure using 1k. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (19.0 mg, 85%) as a colorless oil.

¹**H NMR** (700 MHz, CDCl₃): δ 7.75 (d, *J* = 7.4 Hz, 2H), 7.49 (t, *J* = 7.4 Hz, 1H), 7.42 (t, *J* = 7.4 Hz, 2H),

5.92 (br, 1H), 4.53–4.41 (m, 2H), 4.11 (ddq, *J* = 13.6, 8.9, 4.9 Hz, 1H), 1.85–1.72 (m, 3H), 1.70–1.64 (m, 1H), 1.60–1.49 (m, 2H), 0.97 (t, *J* = 7.4 Hz, 3H);

¹³**C NMR** (151 MHz, CDCl₃): δ 167.6, 134.9, 131.5, 128.7, 126.9, 84.0 (d, J = 164.7 Hz), 50.8, 30.9 (d, J = 164.7 Hz), 50.8, 30.9

4.4 Hz), 28.4, 27.2 (d, *J* = 19.8 Hz), 10.5;

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.8 – –218.2 (m, 1F);

HRMS (ESI): Calc'd for C₁₃H₁₉FNO [M+H]⁺: 224.1445, found: 224.1450.



N-(4-Fluoro-2-methylpentyl)-N-formylbenzamide (21): The title compound was prepared according to the representative procedure using 11. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (12.6 mg, 50%) as a colorless oil as a 1:1 mixture of

diastereomers.

¹**H** NMR (400 MHz, CDCl₃): (400 MHz, 1H), 8.95 (s, 1H), 7.59–7.47 (m, 10H), 4.91–4.68 (m, 2H), 3.88– 3.74 (m, 4H), 2.25–2.10 (m, 2H), 1.80–1.49 (m, 3H), 1.42–1.26 (m, 1H), 1.34 (dd, *J* = 23.7, 6.1 Hz, 3H), 1.33 (dd, *J* = 23.7, 6.1 Hz, 3H), 1.00 (d, *J* = 6.6 Hz, 3H), 0.99 (d, *J* = 6.6 Hz, 3H);

¹³C NMR (101 MHz, CDCl₃): δ 172.7 (2C), 164.7, 164.6, 133.84, 133.79, 132.4, 132.3, 129.14, 129.13, 128.94, 128.92, 89.6 (d, *J* = 164.6 Hz), 88.7 (d, *J* = 165.0 Hz), 46.0, 45.8, 41.90 (d, *J* = 20.7 Hz), 41.86 (d, *J* = 20.6 Hz), 29.7 (d, *J* = 4.0 Hz), 29.1 (d, *J* = 2.9 Hz), 21.8 (d, *J* = 22.6 Hz), 21.4 (d, *J* = 22.8 Hz), 18.3, 17.6;
¹⁹F NMR (376 MHz, CDCl₃): δ -169.8 - -170.3 (m, 1F), -172.6 - -173.1 (m, 1F);

HRMS (ESI): Calc'd for C₁₄H₁₈FNO₂Na [M+Na]⁺: 274.1214, found: 274.1212.



trans-N-(2-(2-Fluoroethyl)cyclohexyl)benzamide (2m): The title compound was prepared according to the representative procedure using 1m. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (10.7 mg, 43%) as a colorless oil.

¹**H** NMR (400 MHz, CDCl₃): δ 7.76 (d, *J* = 7.3 Hz, 2H), 7.50 (t, *J* = 7.3 Hz, 1H), 7.44 (t, *J* = 7.3 Hz, 2H),

5.92 (br, 1H), 4.52 (ddd, *J* = 47.4, 7.2, 3.7 Hz, 2H), 3.83 (dq, *J* = 10.6, 3.9 Hz, 1H), 2.16–2.05 (m, 2H), 2.00–

1.96 (m, 1H), 1.81–1.74 (m, 2H), 1.63–1.09 (m, 6H);

¹³C NMR (151 MHz, CDCl₃): δ 167.2, 135.0, 131.5, 128.7, 127.0, 82.4 (d, *J* = 163.6 Hz), 53.0, 40.1 (d, *J* = 3.3 Hz), 34.2, 33.8 (d, *J* = 19.5 Hz), 31.6, 25.7, 25.4;

¹⁹**F NMR** (400 MHz, CDCl₃): δ –217.5 – –217.9 (m, 1F);

HRMS (ESI): Calc'd for C₁₅H₂₁FNO [M+H]⁺: 250.1602, found: 250.1595.



Methyl (*S*)-2-Benzamido-5-fluoropentanoate (2n): The title compound was prepared according to the representative procedure using 1n. Purification by preparative thin-layer chromatography (25% EtOAc/hexanes) provided the title compound (17.2 mg, 68%) as a white solid.

Melting Point: 63–65 °C;

Optical Rotation: $[\alpha]^{22}_{D} = +17 (c \ 0.67, \text{CHCl}_{3});$

¹**H NMR** (600 MHz, CDCl₃): δ 7.80 (d, *J* = 7.4 Hz, 2H), 7.52 (t, *J* = 7.4 Hz, 1H), 7.44 (t, *J* = 7.4 Hz, 2H), 6.78 (d, *J* = 7.4 Hz, 1H), 4.87 (dt, *J* = 7.4, 5.3 Hz, 1H), 4.54–4.41 (m, 2H), 3.79 (s, 3H), 2.16–2.10 (m, 1H),

1.95-1.89 (m, 1H), 1.88-1.72 (m, 2H);

¹³C NMR (151 MHz, CDCl₃): δ 172.9, 167.2, 133.9, 132.0, 128.8, 127.2, 83.4 (d, *J* = 165.5 Hz), 52.7, 52.2, 28.9 (d, *J* = 4.7 Hz), 26.6 (d, *J* = 20.2 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –218.4 (tt, *J* = 47.3, 25.7 Hz, 1F);

HRMS (ESI): Calc'd for C₁₃H₁₇FNO₃ [M+H]⁺: 254.1187, found: 254.1185.



Methyl (S)-1-Benzoyl-5-oxopyrrolidine-2-carboxylate (10): The title compound was prepared according to the representative procedure using **10**. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (11.4 mg, 46% yield). Spectroscopic data is fully consistent with previously reported data (*60*).



1-Methylpyrrolidine-2,5-dione (1p): The title compound was prepared according to the representative procedure using **1p**. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (9.2 mg, 81% yield). Spectroscopic data is fully consistent with previously reported data (*10*).



1-Methylpiperidine-2,6-dione (1q): The title compound was prepared according to the representative procedure using **1p**. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (7.1 mg, 56% yield). Spectroscopic data is fully consistent with previously reported data (*10*).



4-Fluoro-N-pivaloylbutanamidebenzamide (2r): The title compound was prepared according to the representative procedure using 1r. Purification by preparative thin-layer chromatography (30%)

EtOAc/hexanes) provided the title compound (8.1 mg, 43% yield) as a white solid.

Melting Point: 63–65 °C;

¹**H NMR** (600 MHz, CDCl₃): δ 8.07 (br, 1H), 4.51 (dt, *J* = 47.2, 5.9 Hz, 2H), 3.01 (t, *J* = 7.2 Hz, 2H), 2.05 (ddd, *J* = 25.8, 7.2, 5.9 Hz, 2H), 1.25 (s, 9H);

¹³**C NMR** (151 MHz, CDCl₃): δ 177.2, 175.3, 83.2 (d, *J* = 165.1 Hz), 40.2, 33.4 (d, *J* = 5.2 Hz), 27.2, 25.0 (d, *J* = 20.3 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –219.1 (tt, *J* = 47.2, 25.8 Hz, 1F);

HRMS (ESI): Calc'd for C₉H₁₇FNO₂ [M+H]⁺: 190.1238, found: 190.1245.



N-(2,4-Difluorobutyl)benzamide (2s): The title compound was prepared according to the representative procedure using 1s. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (6.0 mg, 28% yield) as a waxy white solid.

¹**H NMR** (400 MHz, CDCl₃): δ 7.79 (d, J = 7.3 Hz, 2H), 7.52 (t, J = 7.3 Hz, 1H), 7.45 (t, J = 7.3 Hz, 2H),

6.52 (br, 1H), 4.90 (dtt, *J* = 48.4, 7.2, 3.6 Hz, 1H), 4.62 (dt, *J* = 47.3, 5.9 Hz, 2H), 3.89 (dddd, *J* = 28.2, 14.7,

6.6, 3.0 Hz, 1H), 3.66-3.54 (m, 1H), 2.20-1.93 (m, 2H);

¹³**C NMR** (101 MHz, CDCl₃): δ167.8, 134.1, 131.9, 128.8, 127.1, 89.8 (dd, *J* = 169.6, 3.6 Hz), 79.7 (dd, *J* = 165.6, 5.1 Hz), 43.6 (d, *J* = 20.1 Hz), 33.5 (t, *J* = 20.1 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –188.7 – –189.1 (m, 1F), –220.3 – –220.7 (m, 1F);

HRMS (ESI): Calc'd for $C_{11}H_{14}F_2NO [M+H]^+$: 214.1038, found: 214.1038.



N-(3,4-Difluorobutyl)benzamide (2t): The title compound was prepared according to the representative procedure using 1t. Purification by preparative thin-layer chromatography (50% EtOAc/hexanes) provided the title compound (4.7 mg, 22% yield) as a waxy white solid.

¹**H NMR** (700 MHz, CDCl₃) 7.78 (d, *J* = 7.4 Hz, 2H), 7.52 (t, *J* = 7.4 Hz, 1H), 7.46–7.41 (m, 2H), 6.45 (br,

1H), 4.98–4.74 (m, 1H), 4.69–4.41 (m, 2H), 3.73–3.62 (m, 2H), 2.14–1.98 (m, 2H);

¹³C NMR (176 MHz, CDCl₃) δ 167.9, 134.4, 131.8, 128.8, 127.0, 90.8 (dd, J = 172.6, 19.7 Hz), 84.0 (dd, J = 172.6, 19.7 Hz), 84.0

174.5, 22.4 Hz), 36.5 (d, *J* = 4.3 Hz), 36.5 (dd, *J* = 20.2, 6.0 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –189.5 – –190.0 (m, 1F), –229.4 – –229.8 (m, 1F);

HRMS (ESI): Calc'd for $C_{11}H_{14}F_2NO[M+H]^+$: 214.1038, found: 214.1038.



Methyl ((*S*)-2-Benzamido-5-fluoropentanoyl)-*L*-valinate (4a): The title compound was prepared according to the representative procedure using **3a** with the following modifications: reaction time of 15 h at room temperature. Purification by preparative thin-layer chromatography (20% to 50% EtOAc/hexanes) provided the title compound (17.5 mg, 50% yield) as a white amorphous solid along with recovered **3a** (8.7 mg, 25%). **Optical Rotation:** $[\alpha]^{22}_{D} = -13$ (*c* 0.47, CHCl₃);

¹**H NMR** (600 MHz, CDCl₃): δ 7.80 (d, *J* = 7.4 Hz, 2H), 7.50 (t, *J* = 7.4 Hz, 1H), 7.80 (t, *J* = 7.4 Hz, 2H), 7.04 (d, *J* = 8.3 Hz, 1H), 6.89 (d, *J* = 8.5 Hz, 1H), 4.84 (q, *J* = 8.5 Hz, 1H), 4.51 (dd, *J* = 8.3, 5.0 Hz, 1H), 4.50 (dt, *J* = 47.4, 5.7 Hz, 2H), 3.75 (s, 3H), 2.21–2.14 (m, 1H), 2.09 (ddt, *J* = 13.4, 9.4, 6.1 Hz, 1H), 1.97–1.76 (m, 3H), 0.90 (d, *J* = 6.9 Hz, 3H), 0.88 (d, *J* = 6.9 Hz, 3H);

¹³C NMR (151 MHz, CDCl₃): δ 172.1, 171.6, 167.5, 133.8, 132.0, 128.7, 127.2, 83.9 (d, *J* = 164.8 Hz), 57.6, 53.0, 52.3, 31.1, 29.1 (d, *J* = 4.1 Hz), 26.6 (d, *J* = 20.0 Hz), 19.1, 17.8;

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.1 – –217.5 (m, 1F);

HRMS (ESI): Calc'd for C₁₈H₂₅FN₂O₄Na [M+Na]⁺: 375.1691, found: 375.1692.



Methyl (S)-2-((S)-2-Benzamidopropanamido)-5-fluoropentanoate (4b): The title compound was prepared according to the representative procedure using **3b** with the following modifications: reaction time of 15 h at room temperature. Purification by preparative thin-layer chromatography (20% to 50% EtOAc/hexanes) provided the title compound (12.3 mg, 38% yield) as a white amorphous solid along with recovered **3b** (12.7 mg, 40%).

Optical Rotation: $[\alpha]_{D}^{22} = +7.1$ (*c* 0.63, CHCl₃);

¹H NMR (600 MHz, CDCl₃, *ca*. 1:1 mixture of rotamers): δ 7.81–7.79 (m, 2H), 7.52–7.49 (m, 1H), 7.45–7.40 (m, 2H), 7.33 (d, J = 7.9 Hz, 0.5H), 7.23 (d, J = 7.9 Hz, 0.5H), 7.09 (d, J = 7.3 Hz, 1H), 4.83 (dq, J = 14.3, 7.3 Hz, 1H), 4.63–4.59 (m, 1H), 4.49–4.47 (m, 0.5H), 4.42–4.39 (m, 1H), 4.34–4.32 (m, 0.5H), 3.75 (s, 1.5H), 3.68 (s, 1.5H), 2.03–1.98 (m, 1H), 1.86–1.66 (m, 3H), 1.495 (d, J = 7.0 Hz, 1.5H), 1.491 (d, J = 7.0 Hz, 1.5H); ¹³C NMR (151 MHz, CDCl₃, mixture of rotamers): δ 172.6, 172.5, 172.4, 172.3, 167.5, 167.4, 133.9 (2C), 131.9 (2C), 128.7 (2C), 127.23, 127.20, 82.23 (d, J = 165.7 Hz), 82.20 (d, J = 165.7 Hz), 52.61, 52.58, 52.1,

52.0, 49.29, 49.26, 28.33 (d, *J* = 4.7 Hz), 28.30 (d, *J* = 4.7 Hz), 26.61 (d, *J* = 20.2 Hz), 26.55 (d, *J* = 20.2 Hz), 18.7, 18.6;

¹⁹**F NMR** (376 MHz, CDCl₃): δ –218.4 – –218.9 (m, 1F);

HRMS (ESI): Calc'd for C₁₆H₂₁FN₂O₄Na [M+Na]⁺: 347.1378, found: 347.1379.



Methyl ((S)-2-Benzamido-5-fluoropentanoyl)-L-valyl-L-alaninate (4c): The title compound was prepared according to the representative procedure using 3c with the following modifications: reaction time of 15 h at room temperature. Purification by preparative thin-layer chromatography (5% MeOH/CH₂Cl₂) provided the title compound (16.5 mg, 39% yield) as a white amorphous solid along with recovered 3c (10.4 mg, 25%).

Optical Rotation: $[\alpha]^{22}_{D} = -41$ (*c* 0.39, MeOH);

¹**H NMR** (700 MHz, CD₃OD): δ 7.85–7.84 (m, 2H), 7.55–7.53 (m, 1H), 7.47–7.45 (m, 2H), 4.64 (dt, *J* = 8.8, 4.7 Hz, 1H), 4.47 (d, *J* = 47.6 Hz, 2H), 4.39 (dt, *J* = 13.5, 6.7 Hz, 1H), 4.24 (dd, *J* = 6.6, 4.2 Hz, 1H), 3.69 (s, 3H), 2.08 (dt, *J* = 12.1, 5.5 Hz, 1H), 2.01 (tt, *J* = 10.4, 5.3 Hz, 1H), 1.90–1.77 (m, 3H), 1.39 (br s, 3H), 0.99 (br s, 9H);

¹³C NMR (176 MHz, CD₃OD): δ 174.4, 174.2, 173.3, 170.5, 135.2, 132.9, 129.6, 128.5, 84.4 (d, *J* = 164.3 Hz), 59.8, 55.0, 52.6, 49.4, 32.3, 28.9 (d, *J* = 5.3 Hz), 28.3 (d, *J* = 20.0 Hz), 19.6, 18.6, 17.3;

¹⁹**F NMR** (376 MHz, CD₃OD): δ –219.7 (tt, J = 47.7, 24.8 Hz, 1F);

HRMS (ESI): Calc'd for C₂₁H₃₀FN₃O₅Na [M+Na]⁺: 446.2062, found: 446.2060.



Methyl ((*S*)-1-Benzoyl-5-oxopyrrolidine-2-carbonyl)-*L*-valinate (4d): The title compound was prepared according to the representative procedure using 3d with the following modifications: 0.2 mmol scale with a reaction time of 15 h at room temperature. Purification by preparative thin-layer chromatography (20% to 50% EtOAc/hexanes) provided the title compound (52.6 mg, 76% yield) as a white solid.

Melting Point: 155–158 °C;

Optical Rotation: $[\alpha]^{22}{}_{D} = -256 (c \ 0.46, \text{CHCl}_3);$

¹**H NMR** (600 MHz, CDCl₃): δ 7.62–7.61 (m, 2H), 7.52–7.48 (m, 1H), 7.40 (t, *J* = 7.8 Hz, 2H), 4.83 (dd, *J* = 8.3, 3.6 Hz, 1H), 4.57 (dd, *J* = 8.9, 4.8 Hz, 1H), 3.75 (s, 3H), 2.89–2.83 (m, 1H), 2.53 (ddd, *J* = 17.8, 9.1, 4.3 Hz, 1H), 2.36–2.25 (m, 2H), 2.23–2.17 (m, 1H), 0.95 (d, *J* = 6.9 Hz, 3H), 0.93 (d, *J* = 6.9 Hz, 3H);

¹³C NMR (151 MHz, CDCl₃): δ 174.1, 172.4, 170.9, 170.4, 134.0, 132.2, 129.0, 128.0, 60.0, 57.5, 52.4, 32.1, 31.4, 22.2, 19.0, 17.8;

HRMS (ESI): Calc'd for $C_{18}H_{23}N_2O_5 [M+H]^+$: 347.1601, found: 347.1599.

3.2. Representative Procedure for the Silver-Mediated Difluorination of Enamides



To a 1-dram vial containing a solution of **10a** (18.7 mg, 0.100 mmol) in 1:1 acetone: H₂O (0.5 mL) was added Selectfluor[®] (142 mg, 0.400 mmol) followed by AgBF₄ (4.9 mg, 0.025 mmol). The resulting mixture was stirred at room temperature. After 15 h, the reaction mixture was partitioned with EtOAc (0.5 mL) and H₂O (0.5 mL) and the phases were separated. The aqueous phase was extracted with EtOAc (1.5 mL × 3) and the combined organic layers were concentrated under reduced pressure. The crude residue was purified by preparative thin-layer chromatography (25% EtOAc/hexanes) to provide *N*-(4,4-difluorobutyl)-*N*-formylbenzamide (**12a**) (18.7 mg, 78%) as a colorless oil.

¹**H NMR** (600 MHz, CDCl₃): δ 8.93 (s, 1H), 7.59 (d, *J* = 7.1 Hz, 1H), 7.54–7.49 (m, 4H), 5.87 (tt, *J* = 56.5, 4.2 Hz, 1H), 3.93 (t, *J* = 7.2 Hz, 2H), 1.96–1.82 (m, 4H);

¹³C NMR (151 MHz, CDCl₃): δ 178.6, 83.9 (d, J = 164.6 Hz), 39.1, 38.8, 27.9 (d, J = 19.9 Hz), 27.7, 25.9 (d, J = 4.4 Hz) ¹³C NMR (151 MHz, CDCl₃) δ 172.4, 164.3, 133.5, 132.5, 129.2, 128.9, 116.8 (t, J = 239.2 Hz), 39.9, 31.8 (t, J = 21.5 Hz), 20.9 (t, J = 5.5 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –115.3 (dt, *J* = 56.5, 16.9 Hz, 2F);

HRMS (ESI): Calc'd for C₁₁H₁₄F₂NO [M–CO+H]⁺: 214.1038, found: 214.1038.



N-(4,4-Difluoro-3-methylbutyl)-*N*-formylbenzamide (12b): The title compound was prepared according to the representative procedure using 10b. Purification by preparative thin-layer chromatography (20% EtOAc/hexanes) provided the title compound (13.9 mg, 54%) as a colorless oil.

¹**H** NMR (500 MHz, CDCl₃): δ 8.93 (s, 1H), 7.58 (t, J = 6.9 Hz, 1H), 7.54–7.48 (m, 4H), 5.68 (td, J = 56.7,

3.5 Hz, 1H), 3.94 (t, *J* = 7.5 Hz, 2H), 2.01–1.90 (m, 2H), 1.59–1.53 (m, 1H), 1.11(d, *J* = 6.9 Hz, 3H);

¹³C NMR (126 MHz, CDCl₃): δ 172.4, 164.3, 133.5, 132.5, 129.2, 128.9, 118.8 (t, *J* = 242.5 Hz), 38.4, 35.7 (t, *J* = 19.9 Hz), 28.3 (t, *J* = 4.4 Hz), 12.6 (t, *J* = 5.2 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ -122.4 (ddd, *J* = 56.6, 29.6, 14.7 Hz, 2F);

HRMS (ESI): Calc'd for $C_{13}H_{16}F_2NO [M-CO+H]^+$: 256.1144, found: 256.1143.



N-(5,5-Difluoropentyl)-*N*-formylbenzamide (12c): The title compound was prepared according to the representative procedure using 10c. Purification by preparative thin-layer chromatography (20% EtOAc/hexanes) provided the title compound (15.6 mg, 61%) as a colorless oil. ¹H NMR (500 MHz, CDCl₃): δ 8.92 (s, 1H), 7.59–7.56 (m, 1H), 7.53–7.48 (m, 4H), 5.82 (tt, *J* = 56.8, 4.4 Hz,

1H), 3.88 (t, J = 7.4 Hz, 2H), 1.95–1.83 (m, 2H), 1.72 (quint, J = 7.4 Hz, 2H), 1.53 (quint, J = 7.4 Hz, 2H);

¹³C NMR (126 MHz, CDCl₃): δ 172.5, 164.4, 133.6, 132.4, 129.1, 128.9, 117.1 (t, *J* = 238.8 Hz), 40.2, 33.7 (t, *J* = 21.0 Hz), 27.5, 19.6 (t, *J* = 5.6 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –115.3 (dt, *J* = 56.7, 17.5 Hz, 2F);

HRMS (EI): Calc'd for $C_{13}H_{15}F_2NO_2[M]^+$: 255.1065, found: 255.1070.

4. Mechanistic Studies



According to the representative procedure, aldehyde **D** (61) was used as a starting material. Triphenylmethane was used as an internal standard and ¹H NMR analysis showed the formation of **2a** in 55% yield.



To a 1-dram vial containing **1a** (94.5 mg, 0.5 mmol), AgBF₄ (390 mg, 2.00 mmol) and Selectfluor[®] (710 mg, 2.00 mmol) was added 1:9 acetone: H₂O (2.5 mL), and the resulting mixture was stirred at room temperature. After 16 h, the reaction mixture was partitioned with EtOAc (2.5 mL) and H₂O (0.5 mL) and the phases were separated. The aqueous phase was extracted with EtOAc (5.0 mL \times 3) and the combined organic layers were concentrated under reduced pressure. The crude residue was purified by preparative thin-layer chromatography (50% EtOAc/hexanes) to provide *N*-(4-fluorobutyl)benzamide (**6**) (52.7 mg, 54%) as a waxy white solid.

¹**H NMR** (700 MHz, CDCl₃): δ 7.76 (d, *J* = 7.5 Hz, 2H), 7.49 (t, *J* = 7.5 Hz, 1H), 7.42 (t, *J* = 7.5 Hz, 2H), 6.31 (br, 1H), 4.50 (dt, *J* = 46.9, 5.6 Hz, 2H), 3.51 (q, *J* = 6.5 Hz, 2H), 1.86–1.72 (m, 4H);

¹³C NMR (176 MHz, CDCl₃): δ 167.6, 134.6, 131.4, 128.5, 126.8, 83.7 (d, *J* = 164.7 Hz), 39.5, 27.8 (d, *J* = 19.9 Hz), 25.7 (d, *J* = 4.4 Hz);

¹⁹**F NMR** (376 MHz, CDCl₃): δ –217.4 (tt, *J* = 47.5, 26.2 Hz, 1F);

HRMS (ESI): Calc'd for C₁₁H₁₄FNONa [M+Na]⁺: 218.0952, found: 218.0952.



According to the representative procedure, aldehyde 7 (62) was used as a starting material. Triphenylmethane was used as an internal standard and ¹H NMR analysis showed theformation of **8** (63) in 70% yield.



According to the representative procedure, carboxylic acid 9 (10) was used as a starting material. Triphenylmethane was used as an internal standard and ¹H NMR analysis showed the formation of 6 in 23% yield.

5. Electrochemical Measurement

Non-aqueous electrochemical experiments were conducted under an Ar atmosphere in $0.1 \text{ M NBu}_4\text{PF}_6$ electrolyte in acetonitrile. Cyclic voltammetry experiments were performed using an Epsilon potentiostat from Bioanalytical Systems, Inc. The working electrode was a 3.0 mm diameter glassy carbon disk (from Bioanalytical Systems, Inc.) and was polished between every scan with 0.05-micron alumina powder on a felt pad. The counter electrode was a platinum wire. A silver wire in porous Vycor tip glass tube filled with 0.1 M NBu_4PF_6 in acetonitrile was used as a pseudo-reference electrode. At the conclusion of the series of experiments, the pseudo-reference potentials were referenced against ferrocene/ferrocenium as an external standard. The scan rate for all cyclic voltammograms was 100 mV/sec unless otherwise noted. All scans were compensated for internal resistance. Data measured with respect to Fc/Fc⁺ and reported to SCE.



Fig. S1, Cyclic voltammograms of 1a (1mM) and Ar background in 0.10 M NBu₄PF₆ in acetonitrile. Data was collected with a scan rate of 100 mV/s.

6. NMR Studies

6.1. Interaction of AgBF₄ with Selectfluor[®]

Procedure: To a 4 ml vial containing Selectfluor[®] (35.4 mg, 0.100 mmol) and AgBF₄ (19.4 mg, 0.100 mmol) was added 1:9 (v/v) Acetone- $d6/D_2O$ (1.0 ml). The resulting solution was allowed to stir at 40 °C for 1 h. The contents of the reaction vial were then transferred into a NMR tube and an NMR spectrum was taken directly afterwards to measure consumption of Selectfluor[®].



Fig. S2. ¹⁹F NMR monitoring of Selectfluor[®] consumption in the presence of AgBF₄

6.2. Interaction of AgBF₄ with 1a.

Procedure: To a 4 ml vial containing **1a** (18.9 mg, 0.100 mmol) and $AgBF_4$ (19.4 mg, 0.100 mmol) was added 1:9 (v/v) Acetone- d_6/D_2O (1.0 ml). The contents of the reaction vial were then transferred into a NMR tube and spectroscopic data was collected right after. The same procedure was followed with varying amounts of $AgBF_4$. The residual signal of acetone was used as internal reference.



Fig. S3. $^1\!\mathrm{H}$ NMR monitoring of interaction of $AgBF_4$ and 1a

7. NMR Spectra






































S42















S48







S51







S54

























S66


















References and Notes

- 1. A. H. Hoveyda, A. R. Zhugralin, The remarkable metal-catalysed olefin metathesis reaction. *Nature* **450**, 243–251 (2007). <u>doi:10.1038/nature06351</u> <u>Medline</u>
- G. C. Vougioukalakis, R. H. Grubbs, Ruthenium-based heterocyclic carbene-coordinated olefin metathesis catalysts. *Chem. Rev.* 110, 1746–1787 (2010). <u>doi:10.1021/cr9002424</u> <u>Medline</u>
- 3. P. S. Bailey, The Reactions Of Ozone With Organic Compounds. *Chem. Rev.* 58, 925–1010 (1958). doi:10.1021/cr50023a005
- S. K. Silverman, P. J. Hergenrother, Combinatorial chemistry and molecular diversityTools for molecular diversification and their applications in chemical biology. *Curr. Opin. Chem. Biol.* 10, 185–187 (2006). doi:10.1016/j.cbpa.2006.04.024
- 5. E. Vitaku, D. T. Smith, J. T. Njardarson, Analysis of the structural diversity, substitution patterns, and frequency of nitrogen heterocycles among U.S. FDA approved pharmaceuticals. J. Med. Chem. 57, 10257–10274 (2014). <u>doi:10.1021/jm501100b</u> <u>Medline</u>
- 6. S. A. Lawrence, *Amines: Synthesis, Properties and Applications* (Cambridge Univ. Press, 2004).
- 7. A. P. Shawcross, S. P. Stanforth, Reaction of *N*-nitroaryl-1,2,3,4-tetrahydroisoquinoline derivatives with oxygen. *J. Heterocycl. Chem.* 27, 367–369 (1990). <u>doi:10.1002/jhet.5570270249</u>
- 8. G. Han, M. C. McIntosh, S. M. Weinreb, A convenient synthetic method for amide oxidation. *Tetrahedron Lett.* **35**, 5813–5816 (1994). <u>doi:10.1016/S0040-4039(00)78191-3</u>
- G. Cocquet, C. Ferroud, A. Guy, A Mild and Efficient Procedure for Ring-Opening Reactions of Piperidine and Pyrrolidine Derivatives by Single Electron Transfer Photooxidation. *Tetrahedron* 56, 2975–2984 (2000). doi:10.1016/S0040-4020(00)00048-X
- R. Ito, N. Umezawa, T. Higuchi, Unique oxidation reaction of amides with pyridine-N-oxide catalyzed by ruthenium porphyrin: Direct oxidative conversion of N-acyl-L-proline to N-acyl-L-glutamate. J. Am. Chem. Soc. 127, 834–835 (2005). <u>doi:10.1021/ja045603f</u> Medline
- M. Kaname, S. Yoshifuji, H. Sashida, Ruthenium tetroxide oxidation of cyclic N-acylamines by a single layer method: Formation of ω-amino acids. *Tetrahedron Lett.* 49, 2786–2788 (2008). doi:10.1016/j.tetlet.2008.02.127
- T. J. Osberger, D. C. Rogness, J. T. Kohrt, A. F. Stepan, M. C. White, Oxidative diversification of amino acids and peptides by small-molecule iron catalysis. *Nature* 537, 214–219 (2016). doi:10.1038/nature18941 Medline
- C. Yu, M. A. Shoaib, N. Iqbal, J. S. Kim, H. J. Ha, E. J. Cho, Selective Ring-Opening of *N*-Alkyl Pyrrolidines with Chloroformates to 4-Chlorobutyl Carbamates. *J. Org. Chem.* 82, 6615–6620 (2017). doi:10.1021/acs.joc.7b00681 Medline

- 14. E. P. Gillis, K. J. Eastman, M. D. Hill, D. J. Donnelly, N. A. Meanwell, Applications of Fluorine in Medicinal Chemistry. J. Med. Chem. 58, 8315–8359 (2015). doi:10.1021/acs.jmedchem.5b00258 Medline
- 15. J. Wang, M. Sánchez-Roselló, J. L. Aceña, C. del Pozo, A. E. Sorochinsky, S. Fustero, V. A. Soloshonok, H. Liu, Fluorine in pharmaceutical industry: Fluorine-containing drugs introduced to the market in the last decade (2001-2011). *Chem. Rev.* **114**, 2432–2506 (2014). <u>doi:10.1021/cr4002879</u> <u>Medline</u>
- 16. S. Purser, P. R. Moore, S. Swallow, V. Gouverneur, Fluorine in medicinal chemistry. *Chem. Soc. Rev.* 37, 320–330 (2008). <u>doi:10.1039/B610213C Medline</u>
- 17. K. Müller, C. Faeh, F. Diederich, Fluorine in pharmaceuticals: Looking beyond intuition. *Science* **317**, 1881–1886 (2007). <u>doi:10.1126/science.1131943</u> <u>Medline</u>
- P. A. Champagne, J. Desroches, J.-D. Hamel, M. Vandamme, J.-F. Paquin, Monofluorination of Organic Compounds: 10 Years of Innovation. *Chem. Rev.* 115, 9073–9174 (2015). <u>doi:10.1021/cr500706a Medline</u>
- 19. B. Lantaño, A. Postigo, Radical fluorination reactions by thermal and photoinduced methods. *Org. Biomol. Chem.* **15**, 9954–9973 (2017). <u>doi:10.1039/C7OB02402A Medline</u>
- 20. M. Murakami, N. Ishida, β-Scission of Alkoxy Radicals in Synthetic Transformations. *Chem. Lett.* **46**, 1692–1700 (2017). <u>doi:10.1246/cl.170834</u>
- 21. M. Yan, J. C. Lo, J. T. Edwards, P. S. Baran, Radicals: Reactive Intermediates with Translational Potential. J. Am. Chem. Soc. 138, 12692–12714 (2016). doi:10.1021/jacs.6b08856 Medline
- 22. J. Sperry, The Oxidation of Amides to Imides: A Powerful Synthetic Transformation. Synthesis 2011, 3569–3580 (2011). doi:10.1055/s-0030-1260237
- 23. H. N. Po, Heterocyclic and macrocyclic amine complexes of silver(II) and silver(III). *Coord. Chem. Rev.* **20**, 171–195 (1976). <u>doi:10.1016/S0010-8545(00)80324-0</u>
- 24. H. Zhao, X. Fan, J. Yu, C. Zhu, Silver-catalyzed ring-opening strategy for the synthesis of βand γ-fluorinated ketones. J. Am. Chem. Soc. 137, 3490–3493 (2015). doi:10.1021/jacs.5b00939 Medline
- 25. S. Ren, C. Feng, T.-P. Loh, Iron- or silver-catalyzed oxidative fluorination of cyclopropanols for the synthesis of β-fluoroketones. *Org. Biomol. Chem.* **13**, 5105–5109 (2015). <u>doi:10.1039/C5OB00632E Medline</u>
- Q. Tian, B. Chen, G. Zhang, Silver-initiated radical ring expansion/fluorination of ethynyl cyclobutanols: Efficient synthesis of monofluoroethenyl cyclopentanones. *Green Chem.* 18, 6236–6240 (2016). doi:10.1039/C6GC02656G
- 27. Y. Deng, N. I. Kauser, S. M. Islam, J. T. Mohr, Ag ^{II} -Mediated Synthesis of β-Fluoroketones by Oxidative Cyclopropanol Opening. *Eur. J. Org. Chem.* 2017, 5872–5879 (2017). doi:10.1002/ejoc.201700899
- L. Wang, Z. Zha, W. Qu, H. Qiao, B. P. Lieberman, K. Plössl, H. F. Kung, Synthesis and evaluation of 18F labeled alanine derivatives as potential tumor imaging agents. *Nucl. Med. Biol.* **39**, 933–943 (2012). <u>doi:10.1016/j.nucmedbio.2012.03.007</u> <u>Medline</u>

- 29. F. Yin, Z. Wang, Z. Li, C. Li, Silver-catalyzed decarboxylative fluorination of aliphatic carboxylic acids in aqueous solution. J. Am. Chem. Soc. 134, 10401–10404 (2012). doi:10.1021/ja3048255 Medline
- 30. Z. Antosova, M. Mackova, V. Kral, T. Macek, Therapeutic application of peptides and proteins: Parenteral forever? *Trends Biotechnol.* 27, 628–635 (2009). doi:10.1016/j.tibtech.2009.07.009 Medline
- P. Vlieghe, V. Lisowski, J. Martinez, M. Khrestchatisky, Synthetic therapeutic peptides: Science and market. *Drug Discov. Today* 15, 40–56 (2010). doi:10.1016/j.drudis.2009.10.009 Medline
- 32. A. A. Kaspar, J. M. Reichert, Future directions for peptide therapeutics development. *Drug Discov. Today* **18**, 807–817 (2013). <u>doi:10.1016/j.drudis.2013.05.011</u> <u>Medline</u>
- 33. S. Sengupta, G. Mehta, Late stage modification of peptides via C H activation reactions. *Tetrahedron Lett.* 58, 1357–1372 (2017). doi:10.1016/j.tetlet.2017.02.069
- 34. Selectfluor is reported to react with AgNO₃ in acetone- $dd/D_2O(35)$.
- 35. N. R. Patel, R. A. Flowers 2nd, Mechanistic study of silver-catalyzed decarboxylative fluorination. J. Org. Chem. 80, 5834–5841 (2015). doi:10.1021/acs.joc.5b00826 Medline
- 36. V. Romanov, C.-K. Siu, U. H. Verkerk, H. El Aribi, A. C. Hopkinson, K. W. M. Siu, Binding energies of the silver ion to alcohols and amides: A theoretical and experimental study. J. Phys. Chem. A 112, 10912–10920 (2008). doi:10.1021/jp8055653 Medline
- 37. V. Romanov, C.-K. Siu, U. H. Verkerk, A. C. Hopkinson, K. W. M. Siu, Bond dissociation energies of solvated silver(I)-amide complexes: Competitive threshold collision-induced dissociations and calculations. J. Phys. Chem. A 114, 6964–6971 (2010). doi:10.1021/jp102470x Medline
- 38. C. R. Pitts, S. Bloom, R. Woltornist, D. J. Auvenshine, L. R. Ryzhkov, M. A. Siegler, T. Lectka, Direct, catalytic monofluorination of sp³ C-H bonds: A radical-based mechanism with ionic selectivity. J. Am. Chem. Soc. 136, 9780–9791 (2014). doi:10.1021/ja505136j Medline
- S. Singh, C.-M. Martinez, S. Calvet-Vitale, A. K. Prasad, T. Prangé, P. I. Dalko, H. Dhimane, Synthesis and Conformational Analysis of Fluorinated Pipecolic Acids. *Synlett* 23, 2421–2425 (2012). doi:10.1055/s-0032-1316770
- N. A. Meanwell, Synopsis of some recent tactical application of bioisosteres in drug design. J. Med. Chem. 54, 2529–2591 (2011). doi:10.1021/jm1013693 Medline
- 41. C. D. Sessler, M. Rahm, S. Becker, J. M. Goldberg, F. Wang, S. J. Lippard, CF₂H, a Hydrogen Bond Donor. J. Am. Chem. Soc. 139, 9325–9332 (2017). doi:10.1021/jacs.7b04457 Medline
- 42. C. J. Li, L. Chen, Organic chemistry in water. *Chem. Soc. Rev.* **35**, 68–82 (2006). doi:10.1039/B507207G Medline
- 43. M. O. Simon, C. J. Li, Green chemistry oriented organic synthesis in water. *Chem. Soc. Rev.* 41, 1415–1427 (2012). doi:10.1039/C1CS15222J Medline

- 44. Y. Otani, O. Nagae, Y. Naruse, S. Inagaki, M. Ohno, K. Yamaguchi, G. Yamamoto, M. Uchiyama, T. Ohwada, An evaluation of amide group planarity in 7-azabicyclo[2.2.1]heptane amides. Low amide bond rotation barrier in solution. *J. Am. Chem. Soc.* 125, 15191–15199 (2003). doi:10.1021/ja036644z Medline
- 45. Z. Mucsi, G. A. Chass, I. G. Csizmadia, Amidicity change as a significant driving force and thermodynamic selection rule of transamidation reactions. A synergy between experiment and theory. *J. Phys. Chem. B* **112**, 7885–7893 (2008). <u>doi:10.1021/jp8023292</u> <u>Medline</u>
- 46. W. Xie, M. Zhao, C. Cui, Cesium Carbonate-Catalyzed Reduction of Amides with Hydrosilanes. *Organometallics* **32**, 7440–7444 (2013). <u>doi:10.1021/om400951n</u>
- Y. Kuninobu, H. Ida, M. Nishi, M. Kanai, A meta-selective C-H borylation directed by a secondary interaction between ligand and substrate. *Nat. Chem.* 7, 712–717 (2015). doi:10.1038/nchem.2322 Medline
- J. Zhu, Y. Zhang, F. Shi, Y. Deng, Dehydrogenative amide synthesis from alcohol and amine catalyzed by hydrotalcite-supported gold nanoparticles. *Tetrahedron Lett.* 53, 3178–3180 (2012). doi:10.1016/j.tetlet.2012.04.048
- D. A. Gruzdev, S. A. Vakarov, G. L. Levit, V. P. Krasnov, N-Tosyl-(S)-Prolyl Chloride in Kinetic Resolution of Racemic Heterocyclic Amines. *Chem. Heterocycl. Compd.* 49, 1795–1807 (2014). doi:10.1007/s10593-014-1432-4
- 50. K. Ishihara, S. Ohara, H. Yamamoto, 3,4,5-Trifluorobenzeneboronic Acid as an Extremely Active Amidation Catalyst. J. Org. Chem. 61, 4196–4197 (1996). doi:10.1021/jo9606564 Medline
- A. Takács, Z. Kabak-Solt, G. Mikle, L. Kollár, Alkoxycarbonylpiperidines as N-nucleophiles in the palladium-catalyzed aminocarbonylation. *Monatsh. Chem.* 145, 1473–1478 (2014). doi:10.1007/s00706-014-1254-1
- 52. S. C. Ghosh, J. S. Y. Ngiam, A. M. Seayad, D. M. Tuan, C. W. Johannes, A. Chen, Tandem oxidative amidation of benzyl alcohols with amine hydrochloride salts catalysed by iron nitrate. *Tetrahedron Lett.* 54, 4922–4925 (2013). doi:10.1016/j.tetlet.2013.07.005
- U. Košak, B. Brus, S. Gobec, Straightforward synthesis of orthogonally protected piperidin-3-ylmethanamine and piperidin-4-ylmethanamine derivatives. *Tetrahedron Lett.* 55, 2037–2039 (2014). doi:10.1016/j.tetlet.2014.02.034
- 54. Z. Tu, S. M. N. Efange, J. Xu, S. Li, L. A. Jones, S. M. Parsons, R. H. Mach, Synthesis and in vitro and in vivo evaluation of ¹⁸F-labeled positron emission tomography (PET) ligands for imaging the vesicular acetylcholine transporter. *J. Med. Chem.* **52**, 1358–1369 (2009). <u>doi:10.1021/jm8012344 Medline</u>
- 55. P. Zajdel, G. Nomezine, N. Masurier, M. Amblard, M. Pawłowski, J. Martinez, G. Subra, A new highly versatile handle for chemistry on a solid support: The pipecolic linker. *Chemistry* 16, 7547–7553 (2010). doi:10.1002/chem.201000313 Medline
- 56. Y. Sasano, S. Nagasawa, M. Yamazaki, M. Shibuya, J. Park, Y. Iwabuchi, Highly chemoselective aerobic oxidation of amino alcohols into amino carbonyl compounds. *Angew. Chem. Int. Ed.* **53**, 3236–3240 (2014). <u>doi:10.1002/anie.201309634</u> <u>Medline</u>

- 57. N. Gigant, L. Chausset-Boissarie, M.-C. Belhomme, T. Poisson, X. Pannecoucke, I. Gillaizeau, Copper-catalyzed direct arylation of cyclic enamides using diaryliodonium salts. *Org. Lett.* 15, 278–281 (2013). <u>doi:10.1021/ol303400s Medline</u>
- 58. L. Cui, Y. Peng, L. Zhang, A two-step, formal [4 + 2] approach toward piperidin-4-ones via Au catalysis. J. Am. Chem. Soc. 131, 8394–8395 (2009). doi:10.1021/ja903531g Medline
- 59. N. Gigant, I. Gillaizeau, Construction of nitrogen-fused tetrahydroquinolines via a domino reaction. *Org. Lett.* 14, 4622–4625 (2012). doi:10.1021/ol3020732 Medline
- 60. M. S'kof, J. Svete, M. Kmetič, S. Golič-Grdadolnik, B. Stanovnik, Stereoselective Amination of 5-Substituted γ-Lactones and γ-Lactams A Convenient Route for the Preparation of 5-Substituted (3S,5S)-3-Acetylaminotetrahydrofuran-2-ones and (3S,5S)-3-Acetylaminopyrrolidin-2-ones. *Eur. J. Org. Chem.* **1999**, 1581–1584 (1999). doi:10.1002/(SICI)1099-0690(199907)1999:7<1581:AID-EJOC1581>3.0.CO;2-2
- 61. A. Boto, R. Hernández, E. Suárez, Tandem radical decarboxylation-oxidation of amino acids: A mild and efficient method for the generation of *N*-acyliminium ions and their nucleophilic trapping. *J. Org. Chem.* 65, 4930–4937 (2000). doi:10.1021/jo000356t <u>Medline</u>
- 62. M. Rubinshtein, C. R. James, J. L. Young, Y. J. Ma, Y. Kobayashi, N. C. Gianneschi, J. Yang, Facile procedure for generating side chain functionalized poly(α-hydroxy acid) copolymers from aldehydes via a versatile Passerini-type condensation. *Org. Lett.* **12**, 3560–3563 (2010). doi:10.1021/ol101433v Medline
- 63. Z. Li, Z. Wang, L. Zhu, X. Tan, C. Li, Silver-catalyzed radical fluorination of alkylboronates in aqueous solution. *J. Am. Chem. Soc.* **136**, 16439–16443 (2014). <u>doi:10.1021/ja509548z</u> <u>Medline</u>